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Fungal rock phosphate solubilization using sugarcane bagasse

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Abstract The effects of different doses of rock phosphate (RP), sucrose, and $(NH_4)_2SO_4$ on the solubilization of RP from Araxá and Catalão (Brazil) by *Aspergillus niger*, *Penicillium canescens*, *Eupenicillium ludwigii*, and *Penicillium islandicum* were evaluated in a solid-state fermentation (SSF) system with sugarcane bagasse. The factors evaluated were combined following a $2^3 + 1$ factorial design to determine their optimum concentrations. The fitted response surfaces showed that higher doses of RP promoted higher phosphorus (P) solubilization. The addition of sucrose did not have effects on P solubilization in

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Bolsista CNPq-Brasil, Laboratório de Associações Micorrízicas, Instituto de Biotecnologia Aplicada à Agricultura (BIOAGRO), Av. P. H. Rolfs, s/n, Campus, Viçosa, MG 36570-000, Brazil e-mail: mdcosta@ufv.br most treatments due to the presence of soluble sugars in the bagasse. Except for A. niger, all the fungi required high (NH₄)₂SO₄ doses to achieve the highest level of P solubilization. Inversely, addition of (NH₄)₂SO₄ was inhibitory to P solubilization by A. niger. Among the fungi tested, A. niger stood out, showing the highest solubilization capacity and for not requiring sucrose or (NH₄)₂SO₄ supplementation. An additional experiment with A. niger showed that the content of soluble P can be increased by adding higher RP doses in the medium. However, P yield decreases with increasing RP doses. In this experiment, the maximal P yield (approximately 60 %) was achieved with the lower RP dose (3 g L^{-1}). Our results show that SSF can be used to obtain a low cost biofertilizer rich in P combining RP, sugarcane bagasse, and A. niger. Moreover, sugarcane bagasse is a suitable substrate for SSF aiming at RP solubilization, since this residue can supply the C and N necessary for the metabolism of A. niger within a range that favors RP solubilization.

Introduction

Although phosphorus (P) is abundant in many soils, it is one of the main limiting nutrients for plant growth because of its retention in soil particles, making it unavailable to plants. Frequent applications of soluble inorganic forms of P are therefore necessary to achieve satisfactory plant growth.

Soluble phosphate fertilizers are obtained from the chemical treatment of rock phosphate (RP) with acid at high temperatures. One of the drawbacks of this technology

is the high cost involved in the process (Nahas et al. 1990). Moreover, it brings about unwanted environmental consequences due to the release of contaminants previously retained in the RP and the production of gases and byproducts resulting from the acid treatment (Vassilev and Vassileva 2003). The direct application of RPs may be an alternative to reduce the costs of phosphate fertilization and minimize pollution. However, due to the low reactivity of the majority of these rocks, their direct application is not, in many cases, adequate to satisfy the needs of plants on either a short- or long-term basis (Novais and Smith 1999).

The use of microorganisms capable of solubilizing RPs is a promising alternative for obtaining soluble P at a low cost and with reduced environmental damage. Several studies have been conducted to develop techniques for using microorganisms for large-scale phosphate solubilization (Nahas et al. 1990; Vassilev et al. 1996; Vassilev et al. 2001; Vassilev et al. 2006; Vassileva et al. 1998). Among these techniques, solid-state fermentation (SSF) has been shown to be efficient at providing appropriate conditions for the action of phosphate-solubilizing microorganisms. The use of SSF processes to produce soluble P from RP has been evaluated using agro-industrial residues such as olive cake (Vassilev et al. 2006; Vassileva et al. 1998) and sugar-beet (Vassilev et al. 1996; Vassilev et al. 2006). Seventy-six percent of the RP was solubilized by Aspergillus niger in an SSF system with sugar-beet residues as the substrate. In the same study, the fermentation product, rich in soluble P, was used as a fertilizer in Trifolium repens plants and resulted in increased plant growth and P content (Vassilev et al. 1996).

