

FORUM REVIEW ARTICLE

Hydrogen Sulfide as an Oxygen Sensor

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Abstract

Significance: Although oxygen (O_2)-sensing cells and tissues have been known for decades, the identity of the O_2 -sensing mechanism has remained elusive. Evidence is accumulating that O_2 -dependent metabolism of hydrogen sulfide (H_2S) is this enigmatic O_2 sensor. **Recent Advances:** The elucidation of biochemical pathways involved in H_2S synthesis and metabolism have shown that reciprocal H_2S/O_2 interactions have been inexorably linked throughout eukaryotic evolution; there are multiple foci by which O_2 controls H_2S inactivation, and the effects of H_2S on downstream signaling events are consistent with those activated by hypoxia. H_2S -mediated O_2 sensing has been demonstrated in a variety of O_2 -sensing tissues in vertebrate cardiovascular and respiratory systems, including smooth muscle in systemic and respiratory blood vessels and airways, carotid body, adrenal medulla, and other peripheral as well as central chemoreceptors. **Critical Issues:** Information is now needed on the intracellular location and stoichiometry of these signaling processes and how and which downstream effectors are activated by H_2S and its metabolites. **Future Directions:** Development of specific inhibitors of H_2S metabolism and effector activation as well as cellular organelle-targeted compounds that release H_2S in a time- or environmentally controlled way will not only enhance our understanding of this signaling process but also provide direction for future therapeutic applications. *Antioxid. Redox Signal.* 22, 377–397.

“Nothing in Biology Makes Sense Except in the Light of Evolution”

—Theodosius Dobzhansky (29)

Introduction

IN THE BEGINNING, there was no oxygen (O_2). Energy in the form of reducing equivalents flowed outward across the Earth's crust through “pores,” hydrothermal vents, and volcanoes. Much of this was sulfide, which was then oxidized in the mildly oxidizing atmosphere. Iron also traversed this boundary and its ability to associate with sulfur in the form of iron sulfur clusters enabled both the structural organization of compartment boundaries and the catalytic control of electron transfer (25). The energy could now be harnessed, and life had begun. For the next 3 billion years, organisms continued to evolve and develop more sophisticated methods to control this energy. During this period, rain leached sulfur from the land and as it flowed into the oceans the sulfur became reduced, creating a euxinic environment that was both sulfidic and hypoxic (Fig. 1). Eukaryotic cells arose in this environment from the combination of a sulfide-reducing Archaea and a sulfide-oxidizing α proto-bacterium. This union arguably

enabled sulfur cycling and energy transfer between the cytoplasm and mitochondrion (146). For 500 million years, eukaryotic cells continued to develop and thrive in this environment as evidenced by the fact that the oldest known microfossil was a sulfide-oxidizing organism (180). As photosynthesis began to create O_2 -rich areas “oases” in the seas, organisms living in or around these environments now had to develop strategies to detoxify this unusually reactive gas. It was also soon realized that O_2 's reactivity could be harnessed, and this culminated with O_2 as the ultimate electron acceptor in the electron transport chain. While the increase in ambient O_2 promoted oxidative metabolism, it presented another problem: reduced sulfide disappeared. Organisms were now obligated to change to carbon-based substrates as an energy source. On occasion, ambient concentrations of O_2 and hydrogen sulfide (H_2S) were again reversed and while this euxinic environment produced mass extinctions (51), it likely ensured the survival of organisms whose genome had not only retained, but also passed on the capacity

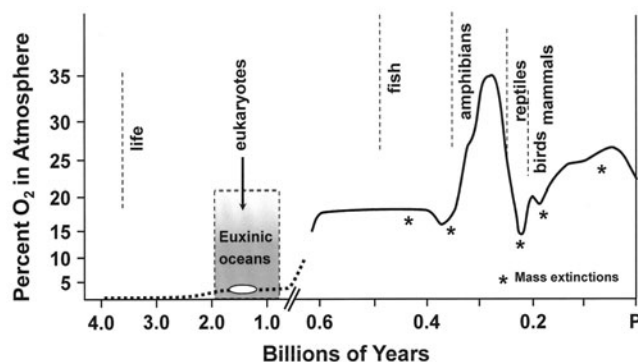


FIG. 1. Oxygen (O_2) and sulfide during evolution. Around 2.5 billion years ago, atmospheric O_2 began to increase and it was $\sim 2\%$ or ~ 15 mmHg (but undoubtedly much lower in the oceans) when the eukaryotes first appeared. Eukaryotic evolution continued for the next 500 million years in this sulfidic and very hypoxic (euxinic) environment. During this time, many of the metabolic pathways for sulfide metabolism were established and they continue to this day as the basis for O_2 sensing. Periodic reciprocal oscillations in O_2 and hydrogen sulfide (H_2S) probably contributed to mass extinctions (*). Drawn from Olson (121).

for sulfide metabolism. Remnants of this early life are present these days, even in modern aerobic eukaryotes, as sulfide is still preferred compared with carbon-based substrates in the electron transport chain [11, reviewed in Olson (121)].

O_2 -Sensing Tissues

The vast majority of modern-day animals, and especially vertebrates, now depend on O_2 and monitoring “sensing” O_2 availability is key for survival. O_2 sensors can be divided into four “reporting” levels: external, internal, tissue (vascular), and intracellular.

External chemoreceptors monitor ambient O_2 . Aquatic vertebrates are especially susceptible to ambient O_2 because of the lower solubility (1/30 that of air), low diffusivity (200,000 times slower), 60-fold higher viscosity, and wide swings in O_2 availability seasonally, daily, and spatially, even within a few meters (8). Chemoreceptor neuroepithelial cells (NEC) on the external surfaces of fish gills are employed to continuously monitor water partial pressure of oxygen (PO_2) (66, 109). Similar neuroepithelial-like cells are found in clusters (neuroepithelial bodies [NEB]) near airway bifurcations in lungs of newborn mammals where they may be important in the transition away from the relatively hypoxic uterine environment during and shortly after birth (72). External O_2 sensors other than NEB are relatively uncommon in terrestrial vertebrates. These are replaced by internal O_2 sensors that are better suited to monitor blood O_2 status and changes in O_2 availability (such as in borrows or with increasing altitude) if needed.

The first and second gill arches of fish are heavily invested with internal arterial-facing NEC, and these are the antecedents of glomus cells in the carotid body and aortic arch. The first gill arch and the mammalian carotid body arise from the third embryonic arch, and the second gill arch and aortic bodies arise from the fourth embryonic aortic arch; NEC and

type I glomus cells of the carotid body are so similar at the ultrastructural level that there is little doubt of their lineage [reviewed in Jonz and Nurse (66)]. Mammalian adrenal medullary cells and homologous chromaffin cells in fish that line systemic veins secrete catecholamines in response to hypoxemia (119, 138) and may monitor tissue O_2 extraction. The adrenal medullary cells may be especially important in monitoring arterial O_2 in the newborn until the carotid bodies become fully functional (66).

The ability of blood vessels themselves to respond to O_2 is important in matching perfusion to metabolic demand or in the case of the respiratory organs to maintain normal ventilation/perfusion ratios. It is commonly assumed that to accomplish this, systemic vessels dilate in response to hypoxia and pulmonary vessels constrict (155) but this is obviously not consistent throughout vertebrates or even within mammals (129, 143).

Finally, it is evident that individual cells not only monitor their own O_2 status but also have biochemical “contingency plans” to adjust metabolic demand and energy utilization should O_2 levels fall. The scope of these contingency plans is highly variable from a few short minutes of survival in highly active mammalian tissues such as the brain and the heart to extended (weeks–months) anoxemia that is tolerated by lower vertebrates such as the crucian carp and turtle (48). Initial sensing and coping mechanisms reflected in response to ischemia/reperfusion injury and pre- and postconditioning are only briefly discussed here in the context of acute O_2 sensing. Chronic responses involving metabolic control are described in an excellent review by Clanton *et al.* (22) and are beyond the scope of the present discussion.

Acute O_2 Sensing

Despite intense interest, the identity of the O_2 “sensor” remains enigmatic. A number of mechanisms have been suggested, but none has achieved unequivocal support. Their pros and cons have been extensively reviewed (22, 37, 47, 54, 144, 155, 184, 187). The intent of the present review is to examine the evidence suggesting that the O_2 -dependent metabolism of H_2S is an O_2 sensor.

H_2S and O_2 Sensing

There are a number of different opinions regarding how (or if) H_2S senses O_2 (Fig. 2). These include (i) direct inhibition of oxidative phosphorylation; (ii) as a downstream effector of a different O_2 sensor, for example, carbon monoxide (CO); or (iii) inverse coupling between O_2 availability and O_2 -dependent metabolism of endogenously generated H_2S .

H₂S as a direct inhibitor of oxidative phosphorylation

It is well known that H_2S inhibits cytochrome C oxidase (CCO) and it has been proposed that this effect, while it mimics hypoxia, is merely an artifact of H_2S poisoning (15, 155, 181) (Fig. 2A). Indeed, H_2S concentrations above 20–40 μM inhibit complex IV (11). However, recent studies have shown that H_2S exerts a biphasic effect on respiration. H_2S stimulates respiration at concentrations of approximately 3 μM when complexes II–IV are intact, whereas the inhibitory effect is only mediated through complex IV at concentrations exceeding 30–100 μM (110). There are other

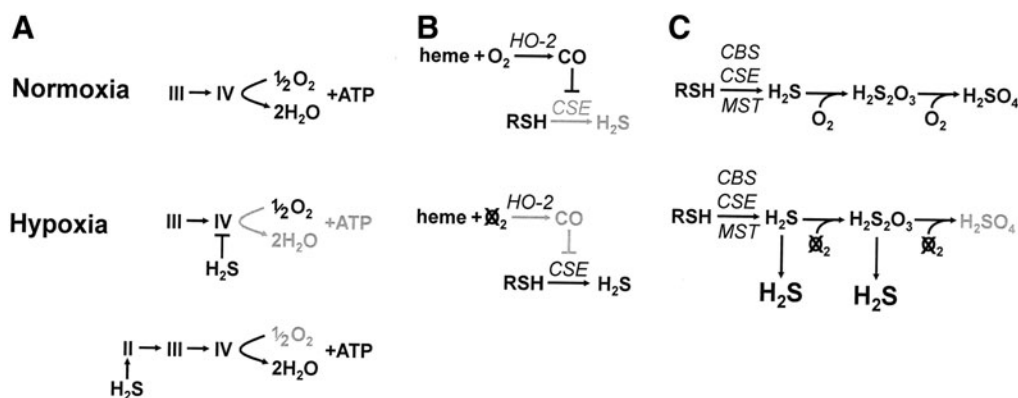


FIG. 2. Mechanisms of H₂S signaling in hypoxia. (A) Exogenous H₂S reversibly inhibits complex IV, thereby mimicking hypoxia. Elevated H₂S could also feed electrons into complex II and by driving oxidative phosphorylation, deplete cellular O₂. (B) H₂S as a downstream effector of carbon monoxide (CO) mediated O₂ sensing. In normoxia, CO is constitutively produced from heme by hemoxygenase 2 (HO-2) and inhibits cystathionine γ -lyase (CSE), thereby preventing H₂S production from thiols (RSH). CO can no longer be produced during hypoxia, and the inhibitory effect on CSE is lost while permitting H₂S biosynthesis. (C) Direct inverse coupling between H₂S synthesis and O₂ availability. In normoxia, H₂S is constitutively generated from RSH in the transsulfuration pathway by CSE, cystathionine β -synthase (CBS), and 3-mercaptopyruvate sulfurtransferase (3-MST) and oxidized to thiosulfate (H₂S₂O₃) and then sulfate (H₂SO₄), which is then excreted. Hypoxia does not affect H₂S production from RSH, but it prevents oxidation of H₂S to thiosulfate, thereby increasing H₂S concentration. In addition, previously formed thiosulfate can no longer be oxidized to sulfate and in the increased reducing environment in the mitochondrial matrix, this thiosulfate is reduced to H₂S and sulfite (not shown). Production of H₂S from thiosulfate is likely to be faster than H₂S production from RSH and may be the initial event in hypoxic signaling.

