High-temperature electrical and thermal transport properties of fully filled skutterudites RFe_4Sb_{12} (R = Ca, Sr, Ba, La, Ce, Pr, Nd, Eu, and Yb)

P. F. Qiu,^{1,2} J. Yang,¹ R. H. Liu,^{1,2} X. Shi,^{1,a)} X. Y. Huang,¹ G. J. Snyder,³ W. Zhang,⁴ and L. D. Chen¹

¹CAS Key Laboratory of Materials for Energy Conversion, Shanghai Institute of Ceramics, Chinese Academy of Sciences, 1295 DingXi Road, Shanghai 200050, China

²Graduate University of Chinese Academy of Sciences, Beijing 100049, China

³Department of Materials Science, California Institute of Technology, 1200 East California Boulevard, Pasadena, California 91125, USA

⁴State Key Laboratory of High Performance Ceramics and Superfine microstructure, Shanghai Institute of Ceramics, Chinese Academy of Sciences, 1295 Dingxi Road, Shanghai 200050, China

(Received 15 November 2010; accepted 5 January 2011; published online 23 March 2011)

Fully filled skutterudites RFe_4Sb_{12} (R = Ca, Sr, Ba, La, Ce, Pr, Nd, Eu, and Yb) have been prepared and the high-temperature electrical and thermal transport properties are investigated systematically. Lattice constants of RFe₄Sb₁₂ increase almost linearly with increasing the ionic radii of the fillers, while the lattice expansion in filled structure is weakly influenced by the filler valence charge states. Using simple charge counting, the hole concentration in RFe₄Sb₁₂ with divalent fillers (R = Ca, Sr, Ba, Eu, and Yb) is much higher than that in RFe₄Sb₁₂ with trivalent fillers (R = La, Ce, Pr, and Nd), resulting in relatively high electrical conductivity and low Seebeck coefficient. It is also found that RFe₄Sb₁₂ filled skutterudites having similar filler valence charge states exhibit comparable electrical conductivity and Seebeck coefficient, and the behavior of the temperature dependence, thereby leading to comparable power factor values in the temperature range from 300 to 800 K. All RFe₄Sb₁₂ samples possess low lattice thermal conductivity. The correlation between the lattice thermal resistivity W_L and ionic radii of the fillers is discussed and a good relationship of $W_L \sim (r_{cage} - r_{ion})^3$ is observed in lanthanide metal filled skutterudites. CeFe₄Sb₁₂, PrFe₄Sb₁₂ and NdFe₄Sb₁₂ show the highest thermoelectric figure of merit around 0.87 at 750 K among all the filled skutterudites studied in this work. © 2011 American Institute of *Physics*. [doi:10.1063/1.3553842]

I. INTRODUCTION

Filled skutterudites have been known as one of the best thermoelectric materials due to their excellent electrical performance and low lattice thermal conductivity.¹ The general structure formula of filled skutterudites is R_vM₄Sb₁₂, where R can be alkaline, alkaline-earth, lanthanide, or actinide metals, y is the filling fraction, and M is Fe, Ru, Os, Co, Rh, or Ir. Investigation on the thermoelectric properties in filled skutterudites has been initially started in 1990s.²⁻¹⁰ Earlier studies are mainly focused on p-type Fe_xCo_{4-x}Sb₁₂-based compounds and the values of the dimensionless thermoelectric figure of merit (ZT) had been reported above unity.^{5,6,9,11,12} The progress in n-type filled skutterudites was rather slowly until partially filled skutterudites R_vCo₄Sb₁₂ with high filling fraction of the fillers were synthesized and reported with good thermoelectric performance at elevated temperature.^{13,14} After that, improving the thermoelectric figure of merit in ntype skutterudites gained keen interest in the next decade. Various n-type single element filled RyCo₄Sb₁₂ skutterudites with high ZT values were reported,^{15,16} showing great potentials as thermoelectric power generator used at high temperatures. Moreover, recently, filling two or more types of fillers into the voids in skutterudite have been shown to be able to further reduce lattice thermal conductivity and improve ZT to 1.4 in double-filled and 1.7 in multiple-filled n-type skutterudites.^{17–20} Although there are a series of recent study reported with very high ZTs in n-type partially filled skutterudites, only a few literatures focusing on high temperature thermoelectric properties in p-type skutterudites have been published in the past decade. Furthermore, the ZT values in p-type filled skutterudites are confirmed in experiment around 1.0, much lower than n-type materials.^{21–23} The stagnation of low ZT value in p-type skutterudites limits the further development of thermoelectric modules and devices, since both excellent n- and p-type materials are required for high efficiency thermoelectric technology. Therefore, searching for p-type skutterudites with comparable ZT values to n-type materials is urgent for real industry applications.

Recently, didymium-filled Fe-based p-type skutterudites were reported to exhibit high thermoelectric performance, indicating a continual interest in p-type skutterudites.²⁴ Because the formation of n-type double- and multiple-filled skutterudites has shown very effective to reduce lattice thermal conductivity and improve thermoelectric figure of merit, it might be also suitable for p-type skutterudites to achieve high *ZTs*. Therefore, systematic and deep understanding on the thermoelectric properties of single element filled skutterudites is necessary and urgent for the design and optimization in p-type double and multiple-filled skutterudites. Fully filled

0021-8979/2011/109(6)/063713/8/\$30.00

^{a)}Electronic email: xshi@mail.sic.ac.cn.

skutterudites RFe₄Sb₁₂ is one of the most important p-type skutterudites. The first RFe₄Sb₁₂ compound was initially reported in 1980.²⁵ After that, various RFe₄Sb₁₂ ternary compounds with different filler elements were synthesized and reported with many interesting physical properties.^{3,4,6,8,10,26–38} However, the thermoelectric properties were only studied in a few RFe₄Sb₁₂ compounds experimentally.^{3,6,10,22,30,31} Furthermore, high temperature thermoelectric properties, which are crucial to the thermoelectric power generator, have been rarely reported in RFe₄Sb₁₂.^{6,10,31} Due to the lack of a broad systematic study, detailed understanding of the electrical and thermal transport properties in RFe₄Sb₁₂ system is still not available, which hinders the further optimization to achieve high ZT. Therefore, it is essential to gain a comprehensive understanding of electrical and thermal transport properties in single element filled RFe₄Sb₁₂ skutterudites.

