



Hierarchical electrospun nanofibers treated by solvent vapor annealing as air filtration mat for high-efficiency PM2.5 capture

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ABSTRACT Recently, air pollution has become more serious and started to have a dramatic effect on the health of humans in many large cities. Generally, outdoor personal protection, such as commercial masks, cannot effectively prevent the inhalation of many pollutants. Particulate matter (PM) pollutants are a particularly serious threat to human health. Here we introduce a new efficient air filtration mat that can be used for outdoor protection. The new efficient air filter's nanocomposite materials were successfully fabricated from poly(ϵ -caprolactone)/polyethylene oxide (PCL/PEO) using an electrospinning technique and solvent vapor annealing (SVA). SVA treatment endows the wrinkled fiber surface and enhances the PM2.5 capture capacity of protective masks. This nanowrinkled air filtration mat can effectively filter PM2.5 with a removal efficiency of 80.01% under seriously polluted conditions (PM2.5 particle concentration above 225 mg m⁻³). Our field test in Qinhuangdao indicated that the air filtration mat had a high PM2.5 removal efficiency under thick haze. Compared to commercial masks, the fabricated SVA-treated PCL/PEO air filter mat demonstrated a simpler and eco-friendly preparation process with excellent degradation characteristics, showing wide potential applications with a high filtration efficiency.

Keywords: electrospinning nanofiber mat, solvent vapor annealing, air filtration, PM2.5 removal

INTRODUCTION

In the last several years, environmental issues have drawn extensive attention owing to the dangerous nature of their impact on human health and the environment [1–4]. Particulate matter (PM) is a very complex mixture of

pollutants with extremely fine particles and small liquid droplets [5–8]. PM2.5 is particularly harmful because these small particles can penetrate the lungs and the bronchi [9–13]. In indoor environments, these particles can be filtered out by ventilation or central air-conditioning systems, whereas the individual protection used by people outdoors is not effective because the PM2.5 removal efficiency of most commercial masks is low because of the micro-sized diameter of the particles. Thus, the efficient and convenient removal of these PM pollutants from the air has become an urgent and challenging concern for researchers [14]. For instance, the design and development of efficient filtration materials that can remove PM2.5 particles with high recyclability are a central focus. It is well documented that filtration fibers with high porosity, high accumulated density, and small fiber diameters exhibit a higher PM2.5 removal efficiency [15].

It is well known that electrospinning is a relatively simple method of producing continuous fibers with nano- and submicron-sized diameters [16]. The fibers prepared by electrospinning exhibit high porosity, fine pore size, staggered pore structure, small pore size, and controllable diverse architecture and thickness, which makes them ideal candidates for air filtration materials [17–20]. Based on these properties, many types of electrospun fibrous membranes, including polyurethane (PU), polystyrene, polyvinyl alcohol, polyvinyl acetate, Nylon 6 nanofibers, and polyacrylonitrile (PAN) nanofibers, have been successfully fabricated and used as air filtration media [21–27]. However, these nanofibers usually have low breathability and low recyclability. Cui

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et al. [15] reported on a transparent air filter prepared by electrospinning with high-efficiency PM_{2.5} capture. Pan *et al.* [28] designed and prepared porous bead-on-string polylactic acid (PLA) fibrous mat-type materials by electrospinning for PM filtration. Some research has illustrated that the morphological characteristics of fibers have a significant impact on the fibers' filtration performance, such as their diameter, surface structure, packing density, and volume fraction [29–31]. The above-mentioned nanofibers are generally composed of single elements with mostly smooth surface structures, circular cross sections, and very small diameters, which is beneficial for efficiently capturing PM_{2.5} [32–34]. However, multicomponent polymer nanofibers generally show stronger controllability than single-component polymer nanofibers in terms of strength, recyclability, and specific surface area [35]. For example, PAN, polyvinyl chloride (PVC), PU, polyethylene terephthalate nanofiber membranes enabled with PU, fly ash particles, and cyclodextrin, respectively, improved the mechanical properties primarily because of various intermolecular weak interactions [22,36–38]. At the same time, nanofibers exhibited high filtration efficiency and were found to be potentially suitable for the removal of certain air pollutants, along with biological contaminants owing to their high specific surface area [39]. Thus, multicomponent polymers are preferred in electrospinning to prepare air filter mats owing to their excellent chemical and thermal stability of synthetic polymers along with controllable fluid properties and surface structure. Researchers often focused on the key point of improving the air filtration efficiency by adjusting the thickness of the nanofibers and neglected the more important key point of regulating the surface structure of nanofibers [40,41].

In order to overcome the challenges that were previously described, we designed and successfully prepared electrospun poly(ϵ -caprolactone)/polyethylene oxide (named PCL/PEO) nanofiber mat composite membrane materials, which were then treated with solvent vapor annealing (SVA) to form nanowrinkled nanostructures to remove PM_{2.5}. The selection of the particular PCL/PEO material used in this study was based on several considerations. First, PCL can interact with acetone solvents and its presence is a prerequisite for the preparation of wrinkled structures in PCL/PEO fibers. Secondly, PCL is an ecofriendly and inexpensive material that is easy to obtain. In this study, the motivation behind SVA treatment of electrospun mats originated primarily from the modification and regulation of micro/nanos-

tructures of membranes for potential incremental improvement of filter efficiency. In the current SVA process that is used to fabricate PCL/PEO materials, acetone, as a poor solvent for PCL/PEO nanofibers, can delocalize the amorphous chains very efficiently and allow them to be redeposited onto the preexisting crystal edges, growing nanowrinkled nanostructures on the surface of the PCL/PEO nanofibers [42–45]. The fabricated SVA-treated air filtration mat demonstrated a high PM_{2.5} removal efficiency due to nanowrinkled nanostructures, low resistance to air flow, and light weight. The PM_{2.5} removal efficiency of the nanofiber mat reached 80.01% by adjusting the surface structure and fiber thickness, and the PM_{2.5} removal efficiency of commercial masks was thus increased from 7% to nearly 80% under extremely hazardous air quality conditions.

EXPERIMENTAL SECTION

Materials

The experimental materials, PCL ($M_n = 80 \text{ kg mol}^{-1}$; Sigma-Aldrich (Shanghai) Trading Co. Ltd.), PEO ($M_n = 600 \text{ kg mol}^{-1}$; Sigma-Aldrich (Shanghai) Trading Co. Ltd.), trichloromethane (analytical reagent grade; Fisher Scientific, Waltham), and acetone (analytical reagent grade; Fisher Scientific), were used directly without further purification.

