Glucan-like synthetic oligosaccharides: iterative synthesis of linear oligo- β -(1,3)-glucans and immunostimulatory effects

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Small reducing and linear oligo- β -(1,3)-glucans, which are able to act as phytoallexin elicitors or as immunostimulating agents in anticancer therapy, were synthesized according to an iterative strategy that involved a unique key monosaccharidic donor. To avoid anomeric mixtures, the reducing entity of the target oligomers was first locked with benzyl alcohol and further selective deprotection of the 3-OH with DDQ afforded the desired building block as an acceptor. The latter was then used in a second cycle of glycosylation/deprotection to afford the desired disaccharide, and successive reiterations of this process provided the desired oligomers. Unusual conformational behaviors were observed by standard NMR sequences and supported by NOESY studies. Finally, removal of protecting groups afforded free tri-, tetra-, and pentaglucosides in good overall yields. Two oligosaccharides representing linear laminaritetraose and laminaripentaose were compared to the recently described β -(1,3)-glucan phycarine. Following an intraperitoneal injection, the influx of monocytes and granulocytes into the blood and macrophages into the peritoneal cavity was comparable to that caused by phycarine. Similarly, both oligosaccharides stimulated phagocytic activity of granulocytes and macrophages. Using ELISA, we also demonstrated a significant stimulation of secretion of IL-1B. Together these results suggest that the synthetic oligosaccharides have similar stimulatory effects as natural β -(1,3)-glucans.

Key words: carbohydrates/glycosides/glycosylation/immunostimulation/oligosaccharides

Introduction

 β -(1,3)-Glucans were originally identified as essential constituents of the cell wall in fungi (Kobayashi *et al.*, 1974) or as major storage source in brown seaweed (Black *et al.*, 1951). These renewable polysaccharides are now classified as biological response modifiers. In this context, Albersheim has isolated and characterized a β -glucan fragment able to elicit the biosynthesis of phytoallexins, which are natural antibiotics in plants (Sharp *et al.*, 1984a,b). Then, interest in poly- β -(1,3)-glucosides rapidly reached the human health field since immunostimulating and antitumoral properties were established (Vetvicka and Yvin, 2004; Bohn and BeMiller, 1995). In addition, numerous reports demonstrated that these polymers could also enhance the resistance of hematopoiesis in gamma-irradiated mice (Hofer and Popisil, 1997).

In connection with biological activities of β -(1,3)-glucans, Bohn and BeMiller (1995) published a survey devoted to knowledge accumulated until the mid-1990s about structurefunctional activity relationships. They emphasized the importance of the β -(1,3)-glucan backbone but also many noticeable contradictory data that were previously published on the influence of molecular weight, water solubility, degree of 6-O-substitution by glucopyranosyl units, global chain conformation, and intermolecular associations (Falch et al., 2000) on antitumor activity and on mechanisms involved by these glucans used as biological response modifiers. Nevertheless, it is generally assumed that β -(1,3)-glucans are able to stimulate the nonspecific immune system against bacterial, viral, mycotic, and microparasitic infections as well as against malignant cell growth. On the other hand, current studies describing cellular recognition events finally elucidated the molecular role of these polysaccharides and especially their interactions with two specific binding sites, that is, the lectin domain of the complement receptor type 3 (CR3) (Ross et al., 1999; Xia et al., 1999; Yan et al., 1999) and the dectin-1 molecule, located on macrophages and white blood cells (Brown and Gordon, 2001). All these biological activities explain why purification (Lépagnol-Décamps et al., 1998) and structural elucidation of laminarins from various sources (Dong et al., 2002; Lowman et al., 2003; Read et al., 1996; Schmid et al., 2001), their chemical derivatization to improve their properties (de Nooy et al., 2000), and their enzymatic and/or chemical synthesis (Viladot et al., 1998) are still topics of significance. In addition, the biological variability among individual batches of natural glucans made the possibility of immunologically highly active oligosaccharides extremely important.

Among existing synthetic approaches, a number are connected with the preparation β -(1,6)-branched oligo- β -(1,3)-glucans (Amaya *et al.*, 2001; He *et al.*, 2002; Ning *et al.*, 2002, 2003; Takeo and Tei, 1986; Takahashi *et al.*, 2002; Tanaka *et al.*, 2003; Zeng and Kong, 2003; Zeng *et al.*, 2002; Zhao *et al.*, 2003), but few methods were proposed for the synthesis of linear laminara-oligosaccharides. In fact, only unbranched methyl oligoglucosides were obtained (Takeo *et al.*, 1993). In this context, and considering the

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newly discovered activities of small fragments of laminarins (Miyanishi *et al.*, 2003), we disclose herein with full details of synthesis and characterization an innovative route to linear and reducing oligo- β -(1,3)-glucans and their immunostimulatory effects, which were found to be as good as the natural polymers. Our synthetic methodology is based on an iterative two-step elongation of the glycosidic chain that involves only one monosaccharidic building block. The latter was designed so that it could successively act as a donor and as an acceptor. Reiteration of this glycosylation/deprotection strategy leads to laminaribi-, laminaritri-, laminaritetra-, and laminaripentaose 1–4, respectively (Figure 1). We also described unusual conformational changes for several intermediates observed during this work.

 β -(1,3)-Glucans have been extensively studied for their immunological and pharmacological effects. More than 900 papers describing the biological activities of β -(1,3)-glucans exist. Thus far, strong immunostimulating effects of β -(1,3)-glucans have been demonstrated in all tested animal species. However, just like any other natural product, various glucans cannot be 100% purely prepared, do not have a constant molecular weight, and suffer from significant batch-to-batch differences, all of which can affect some of their clinical effects. It is important, therefore, to evaluate the possibility of using synthetic oligosaccharides based on structure of β -(1,3)-glucans.

Results

The shorter route to laminaribiose 1 lies on direct glycosylation of the well-known diacetone glucose (5) (He et al., 2002; Ning et al., 2002; Zeng and Kong, 2003; Zeng et al., 2002; Zhao et al., 2003). Unfortunately, in our hands, this coupling using a peracylated thioglucosides or trichloroacetimidate gave the desired intermediate but invariably in admixture with a gentiobiose precursor, resulting from acid-catalyzed intramolecular transketalation and further 6-O-glycosylation, even with the more stable dicyclohexylidyl glucose as an acceptor. To avoid this side reaction, we reinvestigated our approach and assumed that the use of a unique intermediate, able to act alternatively as a donor for β -glucosidation and as a precursor of 3-OH acceptor, would gain in efficacy and so is highly desirable. This led us to choose the 2-naphthylmethyl (NAP) group (Borbás et al., 2002; Gaunt et al., 1998; Sarkar et al., 1997, 2000; Xia et al., 2000, 2001; Wright et al., 2001) for an orthogonal protection of the 3-position on the basis of its higher stability under acidic conditions than that of its *p*-methoxybenzyl

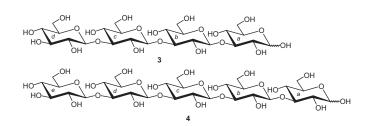
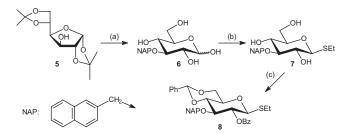


Fig. 1. Chemical structures of linear oligomers of β -(1'3)-glucans.



Conditions: (a) 1. NAPBr, NaH; 2. H₂O, THF, IR 120 (H⁺-form) (80% for 2 steps); (b) 1.

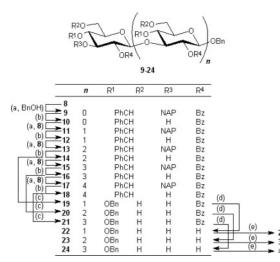
Ac2O, AcONa; 2. EtSH, BF3.OEt2; 3. NaOMe, MeOH (67% for 3 steps); (c) 1.

PhCH(OMe)₂, CSA; 2. BzCl, Pyr (84% for 2 steps).

Scheme I. Synthesis of the key building block 8. Conditions: (a) 1. NAPBr, NaH; 2. H_2O , THF, IR 120 (H⁺-form) (80% for 2 steps); (b) 1. Ac_2O , AcONa; 2. EtSH, BF₃.OEt₂; 3. NaOMe, MeOH (67% for 3 steps); (c) 1. PhCH(OMe)₂, camphorsulfonic acid (CSA); 2. BzCl, Pyr (84% for 2 steps).

counterpart. Therefore, the synthesis began with Williamson etherification of **5** followed by resin-assisted hydrolysis of ketal groups (Scheme I). The resulting product **6** gave the thioglucopyranoside **7** by running through three reactions (acetylation, Ferrier thioglycoside synthesis, and methanolysis) without intermediate chromatographic purification. Subsequent acidic transacetalization of benzaldehyde dimethylacetal followed by 2-*O*-benzoylation afforded the target compound **8**. This key building block was efficiently obtained on a 100-g scale in an average yield near 90% for each of the seven steps.

With this central product in hand, we could explore the elongation of the saccharidic chain. Practically, 8 reacted first with benzyl alcohol under activation by N-iodosuccinimide (NIS) and a catalytic amount of triethylsilyl trifluoromethanesulfonate (TESOTf), and afforded the required β -glucoside 9 in 89% yield (Scheme II). Further deprotection of 9 on oxidative cleavage of the 3-O-NAP group with 2,3-dichloro-5,6-dicyano-1,4-quinone (DDQ) in the presence of methanol (Borbás et al., 2002; Xia et al., 2000, 2001; Wright et al., 2001) gave 10 in 85% yield without alteration of the 4,6-O-benzylidene ring. Subsequently, a second cycle of the iterative process starting from 8 as donor and 10 as the second acceptor provided successively 11 and 12 with a similar efficiency. Precursors of laminaritriose 13 and 14 were further prepared from 8 and the previously obtained disaccharide 12. A fourth and a fifth cycle gave the tetraand pentaglucosides 15, 16 and 17, 18, respectively, in good overall yields. It is interesting to note that glycosylation of the tetrasaccharidic acceptor 16 was significantly improved by substituting TESOTf by tin(II) triflate (Gelin et al., 2000). Finally, to improve these results, we also performed the iterative approach by substituting the naphthylmethyl group by the common allyl one. Although the synthesis of the corresponding key donor was quite efficient, troubles and lost of reproducibility were rapidly observed for both deprotection and glycosylation steps with tri- and higher oligoglucosides (Jamois, 2003). In fact, fine-tuning of reactivity between designed donor 8 and acceptors was best achieved using the NAP protection instead of another one.



Conditions : (a) NIS, TESOTf or Sn(OTf)₂ (9: 89%; 11: 83%; 13: 86%; 15: 88%; 17: 74%); (b) DDQ, CH₂Cl₂, MeOH (10: 85%; 12: 79%; 14: 78%; 16:80%; 18:86%) ; (c) H₂O, MeOH, CSA or PTSA, H₂O (19: 77%; 20: 73%; 21: 76%); (d) MeONa, MeOH (22: 100%; 23: 99%; 24: 99%); (e) H₂, Pd(OAc)₂, EtOH, H₂O (2: 95%; 3: 96%; 4: 94%).

Scheme II. Preparation of protected oligo-β-(1,3)-glucans 9–24 and free oligoglucosides 2–4. Conditions: (a) NIS, TESOTf or Sn(OTf)₂ (9: 89%; 11: 83%; 13: 86%; 15: 88%; 17: 74%); (b) DDQ, CH₂Cl₂, MeOH (10: 85%; 12: 79%; 14: 78%; 16:80%; 18:86%) ; (c) H₂O, MeOH, ČSA or p-Toluenesulfonic acid, H₂O (19: 77%; 20: 73%; 21: 76%); (d) MeONa, MeOH (22: 100%; 23: 99%; 24: 99%); (e) H₂, Pd(OAc)₂, EtOH, H₂O (2: 95%; 3: 96%; 4: 94%).

At this stage of our work, our attention was directed toward the characterization of these new synthons. However, within a molecule, only few structural differences really exist between the glucosyl residues and more particularly within the tetra- and the pentamers. This involved some uncertainty about assignment of nuclear magnetic resonance (NMR) signals. To meet this challenge, application of standard 2D NMR techniques (correlation spectroscopy [COSY] ¹H-¹³C correlation) but also COSY long-range sequences that allowed connection between H-1 and other protons of the same entity were necessary. Moreover, the signal induced by the free 3-OH in **12**, **14**, and **16** was an excellent starting information to elucidate the spectra (Tables I and II).

