How reproducible are surface areas calculated from the BETequation?

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128 **To the editor:**

129 The Brunauer-Emmett-Teller (BET) equation is arguably one of the most used equations in physical 130 chemistry and porosimetry. Since its conception in the 1930s¹ to estimate open surfaces whilst 131 working with adsorbents of the time such as Fe/Cu catalysts, silica gel, and charcoal, it has found widespread use in the characterisation of synthetic zeolites.² Furthermore, it gained considerable 132 133 momentum following the discovery of more complex porous materials such as mesoporous silicas,³ porous coordination polymers (PCPs),⁴ metal-organic frameworks (MOFs),⁵ and covalent organic 134 frameworks (COFs).⁶ Novel porous materials are of significant academic and industrial interest due 135 to their applications in gas storage and separation,⁷⁻¹⁰ catalysis,¹¹ and drug delivery,¹² and the BET 136 137 area is their *de facto* standard for the characterisation. It has been recognized by the International 138 Union of Pure and Applied Chemistry (IUPAC) as "the most widely used procedure for evaluating the surface area of porous and finely-divided materials", ^{13,14} and it has been an International 139 140 Organization for Standardization (ISO) standard for surface area determination since 1995.¹⁵ Whilst 141 concerns over the applicability of the BET theory for microporous materials are important, it remains, 142 arguably, the most important figure of merit for porous materials. Given the broad use of the BET equation, it is not surprising to see that much has been written on the *applicability* and the *accuracy* 143 144 of the BET theory - that is, its model of the adsorption process - and on the reproducibility of the raw data, *i.e.* the adsorption isotherm.^{16–20} 145

146 The advent of materials with more complex pore networks and dynamic frameworks through 147 material design strategies such as reticular chemistry has boosted interest in BET theory (Figure S1) and given rise to reported BET areas in excess of 8,000 m² g⁻¹.^{8,21,22} Often, these modern 148 materials have complex adsorption isotherms that are more problematic or ambiguous to fit to the 149 150 BET model, e.g. several steps can occur due to different pore types and/or flexibility being present in the material.²³ Whilst adsorption rigs capable of ultra-low pressure (<10⁻⁵ mbar) recordings have 151 been developed, reliance on manual calculations of BET areas remains commonplace. In this 152 153 context, 'manual' refers to the judicious selection of the optimal pressure range by a scientist, be it through a self-developed spreadsheet or commercial software. This raises the question of the 154 155 reproducibility of BET calculations from the same measured isotherm but from different assessors.

The eponymously named Rouquerol criteria (**Section S2**, Supplementary Information) aim to ensure good practice in identifying a valid fitting range, and, as such, they have found widespread acceptance in the literature and have been adopted in both IUPAC and ISO standards.^{13–15,17,18,24,25} Despite this safeguard, we herein propose that current BET area calculations are many times irreproducible for two reasons: first, the Rouquerol criteria are indeterminate in identifying the correct

161 fitting region, as they apply to multiple regions simultaneously. Second, even if they were 162 determinate, they are too cumbersome and lengthy to be systematically implemented and are 163 therefore often neglected in practice.

164 To prove our hypothesis and to assess the current spread of BET calculation results, we have 165 shared a set of 18 experimental isotherms representing four classes of porous materials (zeolites, 166 mesoporous silicas, MOFs, and COFs) with 60 laboratories with expertise in adsorption science and synthesis of porous materials. In this round-robin exercise, we asked the researchers to calculate 167 168 the BET areas in the way they saw most fit. More details about the specific materials and the 169 adsorption isotherms, sampled both from our laboratory and from the NIST/ARPA-E database,²⁶ are included in the Supplementary Information, Section S12. To avoid any recognition bias, all 170 171 isotherms were anonymised and scaled off arbitrarily.

172 In parallel, we have developed a computational approach to calculating BET areas that only 173 requires the adsorption isotherm as input data. The BET Surface Identification (BETSI) algorithm, 174 steps through all possible fitting regions and outputs a full distribution of BET areas that are 175 consistent under the Rouquerol criteria. We further propose an addition to the criteria that makes, 176 for the first time, an unambiguous assignment of BET areas from an adsorption isotherm possible: 177 the ideal fitting range ends on the highest permissible pressure point under all criteria, representing 178 the end of the bulk adsorptive activity of the material, *i.e.* the isotherm knee. Further, it is chosen as 179 having the lowest percentage error under the last Rouquerol criterion. Further details on the BETSI 180 algorithm and the extension of the Rouquerol criteria can be found in Section S3, and a more 181 detailed description in Section S14. The source code is fully published under GitHub 182 https://github.com/fairen-group/betsi-gui.

