

1 High Carbon Use Efficiency in Soil Microbial Communities Is Related to Balanced  
2 Growth, not Storage Compound Synthesis

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23

24 ABSTRACT

25           The efficiency with which microbes use substrate (Carbon Use Efficiency or CUE) to make new  
26 microbial biomass is an important variable in soil and ecosystem C cycling models. It is generally  
27 assumed that CUE of microbial activity in soils is low, however measured values vary widely. It is  
28 hypothesized that high values of CUE observed in especially short-term incubations reflect the build-up  
29 of storage compounds in response to a sudden increase in substrate availability and are therefore not  
30 representative of CUE of microbial activity in unamended soil.

31           To test this hypothesis, we measured the  $^{13}\text{CO}_2$  release from six position-specific  $^{13}\text{C}$ -labeled  
32 glucose isotopomers in ponderosa pine and piñon-juniper soil. We compared this position-specific  $\text{CO}_2$   
33 production pattern with patterns expected for 1) balanced microbial growth (synthesis of all compounds  
34 needed to build new microbial cells) at a low, medium, or high CUE, and 2) synthesis of storage  
35 compounds (glycogen, tri-palmitoyl-glycerol, and polyhydroxybutyrate).

36           Results of this study show that synthesis of storage compounds is not responsible for the  
37 observed high CUE. Instead, it is the position-specific  $\text{CO}_2$  production expected for balanced growth and  
38 high CUE that best matches the observed  $\text{CO}_2$  production pattern in these two soils. Comparison with  
39 published studies suggests that the amount of glucose added in this study is too low and the duration of  
40 the experiment too short to affect microbial metabolism. We conclude that the hypothesis of high CUE  
41 in undisturbed soil remains viable and worthy of further testing.

## 42 1. Introduction

43 Heterotrophic microbes use organic carbon (C) compounds to synthesize cellular compounds  
44 while releasing some substrate-C as CO<sub>2</sub>. Which compounds are synthesized depends on the physiology  
45 of the cells (active growth and division, survival when substrate availability is low, dormancy). It is  
46 currently not possible to determine directly the compounds that are produced. It seems plausible that  
47 the microbial community consists of cells in all possible physiological states at any time, unless there are  
48 synchronizing events, such as a simultaneous depletion of substrate in all soil niches or a sudden  
49 increase in substrate availability. The C Use Efficiency (CUE; biomass-C synthesized per substrate-C  
50 consumed; mol C / mol C) of the soil microbial community is an important ecosystem variable that  
51 influences what proportion of organic C utilized is released to the atmosphere as CO<sub>2</sub> or potentially  
52 remains in the soil as organic matter in living cells or dead soil organic matter (Billings & Ballantyne,  
53 2013; Bradford, 2013; Hagerty et al., 2014). Indirectly, CUE also determines whether nutrients such as  
54 nitrogen (N) or phosphate are immobilized or mineralized (Manzoni et al., 2012; Sinsabaugh et al.,  
55 2013). Consequently, an improved understanding of CUE is important for soil C and N cycling models  
56 (Allison et al., 2010; Hagerty et al., 2014; Li et al., 2014; Manzoni et al., 2012; Wieder et al., 2013). The  
57 CUE is a function of the cellular demand for energy and biosynthesis, and therefore a function of the  
58 physiological state and compounds that are being produced. When only energy is required (such as for  
59 cell maintenance), CUE is close or equal to zero (Amthor 2000; Chapman & Gray, 1986).

60 Because of low C availability in soil and the supposedly recalcitrant nature of soil organic matter,  
61 the CUE of the microbial community is often assumed to be low (Anderson & Domsch, 2010; Manzoni et  
62 al., 2012; Reischke et al., 2015; Sinsabaugh et al., 2013). The limited substrate available is used to satisfy  
63 energy demands for cell maintenance with little left for growth. However, many studies find high values  
64 of CUE (0.6 and higher; e.g., Brant et al., 2006; Dijkstra et al., 2011a,b; Frey et al., 2013; van Groenigen  
65 et al., 2013; Hagerty et al., 2014; Steinweg et al., 2008; Thiet et al., 2006; Tucker et al., 2013; Ziegler et

66 al., 2005). The average CUE observed in soil is 0.55 (Manzoni et al., 2012; Sinsabaugh et al., 2013). This  
67 value is remarkably close to the average maximum value of CUE observed in pure culture studies (~0.6;  
68 Blagodatskaya et al., 2014; Roels, 1980; Sinsabaugh et al., 2013), but below the theoretical  
69 thermodynamic maximal CUE of growth on glucose (0.88 – 1.0; Gommers et al., 1988; Heijnen, 2010;  
70 Heijnen & van Dijken, 1992; Manzoni et al., 2012; Roels, 1980; Xiao & van Briesen, 2006). The average  
71 CUE for soil is much higher than that found in aquatic ecosystems (~0.3; Hobbie & Hobbie, 2013;  
72 Manzoni et al., 2012; Sinsabaugh et al., 2013). This large discrepancy in CUE raised concerns (Hobbie &  
73 Hobbie 2013; Sinsabaugh et al., 2013), prompting a critical evaluation of methods used to determine  
74 community CUE (Sinsabaugh et al., 2013).

75         The measurement of CUE often involves adding (<sup>13</sup>C-enriched) substrates. It is suggested that  
76 high substrate additions alter CUE, either increasing (van Groenigen et al., 2013; Sinsabaugh et al., 2013)  
77 or decreasing it (van Groenigen et al., 2013; Russell, 2002). Specifically for short-term experiments, it is  
78 hypothesized that high CUE values may not represent microbial balanced growth (that is, the synthesis  
79 of all compounds to build new cells), but instead may be the result of rapid uptake of substrate followed  
80 by synthesis of storage compounds (Blagodatskaya et al., 2014; Hill et al., 2008; Nguyen & Guckert,  
81 2001; Reischke et al., 2014, 2015; Sinsabaugh et al., 2013). Although this still represents an increase in  
82 biomass, for a sound understanding of C cycling in soil ecosystems, it is important to distinguish  
83 between CUE during long-term microbial activity and that where microbes temporarily allocate C to  
84 storage synthesis associated with a sudden and temporary increase in substrate availability (Sinsabaugh  
85 et al., 2013). Microbial cells can store substrate as starch, glycogen, trehalose, extracellular  
86 polysaccharides (Wilson et al., 2010), polyhydroxyalkanoates and storage lipids (Lu et al., 2009; Olsson &  
87 Johansen, 2000). However, measurements of storage synthesis in soil have not been made.

88         In this study we evaluate four mutually exclusive hypotheses: 1) the microbial community uses  
89 substrate for maintenance only (CUE = 0); 2) the microbial community exhibits balanced growth but an

90 overall low CUE (CUE = 0.3 as suggested by Sinsabaugh et al., 2013), 3) the microbial community exhibits  
91 a high CUE but “unbalanced” growth where biosynthesis is limited to storage compound production  
92 (glycogen, lipids, or polyhydroxybutyrate), and 4) the microbial community exhibits balanced growth at  
93 high CUE (0.6; close to the maximal CUE in pure culture studies).