Table I. Chemical shifts of anomeric centers for compounds 11-18

	δ (ppm)				
Compound	C-1a	C-1b	C-1c	C-1d	C-1e
11	99.6	100.6	_	_	_
12	99.6	100.4			
13	99.5	97.7	98.3		
14	99.5	98.0	98.2		
15	99.5	98.4	96.9	99.0	
16	99.5	98.5	97.0	98.8	
17	99.4	98.1 ^a	97.2 ^a	96.8 ^a	98.6
18	99.1	98.1 ^b	97.2 ^b	96.8 ^b	98.4 ^b

^{a,b}Signals may be interchanged.

As expected, the data first shown that C-1 chemical shifts and ${}^{3}J_{1,2}$ values observed for disaccharides 11 and 12 are relevant for two β -glucosidic linkages ($\delta_{C-1} \sim 99-100$ ppm, ${}^{3}J_{1,2} \sim$ 7.5 Hz). However, from three to five glucopyranosyl residues, we were intrigued by significant modifications: (1) upfield chemical shifts of 1-2 ppm for all C-1 but C-1a, especially for anomeric center of the nonterminal entities; and (2) values of coupling constant of the middle units than that observed for the terminal ones ($\Delta J_{1,2} \sim -2$ to -3 Hz, $\Delta J_{2,3} \sim -1$ to -2 Hz, $\Delta J_{3,4} \sim -0.5$ to -1 Hz). As a consequence, NMR data of compounds 11 and 12 seem to be quite acceptable for a β -D- $Glcp-(1,3)-\beta$ -D-Glcp chain, whereas those compiled for the central core of tri-, tetra-, and pentasaccharides 13-18 could be in accordance with neither β -linkages nor α -connections or orthoesters as potential undesirable products. To explain these results, we first expected that desired β -glucosidic linkages were efficiently formed due to the benzoyl-directed protection and then assumed that the usual ${}^{4}C_{1}$ conformation of middle glucopyranosyl residues in 13-18 undergoes a strong ring distortion. Because NMR data of 3-O-etherified derivatives 13 and 15 present tight similarity with that of compounds 14 and 16 bearing a free 3-OH, this phenomenon cannot be connected with the 3-O-NAP protecting group. We thus hypothesized that important sterical effects are involved by benzylidene rings and that minimization of these contributions requires conformational adjustment for each nonterminal units. For further confirmation, nuclear Overhauser effect (NOE) spectroscopy (NOESY) experiments were carried out on the trisaccharidic derivative 14. Besides anticipated NOE correlations between H-1c and H-3b, irradiation of H-1c and H-1a signals resulted in NOE enhancements of H-5c and H-5a, and to a lower extent, of H-3c and H-3a, respectively. Although irradiating H-1b involved positive NOE effects with H-3a, no other significant correlation could be established, neither with H-3b nor with H-5b. Consequently, these observations are indicative of a longer distance between H-1b and H-5b than that observed for the corresponding protons in a and c units. These results corroborate well the ring distortion observed only for the sugar ring located in the core of the trisaccharide 14 and are relevant to an average conformation on the pseudorotational itinerary between ${}^{4}C_{1}$ to ${}^{1,4}B$ or $B_{2,5}$ through ${}^{4}H_{5}$ and ${}^{1}S_{5}$ (Bentley, 1972).

To demonstrate the impact of acetal protecting groups on this spatial behavior, the benzylidene rings were first removed under acidic conditions and products 14, 16, and 18 gave the corresponding derivatives 19, 20, and 21, respectively. As expected, these latter compounds were characterized by NMR data usually observed for β -glucopyranosides whatever the location of the glucosyl unit along the chain. Successive Zemplén transesterification and hydrogenolysis finally gave the required free tri-, tetra-, and pentasaccharides 2, 3, and 4.

In conclusion, the key monoglucopyranosidic donor **8** was readily obtained in large scale. The efficiency and versatility of the designed iterative process thus demonstrated allowed the synthesis of structurally well-defined short, linear, reducing oligo- β -(1,3)-glucans and so represents an alternative to enzymatic hydrolysis of natural laminarins. More particularly, our study highlighted fine-tuning of reactivity between 3-*O*-NAP glucosyl donor and glucosidic acceptors so that assembly of linear oligoglucans up to the pentasaccharide level was attained. It was also interesting

 Table II. ³J coupling constants for compounds 11–16

	$J(\mathrm{Hz})$															
Compound	1a,2a	1b,2b	1c,2c	1d,2d	2a,3a	2b,3b	2c,3c	2d,3d	3a,4a	3b,4b	3c,4c	3d,4d	4a,5a	4b,5b	4c,5c	4d,5d
11	7.6	7.2			9.0	7.7			9.1	9.2			9.5	9.6		_
12	7.5	7.3			8.2	9.1			9.0	9.8			9.5	9.6		
13	7.9	4.3	7.6		8.9	4.1	8.0		9.3	8.1	9.8		9.6	9.0	9.5	
14	7.9	4.3	7.5		8.8	4.0	8.6		9.2	8.5	9.3		9.7	9.0	9.4	
15	7.8	3.1	5.4	7.4	8.6	ND	5.3	7.9	9.0	ND	8.8	9.2	9.2	ND	8.8	8.7
16	7.9	5.6	5.3	7.6	9.2	5.2	5.1	8.7	9.0	8.5	8.5	9.2	9.2	ND	9.1	9.4

ND: not determined.

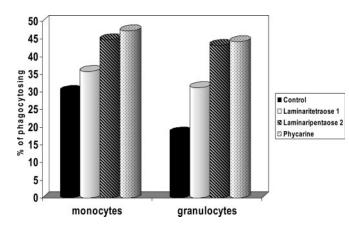


Fig. 2. Potentiation of phagocytosis of synthetic microspheres (HEMA particles) by IP-injected phycarine or two oligosaccharides. Monocytes and granulocytes with three or more HEMA particles were considered positive.

to note that the protected intermediates underwent unexpected conformational changes for each nonterminal residues lying on steric hindrance involved by benzylidene rings. This effect had however no consequence on further glycosylation steps.

The effects of various glucans on macrophages are well established. However, to demonstrate that synthetic oligosaccharides really exhibit an immunomodulatory characteristics, an evaluation of phagocytosis is necessary. We measured the effects of different doses of phycarine on phagocytosis of synthetic 2-hydroxyethyl methacrylate (HEMA) microspheres in peripheral blood (Figure 2) and peritoneal cavity (Figure 3). The internalization of synthetic particles was notably more comparable to phycarine application, and at higher doses (250 μ g/mouse) caused significant elevation even 4 days after injection (data not shown). We also measured the effects of administration of these two oligosaccharides on cellularity in both peripheral blood and peritoneum. Data summarized in Figures 4 and 5 show a significant increase in the number of neutrophils and peritoneal macrophages.

We also examined the effect of a single intraperitoneal injection of oligosaccharides on systemic in vivo release of IL-1 β . Peripheral blood was isolated at three different intervals after the injection and the obtained serum was stored at -80°C for no more than 1 week. The data summarized in Figure 6 show significant elevation in levels of IL-1 β at

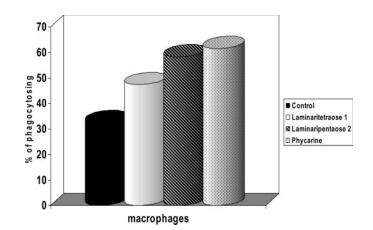


Fig. 3. Potentiation of phagocytosis of synthetic microspheres (HEMA particles) by IP-injected phycarine or two oligosaccharides. Peritoneal macrophages with three or more HEMA particles were considered positive.

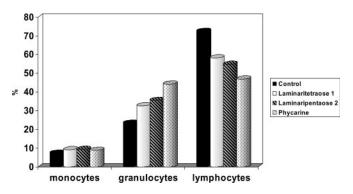


Fig. 4. Effect of IP injecti on of $100 \,\mu g$ phycarine or oligosaccharides on differential counts in peripheral blood.

every tested interval, the level of secretion was highest in case of phycarine.

In the final phase of the series of experiments, mice challenged withPtas64 mammary tumors were tested for a therapeutic response to daily intraperitoneal injections of either phycarine or individual oligosaccharides (Figure 7). This experiment was repeated three times with similar results. To evaluate the possible effects of lipopolysaccharide contamination, we also used lipopolysaccharide-free

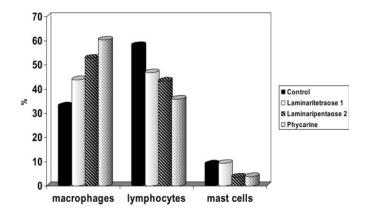


Fig. 5. Effect of IP injection of $100 \,\mu g$ phycarine or oligosaccharides on differential counts in peritoneal cavity.

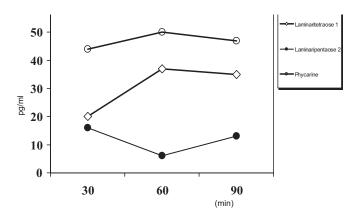


Fig. 6. Effect of IP injection of $100 \ \mu g$ phycarine or oligosaccharides on levels of IL-1 β in peripheral blood.

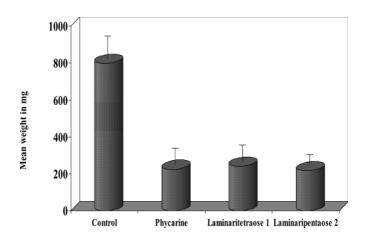


Fig. 7. Phycarine or oligosaccharide therapy of Balb/c mice with Ptas64 mammary carcinoma. Data from three experiments are shown. After 2 weeks of therapy, and for each experiment, four groups of mice were tested for a response for phycarine or oligosaccharides by weighing the tumors. Additionally, individual groups were given daily IP injections of 100 μ g of tested substance, respectively. The control group of mice received daily IP PBS. Each value represents the mean \pm SD.

phycarine with identical results (data not shown). This data showed the strong inhibition of tumor growth by both phycarine and oligosaccharides.

Discussion

Branched and linear β -(1,3)-glucans originated from numerous sources, such as mushrooms, yeast, or seaweed and are well known biological response modifiers. However, their rate of success is often varying and unpredictable (Kimura et al., 1994; Tanabe et al., 1990). This can be at least partly ascribed to the differences between individual batches and impurities due to the imperfect isolation processes. The possibility of using higly active biological synthetic oligosaccharides, based on know glucan structure and compared to some kind of benchmark biological activities, is therefore important. The pharmacokinetic parameters of β -(1,3)-glucans are extremely important for calculating pharmaceutical dose levels. Although large glucans appear to be better because of their resistance to glomerular filtration and saturation of liver clearance, glucans that are too large have a higher potential for producing undesirable side effects.

Without any doubts, some commercially available oligosaccharides corresponding to our oligosaccharides exist on the market. However, our evaluation of the commercial preparations revealed presence of significant levels of impurities (e.g., hexasaccharides in samples of pentasaccharides or presence of laminaripentaoses in samples of laminarintetraoses), mostly due to the use of enzymatic hydrolysis of higher polymers. This mixture of oligosaccharides is virtually impossible to separate completely. On the other hand, no traces of other oligosaccharides were detected in our samples when tested by mass spectroscopy. Therefore only our synthetic oligosaccharides have been used throughout the present study.

For comparison of immunostimulating properties, we used phycarine, a recently described seaweed-derived glucan (Vetvicka and Yvin, 2004). Structural analysis showed that phycarine (laminarin) is an essentially linear glucan composed of ~33 glucopyranose units joined by acetalic β -(1,3) linkages with molecular weight of 5000 Da with purity higher than 97%.

Glucans are well known to stimulate phagocytosis (Abel *et al.*, 1989; Yan *et al.*, 2003). Therefore, one of the first tests of the immunological characteristics is the phagocytosis. We used synthetic microspheres based on HEMA. These HEMA particles, in contrast to yeast particles, have a slight negative charge and therefore do not specifically adhere to the cell surface, which guarantees that only actively phagocytosing cells will internalize these inert particles (for review see Vetvicka and Formusek, 1987). In both peritoneal macrophages and peripheral blood cells, we found significant stimulation of phagocytosis at the level comparable with phycarine. In addition, the influx of macrophages into the peritoneal cavity lasted at least 3 days (data not shown), which further shows the activation of macrophages by these oligosaccharides.

