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Highly Oxygenated Limonoids and Lignans from *Phyllanthus flexuosus*



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Abstract Two new highly oxygenated limonoids, flexuosoids A (1) and B (2), and three new arylnaphthalene lignan glycosides, phyllanthusmins D–F (3–5), were isolated from the roots of *Phyllanthus flexuosus*, in addition to three known lignans, phyllanthusmin C, arabelline, and (+)-diasyringaresinol. Their structures were elucidated on the basis of detailed spectroscopic analysis and chemical methods. Compounds 1 and 2, two new decaoxygenated limonoids with a C-19/29 lactol bridge and heptaoxygenated substituents at C-1, C-2, C-3, C-7, C-11, C-17, and C-30, represent the second example of limonoids in the Euphorbiaceae family. Most of the isolates were tested for their antifeedant, anti-herpes simplex virus 1, and cytotoxic activities. The new limonoids 1 and 2 showed promising antifeedant activity against the beet army worm (*Spodoptera exigua*) with EC₅₀ values of 25.1 and 17.3 μ g/cm², respectively. In addition, both of them displayed moderate cytotoxicity against the ECA109 human esophagus cancer cell line, along with the known lignan glycoside, phyllanthusmin C, with the IC₅₀ values of 11.5 (1), 8.5 (2), and 7.8 (phyllanthusmin C) μ M, respectively.

Keywords Phyllanthus flexuosus · Euphorbiaceae · Limonoids · Lignan glycosides · Antifeedant · Antiviral · Cytotoxicity

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1 Introduction

The genus *Phyllanthus* (Euphorbiaceae), comprising about 600 species, is widespread throughout the tropical and subtropical countries of the world, with about 30 species growing in the South of China. Several species have been used as traditional medicines [1], and flavonoids, alkaloids, sesquiterpenoids, triterpenoids, lignans, and tannins have been reported from the genus [2–6]. Among these, some sesquiterpenoids showed a wide range of biological properties including cytotoxic and antiviral effects [5–7].

Phyllanthus flexuosus (Sieb. et Zucc.) Muell. Arg (Euphorbiaceac), a shrub up to 3 m height, mainly grows in the southern parts of the Yangtze River, People's Republic of China. The whole plants have been utilized medicinally for treating infectious diseases by the Dai people in Yunnan Province, China [8]. Chemical studies on the leaves and stem barks of *P. flexuosus* have revealed the occurrence of triterpenoids, tannins and coumarins [9–12]. Previously we



Fig. 1 Structures of compounds from the roots of P. flexuosus

obtained three phenylacetylene-bearing tricyclic diterpenes from this species, among which phyllanflexoids A was the first example of phenylacetylene-bearing 18-nor-diterpenoid glycoside [13]. As a part of our continuing research for new bioactive secondary metabolites from the genus *Phyllanthus* [6, 14–17], five new compounds, including two highly oxygenated limonoids, flexuosoids A (1) and B (2), and three arylnaphthalene lignan glycosides, phyllanthusmins D–F (3–5), were isolated from the roots of *P. flexuosus*, together with three known lignans. Their structures were elucidated by extensive spectroscopic analysis and chemical method. Most of the isolated compounds were evaluated for their antifeedant, anti HSV-1, and cytotoxic activities, and the results obtained are discussed herein.

2 Results and Discussion

The MeOH extract of the air-dried roots of *P. flexuosus* was applied to repeated column chromatography over

macroporous resin D101, Sephadex LH-20 and silica gel, followed by semi-preparative HPLC, to afford five new compounds (1–5), in addition to three known lignans. The known compounds were identified as phyllanthusmin C [18], arabelline [19], and (+)-diasyringaresinol [20] by comparison of their spectroscopic data with literature values (Fig. 1).

Compound 1 was obtained as a white amorphous powder. The negative HRESIMS (m/z 505.2072 [M – H]⁻, calcd for C₂₆H₃₃O₁₀, 505.2073) and the ¹³C NMR (DEPT) spectra of 1 indicated a molecular formula of C₂₆H₃₄O₁₀, requiring for 10 degrees of unsaturation. The IR spectrum of 1 indicated the presence of hydroxy (3424 cm⁻¹) and carbonyl (1691 cm⁻¹) groups. The ¹³C NMR (DEPT) data for 1 (Table 1) revealed 26 carbon signals, including characteristic signals due to a hemiacetal ring unit (δ_C 97.36 and 64.81) [21], a β -substituted furan ring (δ_C 128.55, 141.25, 110.63 and 144.33) [22], a trisubstituted double bond (δ_C 153.48 and 121.76) and a ketone (δ_C 214.73). The hemiacetal ring [H₂-19: δ_H 4.20 and 4.52

Table 1 ¹³C NMR spectroscopic data of 1 and 2 (methanol- d_4 , δ in ppm)

