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Identification of Broad-Based HIV-1 Protease Inhibitors From Combinatorial Libraries

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Abstract

Clinically approved inhibitors of HIV-1 protease function via a competitive mechanism. A particular vulnerability of competitive inhibitors is their sensitivity to increases in substrate concentration, as may occur during virion assembly, budding and processing into a mature, infectious viral particle. Advances in chemical synthesis have led to the development of new chemical libraries with high diversity using rapid in-solution syntheses. These libraries have been previously shown to be effective at disrupting protein-protein and protein-nucleic acid interfaces. We have screened 44,000 compounds from such a library to identify inhibitors of HIV-1 protease. One compound was identified that inhibits wild type protease, as well as a drug-resistant protease with 6 mutations. Moreover, analysis of this compound suggests an allosteric, non-competitive mechanism of inhibition and may represent a starting point for an additional strategy for anti-retroviral therapy.

Keywords

enzyme; non-competitive inhibitors; kinetics; high-throughput screening

INTRODUCTION

HIV infection continues to be a worldwide health crisis, with over thirty-three million infected people worldwide [1]. Despite improvements in antiretroviral therapeutic development, drug resistance remains a major obstacle to effective control of infection in

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AUTHOR CONTRIBUTION

Rolf Muller, Jeremiah Savage, and Ying C. Lin screened the combinatorial chemical library for HIV-1 protease inhibitory activity. Ying C. Lin and Jeremiah Savage produced the protease for the biochemical assays and protease screening. Sukwon Hong, Wei Jin, and Landon R. Whitby synthesized, determined the composition and purity of the library as well as deconvoluted the library. Michael J. Giffin and Max W. Chang designed and performed all biochemical analyses on the selected protease inhibitors. Max W. Chang performed all docking studies. John H. Elder, Dale L. Boger, and Bruce E. Torbett were involved in the design and interpretation of the results. Michael J. Giffin, Max W. Chang, and Bruce E. Torbett were primarily involved in writing the manuscript. Bruce E. Torbett and Max W. Chang edited the final manuscript.

HIV-infected patients. Numerous advances and improvements have been made in drugs targeting the viral protease, required for maturation of virions into infectious particles [2]. However, common mutations associated with protease inhibitor drug resistance appear in drug-experienced patients, and in certain subtypes of the virus in drug-naïve patients as well [3], often leading to virologic failure and the onset of disease progression. Drug resistance complicates the use of therapeutics in the treatment of HIV infection, necessitating an ongoing search for novel therapeutics targeting the viral protease.

HIV-1 protease is a 22-kDa homodimeric aspartic protease consisting of two 99-residue polypeptide chains that self-assemble to form the enzymatically active dimer. Currently, all FDA-approved protease inhibitors (PIs) are in the same mechanistic class, i.e. competitive inhibitors that bind the active site of the protease, preventing the association of the protease with substrates and resulting in disruption of virion maturation [2]. One drawback of competitive inhibitors is that similar active site mutations can deleteriously affect small molecule binding in the active site, leading to increased risk of cross-resistance to other competitive inhibitors. An additional potential pitfall is that competitive inhibitors are sensitive to substrate concentrations [4]. An alternative to competitive inhibitors has been the identification of inhibitors that target non-active site regions of the protease, such as the dimer interface [5, 6], flaps [7, 8], or other non-substrate active site regions [9]. Moreover, inhibitors that utilize non-competitive, uncompetitive, or mix-mode mechanisms have also been identified [5–8, 10, 11]. A potential advantage of a non-competitive mechanism will be the insensitivity to substrate concentrations, which may better maintain a therapeutic threshold in the substrate-rich virion. The improved therapeutic threshold and potential insensitivity to current resistance mutations may be more efficacious at inhibiting viral replication and the emergence of drug-resistance.

