

Highly sensitive detection of methane by near-infrared laser absorption spectroscopy using a compact dense-pattern multipass cell

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Abstract

Highly sensitive detection of atmospheric methane (CH₄) was performed by long optical pathlength absorption spectroscopy based on a novel compact dense-pattern multipass cell (DP-MPC) in conjunction with a fiber-coupled distributed feedback diode laser operating at 1.653 μm. Wavelength modulation spectroscopy approach was used and a minimum detectable concentration (1σ) of 100 ppb was obtained with a lock-in time constant of 1ms. A measurement precision of < 79 ppb was achieved by average of 5 laser scans in 1 s. This newly developed DP-MPC realized 215 times multiple reflections between two very cheap and robust silver coated concave spherical mirrors separated by a distance of 12 cm (forming an absorption cell of 280 cm³ volume), offering an effective optical path length of 26.4 m, which is very suitable for applications to trace gas sensing in harsh environment, and weight-limited unmanned aerial vehicle (UAV) - or balloon-embedded field observations.

1. Introduction

High sensitivity and high precision monitoring of atmospheric methane (CH₄) concentration has become increasingly important for advanced research on carbon cycle, greenhouse gas budgets and perturbations [1-2]. CH₄ is well known as one of the most important greenhouse gases playing an important role in global warming and climate change. Although the concentration of methane in natural air (~1.8 ppm) is roughly 200 times lower than that of CO₂ (~400 ppm), methane presents a global warming potential 25 times greater than CO₂ and may account for 15 percent of anticipated global warming [3]. Methane is also involved in complex feedback mechanisms in atmospheric chemistry and can serve as makers of large-scale dynamics of the earth's atmosphere [4]. It was recognized that methane emissions originate from both natural sources (such as plants oxidation in anaerobic conditions, oceans, wetlands) and anthropogenic sources (like domestic animals, industries, biomass burning, fossil fuel production, etc) [5]. For instance, irrigated rice cultivation is thought to be a major source of atmospheric CH₄ and may constitute 10 to 30% of the total methane emitted into the atmosphere [6]. The atmospheric methane concentrations have increased from a 700 ppb level to 1.8 ppm since 1750 with a rate of about 1% per year [3,7], but the growth rate slowed in the 1990s, the methane burden was nearly constant during the period of 1999-2006. Yet strong growth resumed in 2007. The reasons for these observed changes remain poorly understood because of limited knowledge of its global sources and sinks controlling the global methane budget [1]. Strange craters and burning ice are another examples of mysterious recent happenings on the planet, which are relevant to methane [Scientific American, August 2, 2014]. In addition, methane is a widely used flammable and

explosive fuel, which can lead to a fatal accident. These important various applications in environmental science, climate change study and industrial monitoring have been a key driving force for the development of robust, sensitive, selective, accurate and precise measurement infrastructure for monitoring CH₄ concentration in the atmosphere.

Various techniques have been developed for methane detection. Laser absorption spectroscopy based techniques offer the unique advantages for fast, self-calibration, highly selective and highly sensitive *in situ* quantification of CH₄ without any sample preparation [8-14]. Based on the Beer–Lambert absorption law, the detection sensitivity by absorption spectroscopy is proportional to the molecule absorption path length and the molecule absorption line intensity. In general, sensitive detection scheme is based on using molecular fundamental vibration-rotation transition in the mid-infrared (mid-IR) in conjunction with a traditional multipass cell to achieve the necessary detection sensitivity (of ppb or ppt [12,13]). Three types of multipass cells are most commonly applied: White cells [15,16], Herriott cells [17], and Chernin [18]. Despite the high detection sensitivity achieved, the sensors based on mid-infrared laser source coupled to a traditional multipass cell present some drawbacks: (1) mid-infrared lasers and detectors are costly compared to the devices operating in the near-IR (despite that their lifetime are not as long as those in the near-IR), (2) optical fiber coupling technology is not yet mature and readily available for realization of cost-effective compact and robust sensor platform. Moreover, for the compact Herriott configuration using conventional design, the number of the light passes inside the cell is limited by the overlapping of light spots on the cell mirrors, which limit the achievable optical path length within a compact cell size. A variation of the Herriott cell, called astigmatic mirror cell has been developed to spread the light spots over the entire mirror surfaces to avoid light beam overlapping [19]. This configuration greatly increases the number of light spots achievable and hence the light pass number in a smaller cell volume than Herriott cell. However, the manufacturing of an astigmatic mirror is more difficult than a concave spherical mirror, which will increase significantly the cost. Recently, sub-ppb detection of C₂H₆ using a novel dense-pattern multipass cell (DP-MPC) was reported by Karol Krzempek et al. [20]. The DP-MPC is developed by Sentinel Photonics Inc. [21] offering 57.6 m effective optical path length with 459 times light passes between two spherical mirrors separated by 13 cm, 5 concentric cycles of light spots were realized on the mirror without light overlapping. Wei Ren et al. also reported high sensitivity detection of CH₄ and N₂O by using the same 57.6 m ultra-compact multipass gas absorption cell [22]. Such a novel DP-MPC needs however to use dielectric-coated mirror to realize high reflectivity with a wavelength-dependent narrow spectral bandwidth.

