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High-mobility hydrogenated polycrystalline In_2O_3 (In_2O_3 :H) thin-film transistors

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Oxide semiconductors have been extensively studied as active channel layers of thin-film transistors (TFTs) for electronic applications. However, the field-effect mobility (μ_{FE}) of oxide TFTs is not sufficiently high to compete with that of low-temperature-processed polycrystalline-Si TFTs (50-100 cm²V⁻¹s⁻¹). Here, we propose a simple process to obtain high-performance TFTs, namely hydrogenated polycrystalline In₂O₃ (In₂O₃:H) TFTs grown via the low-temperature solid-phase crystallization (SPC) process. In₂O₃:H TFTs fabricated at 300 °C exhibit superior switching properties with $\mu_{FE} = 139.2 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$, a subthreshold swing of 0.19 Vdec⁻¹, and a threshold voltage of 0.2 V. The hydrogen introduced during sputter deposition plays an important role in enlarging the grain size and decreasing the subgap defects in SPC-prepared In₂O₃:H. The proposed method does not require any additional expensive equipment and/or change in the conventional oxide TFT fabrication process. We believe these SPC-grown In₂O₃:H TFTs have a great potential for use in future transparent or flexible electronics applications.

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ide-bandgap oxide semiconductors (OSs) have been extensively studied as active channel layers of thinfilm transistors (TFTs) for next-generation flat-panel displays^{1,2}, nonvolatile memories³, inverters⁴, various sensors^{5,6}, Schottky devices^{7,8}, and so on. Among OSs, amorphous In-Ga-Zn-O (a-IGZO) TFTs have now become the backplane standard for active-matrix liquid-crystal displays (AMLCDs) and active-matrix organic light-emitting diode (AMOLED) displays because of their reasonable field-effect mobility ($\mu_{\rm EF}$) of over 10 cm² V⁻¹ s⁻¹, extremely low leakage current, low process temperature (<350 °C), and large-area scalability^{9,10}. Although the $\mu_{\rm FE}$ value of a-IGZO TFTs is more than ten times higher than that of hydrogenated amorphous Si (a-Si:H) TFTs $(<1 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1})$, it is not sufficiently high to compete with that of low-temperature-processed polycrystalline-Si (LTPS) TFTs $(50-100 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1})^{11}$. The main disadvantages of LTPS TFTs are a relatively high process temperature (450-550 °C) and an expensive crystallization process. The high $\mu_{\rm FE}$ values of OS TFTs mean these devices could be used in fields that have been dominated by LTPS TFTs and in transparent and flexible devices that are incompatible with Si. Numerous types of approaches to enhance the $\mu_{\rm FE}$ value of OS TFTs have been investigated, including cation composition^{12,13}, multiple chan-nel structures^{14,15}, dual-gate architecture^{16,17}, metal capping layer structures^{18,19}, post treatment^{20,21}, and their combination. Among these, cation composition control is the most promising method. It requires no extra complex process for integrating OS TFTs. In-rich OSs have been studied extensively. The large spatial spread of the In 5s orbital with a large overlap can provide a facile electron transport path with a low electron effective mass²². However, undoped In₂O₃ films exhibit a high background electron concentration $(10^{19}-10^{21} \text{ cm}^{-3})^{23,24}$, which is attributable to the presence of native defects, such as oxygen vacancies, making it difficult to control the threshold voltage of the TFTs^{25,26}. To suppress the carrier concentration in In₂O₃, elements such as Ga, Hf, Si, Al, and W were added, which have large bond dissociation energies with oxygen¹². Many AOS TFTs have been explored with multicomponent oxide semiconductors, such as In-W-Zn-O²⁷, Al-In-Sn-Zn-O²⁸, and In-Ga-Zn-Sn-O²⁹. However, multicomponent oxides complicate the composition control of the deposited film. Moreover, multimetal cations cause potential fluctuation near the conduction band minimum, which might hinder electron transport³⁰.

