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Immobilization of Antibodies and Enzymes on 3-Aminopropyltriethoxysilane-Functionalized Bioanalytical Platforms for Biosensors and Diagnostics

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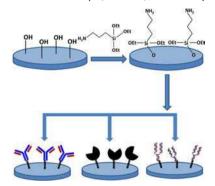
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1. INTRODUCTION

Biosensing and diagnostic platforms with high sensitivity, specificity, and fast response time are based on immobilized biomolecules such as antibodies (Abs), aptamers, enzymes, nucleic acids, receptors, and whole cells for the detection of target analytes. Such sensing biomolecules should be bound to the surface of a signal transducer with a required specific chemical, electrical, or optical property. The biological recognition event generates a quantifiable signal, which is equated to the amount or concentration of the analyte. Thus, the biomolecule immobilization plays a crucial role in achieving high sensitivity and selectivity with prolonged device lifetime. The substrate materials such as electrodes, microtiter plates (MTPs), nanoparticles (NPs), nanotubes, graphene, etc., for biomolecule immobilization must be modified to introduce functional groups that bind to biomolecules with high bonding strength, high activity, good orientation, and excellent long-term stability.

Chemisorbed monolayers of silane compounds, a class of selfassembled monolayers (SAM), on glass or oxidized silicon surfaces with active silanol (Si-OH) groups were first reported by Sagiv.¹ For biosensing applications, silane SAMs with terminal

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amino, epoxy, or chloro groups are necessary for attaching biomolecules. Such silane derivatives also provide a means to tailor adhesion, biocompatibility, charge, hydrophilicity, and hydrophobicity of a surface. 3-Aminopropyltriethoxysilane (APTES) has been used extensively for the functionalization of bioanalytical platforms as its role in surface modification has been intensively investigated and well understood.² Table 1 shows

Table 1. Properties of APTES

IAPAC name	1-propanamine, 3-(triethoxysilyl)-
synonyms	(γ-aminopropyl)triethoxysilane, (3-aminopropyl) triethoxysilane, 1-propanamine, 3-(triethoxysilyl)-; 3- (triethoxysilyl)-1-propanamine; 3-(triethoxysilyl) propylamine; triethoxy(3-aminopropyl)silane; A-1100; A- 1112; AGM 9 (VAN); NUCA 1100; Silane 1100; Silicone A- 1100 and UC-A 1100
molecular formula	$C_9H_{23}NO_3Si$
appearance	colorless liquid
molar mass	$221.37 \text{ g mol}^{-1}$
density	0.946 g mL ⁻¹ at 25 °C
melting point	-70 °C
boiling point	217 °C
vapor pressure	0.02 hPa @ 20 °C
water solubility	$7.6 \times 10^{5} \text{ mg L}^{-1}$ @ 25 °C
hazard	toxic; the target organs of APTES are nerves, liver, and kidney

some general properties of APTES. The APTES-amino groups also facilitate the formation of siloxane bonds with surface Si-OH,^{3,4} which obviates the need of a curing step after surface modification. One of the classical examples is the reaction of the Si-OH on a glass/silica surface using APTES. Silicon and silicon derivatives, metals, and polymers are commonly used to serve as a solid support in bioanalytical platforms. In some cases, they are modified with NPs, nanotubes, graphene, quantum dots (QDs), etc., to form nanocomposites. The amino-functionalized silane then forms covalent bonds with the amino (glutaraldehyde (GLD)-mediated reaction) or carboxyl groups (1-ethyl-3-[3-(dimethylamino)propyl]-carbodiimide (EDC)-mediated crosslinking) of biomolecules. However, it is somewhat difficult to control this immobilization step as the protein's surface is rich in hydrophilic amino acids, that is, Lys, Glu, and Asp. The use of thiol-based immobilization strategies enables the site-specific binding of biomolecules. Yet it is not efficient as biomolecules have a limited number of cysteine residues on their surface. Hence, these immobilization strategies offer more site-specific reactions, but the protein density is low and a thiol silane, for example, 3-mercaptopropyltrimethoxysilane, should be used to enable the formation of a disulfide bond between the biomolecule and thiol silane.

Nevertheless, silanized platforms with APTES offer biomolecule stability, high immobilization density, reduced biofouling, leach-proof biomolecule binding, and enhanced analytical performance. APTES serves as superglue to anchor the target biomolecule on a solid surface and also acts as a spacer, allowing more steric freedom to the biomolecule during the immobilization step for higher specific activity. These platforms are advocated for the fabrication of commercial glucose biosensors and other biosensors for detecting important molecules, biomarkers, pathogens, and environmental contaminants. APTES has also been widely used in different assay platforms: enzyme-linked immunosorbent assays (ELISA), biosensors and immunosensors (IMS) based on surface plasmon resonance (SPR)-based assays, microgravimetric assays using microcantilevers and piezoelectric quartz crystals, surface acoustic wavebased assays, and nanomaterial-based assays. During the last two decades, different strategies have been developed for the immobilization of Abs and enzymes, involving covalent, oriented, covalent-oriented, site-specific, affinity tags, and the peptide nucleic acids (PNA)- and deoxyribonucleic acid (DNA)-directed ones for grafting organosilanes on silica and metal oxides.⁵ In biosensor technology, a direct electron transfer between the enzyme and its underlying electrode is of considerable interest toward the development of simple "reagentless" biosensors or biodevices with remarkably improved detection sensitivity and selectivity.

This work provides a critical review of the immobilization of Abs and enzymes on APTES-functionalized bioanalytical platforms. The important role of immobilization strategies will be discussed along with their impact on sensitivity, detection limit, operational and storage stability, and reusability toward the development of potential and commercial biosensing devices.

2. PREPARATION OF APTES-FUNCTIONALIZED BIOANALYTICAL PLATFORMS

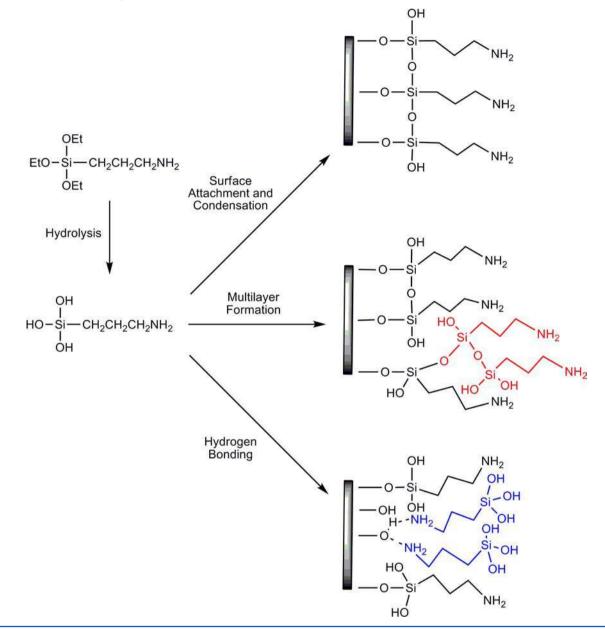
2.1. General Aspects of Silanization

APTES can be deposited on solid materials, electrode materials, nanomaterials, and nanocomposites under variable conditions of concentration, solvent, temperature, and time. In addition, curing conditions such as air/heat drying might be necessary depending upon the intended application. Pertinent information on the thickness, morphology, and conformation of the APTES layer reported in the literature is often different and conflicting.

Silanization occurs on any substrate, normally with chemically active hydroxyl groups for silane grafting.⁶ Some substrates (e.g., silica, agarose, etc.) already have hydroxyl groups, while others require a pretreatment step using KOH/NaOH, acid, piranha solution, or plasma treatment to introduce hydroxyl groups. For instance, silanization with APTES for controlling the interactions between inert stainless steel for the preparation of working or counter electrodes and biocomponents generally involves a pretreatment step that may involve heat, acid etching, and/or chemical treatment.⁷ Phosphoric acid chemisorption has been demonstrated for silanization of TiO₂ films.⁸ APTES assembling is enhanced via phosphoric acid preadsorption on the anatase TiO₂ film as compared to pristine TiO₂ with no phosphoric acid treatment.

In general, APTES reacts with the free hydroxyls of an oxidized substrate by $S_N 2$ exchange with loss of ethanol. APTES bonds to a substrate in three different ways (Scheme 1). Surface attachment and condensation lead to horizontal polymerization when a surface-bound APTES molecule forms siloxane with its neighboring surface-bound APTES.^{6,9} In vertical polymerization, the surface-bound APTES reacts with a nearby APTES in solution. $^{10}\ \mathrm{In}$ the third scenario, where the substrate possesses surface hydroxyl groups, the amine of APTES forms a hydrogen bond with the metal surface or becomes protonated by abstracting protons from the surface.¹¹ Therefore, a mixture of neutral amine and protonated and hydrogen-bonded amines coexists on the solid surface; the former orients away from the surface for biomolecule immobilization, whereas the latter is close to the surface. The amine group is transformed with various curing conditions and substrate-dependent. For the substrate

Scheme 1. Different Bonding Modes for APTES on a Substrate Surface



without oxide groups, amines react quickly with carbon dioxide of the atmosphere to form bicarbonates and then imines in the presence of heating.

The silanization mechanism is complex with adsorption and chemical sorption contributing to the rate-determining steps. Initially, silanization is fast and governed by the Langmuir model leading to a SAM followed by a slow kinetic saturation process.^{12,13} Silanization is classified as wet chemistry or chemical vapor deposition (CVD). The former is performed in liquid medium and classified as organic, aqueous, or evaporation-based, depending upon the choice of solvents. Silanization is initiated via the hydrolysis of ethoxy terminal groups of APTES by the moisture (organic phase and evaporation-based approaches) or by water (aqueous phase). The CVD approach is based upon the deposition of silane vapor on a substrate surface.

Another popular and commercially available silane, (3glycidoxypropyl)-trimethoxysilane (GOPS), is also frequently used for covalent bonding of biomolecules. Epoxy groups are very stable in neutral pH, and under aqueous reaction conditions, are reactive toward nucleophiles such as amines, thiols, or acids. Immobilization of biomolecules to an epoxy-modified substrate occurs via a two-step procedure: adsorption followed by covalent reaction between amino acid side groups (amino, thiol, or hydroxyl) and epoxy groups. However, high ionic strength is necessary to promote biomolecule adsorption using GOPS, which could denature proteins/fragile enzymes,¹⁴ limiting its widespread applications.

2.2. Wet Chemistry

2.2.1. Organic Silanization. An anhydrous solvent with a negligible trace amount of water is required in the preparation for controlling the degree of polymerization of APTES at the surface and in solution. APTES concentrations above 10% result in the formation of oligomers and polymers, whereas 0.4% APTES allows for monolayer coverage.¹⁵ Silanization at 60–90 °C disrupts noncovalent interactions including hydrogen bonding to minimize the density of silane molecules that are weakly adhered to the surface. Extensive rinsing after silanization with toluene,

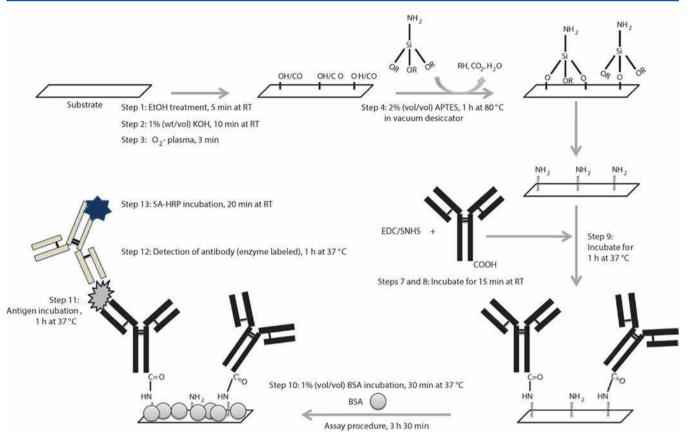


Figure 1. Treatment of a substrate with KOH and O_2 plasma prior to APTES functionalization as part of an APTES-based sandwich ELISA procedure. Reproduced with permission from ref 60. Copyright 2011 Nature America Inc.

ethanol, or even water removes such weakly bonded silane molecules and hydrolyzes residual alkoxy linkages in the silanized layer. The drying—curing step, if performed at 110 °C, promotes the formation of siloxane linkages and converts ammonium ions to more reactive, neutral amines. However, the amine groups are dissociated at higher temperature and are completely eliminated at 150 °C.

To date, toluene-diluted APTES is widely used for the silanization (25 and 75 °C) of different materials ranging from plastics to metals.^{16–19} The grafting efficiency of different APTES concentrations, prepared in anhydrous toluene under nitrogen atmosphere on plasma-activated glass slides, has been demonstrated for the enhancement of cellular-adhesion.⁶ A monolayer of APTES is expected at room temperature with toluene as the reaction solvent (15 min at room temperature followed by air curing or 15 min at 200 °C). The more forcing conditions, such as increasing reaction and curing time, end up with the formation of multilayers.¹⁵

The ultraviolet/zone-activated glass is silanized by incubating with 1% APTES in anhydrous toluene at 60 °C.²⁰ Metallic NPs, however, require an additional step to generate an oxide layer on the particle surface. For example, Pb–Sn hybrid nanostructures are heated to generate an oxide layer for subsequent salinization with APTES (10% in toluene for 1 h).²¹ Similarly, TiO₂ surfaces are silanized using APTES reconstituted in toluene.²² Fe₃O₄ NPs are silanized by treatment with APTES reconstituted in toluene and DMF followed by incubation for 24 h with continuous stirring.²³ The silanization procedure has been extended to flat silicon surfaces,²⁴ mesoporous silica nanofibers,²⁵ monodispersed silica nanospheres,^{26,27} Ag nanotriangles,²⁸ attapulgite,²⁹

agarose,³⁰ zirconium,³¹ multiwalled carbon nanotubes (MWCNTs),³² and quartz fibers.³³

Acetone is another popular solvent to silanize carbon nanotubes (CNT, 30 min followed by drying at 60 $^{\circ}$ C),³⁴ silicon dioxide (5% APTES in acetone),³⁵ Pt electrodes (0.2% APTES in acetone, 12 h at 120 °C),³⁶ glassy carbon electrodes (6.2.014 FLS in acetone, 12 h at 120 °C),³⁶ glassy carbon electrodes (GCE, 10% APTES in acetone for 5 h),³⁷ and optical grating coupler sensor chips.³⁸ Benzene,³⁹ anhydrous hexane,^{40,41} dimethyl sulfoxide (DMSO)/ethanol,⁴² ethanol (under reflux),^{43–48} chloroform,^{49,50} methanol,⁵¹ and glacial acetic acid/methanol⁵² are also useful for silanization. The APTES surface density is governed by solvent viscosity, which in turn affects the number of siloxane bridges formed. Hydrocarbons like isooctane and pentane are better solvents as they result in significantly higher APTES surface density. Pentane allows for greater mobility of the surface-bound silane due to its low viscosity, enabling the formation of more siloxane bonds. In contrast, the surface density of APTES molecules is not optimal when aromatic solvents such as toluene are used.53 Therefore, the choice of organic solvent plays a critical step in silanization that can significantly affect a specific bioanalytical application.

APTES monolayer films prepared in a dry organic solvent must be cured under mild conditions and incubated in water before use. In contrast, thick films need to be treated in an oven or incubated in ambient conditions for a few days. APTES silanization of iron oxide NPs using 0.2% APTES in a mixture of methanol/toluene is not suitable at 30 °C. The highest APTES functionalization density is achieved at 70 °C using 2% of APTES.⁵⁴ If silanization is conducted at a lower reaction temperature and a lower silane concentration, it takes more than a day to get a silanized surface with a lower APTES functionalization density. Only a minute amount of polymerized APTES is formed if silanization with APTES (10 mM) is conducted in dry toluene under reflux at 70 $^\circ$ C for 24 h.⁵⁵

2.2.2. Aqueous Silanization. Water plays a dual role as a solvent and catalyst for the hydrolysis of silane, a prerequisite for silane grafting. In aqueous solution, APTES is hydrolytically unstable as the ethoxy groups are easily hydrolyzed to produce ethanol and trisilanol groups. At neutral pH, the half-life of APTES is 56 h (at 10 °C) and 8.4 h (at 24 °C), respectively.⁵⁶ However, the Si-C bond will not further hydrolyze, and the aminopropyl group remains intact. The transient Si-OH groups also condense with other available Si-OH groups. In aqueous solution, the amine groups of APTES are positively charged.⁵⁷ Yet if the substrate is negatively charged, for example, silicon oxide,58 APTES will electrostatically bind to the substrate surface. For silanization in water or ethanol, the APTES molecules are in the form of a monolayer even after a day. Thick APTES layers can be obtained after deposition from water using high APTES concentrations that could be attributed to different orientations of APTES at the surface.⁵⁹

Different polymeric substrates can be pretreated with 1% (w/v) KOH and oxygen plasma (e.g., using a reactive ion etcher) to generate the desired hydroxyl groups for silanization. Subsequently, these resulting polymers are incubated with 2% (v/v) APTES in deionized water for 1 h (Figure 1).⁶⁰ This simple treatment obviates the use of complicated oxygen plasma for silanization.⁶¹ Recently, this procedure has been proven in SPR-based immunoassays (IAs) using Au-coated SPR chips.^{62,63} The Au SPR chip is simply treated with a piranha solution for 2 min to generate the desired surface hydroxyl groups and incubated in 2% (v/v) APTES in water for 1 h. A typical mixture of the piranha solution is 3:1 (v/v) concentrated sulfuric acid to 30% H₂O₂ solution.

Of interest is aqueous silanization of an aluminized plastic surface by dip-coating with APTES:water (1:5) at pH 7 for 30 min, followed by incubation at 100 °C,⁶⁴ whereas polydime-thylsiloxane (PDMS) and plastic are salinized with 5% (v/v) APTES.⁶⁵ SnO₂ surfaces can also be activated and coated with silanes prepared in deionized water.⁶⁶ For a diblock copolymer functionalized surface, an M5 4-(2-hydroxyethyl)-1-piperazinee-thanesulfonic acid (HEPES)-based buffer solution serves as the solvent with APTES simply drop-coated on the surface.⁶⁷ The effect of silanization on poly(methyl methacrylate) (PMMA)– alumina bonding of 3-methacryloxypropyl-trimethoxysilane, APTES, and *N*-(2-aminoethyl)-APTES, reconstituted in water, has been discussed elsewhere.⁶⁸

Aqueous silanization has also been advocated for various nanoparticulate surfaces due to its simplicity. Magnetic NPs are silanized using a water-doped organic solvent (~2% water)⁶⁹ or 10% (v/v) APTES in water.⁷⁰ Similarly, Si⁷¹ and Au⁷² NPs are incubated using various concentrations of APTES in ethanol:water (95:5) or only in deionized water.

The kinetics of silanization is described as a function of time and solvent. APTES (26.5 mM aqueous solution) is ideal for homogeneous coating.⁷³ The APTES adsorption and desorption kinetics for Si can be observed by solid/liquid ellipsometry. At 0.4% APTES, APTES first adsorbs to the silicon¹⁵ by noncovalent means and then reacts covalently with the surface.⁷⁴ Under mild curing, only the close one or two APTES monolayers on the silicon are stable, whereas the remaining APTES layers are weakly bound and thus removed. For APTES silanization of Fe₃O₄ NPs, the initial binding depends on reaction temperature and APTES concentration.⁵⁴ However, the former is more important than the latter for silanization. For instance, the initial grafting density (within 1 h at 30 °C) is 64 mg g⁻¹ at 0.2% of APTES loading as compared to 224 mg g⁻¹ at 2% APTES loading. At 70 °C with 0.2% APTES loading, the APTES functionalization density is 236 mg g⁻¹. The overall reaction kinetics is complex involving both adsorption and chemical sorption.

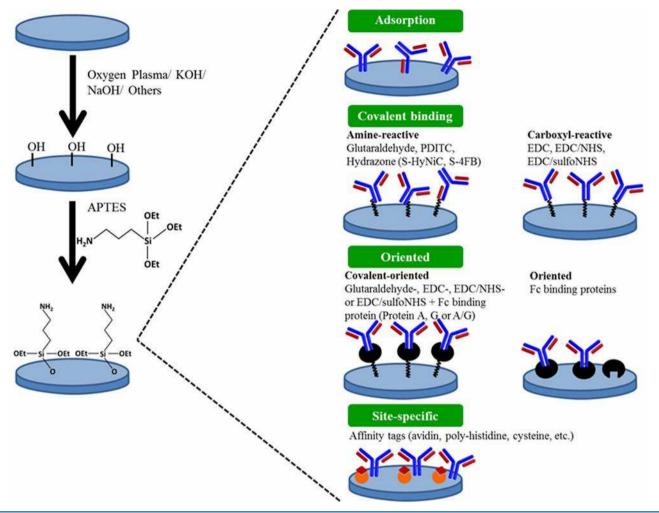
2.3. Chemical Vapor Deposition

CVD is applicable for mass production of silanized substrates. It obviates the limitations of other approaches such as the presence of trace water in solution and the purity of silane. An activated substrate is simply placed in an environment saturated with silane vapor, leading to substrate silanization. As the silane polymers have negligible vapor pressure at the reaction temperatures, they do not get deposited on the target, thereby resulting in smooth monolayers. However, the attachment mechanism of silane molecules to the substrate is uncertain. The APTES layer thickness on silica is 5 and 6 Å at the deposition temperature of 70 and 90 °C, respectively, corresponding to the length of an APTES molecule, illustrating the formation of a monolayer on the substrate.⁷⁵

There are two types of CVD procedures, concentrated and dilute, which are dependent upon the APTES vapor density. As an example, the APTES film thickness on a piranha-activated and oven-dried glass surface is directly correlated to the vapor density.⁷⁶ Silanization in the vapor phase using a vacuum plasma deposition epitaxial system is pioneered by Jonsson and Olofsson.⁷⁷ There is higher heterogeneity in the APTES film formed at high-density vapor, leading to the development of several CVD procedures. Nevertheless, a simplified CVD setup is more practical where APTES is incubated along with the substrate at high temperatures inside vacuum desiccators, resulting in a cheaper alternative known as low vapor density CVD.^{78–80} The vapor deposition system offers stringent control over the vapor density, thickness, and homogeneity of the silane layer. Krumdieck and Raj⁸¹ described one of the first mechanical low-pressure metalorganic CVD (MOCVD) systems for surface functionalization,^{82,83} enabling the use of MOCVD for silanization.⁸⁴ Olofsson's mechanical CVD can be improved further by using novel mechanical components.⁸⁵ Consequently, plasma enhanced-CVD (PECVD) has been adopted as a potential CVD technique for surface functionalization of pointof-care (POC) diagnostic platforms.⁸⁶

2.4. Characterization of APTES-Functionalized Surface

The morphology and thickness of APTES layers are dependent on the conditions of deposition and curing, and it is of great interest to characterize the structure and chemistry of APTES at interfaces. For APTES freshly deposited on an unreactive, Fourier transform infrared (FTIR)-featureless Au surface, the FTIR signature of APTES has been well described in the literature.¹⁵ In brief, the peaks at 1132 and 1194 cm⁻¹ are characteristics of Si-OCH₂CH₃ groups, while a low frequency shoulder on the peak at 1132 cm⁻¹ indicates partial polymerization or Si-O-Si bond formation. Curing conditions after deposition also affect the FTIR signature, as exemplified by a new peak at 1034 cm⁻¹ that shows the formation of siloxane bonds when APTES on Au is exposed to air for 24 h.⁸⁷ However, a strong peak at 1126 cm⁻¹ indicates the presence of an enormous Si--OCH₂CH₃ on the Au surface. Three prominent peaks (1475, 1578, and 3300 cm^{-1}) reveal the existence of an amine bicarbonate salt formed by interaction with carbon dioxide and water vapor in the atmosphere.⁸⁸ The APTES adsorption on Scheme 2. Schematic of Most Commonly Used Ab Immobilization Strategies on APTES-Functionalized Bioanalytical Platforms



silicon dioxide with some covalent attachment or cross-linking has been reported for the deposition of APTES on silicon dioxide in dry solvents.¹⁵

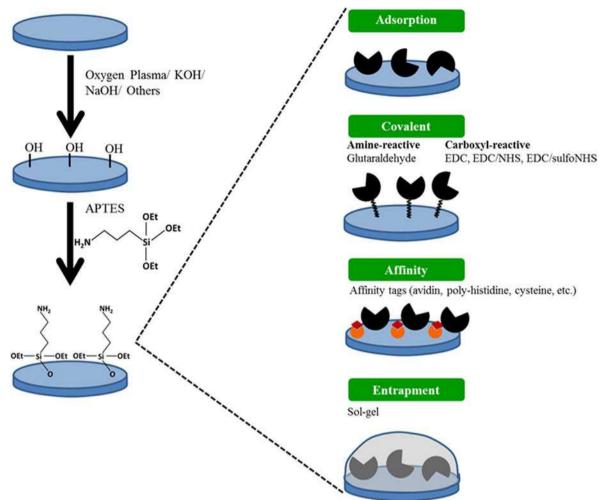
A mixture of protonated and neutral amine groups coexists on silicon dioxide or aluminum oxide when APTES is freshly deposited.⁸⁹ Protonated and hydrogen-bonded APTES amines are close to the silicon dioxide surface and interact with it via electrostatic and hydrogen bonds, while free amines and carbon are away from the surface. Longer reaction times result in multilayers followed by the formation of macroscopic aggregates of APTES. Irregular shapes of APTES samples with a dense, weblike structure should be expected from toluene deposition under reflux conditions.

Several different techniques can be used for probing the surface roughness before and after silanization and bioconjugation. Such techniques also enable the estimation of protein immobilization density and morphology. The analysis of atomic force microscopy (AFM) height histograms provides useful information about average height and roughness of an organic layer.⁹⁰ The average height of an APTES layer on Si₃N₄ is 0.80 \pm 0.02 nm, in agreement with Kim et al. (0.8–1.3 nm)⁹¹ with a surface roughness of 0.23 \pm 0.01 nm. Noncontact AFM is often used for probing the 3-D structure of bound biomolecules. Although the specific biomolecular binding can be quantified, AFM does not provide pertinent information related to biomolecule surface density. Such information can be obtained

by other techniques such as SPR, ellipsometry, or isotope-labeling. $^{92-95}$

The protein/amino-organosilane/interface structure can be confirmed by X-ray photoelectron spectroscopy (XPS), whereas homogeneous protein binding is probed by a near-field scanning optical microscope in reflection and fluorescence mode. A surface analytical method such as XPS can be used to identify and quantify the presence of an immobilized biomolecule film from the increase of the signals in the N 1s, C 1s, and O 1s regions. The thickness of protein coatings is measured by angle-resolved XPS (ARXPS) and evaluated according to Paynter et al.⁹⁶ The photoelectrons from amine (NH2, 399.8 eV) are taken as characteristic for the APTES film, while protonated amino groups $(NH_3^+, 401.4 \text{ eV})$ of the N 1s core-level show the protein layer. Surface coverage with proteins can be determined with ARXPS^{97,98} or microgravimetry.⁹⁸ Time-of-flight secondary ion mass spectrometry (ToF-SIMS) is another technique to analzye immobilized biomolecules.^{99,100} The disulfide bond signal in ToF-SIMS is directly related to the activity of adsorbed protein coatings. A pertinent model with combined information from AFM, ARXPS, and ToF-SIMS can provide for insight into the vertical structure with respect to the surface density of biomolecules.

Contact angle measurement is extremely sensitive to contamination; therefore, purified liquids and very clean solid surfaces are required to attain good reliability and reproducibility Scheme 3. Schematic of Most Commonly Used Enzyme Immobilization Strategies on APTES-Functionalized Bioanalytical Platforms



to within a few degrees. With a contact angle of less than 90°, the solid surface is considered hydrophilic, and, beyond this value, the solid surface is considered hydrophobic. In general, the silicon surface modified with APTES is hydrophobic as exemplified by a contact angle of ~50° obtained for silicon modified with APTES-GLD.¹⁰¹

There is considerable use of APTES and related organosilanes in the synthesis of bonded silica chromatographic stationary phases, ¹⁰² a subject that is beyond the scope of this Review. FTIR and solid-state NMR spectroscopy are useful for probing bonding methodology, surface coverage, and silanol conversion efficiencies with supercritical-CO₂ as reaction solvent.¹⁰³

3. ANTIBODY-ENZYME IMMOBILIZATION

Various enzyme/Ab immobilization strategies, based on physical adsorption, covalent binding, affinity, entrapment, and other chemistries, will be described along with their advantages and limitations (Schemes 2 and 3).

Tables 2 and 3 summarize pertinent applications that use APTES for the immobilization of Abs and enzymes, respectively.

3.1. Covalent

The amine group of APTES enables the covalent bonding of biomolecules based on the use of a functional linker. GLD, a homobifunctional agent, cross-links the amino groups of APTES-functionalized surfaces to the biomolecule amino groups.³⁶

Aldehyde groups form imine bonds with amine groups of lysine common to almost every protein or enzyme to form reversible Schiff bases. If necessary, the carbon—nitrogen double bond can be reduced by a reducing agent to produce stable secondary amine linkages. In aqueous solution, GLD might possess different structures, which in turn affect immobilization reactivity. A dimer GLD form enables fast protein immobilization as compared to its monomer counterpart.¹⁰⁴ In a low ionic medium, immobilization proceeds via ionic exchange with the amino groups on the support followed by the covalent attachment. In a high ionic medium, the biomolecule is directly immobilized by covalent bonding with slower immobilization kinetics.