Nanocomposites fabrication

PCL/PEO nanofiber mats were fabricated *via* an improved electrospinning method. In brief, 0.21 g of PEO powder and 0.5 g of PCL pellets were dissolved in 6.3 mL of chloroform solution and then stirred for 24 h at room temperature to obtain a uniform and transparent spinning precursor solution [45]. Next, the solution was allowed to stand for 1 h to drive the air bubbles out of the solution. After that, the uniform solution was transferred into a syringe. We chose three typical electrospun fiber samples with the following parameters (needle tip voltage, working distance, feed rate of solutions): 25 kV, 25 cm, 0.3 mL h⁻¹ (Sample 1), 15 kV, 25 cm, 1 mL h⁻¹ (Sample 2), and 15 kV, 15 cm, 1 mL h⁻¹ (Sample 3). The environmental parameters included humidity of 18% and a temperature of 15°C during the electrospinning process. About 300 mL of acetone was poured into a 5 L desiccator, which was maintained at a constant temperature of 22–23°C. The samples were put on a ceramic plate inside the desiccator and fumigated in 100% acetone vapor [46–48]. At the same time, the samples were fumigated in acetone vapor inside the desiccator for 5

days. After 5 days, the samples were taken out of the desiccator, dried under vacuum for 24 h at room temperature, and evaluated for further characterization.

Batch filter tests for PM_{2.5} removal

For the filtration experiment, the filter properties of the fabricated nanocomposite materials were investigated, especially the PM_{2.5} removal efficiency. In order to find the most efficient air filter materials, commercial masks made of different materials and electrospun fiber mat samples from different production methods were investigated for PM_{2.5} capture at room temperature. In this study, PM_{2.5} was generated by burning moxa stick incense for a period of time in a confined room. The burning smoke contained PM_{2.5} above 270 $\mu\text{g m}^{-3}$, and the smoke had a very complex composition that included CO, CO₂, and certain volatile organic compounds, such as benzene, alcohols, amines, phenols, and long-chain hydrocarbons [49–53]. This smoke had a composition similar to that of common haze, so it can be used as a model to simulate haze in the laboratory [54,55]. The current environmental PM_{2.5} concentration was recorded, and then the nanofiber mat was fixed with a rubber band on the PM_{2.5} mask filtration efficiency measuring device, with count data collected every 5 s with a total test time of 90 s. To ensure the accuracy of the results, each sample was tested 5 times and averaged. For other parameters of the fiber mats and commercial masks of different materials, similar experimental procedures were adopted. The change in PM_{2.5} concentrations was evaluated by measuring the PM_{2.5} mask filtration efficiency (Inovafitness Electronic Technology Co., Ltd.) at room temperature using an air flow of 600 L min⁻¹. For the recycling experiments, the filtered fiber mats were washed thoroughly with deionized water and ethanol 4–5 times and were found to maintain a relatively high filtering efficiency [56].

Characterization

The microstructure of all the obtained composite materials was characterized using field-emission scanning electron microscopy (FE-SEM; Hitachi S-4800-II) with 5–15 kV accelerating voltage and transmission electron microscopy (TEM; Hitachi HT7700, High-Technologies Corp.). X-ray photoelectron spectroscopy (XPS) was performed with a Thermo Fisher Scientific ESCALab 250Xi with an Al-K α source, and X-ray diffraction (XRD) characterization was accomplished with an X-ray diffractometer equipped with a CuK α X-ray radiation source and a Bragg diffraction setup (SmartLab, Rigaku). The

thermal stability of the samples was measured on a simultaneous thermal analyzer using thermogravimetry-differential scanning calorimetry (TG-DSC) in a nitrogen atmosphere (Netzsch STA 409 PC Luxx, Germany). The change in particle concentration was measured using a PM_{2.5} mask filtration efficiency measuring device (Inovafitness Electronic Technology Co., Ltd.).

RESULTS AND DISCUSSION

Structural characterization of the air filtration mats

Fig. 1 presents the steps of the entire experimental procedure, including the preparation of the precursor solutions, the electrospinning process, the SVA treatment, and the final PM_{2.5} removal test. Simply stated, the PCL/PEO nanofibers were obtained by electrospinning, and the surface of the fibers was wrinkled by SVA treatment. Finally, the treated fibers were subjected to PM_{2.5} filtration experiments. The morphologies and nanostructures of all the prepared composite materials before and after SVA treatment were investigated using SEM.

A typical SEM image of the nanowrinkled air filtration mat fabricated from a PCL/PEO chloroform solution is presented in Fig. 2, which is similar to examples from previous reports [57,58]. The image in Fig. 2a distinctly demonstrates that the electrospun mat shows a uniform and smooth fiber with a fiber diameter distribution concentrated at 1–1.5 μm (Fig. 2a'). The fibers' surface exhibits uniform wrinkled nanostructures (Fig. 2b) with an increased specific surface area when the electrospun fiber mat was treated *via* the SVA method [59–61]. Additionally, the diameter of the SVA-treated fibers shows the range of 3–5 μm (Fig. 2b'), indicating that the nanostructures of the fibers changed clearly with the SVA-treated process.

Electrospinning liquid jets generate considerable stretching forces on nanofibers of about 2 μm diameter, which consist of dislocated sheets and loose amorphous chains. The primary PCL/PEO fibers were dried at 25°C for 24 h under vacuum and then aged at 20–25°C for 48 h prior to the SVA test. The used aged temperature must be lower than the PCL/PEO crystallization temperature to keep organized fiber nanostructures; thus, the primary fibers were aged at 20–25°C for further crystallization [62–64]. The PCL/PEO fibers that were SVA-treated exhibited clear lamella-like petals. The acetone vapor appeared to first selectively diffuse into the amorphous region because the chain packing in the crystallization zone was loose in the amorphous region, and many lamella-like petals grew on the annealed PCL/PEO