Sugarcane bagasse is an abundant agro-industrial waste in Brazil and corresponds to, approximately, 30 % of the sugarcane dry weight. It has desirable characteristics for SSF, such as high carbon and reduced ash contents compared to other crop residues (Soccol and Vandenberghe 2003). Sugarcane bagasse does not agglomerate after moistening with culture medium, allowing better heat and mass transfer during SSF (Kumar et al. 2003). Kumar et al. (2003) obtained 20.2 g citric acid in an SSF system with 100 g of sugarcane bagasse inoculated with *A. niger*. Sugarcane bagasse is, therefore, a promising alternative for SSF-based RP solubilization, given that the production of organic acid is the main mechanism of phosphate solubilization by microorganisms (Banik and Dey 1982; Reyes et al. 2001).

In some cases, the use of different agro-industrial residues in SSF systems requires the supplementation with carbon (C) and nitrogen (N) sources in addition to micronutrients. Depending on the C and N sources used, microorganisms will solubilize RPs at different efficiencies. Generally, $N-NH_4^+$ in the medium results in higher levels of phosphate solubilization than $N-NO_3^-$ (Ahuja et al. 2007; Chuang et al. 2007). The effects of the C source on phosphate solubilization are variable and frequently depend on the microorganism being tested (Reyes et al. 1999; Ahuja et al. 2007). The optimal C and N concentrations in SSF systems for RP solubilization are not known. Different waste materials with distinct chemical compositions require appropriate C and N concentrations for achieving high levels of RP solubilization.

Another factor that influences microbial phosphate solubilization is the P source concentration. For Yichang RP (8.35 % P), Xiao et al. (2008) showed that the RP concentration that provided the highest level of solubilization in liquid culture medium was 2.5 g L⁻¹. However, due to the high variability in RP composition and reactivity, specific studies are necessary to define the appropriate concentration for each RP.

Thus, the aims of this study were to evaluate the solubilization of Araxá and Catalão RPs by fungal isolates in an SSF system using sugarcane bagasse as the substrate, and to study the effects of different doses of RP, sucrose, and $(NH_4)_2SO_4$ on the solubilization process.

Materials and methods

Microorganisms

The isolates *A. niger* FS1, *Penicillium canescens* FS23, *Eupenicillium ludwigii* FS27, and *Penicillium islandicum* FS30 used in this study were maintained at 28 °C on Petri dishes containing potato-dextrose-agar (PDA). The isolates belong to the Collection of Phosphate Solubilizing Fungi, Institute of Biotechnology Applied to Agriculture (BIO-AGRO), Federal University of Viçosa, Viçosa, MG, Brazil.

Culture media and fermentation conditions

N, P, and soluble sugar contents of the sugarcane bagasse used as substrate for the SSF system, in $g kg^{-1}$, were 3.4, 0.6, and 300, respectively. The bagasse was ground into fragments of up to 2 mm in length and added at a ratio of 10 % (w/v) to 125-mL Erlenmeyer flasks containing 50 mL of modified Czapek's solution (0.5 g L^{-1} MgSO₄. 7H₂O, 0.5 g L⁻¹ KCl, 0.01 g L⁻¹ FeSO₄). Araxá (32 % P₂O₅) and Catalão (34 % P₂O₅) RPs were added separately to the fermentation medium at concentrations of 0.5, 1.75, and 3 g L^{-1} . (NH₄)₂SO₄ at 0, 1, and 2 g L^{-1} and sucrose at 0, 5, and 10 g L^{-1} were also tested. These different doses of RPs, sucrose, and (NH₄)₂SO₄ were combined following the experimental design displayed in Table 1. After sterilization at 121 °C for 30 min, the flasks were inoculated with five 7-mm-diameter disks containing fungal mycelium taken from the edges of five-day-old colonies and

Table 1 Solubilized P from Araxá or Catalão rock phosphates (RP) by fungal isolates under different combinations of RP, $(NH_4)_2SO_4$, and sucrose doses