problems with signaling *via* inhibition of complex IV. First, it puts the cart before the horse. It is unclear what factor(s) would cause H₂S concentration to increase to this level. In addition, H₂S concentrations > 1 μ M would be expected to produce an obvious odor. Second, if H₂S is from an endogenous source, this should result in a fatal positive feedback. If initial, low-level H₂S concentrations partially inhibited CCO, this would further reduce the rate of H₂S oxidation and H₂S concentrations would continue to increase to the point where CCO is completely and irreversibly inhibited. While partial CCO inhibition could be considered a general mechanism for shutting down metabolism and conserving resources, it would not explain how O₂-sensing cells are activated by hypoxia. Third, arguments can be made against H₂S inhibition of O₂ consumption based on mass balance of sulfur and O₂ metabolism. A 70 kg adult male ingests ~26 mmol of sulfur amino acids per day, and total sulfur intake is probably ~40 mmol per day (62, 118). O₂ consumption is ~250 ml/min (179), or 360 L/day, which is equivalent to 15 mol of O₂ per day. Assuming all ingested sulfur is ultimately excreted as sulfate (SO₄); the rate of O₂ utilization to sulfur excretion (4O/S) is 250:1. It is difficult to conceive how sulfide could increase to the point where it has a significant impact on O₂ consumption and, therefore, ATP production. Fourth, inhibition of CCO by exogenous H₂S (15) is unlikely to reflect the *in vivo* situation where H₂S production and metabolism is most likely spatially restricted (122).

H₂S as a downstream effector of CO-mediated O₂ sensing

The relationship between hypoxia and H₂S signaling in the carotid body, adrenal medulla, and cerebral vasculature has been suggested to be a secondary event downstream of the

initial O₂ sensor, CO (Fig. 2B) (60, 139). In the carotid body, it has been proposed that hypoxia inhibits constitutive CO production by hemoxygenase 2 (HO-2), which then relieves the heretofore tonic inhibitory effect of CO on cystathionine γ -lyase (CSE) and H₂S levels increase, closing large-conductance Ca²⁺-dependent K⁺ channels (BK_{Ca}) (87, 162, 163) or TWIK-related acid-sensitive potassium (TASK) (15) channels. This depolarizes glomus cells and releases neurotransmitter. In this sense, H₂S is a mediator of the actual O₂ sensor, CO (139). The role of H₂S in these studies is supported by observations that CSE knockout mice or the CSE inhibitor, propargyl glycine (PPG), inhibits hypoxic responses in both the carotid body and adrenal medulla (97, 136). However, it is unclear how or whether CO actually inhibits CSE (112). In addition, HO-2 null mice have an intact chemoreceptor response to hypoxia (131).

The inhibitory effects of CO on CBS have also been proposed to mediate hypoxic vasodilation in the cerebral cortical circulation (111). Here, HO-2 that is present in neurons and endothelia constitutively generates substantial quantities of CO which inhibits CBS located in the endfeet of astrocytes in contact with the vessel wall. In normoxia, CO inhibition of CBS inhibits release of vasodilator H₂S; whereas the inhibition is lost in hypoxia, resulting in vasodilation.

The anatomical arrangement of CBS and HO-2 is likely unique to the cerebral vasculature, as it does not appear to be found in other systemic vessels. Since CBS controls sulfur flux into the trans-sulfuration pathway and away from the remethylation cycle (60), it is unclear what impact CO in the brain has on overall sulfur metabolism or remethylation, both of which appear to be highly susceptible to O₂ availability in this model. In addition, inhibitors of CSE, but not CBS inhibit hypoxic vasodilation in other systemic vessels such as the rat aorta (125). Obviously, additional studies are required to

elucidate the relationship between CO and H₂S in the carotid and other systemic vessels as well as in the pulmonary circulation.

O₂ sensing by O₂-dependent H₂S inactivation

We proposed that there was a direct metabolic coupling between H₂S and O₂ (125). In this mechanism, H₂S is constitutively generated through sulfur metabolism but intracellular H₂S concentration is maintained at low levels through concomitant oxidation (Fig. 2C). H₂S signaling is achieved through the inability of H₂S oxidation to keep pace with H₂S production during hypoxia. Since we initially proposed this mechanism, a considerable amount of evidence has accumulated that not only supports this hypothesis but also expands the scope of H₂S-mediated O₂ sensing. This includes evidence that (i) H₂S production is O₂ independent, whereas there are numerous effectors of O₂-dependent H₂S metabolism; (ii) O₂-dependent H₂S metabolism is regulated at physiologically relevant P_{O₂}; (iii) the effects of exogenous H₂S mimic hypoxia and compounds that inhibit or augment H₂S production inhibit or augment hypoxic responses, respectively; (iv) H₂S acts on effector mechanisms which are known to mediate hypoxic responses; and, (v) as stated in the introduction, the reciprocal relationship between O₂ and H₂S has been inexorably intertwined throughout evolution, albeit with the ironic twist that the molecule that was once used as the primary energy source is now the reporter for the molecule that replaced it. Points i–iv are considered in the next sections.

Metabolic Relationships Between H₂S and O₂

The next few paragraphs and Figure 3 illustrate the basic pathways of H₂S production and metabolism. Not surprisingly, essentially all H₂S production is independent of O₂. Conversely, O₂ is directly involved in nearly all aspects of H₂S degradation and indirectly involved in other aspects of H₂S metabolism.

H₂S production

H₂S can be generated from the classical trans-sulfuration pathways or through reduction of sulfur in persulfides (R-S_n, where *n* = 2–8). Trans-sulfuration mechanisms have been (and still are) examined in considerable detail, whereas persulfide reduction is only beginning to be understood in the context of H₂S production.

There are four enzymes: cystathionine β-synthase (CBS), CSE (also known as CGL) and tandem catalysis by cysteine aminotransferase (CAT) and 3-mercaptopyruvate sulfurtransferase (3-MST). CAT initially transfers the sulfur from cysteine to α-ketoglutarate, forming 3-mercaptopyruvate. The sulfur is then transferred to 3-MST, forming a persulfide, which is subsequently released as H₂S by a reducing agent such as thioredoxin (Trx) or dihydrolipoic acid.

It is evident that L-cysteine and L-homocysteine account for most H₂S production. CBS, CSE, and CAT/3-MST are cytosolic enzymes, whereas CAT and 3-MST are also present in mitochondria (71). 3-MST is especially abundant in the mitochondrial matrix (103), presumably enabling access to a threefold higher cysteine concentration here than in the cytosol (42). Hypoxia appears to increase CBS in the mito-

chondria (164) and stress-related stimuli can translocate CSE to the mitochondria, although this may take hours (42); additional details are provided in sections “CBS Translocation to Mitochondria” and “CSE Translocation to Mitochondria.” Initially, it was believed that CBS was predominantly found in the brain and CSE was found in the cardiovascular system [reviewed in Kimura (73)], although a broader distribution is now becoming apparent (6, 73, 129). H₂S can also be derived from D-cysteine; however, this appears limited to the brain and kidney, where it protects the former from oxidative stress and the latter from re-perfusion injury (147). CBS, CSE, and CAT are pyridoxal 5′ phosphate-dependent, enzymes. S-adenosylmethionine allosterically activates CBS (152) and although CBS contains a heme group that can be inhibited by CO, neither nitric oxide (NO) nor O₂ appears effective at physiological concentrations (4). CSE and both cytosolic and mitochondrial CAT activity are inhibited by calcium, independent of calmodulin (104, 105).

H₂S metabolism (inactivation)

Theoretically, mitochondrial metabolism is a far more efficient and controllable means of regulating H₂S than simple diffusion out of the cell (122). In fact, there is considerable evidence showing that mitochondria efficiently oxidize H₂S [reviewed in Olson (121)]. Sulfide:quinone oxidoreductase (SQR), 3-MST, rhodanase (Rde), thiosulfate reductase (TR), sulfur dioxygenase (ETHE1), and sulfite oxidase (SO) are mitochondrial enzymes that oxidize H₂S to sulfate (SO₄²⁻) for subsequent excretion. Sulfite (SO₃²⁻) and thiosulfate (S₂O₃²⁻) are intermediates. SQR is bound to the inner mitochondrial membrane and intimately associated with the respiratory chain “supercomplex” (58). This provides a close association with an O₂-sensing process, as the latter also appears to be a mitochondrial event (155). Not surprisingly, 3-MST, Rde, and TR are abundant in the mitochondrial matrix and, to a lesser extent, the intermembrane space (78).

H₂S oxidation begins with H₂S binding to the highly conserved Cys-Cys disulfide bridge of SQR. The sulfide is oxidized to elemental sulfur forming SQR persulfide (sulfane sulfur) with the now-reduced SQR cysteine (SQR-SSH). Two H₂S and two SQR are involved: One H₂S sulfur is transferred to the mitochondrial sulfur dioxygenase, ETHE1, where it is oxidized to sulfite and the second sulfur is transferred from the SQR to sulfite by sulfur transferase producing thiosulfate. One electron from each of the two H₂S are fed *via* the quinone pool (Q) into the respiratory chain (98) and they ultimately reduce O₂ at complex IV. SQR is bound to the inner mitochondrial membrane, and it is believed that sulfur is shuttled from SQR by an as-yet unidentified mobile carrier, possibly glutathione (GSH), dihydrolipoate, Trx, or sulfite (59, 63, 165). SQR is found in all tissues except the brain (58, 82, 89) (but see “2. ETHE1” section), which may account for the pronounced sensitivity of the nervous system to H₂S toxicity. The capacity of cells to oxidize sulfide appears to be greater than the estimated rate of sulfide production (11, 44) and in *Caenorhabditis elegans* sqrd-1, the gene encoding SQR is increased eightfold after exposure to H₂S (108). Thus, it is expected that intracellular H₂S concentrations are very low under normoxic conditions.