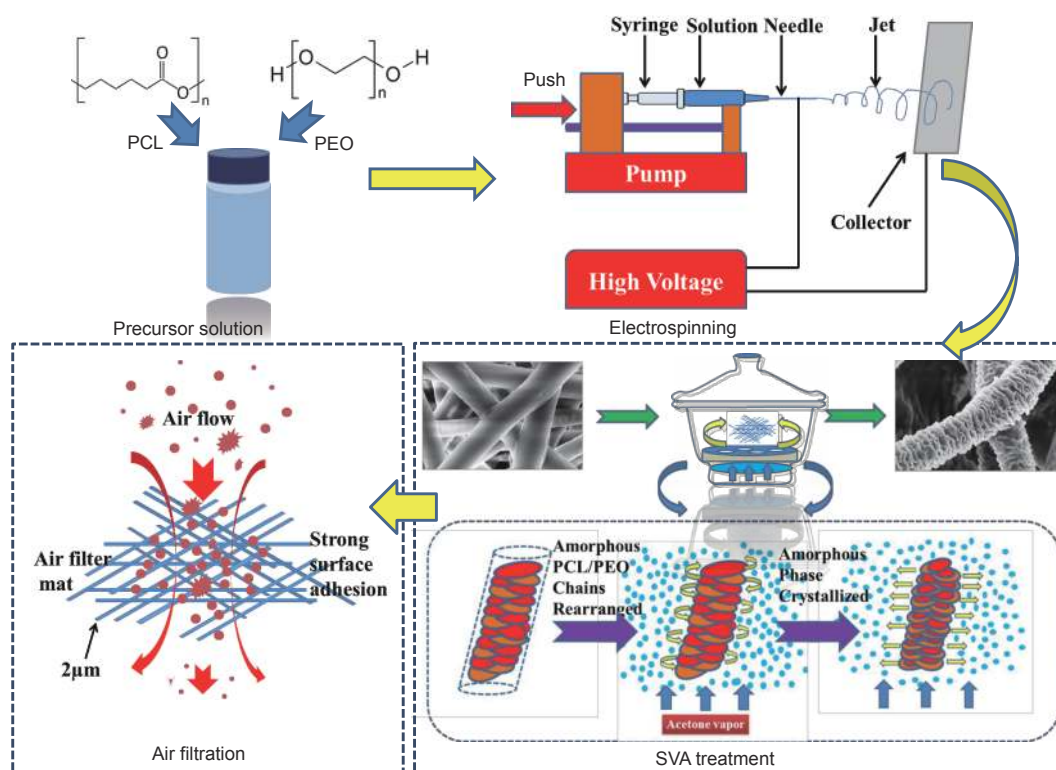


Figure 1 A schematic illustration of the preparation and filtration of the obtained electrospun nanowrinkled air filtration mat.

nanofibers due to those amorphous chains with low availability and low mobility, which can cause rearrangement during annealing [45]. First, the acetone vapor selectively mobilized kinetically stagnant amorphous chains as the liquid jet quickly dried. Then, these amorphous chains crystallized at the edges of preexisting crystals with the action of acetone. Finally, a wrinkled structure with a high density grew on the fiber surface.

SVA processing time and temperature were found to be extremely important factors affecting fiber morphology. SEM images of the PCL/PEO fibers were obtained for different SVA treatment durations at room temperature in ceramic desiccators. Images of Sample 1 dried for 24 h under vacuum and aged 48 h after different durations of annealing [65–67], shown in Fig. 3a–f, shows diameters of about 2 μm . The prepared electrospun PCL/PEO nanofibers with a diameter of approximately 2 μm shows a smooth surface in Fig. 3a. Compared to the fibers treated with SVA, a slight wrinkle-like structure appeared after one day of annealing (Fig. 3b). By comparison, PCL/PEO fibers undergoing 2 days (Fig. 3c) of SVA treatment generated secondary hierarchical nanostructures that were more pronounced and denser, and the PCL/PEO

fibers undergoing 3 days of SVA treatment exhibited a deep and an uneven layered structure (Fig. 3d). Compared with Fig. 3d, Fig. 3e illustrates a deeper and uniform layered structure. With 5 days of SVA treatment, clear lamella-like petal structures appear on the fiber surface, uniform and dense, indicating that longer aging time results in more perfect crystal growth (Fig. 3f).

In addition, the PCL/PEO fibers (Sample 1) prepared for different SVA treatment durations were characterized by XRD, which was important for self-assembled composite materials [68–75]. The XRD pattern of PCL (Fig. 4a) shows characteristic diffraction peaks at 21.4° and 23.7°. The PEO diffraction pattern exhibits two intense characteristic peaks located at 19.3° and 23.6°, whereas PCL/PEO components show three main peaks at 19.3°, 21.4°, and 23.6°, which indicate that the PCL/PEO components contain both PCL and PEO. Fig. 4b shows the diffraction patterns of the prepared air filtration mat with different SVA treatment durations. The characteristic peak located at 21.4° shows a sharper tendency with the increment of SVA time, which indicates a change in the crystallinity of PCL/PEO composites. This is a better proof of the structural phenomenon in Fig. 3. In addition,

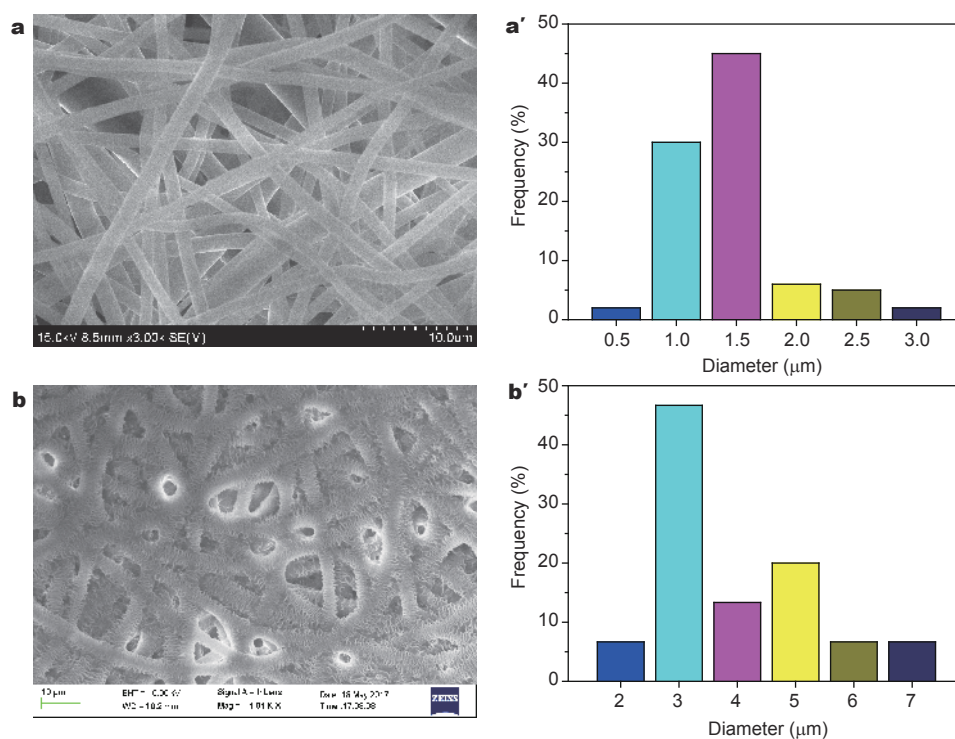


Figure 2 (a) SEM images and fiber diameter size distributions of the prepared PCL/PEO nanofibers mat (Sample 1) by electrospinning (a and a') and subsequent by SVA treatment for 5 day (b and b').