The binding of Abs by their amino groups might lead to improper orientation of Abs by masking antigen-binding sites at the amino terminal. This shortcoming can be circumvented by using heterobifunctional cross-linkers and fragment crystallizable (F_c)-binding proteins (FBP). The most widely used heterobifunctional cross-linker combination is EDC and *N*-hydroxysuccinimide (NHS)/*N*-hydroxysulfosuccinimide (SNHS), where EDC of the EDC-NHS/SNHS complex first binds to the carboxyl-terminal of the Abs (see Figure 1).⁶⁰ The zerolength cross-linker EDC-NHS/SNHS leaves the activated surface after an amide bond between NH₂-APTES and the COOH-Ab is formed. Commercial manufacturers often

Table 2. APTES-Based Antibody Immobilization Strategies and Their Applications in Biosensors and Diagnostics

description of APTES-based antibody immobilization strategy	biochip substrate	application	re
glass – APTES – GLD – coating Abs (anti-MYO, anti-CK-MB, anti-TrI, anti-FABP)	glass capillaries	detection of human myocardial proteins by chemiluminescence IA	16
SU-8 photoresist – cerium ammonium nitrate treatment – APTES – GLD – anti-CRP Ab	SU-8 photoresist	immobilization of IgGs on SU-8 photoresist surfaces	17
silicon nitride – APTES – GLD – Abs against bovine PrP	resonator (silicon nitride deposited on silicon wafer)	detection of PrP using nanomechanical resonator arrays and secondary mass labeling	18
field effect transistors (FET) – APTES – GLD – antialpha fetoprotein	FET	detection of tumor marker antialpha fetoprotein using stable Ab-modified FET	19
glass – APTES – Au NPs – Ag – IgG	microscope glass slide	Au NP-based silver-enhanced IAs	20
quartz optical fibers – APTES – carbonyldiimidazole–polyclonal Ab against Newcastle disease virus (NDV) – NDV	quartz optical fibers	detection of NDV	33
Pt – APTES – GLD – anti-MA Ab	Pt	development of IMS to determined MA	36
GCE – APTES – Au NPs – <i>Staphylococcal</i> Protein A – anti-GA IgG	GCE	detection of GA in the rice grain using SWASV technique-based IMS	37
OGC sensor chip – APTES – GLD – anti-Salmonella Ab	OGC sensor chip	detection of Salmonella with a direct-binding OGC IMS	38
IrO _x electrodes – APTES – GLD – anti-CEA/anti-AFP	IrO _x electrodes	simultaneous electrochemical immunosensing of CEA and AFP	41
microchips on glass slides – APTES – GLD – goat IgG	microchips (with dual-ring working and counter electrodes, and a sensing cavity chamber) on glass slides	development of electrochemical enzyme IA microdevice using dual ring electrodes with a sensing cavity chamber	43
$LiTaO_2 - SiO_2 - APTES - GLD - HBsAg$	LiTaO ₂	a Love-mode SAW IMS was employed for the real-time detection of HBsAbs in whole blood samples	44
silicon – APTES – GLD – avidin – Ab GP3 (mouse monoclonal biotin- labeled anti-M13)	Si wafer	detection of phage M13K07 using optical protein-chip	45
silicon – APTES – GLD – anti-human cTnT Ab	SiNW	detection of cTnT using complementary metal—oxide semiconductor (CMOS) SiNW sensor arrays	46
nanoporous Al – APTES – GLD – anti-SEB Ab	nanoporous Al	detection of SEB by time-resolved EIS	24
silica – APTES – GLD – goat antihuman IgG Ab	silica monolithic capillary	detection of human IgG in human serum using immunoaffinity monolithic capillary microextraction coupled MCN-ICP-MS	47
nterdigitated electrodes precoated with sol—gel derived BST film — preinactivated dengue virus — APTES — GLD — antidengue virus Ab	BST-coated interdigitated electrodes	detection of dengue infection using sol—gel derived BST thin film and interdigitated electrodes	48
Nb/NbOxHy – APTES – GLD – k47 Ab	Nb/NbOxHy	detection of atrazine	50
COC – APTES –PDITC – goat antihuman IgG	COC (Zeonor and Zeonex)	Ab immobilization on COC for sandwich IA	51
COC – APTES – toluene-2,4-diisocyanate – goat antihuman IgG			
COC – APTES – succinimidyl-4-(N-maleimidomethyl)cyclohexane-1- carboxylate – goat antihuman IgG			
glass microcapillary – APTES – TNT analogue, trinitrobenzenesulfonic acid – CyS-labeled anti-TNT Ab	borosilicate glass microcapillary	detection of trace levels of TNT in seawater using microcapillary reversed- displacement IMS	52
Si NPs – APTES – EDC-sulfoNHS activated mouse IgG	dye-doped Si NPs	elucidation of the surface amine-silanization reaction profile on fluorescent Si NPs	71
Si – APTES – GLD – human IgG	Si slides	covalent immobilization of proteins for imaging ellipsometry-based biosensor	10
PS and Au – APTES – EDC, EDC/SNHS and EDC/NHS – Ab	PS	comparison of various EDC-based Ab immobilization strategies on APTES- functionalized immunodiagnostic platforms	10
microring resonator sensor chips (made from silicon) – APTES – anti(human thrombin)	microring resonator sensor chips	attachment of Abs to biosensor surfaces for label-free IA	10
silicon – APTES – GLD – IgG	monocrystalline silicon wafer	preparation of IMS bioselective matrix	10
glass – APTES (5%) – GLD – Ab	glass	preparation of uniform Ab-bound glass	10
PS – GNPs in APTES – EDC-activated Ab	PS	development of graphene-based IA for LCN2	17
polystyrene – APTES – cross-linker – (N-(γ- maleimidobutyryloxy] succinimide ester – IgG	polystyrene	development of modified polystyrene surface for ELISA development	177

description of APTES-based antibody immobilization strategy	biochip substrate	application	ref
mesoporous Si NPs - APTES - GLD - anti-CEA Ab	Si NPs	development of ultrasensitive chemoluminescence IMS using HRP-labeled mesoporous Si NPs	180
polystyrene – APTES – 4,4-bis(4-hydroxyphenyl) valeric acid-BSA conjugate	polystyrene	development of IA for hapten-protein conjugates on modified polystyrene surface of microtiter plate	181
glass – Au – Si NPs/APTES/EDC/NHS – Protein A – antifibrinogen Ab	glass	detection of fibrinogen by LSPR	186
Au SPR chip – KOH – APTES – anti-HFA	Au SPR chip	SPR-based IA biosensor was designed for the detection of HFA	187
glass – APTES – GLD – goat antihuman IgG/goat antirabbit IgG	glass slides	development of an optical IMS based on SELS	188
silicon microring – 2% APTES – GLD – anti-IgE or antithrombin aptamers	Si microring	SOI format to detect human IgE and human thrombin	190
Si microring – APTES – succinimidyl hydrazinonicotinamide – succimidyl 4- formylbenzoate – anti-CEA	Si microrings	development of a Si microring resonator for label-free detection of antigens	191
glass – APTES – photoresist – biomolecule	glass	development of APTES-functionalized nanoporous polymer gratings	192
guided-mode resonance filter (silicon nitride deposited on adhesive polymer resin-functionalized glass) – APTES – GLD – Abs against cardiac markers	guided-mode resonance filter biosensor chips	detection of cardiac markers in human serum using a GMR biosensor	193
fiber optics – APTES – GLD – BPA-BSA	fiber optics	an AFOB for the detection of BPA leaching risk using fluorescent-labeled anti- BPA monoclonal Abs	194
silica capillary – APTES – GLD Abs for E. coli O157:H7	silica capillary	the detection of E. coli O157:H7 by monitoring the side reaction of p-nitrophenol	195
titania sol-gel films - APTES - GLD - Abs against <i>E. coli</i> O157:H7	titania	an optical biosensor based on FTIR spectroscopy for the detection of pathogenic $E.\ coli$	197
OGC sensor chip - APTES - GLD anti-Salmonella Abs	OGC sensor chip	the detection of <i>Salmonella</i> using a direct binding OGC IMS based on optical waveguide light mode spectroscopy	38
glass – APTES – streptavidin – biotinylated antimouse IgG-Fc specific/ biotinylated mono- or poly clonal anti-PSA Abs	glass	development of an integrated microfluidic processor for automated microfluidic IAs based on solid phase capture of Abs	198
glass – APTES – EDC/NHS – IgG	glass	development of multiplexed protein microarrays in a single microfluidic channel	201
glass – APTES – streptavidin-coated beads – biotinylated capture Ab	glass substrate	development of on-chip IA using electrostatic self-assembly of streptavidin-coated beads on silanized glass substrates	209
PZT glass cantilever – APTES – EDC – sulfoNHS activated anti- <i>E. coli</i> O157: H7 Ab	PZT cantilever	detection of E. coli O157:H7 using PZT cantilever sensor	215
PZT cantilever – APTES – EDC-sulfoNHS – Ab	PZT cantilever	the label-free detection of femtogram levels of <i>E. coli</i> and other microbes was demonstrated in flowing liquid samples using PZT microcantilever-based IMS	216
PZT cantilever – APTES – EDC-sulfoNHS activated anti-BA	PZT cantilever	detection of BA spores	217
Pt electrode – APTES – Nafion – BSA solution containing GLD and GOx – perfluorocarbon polymer	Pt	development of long-term lifetime amperometric glucose sensor	221
ITO – APTES – GLD – Ab	ITO	development of a disposable IMS for the detection of HER-3 cancer biomarker	231
ITO electrode – APTES – Au NPs – VEGF Ab fragment	ITO electrode	electrochemical detection of VEGFs using VEGF Ab fragments modified Au NPs/ ITO electrode	232
Au PCB – APTES – 1,4-penylene diisothiocyanate – antiaflatoxin M1 Ab	Au	development of electrochemical immunochip sensor for aflatoxin detection	242
sol-gel derived BST thin film – APTES – GLD – antihuman IgG	BST thin film	detection of human IgG using sol-gel derived BST thin film	247
CdS-G-agarose hybrid – APTES – GLD – AFP	CdS-G-agarose hybrid	electroluminescence IMS for ultrasensitive detection of AFP	253
MEG – APTES – EDC/NHS-activated Ab	MEG grown on silicon carbide	development of a MEG-based IMS for hCG	254
GCE – Nafion-cysteine-graphene sheets – Au NPs – anti-GRP	GCE	detection of ProGRP using glucose oxidation as a means to detect other biomolecules	260
stainless steel – APTES – Au NPs – monoclonal Ab	stainless steel	development of impedimetric IMS for doxorubicin	262
Fe_3O_4 NPs - APTES - GLD - IgG	Fe ₃ O ₄ NPs	immobilization of Abs on Fe ₃ O ₄ NPs	263
Fe ₃ O ₄ NPs - APTES - GLD - fluorescein-labeled antimouse IgG	Fe ₃ O ₄ NPs	development of Fe ₃ O ₄ NPs for biorecognition	264
Au – Fe ₃ O ₄ /CdSe–CdS NP/polyelectrolyte nanocomposite – Au NPs – anti- CEA Ab	Au	ECL detection of CEA using Fe ₃ O ₄ /QDs	268
Fe ₃ O ₄ NPs – APTES – GLD – monoclonal Abs	Fe ₃ O ₄ NPs	detection of Salmonella using personal glucose meters	270

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description of APTES-based antibody immobilization strategy	biochip substrate	application	ref
$Fe_3O_4 NPs - APTES - GLD - Ab$	Fe ₃ O ₄ NPs	development of Fe ₃ O ₄ NPs-based ELISA to detect multiresidue organophosphorous pesticides	271
polycarbonate	polycarbonate	development of self-assembled nanoparticulate film for covalent attachment of Abs to plastic	272
Si NPs – APTES – carboxyl-functionalized Au-coated glass	Si NPs	development of a local SPR-based nanochip	273
SiNWs/NPs - APTES - EDC - monoclonal Ab CD54	SiNWs and NPs	comparison of Ab-functionalized SiNWs and Si NPs by measuring their specific binding with THP-1 monocyte cells using flow cytometry	275
silicon – APTES – GLD – goat antihuman IgG	Si	detection of proteins by heterostructured ZnO/Au NPs-based resonant Raman scattering	283
glass – APTES – anti-HSA Ab	glass	detection of HSA using biotinylated polyclonal anti-HSA Ab bound to streptavidin-conjugated CdSe/ZnS QD	284
Fe ₃ O ₄ NPs – APTES – Au – Ab of CA125	Fe ₃ O ₄ NPs	ECL sensor using QDs for the detection of carbohydrate antigen 125	285
ITO – graphene – Fe ₃ O ₄ NPs – Ab for PSA	ITO	ECL sensor using QDs for the detection of PSA	286
Si – APTES – anti-CEA monoclonal Ab/anti-CEA labeled with Alexa 647	Si	patterning of biomolecules on silicon-based analytical devices	288
epoxy-functionalized mPDGs – APTES/octadecyltrichlorosilane – monoclonal Ab	mPDGs	development of mPDGs for the fluorescent detection of human IL-10	289
quartz – APTES (10%) – GLD – Ab	quartz fibers	detection of T2 mycotoxin	295
silicon – APTES (CVD) – GLD – Ab	silicon	characterization of the biosensor surface	296
N-vinyl-2-pyrrolidone/allyl alcohol modified polyethylene – APTES – high affinity heparin oligosaccharides	N-vinyl-2-pyrrolidone/allyl alcohol modified polyethylene	immobilization of heparin oligosaccharides and determining its anticoagulant activity	297
Au – APTES – (GLD – albumin/hexamethyelenediamine) – 2,4- dichlorophenoxyacetic acid	Au	characterization of monoclonal Abs to 2,4-dichlorophenoxyacetic acid	298
piezoelectric quartz crystal – APTES – GLD – protamine	piezoelectric quartz crystal AT-cut	determination of IgM	299
piezoelectric quartz crystal – tresyl chloride – protamine			
piezoelectric quartz crystal – APTES – terapathaldicarboxaldehyde – Protein G – IgG	piezoelectric quartz crystal	functionalization of piezoelectric crystal with APTES for immunobiosensing	300
Au-coated glass chip – APTES – GLD – anti-HSA/anti-atrazine Ab – HSA/ atrazine	Au-coated glass chip	fabrication of APTES-functionalized SPR chip for immunobiosensing	301
silicon – APTES – oxidized anti-HBsAg IgG – HBsAg	silicon wafer	AFM studies of immobilized Abs	302
silicon – APTES – oxidized rabbit anti-HBsAg IgG (aldehydes generated on the carbohydrate side chains at the C-terminal of IgG using sodium periodate- oxidized reactions) – HBsAg	silicon wafer	site-directed immobilization of IgG on APTES-functionalized silicon	303
Au-coated piezoelectric quartz crystal – APTES – GLD – antistaphylococcal enterotoxin C $_2$ Ab – staphylococcal enterotoxin C $_2$	Au-coated piezoelectric quartz crystal	detection of staphylococcal enterotoxin C ₂	304
glass wool fibers – APTES – ${\rm NaIO_4}$ oxidized goat antibiotin Ab – biotinylated ALP	glass wool fibers	development of more efficient and effective glass-base immunosupports	305
fused silica fiber – APTES – GLD – theophylline Ab – theophylline	fused silica fiber	development of immunoaffinity solid-phase microextraction coating for biological sample analysis	306
optical silica fiber	optical silica fiber	comparing various antigen coupling methods for optical fiber-based IMSs	307
nicroporous silicon membranes – APTES – PDITC – Cy5-labeled goat antihuman IgG	microporous Si membranes	immobilization of Ab on silicon membranes containing arrays of micropores	308
quartz crystals – APTES – GLD – cymbidium mosaic protexvirus/ odontoglossum ringspot tobamovirus Ab	circular AT-cut 10 MHz Quartz crystal microbalances with Au electrodes	detection of cymbidium mosaic protexvirus and odontoglossum ringspot tobamovirus	309
polished Si wafer – APTES – GLD-linked chitosan – polyclonal Ab	polished Si wafer	investigation of blood protein adsorption onto chitosan	310
Si microchip – APTES – GLD – antiatrazine Ab	Si microchip	development of microfluidic enzyme IA	311
glass – APTES – NaIO ₄ oxidized Abs	glass slides	development of Ab microarrays	312
Si – mixed silanes (APTES and methyltriethoxysilane) – GLD – human IgG	Si wafer	immobilization of proteins on silicon surface modified with mixed silanes layer for imaging ellipsometry-based biosensor	313

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description of APTES-based antibody immobilization strategy	biochip substrate	application	ref
glass – APTES – GLD – antimouse IgG peroxidase conjugates	glass	polymer grafting on glass for protein chip applications	314
glass – APTES – agarose gel – $NaIO_4$ – GLD – IgG	agarose gel on glass slides	enzyme IA on microarray that was prepared on the basis of the oxidation of agarose gel	315
glass – APTES – GLD – IgG ethyl parathion Abs	glass beads	regeneration of Abs for repeated use in IMS	316
silicon nitride – APTES – aldehyde – human transferrin/6x His-HrpW – primary Ab – secondary Ab	silicon nitride	development of silicon nitride surface modification strategy for developing biosensors	317
silica – APTES – FITC – anti-HBsAg (mouse monoclonal IgG1)	silica sphere	preparation of amine-functionalized FITC-doped Si NPs	318
Au-coated Si – APTES – EDC-Neutravidin – biotinylated human IgG	Au-coated Si chip	quantification of human IgG immobilized on Au-coated Si chip	319
glass – APTES – GLD – goat antihuman IgG	glass slide	development of signal enhanced QCM-based IA using colloidal Au immunocomplex	320
silicon – APTES – NaIO ₄ oxidized IgG	Si wafer	immobilization of IgG on APTES-functionalized Si wafer surfaces	321
glass – APTES – chicken IgG	glass cover slides	immobilization of proteins on the modified glass surfaces	322
glass – APTES – GLD – rabbit IgG	glass slides	optimization of printing buffer for protein microarrays	323
silicon – APTES – GLD – human IgG	silicon wafer	detection of ultrathin biological films by ellipsometry	324
piezoelectric cantilever – APTES – EDC-sulfoNHS-Ab complex	piezoelectric cantilever (PZT layer bonded to borosilicate glass or fused silica layer)	detection of biologics in flowing liquid samples	216
silica – APTES	silica	monitoring of siloxane monolayer formation on silica using a fiber Bragg grating	325
silicon – APTES – GLD – antihuman IgG	Si wafer	reduction of morphological anomalies in protein arrays using non-nucleophilic additives	326
glass – APTES – GLD – antimouse IgG-FITC labeled	glass slide	development of protein chips using APTES	327
SiO ₂ – APTES – GLD – anti-turboGFP Ab	SiO ₂ substrates	precise micro-/nanoscale patterning of functional biomolecules	328
glass – APTES – rabbit γ-globulins	glass capillaries	development of capillary waveguide fluoroIMS	329
SiO ₂ – APTES – anti-M13K07 IgG1 FITC-labeled	SiO_2 surface of differential capacitive biosensor	Ab immobilization on ${\rm SiO}_2$ and polyester surfaces for biomedical applications by light-guided surface engineering	330
PDMS - APTES - GLD - antiglutathione S-transferase Ab	PDMS column	development of a miniaturized, polymer-supported immunoprecipitation method for the on-chip purification of proteins from complex mixtures	331
$\mathrm{SiO_2}-\mathrm{APTES}-\mathrm{biotin}\mathrm{\cdot NHS}-\mathrm{streptavidin}-\mathrm{biotinylated}$ antirabbit IgG	Au/SiO ₂ template substrates	creation of patterned biofunctional surfaces using chemically modified ${\rm Au}/{\rm SiO}_2$ template substrates	332
waveguide glass – APTES – cholic acid/cysteine/cyanuric chloride complex – hapten – 6HEX labeled anticholic acid Ab	waveguide glass	detection of protease activity using the Ab–hapten complexes attached to waveguide glass	333
porous silica – APTES – GLD – FITC-labeled rabbit IgG	porous silica	microcontact printing of proteins on nanoporous silica for POC immunoassays	334
interdigitated microelectrode ${\rm SiO}_2$ sensor chip – APTES – GLD – anti-Japanese encephalitis virus (JEV) Abs	interdigitated microelectrode sensor chip	comparison of different methods to immobilize anti-JEV Abs on interdigitated microelectrode sensor for electrochemical detection	335
sensor chip – APTES – anti-JEV Abs			
sensor chip – APTES – GLD – Protein A – anti-JEV Abs			
quartz glass – APTES – benzophenone – UV light – protein (IgG and BSA)	Si	photoinduced protein immobilization on quartz glass slides using benzophenone	336
silicon nanofiber - APTES - EDC/NHS - anti-CA15.3	Si	Debye screening manipulation for quantitative spatial analysis of induced charge on a nanosensor surface	337
Cr - APTES - GLD - Ab/Ag	Cr	reusable chromium-coated QCM for immunosensing	338
$TiO_2 - APTES - GLD - protein$	TiO ₂	development of highly sensitive microarray	339
silicon nitride – APTES – bis(sulfosuccinimidyl)suberate – antiglutathione-S- transferase Ab	silicon nitride	biofunctionalization of cantilever for adhesion studies	340
glass – APTES – GLD – antiglyphosate Ab	glass	detection of glyphosate by UV spectroscopy and fluorescence	341
Fe ₃ O ₄ NPs – poly(allylamine hydrochloride) – CdSe-CdS QD – APTES – Au – Au NPs – Ab	Fe ₃ O ₄ NPs	magnetic electroluminescence using magnetic NPs for efficient immunosensing of a cancer biomarker	342
interdigitated sensor - APTES - GLD - Protein A - anti-JEV	interdigitated sensor	label-free detection of viral antigens	343

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description of APTES-based antibody immobilization strategy	biochip substrate	application	ref
SiNW – APTES – GLD – anti-CRP Ab	SiNW	development of SiNW array for CRP detection in serum	274
Au printed circuit board – APTES – GLD – anti-CRP Ab	Au printed circuit board	detection of CRP using glucosamine-Cu cubes	344
SiNW – APTES – poly(sodium 4-styrenesulfonate) – poly(allylamine hydrochloride) – antiavian influenza Ab	SiNW	development of a dual-gate FET	345
titania – APTES – GLD – anti-PSA Ab	titania	to improve the protein immobilization efficiency of titania	346
Zeonor – APTES – dextran – NaIO ₄ – anti-CRP Ab	Zeonor	development of a polymer chip with silane-dextran chemistry for POC diagnostics	347
polymer – APTES – EDC/SNHS – Ab	polymeric solid supports (PS), polycarbonate, poly (methyl methacrylate), cyclic olefin polymers (i.e., Zeonor and Zeonex)	development of rapid and highly sensitive chemiluminescent IA for human fetuin A	348
Zeonor – APTES – Ab	Zeonor	characterizing the biomolecular adsorption in polymeric microfluidic channels	349
fused silica capillary – hexadecyltrimethylammonium bromide, water, and TEOS and APTES mixed with ethanol – GLD – monoclonal Ab	fused silica capillary	characterization of a highly selective immunosorbent that is coupled online to nanoLC-UV was presented and used for the selective extraction of microcystin- LR from blue-green algae extracts	350
silicon surfaces – APTES – GLD – Ab	Si surfaces	evaluation of surface free energies of Ab-bound surfaces, where Abs were bound to APTES-functionalized Si surfaces	351
photonic crystal – APTES – PDITC – Protein A – IgG Ab	nanoporous hydrogel photonic crystals	development of a prospective label-free detection system for IgG Ab	352
PDMS – APTES – GLD – Ab	PDMS chip	comparison of two acoustic wave biosensors for IA applications	353
TiO ₂ nanotube arrays-based electrodes – APTES – EDC/NHS – Ab	TiO ₂ nanotube arrays-based electrodes	development of a signal-amplified electrochemical IMS	354
CMOS chip – APTES – GLD – Ab	CMOS chips	development of a capacitive biosensor array on a CMOS chip using magnetic microbeads and on-chip electromagnetic manipulation	355
fiber optic sensor – APTES – GLD – captured BPA – BSA	fiber optic sensor	development of optofluidics-based IMS for BPA	194
photonic crystal — $\rm O_2$ plasma — APTES in ethanol—carboxymethyldextran — EDC/NHS — Ab	photonic crystal (titania and silica layers)	development of a label-free photonic crystal biosensor for the detection of cTnI	356
ITO-coated glass plate – 3-mercapto propionic acid capped Au NP – BSA	ITO-coated glass plate	liquid crystal-based biosensor for BSA	357
PS MTP – Ab diluted in 1% APTES (1:1, v/v)	PS	development of very highly sensitive sandwich ELISA procedure for IVD	178
polystyrene MTP – KOH – EDC-activated Ab diluted in 1% APTES (1:1, v/v)	PS	development of rapid sandwich ELISA procedure to detect HFA in 30 min	179
ITO substrate – H_2O_2 +NH ₄ OH – APTES – GLD – anti-RBP Ab	ІТО	development of a disposable biosensor for the very highly sensitive detection of RBP with attogram sensitivity	358

Table 3. APTES-Based Enzyme Immobilization Strategies and Their Applications in Biosensors and Diagnostics

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APTES-based enzyme immobilization strategy employed	biochip substrate	application	
ass microchannel – APTES – poly(maleic anhydride- <i>alt-a</i> -olefin)/GLD – FITC-labeled HRP	glass microchannel	development of a method to incorporate enzymes into microfluidic system channels	35
CE – GLD-cross-linked GOx – APTES	GCE	development of a rapid and highly simplified procedure for reagentless glucose electrochemical biosensor	11
CE – graphene/MWCNTs (in APTES) – EDC-activated GOx	GCE	comparison of graphene versus MWCNTs for electrochemical glucose biosensing	11
mposite sol–gel glass – APTES + 2-(3,4-epoxycyclohexyl)-ethyltrimethoxysilane – GOx	composite sol-gel glass	detection of glucose using encapsulated GOx within sol-gel glass	17
: electrode–permselective hybrid sol–gel film (APTES + TEOS + triethoxy-1 <i>H</i> ,1 <i>H</i> ,2 <i>H</i> ,2 <i>H</i> - tridecafluoro- <i>n</i> -octylsilane + BSA + chitosan) – GLD + GOx	Pt	development of amperometric glucose biosensor using permselec- tive hybrid sol-gel film	17
CE – graphene (in APTES) – EDC-activated GOx	GCE	development of GOx-bound graphene-functionalized GCE for mediatorless amperometric glucose sensing	2
CE – APTES – GLD-cross-linked GOx	GCE	demonstrating the effect of APTES concentration on the	2
CE – APTES – MWCNTs (in dimethylformamide) – GLD-cross-linked GOx		MWCNTs-based reagentless glucose biosensing	
CE – APTES – MWCNTs (in APTES) – GLD-cross-linked GOx			
CE – MWCNTs in APTES – GLD – GOx	GCE	development of GOx-bound APTES-functionalized MWCNTs for amperometric glucose biosensing	2
ectrode/polymer-coated electrode – APTES – GLD/EDC – GOx	electrode/polymer-coated electrode	development of long-term analyte sensor	2
a electrode-vapor-liquid-solid grown silica nanowires - APTES - EDC/NHS - GOx	silica nanowires functionalized Au electrode	development of silica nanowires-based electrochemical glucose biosensor	2
a/platinized Au electrode – APTES + <i>p</i> -benzoquinone + chitosan + GOx	Au/platinized Au	development of a novel one-pot electrodeposition strategy of using APTES-chitosan hybrid gel film to immobilize GOx for electro- chemical glucose detection	2
CE – MWCNTs – Nafion – APTES – POx	GCE	detection of putrescine	2
CE – MWCNTs – PDDA – Nafion – APTES – POx	CGE	improved detection of putrescine using PDDA to aid in dispersion of film materials	2
$a - APTES/CNTs - Fe_3O_4 NPs - xanthine oxidase$	Au	disposable Au screen-printed electrodes for the detection of xanthine	2
O – APTES – bis(sulfosuccinimidyl)suberate – uricase	ITO	detection of uric acid	2
e₃O₄ NPs − serratiopeptidase	Fe ₃ O ₄ NPs	increased anti-inflammatory effect of carrageenan-induced paw edema in rats with immobilized enzymes	2
ica-coated Fe ₃ O ₄ NPs – APTES – GLD – trypsin	silica-coated Fe ₃ O ₄ NPs	development of trypsin bound hairy polymer chains hybrid ${\rm Fe_3O_4}$ NPs	2
nO nanocrystals – APTES + TEOS – GLD – HRP	ZnO nanocrystals	investigating the effects of ZnO morphology on HRP immobiliza- tion and its catalytic activity	2
'O – ZnO – QDs – DNA/biotin – HRP/streptavidin	ITO	enhanced DNA sensor with a QD/ZnO composite	2
uartz slide – APTES – AuAg bimetallic NPs – GOx	quartz slide	preparation of AuAg bimetallic NPs and their use for glucose sensing	3
erlite – APTES – GLD – coimmobilization of cholesterol oxidase and HRP	perlite	development of covalent strategy for the coimmobilization of cholesterol oxidase and HRP	3
rconium – APTES – GLD – α -amylase	zirconium	development of α -amylase-bound zirconium dynamic membranes	3
esoporous silica – APTES – GLD – bovine carbonic anhydrase	SBA-15	development of bovine carbonic anhydrase-bound mesoporouse silica for hydration and sequestration of $\rm CO_2$	
ontrolled pore glass (CPG) - APTES - GLD - catalases	CPG	development of catalase-bound CPG and its use in batch and plug- flow reactors	
gano-functionalized mesoporous silica (FDU-12) – APTES – cellulase	FDU-12	mesoporous silica platform for cellulose-immobilized systems	
itosan–acetic acid – APTES – hexadecyltrimethylammonium bromide – GLD – carbonic anhydrase	chitosan	development of carbonic anhydrase-bound platform and its use to accelerate the hydration of CO ₂ for biomimetic CO ₂ sequestration in an aqueous solution	3
acroporous silica gels – APTES – 0.1 N HCl–0.1 N NaOH – GLD – glucoamylase	macroporous silica gels	development of glucoamylase-bound porous silica	3
PG – APTES – GLD – cellulose/D-xylanase	CPG	immobilization of cellulase and D-xylanase on CPG	3

porous glass beads — APTES — GLD — protocatechuate 3,4-dioxygenase porous glass beads immobilization of protocatechu		
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$ soil - HNO_3 - acetone - APTES - GLD - enzymes (laccase, GOx, tyrosinase, protease, β-D- soil and clay development of immobilization and clays } $	strategy for binding enzymes to soils	369
porous glass — APTES — GLD — glucoamylase preparation of APTES-treated p glucoamylase immobilization		370
solid support – HNO ₃ (except hydroxyapatite) – APTES – GLD – amyloglucosidase/α-amylase CPG (many types)/Micropil (various types)/Uni-sphere uncoated alumina/Lichrospher/Kromasil/hydroxyapatite (various types)	ization on various inorganic supports	371
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CPG – APTES – GLD – laccase CPG immobilization of laccase on C	CPG	373
CPG – APTES – GLD/EDC – Dextranase or immobilization of dextranase or	n CPG and study of kinetics	374
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CPG – APTES – GLD – laccase CPG immobilization of laccase on C	CPG	376
Al ₂ O ₃ - APTES - GLD - enzyme (urease, GOx, acetylcholine esterase, and <i>a</i> -chymotrypsin) Al ₂ O ₃ development of enzyme monol sensing	layer-functionalized FET for bio-	377
ceramic membrane – APTES – GLD – glucoamylase ceramic membrane development of surface-modifie immobilization	ed ceramic membrane for enzyme	378
silica support – APTES/2-hydroxyethylmethacralate – GLD – alcalase 2T silica derivatives studying the effect of various silica derivatives and stabilization of the enzymetry of	ica derivatives on the immobilization ne	379
mesoporous silica – APTES/ethanolamine – glyoxal – enzyme (lysozyme and catalase) mesoporous silica continuous surface modification	n for enzyme immobilization	380
nanoporous silica (SBA-15) – APTES – penicillin G acylase SBA-15 immobilization of penicillin G a silica	cylase on functionalized nanoporous	381
mesoporous silica — APTES — GLD — penicillin G acylase mesoporous silica design of mesoporous materials	for immobilizing penicillin G acylase	382
high-porosity mullite advanced ceramic material monolith – carbon nanofiber coating – APTES – GLD high-porosity mullite advanced ceramic material development of structured reac – polyethylenimine – enzyme (β-galactosidase and lipase) development of structured reac	ctors for enzyme immobilization	383
magnesium silicate (Florisil) – APTES – GLD – bovine liver catalase Florisil immobilization of catalase and s silicate	study of its kinetics onto magnesium	384
Na—Y zeolite — APTES — Au NPs zeolite immobilization of fungal protea microspheres and study of its		385
stainless steel – APTES – GLD – lysozyme stainless steel covalent immobilization of lyso	ozyme on steel	386
SiO ₂ – APTES – GLD – glutamate DHD SiO ₂ XPS-based analysis of glutamate	e DHD immobilization	387
mesoporous silica (pretreated with methanesulfonic acid) – APTES – acrolein glycol acteal + pyridine – mesoporous silica covalent immobilization of triacyl glycerol lipase mesoporous silica supports	cylglycerol lipase onto functionalized	388
activated carbon – APTES – GLD – laccase advantages of laccase immobiliz	zation	389
nano-SiO ₂ – APTES – novel carriers with reactive aldehyde groups – porcine pancreatic lipase nano-SiO ₂ covalent immobilization of triac nanoscale SiO ₂ spheres	cylglycerol lipase onto functionalized	390
silicic acid + APTES – amino-chemically surface modified gel – GLD – invertase porous silica matrix preparation of chemically surface immobilization	ce modified gel for enzyme	391
nanoporous silica – APTES – GLD – enzyme nanoporous silica preparation of single enzyme N	JPs in nanoporous silica	392
porous glass – APTES – GLD – enzyme (soybean peroxidase/HRP) porous glass development of peroxidase-bou	ind glass beads for phenol removal	393
TiO ₂ surfaces – APTES – ascorbic acid – enzyme (HRP and GOx) TiO ₂ characterization of enzyme imm	nobilization on TiO ₂ surfaces	394
porous silicon dioxide – APTES – GLD – GOx porous silicon dioxide characterization of enzyme imm materials	nobilization on silicon-based	395, 396
mesoporous silica monolith – tetramethoxysilane: APTES (5:6 v/v) – GLD – penicillin acylase mesoporous silica monolith covalent binding of penicillin a	cylase in mesoporous silica	397
glass beads – APTES – triethylamine + dry cyclohexane + phthaloyl chloride – α -amylase glass beads covalent binding of α -amylase glass beads	on functionalized glass beads	398
SBA-15 mesoporous molecular sieve – APTES – penicillin acylase SBA-15 mesoporous molecular sieve immobilization of penicillin acy	lase on SBA-15	399