Figure 1 shows the comparison between BET areas calculated by researchers in the round-183 184 robin evaluation and using BETSI. Bar a few exceptions, virtually no two groups of experts reported 185 identical BET areas for any given isotherm. The results are fully tabulated and graphically 186 represented in Section S4 and Section S5 respectively. We observed a spread of at least 300 m² 187 g⁻¹ for each isotherm; however, that number was significantly higher for some individual isotherms. For NU-1104, a modern MOF with substantial porosity²² the highest estimate of 9,341 m² g⁻¹ and the 188 lowest estimate of 1,757 m² g⁻¹ differed by an astonishing 7,584 m² g⁻¹, making the highest estimate 189 190 more than five times higher than the lowest one. Most groups (90%) reported using the Rouguerol 191 criteria in their manual calculation, 23% used a commercial software package, and 6% used a self-192 developed code. Full details on each individual group's methods can be found in Section S13.



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Figure 1 | Round-robin results and BETSI results. Distribution of BET areas from identical isotherms as
 calculated by 60 laboratories with expertise in adsorption science and synthesis of porous materials in red.
 Superimposed are normalised probability distribution functions obtained by kernel density estimation.
 Predictions under BETSI are shown in blue alongside, and the 'optimal' BET area in yellow.

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199 Under BETSI, on the other hand, whilst multiple BET areas are passed as valid, the spread of 200 values was considerably narrower than that obtained by manual calculation (Figure 1; for full BETSI 201 results, see Section S6 and further comparative data Section S7, Section S8, and Section S9). 202 From this, both our first and second hypotheses are substantiated: since BETSI calculates all valid 203 BET areas, it proves that the Rouquerol criteria by themselves are indeterminate and that even full 204 compliance does not guarantee an unambiguous answer. Besides, since the spread of all valid BET 205 areas is narrower than that obtained in the round-robin exercise, it demonstrates how the manual and systematic implementation of the Rouguerol criteria is difficult and often neglected in practice. 206 For instance, in the case of NU-1104, the range of estimates decreases from 7,500 m² g⁻¹ in the 207 social study to 235 m² g⁻¹ under BETSI. 208

Interestingly, some isotherms returned under BETSI much larger spreads of results than others, suggesting that they BET model does not describe them as naturally and thus they were more susceptible to problems associated with the Rouquerol criteria; a trend that was mirrored in the round-robin evaluation. To further investigate the goodness of the isotherm fittings, we define the *BETSI Variation Coefficient* as the relative standard deviation of BETSI results, and the *Pass Rate* as the number of BET fits that pass under the Rouquerol criteria as a fraction of all potential fits. 215 Further, the Hit Rate expresses the fractional number of BET areas calculated in the round-robin 216 exercise that lie within the BETSI range. Figure 2 demonstrates the correlation between the Pass 217 Rate, the BETSI Variation Coefficient, and the Hit Rate. Simply put, the more BET fits are valid, the 218 greater the spread of possible BET areas is, and the more likely researchers are to satisfy the 219 Rouquerol criteria in manual calculations; an alternative representation can be found in Section S10. 220 From Figure 2, we classify adsorption isotherms into three broad categories, types A, B and C. 221 Whilst it is difficult to generalise about the shape of these isotherms, we offer some discussion about 222 common features in **Section S11**. Type A isotherms fit the BET model 'best'. Under BETSI, they 223 have a relatively high Pass Rate and return a fairly narrow spread of results. Examples include 224 materials such as AI fumarate, NU-1000, Zeolite-13X and MCM-41. Hit Rates greater than 70% are 225 generally observed for these materials, suggesting that the majority of researchers did not struggle with the fittings. Type B isotherms only fit the BET model over a very limited range. These have 226 227 extremely low Pass Rates, meaning that only a few BET fits are valid, which in turn will be spread 228 narrowly. Examples include MOF-5, DMOF-1, NU-1104, HKUST-1, and NU-1105. For the latter, out 229 of 9,409 hypothetical 10-point fits (the minimum point requirement for BET fits), only one is 230 permissible under the Rouquerol criteria. Such prohibitively low Pass Rates make the correct BET 231 assignment by hand virtually impossible and demonstrate the need for computational support. Type C isotherm fittings are arguably the most problematic. They have high Pass Rates and, 232 233 concomitantly, they return large spreads of BET results. Typical materials that fit into this category 234 are MIL-101, MIL-100, TPB-DMTP-COF and PCN-777. It is for these materials that the necessity to 235 extend the Rouquerol criteria is demonstrated and the BETSI algorithm makes an unambiguous BET 236 assignment possible.



