

HfO₂ as gate dielectric on Ge: Interfaces and deposition techniques

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Abstract

To fabricate MOS gate stacks on Ge, one can choose from a multitude of metal oxides as dielectric material which can be deposited by many chemical or physical vapor deposition techniques. As a few typical examples, we will discuss here the results from atomic layer deposition (ALD), metal organic CVD (MOCVD) and molecular beam deposition (MBD) using HfO₂/Ge as materials model system. It appears that a completely interface layer free HfO₂/Ge combination can be made in MBD, but this results in very bad capacitors. The same bad result we find if HfGe_y (Hf germanides) are formed like in the case of MOCVD on HF-dipped Ge. A GeO_x interfacial layer appears to be indispensable (if no other passivating materials are applied), but the composition of this interfacial layer (as determined by XPS, TOF-SIMS and MEIS) is determining for the *C/V* quality. On the other hand, the presence of Ge in the HfO₂ layer is not the most important factor that can be responsible for poor *C/V*, although it can still induce bumps in *C/V* curves, especially in the form of germanates (Hf–O–Ge). We find that most of these interfacial GeO_x layers are in fact sub-oxides, and that this could be (part of) the explanation for the high interfacial state densities. In conclusion, we find that the Ge surface preparation is determining for the gate stack quality, but it needs to be adapted to the specific deposition technique.

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Keywords: Hafnium oxide; Germanium; Metal–oxide–semiconductor structures; Gate stack; Atomic layer deposition; Metal organic vapor deposition; Molecular beam deposition

1. Introduction

Germanium as alternative, high mobility MOS channel material requires a deposited gate dielectric, because of the unstable character of thermally grown GeO₂ [1]. Initially, the high-k metal oxides and their deposition processes as developed for Silicon CMOS seemed to offer a quick solution for this problem. Interfacial passivation for Si/high-k systems was basically “just” a matter of creating and maintaining a minimum of SiO₂ and preventing silicide formation. For Ge however this is less straightforward due to the unstable character of GeO₂. Because of this, several alternative passivation techniques have been proposed and tested such as nitridation by NH₃ annealing [2] or SiH₄ annealing [3] which is similar to Si deposition [4]. Although the best Ge MOS transistor results so far published have been obtained in Si-passivated structures, in this overview,

we will limit ourselves to interfacial passivation based on Ge oxide or oxide–nitride layers formed either ex situ or in situ, in combination with HfO₂ layers deposited by either atomic layer deposition (ALD), metal organic chemical vapor deposition (MOCVD) or molecular beam deposition (MBD). We will show that GeO₂-like interfacial layers can be better than expected, and that other factors can be responsible for poor *C/V* characteristics.

2. Experimental methods

HfO₂ deposition was done by atomic layer deposition (ALD) in an ASM Pulsar[®] 2000 reactor from HfCl₄ and H₂O at 300 °C. This Pulsar[®] reactor is part of a 200 mm reactor cluster Polygon[®] 8200, including amongst others an Epsilon[®] style nitride module which is basically a lamp heated, single wafer horizontal flow reactor with a quartz tube equipped with NH₃, O₂ and other gases. MOCVD is performed in a 200 mm Applied Materials Centura[®] MOCVD reactor module. We use tetrakis

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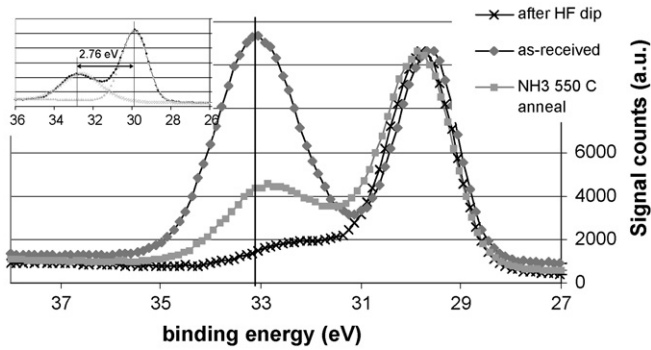


Fig. 1. XPS analysis of Ge substrate: Ge 3d spectrum of Ge surface as-received, after an HF dip and after NH₃ anneal at 550 °C. The inset shows a de-convolution of the NH₃-annealed sample spectrum.

di-ethyl amido hafnium (TDEAH) and O₂ as precursor gases at a standard temperature of 485 °C. Part of this Centura cluster is an ISSG rapid thermal oxidation chamber. Molecular beam deposition (MBD) is carried out in an 2 in. ultra-high vacuum (UHV) MBE tool. For MBD, the native oxide was desorbed under UHV for 15 min at 360 °C (until clear (2 × 1) Ge surface reconstruction was visible in RHEED) and HfO₂ was grown with atomic Hf and O beams at 225 °C. O was supplied from a remote RF plasma source.