Treatment	RP	$(NH_4)_2SO_4$	Sucrose	Solubilized P^b (mg L^{-1})								
	$(g L^{-1})$	$(g L^{-1})$	$(g L^{-1})$	Araxá	RP	Р			Catalão RP			
				A. niger	· P. canescens	E. ludwigii	P. islandicum	A. niger	P. canescens	E. ludwigii	P. islandicum	
1	0.5	0	0	49.0	7.5	-19.1	-19.8	43.9	2.8	-10.6	-9.9	
2	3	0	0	239.3	13.0	-8.8	-8.7	259.6	14.9	-6.4	-4.3	
3	0.5	2	0	50.3	-12.0	-7.2	-14.5	56.1	-0.9	6.5	-4.2	
4	3	2	0	114.0	26.0	11.6	-3.7	130.6	24.3	19.0	6.2	
5	0.5	0	10	34.9	1.4	-19.6	-15.6	50.8	-4.5	-9.9	-8.5	
6	3	0	10	148.8	9.1	-0.3	-17.0	197.8	22.5	-0.7	0.3	
7	0.5	2	10	30.7	-10.3	-8.7	-17.2	47.8	0.0	3.7	-6.1	
8	3	2	10	108.4	27.5	14.2	12.4	139.8	5.3	21.4	11.1	
9 ^a	1.75	1	5	75.6	8.5	-1.5	-5.6	78.1	8.2	13.2	-1.2	

The first four columns show the combinations of the factors in each treatment in the experimental design

^a Central point

^b Solubilized P was calculated by the difference between soluble P in the inoculated samples and in the uninoculated flasks (Soluble P in uninoculated flasks (mg L^{-1}): Araxá RP, 24.6; Catalão RP, 14.8. Part of this P (13.3 mg L^{-1}) is derived from the sugarcane bagasse.)

incubated at 30 °C for 10 days. Uninoculated flasks were also incubated to determine the soluble P in the bagasse and the P solubilized by abiotic processes.

Analytical methods

After the incubation, the aqueous phase of the fermentation medium was extracted by manual compression of the bagasse into a funnel covered with quantitative filter paper. The resulting filtrate was used to determine P concentrations by the ascorbic acid method in a spectrophotometer at wavelength of 725 nm (Braga and Defelipo 1974). Solubilized P was calculated by the difference between soluble P in the inoculated samples and in the uninoculated flasks. P yield was expressed as the percentage of solubilized P obtained in relation to the total P content in the RPs. The pH of the filtrate was measured and the titratable acidity determined by titrating 5 mL of the filtrate up to pH 7.0 with 0.1 M NaOH using bromothymol blue as indicator.

Experimental design and statistical analyses

To determine the optimum combination of the three factors evaluated (RP, $(NH_4)_2SO_4$, and sucrose doses), a $2^3 + 1$ factorial design with three replications at the central point was used. The experimental design with the combinations of the factors evaluated is shown in Table 1. Altogether, eight experiments were conducted, corresponding to the combinations of the four fungal isolates with the two RPs evaluated (Table 1).

The response surfaces were fitted through the least squares method using the statistical software Minitab 15. Variance analysis with a lack of fit test for the response surface was performed, and the coefficients were tested using a t test up to 10 % probability. The largest response surface model adopted was:

$$y_{ijkl} = \beta_0 + \beta_1 RP_i + \beta_2 AS_j + \beta_3 SC_k + \beta_4 RP_i AS_j + \beta_5 RP_i SC_k + \beta_6 AS_i SC_k + e_{iikl},$$

where:

- y_{ijkl} Value of solubilized P observed with the combination of RP_i, AS_i and SC_k used in replication 1
- β_0 Regression constant
- β_1 Regression coefficient of the linear effect of RP (rock phosphate)
- β_2 Regression coefficient of the linear effect of AS (ammonium sulfate)
- β_3 Regression coefficient of the linear effect of SC (sucrose)
- β_4 Regression coefficient of the effect of the interaction between RP and AS
- β_5 Regression coefficient of the effect of the interaction between RP and SC
- β_6 Regression coefficient of the effect of the interaction between AS and SC
- e_{ijkl} Effect of the experimental error associated with the observation y_{ijkl}

To determine the fitted response surface with the presence of only significant coefficients, those coefficients that were not significant by the *t* test (p < 0.10) were removed one at a time, starting with the interactions. The coefficient of determination (\mathbb{R}^2) was also considered for verifying the adjustment of the model. To visualize the data, surface response plots were built.