In this paper, we report an alternative solution using cheap metal coating to realize a highly compact DP-MPC. The novel DP-MPC consists of two silver coated concave spherical mirrors separated by a distance of 12 cm and it offers an effective optical pathlength of 26.4 m. Silver (or gold) coated mirror is less sensitive to harsh environmental contamination and requests less attentive maintenance. This low cost, compact and wavelength-independent widely usable DP-MPC is very suitable for various field applications, in particular for gas sensing in harsh environment, and weight-limited unmanned aerial vehicle (UAV) or balloon-embedded observations.

2. Experimental details

A schematic diagram of the experimental setup using a compact DP-MPC is shown in Fig. 1. A fiber-coupled distributed feedback (DFB) diode laser (NLK1E5GAAA, NEL) operating at 1.653 μm and room temperature was directly connected to the DP-MPC. The laser was tunable from 6040 to 6054 cm^{-1} , suitable for detection of methane by probing one of the strong absorption lines located at 6046.95 cm^{-1} in the near infrared. The laser current and temperature were controlled with a commercial diode laser controller (ILX Lightwave LDC-3724). Coarse and fine wavelength tunings were performed by

changing the laser temperature and current, respectively. Wavelength modulation and second harmonic detection was applied to enhance the detection sensitivity [23].

Laser wavelength scan was realized by feeding an external triangular voltage ramp from a function generator (SPF05, NANJING SAMPLE INSTRUMENT TECHNOLOGIES CO., LTD) to the laser diode current which sweeps the laser wavelength back and forth across the absorption line at a rate of 5 Hz. Wavelength modulation was achieved by adding a 11 kHz sine wave to the DFB laser diode current. The sine form wave was supplied by the sinusoidal signal output of a lock-in amplifier (Stanford Research Systems, Model SR 830 DSP). The voltage ramp and the sine wave were combined with a home made adder and then fed to the laser driver. The laser beam was collimated with a fiber-coupled collimator ($f \sim 4.8$ mm) and subsequently injected into the compact DP-MPC. The output beam was focused on to a photodetector (New Focus, 2011-FC) with a lens ($f = 40$ mm). A lock-in amplifier was used for demodulation of the absorption signal from the photodetector at a frequency of $2f$ (where $f = 11$ kHz is the modulation frequency of the sine wave). The time constant of the lock-in amplifier was set to 1 ms in combination with an 18 dB/octave slope (leading to a detection bandwidth of 9.375 Hz). The demodulated signal was subsequently digitalized by a DAQ card (NI-USB-6212) and displayed on a laptop via a LabVIEW interface. The signal of the photodetector was also acquired for laser power monitoring.

