

## Technical Notes and Comments

### Gas Exchange in Rivers and Estuaries: Choosing a Gas Transfer Velocity

PETER A. RAYMOND<sup>1,\*</sup> and JONATHAN J. COLE<sup>2</sup>

<sup>1</sup> *The Ecosystems Center, Marine Biological Laboratory, Woods Hole, Massachusetts 02543*

<sup>2</sup> *Institute of Ecosystem Studies, Millbrook, New York 12545*

We are writing this comment to call attention to large uncertainties in the estimates of CO<sub>2</sub> flux from rivers and estuaries, a topic that has been receiving considerable attention recently (Raymond et al. 1997, 2000; Cai and Wang 1998; Frankignoulle et al. 1998). It is our view that there are too few direct measurements of the physical component of gas exchange (e.g., the gas transfer velocity, piston velocity, gas exchange coefficient, or *k*) for rivers and estuaries. While studies in streams, lakes, and marine systems have progressed to the point where the gas transfer velocity can be partially predicted from physical forcing functions (O'Connor and Dobbins 1958; Cole and Caraco 1998; Wanninkhof and McGillis 1999), this is not yet the case for estuaries and rivers. A comparison of gas transfer velocity measurements in estuaries and rivers reveals a general lack of agreement among studies and physically-based predictive models. Until we have a better understanding of the magnitude and causes of variation in estuarine gas transfer velocity estimates, it will be difficult to use gas exchange in rivers or estuaries to accurately mass-balance gases of interest.

The exchange of CO<sub>2</sub> between an aquatic ecosystem and the overlying atmosphere is an area of intense interest for several reasons: aquatic systems can be significant CO<sub>2</sub> sources or sinks on a global or regional scale (Kling et al. 1991; Quay et al. 1992; Sarmiento and Sundquist 1992; Cole et al. 1994); the magnitude and direction of an ecosystem's CO<sub>2</sub> flux can provide important clues about metabolism in a given system (Depetris and Kempe 1993; Gattuso et al. 1993; Hamilton et al. 1995;

Raymond et al. 1997; Boehme et al. 1998); and can suggest ways in which the aquatic system is linked to its watershed (Dillon and Molot 1997; Jones and Mulholland 1998; del Giorgio et al. 1999). Estuaries and rivers receive inorganic nutrients and organic C from land and both are important systems where terrestrial nutrients and organic C are processed before entering the ocean. Some estuaries, or parts of estuaries are dominated by autotrophic C inputs while others are clearly net heterotrophic and are considered strong exporters of CO<sub>2</sub> to the atmosphere (Kemp et al. 1997; Smith and Hollibaugh 1997; Cai and Wang 1998; Frankignoulle et al. 1998; Raymond et al. 2000). Developing accurate estimates of the gas transfer velocity in estuaries has become an important research goal in recent years (Marino and Howarth 1993; Clark et al. 1994; Carini et al. 1996).

The flux of CO<sub>2</sub> depends mainly on two factors: the concentration gradient between the surface water and the air and the physical transfer or turbulent energy at this interface (MacIntyre et al. 1995). Thus,

$$\text{Flux} = k \alpha ((p\text{CO}_2 \times K_h) - [\text{CO}_2]_{\text{sat}})$$

Where  $[\text{CO}_2]_{\text{sat}}$  is the concentration of CO<sub>2</sub> in water at equilibrium with the overlying atmosphere;  $K_h$  is Henry's constant for CO<sub>2</sub> at a given temperature and salinity;  $p\text{CO}_2$  is the partial pressure of CO<sub>2</sub> in the surface water;  $k$  is the gas transfer velocity for CO<sub>2</sub>; and  $\alpha$  is the coefficient of chemical enhancement, which affects CO<sub>2</sub> flux only at high pH and low values of  $k$  (Degrandpre et al. 1995; Wanninkhof and Knox 1996). In the normal range of estuarine pH values, this equation shows that the flux of CO<sub>2</sub> is governed by the concentration gradient between air and water, and  $k$ . To make comparisons,  $k$  is often reported as  $k_{600}$ , which is the  $k$  for CO<sub>2</sub> at 20°C in freshwater, that is,  $k$  at a Schmidt number of 600. The methodology for measuring CO<sub>2</sub> concentrations in both air and water has progressed to the stage where it is now possible to make continuous, accurate measurements of the concentration gradient by a variety of means in almost any type of aquatic system (Goyet and Snover 1993; Millero 1995; Oudot et al. 1995; Sellers et al. 1995; Carignan 1998; Murphy 1998). While there are a number of measurement options and equations for choosing a value for  $k_{600}$ , the estimation of  $k_{600}$  is by far the most problematic term in the flux equation.