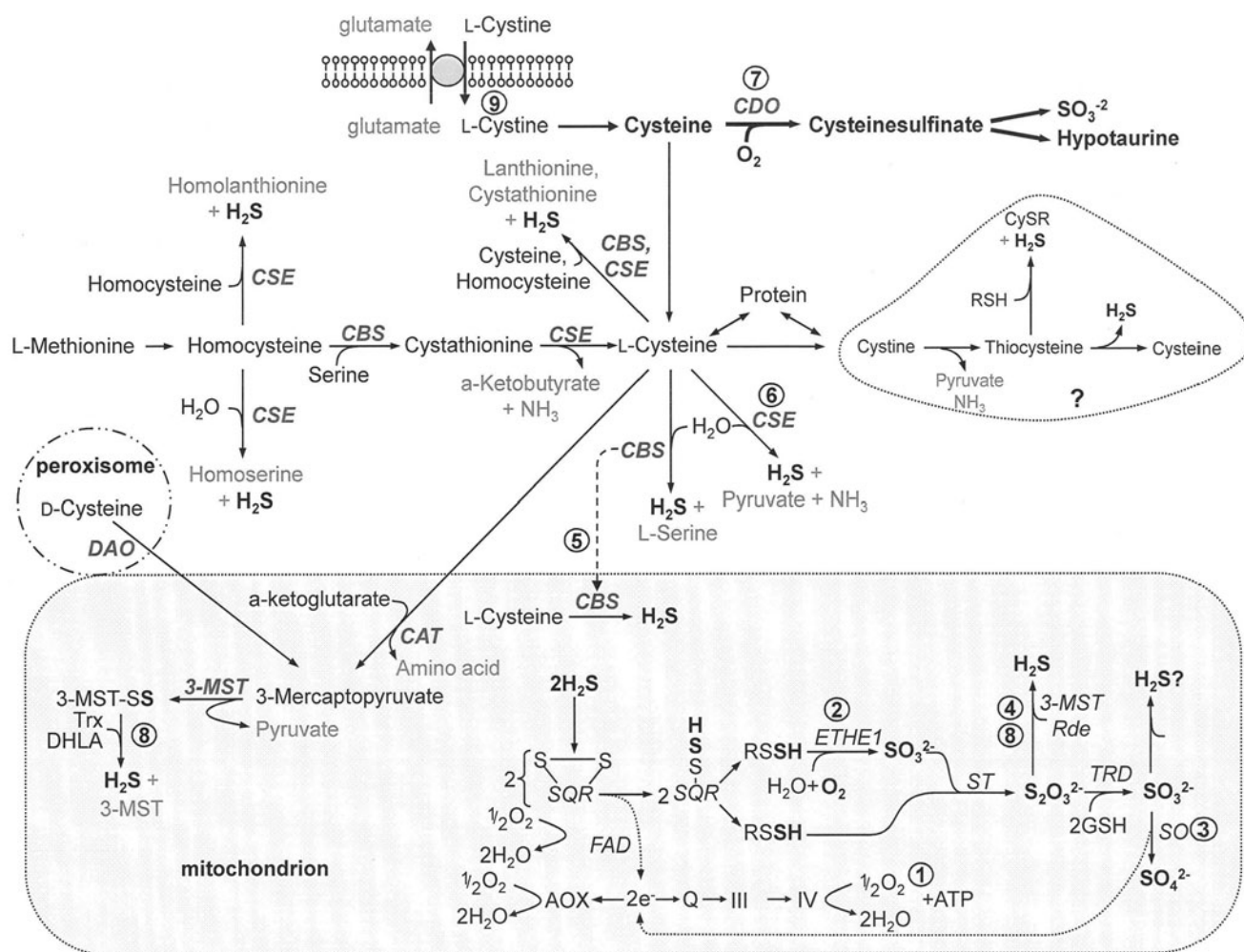


FIG. 3. Pathways for H₂S production and degradation. H₂S is synthesized from homocysteine or cysteine by the cytosolic enzymes cystathionine β -synthase (CBS) and cystathionine γ -lyase (CSE), or the tandem action of cysteine aminotransferase (CAT) and 3-mercaptopyruvate sulfurtransferase (3-MST) in both the cytosol and mitochondria. Other potential mechanisms for H₂S biosynthesis have been described in invertebrates (68) (dotted enclosure) but have yet to be confirmed in mammals. 3-mercaptopyruvate can also be produced from D-cysteine in the peroxisome by D-amino acid oxidase (DAO). During hypoxia, CBS migrates from the cytosol to the mitochondrial matrix and in the absence of Lon protease degradation, generates H₂S from the abundance of cysteine in the matrix. Oxidation of H₂S in the mitochondria is initiated by an interaction with the enzyme sulfide:quinone oxidoreductase (SQR), producing two SQR persulfides that are then transferred to a mobile sulfide carrier (RSSH), one of which is oxidized to sulfite (SO₃²⁻) by mitochondrial sulfur dioxygenase (ETHE1). Sulfur transferase (ST) transfers the other persulfide and forms thiosulfate (S₂O₃²⁻). H₂S may be regenerated from thiosulfate by 3-MST or rhodanase (Rde), or the thiosulfate may be oxidized to sulfate (SO₄²⁻) by the sequential actions of thioredoxin (Trx) reductase (TRD) in combination with glutathione (GSH) and sulfite oxidase (SO). Oxidation of H₂S donates electrons to the respiratory chain (Q, III, IV) that has been shown in invertebrates and mammals to result in ATP production and O₂ consumption. An alternative oxidase (AOX) that oxidizes H₂S without concomitant ATP production has also been observed in invertebrates. H₂S can also be regenerated from thiosulfate by 3-MST or Rde in the presence of other reducing disulfides such as Trx or dihydrolipoic acid (DHLA). Circled numbers indicate actual or potential hypoxia-sensitive sites (see sections “Metabolic Relationships Between H₂S and O₂” and “Effectors of H₂S Metabolism” for details). Modified from Olson (121).

In normoxia, most thiosulfate is further metabolized to sulfate by TR and SO. Although sulfur excretion as thiosulfate would conserve O₂, apparently there is little need for this when O₂ is plentiful and thiosulfate excretion in vertebrates is generally low. In fact, it is not clear why most vertebrates, especially terrestrial ones, would be bothered with this pathway at all, except that it may be an important avenue for regeneration of H₂S during hypoxia (124). A number of reviews (69, 73, 121, 152, 154) provide additional details of H₂S metabolism.

Effectors of H₂S Metabolism

It is now evident that there are a number of mechanisms with which O₂ can potentially influence H₂S concentration, thereby contributing to O₂ sensing. Some act over a relatively short timescale and contribute to acute hypoxic responses while others, operating on a longer timescale, may bias the background levels of H₂S. A number of examples are presented where H₂S or its metabolites are increased due to deficiencies in enzyme activity. Since O₂ is also involved in

these pathways, it is presumed that hypoxia will have a qualitatively similar effect, although this has rarely been examined. Additional factors may also indirectly contribute to the O₂-sensing process. These mechanisms are briefly described next; the reader is referred to corresponding numbers in Figure 3.

Rapid effectors of H₂S concentration

①. Electron transport: The absence, or reduced availability, of the terminal electron acceptor (O₂) prevents electron flow down the respiratory chain, enabling H₂S that was derived from trans-sulfuration to accumulate. This was our initial hypothesis of H₂S-mediated O₂ sensing (125).

②. ETHE1: The mitochondrial dioxygenase, ETHE1, uses molecular O₂ and water to oxidize the mobile persulfide from SQR to form sulfite. Inhibition of this pathway will prevent H₂S binding to SQR and enable H₂S and thiosulfate to accumulate. This has been shown in human and animal models of ETHE1 deficiencies (28, 36, 46, 166).

③. Sulfite oxidase: SO in the mitochondrial intermembrane space catalyzes the oxidation of sulfite to sulfate by transferring an atom of O₂ from water to sulfite and in the process, the enzyme undergoes a two-electron reduction (140). These electrons are then transferred from SO to cytochrome C and then into the electron transport chain, effectively inversely coupling sulfite concentration to O₂ availability. Humans with SO deficiency present with elevated urinary thiosulfate (113). Interestingly, SO is expressed in lung alveolar cells but not vessels (107).

④. Thiosulfate reduction: Thiosulfate is a simple persulfide and any factor that promotes its accumulation, for example, inhibition of electron transport, ETHE1, or SO can potentially produce H₂S. Thiosulfate is easily reduced to SO₂²⁻ and H₂S by endogenous reductants in the mitochondria, and this is catalyzed by 3-MST or TR (also known as Rde) (103, 178). This has been demonstrated in a variety of mammalian and non-mammalian tissues (124) and is likely favored during hypoxia when thiosulfate can accumulate in the mitochondria and the matrix becomes selectively reduced (183). This mechanism is predicted to be relatively rapid and may be considered the initial event in O₂ sensing; it would also conserve biologically relevant thiols by recycling sulfur.

Medium- to long-term effectors of H₂S concentration

⑤. CBS translocation to mitochondria: The mitochondrial heat shock protein (mtHsp 70) transports CBS from the cytosol to the mitochondrial matrix, where under normoxic conditions it is degraded by Lon protease. Oxygenation of the prosthetic heme group in the CBS appears to be key for Lon protease degradation (164). Loss of this O₂ in hypoxia prevents CBS degradation, enabling CBS to accumulate. Mitochondrial CBS concentration doubles within 10 min of hypoxia and increases sixfold within 1 h. Normoxia rapidly restores CBS to control levels within 5 min of normoxic reperfusion. This correlates with a hypoxic increase in mitochondrial, but not cytosolic H₂S production, which could be blocked by aminooxyacetate (AOA). H₂S generated by mitochondrial CBS prevents Ca²⁺-mediated cytochrome C release from mitochondria, thereby preventing mitochondrial swelling, and decreases generation of reactive oxygen species (ROS). This has been proposed to explain the effects

of exogenous H₂S on myocardial and hepatic ischemia/reperfusion injury (164).

⑥. CSE translocation to mitochondria: Hypoxia stimulates CSE translocation from the cytosol to the mitochondria in vascular smooth muscle cells to take advantage of a threefold increase in cysteine concentration and presumably generate ATP from H₂S, supposedly providing some protection from hypoxia (42). However, there are several problems with this theory: First, anaerobic metabolism is sufficient for energy production in vascular smooth muscle, even during hypoxia (35), and second, electron transport cannot continue in the absence of O₂. However, the H₂S formed by CSE translocation could obviously contribute to O₂ sensing and hypoxic vasodilation.