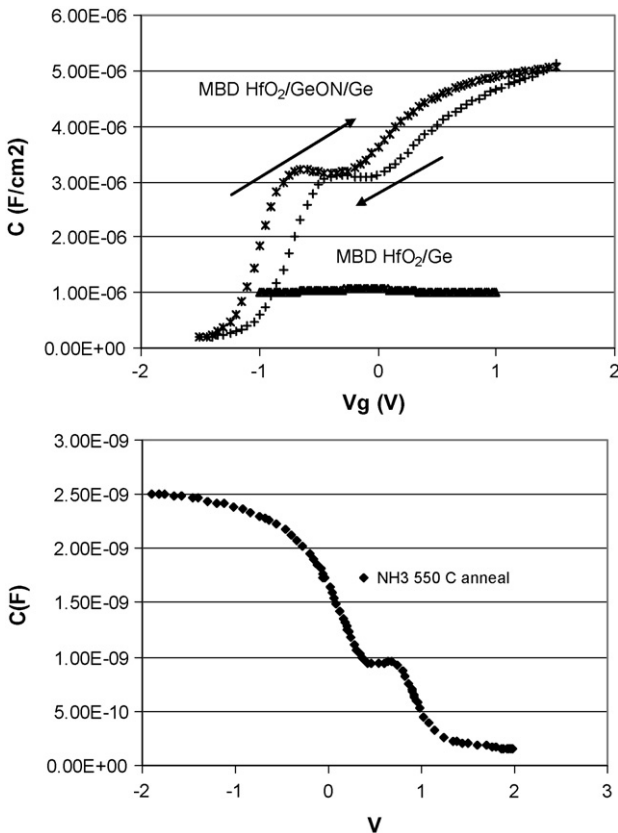


Fig. 2. C/V curves of TiN metal dot capacitors on MBD (upper panel) and MOCVD (lower panel) deposited HfO₂. The samples in the upper panel contain 4.5 nm HfO₂ on either HF-dipped Ge (lower curve) or on a GeON layer (upper curves). The latter ones, measured at 100 kHz, show a shift in the flat-band voltage (hysteresis) of 200–250 mV for back-and-forth voltage sweeps.

3. Results

Ge wafers out-of-the-box are covered by a ~1.7 nm thick native oxide (as measured by X-ray photo-electron spectroscopy, XPS), comprising mainly GeO₂, see the XPS Ge 3d spectrum showing the Ge⁰ and Ge⁴⁺ states in Fig. 1 (“as-received”). Stripping the native oxide in an H₂O/HF solution leaves ~0.3 nm GeO_x (same figure). De-convoluting the spectra (see by way of example the inset in Fig. 1, showing the de-convoluted spectrum after NH₃ anneal) allows to estimate the binding energy shift between the elemental and the oxidized Ge peak. In the as-received sample, this shift is 3.38 eV, which is very close to the shift of 3.4 eV assigned to Ge⁴⁺ (GeO₂) as determined by Schmeisser et al. [5]. The Ge sub-oxide peak in the HF dipped sample shows a shift of ~2.18 eV. According to [5], this would mean this 0.3 nm sub-oxide layer is comprised of a mixture of Ge²⁺ and Ge³⁺ with respective shifts of 1.8 and 2.6 eV.

