

## FURTHER FLAVONOL GLYCOSIDES OF *EMBELIA SCHIMPERI* LEAVES

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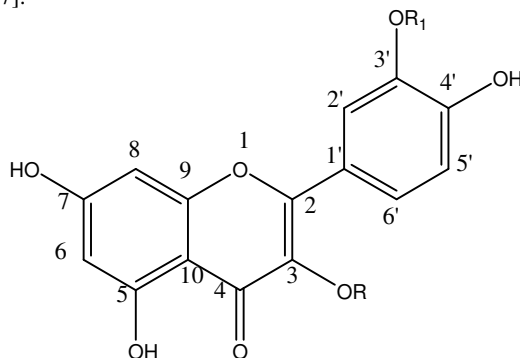
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**ABSTRACT.** Fractionation of the methanolic extract of *Embelia schimperi* leaves has led to the isolation of two novel flavonol glycosides. The compounds were characterized as isorhamnetin 3-*O*-β-galactosyl (1→4)-β-galactoside and quercetin 3-*O*-[α-rhamnosyl (1→2)] [α-rhamnosyl (1→4)]-α-rhamnoside. Also reported from the same extracts were known compounds quercetin, myricetin, quercetin 3-*O*-α-rhamnoside, quercetin 3-*O*-β-glucoside, quercetin 3-*O*-rutinoside, myricetin 3-*O*-β-xyloside, isorhamnetin 3-*O*-β-glucoside and myricetin 3-*O*-β-glucoside. Their structural elucidation was accomplished using spectral measurements and chemical methods.

**KEY WORDS:** *Embelia schimperi*, Flavonol glycosides, Isorhamnetin 3-*O*-β-galactosyl (1→4)-β-galactoside, Quercetin 3-*O*-[α-rhamnosyl (1→2)] [α-rhamnosyl (1→4)]-α-rhamnoside

## INTRODUCTION

*Embelia schimperi* Vatke is one of the five Myrsinaceae species endemic to Kenya and is widely used in traditional medicine as an anthelmintic and anti-microbial [1]. Phytochemically, the plant is typified by the presence of benzoquinone derivatives; 2,5-dihydroxy-3-undecyl-1,4-benzoquinone (trivial name: embelin) and 2,5-dihydroxy-3-tridecyl-1,4-benzoquinone (trivial name: rapanone) [2, 3]. In a recent study [4], a novel flavonol glycoside along with four known ones has been isolated from this source. In continuing investigation of a methanolic extract of the plant leaves, we now report the isolation and characterization of two new flavonol glycosides (**1** and **2**). Also being reported are eight known ones identified as isorhamnetin 3-*O*-β-glucoside (**3**), quercetin 3-*O*-α-rhamnoside (**4**), quercetin (**5**), myricetin (**6**), quercetin 3-*O*-β-glucoside (**7**), myricetin 3-*O*-β-glucoside (**8**), myricetin 3-*O*-β-xyloside (**9**) and quercetin 3-*O*-rutinoside (**10**) [5-7].



**1** R = galactosyl (1→4)-galactoside, R<sub>1</sub> = Me; **2** R = [rhamnosyl (1→2)][rhamnosyl (1→4)]-rhamnoside, R<sub>1</sub> = H; **3** R = glucose, R<sub>1</sub> = Me; **4** R = rhamnose R<sub>1</sub> = H.