Position	1^{a}		2^{a}		
_	29- <i>exo</i>	29-endo	29- <i>exo</i>	29-endo	
1	75.84, CH	75.76, CH	74.36, CH	74.36, CH	
2	68.60, CH	68.81, CH	67.88, CH	67.99, CH	
3	75.78, CH	79.27, CH	76.69, CH	79.83, CH	
4	42.96, C	43.17, C	42.65, C	42.58, C	
5	27.69, CH	25.06, CH	29.15, CH	27.76, CH	
6	25.62, CH ₂	27.77, CH ₂	25.70, CH ₂	$26.67, CH_2$	
7	69.54, CH	69.79, CH	69.61, CH	69.88, CH	
8	52.89, C	52.95, C	52.96, C	53.01, C	
9	46.06, CH	46.36, CH	46.06, CH	46.39, CH	
10	43.41, C	43.51, C	43.41, C	43.41, C	
11	214.73, C	214.89, C	214.71, C	214.89, C	
12	44.31, CH ₂	44.31, CH ₂	44.37, CH ₂ ,	44.52, CH ₂	
13	53.03, C	52.96, C	53.15, C	52.90, C	
14	153.48, C	153.55, C	153.55, C	153.60, C	
15	121.76, CH	121.69, CH	121.89, CH	121.80, CH	
16	44.45, CH ₂	44.39, CH ₂	44.59, CH ₂	44.45, CH ₂	
17	84.21, C	84.21, C	84.33, C	84.33, C	
18	26.12, CH ₃	26.05, CH ₃	26.21, CH ₃	26.15, CH ₃	
19	64.81, CH ₂	59.49, CH ₂	65.17, CH ₂	59.32, CH ₂	
20	128.55, C	128.55, C	128.79, C	128.79, C	
21	141.25, CH	141.25, CH	141.34, CH	141.34, CH	
22	110.63, CH	110.63, CH	110.73, CH	110.73, CH	
23	144.33, CH	144.33, CH	144.42, CH	144.42, CH	
28	20.45, CH ₃	19.73, CH ₃	19.75, CH ₃	19.32, CH ₃	
29	97.36, CH	97.09, CH	96.88, CH	96.80, CH	
30	71.92, CH ₂	72.01, CH ₂	72.04, CH ₂	72.08, CH ₂	
<u>CO</u> Me			173.31, C	173.25, C	
CO <u>Me</u>			21.23, CH ₃	21.18, CH ₃	

^a Data were measured at 125 MHz

(each 1H, J = 12.9 Hz); H-29: $\delta_{\rm H}$ 4.83] [21] and the β substituted furan ring protons ($\delta_{\rm H}$ 6.42, 7.52, 7.47) [22] were also distinguished from the ¹H NMR spectrum (Table 2), in addition to two methyl singlets at $\delta_{\rm H}$ 0.97 and 1.03. The aforementioned NMR data for 1 were closely related to those of 3α -deacetylamoorastatin [23]. However, a major difference between these two compounds was observed as the oxidation of a methyl and a methylene in 3α -deacetyl-amoorastatin core to an oxymethylene ($\delta_{\rm C}$ 71.92, $\delta_{\rm H}$ 3.60, 3.89 (each 1H, d, J = 10.7 Hz) and an oxymethine [$\delta_{\rm C}$ 68.60, $\delta_{\rm H}$ 4.54 (dd, J = 4.4, 3.4 Hz] in 1, respectively. Moreover, the 14,15-epoxide and the C-17 methane in 3α -deacetyl-amoorastatin were replaced with a $\Delta^{14(15)}$ double bond ($\delta_{\rm C}$ 153.48 and 121.76) and an oxygenbearing quaternary carbon ($\delta_{\rm C}$ 84.21, C-17) in 1, respectively. The above data were further assigned unambiguously by HSQC, ¹H-¹H COSY and HMBC analysis (Fig. 2

and Electronic supplementary material). The ¹H-¹H COSY spectrum verified the presence of -CH(O)(1)-CH(O)(2)-CH(O)(3)-, -CH(5)-CH₂(6)-CH(O)(7)-, and =CH(15)- $CH_2(16)$ fragments in 1 (bold lines in Fig. 2). The HMBC correlations (Fig. 2) from the oxymethylene at $\delta_{\rm H}$ 3.60 and 3.89 (H₂-30) to $\delta_{\rm C}$ 69.54 (C-7)/ $\delta_{\rm C}$ 52.89 (C-8)/ $\delta_{\rm C}$ 46.06 (C-9)/ $\delta_{\rm C}$ 153.48 (C-14), and from the oxymethine at $\delta_{\rm H}$ 4.54 (H-2) to $\delta_{\rm C}$ 75.84 (C-1)/ $\delta_{\rm C}$ 43.96 (C-4)/ $\delta_{\rm C}$ 43.41 (C-10) indicated that C-30 and C-2 of 1 were oxygenated, respectively. The $\Delta^{14(15)}$ double bond was confirmed by the HMBC correlations of H₂-12, H₂-16 and Me-18 to the olefinic carbon at $\delta_{\rm C}$ 153.48 (C-14). Moreover, HMBC correlations from H-15, H₂-16, Me-18, H-21 and H-22 to the oxygen-bearing quaternary carbon ($\delta_{\rm C}$ 84.21, C-17) revealed the hydroxylated C-17 of D ring, on which the furan ring was located in 1. Other HMBC correlations confirmed the planar structure of 1 as shown.

The relative stereochemistry of 1 was fixed by the proton coupling constants observed in the NMR spectrum and ROESY experiment (Fig. 3). The ROESY correlations of H-5/H-9, H-1/H-9, H-5/Me-28, H-3/Me-28, H-3/H-5, and H-2/H-19 supported the relative configuration of the fused A/B rings as shown in Fig. 1, and the hemiacetal ring between C-19 and C-29 was on the same side to those of 1-OH and 3-OH. The small values of $J_{1,2}$ (4.4 Hz) and $J_{2,3}$ (3.4 Hz) indicated the equatorial H-1, H-2 and H-3, corresponding to β -configurations for 1-OH, and 3-OH, and α configurations for 2-OH. In the ¹H NMR spectrum of **1**, the broad singlet H-7 suggested an α -configuration of 7-OH, which was further confirmed by ROESY correlations of H-29/H-19, H-19/H-30 and H-30/H-7. The 17-OH was assigned as β by a ROESY correlation of H-22 with Me-18. In addition, it is noted that compound 1 was isolated as a mixture of two tautomers in a ratio of 4:1 as estimated by ¹H NMR spectrum. The ¹H and ¹³C NMR spectral features of the mixture (Tables 1, 2) showed two sets of H-atom and C-atom resonances (partially overlapped) [24]. This could be explained by the equilibrium due to the hemiacetal unit. Acetylation of 1 yielded the 1,2,3,7,29,30-hexa-acetylated (1a) and 2,3,7,29,30-penta-acetylated (1b) adducts of 1, whose structures (Fig. 1) were characterized by detailed spectroscopic analysis (see Electronic supplementary material). The 29-exo configurations for both 1a and 1b were supported by their chemical shifts of H-3 ($\delta_{\rm H}$ 5.41), since H-3 resonated at $\delta_{\rm H}$ 4.9–5.1 for 29-endo and 5.3–5.6 for 29-exo [25]. Based on the above evidence, compound 1 was determined as shown in Fig. 1 and named as flexuosoid A.