In this paper, a series of RFe_4Sb_{12} samples with various fillers (R = Ca, Sr, Ba, La, Ce, Pr, Nd, Eu, and Yb) were synthesized and the thermoelectric properties were measured in the temperature range from 300 to 800 K. Systematic evaluation and discussion related to the effects of these fillers on the crystal structure as well as thermoelectric properties were carried out.

II. EXPERIMENT

High purity elements were weighed out in the atomic ratio of R: Fe: Sb = 1: 4: 12, and then sealed in evacuated quartz ampoules which were coated with carbon. The sealed ampoules were heated slowly up to 1350 K for 10 hs, then quenched into a water bath and annealed at 773-973 K for 7 days. To form dense pellets, the obtained ingots were manually ground into fine powders (<100 um) and then sintered by Spark Plasma Sintering (SPS) at 733-833 K for 10-15 mins. High-density (>97%) of the theoretical density) were obtained for all samples. X-ray diffraction (XRD) analysis (Rigaku, Rint2000) and electron probe microanalysis (EPMA, Shimadzu 8705QH2) were used to examine the purity and chemical composition of the samples. Electrical conductivity (σ) and Seebeck coefficient (S) were measured using the ZEM-3 (ULVAC Co. Ltd.) apparatus under Helium atmosphere from 300 to 800 K. The thermal diffusivity and specific heat were measured in argon atmosphere using laser flash method (NETZSCH LFA 427) and Shimadzu DSC-50, respectively. The density of the samples was measured using Archimedes method. Total thermal conductivity (k) was calculated by multiplying the measured values of thermal diffusivity, specific heat, and density. Hall coefficients (R_H) at 300 K were measured in a Physical Property Measurement System (Quantum Design) by sweeping the magnetic field up to 3 T in both positive and negative directions.

III. RESULTS AND DISCUSSION

A. Phase structure, composition and lattice constant

1. XRD and EPMA results

Figure 1 shows the powder x-ray diffraction (XRD) patterns for all RFe_4Sb_{12} samples. All peaks are identified as belonging to the $LaFe_4P_{12}$ structure. EPMA analyses show

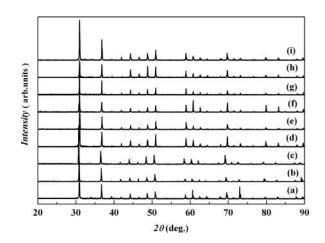
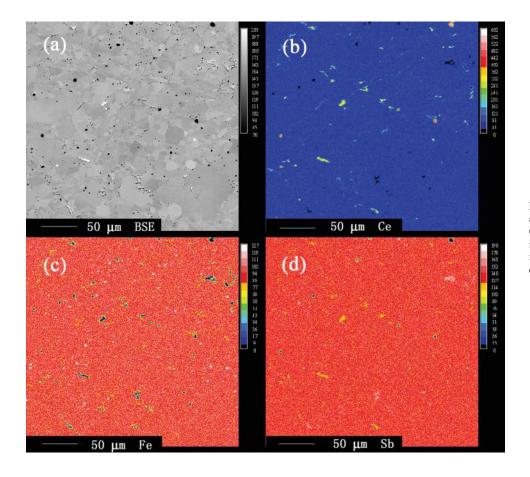


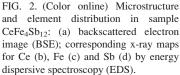
FIG. 1. XRD patterns for (a) $CaFe_4Sb_{12}$, (b) $SrFe_4Sb_{12}$, (c) $BaFe_4Sb_{12}$, (d) $LaFe_4Sb_{12}$, (e) $CeFe_4Sb_{12}$, (f) $PrFe_4Sb_{12}$, (g) $NdFe_4Sb_{12}$, (h) $EuFe_4Sb_{12}$, and (i) $YbFe_4Sb_{12}$.

that all samples contain trace amount of RSb₂ and Sb (less than 3%), which is not detected by XRD. Figure 2 shows the microstructure and element distribution of CeFe₄Sb₁₂. The energy dispersive spectroscopy (EDS) results indicate that all elements are distributed homogeneously in the matrix. The actual chemical compositions for all RFe₄Sb₁₂ samples determined by EPMA are shown in Table I. The final filling fraction of the fillers in all samples are less than 100%, being consistent with the published literatures.^{32,36,38} Although metastable Fe₄Sb₁₂-based filled skutterudites with the entire sequence of lanthanides have been prepared using multilayer precursor approach,⁸ our attempt to synthesize lanthanides from Gd to Tm filled skutterudites using normal meltingquenching-annealing method was unsuccessful. The possible reason could be that the chemical bonds between the fillers and [Fe₄Sb₁₂] framework are too weak to form thermodynamically stable materials under equilibrium conditions because of their very small ionic radius.²⁸