⑦. Cysteine dioxygenase: Cytosolic cysteine dioxygenase (CDO) irreversibly catalyzes the oxidation of cysteine to cysteinesulfinate, effectively eliminating sulfur from entering the H₂S pool (154). CDO activity is dynamically regulated by cysteine (as much as 450-fold), and this is important in the detoxification of excess dietary or metabolic cysteine. Conversely, the H₂S-forming trans-sulfuration pathway is not similarly regulated (154). In the absence of CDO, sulfur is redirected through the desulfuration pathway, which then increases thiosulfate and H₂S production (142, 173). Since O₂ is the only other substrate in CDO-mediated cysteine oxidation, it is likely that hypoxia will also impair cysteine oxidation and favor H₂S production. While this is not likely a rapid response, it could place a long-term bias on chronic O₂ sensing. The striking similarities between CDO and ETHE1 pathologies support sulfide/H₂S as a common factor (173). Apart from this, the inability of CDO to handle a large transient cysteine load may partially explain how hypoxic responses are augmented by exogenous cysteine (see “Physiological Evidence for H₂S-Mediated O₂ Sensing” section).

Indirect O₂ effects

⑧. 3-MST and Trx catalytic-site cysteines: Many enzymes contain cysteine in the catalytic site and since they generally have a low pK_a, they are redox active (116). One example is the catalytic cysteine in 3-MST (Cys²⁴⁷; rat), which, being exposed, is readily oxidized to a sulfenyl (R-SO) by O₂, peroxide (H₂O₂), or other oxidants. This inactivates the enzyme. The sulfenyl can be reduced by reduced Trx (117), and reactivated. Monomeric Rat 3-MST can also dimerize by mild oxidation of two other exposed cysteines, Cys¹⁵⁴ and Cys²⁶³, which also inactivates the enzyme. A defect in 3-MST activity, presented clinically as mercaptolactate-cysteine disulfidias, is believed to be associated with deficient H₂S production (117). It has been proposed that these three cysteines enable 3-MST to serve as an effective antioxidant (117). However, this could also be considered a key component of the O₂-sensing mechanism. Since hypoxia increases ROS in the cytosol while decreasing ROS in the mitochondrial matrix of both pulmonary and systemic arterial smooth muscle cells (183), this would be expected to augment mitochondrial H₂S production while inhibiting production in the cytosol. Parenthetically, the now-oxidized Trx can be reduced by Trx reductase using nicotinamide adenine dinucleotide phosphate (NADPH), which may be an overlooked but key explanation for why NADPH has been central in many O₂-sensing theories (52).

⊙. System X_c⁻: System X_c⁻ is a cystine/glutamate antiporter that transports L-cystine into cells in exchange for glutamate (12). While system X_c⁻ is generally regarded as a mechanism to ultimately increase intracellular L-cysteine for subsequent glutathione synthesis and antioxidant protection, it could also provide a substrate for H₂S formation. In this regard, system X_c⁻ is upregulated in murine neural stem cells by hypoxic preconditioning, although the preconditioning periods were quite long, 45 min and 4 h (149). H₂S also increases X_c⁻ transporter activity in primary cultures of rat cortical neurons, which increases cystine uptake and intracellular cysteine (74). This could potentially provide a positive-feedback enhancement of H₂S production in hypoxia as well. It is also interesting that the highest activity of system X_c⁻ in the central nervous system is found in the hippocampus (149), which was one of the first areas in the central nervous system where the physiological effects of H₂S (selective enhancement of NMDA receptor-mediated responses and facilitation of hippocampal long-term potentiation) were demonstrated (1).

Inverse Relationship Between O₂ and H₂S

The fact that O₂ and H₂S do not typically coexist in tissues is pretty much a universal finding. H₂S is rapidly consumed in gills and gill mitochondria from sulfide-adapted mussel, *Geukensia demissa*, in normoxic conditions but reduced 50-fold by anoxia (79). In rat tissues, O₂ either inhibits H₂S production or it may be associated with H₂S consumption and both are reversed by hypoxia (30, 44). Similar observations have been made in a variety of tissues from numerous vertebrates (32, 89, 94, 126, 128, 186). A compelling argument for H₂S-mediated O₂ sensing can be made by comparing bovine and sea lion lungs. While both tissues exhibit an identical reciprocal relationship between H₂S production/consumption and O₂, both hypoxia and H₂S constrict bovine

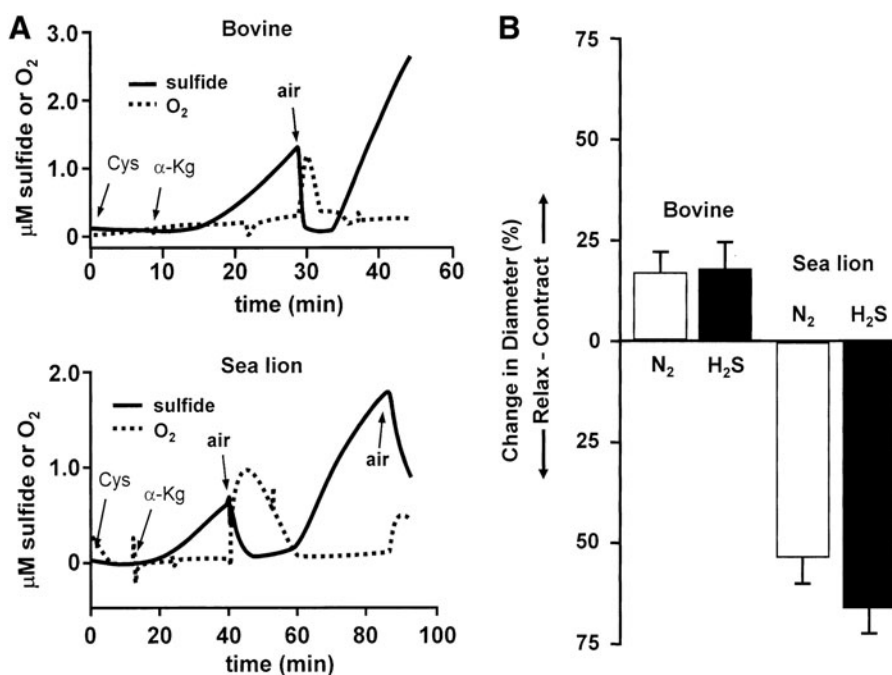
pulmonary arteries but these stimuli dilate pulmonary arteries from the sea lion (Fig. 4) (129).

O₂-Dependent H₂S Metabolism Occurs at Physiologically Relevant P_{O₂}

There is relatively little information on the specific P_{O₂}s at which H₂S oxidation becomes impaired. We used polarographic H₂S sensors to monitor the rate of tissue and mitochondrial H₂S consumption at carefully controlled P_{O₂} (129). This can then be compared with the P_{O₂} that produces hypoxic vasoconstriction or the O₂ sensitivity of other putative O₂-sensing mechanisms. As shown in Figure 5A, the efficiency of H₂S oxidation by bovine lung homogenate, bovine pulmonary arterial smooth muscle cells, or purified bovine heart mitochondria begins to fail when P_{O₂} falls below 30 mmHg; and half-maximal inhibition (P₅₀) of H₂S oxidation occurs around 4–7 mmHg for tissue and <1 mmHg for isolated mitochondria. These P₅₀s are physiologically relevant, as the corresponding P_{O₂}s are routinely encountered during hypoxia (182). Not surprisingly, the P₅₀ for H₂S oxidation of bovine lung or pulmonary arterial smooth muscle is essentially identical to the P₅₀ of hypoxic pulmonary vasoconstriction of bovine pulmonary arteries and also similar to that of lamprey and New Zealand hagfish dorsal aortas. Since mitochondria function at a P_{O₂} well below cytosolic P_{O₂}s, their P₅₀ is also correspondingly lower as is the P₅₀ for hypoxic vasoconstriction of the hypoxia-tolerant hagfish dorsal aorta. A slightly higher P₅₀ (7.5 mmHg) has been observed for H₂S oxidation in mitochondria of *G. demissa* (79), which may reflect the ability of these animals to live in sulfidic environments.

The carotid body has an unusually high metabolic rate and O₂ sensitivity (16). Although O₂-dependent H₂S metabolism has not been measured in the carotid body, its corollary, O₂-dependent H₂S production has been measured (136) and it also correlates exceptionally well with O₂ sensitivity (Fig. 5B).

FIG. 4. Comparison of the effects of O₂ on tissue H₂ production (A) and vessel response (B) in bovine and sea lion tissues. (A) Continuous recording of O₂-dependent H₂S production by homogenized bovine and sea lion lungs measured in real time with amperometric electrodes. Cysteine (Cys, 1 mM) and α -ketoglutarate (α -Kg, 1 mM) were added at arrows. H₂S production steadily increased until samples were exposed to O₂ (air), at which time the H₂S was consumed. H₂S production resumed when the O₂ was consumed. (B) Effects of hypoxia (N₂) and H₂S (3 \times 10⁻⁴ M) on U-46619 (10⁻⁶ M) precontracted bovine and sea lion resistance pulmonary arteries (RPA; <400 μ M dia). Both stimuli contracted bovine but relaxed sea lion RPA. Redrawn from Olson *et al.* (129).