To identify inhibitors that may target other features of the protease structure to inhibit its function, we screened a library of compounds previously shown to provide inhibitors of protein-protein and protein-nucleic acid interactions [12, 13]. One compound, compound 1, was found to inhibit wild type protease, from the NL4-3 strain of HIV-1, in the low micromolar range. Moreover, compound 1 also inhibited a multidrug resistant (MDR) protease containing 6 mutations associated with PI resistance [14]. The kinetics findings for wild type protease demonstrated a mechanism of inhibition consistent with non-competitive inhibition, and cross-competitive inhibition studies with compound 1 and Pepstatin A, a competitive inhibitor, implicated a non-active site binding affect. Taken together, these findings suggest that compound 1 functions as a non-competitive, allosteric protease inhibitor.

MATERIALS AND METHODS

Enzyme Activity Assays

HIV-1 protease enzymatic activity was assayed as described previously [15], using the fluorescently labeled anthranilyl protease substrate Abz-Thr-Ile-Nle-p-nitro-Phe-Gln-Arg-NH₂ (H-2992, Bachem, CA) [16]. In brief, bacterially purified HIV-1 protease was mixed with inhibitor compounds in a reaction buffer containing 25 mM MES, pH 5.6, 200 mM NaCl, 5% DMSO, 5% glycerol, 0.0002% Triton X-100, and 1 mM dithiothreitol in a prewarmed 96-well plate. All clones used for protease bacterial expression were generated from NL4-3 wild type or a multidrug resistant (MDR) protease containing 6 mutations (L24I/M46I/F53L/L63P/V77I/V82A), termed 6X, associated with resistance to saquinavir, nelfinavir, ritonavir, and TL3, as described [14]. The enzyme-substrate reaction was started by the addition of fluorescently labeled substrate and the reaction progress was measured by fluorescence intensity using an FLx-800 fluorescence plate reader (BioTek, VT). For IC₅₀

determinations, final reaction concentrations were 25 nM protease, 30 μ M substrate (the approximate K_m), and 0.001–600 μ M of inhibitor.

Chemical Library

The Boger laboratory has established and reported previously on a collection of chemical libraries [17] consisting of approximately 66,000 compounds prepared by using solution phase technology with liquid–liquid acid–base extraction purification, evaluated for composition and purity, and stored for further assessment [18, 19]. Figure 1 shows a representative diagram of chemical scaffolds and substitutions used to generate the library. From the original library, 44,000 compounds were evaluated in our study. The lead compounds 1 and 2 (Figure 2) identified from the screening were synthesized, then evaluated for composition and purity, below, before use [18, 19].

Compound 1. ^1H NMR (400 MHz, DMSO- d_6 , 25 $^\circ\text{C}$) δ 11.84 (s, 1H), 10.74 (s, 1H), 9.86 (s, 1H), 8.39 (d, J = 1.9, 1H), 8.14 (dd, J = 8.3, 1.6 Hz, 2H), 8.08 (s, 1H), 7.92 (d, J = 8.8, 1H), 7.83 (dd, J = 8.9, 1.9, 1H), 7.39 (s, 1H), 3.84 (s, 3H), 1.49 (s, 9H); MS-ESI (m/z) calcd for $[\text{C}_{24}\text{H}_{22}\text{N}_4\text{O}_7\text{S}_2+\text{Cl}]^-$ 577.1; found: 577.1.

Compound 2. ^1H NMR (400 MHz, DMSO- d_6 , 25 $^\circ\text{C}$) δ 11.81 (s, 1H), 10.85 (s, 1H), 8.41 (d, J = 1.8 Hz, 1H), 8.08 (dd, J = 15.1, 1.4 Hz, 2H), 8.03 (s, 1H), 7.94 (d, J = 8.6 Hz, 1H), 7.84 (dd, J = 8.7, 1.6 Hz, 1H), 7.69 (m, 2H), 7.55 (m, 1H); MS-ESI (m/z) calcd for $[\text{C}_{18}\text{H}_{12}\text{N}_4\text{O}_5\text{S}_2+\text{H}]^+$ 429.0; found: 429.0.