2. Lattice constant

Tang et al. reported that $Fe_xCo_{4-x}Sb_{12}$ is stable in the composition range of $x = 0 \sim 0.7$,³⁹ and the lattice constant (a) of this system obeys Vegard's law. The lattice constant of a "Fe₄Sb₁₂" unit (a_0) could be extrapolated to be 9.125 Å under Vegard's law based on the reported data.^{1,39} Table I shows a of all RFe₄Sb₁₂ samples determined by an analytical extrapolation function of $(\cos^2\theta/\sin\theta + \cos^2\theta/\theta)/2$ using high-angle XRD data. Obviously, the increased a values in RFe_4Sb_{12} are due to the extra lattice expansion by filling guest atoms into the voids in skutterudites. In order to identify the relationship, ionic radii of the fillers (r_{ion}) dependence of a for all RFe₄Sb₁₂ samples is shown in Fig. 3. Here, we use the r_{ion} values of the fillers with a coordinate number of 12 for alkaline earths $(Ca^{2+}, Sr^{2+} \text{ and } Ba^{2+})$ and lantha-nides $(La^{3+}, Ce^{3+} \text{ and } Nd^{3+})$.⁴⁰ There are no 12-coordinated r_{ion} values for Pr^{3+} , Eu^{2+} and Yb^{2+} , and the data used here are the extrapolated results using the relationship between the coordination number and effective r_{ion} in Ref. 40. Although the oxidation states could be +3 for both Eu and Yb, it had been reported and confirmed that the valence





charge states for Eu and Yb in skutterudites are close to +2 in filled skutterudites.^{28,31,41,42} These relatively large r_{ion} values of divalent Eu and Yb ion may enable Eu and Yb to form stable filled skutterudites.⁴³ For comparison, the reported *a* values for NaFe₄Sb₁₂, KFe₄Sb₁₂ and SmFe₄Sb₁₂ in literatures^{36,38} are also included in Fig. 3. A reasonably good correlation between expanded lattice constant and ion radius in RFe₄Sb₁₂ samples was observed. The lattice constant of the filled materials increases monotonously with increasing r_{ion} because larger fillers could expand lattice structure more significantly. The lattice constant of SmFe₄Sb₁₂ is around 9.1298 Å, close to the calculated value of "Fe₄Sb₁₂." If the ionic radii of the filler is less than Sm, the lattice compara-

ble or smaller than "Fe₄Sb₁₂", leading the filled structure thermodynamically unstable. This is further confirmed by our data that the synthesis of filled skutterudites with the fillers from Gd to Tm is unsuccessful. In addition, our data shows that the lattice constant in RFe_4Sb_{12} is mainly determined by the ionic radius of the fillers, but not sensitive to the filler valence state (see Fig. 3).

Figure 3 also shows the relative increment of lattice constant ($\Delta a = (a - a_0)/a_0$) as a function of ionic radii of the fillers. Δa is less than 1% among all the fully filled skutterudites studied here. The slight deviation of lattice constant might scarcely affect the symmetry of the crystal structure, and thus exert little influence on the electrical transport properties. This provides a platform to discuss the effect of

TABLE I. Nominal and actual chemical composition, lattice constant, calculated chemical carrier concentration, and room temperature Hall carrier concentration, Seebeck coefficient, electrical conductivity, and lattice thermal conductivity for all RFe₄Sb₁₂ samples.

Nominal composition	EMPA composition	Lattice constant (Å)	Chemical carrier concentration $(10^{21} \times \text{cm}^{-3})$	$1/R_{H}e$ (10 ²¹ × cm ⁻³)	S (μV/K)	$\sigma (\times 10^5 \mathrm{Sm}^{-1})$	$K_L (\mathrm{Wm}^{-1}\mathrm{K}^{-1})$
LaFe ₄ Sb ₁₂	La _{0.89} Fe ₄ Sb _{12.02}	9.148	3.53	1.38	78.3	2.44	1.63
CeFe ₄ Sb ₁₂	Ce _{0.91} Fe ₄ Sb _{11.97}	9.140	3.25	3.86	79.4	2.23	1.28
PrFe ₄ Sb ₁₂	Pr _{0.90} Fe ₄ Sb _{12.00}	9.138	3.41	7.19	81.5	2.40	1.09
NdFe ₄ Sb ₁₂	Nd _{0.85} Fe ₄ Sb _{12.00}	9.136	3.47	4.66	83.7	2.37	0.96
EuFe ₄ Sb ₁₂	Eu _{0.96} Fe ₄ Sb _{11.94}	9.170	5.24	3.43	70.8	2.97	1.79
YbFe ₄ Sb ₁₂	Yb _{0.94} Fe ₄ Sb _{12.07}	9.159	5.70	2.54	77.9	3.36	1.18
CaFe ₄ Sb ₁₂	Ca _{0.98} Fe ₄ Sb _{12.08}	9.159	5.52	1.34	73.9	3.14	1.61
SrFe ₄ Sb ₁₂	Sr _{0.94} Fe ₄ Sb _{12.08}	9.181	5.68	3.36	72.5	3.05	2.07
$BaFe_4Sb_{12}$	$Ba_{0.92}Fe_4Sb_{12.04}$	9.201	5.64	1.61	69.4	3.08	2.15

Downloaded 22 Apr 2011 to 131.215.220.185. Redistribution subject to AIP license or copyright; see http://jap.aip.org/about/rights_and_permissions

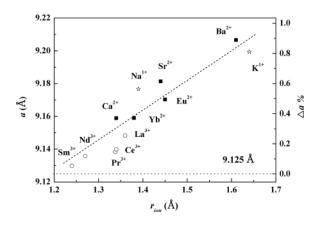


FIG. 3. Lattice constant and the relative increment of lattice constant Δa ($\Delta a = (a - a_0)/a_0$) as a function of ionic radii (12-coordinated) of the fillers in RFe₄Sb₁₂ (R = Ca, Sr, Ba, La, Ce, Pr, Nd, Eu, and Yb). Dashed line is guide to the eyes.

different fillers on the electrical transport properties in the following part.