It is hypothesized that the immunostimulating actions of β -glucans are, at least in part, caused by potentiation of a synthesis and release of several cytokines such as TNF α , IFN γ , and IL-1 β (Ohmura *et al.*, 2001; Seljelid *et al.*, 1989). This cytokine-stimulating activity is thought to be dependent

on the triple helix conformation (Falch et al., 2000). Most glucans, including oligomer prepared from laminarin (Miyanishi et al., 2003), have been shown to stimulate TNF α both in vivo and in vitro (Ross *et al.*, 1999). The stimulation of TNF α was often found to be secondary effect of a massive release of IL-1 (Ohmura et al., 2001). Therefore we compared the levels of IL-1 after injection of these synthetic oligosaccharides and phycarine. The data obtained not only showed similar stimulation but also correlated with findings of effects of lentinan in treatment of human cancer (Ross et al., 1999) and stimulation caused by paramylon (Kondo et al., 1992). Our data suggest that the triple helix conformation generally observed for glucans with degree of polymerization near 30 is not necessary for tetra- and penta- β -(1,3)-glucans to obtain immunostimulating effects. The biological effect of these oligosaccharides was further evaluated by measuring the inhibition of breast cancer growth as determined by the weight of the tumor. This was done after 2 weeks of therapy with either oligosaccharides or phycarine. In every case there was a significant inhibition of cancer growth.

Only very limited knowledge of glucan-based oligosaccharides and their immunological properties exist. Thus far, only laminarin-based oligomer (Miyanishi *et al.*, 2003) and β - and α -(1,3)-glucohexaoses were described as having immunostimulating characteristics comparable to parental β -glucan. The current spike of interest in commercial use of β -glucans clearly shifts the attention on the prospect of 99% pure synthetic oligomers without any contaminating biological impurities.

Material and methods

General methods

All reactions were performed under nitrogen. Thin-layer chromatography (TLC) analyses were carried out on precoated nonactivated plates (Merck 60 F_{254} ; Darmstadt, Germany) with detection by UV absorption (254 nm) when applicable and charring with 5% H_2SO_4 in EtOH. For column chromatography, Merck 60H (5–40 µm) silica gel was used. Optical rotations were determined with a Perkin-Elmer 341 polarimeter at 20°C using a 1-dm cell. ¹H, ¹³C, NMR spectra, and 2D COSY, heteronuclear multiple band correlation and NOESY experiments were recorded on a Brüker ARX 400 spectrometer; chemical shifts (δ) are given in ppm. Microanalyses and recording of mass spectra were performed by the Centre Régional de Mesures Physiques de l'Ouest (CRMPO, University of Rennes 1, France).

Compound 6

For compound **6** (3-*O*-NAP-D-glucopyranose), a 60% suspension of NaH in oil (21.7 g, 543 mmol) was added at 0°C to a solution of **5** (117.7 g, 452 mmol) and NAP bromide (100 g, 452 mmol) in dry N,N-Dimethylformamide. The mixture was allowed to reach room temperature and stirred for 3 h. The excess of NaH was then neutralized with MeOH and the product crystallized out by adding iced water (2 L) under vigorous agitation. The supernatant was removed and the solid taken up into CH_2Cl_2 . After decanting, the organic layer was dried (MgSO₄) and concentrated.

The product thus obtained was dissolved in acetone (270 ml) and water (270 L) and resin IR 120 (H⁺ form) (420 g) were successively added. The reaction media was heated at 60°C for 2 days. The resin was filtered off and washed with MeOH, and the filtrate was neutralized by adding few drops of an aqueous solution of sodium bicarbonate. After removal of the solvents, the resulting solid was washed with toluene to give **2** (116 g, 80%) : TLC (CH₂Cl₂/MeOH, 9:1): $R_{\rm f} = 0.5$; ¹H-NMR (CD₃OD, 400 MHz): δ 7.78–7.31 (m, NAP), 4.97 [d, ²J 11.3 Hz, 1 H, CH₂(NAP)β], 4.93 [d, 1 H, CH₂(NAP)β] and Table III; ¹³C-NMR (CD₃OD, 100 MHz): δ 138.0, 134.8, 134.4 [C-*q* (NAP)], 128.9, 128.8, 128.6, 127.5, 127.4, 127.3, 126.9, 126.8 (NAP), 76.3 [CH₂(NAP) α^a], 73.0 [CH₂(NAP) β] and Table IV; (a): value can be inverted with those listed in Table IV; electrospray ionization mass spectrometry (ESI-MS): ([M+Na]⁺, C₁₇H₂₀O₆Na): *m*/*z* calculatedd 343.1158, found 343.1159.

Compound 7

For compound 7 (Ethyl 3-O-NAP-1-thio- β -D-glucopyranoside), compound 6 (116 g, 362 mmol), sodium acetate (59.4 g, 724 mmol) and acetic anhydride (680 ml, 7.24 mol) were successively introduced in a flask. The mixture was refluxed for 2 h and then poured onto water (5 L). After stirring overnight, the solid was filtered off, washed with a 5% aqueous solution of sodium bicarbonate, filtered off again, washed with water, and dried (MgSO₄).

To the resulting product dissolved in CH_2Cl_2 (885 ml) were successively added at 0°C ethanethiol (20.5 ml, 399 mmol)

Table III. ¹H-NMR chemical shifts and ³J coupling constants for monosaccharidic compounds 6–10

	δ (ppm) (<i>J</i> , Hz)					
Compound	H-1 $(J_{1,2})$	H-2 $(J_{2,3})$	H-3 (J _{3,4})	H-4 $(J_{4,5})$	H-5 (J _{5,6})	H-6 $(J_{6,6'})$	H-6' $(J_{5,6'})$
6β	4.42 (7.8)	3.18-3.24 (8.8)	3.32 (9.0)	3.38 (9.1)	3.18-3.24 (2.4)	3.77 (11.9)	3.57 (5.9)
6α	5.02 (3.6)	ND	ND	ND	ND	ND	ND
7	4.36 (9.6)	3.53 (8.8)	3.46 (8.9)	3.62 (9.2)	3.37 (3.4)	3.87 (12.0)	3.75 (4.9)
8	4.59 (10.0)	5.36 (8.9)	3.93 (9.0)	3.88 (9.2)	3.56 (5.0)	4.41 (10.4)	3.84 (10.1)
9	4.58 (7.8)	5.40 (8.4)	3.84 (9.1)	3.90 (9.2)	3.45 (5.0)	4.40 (10.4)	3.86 (10.0)
10	4.70 (7.9)	5.26 (9.2)	3.99 (9.3)	3.68 (9.6)	3.49 (5.1)	4.42 (10.4)	3.85 (10.0)

ND: not determined.

Table IV. ¹³C-NMR chemical shifts for monosaccharidic compounds 6–10

	δ (ppm)	1				
Compound	C-1	C-2	C-3	C-4	C-5	C-6
6β	98.3	76.5 ^a	86.4	71.6	78.0 ^a	62.8
6α	94.2	71.7 ^b	83.6	73.1 ^b	74.0 ^b	62.7
7	86.7	73.2	85.0	70.1	79.5	62.7
8	84.4	71.9	79.0	81.8	70.8	68.7
9	100.0	73.3	72.6	81.8	66.3	68.8
10	99.8	74.8	72.4	80.9	66.3	68.7

^{a,b}Values can be inverted among themselves or with CH₂(NAP).

and boron trifluoride etherate (50.1 ml, 3.99 mmol). After 2 h stirring at 0°C, the organic solution was washed with a 5% aqueous solution of sodium bicarbonate abd water, then dried (MgSO₄) and concentrated.

To a methanolic solution of sodium methanolate, prepared from MeOH (200 ml) and sodium (703 mg, 306 mmol), was added a solution of the previously obtained intermediate in toluene (100 ml). After 4 h at room temperature, the reaction was quenched with IR 120 (H⁺ form) resin, filtered, and concentrated. The resulting crude oil was then poured onto heptane (1 L). After filtration, the resulting solid was taken up with CH_2Cl_2 (600 ml), and the by-products were removed by washings with water (600 ml) at 40°C. The organic layer was recovered, dried, and evaporated to give 7 (88.3 g, 67% overall yield): TLC (CH₂Cl₂/MeOH, 9:1): $R_{\rm f}$ = 0.5. mp: 78–80°C; ¹H-NMR (CDCl₃, 400 MHz): δ 7.86– 7.82 (m, 4 H, NAP), 7.51–7.47 (m, 3 H, NAP), 5.17 [d, 1 H, ²J 11.8 Hz, CH₂(NAP)], 4.94 [d, 1 H, CH₂(NAP)], 2.73 (qd, 2 H, ²J 7.4 Hz, ³J 7.5 Hz, SCH₂CH₃), 2.68 (sl, 1 H, OH), 2.56 (s, 1 H, OH), 2.22 (sl, 1 H, OH), 1.33 (t, 3 H, SCH₂CH₃) and Table III; 13 C-NMR (CDCl₃, 100 MHz): δ 135.8, 133.4, 133.1 [C-q (NAP)], 128.6, 128.0, 127.8, 127.0, 126.3, 126.2, 125.9 (NAP), 74.9 [CH₂(NAP)], 24.7 (SCH₂CH₃), 15.5 (SCH₂CH₃) and Table IV; ESI-MS: $([M+Na]^+, C_{19}H_{24}O_5SNa): m/z$ calculated 387.1242, found 387.1244; ([M+K]⁺, C₁₉H₂₄O₅SK): *m/z* calculated 403.0982, found 403.0997.

Compound 8

To a solution of compound 7 (50 g, 137 mmol) in acetonitrile (300 ml) were successively added benzaldehyde dimethylacetal (31 ml, 206 mmol) and anhydrous camphorsulfonic acid (6.4 g, 27 mmol). The mixture was heated at 55°C for 2 h, cooled to room temperature, and neutralized with triethylamine. After concentration, the residue was dissolved in CH_2Cl_2 and poured onto MeOH (1 L). After 1 night at 4°C, the intermediate was filtered off and washed with cold MeOH.

The crude product was further submitted to benzoylation under standard conditions using pyridine (300 ml) and benzoyl chloride (37 ml, 321 mmol). After stirring overnight at room temperature, the reaction media was poured onto MeOH (1 L). The resulting solid was filtered off, washed with MeOH, and dried to afford **8** (ethyl 2-*O*-benzoyl-4, 6-*O*-benzylidene-3-*O*-NAP-1-thio- β -D-glucopyranoside) (50 g,

84% overall yield): TLC (toluene/EtOAc, 9:1): $R_f = 0.6$; mp: 137–139°C; $[\alpha]^{20'}_{D}$ + 17 (*c* 1.0; CH₂Cl₂); ¹H-NMR (CDCl₃, 400 MHz): δ 7.95–7.92 (m, 2 H, H arom.), 7.69–7.36 (m, 12 H, H arom.), 7.22 (dd, J 1.4 Hz, J 8.4 Hz, 1 H, H arom.), 5.64 (s, 1 H, CHPh), 4.98 [d, ²J 12.2 Hz, 1 H, CH₂(NAP)], 4.86 (d, 1 H, CH₂(NAP)], 2.77–2.64 (m, 2 H, SCH₂CH₃), 1.21 (t, J7.4 Hz, $\overline{3}$ H, SCH₂CH₃) and Table III; ^{13}C -NMR $(CDCl_3, 100 \text{ MHz}): \delta 165.3 \text{ (C = O)}, 137.3, 135.3, 133.3,$ 133.1, 133.0 (C-q arom.), 130.0, 129.2, 128.4, 128.1, 127.9, 127.7, 127.1, 126.3, 126.1, 126.0, 125.8 (C arom.), 101.4 (CHPh), 74.3 [CH₂(NAP)], 24.1 $(SCH_2CH_3),$ 14.9(SCH₂CH₃), and Table IV; ESI-MS: $([M+Na]^+,$ C₃₃H₃₂O₆SNa): *m*/*z* calculated 579.1817, found 579.1820; $([M+K]^+, C_{33}H_{32}O_6SK)$: *m/z* calculated 595.1557, found 595.1559; ($[M+Na+CH_3OH]^+$, $C_{34}H_{36}O_7SNa$): *m/z* calculated 611.2080, found 611.2066. Analytical calculated for C₃₃H₃₂O₆S (556.18): C, 71.20; H, 5.79; S, 5.76; found: C, 71.70; H, 5.74; S, 6.12.