Michaelis-Menten Kinetics Measurements

For Michaelis-Menten kinetics measurements, protease substrate was titrated from 1 to 200 μ M. To assay for promiscuous inhibition, 0.001–0.01% Triton X-100 was added to the reaction. To assess inhibitor specificity horseradish peroxidase (Sigma) activity was assayed in 14 mM potassium phosphate, pH 6.0, 0.5% hydrogen peroxide with an enzyme concentration of 5 nM. The reaction was initiated with the addition of the substrate o-phenylenediamine at concentrations from 20 μ M to 100 μ M in 100 mM sodium phosphate and 50 mM sodium citrate, pH 5.0. Kinetics constants were determined by nonlinear regression of initial reaction velocities as a function of inhibitor concentration using Prism v5.0c (GraphPad Software, San Diego, CA). IC_{50} values were fit with the following equation:

$$Y = \frac{100}{1 + 10^{((\log \text{IC}_{50} + X) \times \text{HillSlope})}}$$

Michaelis-Menten kinetics constants were fit to the following equation:

$$Y = \frac{V_{\max} \times X}{K_m + X}$$

K_i constants for compounds 1 and 2 were fit to the following equation:

$$V_{\max i} = \frac{V_{\max}}{1 + \frac{I}{K_i}}$$

Cross-competitive Inhibitor Measurements

A variation of Yonetani and Theorell analysis was used to evaluate the binding mode of compound 1 [20, 21]. The use of the variation of Yonetani and Theorell analysis, as discussed by Martinez-Irujo et al [21], takes into account the binding interactions on an enzyme of competitive and non-competitive inhibitors. The cross-competitive inhibitor assessment was accomplished by varying the concentration of Pepstatin A (Roche), a competitive inhibitor, at a fixed concentration of compound 1, a non-competitive inhibitor, while keeping the substrate and protease concentrations constant. The experimental conditions for assessing protease function were identical to those used to determine IC_{50} . Pepstatin A was used at concentrations from 0.6 to 3.0 μ M, while compound 1 was held constant at 45, 30, 20, or 0 μ M. The determination of the interaction term, γ , that defines the degree to which binding of one inhibitor influences the binding of the second inhibitor, was determined utilizing Prism v5.0c (Graphpad Software, San Diego, CA) using the following equation [21]:

$$\frac{v_0}{v_{1,2}} = 1 + \frac{[I_1]}{(IC_{50})_1} + \frac{[I_2]}{(IC_{50})_2} + \frac{[I_1][I_2]}{\gamma(IC_{50})_1(IC_{50})_2}$$

Docking Studies of Compound 1

The docked conformation of compound 1 with HIV protease was generated using AutoDock Vina 1.02 [22]. The high resolution HIV-1 protease structure 2HS1 was chosen as the receptor. Two overlapping search spaces were used, each measuring $25 \times 32 \times 40$ Å, which together spanned chain A of this structure. The darunavir molecule bound in the active site was preserved. In each docking run, 9 conformations were reported, only the most favorable is detailed below. Three-dimensional coordinates for the ligand were determined using Corina [23]. Other docking parameters were kept to their default values.

RESULTS AND DISCUSSION

We have screened a library of compounds previously shown to inhibit protein-protein and protein-nucleic acid interactions [12, 13, 17, 24]. A chemically diverse library of 44,000 compounds was synthesized using a solution phase combinatorial synthesis as previously described [13, 18, 19, 25]. To facilitate synthesis and screening, some compounds were synthesized as part of a screened mixture, with some mixtures containing up to 10 related but distinct compounds. A representative group of compounds from which the lead compounds emerged is shown in Figure 1.

Compounds were screened initially for the ability to inhibit the wild type HIV-1 protease, obtained from the NL4-3 virus, in a real-time kinetics assay using a fluorogenic substrate at a concentration equal to the K_m . Compounds that had significant effects on the baseline fluorescent signal, which was designated as 10% above baseline independent of the substrate peptide, were excluded from further screening. Assay conditions were chosen to reduce the possibilities of false positives resulting from promiscuous inhibitors, including minimizing compound aggregate formation by the inclusion of detergent and reducing compound incubation time [26, 27]. Compound groups that showed greater than 50% inhibition of wild type protease at a compound concentration of 20 μ M were then tested against 6X protease, a multidrug resistant (MDR) protease containing 6 mutations (L24I/M46I/F53L/L63P/V77I/V82A) associated with resistance to saquinavir, nelfinavir, ritonavir, and TL3, identified here as 6X [14].