\* Corresponding author: tele: 508/289-7695; e-mail: praymond@mbl.edu.

Because the number of studies that directly measured  $k_{600}$  are few, authors generally rely on predictive equations to constrain  $k_{600}$  for their system. Theoretically  $k_{600}$  is constrained by turbulent mixing in the surface aqueous boundary layer. Research suggests that in lakes and oceans the dominant source of turbulence in the surface aqueous boundary layer is controlled by wind stress and therefore  $k_{600}$  is a function of wind-speed (Broecker and Peng 1974; Liss and Merlivat 1986; Wanninkhof 1992; Cole and Caraco 1998; Wanninkhof and McGillis 1999), although the relationship breaks down at low wind speeds (Clark et al. 1994; MacIntyre et al. 1995; Cole and Caraco 1998). It should be noted that other processes such as rainfall (Ho et al. 1997) and the formation of thin films (Frew 1997) affect the gas transfer velocity. In shallower streams and rivers the source of turbulence in the surface micro-layer is also due to bottom stress (the turbulence created by friction between water passing over the bottom) and  $k_{600}$  is a function of the stream depth and flow velocity (O'Connor and Dobbins 1958; Owens et al. 1964; Langbein and Durum 1967). For example, shallow fast running streams generally have higher  $k_{600}$ s than slow sluggish streams.

Estuaries represent a unique situation because they may fit in either category. They can be shallow with high tidal velocities, or deep with low tidal velocities. Cerco (1989) concluded, based on theoretical predictive models, that in estuaries the dominant source of turbulence at the surface aqueous boundary layer can originate from both wind and water-shear produced turbulence, and therefore depends on the depth, mean tidal velocity, and wind regime of a given estuary. Recent measurements of  $k_{600}$  in estuaries and across systems warrant a new look at the expected magnitude of  $k_{600}$  in estuaries. The objective of this comment is to review the predictive equations and measurements of  $k$  in rivers and estuaries in an attempt to predict the dominant source of turbulence in estuaries and evaluate if the current models are accurate enough to utilize gas exchange measurements in current inorganic carbon studies.

Figure 1 displays results of predictive equations for  $k_{600}$  using wind speed and bottom turbulence. According to Fig. 1, wind stress dominates turbulence in the surface aqueous boundary layer for all systems with depths greater than 10 m, or at wind-speeds greater than 8 m s<sup>-1</sup>; at all other depths and wind speeds either wind stress or bottom stress may dominate. Wind data compiled for 39 coastal cities in the United States shows a mean daily wind speed of 4.6 ± 0.28 m s<sup>-1</sup> (95% CI) and ranges from 3 to 7.7 m s<sup>-1</sup> among sites (wind data from www.noaa.gov). The median and modal wind