The used compact DP-MPC consisted of two 2" silver coated concave spherical mirrors separated by a distance of 12 cm, which allows laser beam passing 215 times between two spherical mirrors (leading to an effective optical pathlength of 26.4 m). The curvature R of the used spherical mirrors was 100 mm. With a careful optical design and alignment, a novel dense pattern was configured on the mirror with the objective to maximize light reflection number inside the cell while minimizing light spot overlapping to reduce etalon fringe effects. Such silver (or gold) coated spherical mirrors, instead of dielectric-based spherical mirror, offer not only a low cost DP-MPC device, but also a wide spectral response range from the visible to the middle infrared. A home-developed Labview software based program was used to determine and optimize optical design of the DP-MPC according to the ray tracing results [24]. Fig. 2 shows a photograph of a red laser beam patterns on one mirror of the DP-MPC, in comparison with the corresponding patterns by the design simulation. A laser power of 11.8 mW was injected to the DP-MPC, and the output power was measured to be 116 μ W, which allows one to deduce a reflectivity of the mirrors of about 97.8% at 1.65 μ m. The two mirrors were mounted on an optical board with a dimension of $200 \times 76 \times 10$ mm³ and sealed using a quartz glass tube with an inner diameter of 55 mm. The weight of the DP-MPC with body material in stainless was 3.5 Kg. The sample volume of the DP-MPC is ~ 280 cm³, much smaller than a traditional concave spherical mirror based Herriott or White MPC (of ~ 700 cm³ with 80 times reflections), it significantly reduces gas sampling time which is a very important issue that should be carefully addressed for sensing of chemically reactive short-lived gas where the gas residence time in absorption cell should be as short as possible.

3. Measurements and results

The main objective of the present work was to develop a compact and low cost sensitive CH₄ sensor operating at atmospheric pressure. Selection of the absorption line for methane detection was made based on HITRAN04 database [25]. The best CH₄ absorption lines within the spectral tuning range of the used 1.653 μ m diode laser are the R₃ triplet of the 2 ν_3 band near ~ 6046.95 cm⁻¹, they have the highest line intensities and are interference free from other molecules (such as H₂O and CO₂). These R₃ triplet of the 2 ν_3 band at ~ 6046.95 cm⁻¹ were thus selected as the target. According to the HITRAN04 database, the R₃ triplet consists of 3 lines: F₁, F₂ and A₂, which overlap and not resolved even at low pressure under Doppler broadening conditions. The measurement was carried out at ambient pressure,

(though it may get a better signal to noise ratio when working at a reduced pressure [26], it requests the use of additional mechanical pump and pressure controller, which increases the sensor size and its cost). In the present work, the laser center current I_0 was set at 57 mA with a temperature set at 23 °C for targeting the selected CH₄ absorption lines around 6046.95 cm⁻¹. The amplitude of the triangular ramp was 1.5 V (3 V peak to peak), which scanned the laser diode current from 27 to 87 mA (20 mA/V). Fig. 3 shows a calibration curve of the laser wavelength as a function of the laser current at temperature of 23 °C, and an absorption spectrum of 200 ppm CH₄ which was simulated based on HITRAN04 database [25]. The laser wavelength was measured with a wavelength meter (Burleigh, WA-1500).

In wavelength modulation spectroscopy, amplitude of the 2nd harmonic signal is dependent on the modulation amplitude [27]. To determine the optimum modulation amplitude, 2nd harmonic signals of 200 ppm CH₄ measured at different modulation amplitude were recorded. The measurement was carried out at normal atmosphere. The 2nd harmonic signal amplitude of CH₄ as a function of the modulation amplitude is shown in Fig. 4. The optimum modulation amplitude of the sine wave for CH₄ detection was found to be 0.3 V, and the corresponding laser wavelength modulation amplitude was $\Delta\nu_m \approx 0.15 \text{ cm}^{-1}$. According to the HITRAN04 database, the 200 ppm CH₄ line width (HWHM) of the targeted absorption lines was $\Delta\nu_D \approx 0.068 \text{ cm}^{-1}$. This result agrees with the theoretical optimum modulation amplitude $\Delta\nu_m = 2.2 \Delta\nu_D$ for wavelength modulation with 2nd harmonic demodulation [23,27]. As such a modulation amplitude of 0.3 V was used in the following measurements. The 2nd harmonic signal is also dependent on the laser power, therefore the signal from the photodetector was recorded for monitoring the laser power fluctuations and the $2f$ signal amplitude was normalized to the laser power before concentration retrieval.