Compound **2** was obtained as a white amorphous powder and possessed a molecular formula $C_{28}H_{36}O_{11}$, as deduced from the HRESIMS (*m*/*z* 547.2182 [M - H]⁻). The ¹H and ¹³C NMR data of **2** were similar to those of **1**, except for the appearance of an additional acetyl group [$\delta_{\rm H}$

Position	1 ^a		2ª		
	29- <i>exo</i>	29-endo	29- <i>exo</i>	29-endo	
1	4.35, d (4.4)	4.35, d (4.4)	4.37, d (4.6)	4.37, d (4.6)	
2	4.54, dd (4.4, 3.4)	4.50 ^b	4.59 ^b	4.54 ^b	
3	3.93, d (3.4)	3.42, d (3.2)	5.50, d (4.8)	4.97, d (4.8)	
5	2.63, dd (13.9, 3.0)	2.48 ^b	2.64, dd (10.9, 2.9)	2.71 ^b	
6	1.78, dd (13.7, 3.5)	1.81, br. d	1.76, dd (10.9, 3.5)	1.78 ^b	
	1.90, br. d (14.5)	2.48 ^b	1.88, br. d (14.1)	2.45 ^b	
7	4.05, br. d	4.02, br. d	4.04, br. d	4.00, br. d	
9	4.01, s	4.03, s	4.01, s	4.03, s	
12	1.99, d (19.2)	1.99, d (19.2)	3.25, d (19.3)	3.25, d (19.3)	
	3.25, d (17.9)	3.25, d (17.9)	1.89, d (19.1)	1.89, d (19.1)	
15	5.73, br. d	5.73, br. d	5.73, br. d	5.73, br. d	
16	2.52, dd (16.3, 3.3)	2.52, dd (16.3, 3.3)	2.51, dd, (16.3, 3.5)	2.51, dd, (16.3, 3.5)	
	3.25, d (17.9)	3.25, d (17.9)	3.24, d (16.2)	3.24, d (16.2)	
18	1.03, s	1.03, s	1.04, s	1.02, s	
19	4.20, d (12.9)	4.52 ^b	4.18, d (10.2)	4.38 ^b	
	4.52, d (12.9)	4.56 ^b	4.56 ^b	4.38 ^b	
21	7.52, s	7.52, s	7.53, s	7.53, s	
22	6.42, s	6.42, s	6.43, s	6.43, s	
23	7.47, s	7.47, s	7.47, s	7.47, s	
28	0.97, s	1.00, s	0.82, s	0.83, s	
29	4.83, s	4.62, s	4.80, s	4.72, s	
30	3.60, d (10.7)	3.61, d (9.8)	3.60, d (11.1)	3.63 ^b	
	3.89, d (10.7)	3.93, d (9.8)	3.87, d (11.1)	3.92, d (11.1)	
CO <u>Me</u>			2.09, s	2.08, s	

Table 2 ¹H NMR spectroscopic data of **1** and **2** (methanol- d_4 , δ in ppm)

^a Data were measured at 500 MHz

^b Overlapping ¹H NMR signals are reported without designated multiplicity

2.08/2.09 (3H, s) and $\delta_{\rm C}$ 173.31/173.25 (CO), 21.23/21.18 (CH₃)] and the down-field shift of C-3 and H-3 to $\delta_{\rm C}$ 76.69 and $\delta_{\rm H}$ 5.50, respectively, suggesting that the hydroxy group at C-3 in **1** was acetylated in **2**. This was confirmed by HMBC correlation of $\delta_{\rm H}$ 5.50 (H-3) with the acetyl carbonyl carbon at $\delta_{\rm C}$ 173.31 (Fig. 1S, Electronic

supplementary material). In the ROESY spectrum, correlations of H-5/Me-28, H-5/H-9, H-19/H-29 and H-22/Me-18 supported the relative configurations of rings system of



Fig. 2 Key $^{1}\text{H}^{-1}\text{H}$ COSY (–) and HMBC (\rightarrow) correlations of 1





Fig. 3 Key ROESY (\leftrightarrow) correlations of 1

Table 3 ¹H and ¹³C NMR spectroscopic data of **3** and **4** (methanol- d_4 , δ in ppm)