Compound families showing greater than 50% inhibition against both wild type and 6X proteases were then deconvoluted and synthesized as individual compounds. These compounds were then tested individually against wild type and MDR 6X proteases. Individual compounds again showing greater than 50% inhibition at a concentration of 20 μM were selected for more detailed kinetics analyses. One compound, compound 1 (Figure 2), was found to inhibit wild type protease in the low micromolar range (Figure 3). Furthermore, compound 1 also showed low micromolar inhibition of the 6X protease. The half maximal inhibitory concentrations, IC_{50} , were determined against wild type and 6X proteases and found to be 17 μM against wild type protease and 11 μM against the 6X protease (Figure 3). Thus, compound 1 is effective in inhibiting both wild type and a MDR 6X protease at a similar IC_{50} .

To address whether compound 1 was a general enzymatic inhibitor we evaluated whether the compound altered horseradish peroxidase function at various concentrations. No measurable effect on the Michaelis-Menten kinetics of the reaction was observed at any of the compound concentrations evaluated, suggesting that compound 1 is not a general enzymatic inhibitor (data not shown). Furthermore, the inhibitory activity of compound 1 on HIV-1 protease was not abrogated by the addition of non-ionic detergents further strengthening the case that compound 1 is not a promiscuous inhibitor (data not shown) [26, 27].

In order to identify the minimal chemical moieties necessary for PI activity, we synthesized a derivative library based on compound 1 and screened each fragment against wild type protease independently. While the majority of the derivatives showed significantly reduced inhibition of protease activity with IC_{50} s ranging from 20 to greater than 1000 μM , one derivative, compound 2 (Figure 2), showed more potent inhibitory activity. Compound 2 is similar to compound 1, but with the Boc and methyl groups removed. When the inhibitory activity of compound 2 was compared to compound 1, it showed a slight decrease in both the IC_{50} and K_i values as determined with the fluorogenic protease substrate assay (Figures 3 and 4). Therefore, both compounds were active on wild type protease and compound 1 demonstrated activity against the MDR 6X protease mutant.

We next determined the effect of compounds 1 and 2 on the K_m and V_{max} of wild type protease with reactions performed within a range of substrate concentrations from 1 to 200 μM , centered around the K_m of 30 μM , at several inhibitor concentrations, from 0 to 30 μM . The values for the initial velocities were then fit to a Michaelis-Menten model using nonlinear regression to determine the dose-dependent effects of the compounds on the K_m and V_{max} for HIV protease, as shown in Figure 4, panel A, compound 1, and panel B, compound 2. When we measured protease activity as a function of both substrate concentration and inhibitor concentration, and used nonlinear regression to fit the resulting initial velocities to a Michaelis-Menten model, we observed a curvilinear response in V_{max} as a function of increasing concentration of both compounds (Figure 4, panel C). The results are consistent with a non-competitive mechanism of inhibition.

To glean further insights into the underlying molecular process of protease inhibition by compound 1, we utilized a variation of Yonetani and Theorell analysis [20, 21] to evaluate the binding mode. Since the inhibition by compound 1 is consistent with a non-competitive mechanism, which might predict that substrate may still bind to the protease active site when compound 1 is bound, Pepstatin A was used as a cross-competitive inhibitor for analysis. This method of inhibitor cross-competitive analysis allows determination of the degree to which the binding of compound 1 to protease influences the binding of the second inhibitor to the active site [4, 20, 28]. The choice of Pepstatin A was based on the ability to inhibit protease [29], the well established biochemical and structural reports of its binding location