Recently, crystalline OSs have been proposed to enhance the carrier mobility because the disorder-induced subgap states can be suppressed via lattice ordering. Yang et al. reported a $\mu_{\rm FE}$ value of 60.7 cm² V⁻¹ s⁻¹ for a TFT obtained using polycrystalline In–Ga–O annealed at 700 °C³¹. Although high annealing temperatures result in better electrical properties of the oxide active channel layer, such high temperatures are unsuitable for device application on glass or plastic substrates. Our group reported a $\mu_{\rm FE}$ value of 50.6 cm² V⁻¹ s⁻¹ for a TFT obtained using hydrogenated polycrystalline In–Ga–O formed via solid-phase crystallization (SPC) at 300 °C³².

This study proposes a simple material and a simple process to obtain high-performance TFTs, namely hydrogenated polycrystalline In₂O₃ (In₂O₃:H) TFTs grown via the low-temperature SPC process. In₂O₃:H TFTs fabricated at 300 °C exhibit superior switching properties with $\mu_{\rm FE} = 139.2 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, a subthreshold swing (SS) of 0.19 V dec⁻¹, and a threshold voltage ($V_{\rm th}$) of 0.2 V. The hydrogen introduced during sputter deposition plays an important role in enlarging the grain size and decreasing the subgap defects in SPC-prepared In₂O₃:H. The proposed method has great potential for future transparent or flexible electronics applications.

Results

Structural properties of the In2O3 and In2O3:H films. Figures 1a, b show the XRD patterns of the 50-nm-thick In₂O₃ and In₂O₃:H films deposited at various $R[H_2]$ values and at a constant $R[O_2]$ of 1%. For the as-deposited films (Fig. 1a), the In₂O₃ film without H₂ introduction exhibited a clear crystalline nature with the (222) preferred orientation of the cubic bixbyite In₂O₃ crystal. There was no noticeable peak for the In_2O_3 :H films deposited at $R[H_2]$ values of 3 and 5%. This result indicates that H₂ addition suppresses the growth of crystallites during deposition. After annealing at 250 °C in nitrogen for 1 h, the amorphous phase of In₂O₃:H changed to the crystalline one with the (222) preferred orientation. The angles of the diffracted peaks are in good agreement with the In2O3 powder data (ICSD code: 14388). Moreover, the crystallized films exhibited smaller fullwidth at half-maximum values of the (222) reflection than the film deposited without hydrogen introduction, indicating larger crystallite sizes and smaller strains in the In₂O₃:H films.

Figure 1c-h depict the EBSD images along the normal direction for the In₂O₃ and In₂O₃:H films with and without annealing at 250 °C in nitrogen. For the as-deposited films (Fig. 1c-e), a randomly oriented small grain structure embedded in the amorphous matrix can be observed in the In₂O₃ film without H₂ introduction. The grain structure disappeared upon increasing the $R[H_2]$ value to 5%. In contrast, a huge grain structure appeared for In_2O_3 :H deposited at $R[H_2] = 5\%$ after annealing at 250 °C (Fig. 1h), indicating SPC occurrence. This is consistent with the results of the XRD analysis shown in Fig. 1a, b. The corresponding area fractions of each lateral grain size are shown in Fig. 1i, j. The detected minimum grain size is around 15 nm, which is limited by the electron beam size of the EBSD measurements. For the as-deposited films (Fig. 1i), all films showed the maximum area fraction for a grain size of 15 nm; however, a small proportion of the area fraction with a grain size of ~70 nm was detected in the In₂O₃ film, indicating nuclei in the as-deposited film. After annealing at 250 °C (Fig. 1j), the peak of the area fraction shifted toward a larger grain size as $R[H_2]$ increased, and the In_2O_3 :H film deposited at $R[H_2] = 5\%$ showed a maximum area fraction of 23% at a grain size of 140 nm. Furthermore, as $R[H_2]$ increased from 0 to 5%, the area fraction of the minimum grain size below 15 nm significantly decreased, and only a few small grains were in between the large grains, as shown in Fig. 1h. Similar results were observed for films annealed at 250 °C in ambient air (shown in Supplementary Fig. 1). The EBSD results show that the nuclei density in the as-deposited film was suppressed by introducing hydrogen during sputtering. Because of the reduction in the nuclei density in the initial In₂O₃:H film, the grain size of the In₂O₃:H film could be enlarged through SPC. Thus, the XRD and EBSD results indicate that controlling the crystallinity and nuclei density in the as-deposited film are key factors to achieve high-quality In₂O₃:H films.