B. Electrical transport properties

The temperature dependence of electrical conductivity (σ) and Seebeck coefficient (*S*) are shown in Fig. 4. Large σ values, with the magnitude of 10⁵ S/m, are observed in all samples. In the whole temperature range investigated here, σ for all samples decreases monotonously with increasing temperature, which is typical heavily doped semiconducting behavior. All samples exhibit positive *S* values, showing that the majority of charge carriers are holes. Since Fe atom has one electron less than Co atom, a "Fe₄Sb₁₂" unit is 4-electron

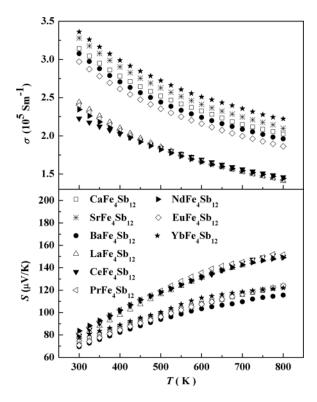


FIG. 4. Temperature dependence of electrical conductivity and Seebeck coefficient for all RFe_4Sb_{12} samples (R = Ca, Sr, Ba, La, Ce, Pr, Nd, Eu, and Yb).

deficiency as compared with the binary semiconducting skutterudite Co₄Sb₁₂. Though the fillers in the voids donate their almost all valence electrons to the skutterudite framework, it is still insufficient to compensate the holes in fully filled skutterudites RFe₄Sb₁₂ because the oxidation states of the fillers are usually around +2 or +3. Hence, RFe₄Sb₁₂ system is expected to possess high hole density and positive S value. Table I shows the measured Hall carrier concentration p_H $(p_H = 1/R_H e$, where e is the elementary charge) at room temperature for all samples. Here, the Hall number is assumed to be 1 in the single carrier model. All samples are hole dominated with the carrier concentration on the order of 10^{21} cm⁻³. The measured hole concentrations obviously deviate from the calculated values by simple charge counting based on the measured chemical compositions. Similar phenomenon has been also reported by Schnelle et al.³⁶ One possible reason might be related to the multiple bands near the Fermi level in RFe₄Sb₁₂ materials.³⁶ In this case, the understanding of the measured Hall coefficient is very complicated (there is no simple relationship between free carrier concentration pand R_H) and would require further systematic and detailed study. In addition, strong magnetic effects in RFe₄Sb₁₂ compounds may also contribute certainly to the anomalous Hall effect.³⁴ Therefore, in the following part, the calculated chemical carrier concentration (Table I) using simple electron counting are used for further discussion.

As shown in Fig. 4, RFe_4Sb_{12} with divalent fillers (R = Ca, Sr, Ba, Eu, and Yb) exhibit higher σ and lower *S* than those in the trivalent lanthanide (R = La, Ce, Pr, and Nd) filled skutterudites. Because the divalent fillers in the voids donate one electron less than the trivalent fillers, the filled skutterudites (RFe_4Sb_{12}) with divalent fillers will possess higher hole densities and σ , and lower *S* values. The maximum *S* values at high temperature in RFe_4Sb_{12} with divalent fillers are still higher than 100 μ V/K. These high *S* values in Fe_4Sb_{12} -based skutterudites might be related to the large effective mass of the holes due to the localized flat features of Fe-3d states near the Fermi level.^{4,44}

The electronic structure calculation by L. Nordström et al. suggested that CeFe₄Sb₁₂ is an intrinsic semiconductor where the 4f electron of Ce is completely donated into $[Fe_4Sb_{12}]$ framework.^{4,44} Simple electron counting also shows CeFe₄Sb₁₂ should be a semiconductor and the effective charge state of Ce is +4 if Ce-4f electron is completely transferred into [Fe₄Sb₁₂] framework. However, our data indicate that CeFe₄Sb₁₂ exhibits metallic behavior with very high electrical conductivity, i.e., σ decreases with increasing temperature. This is consistent with other experimental results reported previously.^{3,21} Therefore, experimental data indicate that the effective charge state of Ce is not +4 in fully filled skutterudites, inconsistent with the theoretical calculations in Ref. 4. Hall measurement for Ce filled skutterudites by Chen et al. also demonstrated that the effective charge state of Ce is about +3,⁷ similar to the value in ntype filled skutterudites.⁴⁵ The difference between the experiment and theoretical calculations might be due to the difficulty of dealing with the strong-correlated 4f electron of Ce in the local density approximation. This problem may also exist in other p-type lanthanides filled skutterudites.

Interestingly, as shown in Fig. 4, RFe₄Sb₁₂ with trivalent fillers (R = La, Ce, Pr, and Nd) exhibit very similar σ and S values in the whole temperature range. In addition, RFe₄Sb₁₂ with divalent fillers (R = Ca, Sr, Ba, Eu, and Yb) also show similar σ and S values although the deviations are large. In general, σ and S are predominantly determined by the hole concentration and band structure near the Fermi level. Our data likely indicate that the fillers, particularly lanthanides La, Ce, Pr, and Nd, show similar effect on the band edge near the Fermi level. Then, the similar electronic structure of these several lanthanides filled RFe4Sb12 would lead to small difference in the observed σ and S. The band structures of CeFe₄Sb₁₂ and LaFe₄Sb₁₂ were calculated by K. Nouneh et al.⁴⁴ It was shown that the primary valence band of both CeFe₄Sb₁₂ and LaFe₄Sb₁₂ are the hybridized Sb-5p orbital and Fe-3d orbital. The 4f and 5d orbitals of Ce only contribute to the conduction band. Optical reflectivity measurement and electronic structure calculation in RFe_4Sb_{12} (R = Ca, Ba, and Yb) Ref. 35 also suggested that they possess similar valence band structure. All these results seem to be consistent with our arguments mentioned above. Systematic study on the band structures in RFe₄Sb₁₂ is needed to get a fully comprehensive understanding of the observed experimental results.