Elongation chain: general procedure

Glycosidic coupling. NIS (1.1 or 1.2 equiv.) and a Lewis acid [0.1 eq, either TMSOTf, TESOTf, or $Sn(OTf)_2$] were successively added to a solution of donor **8** (1.1 equiv.) and glycosyl acceptor (1 equiv.) in the presence of 4-Å molecular sieves under nitrogen. The mixture was stirred at 0°C and, when TLC monitoring indicated completion of the reaction, quenched by addition of triethylamine. After filtration through a bed of celite, the solution was diluted with CH_2Cl_2 , washed with a 10% aqueous solution of sodium thiosulfate and with water, dried (MgSO₄), and concentrated. The expected product was finally purified by flash chromatography.

Deprotection of NAP group. After dissolution of the resulting coupling compound (1 equiv.) in $CH_2Cl_2/MeOH$ (4:1, v/v), DDQ (3 equiv.) was added and the mixture then stirred at room temperature. After completion of the reaction, the media was diluted with CH_2Cl_2 , washed with a 5% aqueous solution of sodium bicarbonate, and washed again with water. The organic layer was dried (MgSO₄) and concentrated, and the pure target product was obtained after purification over silica gel.

Compound 9

This compound was obtained as described in procedure A, starting from 8 (312 mg, 0.56 mmol) in CH_2Cl_2 (2 ml) containing molecular sieves (100 mg), using benzyl alcohol (69.6 µl, 0.67 mmol) as an acceptor and NIS (139 mg, 0.62 mmol) and TESOTf (12.7 µl, 0.06 mmol) as a promotor. After 1 h stirring, work-up and chromatography (toluene/ EtOAc, 17:3) enable collecting 9 (benzyl 2-O-benzoyl-4,6-*O*-benzylidene-3-*O*-NAP- β -D-glucopyranoside) (300 mg, 89%): TLC (toluene/EtOAc, 9:1): $R_f = 0.6$; mp: 142–144°C; $[\alpha]^{20'}$ $v_{\rm D} - 6 (c 1.0, \text{CH}_2\text{Cl}_2); {}^1\text{H-NMR} (\text{CDCl}_3, 400 \text{ MHz}): \delta\delta$ 7.89–7.86 (m, 2 H, H arom.), 7.67–7.07 (m, 17 H, H arom.), 5.62 (s, 1 H, CHPh), 4.95 [d, ²J 12.4 Hz, 1 H, CH₂(NAP)], 4.84 [d, 2 H, CH₂Ph, CH₂(NAP)], 4.58 (d, ²J 13.6 Hz, 1 H, CH₂Ph) and Table III; ${}^{13}C$ -NMR (CDCl₃, 100 MHz): δ 165.1 (C = O), 137.3, 136.8, 135.3, 133.1, 133.0, 132.9 (C-q arom.), 129.9, 129.8, 129.1, 128.4, 128.3, 128.2, 128.1, 127.9, 127.8, 127.7, 127.6, 127.0, 126.2, 126.1, 125.9, 125.8 (C arom.), 101.4 (CHPh), 74.0 [CH₂(NAP)], 70.5 (CH₂Ph) and Table IV; ESI-MS: ([M+Na]⁺, $C_{38}H_{34}O_7Na$): *m/z* calculated 625.2202, found 625.2194; ([M+K]⁺, $C_{38}H_{34}O_7K$): *m/z* calculated 641.1942, found 641.1949; ([M+Na+CH₃OH]⁺, $C_{39}H_{38}O_8Na$): *m/z* calculated 657.2464, found 657.2438.

Compound 10

For compound 10 (benzyl 2-O-benzoyl-4,6-O-benzylidene- β -D-glucopyranoside), the deprotection reaction was performed according to procedure B starting from 8 (420 mg, 0.70 mmol) in solvent (8.5 ml) and using DDQ (475 mg, 2.09 mmol). After 5.5 h stirring, work-up and chromatography (toluene/EtOAc, 9:1) gave 9 (274 mg, 85%): TLC (toluene/EtOAc, 9:1): $R_{\rm f} = 0.3$; mp: 147–149°C; $[\alpha]^2$ ′_D – 84 (c 1.0, CH₂Cl₂); ¹H-NMR (CDCl₃, 400 MHz): δ 8.03–8.00 (m, 2 H, H arom.), 7.62–7.58 (m, 1 H, H arom.), 7.52–7.43 (m, 4 H, H arom.), 7.41–7.36 (m, 3 H, H arom.), 7.23–7.19 (m, 5 H, H arom.), 5.57 (s, 1 H, CHPh), 4.89 (d, ²J 12.5 Hz, 1 H, CH₂Ph), 4.66 (d, 1 H, CH₂Ph), 2.84 (s, 1 H, OH-3) and Table III; ¹³C-NMR (CDCl₃, 100 MHz): δ 166.0 (C = O), 137.0, 136.7, 133.4 (C-q arom.), 130.1, 129.6, 129.4, 128.4, 127.9, 127.8, 126.4 (C arom.), 102.0 (CHPh), 70.6 (CH₂Ph) and Table IV; ESI-MS: $([M+Na]^+, C_{27}H_{26}O_7Na)$: *m/z* calculated 485.1576, found 485.1574; $([M+K]^+, C_{27}H_{26}O_7K)$: m/z calculated 501.1316, found 501.1312; $([M+Na+CH_3OH]^+, C_{28}H_{30}O_8Na)$: *m/z* calculated 517.1838, found 517.1832.

Compound 11

Disaccharide 11 (benzyl 2-O-benzoyl-4,6-O-benzylidene-3-O-NAP-β-D-glucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylidene- β -D-glucopyranoside) was synthesized according to procedure A, using 8 (802 mg, 1.44 mmol) and 10 (606 mg, 1.31 mmol) in CH₂Cl₂ (7.0 ml) containing molecular sieves (200 mg), NIS (324 mg, 1.44 mmol), and TESOTf (29.6 µl, 0.13 mmol). After 50 min at room temperature, work-up and chromatography (toluene/EtOAc, 19:1) afforded 11 (1.036 g, 83%): TLC (toluene/EtOAc, 4:1): $R_f = 0.6$; mp: 202–204°C; $[α]^{20}_{D}$ – 1 (*c* 1.0, CH₂Cl₂); ¹H-NMR (CDCl₃, 400 MHz): δ 7.74–7.02 (m, 32 H, H arom.), 5.53 (s, 1 H, CHPh), 5.34 (s, 1 H, CHPh), 4.84 [d, ²J 12.4 Hz, 1 H, CH₂(NAP)], 4.74 (d, 1 H, CH₂(NAP)], 4.74 (d, ²J 12.2 Hz, 1 H, CH_2Ph), 4.50 (d, 1 H, CH_2Ph), and Table V; ¹³C-NMR (CDCl₃, 100 MHz): δ 164.7, 164.6 (C = O), 137.3, 137.2, 136.7, 135.3, 133.0, 132.9, 132.8, 132.7 (C-q arom.), 129.8–125.7 (C arom.), 101.5, 101.1 (CHPh), 73.5^a [CH₂(NAP)], 70.3 (CH₂Ph) and Table VI; (a): value can be inverted with those listed in Table VI. Analyitcal calculated for C₂₈H₃₈O₁₉: C, 72.79; H, 5.48; found: C, 73.12; H, 5.48.

Table V. ¹H-NMR chemical shifts and ³J coupling constants for disaccharides 11 and 12; trisaccharides 13, 14, and 19; and tetrasaccharidic derivatives 15 and 16

		δ (ppm) (³ J	, Hz)					
Product		H-1 $(J_{1,2})$	H-2 $(J_{2,3})$	H-3 (J _{3,4})	H-4 $(J_{4,5})$	H-5 (J _{5,6})	H-6 (J _{6,6')}	H-6' (J _{5,6')}
11	Unit a	4.56 (7.6)	5.31 (9.0)	4.11 (9.1)	3.84 (9.5)	3.49 (4.9)	4.36 (10.4)	3.80 (10.0)
	Unit b	4.85 (7.2)	5.28 (7.7)	3.74 (9.2)	3.90 (9.6)	3.37 (4.9)	4.16 (10.4)	3.68 (10.0)
12	Unit a	4.52 (7.5)	5.25 (8.2)	4.08 (9.0)	3.77 (9.5)	3.44 (4.8)	4.30 (10.4)	3.73 (10.0)
	Unit b	4.83 (7.3)	5.03 (9.1)	3.76 (9.8)	3.57 (9.6)	3.28 (4.9)	4.10 (10.4)	3.60 (10.0)
13	Unit a	4.45 (7.9)	4.86 (8.9)	4.07 (9.3)	3.18 (9.6)	3.37 (4.9)	4.31 (10.3)	3.64 (10.0)
	Unit b	4.86 (4.3)	5.08 (4.1)	3.97 (8.1)	4.07 (9.0)	3.55 (3.9)	4.11 (9.1)	3.52 (9.7)
	Unit c	5.06 (7.6)	5.34 (8.0)	3.85 (9.8)	3.91 (9.5)	3.50 (4.4)	4.22 (10.4)	3.72 (10.1)
14	Unit a	4.47 (7.9)	4.95 (8.8)	4.09 (9.2)	3.21 (9.7)	3.38 (4.9)	4.31 (10.5)	3.65 (10.3)
	Unit b	4.90 (4.3)	5.13 (4.0)	4.01 (8.5)	4.07 (9.0)	3.56 (3.7)	4.12 (10.4)	3.47 (9.7)
	Unit c	5.11 (7.5)	5.20 (8.6)	3.96 (9.3)	3.67 (9.4)	3.47 (4.8)	4.22 (10.3)	3.69 (10.2)
19	Unit a	4.41 (7.9)	4.85 (9.4)	3.75-4.86 (ND)	3.34-3.42 (ND)	3.34-3.42 (ND)	3.75-4.86 (ND)	3.53-3.64 (ND)
	Unit b	4.61 (8.0)	4.82 (9.5)	3.75-4.86 (ND)	3.34-3.42 (9.8)	3.27 (2.1)	3.75-4.86 (ND)	3.53-3.64 (5.7)
	Unit c	4.52 (8.1)	4.75 (9.0)	3.15-3.22 (ND)	3.15-3.22 (ND)	3.15-3.22 (ND)	3.75-4.86 (ND)	3.53-3.64 (ND)
15	Unit a	4.47 (7.8)	4.98 (8.6)	4.04 (9.0)	3.35 (9.2)	3.37-3.59 (4.3)	4.34 ^a (10.3)	3.68–3.75 ^b (ND)
	Unit b	4.75 (3.1)	4.75–4.77 (ND)	3.89-3.92 (ND)	3.37-3.59 (ND)	3.37-3.59 (ND)	4.09–4.12 ^a (ND)	3.68–3.75 ^b (ND)
	Unit c	4.95 (5.4)	5.12 (5.3)	4.00-4.03 (8.8)	3.95 (8.8)	3.37-3.59 (ND)	4.09–4.12 ^a (ND)	3.37–3.59 ^b (ND)
	Unit d	4.98 (7.4)	5.33 (7.9)	3.82 (9.2)	3.87 (8.7)	3.37-3.59 (4.9)	4.19 ^a (10.4)	3.37–3.59 ^b (ND)
16	Unit a	4.47 (7.9)	4.99 (9.2)	4.04 (9.0)	3.35 (9.2)	3.37-3.47 (4.6)	4.34 ^c (10.4)	3.72 ^d (9.9)
	Unit b	4.76 (5.6)	5.15 (5.2)	3.91 (8.5)	3.37-3.47 (ND)	3.37-3.47 (ND)	4.09–4.12 ^c (ND)	3.67 ^d (10.1)
	Unit c	4.97 (5.3)	4.82 (5.1)	4.05 (8.5)	3.95 (9.1)	3.54-3.58 (ND)	4.09–4.12 ^c (ND)	3.37–3.47 ^d (ND)
	Unit d	5.02 (7.6)	5.16 (8.7)	3.89 (9.2)	3.60 (9.4)	3.37-3.47 (4.9)	4.18 ^c (10.4)	3.51 ^d (10.0)

^{a,b,c,d}Signals may be interchanged.

ND: not determined.