The MBD (Fig. 2) and MOCVD (not shown) layers typically resulted in flat C–V curves and suffered from very high leakage (not shown). The ALD layers on the other hand showed not

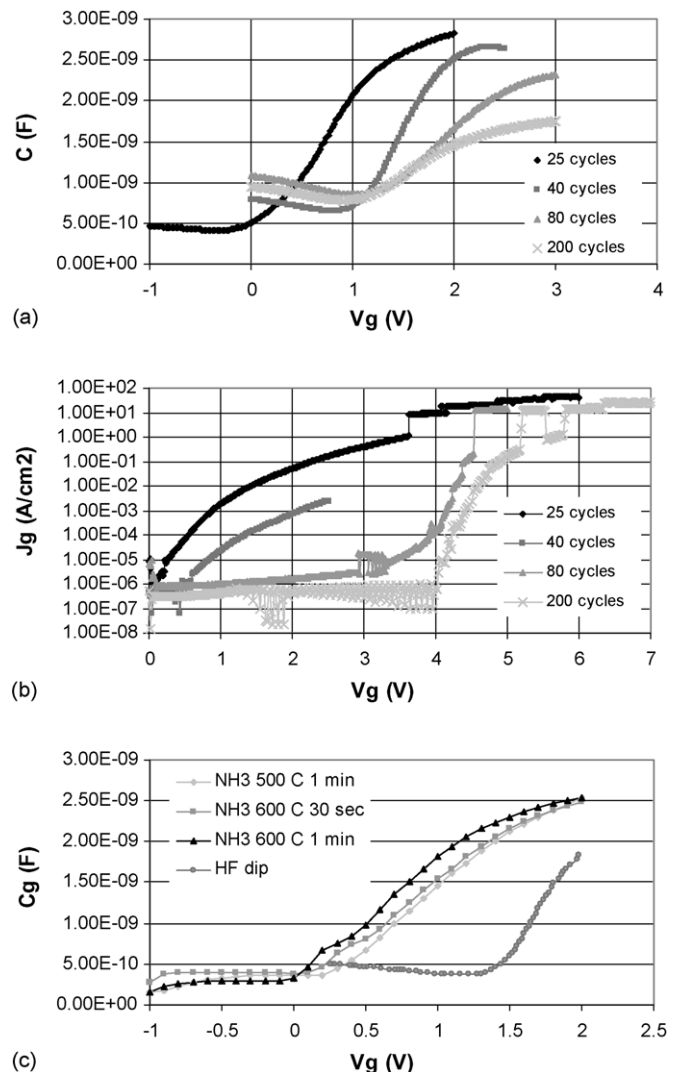


Fig. 3. C/V (a) and I/V (b) analysis of ALD HfO₂ on HF dipped and NH₃-annealed (c) Ge.

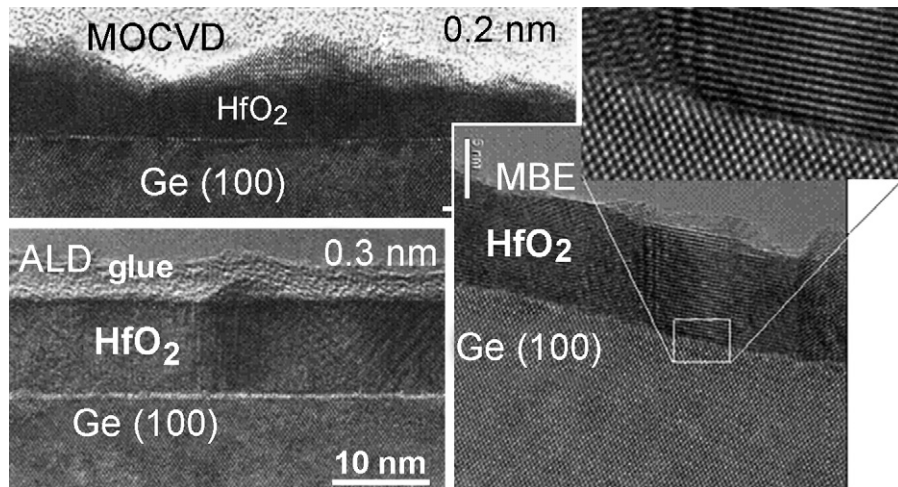


Fig. 4. x-TEM images of HfO₂ layers on Ge, deposited by MOCVD, ALD and MBD. The interfacial layer thickness is also indicated.

perfect but still quite reasonable C – V and I – V curves with a capacitance and leakage that scale with the thickness of the HfO₂ layer, see Fig. 3.