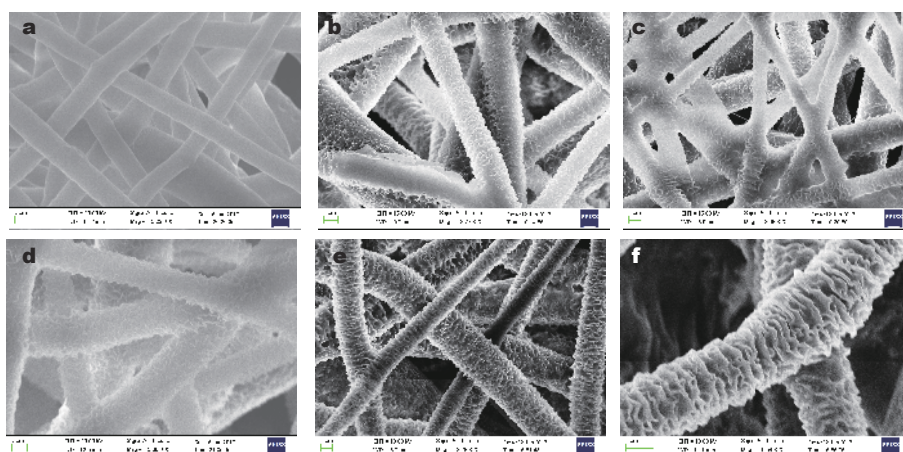


Figure 3 SEM images of the prepared electrospun PCL/PEO nanofibers (a, Sample 1) and SVA treatment at different time intervals: (b), one day; (c), 2 days; (d), 3 days; (e), 4 days; (f), 5 days.

Fig. 5 demonstrates the thermograms of PCL/PEO primary fibers and PCL/PEO fibers with 5 days of SVA treatment (Sample 1_{SVA}). More specifically, both samples began to decompose from 280°C and stabilized after 450°C, which may be the result of various oxygenated chemical groups and alkyl chain pyrolysis in high-temperature environments. The quality retention rates

demonstrate the values of 8.97% and 4.27% for PCL/PEO nanofiber mats before and after SVA treatment. According to the above TG results, the SVA-treated PCL/PEO fibers show a slight decrease in the thermal stability compared to PCL/PEO fibers because of the surface lamella-like petal structures that originate from more uniform crystallinity and nanostructures of high regular-

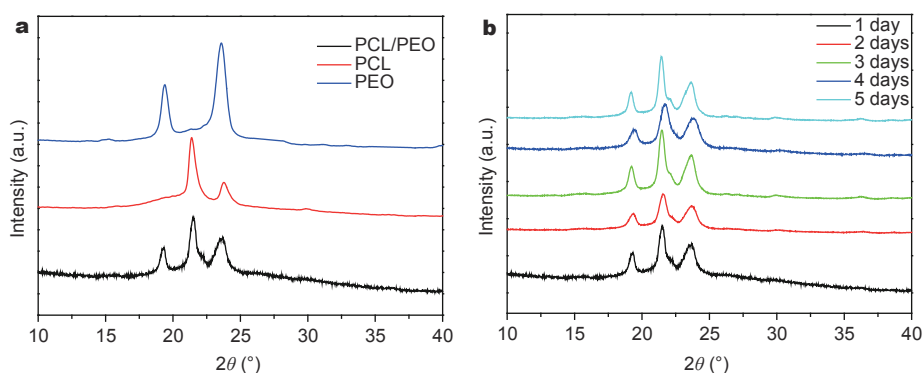


Figure 4 XRD patterns of the prepared PCL/PEO nanofibers (a) and with different subsequent SVA treatment time intervals (b).

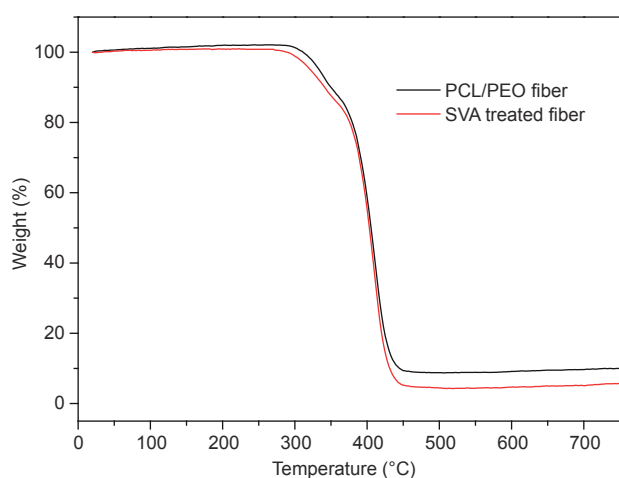


Figure 5 TG curves and infrared spectra of the PCL/PEO nanofiber mat by electrospinning and subsequent SVA treatment time of 5 days.

ity after SVA treatment [45,46]. It is important to note that PCL crystallite growth is apparently restricted by the PEO phase with a rigid and more densely packed semicrystalline structure. In this case, the spherulitic growth of PCL is suppressed because of the spatial confinement of the semicrystalline PEO phase. Meanwhile, thicker fibers favor the spherulitic growth of PCL, resulting in a PCL phase that can crystallize simultaneously throughout the surfaces of electrospun fibers [57].