For the isolate *A. niger* FS1, an additional experiment was conducted using the same SSF conditions defined previously while varying the doses of the two RPs (3.0, 4.5, 6.0, 7.5, and 9.0 g L⁻¹). In this experiment, neither (NH₄)₂SO₄ nor sucrose was added. This experiment was conducted using an entirely randomized design with three replications at the central point (6.0 g L⁻¹) followed by regression analysis. The Pearson correlation coefficients between the variables studied were also analyzed. The experimental design and the statistical analyses were done using the statistical software Minitab 15.

Results

Effects of different sucrose, (NH₄)₂SO₄ and RP doses

The fungal isolates showed distinct solubilization potentials and, depending on the medium supplementation treatments, distinct behaviors were observed for a single isolate when grown in the presence of different RPs (Table 1, 2). Negative values of solubilized P were observed for the treatments (Table 1) in which the fungus immobilized P derived from sugarcane bagasse and abiotic RP solubilization, being not able to solubilize significant P amounts from the RPs. For all the combinations of isolates and RPs, the highest dose of RP added to the SSF system (3.0 g L⁻¹) promoted higher levels of solubilized P (Table 1). (NH₄)₂SO₄ supplementation influenced RP solubilization for most of the treatments evaluated in this study (Table 2), however, the response to this compound varied depending on the fungal isolate and RP combination tested. Sucrose improved the level of solubilized P only when *P. islandicum* FS30 was combined with Catalão RP (Table 2).

Aspergillus niger FS1 promoted the highest levels of solubilized P for both RPs (Table 1). A positive effect of increasing RP doses on the levels of solubilized P was observed for both RPs in the treatments with A. niger FS1 (Table 2; Figs. 1a, 2a). The regression equations for A. niger FS1 shows that, although $(NH_4)_2SO_4$ addition was not significant by itself, the nitrogen source doses in the medium correlated negatively with RP doses. Since the RP coefficient is positive, the negative coefficient corresponding to the interaction between RP and (NH₄)₂SO₄ in the equations indicates that higher P solubilization is achieved combining high RP doses with low (NH₄)₂SO₄ (Table 2). This effect is clearly shown in the surface response graphs (Figs. 1a, 2a). Moreover, a negative correlation between (NH₄)₂SO₄ concentration and titratable acidity was observed for both Araxá and Catalão RPs (-0.76, p < 0.01, and -0.70, p < 0.05, respectively).

 $(NH_4)_2SO_4$ addition promoted different patterns of solubilized P in the treatments with *P. canescens* FS23 and Araxá RP (Fig. 1b). At lower RP doses, as $(NH_4)_2SO_4$ concentrations increased, phosphate solubilization was reduced. Inversely, at higher RP doses, a positive effect of increasing $(NH_4)_2SO_4$ concentrations was recorded (Fig. 1b), achieving a maximum yield of solubilized P of 7 % for Araxá RP (Table 1). In the medium supplemented with Catalão RP, only the RP doses influenced P solubilization by *P. canescens* FS23 (Table 2). Higher yields of solubilized P were obtained with increasing RP doses in the

Table 2 Regression equation for the levels of solubilized P by the fungal isolates as a function of Araxá or Catalão rock phosphate (RP), $(NH_4)_2SO_4$ (AS) and sucrose (SC) doses

Isolate	Regression equation	\mathbb{R}^2	Figure
Araxá RP			
A. niger FS1	$\hat{Y} = 5.7 + 60.8^{**} RP + 7.4 AS - 16.3^{\dagger} RP AS$	0.85	<u>1</u> a
P. canescens FS23	$\hat{Y} = 3.3 + 2.6 \text{ RP} - 10.9^{**} \text{ AS} + 6.3^{**} \text{ RP} \text{ AS}$	0.94	1b
E. ludwigii FS27	$\hat{Y} = -23.5 + 7.1^{**} RP + 7.2^{**} AS$	0.91	1c
P. islandicum FS30	$\hat{Y} = -17.3 + 1.9 \text{ RP} - 0.6 \text{ AS} + 3.1^{\dagger} \text{ RP} \text{ AS}$	0.68	1d
Catalão RP			
A. niger FS1	$\hat{Y} = 0.8 + 72.5^{**} RP + 12.1 AS - 19.6^{*} RP AS$	0.89	<mark>2</mark> a
P. canescens FS23	$\hat{Y} = -4.1 + 6.9^{**} RP$	0.70	-
E. ludwigii FS27	$\hat{Y} = -11.7 + 4.4* \text{ RP} + 9.7** \text{ AS}$	0.77	2 b
P. islandicum FS30	$\hat{Y} = -9.8 + 1.9^{*} \text{ RP} + 1.4 \text{ AS} - 0.1 \text{ SC} + 1.3^{*} \text{ RP} \text{ AS} + 0.2^{\dagger} \text{ RP} \text{ SC}$	0.97	3