Position	3 ^a		4 ^b		
	$\delta_{\rm H} (J \text{ in Hz}) \delta_{\rm C}$		$\delta_{\rm H} (J \text{ in Hz}) \delta_{\rm C}$		
1		146.14/146.13 ^c , C		146.5, C	
2		119.85/119.76 ^c , C		120.71/120.72 ^c , C	
3		132.19/132.10 ^c , C		132.70/132.72 ^c , C	
4		137.47/137.40 ^c , C		137.5, C	
4a		131.69/131.66 ^c , C		132.3, C	
5	6.89 ^c , s	106.75/106.70 ^c , CH	7.13 ^c ,s	107.7, CH	
6		151.51/151.47°, C		151.7, C	
7		153.14/153.11°, C		150.6, C	
8	8.03, s	102.63/102.59 ^c , CH	8.25, s	106.5, C	
8a		128.64/128.61°, C		128.5, C	
1'		129.74/129.80 ^c , C		130.2, C	
2'	6.64, d (1.5)	111.65/111.71°, CH	6.81 ^c , d (1.7)	111.86/111.89°, CH	
3'		148.86/148.90 ^c , C		149.2, C	
4′		148.81/148.90 ^c , C		149.2, C	
5'	6.87 ^c , d (8.0)	108.84/108.88 ^c , CH	6.96 ^c , d (7.8)	109.18/109.20 ^c , CH	
6′	6.59 ^c , dd (8.0, 1.5)	124.71/124.74 ^c , CH	6.77 ^c , dd (7.8, 1.7)	124.9, CH	
2a	5.48°, d (15.2)	69.0, CH ₂	5.47 ^c , d (15.1)	69.4, CH ₂	
	5.39 ^c , d (15.2)		5.62 ^c , d (15.1)		
3a		172.09/172.05 ^c , C		172.3, C	
6-OMe	3.63 ^c , s	56.0, CH ₃			
7-OMe	3.94, s	56.7, CH ₃	3.76, s	56.3, CH ₃	
O-CH2-O	6.02 ^c , s	102.6, CH ₂	6.06 ^c , s	102.8, CH ₂	
	6.00 ^c , s		6.05 ^c , s		
1″	4.74, d (7.7)	106.9, CH	4.76 ^c , d (7.4)	107.3, CH	
2″	4.12, dd (8.6, 7.7)	72.2, CH	3.99, dd (9.3, 7.4)	72.9, CH	
3″	3.83 ^d	81.8, CH	3.65, dd (9.3, 3.5)	74.4, CH	
4″	5.16, br. s	72.5, CH	3.86, br. s	70.0, CH	
5″	3.98 ^d	65.4, CH ₂	3.50, d (12.5)	67.8, CH ₂	
	3.52 ^d		3.93, dt (12.5, 1.9)		
1‴	4.49, d (7.0)	106.5, CH	5.31, d (7.5)	101.3, CH	
2‴	3.66, dd (8.8, 7.0)	72.6, CH	3.60, dd (9.2, 7.6)	74.9, CH	
3‴	3.54 ^d	74.1, CH	3.55, t (9.2)	78.2, CH	
4‴	3.81, br. s	69.5, CH	3.41, t (9.2)	71.5, CH	
5‴	3.86 ^d	67.0, CH ₂	3.75 ^d	78.0, CH	
	3.52 ^d				
6‴			3.70, dd (11.9, 6.2)	62.8, CH ₂	
			3.97, dd (11.9, 2.0)		
<u>CO</u> Me		172.5, C			
CO <u>Me</u>	2.15, s	21.2, CH ₃			

^a Data were measured at 500 and 125 MHz for ¹H and ¹³C, respectively

 $^{\rm b}\,$ Data were measured at 600 and 150 MHz for $^1{\rm H}$ and $^{13}{\rm C},$ respectively

^c Existing as very close pairs at room temperature (21 °C)

^d Overlapping ¹H NMR signals are reported without designated multiplicity

2 as shown in Fig. 1. Moreover, the broad singlet H-7 in the ¹H NMR spectrum and the ROESY correlations of H-30/H-7 of **2** revealed that 7-OH had the α -configuration.

The ROESY correlation of H-22 with Me-18 supported the assignment of the β -configuration to 17-OH, and the correlations of H-1/H-9 and H-3/H-28 relieved the β -



Fig. 4 Key HMBC (\rightarrow) correlations of 3

configuration to 1-OH and 3-OH. Due to the severe overlapping of H₂-19 and H-2 signals in **2**, the assignment of α orientation for hydroxy group at C-2 was achieved by acetylation of **2** to give 1,2,3,7,29,30-hexa-acetyl flexuosoid A (**1a**) and 2,3,7,29,30-penta-acetyl flexuosoid A (**1b**). In the ¹H NMR spectrum of **1a**, the small values of $J_{1,2}$ (4.4 Hz) and $J_{2,3}$ (4.4 Hz) revealed the equatorial H-1, H-2, and H-3 in **2**. Furthermore, the 29-*exo* configuration for **2** was supported by chemical shifts of H-3 of **1a** and **1b**, the acetylated products of **2**. Thus, **2** was assigned as shown in Fig. **1** and named as flexuosoid B.