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238 Figure 2 | Isotherm classifications. Plot of the BETSI Variation Coefficient (relative standard deviation of 239 BETSI results) against the Pass Rate (fraction of valid fits against all hypothetical ones). Bubble size scales 240 with the Hit Rate, the fraction of results from the social study that lie within the BETSI range. Red symbols 241 have a Hit Rate of zero. Note the positive correlation between all three parameters. Isotherm fit classifications. 242 Type A fits have a relatively wide fitting window, within which multiple fits are possible, but return a relatively 243 narrow spread of BET results. Type B fits have a narrow fitting window and concomitantly return a narrow set 244 of spread of results. Type C fits have wide fitting windows, which translates to multiple passable fits and a wide 245 spread of permissible BET areas.

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In conclusion, BET theory is a great success story. Developed in the 1930s for open surfaces, it continues to be applied to modern adsorbents with complex porosities. Despite the advances from classical density functional theory (DFT) methods, the BET area will likely continue playing a crucial role in porosimetry for decades to come, with impacts in energy research, transport, medical applications and climate-change mitigation. In light of these future developments, it will become increasingly important to share critical scientific metrics reliably to find a common language to report both academic and industrial progress.

Here, we have demonstrated the difficulties in unambiguously determining BET areas from adsorption isotherms, which in turn affect the assessment of material quality and reproducibility. These problems arise from imperfect and insufficient manual calculations and can only be met using modern computational methods. BETSI is a step towards greater transparency and critical

assessment in reporting BET areas. We stress here that it is neither the function nor the purpose of BETSI to eliminate doubt and treat a particular BET area as 'true'. Researchers should remain aware of the limitations of BET theory when applied to microporous adsorbents in general and when BET areas are reported, the pressure range and number of points used should always be stated. We further recommend here that isotherms must be reported transparently and in detail, *i.e.* semi-log representation to show the low-pressure regions. The 'experiment' is the adsorption isotherm – not the BET area.

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266 Online Content

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270 Isotherm data reported with this paper are included in the NIST/ARPA-E Database of Novel and

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273

274 Bibliography

- Brunauer, S., Emmett, P. H. & Teller, E. Adsorption of Gases in Multimolecular Layers. J.
 Am. Chem. Soc. 60, 309–319 (1938).
- Cid, R., Arriagada, R. & Orellana, F. Zeolites surface area calculation from nitrogen adsorption data. *J. Catal.* **80**, 228–230 (1983).
- Beck, J. S. *et al.* A New Family of Mesoporous Molecular Sieves Prepared with Liquid
 Crystal Templates. *J. Am. Chem. Soc.* **114**, 10834–10843 (1992).
- Kitagawa, S., Kitaura, R. & Noro, S. I. Functional porous coordination polymers. *Angew. Chemie Int. Ed.* 43, 2334–2375 (2004).
- Zhou, H. C., Long, J. R. & Yaghi, O. M. Introduction to metal-organic frameworks. *Chemical Reviews* vol. 112 673–674 (2012).
- 285 6. Diercks, C. S. & Yaghi, O. M. The atom, the molecule, and the covalent organic framework.
 286 Science 355, (2017).
- 287 7. Li, J., Sculley, J. & Zhou, H. Metal-Organic Frameworks for Separations. *Chem. Rev.* 112, 869–932 (2012).
- 289 8. Farha, O. K. *et al.* De novo synthesis of a metal-organic framework material featuring
 290 ultrahigh surface area and gas storage capacities. *Nat. Chem.* 2, 944–948 (2010).
- 291 9. Li, B., Wen, H.-M., Zhou, W. & Chen, B. Porous Metal–Organic Frameworks for Gas
 292 Storage and Separation: What, How, and Why? *J. Phys. Chem. Lett.* 5, 3468–3479 (2014).
- Moghadam, P. Z. *et al.* Computer-aided discovery of a metal-organic framework with superior oxygen uptake. *Nat. Commun.* 9, 1378–1385 (2018).
- 295 11. Corma, A., García, H. & Llabrés i Xamena, F. X. Engineering Metal Organic Frameworks for
 296 Heterogeneous Catalysis. *Chem. Rev.* 110, 4606–4655 (2010).
- Horcajada, P. *et al.* Metal-organic frameworks in biomedicine. *Chem. Rev.* 112, 1232–1268 (2012).
- 299 13. Thommes, M. et al. Physisorption of gases, with special reference to the evaluation of