Electrical and optical properties of the In₂O₃ and In₂O₃:H films. Figure 2a–d show the carrier concentration (N_e) and Hall mobility (μ_H) of the 50-nm-thick In₂O₃ and In₂O₃:H films deposited at various $R[H_2]$ values as a function of the annealing temperature (T_{ann}). Koida et al. reported that both the N_e and μ_H of the SPC-prepared In₂O₃:H films decrease dramatically for the films annealed in vacuum at $T_{ann} > 400$ °C due to the desorption of the H₂O and H₂ gases from the films and additional microscopic defects inside the grains³³. The electrical properties of OSs are strongly affected by the annealing atmosphere³⁴; thus, annealing treatments at temperatures ranging from 150 to 350 °C in nitrogen and ambient air were examined. For the as-deposited films, N_e increased from 7.1 × 10¹⁹ to 5.7 × 10²⁰ cm⁻³ upon increasing $R[H_2]$ from 0 to 5%, as shown in Fig. 2a. Since

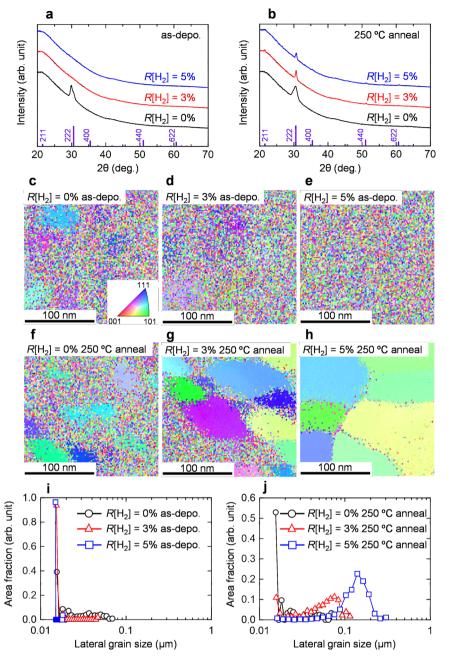


Fig. 1 Structural properties of the ln_2O_3 and ln_2O_3:H films. XRD patterns of the ln_2O_3 and ln_2O_3 :H films deposited at different $R[H_2]$ values and at a constant $R[O_2]$ value of 1% **a** before and **b** after annealing at 250 °C in N_2 . EBSD images of the ln_2O_3 and ln_2O_3 :H films deposited at different $R[H_2]$ values **c**-**e** before and **f**-**h** after annealing at 250 °C in N_2 . Area fraction of each grain size obtained from the ln_2O_3 and ln_2O_3 :H films deposited at different $R[H_2]$ values **i** before and **j** after annealing at 250 °C in N_2 .

hydrogen acts as a shallow donor in $In_2O_3^{35}$, the increase in the N_e of the as-deposited In_2O_3 :H film upon increasing $R[H_2]$ is attributable to hydrogen doping effects. The In_2O_3 film deposited without hydrogen introduction exhibited an almost constant N_e value over the whole range of investigated T_{ann} values irrespective of the annealing atmosphere, as shown in Fig. 2a, c. In contrast, there was a strong dependence of N_e on the annealing atmosphere for the In_2O_3 :H film annealed in N_2 gradually decreased with increasing T_{ann} (Fig. 2a), whereas the N_e of the In_2O_3 :H film annealed in air rapidly decreased for $T_{ann} > 200$ °C (Fig. 2c). In addition, the N_e reduction was remarkable in the In_2O_3 :H film deposited at a higher $R[H_2]$ value. As a result, an appropriate N_e value of 2.0×10^{17} cm⁻³ for TFT fabrication was obtained at $T_{ann} = 300$ °C for the In_2O_3 :H film deposited at

 $R[H_2] = 5\%$; this N_e value is over two orders of magnitude lower than that of the In₂O₃ film deposited without hydrogen introduction (3.0 × 10¹⁹ cm⁻³). Such a large decrease in the N_e value of the In₂O₃:H films has not been reported before. Although adding H₂ induced the formation of free carriers in the asdeposited films, the N_e of the films could be reduced via the relatively low-temperature SPC process and became comparable to that of single-crystalline epitaxial In₂O₃ films deposited at 650 °C (~1 × 10¹⁷ cm⁻³)³⁶.