The response of the $2f$ signal amplitude to different CH₄ concentration levels (diluted in N₂) was measured and displayed in Fig.5. CH₄ concentrations in the range of 7–80 ppm were generated by diluting a 1000 ppm CH₄:N₂ calibrated gas mixture sample with pure N₂. Two flow meters were used in this calibration process. One flow meter with a flow range of 0-50 ccm was used to control the flow rate of the 1000 ppm CH₄:N₂ calibrated gas sample; another one with a flow range of 0-1000 ccm was used to control the flow rate of pure N₂. In the measurements, absolute concentration of the diluted CH₄ was determined by direct absorption spectroscopy in a 100 m multipass cell (Aerodyne Research, Inc. model 5612). The $2f$ signals for each calibrated CH₄ concentration were averaged 100 times. The linear response of the $2f$ signal to CH₄ concentration was confirmed by fitting the data with a linear slope as shown in Fig. 5. The R squared value for the linear fit was >0.99 with a dynamic measurement range of $\sim 10^3$.

For evaluation of the measurement precision and the stability of the compact DP-MPC based CH₄ sensor, time series measurements of ambient CH₄ sealed in the MPC (so having constant concentration) was performed during 2.5 hours. The measurements were performed by averaging 5 scans with a scan rate of 5 Hz, namely, each point of the signal was obtained in 1 second. 9000 data points were obtained during 2.5 hours' continuous measurement. Histogram plot of measured data is shown in fig. 6, which can be helpful to evaluate the measurement precision. The data distribution is a gaussian profile and the measurement precision, determined by the half width at half maximum (HWHM) was found to be 79 ppb with 1 ms lock-in time constant and 1 s average time. The stability of the compact DP-MPC based CH₄ sensor was determined by an Allan analysis. Fig. 7 shows the Allan deviation obtained from the continuous time series measurements. The Allan analysis showed that the sensor allows an averaging time of up to 70 s and the corresponding minimum detectable concentration of CH₄ can be down to 15 ppb. These high sensitivity and high measurement precision illustrate the high performance of the developed compact DP-MPC based CH₄ sensor.

Measurement of CH₄ in ambient air was performed to evaluate the performance of the developed

CH₄ sensor. Fig. 8 shows the acquired $2f$ signal of CH₄ in the ambient air (without average), the amplitude of the $2f$ signal is 174 μV . According to the calibrated curve shown in Fig. 5, the corresponding CH₄ concentration is 1.8 ppm, in good agreement with the typical ambient value of CH₄. The noise level was determined by the standard deviation, deduced from non-absorption wing of the CH₄ spectrum, was found to be $\sim 10 \mu\text{V}$ (1σ), which corresponds to a signal-to-noise ratio (SNR) of ~ 17 . From these results, a minimum detectable concentration of CH₄ of 100 ppb was deduced, and a noise equivalent absorption sensitivity (NEAS) of (1σ) $3.6 \times 10^{-9} \text{ cm}^{-1} \text{ Hz}^{-1/2}$ was obtained at atmospheric pressure. Two days' continuous measurements of CH₄ in ambient air (Hefei, China) were performed. During the measurement, a dryer and a particle filter were placed before the inlet of the DP-MPC to remove humidity and aerosols. The time series continuous measurements of ambient CH₄ concentration are shown in Fig. 9, in which the data were averaged in 1 minute (averaging 300 times laser scans). Measured results shown that the highest CH₄ concentration occurs at ~ 7 am, and the lowest concentration occurs at ~ 11 am, respectively. Two consecutive days' measurements represent the same CH₄ concentration variation trends. The averaged concentration of a day was found to be 2.01 ppm and 1.99 ppm for a two days' measurement, respectively.

4. Conclusions

High sensitivity and precision monitoring of CH₄ using a 1.653 μm near infrared diode laser was achieved with the help of a novel compact DP-MPC which involved cheap and easily manufactured silver coated concave spherical mirrors to realize a 26.4 m effective optical pathlength in a cell with a base length of 12 cm. The overall dimension of the 26.4 m DP-MPC is $20 \times 7.6 \times 10.5 \text{ cm}^3$ leading to a sample volume of 280 cm^3 . Furthermore, such metal coated mirror based DP-MPC is less sensitive to harsh environmental contamination and require less attentive maintenance. Such a sensitive, low cost, long lifetime, compact and robust, handheld CH₄ sensor is very suitable for application in harsh environment (leaking detection in nature gas industries and coal mines), in agricultural field monitoring and UAV- or balloon-embedded observation.