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Table 3. continued

APTES-based enzyme immobilization strategy employed	biochip substrate	application	ref
ceramic monoliths – APTES – GLD – β -galactosidase	ceramic monoliths	immobilization of β -galactosidase on ceramic monoliths	400
SBA-15 – APTES – GLD – enzyme (α -chymotrypsin and lipase)	SBA-15	development of cross-linked enzyme aggregates in SBA-15	401
Si substrates (crystalline Si, porous Si, silicon nitride) – APTES – GLD – eta -lipase	silicon substrates	covalent binding of lipase on semiconducting materials	402
organic—inorganic hybrid silica monoliths — TEOS + APTES + ethanol + cetyltrimethyl ammonium bromide + H_2O – GLD – trypsin	organic-inorganic hybrid silica monoliths	development of immobilized trypsin reactor based on hybrid silica monoliths	403
SBA-15 – APTES – pepsin	SBA-15	preparation of active biocatalysts by immobilizing pepsin in mesoporous SBA-15 and investigating the reusability of the bioreactor	404
silicone nanofilament coated-glass slides – O₂ plasma – [2-(carbomethoxy)-ethyl]trichlorosilane – APTES – DY-647-NHS labeled α-chymotrypsin and lysozyme	silicone nanofilament coated-glass slides	use of superhydrophobic silicone nanofilaments for selective protein enrichment and purification applications	405
$SiO_2 - NH_3$ and H_2O_2 water solution - APTES - GLD - GOx	SiO ₂	characterization of GOx immobilization on SiO_2 surfaces	406
Si NPs – APTES – GLD – enzyme (GOx, invertase, esterase, and urease)	silica microparticles	determination of optoelectronic properties of a nanostructure system composed of enzyme-functionalized silica microparticles	407
silicon – APTES – dendrimer porphyrin – EDC/NHS – GOx	Si	development of a dendrimer-coated surface for GOx binding and analysis of protein activity	408
mesoporous silica SBA-15 and MCF – APTES – succinic anhydride – EDC – alkaline serine endopeptidase	SBA-15 and MCF	covalent binding of alkaline serine endopeptidase on SBA-15 and MCF	409
porous alumina membranes – H_2O_2 – APTES – GLD – glucose-6-phosphate DHD	porous alumina membranes	development of a method for the covalent binding of human	410
porous alumina membranes — $\rm H_2O_2$ — APTES — N,N-dimethyl-formamide + N-succinimidyl 3-maleimidopropionate — human CYP2E1		CYP2E1 and glucose-6-phosphate DHD in porous alumina membranes for xenobiotic metabolic studies	
mesoporous silica foams – APTES – irradiation in microwave reactor – penicillin acylase	mesoporous silica foams	improvement of microwave-assisted covalent binding of penicillin acylase using macromolecular crowding with glycine quenching	411
carbon felt – APTES – various coupling agents (GLD; ascorbic acid; 1,4-benzoquinone; terephththaloyl chloride + cyanuric acid; dimethylauberimidate dihydrochloride + <i>N</i> -(ɛ-maleimidocaproyloxy) succinimide ester + disuccinimidyl suberate) – tyrosinase	carbon felt	covalent binding of tyrosinase onto carbon felt using various coupling agents	412
GCE – APTES – GLD – acetylcholineesterase –choline oxidase	GCE	amperometric detection of paraoxon ethyl, aldicarb, and sarin using acetylcholineesterase-choline oxidase bound to Au–Pt bimetallic NP modified GCE	413
silica – TEOS and APTES in water/oil microemulsion – GLD – penicillin G acylase	silica	synthesis of aminopropyl-functionalized silicas by water/oil micro- emulsion and their use for immobilizing penicillin G acylase	414
Si wafer – APTES – sulfosuccinimidyl 4-(<i>N</i> -maleimidomethyl)-cyclohexane-1-carboxylate – rat phenol sulfotransferase	Si wafer	analysis of immobilized enzyme on planar surfaces in a microflow reactor	415
$SiO_2 - APTES - GLD - GOx$	SiO ₂	development of molecular-thin functional microarrays of GOx	416
Ta ₂ O ₃ - APTES - GLD - acetylcholine esterase	Ta_2O_3	investigation of the feasibility of transducing molecular-recognition into a measurable potentiometric signal	417
silica gel – APTES – GLD – β -galactosidase	silica gel	investigating the pretreatment of β -galactosidase with lactose to prevent the formation of covalent bonds near the active site during immobilization	418
silica gel – APTES – GLD – GL-7-ACA acylase	silica gel	covalent binding of GL-7-ACA acylase on silica gel	419
ZrO ₂ microtubes – APTES – EDC-activated lysozyme	ZrO ₂ microtubes	fabrication of ZrO_2 microtubes with specific surface functionalization for bacteria filtration and digestion	420
SBA-15 – APTES – GLD – laccase	SBA-15	development of laccase-bound mesostructure silica material and it use for the degradation of naphthalene	421
MWCNTs - APTES - GLD - organophosphate hydrolase	CNTs	development of organophosphate hydrolase-bound CNTs with enhanced stability for the amperometric detection of paraoxon	422
macroporous silica gel – APTES – GLD – papain	macroporous silica gel	development of highly stable papain-bound macroporous silica gel for industrial application	423
silica-coated superparamagnetic Fe ₃ O ₄ microspheres – TEOS + APTES – GLD – penicillin G acylase	silica-coated superparamagnetic $\mathrm{Fe_3O_4}$ microspheres	development of penicillin G acylase-bound silica-coated super- paramagnetic Fe ₃ O ₄ microspheres	424

APTES-based enzyme immobilization strategy employed

aluminum-coated optical fiber nanoprobes - APTES - GLD - lactate dehydrogenase

monolithic hybrid silica - TEOS + APTES - GLD - trypsin

aluminum oxide spherical beads - APTES - GLD - laccase

SiO₂ surfaces - plasma oxide activation - APTES - GLD - GOx

application	ref
development of a nanosensor to evaluate the effect of metabolic agents on cancer metabolism and survival	425
development of device to enrich and digest proteins apart from performing online sample buffer exchange	426
development of contactless, low-cost, and rapid drug screening methodology by inkjet printing	427
demonstrating the potential of immobilized laccase to polymerize technical lignosulfonates	428
covalent binding of catalase onto fluorisil using a spacer-arm and its application to batch and plug-flow type reactor systems	429
development of a stable and efficient aldolase immobilization technique using consecutive microwave irradiation at low temperature	430
development of effective lipase-bound glass beads for hydrolysis of	431

		technical lignosulfonates	
fluorisil – APTES – GLD – 6-amino hexanoic acid – 1-cyclohexyl-3-(2-morpholinoethyl) carbodiimide metho- <i>p</i> -toluene sulfonate – catalase	fluorisil	covalent binding of catalase onto fluorisil using a spacer-arm and its application to batch and plug-flow type reactor systems	429
mesocellular siliceous foam – p-benzoquinone – 2-deoxy-D-ribose-5-phosphate aldolase – microwave reactor (output power 30 W) coupled with a cooling module	mesocellular siliceous foam	development of a stable and efficient aldolase immobilization technique using consecutive microwave irradiation at low temperature	430
glass beads - APTES - GLD - lipase	glass beads	development of effective lipase-bound glass beads for hydrolysis of	431
glass beads – APTES – GLD – sol–gel encapsulation (using aqueous polyvinyl alcohol, aqueous sodium fluoride, and isopropyl alcohol, APTES + TEOS) – lipase		Naproxen methyl ester	
$Fe_3O_4 NPs - APTES - GLD - GOx$	Fe ₃ O ₄ NPs	synthesis and characterization of enzyme-magnetic NP complexes	267
porous glass – APTES – sodium borate buffer + HRP + sodium periodate	porous glass	development of a novel glass-based support incorporating ${\rm TiO}_2$ and its use for immobilizing HRP to treat aqueous halogenated phenolic waste	432
silica NPs – APTES – GLD – lipase	silica NPs	development of lipase-bound silica NPs for the enzyme-catalyzed resolution of racemic naproxen 2,2,2-trifluoroethyl thioester	433
ordered mesoporous SBA-15 - APTES - porcine pancreatic lipase	SBA-15	development of lipase-bound ordered mesoporous SBA-15 with improved adsorption	434
carbon felt – APTES – GLD – tyrosinase (under ultrasonic treatment)	carbon felt	development of tyrosinase-bound carbon felt-based working electrode for the amperometric detection of phenolic compounds	435
mesoporous SBA-15 – APTES – HAuCl_4 (lead to the formation of Au NPs) – human carbonic anhydrase	SBA-15	development of a biocatalyst by binding human carbonic anhydrase on Au NPs assembled onto mesoporous SBA-15 and its use for industrial-scale $\rm CO_2$ sequestration	436
short rod-shaped mesoporous silica – APTES – porcine pancreatic lipase	short rod-shaped mesoporous silica	development of porcine pancreatic lipase immobilization strategy for binding to the synthesized amino-functionalized short rod-shaped mesoporous silica	437
PDMS – O ₂ plasma – APTES – GLD – trypsin	PDMS	development of a versatile and easy-to-use site-selective surface modification strategy using microcontact printing and the selectivity of enzymatic degradation	438
magnetic SBA-15 – APTES – 2-bromoisobutyryl bromide – GMA/CuBr/CuBr_2/Bpy/DMF/H_2O – glucoamylase	magnetic SBA-15	development of reversible glucoamylase immobilization onto Cu(II) metal–ligand functionalized magnetic SBA-15	439
magnetic SBA-15 – APTES – 2-bromoisobutyryl bromide – GMA/CuBr/CuBr_/Bpy/DMF/H_2O – imidazole – glucoamylase			
magnetic SBA-15 – APTES – 2-bromoisobutyryl bromide – GMA/CuBr/CuBr_/Bpy/DMF/H_2O – iminodiacetic acid – glucoamylase			
Co-based silica core/shell NPs - EDC - GOx	Co-based silica core/shell NPs	development of immobilization strategy for binding GOx onto cobalt-based silica core/shell NPs	440
glass-ceramic support – APTES – GLD – invertase	glass-ceramic support	development of invertase-bound glass-ceramic support and analyz- ing its performance in batch bioreceptor	441
long period grating fibers $-$ H ₂ SO ₄ $-$ APTES $-$ GOx	long period grating fibers	detection of glucose using GOx-bound long period grating fibers	442
graphite nanosheets – Fe_3O_4 NPs – APTES – GLD – glucoamylase	GNSs	development and characterization of glucoamylase-bound graphite nanosheets–Fe ₃ O ₄ NPs hybrid	443
PDMS-coated glass microchannel – APTES – GLD – β -galactosidase PDMS-coated glass microchannel – MWCNTs – APTES – EDC·HCl + β -galactosidase	PDMS-coated glass microchannel	development of a β -galactosidase immobilized microreactor and its use for the continuous synthesis of lactulose	444

biochip substrate

aluminum-coated optical fiber nanoprobes

monolithic hybrid silica

aluminum oxide spherical beads

SiO₂

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APTES-based enzyme immobilization strategy employed	biochip substrate	application	ref
PDMS-coated glass microchannel – APTES – GLD – SWCNTs-DNA + β -galactosidase			
ceramic membrane surface – APTES – surface corona discharge induced plasma chemical process – CVD – glucoamylase	ceramic membrane surface	development of glucoamylase-bound ceramic membrane surfaces	445
TiO ₂ NPs – APTES – Au NPs – ferrocene-labeled secondary Ab – GOx	TiO ₂ NPs	development of GOx and ferrocene labeled Ab bound ${\rm Au}/{\rm TiO}_2$ nanocomposites for the electrochemical detection of progastrin releasing-peptide	260
palygorskite (also called attapulgite) – APTES – GLD – linoleic acid isomerase	palygorskite	preparation and characterization of linoleic acid isomerase-bound palygorskite	446
TiO_2 nanofiber membrane – APTES – GLD – Protex 6L (an alkaline protease)	TiO ₂ nanofiber membrane	use of Protex 6L-bound electrospun ${\rm TiO_2}$ nanofiber for the synthesis of sucrose monolaurate	447
SiO_2 deposited Au electrodes – APTES – GLD – GOx	SiO ₂ deposited Au	development of amperometric glucose sensor	448
silicon capillary – APTES – GLD – GOx	silicon capillary	development of a micromachined enzyme reactor for a flow injection analysis system for glucose detection	449
microporous silica support – APTES – GLD – GOx	microporous silica support	development of GOx bound microporous silica support	450
Fe_3O_4/SiO_2 NPs - APTES - GLD - GOx	Fe ₃ O ₄ /Si NPs	preparation of GOx bound Fe ₃ O ₄ /Si NPs	451
Pt disk electrode – chitosan/silica sol-gel hybrid membranes (prepared by mixing chitosan with APTES) + Au NPs + K ₃ Fe(CN) ₆ + HRP	Pt	development of $\rm H_2O_2$ biosensor by entrapping HRP in membranes doped with $\rm K_3Fe(CN)_6$ and Au NPs	452
glass microbeads – APTES – GLD – GOx	glass microbeads	development of microfluidic device involving heterogeneous microbeads and microarray-based detection	453
porous Celite beads – APTES – GLD – HRP	porous Celite beads	investigation of enzyme-catalyzed conversion of phenol in a HRP- bound membraneless electrochemical reactor	454
diatomaceous earth support Celite R-633 - APTES - GLD or glyoxal-laccase	diatomaceous earth support Celite R-633	covalent immobilization of laccase on Celite R-633 and it use to eliminate nonylphenol, BPA, and triclosan	455
silica-gel carrier – APTES – laccase	silica-gel carrier	development of highly stable silica-gel bound laccase for continuous analysis	456
dendritic composite magnetic particles (Fe ₃ O ₄ + APTES + TEOS) – GLD – porcine pancreatic lipase dendritic composite magnetic particles – porcine pancreas lipase (PPL)	dendritic composite magnetic particles	preparation of PPL-bound dendritic composite magnetic particles	457
modified palygorskite supports – APTES – lipase	modified palygorskite supports	development of lipase-bound modified palygorskite supports	458
mesoporous silica SBA-15 – APTES – lipase	SBA-15	preparation of lipase-bound SBA-15	459
palygorskite support – APTES – GLD – lipase	palygorskite support	development of lipase-bound palygorskite support	460
magnetic sol-gel composite supports – APTES + TEOS – GLD – lipase	magnetic sol-gel composite supports	preparation of lipase-bound magnetic sol-gel composite supports	461
Fe ₃ O ₄ NPs – APTES – GLD – lipase	Fe ₃ O ₄ NPs	development of lipase-bound magnetic Fe ₃ O ₄ NPs for the enzymatic synthesis of Diltiazem intermediate	462
mesoporous ZSM-5 zeolites – APTES – lipase	mesoporous ZSM-5 zeolites	preparation of lipase-bound mesoporous ZSM-5 zeolites	463
silica-coated Fe ₃ O ₄ NPs – APTES – lipase	silica-coated Fe ₃ O ₄ NPs	preparation of silica-coated ${\rm Fe}_3{\rm O}_4$ NPs with enhanced stability and catalytic activity	464
silicone elastomer – plasma (Ar followed by ultrapure H2O vapor) – APTES – GLD – oxalate oxidase	silicone elastomer	development of oxalate oxidase-bound silicone elastomer	465
AlGaN/GaN solution gate FETs – APTES – GLD + penicillinase G (in the presence of sodium cyanoborohydride	AlGaN/GaN solution gate FETs	development of penicillinase G-bound AlGaN/GaN solution gate FETs	466
tetramethoxysilane + APTES + pepsin	sol-gel	development of sol-gel immobilized pepsin-based strategy for the purification of chitosan	467
$\rm Fe_3O_4~NPs$ – APTES – GLD – peroxidase – entrapment in PEG hydrogel microparticles by photopatterning	Fe ₃ O ₄ NPs	assay development strategy based on the entrapment of enzyme- linked Fe ₃ O ₄ NPs within PEG hydrogel microparticles prepared by photopatterning	468
3-D poly(ethylene- <i>alt</i> -maleic anhydride) copolymer film – APTES – α -amylase poly(octadecene- <i>alt</i> -maleic anhydride) copolymer film – APTES – α -amylase	maleic anhydride copolymers	development of an immobilization strategy for binding α -amylase onto reactive copolymer films	469
silica capillary – APTES + methyltriethoxysilane – 1 M DMF solution of succinic anhydride – EDC/ NHS – cucumicin	silica capillary	immobilization of cucumicin in nanostructured silica capillary	470
NHS – cucumicin Co–Cr–Mo surface – APTES – GLD – trypsin	Co-Cr-Mo surface	immobilization of trypsin to silanized Co–Cr–Mo surface	471

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Table 3. continued

APTES-based enzyme immobilization strategy employed	biochip substrate	application	re
magnetic silica microspheres - TEOS + APTES - GLD - trypsin	magnetic silica microspheres	preparation of magnetic silica microspheres and their use for on-chip proteolysis systems	472
BDD thin film substrate – ZnO nanorod microarray – APTES + TEOS – GLD – tyrosinase	BDD thin film substrate	development of tyrosinase-bound ZnO nanorod microarrays on boron-doped nanocrystalline diamond to detect phenolic compounds	473
silica-coated fiberglass core in microchip – APTES + TEOS – GLD – trypsin	silica-coated fiberglass core in microchip	preparation of trypsin-bound silica-coated microchip for effective proteolysis	474
GCE – APTES – EDC-activated GOx	GCE	development of mediatorless electrochemical glucose sensing	475
GCE – graphene/poly-L-lysine (in APTES) – EDC-activated GOx		procedures based on the leach-proof covalent binding of GOx to GCE	
GCE – (GOx + HRP) GLD-cross-linked – APTES			
GCE – (GOx + HRP) GLD-cross-linked–graphene/MWCNTs/chitosan (in APTES)			
nanoporous alumina membranes – APTES + triethylamine – <i>N,N'</i> -carbonyldiimidazole + triethylamine – glutathione transferase	e nanoporous alumina membranes	characterization of glutathione transferase immobilized on nano- porous alumina membrane	476
Fe ₃ O ₄ /Si NPs - APTES - GMA - phospholipase A ₁	Fe ₃ O ₄ /Si NPs	development of phospholipase A ₁ on Fe ₃ O ₄ /SiO _x -g-P(GMA) NPs and its use for oil degumming	477
superparamagnetic Fe ₃ O ₄ NPs – APTES – penicillin G acylase + GLD	superparamagnetic Fe ₃ O ₄ NPs	development of penicillin G acylase magnetic cross-linked enzyme aggregates	478
Si (GLD/terephtalaldehyde/1,4-phenylene diisocyanate/1,4-phenylene diisothiocyanate/1,3-phenylene diisothiocyanate) – glucose-6-phosphate DHD	Si	determining the effect of cross-linking agent on the activity and thermostability of glucose-6-phosphate DHD	479
silica monolith capillary – prepolymer sol–gel (polyethylene glycol + TEOS + acetic acid) – APTES – GLD – lipase	- silica monolith	preparation of lipase-bound silica monolith in a fused silica capillary to develop a microreactor for performing lipid transformations	480
diatom-biosilica particles – plasma – APTES – GLD – tyrosinase	diatom-biosilica particles	development of tyrosinase-bound diatom biosilica for the biode- gradation of phenolic compounds	481
Au electrode – electrodeposition of Au NPs-Pt NPs – APTES – GLD – lysine oxidase	Au	detection of lysine using lysine oxidase-bound Au NPs-Pt NPs modified Au electrode	482
Fe ₃ O ₄ NPs - APTES - GLD - amyloglucosidase	Fe ₃ O ₄ NPs	preparation of amyloglucosidase-bound Fe ₃ O ₄ NPs	483
Fe ₃ O ₄ NPs – APTES – GLD – serratiopeptidase	Fe ₃ O ₄ NPs	preparation and characterization of serratiopeptidase-bound $\mathrm{Fe_3O_4}$ NPs	265
Fe ₃ O ₄ NPs - APTES - GLD - lipase	Fe ₃ O ₄ NPs	preparation of lipase-bound Fe ₃ O ₄ NPs and their use for the enrichment of polyunsaturated fatty acids	484
capillary – APTES – GLD – trypsin	capillary	screening of trypsin inhibitors by capillary electrophoresis using a trypsin-bound microreactor	485
MWCNTs-conc H_2SO_4 + HNO_3 (3:1 v/v) - APTES - succinic acid anhydride - lipase B	MWCNTs	development of MWCNTs-lipase B nanoconjugates for multiple usages in nonaqueous biocatalysis	486
mesoporous SBA-15 - APTES - GLD - hydrolase	SBA-15	development of hydrolase-bound mesoporous SBA-15 for copreci- pitation of cholesterol	487
TiO ₂ NPs/TiO ₂ -functionalized membranes – APTES – GLD – laccase TiO ₂ NPs/TiO ₂ -functionalized membranes – APTES – laccase – GLD TiO ₂ NPs/TiO ₂ -functionalized membranes – laccase TiO ₂ NPs/TiO ₂ -functionalized membranes – APTES – laccase	$\rm TiO_2~NPs$ and $\rm TiO_2\mathchar`-functionalized$ membranes	immobilization of laccase on TiO ₂ NPs and TiO ₂ -functionalized membranes	488
BDD electrode – KOH – APTES – GLD – GOx	BDD	demonstration of direct electron transfer between GOx and BDD, and detection of glucose	246

recommend the use of EDC-SNHS/EDC-NHS in their immobilization protocols; however, EDC alone still yields positive results.¹⁰⁵ Notice also that the NHS ester readily reacts with the amine group of lysine to form stable amide bonds, NHS ester is unstable in aqueous conditions, and ester hydrolysis competes with biomolecule attachments. NHS derivatives in solution at pH 8.0 have a half-life of only 1 h.¹⁰⁶ Nevertheless, heterobifunctional cross-linking is applicable for a variety of IAs, using ELISA, SPR, microarrays, advanced optical platforms, microfluidics, micro-, and nanomaterials.

The covalent binding of Abs on APTES-functionalized platforms has also been performed using 1,4-phenylene diisothiocyanate (PDITC).⁵¹ This homobifunctional cross-linker has on its on a phenyl ring two isothiocyanate groups that can bind to amines. The former reacts with the free amine groups on the APTES-functionalized platform to form a thiourea linkage, while the other binds to the Ab amine groups. The hydrazone bond has also been used for covalent binding of Abs.¹⁰⁷ The Abs are bound to succinimidyl 4-formylbenzoate (S-4FB) via the aldehyde moiety of S-4FB. Thereafter, S-4FB-bound Abs are covalently attached to APTES-functionalized surfaces via a hydrazone bond by the activation of succinimidyl 6-hydrazinonicotinamide acetone hydrazine (S-HyNic).

The covalent attachment of Abs is widely performed on surfaces rich in silicon, for example, silicates. The Abs covalently bound on monocrystalline silicon are highly stable and retain functional activity for over 2 months.¹⁰⁸ Similarly, highly uniform Ab immobilization has been achieved on APTES-functionalized glass.¹⁰⁹ This procedure involves the reduction of an imine bond formed between aldehyde and amine at reaction sites to aziridine, which stabilizes the complex and prevents protein leaching.

The leach-proof covalent binding of enzyme to electrodes with a prolonged shelf life is highly critical for the development of commercially viable POC glucose biosensors. The aldehyde groups of biomolecules bind covalently with the amine groups of APTES on the substrate via the formation of Schiff base.¹¹⁰ The GLD coupling procedure can be performed under ambient conditions, which are ideal for developing biosensors with high activity and stability. The water-soluble EDC-mediated reactions convert the carboxyl group into unstable O-acylisourea, which forms an amide bond with the amine group of APTES. As a few examples, the preparation of a glucose oxidase (GOx)-bound GCE involves the drop-casting of 10 mg mL⁻¹ GOx cross-linked via GLD on 1% (w/v) KOH-pretreated GCE, followed by the immediate drop-casting of 4% APTES to form a highly stable GOx-bound GCE.¹¹¹ This strategy is also applicable for covalent immobilization of GOx on a graphene-functionalized GCE.¹¹² The GCE pretreated with 1% KOH is allowed to react with graphene (1 mg mL⁻¹ dispersed in 0.125% APTES). APTES plays a dual role in this procedure, acting as a surface modification agent that introduces amino groups on the GCE and graphene, besides its role as a dispersion agent. Thereafter, EDC-activated GOx is bound covalently to the graphenemodified GCE by heterobifunctional cross-linking. The incorporation of graphene on GCE significantly increases the active surface area for enzyme immobilization, resulting in enhanced response signal. Similarly, the same GOx immobilization strategy can be extended to the MWCNT-functionalized GCE.

3.2. Oriented

Most of the oriented Ab immobilization strategies employ FBP, such as Protein A,^{113,114} Protein G,^{115,116} and Protein A/G¹¹⁷ that bind Abs through the F_c region, leaving the antigen-binding

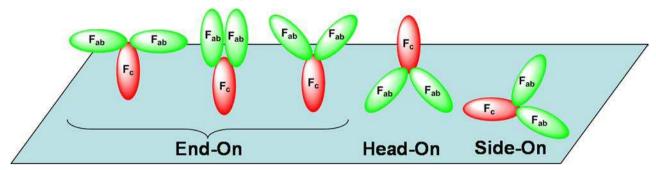
sites at the fragment antigen-binding (F_{ab}) region free for antigen binding. The native Protein A is a cell wall component produced by several strains of Streptococcus aureus, but the recombinant Protein A is produced in E. coli. With a MW of ~46 kDa and isoelectric point (pI) of 5.16, it contains five high affinity ($K_{\rm a}$ = 10^8 mol^{-1}) binding sites, which are capable of binding to the F_c region of immunoglobulin G (IgG) from several species such as rabbit and human. A Protein A molecule can bind to a maximum of two Abs due to spatial constraints.¹¹⁸ However, Protein A and IgG binding is not equivalent for all species and even for the subclasses of a particular IgG. As an example, Protein A covalently bound on a silanized Au-coated SPR chip by EDC-SNHS can be useful for the oriented immobilization of antihuman fetuin A (anti-HFA) Ab.⁶³ This orientation strategy results in a superior analytical performance as compared to covalent, noncovalent, Protein A noncovalently bound, and commercial CM5 (carboxymethyl dextran) SPR chip-based Ab immobilization strategies. Although these intermediate proteins have been widely used for direct binding assays, such as those based on SPR, microcantilevers, and quartz crystal microbalance (QCM), they are not widely employed for sandwich IAs. This could be rationalized that the secondary Abs can bind to the vacant binding sites on these intermediate proteins, which leads to a higher background signal and nonspecific binding. The blocking of these vacant sites further adds to the assay cost. To date, the oriented immobilization strategies are not widely used for enzyme immobilization in biosensor construction.

Similarly, Protein G is another FBP, which is a bacterial cell wall protein from group G Streptococci. The native Protein G has two IgG binding domains apart from sites for albumin and cell surface binding, but these nonspecific binding sites have been eliminated from recombinant Protein G (MW = \sim 22 kDa and pI = 4.5) produced commercially from *E. coli*. Protein G has higher affinity than Protein A for most mammalian IgGs, but it does not show any binding to human IgM, IgA, and IgD. Despite having tremendous similarity in the tertiary structures, Protein A and G have different amino acid compositions. The optimal binding of Protein A and G occurs at pH of 8.2 and 5, respectively, but is still sufficient at pH 7.0-7.6. Because of the inherent differences in the binding properties of these FBPs, assessment of their binding to specific capture Abs is a prerequisite for a particular bioanalytical application. Indeed, this problem has been solved by the production of Protein A/G, which results from the gene fusion of the F_c-binding domains of Protein A and G. It binds to all human IgG subclasses with the exception of IgA, IgE, IgM, and to a lesser extent IgD. Besides its stronger affinity, the binding is less pH-dependent with optimum Ab binding at pH 5-8. Apart from the FBPs, Protein L (MW = \sim 36 kDa and pI = 4.5) isolated from *Peptostreptococcus magnus* has been employed as an intermediate protein.^{119,120} Protein L binds to the kappa light chains of a wider range of IgG classes and subclasses without interfering with the antigen-binding sites of the Ab. Furthermore, it binds to the F_{ab} fragments and single chain variable fragments.