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## RESULTS AND DISCUSSION

Compound **1** afforded UV data, which suggested free hydroxyls at C-5, C-7, and C-4 [8-10]. The assignment of the  $^1\text{H}$  and  $^{13}\text{C}$  NMR data supported by NOESY and HMBC correlations confirmed the aglycone as isorhamnetin (3'-*O*-methylquercetin) substituted at C-3. Acid hydrolysis gave galactose as the sugar residue. The FAB mass spectrum results indicated galactosylgalactoside bios as evidenced by the cleavage behaviour (see experimental section). In the  $^{13}\text{C}$  NMR spectrum a signal due to C-4'' of residual galactose shifted downfield at  $\delta$  82.40 in comparison to isorhamnetin 3-*O*- $\beta$ -glucoside (**3**), suggesting interglycosidic linkage at this position. This was corroborated by HMBC correlation between the anomeric proton ( $\delta$  4.65) of terminal galactose and a carbon ( $\delta$  82.40) assigned to C-4'' of the residual galactose and further confirmed by NOESY cross peaks between the terminal galactose anomeric proton ( $\delta$  4.65) and the C-4'' proton ( $\delta$  3.68). Therefore compound **1** was concluded as isorhamnetin 3-*O*- $\beta$ -galactosyl (1 $\rightarrow$ 4)- $\beta$ -galactoside.

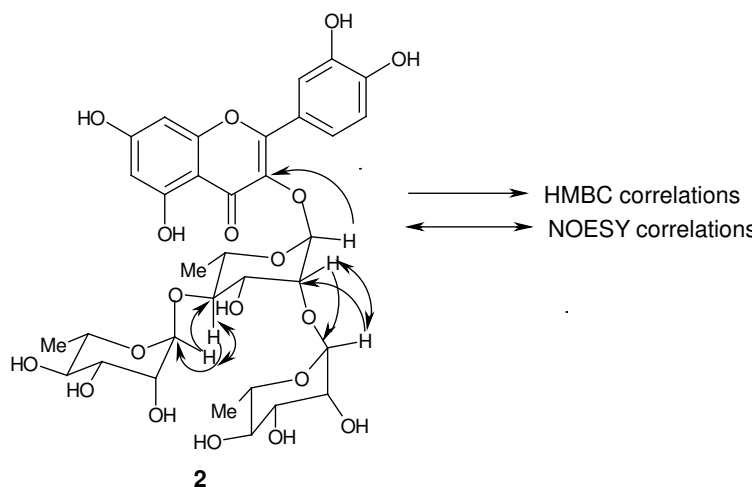


Figure 1. Pertinent HMBC and NOESY correlations observed with compound **2**.

Compound **2** showed a molecular ion peak at  $m/z$  741  $[\text{M}+\text{H}]^+$ , analyzing for  $\text{C}_{33}\text{H}_{40}\text{O}_{19}$ . Acid hydrolysis afforded rhamnose as the sugar residue. Its  $^1\text{H}$  NMR data showed a 2H AX and a 3H ABX system characteristic of quercetin moiety [11] while the UV data recorded in MeOH and with addition of common shift reagents [8, 9] suggested that the residual sugar was attached at C-3 position of the aglycone, a fact further supported by HMBC correlations as outlined in Figure 1. From the mass spectrum data, fragments at  $m/z$  595  $[\text{M}-146 + \text{H}]^+$  (loss of rhamnose), 449  $[\text{M}-2 \times 146 + \text{H}]^+$  (loss of two rhamnoses), 303  $[\text{M}-3 \times 146 + \text{H}]^+$  (loss of three rhamnoses), 339  $[3 \times \text{rhamnose}]^+$ , 293  $[2 \times \text{rhamnose}]^+$  (Figure 2) together with three anomeric protons of rhamnose units at  $\delta$  5.10 (d,  $J = 1.1$  Hz), 4.75 (d,  $J = 1.2$  Hz) and 4.50 (d,  $J = 1.0$  Hz) indicated the presence of three sugar residues in the molecule. Furthermore, a comparative analysis of the  $^{13}\text{C}$  NMR chemical shift of the sugar unit with those of quercetin 3-*O*- $\alpha$ -rhamnoside (**4**) showed glycosylation shift for C-2'' (11.80 ppm) and C-4'' (10.30 ppm) in the residual rhamnose unit, thus suggesting the presence of 2'',4''-substituted rhamnose unit bearing two other terminal ones. The signal at  $\delta$  82.30 attributed to C-2'' of the residual rhamnose suggested rhamnosyl (1 $\rightarrow$ 2)-rhamnoside moiety [12] while the signal at  $\delta$  81.60 attributed to C-4'' of the same rhamnose

indicated a rhamnosyl (1→4)-rhamnoside moiety [13], a fact further supported by NOESY and HMBC spectra correlations (Figure 1). On this basis compound **2** was established as quercetin 3-*O*-( $\alpha$ -rhamnosyl (1→2)) [ $\alpha$ -rhamnosyl (1→4)]- $\alpha$ -rhamnoside.