in the active site [30], that it has a competitive inhibition mechanism [31, 32], and the reported use of Acetyl-pepstatin A for inhibitor cross-competitive studies for non-active site inhibitors [5]. The graphical findings from a representative cross-competitive study utilizing compound 1 and Pepstatin A is shown Figure 5. In each case, non-parallel lines were obtained which converged at the x-axis, which is consistent with the interpretation that compound 1 and Pepstatin A may bind independent sites [4, 20, 21]. The interaction term, γ , that defines the degree to which the binding of one inhibitor to the enzyme influences the binding of the second inhibitor can be determined through interpolation of the x-intercept or calculated [4, 20, 21]. A small γ value (<1) signifies a synergistic interaction between the inhibitors, whereas a large γ value (>1) indicates mutual antagonism, and in the case that $\gamma = 1$, the inhibitors bind to the enzyme in an independent manner. Calculation of γ yielded approximately 1, consistent with compound 1 binding to a protease site independent of Pepstatin binding in the active site. These findings are consistent with compound 1 binding and providing inhibition through a site independent of the active site.

Given our findings from the inhibitor cross-competitive study indicating that compound 1 was not binding in the active site, we investigated whether compound 1 functions as a dimerization inhibitor. A number of compounds have been reported to promote inhibition through disruption of protease dimerization [5, 6]. To address whether compound 1 disrupts dimerization we utilized a tethered homodimeric protease, formed by a direct repeat of protease monomers linked by a 5 amino acid sequence [33]. The IC_{50} of compound 1 was found to be similar for both the non-covalent wild type protease dimer and the covalently tethered dimer protease, as shown in Figure 6. Since compound 1 was active against the protease-tethered dimer, this implies that dimerization disruption is not required for inhibitory activity.

As compound 1 was shown to have a distinct binding location from Pepstatin A and not promote dimer interface disruption, possible binding modes were explored using molecular docking. Focusing the search on the outside surface of the protein, a low-energy conformation was discovered which placed compound 1 in a long, solvent exposed cleft, termed the exo site [34], as shown in Figure 7. The exo site is composed of distinct regions that include the elbow, cantilever, and fulcrum components of the protease. Molecular dynamic simulations of protease flap movement relative to the exo site has indicated that the exo site is compressed when the flaps are open and is extended when the flaps are closed [34]. Moreover, the exo site has been shown, via a fragment-based screen, to accommodate small molecules [35]. The predicted binding energy from the compound 1 docking simulation (-7.2 kcal/mol) corresponds to a K_i of 5.2 μ M, very close to the experimentally observed K_i value of 6.1 μ M. Together with the biochemical findings presented herein, the docked compound 1 conformation supports a plausible allosteric binding mechanism which is consistent with structural data [35]. It is tempting to speculate that binding of compound 1 to the exo site influences flap dynamics, perhaps by locking the flaps closed and rendering the protease unable to bind substrate. A number of recent reports have implicated novel compounds that disrupt flap movement, thereby altering enzymatic function [36, 37].

Currently, all approved protease inhibitors are competitive inhibitors, which target the active site. Given the rise in protease inhibitor resistant HIVs, new inhibitors with novel inhibitory mechanisms are needed. Non-active site, allosteric inhibitors may avoid mediating selective pressure associated with active site inhibitors, which result in drug resistance mutations. The identification of compound 1 from a novel library of diverse compounds was found to inhibit both wild type and a multidrug resistant protease, through a non-competitive, allosteric mechanism. This compound provides a rationale starting point from which to chemically investigate novel inhibitory mechanisms that may provide another avenue of viral suppression.

Abbreviations used

FDA	Food and Drug Administration
MDR	multidrug resistant
MS-ESI	mass spectrometry–electrospray ionization
PI	protease inhibitor
6X	HIV protease containing L24I/M46I/F53L/L63P/V77I/V82A mutations