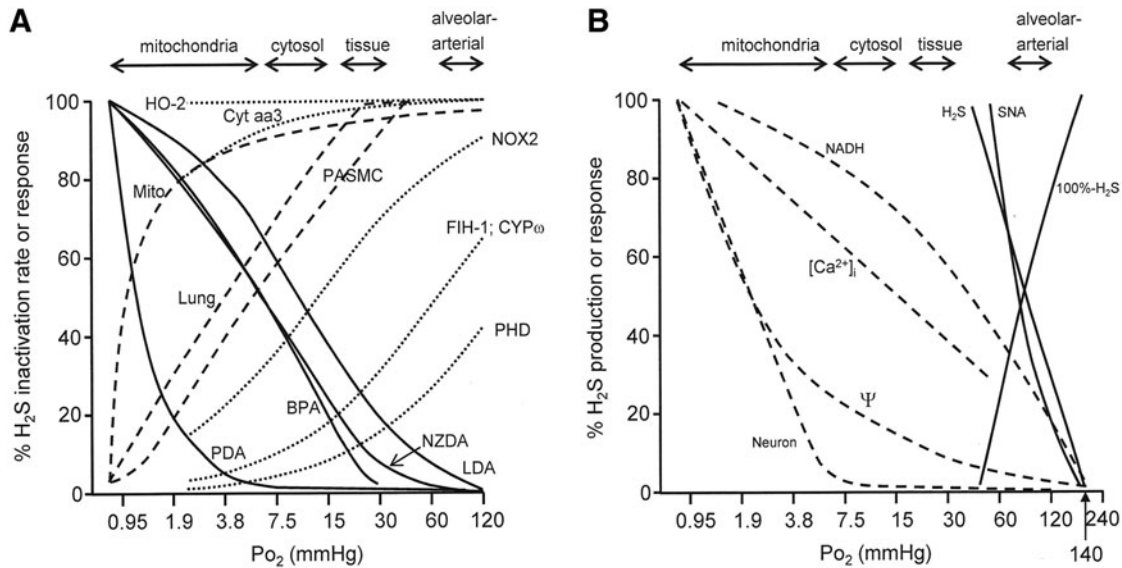


FIG. 5. Effect of O₂ on H₂S metabolism and tissue response. (A) Comparison of O₂ sensitivity of H₂S consumption by homogenized bovine lung (Lung), pulmonary arterial smooth muscle cells (PASMC), and bovine heart mitochondria (Mito) to hypoxic vasoconstriction in bovine pulmonary arteries (BPA), lamprey dorsal aorta (LDA), and dorsal aortas from New Zealand and Pacific hagfish (NZDA and PDA, respectively). O₂ sensitivity of tissue H₂S consumption is similar to O₂ sensitivity in vessels from O₂-sensitive vertebrates (bovine, lamprey, and New Zealand hagfish), whereas O₂ sensitivity in Pacific hagfish aortas is considerably lower commensurate with their tolerance to hypoxia. *Double-headed arrows* above figure show approximate ranges of O₂ tension in different tissue and cellular compartments. The partial pressure of oxygen (Po₂) values at which H₂S metabolism is impaired are at the low end of cytosolic and mitochondrial Po₂ and would be expected during hypoxia. *Thin lines* indicate efficacy of other putative O₂-sensing mechanisms, HO-2, mitochondrial cytochromes a and a3 (Cyt aa3), nicotinamide adenine dinucleotide phosphate (NADPH) oxidase (NOX2), factor inhibiting hypoxia-inducible factor-1 α (FIH-1), cytochrome P450 monooxygenase ω (CYP ω), and prolyl hydroxylase domain proteins involved in O₂-dependent hydroxylation of protein residues on hypoxia-inducible factor (HIF)-1 α (PHD). (B) Comparison of O₂ sensitivity of afferent sinus nerve activity from the carotid body (SNA) to H₂S production (H₂S; or its inverse, 100%-H₂S) and components of intracellular signaling in the carotid body. The P₅₀ for carotid activation is essentially identical to the P₅₀ for H₂S production, which is more evident when the latter is expressed as the inverse (100%-H₂S). The P₅₀ for intracellular excitation events, mitochondrial NADH, intracellular calcium ([Ca²⁺]_i), mitochondrial transmembrane potential (Ψ), or activation of sympathetic neurons (Neuron). Carotid sinus nerve activation and H₂S production are well below the P₅₀s of the intact carotid body or H₂S production and more in line with other tissues in (A). (A) Adapted from Refs. (102, 127, 129, 182). (B) Adapted from Refs. (16, 136).

Mechanism of H₂S Signaling

Sulfur is a highly reactive molecule that is able to bind with its own species as well as with nitrogen and O₂ moieties. This has led to a wide variety of potential sulfide signaling mechanisms, some of which are well established while others are only speculative. These are briefly described in the next few paragraphs; detailed reviews can be found in Refs. (75, 86, 134, 135, 167, 168, 181, 196). The downstream consequences of these initial signaling events on ion channel permeability, transport processes, enzyme activity, or structural proteins are not considered.

Reactions with sulfur

Cysteine is important in protein structure as well as an active component in many enzyme reactive centers; cysteine sulfur is also the most reactive nucleophile in the cell. Thus, it is susceptible to modification by a variety of molecules, including sulfide. Low-molecular-weight sulfides, namely H₂S and polysulfides (H₂S_n, where $n = 2-8$), can modify cysteine sulfur by either creating polysulfides or breaking disulfide bonds. Sulfide can also reverse the effects of cysteine nitrosylation and sulfonylation/sulfinylation by displacing ni-

trogen and O₂. These sulfur modifications are described next and shown in Figure 6. It should be noted that many of these reactions also generate other signaling molecules (H₂O₂, HO₂[•], HNO, and H₂S₂) that may have additional (and perhaps greater) stimulatory effects.

Sulphydration. The term "sulphydration" was initially proposed by Mustafa *et al.* (114) to describe the mechanism of H₂S signaling by which H₂S combined with a protein thiol to form a persulfide (RSSH). Unlike nitrolylation, this was proposed to activate enzymes such as GAPDH and the authors reported that as many as 30% of protein thiols were sulphydrated. The term "sulphydration" is probably more correctly denoted as "sulfuration" (169). Sulfurated proteins and a variety of potential metabolic pathways for their production were previously described by Toohey (167), who designated the reactive sulfur as "sulfane" sulfur [see also Toohey (168)]. This further confounds the nomenclature, as "sulfane" is now the IUPAC name for H₂S (86). More importantly, it has recently been suggested that sulfuration of protein cysteine by H₂S is not possible, because the H₂S sulfur is in its most reduced form and, therefore, incapable of oxidizing cysteine sulfur (50, 75, 169, 193, 196). However,

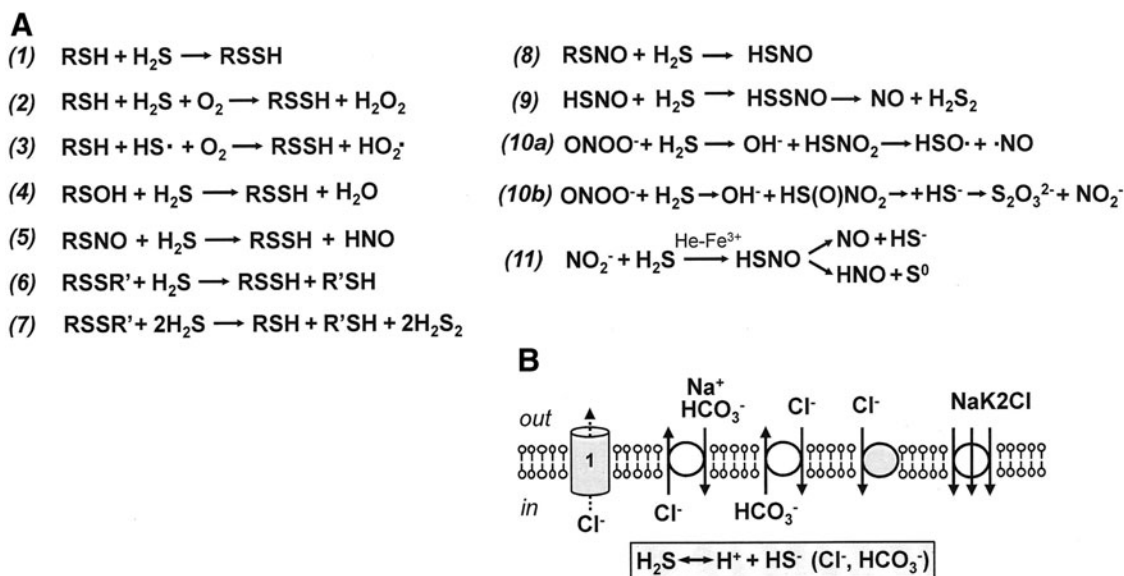
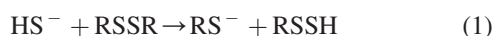


FIG. 6. Mechanisms of sulfide signaling. (A) Sulfhydration of a protein thiol. The generic mechanism proposed by Mustafa *et al.* (115); (1) is generally believed to be not possible, as the reduced sulfide is incapable of oxidizing cysteine sulfur. However, there are a number of other possible mechanisms, including spontaneous oxidation of H₂S (2), metal-catalyzed oxidation of H₂S with a radical intermediate (3), reaction with sulfenic acids (4), reaction with S-Nitrosothiols (5), or reduction of the disulfide bond (the latter may not be thermodynamically favorable) (193). It should be noted that reactions (4) and (5) can also proceed without sulfhydryl formation, thereby restoring the original thiol. Two-step reduction of a disulfide bond involving two H₂S (7). Reactions with nitrogenous compounds include an H₂S reaction with S-nitrosols to produce thionitrous acid (HSNO) (8), an HSNO reaction with excess H₂S to form HSSNO (SSNO⁻), which slowly decomposes to nitric oxide (NO), and H₂S₂ (possibly *via* a disulfide radical, SS^{•-}, intermediate, not shown) (9), scavenging peroxyxynitrite to form thionitrite (HSNO₂), which can produce HSO[•] and [•]NO under anerobic conditions (10a) or produce sulfinyl nitrite [HS(O)NO], which under aerobic conditions can react with additional H₂S and form S₂O₃²⁻ and NO₂⁻ (10b), or iron-catalyzed reduction of nitrite (NO₂⁻) to form NO or nitroxyl (HNO) (11). Many of the reactions in (A) also generate other signaling molecules (H₂O₂, HO₂[•], and HNO) that may have additional and perhaps greater stimulatory effects. (B) At physiological pH, from half to two-thirds of dissolved H₂S exists as the anion, HS⁻ in intracellular and extracellular fluid, respectively. This HS⁻ can compete with a variety of chloride (Cl⁻) and bicarbonate (HCO₃⁻) anion channels and transporters, thereby affecting transmembrane potential.

sulfhydration can be achieved through previous oxidation of H₂S or H₂S reduction of S-nitrosols or sulfenyl sulfur (193).

Disulfide bond cleavage. Certain protein disulfide bonds may serve as specific “receptors” for H₂S. Two H₂S molecules can reduce disulfide bonds in a two-step reaction; although a sulfhydrated intermediate is formed, it does not appear to be stable (161, 196). H₂S is proposed to initiate a nucleophilic attack in which 2 HS⁻ are involved in a two-step reaction:



This reaction appears specific for H₂S as opposed to other thiols (cysteine, glutathione, *etc.*), because H₂S can penetrate deeply into the protein and H₂S may have some property in excess of its reducing ability to interact with these bonds (161). Unlike sulfhydration, disulfide bonds readily reform when H₂S is removed (196). H₂S appears to use this mechanism to close K_{ATP} channels by breaking disulfide bonds in the SUR subunit (65). In reaction 2 (above), the disulfide

product, HSSH can also act as an oxidant and interact with other proteins (50).

Reduction of sulfenyl sulfur and S-nitrosothiols. Similar to the mechanism of sulfhydration (above), H₂S could signal by reducing sulfenyl or S-nitroso groups without formation of a persulfide. This would restore the original thiol. H₂S can also release NO from nitrosothiols, thereby initiating the NO signaling cascade (130).