The interface between the MBD HfO₂ grown straight on the (2×1) reconstructed Ge surface appears to be very sharp without any noticeable interfacial layer according to cross-sectional TEM (x-TEM) (Fig. 4). The HfO₂ layer surface shows some roughness which seems to be correlated to crystallization. XPS of the Ge 3p core electrons (Fig. 5, middle panel) shows a clean spectrum with the Ge 3p_{1/2} and Ge 3p_{3/2} peaks, without any additional bump that could indicate oxidized Ge. Also TOFSIMS (Fig. 7, middle panel) shows a sharp interface between the Ge substrate and the HfO₂ layer, with almost no out-diffusion of Ge into the HfO₂.

The MOCVD layer on the HF-dipped Ge substrate shows a very rough surface in the x-TEM images in Fig. 4 while the ALD layer (also on HF-dipped Ge) is very smooth. In contrast to this, both the ALD and MOCVD layers grown on HF-dipped Ge show a thin, smooth and well-defined interfacial layer of 0.2–0.3 nm.

The Ge 3s spectrum of the ALD layer (XPS, Fig. 5, upper panel) can be resolved completely with only two peaks, one for the metallic Ge of the substrate at 182.2 eV and another one at 184.9 eV, which corresponds to the Ge⁴⁺ oxidation state, i.e. GeO₂. Similarly to the MBD layer, TOFSIMS (Fig. 7, lower panel) cannot find almost any Ge inside the HfO₂ layer, even not after a 600 °C O₂ anneal. According to medium ion scattering spectroscopy (MEIS, Fig. 6, panel a) the ALD interfacial layer is mainly composed of Ge and O (with quite some Cl contamination), with only a minor trace of Hf. The MOCVD HfO₂ layer on HF-dipped Ge is quite different: TOFSIMS (Fig. 7, upper panel) shows strong Ge in-diffusion from the substrate into the HfO₂ layer. According to MEIS (Fig. 6, panel b), the HfO₂ layer indeed looks more like a mixed HfO₂/GeO₂ layer, and at the interface there is a strong indication for a Hf Germanide phase (Hf–Ge mix, without the presence of O).

A possible passivation of the surface is nitridation of the Ge surface such as by annealing in NH₃ [2]. Fig. 1 shows an XPS

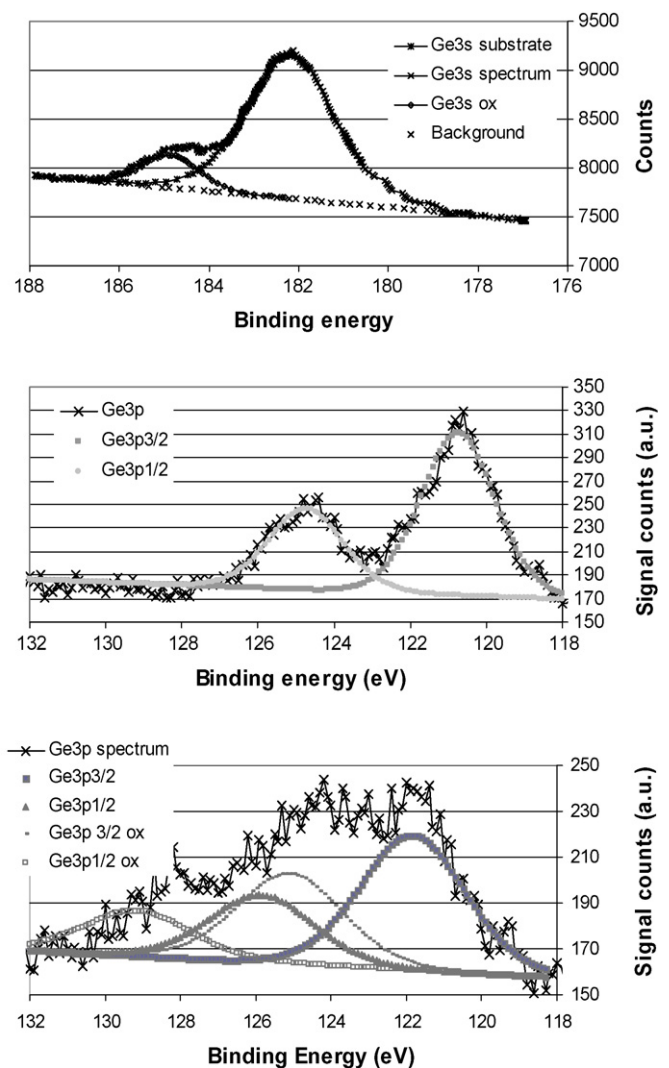


Fig. 5. XPS Ge 3p spectrum of ALD HfO₂ on Ge after HF dip (upper), MBD HfO₂ on Ge (middle) and on GeON/Ge (lower figure).