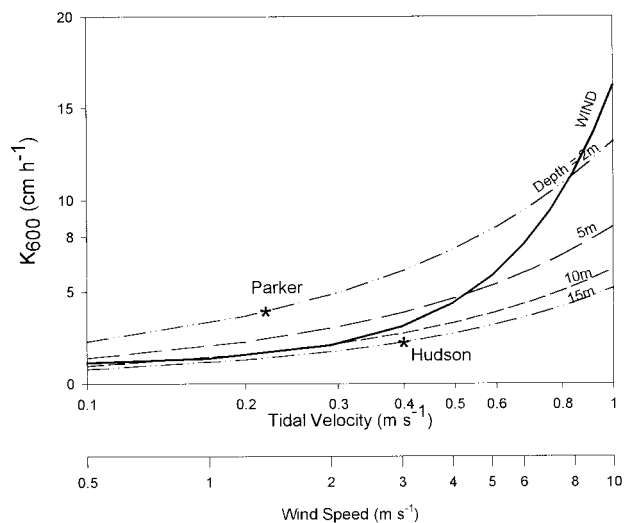


Fig. 1. Values of  $k_{600}$  based on predictive equations (after Cerco 1989). The solid line represents values of  $k_{600}$  using the average output from equations by Cole and Caraco (1998) and Wanninkhof (1992), which are based on wind-dominated turbulence and were generated through literature review. The dashed lines are the mean output from equations by Langbein and Durum (1967), O'Connor and Dobbins (1958), and Owens et al. (1964), which are based on turbulence generated from bottom stress and were originally developed for stream systems. According to the stream equations, turbulence in the surface aqueous boundary layer is a function of water depth and tidal velocity. For the stream equations, we show the output from systems with a mean depth of 2, 5, 10, and 20 m at varying average tidal velocities. The \* locate the Hudson and Parker Rivers on their respective depth and tidal velocity curve.

speeds are 4.7 and 4.74 m s<sup>-1</sup>, respectively, in this data set. Grouping the Hudson River and Parker River average tidal velocities (0.38 and 0.22 m s<sup>-1</sup>, respectively) with the tidal velocities reported for 7 estuaries in Officer (1976) produced an average tidal velocity of 0.34 ± 0.18 m s<sup>-1</sup>. Thus according to Fig. 1, unless an estuary has either above average winds, is exceptionally shallow, or has rapid tidal velocities we would expect  $k_{600}$  to be in the range of 3–7 cm h<sup>-1</sup>.

The predictive equations are in reasonably good agreement with purposeful gas tracer experiments (i.e., SF<sub>6</sub> additions) conducted in two estuaries, the upper-Parker estuary (Carini et al. 1996) and the tidal-freshwater Hudson River (Clark et al. 1994). We believe purposeful gas tracer experiments are the most applicable studies for comparison because they measure average gas transfer velocities over long time scales. The Parker River is a macrotidal estuary with tides of 2.9 m (Vallino and Hopkinson 1998), and the Carini et al. (1996) study was done at depths of ~1.9 m, widths of < 100 m, and tidal velocities of ~0.2 m s<sup>-1</sup> (Vallino personal communication). The Hudson River study was conducted in the tidal freshwater Hudson at depths of

TABLE 1. Average gas transfer velocities corrected to a Schmidt number of 600 ( $k_{600}$ ) for rivers and estuaries. We limited our search to systems with depths greater than 1 m. In order to convert reported  $k$  for  $O_2$  and radon studies to  $k_{600}$  we employed equations offered in Wanninkhof (1992). For the Pee Dee River and South San Francisco Bay natural gas tracer experiments we used the equation in Elsinger and Moore (1983) to estimate the diffusivity of radon at different temperatures.

System and Study	Type of Study	$k_{600}$ (cm h <sup>-1</sup> )	
		Range	Average
Hudson River, Clark et al. (1994)	Purposeful gas tracer	1.5–9.0	4.8
Parker River, Carini et al. (1996)	Purposeful gas tracer	1.4–6.1	3.8*
South San Francisco Bay, Hammond and Fuller (1979)	Natural gas tracer ( <sup>222</sup> Rn)	1.0–6.7	4.3
South San Francisco Bay, Hartman and Hammond (1984)	Floating Dome	1.1–12.8	5.7
Amazon and Tributaries, Devol et al. (1987)	Floating Dome	2.4–9.9	6.0
Narragansett Bay, Roques (1985)	Floating Dome	4.5–11.0	7.4
Hudson River, Marino and Howarth (1993)	Floating Dome	3.3–26.0	11.6
Pee Dee River, Elsinger and Moore (1983)	Natural gas tracer ( <sup>222</sup> Rn)	10.2–30	12.6
Hudson River, Clark et al. (1992)	Natural gas tracer (CFC)	2.0–4.0	3.0
This study	Predictive Equations	3.0–7.0	

\* Data from rain event are excluded.

~10–15 m, widths of ~2 km, and tidal velocities of 0.38 m s<sup>-1</sup> (Clark et al. 1994). We used the data in Fig. 1c of Carini et al. (1996) which excluded a high rain event during one of the days of the Parker River study. The turbulence created by raindrops increases gas exchange (Ho et al. 1997).