Regarding the Hall mobility of the films, In_2O_3 without hydrogen introduction exhibited an almost constant μ_H value over the entire range of investigated T_{ann} values irrespective of the annealing atmosphere, as shown in Fig. 2b, c. Upon annealing in N₂ at $T_{ann} = 200$ °C, the μ_H of In_2O_3 :H increased to 78.6 cm² V⁻¹ s⁻¹ at

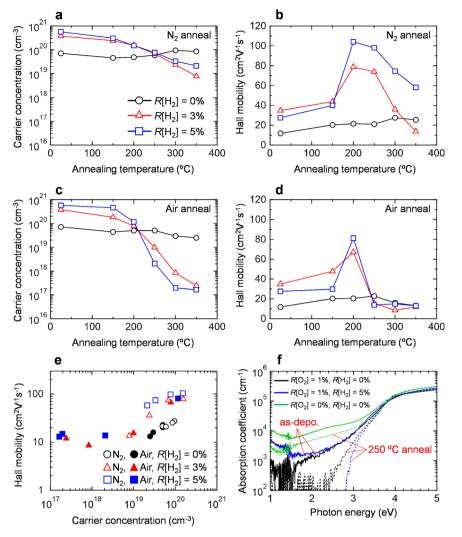


Fig. 2 Electrical and optical properties of the In₂O₃ and In₂O₃:H films. a N_e and **b** μ_H of In₂O₃ and In₂O₃:H deposited at different $R[H_2]$ values as a function of the annealing temperature in N₂. **c** N_e and **d** μ_H of In₂O₃ and In₂O₃:H deposited at different $R[H_2]$ values as a function of the annealing temperature in ambient air. **e** The relationship between μ_H and N_e for the films annealed at $T_{ann} > 200$ °C in N₂ and ambient air. **f** Optical absorption spectra of the 50-nm-thick In₂O₃ and In₂O₃:H films deposited at different $R[H_2]$ values before and after annealing at 250 °C in ambient air. In₂O₃ films deposited without oxygen and hydrogen are also shown for comparison.

 $R[H_2] = 3\%$ and 104.0 cm² V⁻¹ s⁻¹ at $R[H_2] = 5\%$, indicating that the SPC started at a T_{ann} value between 150 and 200 °C. Furthermore, the increased μ_H is attributable to the increased grain size, as shown in Fig. 1f–h. As T_{ann} increased, the μ_H of In₂O₃:H gradually decreased (Fig. 2b). Upon annealing in air (Fig. 2d), the maximum μ_H of In₂O₃:H decreased slightly to 67.1 cm² V⁻¹ s⁻¹ at $R[H_2] = 3\%$ and 81.2 cm² V⁻¹ s⁻¹ at $R[H_2] = 5\%$, and the decrease in μ_H for $T_{ann} > 250$ °C was confirmed.

To understand the transport properties of the In₂O₃:H films after SPC, the relationship between $\mu_{\rm H}$ and $N_{\rm e}$ for the films annealed in the range of temperatures between 200 °C and 350 °C in N₂ and ambient air was summarized, as shown in Fig. 2e. The $N_{\rm e}$ of the In₂O₃:H film could be controlled by up to three orders of magnitude. Moreover, for the $N_{\rm e}$ range between 10¹⁹ and 10²⁰ cm⁻³, $\mu_{\rm H}$ increases with increasing $R[{\rm H_2}]$, which is attributable to the suppression of grain boundary scattering due to the increasing grain size. For all the films, $\mu_{\rm H}$ increased with increasing $N_{\rm e}$. In general, the grain boundaries have a small impact on $\mu_{\rm H}$ in transparent conductive oxides with high $N_{\rm e}$ (>10²⁰ cm⁻³) because electrons can tunnel through the narrow width (<1 nm) of the grain barriers at high $N_{\rm e}$ values (>10²⁰ cm⁻³). However, grain boundary scattering is a dominant

factor that limits the $\mu_{\rm H}$ in films with lower $N_{\rm e}^{37}$. Thus, the observed decrease in $\mu_{\rm H}$ with decreasing $N_{\rm e}$ (Fig. 2e) is due to grain boundary scattering. Although the $\mu_{\rm H}$ of In₂O₃:H decreased for $N_{\rm e} < 10^{19} \,{\rm cm}^{-3}$, carriers (with number in the range of $10^{19} - 10^{20} \,{\rm cm}^{-3}$) will be generated at the In₂O₃:H/gate insulator interface of the TFTs when applying a voltage to the gate¹⁵; thus, a high $\mu_{\rm FE}$ can be expected in In₂O₃:H TFTs.