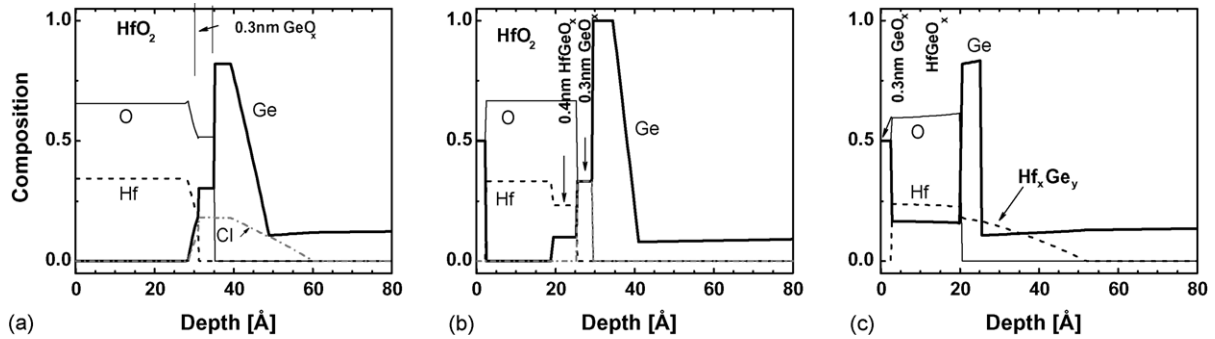


Fig. 6. MEIS spectra fittings for ALD (a) and MOCVD (b) HfO₂ on HF-dipped Ge, and MOCVD HfO₂ on NH₃-annealed Ge (c).

spectrum of the Ge surface after a 1 min NH₃ 550 °C anneal, which results in a Ge–O–N type of bond with a reduced peak shift of the oxidized Ge peak as compared to the fully oxidized Ge⁴⁺. For MOCVD HfO₂ deposition on such a surface, TOFSIMS

shows that the Ge in-diffusion in the HfO₂ layer is reduced with an order of magnitude (Fig. 7, upper panel). MEIS shows a strong change in the MOCVD case, in that the Hf germanide interface now is replaced by a HfGeO_x layer with a graded Hf content (Fig. 6, panel c). Capacitors on these MOCVD layers on nitrated Ge show a pretty decent *C/V* shape albeit with some bump in depletion that could be due to interfacial states (Fig. 2, lower panel). Also their leakage current is considerably reduced (not shown) in contrast to the poor behavior on HF-dipped Ge. For *C/V* on ALD oxide layers (Fig. 3c), we notice no specific changes in the *C/V* shape except a strong shift in *V_{FB}*, likely due to fixed charges because of the presence of N [6].