94 We conducted an incubation experiment with six position-specific <sup>13</sup>C-labeled glucose  
95 isotopomers and two soils from northern Arizona, USA. We compared the observed pattern of position-  
96 specific CO<sub>2</sub> production with patterns predicted for balanced microbial growth at varying CUE (CUE = 0,  
97 0.3, or 0.6) and storage synthesis (glycogen, tri-palmitoyl-glycerol - TPG - and polyhydroxybutyrate -  
98 PHB). By comparing our experimental methods and results with published studies of responses of  
99 microbial growth to substrate addition, we test a fifth hypothesis that the increase in substrate  
100 availability changed the CUE of the microbial community. We show that the observed position-specific  
101 CO<sub>2</sub> production resembles patterns expected for balanced growth at high CUE, does not match CO<sub>2</sub>  
102 production patterns of any combination of low or medium CUE and storage compound synthesis.  
103 According to current published research results, these results were not affected by the change in  
104 substrate availability.

## 105 2. Materials and methods

### 106 2.1 Experimental procedures

107 We collected soil (0-10 cm depth) from two locations along the C. Hart Merriam Elevation  
108 Gradient ([www.nau.edu/Ecoss/](http://www.nau.edu/Ecoss/)) near Flagstaff, Arizona. The highest site (2340 m elevation, mean  
109 annual temperature (MAT) 8°C, mean annual precipitation (MAP) 660 mm) was a small open area in a  
110 ponderosa pine (*Pinus ponderosa*) stand covered with blue grama (*Bouteloua gracilis*). Soil was a Mollic  
111 Eutroboralf (C content 1.5%, N content 0.11%; Dijkstra et al., 2006). The second site (2020 m elevation,  
112 MAT 10°C, MAP 380 mm) was an intercanopy space in a piñon-juniper stand (*Pinus edulis*, *Juniperus*  
113 *monosperma*) also covered with blue grama grass. Soil type was a Calcic Haplustand (C content 1.7%, N  
114 content 0.16%; Dijkstra et al., 2006). Soil was sieved (2 mm mesh) and stored at 4°C until used in the fall  
115 of 2012.

116 We weighed 40 g of sieved soil into a specimen cup and placed in a Mason jar (473 ml) equipped  
117 with an airtight lid and septum (n = 4). Soil moisture content was adjusted to field capacity (0.272 and  
118 0.300 g water g<sup>-1</sup> soil dry weight for respectively ponderosa pine and piñon-juniper soil) and incubated  
119 overnight in the dark at room temperature (21°C). The next morning, jars were opened, headspace  
120 atmosphere was replaced with lab air, and, after closing the jar, 10 ml of pure CO<sub>2</sub> was added to the  
121 headspace. This addition of pure CO<sub>2</sub> was needed to have enough CO<sub>2</sub> in 10 ml headspace gas samples  
122 for the Picarro 2101-*i* CO<sub>2</sub> isotope spectrometer (Picarro Inc, Sunnyvale, CA) to measure isotope ratios  
123 within the calibrated range of concentrations (Dijkstra et al., 2011a). After 30 min and before glucose  
124 isotopologue addition, a 10 ml headspace gas sample was taken (time zero).

125 We used glucose (<sup>13</sup>C-labeled in C<sub>1</sub>, C<sub>2</sub>, C<sub>3</sub>, C<sub>4</sub>, C<sub>5</sub>, C<sub>6</sub> and uniformly (U) labeled) as the metabolic  
126 tracer (99 atom fraction %; Cambridge Isotope Laboratories, Andover, Massachusetts). Two ml of a 1.79  
127 mM glucose isotopomer solution was added to each specimen cup (0.536 μmol glucose-C g<sup>-1</sup> soil; n = 4).  
128 Because of the large number of isotopologues, replicates, each consisting of seven glucose isotopologue

129 incubations, were done on successive days. Ten ml headspace gas samples were taken 20, 40, and 60  
130 min after tracer addition and analyzed for isotope composition with the Picarro CO<sub>2</sub> isotope analyzer.  
131 The isotope composition of headspace CO<sub>2</sub> was expressed as atom fraction excess (%; Coplen, 2011) and  
132 plotted against time. We determined the slope of atom fraction excess (calculated as the difference  
133 between the atom fraction at t=1 and the atom fraction at t=0) for the period that the CO<sub>2</sub> production  
134 rate was constant (40 min, Fig. 3a) and calculated the ratio of position-specific CO<sub>2</sub> production rates as  
135 follows:

$$136 \quad \frac{Cx}{CU} = \frac{{}^{13}\text{CO}_2 \text{ production from } x\text{-}^{13}\text{C glucose}}{{}^{13}\text{CO}_2 \text{ production from } U\text{-}^{13}\text{C glucose}} \quad (1)$$

137 where x stands for each of the six C-atoms in glucose and U for the uniformly labeled glucose  
138 isotopologue. Ratios were calculated for each replicate.

139

## 140 2.2 Modeling

141 We used the metabolic model of the central C metabolic network (CCMN) as described in  
142 Dijkstra et al. (2011a) with small modifications (Fig. 1). In short, this model included glycolysis, pentose  
143 phosphate pathway (PP-pathway), TCA cycle, pyruvate carboxylase as the anaplerotic reaction, and  
144 eight precursor-consuming biosynthesis reactions. The model assumed that glucose is the only substrate  
145 for cell metabolism. The model included consumption of CCMN intermediates for biomass synthesis.  
146 The model did not make any assumptions about microbial growth rates or CUE, but assumed that all  
147 biomass synthesis reactions consumed precursors in constant proportion (in other words a constant  
148 chemical composition of the cell). The proportional precursor demand (demand for precursors to enable  
149 balanced growth) was br1 : br2 : br3 : br4 : br5 : br6 : br7 : br8 = 1 : 0.32 : 9.29 : 11.63 : 13.20 : 5.01 :  
150 7.44 : 4.89 representative of Gram-negative bacteria, estimated from pure culture studies (Dijkstra et  
151 al., 2011a). We compared the position-specific CO<sub>2</sub> fluxes for Gram-negative bacteria with CO<sub>2</sub> fluxes for  
152 Gram-positive bacteria and fungi which have slightly different precursor demands and chemical

153 composition (see below; Dijkstra et al., 2011a). The model calculated the fate of each C atom in glucose  
154 and the probability that it is released as CO<sub>2</sub> by pyruvate dehydrogenase, isocitrate dehydrogenase, α-  
155 ketoglutarate dehydrogenase or phosphogluconate dehydrogenase. Carbon use efficiency of the  
156 microbial community was calculated from the model as

$$157 \quad CUE = 1 - \frac{\sum(r5,r6,r7,r9)}{6 \times r1} \quad (\text{eq. 2}).$$

158 with *r5*, *r6*, *r7*, *r9* as the rate of the CO<sub>2</sub>-releasing reactions *r5*, *r6*, *r7*, and *r9* respectively, and *r1* as the  
159 rate of substrate uptake *r1* (Fig. 1). For further details, see Dijkstra et al. (2011a, b), van Groenigen et al.  
160 (2013), and supplemental information with Hagerty et al. (2014).