In our demonstration experience, a minimum detectable concentration of 100 ppb for CH₄ detection was obtained with a lock in time constant of 1 ms. A measurement precision of 79 ppb was obtained in 1 second (Averaging 5 times of laser scans). The detection sensitivity can be still improved by further averaging the data, locking the laser wavelength to the targeted absorption line center and increasing time constant of lock-in amplifier, or use filter technique like Kalman Filter [28, 29]. Such DP-MPC based on silver (or gold) coated spherical mirrors is operational in a wide spectral response range which allows one to couple the DP-MPC to quantum cascade laser in the mid-IR.

Acknowledgments

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Figure captions

Fig. 1 Schematic diagram of the developed CH₄ sensor (bottom), and photograph of the compact multipass gas absorption cell (top)

Fig. 2 Photograph of the multipass gas absorption cell beam pattern visualized using a red diode laser beam (left) and the corresponding software simulated pattern (right).

Fig. 3 DFB diode laser wavelength as a function of the injection current at a temperature of 23 associated with a simulated absorption spectrum of 20 ppm CH₄.

Fig. 4 Measured $2f$ signal amplitude of 100 ppm CH₄ as a function of the laser modulation amplitude.

Fig. 5 Measured $2f$ signal as a function of the CH₄ concentration

Fig. 6 Histogram plot obtained from time series measurements of 2.2 ppm CH₄ sealed in the DP-MPC.

Fig. 7 Allan deviation from time series measurements of CH₄ sealed in the DP-MPC.

Fig. 8 $2f$ absorption spectrum of ambient CH₄ at normal atmospheric pressure.

Fig. 9 Ambient CH₄ concentration variation during two days continuous measurements.