Figure 5 shows the power factor $(PF = S^2\sigma)$ as a function of temperature for all samples. Due to the comparable electrical transport properties, RFe₄Sb₁₂ with the trivalent fillers possess similar *PF* values in the whole temperature range. Except YbFe₄Sb₁₂, *PF* values of RFe₄Sb₁₂ with divalent fillers are lower than those with trivalent fillers because of their smaller *S* values. The maximum value around 34 μ W/cmK² is obtained in CeFe₄Sb₁₂, PrFe₄Sb₁₂, and YbFe₄Sb₁₂ at 750 K.

C. Thermal transport properties

35

1. Thermal conductivity and lattice thermal conductivity

Figure 6(a) shows the temperature dependence of thermal conductivity (κ) for all RFe₄Sb₁₂ samples. Here, the measured specific heat (C_p) of CeFe₄Sb₁₂ is used for calcu-

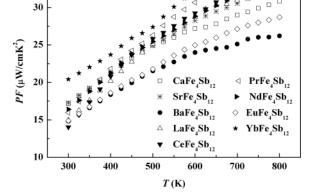


FIG. 5. Temperature dependence of power factor for all RFe_4Sb_{12} samples (R = Ca, Sr, Ba, La, Ce, Pr, Nd, Eu, and Yb).

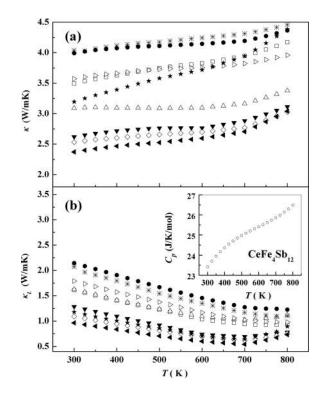


FIG. 6. Temperature dependence of (a) thermal conductivity and (b) lattice thermal conductivity for all RFe_4Sb_{12} samples ($R = Ca: \Box$, Sr: *, $Ba: \bullet$, $La:\Box$, $Ce: \blacktriangledown$, $Pr: \lhd$, $Nd: \blacktriangleright$, $Eu:\Box$, and $Yb: \Box$). The inset shows the specific heat for sample $CeFe_4Sb_{12}$ above room temperature.

lating κ for all samples because of their almost identical C_p values above room temperature. As shown in the insert of Fig. 6, C_p of sample CeFe₄Sb₁₂ increases with increasing temperature. It reaches 26.2 J/g/mol at 800 K, about 15% higher than that at 300 K. In the whole temperature range investigated here, all the κ values increase with increasing temperature. The total κ is a sum of two contributions: lattice thermal conductivity (κ_L) and carrier thermal conductivity (κ_c). According to the Wiedemann–Franz law, κ_c can be estimated by $\kappa_c = L_0 T \sigma$ under single carrier approximation (where the Lorenz number L_0 has an empirical value of $2.0 \times 10^{-8} \text{ V}^2 \text{K}^{-2}$ for skutterudite compounds^{9,11}). Consequently, κ_L can be obtained by subtracting the carrier part from the measured κ . This assessment of κ_L is only an approximation due to the lack of precise L_0 value and its uncertain temperature dependence for this degenerate semiconducting system. However, this method can provide a mechanism for comparing the existing κ_L values for all RFe_4Sb_{12} samples. Figure 6(b) shows the temperature dependence of κ_L for all samples. The lowest κ_L value in RFe₄Sb₁₂ is around 1.0 W/mK at room temperature and 0.6 W/mK at 700 K, comparable to or lower than α -SiO₂. One possible mechanism for such low κ_L in filled skutterudites is the "rattling" effect of the loosely bonded guest fillers in the skutterudite voids. These guest fillers interact strongly with the low-frequency acoustic phonons, leading to partial energy of the phonons being trapped in the excited rattling state. This part of phonons would later decay into phonons incoherent with those of the absorbed phonons, and thus contribute to the lowed κ_L .^{17,46,47}

As shown in Fig. 6(b), below 700 K, κ_L for all samples decreases with increasing temperature, indicating that phonon transport is hindered by the dynamical process of Umklapp scattering. κ_L for RFe₄Sb₁₂ with trivalent fillers (R = La, Ce, Pr, and Nd) reach the minimum values around 700 K, then slightly increase with increasing temperature. This is due to the additional heat carried by the electron-hole pair generated from intrinsic excitation at high temperatures (bipolar diffusion). In fact, bipolar diffusion is an inherent feature in p-type filled skutterudites because of its narrow bandgap. In order to improve ZT in p-type filled skutterudites, one would like to reduce the bipolar diffusion effect to make κ_L continually decreasing at high temperature. One possible method is to increase the hole concentration to shift the turning point of κ_L to higher temperature. As shown in Fig. 6(b), κ_L for RFe₄Sb₁₂ (R = Ca, Sr, Ba, and Eu) decreases monotonously in the entire temperature range due to their larger hole concentration. However, the enhanced carrier concentration definitely increases κ_c and deteriorates S simultaneously, which could lower the overall thermoelectric performance. Therefore, other approaches (e.g., enlarging bandgap) are recommended to reduce the effect of bipolar diffusion to achieve higher ZT values.