** Significant by the *t* test (p < 0.01)

* Significant by the *t* test (p < 0.05)

[†] Significant by the *t* test (p < 0.10)

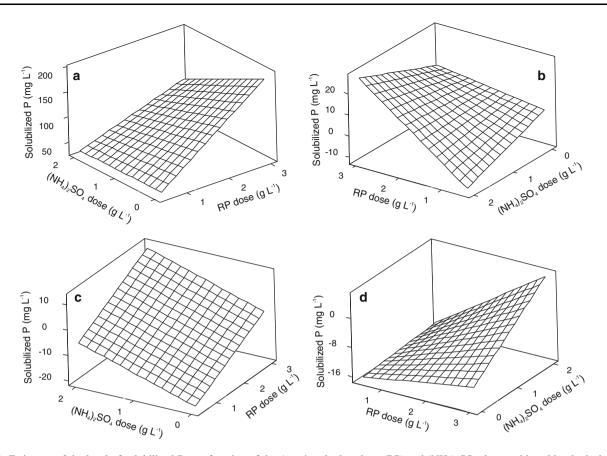


Fig. 1 Estimates of the level of solubilized P as a function of the Araxá rock phosphate (RP) and $(NH_4)_2SO_4$ doses achieved by the isolates **a** *A. niger* FS1, **b** *P. canescens* FS23, **c** *E. ludwigii* FS27 and **d** *P. islandicum* FS30

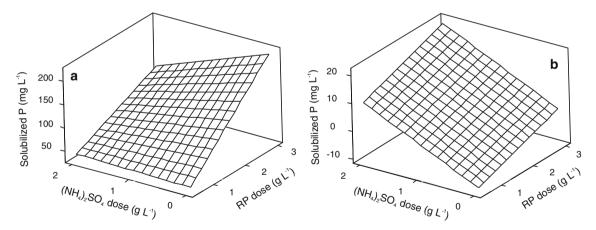


Fig. 2 Estimates of the level of solubilized P as a function of the Catalão rock phosphate (RP) and $(NH_4)_2SO_4$ doses achieved by the isolates **a** *A. niger* FS1 and **b** *E. ludwigii* FS27

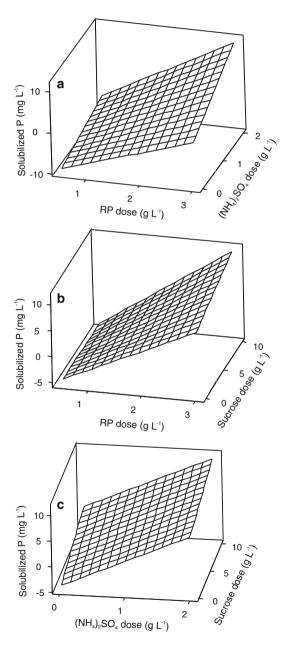
growth medium (Table 1), however, the highest P yield corresponded to only 5 % of the P present in Catalão RP.

Eupenicillium ludwigii FS27 showed similar patterns of solubilization for the two RPs evaluated. Increasing RP and $(NH_4)_2SO_4$ doses led to higher P solubilization, with no interactions between RPs and N source (Table 2; Figs. 1c,

2b). The highest P yield obtained corresponded to 3 and 5 % of the P present in Araxá and Catalão RPs, respectively.

In the medium with Araxá RP, inoculated with *P. islandicum* FS30, a positive interaction between RP doses and increasing $(NH_4)_2SO_4$ concentrations was

observed (Table 2; Fig. 1d). The highest P yield achieved by *P. islandicum* FS30 corresponded to, approximately, 3 % of the total P in the Araxá RP. When *P. islandicum* FS30 was grown in media with Catalão RP, effects of the RP doses and of the interaction between RP and $(NH_4)_2SO_4$ and of RP and sucrose could be observed (Table 2). All the interactions were positive, showing that the combination of higher Araxá RP, $(NH_4)_2SO_4$, and sucrose doses allowed for increased levels of solubilized P (Fig. 3).