14, and 19; and tetrasaccharidic derivatives 15 and 16
13,
trisaccharrides
12
C-NMR chemical shifts for disaccharides 11 and 12;
Table VI. ¹³ C-

	δ (ppm)	()																						
	Unit a						Unit b						Unit c						Unit d					
Product	C-1		C-3	C-4	C-2 C-3 C-4 C-5 C-6	C-6	C-1	C-2	C-3	C-4	C-2 C-3 C-4 C-5	C-6	C-1	C-2	C-3	C-4	C-1 C-2 C-3 C-4 C-5 C-6	C-6		C-1 C-2 C-3 C-4 C-5	C-3	C-4	C-5	C-6
=	9.66	73.5 ^a	77.8	79.3	66.5	68.7 ^b	100.6	73.7	78.2	80.9	66.0	68.8 ^b			I					I	I	I	I	I
12	9.66	73.5	78.0	79.3	66.5	68.8	100.4	75.3	72.6	80.5	66.1	68.7												
13	99.5	74.3	74.1	78.7	66.5	68.7	97.9	72.6	76.2	77.6	65.4	68.8	98.3	73.3	78.2	81.4	66.1	68.8						
14	99.5	74.3	74.4	78.7	66.5	68.7	98.0	72.7	76.3	77.6	65.4	68.8	98.2	74.7	72.5	80.9	66.1	68.7						
19	100.9	74.7	82.7	9.9 ^d	77.8 ^e	62.4 ^f	102.3	74.4	83.9	70.1 ^d	77.9 ^e	62.4 ^f	101.9	75.4	76.1	71.5	78.3	62.2 ^f						
15	99.5	74.0	75.1	78.8	65.5 ^a	68.7 ^b	98.4	73.5	74.2	77.4	65.5 ^a	68.9 ^b	9.66	72.5	76.9	78.2 ^c	66.1 ^a	68.7 ^b	0.06	73.4	78.3 ^c	81.3	66.4 ^a	68.7 ^b
16	99.5	73.9	75.1	78.8	65.5 ^d	68.6	98.5	73.5	74.3	77.4	65.5 ^d	68.6	97.0	72.5	76.8	78.2	66.0 ^d	68.6	98.8	74.8	72.4	80.7	66.4 ^d	68.6

^{a,b,c,d,e,f}Signals may be interchanged

Compound 12

According to procedure B, compound **11** (217 mg, 0.23 mmol) and DDQ (154 mg, 0.68 mmol) were stirred for 4.5 h in solvent (4.3 ml) and further work-up and chromatography (toluene/ EtOAc, 9:1) provided 12 (benzyl 2-O-benzoyl-4,6-Obenzylidene-β-D-glucopyranosyl-(1,3)-2-O-benzoyl-4,6-Obenzylidene-β-D-glucopyranoside) (146 mg, 79%): TLC (toluene/EtOAc, 4:1): $R_{\rm f} = 0.4$; mp: 219–221°C; $[\alpha]^{20}_{\rm D} - 46$ (c 1.0, CH₂Cl₂); ¹H-NMR (CDCl₃, 400 MHz): δ 7.70–6.96 (m, 25 H, H arom.), 5.47 (s, 1 H, CHPh), 5.27 (s, 1 H, CHPh), 4.69 (d, ²J 12.5 Hz, 1 H, CH₂Ph), 4.44 (d, 1 H, CH₂Ph), 2.55 (s, 1 H, OH-3b) and Table V; 13 C-NMR $(CDCl_3, 100 \text{ MHz})$: δ 165.6, 164.7 (C = O), 137.2, 137.0, 136.6, 133.1, 133.0 (C-q arom.), 129.8–126.2 (C arom.), 101.7 (CHPh), 101.5 (CHPh), 70.3 (CH₂Ph) and Table VI. Analytical calculatd for C47H44O13: C, 69.11, H, 5.43; found: C, 68.73, H, 5.51.

Compound 13

Compound 13 (benzyl 2-O-benzoyl-4,6-O-benzylidene-3-O-NAP-β-D-glucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylideneβ-D-glucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylidene-β-D-glucopyranoside) was prepared as described in procedure A, starting from 8 (503 mg, 0.90 mmol) in CH₂Cl₂ (10 ml), containing molecular sieves (1.0 g), and using disaccharide 12 (670 mg, 0.82 mmol), NIS (203 mg, 0.90 mmol), and TESOTf (19 µl, 0.09 mmol). After 50 min stirring, work-up and chromatography (toluene/EtOAc, 19:1.9:1) enabled collecting **13** (928 mg, 86%); TLC (toluene/EtOAc, 17:3): $R_{\rm f} = 0.5$; mp: 118–120°C; $[\alpha]^{20}_{\rm D} + 12$ (*c* 1.0, CH₂Cl₂); ¹H-NMR (CDCl₃, 400 MHz): δ 7.90–7.05 (m, 42 H, H arom.), 5.47 (s, 1 H, CHPh), 5.46 (s, 1 H, CHPh), 4.91 [d, ²J 12.3 Hz, 1 H, CH₂(NAP)], 4.80 (d, 1 H, CH₂(NAP)], 4.75 (d, ²J 12.5 Hz, 1 H, CH₂Ph), 4.57 (s, 1 H, CHPh), 4.49 (d, ²J 12.6 Hz, 1 H, CH₂Ph) and Table V; 13 C-NMR (CDCl₃, 100 MHz): δ 165.1, 164.8, 164.5 (C = O), 137.4, 137.4, 137.2, 136.8, 135.4, 133.4, 133.2, 133,1, 132,9 (C-q. arom.), 129.9–125.4 (C arom.), 101.9 (CHPh), 101.2 (CHPh), 100.6 (CHPh), 73.87 [CH₂(NAP)], 70.22 (CH₂Ph) and Table VI; ESI-MS: $([M+Na]^+, C_{53}H_{70}O_{19}Na): m/z$ calculated 1333.4409, found 1333.4407.

Compound 14

For compound 14 (benzyl 2-O-benzoyl-4,6-O-benzylideneβ-D-glucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylideneβ-D-glucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylidene- β -D-glucopyranoside), the deprotection reaction was performed according to procedure B starting from 13 (621 mg, 4.7 mmol) in solvent (12.5 ml) and using DDQ (323 mg, 1.42 mmol). After stirring for 7 h, work-up and chromatography (toluene/EtOAc, 9:1'17:3) gave 13 (889 mg, 78%): TLC (toluene/EtOAc, 17:3): $R_f = 0.3$; mp: 138–140°C; $[\alpha]^{20}$ + 12 (c 0.75, CH₂Cl₂); ¹H-NMR (CDCl₂, 400 MHz): δ 8.02–7.06 (m, 35 H, H arom.), 5.47 (s, 1 H, CHPh), 5.42 (s, 1 H, CHPh), 4.76 (d, ²J 12.6 Hz, 1 H, CH₂Ph), 4.62 (s, 1 H, CHPh), 4.32 (d, 1 H, CH₂Ph), 2.64 (d, J 3.7 Hz, 1 H, OH-3c) and Table V; ¹³C-NMR (CDCl₃, 100 MHz): δ 165.9, 164.8, 164.6 (C = O), 137.3, 137.2, 137.0, 136.8,133.6, 133.3, 133.2 (C-q arom.), 130.1–125.4 (C arom.),

102.0 (*C*HPh), 101.8 (*C*HPh), 100.6 (*C*HPh), 70.3 (*C*H₂Ph) and Table VI; ESI-MS: ($[M+Na]^+$, $C_{67}H_{62}O_{19}Na$): *m/z* calculated 1193.3783, found 1193.3771; ($[M+K]^+$, $C_{67}H_{62}O_{19}K$): *m/z* calculated 1209.3522, found 1209.3505; ($[M-H+2Na]^+$, $C_{67}H_{61}O_{19}Na_2$): *m/z* calculated 1215.3602, found 1215.3547.

Compound 15

According to procedure A, the required tetrasaccharide 15 2-O-benzoyl-4,6-O-benzylidene-3-O-NAP-β-D-(benzyl glucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylidene-β-Dglucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylidene-β-Dglucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylidene-β-Dglucopyranoside) was obtained starting from donor 8 (323 mg, 0.58 mmol) in CH₂Cl₂ (10 ml), containing molecular sieves (1.0 g), trisaccharide 14 (617 mg, 0.53 mmol), NIS (142 mg, 0.63 mmol), and TESOTf (10 µl, 0.05 mmol). After 1 h stirring and work-up, a chromatographic purification (toluene/ EtOAc, 19:1.9:1) yielded **15** (770 mg, 88%): TLC (toluene/ EtOAc, 4:1): $R_{\rm f} = 0.5$; mp: 132–134°C; $[\alpha]^{20}{}_{\rm D} + 28$ (c 1.0, CH₂Cl₂); ¹H-NMR (CDCl₃, 400 MHz): δ 7.78–7.04 (m, 52 H, H arom.), 5.53 (s, 1 H, CHPh), 5.43 (s, 1 H, CHPh), 4.89 [d, ²J 12.5 Hz, 1 H, CH₂(NAP)], 4.82 (s, 1 H, CHPh), 4.79 [d, 1 H, CH₂(NAP)], 4.76 (d, ²J 11.4 Hz, 1 H, CH₂Ph), 4.74 (s, 1 H, CHPh), 4.50 (d, 1 H, CH₂Ph) and Table V; ^{13}C -NMR (CDCl₃, 100 MHz): δ 165.1, 164.7, 164.6, 164.5 (C = O), 137.4, 137.3, 137.2, 137.1, 136.7, 135.4, 133.4, 133.3, 133.1, 133.0, 132.9 (C-q arom.), 129.8-125.3 (C arom.), 101.8 (CHPh), 101.2 (CHPh), 101.1 (CHPh), 100.8 (CHPh), 73.8 [CH₂(NAP)], 70.2 (CH₂Ph) and Table VI. Analytical calculated for C₉₈H₈₈O₂₅: C, 70.66, H, 5.32; found: C, 70.57, H, 5.23.

Compound 16

The reaction was performed according to procedure B starting from 15 (3.79 g, 2.28 mmol) in solvent (55 ml) and using DDQ (1.55 mg, 6.83 mmol). After for 7.5 h at room temperature, work-up and chromatographic purification (toluene/EtOAc, 17:3) afforded the required compound 16 (benzyl 2-O-benzoyl-4,6-O-benzylidene-β-D-glucopyranosyl-(1,3)-2-*O*-benzoyl-4,6-*O*-benzylidene-β-D-glucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylidene-β-D-glucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylidene-β-D-glucopyranoside) (2.78 g, 80%); TLC (toluene/EtOAc, 4:1): $R_f = 0.3$; mp: 162–164°C; $[\alpha]^{20}$ D – 9 (c 1.0, CH₂Cl₂); ¹H-NMR (CDCl₃, 400 MHz): δ 7.91-7.04 (m, 45 H, H arom.), 5.51 (s, 1 H, CHPh), 5.39 (s, 1 H, CHPh), 4.83 (s, 1 H, CHPh), 4.75 [d, ²J 10.8 Hz, 1 H, CH₂(NAP)], 4.74 (s, 1 H, CHPh), 4.50 (d, 1 H, CH₂(NAP)], 2.73 (d, J 3.8 Hz, 1 H, OH-3d) and Table V; ¹³C-NMR (CDCl₃, 100 MHz): δ 165.8, 164.7, 164.6, 164.6 (C = O), 137.3, 137.2, 137.1, 137.0, 136.7, 133.6, 133.4, 133.1, 133.0 (C-q arom.), 129.9–125.3 (C arom.), 101.8 (CHPh), 101.7 (CHPh), 101.1 (CHPh), 100.7 (CHPh), 70.2 (CH₂Ph) and Table VI. Analytical calculated for C₈₇H₈₀O₂₅: C, 68.50, H, 5.29; found: C, 68.09, H, 5.27.