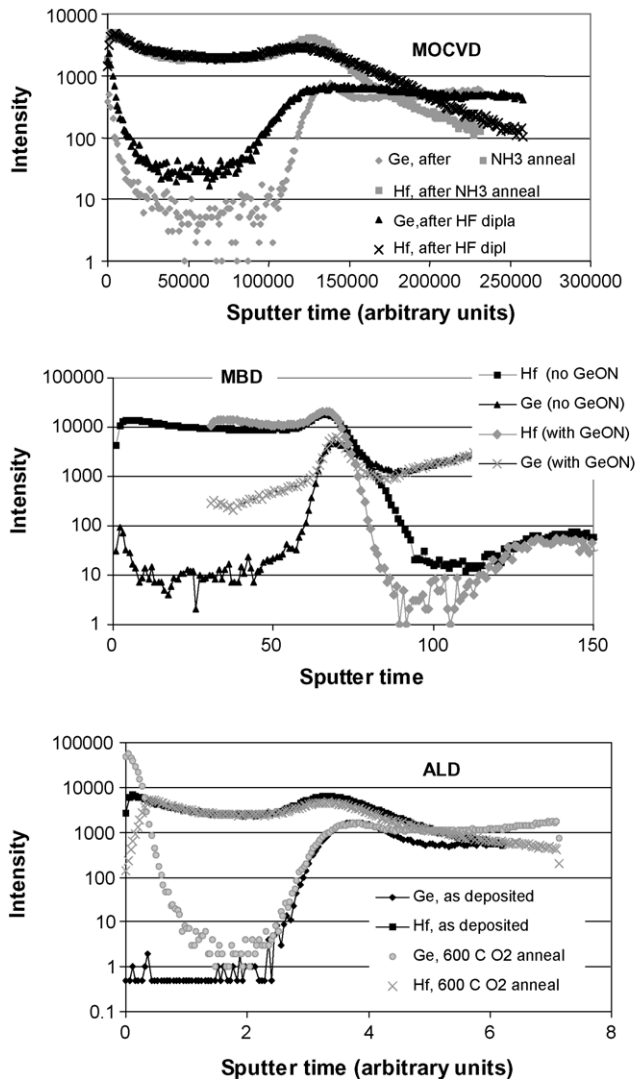


Fig. 7. TOFSIMS profiles of Hf and Ge in HfO₂ layers: MOCVD (upper panel, HfO₂ on Ge after HF dip and after NH₃ 550 °C anneal), MBD (middle panel, HfO₂ on Ge, HfO₂ on GeON/Ge), ALD (lower panel, as-deposited HfO₂ on HF-dipped Ge, and after 600 °C O₂ anneal).

In the case of MBD, we deposited in situ a GeO_xN_y layer (further-on denoted as GeON) with combined beams of Ge and/or O atoms [7] immediately followed by the HfO₂ layer. Similar to MOCVD, the interfacial passivating layer has a dramatic effect on improving the *C/V* characteristics, see Fig. 2, from which an interfacial state density $N_{it} \sim 5 \times 10^{12} \text{ cm}^{-2}$ can be extracted. Switching the device from inversion to accumulation and back to inversion shows a flat-band voltage shift of about 200–250 mV (measurement frequency is 100 kHz). This hysteresis effect could be due to trapping/de-trapping defects in either the HfO₂ bulk or at the interface. In this respect, it is interesting to note that TOFSIMS shows quite some Ge in this HfO₂ layer, in contrast to HfO₂ deposited straight on bare Ge after an oxide desorption step (Fig. 7, middle panel). We speculate the presence of Hf in the interfacial layer is due to the unstable nature of the GeON layer which “dissolves” in the growing HfO₂ layer. From XPS (Fig. 5, lower panel, Ge 3p spectrum) we know this GeON layer is not GeO₂ but some form of Ge sub-oxide (Ge¹⁺, Ge²⁺, Ge³⁺), the exact oxidation state of which however is difficult to determine on the 3p peak. The 3d peak would be better suited for this, if it were not for an interfering overlap between the Ge 3d and the Hf 4f from the HfO₂ overlayer.

In another experiment, we deposited a tiny amount of HfO₂ (0.3–1.5 nm) by means of either ALD or MOCVD on Ge and annealed these in O₂ at 550 °C for 15 min [8]. Careful AR-XPS showed a homogeneous, mixed oxide layer and a Ge⁴⁺ peak shifting in the direction of sub-oxides. A similar shift of the Hf peak shows that this newly formed bond is a Hf germanate (Hf–O–Ge). *C/V* and *I/V* measurements on these layers showed high leakage and – more important – an interfacial density of states that increases with the amount of Hf–O–Ge bonds.

4. Discussion

A first striking feature is the different behavior of HfO₂ layers deposited with different techniques on a bare Ge surface. The abrupt interface in the MBD case (x-TEM) probably involves a high density of dangling bonds because of the mismatch between the two crystalline structures of Ge and HfO₂ (if not epitaxial), which could explain the poor *C/V*. In the case of MOCVD on HF-dipped Ge, the formation of a metallic-like germanide observed in MEIS can cause Fermi level pinning, so hindering modulation of the surface potential. None of these fatal errors occurs in the ALD case, which consequently results in decent *C/V*. It is worthwhile to mention here that the interface in the ALD case is characterized as GeO₂ by XPS.

By inserting a “passivation” layer between the HfO₂ and the Ge, these fatal errors can be prevented: for MOCVD, the GeON layer made by NH₃ anneal prevents the formation of Hf germanide (as indicated by MEIS measurements), whereas the in situ deposited amorphous GeON layer in MBD prevents the formation of HfO₂ crystalline phases in direct contact with the Ge crystalline phase, which helps saturating any dangling bonds. Electron spin resonance [9] was not able to show significant ($\leq 10^{12}$ cm⁻²) densities of dangling bond type of defects in GeO_x interfaces made in various ways.