2. Correlation between lattice thermal conductivity and ionic radii of the trivalent fillers

Since all the samples have similar filling fraction (y) (as shown in Table I), it is possible to evaluate the effect of different fillers on the κ_L reduction systematically. In filled skutterudites, based on the assumption that the effect of the additional κ_L reduction by the fillers is a localized behavior,⁴⁸ total lattice thermal resistivity W_L (= 1/ κ_L) could be written as $W_L = W_{matrix} + W_{PD} + W_R$, where W_{matrix} is the thermal resistivity of the $[Fe_4Sb_{12}]$ framework, W_{PD} is the thermal resistivity contributed by point defect, and W_R is the thermal resistivity contributed by resonant scattering due to the local rattling of guest fillers, respectively. In principle, W_{PD} includes the effect of structure strain (W_S) and mass fluctuation (W_M) between the void and filler.⁴⁹ Here, the data are compared among these nearly fully filled skutterudites in an attempt to get a systematic understanding on the κ_L reduction. The additional effect of strain fluctuation on the κ_L reduction might be neglected because of the small deviation (less than 1%) of lattice constant among all the fully filled skutterudites mentioned above. Since the atomic weight (M) for the rare earth atoms La, Ce, Pr, Nd, and Eu are very close, W_M for RFe₄Sb₁₂ (R = La, Ce, Pr, Nd and Eu) are also similar to each other. Therefore, in these fully filled skutterudites, phonon resonant scattering might be the only mechanism to make κ_L values in RFe₄Sb₁₂ (R = La, Ce, Pr, Nd, and Eu) different with each other. As we know, the chemical bonds between the filler and skutterudite framework are mainly ionic,⁴⁵ thus the bond energy is approximately in proportion to $1/\Delta r$.⁵⁰ Thus, for a resonant scattering, W_R is almost in proportion to $1/(\Delta r)$ ³, where Δr is the distance between the positive charge of filler ion and the negative charge of its nearest Sb anion, and thus equal to $(r_{cage} - r_{ion})$. Here, r_{cage} is the Sb-icosahedron void radius in RFe₄Sb₁₂ and an average value of 1.94 Å is used in this paper.¹ Figure 7

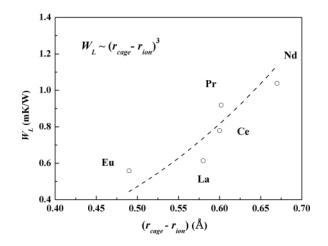


FIG. 7. Room temperature W_L as a function of $(r_{cage} - r_{ion})$ for RFe₄Sb₁₂ (R = La, Ce, Pr, Nd, and Eu). Dashed line represents the relationship of $W_L \sim (r_{cage} - r_{ion})$ (Ref. 3).

shows W_L at 300 K as a function of $(r_{cage} - r_{ion})$ for RFe₄Sb₁₂ (R = La, Ce, Pr, Nd, and Eu) samples. A good relationship of $W_L \sim (r_{cage} - r_{ion})$ (Ref. 3) is observed in these fully filled skutterudites. Larger $(r_{cage} - r_{ion})$ value indicates that the fillers are more loosely bonded with [Fe₄Sb₁₂] framework. The stronger "rattling" effect can greatly scatter heat carrying phonons, thereby enhancing W_L . For example, NdFe₄Sb₁₂ has the largest $(r_{cage} - r_{ion})$ value investigated here. At the same time, it has the highest W_L among all the samples.

D. Dimensionless thermoelectric figure of merit ZT

Figure 8 shows the dimensionless figure of merit (*ZT*) calculated based on the measured σ , *S* and κ . As a result of larger *S* and lower κ , RFe₄Sb₁₂ with trivalent fillers (R = La, Ce, Pr, and Nd) exhibit higher *ZT* values than those with divalent fillers (R = Ca, Sr, Ba, Eu, and Yb). CeFe₄Sb₁₂, PrFe₄Sb₁₂, and NdFe₄Sb₁₂ exhibit the maximum *ZT* value, about 0.87 at 750 K.

Figure 9 plots the ZT values as a function of the filler ionic radii for all RFe_4Sb_{12} samples at 300 K and 800 K,

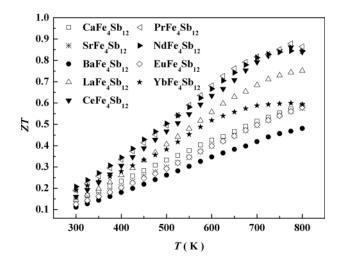


FIG. 8. Temperature dependence of ZT for all RFe_4Sb_{12} compounds (R = Ca, Sr, Ba, La, Ce, Pr, Nd, Eu, and Yb).

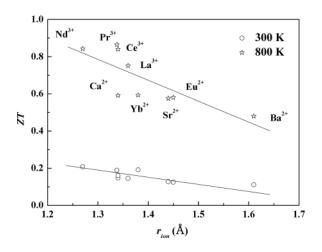


FIG. 9. *ZT* as a function of r_{ion} for all RFe₄Sb₁₂ compounds (R = Ca, Sr, Ba, La, Ce, Pr, Nd, Eu, and Yb) at 300 K and 800 K. The dashed lines are guide to the eyes.

respectively. A general trend, i.e., small r_{ion} corresponds with higher ZT, can be observed. This result might provide a straightforward image to design or search novel compositions to realize high ZT values in filled skutterudites.