3.3. Site-Specific

The ability to introduce desired amino acids and functional groups at a specific position in Abs and their derivatives enables the development of site-specific immobilization strategies.^{121,122} It involves the insertion or chemical binding of an amino acid, an oligopeptide, or an affinity tag at the specific site of an Ab or its derivatives that binds to the complementary affinity molecule immobilized on a matrix surface. Commercially available tags include polyhistidine,¹²³ avidin–biotin,¹²⁴ and FLAG

Scheme 4. Different Possible Orientations of IgG on a Substrate



tags.^{125–127} Free amino acids containing the sulfhydryl group (such as cysteine) can also be used for Ab immobilization on Au surfaces via self-assembly.^{128,129} The incorporation of cysteine on the Ab fragment apart from the antigen recognition site enables the use of sulfhydryl linkers that leads to more efficient solidphase immobilization.^{130,131} The cysteine tag also broadens the choice of solid supports as Abs can be immobilized on various substrates including cyanogen bromide (CNBr)-activated surfaces.¹³² It provides the robustness to the tagged Ab required for leach-proof immobilization..¹³³ The tagging-based strategies are highly efficient. However, tagging of Abs adversely affects their functional activity due to the conformational change, thereby resulting in reduced binding efficiency.¹³⁴

3.4. PNA and DNA-Directed

Complementary PNAs are chemically synthesized singlestranded nucleic acids that feature an amino acid derivative of alkylamide instead of the normal phosphodiester backbone.^{135,136} PNAs can form nonionic and achiral PNA–PNA, PNA–RNA, and PNA–DNA constructs that are resistant to enzymatic hydrolysis and temperature, attractive properties that may allow for the use of PNA in immobilization strategies. PNAconjugated Ab has been immobilized on surfaces grafted with the antisense oligonucleotide or PNA by hybridization.^{137,138} The ability to control surface-Ab distance through PNA monomers provides strong site specificity and orientation for Ab capturing.^{139–141} Hybridized oligonucleotide tags are also attractive as they can be separated from their complementary strands by changing the pH or divalent ion strength.¹⁴²

3.5. Noncovalent Attachment

Protein/enzyme molecules can be attached to different interfaces with various mechanisms and forces acting between protein and surface, including hydrophobic, electrostatic, and van der Waals forces.^{143–145} Generally proteins tend to adhere onto hydrophobic surfaces due to the release of water dipoles from the hydrophobic surface to the bulk solution during protein adsorption. In addition, the protein concentration plays an important role in the steady-state adsorption of proteins only on hydrophilic surfaces,¹⁴⁶ not hydrophobic ones.¹⁴⁷ Proteins can undergo conformational changes and unfold due to different forces acting during the adsorption process, affecting their biological activity. "Hard" proteins (lysozyme, β -lactoglobulin, α chymotrypsin, etc.) upon adsorption retain their dissolved state without spreading. In contrast, "soft" proteins (bovine serum albumin (BSA), IgG, fibrinogen, or α -lactoglobulin) can spread on the surface upon adsorption. "Soft" proteins tend to adsorb onto various surfaces as compared to "hard" proteins that adsorb onto hydrophilic surfaces only when there is electrostatic attraction between protein molecules and the surface.¹⁴⁸ Surface

roughness also affects adsorption and the morphology of adsorbed protein. For example, larger amounts are adsorbed to the hydrophobic surfaces as compared to the hydrophilic ones. Only on smooth hydrophilic or hydrophobic surfaces does collagen form elongated assemblages with small or high surface features, respectively.¹⁴⁹ Random orientation is often encountered with protein adsorption; that is, there are a few different possible orientations of IgG on the surface (Scheme 4).

Review

Amine groups on APTES-functionalized surfaces with positive charge at neutral pH can interact with negatively charged enzymes via electrostatic interaction. This is the simplest approach to immobilize enzymes onto solid surfaces for applications that require no directional orientation of the immobile biomolecules. The mechanisms involved in adsorption include ion exchange, ion pairing, hydrophobic bonding, and polarization of π -electrons. As an example, the negatively charged DNA backbone binds strongly to the positively charged APTEScoated mica.¹⁵⁰ Resulting structures are stable enough for AFM imaging in air or buffer. Of notice is an important medical application using amine-terminated silicon surfaces, that is, hydrogen-terminated silicon surface, as an immobilization carrier for bone morphogenetic protein 2 (BMP2).¹⁵¹ However, immobilized enzymes can be often desorbed by simple change of pH or ionic strength.

Hydrophobic interaction is simple and useful for the binding of lipophilic biomolecules, for example, membrane-bound dehydrogenase enzymes. The main driving force for adsorption can be attributed to the hydrophobic interaction between such lipophilic membrane-bound enzymes and the alkane moiety of silanes.¹⁵² This is a fast, direct, and simple procedure for lipophilic biomolecules; however, denaturation of adsorbed biomolecules with random orientation is often encountered.

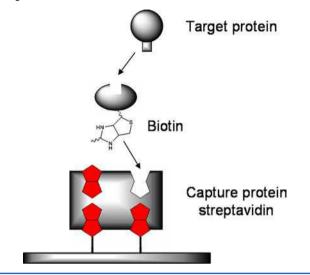
Protein surface density is higher for adsorbed proteins than for covalently bound ones. Considering IgG molecule (156 kDa) with nominal dimensions of 14.5 × 8.5 × 4 nm, the area covered by singular IgG molecule with side-on orientation is minimal and about 2.6 mg m⁻².¹⁵³ With adsorption on the GOPS and APTES modified surface, respectively, the protein surface density is 1.55 \pm 0.05 to 2.00 \pm 0.04 mg m⁻² as compared to 0.20 \pm 0.02 to 0.55 \pm 0.05 mg m⁻² for covalently bound ones.¹⁵⁴ The protein structure is unaffected by the duration of incubation of the silanized substrate with the IgG solution.

3.6. Affinity

The specific molecular interaction between the avidin and biotin pair ($K_{\rm d} \sim 10^{-15}$ M) is a well-known phenomenon and has been extensively advocated for biomolecule immobilization as the cubic shape avidin possesses for four biotin binding sites. Conceptually, avidin is covalently attached to APTES-surfaces by either GLD-mediated or EDC-mediated reactions as previously

described. Avidin then acts as a biocompatible linker between APTES-surfaces and biotinylated biomolecules. Although avidin/streptavidin is randomly attached to the APTES-modified surface, the interaction between biotinylated biomolecules and avidin/streptavidin is more specific to effect oriented immobilization of the target biomolecule as shown in Scheme 5. Vice

Scheme 5. Immobilization of a Protein by Affinity through a Streptavidin/Biotin Interaction



versa, the APTES-functionalized surface can be biotinylated and binds to avidin. Several biotinylated biomolecules are commercially available, and various protocols have been established for biotinylation of biomolecules.^{155–160} Biotin is a small compound (244.3 Da); therefore, biotinylation does not affect biomolecule

conformation and biological activity. This procedure is more popular for the immobilization of Abs as compared to enzymes. Besides avidin—biotin, recombinant proteins can be selectively attached to an artificial polypeptide scaffold system via heterodimeric association of a separate leucine zipper pair consisting of 43 amino acids with an affinity constant of 10^{-15} M.¹⁶¹

The capture ligand approach involving the use of protein transsplicing mechanism deserves a brief comment here because this approach can offer some potential in biomolecule immobilization. Intein fragments exhibits protein splicing activity, and the intein self-processing domain can split into two fragments: the former is incorporated to the target biomolecule, whereas the latter is attached to the APTES-modified surface. The process is lengthy (16 h), which is similar to other intein-mediated approaches.¹⁶² Nevertheless, this approach has been employed for the immobilization of proteins onto cysteine-functionalized substrates.^{163,164} The main drawback of this procedure is the requirement of a relatively high proteins concentration (100 mM) to effect direct ligation to cysteine-modified biosensor surfaces.

The hexahistidine tag is most commonly used in high throughput protein purification^{165,166} as well as for the immobilization of histidine-tagged proteins on electrode surfaces.^{167–169} However, specificity and stability of this tag are lower as compared to other affinity methods, and the preparation involves many reagents and steps.¹⁷⁰ Recently, Sanz et al.¹⁷¹described an elegant immobilization procedure for proteins tagged to the affinity polypeptide C-LytA, a choline-binding protein (CBP) of the pneumococcal cell wall,¹⁷² on functionalized graphite electrodes. In brief, a SAM of choline-functionalized thiol chains is synthesized over thiocarboxylic acid-treated Au, which displays selective affinity for β -galactosidase tagged to

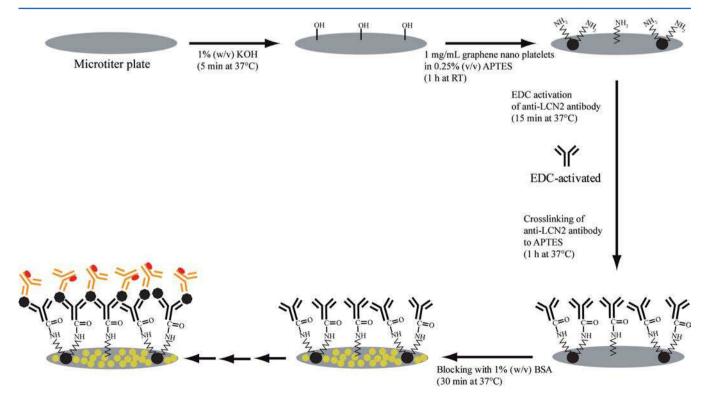


Figure 2. APTES-based sandwich ELISA procedures using graphene nano platelets-based Ab immobilization for the detection of human lipocalin-2 (LCN2). Reproduced with permission from ref 176. Copyright 2014 Elsevier B.V.

C-LytA (CLyt-bGal protein). The C-LytA system exhibits higher buffer compatibility and is not subject to interference with other metalloproteins. The binding is also completely reversible, which enables the reuse of the electrode and opens the possibility of developing other tailor-made supports.

3.7. Entrapment

Silica sol–gel films have been used to encapsulate enzyme and redox mediators. A porous, silane surface using APTES, GOx, and [2-(3,4-epoxycyclohexyl)ethyl]trimethoxysilane was deposited on Pt for glucose detection.¹⁷³ The thick sol–gel film adversely affected detection time. The Pt electrode required high detection potentials (+0.7 V vs Ag/AgCl), which compromises detection linearity at higher glucose concentrations. Methyl-trimethoxysilane (MTMOS), tetraethoxysilane (TEOS), APTES, and GOPS can be assembled onto highly oriented pyrolytic graphite (HOPG) for glucose detection.¹⁷⁴ The best matrix for GOx immobilization is GOPS, followed by APTES, MTMOS, and TEOS. The chemical structure and composition of the sol–gel mixture changes the roughness, size, and distribution of pores in the film, which affect the response time, sensitivity, and detection limit of glucose.

A sol-gel approach has been attempted for making permselective organic-inorganic hybrid films consisting of three organosilanes (APTES, TEOS, triethoxy-1H,1H,2H,2H-tridecafluoro-*n*-octylsilane) and two biomacromolecules, BSA and chitosan, deposited on Pt.¹⁷⁵ 3D sol-gel Si-OH condensation networks cross-linked by APTES and coupled with molecular interactions between different components of the films are a suitable environment for H₂O₂ permeation. This biosensing scheme offers a detection limit of 32 μ M and a response time of only 3 s. However, the long-term stability of the biosensor is questionable as the glucose signal decreases to 76%, while the ascorbic acid interference signal increases to 126% of initial responses after only 5 days of storage.

4. APPLICATIONS IN BIOSENSORS AND DIAGNOSTICS

4.1. Microtiter Plate Formats

The immobilization of biomolecules on MTPs is generally achieved by silanization on pretreated plates with available surface hydroxyl groups. Highly sensitive and rapid sandwich ELISA procedures have been developed for the detection of disease biomarkers, such as HFA, human albumin (HA), and human lipocalin-2 (LCN2). The resulting microtiter plate format is capable of detecting HFA with linearity of 9 pg m L^{-1} to 20 ng mL^{-1} and sensitivity of 39 pg mL^{-1} , that is, ~16-fold more sensitive and 3-fold rapider than the established commercial ELISA (Figure 2).¹⁷⁶ HFA is a specific biomarker for hepatocellular carcinoma and atherosclerosis. Doubtlessly, this generic Ab immobilization procedure can be extended to other commercially available substrates such as polystyrene, poly-(methyl methacrylate), cellulose acetate, polycarbonate, and cyclic olefin polymers (Zeonex). Adapting the same silanization protocol, improvements in other IA formats for chemiluminescent enzyme IA and SPR IA could also be anticipated.

Further improvement and simplification of this immobilization procedure is feasible by using only EDC as the heterobifunctional cross-linker for HFA, HA, and LCN2.¹⁰⁵ The analytical sensitivity of the IA increases by 80-fold with the addition of graphene nanoplatelets (GNPs) by virtue of higher Ab immobilization density associated with the increased surface area. The use of GNPs does not lead to any additional processing steps or increase in IA time or complexity. The assay can detect LCN2 in plasma, serum, and whole blood without any interference from endogenous physiological substances. Argon plasma is an alternative for generating reactive hydroxyl moieties on the MTPs for APTES-functionalization and subsequent binding of rabbit IgG via cross-linkers.¹⁷⁷ After immobilization, about 95% of covalently bound IgGs are retained after extensive washing on the plasma-treated polystyrene MTP in comparison to an untreated MTP. The resulting plasma-treated MTPs exhibit similar or even better analytical performance as compared to commercially available preactivated MTPs.

Recently, a simple immobilization procedure involving the dilution of a capture Ab in the ratio of 1:1 (v/v) in 1% (v/v) APTES to form a stable complex via ionic and hydrophobic interactions was demonstrated.¹⁷⁸ The complex is covalently bonded to a polystyrene-based MTP, pretreated with KOH. This IA platform shows 51-fold increased sensitivity than the commercial IVD kits for HFA, C-reactive protein (CRP), HA, and LCN2. Moreover, the Ab-bound MTPs, stored at 4 °C in 0.1 M PBS, are stable for 2 months, attesting the leach-proof and functionally stable binding of capture Abs.

The Ab can also be preactivated with EDC and admixed with 1% (v/v) APTES in the ratio of 1:1 (v/v) to form a stable complex. Similarly, the complex is covalently bonded to a polystyrene-based MTP, pretreated with KOH toward the development of a prospective rapid IVD procedure for detecting HFA in just 30 min.¹⁷⁹ Both of the aforementioned procedures exhibit high precision equivalent to that of the commercial sandwich ELISA in the real sample matrices.

Sensitive detection of carcinoembryonic antigen (CEA) is feasible using horse radish peroxidase (HRP)-functionalized mesoporous silica NPs (MSN) as labels.¹⁸⁰ The HRP-functionalized MSN is prepared by simultaneous coimmobilization of HRP and anti-CEA Ab onto the APTES/GLD-functionalized MSN. The maximum chemiluminescent intensity is obtained at the HRP–CEA mass ratio of 10:1. The use of HRPfunctionalized MSN enhances chemiluminescent signal and sensitivity, 10-fold higher than the conventional IA with the anti-CEA monoclonal Ab simply bound to the 96-well MTP by passive adsorption.

The highly sensitive detection of bisphenol A (BPA) was demonstrated using direct hapten-coated polystyrene MTPbased competitive indirect ELISA (ciELISA).¹⁸¹ The H₂SO₄/ HNO₃ pretreated polystyrene MTP is functionalized with APTES and subsequently bound to the carboxyl group of 4,4bis(4-hydroxyphenyl) valeric acid (BVA), an analogue of BPA, by the *N*,*N'*-dicyclohexylcarbodiimide (DCC) method. This is followed by the addition of BPA, anti-BPA Ab, and goat antimouse IgG-HRP conjugate, and the 3,3',5,5'-tetramethylbenzidine (TMB) reaction substrate for HRP. The direct haptencoated ciELISA detects BPA with a limit of detection of 0.27 ng mL⁻¹ as compared to 15.90 ng mL⁻¹ for the hapten-protein coated format. The assay is capable of detecting BPA in tap water and seawater samples with recoveries between 70% and 142%.

4.2. Optical Biosensors

Optical-based biosensors have been designed for the monitoring of biomolecular interactions in real-time because they are usually label-free (not requiring enzymes or fluorescent dyes), easy to fabricate, and easily multiplexed to enable an array-based format for high throughput efficiency. SPR, FTIR, and refractive index (RI) are the most common techniques used in conjunction with optical biosensors. However, immobilization of the biomolecules to obtain biosensors with high sensitivity remains problematic.

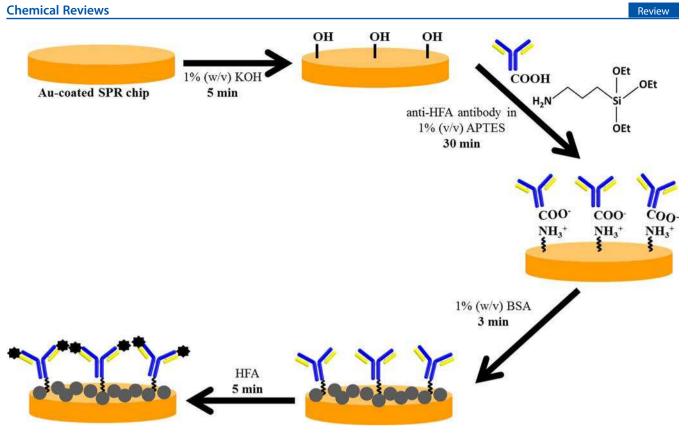


Figure 3. Schematic of the developed SPR-based IA procedure for human fetuin A (HFA). Reproduced with permission from ref 187. Copyright 2014 Royal Society of Chemistry.

These problems have been addressed in many examples by the use of APTES to minimize the sensor preparation time, reagents, and required steps.

In SPR, p-polarized laser light reflects at the interface of an optically dense and less dense medium to generate an evanescent field at a specific angle, which is apparent by collective motions of conducting electrons (plasmons) in a thin metal layer (30 nm). Au films are commonly used due to their lower dielectric constant, biocompatibility, and inertness. Again, SAM has been shown to be immensely useful for this sensor type.¹⁸² The SPR Au chip can be cleaned with piranha treatment for a few minutes, which oxidizes the Au surface, as shown by the contact angle of about 9.3°. The further washing of chips with deionized water leads to the adsorption of water molecules and generates the desired hydroxyl groups,^{183,184} as apparent from an increase in contact angle to $28-32^{\circ}$. After silanization, the contact angle of water for the SAM of APTES on Au surface is $59 \pm 1^{\circ}$.

Conventional SPR biosensors are difficult to realize in a largescale array format because they operate in total internal reflection mode. However, localized SPR (LSPR) using surface modified silica NPs on an Au layer can overcome some of these issues as it operates using the transmission mode. An optical sensor was developed for the detection of fibrinogen in the range of 10 ng mL⁻¹ to 10 μ g mL⁻¹. In this configuration, APTES modified silica NPs are attached through EDC to form a dense monolayer (100 nm) on an Au-coated glass substrate.¹⁸⁶ The silica NP layer is then capped with another layer of Au before the attachment of Protein A through EDC and NHS. Antifibrinogen Ab is then attached to the Protein A to detect fibrinogen. The Ab–antigen interaction is monitored by following the absorbance increase at 550 nm after 30 min (BSA as a control shows no absorption increase). Recently, an SPR-based IA biosensor was designed for the detection of HFA using a simplified Ab immobilization procedure.¹⁸⁷ In this system, the anti-HFA capture Ab is simply added to 1% APTES and deposited for 30 min on a KOH-treated Au-coated SPR chip (Figure 3). The Ab bound chip is first blocked with BSA before interacting with HFA for 5 min. The IA detects HFA in the range of 0.3-20 ng mL⁻¹, and the chip could be regenerated with glycine-HCl for at least 35 repeated HFA analyses, indicating the leach-proof capture Ab binding to the chip. The detection of HFA spiked into real sample matrixes with this system compares well with commercial HFA sandwich ELISA. The chips are very stable after 2 months of storage without biological activity loss. Considering the simplicity and rapidity of this procedure, the chip can be easily prepared 30 min before the intended use.

A cost-effective, label-free, and simple optical immunosensing system is desired for the measurement of surface enhanced light scattering (SELS) signals of solid supports, by simultaneous scanning the excitation and the emission of the response signal.¹⁸⁸ The goat antihuman IgG, goat antirabbit IgG, and rabbit antihuman IgG, respectively, are bound to the APTES-GLD functionalized glass slides, which detect human IgG, rabbit IgG, and human IgG in the linear ranges of $0.1-100 \ \mu g \ mL^{-1}$ (LOD 50 ng mL⁻¹), and $0.1-10 \ \mu g \ mL^{-1}$ (LOD 10 ng mL⁻¹), respectively. The Ab-bound sensor is highly specific and not subject to any interference from other proteins. It can be effectively regenerated for repeated use by treatment with carbamide solution.

Of notice is the use of a silicon-on-insulator (SOI) optical biosensor based on a microring resonator (cavity radius of $5 \,\mu$ m) for specific protein binding detection. This format exhibits

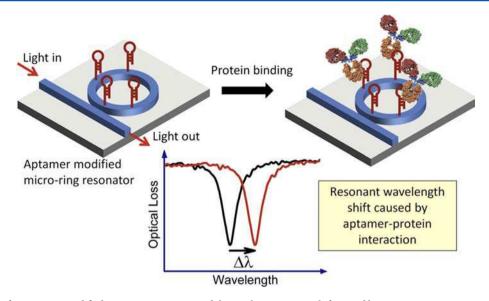


Figure 4. Schematic of an aptamer modified microring resonator and the resultant resonant shift caused by an aptamer—protein interaction. Reproduced with permission from ref 190. Copyright 2013 Elsevier B.V.

advantages as compared to the system using traditional passive photonic components because its dimensions are reduced drastically. Silicon surfaces are silanized using 1% APTES followed by the coupling of biotin using EZ link-NHS-LC-biotin for the detection of avidin in the range of 50 ng mL⁻¹ to 5 μ g mL^{-1} . The measurement is based on the shift in resonance wavelength (as low as 7.5 pm) due to changes in the dielectric surroundings of the cavity. This SOI format when coupled with complementary metal-oxide semiconductor fabrication offers the high throughput and resolution required for practical and routine applications.¹⁸⁹ This similar SOI format is extended to detect human IgE and human thrombin with a detection limit of 33 pM and 1.4 nM, respectively. The system consists of an array of three sensor microrings and one reference microring. The silicon microring (cavity radius of $5 \mu m$) device is treated with 2% APTES followed by GLD activation for immobilization of anti-IgE and antithrombin aptamers, respectively. This sensing approach is advantageous in the development of POC diagnostic devices due to its stability and specificity.

Resonant wavelength red-shifts (increase in the effective RI) are used to determine the IgE and thrombin concentrations after 30 min of binding (Figure 4). The detection of IgE is more sensitive than human thrombin due to the significantly higher affinity of binding between IgE and its aptamer. The system, when designed by using the two different aptamers attached to the individually addressable microrings, is very successful in terms of selectivity when challenged with a mixture of IgE and thrombin.¹⁹⁰ Another application of the SOI microring optical resonator arrays is the label-free detection of CEA in undiluted fetal bovine serum in less than 30 min.¹⁹¹ The anti-CEA Abs are bound to the substrate by hydrazone bond formation chemistry. It detects CEA in the range of 0-121 ng mL⁻¹ with LODs of 2 and 25 ng mL⁻¹ in buffer and serum, respectively. Therefore, it can monitor the entire clinically relevant range of 5-100 ng mL⁻¹. The anti-CEA Ab-bound surface is regenerated effectively with glycine for repeated use and provides high accuracy as verified by standard ELISA.

Ultrasensitive, label-free detection of human cardiac troponin-T (cTnT) is feasible by using complementary metal—oxide semiconductor-compatible silicon nanowire (SiNW) sensor arrays.⁴⁶ The SiNW-sensor arrays are fabricated on an N-doped silicon-SOI substrate with a 145 nm buried oxide layer. The SiNW device has 36 clusters of five individually addressable nanowires each. The SiNW array chips functionalized with APTES-GLD and anti-cTnT Abs enable cTnT detection in assay buffer and undiluted human serum down to 1 and 30 fg mL⁻¹, respectively. This remarkable LOD is 3 orders of magnitude lower than the available ELISA-based assay formats.

Label-free biosensors containing APTES functionalized nanoporous (20-100 nm) polymeric gratings have also been designed to detect biomolecules. Application of gratings for biosensing has been hindered because the as-prepared polymers with optical properties encounter problems immobilizing biomolecules. Modifying the traditional holographic polymerdispersed liquid crystal (H-PDLC) system by adding APTES directly to the photopolymer suspension enabled the gratings to capture biomolecules. The concentration of APTES chosen (10%) was critical in maintaining a high index modulation (0.07), which enhanced the signal-to-noise ratio. Changes in the RI before and after the additions of the target analytes resulted in the sensor's capability of detecting biomolecules such as biotin in the range of $0.05-0.5 \text{ mg mL}^{-1}$. The optical biosensor is also capable of detecting sequential biomolecular interactions as demonstrated by a series of additions of biotin, streptavidin, biotinylated antirabbit IgG, and finally rabbit IgG, whereas goat IgG as a control for the last step shows no response.¹⁹²

A guided-mode resonance (GMR) biosensor was developed for the rapid detection of cardiac biomarkers, that is, cardiac troponin I (cTnI), creatine kinase MB (CK-MB), and myoglobin (MYO), in human serum in 30 min.¹⁹³ The dose–response ranges and LODs for cTnI, CK-MB, and MYO were 0.05–10 ng mL⁻¹ and 0.05 ng mL⁻¹, 0.1–10 ng mL⁻¹ and 0.1 ng mL⁻¹, and 0.03–1.7 μ g mL⁻¹ and 35 ng mL⁻¹, respectively. The capture Abs are bound to GLD by cross-linking. The immobilization of Abs is evaluated by sandwich APTES-functionalized Si₃N₄ surface ELISA and sandwich immunogold assays. The developed technology holds great promise to develop low-cost and portable biosensors.

An all-fiber optofluidics-based bioassay platform (AFOB) has been attempted for the highly sensitive, selective, and rapid

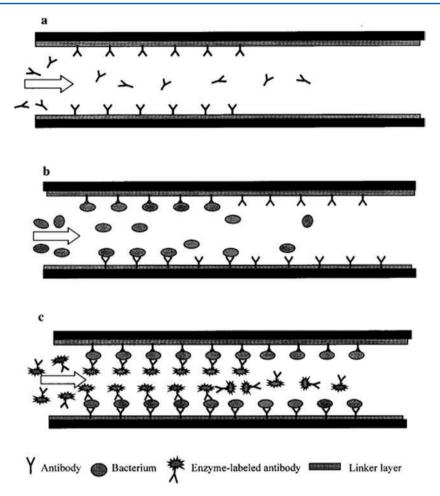


Figure 5. Schematic diagram of a fused-silica capillary column for the detection of *E. coli* O157:H7: (a) immobilization of Abs via APTES and GLD, (b) *E. coli* capture, and (c) formation of immunocomplex by the addition of alkaline phosphatase-labeled anti-*E. coli* O157:H7 Ab. Reprinted with permission from ref 195. Copyright 2001 American Chemical Society.

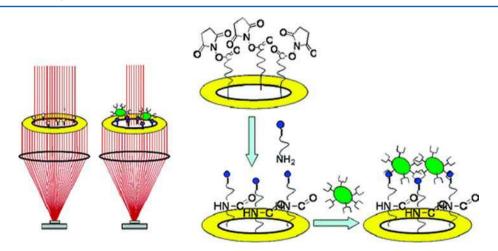


Figure 6. An optical biosensor was designed for the label-free detection of *E. coli* O7:K1. (left) Difference in transmission of laser beam between a reference well with only immobilized Abs and a well containing both Abs and *E. coli* cells blocking a part of the laser beam. (right) Schematic of the coupling of Abs to the activated glass slide followed by immobilization of *E. coli*. Reprinted with permission from ref 196. Copyright 2006 American Chemical Society.

assessment of BPA leaching risk.¹⁹⁴ It involves the covalent binding of BPA-BSA on the APTES and GLD-functionalized fiber optic sensor and the detection of BPA using fluorescent-labeled anti-BPA monoclonal Abs. With linearity of $0.5-100 \ \mu g$ L⁻¹ and LOD of $0.06 \ \mu g$ L⁻¹, the sensor is proven for the

detection of BPA in a spiked environmental sample matrix with recoveries in the range of 90-120%.

Optical biosensors have also been designed to detect whole microorganisms rather than just small molecules such as proteins. The detection of pathogenic microorganisms is critical for controlling foodborne illnesses. Conventional methods based on culture and colony counting are time-consuming, expensive, complicated, and may require signal amplification or labeling. Detection by fluorescence or SPR has led to rapid analysis using a variety of optical biosensors. In one example, a fused-silica capillary column is treated with APTES, and then Abs to virulent E. coli O157:H7 are attached through GLD coupling. E. coli in a sample is then bound to the Abs on the column (Figure 5). A "sandwich" immunocomplex is formed by the addition of alkaline phosphatase-labeled anti-E. coli O157:H7 Ab, which binds to the E. coli bound in the previous step. The absorbance of pnitrophenol, the product of the enzyme and *p*-nitrophenyl phosphate reaction, can be monitored at 400 nm. The biosensing approach is capable of detecting *E. coli* in the range of $10^2 - 10^6$ cfu mL⁻¹ with the total assay being completed within 1.5 h. Its selectivity was confirmed by testing with other microorganisms such as S. typhimurium, L. monocytogenes, and C. jejuni.¹⁹

An optical biosensor was designed for the label-free detection of *E. coli* O7:K1 cells using laser transmission intensity measurements. In brief, specific Abs for *E. coli* are coupled to the wells of a CodeLink activated glass slide, followed by binding of the *E. coli* cells (Figure 6). Detection is accomplished using a laser diode (635 nm), and the transmission intensity difference between the reference (only Ab) and the sample is calculated. The assay system detects as few as 45 cells within 1 h.¹⁹⁶

In another label-free system, an optical biosensor based on FTIR spectroscopy is useful for the detection of pathogenic E. coli. Titania thin films prepared by the sol-gel method are treated with 0.2 M APTES and GLD for the attachment of Abs against E. coli O157:H7. FTIR spectroscopy provides mid-IR fingerprints of the E. coli pathogen in a sample, which could be detected at concentrations as low as 1×10^2 CFU mL⁻¹ within 30 min. Specificity of the optical biosensor was validated by using E. coli K12 as a control, which did not give a response.¹⁹⁷ For the label-free detection of phage M13K07,45 a silicon wafer is sequentially functionalized with APTES, activated with GLD, and coupled to avidin. Thereafter, mouse monoclonal biotinanti-M13 Ab is immobilized, which detects phage M13K07 based on the increase in thickness of the phage M13K07 layer corresponding to the increase in its concentration from $0.1 \times$ 10^{10} to 2.5×10^{10} pfu mL⁻¹. The developed optical chip exhibits a sensitivity of 10^9 pfu mL⁻¹.