Compound 3, a white amorphous powder, possessed a molecular formula C₃₃H₃₄O₁₆, as deduced by the HRE-SIMS at m/z 685.1763 [M - H]⁻ (calcd for C₃₃H₃₃O₁₆, 685.1768). The IR spectrum showed the presence of hydroxy (3423 cm^{-1}) and benzene rings (1622, 1507, and)1435 cm⁻¹). The ¹³C NMR and DEPT spectra (Table 3) displayed the presence of 33 carbon resonances, due to two carboxyls ($\delta_{\rm C}$ 172.09 and 172.5), 16 aromatic carbons ($\delta_{\rm C}$ 102–153), one methylenedioxy ($\delta_{\rm C}$ 102.6), one oxymethylene ($\delta_{\rm C}$ 69.0), two methoxys ($\delta_{\rm C}$ 56.7 and 56.0), two pentosyl moieties and an acetyl methyl ($\delta_{\rm C}$ 21.2). In the ¹H NMR spectrum, two singlet aromatic protons at $\delta_{\rm H}$ 6.89 and 8.03 (each 1H, s) and a set of ABX coupled signals [$\delta_{\rm H}$ 6.87 (1H, d, J = 8.0 Hz), 6.64 (1H, d, J = 1.5 Hz), 6.59 (1H, dd, J = 8.0, 1.5 Hz)] arising from an 1,3,4-trisubstituted benzene ring were observed, in addition to characteristic proton signals due to a methylenedioxy ($\delta_{\rm H}$ 6.02 and 6.00, each 1H, s), two methoxys [$\delta_{\rm H}$ 3.94, 3.63 (each 3H, s)], and two anomeric protons [$\delta_{\rm H}$ 4.74 (d, J = 7.7 Hz) and 4.49 (d, J = 7.0 Hz)]. Acid hydrolysis of **3** afforded L-arabinose as the soly monosaccharide, which was confirmed by GC analysis of its corresponding trimethylsilylated L-cysteine adducts. The aforementioned data of 3 were closely related to those of phyllanthusmin C, a known arylnaphthalene lignan glycoside from Phyllanthus oligospermus [18]. The major difference was the presence of one more arabinosyl and an additional acetyl group in 3,

compared with phyllanthusmin C. All signals of 3 were assigned completely on the basis of HSQC, HMBC and ¹H-¹H COSY experiments. In the HMBC spectrum of **3** (Fig. 4), correlation of an anomeric proton at $\delta_{\rm H}$ 4.74 with C-1 ($\delta_{\rm C}$ 146.1) confirmed the linkage of an arabinosyl moiety at C-1 of lignan aglycone. Subsequently, HMBC correlations of the additional arabinosyl anomeric proton at $\delta_{\rm H}$ 4.49 with the inner arabinosyl C-3' ($\delta_{\rm C}$ 81.8) and the inner arbinosyl H-4" ($\delta_{\rm H}$ 5.16) with the acetyl carbonyl carbon ($\delta_{\rm C}$ 172.5), indicated that the additional arabinosyl and acetyl units were linked to the inner arabinosyl C-3' and C-4' in 3, respectively. This resulted in deshielded resonances of the inner arabinosyl C-3' and C-4' of 3 by 7.4 and 2.8 ppm, respectively, and shielded resonances of C-5' by 2.2 ppm, compared with those of phyllanthusmin C. On the basis of these observations and other HMBC correlations (Fig. 4 and Electronic supplementary material), the structure of phyllanthusmin D (3) was established.

Compound 4 possessed a molecular formula of $C_{31}H_{32}O_{16}$, on the basis of HRESIMS (*m/z* 661.1766 $[M + H]^+$). Acid hydrolysis of 4 afforded L-arabinose and D-glucose as sugar residue, which was confirmed by GC analysis of its corresponding trimethylsilated L-cysteine adducts. The ¹H and ¹³C NMR spectroscopic data (Table 3) of 4 showed close resemblance with those of phyllanthusmin C, except for the presence of only one methoxy group and an additional set of signals assignable to a β -glucosyl moiety [$\delta_{\rm H}$ 5.31 (d, J = 7.5 Hz, $\delta_{\rm C}$ 101.3, 74.9, 78.2, 71.5, 78.0 and 62.8] in 4. In the HMBC experiment (Fig. 2S, Electronic supplementary material), correlations of H₂-2a [$\delta_{\rm H}$ 5.47, (d, J = 15.1); 5.62, (d, J = 15.1] and $\delta_{\rm H} 8.25$ (H-8) with the quaternary carbon at $\delta_{\rm C}$ 146.5 (C-1) allowed assignment of C-1 and H-8. The HMBC correlation from the arabinosyl anomeric proton $(\delta_{\rm H} 4.76, \text{Ara-1}'')$ to C-1 $(\delta_{\rm C} 146.5)$ indicated that the arabinosyl moiety was attached to C-1 of the lignan core. Moreover, HMBC correlation of the anomeric proton ($\delta_{\rm H}$ 5.31, Glc-1^{'''}) with C-7 ($\delta_{\rm C}$ 150.6) supported linkage of the additional glucosyl moiety to C-7, which was further confirmed by the ROESY correlation between the anomeric proton ($\delta_{\rm H}$ 5.31, Glc-1^{'''}) with H-8 ($\delta_{\rm H}$ 8.25). On the basis of these observations, the structure of 4 was established as phyllanthusmin E.

Compound **5** was isolated as a white amorphous powder. Its molecular formula was deduced to be $C_{38}H_{44}O_{21}$ by HRESIMS (*m/z* 871.2077 [M – H][–]). Acid hydrolysis of **5** afforded L-arabinose, D-glucose and D-galactose as sugar residue, which was confirmed by GC analysis of their corresponding trimethylsilylated L-cysteine adducts. The ¹H and ¹³C NMR data (Table 4) of **5** were very similar to those of phyllanthusmin C, except for the appearance of two additional hexosyl units attributable to one glucosyl [δ_H 5.55 (d, J = 7.9 Hz), δ_C 105.5, 75.3, 79.0, 72.3, 78.9