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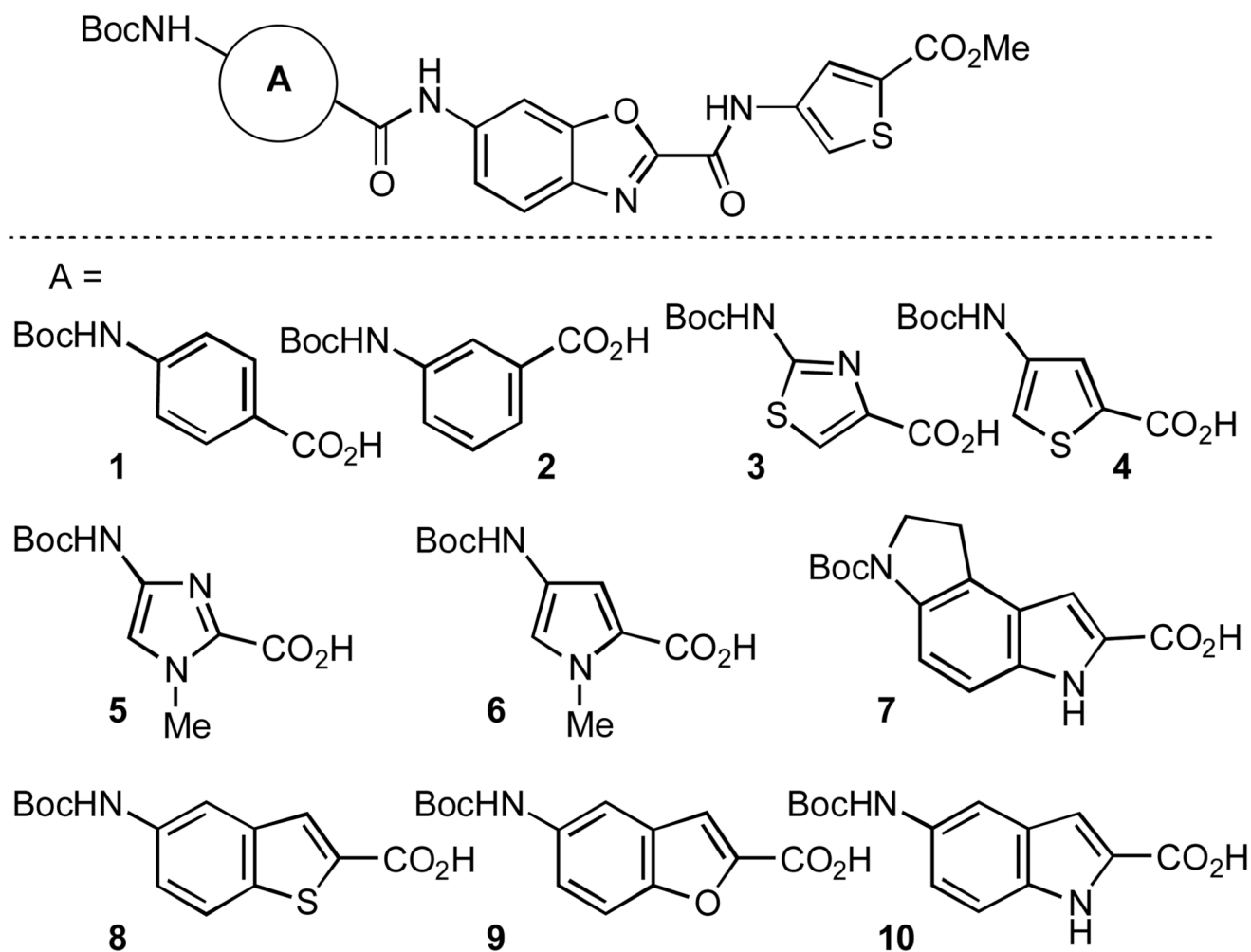


Figure 1. Representative group of chemical substituents that are used in the reaction in the context of the compound scaffold

Each substituent is found at circle site labeled A, in this case generating 10 different compounds. For additional information see Materials and Methods and [18, 19].

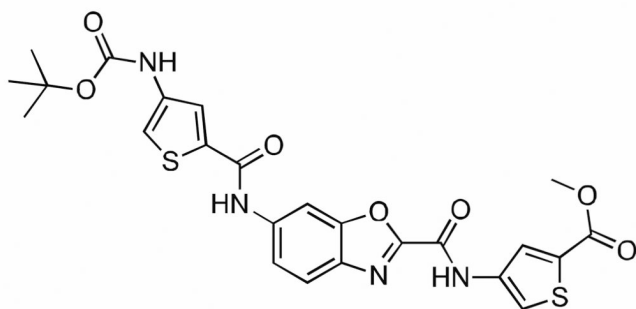
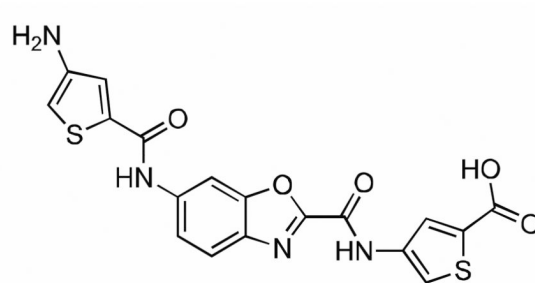
A**B**