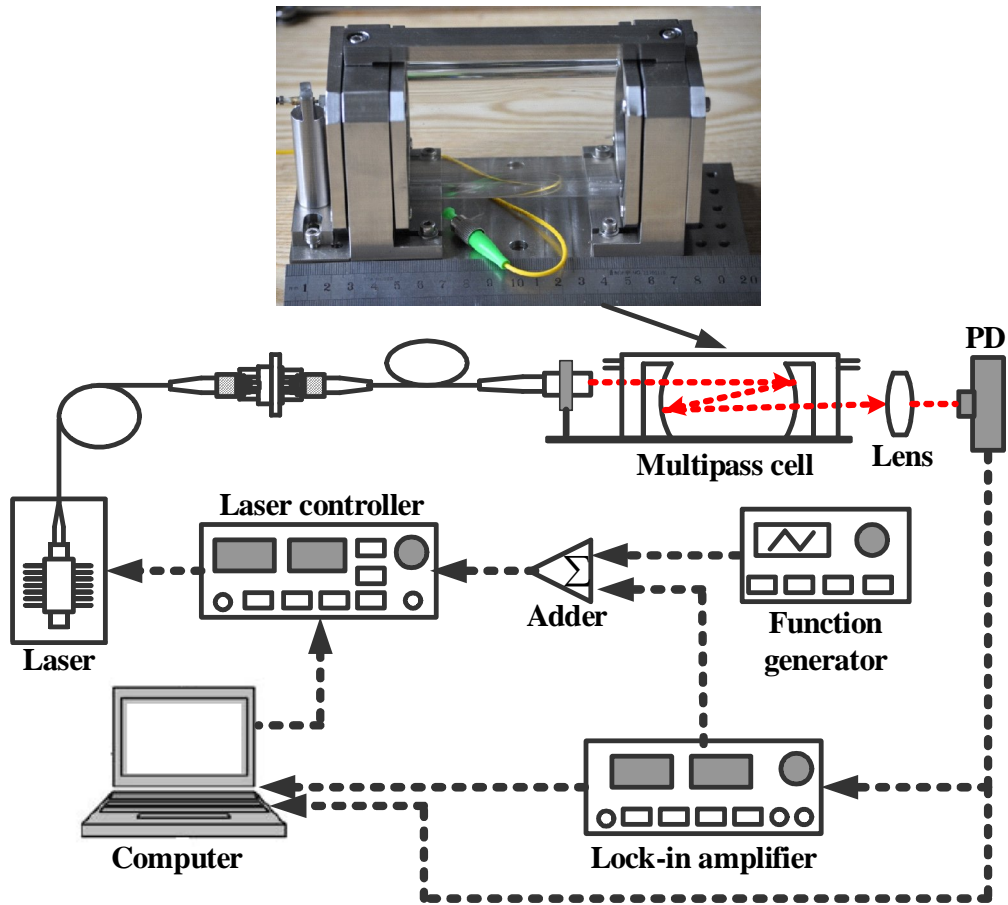


Figure 1

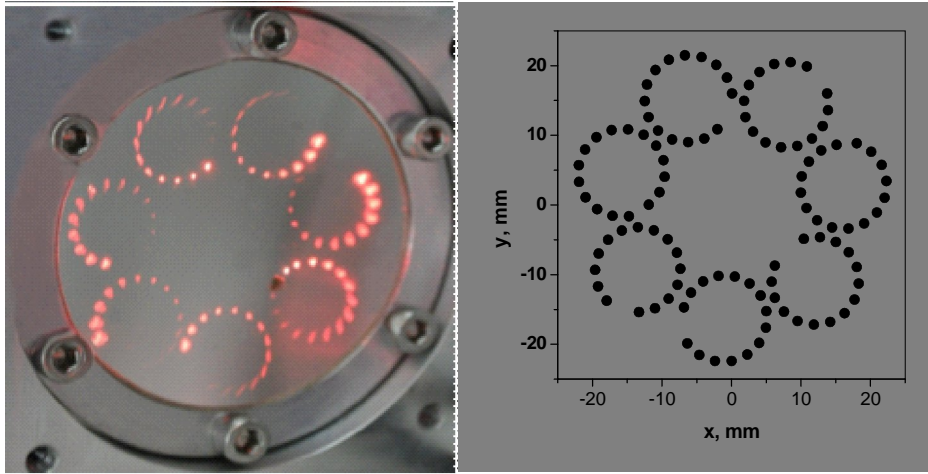


Figure 2

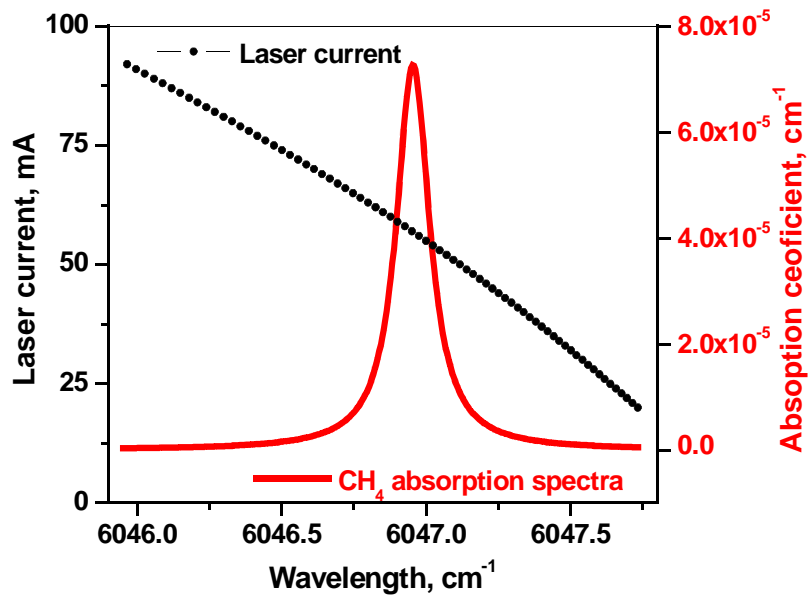


Figure 3