PM_{2.5} removal performances of air filtration mats

The filtration properties of the fabricated PCL/PEO mats were examined in the PM_{2.5} filtration experiments. In this experiment, PM_{2.5} is generated by burning moxa stick incense. The burning moxa stick contains PM_{2.5} smoke with CO, CO₂, and many volatile organic compounds that are similar to PM_{2.5} in real haze that forms over big cities. The filtration experiments were

performed on the PCL/PEO air filtration mats, and the results, test equipment, and the test methods are presented in Fig. 6. In this study, the air filtration mat performance was evaluated using a PM_{2.5} mask filtration efficiency measuring device that consists of four parts: an air inlet, a measuring cavity, a fan, and an air outlet (Fig. 6a). When the fan switch is turned on, the air flow containing PM_{2.5} enters the measuring cavity at a flow rate of 600 L min⁻¹. At the beginning of the experiment, the real-time PM_{2.5} particle concentrations could be recorded by the sensors in the measuring cavity showing a stable number. Then, the test sample was sealed with a rubber band on the top of the measuring cavity immediately (Fig. 6c), and the particle density was measured and recorded every 5 s over a total test time of 3 min. To ensure the accuracy of the experimental results, each experiment was repeated 5 times and averaged. The removal of PM_{2.5} was related to the volume of the spinning precursor solution. To determine the volume, Sample 1 was selected with different precursor solution volumes to test the filter performance. Fig. 6d shows the filtration curves of the PCL/PEO air filtration mat with precursor solution volumes of 2, 2.5, 3, 3.5, and 4 mL, respectively. In order to evaluate the air filtration mat before SVA treatment, the PM_{2.5} removal rates were obtained for several different volumes: 57.31% removal at 2 mL, 65% removal at 2.5 mL, 68.5% removal at 3 mL, 67.43% removal at 3.5 mL, and 67.31% removal at 4 mL. After SVA treatment, the values were 68.89% at 2 mL, 69.01% at 2.5 mL, 71.37% at 3 mL, 70.72% at 3.5 mL, and 69.53% at 4 mL (Fig. 6d). The optimum volume was found to be 3 mL, which had sufficient air penetration and resulted in the best PM_{2.5} removal rate. Therefore, the volume of 3 mL was chosen for follow-up experiments. The PM_{2.5} removal efficiencies of the air filtration mat with different volumes and the comparison

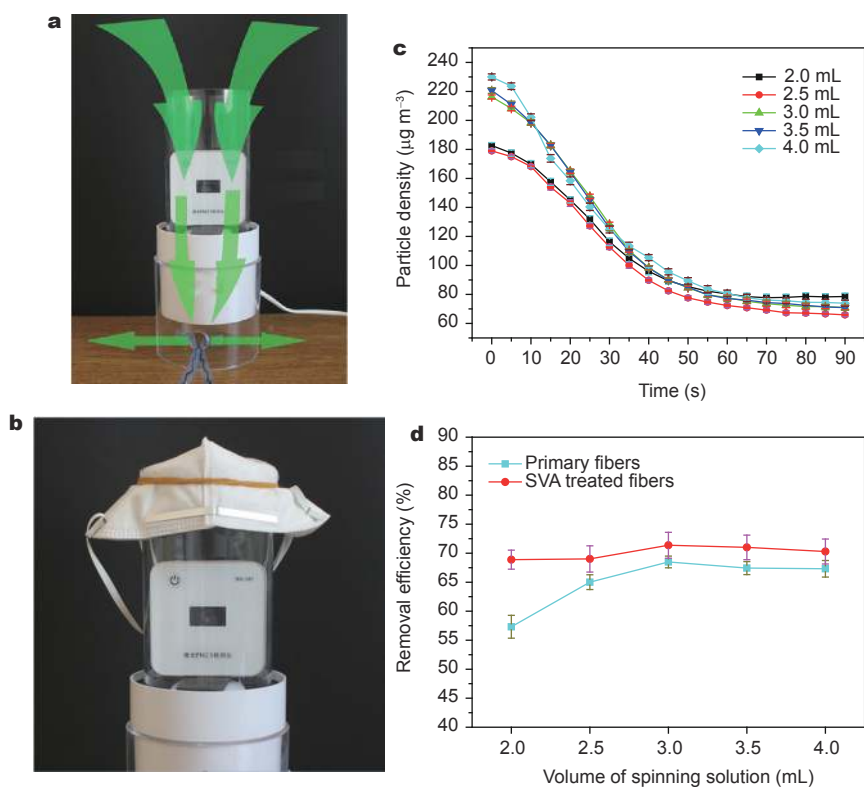


Figure 6 Schematics of the air filtration mat that captured PM_{2.5} by air flow (a) and working state with mask filtration (b); (c) PM_{2.5} filtration curves for the air filtration mat (Sample 1) with different volumes of spinning precursor solution; (d) PM_{2.5} removal efficiency plots for primary fibers and SVA-treated fibers.

of PM_{2.5} removal efficiencies of the air filtration mat before and after SVA treatment are presented in Fig. 6d. Another important factor that affects the PM_{2.5} removal performance is the surface structure of the fiber. Compared to the primary fibers, the SVA-treated fibers with the same parameters of 15 kV, 25 cm, and 1 mL h⁻¹ and the same volume of 3 mL show significantly increased PM_{2.5} removal efficiencies by about 6.37% with increasing the specific surface area.

Besides the volume of the spinning precursor solution and the surface structure of the fiber, another important parameter for the air filtration mat is the filter's dimension, which also affects the PM_{2.5} removal efficiency [76–78]. In this study, we used an air filtration mat with two other parameters (Sample 2 and Sample 3) to test the PM_{2.5} removal efficiency. The average fiber diameters of Sample 2 and Sample 3 are of 2.5 ± 0.4 and 4 ± 0.5 μm, respectively, as shown in Fig. 7a' and c'. As shown in Fig. 7b' and d', the average fiber diameters of Sample 2_{SVA} (Sample 2 with 5 days of SVA treatment) and Sample 3_{SVA} (Sample 3 with 5 days of SVA treatment) are, respectively, of 6 ± 1.1 and 4.5 ± 1.2 μm, indicating that

the fiber diameter slightly increased and became relatively uniform after SVA treatment.