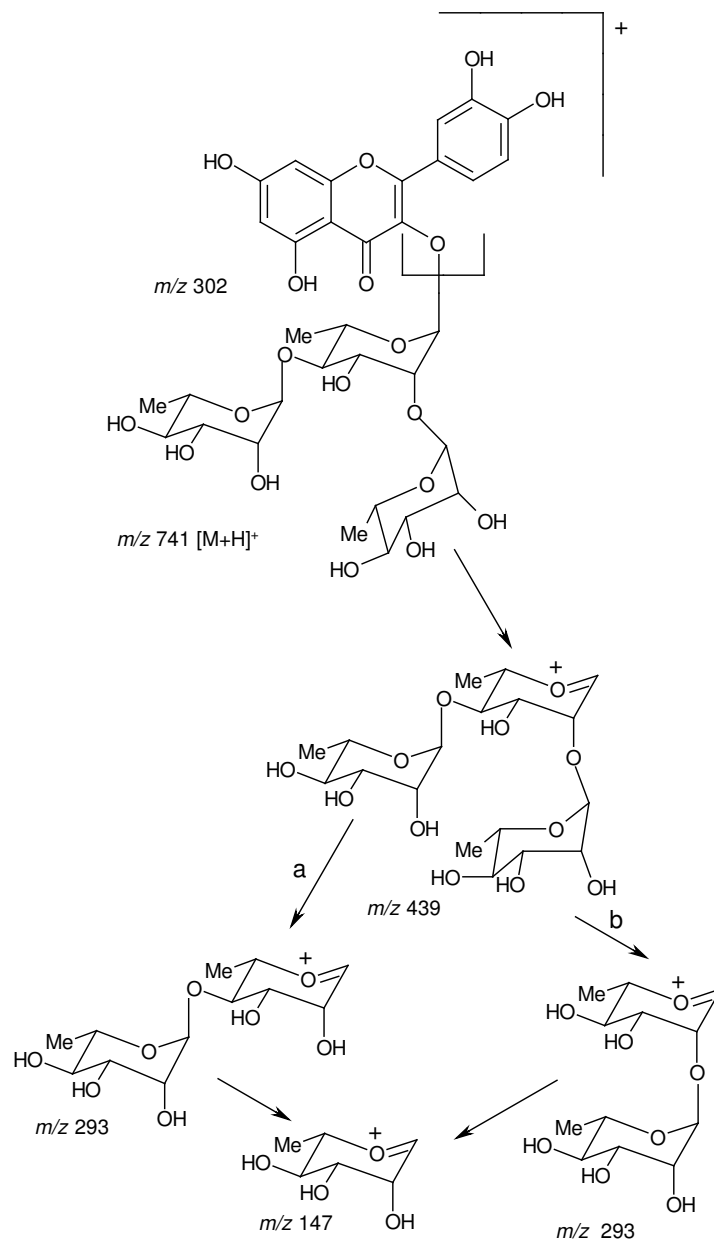


Figure 2. Fragmentation of compound **2** during FAB.

Table 1. <sup>13</sup>C NMR data of compounds **1-4**.