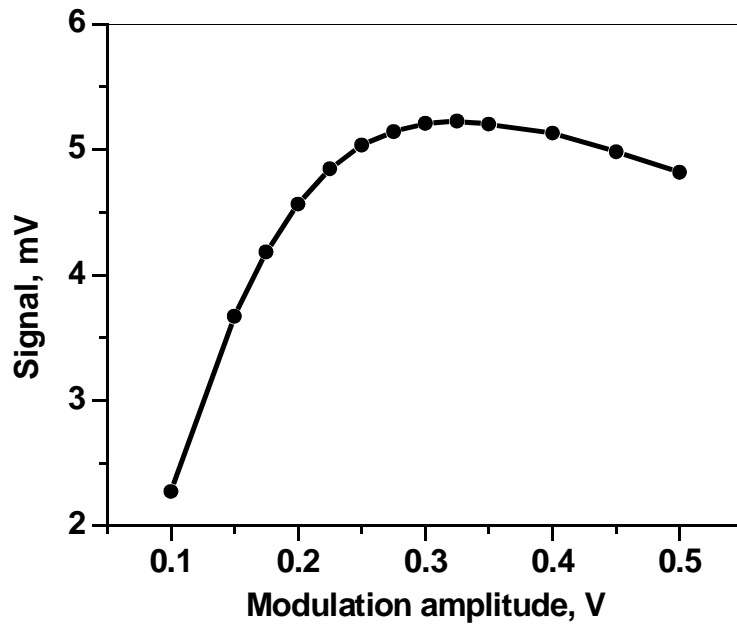


Figure 4

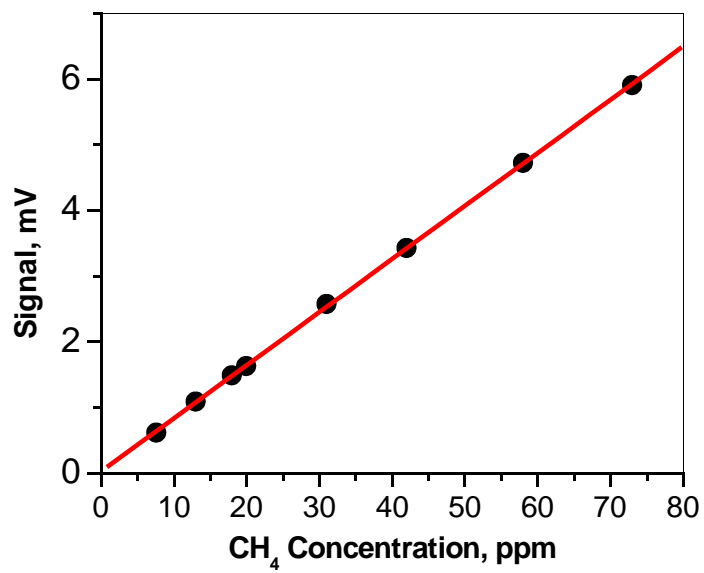


Figure 5

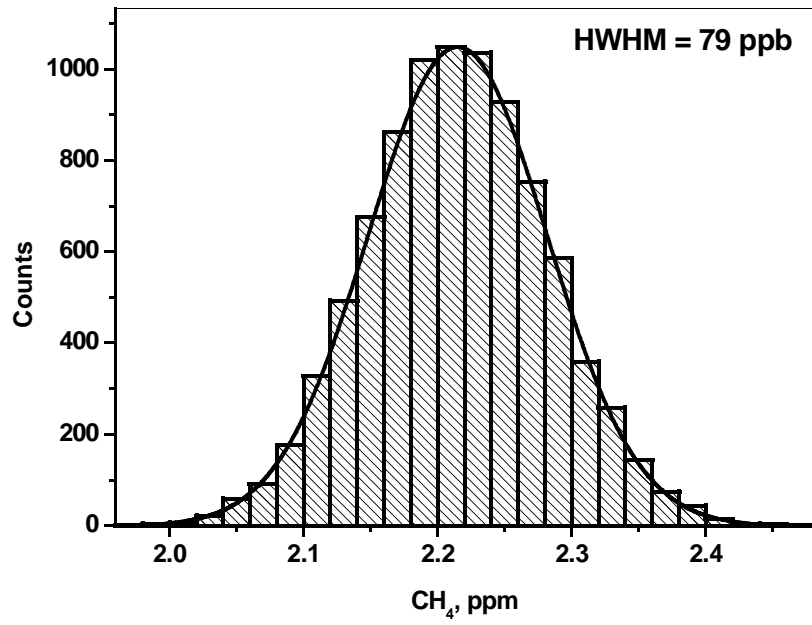


Figure 6

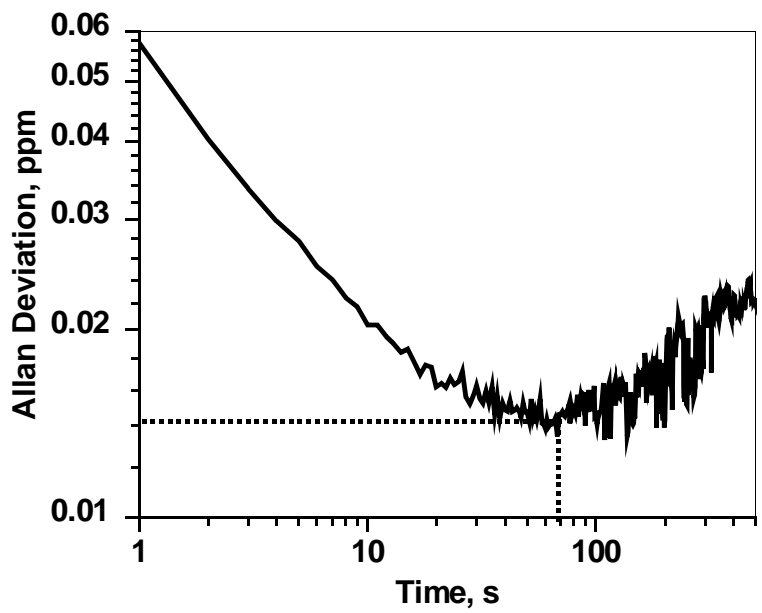