Reactions with NO and related compounds

With NO. A number of *in vitro* studies in a marine echinuran worm and fish (67), mammalian heart (191, 192), and RAW246.7 cells or liver homogenate (185) have shown that addition of H₂S (as a sulfide salt or slow-releasing donor) to an NO donor (often sodium nitroprusside [SNP]) elicits a response in the tissue that is either greater than or different from the parent compounds. It has been proposed that this new signaling molecule is nitroxyl (HNO/NO⁻), a one-electron reduced and protonated form of NO (191, 192), and/or a S-nitrosothiol (thionitrous acid [HSNO]) (40, 185). The manner in which H₂S and NO react has not yet been resolved, and a number of possibilities have been proposed (75). Neither HNO nor HSNO can be produced by the direct

interaction of $\text{H}_2\text{S}/\text{HS}^-$ and NO as the former is a diamagnetic reductant or nucleophile, whereas NO is a paramagnetic free radical (43, 75, 86). However, one-electron reduction of $\text{H}_2\text{S}/\text{HS}^-$ to produce the thiyl radical (HS^\bullet) or a one-electron oxidized species of NO, such as nitrite (NO_2^-), can react to produce HSNO/SNO^- (40). HSNO/SNO^- is reported to be further metabolized to NO^+ , NO, and NO^- with a variety of distinct stimulatory functions, and since it can diffuse freely across membranes, it may facilitate transnitrosation of proteins (40). SNP appears to be a poor choice of an NO donor, as it directly reacts with H_2S and produces HNO and although this is a source of HNO, the chemistry does not appear to be similar to $\text{H}_2\text{S}/\text{NO}$ interactions in biological systems (38). HSNO/SNO^- can further react with additional H_2S and form H_2S_2 (40). Recently, Cortese-Krott *et al.* (23) showed that excess H_2S (HS^- ; from Na_2S or Gyy4137) interacted with S-nitroso-N-acetyl-DL-penicillamine (SNAP) and formed nitropersulfide anion (SSNO^-). This was more stable than SNO ($t_{1/2} > 30 \text{ min vs. } < 2 \text{ min}$), prolonged the release of NO from SSNO, and sustained vasodilator activity. The remaining hydropersulfide (HSS^-) can also mediate a variety of biological functions as described earlier, but again, the physiological relevancy of this is questionable.

With peroxyxynitrite. H_2S can scavenge peroxyxynitrite and form thionitrite (HSNO_2), which can then produce HSO^\bullet and $^\bullet\text{NO}$ under anerobic conditions or sulfinyl nitrite [HS(O)NO] under aerobic conditions (39). The latter can react with additional H_2S and form $\text{S}_2\text{O}_3^{2-}$ and NO_2^- .

Reduction of nitrite. Nitrite (NO_2^-) has been proposed to be a regulator of hypoxic signaling *via* a heme-iron catalyzed reduction to NO, with the latter then mediating hypoxic vasodilation and protection against ischemia/reperfusion injury. However, the biochemistry has been difficult to resolve (176). Myoglobin (Mb) heme appears to be key to hypoxic vasodilation, as it is present in rat aortic smooth muscle and in Mb null rats ($\text{Mb}^{-/-}$) hypoxic vasodilation is impaired (170). Recently, Miljkovic *et al.* (106) showed that H_2S is a rapid mediator of nitrite reduction by providing reducing equivalents to Fe^{3+} , thereby generating both NO and its reduced congener, nitroxyl (HNO). Vasodilation produced by acute hypoxia in mice *in vivo* appears to be independent of NO production, and it is associated with reduced plasma nitrite concentration and increased RSNOs (174). This is consistent with H_2S , as the initial hypoxic signal that mediates downstream increases in both H_2S_2 and HNO as shown in Figure 7. It also explains, parenthetically, why nitrite is an antidote for H_2S poisoning.

As an anion

Chloride is accumulated against an electrochemical gradient by a variety of cells, including vascular smooth muscle, and it is an important regulator of vasoactivity (21, 99). Metabolically generated bicarbonate (HCO_3^-) can interact with Cl^- antiporters and produce either vasodilation or vasoconstriction (95, 96). With the pK_{a1} of ~ 6.9 for the reaction $\text{H}_2\text{S} \leftrightarrow \text{HS}^- + \text{H}^+$, half of intracellular sulfide and more than two-thirds of extracellular sulfide is also an anion. Although H_2S freely diffuses across lipid membranes (100), it is conceivable that HS^- can also substitute for, or compete

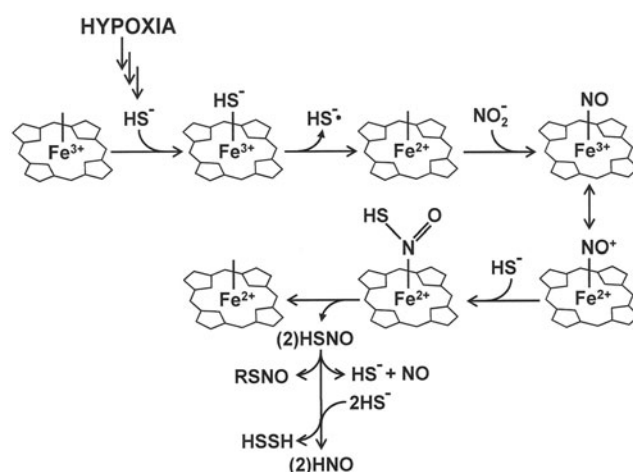


FIG. 7. Proposed mechanism of H_2S signaling in myoglobin-mediated formation of vasoactive molecules. H_2S produced during hypoxia reduces ferric (Fe^{3+}) myoglobin to ferrous (Fe^{2+}) myoglobin, liberating a thiyl radical (HS^\bullet). Ferrous myoglobin then binds NO_2^- and the myoglobin is re-oxidized, forming NO. Reduction of the iron forms a nitrosyl- Fe^{2+} that then binds an additional H_2S which results in the liberation of thionitrus acid (HSNO) and this can also give rise to H_2S , polysulfide (HSSH), NO and its reduced congener, nitroxyl (HNO), and protein S-nitrosothiol (RSNO), all of which could contribute to vascular signaling. Modified from Miljkovic *et al.* (106).

with, the main regulatory anions, Cl^- and HCO_3^- (Fig. 6B). This is supported by observations that HS^- is transported into mammalian red blood by the AE1 antiporter (64), and H_2S -mediated vasodilation appears to involve chloride/bicarbonate channels (76, 84).

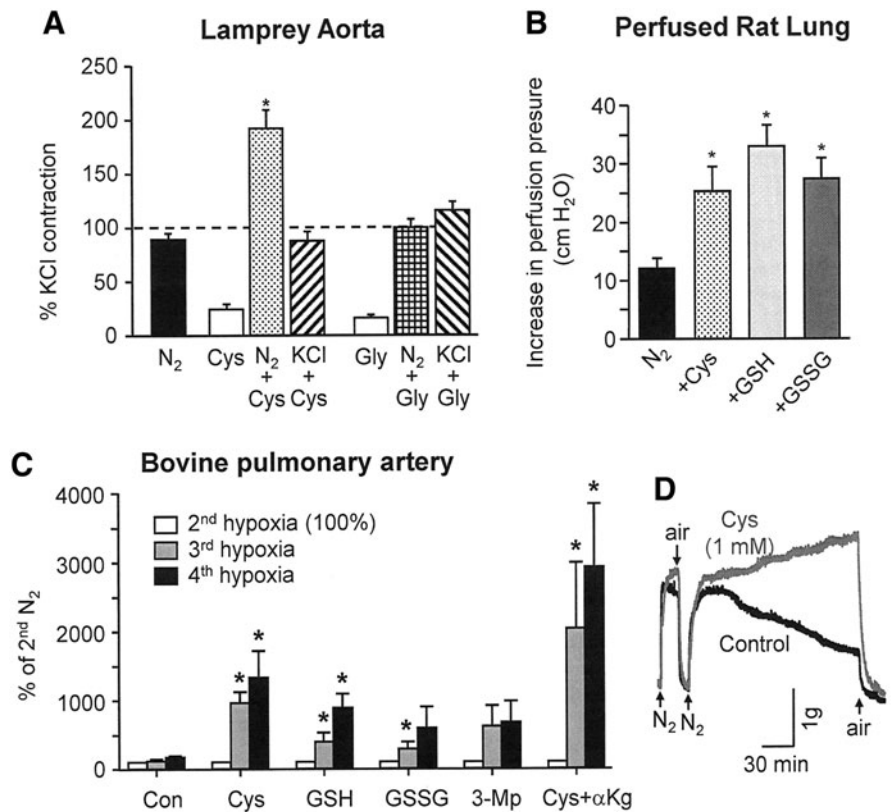
Physiological Evidence for H_2S -Mediated O_2 Sensing

Cardiovascular system

Identical responses to hypoxia and H_2S , be it constriction, dilation, or multi-phasic, have been observed in more than 30 studies on isolated vessels, perfused organs, and intact animals encompassing all classes of vertebrates (27, 94, 123, 125, 128, 129, 150, 151). Not surprisingly, H_2S also dilates the mouse ductus arteriosus (5), which would be expected to keep this vessel patent in the relative hypoxic intrauterine environment. Perhaps the strongest argument for the commonality of H_2S and hypoxia is the observation that both constrict bovine pulmonary arteries, whereas they dilate pulmonary arteries from the sea lion (129) (Fig. 4B).

Compounds that augment or inhibit H_2S production augment or inhibit hypoxic responses. Sulfur donors, especially cysteine, are well known to augment hypoxic responses (Fig. 8). Cysteine increases the magnitude of hypoxic vasoconstriction of isolated lamprey aortas (125), bovine pulmonary arteries (125, 129), and the perfused rat lung (94). Both reduced and oxidized glutathione augment hypoxic vasoconstriction in pulmonary arteries and the perfused rat lung and cysteine plus α -ketoglutarate (presumably utilizing the CAT/3-MST pathway) increases hypoxic vasoconstriction in bovine pulmonary arteries (94, 129). The presence of exogenous cysteine also sustains hypoxic vasoconstriction

FIG. 8. Potential sulfur-donating molecules augment hypoxic vasoconstriction in vertebrates. (A) Cysteine (Cys, 1 mM) nearly doubles a hypoxic contraction but does not affect a KCl (80 mM) contraction in the lamprey aorta. Glycine (Gly, 1 mM) does not affect the hypoxic response. **(B)** Cys (1 mM) and reduced or oxidized glutathione (GSH, GSSG, 1 mM) increase perfusion pressure in the perfused rat lung. **(C)** 1 mM Cys, GSH, GSSG, and Cys + α -Kg enhance consecutive hypoxic contractions of bovine pulmonary arteries (BPA). **(D)** Representative myograph traces demonstrating the ability of Cys (1 mM) to prolong a hypoxic contraction in an isolated BPA. From Olson (120).



(Fig. 8D), enhances hypoxic relaxation of rat aortas (14) and perfused trout gills (150).

Although inhibitors of H₂S biosynthesis have a variety of inherent problems (3, 156), they have been shown to inhibit hypoxic responses of lamprey aorta, bovine pulmonary arteries, rat aorta, and perfused trout gills and rat lungs (94, 125, 150) (Fig. 9). With these caveats, CSE appears to be the major pathway for H₂S production by systemic vessels, and CBS plus perhaps CAT/3-MST contribute to H₂S production in bovine pulmonary vessels, whereas CSE may be important in the rat lung (94).

Respiration

General effects on respiration

Although high levels of H₂S inhibit respiration, an intravascular injection or inhalation of lower concentrations of H₂S mimics hypoxic hyperventilation in fish (126), birds (77), and mammals (7, 53, 55–57, 141, 175). These effects appear to be mediated through both central and peripheral mechanisms.