Compound 17

The target pentasaccharide 17 (benzyl 2-*O*-benzoyl-4, 6-*O*-benzylidene-3-*O*-(2-naphthylmethyl)-β-D-glucopyranosyl-

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(1,3)-2-O-benzoyl-4,6-O-benzylidene-β-D-glucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylidene-β-D-glucopyranosyl-(1,3)-2-O-benzoyl-4,6-O-benzylidene-β-D-glucopyranosyl-(1.3)-2-*O*-benzoyl-4.6-*O*-benzylidene-β-D-glucopyranoside) was synthesized as described in procedure A, starting from 8 (337 mg, 0.61 mmol) in CH_2Cl_2 (18 ml), containing molecular sieves (1.8 g), tetrasaccharide 16 (838 mg, 0.55 mmol), NIS (148 mg, 0.66 mmol) and Sn(OTf)₂ (23 µg, 0.06 mmol). After 2 h stirring, work-up and chromatography (toluene/ EtOAc, 19:1, 9:1 and 17:3) enabled collecting 17 (820 mg, 74%); TLC (toluene/EtOAc, 4:1): R_f 0.6. mp: 162–165°C; $[\alpha]^{20}_{D}$ + 26 (c 1.0, CH₂Cl₂); ¹H=NMR (CDCl₃, 400 MHz): δ 7.85–7.04 (m, 62 H, H arom.), 5.51 (s, 1 H, CHPh), 5.45 (s, 1 H, CHPh), 5.36 (t, $J_{1e,2e}$ and $J_{2e,3e}$ 7.8 Hz, 1 H, H-2e), 5.14 (t, $J_{1c,2c}$ and $J_{2c,3c}$ 4.7 Hz, 1 H, H-2c*), 5.04 (d, 1 H, H-1e), 4.96 (t, $J_{1a,2a}$ and $J_{2a,3a}$ 8.4 H, 1 H, H-2a), 4.94 (d, 1 H, H-1c*), 4.92 (s, 1 H, CHPh), 4.91 [d, ²J 13.0 Hz, 1 H, $CH_{2(}$ NAP)], 4.89 (d, $J_{1d,2d}$ 7.3 Hz, 1 H, H-1d*), 4.83 (t, $J_{1b,2b}$ and $J_{2b,3b}$ 4.3 Hz, 1 H, H-2b*), 4.81 (dd, $J_{2d,3d}$ 5.9 H, 1 H, H-2d*), 4.80 (d, 1 H, CH₂(NAP)], 4.79 (d, 1 H, H-1b*), 4.75 (d, ²J 12.6 Hz, 1 H, CH₂Ph), 4.72 (s, 1 H, CHPh), 4.69 (s, 1 H, CHPh), 4.49 (d, 1 H, CH₂Ph), 4.46 (d, 1 H, H-1a), 4.33 (dd, $J_{5a,6a}$ 4.6 Hz, $J_{6a,6'a}$ 10.2 Hz, 1 H, H-6a), 4.21 (dd, $J_{5,6}$ 4.8 Hz, $J_{6,6'}$ 10.4 Hz, 1 H, H-6), 4.14– 4.12 (m, 2 H, H-6), 4.09–3.99 (m, 2 H, H-3d*, H-6), 4.06 (t, $J_{3a,4a}$ 9.0 Hz, 1 H, H-3a), 4.01 (dd, $J_{3c,4c}$ 10.4 Hz, 1 H, H-3c*), 3.99 (t, $J_{4c,5c}$ 8.5 Hz, 1 H, H-4c*), 3.91–3.86 (m, 1 H, H-3b*), 3.89 (t, J_{3e,4e} and J_{4e,5e} 8.3 Hz, 1 H, H-4e), 3.86 (dd, $J_{3e,4e}$ 8.5 Hz, 1 H, H-3e), 3.72 (t, $J_{5,6'}$ and $J_{6,6'}$ 10.8 Hz, 1 H, H-6'), 3.69 (t, $J_{5a,6'a}$ and $J_{6a,6'a}$ 10.7 Hz, 1 H, H-6'a), 3.67 (t, $J_{3b,4b}$ and $J_{4b,5b}$ 8.4 Hz, 1 H, H-4b*), 3.59–3.36 (m, 8 H, H-5a, H-5b, H-5c, H-5d, H-5e, 3 H-6), 3.24 (t, $J_{3d,4d}$ and $J_{4d,5d}$ 8.6 Hz, 1 H, H-4d*), 3.20 (t, 1 H, H-4a, $J_{3a,4a}$ and $J_{4a,5a}$ 9.3 Hz), asterisks indicate that assignment of these signals to units b, c, and d can be inverted; ¹³C-NMR (CDCl₃, 100 MHz): δ 165.1, 164.8, 164.7, 164.6, 164.5 (C = O), 137.4, 137.3, 137.2, 136.7, 135.4, 133.6, 133.5, 133.4, 133.1, 133.0, 132.8 (C-q arom.), 129.8–125.3 (C arom.), 102.0 (CHPh), 101.3 (CHPh), 101.1 (CHPh), 100.9 (CHPh), 100.6 (CHPh), 99.4 (C-1a), 98.6 (C-1e), 98.1 (C-1b*), 97.2 (C-1c*), 96.8 (C-1d*), 81.3 (C-4e), 78.7 (C-4a), 78.3 (C-3e), 78.0 (C-4c*), 77.9 (C-4d*), 77.3 (C-4b*), 76.6 (C-3c*), 74.5 (C-3a, C-3b*), 74.1 (C-2a, C-3d*), 73.8 (C-8e), 73.5 (C-2d*), 73.3 (C-2e), 72.9 (C-2b*), 72.4 (C-2c*), 70.2 (CH₂Ph), 68.7, 68.6, 68,5 (C-6a, C-6b, C-6c, C-6d, C-6e), 66.40 66.1, 65.6, 65.5, 65.4 (C-5a, C-5b, C-5c, C-5d, C-5e), asterisk indicates that assignment of these signals can be inverted. Analytical calculated for C₁₁₈H₁₀₆O₃₁: C, 70.16, H, 5.29; found: C, 70.23, H, 5.30.

Compound 18

The reaction was performed according to procedure B starting from **17** (2.0 g, 0.99 mmol) in solvent (20 ml) and using DDQ (675 mg, 2.97 mmol). After 5 h stirring and work-up, chromatographic purification (toluene/ EtOAc, 17:3, 4:1) provided **18** (benzyl 2-*O*-benzoyl-4,6-*O*-benzylidene- β -D-glucopyranosyl-(1,3)-2-*O*-benzoyl-4,6-*O*-benzylidene- β -D-glucopyranosyl-(1,3)-2-*O*-benzoyl-4,6-*D*-benzylidene- β -D-glucopyranosyl-(1,3)-2-*O*-benzoyl-4,6-*D*-benzylidene- β -D-glucopyranosyl-(1,3)-2-*D*-benzoyl-4,6-*D*-benzylidene- β -D-glucopyranosyl-(1,3)-2-*D*-benzoyl-4,6-*D*

(toluene/EtOAc, 4:1): $R_{\rm f} = 0.4$; mp: 176–179°C; $[\alpha]^{20}_{\rm D} + 3$ (c 1.0, CH₂Cl₂); ¹H-NMR (CDCl₃, 400 MHz): δ 7.96–7.04 (m, 55 H, H arom.), 5.51 (s, 1 H, CHPh), 5.42 (s, 1 H, CHPh), 5.20 (dd, $J_{1e,2e}$ 7.6 Hz, $J_{2e,3e}$ 10.4 Hz, 1 H, H-2e), 5.18 (t, $J_{1e,2e}$ and $J_{2c,3c}$ 5.2 Hz, 1 H, H-2c*), 5.09 (d, 1 H, H-1e), 4.97 (d, 1 H, H-1c*), 4.94 (dd, $J_{1a,2a}$ 7.8 Hz, $J_{2a,3a}$ 8.4 Hz, 1 H, H-2a), 4.92–4.91 (m, 2 H, H-1d*, H-2d*), 4.90 (s, 1 H, CHPh), 4.84 (t, $J_{1b,2b}$ and $J_{2b,3b}$ 4.5 Hz, 1 H, H-2b*), 4.80 (d, 1 H, H-1b*), 4.76 (d, ${}^{2}J$ 13.1 Hz, 1 H, CH₂Ph), 4.73 (s, 1 H, CHPh), 4.68 (s, 1 H, CHPh), 4.50 (d, 1 H, CH₂Ph), 4.46 (d, 1 H, H-1a), 4.33 (dd, *J*_{5,6} 4.7 Hz, *J*_{6,6}, 10.3 Hz, 1 H, H-6), 4.21 (dd, J_{5,6} 4.9 Hz, J_{6,6}, 10.4 Hz, 1 H, H-6), 4.13-3.88 (m, 9 H, H-3a, H-3b, H-3c, H-3d, H-3e, H-4c*, 3 H-6), 3.67 (t, $J_{3c,4c}$ and $J_{4c,5c}$ 8.7 Hz, 1 H, H-4c*), 3.65 (t, $J_{3e,4e}$ and $J_{4e,5e}$ 9.3 Hz, 1 H, H-4e), 3.72–3.34 (m, 10 H, H-5a, H-5b, H-5c, H-5d, H-5e, 5 H-6), 3.23 (t, J_{3d,4d} and J_{4d,5d} 10.0 Hz, 1 H, H-4d*), 3.20 (t, $J_{3a,4a}$ and $J_{4a,5a}$ 9.5 Hz, 1 H, H-4a), 2.69 (d, J 3.4 Hz, 1 H, OH), asterisks indicate that assignment of these signals to units b, c, and d can be inverted; ¹³C-NMR (CDČl₃, 100 MHz): δ 165.9, 164.8 164.7, 164.6, 164.5 (C = O), 137.3, 137.2, 137.0, 136.7, 133.8, 133.6,133.5, 133.2 (C-q arom.), 129.9–125.3 (C arom.), 102.0 (CHPh), 101.8 (CHPh), 101.4 (CHPh), 100.9 (CHPh), 100.6 (CHPh), 99.4 (C-1a), 98.4 (C-1e), 98.1 (C-1b*), 97.2 (C-1c*), 96.8 (C-1d*), 80.8 (C-4e), 78.7 (C-4a), 78.0 (C-4c*, C-4d*), 77.3 (C-4b*), 76.6 (C-3c*), 74.7 (C-2e), 74.5 (C-3a, C-3b*), 74.3 (C-3d*), 74.1 (C-2a), 73.4 (C-2d*), 72.9 (C-2b*), 72.5 (C-3e), 72.4 (C-2c*), 70.2 (CH₂Ph), 68.7, 68.6, 68.5 (C-6a, C-6b, C-6c, C-6d, C-6e), 66.4, 66.0, 65.6, 65.4 (C-5a, C-5b, C-5c, C-5d, C-5e). Analytical calculated for C₁₀₇H₉₈O₃₁: C, 68.36, H, 5.25; found: C, 68.05; H, 5.25.

Deprotection steps

Removal of benzylidene groups. To a solution of oligosaccharide (1 equiv.) in a acetone/MeOH/water (1:4:1, v/v/v) mixture was added an acid (1 equiv.) and the media was heated to 70°C. When TLC monitoring indicated completion of the reaction, the mixture was cooled to room temperature and neutralized with triethyamine; the solvents were coevaporated with ethanol. The residue was then dissolved in MeOH and poured onto cold light petroleum or cold toluene. The resulting solid was filtered off, washed, and dried. The expected compound was finally purified by flash chromatography.

Debenzoylation. Sodium (0.2 equiv.) was added to a solution of previously obtained oligosaccharide (1 equiv.) in MeOH. The mixture was heated to 50°C, neutralized with acetic acid after completion of the reaction, and concentrated. The crude residue was further dissolved in water and methyl benzoate removed with CH_2Cl_2 . The aqueous layer was then coevaporated with ethanol and the product purified by size exclusion chromatography (SEC) over Sephadex G-15 gel eluting with water. The collected fractions were finally freeze-dried to afford the target benzyl oligoglucoside.

Debenzylation. In a MeOH/water mixture (1:1, v/v) were successively introduced the oligosaccharide (1 equiv.) and palladium acetate. After vigorously stirring for 2 h at room temperature under a hydrogen atmosphere, palladium acetate (10% w) was added again and stirring under hydrogen was kept on 2 h more. After filtration on sintered glass, the organic layer was coevaporated with absolute ethanol. The final product was purified by gel permeation over Sephadex G-15 (water), and freeze-drying of the collected fractions provided the pure free oligosaccharide.