161 Molar flux rates of the CCMN processes (relative to glucose uptake) were estimated by matching  
162 observed and modeled patterns of position-specific CO<sub>2</sub> production. In the previous version of this  
163 model, isotopologue pairs of glucose and pyruvate were used. In this study, we solved the model using  
164 the six isotopologue ratios (eq. 1) of glucose. A model solution was calculated by minimizing the sum of  
165 squares (SS) of the difference between observed and predicted ratios for the six glucose isotopomers by  
166 altering the reaction rates *r9* and *br1* (Fig. 1) using the Excel linear programming tool Solver. A local  
167 minimum of SS was sometimes observed when model calculations were initiated with *r9* and *br1* equal  
168 to zero. For these solutions, R<sup>2</sup> was low and the regression had a negative slope. These results were  
169 avoided when initiating the Solver procedure with *r9* and *br1* greater than 0.5.

170

### 171 *2.3 Calculation of position-specific CO<sub>2</sub> production for balanced growth with low, medium, and high CUE*

172 The metabolic model (Fig. 1) is typically used to find the flux rates through the CCMN processes  
173 and CUE by matching modeled glucose (and pyruvate) isotopomer ratios to ratios observed in soil.  
174 However, the model can also be used in reverse to calculate glucose position-specific CO<sub>2</sub> production  
175 rates (or isotopomer ratios) for hypothetical situations. We used the model to calculate the hypothetical  
176 position-specific CO<sub>2</sub> production rates for CUE equal to 0 (*i.e.*, only synthesis of ATP for maintenance



177 processes), CUE equal to 0.3 (*i.e.*, most likely CUE in soil ecosystems as proposed by Sinsabaugh et al.,  
178 2013), and CUE equal to 0.6 (*i.e.*, CUE similar to the maximum observed in pure culture studies). For CUE  
179 equal to 0.3 or 0.6, position-specific CO<sub>2</sub> production was calculated assuming balanced microbial growth  
180 (synthesis of all compounds to build new microbial biomass plus energy for maintenance). Because the  
181 position-specific CO<sub>2</sub> production patterns were strongly influenced by the activity of the PP-pathway, we  
182 modeled CO<sub>2</sub> production patterns for minimal and maximal PP-pathway activity for Gram-negative  
183 bacteria separately (see above for proportional precursor demand). For maximal PP-pathway activity,  
184 we set the value of *r*<sub>2</sub> (Fig. 1) to zero, so that all flux was directed via the PP-pathway. Substrate was  
185 returned to the glycolysis as fructose-6P and glyceraldehyde-3P. For minimal PP-pathway activity, the  
186 flux of substrate into the PP-pathway was set to that required for growth (*r*<sub>9</sub> = *br*<sub>8</sub>), but no substrate  
187 was cycled through the PP-pathway back into glycolysis. For these two situations, *br*<sub>1</sub> was then manually  
188 changed until the calculated value of CUE equaled the desired value (0, 0.3, or 0.6).

189 Calculation of position-specific CO<sub>2</sub> production patterns for the situation of “unbalanced”  
190 growth (production of reserve compounds only) is described below. To evaluate the sensitivity of  
191 position-specific CO<sub>2</sub> production rates for variation in proportional precursor demand, we compared the  
192 position-specific CO<sub>2</sub> production for the situation of balanced growth, CUE equal to 0.6, and minimal and  
193 maximal activity of the PP-pathway for Gram-negative bacteria (see above), Gram-positive bacteria (1 :  
194 0.47 : 5.01 : 5.53 : 6.28 : 3.27 : 4.28 : 2.54) and fungi (1 : 0.18 : 1.30 : 1.45 : 1.72 : 1.04 : 1.08 : 0.60).

195

#### 196 2.4 Calculation of position-specific CO<sub>2</sub> production for storage compound synthesis

197 We also calculated the CUE and position-specific CO<sub>2</sub> production of cells that synthesize  
198 glycogen (as an example of carbohydrate storage), PHB (an example of polyhydroxyalkanoates), and TPG  
199 (an example of lipid storage) by taking into account the amount of ATP and precursors needed for

200 biosynthesis. Information on synthesis pathway stoichiometry and energy demand was obtained from  
201 MetaCyc.org (Caspi et al., 2014) and is detailed in sections 2.4.1-2.4.3.

202 We manually calculated the position-specific CO<sub>2</sub> production rates for synthesis of glycogen,  
203 TPG, and PHB, again for minimal (all C flows via glycolysis) and maximal activity of the PP-pathway (all C  
204 flows via the PP-pathway). Glycogen synthesis included glucose uptake, phosphorylation, and  
205 polymerization into glycogen with ATP as the energy donor. ATP was provided by the complete  
206 oxidation of a small fraction of glucose to CO<sub>2</sub>. Tri-palmitoyl-glycerol was synthesized with acetyl-CoA as  
207 the precursor for the fatty acids and dihydroxyacetone-P as the precursor for the glycerol backbone. The  
208 ATP needed for this process was produced during the formation of acetyl-CoA via glycolysis and PP-  
209 pathway. The extra energy for desaturation of fatty acids came from complete oxidation of a small  
210 amount of glucose when PP-pathway activity was minimal. When PP-pathway activity was maximal,  
211 enough ATP was produced for fatty acid production and subsequent desaturation reactions. Synthesis of  
212 PHB was accomplished with acetyl-CoA as the only precursor, and all required energy was produced  
213 during acetyl-CoA production via glycolysis or PP-pathway.

214 During the synthesis of acetyl-CoA from glucose with minimal PP-pathway activity, C<sub>3</sub> and C<sub>4</sub> of  
215 glucose were released as CO<sub>2</sub> by pyruvate dehydrogenase. The remainder (acetyl-CoA) was used for TPG  
216 and PHB synthesis (Fig. 2). In this case, C<sub>3</sub> and C<sub>4</sub> of the original glucose molecule were released as CO<sub>2</sub>,  
217 while C<sub>1</sub>, C<sub>2</sub>, C<sub>5</sub> and C<sub>6</sub> of glucose were incorporated in acetyl-CoA and ended up in fatty acids or  
218 polyhydroxybutyrate.

219 The breakdown of glucose to acetyl-CoA via the PP-pathway was more complex as C<sub>1</sub> of glucose  
220 was lost in the first few reactions, and the C<sub>2</sub> and C<sub>3</sub> were rearranged to form fructose-6P and  
221 glyceraldehyde-3P (Fig. 2). As part of the PP-pathway, glucose was decarboxylated and the pentose  
222 sugars were rearranged to fructose-6P and glyceraldehyde-3P in a ratio of 2 : 1. Fructose-6P then broke

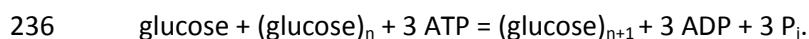
223 into two molecules glyceraldehyde-3P, which mixed with glyceraldehyde-3P retrieved from the PP-  
224 pathway. The next step was the release of CO<sub>2</sub> by pyruvate dehydrogenase to form acetyl-CoA.

225 For energy production, we assumed that 1 NAD(P)H was equivalent to 2.5 ATP, 1 FADH<sub>2</sub>  
226 produced 1.5 ATP, and 1 GTP was equal to 1 ATP and were readily exchangeable.