H₂S mediation of central respiratory centers

Injection of H₂S into the cerebral ventricles produces a K_{ATP} channel-mediated dose-dependent bradycardia and hypotension (90) mimicking the hypoxic diving reflex in mammals. H₂S increases discharge frequency from the pre-Bötzinger (pB) dorsal inspiratory respiratory group, and it may initially produce transient inhibition of the pB by stimulating the nearby parafacial respiratory group (pFRG) (19, 61). Discharge frequency of hypoglossal rootlets in medullary slices of female

neonatal Sprague–Dawley rats is transiently decreased and then increased as H₂S is increased from 100 to 300 μ M and then decreased at 400 μ M H₂S (61). The effect of 100 μ M H₂S was mimicked by 200 μ M cysteine and the CBS activator S-adenosyl-L-methionine (AdoMet) and inhibited by the non-specific inhibitor, hydroxylamine. In a later study, Chen *et al.* (19) found that H₂S increased burst frequency in the pB complex and produced biphasic (initial decrease followed by an increase) responses in slices that contained the pFRG. Lesioning the pFRG removed the initial inhibitory effect of H₂S. A selective micro-injection into the preBötC increases discharge frequency whereas it decreases discharge frequency when injected into the pFRG, thus identifying the effect of H₂S in the two different respiratory centers. In other studies using a similar preparation (132, 133), the authors found that H₂S prevented the inhibitory effect of hypoxia on burst activity and these effects were suppressed by pretreatment with the K_{ATP} channel inhibitor glibenclamide and enhanced by cysteine. Hydroxylamine postponed recovery from hypoxic inhibition and significantly enhanced hypoxia-induced increase in malondialdehyde content in the slices. Hypoxia-induced upregulation of *c-fos* mRNA could be antagonized by SAM but was increased by hydroxylamine. These studies suggest that H₂S helps protect the medullary respiratory centers from hypoxic injury *via* an anti-oxidation effect and by downregulation of *c-fos*.

Although removal of the first pair of gill arches, the site of peripheral chemoreceptors in the rainbow trout, inhibited hypoxic bradycardia, it did not affect hypoxic hyperventilation, nor did addition of the CBS inhibitor, AOA, or the CSE inhibitor, PPG (126). This suggests that fish have central chemoreceptors as well; however, it remains to be determined whether H₂S signaling is involved.

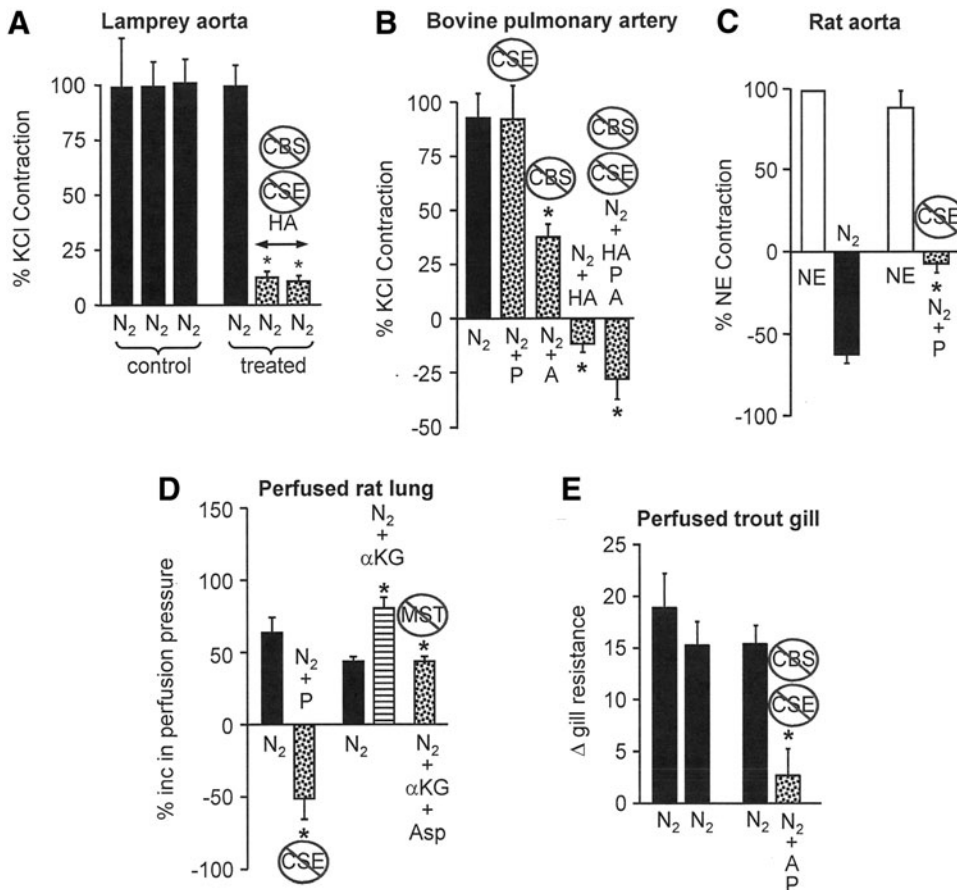


FIG. 9. Inhibiting H₂S production inhibits hypoxic vasoconstriction and vasodilation. Inhibitors of H₂S biosynthesis inhibit hypoxic vasoconstriction in the lamprey aorta (A), and BPA (B) and hypoxic vasodilation of the norepinephrine (NE, 1 μM) precontracted rat aorta (C). Inhibition of H₂S biosynthesis also inhibits hypoxia vasoconstriction in the perfused rat lung (D) and perfused trout gill (E). A, aminooxyacetate a CBS inhibitor (1 mM); Asp, aspartic acid, an inhibitor of MST (10 mM); HA, CBS, and CSE inhibitor hydroxylamine (1 mM); P, propargyl glycine a CSE inhibitor (10 mM); α-Kg, α-ketoglutarate a substrate for MST. From Olson (120).

H₂S mediation of peripheral chemoreceptors, NEC, and carotid body

H₂S stimulates peripheral chemoreceptors (NEC) in the trout gill, and this mediates hypoxic bradycardia (126). H₂S also depolarizes NEC isolated from zebrafish gills, as does hypoxia. Although both CBS and CSE transcripts have been identified in the gills of rainbow trout, the specific location of these enzymes has not been resolved (126). Since NEC are the antecedents of carotid glomus cells (66), it is likely that NEC signaling is homologous to that in their mammalian counterparts.

A number of studies have shown that exogenous H₂S depolarizes the mammalian carotid glomus cells, increases afferent nerve activity, and *in vivo* mimics hypoxic hyperventilation (15, 26, 87, 97, 136, 145). As described in section "H₂S as a Downstream Effector of CO-Mediated O₂ Sensing," this presumably entails closing BK_{Ca} and/or TASK channels. Low doses of exogenous H₂S have a synergistic effect on the hypoxic response and accelerate the sinus nerve response to hypoxia, whereas maximal stimulatory concentrations prevent hypoxic responses (87, 97). This is similar to that observed in blood vessels and suggests that hypoxia and H₂S operate through a common effector mechanism. However, it should be noted that H₂S has been reported to inhibit release of ATP and acetylcholine in the carotid body (41). Since the latter results seem to be inconsistent with all other studies on the carotid body, they need to be confirmed.

CBS and CSE immunoreactivity has been identified in glomus cells in the cat, rat, and wild-type (WT; CSE^{+/+}) mouse but CSE immunoreactivity is absent in CSE null (CSE^{-/-}) mice (41, 87, 97, 136). This has afforded a means of approximating the sources of endogenous production, which can be complicated by the nonspecificity of enzyme inhibitors. Li *et al.* (87) inhibited hypoxia-stimulated afferent nerve activity in the mouse carotid body *in vitro* and blunted hypoxic hyperventilation *in vivo* with either AOA or hydroxylamine, suggesting that H₂S is derived from CBS as PPG and β-cyanoalanine were ineffective. Other studies have suggested that the main source of glomus cell H₂S is CSE, as hypoxic responses were not demonstrated in the presence of CSE inhibitors or in CSE null mice and CBS inhibitors were generally ineffective (26, 97, 136, 145). However, lower inhibitor concentrations of AOA and PPG, when combined, had a greater inhibitory effect than AOA alone, suggesting some degree of CBS activity as well. Perhaps related is the observation that PPG enhances the hyperventilatory response to hypercapnia (26, 136). This suggests that a CBS- or MST-mediated CO₂-sensing pathway was unmasked. If this turns out to be the case, it will add another dimension to H₂S signaling in chemoreceptor cells. Recent studies have shown that the commonly employed CBS inhibitor, AOA, is even a more potent inhibitor of CSE (3) and the role of CBS-mediated H₂S production in chemoreceptor cells needs to be re-evaluated. Nevertheless, it is evident that H₂S is intimately involved in peripheral chemoreceptor function.

Hypoxia (P_{O₂} ~30 mmHg) increases H₂S production in rat carotid bodies (97, 136) (Fig. 5B) and this, but not baseline production, is inhibited by PPG. H₂S production was also lower in CSE^{-/-} than in WT mice, and hypoxia or the CO donor ([Ru(CO₃)Cl₂]₂) increased H₂S production in WT, but not CSE^{-/-} animals. (Conversely, breathing 100% O₂ suppresses H₂S-induced hyperventilation, suggesting that enhanced O₂ increases H₂S metabolism by the glomus cells) (175). However, published rates of H₂S production by glomus cells should be viewed with caution. Assuming a cell with 15% protein and 70% cytosolic water by weight, 1 nmol H₂S/mg cell protein is equivalent to 214 μmol H₂S/L in the cytosol. Using this conversion, the basal tissue production rate of 55 nmol H₂S/h/mg protein reported in mouse glomus cells by Peng *et al.* (136) would increase cytosolic H₂S to 12 mM in 1 h. Doing so would essentially consume all of the glutathione sulfur in a typical cell in 1 h. Peak production of 1300 nmol H₂S/h/mg protein reported by Peng *et al.* (136) in rat carotid bodies would increase cytosolic H₂S to 278 mM in 1 h. This would consume all of the cell's sulfur in less than 15 min.

The effects of H₂S on glomus cells have been attributed to BK_{Ca} channels (87, 162, 163) or TASK channels (15). Li *et al.* (87) showed that in whole cell recordings of mouse glomus cells, both hypoxia and H₂S inhibited BK_{Ca} channels and AOA could inhibit the hypoxic affect. CO reversed the H₂S inhibitory effect on the channels. K_{ATP} channels do not seem to be involved (136).

Calcium influx is a requirement for hypoxic and H₂S activation of glomus cells (15, 87, 97, 136). Hypoxia-, but not H₂S-mediated increases in intracellular calcium are inhibited by PPG and do not occur in CSE^{-/-} mice.

Glomus cells exhibit an unusually high sensitivity to hypoxia (Fig. 5B), and an increase in afferent nerve activity is often observed when P_{O₂} falls below 100 mmHg (16, 97, 136). Glomus activation is often regarded as a sequential process by which the O₂ sensor (CO, H₂S, inhibition of oxidative phosphorylation, i.e., the metabolic hypothesis, or others) initiates potassium channel closure and the resulting cellular depolarization increases calcium entry. This increases intracellular calcium, initiates neurotransmitter release, and activates neuronal afferents. However, recent studies have suggested that this may be a two-step process in which the O₂ sensor initiates an increase in intracellular calcium and a second regulatory step involving either a downstream effector or the activated sensor (or a mechanism coupled to it) acts through protein kinase C (PKC) to modulate the magnitude of responsiveness (34). Although the exact mediators of this process are unknown, it is interesting that H₂S can directly affect both potassium (and even calcium) channels and PKCε (122).