Figure 2f shows the optical absorption (α) spectra of the In₂O₃ and In₂O₃:H films before and after annealing at 250 °C in ambient air. The green line represents the In₂O₃ film deposited without oxygen and hydrogen (only Ar gas), which is shown for comparison. The spectral features that arise as the photon energy exceeds 2.9 eV are due to the absorption associated with the interband transition in In₂O₃, whereas the features that arise when the photon energy is below 1.5 eV are due to free carrier absorption. The absorption in the subgap region (<2.9 eV) dropped as *R*[O₂] increased from 0 to 1%, suggesting that oxygen deficiencies, which give rise to subgap defects, are compensated when sputtering in an oxidizing atmosphere. When hydrogen is added during sputter deposition, the absorption is enhanced in the subgap region for the as-deposited films, especially in the

photon energy region below 1.5 eV, indicating that free electron absorption is increased due to the hydrogen doping effect. On the other hand, after annealing at 250 °C in ambient air (dashed line), the absorption across the subgap of the In₂O₃:H film decreased significantly and was lower than that of the In₂O₃ film. Since subgap defects are generated from native defects, such as oxygen vacancies, as described above, it can be inferred that oxygen vacancies were efficiently reduced in In₂O₃:H via SPC in ambient air. This result is consistent with the Hall effect measurements, where it was found that $N_{\rm e}$ decreased from 5.7 \times 10²⁰ to 2.0 \times 10^{18} cm^{-3} upon annealing in air at 250 °C, as shown in Fig. 2c. The structural, electrical, and optical properties of the In₂O₃:H films show that the hydrogen introduced during sputter deposition plays an important role in enlarging the grain size and decreasing the subgap defects after SPC, increasing $\mu_{\rm H}$ and decreasing $N_{\rm e}$. However, a more detailed study will be necessary to carry out quantitative evaluations of the carrier generation and scattering in the SPC-prepared In₂O₃:H films.

In₂O₃:H TFT characteristics. Figure 3a shows the picture and schematic cross-sectional view of the fabricated In₂O₃:H TFT. All TFTs were fabricated using annealing in ambient air at 300 °C. Figure 3c-e show typical transfer characteristics of the TFTs with

 In_2O_3 and In_2O_3 :H channels deposited at various $R[H_2]$ values. The variations of μ_{FE} , subthreshold swing (SS), threshold voltage $(V_{\rm th})$, and hysteresis (ΔV) were evaluated from ten TFTs on the same substrate. The $\mu_{\rm FE}$ was calculated from the linear transfer characteristics as $\mu_{\rm FE} = Lg_{\rm m}/(WC_{\rm ox}V_{\rm ds})$ at $V_{\rm ds} = 0.1$ V, where $g_{\rm m}$ is the transconductance, C_{ox} is the oxide capacitance of the gate insulator, and V_{ds} is the drain voltage. V_{th} was defined by gate voltage (V_{gs}) at drain current (I_{ds}) of 1 nA, and SS was extracted from V_{gs} , which required an increase in the I_{ds} from 10 to 100 pA. The In_2O_3 TFT without H_2 introduction (Fig. 3c) did not exhibit any switching (conductive behavior). By contrast, the In2O3:H TFT deposited at $R[H_2] = 5\%$ exhibited a switching with an extremely high μ_{FE} of 139.2 cm² V⁻¹ s⁻¹, a SS of 0.19 V dec⁻¹, a $V_{\rm th}$ of 0.2 V, and a ΔV of 0.3 V as shown in Table 1. The TFT with In_2O_3 :H deposited at $R[H_2] = 5\%$ showed a slightly small SS value (0.19 V dec⁻¹) compared with that of the TFT with In_2O_3 :H deposited at $R[H_2] = 3\%$ (0.41 V dec⁻¹). The subgap density of states (D_{sg}) at the Fermi level was calculated as $SS = dV_{gs}/dlo$ $gI_{ds} = \ln 10 \ k_B T/e \ (1 + eD_{sg}/C_{ox})$, where k_B is Boltzmann's constant and e is the elementary electric charge¹⁰. The D_{sg} decreased from 1.3×10^{12} to 4.7×10^{11} cm⁻² eV⁻¹ by increasing the R[H₂] value from 3 to 5%. This result suggests that the disorderoriginated subgap defect states near the conduction band minimum can be reduced in the In2O3:H channel deposited at