Table 4 ¹H and ¹³C NMR spectroscopic data of 5 (methanol- d_4 , δ in ppm)^a

Position	$\delta_{\rm H}~(J~{\rm in}~{\rm Hz})$	$\delta_{\rm C}$	Position	$\delta_{\rm H}~(J~{\rm in}~{\rm Hz})$	$\delta_{ m C}$
1		146.60/146.63 ^b , C	0CH2O	5.93, s	102.3, CH ₂
2		120.46, C		6.03, s	
3		132.30/132.36 ^b , C	Ara-1″	5.35, d (7.7)	107.4, CH
4		136.7, C	2″	4.26 ^c	75.6, CH
4a		131.5, C	3″	4.59 ^c	81.0, CH
5	7.36, s	107.4, CH	4″	4.01, br. d (9.5)	75.7, CH
6		151.6, C	5″	4.63 ^c	68.9, CH ₂
				4.75, t (9.5)	
7		153.2, C	Glc-1‴	5.55, d (7.9)	105.5, CH
8	8.85, s	103.3, CH	2‴	4.08 ^c	75.5, CH
8a		128.9, C	3‴	4.39, t (8.3)	79.0, CH
1'		129.99/130.02 ^b , C	4‴	4.22 ^c	72.3, CH
2'	7.17/7.25, d (1.4)	112.09/112.20 ^b , CH	5‴	4.22 ^c	78.9, CH
3'		148.6, C	6‴	4.24 ^c	68.6, CH ₂
4′		148.5, C		3.68, d (11.3)	
5'	7.16/7.14, d (7.8)	125.0, CH	Gal-1""	5.02, d (7.4)	106.2, CH
6'	7.13/7.08, dd (7.8, 1.4)	109.0, CH	2""	4.49, t (8.2)	73.0, CH
2a	5.70, dd (15.2, 3.7)	68.6, CH ₂	3""	4.06 ^c	75.1, CH
	6.11, dd (15.7, 10.9)		4''''	4.22 ^c	70.2, CH
3a		170.7, C	5''''	4.28 ^c	76.9, CH
6-OMe	3.66, s	55.9, CH ₃	6""	4.56, d (16.3)	63.1, CH ₂
7-OMe	4.17, s	56.9, CH ₃		4.30 ^c	

^a Data were measured at 600 and 150 MHz for ¹H and ¹³C, respectively

^b Existing in pairs at room temperature (21 °C)

^c Overlapping ¹H NMR signals are reported without designated multiplicity

and 68.6] and one galactosyl [$\delta_{\rm H}$ 5.02 (d, J = 7.4 Hz), $\delta_{\rm C}$ 106.2, 73.0, 75.5, 70.2, 76.9 and 63.1] units [26]. Linkage of the additional glucosyl unit to the arabinosyl C-3" position was confirmed by the long range correlations between the glucosyl anomeric proton at $\delta_{\rm H}$ 5.55 (H-1"") and the arabinosyl C-3" ($\delta_{\rm C}$ 81.0). Moreover, HMBC correlation of the galactosyl H-1"" ($\delta_{\rm H}$ 5.02) with the glucosyl C-6" ($\delta_{\rm C}$ 68.6) was also observed (Fig. 2S, Electronic supplementary material). Therefore, the structure of phyllanthusmin F (**5**) was elucidated as shown in Fig. 1.

It was noted that the ¹H and ¹³C NMR spectra of the new arylnaphthalene lignans (**3**–**5**) recorded at room temperature exhibited doubling of the signals, of which most aromatic signals existed as very close pairs, whose chemical shifts are different only at the second decimal places, due to the equilibrium between the two conformational isomers resulting from the slow rotation of sugar unit at room temperature around the glycosidic linkage [19]. The energy barrier about the glycosidic bond was sufficiently high to prevent fast exchange between the two rotamers at room temperature [27].

Most of the isolated compounds were tested for their antifeedant, anti HSV-1 and cytotoxic activities. The two new limonoids **1** and **2** displayed promising antifeedant activity against *Spodoptera exigua* with EC₅₀ values of 25.1 and 17.3 µg/cm², respectively, which were less active than that of the positive control (EC₅₀ = 3.7 µg/cm²), a commericial neem oil containing 1 % azadirachtin, while slightly more active than aphanamixoid A (EC₅₀ = 25.8 µg/cm²), a potential natural product with antifeedant activity [28]. Furthermore, both compounds **1** and **2**, together with the known lignan, phyllanthusmin C, showed moderate cytotoxicity against the ECA109 cell line, with the IC₅₀ values of 11.5, 8.5, and 7.8 µM, respectively, compared to the positive control, 17-AAG (IC₅₀ = 1.1 µM). However, all the tested compounds showed no inhibitory effects against HSV-1.

Limonoids, mainly found in the Meliaceae and Rutaceae families, are of rare occurrence in the Euphorbiaceae family [29]. The new deca-oxygenated flexuosoids A (1) and B (2) represent the second example of limonoids in the Euphorbiaceae family [30]. It is noteworthy that the new limonoids 1 and 2 bear a C-19/29 lactol bridge and oxygen substituents at C-1, C-2, C-3, C-7, C-11, C-17, and C-30 positions, of which the oxygen substituents at C-17 and C-30 are of rare occurrence among the reported limonoids [29]. Furthermore, compounds 1 and 2 showed not only promising antifeedant

activity but also moderate cytotoxicity against the ECA109 human esophagus cancer cell line.