Other studies suggest that H₂S mediates chemoreflex activity under pathological conditions. Heart failure, hypertension, and renal failure activate the carotid body, which contributes to breathing instability and autonomic dysfunction (145). In a rat model of congestive heart failure, PPG inhibition of CSE decreased the hypoxic chemoreflex response and afferent sinus nerve activity, as it did in normal rats. PPG also led to breathing stability, decreased sympathetic nervous system activity, substantially reduced apnea index and breath rate variability, and partially reversed heart rate and systolic blood pressure variability (26).

Role of CO

It has thus far been difficult to determine whether CO is the actual O₂ sensor and H₂S modulates this response or vice versa. As described earlier, a number of studies have suggested that CO is the O₂ sensor and the H₂S effects are modulatory. The CO donor [Ru(CO₃)Cl₂]₂ reverses the stimulatory effect of H₂S on the sinus nerve and the inhibitory effect of H₂S on BK_{Ca} channels and inhibits H₂S production by glomus cells in WT but not CSE^{-/-} mice (87, 136). Conversely, hypoxic hyperventilation is not affected in HO-2 null mice (131). H₂S appears to have more of a direct effect on BK_{Ca} channels, whereas CO appears modulatory (162, 163) and HO inhibitors such as zinc protoporphyrin IX directly bind to H₂S (33). Obviously, this field needs an additional study.

H₂S mediation of peripheral chemoreceptors, adrenal medulla

Chromaffin cells, containing both CBS and CSE, line the posterior cardinal vein and anterior kidney in trout and are homologous to adrenal medullary cells in mammals. They respond to hypoxia by a CBS-mediated release of H₂S and epinephrine into the systemic circulation (137). As with type I glomus cells, catecholamines are released by exogenous H₂S and this requires extracellular calcium, suggesting that H₂S is involved in chromaffin cell depolarization.

Mammalian adrenal medullary chromaffin cells are considered to provide an O₂-sensing role during neonatal development of the carotid bodies (66). CSE immunoreactivity has been found in rat and mice neonatal chromaffin cells and hypoxic stimulation of catecholamine secretion appears to be mediated in mice and rats by H₂S, likely through CSE (136, 195). Exogenous H₂S directly induces catecholamine release from adrenal cells cultured from rats by inhibiting BK_{Ca} channels.

Airway receptors

In spontaneously breathing chickens, low concentrations (0.2%) of inhaled H₂S stimulate ventilation, an effect that may be mediated in part by airway receptors (77, 172). Vagal and sensory neurons in mammalian airways may also be activated by H₂S (158).

Mechanical effects on airway smooth muscle

Hypoxia and H₂S relax tracheal and bronchiolar airway smooth muscle, although there is some species specificity regarding the efficacy of H₂S (20, 80). Hypoxic relaxation is accompanied by a decrease in intracellular calcium and may or may not be mediated by K_{ATP} channels (88, 177). H₂S-mediated relaxation of airways appears to be independent of K_{ATP} channels, prostaglandins, and NO (80). In an elegant study, Castro-Piedras and Perez-Zoghbi (18) showed that H₂S dilated small (~200 μM) airways in lung slices and they observed a considerably lower EC₅₀ (36 μM) than the 300 μM reported by Kubo *et al.* (80). These differences could be due to increased small airway sensitivity to H₂S and/or due to the closed conditions in the study by Castro-Piedras and Perez-Zoghbi (18), which prevented H₂S volatilization. Conversely, much higher, H₂S concentrations (EC₅₀=1.3 mM) produce a dose-dependent contraction in guinea pig main

bronchae and distal trachea that appears to be mediated by activation of vanilloid neurons (171). We (Olson, unpublished) have also observed H₂S relaxation of bovine bronchioles in the range reported by Kubo *et al.* (80).

Other Tissues

Hypoxia and H₂S also relax nonvascular smooth muscle of fish urinary bladder and the gastrointestinal tract, both of these stimuli produce a unique and transient increase in spontaneous contraction frequency and amplitude before the onset of the inhibitory effects (31, 32). H₂S also relaxes human corpus cavernosum and urinary bladder smooth muscle (24, 45). Addition of exogenous cysteine, which presumably enhances or sustains H₂S production, augments hypoxic relaxation of trout urinary bladder (31) and salmon intestine (32). Consistent with findings in other tissues, inhibitors of H₂S biosynthesis inhibit hypoxic relaxation of rainbow trout urinary bladder (31) and rainbow trout and Coho salmon intestine (32).

General Metabolism

In addition to cells and tissues which serve as homeostatic O₂ sensors, there is some evidence that H₂S-mediated O₂ sensing is involved in general cellular metabolism and function. As described earlier, cells can use H₂S as a substrate for ATP production and this can serve as an endogenous stimulator of cellular bioenergetics (49, 157). These possibilities are currently being explored, albeit with limited success thus far, in protecting tissue from ischemia/reperfusion injury, organ preservation before transplantation, protective metabolic depression during bypass surgery or after severe trauma associated with shock, sepsis, and acute lung injury.

H₂S can also produce metabolic depression (akin to “suspended animation”) in small mammals (9, 10), and it has been implicated in torpor and hibernation (159, 160), anapnoea (81), and hypoxia-induced radiation resistance (194). H₂S has also been shown to protect cells from both hypoxia- and hyperoxia-induced redox imbalance and apoptosis (13, 83, 85, 92, 101, 189, 190). The beneficial effects of H₂S in treating ischemia/reperfusion injury and in pre- and post-conditioning have been well documented, and it has been suggested that the beneficial effects of exogenous H₂S are due to mimicking transient hypoxia, which is also used in conditioning (128).

It is also evident that H₂S is involved in long-term O₂ sensing through its inhibitory effects on hypoxia inducible factor (HIF)-1 α expression (70, 148, 149, 188), although opposite effects have also been reported (17, 91, 93). In an elegant study, Ma *et al.* (93) identified a novel genetic pathway where an increase in H₂S produced by prolonged (24 h) hypoxia increases HIF-1 expression in *C. elegans*. They showed that H₂S promotes the interaction of CYSL-1 with EGL-9, thereby removing the inhibitory effect of EGL-9 on HIF-1. This mechanism is independent of EGL-9 hydroxylation of HIF-1 and does not utilize the von Hippel-Lindau pathway for HIF-1 degradation. Interestingly, they (93) also showed that CYSL-1 is in the family of CBS proteins, albeit with insignificant enzymatic activity, and they propose that a mammalian CYSL-1-like CBS protein is involved in protection from reperfusion injury.

Conclusions

O₂-dependent inactivation of H₂S is an effective, highly accurate, and tightly coupled mechanism that is used by a variety of O₂-sensing tissues and cells to detect O₂ availability and to coordinate supply and demand. A variety of observations support this hypothesis. H₂S metabolism appears ubiquitous, even predating oxidative phosphorylation. H₂S is produced by cells independent of O₂, whereas numerous O₂-dependent biochemical pathways affect its destruction and this occurs at physiologically relevant O₂ tensions. Protein cysteines, which are often key regulatory components of protein structure and enzymatic activity, readily react with H₂S and initiate downstream signaling processes. Vertebrate cardiovascular and respiratory systems, the primary effectors of O₂ supply, are replete with examples of H₂S-mediated O₂ sensing and their functions can be enhanced or inhibited by factors that promote or prevent H₂S metabolism. However, challenges remain. Additional information is needed on H₂S metabolism and interaction with sulfides (including H₂S itself), other putative O₂-sensing mechanisms, and effector processes. Better inhibitors of H₂S metabolism are needed as are more specifically targeted and controlled H₂S “donors.” Finally, this field needs to develop into the realm of therapeutic application, which, of course, is one of the goals of scientific investigation.

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Abbreviations Used

α -Kg = α -ketoglutarate
 ψ = mitochondrial transmembrane potential
 3-MST = 3-mercaptopyruvate sulfurtransferase
 AdoMet = S-adenosyl-L-methionine
 AMP = adenosine monophosphate
 AOA = aminooxyacetate
 AOX = alternative oxidase
 BK_{Ca} = large-conductance Ca²⁺-dependent K⁺ channels
 BPA = bovine pulmonary arteries
 CAT = cysteine aminotransferase
 CBS = cystathionine β -synthase
 CCO = cytochrome C oxidase
 CDO = cytosolic cysteine dioxygenase
c-fos = proto-oncogene
 CO = carbon monoxide
 CPA = conductance pulmonary arteries
 CSE = cystathionine γ -lyase
 CYP ω = cytochrome P450 monooxygenase ω
 Cyt aa3 = mitochondrial cytochromes a and a3
 DAO = D-amino acid oxidase
 DHLA = dihydrolipoic acid
 ETHE1 = mitochondrial sulfur dioxygenase
 FIH-1 = factor inhibiting hypoxia inducible factor-1 α
 GSH = reduced glutathione
 GSSG = oxidized glutathione
 H₂S = hydrogen sulfide
 HA = hydroxylamine
 HIF = hypoxia inducible factor
 HNO = nitroxyl
 HO-2 = hemoxygenase-2
 HS(O)NO = sulfinyl nitrite
 HSNO = thionitrous acid
 HSNO₂ = thionitrite
 HS[•] = thiyl radical

K_{ATP} = ATP-dependent potassium channels
 K_{Ca} = Ca²⁺-dependent K⁺ channels
 K_v = voltage-gated potassium channels
 LDA = lamprey dorsal aorta
 Mito = mitochondria
 mtHsp 70 = mitochondrial heat shock protein
 NADPH = nicotinamide adenine dinucleotide phosphate
 NE = norepinephrine
 NEB = neuroepithelial body
 NEC = neuroepithelial cell
 NO = nitric oxide
 NO₂⁻ = nitrite
 NOX2 = NADPH oxidase
 NZDA = New Zealand hagfish dorsal aorta
 O₂ = oxygen
 PASMCM = pulmonary arterial smooth muscle cells
 pB = pre-Bötzing
 PDA = Pacific hagfish dorsal aorta
 pFRG = parafacial respiratory group
 PHD = prolyl hydroxylase domain proteins on HIF-1 α
 PKC = protein kinase C
 P_{O2} = partial pressure of oxygen
 PPG = propargyl glycine
 Rde = rhodanase
 ROS = reactive oxygen species
 RPA = resistance pulmonary arteries
 SNA = sinus nerve activity
 SNP = sodium nitroprusside
 SO = sulfite oxidase
 SQR = sulfide:quinone oxidoreductase
 ST = sulfur transferase
 TASK = TWIK-related acid-sensitive potassium
 TR = thiosulfate reductase
 TRD = Trx reductase
 Trx = thioredoxin
 WT = wild-type