Another successful application is the detection of *Salmonella* using a direct binding optical grating coupler (OGC) IMS based on optical waveguide light mode spectroscopy.³⁸ The procedure involves the rotation of an OGC sensor chip placed in a flow-through cell in a defined angle by a goniometer and the measurement of the photon signals diffracted and propagated inside the waveguide film using two photodiode detectors. The anti-*Salmonella* Ab is bound to the APTES-functionalized sensor chip by GLD. With a LOD of 1.3×10^3 CFU mL⁻¹ for *Salmonella*, the sensor also displays high selectivity for *Salmonella* and can be reused for 10 repeated runs with a coefficient of variability of 2.87% after regenerating the Ab-bound chip with 10 mM HCl.

4.3. Microfluidic/Lab-on-a-Chip Applications

Manipulation of liquid inside a microscale fluidic network reduces consumption of reagents and allows facile reactions between analyte and biomolecule, resulting in reduced assay times. Therefore, IAs using a microfluidic format are designed for rapid and sensitive detection of important antigens in clinical diagnostics. Glass and silicon-based microfluidic devices equipped with SPR, Raman spectroscopy, and electrochemical analysis have been developed extensively. For microfluidic IAs, Ab or antigen is immobilized inside microchannels. Silicon and glass share similar surface chemistry; thus the route to biomolecule immobilization is similar and APTES has been widely used for surface coatings. For example, a chemiluminescence IA in glass capillaries was developed for the detection of MYO, CK-MB, troponin I (TnI), and fatty acid binding protein (FABP) in human plasma.¹⁶ The capillaries were functionalized with APTES and subsequently bound to capture Abs using ultrasound waves to promote diffusion of materials. This significantly improved the silanization and Ab immobilization leading to increased detection response and sensitivity. The LODs for MYO, CK-MB, TnI, and FABP in human plasma are 1.2, 0.6, 5.6, and 4 ng mL⁻¹, respectively. The IA is much better than the sandwich ELISA performed on Maxisorb MTP.

An integrated microfluidic device was developed for solid-phase sandwich IA, based on microparticle labeling (Figure 7).¹⁹⁸

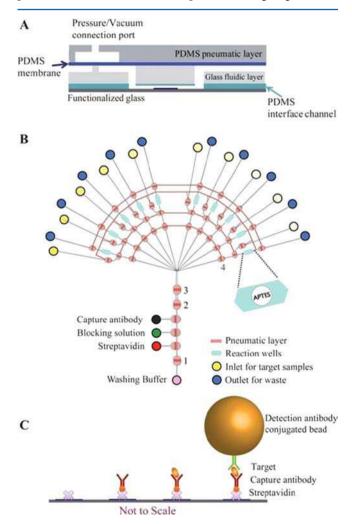


Figure 7. Integrated microfluidic bioprocessor for solid phase capture IAs: (A) cross section of the reaction chamber where PDMS is polydimethylsiloxane; (B) integrated microfluidic system that uses monolithic membrane valves to transport reagents and washing buffer to the reaction chambers; and (C) schematic of the IA consisting of streptavidin, biotinylated capture Ab, target molecules, and detection beads that are pumped to the reaction chamber to form a total immune-complex on patterned APTES. Reproduced with permission from ref 198. Copyright 2011 Royal Society of Chemistry.

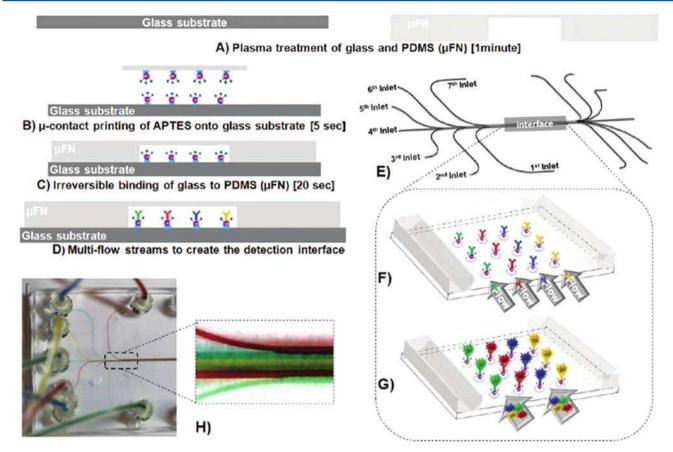


Figure 8. Patterning multiplex protein microarrays in a single microfluidic channel. (A) Plasma treatment of the glass and PDMS substrates. (B) Microcontact printing of APTES onto glass substrate. (C) Forming the irreversibly sealed microfluidic device. (D) Covalent binding of biomolecules using EDC-NHS chemistry. (E) Microfluidic design. (F) Multiple laminar flow streams carrying individual specific biomarkers to create multiplex protein patterns along the main channel. (G) Schematic of a mixture of secondary Abs specific to the primary Abs for a high throughput biomarker assay. (H) Magnified multiple laminar flows in the main channel. Reprinted with permission from ref 201. Copyright 2011 American Chemical Society.

It employs programmable microvalve-control structures for fully automated sample processing. The APTES-patterned glass surface was sequentially bound to streptavidin, biotinylated capture Ab, analyte, and detection Ab-conjugated magnetic beads. The integrated microfluidic device was used for two sandwich IAs, that is, for mouse IgG and prostate-specific antigen (PSA), with LODs of 1.8 and 3 pM, respectively.

PDMS is a silicon-based organic polymer with widespread uses because of its low cost and ease of fabrication.¹⁹⁹ This rubber-like flexible polymer (i.e., elastomer) is also transparent to enable optical imaging. PDMS is hydrophobic in native form, so nonspecific protein is problematic and it lacks functional groups for covalent derivatization. Silanol groups can be introduced by oxygen plasma treatment, but these groups only have limited long-term stability.²⁰⁰ Nevertheless, the patterning of five different Ab arrays in a single microfluidic channel was demonstrated using microcontact printing for APTES-functionalization and Ab immobilization.²⁰¹ The Abs were activated by EDC-NHS and then immobilized on APTES-patterned glass substrate by microcontact printing using PDMS stamps (Figure 8). The multiplex Ab arrays maintained their functionality even at high flow rates and remained stable at sheer stresses of up to 50 dyn cm^{-2} . The functionality was tested by employing a mixture of fluorescent-labeled secondary Abs that were specific to the primary Abs bound to the glass substrate.

Other plastic substrates such as PMMA, polystyrene (PS), and cyclic olefin copolymer (COC) exhibit optical transparency, high

chemical resistance, and low autofluorescence as compared to PDMS. Such materials have gained attention in microfluidics due to low cost of fabrication using injection molding or hot embossing. The surfaces of PS, COC, and PMMA lack ligands for covalent linkage; therefore, various methods have been attempted for surface modification. Besides common oxygenplasma treatment, of interest is to coat the polymer surface with an intermediate polymer layer that has an attractive interaction with biomolecules.²⁰² Another important point is limited biomolecule density immobilized on the microchannel planar surface. Several strategies using 3D structures have been employed inside microfluidic channels for higher biomolecule capture capacity. They include patterning microstructures (e.g., microposts^{203,204} and micropits,²⁰⁵ insertion of porous mem-branes^{206,207}) before the assembly of microfluidic chips. Microbeads,^{208,209} sol–gels,²¹⁰ hydrogels,^{211,212} polymer mono-liths,²¹³ or in situ polymerized membranes²¹⁴ can be packed into enclosed channels to create 3D structures. In general, for silicabased 3D structures such as silica beads and alkoxysilane-based sol-gels, a similar glass/silicon surface immobilization strategy with APTES can be used for biomolecule immobilization.

4.4. Microgravimetric Biosensors

The silanized platforms have been widely employed in microgravimetric biosensors, such as those employing microcantilevers and QCM, for the immobilization of Abs. A millimeter-sized lead zirconium titanate (PZT) glass cantilever was employed for the detection of *E. coli* O157:H7.²¹⁵ The monoclonal anti-*E. coli* O157:H7 Ab was covalently bound to the APTES-functionalized cantilever glass tip by EDC-SNHS. *E. coli* O157:H7 was detected on the basis of the reduction in the resonant frequency of cantilever after the formation of immune complex. The LOD of the developed cantilever biosensor was 700 cells mL^{-1} .

The label-free detection of femtogram levels of biologics was demonstrated in flowing liquid samples using a PZT microcantilever-based IMS.²¹⁶ The cantilevers were functionalized with APTES, with Abs immobilized by the EDC-SNHS cross-linking strategy. It detected very low concentrations of pathogen (E. coli O157:H7, 10 cells mL⁻¹) and protein (BSA, 100 fg mL⁻¹) in about 10 min. The developed IMS was highly selective for E. coli as it detected 15-60 cells mL⁻¹ in a background of 10⁴ mL⁻¹ naturally occurring bacteria. The mass sensitivities of the sensor for BSA and *E. coli* were 0.8 ± 0.2 ag Hz⁻¹ and 1.5 ± 0.3 fg Hz⁻¹. Similarly, a PZT-anchored piezoelectric excited millimeter-sized cantilever (PAEPEMC) IMS was employed for the measurement of Bacillus anthracis (BA) spores (333 spores mL⁻¹).²¹⁷ The results showed that anti-BA Ab functionalized PAEPEMC IMS was highly selective to BA spores. The presence of other Bacillus species did not prevent the binding, but hindered the transport of BA to the sensor.

A Love-mode surface acoustic wave (SAW) IMS was employed for the real-time detection of hepatitis B surface Abs (HBsAbs) in whole blood samples.⁴⁴ The sensor was composed of a SiO₂ guiding layer deposited on a 36° YX LiTaO₂ piezoelectric single crystal substrate that protects the electrodes and traps acoustic energy near the surface. The hepatitis B surface antigen (HBsAg) was bound to the APTES-functionalized interdigitated transducer (IDT) electrodes by GLD crosslinking. The assay detects 10 pg mL⁻¹ to 10 ng mL⁻¹ of HBsAb with mass sensitivity of 0.74 Hz (pg μ L⁻¹)⁻¹. It is applicable for the detection of HBsAb in whole blood samples from vaccinated patients, but not completely immune to hepatitis B virus (HBV), and fully vaccinated counterparts.

4.5. Electrochemical Biosensors

Different strategies using APTES have been applied for the detection of glucose, ^{111,112,173–175,218–225} putrescine, ^{226,227} xanthine, ²²⁸ uric acid, ²²⁹ amino acids, ²³⁰ and cancer markers. ^{231,232} Various electrode materials may influence critical factors for increased biosensor detection limits such as the efficiency of direct electron transfer, molecular recognition, enzyme matrix distribution, or stability. The application of APTES in layer-by-layer (LbL) and SAM, ^{111,112,218–222,233} sol–gel, ^{173,174} or electrodeposited systems^{223–225,234} resulted in different 2D and 3D structures that significantly improve the performance of electrochemical glucose biosensors. The most commonly used APTES-based electrochemical sensing techniques are steady-state amperometry at a fixed detecting potential in mediatorless sensing, ^{112,174,175,219,220,224,226–230} DPV in redox detection of ferrocene-tagged Ab, ²³² and electrochemical impedance spectroscopy.^{230,231}

4.5.1. Modification of Pristine Electrodes. Besides Au and Pt, graphite is commonly used due to its good electrochemical properties, ease-of-handling and low cost, and high surface area. All metals apart from the noble metals are covered by a native oxide layer with a thickness of only several nanometers under atmospheric conditions. These passive oxide layers are terminated by hydroxyl groups and covered with an adsorbed water layer to minimize the surface potential. The presence of oxides consisting of hydroxyl, phenolic, carbonyl, carboxylate,

etc., on electrode surfaces is well recognized. The distribution of oxide types is dependent upon the electrode material, its treatment, exposure to air or electrolytes, etc. Various procedures can be used to modify the oxide thickness, composition, and stability. The resulting structure of the obtained oxide layer is dependent on the method of production, such as thermal oxidation, plasma oxidation, physical and CVD, atomic layer deposition, magnetron sputtering, laser ablation, hydrothermal synthesis, and electrochemical anodization. Selective enrichment of surface hydroxyls is reported for KOH treatment of a vacuum heat-treated surface, while surface carboxyl groups can be selectively enriched by hot HNO3.²³⁵ Freshly prepared borondoped diamond (BDD) electrodes are hydrogen terminated,²³⁶⁻²³⁹ which are converted to oxygen-containing functional groups (i.e., carboxyl, carbonyl, and hydroxyl groups) upon anodic polarization.²⁴⁰ The immobilization of enzymes with APTES on bare electrodes has been performed on GCE and carbon surfaces,^{111,174} noble metals (Au and Pt),^{36,175,222} HOPG,¹⁷⁴ and indium tin oxide (ITO) surfaces.^{230–232,241}

A conductivity-based IMS detects methamphetamine (MA) in human urine in just 15 min with linearity of $1-10 \,\mu g \, mL^{-1}$ and a minimum detectable concentration of 0.5 $\mu g \, mL^{-1.36}$ The procedure involves the binding of anti-MA Ab on APTESfunctionalized Pt electrodes using GLD. Therefore, it is applicable for detecting the critical concentration of MA ($1 \,\mu g \, mL^{-1}$) in human urine with high selectivity. The result obtained agrees well with that of gas chromatography—mass spectrometry, a time-consuming and complex analytical procedure.

Noteworthy is the detection of atrazine in the range of $50-700 \ \mu g \ mL^{-1}$ using a niobium/niobium oxide (Nb/NbOxHy) modified electrode.⁵⁰ The F_{ab} fragment k47 Ab against atrazine is covalently attached to the APTES-functionalized electrode via GLD. In this EIS scheme, a hexane-cyanoferrate(II)/(III) mixture serves as a redox probe with a cathodic polarization voltage of -1.2 V vs saturated calomel electrode (SCE). The mediatorless GOx-bound GCE is applicable for the detection of glucose over the entire pathophysiological glucose range in diabetics (1.3–28.2 mM) and eliminates any interactions between a mediator and electroactive interfering substances for precise blood glucose monitoring.

Optically transparent, extremely smooth, and electrically conducting ITO surfaces have found their application in optical, electroluminescence, and electrochemical biosensing. Despite low electrocatalytic activities, ITO electrodes have a very low and flat background current, which is favorable to noise reduction in an electrochemical biosensor. A SAM of APTES is used to modify the surface of an ITO microelectrode array for subsequent immobilization of uricase using the cross-linker bis(sulfosuccinimidyl)suberate.²²⁹ Figure 9 shows the AFM micrographs of ITO surfaces on glass and the change in roughness of the surface upon immobilization of uricase. The LOD for uric acid is excellent (8.4 μ M); however, this approach is very vulnerable to endogenous electroactive interferences in biological samples.

For the first time, anti-HER-3 Ab was integrated in a disposable IMS using the ITO substrate for detecting HER-3, which is a transmembrane growth factor receptor of the human epidermal growth factor receptor that is associated with cancer cells.²³¹ Figure 10 shows the fabrication of the IMS as bare ITO surfaces are pretreated with a mixture of H_2O_2 and NH_4OH , prior to functionalization with APTES and anti-HER-3 immobilization and BSA blocking. HER-3 down to the fg mL⁻¹ level can be detected using an electrochemical impedance

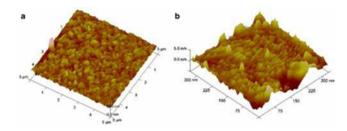


Figure 9. Contact-mode AFM images of (a) APTES/indium tin oxide microelectrode array (ITO- μ EA)/glass and (b) uricase/BS3/APTES/ITO- μ EA/glass. Reproduced with permission from ref 229. Copyright 2013 Springer.

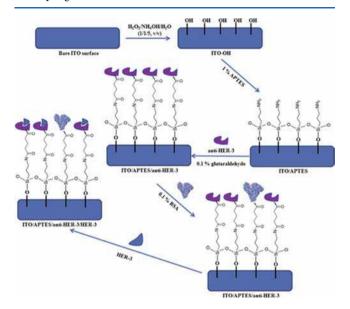


Figure 10. Schematic for IMS using anti-HER-3 Ab immobilized on an indium tin oxide (ITO) surface for HER-3, a type of transmembrane growth factor receptor of the human epidermal growth factor receptor associated with cancer cells. Reproduced with permission from ref 231. Copyright 2014 Elsevier B.V.

spectroscopy technique. However, attempts to monitor and reveal binding characteristics between HER-3 and anti-HER-3 using single frequency impedance are not conclusive due to interferences at a given single frequency.

Two tumor biomarkers, CEA and AFP, can be detected simultaneously using two ITO electrodes (1 mm diameter) patterned on a glass substrate at a spatial separation of 2.5 mm.⁴¹ Again, the electrode surface is functionalized with APTES for subsequent immobilization of their corresponding Abs via GLD cross-linking. The biomarkers are detected by electrochemical ELISA using streptavidin-labeled alkaline phosphatase. The response signal from electrochemical oxidation of alkaline phosphatase-generated hydroquinone provides a dynamic range of 25–150 and 10–200 ng mL⁻¹ for CEA and AFP detection with corresponding LODs of 1.2 and 1 ng mL⁻¹.

Electrochemical detection of goat IgG with a dynamic range of 118 fg mL⁻¹ to 1.18 ng mL⁻¹ is feasible using microchips with a dual-ring working and counter electrodes and a sensing cavity chamber fabricated on glass slides.⁴³ Goat IgG attached to APTES-functionalized glass using GLD binds to alkaline phosphatase (ALP)-conjugated antigoat IgG. Thereafter, goat IgG concentrations in samples can be determined by competitive ELISA using *p*-aminophenyl phosphate (PAPP) as an enzyme

substrate. The enzymatic conversion of PAPP to *p*-aminophenol (PAP) is analyzed by differential pulse voltammetry (DPV).

An electrochemical immunochip sensor involving the binding of antiaflatoxin M1 Abs to an APTES-functionalized Au microelectrode array (MEA) is useful for the detection of aflatoxin M₁ in milk.²⁴² The detection scheme is based on competitive ELISA using TMB/H₂O₂ and HRP as the enzyme label. The sensor exhibits a dynamic detection range of 10–100 ng mL⁻¹ and a LOD of 8 ng mL⁻¹, significantly lower than the legislative limit of 50 ng mL⁻¹. Similarly, anti-SEB Ab bound to the APTES-functionalized nanoporous aluminum surface via GLD activation has been described for detection of *Staphylococcus* enterotoxin B (SEB) as low as 10 pg mL⁻¹ in 15 min.²⁴³ The measurement is based on time-resolved electrochemical impedance spectroscopy (EIS).

SPR Au chips can be reused by treatment with 12 M HCl for 10 min and O₂-plasma scrapping for 5 min.²⁴⁴ In acidic milieu, hydroxyl (OH⁻), sulfate (SO₄²⁻), and chloride (Cl⁻) generated by the dissolution of siloxane bonds destabilize the siloxane framework in the presence of water.²⁴⁵ However, the acidic environment also favors the polymerization of silanes, resulting in incomplete regeneration. Therefore, a second treatment step involving O₂-plasma scrapping is mandatory for complete regeneration of the Au surface.

APTES has been used to modify a planar BDD electrode, pretreated with KOH to serve as a biosensing platform for biomolecule immobilization with GOx to achieve direct electron transfer (DET).²⁴⁶ The amino groups (mainly from lysine) of GOx and APTES were cross-linked by GLD to form a stable enzyme layer on the BDD surface. Cyclic voltammetry (CV) of GOx immobilized shows a pair of well-defined redox peaks in neutral phosphate buffer solution, corresponding to DET of FAD/FADH₂-GOx. This simple scheme for the achievement of DET is favorably compared to other complex sensing schemes involving the use of nanomaterials, redox polymers, and nanowires with respect to simplicity and reproducibility.

4.5.2. Film-Modified Electrodes. The benefits of using APTES in LbL deposition of a four-layered membrane are demonstrated in the development of an amperometric glucose biosensor consisting of APTES, Nafion, GOx, and perfluorocarbon polymer (PFCP) on a single-chip electrode set.²²¹ The PFCP layer acts as a glucose diffusion limiting membrane, while the Nafion layer prevents the diffusion of potential interference species. The biosensor exhibits a dynamic range of 2.8–167 mM, sensitivity of 2.21 nA mM⁻¹, and a response time of 15 s. Electroactive ascorbic acid, uric acid, and p-acetaminophen provoke no signal response, and the results obtained by the glucose biosensor agree well with those of the glucose DHD reference method-based analyzer, commonly used for the determination of diabetic urine glucose concentrations. Of interest also is the construction of a reference electrode in amperometric glucose sensing based on such PFCP properties.²²² A quasi-Ag/AgCl electrode coated with APTES and PFCP is invulnerable to interference species and provides adequate Nernst response under controlled ionic strength conditions. Unlike the standard Ag/AgCl reference electrode, this quasi-Ag/AgCl reference electrode can be easily integrated in miniaturized electrode arrays, each consisting of a working electrode, a reference electrode, and a counter electrode.

APTES-chitosan (CS) hybrid gel film was electrodeposited on Au or platinized Au (Pt-nano/Au) electrodes for the immobilization of GOx.²²⁴ The platinized Au electrode with an enhanced surface area displays broader linearity for glucose detection. In such a study, an electrochemical QCM is used to monitor GOx–APTES–CS electrodeposition for optimized biosensor construction. A sensor based on the nonfaradic process, employing only deionized water as electrolyte, is useful for the detection of dengue infection.⁴⁸ The IMS, precoated with a sol–gel derived barium strontium titanate (BST) thin film, was functionalized with APTES and GLD for immobilization of preinactivated dengue virus. The detection of dengue Ab molecules in human serum is monitored by the impedance or current change. The same concept was extended for the detection of human IgG with a LOD of 40 ng mL⁻¹ using a sol–gel derived amorphous BST thin film and a highly diluted NaCl solution as electrolyte.²⁴⁷

4.5.3. Graphene and CNT Modification. The use of graphene and CNTs can enhance the electrochemical active surface area and electroconducting properties of the modifying layer.^{112,219–221,227–229,231} The graphene-functionalized GCE²¹⁸ and graphene/MWCNT-functionalized GCE¹¹² have been employed for mediatorless electrochemical glucose sensing. While both GCEs can detect the entire pathophysiological glucose range of blood glucose, the graphene/MWCNT-functionalized GCE exhibits a 4-fold higher current signal due to DET. GCEs and carbon-fiber microelectrodes with MWCNTs are fabricated into a high-performance electrochemical glucose microbiosensor with a detection limit of 25 μ M, displaying a significant decrease in the diffusion current as compared to the bare electrode. The presence of APTES induces solubilization of MWCNTs and serves as an immobilization matrix for GOx.

Xanthine oxidase is immobilized on Fe₃O₄ NPs modified with APTES/CNTs for the determination of xanthine.²²⁸ This magnetic, disposable Au screen-printed electrode exhibits a remarkable LOD of 60 nM for xanthine. However, this sensor exhibits a narrow linear range (0.25–3.5 μ M), which can be attributed to a relatively low enzyme loading and low reproducibility. More importantly, the electrode exhibits high reactivity toward electroactive uric and ascorbic acids. The inability to avoid detection of interfering molecules is a major challenge in developing any new electrode materials for biosensing applications.

Putrescine is associated with cancer markers,^{248,249} plant stress markers,²⁵⁰ bacterial vaginosis,²⁵¹ and food spoilage.²⁵² The GCE fabricated with MWCNTs, Nafion, putrescine oxidase (POx), and APTES offers a fast response time of 10 s and a detection limit of 500 nM.²²⁶ Again, APTES helps in the solubilization of MWCNTs in solution with EtOH and Nafion prior to casting on the GCE. MWCNTs can also be dispersed in poly(diallyldimethylammonium) chloride (PDDA) for the construction of an improved putrescine biosensor with greater sensitivity and selectivity.²²⁷ The level of putrescine in mouse plasma detected by the biosensor agrees well with the standard high performance liquid chromatography (HPLC) method.

The enantioselective discrimination of chiral, biologically active synthetic and natural compounds by APTES-modified ITO electrodes is an interesting research subject.²³⁰ Among the amino acids, tryptophan (Trp) as a precursor of melatonin and serotonin has a very important role in mental health. In this context, a modified ITO electrode (ITO/APTES/GO/HSA) using human serum albumin (HSA) differentiates D- and L-Trp enantiomers as exemplified by two well-defined irreversible oxidation peaks at +0.86 and +1.26 V. The modification of the ITO surface by subsequent addition of APTES and GO increases the polarization window, enabling the differentiation of the two tryptophan enantiomers. Besides its long-term stability of only 10 days, the interfering effect of other amino acids or endogenous compounds remains to be investigated, a subject of future endeavor. Nevertheless, this approach offers several advantages such as easiness of preparation, accuracy, and low material cost.

Ultrasensitive detection of AFP²⁵³ involves four sequential steps: (i) coating a graphene-CdS QDs-agarose composite on GCE, (ii) APTES-functionalization, (iii) activation with GLD, and (iv) binding of anti-AFP Ab (Figure 11). The detection of

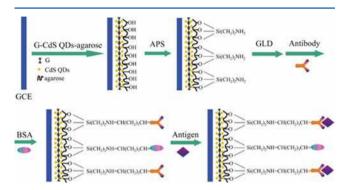


Figure 11. Fabrication of electrochemiluminescence IMS based on graphene (G)-CdS QD-agarose composite for the ultrasensitive detection of alpha fetoprotein; glassy carbon electrode (GCE), 3-aminopropyltriethoxysilane (APS), and GLD. Reproduced with permission from ref 253. Copyright 2012 Elsevier B.V.

AFP is followed by a decrease in electrochemiluminescence intensity after the immune complex formation with a linear range and a LOD of 0.0005-50 pg mL⁻¹ and 0.2 fg mL⁻¹, respectively. It is highly specific for AFP, as demonstrated by the detection of AFP in the presence of several nonspecific proteins. The assay is applicable for the detection of AFP in serum and saliva samples of patients with recoveries ranging from 89% to 100%. Of interest is the use of multilayer epitaxial graphene (MEG) for the detection of human chorionic gonadotropin (hCG).²⁵⁴ MEG was grown on silicon carbide substrates and modified by the Fenton reaction to yield a hydroxylated MEG layer. It was functionalized with APTES and then bound to anti-hCG Ab by EDC-NHS chemistry to provide a LOD of 0.62 ng mL⁻¹ and a linear range of 0.62–5.62 ng mL⁻¹.

4.6. Nanotechnology-Based Applications

Emerging nanomaterials play an important role in biosensors and diagnostics by offering a high surface area, functionalized platform for increased biomolecular loading, and resulting in low LOD for bioassays through signal amplification associated with their optical/electrical properties. Fe₃O₄ NPs with super paramagnetic properties can be easily separated from the reaction medium to facilitate product isolation/purification and the reuse of biocatalysts. Au,²⁵⁵ Fe,²⁵⁶ Si,²⁵⁷ and Zn NPs,²⁵⁸ which can be synthesized to precise dimensions and are fairly inexpensive to produce, are attached via APTES as part of covalent attachment systems for enzymes/proteins. Despite the numerous advantages of using nanomaterials for in vitro applications, their use in in vivo application is limited due to toxic effects in the human body. The incompatibility of NPs inside the body can lead to metal agglomeration that will significantly decrease diagnostic sensitivity.²⁵⁹

4.6.1. Gold. Au NPs with different nanoscale sizes can easily be dispersed in water and functionalized with enzymes with appropriate cross-linking agents. An IMS for progastrin releasing-peptide (ProGRP), a biomarker for small cell lung cancer, uses glucose oxidation as a means to detect other

biomolecules. Modifying the GCE with Nafion-cysteinegraphene sheets followed by Au NPs increases the surface area to enable higher loading density for anti-GRP.²⁶⁰ GOx and ferrocene-labeled secondary Abs are conjugated onto APTESfunctionalized Au-TiO₂ NPs to produce a tracer (Figure 12). In

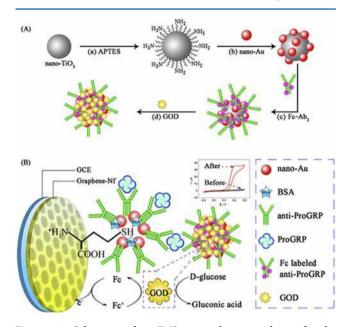


Figure 12. Schematic of an IMS using glucose oxidation for the detection of progastrin releasing-peptide (ProGRP). The IMS consists of GCE with Nafion-cysteine-graphene sheets loaded with anti-GRP conjugated on Au NPs. GOx and ferrocene-labeled secondary antibodies (Fc-Ab₂) are conjugated onto APTES-functionalized Au-TiO₂ NPs to produce a tracer; bovine serum albumin (BSA) and Nafion (Nf). Reproduced with permission from ref 260. Copyright 2011 Elsevier B.V.

the presence of ProGRP, a sandwich complex is obtained between the GCE and the tracer. Exposure to glucose results in oxidation of glucose, which can be measured by CV, giving a quantitative measurement of ProGRP with a LOD of 3.0 pg mL⁻¹.

An Au/ITO electrode suited for the detection of vascular endothelial growth factors (VEGF) offers a LOD of 100 pg mL^{-1} .²³² The modified sandwich-type electrochemical IA platform takes advantage of the ITO electrodes low background

current with the high electrocatalytic activity of Au NPs. VEGF Abs are cleaved into two half-fragments by 2-mercaptoethylamine—HCl and immobilized onto the Au NPs. VEGF target molecules bind to the Ab fragments to form antigen—Ab complexes (Figure 13). Subsequently, ferrocene-tagged Abs react with these complexes and release electrons at an appropriate applied potential for detection by DPV.

The rapid detection of gibberellic acid (GA) in rice grain uses the coupling of electrochemical IMS with square wave anodic stripping voltammetry (SWASV) involving Cu ion-labeled antigen in a competitive immunoreaction.³⁷ The GCE is sequentially functionalized with APTES, Au NPs, anti-GA Ab, and milk. For the detection of GA, the IMS exhibits a linear range of $1-150 \ \mu g \ mL^{-1}$ and provides reliable results as corroborated by HPLC and ELISA.