2.1 Experimental Section

2.1.1 General Experimental Procedures

Optical rotations were performed on a P-1020 polarimeter (JASCO, Tokyo, Japan). IR spectra were measured on a Bruker Tensor 27 spectrometer with KBr pellets. 1D and 2D NMR spectra were run on Bruker DRX-500 and AV-600 instruments operating at 500 and 600 MHz for ¹H, and 125 and 150 MHz for ¹³C, respectively. Coupling constants are expressed in Hertz and chemical shifts are given on a ppm scale with tetramethylsilane as internal standard. The MS data were recorded on a VG Auto Spec-3000 spectrometer (VG, Manchester, U.K.) with glycerol as the matrix. HRESIMS were recorded on an API Qstar Pulsa LC/TOF spectrometer. GC analysis was run on a Shimadzu GC-14C gas chromatograph. Column chromatography (CC) was performed with macroporous resin D101 (Mitsubishi Chemical Industry, Ltd., Tokyo, Japan), silica gel (200-300 mesh, Qingdao Haiyang Chemical Co., Ltd., Qingdao, People's Republic of China), Sephadex LH-20 (25-100 µm, Pharmacia Fine Chemical Co. Ltd. Japan). Thin-layer chromatography (TLC) was carried out on silica gel H-precoated plates (Qingdao Haiyang Chemical Co., Ltd., Qingdao, People's Republic of China) with CHCl₃/MeOH/H₂O (8.5:1.5:0.1, 8:2:0.2 or 7:3:0.5, v/v). Spots were detected by spraying with 10 % H₂SO₄ in EtOH followed by heating. Semi-preparative HPLC separation was performed on an Agilent 1260 liquid chromatography with a 5 μ m Waters Sunfire-C18 column (10 \times 250 mm, Waters, Sunfire TM, USA). GC analysis was run on Agilent Technologies HP5890 gas chromatography equipped with an H₂ flame ionization detector. The column was 30QC2/AC-5 quartz capillary column (30 m \times 0.32 mm) with the following conditions: column temperature: 180-280 °C; programmed increase, 3 °C/min; carrier gas: N₂ (1 mL/min); injection and detector temperature: 250 °C; injection volume: 4 μ L, split ratio: 1/50.

2.1.2 Cell Lines and Biochemicals

African green monkey kidney cells (Vero, ATCC CCL81), provided by Wuhan Institute of Virology, Chinese Academy of Sciences, were propagated in DMEM supplemented with 10 % heat-inactivated FBS. The constituents of the maintenance medium were the same as those of the growth medium except that only 2 % FBS was added. The cells were cultured at 37 °C in a humid atmosphere with 5 % CO₂. HSV-1 strain F (ATCC VR-733), obtained from Hong Kong University, was propagated in Vero cells and stored at -80 °C until use. The human esophagus cancer cell ECA109 was cultured in RPMI 1640 containing 10 % heat inactivated FBS and 100 U/mL penicillin/streptomycin in a humidified incubator in a 5 % CO₂ atmosphere at 37 °C.

2.2 Plant Material

The roots of *P. flexuosus* were collected from Yunnan Province, People's Republic of China, in June 2010. Voucher specimens (HITBC_015077) were deposited at the State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, and were identified by Mr. Shi-Shun Zhou from Xishuangbanna Tropical Botanical Garden, Chinese Academy of Sciences.

2.3 Extraction and Isolation

The air-dried and powdered roots of P. flexuosus (2.0 kg) were extracted with MeOH (3 times, 3 h each time) under reflux at 60 °C. The methanol extracts (320 g) were subjected to macroporous resin D101 column chromatography (CC) and eluted with the following gradient: MeOH/H₂O (3:7, 6:4, 9:1) and finally MeOH, to give 4 fractions (Fr. 1-4). Fr. 2 (45.0 g) and Fr. 3 (16.6 g) were subjected separately to repeated CC on silica gel (CHCl₃/MeOH/H₂O, 9:1:0.1-7:3:0.5) and Sephadex LH-20 (MeOH/H₂O, 1:9-8:2) to yield 3 (13 mg), phyllanthusmins C (20 mg), arabelline (12 mg), (+)-diasyringaresinol (8 mg) and subfraction F2A (32 mg). Subfraction 2A was purified by semi-preparative HPLC (MeCN/H₂O, 25:75) to afford 4 (4 mg). Fr. 3 was repeatedly chromatographed over MCI-gel CHP-20P (MeOH/H₂O, 5-45 % with a 5 % increment) and silica gel (CHCl₃/MeOH/H₂O, 9:1:0.1-6:4:1) to get subfractions F3A (170 mg) and F3B (180 mg). Subfraction F3A was purified by semi-preparative HPLC (MeCN/H₂O, 25:75) to give 5 (4 mg). Subfraction F3B was purified by semi-preparative HPLC (MeCN/H₂O, 14:86) to furnish 1 (14 mg) and 2 (25 mg).

2.3.1 Flexuosoid A (1)

White amorphous powder; $[\alpha]_D^{17} - 41.2$ (*c* 0.30, MeOH); UV (MeOH) λ_{max} (log ε): 203.8 (3.06) nm; IR (KBr) v_{max} 3424, 2965, 2931, 1691, 1629, 1455, 1382, 1263, 1159, 1137, 1058 cm⁻¹; ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (pos. ion mode) *m/z* 529 [M + Na]⁺; negative HR ESIMS *m/z* 505.2072 [M - H]⁻ (calcd for C₂₆H₃₃O₁₀, 505.2073).

2.3.2 Flexuosoid B (2)

White amorphous powder; $[\alpha]_D^{18}$ –41.7 (*c* 0.68, MeOH); UV (MeOH) λ_{max} (log ε) 204.0 (3.95) nm; IR (KBr) v_{max}

3424, 2967, 2933, 1695, 1455, 1380, 1257, 1037 cm⁻¹; ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (neg. ion mode) m/z 583 [M + Cl]⁻; negative HRESIMS m/z 547.2182 [M - H]⁻ (calcd for C₂₈H₃₅O₁₁, 547.2179).

2.3.3 Phyllanthusmin D (3)

White amorphous powder; $[\alpha]_D^{16} + 10.5$ (*c* 0.07, MeOH); UV (MeOH) λ_{max} (log ε): 201.0 (4.86), 260.4 (4.94) nm; IR (KBr) v_{max} 3423, 2908, 1742, 1622, 1507, 1480, 1455, 1435, 1386, 1377, 1342, 1264, 1216, 1169, 1080, 1025, 1012 cm⁻¹; ¹H and ¹³C NMR data, see Table 3; ESIMS (neg. ion mode) *m/z* 721 [M + Cl]⁻; negative HRESIMS *m/z* 685.1763 [M - H]⁻ (calcd for C₃₃H₃₃O₁₆, 685.1768).