The PM_{2.5} filtration column chart of different air filtration mats is shown in Fig. 8. PM_{2.5} removal efficiencies reach 77.92% and 77.97% for Sample 2 and Sample 3, respectively. In the controlled experiment of air filtration nanocomposite mats with different fiber diameters, the removal performances of PM_{2.5} were significantly different. Sample 3 with 15 kV, 15 cm, and 1 mL h⁻¹ had the characteristics of uneven diameter, high porosity, and high packing density, so as to obtain a higher PM_{2.5} removal efficiency. For the SVA-treated fibers shown in Fig. 8a, Sample 2_{SVA} had a PM_{2.5} removal efficiency of 77.14% and Sample 3_{SVA} had a PM_{2.5} removal efficiency of 80.01%. Therefore, the PM_{2.5} removal efficiency of Sample 3_{SVA} was optimal, corresponding to the experimental results shown in Fig. 8a. Sample 3_{SVA} has the highest PM_{2.5} removal rate, since crude fibers serve as a kind of framework, playing a supporting role in the deposition of finer fibers in the fiber membrane, increasing the looseness between fibers, and improving the porosity of the fiber membrane [47].

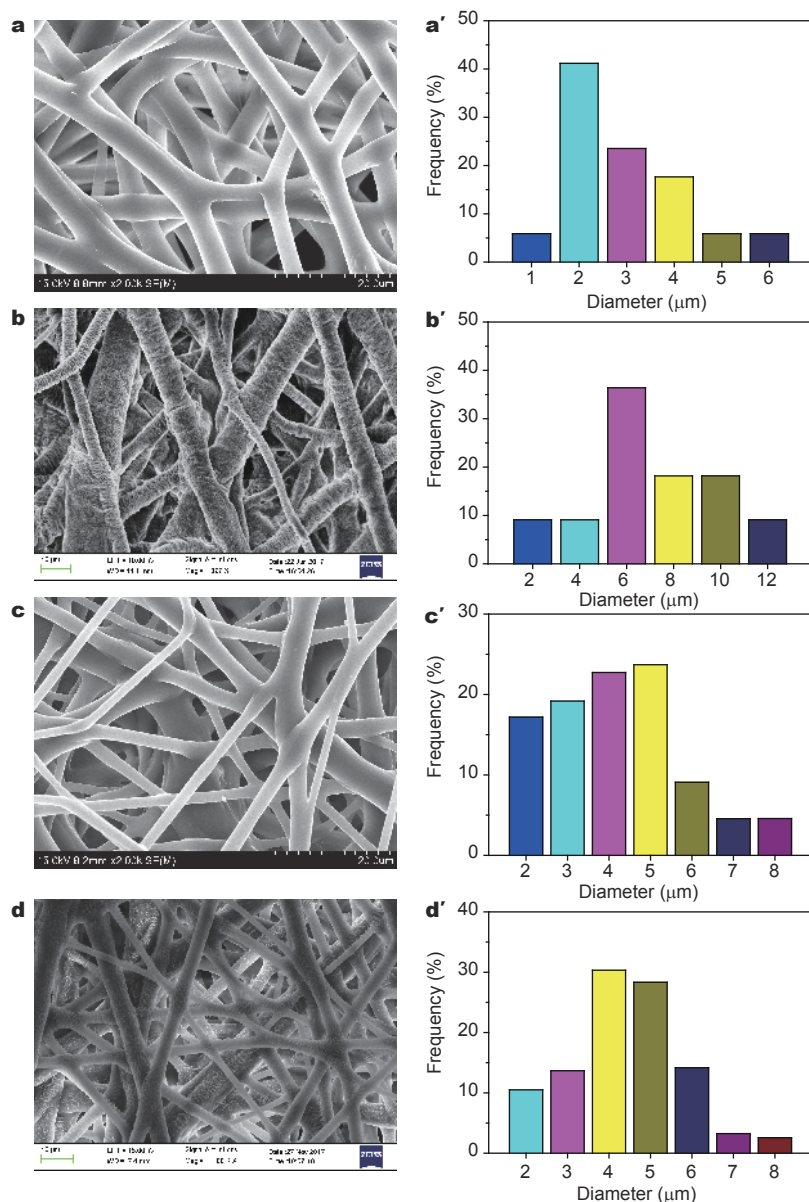


Figure 7 SEM images and fiber diameter size distributions of the prepared PCL/PEO nanofiber mat by electrospinning (Sample 2, a and a'; Sample 3, c and c') and subsequently with SVA treatment at a time of 5 days (Sample 2, b and b'; Sample 3, d and d').

In addition, in order to study the efficiency of our air filtration mat (Sample 3_{SVA}) on a truly contaminated hazy day, we conducted a field test on 29 December 2017 in Qinhuangdao, China. The concentration of PM_{2.5} on that day was 250 μg m⁻³, which is considered serious air pollution. The Sample 3_{SVA} air filtration mat achieved a PM_{2.5} removal efficiency of 73.75%. An error bar represents the standard deviation of four replicate measurements. In this experiment, by adjusting the spinning parameters of the air filtration mat, the PM_{2.5}

removal efficiency was increased by about 9%. For Sample 2 and Sample 3, with SVA treatment, the PM_{2.5} removal efficiencies increased by 2.63% and 2.09%, respectively. As a result, we designed and successfully prepared a high-efficiency air filtration mat using three steps: quantitative precursor spinning solution with 3 mL, suitable electrostatic spinning parameters with 15 kV, 15 cm, and 1 mL h⁻¹, and SVA treatment under appropriate conditions.

To further compare the PM_{2.5} filter performance of the

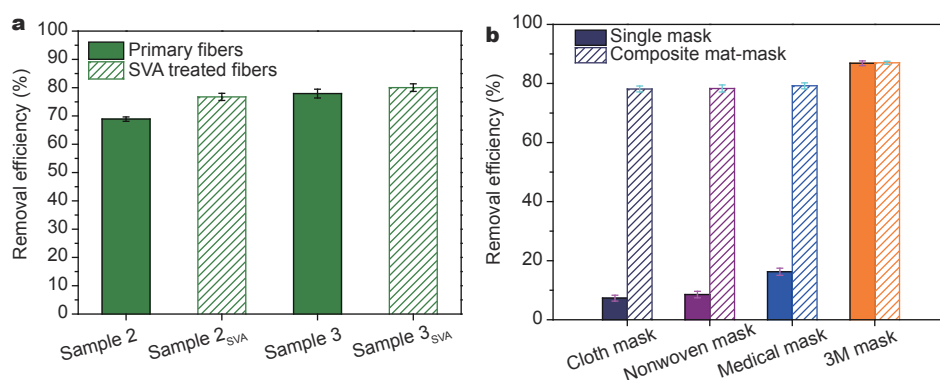


Figure 8 (a) Comparison chart of PM_{2.5} removal efficiency of Sample 2 and Sample 3 with/without SVA treated fibers. (b) Comparison chart of PM_{2.5} removal efficiency of Sample 3 deposited on different commercial masks.