Signal enhancement with Au NPs is demonstrated in the design of evanescent field coupled waveguide mode sensors.²⁶¹ Au NP-conjugated streptavidin is immobilized onto silica-based sensor chips functionalized with different surface chemistries to investigate the effects of surface roughness induced by such functionalization. APTES/GLD-modified surfaces are suitable for small molecules with diameters ~5 nm due to the resulting surface roughness, whereas carbonyldiimidazole-modified surfaces are a better choice for the direct immobilization of biomolecular assemblies without any intermediate reactions.

An inexpensive and disposable impedimetric IMS uses stainless steel electrodes modified with APTES and electrodeposited Au NPs for the detection of doxorubicin.²⁶² Au NPs enhance the conductivity and sensitivity of the electrode and allow for the passive adsorption of doxorubicin-specific monoclonal Abs onto the electrode surface. The sensor detects doxorubicin concentrations in two ranges, that is, 2.5–30 pg mL⁻¹ and 30–1000 pg mL⁻¹, with a LOD of 1.7 pg mL⁻¹. The analysis of spiked human serum samples shows recoveries in the range of 88–122.2%.

4.6.2. Iron. Fe_3O_4 NPs are a popular substrate for immobilization of enzymes because enzymes can be separated and reused for subsequent analysis with the aid of a magnetic field. The abundance of oxide groups on the iron surface enables APTES to bind efficiently to the NP to effect cross-linking. Several studies use Fe_3O_4 NPs as means to study the binding properties of Abs in relation to particle size and reaction conditions. Fe_3O_4 NPs with different coating agents can be used to establish the optimal conditions for the binding of goat IgG to the NP surface in which complementary fluorescein isothiocya-

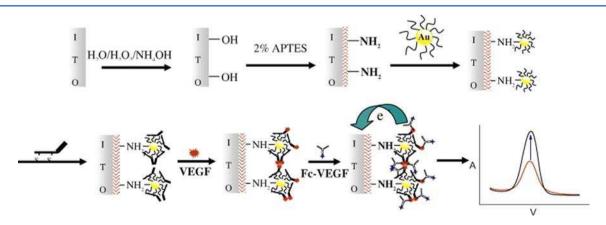


Figure 13. Schematic for the fabrication of an indium tin oxide (ITO) IMS for vascular endothelial growth factors (VEGF) in which ferrocene-tagged Abs release electrons under an applied voltage for detection by DPV. Reproduced with permission from ref 232. Copyright 2010 Elsevier B.V.

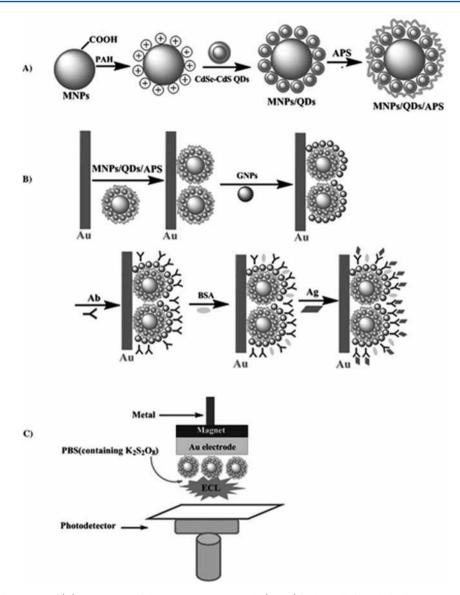


Figure 14. Schematic illustration of (A) preparation of the magnetic nanoparticle (MNP)/CdSe-CdS/APS hybrid nanostructure, (B) fabrication of the electrochemiluminescence (ECL) IMS, and (C) ECL detection by the IMS; poly(allylamine hydrochloride) (PAH), 3-aminopropyltriethoxysilane (APS), QD, gold nanoparticle (GNP), and phosphate buffered saline (PBS). Reproduced with permission from ref 268. Copyright 2011 Wiley-VCH Verlag GmbH & Co. KGaA.

nate (FITC) labeled antigoat IgG is used to probe the presence of surface immobilized goat IgG by fluorescence microscopy.²⁶³ APTES functionalization is the most optimal immobilization technique for IgG on Fe₃O₄ NPs.

Two studies investigated the link between the number of Abs bound to Fe_3O_4 NPs and activity. The immobilization of IgG Abs on Fe_3O_4 NPs with an average size of 32 nm results in about 36% of the theoretical limit for surface coverage, corresponding to 34 bound Ab molecules, as determined by quantitative fluorescence assays.²⁶⁴ In the second study, 17 molecules of serratiopeptidase per Fe_3O_4 NPs (average size 13 nm) with 67% enzymatic activity is determined by mass measurement. The immobilized enzyme enhances the anti-inflammatory effect on carrageenan-induced paw edema in rats at much lower doses than the free enzyme.²⁶⁵ The decreased enzymatic activity could be attributed to covalent attachment of amino groups of APTES to the catalytic sites of the enzyme or changes in conformation of the native enzyme structure by overcrowding effects.²⁶⁶ A similar phenomenon is noted for GOx immobilized on different Fe_3O_4 NPs (5, 26, and 51 nm) with APTES and GLD.²⁶⁷ The immobilized enzyme retains only 15-23% of the free enzyme activity. Smaller NPs exhibit greater activity loss over repeated runs due to a poor recovery of the enzyme.

Highly sensitive detection of CEA is feasible using a magnetic electrochemiluminescent (ECL) $Fe_3O_4/CdSe-CdS$ NP/polyelectrolyte nanocomposite.²⁶⁸ The nanocomposite (Fe_3O_4/QDs) is functionalized with APTES via ionic interactions to form $Fe_3O_4/QDs/APTES$, which was then bound to a gold electrode (Figure 14). Au NPs were bound to the surface of $Fe_3O_4/QDs/APTES$ -functionalized electrodes, followed by anti-CEA Ab and BSA immobilization. As the CEA concentration increases, the ECL signal decreases. The detection of CEA occurs in a linear range of 0.064 pg mL⁻¹ to 10 ng mL⁻¹ with a LOD of 0.032 pg mL⁻¹. The determination of CEA spiked in normal serum shows a recovery of 93.1–107.5%, and its results correlate well with conventional ELISA.

A clever use of the magnetic separation property of Fe_3O_4 NPs is demonstrated when trypsin is immobilized on hairy non-cross-

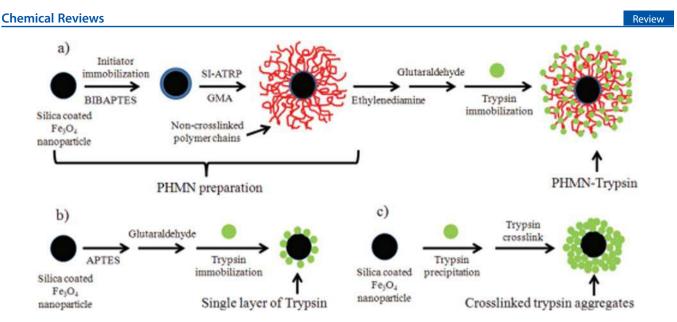


Figure 15. Schematic for the immobilization of trypsin on hairy non-cross-linked polymer chains attached to Fe₃O₄ NPs (PHMN); surface initiated atom transfer radical polymerization (SI-ATRP), 3-(2-bromoisobutyramido)propyl(triethoxy)-silane (BIBAPTES), and glycidyl methacrylate (GMA). Reprinted with permission from ref 269. Copyright 2012 American Chemical Society.

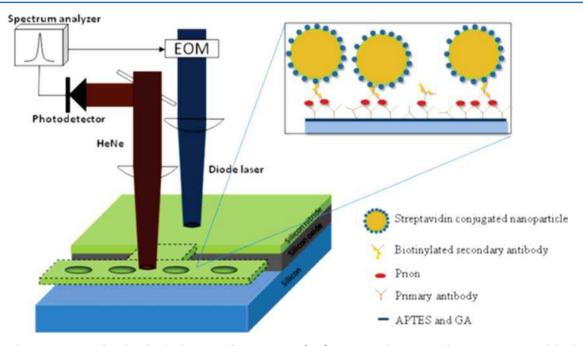


Figure 16. A dynamic resonance-based IA for the detection of protein prions (PrP). Streptavinidin conjugated Fe_3O_4 NPs improved the detection of PrP to 2 ng mL⁻¹; GLD (GA) and electro-optic modulator (EOM). Reprinted with permission from ref 18. Copyright 2008 American Chemical Society.

linked polymer chains with hybrid magnetic NPs (Figure 15).²⁶⁹ The materials facilitate ¹⁸O labeling for protein quantification with H_2 ¹⁸O, with the Fe₃O₄ NPs easily separated from the solution to prevent ¹⁸O to ¹⁶O back-exchange after labeling. The polymer chains provide a large number of binding sites and scaffolds to support 3D trypsin immobilization for increased enzyme density and accessibility over conventional free trypsin digestion in proteomic samples.

A facile and sensitive bioanalytical procedure is demonstrated for the detection of *Salmonella* as low as 10 cfu mL⁻¹ in milk using a personal glucose meter (PGM).²⁷⁰ Magnetic NP clusters are functionalized with APTES and GLD followed by the immobilization of monoclonal anti-*Salmonella* Abs. These clusters then bind to *Salmonella* in milk and are magnetically separated from solution using a magnet. Subsequently, the *Salmonella* complexes are detected by binding to polyclonal Abfunctionalized invertase and dispersing in a 0.5 M sucrose solution at 40 °C. A PGM measures the amount of glucose produced from the hydrolysis of sucrose by invertase.

Fe₃O₄ NPs can improve the detection limits of IAs as well. A dynamic resonance-based technique is used to detect prion proteins (PrP) that cause neurodegenerative diseases when proteins undergo conformational changes.¹⁸ When PrP is immobilized onto a monoclonal Ab that is conjugated onto a resonator's surface via APTES/GLD, LOD increases from 20 to 2 μ g mL⁻¹. Binding streptavidin-functionalized Fe₃O₄ NPs further improves PrP detection to 2 ng mL⁻¹ (Figure 16). In this case, the magnetic properties of the NPs do not play any role as a

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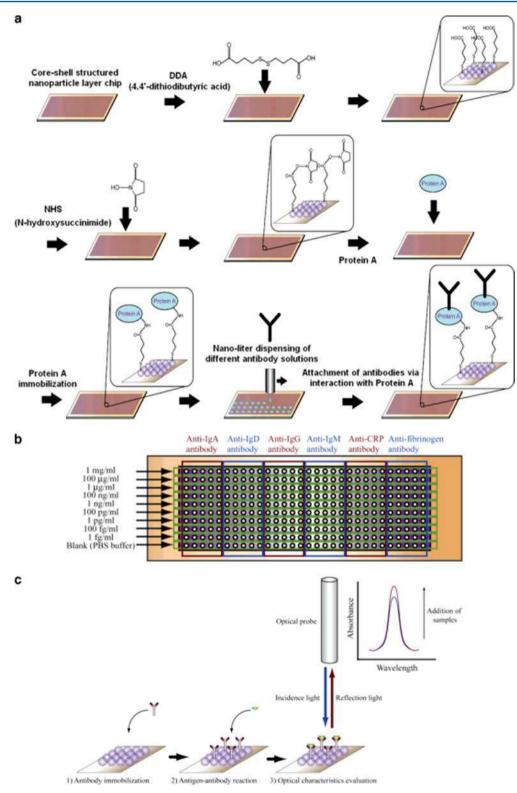


Figure 17. Schematic for the fabrication of a label free localized surface plasmon resonance (LSPR)-based bioanalysis method for developing multiarray optical nanochips: (a) Immobilization of Abs on the surface of the nanochip. (b) Six kinds of Abs and antigens were spotted onto the surface at different concentration. (c) Experimental setup for the multiarray LSPR-based nanochip. Immunoglobulin A (IgA), immunoglobulin D (IgD), immunoglobulin G (IgG), immunoglobulin M (IgM), C-reactive protein (CRP), and phosphate buffered saline (PBS). Reprinted with permission from ref 273. Copyright 2006 American Chemical Society.

separation technique, but amplify the signal given from optical frequency measurement. A magnetic ELISA to detect multiresidue organophosphorus pesticides uses Abs immobilized on Fe_3O_4 NPs coated with APTES.²⁷¹ The abundance of functional groups on Fe_3O_4 NPs enables efficient binding of Abs on the surface of the NP to improve sensitivity and specificity over conventional ELISA.

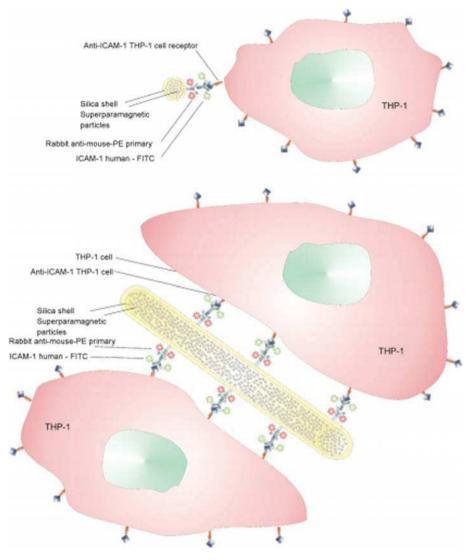


Figure 18. Increased number of binding sites for human acute monocytic leukemia THP-1 monocyte cells and intracellular adhesion molecule (ICAM-1) secondary Abs available along a Silica nanowire in comparison to Silica NPs; phycoerythrin (PE) and fluorescein isothiocyanate (FITC). Reproduced with permission from ref 275. Copyright 2011 Wiley-VCH Verlag GmbH & Co. KGaA.

4.6.3. Silicon. Silicon nanomaterials are functionalized with APTES to introduce amino functionalities for enzyme binding. The hydrolysis of silylethoxy group forms reactive Si-OH groups that condense on the surface of silica to form a lateral siloxane network.⁹¹ The orientation of the amino groups points away from the silicon surface, a favorable orientation for binding enzymes. Nanostructured silicon NP/polycarbonate films can be prepared by immersing polycarbonate in alternating solutions PDDA and 40 nm colloidal silica.²⁷² After functionalization with APTES and GLD, Abs are covalently attached onto the film surface. The efficiency of binding IgG Abs to the APTESfunctionalized silicon NP films is only 33% due to steric effects in binding the larger IgG structure over the smaller APTES molecule. However, the introduction of silicon NPs does increase the surface area of the film, which increases protein loading over native polycarbonate film. A sensitive and quantitative IA for human chorionic gonadotropin is demonstrated with the modified films. Similarly, improved sensitivity and signal amplification associated with increased surface area effects is noticed for a chemiluminescence IA with HRP on

APTES-functionalized mesoporous silicon NPs for the detection of CEA Ab.¹⁸⁰

The deposition layer of APTES on NPs can influence the activity of Abs. Dye-doped silicon NPs can be functionalized with APTES, in which the encapsulated cyanine-based near-infrared dye functions as a probe of the NP surface environment to identify the optimal deposition parameters for the APTES monolayer.⁷¹ An appropriate reaction time is required to form a stable amine layer, but higher reaction times results in disordered cross-linking. By using an IgG plate-based direct binding assay, the optimum APTES monolayer results in the binding of a maximum number of IgG-conjugated silicon NPs to the immobilized anti-IgG Abs.

Silicon nanomaterials have been used to fabricate highly sensitive diagnostic devices. One example is the fabrication of a label free LSPR-based bioanalysis method for developing multiarray optical nanochips (Figure 17).²⁷³ The multiarray LSPR nanochip is constructed with Abs immobilized on 300 nanospots of a sensing surface consisting of APTES-functionalized silicon NPs. The antigen concentration is determined from the peak absorption intensity of the LSPR spectra, providing a

detection limit of 100 pg mL⁻¹. In another example, SiNW field effect transistors (FETs) can be fabricated to be responsive to CRP, a marker of inflammation and possible indicator of future progression of cancer.²⁷⁴ The FETs are connected to an automated fluidic control system where the conjugation of CRP to the anti-CRP (immobilized on the SiNWs via APTES/GLD) results in a change in the detected current.

The effects of nanomaterial aspect ratio can significantly increase the sensitivity of detection. Specific binding between THP-1 monocyte cells and ICAM-1 secondary Ab is measured by flow cytometry for both silicon NPs and nanowires.²⁷⁵ Silica nanowires are better than silica NPs due to the aspect ratio, which can be easily distinguished from NPs when utilized as carriers for immunodetection. A NP forms a single bond to a cell/cell receptor, whereas a wire binds at several points to maximize the functional binding of the nanomaterials along their principal axis (Figure 18). Similarly, GOx is immobilized on SiO₂ nanowires deposited on an Au electrode (Figure 19),²²³ where the SiO₂ nanowires increase the effective surface area resulting in higher enzyme loading, better sensitivity, and lower glucose detection limit (0.6 mM).



Figure 19. Schematic for the immobilization of GOx on SiO_2 nanowires (SiO_2NW) deposited on an Au surface. Reproduced with permission from ref 223. Copyright 2011 Royal Society of Chemistry.

4.6.4. Quantum Dots. QD materials in various morphologies and sizes are applicable for the evaluation of enzyme activity with respect to structural and spectroscopic characteristic of the substrate to which the enzyme is immobilized.²⁷⁶⁻²⁷⁸ Although organic fluorophores such as fluorescein and rhodamine are used extensively for Ab labeling, QDs are beginning to gain greater use in biosensing applications. A major advantage of QDs is its sharply defined emission spectra, which enable a single light source to excite multiple QDs simultaneously without the signal overlap in multiarray applications.²⁷⁹ Strategies to conjugate QDs to biomolecules can range from functionalization of the biomolecule with thiol groups that can bind to the QD surfaces, and electrostatic interactions between negatively charged QDs and positively charged biomolecules.²⁸⁰ Covalent attachment of biomolecules is achieved by functionalizing the QD surface with surfactant molecules to give COOH, NH2, or SH groups. APTES is one such cross-linking agent used in this process.²⁸¹

ZnO nanocrystals with different morphologies can be prepared by varying the ratio of MeOH to water in the hydrothermal synthesis reaction system. Subsequently, these nanocrystals are functionalized with amino groups using APTES and tetraethyl orthosilicate, and then with HRP immobilized via GLD cross-linking.²⁸² 3D ZnO nanomultipods are more appropriate for the enzyme immobilization in comparison to NPs and nanodisks. The increase in activity of the enzymes on multipods could be attributed to less conformation changes in the HRP.

The detection of proteins has been achieved by an Ab-based sandwich assay with ZnO/Au NPs probes.²⁸³ ZnO NPs are considered nontoxic and biosafe materials with distinct Raman signatures. The addition of small amounts of Au NPs to ZnO enhances the ZnO Raman scattering intensity. Initially, an Ab specific for the target protein is bound to the silicon surface. An antigen then binds to the first Ab followed by the addition of a secondary Ab (attached with a ZnO/Au NPs probe), which specifically recognizes and binds to the immune complex formed between the first Ab and the target protein. Good selectivity and sensitivity for the quantitative detection of proteins are facilitated by the amplified Raman scattering signal of the ZnO substrate.

QD-labeled secondary Ab can detect human IgG in human serum using immunoaffinity monolithic capillary microextraction coupled with microcentric nebulizer inductively coupled plasma mass spectrometry (MCN-ICP-MS).⁴⁷ Goat antihuman IgG is immobilized on APTES–silica hybrid monolithic capillary and binds to human IgG in a sample. Subsequently, goat antihuman IgG labeled with CdSe@ZnS QDs is added to bind to the human IgG. The linear detection range is 0.2–10 μ g L⁻¹ (LOD 0.058 μ g L⁻¹) and 0.5–10 μ g L⁻¹ (LOD 0.097 μ g L⁻¹) based on Cd and Se signals.

A simple CdSe/ZnS QD-based optical IMS has been described for the detection of HSA in the range of 2×10^{-1} to 2×10^{-4} mg mL⁻¹ with a LOD of 3.2×10^{-5} mg mL⁻¹.²⁸⁴ The optical system is comprised of a 405 nm diode laser, an optical lens, a Siphotodiode, and a 515 nm long pass filter. The monoclonal anti-HSA Ab is immobilized on APTES-functionalized glass by adsorption followed by blocking with BSA, binding of HSA, and detection using biotinylated polyclonal anti-HSA Ab bound to streptavidin-conjugated CdSe/ZnS QD (Figure 20).

An ECL IMS uses several different nanomaterials previously discussed in this Review to produce an efficient sensor for carbohydrate antigen 125 (CA125) in clinical serum samples.²⁸⁵ Fe₃O₄ NPs are functionalized with APTES followed by reduction of HAuCl₄ to form a core-shell carrier for the Ab of CA125, while graphene loaded with QDs serves as the site of immobilization for secondary Abs. Using microfluidic strategies, the synergistic effects of high surface area and conductivity greatly enhance the ECL signal for a low detection limit (1.2 mU mL^{-1}), with the Fe₃O₄ NPs facilitating the recovery of the IMS. Similar strategies are employed for another ECL immunosenor for the detection of prostate specific antigen (PSA).²⁸⁶ A sandwich-type IMS uses CdTe QDs on silicon NPs, in conjunction with high surface Fe₃O₄ NPs-coated graphene sheets on ITO, to enhance the ECL signal for the detection of PSA down to 0.72 pg mL^{-1} .

CdTe QDs were cross-linked on ZnO using APTES and mercaptopropionic acid as part of a biosensor for the detection of DNA.²⁸⁷ This ZnO composite, showing enhanced photoelectric activity attributed to appropriate band alignment between ZnO and the QDs, is immobilized on an ITO surface with biotinlabeled DNA (Figure 21). When the target DNA and assistant DNA interact with the biotin-labeled DNA, a so-called Yjunction structure is produced, which induces a restriction endonuclease-aided target recycling that results in the release of biotin. As the number of binding sites for HRP-labeled streptavidin decreases due to the loss of biotin, the amount of H₂O₂ consumed by a concurrent HRP-mediated electrochemical reduction will also decrease. With less H_2O_2 consumed, an enhanced photoelectrical response for DNA (LOD = 9.3×10^{-16} M) can be achieved using the available H_2O_2 as an electron acceptor.

4.7. Other Detection Methods and Biomolecule Patterning

A fiber-optic-based evanescent wave IMS was developed for the detection of Newcastle disease virus (NDV) based on a sandwich IA.³³ The capture anti-NDV polyclonal Ab was covalently bound to APTES-functionalized quartz fibers using carbonyldiimidazole

(b)

(c)

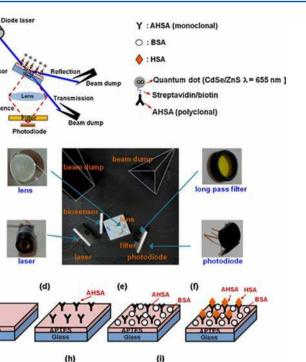


Figure 20. (a) Schematic and (b) real picture of a QD-based optical biosensing system. Schematics of the glass surface modification including (c) APTES deposition; (d) antihuman serum albumin (AHSA) Ab immobilization on APTES; (e) bovine serum albumin (BSA) blocking; and (f) human serum albumin (HAS) immobilization. Schematics of (g) before and (h) after the bonding between QD-streptavidin and biotin-AHSA to form the AHSA-QD complex, and (i) the conjugation of the AHSA-QD complex with the HAS precaptured on the biosensors. Reproduced with permission from ref 284. Copyright 2012 Elsevier B.V.

in dimethyl sulfoxide as the cross-linking agent. The binding of NDV was detected using the fluorescein-labeled anti-NDV as the detection Ab. The developed IMS detected NDV in the dynamic range of 2-250 ng NDV in just 15 min with a LOD of 0.5 ng (5 ng mL⁻¹ for 0.1 mL sample).

The fabrication of rigid Ab-modified FETs and their use for the detection of alpha-fetoprotein (AFP), a tumor cancer biomarker, were also reported.¹⁹ The anti-AFP Abs were immobilized on an APTES-functionalized surface by homobifunctional cross-linking through GLD. The Ab-modified FETs attained increased sensitivity, high reproducibility, and selective analyte detection. The quantitative detection of AFP in the range of 10 ng mL⁻¹ to 1 μ g mL⁻¹ was attained in the presence of other nonspecific proteins.

The biomolecule patterning on analytical devices has also been demonstrated, maintaining their 3D biological functionalization on silicon-based analytical devices.²⁸⁸ It involves the generation of a scaffold on a silicon substrate by the polycondensation of APTES followed by the binding of proteins and nucleic acid sequences, as demonstrated by IA and nucleic acid assays. Similarly, epoxy-functionalized magnetic poly(divinylbenzene-*co*-glycidyl methacrylate) colloidal particles (mPDGs) were prepared by microcontact printing and employed for the

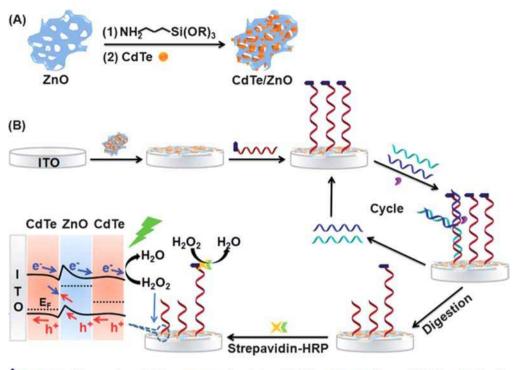
detection of recombinant human interleukin (IL)-10 protein.²⁸⁹ APTES and octadecyltrichlorosilane were patterned on a glass surface by microcontact printing and subsequently bound to prepared mPDGs. The monoclonal antihuman IL-10 Ab was conjugated to the functionalized mPDGs and used for the capture of human IL-10, with the detection performed using fluorescent-labeled antihuman IL-10 Ab.

5. CRITIQUES AND OUTLOOK

Various conditions have been reported for the deposition of APTES on different solid materials, electrode materials, and nanomaterials toward the development of biosensors and diagnostic platforms. Regardless of the choice of the solid substrate, its surface must be conditioned to have sufficient hydroxyl groups and improve the silane grafting density and reproducibility of silanization. Surface treatment steps involving oxidation (such as treatment by piranha, UV light, ozone, and oxygen-plasma) and treatments in KOH/NaOH or even hot water lead to higher silane grafting density as they increase the density of hydroxyl groups on the surface. The cleaning of the surface makes it hydrophilic and hydroxylated as it is covered by water molecules. Therefore, the freshly cleaned surface must be dried to ensure reproducible results. Besides APTES, N-(2aminoethyl)-3-aminopropyltrimethoxysilane is also widely used to modify hydroxylated supports such as silica and glass. Deposition of APTES in water often results in thin layers that initially bind to a negative surface via electrostatic interaction. However, lengthy incubation of a substrate like silicon for several hours results in a thick multilayer. The presence and absence of surface oxide groups changes the reactivity of the APTES. A mixture of neutral amine and protonated and hydrogen-bonded amine coexists on the solid surface; however, the former orients away from the surface and is available for biomolecule immobilization, while the latter is close to the surface. The mechanisms of silanization on surface oxide have not been fully understood because there is no direct analysis (in situ).

Immobilization of a sensing molecule also deserves a brief comment here because several techniques can be used for biomolecule immobilization. Although protonated amine can display electrostatic interaction with the negative moiety of the biomolecule, the immobilized biomolecule is not stable and easily escapes from the APTES bound-surface. GLD activation of amine is popular, followed by NHS ester and carbodiimide. A downside of amine cross-linking is that multiple bonds can be formed due to numerous lysine residues on a protein surface, which leads to conformational changes and/or blocking of the biomolecule's active sites. APTES can be conjugated with a spacer molecule like polyethylene glycol (PEG) or a hydrophilic polymer to reduce steric hindrance and the possibility of biomolecule denaturation.²²⁸ In addition, when the sensing platform involves an Ab molecule or a specific enzyme, affinity immobilization using avidin-biotin interaction might be a better choice to preserve the functional group of the Ab. A typical example is the control of a well-defined 3D nanostructure of HRP via a LbL assembly of avidin and biotin-HRP.²⁹⁰

Both electrostatic and covalent immobilization procedures do not effectively control the orientation of the biomolecule; therefore, many active sites of the immobile biomolecule become inaccessible after immobilization. In contrast, the immobilization of biomolecules using Ab–Protein A/G interaction or DNA hybridization can preserve the biomolecule functionality by the orientated immobilization and increased stability. Protein A or G specifically binds the F_c region of the Ab, and this procedure



Biotin-capture DNA VVVV Assistant DNA VVVV Target DNA Mbo I

Figure 21. Schematic of a biosensor for the detection of DNA assisted by concurrent horseradish peroxidase (HRP)-mediated electrochemical reduction reaction of H₂O₂. Reproduced with permission from ref 287. Copyright 2014 Royal Society of Chemistry.

provides highly improved antigen detection of *Salmonella* over GLD cross-linking.²⁹¹ Nevertheless, GLD or carbodiimide activation has been proven applicable for the immobilization of GOx and other enzymes. This Review focuses on enzymes and Abs in biosensing and diagnostic platforms; however, aptamers can be potential molecular recognition elements because they are relatively inexpensive and more stable than Abs.²⁹² Aptamers are oligonucleotides, which selectively bind to the targeted proteins with high affinity. Nevertheless, Ab still remains as the bioanalytical workhorses due to its exceptional specificity toward its binding partner regardless of the high cost and short shelf life.

The incorporation of nanomaterials into biosensing interfaces offers improved stability, minimal surface fouling, higher sensitivity, cost-effectiveness, and capability for multiplex analysis. Popular examples include Au NPs, CNTs, nanowires, and graphene due to their beneficial contribution in biosensors and diagnostic platforms. Electrodes modified with metal NPs with ultrahigh surface areas can be used to enhance the amount of immobilized biomolecules in the construction of sensors to ultimately lower the detection limit for target analytes. All oxide surfaces can be functionalized with APTES for subsequent bioconjugation. APTES is also useful for the preparation of CNT and graphene suspensions, which can be drop-casted on any surface. Another elegant approach is the preparation of uniform thin coatings of MWCNTs by electrophoretic deposition of MWCNT suspension on APTES-functionalized silicon.²⁹³ It is of importance to design an affinity tag for the reversible binding of biomolecules with high density on carbon-based substrates such as graphene or CNTs. Therefore, such biosensors can be effectively regenerated and reused after the loss of enzyme activity by washing the electrode with a pertinent buffer and providing the activated enzyme. New nanomaterials and nanostructures have been explored for a plethora of analytical

applications. However, more attention must focus on the quantum size effect, surface effect, macro-quantum tunnel effect, and small size effect that are apparent in case of nanomaterials.

Although specific binding between streptavidin has been exploited in various analytical platforms, the binding event is irreversible. Therefore, the biomolecular immobilized platform based on streptavidin—biotin interactions cannot be reused, which leads to higher analysis cost. Of interest is desthiobiotin,²⁹⁴ a biotin analogue with a lower binding constant for streptavidin as compared to biotin. The biomolecular multilayer formed on the surface can be washed away by using biotin solution to pave the way for repeated uses of the APTES-avidin modified surface.