Responses to increasing doses of RP

To determine to what extent P solubilization continues to respond positively to increases in RP doses, the isolate *A. niger* FS1 was selected for another experiment. For both Araxá and Catalão RPs, increasing RP doses led to higher levels of solubilized P. However, the percentage of P in the RPs that was solubilized decreased as the RP doses increased (Fig. 4). In this medium, neither the titratable acidity nor the final pH varied with the RP doses added. For both RPs, the titratable acidity and pH were, on average, 30 mmol H⁺ L⁻¹ and 2.2, respectively.

Discussion

Effects of different sucrose, (NH₄)₂SO₄ and RP doses

Our results showed that RP dose was the most important factor affecting the level of solubilized P at the end of fungal growth under the tested treatments. Higher RP doses promoted higher solubilized P. Sucrose supplementation was only required for a single treatment. At the proportion

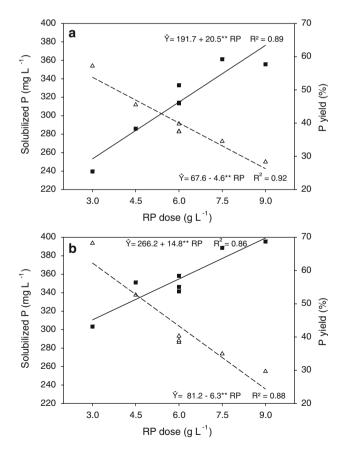


Fig. 3 Estimates of the level of solubilized P by the isolate *P. islandicum* FS30 as a function of the Catalão rock phosphate (RP), $(NH_4)_2SO_4$ and sucrose doses: for a fixed dose of sucrose at 10 g L⁻¹ (**a**), for a fixed dose of N at 2 g L⁻¹ (**b**) and for a fixed dose of RP at 3 g L⁻¹ (**c**)

Fig. 4 Solubilized P (*filled square*) and yield of solubilized P (*triangle*) by the isolate *A. niger* FS1 as a function of the Araxá (**a**) or Catalão (**b**) rock phosphates. **Significant by the *t* test (p < 0.01)

of bagasse used, about 30 g L⁻¹ of soluble sugars were added to the fermentation medium. These sugars derived from the bagasse must have been sufficient to supply the necessary C for fungal metabolism and RP solubilization, explaining the lack of significant effects of sucrose supplementation on RP solubilization. For instance, Vassilev et al. (1996) obtained up to 76 % of RP solubilization using Czapek's solution containing 30 g L⁻¹ of sucrose. This fact represents an advantage of using sugarcane bagasse as a substrate for SSF, considering that C addition to SSF systems can be costly.

 $(NH_4)_2SO_4$ effects were variable among the treatments and the fungal isolates showed distinct requirements for N in the growth medium. For A. niger FS1, (NH₄)₂SO₄ decreased P solubilization and titratable acidity. Similar results were observed for Penicillium purpurogenum, which produced less organic acids and solubilized less P when higher (NH₄)₂SO₄ doses were added to the liquid culture (Scervino et al. 2011). Nahas et al. (1990) observed that higher growth is followed by lower RP solubilization, pH reduction, and acid production when NH₄NO₃ is added to culture medium. This is attributed to the fact that nitrogen stimulates the synthesis of cellular materials by the fungus (Hang et al. 1977), and the fungal cells will, therefore, devote less carbon to the production of acids. The bagasse used in this work could supply about 0.34 g L^{-1} of N to the fermentation medium. This value is three times larger than that found to be optimal for citric acid production by A. niger (Ikram Ul et al. 2005). Thus, we speculate that the negative correlation observed between (NH₄)₂SO₄ dose and titratable acidity results from the inhibition of organic acids production when more N is added to the fermentation medium. As a consequence, RP solubilization decrease because organic acids are important factors involved in the solubilization process. Thus, our results indicate that the available N in the sugarcane bagasse may be sufficient to maintain fungal metabolism within a range that favors RP solubilization.