- surface area and pore size distribution (IUPAC Technical Report). *Pure Appl. Chem.* 87,
 1051–1069 (2015).
- Sing, K. S. W. *et al.* Reporting Physisorption Data for Gas/Solid Systems with Special
 Reference to the Determination of Surface Area and Porosity. *Pure Appl. Chem.* 57, 603–
 619 (1985).
- ISO [International Organization for Standardization]. Determination of the specific surface
 area of solids by gas adsorption BET method (ISO 9277:2010(E)). (2010)
 doi:10.1007/s11367-011-0297-3.
- 30816.Ambroz, F., Macdonald, T. J., Martis, V. & Parkin, I. P. Evaluation of the BET Theory for the
Characterization of Meso and Microporous MOFs. Small Methods 2, 1800173 (2018).
- Gómez-Gualdrón, D. A., Moghadam, P. Z., Hupp, J. T., Farha, O. K. & Snurr, R. Q.
 Application of Consistency Criteria To Calculate BET Areas of Micro- And Mesoporous
 Metal-Organic Frameworks. *J. Am. Chem. Soc.* **138**, 215–24 (2016).
- Walton, K. S. & Snurr, R. Q. Applicability of the BET method for determining surface areas
 of microporous metal-organic frameworks. *J. Am. Chem. Soc.* **129**, 8552–8556 (2007).
- Park, J., Howe, J. D. & Sholl, D. S. How Reproducible Are Isotherm Measurements in MetalOrganic Frameworks? *Chem. Mater.* 29, 10487–10495 (2017).
- Sinha, P. *et al.* Surface Area Determination of Porous Materials Using the Brunauer Emmett-Teller (BET) Method: Limitations and Improvements. *J. Phys. Chem. C* 123, 20195–
 20209 (2019).
- Furukawa, H., Cordova, K. E., O'Keeffe, M. & Yaghi, O. M. The Chemistry and Applications
 of Metal-Organic Frameworks. *Science* 341, 1230444–1230444 (2013).
- Wang, T. C. *et al.* Ultrahigh Surface Area Zirconium MOFs and Insights into the Applicability
 of the BET Theory. *J. Am. Chem. Soc.* **137**, 3585–3591 (2015).
- Fairen-Jimenez, D. *et al.* Opening the gate: Framework flexibility in ZIF-8 explored by
 experiments and simulations. *J. Am. Chem. Soc.* 133, 8900–8902 (2011).
- Rouquerol, J., Llewellyn, P. & Rouquerol, F. Is the BET equation applicable to microporous adsorbents? *Stud. Surf. Sci. Catal.* 160, 49–56 (2007).
- Rouquerol, J., Rouquerol, F., Llewellyn, P., Maurin, G. & Sing, K. S. W. Adsorption by *Powders and Porous Solids: Principles, Methodology and Applications: Second Edition. Adsorption by Powders and Porous Solids: Principles, Methodology and Applications: Second Edition* (Academic Press, 2013). doi:10.1016/C2010-0-66232-8.
- 332 26. D.W. Siderius, V.K. Shen, R.D. Johnson III, and R.d. van Zee, Eds., NIST/ARPA-E
 333 Database of Novel and Emerging Adsorbent Materials, National Institute of Standards and
 334 Technology, Gaithersburg MD, 20899, https://dx.doi.org/10.18434/T43882, (retrieved Augu.
 335 vol. 91.

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