2.3.4 Phyllanthusmin E (4)

White amorphous powder; $[\alpha]_D^{27}$ –48.7 (*c* 0.28, MeOH); UV (MeOH) λ_{max} (log ε): 203.2 (4.24), 259.0 (4.55) nm; IR (KBr) v_{max} 3442, 2923, 1630, 1535, 1468, 1416, 1385, 1203, 1169, 1128, 1073, 1039 cm⁻¹; ¹H and ¹³C NMR data see Table 3; ESIMS (neg. ion mode) *m*/*z* 695 [M + Cl]⁻; positive HRESIMS *m*/*z* 661.1766 [M + H]⁺ (calcd for C₃₁H₃₃O₁₆, 661.1763).

2.3.5 Phyllanthusmin F (5)

White amorphous powder; $[\alpha]_D^{22}$ -40.6 (*c* 0.30, MeOH); UV (MeOH) λ_{max} (log ε): 202.8 (4.41), 262.2 (4.56) nm; IR (KBr) ν_{max} 3426, 2925, 1730, 1627, 1507, 1480, 1458, 1435, 1386, 1341, 1265, 1230, 1168, 1075, 1049, 1009 cm⁻¹; ¹H and ¹³C NMR data see Table 4; ESIMS (neg. ion mode) *m/z* 871 [M + Cl]⁻; negative HRESIMS *m/z* 871.2077 [M + Cl]⁻ (calcd for C₃₈H₄₄O₂₁Cl, 871.2063).

2.3.6 Acetylation of Compounds 1 and 2

See Electronic supplementary material.

2.3.7 Acid Hydrolysis of Compounds 3-5

See Electronic supplementary material.

2.4 Antifeedant Activity Assay

The insects, beet armyworm (*Spodoptera exigua*) purchased from the Pilot-Scale Base of Bio-Pesticides, Institute of Zoology, Chinese Academy of Sciences. A dualchoice bioassay modified from previous methods was performed for antifeedant tests [31]. The larvae were reared on an artificial diet in the laboratory under controlled photoperiod (light:dark. 12:8 h) and temperature $(25 \pm 2 \text{ °C})$. Larvae were star ved 4–5 h prior to each bioassay. Fresh leaf discs were cut from Brassica chinensis, using a cork borer (1.1 cm in diameter). Treated leaf discs were painted with 20 µL of acetone solution containing the test compounds, and control leaf discs with the same amount of acetone. After air drying, two tested leaf discs and two control ones were set in alternating position in the same Petri dish (90 mm in diameter), with moistened filter paper at the bottom. Two-thirds of instars were placed at the center of the Petri dish. Five replicates were run for each treatment. After feeding for 24 h, areas of leaf discs consumed were measured. The antifeedant index (AFI) was calculated according to the formula AFI = [(C - T)/(C + T)] × 100, where C and T represent the control and treated leaf areas consumed by the insect. The insect antifeedant potency of the test compound was evaluated in terms of the ED₅₀ value (the effective dosage for 50 % feeding reduction) which was determined by Probit analysis for the insect species. The positive control was served as commercial neem oil containing 1 % azadirachtin produced by Kunming Rixin Dachuan Technology Co., Ltd.

2.5 Cytotoxicity Assay

The cytotoxicity of the test compounds on Vero cells was determined by an MTT assay. Vero cells were seeded in 96-well plates at a density of 1×10^4 cells/well, and incubated at 37 °C in a 5 % CO₂ atmosphere for 24 h, until 90 % or greater confluence of the monolayers was reached. Increasing concentrations of the test compounds were added to cells, with a replicate number of three wells per concentration and 17-AAG served as positive control. After a 2-day incubation period in such conditions, a MTT solution was added (final concentration 0.5 mg/mL) and the plates were incubated for another 4 h to allow formazan production. The solid precipitate was dissolved with DMSO and the absorbance at 570 nm was measured using a 96-well Spectrophotometer (Bio-Rad 550) with a reference wavelength of 630 nm. The cytotoxicity of the test compounds on the ECA109 human esophagus cancer cell and Vero cell lines was determined by an MTT assay. The ECA109 human esophagus cancer cell line was cultured in RPMI 1640 containing 10 % heat inactivated FBS and 100 U/mL penicillin/streptomycin in a humidified incubator in a 5 % CO₂ atmosphere at 37 °C. Cells (5 \times 10³/ well) were plated in 96-well plates in 100 µL medium, cultured overnight and exposed to a range of concentrations of compounds for 48 h. After the addition of 20 µL MTT solution (5 mg/mL) per well, the plates were incubated for 4 h, the medium were removed, the formazan crystals were solubilized in 100 µL DMSO per well and the absorbance values were read at 570 nm.

2.6 Anti-HSV-1 Assay

Vero cells were seeded in 96-well plates at a density of 1×10^4 cell/well. Confluent cell monolayers were treated with increasing non-cytotoxic concentrations of the plant extract. Four wells were used for each concentration. Afterwards, the cells were infected with HSV-1 (100 TCID₅₀), incubated at 37 °C in a 5 % CO₂ humidified atmosphere and observed daily for cytopathic effect (CPE) using a light microscope. Acyclovir at concentration 20 µg/mL served as positive control. The EC₅₀ value was calculated by MTT method.

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Conflict of interest The authors declare no conflict of interest.

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