6. CONCLUSIONS

In brief, APTES offers several advantages in analytical applications: (a) easy-to-prepare, (b) stable monolayers, and (c) uniform immobilization surface. APTES also protects the biomolecules from the sensor surface, preventing possible denaturation and nonspecific adsorption. Various surface coatings including Au NPs, polymers (poly(vinyl alcohol), PEG, polyacrylamide, etc.), and biopolymers (dextran, CS, and protein aggregates) help to reduce nonspecific adsorption. Moreover, it is desirable to incorporate multiple functions, that is, a functional group for biomolecule immobilization and an antibiofouling layer for reduced nonspecific protein adsorption in a single copolymer. APTES can be modified for its incorporation in such smart materials, and this is a subject of future endeavors. Improvement of detection limits and the ability to regenerate the sensor are two key prerequisites for the successful deployment of biosensors and diagnostic platforms for diversified applications in analytical, bioanalytical, and clinical chemistry. There is a wide variety of cross-linkers available commercially, which can bind the free amino groups on APTES-

functionalized substrates to the various functional groups, such as carboxyl, amino, sulfhydryl, hydroxyl, and carbonyl, present on the biomolecule. Therefore, in the coming years, we will witness more rapid, efficient, robust, and highly simplified Ab immobilization chemistries, which will significantly improve the bioanalytical performance of biosensors and diagnostics apart from decreasing the cost of analysis.

The APTES immobilization technology is anticipated to be widely applied in electrochemical, SPR-based and electrochemical-QCM-based sensors, and various IA formats, such as those based on smartphone-based detection. Last, it would have potential utility in the development of a highly integrated multiparameter assay, where biomolecules such as Abs, proteins, and aptamers could be coimmobilized into the microfluidic channel for more sophisticated multianalyte assays using multiple detectors, for example, SPR, fluorescence, and electrochemical detectors. We anticipate substantial innovation in this area in the "not-too-long distant" future. Last, the biomoleculebound APTES-functionalized surfaces would be appropriate for in vivo applications based on their high stability, less biofouling, and leach-proof binding of biomolecules. However, this needs to be investigated in more depth in animal models before they can be used for in vivo applications.

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Notes

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Biographies



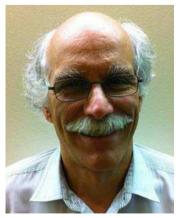
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ABBREVIATIONS

Abs an	ntibodies
AFOB all	l-fiber optofluidics-based platform
AFM at	omic force microscopy
AFP a-	fetoprotein
ALP all	kaline phosphatase
APTES 3-	aminopropyltriethoxysilane
ARXPS an	ngle-resolved X-ray photoelectron spectrosco-
ру	
BA Ba	accillus anthracis
	oron-doped diamond
BMP2 bo	one morphogenetic protein 2
BPA bi	sphenol A
BSA bo	ovine serum albumin
BST ba	arium strontium titanate
BVA 4,4	4-bis(4-hydroxyphenyl) valeric acid
CA125 ca	rbohydrate antigen 125
CBP ch	noline binding protein
CEA ca	rcinoembryonic antigen
CiELISA co	ompetitive enzyme-linked immunosorbent
as	say
CK-MB cr	eatinine kinase MB
CMOS co	omplementary metal—oxide semiconductor
CNBr cy	

CNT	carbon nanotubes
COC	cyclic olefin copolymer
CPG	controlled pore glass
CRP	C-reactive protein
CS	chitosan
CV	cyclic voltammetry
CVD	chemical vapor deposition
cTNT	cardiac troponin-T
DCC	<i>N,N'</i> -dicyclohexylcarbodiimide
DET	direct electron transfer
DHD	D-fructose dehydrogenase
DMSO	
	dimethyl sulfoxide
DNA	deoxyribonucleic acid
DPV	differential pulse voltammetry
EDC	1-ethyl-3-[3-(dimethylamino)propyl]-carbodii-
FIC	mide
EIS	electrochemical impedance spectroscopy
ELISA	enzyme-linked immunosorbent assays
FABP	fatty acid binding protein
FBP	Fc-binding protein
FETs	field effect transistors
FITC	fluorescein isothiocyanate
FTIR	Fourier transform infrared spectroscopy
GA	gibberellic acid
GCE	glassy carbon electrode
GLD	glutaraldehyde
GMA	glycidyl methacrylate
GMR	guided-mode resonance
GNPs	graphene nanoplatelets
GOPS	(3-glycidopropyl)-triethoxysilane
GOx	glucose oxidase
HA	human albumin
HbsAbs	hepatitis B surface antibodies
HBsAg	hepatitis B surface antigen
HBV	hepatitis B virus
hCG	human chorionic gonadotropin
HEPES	4-(2-hydroxyethyl)-1-piperazineethanesulfonic
	acid
HFA	human fetuin A
HOPG	highly oriented pyrolytic graphite
H-PDLC	holographic polymer-dispersed liquid crystal
HPLC	high performance liquid chromatography
HRP	horseradish peroxidase
HSA	human serum albumin
IA	immunoassay
ICAM	intracellular adhesion molecule
IDT	interdigitated transducer
IgG	immunoglobulin G
IL	interleukin
IMS	immunosensors
ITO	indium tin oxide
JEV	Japanese encephalitis virus
LBL	layer-by-layer
LCN2	human lipocalin-2
LOD	limit of detection
LOD	localized surface plasmon resonance
MA	methamphetamine
MCN-ICP-MS	
WICIN-ICP-IMIS	, <u>,</u>
MEA	plasma mass spectrometry
MEA	microelectrode array
MEG	multilayer epitaxial graphene
MWCNT	multiwalled carbon nanotubes

mPDGs	<pre>magnetic poly(divinylbenzene-co-glycidly</pre>
	methacrylate)
MTMOS	methyltrimethoxysilane
MSN	mesoporous silica nanoparticles
MTP	microtiter plate
MYO	myoglobin
NDV	Newcastle disease virus
NHS	N-hydroxysucinimide
NMR	nuclear magnetic resonance
NPs	nanoparticles
OGC	optical grating coupler
PAEPEMC	piezoelectric excited millimeter-sized cantilever
PAP	<i>p</i> -aminophenol
PAPP	<i>p</i> -aminophenyl phosphate
PDDA	poly(diallyldimethylammonium) chloride
PDMS	polydimethylsiloxane
PDITC	1,4-phenylene diisothiocyanate
PECVD	plasma enhanced-chemical vapor deposition
PEG	polyethylene glycol
PFCP	perfluorocarbon polymer
PGM	personal glucose meter
PMMA	poly(methyl methacrylate)
PNA	peptide nucleic acid
POC	point-of-care
POx	putrescine oxidase
ProGRP	progastrin releasing peptide
PrP	prion proteins
PS	polystyene
PSA	prostate specific antigen
PZT	lead zirconium titanate
QCM	quartz crystal microbalance
QD	quantum dot
RĨ	refractive index
S-4FB	succinimidyl 4-formylbenzoate
SAM	self-assembled monolayers
SAW	surface acoustic wave
SCE	saturated calomel electrode
SEB	Staphylococcus enterotoxin B
SELS	surface enhanced light scattering
S-HyNic	succinimidyl 6-hydrazinonicotinamide acetone
,	hydrazine
SiNW	silicon nanowire
SOI	silicon-on-insulator
SPR	surface plasmon resonance
SulfoNHS	sulfo-N-hydroxysucinimide
SWASV	squave wave anodic stripping voltammetry
TEOS	tetraethoxysilane (or tetraethyl orthosilicate)
TMB	3,3',5,5'-tetramethylbenzidine
ToF-SIMS	time-of-flight secondary ion mass spectrometry
Trp	tryptophan
TrI	troponin I
VEGF	vascular endothelial growth factors
XPS	X-ray photoelectron spectroscopy
	, r

REFERENCES

- (1) Sagiv, J. J. Am. Chem. Soc. 1980, 102, 92.
- (2) Vansant, E. F.; Van Der Voort, P.; Vrancken, K. C. Characterization and Chemical Modification of the Silica Surface; Elsevier Science B.V.: Amsterdam, 1995.
- (3) White, L. D.; Tripp, C. P. J. Colloid Interface Sci. 2000, 227, 237.
- (4) Blitz, J. P.; Murthy, R. S. S.; Leyden, D. E. J. Am. Chem. Soc. 1987, 109, 7141.
- (5) Weetall, H. H. Appl. Biochem. Biotechnol. 1993, 41, 157.

	Review
 (6) Howarter, J. A.; Youngblood, J. P. Langmuir 2006, 22, (7) Landoulsi, J.; Genet, M. J.; Karat, K. E.; Richard, C.; Rouxhet, P. G. Silanization with APTES for Controlling the Between Stainless Steel and Biocomponents: Reality vs Exp 	; Pulvin, S.; Interactions vectation. In
Biomaterials-Physics and Chemistry; Pignatello, R., Ed.; InT 2011; pp 99–126.	
(8) Weng, Y. J.; Hou, R.; Xie, D.; Wang, J.; Huang, N. <i>Key</i> 200 7, 330, 865.	Eng. Mater.
 (9) Yamaura, M.; Camilo, R.; Sampaio, L.; Macedo, M.; Nal Toma, H. J. Magn. Magn. Mater. 2004, 279, 210. (10) Asenath Smith, E.; Chen, W. Langmuir 2008, 24, 12- (11) Gandhiraman, R. P.; Volcke, C.; Gubala, V.; Doyle, Desmonts, L.; Dotzler, C.; Toney, M. F.; Iacono, M.; No Daniels, S. J. Mater. Chem. 2010, 20, 4116. 	405. C.; Basabe-
(12) Rozlosnik, N.; Gerstenberg, M. C.; Larsen, N. B. Lang 19, 1182.	zmuir 2003 ,
 (13) Fadeev, A. Y.; Helmy, R.; Marcinko, S. Langmuir 200 (14) Mateo, C.; Fernández-Lorente, G.; Abian, O.; Lafuente, R.; Guisán, J. M. Biomacromolecules 2000, 1, 739. (15) Vandenberg, E. T.; Bertilsson, L.; Liedberg, B.; Erlandsson, R.; Elwing, H.; Lundström, I. J. Colloid Interface 	Fernández- Uvdal, K.;
147, 103. (16) Torabi, F.; Mobini Far, H. R.; Danielsson, B.; Kh	ayyami, M.
Biosens. Bioelectron. 2007, 22, 1218. (17) Blagoi, G.; Keller, S.; Johansson, A.; Boisen, A.; Dufv Surf. Sci. 2008, 255, 2896.	va, M. Appl.
 (18) Varshney, M.; Waggoner, P. S.; Tan, C. P.; Aubin, K.; R. A.; Craighead, H. G. Anal. Chem. 2008, 80, 2141. 	Montagna,
(19) Hideshima, S.; Sato, R.; Kuroiwa, S.; Osaka, T. <i>Biosens</i> .	Bioelectron.
2011 , <i>26</i> , 2419. (20) Gupta, S.; Huda, S.; Kilpatrick, P. K.; Velev, O. D. <i>A</i>	Anal. Chem.
2007 , <i>79</i> , 3810. (21) Ma, L.; Wang, C.; Hong, Y.; Zhang, M.; Su, M. Anal. (Chem. 2010 ,
 82, 1186. (22) Killian, M. S.; Krebs, H. M.; Schmuki, P. Langmuir 201 (23) Huang, Y. F.; Wang, Y. F.; Yan, X. P. Environ. Sci. Tee 44, 7908. 	
(24) van der Maaden, K.; Sliedregt, K.; Kros, A.; Jiskoot, W J. Langmuir 2012 , 28, 3403.	.; Bouwstra,
 (25) Li, W.; Yuan, R.; Chai, Y. J. Phys. Chem. C 2010, 114 (26) Yuan, L.; Hua, X.; Wu, Y.; Pan, X.; Liu, S. Anal. Chem. 	
6800. (27) Yuan, P.; Southon, P. D.; Liu, Z.; Green, M. E. R.; H	Hook, J. M.;
Antill, S. J.; Kepert, C. J. J. Phys. Chem. C 2008, 112, 15742 (28) Yang, Y. J.; Tao, X.; Hou, Q.; Chen, J. F. Acta Bioma	
3488.	
(29) Xue, A.; Zhou, S.; Zhao, Y.; Lu, X.; Han, P. J. Hazard. N 194, 7.	
(30) Sanchez-Vaquero, V.; Satriano, C.; Tejera-Sanchez, N Mendez, L.; Garcia Ruiz, J. P.; Manso Silvan, M. <i>Biointerpha</i> 23.	
 (31) Luo, K.; Zhou, S.; Wu, L.; Gu, G. Langmuir 2008, 24 (32) Su, F.; Lu, C.; Chen, H. S. Langmuir 2011, 27, 8090. (33) Lee, W. L.; Thompson, H. G. Can. J. Chem. 1996, 74 (34) Hu, P.; Huang, C. Z.; Li, Y. F.; Ling, J.; Liu, Y. L.; Fei, P. Anal. Chem. 2008, 80, 1819. 	4, 707.
(35) Skottrup, P. D.; Hansen, M. F.; Lange, J. M.; Der Svendsen, W. E.; Jakobsen, M. H.; Dufva, M. Acta Biomat	
3936. (36) Yagiuda, K.; Hemmi, A.; Ito, S.; Asano, Y.; Fushinuk	i, Y.; Chen,
CY.; Karube, I. <i>Biosens. Bioelectron.</i> 1996 , <i>11</i> , 703. (37) Li, J.; Xiao, L. T.; Zeng, G. M.; Huang, G. H.; Shen, G. J	L.; Yu, R. Q.

- J. Agric. Food Chem. 2005, 53, 1348.
- (38) Kim, N.; Park, I. S.; Kim, W. Y. Sens. Actuators, B: Chem. 2007, 121, 606.
- (39) Shiku, H.; Uchida, I.; Matsue, T. Langmuir 1997, 13, 7239.
- (40) Balasundaram, G.; Sato, M.; Webster, T. J. Biomaterials 2006, 27, 2798.
- (41) Wilson, M. S. Anal. Chem. 2005, 77, 1496.

(42) Faure, A. C.; Hoffmann, C.; Bazzi, R.; Goubard, F.; Pauthe, E.; Marquette, C. A.; Blum, L. J.; Perriat, P.; Roux, S.; Tillement, O. ACS *Nano* **2008**, *2*, 2273.

- (43) Dong, H.; Li, C. M.; Zhou, Q.; Sun, J. B.; Miao, J. M. Biosens. Bioelectron. 2006, 22, 621.
- (44) Lee, H. J.; Namkoong, K.; Cho, E. C.; Ko, C.; Park, J. C.; Lee, S. S. Biosens. Bioelectron. **2009**, *24*, 3120.
- (45) Qi, C.; Feng, J.; Wang, Z.-H.; Meng, Y.-H.; Yan, X.-Y.; Jin, G. *Chin.* J. Biotechnol. **2006**, 22, 856.
- (46) Chua, J. H.; Chee, R.-E.; Agarwal, A.; Wong, S. M.; Zhang, G.-J. Anal. Chem. **2009**, *81*, 6266.
- (47) Chen, B.; Peng, H.; Zheng, F.; Hu, B.; He, M.; Zhao, W.; Pang, D. J. Anal. At. Spectrom. **2010**, 25, 1674.
- (48) Fang, X.; Tan, O. K.; Tse, M. S.; Ooi, E. E. Biosens. Bioelectron. 2010, 25, 1137.
- (49) Gan, S.; Yang, P.; Yang, W. Biomacromolecules 2009, 10, 1238.
- (50) Helali, S.; Abdelghani, A.; Hafaiedh, I.; Martelet, C.; Prodromidis, M. I.; Albanis, T.; Jaffrezic-Renault, N. *Mater. Sci. Eng., C* **2008**, *28*, 826.
- (51) Raj, J.; Herzog, G.; Manning, M.; Volcke, C.; MacCraith, B. D.; Ballantyne, S.; Thompson, M.; Arrigan, D. W. *Biosens. Bioelectron.* **2009**, 24, 2654.
- (52) Charlesa, P. T.; Rangasammya, J. G.; Andersona, G. P.; Romanoskib, T. C.; Kusterbecka, A. W. Anal. Chim. Acta 2004, 525, 199.
- (53) Behringer, K. D.; Blümel, J. J. Liq. Chromatogr. Relat. Technol. **1996**, 19, 2753.
- (54) Liu, Y.; Li, Y.; Li, X. M.; He, T. Langmuir 2013, 29, 15275.
- (55) Song, Y.-Y.; Hildebrand, H.; Schmuki, P. Surf. Sci. 2010, 604, 346.
- (56) Hernandez, O. 3-Aminopropyltriethoxysilane (APTES); UNEP Publications: Arona, Italy, 2003.
- (57) Herder, P.; Vågberg, L.; Stenius, P. Colloids Surf. 1989, 34, 117.
- (58) Iler, R. K. The Chemistry of Silica: Solubility, Polymerization, Colloid and Surface Properties, and Biochemistry; Wiley Interscience: New York, 1979.
- (59) Ishida, H. Polym. Compos. 1984, 5, 101.
- (60) Dixit, C. K.; Vashist, S. K.; MacCraith, B. D.; O'Kennedy, R. Nat. Protoc. 2011, 6, 439.
- (61) Vashist, S. K. Biosens. Bioelectron. 2013, 40, 297.
- (62) Dixit, C. K.; Vashist, S. K.; MacCraith, B. D.; O'Kennedy, R. Analyst 2011, 136, 1406.
- (63) Vashist, S. K.; Dixit, C. K.; MacCraith, B. D.; O'Kennedy, R. Analyst **2011**, *136*, 4431.
- (64) Avseenko, N. V.; Morozova, T. Y.; Ataullakhanov, F. I.; Morozov, V. N. *Anal. Chem.* **2002**, *74*, 927.
- (65) Aran, K.; Sasso, L. A.; Kamdar, N.; Zahn, J. D. *Lab Chip* **2010**, *10*, 548.
- (66) Cheng, Y.; Xiong, P.; Yun, C. S.; Strouse, G. F.; Zheng, J. P.; Yang, R. S.; Wang, Z. L. *Nano Lett.* **2008**, *8*, 4179.
- (67) Park, J. H.; Sun, Y.; Goldman, Y. E.; Composto, R. J. Langmuir 2010, 26, 10961.
- (68) Chaijareenont, P.; Takahashi, H.; Nishiyama, N.; Arksornnukit, M. Dent. Mater. J. **2012**, 31, 610.
- (69) Bruce, I. J.; Sen, T. Langmuir 2005, 21, 7029.
- (70) Roque, A. C.; Bicho, A.; Batalha, I. L.; Cardoso, A. S.; Hussain, A. J. Biotechnol. **2009**, *144*, 313.
- (71) Roy, S.; Dixit, C. K.; Woolley, R.; MacCraith, B. D.; O'Kennedy, R.; McDonagh, C. *Langmuir* **2010**, *26*, 18125.
- (72) Roy, S.; Dixit, C. K.; Woolley, R.; O'Kennedy, R.; McDonagh, C. *Langmuir* **2011**, *27*, 10421.
- (73) Nguyen, Q. X.; Belgard, T. G.; Taylor, J. J.; Murthy, V. S.; Halas, N. J.; Wong, M. S. *Chem. Mater.* **2012**, *24*, 1426.
- (74) Engelhardt, H.; Orth, P. J. Liq. Chromatogr. 1987, 10, 1999.
- (75) Zhu, M.; Lerum, M. Z.; Chen, W. Langmuir 2011, 28, 416.
- (76) Wang, W.; Vaughn, M. W. Scanning 2008, 30, 65.
- (77) Jönsson, U.; Olofsson, G.; Malmqvist, M.; Rönnberg, I. *Thin Solid Films* **1985**, 124, 117.
- (78) Singh, S.; Sharma, S. N.; Govind; Shivaprasad, S. M.; Lal, M.; Khan, M. A. J. Mater. Sci.: Mater. Med. 2009, 20, S181.
- (79) Lyubchenko, Y.; Shlyakhtenko, L.; Harrington, R.; Oden, P.; Lindsay, S. Proc. Natl. Acad. Sci. U.S.A. 1993, 90, 2137.

- (80) Hong, Z.; Chasan, B.; Bansil, R.; Turner, B. S.; Bhaskar, K. R.; Afdhal, N. H. *Biomacromolecules* **2005**, *6*, 3458.
- (81) Krumdieck, S. P.; Raj, R. Chem. Vap. Deposition 2001, 7, 85.
- (82) Krumdieck, S. P.; Sbaizero, O.; Bullert, A.; Raj, R. Surf. Coat. Technol. 2003, 167, 226.
- (83) Krumdieck, S. J. Aust. Ceram. Soc. 2012, 48, 69.
- (84) Metwalli, E.; Haines, D.; Becker, O.; Conzone, S.; Pantano, C. G. J. Colloid Interface Sci. **2006**, 298, 825.
- (85) Price, J.; Wu, S.-Y. Plasma enhanced CVD. US 4692343, September 8, 1987.
- (86) Gandhiraman, R. P.; Gubala, V.; O'Mahony, C. C.; Cummins, T.; Raj, J.; Eltayeb, A.; Doyle, C.; James, B.; Daniels, S.; Williams, D. E. *Vacuum* **2012**, *86*, 547.
- (87) Culler, S.; Ishida, H.; Koenig, J. J. Colloid Interface Sci. 1986, 109, 1.
 (88) Naviroj, S.; Culler, S. R.; Koenig, J. L.; Ishida, H. J. Colloid Interface Sci. 1984, 97, 308.
- (89) Chiang, C.-H.; Liu, N.-I.; Koenig, J. L. J. Colloid Interface Sci. 1982, 86, 26.
- (90) Luzinov, I.; Julthongpiput, D.; Liebmann-Vinson, A.; Cregger, T.; Foster, M. D.; Tsukruk, V. V. *Langmuir* **2000**, *16*, 504.
- (91) Kim, J.; Seidler, P.; Wan, L. S.; Fill, C. J. Colloid Interface Sci. 2009, 329, 114.
- (92) Jeyachandran, Y. L.; Mielczarski, J. A.; Mielczarski, E.; Rai, B. J. Colloid Interface Sci. 2010, 341, 136.
- (93) Ouerghi, O.; Touhami, A.; Othmane, A.; Ouada, H. B.; Martelet, C.; Fretigny, C.; Jaffrezic-Renault, N. *Biomol. Eng.* **2002**, *19*, 183.
- (94) Lu, B.; Smyth, M. R.; O'Kennedy, R. Analyst 1996, 121, 29R.
- (95) Chen, S.; Liu, L.; Zhou, J.; Jiang, S. Langmuir 2003, 19, 2859.
- (96) Paynter, R.; Ratner, B.; Horbett, T.; Thomas, H. J. Colloid Interface Sci. 1984, 101, 233.
- (97) Scofield, J. J. Electron Spectrosc. 1976, 8, 129.
- (98) Raiteri, R.; Grattarola, M.; Butt, H.-J.; Skládal, P. Sens. Actuators, B: Chem. 2001, 79, 115.
- (99) Dubey, M.; Emoto, K.; Takahashi, H.; Castner, D. G.; Grainger, D. W. Adv. Funct. Mater. **2009**, *19*, 3046.
- (100) Samuel, N. T.; Wagner, M.; Dornfeld, K.; Castner, D. G. Surf. Sci. Spectra **2001**, *8*, 163.
- (101) Wang, Z. H.; Jin, G. J. Immunol. Methods 2004, 285, 237.
- (102) Sandoval, J. E.; Pesek, J. J. Anal. Chem. 1989, 61, 2067.
- (103) Ashu-Arrah, B. A.; Glennon, J. D.; Albert, K. J. Chromatogr., A **2012**, 1236, 42.
- (104) Betancor, L.; López-Gallego, F.; Hidalgo, A.; Alonso-Morales, N.; Mateo, G. D.-O. C.; Fernández-Lafuente, R.; Guisán, J. M. *Enzyme Microb. Technol.* **2006**, *39*, 877.
- (105) Vashist, S. K. Diagnostics 2012, 2, 23.
- (106) Wagner, P.; Hegner, M.; Kernen, P.; Zaugg, F.; Semenza, G. Biophys. J. 1996, 70, 2052.
- (107) Byeon, J. Y.; Limpoco, F. T.; Bailey, R. C. Langmuir 2010, 26, 15430.
- (108) Zaitsev, V. N.; Colomiets, L. I.; Elskaya, A. V.; Skopenko, V. V.; Evans, J. Anal. Chim. Acta **1991**, 252, 1.
- (109) Yoshioka, M.; Mukai, Y.; Matsui, T.; Udagawa, A.; Funakubo, H. J. Chromatogr. **1991**, 566, 361.
- (110) Tatsuma, T.; Okawa, Y.; Watanabe, T. Anal. Chem. 1989, 61, 2352.
- (111) Zheng, D.; Vashist, S. K.; Al-Rubeaan, K.; Luong, J. H. T.; Sheu, F. S. Analyst **2012**, 137, 3800.
- (112) Zheng, D.; Vashist, S.; Dykas, M.; Saha, S.; Al-Rubeaan, K.; Lam, E.; Luong, J. H. T.; Sheu, F.-S. *Materials* **2013**, *6*, 1011.
- (113) Björck, I.; Petersson, B. Å.; Sjöquist, J. Eur. J. Biochem. **1972**, 29, 579.
- (114) Sjöquist, J.; Meloun, B.; Hjelm, H. Eur. J. Biochem. 1972, 29, 572.
- (115) Akerström, B.; Björck, L. J. Biol. Chem. 1986, 261, 10240.
- (116) Akerström, B.; Brodin, T.; Reis, K.; Björk, L. J. Immunol. 1985, 135, 2589.
- (117) Sikkema, W. D. Am. Biotechnol. Lab. 1989, 7, 42.
- (118) Yang, L.; Biswas, M.; Chen, P. Biophys. J. 2003, 84, 509.
- (119) Kastern, W.; Sjöbring, U.; Björck, L. J. Biol. Chem. 1992, 267, 12820.

- (121) Arnau, J.; Lauritzen, C.; Petersen, G. E.; Pedersen, J. Protein Expression Purif. 2006, 48, 1.
- (122) Nogi, T.; Sangawa, T.; Tabata, S.; Nagae, M.; Tamura-Kawakami, K.; Beppu, A.; Hattori, M.; Yasui, N.; Takagi, J. *Protein Sci.* 2008, *17*, 2120.
- (123) Darain, F.; Ban, C.; Shim, Y. B. *Biosens. Bioelectron.* **2004**, *20*, 857. (124) Jones, C. J.; Stoddart, R. W. *Histochem. J.* **1986**, *18*, 371.
- (125) Hopp, T. P.; Prickett, K. S.; Price, V. L.; Libby, R. T.; March, C.
- J.; Pat Cerretti, D.; Urdal, D. L.; Conlon, P. J. Nat. Biotechnol. 1988, 6, 1204.
- (126) Nakajima, H.; Brindle, P. K.; Handa, M.; Ihle, J. N. *EMBO J.* 2001, 20, 6836.
- (127) Zhang, L.; Liu, Y.; Chen, T. Microchim. Acta 2009, 164, 161.
- (128) Bain, C. D.; Troughton, E. B.; Tao, Y. T.; Evall, J.; Whitesides, G. M.; Nuzzo, R. G. *J. Am. Chem. Soc.* **1989**, *111*, 321.
- (129) Bhatia, S. K.; Shriver-Lake, L. C.; Prior, K. J.; Georger, J. H.; Calvert, J. M.; Bredehorst, R.; Ligler, F. S. Anal. Biochem. **1989**, 178, 408.
- (130) Rao, S. V.; Anderson, K. W.; Bachas, L. G. *Microchim. Acta* **1998**, *128*, 127.
- (131) Brogan, K. L.; Wolfe, K. N.; Jones, P. A.; Schoenfisch, M. H. Anal. Chim. Acta 2003, 496, 73.
- (132) Morfill, J.; Blank, K.; Zahnd, C.; Luginbuhl, B.; Kuhner, F.;
- Gottschalk, K. E.; Pluckthun, A.; Gaub, H. E. *Biophys. J.* **2007**, *93*, 3583. (133) Kipriyanov, S. M.; Little, M. *Anal. Biochem.* **1997**, *244*, 189.
- (134) Saerens, D.; Huang, L.; Bonroy, K.; Muyldermans, S. Sensors 2008, 8, 4669.
- (135) Nielsen, P. E.; Egholm, M.; Berg, R. H.; Buchardt, O. Science 1991, 254, 1497.
- (136) Egholm, M.; Buchardt, O.; Christensen, L.; Behrens, C.; Freier, S. M.; Driver, D. A.; Berg, R. H.; Kim, S. K.; Norden, B.; Nielsen, P. E. *Nature* **1993**, *365*, 566.
- (137) Sheng, H.; Ye, B. C. Appl. Biochem. Biotechnol. 2009, 152, 54.
- (138) Silva, T. A.; Ferreira, L. F.; Souza, L. M.; Goulart, L. R.; Madurro,
- J. M.; Brito-Madurro, A. G. Mater. Sci. Eng., C 2009, 29, 539.
- (139) Ray, A.; Norden, B. FASEB J. 2000, 14, 1041.
- (140) Höök, F.; Ray, A.; Nordén, B.; Kasemo, B. Langmuir 2001, 17, 8305.
- (141) Masuko, M. Nucleic Acids Symp. Ser. 2003, 3, 145.
- (142) Zhong, M.; Fang, J.; Wei, Y. Bioconjugate Chem. 2010, 21, 1177.
- (143) Rabe, M.; Verdes, D.; Seeger, S. Adv. Colloid Interface Sci. 2011, 162, 87.
- (144) Vogler, E. A. Biomaterials 2012, 33, 1201.
- (145) Tsapikouni, T. S.; Missirlis, Y. F. *Mater. Sci. Eng., B* **2008**, *152*, 2. (146) Bremer, M. G.; Duval, J.; Norde, W.; Lyklema, J. *Colloids Surf, A* **2004**, *250*, 29.
- (147) van der Veen, M.; Stuart, M. C.; Norde, W. Colloids Surf, B 2007, 54, 136.
- (148) Norde, W.; Lyklema, J. Adv. Colloid Interface Sci. 2012, 179–182, 5.
- (149) Denis, F. A.; Hanarp, P.; Sutherland, D. S.; Gold, J.; Mustin, C.; Rouxhet, P. G.; Dufrêne, Y. F. *Langmuir* **2002**, *18*, 819.
- (150) Zhou, T.; Llizo, A.; Wang, C.; Xu, G.; Yang, Y. *Nanoscale* **2013**, *5*, 8288.
- (151) Kim, T. H.; Kim, M.; Eltohamy, M.; Yun, Y. R.; Jang, J. H.; Kim, H. W. J. Biomed. Mater. Res., Part A **2013**, 101, 1651.
- (152) Weiss, S. A.; Jeuken, L. J. Biochem. Soc. 2009, 37, 707.
- (153) Lee, K. B.; Park, S. J.; Mirkin, C. A.; Smith, J. C.; Mrksich, M. *Science* **2002**, *295*, 1702.
- (154) Awsiuk, K.; Bernasik, A.; Kitsara, M.; Budkowski, A.; Rysz, J.; Haberko, J.; Petrou, P.; Beltsios, K.; Raczkowska, J. *Colloids Surf., B* **2010**, *80*, 63.
- (155) Barat, B.; Wu, A. M. Biomol. Eng. 2007, 24, 283.
- (156) Emerman, A. B.; Zhang, Z. R.; Chakrabarti, O.; Hegde, R. S. *Mol. Biol. Cell* **2010**, *21*, 4325.
- (157) Hermanson, G. T. *Bioconjugate Techniques*; Academic Press: San Diego, CA, 1996.