Carbon No.	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>
1				
2	156.80	157.30	158.70	158.30
3	135.75	134.20	135.40	135.60
4	177.40	177.70	179.40	178.65
5	161.50	161.30	163.04	162.30
6	99.01	98.70	98.86	98.80
7	164.45	164.20	165.91	163.70
8	94.00	93.80	94.73	94.02
9	156.10	156.80	158.20	157.40
10	104.10	104.00	105.80	104.50
1'	120.75	120.70	123.10	122.35
2'	113.90	115.60	116.00	116.60
3'	148.35	147.80	148.40	149.10
4'	147.16	145.20	150.80	148.20
5'	115.65	115.40	114.40	116.00
6'	121.90	121.10	123.80	123.50
1''	103.80	101.20	104.71	102.10
2''	74.30	82.30	75.90	70.50
3''	76.50	78.01	77.08	71.90
4''	82.40	81.60	71.50	71.30
5''	73.00	76.80	76.50	69.70
6''	61.90	18.80	62.60	17.80
1'''	104.80	99.80		
2'''	72.45	70.40		
3'''	74.85	72.90		
4'''	68.70	71.00		
5'''	75.30	70.10		
6'''	60.50	18.30		
1''''		101.10		
2''''		71.40		
3''''		73.50		
4''''		72.00		
5''''		69.80		
6''''		17.70		
3'-OMe	57.60		58.20	

## EXPERIMENTAL

*General.* UV spectra were recorded on a 8452 A Hewlett Packard Array spectrophotometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded in CDCl<sub>3</sub>-DMSO-d<sub>6</sub> mixture and DMSO-d<sub>6</sub> on a Bucker WM instrument operating at 360 and 90 MHz, respectively. EIMS were measured using 70 eV MAT 8200 A Varian Bremen instrument. Polish Academy of Sciences provided FAB-MS spectra. Silica gel for column and TLC plates were impregnated with 2% oxalic acid solution. Semi-preparative high performance liquid chromatography (HPLC) was performed on a Bischoff instrument connected to a 785 programmable absorbance detector and a programmable monitor 8252 dual pen recorder.

*Plant material.* *Embelia schimperi* leaves were collected from Ngong hills in June 1997 and voucher specimens were identified after comparison with authentic samples at Botany Department Herbarium, Nairobi University.

*Extraction and isolation.* The CH<sub>2</sub>Cl<sub>2</sub> defatted powdered leaves (approx. 1 kg) was further extracted with MeOH (2.5 L x 3) for one week. The extracts were combined and freed from the solvent under reduced pressure to give a dark green residue (105 g). A portion of the extract (100 g) was subjected to column chromatography with CH<sub>2</sub>Cl<sub>2</sub>-MeOH gradient to pure MeOH affording 160 fractions of 100 mL each. The composition of the fractions were determined by TLC using solvent system CH<sub>2</sub>Cl<sub>2</sub>-MeOH (9:1, 3:2) and those showing similar profiles were combined to give three major pools. The first pool (fractions 20-80, 25 g) was subsequently subjected to repeated flash chromatography with CH<sub>2</sub>Cl<sub>2</sub>-MeOH (95:5) followed by the same solvent system in the ratio 9:1 to give **5** (29 mg), **6** (55 mg) and **4** (24 mg). Fractions 81-105 (Pool II) was evaporated to give 17 g which upon repeated flash chromatography with CH<sub>2</sub>Cl<sub>2</sub>-MeOH (4:1) and collecting 10 mL each, afforded **7** (66 mg), **8** (34 mg), **3** (65 mg) and **9** (35 mg). Pool III (fractions 106-160, 35 g) mainly from methanol elution was further purified by column chromatography over sephadex LH-20 using different solvent systems. Final purification of the eluates by preparative HPLC on reverse phase (RP-18) using MeOH-H<sub>2</sub>O (7:3) gave **10**, **1** and **2** in 75 mg, 48 mg, and 55 mg, respectively.