An interesting observation is that the mere presence of Ge inside the high-k layer is not really a killer problem: MBD HfO₂ on GeON/Ge clearly shows an important Hf trace inside the HfO₂, but still gives reasonable *C/V* albeit with some bumps. As mentioned, we noticed that Hf–O–Ge (germanate) bonds which were formed in trace amounts of HfO₂ on a thermal GeO₂ layer after a long 550 °C anneal in O₂ also induce bumps in the *C/V* curve, the magnitude of which scales with the amount of Hf. So, it is not unlikely that the presence of Ge inside the HfO₂ can create an enhanced interfacial state density depending on the thermal history. Additionally, the presence of Ge inside the HfO₂ film could be related to the observed hysteresis, but this statement is rather speculative at this time. Finally, although we have not been able to index unambiguously the different Ge oxidation states underneath HfO₂, it appears that also the lower oxidation states of Ge, i.e. the sub-oxides, are able to generate bumps in the *C/V* curves.

5. Conclusion

We have studied MOS capacitors made in HfO₂ on Ge as a materials model system made with several different deposition

techniques. By a close inspection of physical characterization results of the interfacial layers, we have found indications for some different root causes of failing gate stacks, all for HfO₂ layers made directly on pristine Ge surfaces. By introducing either in situ or ex situ a passivation layer that screens the HfO₂ from the Ge, the *C/V* and *I/V* results can be strongly improved. We find that a GeON-like interfacial layer is quite effective in passivating the Ge surface, contrary to what originally was believed. Still, the best interfacial state densities we have seen so far in the HfO₂/Ge system is around 5×10^{12} cm⁻², which needs further improvement. We speculate that a better control of the interface, especially the Ge oxidation state which should be as much as possible Ge⁴⁺, could be very important here.

Whereas there is probably no such thing as the ultimate high-k material, from this HfO₂/Ge model material system it may be clear, that the Ge surface preparation and the high-k material deposition technique must be matched to each other, and cannot be combined arbitrarily.

Acknowledgements

The authors are indebted to the European Commission for financial support in the frame of the ET4US project. We acknowledge useful discussions with Tejas Krishnamohan (Stanford University).

References

- [1] K. Prabhakaran, T. Ogino, Surf. Sci. 325 (1995) 263.
- [2] C.O. Chui, H. Kim, P.C. McIntyre, K.C. Saraswat, Tech. Dig. Int. Electron Devices Meet. 2003 437 (2003).
- [3] N. Wu, Q. Zhang, C. Zhu, D.S.H. Chan, M.F. Li, N. Balasubramanian, A. Chin, D.-L. Kwong, Appl. Phys. Lett. 85 (2004) 4127.
- [4] B. De Jaeger, R. Bonzom, F. Leys, O. Richard, J. Van Steenberg, G. Windrickx, E. Van Moorhem, G. Raskin, F. Letertre, T. Billon, M. Meuris, M. Heyns, Microelectr. Eng. 80 (2005) 26.
- [5] D. Schmeisser, R.D. Schnell, A. Bogen, F.J. Himpsel, D. Rieger, G. Landgren, J.F. Morar, Surf. Sci. 172 (1986) 455.
- [6] M. Houssa, B. De Jaeger, A. Delabie, S. Van Elshocht, V. Afanasiev, J. Autran, A. Stesmans, M. Meuris, M. Heyns, J. Non-Crystal Sol. 351 (2005) 1902.
- [7] A. Dimoulas, G. Mavrou, G. Vellianitis, E. Evangelou, N. Boukos, M. Houssa, M. Caymax, Appl. Phys. Lett. 86 (2005) 32908.
- [8] S. Van Elshocht, M. Caymax, T. Conard, S. De Gendt, I. Hoflijck, M. Houssa, B. De Jaeger, J. Van Steenberg, M. Heyns, M. Meuris, Appl. Phys. Lett. 88 (2006) 141904.
- [9] V.V. Afanas'ev, Y.G. Fedorenko, A. Stesmans, Appl. Phys. Lett. 87 (2005) 032107.