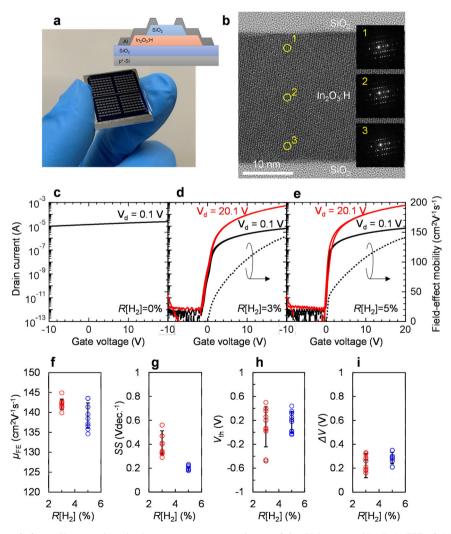


Fig. 3 In₂O₃:**H TFT characteristics.** a Photograph and schematic cross-sectional view of the SPC-prepared In₂O₃:**H TFTs. b** HRTEM image and corresponding SAED pattern of the active layer of the TFT with In₂O₃:**H** channel deposited at $R[H_2] = 5\%$. **c**-**e** Typical transfer characteristics of the TFTs with In₂O₃:**H** channels deposited at various $R[H_2]$ values. Variations of **f** μ_{FE} , **g** SS, **h** V_{th} , and **i** ΔV evaluated from ten TFTs.

Table 1 Summary of the TFT properties.						
R[O ₂] (%)	R[H ₂] (%)	$\mu_{\rm FE}~({ m cm}^2{ m V}^{-1}{ m s}^{-1})$	SS (V dec ⁻¹)	V _{th} (V)	ΔV (V)	$D_{ m sg}$ (cm $^{-2}$ eV $^{-1}$)
1	0	-	-	-		-
1	3	142.0 (1.2)	0.41 (0.09)	0.1 (0.3)	0.2 (0.1)	1.3 × 10 ¹²
1	5	139.2 (3.0)	0.19 (0.02)	0.2 (0.2)	0.3 (0.1)	4.7 × 10 ¹¹

The average values and standard deviations (σ) of the characteristics of ten TFTs on the same substrate. σ are shown in parentheses.

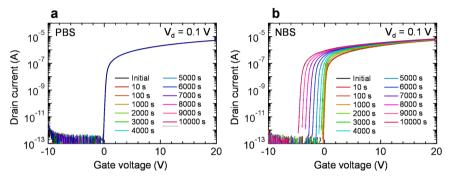


Fig. 4 Reliability of the In₂O₃:H TFT. Changes in the transfer characteristics of the In₂O₃:H (R[H₂] = 5%) TFT during the **a** PBS and **b** NBS tests. The V_{gs} values under the PBS and NBS tests were +20 and -20 V, respectively.

 $R[H_2] = 5\%$, which is confirmed by the optical measurements of the films shown in Fig. 2f.

The resulting transfer performance of the SPC-prepared In₂O₃:H TFT ($R[H_2] = 5\%$) was superior to that of previously reported oxide-based TFTs³⁸. Although the $\mu_{\rm H}$ of the In₂O₃:H ($R[H_2] = 5\%$) films decreased to 14.9 cm² V⁻¹ s⁻¹ with $N_{\rm e}$ of 2.0 × 10¹⁷ cm⁻³ by annealing at 300 °C as shown in Fig. 2c, extremely high $\mu_{\rm FE}$ was obtained from the TFTs because a large number of carriers (10¹⁹–10²⁰ cm⁻³) was generated at the In₂O₃:H/gate insulator interface when applying a voltage to the gate, which allows electrons to tunnel through the narrow width (<1 nm) of the grain barriers at high $N_{\rm e}$ values. The high $\mu_{\rm FE}$ and steep SS of the In₂O₃:H TFTs can be attributed to the high crystallinity of In₂O₃:H, especially near the In₂O₃:H/SiO₂ gate insulator interface.