227

#### 228 2.4.1 Glycogen

229 We used glycogen as an example of a storage compound derived from glucose-6P precursors  
230 (Wilson et al., 2010). ATP cost for making glycogen included glucose uptake (1 ATP – de Kok et al., 2012),  
231 phosphorylation (1 ATP), and transformation to UDP-glucose and polymerization (cost for regenerating  
232 UTP is 1 ATP). For high glucose concentrations, glucose can enter microbial cells through facilitated  
233 transport, but in this analysis, we assumed low glucose concentrations where proton-coupled symport  
234 was more likely (Wilson-O'Brien et al., 2010). Glucose-6P precursor and energy demand for glycogen  
235 synthesis was described with the following reaction equation:



237 The ATP needed for uptake and polymerization of 1 mol glucose into glycogen was produced  
238 from oxidation of 0.0968 mol glucose.

239

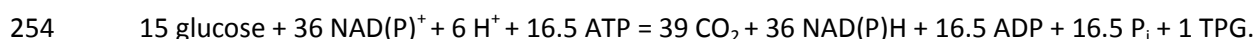
#### 240 2.4.2 Tri-Palmitoyl-Glycerol

241 Storage lipids in fungi and bacteria include triacylglycerides (Alvarez & Steinbuchel, 2002; Kosa &  
242 Ragauskas, 2011). Palmitic acid is a common fatty acid in storage lipids in *Glomus* species (Olsson &  
243 Johansen, 2000). To make 1 mol TPG, 3 mol palmitic acid and 1 mol glycerol were consumed. To produce  
244 3 mol palmitic acid, twelve mol glucose were metabolized to 24 mol acetyl-CoA while releasing 24 mol  
245 CO<sub>2</sub>. An additional 0.5 mol glucose was needed to synthesize dihydroxyacetone-P, which was turned into

246 1 mol glycerol-P and combined with 3 mol palmitic acid to form 1 mol TPG. The stoichiometric reaction  
247 equation for the synthesis of TPG via glycolysis (minimal PP-pathway activity) was:



249 Assuming that ATP and NAD(P)H were fully interchangeable, this reaction produced 0.5 mol ATP per mol  
250 TPG. The introduction of 1.5 double bonds per TPG (Olsson & Johansen, 2000) consumed 1 mol ATP per  
251 mol storage lipids. This required an additional 0.032 mol glucose, which was completely oxidized,  
252 producing ~0.19 mol CO<sub>2</sub>. When substrate was directed via the PP-pathway (maximal PP-pathway  
253 activity), the stoichiometric reaction equation for TPG synthesis was:



255 In this case, sufficient ATP was produced to drive desaturation reactions.

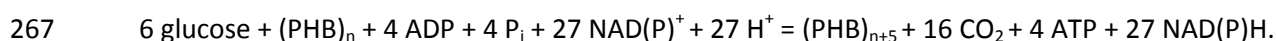
256

### 257 *2.4.3 Polyhydroxybutyrate*

258 Many bacterial species are able to synthesize polyhydroxyalkanoates as C and energy storage  
259 (Lu et al., 2009). Polyhydroxybutyrate is a representative of this class of compounds. For the synthesis of  
260 PHB, glucose was taken up by the cell, phosphorylated, and metabolized to acetyl-CoA. Two acetyl-CoA  
261 molecules were then combined into 3-hydroxy-butanoyl-CoA and polymerized to PHB, consuming 1  
262 NADPH. The stoichiometric reaction equation for PHB synthesis via glycolysis (minimal PP-pathway  
263 activity) was:



265 When glucose was directed into the PP-pathway (maximal PP-pathway activity), the equation for PHB  
266 synthesis was:



268 All ATP required for these reactions was produced during the production of acetyl-CoA.

269

270 *2.5 Statistical Analysis*

271           One-way analysis of variance was used to evaluate differences between soils. To determine  
272 which biochemical scenario (balanced growth with CUE = 0, CUE = 0.3, or CUE = 0.6, synthesis of storage  
273 compounds) best explained observed CO<sub>2</sub> production data, we assessed whether the 95% confidence  
274 interval of observed isotopomer ratios overlapped with model predictions. Correspondence between  
275 observed and predicted position-specific CO<sub>2</sub> production was evaluated using R<sup>2</sup>, and closeness of slope  
276 and intercept to the expected 1 : 1 line.

## 277 3. RESULTS

### 278 3.1 Observed position-specific CO<sub>2</sub> production

279 We measured the position-specific CO<sub>2</sub> production for glucose in ponderosa pine and piñon-juniper soil.  
280 The rate of CO<sub>2</sub> production from glucose isotopomers was constant for 40 minutes (Fig. 3A, results  
281 piñon-juniper soil not shown), after which it started to decline as was seen in previous studies (Dijkstra  
282 et al., 2011b). There were clear and significant differences in CO<sub>2</sub> production from different C atoms  
283 ( $P < 0.05$ ): CO<sub>2</sub> production from C<sub>1</sub> was significantly higher than C<sub>4</sub>, which was higher than C<sub>2</sub> and C<sub>3</sub>, while  
284 C<sub>5</sub> and C<sub>6</sub> were the lowest (Fig. 3B). This pattern was the same for both soils. The CUE derived from  
285 modeling (following Dijkstra et al., 2011a with slight modifications) was not significantly different  
286 between the two soils (0.62 for ponderosa pine soil, 0.61 for piñon-juniper soil).

287

### 288 3.2 Modeled position-specific CO<sub>2</sub> production

#### 289 3.2.1 Balanced growth and varying CUE

290 Carbon use efficiency and activity of the PP-pathway had a large influence on the position-  
291 specific CO<sub>2</sub> production (Fig. 4). With maximal PP-pathway activity and high CUE, most of the CO<sub>2</sub> was  
292 produced from C<sub>1</sub> and C<sub>4</sub>; in contrast, when PP-pathway activity was minimal, most CO<sub>2</sub> was produced  
293 from C<sub>3</sub> and C<sub>4</sub>. With decreasing CUE, these differences became less pronounced. When CUE equaled 0,  
294 substrate was used only for synthesis of ATP and NAD(P)H, and all C positions were released as CO<sub>2</sub> at  
295 the same rate (ratio of C<sub>x</sub> : C<sub>U</sub> = 1 : 6).

296 The compounds produced to make bacterial or fungal cells differed slightly resulting in small  
297 differences in proportional precursor demand between Gram-negative and Gram-positive bacteria and  
298 fungi (Dijkstra et al., 2011a). These differences in proportional precursor demand had only a small effect  
299 on metabolic flux patterns and CUE (Dijkstra et al., 2011a, b; van Groenigen et al., 2013). This was also  
300 true for the position-specific CO<sub>2</sub> production (Fig. 5). For this reason, the results presented for Gram-

301 negative bacteria were considered representative for any combination of fungi, Gram-positive and  
302 Gram-negative bacteria.

303

### 304 *3.2.2 Storage compound synthesis*

305 The CUE of storage compound synthesis was high (Table 1) although dependent on the form of  
306 storage (glycogen > TPG  $\approx$  PHB). For TPG and PHB, the CUE was reduced when PP-pathway activity was  
307 high.