Figure 2. Compounds used in this study, (A) compound 1 and (B) compound 2

Compound 1 was identified through protease – substrate screens of the original chemical library, see text, whereas compound 2 is a derivative of compound 1 with the Boc and methyl groups removed from the ends of the compound.

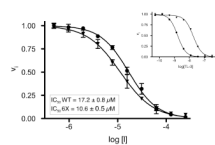


Figure 3. IC₅₀ titration of compound 1 against wild type and the 6X multi-drug resistant proteases

Foreground, evaluation of compound 1, log [I], against wild type (●) and 6X multi-drug resistant (▼) proteases. Inset: For comparison, titration of TL-3, log[TL-3], a protease inhibitor which is effective against the wild type (●) protease, but not the multi-drug resistant 6X protease mutant [14] (▼), is shown. Results from nonlinear regression indicate that the IC₅₀s are within a factor of 2 of each other for wild type and multi-drug resistant 6X protease mutant. IC₅₀ curve fitting was performed as described in the Materials and Methods. ± values indicate the standard error.

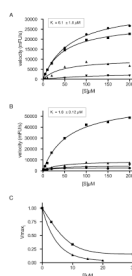


Figure 4. Michaelis-Menten kinetics of compound 1 and 2 against wild type protease
 (A) Compound 1 was used at 0 (●), 3 (■), 10 (▲), and 30 (▼) μM and (B) compound 2 was used as 0 (■), 10 (▲), 15 (▼), and 20 (●) μM over a range of μM substrate concentrations [S] for determination of the Michaelis-Menten kinetics. This is a representative result from 1 of 3 experiments. (C) Nonlinear regression of V_{max} as a function of compound 1 (■) or 2 (▲) concentration. Shown is a representative experiment and standard errors for individual points varied less than 10% of the mean and curve fitting is described in the Materials and Methods. \pm values indicate the standard error.

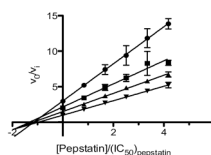


Figure 5. Yonetani and Theorell plot of v/v_i versus concentration of Pepstatin A and compound 1 0 μ M of compound 1 (\blacktriangledown); 20 μ M of compound 1 (\blacktriangle); 30 μ M of compound 1 (\blacksquare); 45 μ M of compound 1 (\bullet) with varying concentrations of Pepstatin A. Each point was in triplicate and this is a representative result from 1 of 4 experiments. Assay conditions and curve fitting described in Materials and Methods.

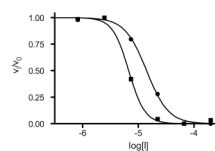


Figure 6. IC_{50} titrations of compound 1 against wild type and tethered dimer proteases
Compound 1 [I] demonstrates similar inhibitory efficacy against wild type (●) and the tethered protease dimer (■). Standard errors for individual points varied less than 10% of the mean and shown is a representative experiment. Assay conditions described in the Materials and Methods

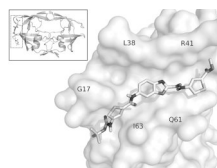


Figure 7. Docked conformation showing compound 1 bound outside of the HIV-1 protease (2HS1) active site

A space-filling rendering of the exo site showing the location of the solvent exposed cleft and binding of compound 1. The exo site is a feature of the protease altered by movement of the flaps. Insert of protease shows the area that is magnified. The predicted binding energy of this conformation was -7.2 kcal/mol, equivalent to a K_i of 5.2 μ M.