When *P. canescens* FS23 was inoculated in the media with Araxá RP at the lower doses, $(NH_4)_2SO_4$ supplementation led to decreased P solubilization. Higher fungal growth due to increased N availability is known to stimulate P consumption from the growth medium (Nahas et al. 1990). Thus, when lower RP doses were added to the SSF system, the lower quantities of P solubilized by the fungus must have been exhausted rapidly. On the other hand, at higher RP doses, $(NH_4)_2SO_4$ favored solubilization. A large Ca-sink is necessary to remove Ca from the medium and to allow the continuity of RP solubilization (Robinson and Syers 1990). In our treatments, when there was a higher amount of Araxá RP to be solubilized, increasing N availability could have stimulated fungal growth and the growing mycelial mass may have become a major sink for Ca generated from RP solubilization, leading to higher soluble P yields.

Among the fungal isolates tested, A. niger FS1 stood out for being the most efficient at solubilizing both Catalão and Araxá RPs. This fungus promoted solubilization percentages of 57 and 58 % of the total P present in Araxá and Catalão RPs, respectively. These values are higher than those obtained by Vassileva et al. (1998) who obtained a 43 %-solubilization of an RP from Morocco (12.8 % P) using A. niger in an SSF system with olive cake as substrate. Aspergillus niger is widely known because of its capacity to produce organic acids at high amounts, such as citrate, oxalate, and gluconate (Magnuson and Lasure 2004). This characteristic highlights A. niger as a potential P solubilizer, taking into account that organic acids are molecules with high capability to release P from insoluble forms (Fox et al. 1990). Moreover, A. niger FS1 demonstrated higher solubilization ability with very simple nutritional requirements. The response surfaces clearly show that, except for A. niger FS1, all the fungal isolates need a combination of high RP and (NH₄)₂SO₄ doses to obtain maximal solubilization (Figs. 1, 2, 3). On other hand, for A. niger FS1, the highest amounts of solubilized P, for both Araxá and Catalão RPs, could be achieved with no $(NH_4)_2SO_4$ or sucrose addition (Table 1). This is extremely interesting and demonstrates that A. niger FS1 is capable of solubilizing part of the P present in the RPs using only the C and N sources present in the sugarcane bagasse.

Responses to increasing doses of RP

A negative correlation between the amount of P solubilized and P yield was observed with increasing RP doses. This trend of reduction of P yield with increasing RP doses can be also calculated from the data of Xiao et al. (2008). The same phenomenon occurs when RP is added to soil and is often a consequence of insufficient acidity for RP solubilization, limited P-sink, and, especially, limited Ca-sink (Robinson et al. 1992; Robinson and Syers 1990). These explanations are also probably valid for the conditions observed in the SSF system. As the acidity remains unchanged among treatments, a fixed amount of RP can be solubilized by this mechanism. Because of this, when the doses of RP is increased, a smaller percentage of the P in the RP will be solubilized. Sinks for Ca and P under SSF correspond mostly to the fungal biomass. No apparent variation in fungal growth could be observed among the treatments tested in our study. Fungal growth was confined to the surface of the SSF medium, forming a thin layer that covered the entire substrate in all the treatments. Thus, when larger RP doses were added, Ca and P sinks, represented by mycelium, may have been quickly saturated, inhibiting the solubilization of a large portion of the added RP.

This inverse relationship between solubilized P and P yield with increasing RP doses opens new avenues for optimizing RP solubilization. Strategies such as removing Ca from the fermentation medium, increasing the area available for fungal growth, and increasing initial inoculum could allow the solubilization of larger quantities of P even at high concentrations of RPs.

Conclusions

Based on our data we can affirm that *A. niger* FS1 is the most efficient fungal isolate for SSF-based RP solubilization system proposed, achieving high P yields with lower nutritional requirements. Sugarcane bagasse can supply the C and N required for fungal RP solubilization. The system proposed could be efficiently used to obtain a low cost biofertilizer rich in P combining RP, sugarcane bagasse, and *A. niger*. Further work is necessary to optimize P yields at high RP doses, searching an optimal equilibrium between soluble P in the product and a good P recovery from the RP.

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