IV. CONCLUSION

In summary, a systematic study has been carried out on RFe₄Sb₁₂ skutterudites filled with alkaline earth (Ca, Sr, and Ba) and lanthanides metals (La, Ce, Pr, Nd, Eu, and Yb). For all RFe_4Sb_{12} samples, lattice constant *a* increases with increasing r_{ion} . σ , S and κ for all samples are measured in the temperature range from 300 to 800 K. Since divalent fillers donate two electrons and trivalent fillers donate three electrons to compensate the holes in $[Fe_4Sb_{12}]$ framework, RFe_4Sb_{12} (R = Ca, Sr, Ba, Eu, and Yb) show large σ and small S values as compared with those in RFe₄Sb₁₂ with trivalent fillers (R = La, Ce, Pr, and Nd). RFe₄Sb₁₂ skutterudites with similar filler valence charge state exhibit very close σ and S values. The maximum power factor value around 34 μ W/cmK² is obtained for CeFe₄Sb₁₂, PrFe₄Sb₁₂, and YbFe₄Sb₁₂ at 750 K. All samples possess low κ_L due to the "rattling" effect of the fillers, which could scatter lowfrequency heat carrying phonons. For RFe₄Sb₁₂ with trivalent fillers (R = La, Ce, Pr, and Nd), the slight increase of κ_L above 700 K can be explained by the bipolar diffusion caused by the intrinsic excitation, which is not observed in RFe_4Sb_{12} with divalent fillers (R = Ca, Sr, Ba, Eu, and Yb). The effect of the ionic radii of the fillers on thermal transport properties is discussed. It is found that κ_L decreases with increasing r_{ion} for samples RFe₄Sb₁₂ (R = La, Ce, Pr, Nd, and Eu). A maximum ZT value of 0.87 is obtained in CeFe₄Sb₁₂, PrFe₄Sb₁₂ and NdFe₄Sb₁₂. In addition, an empirical trend between the ionic radii of the fillers and the thermoelectric figure of merit is proposed. Fillers with small r_{ion} correspond to higher ZT values. This provides a straightforward trend to choose fillers to realize high ZT values in skutterudites. Our work offers a systematic understanding on the thermoelectric transport properties in RFe₄Sb₁₂ and presents much useful information for further design and optimization in Fe_4Sb_{12} -based skutterudites.

ACKNOWLEDGMENTS

This work was partly supported by National Natural Science Foundation of China (Nos. 50802109, 50820145203, 50825205), and the Visiting Professor and Knowledge Innovation Programs of the Chinese Academy of Sciences (Contract No. KJCX2-YW-H20).

- ¹L. D. Chen, *Proceedings of the 21st International Conference of Thermoelectrics* (IEEE, New York, 2002), p. 42.
- ²B. C. Sales, D. Mandrus, and R. K. Williams, Science **272**, 1325 (1996).
- ³D. T. Morelli and G. P. Meisner, J. Appl. Phys. **77**, 3777 (1995).
- ⁴L. Nordström, D. J. Singh, Phys. Rev. B 53, 1103 (1996).
- ⁵J.-P. Fleurial, A. Borshchevsky, and T. Caillat, *Proceedings of the 15th International Conference of Thermoelectrics* (IEEE, New York, 1996), p. 91.
- ⁶J.-P. Fleurial, T. Caillat, and A. Borshchevsky, *Proceedings of the 16th International Conference of Thermoelectrics* (IEEE, New York, 1997), p. 1.
- ⁷B. X. Chen, J. H. Xu, C. Uher, D. T. Morelli, G. P. Meisner, J. -P. Fleurial, T. Caillat, and A. Borshchevsky, *Phys. Rev. B* **55**, 1476 (1997).
- ⁸M. D. Hornbostel, E. J. Hyer, J. H. Edvalson, and D. C. Johnson, Inorg. Chem. **36**, 4270 (1997).
- ⁹B. C. Sales, D. Mandrus, B. C. Chakoumakos, V. Keppens, and J. R. Thompson, Phys. Rev. B 56, 15081 (1997).
- ¹⁰D. M. Rowe, V. L. Kuznetsov, and L. A. Kuznetsova, *Proceedings of the 17st International Conference of Thermoelectrics* (IEEE, New York, 1998), p. 323.
- ¹¹X. F. Tang, L. D. Chen, T. Goto, and T. Hirai, J. Mater. Res. 16, 837 (2001).
- ¹²X. F. Tang, L. D. Chen, T. Goto, T. Hirai, and R. Z. Yuan, J. Mater. Sci. **36**, 5435 (2001).
- ¹³G. S. Nolas, H. Takizawa, T. Endo, H. Sellinschegg, and D. C. Johnson, Appl. Phys. Lett. **77**, 52 (2000).
- ¹⁴L. D. Chen, T. Kawahara, X. F. Tang, T. Goto, T. Hiral, J. S. Dyck, W. Chen, and C. Uher, J. Appl. Phys. **90**, 1864 (2001).
- ¹⁵V. L. Kuznetsov, L. A. Kuznetsova, and D. M. Rowe, J. Phys. Condens. Matter. 15, 5035 (2003) (and references therein).
- ¹⁶Y. Z. Pei, L. D. Chen, W. Zhang, X. Shi, S. Q. Bai, X. Y. Zhao, Z. G. Mei, and X. Y. Li, Appl. Phys. Lett. 89, 221107 (2006) (and references therein).
- ¹⁷J. Yang, W. Zhang, S. Q. Bai, Z. Mei, and L. D. Chen, Appl. Phys. Lett. 90, 192111 (2007).
- ¹⁸S. Q. Bai, Y. Z. Pei, L. D. Chen, W. Q. Zhang, X. Y. Zhao, and J. Yang, Acta Mater. **57**, 3135 (2009).
- ¹⁹X. Shi, H. Kong, C-P. Li, C. Uher, J. Yang, J. R. Salvador, H. Wang, L. D. Chen, and W. Zhang, Appl. Phys. Lett. **92**, 182101 (2008).
- ²⁰X. Shi, J. Yang, J. R. Salvador, M. Chi, J. Y Cho, H. Wang, S. Bai, J. Yang, W. Zhang, and L. Chen, Multiple-filled skutterudites: High thermoelectric figure of merit through separately optimizing electrical and thermal transports, submitted.
- ²¹D. Bérardan, E. Alleno, C. Godart, M. Puyet, B. Lenoir, R. Lackner, E. Bauer, L. Girard, and D. Ravot, J. Appl. Phys. 98, 033710 (2005).
- ²²X. F. Tang, H. Li, Q. J. Zhang, M. Niino, and T. Goto, J. Appl. Phys. 100, 123702 (2006).
- ²³P. N. Alboni, X. Ji, J. He, N. Gothard, and T. M. Tritt, J. Appl. Phys. 103, 113707 (2008).
- ²⁴G. Rogl, A. Grytsiv, E. Bauer, P. Rogl, and M. Zehetbauer, Intermetallic 06, 005 (2009).
- ²⁵D. J. Braun and W. Jeitschko, J. Less-Common Met. 72, 147 (1980).
- ²⁶C. B. H. Evers, W. Jeitschko, L. Boonk, D. J. Braun, Th. Ebel, and U. D. Scholz, J. Alloys. Comp. **184**, 224 (1995).
- ²⁷V. Keppens, D. Mandrus, B. C. Sales, B. C. Chakoumakos, P. Dai, R. Coldea, M. B. Maple, D. A. Gajewski, E. J. Freeman, and S. Bennington, Nature **395**, 876 (1998).
- ²⁸N. R. Dilley, E. J. Freeman, E. D. Bauer, and M. B. Maple, Phys. Rev. B 58, 6287 (1998).
- ²⁹J. W. Kaiser and W. Jeitschko, J. Alloys. Comp. **291**, 66 (1999).
- ³⁰N. R. Dilley, E. D. Bauer, M. B. Maple, S. Dordevic, D. N. Basov, F. Freibert, T. W. Darling, A. Migliori, B. C. Chakoumakos, and B. C. Sales, Phys. Rev. B 61, 4608 (2000).