*Isorhamnetin 3-O-β-galactosyl (1→4)-β-galactoside (1).* Amorphous yellow powder. UV λ<sub>max</sub> (MeOH) nm: 257, 268, 358; (+AlCl<sub>3</sub>) 268, 301, 360, 398; (+AlCl<sub>3</sub>/HCl) 262, 302, 402, 424; (+NaOMe) 272, 324, 420; (+NaOAc) 270, 322, 408; (NaOAc/H<sub>3</sub>BO<sub>3</sub>) 256, 266, 362. IR ν<sub>max</sub> (KBr) cm<sup>-1</sup>: 3510, 3450, 1680, 1650, 1580, 1460, 1410, 1390, 1365, 1250, 1200, 1180, 1050, 900. <sup>1</sup>H NMR (CDCl<sub>3</sub>+one drop DMSO-d<sub>6</sub>) δ: 12.64 (s, OH-5, D<sub>2</sub>O exchange.), 10.20 (br s, OH-7, D<sub>2</sub>O exchange.), 8.30 (br s, OH-4', D<sub>2</sub>O exchange.), 7.90 (d, *J* = 2.3 Hz, H-2'), 7.80 (dd, *J* = 8.2, 2.3 Hz, H-6'), 6.98 (d, *J* = 8.2 Hz, H-5'), 6.50 (d, *J* = 1.8 Hz, H-8), 6.35 (d, *J* = 1.8 Hz, H-6), 3.85 (s, OMe); residual galactose: 5.30 (d, *J* = 7.7 Hz, H-1''), 3.78 (d, *J* = 12.1, 3.2 Hz, H-6''<sub>B</sub>), 3.68 (m, H-4''), 3.60 (dd, *J* = 12.1, 5.4 Hz, H-6''<sub>A</sub>), 3.40 (m, H-3''), 3.25 (m, H-5''), 3.16 (m, H-2''); terminal galactose: 4.65 (d, *J* = 7.60 Hz, H-1'''), 3.97 (m, H-6'''<sub>B</sub>), 3.73 (m, H-6'''<sub>A</sub>), 3.55 (m, H-4'''), 3.45 (m, H-3'''), 3.30 (m, H-5'''), 3.27 (m, H-2'''). <sup>13</sup>C NMR data: see Table 1. EIMS (70 eV): *m/z* (%) 302 (100), 153 (27), 137 (18), 136 (9). FAB-MS (positive ion mode): 641 [M+H]<sup>+</sup>, 479 [M-162+H]<sup>+</sup>, 317 [M-2 x 162+H]<sup>+</sup>, 153, 137.

*Quercetin 3-O-[α-rhamnosyl(1→2)] [α-rhamnosyl(1→4)]-α-rhamnoside (2).* A yellow powder. UV λ<sub>max</sub> (MeOH) nm: 258, 301, 358; (+AlCl<sub>3</sub>) 274, 300, 430; (+AlCl<sub>3</sub>/HCl) 270, 300, 402; (+NaOMe) 272, 328, 414; (+NaOAc) 273, 324, 380, 418; (NaOAc/H<sub>3</sub>BO<sub>3</sub>) 260, 379. IR ν<sub>max</sub> (KBr) cm<sup>-1</sup>: 3500, 1665, 1570, 1510, 1430, 1250, 950. <sup>1</sup>H NMR (99% DMSO-d<sub>6</sub>) δ: 12.80 (s, OH-5, D<sub>2</sub>O exchange.), 10.30 (br s, OH-7, D<sub>2</sub>O exchange.), 9.20-8.50 (br s, OH-3' and 4', D<sub>2</sub>O exchange.), 7.80 (d, *J* = 2 Hz, H-2'), 7.64 (dd, *J* = 8.4, 2 Hz, H-6'), 6.80 (d, *J* = 8.3 Hz, H-5'), 6.40 (d, *J* = 2.2 Hz, H-8), 6.24 (d, *J* = 2.2 Hz, H-6); residual rhamnose: 5.10 (d, *J* = 1.1 Hz, H-1''), 3.70 (d, *J* = 9.3, 7.6 Hz, H-2''), 3.47 (t, *J* = 9.2 Hz, H-3''), 3.36 (t, *J* = 9.1 Hz, H-4''), 3.20 (ddd, *J* = 9.1, 5.3, 2.3 Hz, H-5''), 1.20 (d, *J* = 6.5 Hz, Me-6''); 2''-O-rhamnose: 4.50 (d, *J* = 1.0 Hz, H-1'''), 3.45 (dd, *J* = 9.3, 7.5 Hz, H-2'''), 3.39 (t, *J* = 9 Hz, H-3'''), 3.34 (t, *J* = 9.1 Hz, H-4'''), 3.24 (dd, *J* = 9.1, 5.2, 2.5 Hz, H-5'''), 1.0 (d, *J* = 6.6 Hz, Me-6'''); 4''-O-rhamnose: 4.75 (d, *J* = 1.2 Hz, H-1'''), 3.85 (d, *J* = 3.3 Hz, H-2'''), 3.60 (dd, *J* = 8.5, 3.3 Hz, H-3'''), 3.30 (d, *J* = 11, 6.2 Hz, H-5'''), 3.20 (dd, *J* = 9.5, 7.8 Hz, H-4'''), 0.80 (d, *J* = 6.7 Hz, Me-6'''). <sup>13</sup>C NMR data: see Table 1. EIMS (70 eV): *m/z* (%) 302 (100), 162 (4), 153 (20), 147 (7), 137 (9). FAB-MS (positive ion mode): 741 [M+H]<sup>+</sup>, 595 [M-146+H]<sup>+</sup>, 449 [M-2 x 146+H]<sup>+</sup>, 303 [M-3 x 146+H]<sup>+</sup>, 339, 293, 180, 151, 146, 137.