Compound 19

According to procedure C, compound 14 (2.16 g, 1.85 mmol) in solvent (84 mL) was treated with camphorsulfonic acid (CSA, 429 mg, 1.85 mmol) for 7 h. Work-up and chromatography (CH₂Cl₂/MeOH, 9/1) afforded 19 (benzyl 2-O-benzoyl-β-D-glucopyranosyl-(1,3)-2-O-benzoyl-β-Dglucopyranosyl-(1,3)-2-O-benzoyl-β-D-glucopyranoside) (1.29 g, 77%); TLC (CH₂Cl₂/MeOH, 17:3): $R_{\rm f}$ = 0.4; mp: $176-178^{\circ}C; [\alpha]^{20}_{D} + 15 (c 1.0, MeOH); {}^{1}H-NMR (CD_{3}OD),$ 400 MHz): δ 7.52–6.88 (m, 20 H, H arom.), 4.62 (d, ²J 12.5 Hz, 1 H, CH_2Ph), 4.42 (d, 1 H, CH_2Ph) and Table V; ¹³C-NMR (CD₃OD, 100 MHz): δ 167.2, 166.2, 166.0 (C = O), 138.5, 134.3, 134.1, 133.9 (C-q arom.), 130.9–128.6 (C arom.), 71.4 (CH₂Ph) and Table VI; ESI-MS: ([M+Na]⁺ C₄₆H₅₀O₁₉Na): *m*/*z* calculated 929.2844, found 929.2841; $([M+K]^+, C_{46}H_{50}O_{19}K)$: *m/z* calculated 945.2583, found 945.2523.

Compound 20

According to procedure C, compound 16 (2.88 g, 1.89 mmol) in solvent (84 ml) was submitted to deacetalyzation (CSA: 439 mg, 1.89 mmol) for 3 h. Work-up and chromatography (CH₂Cl₂/MeOH, 9:1) afforded 20 (benzyl 2-O-benzoyl-β-D-glucopyranosyl-(1,3)-2-O-benzoyl-4β-D-glucopyranosyl-(1,3)-2-O-benzoyl-β-D-glucopyranosyl-(1,3)-2-O-benzoyl-β-D-glucopyranoside) (1.62 g, 73%); TLC (CH₂Cl₂/MeOH, 17:3): $R_{\rm f} = 0.4$; mp: 152–154°C; [α]²⁰_D + 20 (*c* 1.0, MeOH); ¹H-NMR (CD₃OD, 400 MHz): δ 7.50–6.88 (m, 25 H, H arom.), 4.79 (d, $J_{1a,2a}$ and $J_{2a,3a}$ 8.7 Hz, 1 H, H-2a), 4.74 (t, $J_{1,2}$ and $J_{2,3}$ 8.8 Hz, 1 H, H-2b or H-2c), 4.72 (t, $J_{1,2}$ and $J_{2,3}$ 8.7 Hz, H-2b or H-2c), 4.71 (t, $J_{1d,2d}$ and $J_{2d,3d}$ 8.6 Hz, 1 H, H-2d), 4.62 (d, ²J 12.5 Hz, 1 H, CH₂Ph), 4.57 (d, $J_{1,2}$ 8.0 Hz, 1 H, H-1b or H-1c), 4.47 (d, $J_{1,2}$ 8.0 Hz, 1 H, H-1b or H-1c), 4.41 (d, 1 H, CH₂Ph), 4.40 (d, 1 H, H-1a), 4.37 (d, 1 H, H-1d), 3.82-3.69 (m, 6 H, H-3a, H-3b or H-3c, H-6), 3.62-3.52 (m, 5 H, H-3b or H-3c, H-6), 3.38-3.05 (m, 9 H, H-3d, H-4a, H-4b, H-4c, H-4d, H-5a, H-5b, H-5c, H-5d); ¹³C-NMR (CD₃OD, 100 MHz): δ 167.1, 166.1, 166.0, 165.8 (C = O), 138.4, 134.3, 134.2, 134.1, 133.9 (C-q arom.), 130.8–128.6 (C arom.), 102.2 (C-1b or C-1c), 101.8, 101.7 (C-1d and C-1b or C-1c), 100.9 (C-1a), 83.9 (C-3a), 82.8, 82.4 (C-3b, C-3c), 78.3, 78.0, 77.8 (C-5a, C-5b, C-5c, C-5d), 76.1 (C-3d), 75.2 (C-2d), 74.6, 74.5, 74.3 (C-2a, C-2b, C-2c), 71.5 (C-4d), 71.4 (*C*H₂Ph), 70.0, 69.8, 69.7 (C-4a, C-4b, C-4c), 62.5, 62.4, 62.4 (C-6a, C-6b, C-6c, C-6d); ESI-MS: $([M+Na]^+, C_{59}H_{64}O_{25}Na): m/z$ calculated 1195.3634, found 1195.3627; $([M+K]^+, C_{59}H_{64}O_{25}K)$: *m/z* calculated 1211.3374, found 1211.3461.

Compound 21

According to procedure C, compound **18** (2.10 g, 1.12 mmol) in solvent (60 ml) was treated with PTSA, H_2O (213 mg, 1.12 mmol) during 4 h. Work-up followed by chromatographic purification (CH₂Cl₂/MeOH, 9:1) afforded **21** (benzyl 2-*O*-benzoyl- β -D-glucopyranosyl-(1,3)-2-*O*-ben-

zoyl-β-D-glucopyranosyl-(1.3)-2-O-benzoyl-β-D-glucopyranosyl-(1.3)-2-O-benzoyl-4β-D-glucopyranosyl-(1.3)-2-Obenzoyl- β -D-glucopyranoside) (1.14 g, 76 %): TLC $(CH_2Cl_2/MeOH, 17:3): R_f = 0.3; mp: 171-174^{\circ}C; [\alpha]^{20}D +$ 21 (c 1.0, MeOH); ¹H-NMR (CD₃OD, 400 MHz): δ 7.48– 6.87 (m, 30 H, H arom.), 4.79–4.52 (m, 5 H, H-2a, H-2b, H-2c, H-2d, H-2e), 4.60 (d, ${}^{2}J$ 12.5 Hz, 1 H, CH₂Ph), 4.53 $(d, J_{1,2} 8.0 \text{ Hz}, 1 \text{ H}, \text{H-1b}, \text{H-1c}, \text{H-1d or H-1e}), 4.39 (d, J_{1,2})$ 7.8 Hz, 1 H, H-1b, H-1c, H-1d or H-1e), 4.39 (d, 1 H, CH₂Ph), 4.38 (d, J_{1a,2a} 7.8 Hz, 1 H, H-1a), 4.31 (d, J_{1,2} 8.2 Hz, 1 H, H-1b, H-1c, H-1d or H-1e), 4.29 (d, $J_{1,2}$ 8.3 Hz, 1 H, H-1b, H-1c, H-1d or H-1e), 3.79–3.72 (m, 6 H, H-3, H-6), 3.67–3.43 (m, 8 H, H-3, H-6), 3.36–3.02 (m, 9 H, H-3e, H-4a, H-4b, H-4c, H-4d, H-4e, H-5a, H-5b, H-5c, H-5d, H-5e); ¹³C-NMR (CD₂OD, 100 MHz): δ 167.1, 166.1, 165.9, 165.8, 165.8 (C = O), 138.4, 134.4, 134.3, 134.2, 134.1, 133.9 (C-q arom.), 130.8-128.5 (C arom.), 102.2, 101.9, 101.8, 101.7 (C-1b, C-1c, C-1d, C-1e), 100.9 (C-1a), 83.9 (C-3a), 82.8, 82.4, 82.3 (C-3b, C-3c, C-3d), 78.4, 78.1, 77.9, 77.8 (C-5a, C-5b, C-5c, C-5d, C-5e), 76.1 (C-3e), 75.2 (C2-e), 74.6, 74.4, 74.3 (C-2a, C-2b, C-2c, C-2d), 71.5 (C-4e), 71.4 (CH₂Ph), 70.0, 69.8, 69.7, 69.6 (C-4a, C-4b, C-4c, C-4d), 62.6, 62.5, 62.4 (C-6a, C-6b, C-6c, C-6d, C-6e); ESI-MS: ([M+Na]⁺,C₇₂H₇₈O₃₁Na): *m*/*z* cal-culated 1461.4425, found 1461.4413; $([M+K]^+, C_{72}H_{78}O_{31}K): m/z$ calculated 1477.4164, found 1477.4237; $([\tilde{M}-H+2Na]^+, C_{72}H_{77}O_{31}Na_2)$: *m/z* calculated 1483.4244, found 1483.4241.

Compound 22

This compound was obtained as described in procedure D, starting from 19 (1.97 g, 2.17 mmol) in MeOH (60 ml) and sodium (75 mg, 3.26 mmol). After 8 h stirring, work-up and chromatography, 22 (benzyl β -D-glucopyranosyl-(1,3)- β -Dglucopyranosyl-(1,3)- β -D-glucopyranoside) (1.29 mg) was isolated in quantitative yield; TLC (EtOAc/i-PrOH/H₂O, 3:3:1): $R_{\rm f} = 0.6$; mp: 176–178°C; $[\alpha]^{20}_{\rm D} - 29$ (*c* 1.0, water); ¹H-NMR (D₂O, 400 MHz): δ 7.34–7.26 (m, 5 H, H arom.), 4.81 (d, ${}^{2}J$ 11.6 Hz, 1 H, CH_{2} Ph), 4.63 (d, $J_{1,2}$ 7.9 Hz, 1 H, H-1b or H-1c), 4.62 (d, 1 H, CH_{2} Ph), 4.61 (d, $J_{1,2}$ 7.9 Hz, 1 H, H-1b or H-1c), 4.42 (d, $J_{1a,2a}$ 8.0 Hz, 1, H-1a), 3.81–3.76 (m, 3 H, H-6), 3.65–3.55 (m, 5 H, H-3a, H-3b, 3 H-6), 3.42– 3.19 (m, 10 H, H-2a, H-2b, H-2c, H-3c, H-4a, H-4b, H-4c, H-5a, H-5b, H-5c); ¹³C-NMR (D₂O, 100 MHz): δ 136.7 (Cq arom.), 129.1, 129.0, 128.8 (C arom.), 103.1, 102.8 (C-1b, C-1c), 101.2 (C-1a), 84.5, 84.3 (C-3a, C-3b), 76.3, 75.9, 75.8 (C-3c, C-5a, C-5b, C-5c), 73.7, 73.5, 73.2 (C-2a, C-2b, C-2c), 71.8 (CH₂Ph), 69.8 (C-4c), 68.4, 68.3 (C-4a, C-4b), 60.9 (C-6a, C-6b, \overline{C} -6c); ESI-MS: ([M+Na]⁺, C₂₅H₃₈O₁₆Na): m/zcalculated 617.2058, found 617.2060; ($[M+K]^+$, $C_{25}H_{38}O_{16}K$): m/z calculated 633.1797, found 633.1801; ([M-H+2Na]⁺, C₂₅H₃₇O₁₆Na₂): *m*/*z* calculated 639.1877, found 639.1876.

Compound 23

This compound was prepared as disclosed in procedure D starting from **20** (2.29 g, 1.96 mmol) in MeOH (70 ml) and using sodium (67 mg, 2.91 mmol). After 8 h stirring, workup and chromatography afforded **23** (benzyl β -D-glucopyranosyl-(1,3)-4 β -D-glucopyranosyl-(1,3)- β -D-glucopyranosyl-(1,3)-? β -D-glucopyranoside) (1.47 g) in 99% yield; TLC (EtOAc/*i*-PrOH/H₂O, 3:3:1): $R_{\rm f} = 0.5$; $[\alpha]^{20}_{\rm D} - 29$ (c 1.0, water); ¹H-NMR (D₂O, 400 MHz): δ 7.35–7.26 (m, 5 H, H arom.), 4.81 (d, ${}^{2}J$ 11.6 Hz, 1 H, CH₂Ph), 4.65 (d, $J_{1,2}$ 8.2 Hz, 1 H, H-1b, H-1c or H-1d), 4.62 (d, $J_{1,2}$ 8.6 Hz, 1 H, H-1b, H-1c or H-1d), 4.62 (d, 1 H, CH_2Ph), 4.61 (d, $J_{1,2}$ 8.0 Hz, 1 H, H-1b, H-1c or H-1d), 4.42 (\bar{d} , $J_{1a,2a}$ 8.0 Hz, 1 H, H-1a), 3.81-3.76 (m, 4 H, H-6), 3.66-3.56 (m, 7 H, H-3a, H-3b, H-3c, H-6), 3.43-3.19 (m, 13 H, H-2a, H-2b, H-2c, H-2d, H-3d, H-4a, H-4b, H-4c, H-4d, H-5a, H-5b, H-5c, H-5d); ¹³C-NMR (D₂O, 100 MHz): δ 136.7 (C-q arom.), 129.1, 129.0, 128.8 (C arom.), 103.1, 102.8, 102.7 (C-1b, C-1c, C-1d), 101.2 (C-1a), 84.5, 84.40 84.2 (C-3a, C-3b, C-3c), 76.3, 75.9, 75.8 (C-3d, C-5a, C-5b, C-5c, C-5d), 73.7, 73.6, 73.5, 73.2 (C-2a, C-2b, C-2c, C-2d), 71.8 (CH₂Ph), 69.8 (C-4d), 68.5, 68.4, 68.3 (C-4a, C-4b, C-4c), 60.9 (C-6a, C-6b, C-6c, C-6d); ESI-MS: $([M+Na]^+, C_{31}H_{48}O_{21}Na)$: m/zcalculated 779.2586, found 779.2580; $([M+K]^+, C_{31}H_{48}O_{21}K)$: m/z calculated 795.2325, found 795.2380.