Figure 3b shows a cross-sectional conventional bright-field HRTEM image and selective area electron diffraction (SAED) pattern obtained from the SPC-prepared In₂O₃:H TFT ($R[H_2] = 5\%$). A clear lattice image was observed over the entire thickness of the In₂O₃:H channel. Moreover, there was a single crystal-like diffraction pattern in the SAED pattern, even in the thin layers, roughly at a distance of 5 nm from the SiO₂ gate insulator without detectable diffuse ring patterns, which would be attributable to an amorphous phase. This observation explains the high $\mu_{\rm FE}$ of 139.2 cm² V⁻¹ s⁻¹ in the In₂O₃:H TFTs, which is comparable to the $\mu_{\rm H}$ of epitaxial single-crystal In₂O₃ films (~160 cm² V⁻¹ s⁻¹)³⁹. In addition, although In₂O₃:H is a polycrystalline film, the standard deviations (σ) of $\mu_{\rm FE}$, SS, $V_{\rm th}$, and ΔV of the In₂O₃:H TFT ($R[H_2] = 5\%$) were 3.0 cm² V⁻¹ s⁻¹, 0.02 V dec⁻¹, 0.2 V, and 0.1 V, respectively, indicating high uniformity of the TFT characteristics (shown in Fig. 3f–i and Supplementary Fig. 2).

To investigate the reliability of the SPC-prepared In_2O_3 :H TFT ($R[H_2] = 5\%$), positive-bias stress (PBS) and negative-bias stress (NBS) tests were carried out under a humidity of 50%. The gate stress voltages for the PBS and NBS were +20 and -20 V, respectively. Figure 4a, b show the changes in the transfer characteristics of the In_2O_3 :H TFT during the PBS and NBS tests. The In_2O_3 :H TFT showed no significant positive shift in $V_{\rm th}$ (only +0.02 V) under the PBS test, indicating the negligible interfacial

trap states in the In₂O₃:H/SiO₂ gate insulator interface as well as the high quality of the In₂O₃:H channel layer. In contrast, a large $V_{\rm th}$ shift of -4.4 V was observed for the NBS test, as shown in Fig. 4b. Furthermore, the $V_{\rm th}$ shift became more significant when the NBS test was conducted at a higher humidity of 70% (shown in Supplementary Fig. 3). Water molecules are coupled to the backchannel of the IGZO TFTs, and excess electrons are donated to the channel under NBS, resulting in a negative $V_{\rm th}$ shift^{40,41}. Applying a passivation layer to the TFTs is effective in minimizing the influence of the atmospheric environment; however, hydrogen can diffuse into the channel through the passivation layer and increase the N_e of the channel⁴². Although the SiO₂ passive layer was applied to the SPC-grown In₂O₃:H TFT, as shown in Fig. 3a, its protection ability was insufficient because the SiO₂ film was deposited via sputtering at RT. Hence, it is believed that the reliability of In₂O₃:H TFTs can be improved by selecting the appropriate passivation layer, such as SiN, Al₂O₃, or Y_2O_3 .

Discussion

In this study, we demonstrate the high-performance polycrystalline In₂O₃:H TFTs using a low-temperature SPC process. To ensure the amorphous state of the as-deposited In₂O₃:H film, a moderate amount of H₂ was introduced into the sputtering system during the In₂O₃:H deposition. The as-deposited amorphous In₂O₃:H film converted into a polycrystalline In₂O₃:H film with a grain size of around 140 nm via low-temperature SPC (at a temperature below 200 °C). As a result, a high $\mu_{\rm H}$ of 104.0 cm² V⁻¹ s⁻¹ was obtained for the In₂O₃:H film. This $\mu_{\rm H}$ value is five times higher than that of the In₂O₃ film without H₂ introduction during sputtering. Furthermore, the N_e of the SPCgrown In₂O₃:H film decreased significantly to 2.0×10^{17} cm⁻³ after annealing at 300 °C in ambient air; this N_e value is two orders of magnitude lower than that of the In₂O₃ film without H₂ introduction. Thus, introducing hydrogen during sputtering followed by annealing in ambient air is an effective method for improving both the crystallinity and the Ne value of In2O3:H films. The obtained In₂O₃:H film was employed as the channel of a TFT, and the resulting In₂O₃:H TFT exhibits an extremely high