*Isorhamnetin 3-O-β-glucoside (3)*. A pale yellow powder. UV  $\lambda_{\max}$  (MeOH) nm: 260, 304, 360; (+AlCl<sub>3</sub>) 268, 302, 402; (+AlCl<sub>3</sub>/HCl) 272, 302, 398; (+NaOMe) 272, 396; (+NaOAc) 270, 320, 360, 406; (NaOAc/H<sub>3</sub>BO<sub>3</sub>) 260, 370. IR  $\nu_{\max}$  (KBr) cm<sup>-1</sup>: 3450, 1650, 1600, 1580, 1440, 1360, 1290, 1140, 1050, 990. <sup>1</sup>H NMR (CDCl<sub>3</sub>+one drop DMSO-d<sub>6</sub>)  $\delta$ : 12.50 (s, OH-5, D<sub>2</sub>O exchang.), 10.25 (br s, OH-7, D<sub>2</sub>O exchang.), 8.60 (br s, OH-4', D<sub>2</sub>O exchang.), 7.82 (d,  $J$  = 1.9 Hz, H-2'), 7.46 (dd,  $J$  = 8.4, 2.0 Hz, H-6'), 6.85 (d,  $J$  = 8.4, 2.0 Hz, H-5'), 6.31 (d,  $J$  = 2.1 Hz, H-8), 6.16 (d,  $J$  = 2.1 Hz, H-6), 3.85 (s, OMe); residual glucose: 5.10 (d,  $J$  = 7.6 Hz, H-1"), 3.80 (dd,  $J$  = 11.6, 3.6 Hz, H-6" <sub>B</sub>), 3.66 (dd,  $J$  = 11.6, 5.8 Hz, H-6" <sub>A</sub>), 3.50 (t,  $J$  = 8.4 Hz, H-4"), 3.45 (m, H-3"), 3.37 (dd,  $J$  = 10.2, 7.7 Hz, H-2"), 3.28 (m, H-4"). <sup>13</sup>C NMR data: see Table 1. EIMS (70 eV):  $m/z$  (%) 316 (100), 245 (10), 150 (5), 128 (10), 83 (57). FAB-MS (positive ion mode): 449 [M+H]<sup>+</sup>, 317 [M-162+H]<sup>+</sup>, 151, 137.