Of the 11 independent estimates of  $k_{600}$  offered by Clark et al. (1994) and Carini et al. (1996), only one is above 8 cm h<sup>-1</sup>, and the average is 4.4 ± 2.4 cm h<sup>-1</sup>. Clark et al. (1994) and Carini et al. (1996) both reported direct relationships between wind and  $k_{600}$ , suggesting that wind is an important source of turbulence at the surface aqueous boundary layer in these two systems.

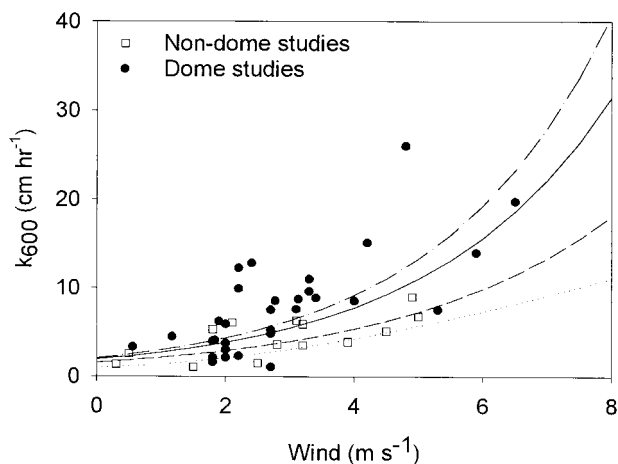


Fig. 2.  $k_{600}$  as a function of wind speed for all data that reported wind in Table 1. The circles are for floating dome studies, while the squares are from purposeful tracer additions and natural tracer studies. The lines are regressions for all data (solid line;  $y = 1.91e^{0.35x}$ ;  $r^2 = 0.53$ ;  $p < 0.0001$ ), the dome data (dot to dashed line;  $y = 2.06e^{0.37x}$ ;  $r^2 = 0.56$ ;  $p < 0.0001$ ), non-dome data (dashed line;  $y = 1.58e^{0.30x}$ ;  $r^2 = 0.46$ ;  $p < 0.001$ ), and average output from equations by Cole and Caraco (1998) and Wanninkhof (1992), which were review papers (dotted line).

The two aforementioned purposeful estuarine gas tracer experiments are not the only applicable studies for comparison. There are three approaches that could be used to obtain  $k_{600}$ : natural tracers, purposeful tracer additions, and floating domes. Table 1 shows results from the three different methods and reports a range of 1 to 26 cm h<sup>-1</sup> for  $k_{600}$  in riverine/estuarine systems. The purposeful gas tracer studies and predictive equations were on the low end of the reported range (Table 1). Combining all data and methods, we see a significant ( $p < 0.001$ ) relationship between wind and  $k_{600}$  (Fig. 2). Splitting the data into dome studies and all other methods reveals that the relationship between dome estimates of  $k_{600}$  and wind is different, and slightly higher, than the estimates produced by the other methods (Fig. 2). Interestingly, the resulting estuarine wind-based equations based on the estuarine studies produce larger  $k_{600}$  than those produced using the general equations of Cole and Caraco (1998) and Wanninkhof (1992), which were derived largely from lakes and bays (Fig. 2). Therefore, the data indicate that at average wind speeds, estuaries may indeed have higher  $k_{600}$  values than average lake and marine systems.

Most of the estuarine measurements have used the floating dome approach (Table 1). The advantages of this method are its ease and simplicity and ability to resolve relatively short-term changes in gas flux. On the other hand, this method has been criticized for theoretical reasons. Mainly that the chamber disturbs the turbulence regime at the boundary layer (Broecker and Peng 1984) and laboratory experiments directly comparing floating dome to other methods have sometimes found little agreement (Belanger and Korzun 1991). Similarly, natural tracers are problematic in estuaries. The largest problem associated with natural tracer studies is that it is extremely difficult to mass bal-

ance any constituent in an estuarine system due to the many possible sources and sinks and complex mixing regimes. Purposeful gas tracer studies are the most difficult to do in field situations, and are rarely performed on estuaries, yet provide the least equivocal answer and are therefore the most promising for estuarine studies. Purposeful gas tracer experiments average exchange over days to weeks and over spatial scales up to several km<sup>2</sup>, in contrast to floating domes that provide a snapshot in time and integrate over < 1 m<sup>2</sup>. Therefore we believe, the purposeful gas tracers currently provide the best estimates of  $k_{600}$  when estimating gas exchange for entire estuaries over long time scales.