- ³¹V. L. Kuznetsov and D. M. Rowe, J. Phys. Condens. Matter. **12**, 7915 (2000).
- ³²E. Bauer, St. Berger, Ch. Paul, M. Della Mea, G. Hilscher, H. Michor, M. Reissner, W. Steiner, A. Grytsiv, P. Rogl, and E. W. Scheidt, Phys. Rev. B 66, 214421 (2002).
- ³³A. Leithe-Jasper, W. Schnelle, H. Rosner, M. Baenitz, A. Rabis, A. A. Gippius, E. N. Morozova, H. Borrmann, U. Burkhardt, R. Ramlau, U. Schwarz, J. A. Mydosh, and Y. Grin, Phys. Rev. B 70, 214418 (2004).
- ³⁴B. C. Sales, R. Jin, D. Mandrus, and P. Khalifah, Phys. Rev. B 73, 224435 (2006).
- ³⁵J. Sichelschmidt, V. Voevodin, H. J. Im, S. Kimura, H. Rosner, A. Leithejasper, W. Schnelle, U. Burkhardt, J. A. Mydosh, Yu. Grin, and F. Steglich, Phys. Rev. Lett. **96**, 037406 (2006).
- ³⁶W. Schnelle, A. Leithe-Jasper, H. Rosner, R. Cardoso-Gil, R. Gumeniuk, D. Trots, J. A. Mydosh, and Y. Grin, Phys. Rev. B 77, 094421 (2008).
- ³⁷M. M. Koza, M. R. Johnson, R. Viennois, H. Mutka, L. Girard, and D. Ravot, Nat. Mater. 7, 805 (2008).
- ³⁸M. Ueda, Y. Kawahito, K. Tanaka, D. Kikuchi, H. Aoki, H. Sugawara, K. Kuwahara, Y. Aoki, and H. Sato, Physica B **403**, 881 (2008).
- ³⁹X. F. Tang, L. D. Chen, T. Goto, and T. Hirai, J. Jpn. Inst. Metals. 63, 1412 (1999).

- ⁴⁰R. D. Shannon, Acta Cryst. A32, 751 (1976).
- ⁴¹A. Grytsiv, P. Rogl, St. Berger, Ch. Paul, E. Bauer, B. Ni, M. M. Abd-Elmeguid, A. Saccone, R. Ferro, and D. Kaczorowski, Phys. Rev. B 66, 094411 (2002).
- ⁴²J. Yang, L. Xi, W. Zhang, L. D. Chen, and J. Yang, J. Electron Mater. 38, 1397 (2009).
- ⁴³Z. G. Mei, J. Yang, Y. Z. Pei, W. Zhang, L. D. Chen, and J. Yang, Phys. Rev. B 77, 045202 (2008).
- ⁴⁴K. Nouneh, Ali. H. Reshak, S. Auluck, I. V. Kityk, R. Viennois, S. Benet, and S. Charar J. Alloys. Compd. **437**, 39 (2007).
- ⁴⁵X. Shi, W. Zhang, L. D. Chen, J. Yang, and C. Uher, Phys. Rev. B 75, 235208 (2007).
- ⁴⁶G. J. Long, R. P. Hermann, F. Grandjean, E. E. Alp, W. Sturhahn, C. E. Johnson, D. E. Brown, O. Leupold, and R. Rüffer, Phys. Rev. B 71, 140302 (2005).
- ⁴⁷E. R. Grannan, M. Randeria, and J. P. Sethna, Phys. Rev. B **41**, 7799 (1990).
- ⁴⁸J. L. Cohn, G. S. Nolas, V. Fessatidis, T. H. Metcalf, and G. A. Slack, Phys. Rev. Lett. 82, 779 (1999).
- ⁴⁹G. A. Slack, Phys. Rev. **105**, 829 (1957).
- ⁵⁰C. Kittel, in *Introduction to Solid State Physics*, 8th ed. (John Wiley and Sons, New York, 2005), p. 63.