Figure 7

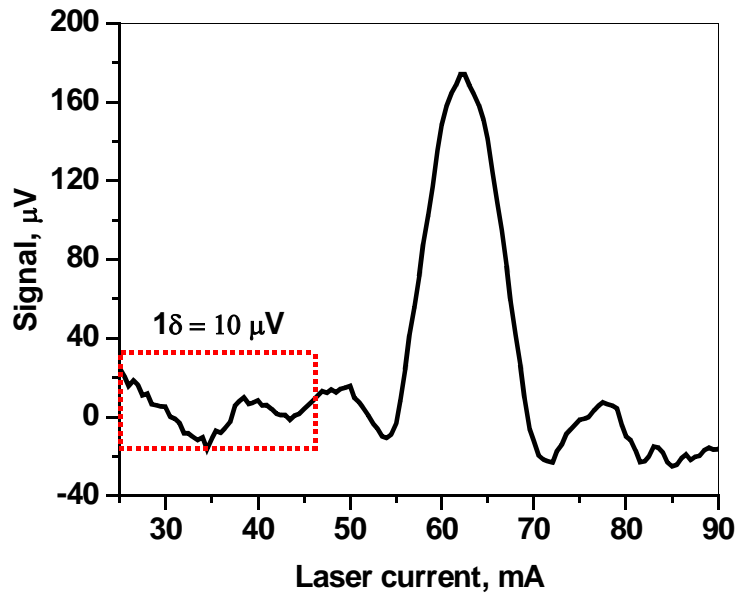


Figure 8

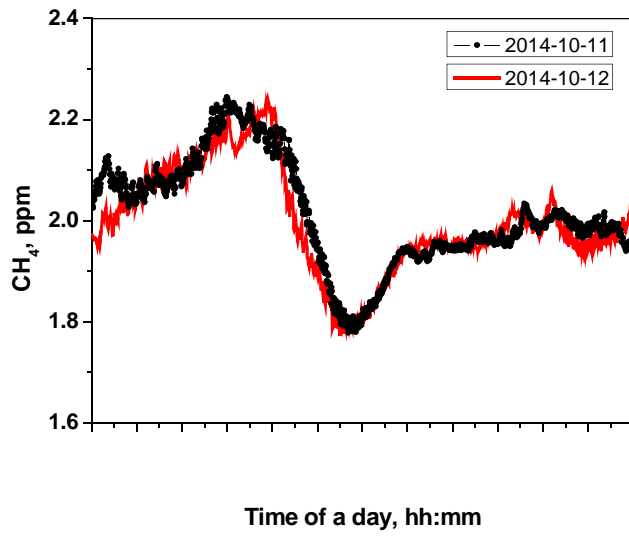


Figure 9

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