308 Complete oxidation of glucose to CO<sub>2</sub> was needed to provide the ATP for glycogen synthesis.  
309 This resulted in a position-specific CO<sub>2</sub> production that was equal for all C atoms (Table 2), similar to the  
310 situation with only maintenance energy (Fig. 4). The ATP required for TPG synthesis was produced  
311 during the breakdown of glucose to acetyl-CoA. Therefore, with minimal PP-pathway activity and all C  
312 flowing via glycolysis, CO<sub>2</sub> was only released during the pyruvate dehydrogenase reaction (Fig. 2 (left); C<sub>3</sub>  
313 and C<sub>4</sub> lost as CO<sub>2</sub>; all other C atoms are incorporated into the fatty acids; Table 2). However,  
314 desaturation of palmitic acid (Olsson & Johansen, 2000) required an additional mol ATP per mol TPG  
315 (0.032 mol glucose, producing  $\sim$ 0.19 mol CO<sub>2</sub> evenly from all six C atoms, and reducing CUE from 0.680  
316 to 0.678). This had only a minor effect on the position-specific CO<sub>2</sub> production. The breakdown of  
317 glucose to acetyl-CoA via the PP-pathway had a higher energy yield and most CO<sub>2</sub> was released from C<sub>1</sub>  
318 and C<sub>4</sub> (Fig. 2; Table 1, 2). The position-specific CO<sub>2</sub> production associated with PHB synthesis resembled  
319 that of TPG (Table 2).

320 We did not model the synthesis of other storage compounds (starch, extracellular  
321 polysaccharides, trehalose, other fatty acids, other polyhydroxyalkanoates). The metabolic pathways for  
322 these compounds were closely related to those for glycogen, TPG or PHB, and would likely result in  
323 similar CO<sub>2</sub> production patterns and CUE. The approach developed here can be used for the synthesis of

324 other hypothetical reserve compounds, at least as long as the details of the biosynthetic pathways are  
325 known.

326

### 327 *3.3 Correlation between measured and modeled position-specific CO<sub>2</sub> production*

328 The correlation between measured and modeled position-specific CO<sub>2</sub> production was low for  
329 CUE = 0, glycogen synthesis, and all cases where PP-pathway activity was low (Table 3). However, the  
330 modeled patterns of position-specific CO<sub>2</sub> production for maximal PP-pathway activity explained  
331 between 75-99% of the variance in the observed data. Although the modeled patterns of CUE = 0.3 and  
332 TPG and PHB synthesis (with maximal PP-pathway activity) exhibited high correlation coefficients (Table  
333 3, Fig. 6), we found the best fit between observed and modeled position-specific CO<sub>2</sub> production for the  
334 hypothetical situation of balanced growth with a CUE = 0.6 and high PP-pathway activity (98-99% of  
335 variance explained; slope and intercept very close to the expected 1 : 1 line).

336 Combining low CUE (CUE = 0 or 0.3) with storage synthesis decreased the correspondence  
337 between modeled and observed position-specific CO<sub>2</sub> production. This is easiest understood by focusing  
338 on one C position, for example C<sub>1</sub>. Assume CUE = 0 combined with sudden synthesis of TPG. In that case,  
339 CO<sub>2</sub> from C<sub>1</sub> during TPG synthesis assuming maximal PP-pathway activity (0.385) is combined with CO<sub>2</sub>  
340 production from C<sub>1</sub> for CUE = 0 (0.167). This will reduce the CO<sub>2</sub> production from C<sub>1</sub>, and increase the  
341 difference between predicted and observed CO<sub>2</sub> production for this C atom (0.44 or 0.43 for ponderosa  
342 pine and piñon-juniper soil respectively). In fact, any combination of medium or low CUE and storage  
343 synthesis resulted in CO<sub>2</sub> production patterns that deviated more from observed patterns than those  
344 associated with storage compound synthesis alone.

345

346



#### 347 4. DISCUSSION

348 Determining the position-specific CO<sub>2</sub> production from <sup>13</sup>C-labeled compounds is a  
349 straightforward and quick way to test biochemically explicit hypotheses for microbial processes,  
350 including storage compound synthesis, in microbial communities. In this study, we tested the mutually  
351 exclusive hypotheses that CUE of microbial substrate use is zero (substrate used for maintenance only -  
352 Hypothesis 1), CUE is low (important role for maintenance energy demand – Hypothesis 2), CUE is high  
353 because of “unbalanced” growth (storage compound production – Hypothesis 3), or CUE is high  
354 associated with balanced growth (Hypothesis 4). Based on the evidence presented, we conclude that the  
355 soil microbial community had a high CUE associated with balanced growth (Hypothesis 4).

356 The CUE observed is in the same range as found on average for soil ecosystems (~0.55) and  
357 higher than that in aquatic ecosystems (~0.3; Manzoni et al., 2012; Sinsabaugh et al., 2013). The two  
358 soils in this study exhibit similar patterns, suggesting that the metabolic processes in these soils are  
359 similar. Studies using a broad range of soils are required to determine whether this is a general pattern  
360 in soils. The results from this study demonstrate that observations of high CUE in earlier studies (e.g.,  
361 Brant et al., 2006; Dijkstra et al., 2011a, b; Frey et al., 2013; van Groenigen et al., 2013; Hagerty et al.,  
362 2014; Hill et al., 2008; Nguyen & Guckert, 2001; Steinweg et al., 2008; Thiet et al., 2006; Tucker et al.,  
363 2013) do not necessarily represent storage compound synthesis as sometimes suggested (Blagodatskaya  
364 et al., 2014; Hill et al., 2008; Nguyen & Guckert, 2001; Reischke et al., 2014, 2015; Sinsabaugh et al.,  
365 2013), but may be related to balanced microbial growth.

366

##### 367 *4.1 High CUE: a consequence of glucose addition?*

368 Although the results from this study exclude storage compound synthesis as an artefact, it does  
369 not eliminate other possible artefacts. The high CUE and balanced growth observed in this experiment  
370 may not be representative of microbial activity in unamended soil but a response to the glucose addition

371 used to measure CUE (Hypothesis 5). Several studies have suggested that glucose additions may alter  
372 microbial growth and CUE, either increasing (van Groenigen et al., 2013; Sinsabaugh et al., 2013) or  
373 decreasing it (van Groenigen et al., 2013; Russell, 2002). In the following, we will discuss the influence of  
374 substrate addition on microbial growth and metabolism, specifically the effect of substrate  
375 concentration and response time.

376 Substrate additions used to determine CUE range from 0.8 nmol glucose-C g<sup>-1</sup> soil (Nguyen &  
377 Guckert, 2001) to 61.5 μmol glucose-C g<sup>-1</sup> soil (Thiet et al., 2006). The glucose applied in this experiment  
378 is at the low end of this range (0.536 μmol glucose-C g<sup>-1</sup> soil), and well within the range of  
379 concentrations found in unamended soils (~1 nmol glucose-C g<sup>-1</sup> soil - Fischer et al., 2007 - to 18 μmol g<sup>-1</sup>  
380 soil - Jones & Darrah, 1995). Yet, high CUE is found in this and other short-term experiments with even  
381 lower glucose additions (0.8 nmol glucose-C g<sup>-1</sup> soil, Nguyen & Guckert, 2001; 1.6 nmol glucose-C g<sup>-1</sup> soil,  
382 Hill et al., 2008). We conclude that there is no evidence to suggest that CUE is high because of  
383 unnaturally high concentrations of substrate.