Table 1 represents a comprehensive list of studies that have measured  $k$  in large rivers and estuaries. The range of reported  $k_{600}$  in these systems is a large potential source of error when calculating gas exchange. In the Hudson River, for example, Raymond et al. (1997) obtained about a two-fold uncertainty in estimating CO<sub>2</sub> exchange from  $k_{600}$  predicted from the Clark et al. (1994) purposeful gas addition study (16 mmol m<sup>-2</sup> d<sup>-1</sup>) and the Marino and Howarth's (1993) floating dome study (37 mmol m<sup>-2</sup> d<sup>-1</sup>). Had they used a  $k_{600}$  of 8 cm h<sup>-1</sup>, the CO<sub>2</sub> flux estimate would be much larger still (71 mmol m<sup>-2</sup> d<sup>-1</sup>), more than 4-fold greater than that based on the purposeful gas tracer study.

Similarly, in order to demonstrate the ramifications of the potential source of error when choosing a  $k$  for CO<sub>2</sub> studies, we applied a  $k_{600}$  of 4 cm h<sup>-1</sup> to recent papers that used an average  $k$  of 8 cm h<sup>-1</sup> or greater. We are not stating that a  $k_{600}$  of 4 cm h<sup>-1</sup> is appropriate for the average estuary, however based on the data available it is a plausible alternative. Frankignoulle et al. (1998) used an average  $k$  of 8 cm h<sup>-1</sup> (based on direct measurements using the floating dome method) to estimate CO<sub>2</sub> flux in European estuaries, and the contribution of estuarine CO<sub>2</sub> evasion to anthropogenic CO<sub>2</sub> emissions of Western Europe. Fluxes of 100 to 500 mmol C m<sup>-2</sup> d<sup>-1</sup> and an evasion contribution of 5% to 10% were reported. If they had used an average  $k$  of 4 cm h<sup>-1</sup> the flux would drop to 50–250 mmol m<sup>-2</sup> d<sup>-1</sup>, and the contribution of estuaries to anthropogenic CO<sub>2</sub> emissions would be 50% less, or 2.5–5%. More importantly, the sources of CO<sub>2</sub> necessary to balance this CO<sub>2</sub> flux would also be 50% less.

Cai et al. (1999) and Cai and Wang (1998) mass balanced carbon and oxygen in the Satilla estuary in order to elucidate the processes that were balancing high CO<sub>2</sub> concentrations and fluxes. For the two studies they used an average  $k$  of 12.5 (Cai and Wang 1998) and 8.75 cm h<sup>-1</sup> (Cai et al. 1999) for CO<sub>2</sub> based on the work of Elsinger and Moore

(1983). These articles argue that in the upper reaches (30–35 km from mouth) of the Satilla, atmospheric CO<sub>2</sub> exchange cannot be balanced by pelagic respiration. In the upper reaches using an average  $k$  of 8 or 12 cm h<sup>-1</sup>, they reported atmospheric CO<sub>2</sub> fluxes of ~220 mmol m<sup>-2</sup> d<sup>-1</sup> for summer months (Table 4 in Cai and Wang 1998). Summertime pelagic respiratory rates in the Satilla are 24.2 mM m<sup>-3</sup> d<sup>-1</sup> (Table 1 in Cai et al. 1999). Assuming a depth of ~4 m (Cai and Wang 1998), pelagic rates of respiration on an areal basis were ~100 mmol m<sup>-2</sup> d<sup>-1</sup>, which is 50% of the total atmospheric flux using an average  $k$  of 8 cm h<sup>-1</sup>. If an average  $k$  of 4 cm h<sup>-1</sup> had been used, the atmospheric flux decreases to ~100 mmol m<sup>-2</sup> d<sup>-1</sup>, resulting in a system that is much closer to balance.