*Quercetin 3-O-α-rhamnoside (4)*. A yellow amorphous powder. UV  $\lambda_{\max}$  (MeOH) nm: 260, 300, 357; (+AlCl<sub>3</sub>) 272, 301, 428; (+AlCl<sub>3</sub>/HCl) 272, 301, 401; (+NaOMe) 271, 330, 414; (+NaOAc) 274, 320, 384; (NaOAc/H<sub>3</sub>BO<sub>3</sub>) 260, 382. IR  $\nu_{\max}$  (KBr) cm<sup>-1</sup>: 3500, 1665, 1580, 1520, 1480, 1450, 1120, and 1020. <sup>1</sup>H NMR (CDCl<sub>3</sub>+ DMSO-d<sub>6</sub>)  $\delta$ : 12.45 (s, OH-5, D<sub>2</sub>O exchang.), 10.50 (br s, OH-7, D<sub>2</sub>O exchang.), 9.20-8.50 (br s, OH-3' and 4', D<sub>2</sub>O exchang.), 7.86 (d,  $J$  = 2.2 Hz, H-2'), 7.70 (dd,  $J$  = 8.4, 2.3 Hz, H-6'), 6.90 (d,  $J$  = 8.4 Hz, H-5'), 6.40 (d,  $J$  = 2 Hz, H-8), 6.20 (d,  $J$  = 2 Hz, H-6); rhamnose: 5.15 (d,  $J$  = 0.9 Hz, H-1"), 3.65 (br d,  $J$  = 3.5 Hz, H-2"), 3.55 (dd,  $J$  = 9.1, 3.1 Hz, H-3"), 3.44 (m, H-4"), 3.35 (m, H-5"), 1.00 (d,  $J$  = 6.5 Hz, Me-6"). <sup>13</sup>C NMR data: see Table 1. EIMS (70 eV):  $m/z$  (%) 302 (100), 153 (21), 137 (30).

*Acid hydrolysis*. Compounds (**1-4**), 15 mg each in a mixture of 8% HCl (2 mL) and MeOH (20 mL) were separately refluxed for 2 h. The reaction mixtures were reduced under pressure to dryness, dissolved in H<sub>2</sub>O (3 mL) and neutralized with NaOH. The neutralized products were subjected to TLC analysis (eluent, EtOAc-MeOH-H<sub>2</sub>O-HOAc, 6:2:1:1). The chromatograms were sprayed with aniline hydrogen phthalate followed by heating at 100 °C. The sugars were identified after comparison with authentic samples.

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#### REFERENCES

1. Kokwaro, J.O. *Medicinal Plants of East Africa*, East African Literature Bureau: Nairobi; **1976**; p 49.
2. Midiwo, J.O.; Manguro Arot, L.O.; Mbakaya, C.L. *Bull. Chem. Soc. Ethiop.* **1988**, 2, 83.
3. Midiwo, J.O.; Manguro Arot, L.O.; Odingo, J.O. *Bull. Chem. Soc. Ethiop.* **1990**, 4, 71.
4. Manguro Arot, L.O.; Williams, L.A.D. *Phytochemistry* **1997**, 44, 1398.
5. Manguro Arot, L.O.; Williams, L.A.D. *Planta Med.* **1996**, 62, 178.
6. Manguro Arot, L.O.; Midiwo, J.O.; Kraus, W. *Phytochemistry* **1996**, 43, 1107.
7. Manguro Arot, L.O.; Midiwo, J.O.; Kraus, W. *Nat.Prod. Lett.* **1996**, 9, 121.

8. Markham, K.R.; Chari, V.M.; Mabry, T.J. in *The Flavonoid Advances in Research*, Harbone, J.B.; Mabry, T.J. (Eds.); Chapman and Hall: London; **1982**; pp 28-52.
9. Mabry, T.J.; Markham, K.R.; Thomas, M.B. *The Systematic Identification of Flavonoids*, Springer-Verlag: Berlin; **1970**; Chapter 1.
10. Markham, K.R. *Techniques in Flavonoids Identification*, Academic Press: London; **1982**; Chapters 1 and 2.
11. Manguro Arot, L.O.; Mukonyi, K.W.; Githiomi, J.K. *Nat. Prod. Lett.* **1995**, 7, 163.
12. Markham, K.R.; Geiger, H.; Jaggy, H. *Phytochemistry* **1992**, 31, 1009.
13. Shao, Y.; Poobrasert, O.; Kenelly, E.J.; Chin, C.; Chi-Tang, H.; Huang, M.; Garrison, S.A.; Cordell, G.A. *Planta Med.* **1997**, 63, 258.