air filter mat, we chose and investigated commercial masks with different texture materials, as seen in Fig. 8b. The results indicated that the PM_{2.5} filter performance of the cloth mask was 7.32%, with values of 8.53%, 16.26%, and 86.85% for the nonwoven mask, medical mask, and the 3M mask, respectively. It could be easily observed that most of the masks, except for the 3M mask, have relatively low PM_{2.5} removal efficiencies and fail to trap PM_{2.5} and protect human life [79]. To meet this challenge, we deposited the prepared air filter mat onto the surface of commercial masks and then tested the filter performance. It should be noted that, after compositing with the air filtration mat, the PM_{2.5} removal efficiency of commercial masks increased significantly (78.14% for cloth masks, 78.31% for nonwoven masks, 79.22% for medical masks, and 87% for 3M masks). In addition, it should be further noted that the SVA-treated PCL/PEO air filter mat showed clear advantages, such as a simple and ecofriendly preparation process, easy degradation nature, controllable nanostructures, low material cost, and wide combinations with different mask substrates, demonstrating potential applications with high filtration efficiency and wide fields compared with the commercial 3M masks. Hence, air filtration mat-anchored masks have shown improved filtration performance. Furthermore, composite materials that were reported in previous work are shown in Table 1, which clearly demonstrated various characteristics of different air filtration mats in the filtration process [24,27,28,36,37,80–82]. For example, Wang *et al.* [28] prepared porous bead-on-string PLA nanofibers by electrospinning, and the obtained nanofibrous membranes exhibited an excellent filtration efficiency with a low pressure drop. Our present study shows significant advantages in the PM_{2.5} removal efficiency, good reusability, and stability. It is expected that, due to

the SVA treatment, the 3D nanowrinkled structure can provide new idea for respiratory protection, air purification, and 3D nanomaterial design.

To further study the ability of air filtration mats to capture PM_{2.5}, we investigated the composition and surface chemistry of PM_{2.5}. The air filtration mat and mask with fibers were sealed in the instrument, which was exposed to outdoor haze air (PM_{2.5} mass concentration larger than 225 mg m⁻³), and the fan was turned on to let the air containing PM_{2.5} pass through for 24 h, trapping PM_{2.5}. Masks with respective capture of PM_{2.5} are shown in Fig. 9a. The surface of the mask with fibers becomes dirty after capturing PM_{2.5}. In order to further study PM_{2.5}, we studied the capture of PM_{2.5} by fibers. From the SEM image, we can see that PM_{2.5} particles are irregular in shape and easily adhere to and agglomerate on the fiber surface (Fig. 9b). TEM images of PM_{2.5} captured by fibers are shown in Fig. 9c. PM_{2.5} aggregates have sizes of around 700 nm. Energy-dispersive spectrometry (EDS) shows that the PM_{2.5} particles contain more complex components, mainly C and O elements and a little Ca, Fe, and other metals (Fig. 9b).

To further study the composition of PM_{2.5}, an XPS analysis was performed on Sample 1 before (Fig. 10a and b) and after (Fig. 10c and d) the filtration of haze. We exposed Sample 1 to outdoor hazy air for 48 h and then carried out an XPS test of PM_{2.5}. Fig. 10a conveys that the C 1s signal mainly is located at 284.8, 285.5, 286.5, and 288.7 eV, corresponding to C–OH, C–O, C=O, and O=C–O bonds, respectively. The XPS spectra with O1s survey data from Sample 1 before filtration of haze are shown in Fig. 10b, which clarifies the position of five peaks at 531.7, 532.7, and 533.5 eV that represent the bonds of C=O, C–O, and –O–H, respectively. In addition, Sample 1, after the filtration of haze, shows that the C 1s

Table 1 Comparative characteristics and filtration performance of different materials reported in the literatures

No.	Materials	Fiber diameter (nm)	Filter media	Advantages	Disadvantage	Ref.
1	Polyacrylonitrile (PAN)	200	Incense smoke	Transparent, environmentally high stability and highly effective removal of PM2.5	Low reusability, easy to aging, low strength, poor wear resistance	[27]
2	Polylactic acid (PLA)	150–300	NaCl aerosol	Controllable morphology, excellent filtration efficiency and a low pressure drop	Poor stability, poor wear resistance and easy to aging	[28]
3	Nylon6 (PA-6)	150	NaCl aerosol	Good stability, good air permeability and good PM2.5 remove	Easy to aging and low intensity	[80]
4	Polyurethanes (PU)	120	Ammonium sulphate	Excellent elasticity, good abrasion resistance and good PM2.5 remove	Poor stability and easy to aging	[24]
5	Nylon6/polyacrylonitrile (N6/PAN)	272	NaCl aerosol	High porosity, high filtration efficiency and good stability.	Poor air permeability, poor cycle performance	[81]
6	Polyvinyl chloride /polyurethane (PVC/PU)	960	NaCl aerosol	Good tensile strength, high abrasion resistance and high filtration efficiency	Poor cycle performance and easy to aging	[37]
7	Polyacrylonitrile /polyurethane (PAN/PU)	175	NaCl aerosol	Superhydrophobicity, controllable morphology, porous structure	Poor air permeability, poor cycle performance	[36]
8	Polyacrylonitrile /silica nanoparticles (PAN/SiO ₂)	600–700	NaCl aerosol	Layer-by-layer assisted stacking structure and high filtration efficiency	Stability is poor, low strength, poor wear resistance	[82]
9	Present work	2,000–5,000	smoke	Adjustable nanostructure, high strength, high stability and high toughness	Sensitive to temperature and solvents	

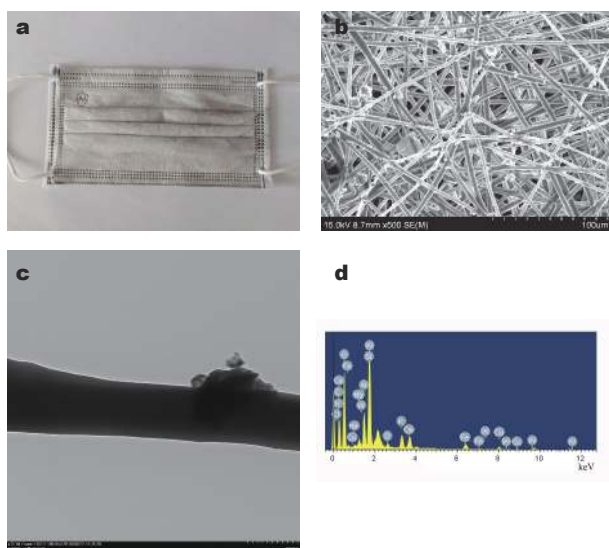


Figure 9 (a) Photo of a commercial mask covered with an electrospun mat (Sample 1) after air filtration; (b, c) TEM and SEM characterization of the morphologies of Sample 1 attached to PM2.5 particles; (d) EDX composition analysis of PM2.5 particles.