384 Anderson & Domsch (2010) and Reischke et al. (2014, 2015) find that glucose addition  
385 stimulates microbial growth, but only after a lag-phase of 8 - 14 h and at high glucose concentrations (>  
386 4.6 - 90 μmol glucose-C g<sup>-1</sup> soil depending on soil type – Anderson & Domsch, 2010; > 41.5 μmol glucose-  
387 C g<sup>-1</sup> soil – Reischke et al., 2014; > 16 μmol glucose-C g<sup>-1</sup> soil – Reischke et al., 2015). These  
388 concentrations are higher than used in this experiment (0.536 μmol glucose-C g<sup>-1</sup> soil), suggesting that  
389 the glucose additions used in this experiment, and those by Nguyen & Guckert (2001) and Hill et al.  
390 (2008), are too low and the incubation duration too short to induce microbial growth.

391 Furthermore, almost immediately after glucose addition, respiration increases, while microbial  
392 growth rates remain unaffected (Reischke et al., 2014, 2015). These observations imply that CUE  
393 decreases during the lag-phase in response to (a large) glucose addition. Similar declines in CUE are  
394 found in pure culture studies where glucose addition rates exceed maximum growth rates or when

395 other nutrients than C limit growth (Russell 2007; Russell & Cook 1992). However, the CUE measured in  
396 this experiment and by Nguyen & Guckert (2001) and Hill et al. (2008) are high, suggesting again that  
397 substrate additions used did not affect microbial metabolism.

398 We conclude, based on existing studies on soil microbial community growth that the glucose  
399 addition in this experiment is too low and incubation duration too short to induce microbial growth.  
400 Furthermore, CUE was high and not low as expected during a lag-phase after glucose addition. Finally,  
401 storage compound synthesis was ruled out as an artefact based on the observed position-specific CO<sub>2</sub>  
402 production patterns. Therefore, we tentatively conclude that the high CUE and balanced growth we  
403 observed is representative of CUE in unamended soil.

404

#### 405 *4.2 High CUE and maintenance energy requirements*

406 The high CUE observed seems in contradiction to the idea that soil is a C-limited environment  
407 where most microbes are not growing or only grow slowly, and where maintenance energy demand  
408 dominates substrate use (Blagodatskaya & Kuzyakov 2013; Reischke et al., 2015). Evidence of actively  
409 dividing microbes is found by Rousk et al. (2011) and Reischke et al. (2014, 2015) in soil without glucose  
410 additions, and in <sup>18</sup>O-H<sub>2</sub>O stable isotope probing experiments (Schwartz, 2007). Blagodatskaya &  
411 Kuzyakov (2013) conclude from extensive literature review that about 0.1-2% of the soil microbial cells  
412 are actively growing and reproducing. This direct evidence of microbial growth in unamended soils  
413 indicates that at least a portion of the microbial community has a high CUE and balanced growth.  
414 Moreover, a low growth rate by itself, expected in C- and nutrient-limited soils, is not necessarily  
415 associated with a low CUE. For example, a 10-fold reduction in growth rate (0.388 h<sup>-1</sup> to 0.044 h<sup>-1</sup>) in *E.*  
416 *coli* pure cultures caused only a moderate reduction of CUE from 0.60 to 0.51 (Kayser et al., 2005).  
417 Likewise, Lin et al. (2009) find no change in CUE for *Geobacillus* growth rates ranging between 0.053 h<sup>-1</sup>  
418 and 0.00078 h<sup>-1</sup>.

419 A high CUE for the entire soil community is only possible if growing microbes with high CUE  
420 dominate microbial activity compared to microbes with low CUE. The community in soil is thought to  
421 consist of actively growing and dividing (0.1-2%), potentially growing (10-40%), and dormant microbes  
422 (remaining fraction; Blagodatskaya & Kuzyakov, 2013). To what degree the high CUE in a small, actively  
423 growing and dividing community is “diluted” by maintenance respiration of the inactive portion of the  
424 community may be calculated as follows. For simplicity, we assume that the active microbial fraction  
425 grows with a CUE near the highest values observed (CUE = 0.7), while the potentially active and dormant  
426 fractions conduct maintenance only (CUE = 0). Price & Sowers (2004) estimate ratios of metabolic rates  
427 of optimal-growth : maintenance : survival (dormancy) as 1 :  $10^{-3}$  :  $10^{-6}$ . Applying these values to a  
428 community with 0.1-2% actively growing and dividing cells means that about 90-98% of substrate  
429 consumed is associated with actively growing microbes, and only 10-2% with potentially active or  
430 dormant microbes. A CUE of 0.7 for the actively growing community would then translate to a total  
431 community CUE of 0.64 (0.1% of community actively growing) or 0.69 (2% active). A similar argument is  
432 presented in Frey et al. (2001).

433

## 434 **5. Conclusions**

435 It is a well-established practice to use uniformly-labeled compounds to study microbial  
436 processes, including CUE. We show here that additional information is obtained by using position-  
437 specific  $^{13}\text{C}$ -labeled compounds. This information can be used to test biochemically explicit hypotheses  
438 related to microbial physiology and biochemistry in soil microbial communities. We conclude that CUE in  
439 two soil microbial communities is high. This high CUE is not related to storage synthesis but to balanced  
440 growth, and appears to be unaffected by the small amount of glucose added.

441 The conclusion that the soil microbial community operates with a high CUE in soil environments  
442 has important and potential far-reaching consequences. It affects how we model microbial activity in

443 soils and think about the relative importance of maintenance processes. As a result, microbial death  
444 (caused by viruses, grazing and predation) becomes more important as a key process in stabilizing  
445 microbial population size and community composition (Hagerty et al., 2014), suggesting a possible top-  
446 down control of microbial production by organisms at higher trophic levels. It also invites a rethinking of  
447 the recalcitrant nature of soil organic matter and its suitability as a microbial substrate, and, as a  
448 consequence, a rebalancing of the role of chemical vs physical protection in soil organic matter  
449 stabilization (von Lützow et al., 2006). We conclude that the hypothesis of high CUE in undisturbed soil  
450 remains viable and worthy of further testing.

451

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456

457

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580

581 TITLES AND LEGENDS TO FIGURES

582 Fig. 1. Model and mass balance equations for calculation of fluxes through the central C metabolic  
583 network (after Dijkstra et al., 2011a). Relative to a previous version, pentose phosphate pathway and  
584 TCA cycle representations are simplified by combining several reactions. These alterations do not  
585 change model outcomes. Flux rates (reactions r2-r11 and biosynthesis reactions br1-br8) are normalized  
586 to glucose uptake rate (r1, set at 100) on a molar basis. Abbreviations: G6P, glucose-6P; F1,6P, fructose-  
587 1,6P2; GAP, glyceraldehyde-P; PYR, pyruvate; ACCO, acetyl-CoA; ICIT, isocitrate;  $\alpha$ KG,  $\alpha$ -ketoglutarate;  
588 OAA, oxaloacetate; RU5P, ribulose-5P; S7P, sedoheptulose-7P; E4P, erythrose-4P.

589  
590 Fig. 2. Diagram of glucose breakdown via glycolysis (left) and PP-pathway (right) to acetyl-CoA and CO<sub>2</sub>  
591 (red circles). Number in circles refer to the C-atom position in the original glucose molecule.