The bottom line is that there is a lack of direct estuarine  $k$  measurements using new accurate gas tracer techniques to predict  $k$  with any certainty. However, based on the predictive equations and purposeful gas tracer experiments it appears that at average wind speeds, tidal velocities, and estuary depth,  $k_{600}$  should be in the range of 3–7 cm h<sup>-1</sup>. We believe researchers should bracket gas exchange using  $k$  values in this range until further research has been conducted. We hope that this comment will produce discussion and future work that will aid in properly constraining  $k_{600}$  in estuarine systems.

#### ACKNOWLEDGMENTS

We thank W. McGillis for a thorough review of this manuscript, and J. Vallino and R. Wanninkhof for comments on a previous version. This work was supported by the Long Term Ecological Research Program of the U.S. National Science Foundation (OCE-9726921) and the Hudson River Foundation. This is a contribution to the Ecosystems Center of the Marine Biological Laboratory and the Institute of Ecosystem Studies.

#### LITERATURE CITED

- BELANGER, T. V. AND E. A. KORZUN. 1991. Critique of floating dome technique for estimating reaeration rates. *Journal of Environmental Engineering* 117:144–150.
- BOEHME, S. E., C. L. SABINE, AND C. E. REIMERS. 1998. CO<sub>2</sub> fluxes from a coastal transect: A time-series approach. *Marine Chemistry* 63:49–67.
- BROECKER, W. S. AND T.-H. PENG. 1974. Gas exchange rates between air and sea. *Tellus* 26:21–35.
- BROECKER, W. S. AND T.-H. PENG. 1984. Gas exchange measurements in natural systems, p. 479–495. In W. Brutsaert and G. H. Jirka (eds.), *Gas Transfer at Water Surfaces*. Reidel Publishing Co., Dordrecht.
- CAI, W.-J., L. R. POMEROY, M. A. MORAN, AND Y. WANG. 1999. Oxygen and carbon dioxide mass balance for the estuarine-intertidal marsh complexes of five rivers in the southeastern U.S. *Limnology and Oceanography* 44:639–679.
- CAI, W.-J. AND Y. WANG. 1998. The chemistry, fluxes, and sources of carbon dioxide in the estuarine waters of the Satilla and Altamaha Rivers, Georgia. *Limnology and Oceanography* 43:657–668.
- CARIGNAN, R. 1998. Automated determination of carbon diox-