signal mainly consists of three peaks at 284.6, 285.3, and 288.7 eV, which represent the bonds of C–OH, C–O, and O=C–O, respectively (Fig. 10c). Moreover, the O 1s peaks are located at 531.7, 532.5, and 533.5 eV, corresponding to C=O, C–O, and –O–H bonds, respectively (Fig. 10d). The overall results show that C and O are the primary elements on the PM2.5 surface with functional groups of C–C, C–O, and C=O [27]. The results of the XPS analysis of PM2.5 correspond to the results of elemental analysis of the energy spectrum (Fig. 9).

CONCLUSIONS

In summary, we have demonstrated that electrospun PCL/PEO air filtration mats can act as highly effective PM2.5 filters because of their lamella-like petals, thickness of uneven fiber diameter, packing density, and surface chemistry. These air filtration mats can effectively prevent PM2.5 from entering the human respiratory tract; thereby they protect humans who carry out activities outdoors on hazy days. An electrospun PCL/PEO air

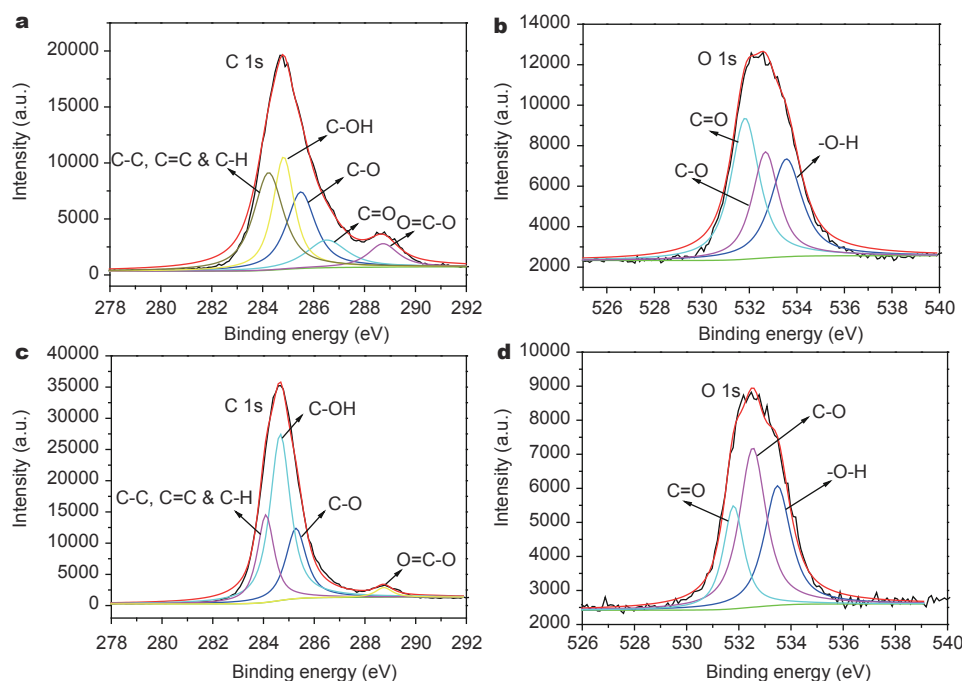


Figure 10 C 1s and O 1s deconvolutions in the XPS profiles of electrospun mat (Sample 1) before (a, b) and after (c, d) air filtration.

filtration mat with a volume of 3 mL spinning precursor fluid and 5 days of SVA treatment can be used under simulated environmental conditions, with a PM_{2.5} removal efficiency maintained at 80.01%. This air filtration mat was also evaluated by a field test in Qinhuangdao with a PM_{2.5} removal efficiency of 73.75%. We found that our air filtration mat integrated with commercial masks can be used as a new filter mask to provide protection in outdoor environments. In addition, this air filtration mat can be regenerated and demonstrate excellent reusability. The present study is expected to open new avenues for the design and preparation of ecofriendly air filtration mats that can be easily integrated into commercial masks, which could filter PM_{2.5} out of the air more efficiently if the SVA treatment method is used.

Received 28 May 2018; accepted 2 July 2018;
published online 8 August 2018

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Acknowledgements This work was supported by the National Natural Science Foundation of China (21473153 and 51771162), Support Program for the Top Young Talents of Hebei Province, China Postdoctoral Science Foundation (2015M580214), Research Program of the College Science & Technology of Hebei Province (ZD2018091), and the Scientific and Technological Research and Development Program of Qinhuangdao City (201701B004).

Author contributions Huang X, Jiao T, and Peng Q performed and designed the project and experiments. Liu Q, Zhang L, Zhou J, Li B, and Peng Q characterized the materials and discussed the results of the experiments. All the authors commented on the final paper.

Conflict of interest The authors declare no conflicts of interest.



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通过溶剂蒸汽处理制备多级静电纺丝纳米纤维作为空气过滤膜实现PM2.5高效捕获

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摘要 空气污染特别是颗粒物(PM)污染,已经威胁到人类的身体健康,因而引起了全世界的高度关注.人们在室外可通过口罩进行个人防护,然而一般的商业口罩起不到好的防护效果.本文利用静电纺丝技术和溶剂蒸汽退火(SVA)方法制备了新型高效的聚(ϵ -己内酯)/聚环氧乙烷(PCL/PEO)空气过滤纳米纤维.通过SVA处理,纤维表面变得褶皱,增强了对PM2.5的捕获效率.在重度污染状况(PM2.5颗粒浓度 $>225 \text{ mg m}^{-3}$)下,这种纳米褶皱空气过滤膜的移除效率达80.01%.秦皇岛雾霾天实地测量表明,空气过滤膜能高效移除PM2.5.与商业口罩相比,本文经过SVA处理后的PCL/PEO空气过滤膜具有制备方法简单、环境友好且易降解的特性,在高效过滤膜领域有潜在应用.