592  
593 Fig. 3. Rates of CO<sub>2</sub> production for six glucose isotopomers in ponderosa pine soil (A; means and S.E.)  
594 and position-specific CO<sub>2</sub> production (relative to U-<sup>13</sup>C labeled glucose; B; means and 95% confidence  
595 interval) for ponderosa pine and piñon-juniper soil. Letters indicate significant differences between C  
596 positions for ponderosa pine (upper case) and piñon-juniper (lower case) soil.

597  
598 Fig. 4. Modeled position-specific CO<sub>2</sub> production (relative to U-<sup>13</sup>C labeled glucose) for CUE = 0, 0.3, and  
599 0.6 for minimal (A) and maximal (B) pentose phosphate pathway activity.

600  
601 Fig. 5. Modeled position-specific CO<sub>2</sub> production (relative to U-<sup>13</sup>C labeled glucose) for CUE = 0.6 for  
602 minimal (A) and maximal (B) pentose phosphate pathway activity for proportional precursor demand  
603 characteristic for Gram-negative bacteria, Gram-positive bacteria, and fungi.

604

605 Fig. 6. Modeled versus observed position-specific CO<sub>2</sub> production (relative to U-<sup>13</sup>C labeled glucose;  
606 means and 95% confidence interval) for ponderosa pine (red squares) and piñon-juniper soil (green  
607 circles) for modeled balanced growth with CUE = 0.6 (A), CUE = 0.3 (B), and tri-palmitoyl glycerol  
608 synthesis (C) with maximal pentose phosphate pathway activity. Dashed lines are the expected 1:1  
609 relationships. Information on regression statistics is available in Table 3.

Table 1: Carbon Use Efficiency of glycogen, tri-palmitoyl-glycerol, and polyhydroxybutyrate synthesis assuming minimal and maximal pentose phosphate pathway activity calculated from stoichiometry of synthesis reactions (eq. 3-6).

	CUE
<b>Minimal pentose phosphate pathway</b>	
Glycogen	0.90
Tri-palmitoyl-glycerol	0.68
Polyhydroxybutyrate	0.67
<b>Maximal pentose phosphate pathway</b>	
Glycogen	0.90
Tri-palmitoyl-glycerol	0.57
Polyhydroxybutyrate	0.56

Table 2: Predicted CO<sub>2</sub> production patterns associated with glycogen, tri-palmitoyl-glycerol and polyhydroxybutyrate synthesis assuming minimal and maximal pentose phosphate pathway activity. <sup>§</sup> and <sup>A</sup> indicate significant differences with observed position-specific CO<sub>2</sub> production for ponderosa pine and piñon-juniper soil respectively.

	C <sub>1</sub>	C <sub>2</sub>	C <sub>3</sub>	C <sub>4</sub>	C <sub>5</sub>	C <sub>6</sub>
<b>Minimal pentose phosphate pathway</b>						
Glycogen	0.167 <sup>SA</sup>	0.167 <sup>SA</sup>	0.167 <sup>SA</sup>	0.167 <sup>SA</sup>	0.167 <sup>SA</sup>	0.167 <sup>SA</sup>
Tri-palmitoyl-glycerol	0.000 <sup>SA</sup>	0.000 <sup>SA</sup>	0.500 <sup>SA</sup>	0.500 <sup>SA</sup>	0.000 <sup>SA</sup>	0.000 <sup>SA</sup>
Polyhydroxybutyrate	0.000 <sup>SA</sup>	0.000 <sup>SA</sup>	0.500 <sup>SA</sup>	0.500 <sup>SA</sup>	0.000 <sup>SA</sup>	0.000 <sup>SA</sup>
<b>Maximal pentose phosphate pathway</b>						
Glycogen	0.167 <sup>SA</sup>	0.167 <sup>SA</sup>	0.167 <sup>SA</sup>	0.167 <sup>SA</sup>	0.167 <sup>SA</sup>	0.167 <sup>SA</sup>
Tri-palmitoyl-glycerol	0.385	0.128	0.128 <sup>§</sup>	0.359 <sup>SA</sup>	0.000 <sup>SA</sup>	0.000 <sup>SA</sup>
Polyhydroxybutyrate	0.375 <sup>§</sup>	0.125 <sup>SA</sup>	0.125 <sup>§</sup>	0.375 <sup>SA</sup>	0.000 <sup>SA</sup>	0.000 <sup>SA</sup>

- 1 Table 3. Correlation of observed position-specific CO<sub>2</sub> production pattern for ponderosa pine (first  
 2 number) and pinon-juniper soil (second number) with modeled CO<sub>2</sub> production patterns for balanced  
 3 growth (CUE = 0, CUE = 0.3, CUE = 0.6), and glycogen, tri-palmitoyl-glycerol, and polyhydroxybutyrate  
 4 synthesis with minimal or maximal pentose phosphate pathway activity.

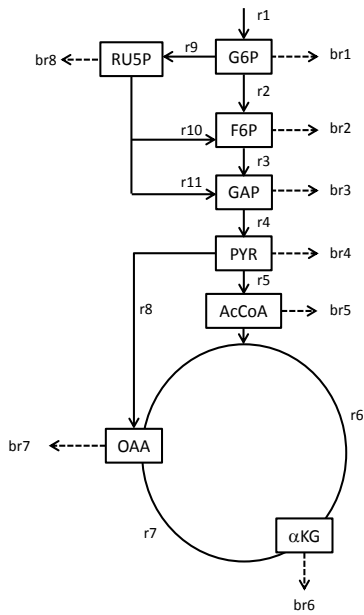
	R <sup>2</sup>	Slope	Intercept
<b>Minimal pentose phosphate pathway</b>			
CUE = 0	0.00 / 0.00	0.00 / 0.00	0.167 / 0.167
CUE = 0.3	0.00 / 0.01	-0.01 / 0.02	0.168 / 0.164
CUE = 0.6	0.00 / 0.04	0.03 / 0.12	0.162 / 0.146
Glycogen	0.00 / 0.00	0.00 / 0.00	0.167 / 0.167
Tri-palmitoyl-glycerol	0.00 / 0.01	-0.10 / 0.21	0.183 / 0.131
Polyhydroxybutyrate	0.00 / 0.01	-0.10 / 0.21	0.183 / 0.131
<b>Maximal pentose phosphate pathway</b>			
CUE = 0	0.00 / 0.00	0.00 / 0.00	0.167 / 0.167
CUE = 0.3	0.91 / 0.95	0.27 / 0.30	0.121 / 0.115
CUE = 0.6	0.98 / 0.99	0.92 / 1.03	0.010 / -0.007
Glycogen	0.00 / 0.00	0.00 / 0.00	0.167 / 0.167
Tri-palmitoyl-glycerol	0.78 / 0.90	1.04 / 1.23	-0.010 / -0.042
Polyhydroxybutyrate	0.75 / 0.88	1.02 / 1.23	-0.007 / -0.041

5

6



Fig. 1



#### Equations

- 1- G6P:  $r_1 = r_9 + r_2 + br_1$
- 2- F6P:  $r_2 + r_{10} = r_3 + br_2$
- 3- GAP:  $r_3 + r_{11} = r_4 + br_3$
- 4- PYR:  $r_4 = r_5 + r_8 + br_4$
- 5- AcCoA:  $r_5 = r_6 + br_5$
- 6-  $\alpha$ KG:  $r_6 = r_7 + br_6$
- 7- OAA:  $r_7 + r_8 = r_6 + br_7$
- 8- RU5P:  $r_9 = r_{10} + r_{11} + br_8$

Fig. 2.