- ide, oxygen, and nitrogen partial pressures in surface waters. *Limnology and Oceanography* 43:969–975.
- CARINI, S., N. WESTON, C. HOPKINSON, J. TUCKER, A. GIBLIN, AND J. VALLINO. 1996. Gas exchange in the Parker estuary, MA. *Biological Bulletin* 191:333–334.
- CERCO, C. F. 1989. Estimating estuarine reaeration rates. *Journal of Environmental Engineering* 115:1066–1070.
- CLARK, J. F., J. SIMPSON, W. M. SMETHIE, AND C. TOLES. 1992. Gas exchange in a contaminated estuary inferred from chlorofluorocarbons. *Geophysical Research Letters* 19:1133–1136.
- CLARK, J. F., R. WANNINKHOF, P. SCHLOSSER, AND H. J. SIMPSON. 1994. Gas exchange rates in the tidal Hudson River using a dual tracer technique. *Tellus* 46:274–285.
- COLE, J. J. AND N. F. CARACO. 1998. Atmospheric exchange of carbon dioxide in a low-wind oligotrophic lake measured by the addition of SF<sub>6</sub>. *Limnology and Oceanography* 43:647–656.
- COLE, J. J., N. F. CARACO, G. W. KLING, AND T. K. KRATZ. 1994. Carbon dioxide supersaturation in the surface waters of lakes. *Science* 265:1568–1570.
- DEGRANDPRE, M. D., W. R. MCGILLIS, N. M. FREW, AND E. J. BOCK. 1995. Laboratory measurements of CO<sub>2</sub> exchange in sea water, p. 375–383. In B. Jahne and E. Monahan (eds.), *Air-Water Gas Transfer*. Aeon-Verlag, Hanau.
- DEL GIORGIO, P. A., J. J. COLE, N. F. CARACO, AND R. H. PETERS. 1999. Linking planktonic biomass structure to plankton metabolism and net gas flux in northern temperate lakes. *Ecology* 80:1422–1431.
- DEPETRIS, P. J. AND S. KEMPE. 1993. Carbon dynamics and sources in the Parana River. *Limnology and Oceanography* 38:382–395.
- DEVOL, A. H., P. E. QUAY, J. E. RICHEY, AND L. A. MARTINELLI. 1987. The role of gas exchange in the inorganic carbon, oxygen, and <sup>222</sup>Rn budgets of the Amazon River. *Limnology and Oceanography* 32:235–248.
- DILLON, P. J. AND L. A. MOLOT. 1997. Effect of landscape on export of dissolved organic carbon, iron, and phosphorus from forested stream catchments. *Water Resources Research* 33:2591–2600.
- ELSINGER, R. J. AND W. S. MOORE. 1983. Gas exchange in the Pee Dee River based on <sup>222</sup>Rn evasion. *Geophysical Research Letters* 10:443–446.
- FRANKIGNOULLE, M., G. ABRIL, A. BORGES, I. BOURGE, C. CANON, B. DELILE, E. LIMBERT, AND J.-M. THEATE. 1998. Carbon dioxide emission from European estuaries. *Science* 282:434–436.
- FREW, N. M. 1997. The role of organic films in air-sea gas exchange, p. 121–163. In P. S. Liss and R. A. Duce (eds.), *The Sea Surface and Global Change*. Cambridge University Press, Cambridge.
- GATTUSO, J.-P., M. PICHON, B. DELESALLE, AND M. FRANKIGNOULLE. 1993. Community metabolism and air-sea CO<sub>2</sub> fluxes in a coral reef ecosystem (Moorea, French Polynesia). *Marine Ecology Progress Series* 96:259–267.
- GOYET, C. AND A. K. SNOVER. 1993. High-accuracy measurements of total dissolved inorganic carbon in the ocean: A comparison of alternate detection methods. *Marine Chemistry* 44:235–242.
- HAMILTON, S. K., S. J. SIPPEL, AND J. M. MELACK. 1995. Oxygen depletion and carbon dioxide and methane production in waters of the Pantanal wetland of Brazil. *Biogeochemistry* 30:115–141.
- HAMMOND, D. E. AND C. FULLER. 1979. The use of Radon-222 to estimate benthic exchange and atmospheric exchange in San Francisco Bay, p. 213–233. In T. J. Conomos (ed.), *San Francisco Bay, the Urbanized Estuary*. California Academy of Science, San Francisco.
- HARTMAN, B. AND D. E. HAMMOND. 1984. Gas exchange across the sediment-water and air-water interfaces in south San Francisco Bay. *Journal of Geophysical Research* 89:3593–3603.
- HO, D. T., L. F. BLIVEN, R. WANNINKHOF, AND P. SCHLOSSER. 1997. The effect of rain on air-water gas exchange. *Tellus* 49:149–158.
- JONES, J. AND P. MULHOLLAND. 1998. Carbon dioxide variation in a hardwood forest stream: An integrative measure of whole catchment soil respiration. *Ecosystems* 1:189–196.
- KEMP, W. M., E. SMITH, M. MARVIN-DIPASQUALE, AND W. R. BOYNTON. 1997. Organic carbon balance and net ecosystem metabolism in Chesapeake Bay. *Marine Ecology Progress Series* 150:229–248.