Compound 24

Debenzoylation was performed according to procedure D and starting from 21 (1.62 g, 1.13 mmol) in MeOH (54 ml) with sodium (39 mg, 1.70 mmol). After 7 h stirring and work-up, SEC purification provided target 24 (benzyl β -D-glucopyranosyl-(1,3)- β -D-glucopyranosyl-(1,3)- β -D-glucopyranosyl-(1,3)-β-D-glucopyranosyl-(1,3)-β-D-glucopyranoside) (1.03 mg, 99%); TLC (EtOAc/i-PrOH/H₂O, 3:3:1): R_f = 0.4; $[\alpha]_{D}^{20} - 21$ (c 1.0, water); ¹H-NMR (D₂O, 400 MHz): δ 7.35–7.28 (m, 5 H, H arom.), 4.81 (d, ²J 11.6 Hz, 1 H, CH₂Ph), 4.66 (d, J_{1,2} 8.0 Hz, 1 H, H-1b, H-1c, H-1d or H-1e), 4.65 (d, J_{1,2} 8.4 Hz, 1 H, H-1b, H-1c, H-1d or H-1e), 4.63 (d, $J_{1,2}$ 9.0 Hz, 1 H, H-1b, H-1c, H-1d or H-1e), 4.63 (d, 1 H, CH_2Ph), 4.61 (d, $J_{1,2}$ 7.4 Hz, 1 H, H-1b, H-1c, H-1d or H-1e), 4.41 (d, $J_{1a,2a}$ 8.0 Hz, 1 H, H-1a), 3.81–3.76 (m, 5 H, H-6), 3.66–3.56 (m, 9 H, H-3a, H-3b, H-3c, H-3d, H-6), 3.43-3.20 (m, 16 H, H-2a, H-2b, H-2c, H-2d, H-2e, H-3e, H-4a, H-4b, H-4c, H-4d, H-4e, H5-a, H-5b, H-5c, H-5d, H-5e); ¹³C-NMR (D₂O, 100 MHz): δ 136.7 (C-q arom.), 129.1, 129.0, 128.8 (C arom.), 103.1, 102.8 (C-1b, C-1c, C-1d, C-1e), 101.2 (C-1a), 84.5, 84.4, 84.2 (C-3a, C-3b, C-3c, C-3d), 76.3, 75.9, 75.8 (C-3e, C-5a, C-5b, C-5c, C-5d, C-5e), 73.7, 73.6, 73.5, 73.2 (C-2a, C-2b, C-2c, C-2d, C-2e), 71.8 (C-7), 69.8 (C-4e), 68.5, 68.4, 68.3 (C-4a, C-4b, C-4c, C-4d), 60.9 (C-6a, C-6b, C-6c, C-6d, C-6e); ESI-MS: ([M+Na]⁺, $C_{37}H_{58}O_{26}Na$): *m/z* calculated 941.3114, found 941.3114; ([M+K]⁺, C₃₇H₅₈O₂₆K): *m/z* calculated 957.2853, found 957.2821; : ($[M-H+2Na]^+$, $C_{37}H_{57}O_{26}Na_2$): *m/z* calculated 963.2934, found 963.2939.

Laminaritriose (2)

The reaction was performed according to procedure E starting from **22** (850 mg, 1.04 mmol) in solvent (20 ml) and in the presence of palladium acetate (85 mg, 0.38 mmol, twice). After stirring for 3.5 h, work-up, and SEC, **2** (0.50 g) was isolated in 95% yield; TLC (EtOAc/*i*-PrOH/H₂O, 3:3:2): $R_{\rm f} = 0.3$; ¹H-NMR (D₂O, 400 MHz): $\delta 5.12$ (d, $J_{1a\alpha,2a\alpha}$ 3.7 Hz, 0.45 H, H-1a α), 4.66 (d, $J_{1,2}$ 8.6 Hz, 1 H, H-1), 4.65 (d, $J_{1,2}$ 8.0 Hz, 1 H, H-1), 4.56 (d, $J_{1a\beta,2a\beta}$ 8.0 Hz, 0.55 H, H-1a β), 3.83–3.59 (m, 9 H), 3.47–3.23 (m, 9 H); ¹³C-NMR (D₂O, 100

MHz): δ 103.1, 102.9, 102.8 (C-1b, C-1c), 96.0 (C-1aβ), 92.3 (C-1aα), 84.7, 84.5, 84.4, 82.5 (C-3a, C-3b), 76.3, 75.9, 75.8, 74.2, 73.7, 73.6, 73.5, 71.5, 71.4, 69.9, 68.4, 68.3, 61.0, 60.8 (C-6). ESI-MS: ([M+Na]⁺, C₁₈H₃₂O₁₆Na): *m/z* calculated 527.1588, found 527.1588; ([M+K]⁺, C₁₈H₃₂O₁₆K): *m/z* calculated 543.1327, found 543.1367.

Laminaritetraose (3)

The required tetrasaccharide was synthesized as disclosed in procedure E with **23** (1.12 g, 1.48 mmol), solvent (20 ml) and catalyst (112 mg, 0.05 mmol, twice). After stirring for 5 h, work-up and chromatographic purification gave **3** (0.95 g, 96%): TLC (EtOAc/i-PrOH/H₂O, 3:3:2): $R_{\rm f} = 0.2$; ¹H-NMR (D₂O, 400 MHz): δ 5.13 (d, $J_{1a\alpha,2a\alpha}$ 3.7 Hz, 0.36 H, H-1a\alpha), 4.67 (d, $J_{1,2}$ 8.9 Hz, 2 H, H-1), 4.65 (d, $J_{1,2}$ 7.8 Hz, 1 H, H-1), 4.57 (d, $J_{1a\beta,2a\beta}$ 8.0 Hz, 0.64 H, H-1a\beta), 3.83–3.59 (m, 12 H), 3.47–3.23 (m, 12 H); ¹³C-NMR (D₂O, 100 MHz): δ 103.1, 103.0, 102.9, 102.8 (C-1b, C-1c, C-1d), 96.0 (C-1a\beta), 92.3 (C-1a\alpha), 84.7, 84.5, 84.3, 84.2, 82.5 (C-3a, C-3b, C-3c), 76.3, 75.9, 75.8, 74.1, 73.7, 73.6, 73.5, 71.5, 71.4, 69.9, 68.4, 68.3, 61.0, 60.8 (C-6); ESI-MS: ([M+Na]⁺, C₂₄H₄₂O₂₁Na): *m/z* calculated 689.2116, found 689.2121; ([M+K]⁺, C₂₄H₄₂O₂₁K): *m/z* calculated 705.1856, found 705.1843.

Laminaripentaose (4)

Final deprotection was performed according to procedure E starting from **24** (960 mg, 1.04 mmol) in solvent (20 ml) and using palladium acetate (192 mg, 0.43 mmol, twice). After 2.5 h stirring, work-up and SEC purification gave the required pentasaccharide **26** (0.81 g) in 94% yield. TLC (EtOAc/*i*-PrOH/H₂O, 3:3:2): $R_{\rm f} = 0.2$; ¹H-NMR (D₂O, 400 MHz): δ 5.12 (d, $J_{1a\alpha,2a\alpha}$ 3.8 Hz, 0.46 H, H-1a α), 4.68 (d, $J_{1,2}$ 8.0 Hz, 1 H, H-1), 4.68 (d, $J_{1,2}$ 7.7 Hz, 1 H, H-1), 4.66 (d, $J_{1,2}$ 7.8 Hz, 1 H, H-1), 4.64 (d, $J_{1,2}$ 7.7 Hz, 1 H, H-1), 4.56 (d, $J_{1a\beta,2a\beta}$ 8.0 Hz, 0.54 H, H-1a β), 3.82–3.58 (m, 15 H), 3.47–3.23 (m, 15 H); ESI-MS: ([M+Na]⁺, C₃₀H₅₂O₂₆Na): *m/z* calculated 851.2645, found 851.2650; ([M-H+2Na]⁺, C₃₀H₅₁O₂₆Na₂): *m/z* calculated 873.2464, found 873.2525.

Animals

Female, 6–8 weeks old BALB/c mice were purchased from Jackson Laboratory (Bar Harbor, ME).

β -(1,3)-glucan

Phycarine was extracted and purified from the marine brown alga *Laminaria digitata* as described in Vetricka and Yvin (2004).

Phagocytosis of peripheral blood cells

Twenty-four hours after IP injection with oligosaccharides, the mice were sacrificed. Peripheral blood from the orbital plexus was collected into heparine (5 IU/ml; Sigma, St.Louis, MO). Phagocytosis of HEMA particles has been done as previously described (Vetricka and Yvin 2004). Following incubation, two blood smears on slides were prepared from two parallel samples, stained with Accustain (Sigma) and evaluated. Differential counts and numbers of phagocytosis cells were established. Cells with at least three engulfted particles were considered positive.

Phagocytosis of peritoneal cells

Peritoneal macrophages were isolated from the peritoneal cavities of mice injected with oligosaccharides. The cells were diluted in RPMI 1640 medium with 5% fetal calf serum (Hyclone, Logan, UT) to 1×10^7 and incubated with HEMA particles as described earlier (Vetricka and Yvin, 2004).

IL- β assay

BALB/c mice were intraperitoneally injected with 100 μ g oligosaccharides. Control mice obtained phosphate buffered saline (PBS) only. After various time intervals (30, 60, and 90 min, respectively), the mice were sacrificed, and blood was collected in Eppendorf tubes. Subsequently, the serum was prepared, collected and stored at -80°C for no more than 1 week.

The levels of IL-1 β in serum samples were evaluated using a commercial kit OptEIA Mouse IL-1 β Set (Pharmingen, San Diego, CA) according to the manufacturer's instructions. The optical density was determined using a STL enzyme-linked immunosorbent assay reader (Tecan U.S., Research Triangle Park, NC) at 450 nm with a correction at 570 nm. Data shown in Figure 6 were calculated from the standard curve prepared by the automated data reduction using linear regression analysis. A standard curve was run with each assay.

Cell lines

The BALB/c mouse-derived mammary tumor cell line Ptas64 was generously provided by Dr. Wei-Zen Wei of the Michigan Cancer Foundation (Wayne State University, Detroit, MI).

Tumor inhibition in vivo

Mice were injected directly into the mammary fat pads with 1×10^{6} /mouse of Ptas64 cells in PBS. The experimental treatment was begun after palpable tumors were found (usually 14 days after injection of cells) and after mice were assigned to experimental groups. Experimental treatment was achieved by daily IP injections of either phycarine or oligosaccharides diluted in PBS. After 2 weeks of treatment, the mice were sacrificed, and tumors were removed and weighed.

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Abbreviations

COSY, correlation spectroscopy; CSA, camphorsulfonic acid; DDQ, 2,3-dichloro-5,6-dicyano-1,4-quinone; ESI-MS, electrospray ionization mass spectrometry; HEMA, 2-hydroxyethyl methacrylate; NAP, 2-naphthylmethyl; NIS, *N*iodosuccinimide; NMR, nuclear magnetic resonance; NOE, nuclear Overhauser effect; NOESY, nuclear Overhauser

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effect spectroscopy; PBS, phosphate buffered saline; SEC, size exclusion chromatography; TESTOf, triethylsilyl trifluoromethanesulfonate; TLC, thin-layer chromatography.

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