 $\mu_{\rm FE}$ of 139.2 cm²V⁻¹s⁻¹, an appropriate $V_{\rm th}$ of 0.2 V, and a small SS of 0.19 V dec⁻¹. HRTEM analysis of the TFT revealed the high crystallinity of cubic bixbyite near the In₂O₃:H/SiO₂ gate dielectric interface, which contributed to the high $\mu_{\rm FE}$ of the TFT. The proposed method does not require any additional expensive equipment and/or change in the conventional oxide TFT fabrication process. Moreover, composition control of binary In₂O₃ films is easier than that of ternary and quaternary semiconductors. We believe that these SPC-grown In₂O₃:H TFTs are promising candidates for use in future transparent or flexible electronics applications.

Methods

Fabrication of In2O3:H TFTs. In2O3:H TFTs were fabricated on a heavily doped ptype Si substrate with a 100-nm-thick thermally grown SiO₂. The doped p-type Si substrate and the SiO₂ were used as the gate electrode and the gate insulator. The 30-nm-thick In₂O₃ and In₂O₃:H channels were deposited via pulsed direct-current (DC) magnetron sputtering without substrate heating from a ceramic In₂O₃ target using a mixture of Ar, O₂, and H₂ gases. The O₂ and H₂ gas flow ratios are $R[O_2] = O_2/(Ar + O_2 + H_2)$ and $R[H_2] = H_2/(Ar + O_2 + H_2)$, respectively. For the Ar + O_2 -sputtered In₂ O_3 film, $R[O_2]$ was set to 1% without H₂ introduction. For the Ar + O_2 + H₂-sputtered In₂O₃ film (In₂O₃:H), R[H₂] varied from 3 to 5%, whereas R[O2] was fixed at 1%. The deposition pressure and DC power were maintained at 0.6 Pa and 50 W, respectively. The base pressure before gas introduction was below 6×10^{-5} Pa. The In₂O₃ and In₂O₃:H films were then annealed in ambient air at 300 °C for 1 h. After annealing, a 100-nm-thick SiO₂ film was deposited via reactive sputtering without substrate heating. This film served as a passive layer. Subsequently, Al source/drain electrodes were deposited via sputtering. Finally, In2O3 and In2O3:H TFTs were annealed at 250 °C in ambient air for 1 h. The In₂O₃, SiO₂, and Al films were deposited through a shadow mask. Both the channel length and the width were 300 µm.

Characterization of the ln₂O₃:H films and TFTs. Structural, electrical, and optical measurements were conducted on the 50-nm-thick In₂O₃ and In₂O₃:H films deposited on a synthetic quartz substrate. The films' structural changes were evaluated through X-ray diffraction (XRD) (Philips corp., X'pert) with CuKa radiation and electron backscattering diffraction (EBSD) (EDAX-TSL Hikari High Speed EBSD Detector). The films' carrier concentrations (N_e) and Hall mobility ($\mu_{\rm H}$) were determined via Hall effect measurements (Accent, HL5500PC) using the van der Pauw geometry at room temperature (RT). The films' optical properties were measured via spectrophotometry (Hitachi, U-4100). The current-voltage characteristics were measured using a semiconductor parameter analyzer (Keysight, E5270B) at RT in the dark. High-resolution transmission electron microscopy (HRTEM) (JEOL, JSM-7001F) analysis was also conducted to observe the microstructure of the In₂O₃:H channel in the TFTs.

Data availability

The authors declare that the all the data supporting the finding of this study are available within this article and its Supplementary Information files and are available from the corresponding author on reasonable request.

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Author contributions

M.F. conceived the concept and supported the device analysis. Y.M. designed the research and experiments. Y.M. performed the film deposition, device fabrication, and their analysis. T.K. supported the electrical and optical measurements. W.Y. designed and fabricated pulse-DC sputtering apparatus for film deposition and set up experimental

environment for TFT fabrication and evaluation, and supported EBSD measurements. All authors contributed to the discussions and commented on the manuscript.

Competing interests

The authors declare no competing interests.

Additional information

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