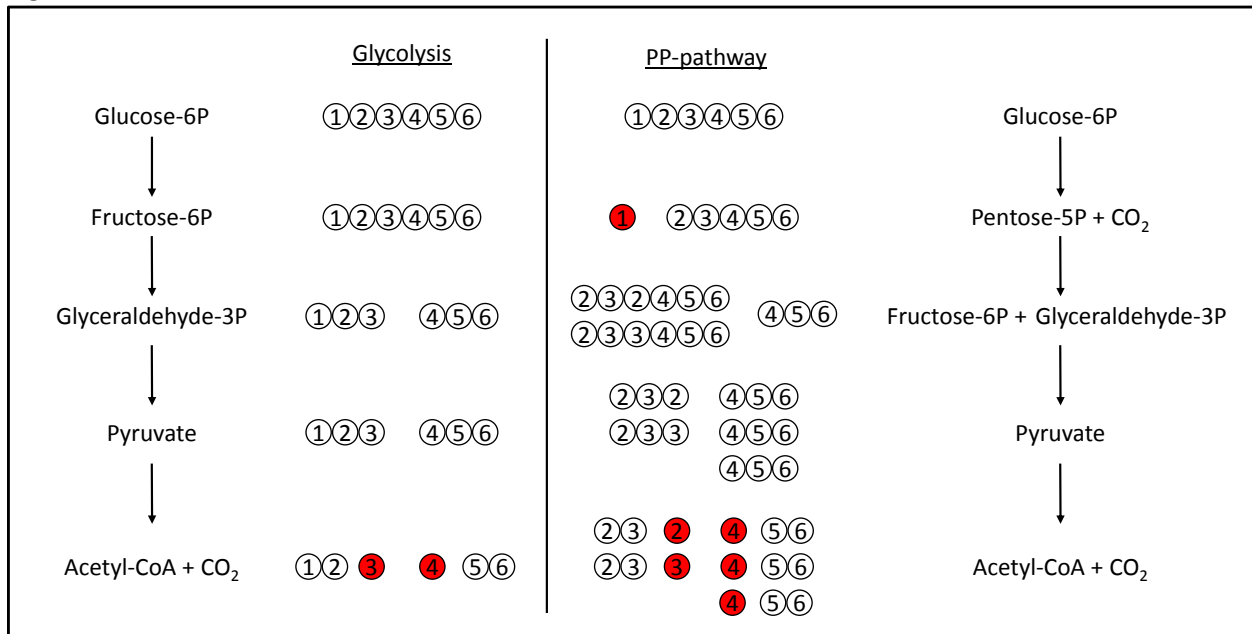


Fig. 3

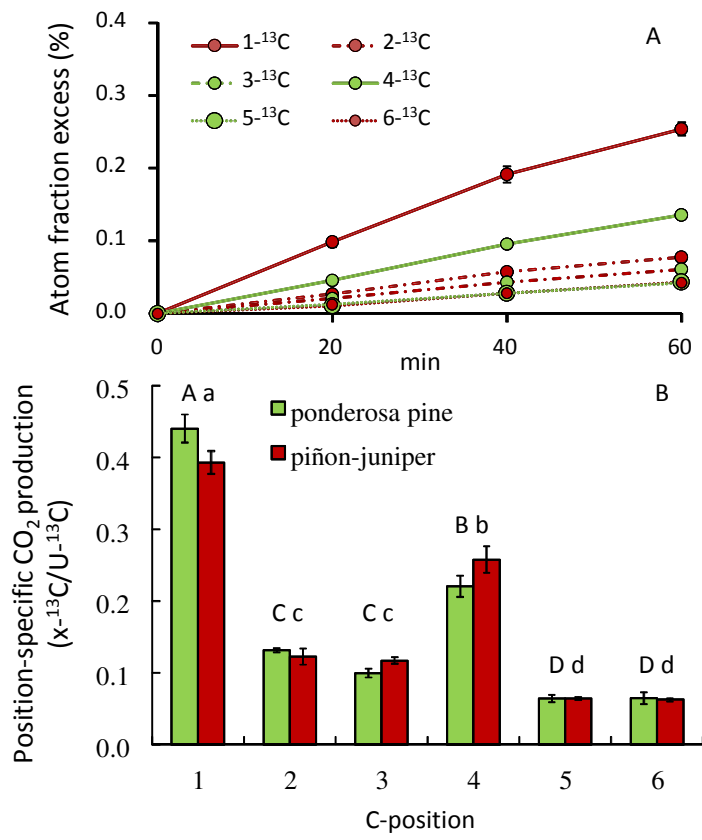


Fig. 4

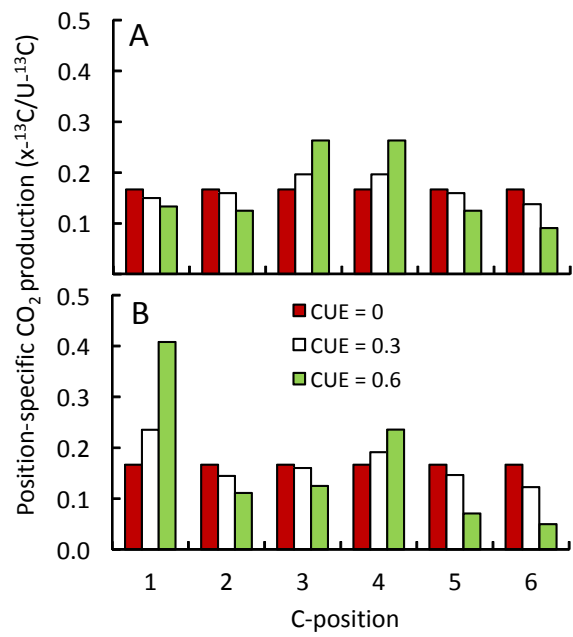


Fig. 5.

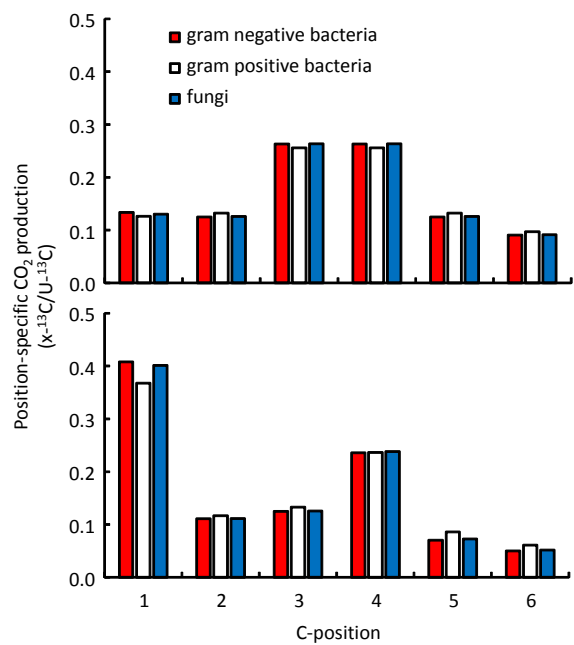


Fig. 6.