- KLING, G. W., G. W. KIPPHUT, AND M. C. MILLER. 1991. Arctic lakes and streams as gas conduits to the atmosphere: Implications for tundra carbon budgets. *Science* 251:298–301.
- LANGBEIN, W. B. AND W. J. DURUM. 1967. *The Aeration of Streams*. U.S. Geological Survey Techniques of Water Resources Investigation.
- LISS, P. S. AND L. MERLIVAT. 1986. Air-sea gas exchange rates: Introduction and synthesis, p. 113–127. In P. Buat-Menard (ed.), *The Role of Air-Sea Exchange in Geochemical Cycling*. D. Reidel Publishing Company, Dordrecht, Boston.
- MACINTYRE, S., R. WANNINKHOF, AND J. P. CHANTON. 1995. Trace gas exchange across the air-water interface in freshwaters and coastal marine environments, p. 52–57. In P. A. Mattson and R. C. Harris (eds.), *Biogenic Trace Gases: Measuring Emissions from Soils and Waters*. Blackwell, New York.
- MARINO, R. AND R. W. HOWARTH. 1993. Atmospheric oxygen exchange in the Hudson River: Dome measurements and comparison with other natural waters. *Estuaries* 16:433–445.
- MILLERO, F. J. 1995. Thermodynamics of the carbon dioxide system in the oceans. *Geochimica et Cosmochimica Acta* 59:661–677.
- MURPHY, P. P. 1998. On obtaining high-precision measurements of oceanic pCO<sub>2</sub> using infrared analyzers. *Marine Chemistry* 62:103–115.
- O'CONNOR, D. AND W. DOBBINS. 1958. Mechanism of reaeration in natural streams. *Transactions of the American Society of Civil Engineers* 123:641–684.
- OFFICER, C. B. 1976. *Physical Oceanography of Estuaries*. Wiley, New York.
- ODOT, C., J. F. TERNON, AND J. LECOMTE. 1995. Measurements of atmospheric and oceanic CO<sub>2</sub> in the tropical Atlantic: 10 years after the 1982–1984 FOCAL cruises. *Tellus* 47:70–85.
- OWENS, M., R. W. EDWARDS, AND J. W. GIBBS. 1964. Some reaeration studies in streams. *International Journal of Air and Water Pollution* 8:469–486.
- QUAY, P. D., B. TILBROOK, AND C. S. WONG. 1992. Oceanic uptake of fossil fuel CO<sub>2</sub>: Carbon-13 evidence. *Science* 256:74–78.
- RAYMOND, P. A., J. E. BAUER, AND J. J. COLE. 2000. Atmospheric CO<sub>2</sub> evasion, dissolved inorganic carbon production, and net heterotrophy in the York River Estuary. *Limnology and Oceanography* 45:1707–1717.
- RAYMOND, P. A., N. F. CARACO, AND J. J. COLE. 1997. Carbon dioxide concentration and atmospheric flux in the Hudson River. *Estuaries* 20:381–390.
- ROQUES, P. F. 1985. Rate and stoichiometry of nutrient remineralization in an anoxic estuary, the Pettaquamscutt River. Ph.D. Dissertation, University of Rhode Island, Narragansett, Rhode Island.
- SARMIENTO, J. L. AND E. T. SUNDQUIST. 1992. Revised budget for the oceanic uptake of anthropogenic carbon dioxide. *Nature* 356:589–593.
- SELLERS, P., R. H. HESSLEIN, AND C. A. KELLY. 1995. Continuous measurements of CO<sub>2</sub> for estimation of air-water fluxes in lakes: An in situ technique. *Limnology and Oceanography* 40:575–581.
- SMITH, S. V. AND J. T. HOLLIBAUGH. 1997. Annual cycle and interannual variability of ecosystem metabolism in a temperate climate embayment. *Ecological Monographs* 67:509–533.
- VALLINO, J. J. AND C. S. HOPKINSON. 1998. Estimation of disper-

- sion and characteristic mixing times in Plum Island Sound estuary. *Estuarine, Coastal and Shelf Science* 46:333–350.
- WANNINKHOF, R. 1992. Relationship between wind speed and gas exchange over the ocean. *Journal of Geophysical Research* 97: 7373–7382.
- WANNINKHOF, R. AND M. KNOX. 1996. Chemical enhancement of CO<sub>2</sub> exchange in natural waters. *Limnology and Oceanography* 41:689–697.
- WANNINKHOF, R. AND W. R. MCGILLIS. 1999. A cubic relationship between air-sea CO<sub>2</sub> exchange and wind speed. *Geophysical Research Letters* 26:1889–1892.

#### SOURCE OF UNPUBLISHED MATERIALS

VALLINO, J. Personal Communication. Marine Biological Laboratory, Woods Hole, Massachusetts.

*Received for consideration, March 28, 2000*  
*Accepted for publication, November 20, 2000*