- (158) Hofmann, K.; Wood, S. W.; Brinton, C. C.; Montibeller, J. A.; Finn, F. M. Proc. Natl. Acad. Sci. U.S.A. **1980**, 77, 4666.
- (159) Hirsch, J. D.; Eslamizar, L.; Filanoski, B. J.; Malekzadeh, N.; Haugland, R. P.; Beechem, J. M.; Haugland, R. P. *Anal. Biochem.* **2002**, 308, 343.
- (160) Hofmann, K.; Titus, G.; Montibeller, J. A.; Finn, F. M. *Biochemistry* **1982**, *21*, 978.
- (161) Zhang, K.; Diehl, M. R.; Tirrell, D. A. J. Am. Chem. Soc. 2005, 127, 10136.
- (162) Kwon, Y.; Coleman, M. A.; Camarero, J. A. Angew. Chem., Int. Ed. 2006, 45, 1726.
- (163) Camarero, J. A.; Kwon, Y.; Coleman, M. A. J. Am. Chem. Soc. 2004, 126, 14730.
- (164) Helms, B.; van Baal, I.; Merkx, M.; Meijer, E. W. *ChemBioChem* **2007**, *8*, 1790.
- (165) Nakanishi, K.; Sakiyama, T.; Kumada, Y.; Imamura, K.; Imanaka, H. *Curr. Proteomics* **2008**, *5*, 161.
- (166) Waugh, D. S. Trends Biotechnol. 2005, 23, 316.
- (167) Schröper, F.; Baumann, A.; Offenhäusser, A.; Mayer, D. Biosens. Bioelectron. 2012, 34, 171.
- (168) Schröper, F.; Baumann, A.; Offenhaüsser, A.; Mayer, D. Chem. Commun. 2010, 46, 5295.
- (169) Ley, C.; Holtmann, D.; Mangold, K. M.; Schrader, J. Colloids Surf., B 2011, 88, 539.
- (170) Millner, P.; Hays, H.; Vakurov, A.; Pchelintsev, N.; Billah, M.; Rodgers, M. Semin. Cell. Dev. Biol. 2009, 20, 34.
- (171) Bello-Gil, D.; Maestro, B.; Fonseca, J.; Feliu, J. M.; Climent, V.; Sanz, J. M. *PLoS One* **2014**, *9*, e87995.
- (172) Lopez, R.; Garcia, E. FEMS Microbiol. Rev. 2004, 28, 553.
- (173) Pandey, P.; Upadhyay, S.; Pathak, H. Sens. Actuators, B: Chem. **1999**, 60, 83.
- (174) Chiorcea-Paquim, A. M.; Pauliukaite, R.; Brett, C. M.; Oliveira-Brett, A. M. Biosens. Bioelectron. 2008, 24, 297.
- (175) Matsuhisa, H.; Tsuchiya, M.; Hasebe, Y. Colloids Surf., B 2013, 111, 523.
- (176) Vashist, S. K. Anal. Biochem. 2014, 446, 96.
- (177) North, S. H.; Lock, E. H.; Cooper, C. J.; Franek, J. B.; Taitt, C. R.; Walton, S. G. *ACS Appl. Mater. Interfaces* **2010**, *2*, 2884.
- (178) Vashist, S. K.; Marion Schneider, E.; Lam, E.; Hrapovic, S.; Luong, J. H. T. Sci. Rep. 2014, 4, 4407.
- (179) Vashist, S. K.; Schneider, E. M.; Luong, J. H. T. Biosens. Bioelectron. 2014, DOI: 10.1016/j.bios.2014.06.058.
- (180) Chen, L.; Zhang, Z.; Zhang, P.; Zhang, X.; Fu, A. Sens. Actuators, B: Chem. 2011, 155, 557.
- (181) Feng, Y.; Ning, B.; Su, P.; Wang, H.; Wang, C.; Chen, F.; Gao, Z. *Talanta* **2009**, *80*, 803.
- (182) Lirtsman, V.; Ziblat, R.; Golosovsky, M.; Davidov, D.; Pogreb, R.; Sacks-Granek, V.; Rishpon, J. *J. Appl. Phys.* **2005**, *98*, 093506.
- (183) Palasantzas, G.; Svetovoy, V.; Van Zwol, P. *Phys. Rev. B* 2009, *79*, 235434.
- (184) Pan, L.; Jung, S.; Yoon, R. H. J. Colloid Interface Sci. 2011, 361, 321.
- (185) Siqueira Petri, D. F.; Wenz, G.; Schunk, P.; Schimmel, T. Langmuir 1999, 15, 4520.
- (186) Endo, T.; Yamamura, S.; Nagatani, N.; Morita, Y.; Takamura, Y.; Tamiya, E. *Sci. Technol. Adv. Mater.* **2005**, *6*, 491.
- (187) Vashist, S. K.; Schneider, E. M.; Luong, J. H. T. Analyst 2014, 139, 2237.
- (188) Zhao, H. W.; Huang, C. Z.; Li, Y. F. Anal. Chim. Acta 2006, 564, 166.
- (189) De Vos, K. M.; Bartolozzi, I.; Bienstman, P.; Baets, R.; Schacht, E. K. Optical biosensor based on silicon-on-insulator microring cavities for specific protein binding detection. In *Proc. SPIE 6447, Nanoscale Imaging, Spectroscopy, Sensing, and Actuation for Biomedical Applications IV, San Jose, CA, February 12, 2007*; Cartwright, A. N., Nicolau, D. V., Eds.; SPIE: San Jose, CA, 2007; p 64470K.
- (190) Park, M. K.; Kee, J. S.; Quah, J. Y.; Netto, V.; Song, J.; Fang, Q.; La Fosse, E. M.; Lo, G.-Q. Sens. Actuators, B: Chem. **2013**, 176, 552.

(191) Washburn, A. L.; Gunn, L. C.; Bailey, R. C. Anal. Chem. 2009, 81, 9499.

- (192) Hsiao, V. K.; Waldeisen, J. R.; Zheng, Y.; Lloyd, P. F.; Bunning, T. J.; Huang, T. J. *J. Mater. Chem.* **2007**, *17*, 4896.
- (193) Kim, W. J.; Kim, B. K.; Kim, A.; Huh, C.; Ah, C. S.; Kim, K. H.; Hong, J.; Park, S. H.; Song, S.; Song, J.; Sung, G. Y. *Anal. Chem.* **2010**, *82*, 9686.
- (194) Long, F.; Zhu, A.; Zhou, X.; Wang, H.; Zhao, Z.; Liu, L.; Shi, H. Biosens. Bioelectron. **2014**, 55, 19.
- (195) Liu, Y.; Li, Y. Anal. Chem. 2001, 73, 5180.
- (196) Acharya, G.; Chang, C. L.; Savran, C. J. Am. Chem. Soc. 2006, 128, 3862.
- (197) Mura, S.; Greppi, G.; Marongiu, M. L.; Roggero, P. P.; Ravindranath, S. P.; Mauer, L. J.; Schibeci, N.; Perria, F.; Piccinini, M.; Innocenzi, P.; Irudayaraj, J. *Beilstein J. Nanotechnol.* **2012**, *3*, 485.
- (198) Kim, J.; Jensen, E. C.; Megens, M.; Boser, B.; Mathies, R. A. Lab Chip **2011**, *11*, 3106.
- (199) McDonald, J. C.; Whitesides, G. M. Acc. Chem. Res. 2002, 35, 491.
- (200) Wang, Y.; Lai, H.-H.; Bachman, M.; Sims, C. E.; Li, G.; Allbritton, N. L. Anal. Chem. 2005, 77, 7539.
- (201) Didar, T. F.; Foudeh, A. M.; Tabrizian, M. Anal. Chem. **2012**, 84, 1012.
- (202) Bai, Y.; Koh, C. G.; Boreman, M.; Juang, Y. J.; Tang, I. C.; Lee, L. J.; Yang, S. T. *Langmuir* **2006**, *22*, 9458.
- (203) Lee, A. G.; Arena, C. P.; Beebe, D. J.; Palecek, S. P. Biomacromolecules **2010**, *11*, 3316.
- (204) Lee, A. G.; Beebe, D. J.; Palecek, S. P. *Biomed. Microdevices* **2012**, 14, 247.
- (205) Ekstrom, S.; Onnerfjord, P.; Nilsson, J.; Bengtsson, M.; Laurell, T.; Marko-Varga, G. Anal. Chem. 2000, 72, 286.
- (206) de Jong, J.; Lammertink, R. G.; Wessling, M. Lab Chip 2006, 6, 1125.
- (207) Peterson, D. S. Lab Chip 2005, 5, 132.
- (208) Tennico, Y. H.; Hutanu, D.; Koesdjojo, M. T.; Bartel, C. M.; Remcho, V. T. *Anal. Chem.* **2010**, *82*, 5591.
- (209) Sivagnanam, V.; Song, B.; Vandevyver, C.; Gijs, M. A. Anal. Chem. 2009, 81, 6509.
- (210) Ji, J.; Zhang, Y.; Zhou, X.; Kong, J.; Tang, Y.; Liu, B. Anal. Chem. **2008**, *80*, 2457.
- (211) He, M.; Herr, A. E. Anal. Chem. 2009, 81, 8177.
- (212) Sung, W. C.; Chen, H. H.; Makamba, H.; Chen, S. H. Anal. Chem. 2009, 81, 7967.
- (213) Peterson, D. S.; Rohr, T.; Svec, F.; Fréchet, J. M. Anal. Chem. 2002, 74, 4081.
- (214) Hisamoto, H.; Shimizu, Y.; Uchiyama, K.; Tokeshi, M.; Kikutani, Y.; Hibara, A.; Kitamori, T. *Anal. Chem.* **2003**, *75*, 350.
- (215) Campbell, G. A.; Mutharasan, R. Anal. Sci. 2005, 21, 355.
- (216) Maraldo, D.; Rijal, K.; Campbell, G.; Mutharasan, R. Anal. Chem. 2007, 79, 2762.
- (217) Campbell, G. A.; Mutharasan, R. Anal. Chem. 2007, 79, 1145.
- (218) Zheng, D.; Vashist, S. K.; Al-Rubeaan, K.; Luong, J. H. T.; Sheu, F. S. *Talanta* **2012**, *99*, 22.
- (219) Zheng, D.; Vashist, S. K.; Al-Rubeaan, K.; Lam, E.; Hrapovic, S.;
- Luong, J. H. T.; Sheu, F.-S. J. Nanopharmaceutics Drug Del. 2013, 1, 64. (220) Luong, J. H. T.; Hrapovic, S.; Wang, D.; Bensebaa, F.; Simard, B.
- *Electroanalysis* **2004**, *16*, 132. (221) Matsumoto, T.; Ohashi, A.; Ito, N.; Fujiwara, H.; Matsumoto, T.
- Biosens. Bioelectron. 2001, 16, 271.
- (222) Matsumoto, T.; Ohashi, A.; Ito, N. Anal. Chim. Acta 2002, 462, 253.
- (223) Murphy-Perez, E.; Arya, S. K.; Bhansali, S. Analyst 2011, 136, 1686.
- (224) Lei, L.; Cao, Z.; Xie, Q.; Fu, Y.; Tan, Y.; Ma, M.; Yao, S. Sens. Actuators, B: Chem. **2011**, 157, 282.
- (225) Lin, J.-J.; Hsu, P.-Y.; Wu, Y.-L.; Jhuang, J.-J. Sensors 2011, 11, 2796.
- (226) Luong, J. H. T.; Hrapovic, S.; Wang, D. Electroanalysis 2005, 17, 47.

- (227) Rochette, J. F.; Sacher, E.; Meunier, M.; Luong, J. H.T. Anal. Biochem. 2005, 336, 305.
- (228) Villalonga, R.; Villalonga, M. L.; Díez, P.; Pingarrón, J. M. J. Mater. Chem. 2011, 21, 12858.
- (229) Puri, N.; Sharma, V.; Tanwar, V. K.; Singh, N.; Biradar, A. M. Prog. Biomater. **2013**, *2*, 1.
- (230) Zor, E.; Patir, I. H.; Bingol, H.; Ersoz, M. Biosens. Bioelectron. 2013, 42, 321.
- (231) Canbaz, M. C.; Sezginturk, M. K. Anal. Biochem. 2014, 446, 9.
- (232) Kim, G. I.; Kim, K. W.; Oh, M. K.; Sung, Y. M. Biosens. Bioelectron. 2010, 25, 1717.
- (233) Vashist, S. K.; Saraswat, M.; Holthšfer, H. *Procedia Chem.* **2012**, *6*, 184.
- (234) Jung, J.; Lim, S. Appl. Surf. Sci. 2013, 265, 24.
- (235) Ray, K. G., III; McCreery, R. L. J. Electroanal. Chem. **1999**, 469, 150.
- (236) Girard, H.; Simon, N.; Ballutaud, D.; Herlem, M.; Etcheberry, A. Diamond Relat. Mater. 2007, 16, 316.
- (237) Martin, H. B.; Argoitia, A.; Landau, U.; Anderson, A. B.; Angus, J. C. J. Electrochem. Soc. **1996**, *143*, L133.
- (238) Xu, J.; Granger, M. C.; Chen, Q.; Strojek, J. W.; Lister, T. E.; Swain, G. M. Anal. Chem. **1997**, *69*, 591A.
- (239) Goeting, C. H.; Marken, F.; Gutiérrez-Sosa, A.; Compton, R. G.; Foord, J. S. *Diamond Relat. Mater.* **2000**, *9*, 390.
- (240) Notsu, H.; Fukazawa, T.; Tatsuma, T.; Tryk, D. A.; Fujishima, A. *Electrochem. Solid-State Lett.* **2001**, *4*, H1.
- (241) Haak, R. M.; Wezenberg, S. J.; Kleij, A. W. Chem. Commun. 2010, 46, 2713.
- (242) Parker, C. O.; Lanyon, Y. H.; Manning, M.; Arrigan, D. W. M.; Tothill, I. E. Anal. Chem. **2009**, *81*, 5291.
- (243) Chai, C.; Takhistov, P. Sensors 2010, 10, 655.
- (244) Leu, G.; Brockhaus, A.; Engemann, J. Surf. Coat. Technol. 2003, 174, 928.
- (245) Bai, S.; Urabe, S.; Okaue, Y.; Yokoyama, T. J. Colloid Interface Sci. 2009, 331, 551.
- (246) Bai, Y.-F.; Xu, T.-B.; Luong, J. H.T.; Cui, H.-F. Anal. Chem. 2014, 86, 4910.
- (247) Fang, X.; Tan, O. K.; Gan, Y. Y.; Tse, M. S. Sens. Actuators, B: Chem. 2010, 149, 381.
- (248) Teti, D.; Visalli, M.; McNair, H. J. Chromatogr., B 2002, 781, 107.
- (249) Harik, S. I.; Sutton, C. H. Cancer Res. 1979, 39, 5010.
- (250) Flores, H. E.; Galston, A. W. Science 1982, 217, 1259.
- (251) Wolrath, H.; Forsum, U.; Larsson, P. G.; Boren, H. J. Clin. Microbiol. 2001, 39, 4026.
- (252) Nagy, L.; Nagy, G.; Gyurcsanyi, R. E.; Neuman, M. R.; Lindner, E. J. Biochem. Biophys. Methods **2002**, 53, 165.
- (253) Guo, Z.; Hao, T.; Duan, J.; Wang, S.; Wei, D. *Talanta* **2012**, *89*, 27.
- (254) Teixeira, S.; Burwell, G.; Castaing, A.; Gonzalez, D.; Conlan, R.; Guy, O. Sens. Actuators, B: Chem. **2014**, 190, 723.
- (255) Zhong, Z.; Patskovskyy, S.; Bouvrette, P.; Luong, J. H. T.; Gedanken, A. J. Phys. Chem. B **2004**, 108, 4046.
- (256) Huang, K. C.; Ehrman, S. H. Langmuir 2007, 23, 1419.
- (257) Menz, W. J.; Shekar, S.; Brownbridge, G. P.; Mosbach, S.; Körmer, R.; Peukert, W.; Kraft, M. J. Aerosol Sci. 2012, 44, 46.
- (258) Wang, L.; Muhammed, M. J. Mater. Chem. 1999, 9, 2871.
- (259) Ladj, R.; Bitar, A.; Eissa, M.; Mugnier, Y.; Le Dantec, R.; Fessi, H.; Elaissari, A. J. Mater. Chem. B **2013**, *1*, 1381.
- (260) Zhuo, Y.; Chai, Y.-Q.; Yuan, R.; Mao, L.; Yuan, Y.-L.; Han, J. Biosens. Bioelectron. **2011**, *26*, 3838.
- (261) Gopinath, S. C.; Awazu, K.; Fujimaki, M.; Shimizu, K.; Mizutani, W.; Tsukagoshi, K. *Analyst* **2012**, *137*, 3520.
- (262) Rezaei, B.; Havakeshian, E.; Ensafi, A. A. Biosens. Bioelectron. 2013, 48, 61.
- (263) Roque, A. C.; Bispo, S.; Pinheiro, A. R.; Antunes, J. M.; Goncalves, D.; Ferreira, H. A. J. Mol. Recognit. 2009, 22, 77.
- (264) Koh, I.; Wang, X.; Varughese, B.; Isaacs, L.; Ehrman, S. H.; English, D. S. J. Phys. Chem. B **2006**, *110*, 1553.

(265) Kumar, S.; Jana, A. K.; Dhamija, I.; Singla, Y.; Maiti, M. Eur. J. Pharm. Biopharm. 2013, 85, 413.

- (266) Branco, R. V.; Estrada Gutarra, M. L.; Freire, D. M.; Almeida, R. V. *Enzyme Res.* **2010**, 2010, 180418.
- (267) Park, H. J.; McConnell, J. T.; Boddohi, S.; Kipper, M. J.; Johnson, P. A. *Colloids Surf*, B **2011**, 83, 198.
- (268) Jie, G.; Wang, L.; Zhang, S. Chem.—Eur. J. 2011, 17, 641.

(269) Qin, W.; Song, Z.; Fan, C.; Zhang, W.; Cai, Y.; Zhang, Y.; Qian, X. Anal. Chem. **2012**, 84, 3138.

- (270) Joo, J.; Kwon, D.; Shin, H. H.; Park, K.-H.; Cha, H. J.; Jeon, S. Sens. Actuators, B: Chem. 2013, 188, 1250.
- (271) Hu, Y.; Shen, G.; Zhu, H.; Jiang, G. J. Agric. Food Chem. **2010**, 58, 2801.
- (272) Cunningham, E.; Campbell, C. J. Langmuir 2003, 19, 4509.
- (273) Endo, T.; Kerman, K.; Nagatani, N.; Hiepa, H. M.; Kim, D. K.;
- Yonezawa, Y.; Nakano, K.; Tamiya, E. *Anal. Chem.* **2006**, *78*, 6465. (274) Lee, M. H.; Lee, D. H.; Jung, S. W.; Lee, K. N.; Park, Y. S.; Seong, W. K. J. Nanomed. Nanotechnol. **2010**, *6*, *78*.
- (275) Prina-Mello, A.; Whelan, A. M.; Atzberger, A.; McCarthy, J. E.; Byrne, F.; Davies, G. L.; Coey, J. M.; Volkov, Y.; Gun'ko, Y. K. *Small* **2010**, *6*, 247.
- (276) Alivisatos, A. P. Science 1996, 271, 933.
- (277) Burda, C.; Chen, X.; Narayanan, R.; El-Sayed, M. A. *Chem. Rev.* **2005**, *105*, 1025.
- (278) Algar, W. R.; Tavares, A. J.; Krull, U. J. Anal. Chim. Acta 2010, 673, 1.
- (279) Alivisatos, A. P.; Gu, W.; Larabell, C. Annu. Rev. Biomed. Eng. 2005, 7, 55.
- (280) Parak, W. J.; Pellegrino, T.; Plank, C. Nanotechnology 2005, 16, R9.
- (281) Zhang, B.; Gong, X.; Hao, L.; Cheng, J.; Han, Y.; Chang, J. Nanotechnology **2008**, *19*, 465604.
- (282) Zhang, Y.; Wu, H.; Huang, X.; Zhang, J.; Guo, S. Nanoscale Res. Lett. 2011, 6, 450.
- (283) Shan, G.; Wang, S.; Fei, X.; Liu, Y.; Yang, G. J. Phys. Chem. B 2009, 113, 1468.
- (284) Tu, M. C.; Chang, Y. T.; Kang, Y. T.; Chang, H. Y.; Chang, P.; Yew, T. R. *Biosens. Bioelectron.* **2012**, *34*, 286.
- (285) Liu, W.; Zhang, Y.; Ge, S.; Song, X.; Huang, J.; Yan, M.; Yu, J. Anal. Chim. Acta **2013**, 770, 132.
- (286) Liu, F.; Zhang, Y.; Ge, S.; Lu, J.; Yu, J.; Song, X.; Liu, S. *Talanta* **2012**, *99*, 512.
- (287) Wang, W.; Hao, Q.; Wang, W.; Bao, L.; Lei, J.; Wang, Q.; Ju, H. Nanoscale **2014**, *6*, 2710.
- (288) Suarez, G.; Keegan, N.; Spoors, J. A.; Ortiz, P.; Jackson, R. J.; Hedley, J.; Borrise, X.; McNeil, C. J. *Langmuir* **2010**, *26*, 6071.
- (289) Eissa, M. M.; Rahman, M. M.; Zine, N.; Jaffrezic, N.; Errachid, A.; Fessi, H.; Elaissari, A. *Acta Biomater.* **2013**, *9*, 5573.
- (290) Rauf, S.; Zhou, D.; Abell, C.; Klenerman, D.; Kang, D. J. *Chem. Commun.* **2006**, 1721.
- (291) Babacan, S.; Pivarnik, P.; Letcher, S.; Rand, A. G. *Biosens*. *Bioelectron*. **2000**, *15*, 615.
- (292) Maehashi, K.; Katsura, T.; Kerman, K.; Takamura, Y.; Matsumoto, K.; Tamiya, E. *Anal. Chem.* **2007**, *79*, 782.
- (293) Sarkar, A.; Daniels-Race, T. Nanomaterials 2013, 3, 272.
- (294) Chen, A.; Kozak, D.; Battersby, B. J.; Forrest, R. M.; Scholler, N.; Urban, N.; Trau, M. *Langmuir* **2009**, *25*, 13510.
- (295) Williamson, M. L.; Atha, D. H.; Reeder, D. J.; Sundaram, P. V. Anal. Lett. **1989**, 22, 803.
- (296) Berney, H.; Roseingrave, P.; Alderman, J.; Lane, W.; Collins, J. K. Sens. Actuators, B: Chem. **1997**, 44, 341.
- (297) Yuan, S.; Szakalas-Gratzl, G.; Ziats, N. P.; Jacobsen, D. W.; Kottke-Marchant, K.; Marchant, R. E. J. Biomed. Mater. Res. 1993, 27, 81.
- (298) Skládal, P.; Minunni, M.; Mascini, M.; Kolář, V.; Fránek, M. J. Immunol. Methods 1994, 176, 117.
- (299) Suri, C. R.; Raje, M.; Mishra, G. C. Biosens. Bioelectron. 1994, 9, 325.
- (300) Suri, C. R.; Mishra, G. C. Biosens. Bioelectron. 1996, 11, 1199.

- (301) Sasaki, S.; Nagata, R.; Hock, B.; Karube, I. *Anal. Chim. Acta* **1998**, 368, 71.
- (302) Qian, W.; Yao, D.; Xu, B.; Yu, F.; Lu, Z.; Knoll, W. Chem. Mater. **1999**, *11*, 1399.
- (303) Weiping, Q.; Bin, X.; Danfeng, Y.; Yihua, L.; Lei, W.; Chunxiao, W.; Fang, Y.; Zhuhong, L.; Yu, W. *Mater. Sci. Eng.*, C **1999**, *8*, 475.
- (304) Gao, Z.; Chao, F.; Chao, Z.; Li, G. Sens. Actuators, B: Chem. 2000, 66, 193.
- (305) Nisnevitch, M.; Kolog-Gulco, M.; Trombka, D.; Green, B. S.; Firer, M. A. J. Chromatogr., B 2000, 738, 217.
- (306) Yuan, H.; Mullett, W. M.; Pawliszyn, J. Analyst **2001**, 126, 1456. (307) Claudon, P.; Donner, M.; Stoltz, J. F. J. Mater. Sci.: Mater. Med. **1991**, 2, 197.
- (308) Herzog, G.; Raj, J.; Arrigan, D. W. M. *Microchim. Acta* **2009**, *166*, 349.
- (309) Eun, A. J.-C.; Huang, L.; Chew, F.-T.; Li, S. F.-Y.; Wong, S.-M. J. Virol. Methods **2002**, *99*, 71.
- (310) Benesch, J.; Tengvall, P. Biomaterials 2002, 23, 2561.
- (311) Yakovleva, J.; Davidsson, R.; Lobanova, A.; Bengtsson, M.; Eremin, S.; Laurell, T.; Emnéus, J. *Anal. Chem.* **2002**, *74*, 2994.
- (312) Kusnezow, W.; Jacob, A.; Walijew, A.; Diehl, F.; Hoheisel, J. D. *Proteomics* **2003**, *3*, 254.
- (313) Wang, Z. H.; Jin, G. Colloids Surf., B 2004, 34, 173.
- (314) Kim, J.-K.; Shin, D.-S.; Chung, W.-J.; Jang, K.-H.; Lee, K.-N.; Kim, Y.-K.; Lee, Y.-S. Colloids Surf., B 2004, 33, 67.
- (315) Wei, Y.; Ning, G.; Hai-Qian, Z.; Jian-Guo, W.; Yi-Hong, W.; Wesche, K.-D. Sens. Actuators, B: Chem. 2004, 98, 83.
- (316) Kandimalla, V. B.; Neeta, N. S.; Karanth, N. G.; Thakur, M. S.; Roshini, K. R.; Rani, B. E.; Pasha, A.; Karanth, N. G. *Biosens. Bioelectron.* **2004**, *20*, 903.
- (317) Diao, J.; Ren, D.; Engstrom, J. R.; Lee, K. H. Anal. Biochem. 2005, 343, 322.
- (318) Liu, M.; Liu, Z. Y.; Lu, Q.; Yuan, H.; Ma, L.; Li, J. H.; Bai, Y. B.; Li, T. J. Chin. J. Chem. **2005**, 23, 875.
- (319) Vashist, S. K.; Raiteri, R.; Tewari, R.; Bajpai, R. P.; Bharadwaj, L. M. J. Phys.: Conf. Ser. **2006**, 34, 806.
- (320) Chu, X.; Zhao, Z.-L.; Shen, G.-L.; Yu, R.-Q. Sens. Actuators, B: Chem. 2006, 114, 696.
- (321) Qian, W.; Xu, B.; Song, Z.; Wu, L.; Yao, D.; Yu, F.; Lu, Z.; Weia, Y. Mol. Cryst. Liq. Cryst. 1999, 337, 277.
- (322) Qin, M.; Hou, S.; Wang, L.; Feng, X.; Wang, R.; Yang, Y.; Wang, C.; Yu, L.; Shao, B.; Qiao, M. *Colloids Surf.*, B **2007**, *60*, 243.
- (323) Liu, Y.; Li, C. M.; Yu, L.; Chen, P. Front. Biosci. 2007, 12, 3768.
- (324) Goyal, D. K.; Pribil, G. K.; Woollam, J. A.; Subramanian, A. Mater. Sci. Eng., B 2008, 149, 26.
- (325) Stanford, C.; Dagenais, M.; Park, J.-H.; DeShong, P. Curr. Anal. Chem. 2008, 4, 356.
- (326) Mace, C. R.; Yadav, A. R.; Miller, B. L. Langmuir **2008**, *24*, 12754. (327) Jang, L. S.; Liu, H. J. Biomed. Microdevices **2009**, *11*, 331.
- (328) Sabella, S.; Brunetti, V.; Vecchio, G.; Torre, A. D.; Rinaldi, R.;
- Cingolani, R.; Pompa, P. P. Nanoscale Res. Lett. 2009, 4, 1222.
- (329) Niotis, A. E.; Mastichiadis, C.; Petrou, P. S.; Christofidis, I.; Siafaka-Kapadai, A.; Misiakos, K.; Kakabakos, S. E. *Anal. Bioanal. Chem.* **2009**, 393, 1081.
- (330) Jayagopal, A.; Stone, G. P.; Haselton, F. R. *Bioconjugate Chem.* 2008, 19, 792.
- (331) Sandison, M. E.; Cumming, S. A.; Kolch, W.; Pitt, A. R. Lab Chip 2010, 10, 2805.
- (332) Briand, E.; Humblot, V.; Landoulsi, J.; Petronis, S.; Pradier, C. M.; Kasemo, B.; Svedhem, S. *Langmuir* **2011**, *27*, 678.
- (333) Wildeboer, D.; Jiang, P.; Price, R. G.; Yu, S.; Jeganathan, F.; Abuknesha, R. A. *Talanta* **2010**, *81*, 68.
- (334) Blinka, E.; Loeffler, K.; Hu, Y.; Gopal, A.; Hoshino, K.; Lin, K.; Liu, X.; Ferrari, M.; Zhang, J. X. *Nanotechnology* **2010**, *21*, 415302.
- (335) Huy, T. Q.; Hanh, N. T. H.; Chung, P. V.; Anh, D. D.; Nga, P. T.; Tuan, M. A. Appl. Surf. Sci. 2011, 257, 7090.
- (336) Wu, X.; Tang, Q.; Liu, C.; Li, Q.; Guo, Y.; Yang, Y.; Lv, X.; Geng, L.; Deng, Y. *Appl. Surf. Sci.* **2011**, 257, 7415.

- (338) Oliver, M. J.; Hernando-Garcia, J.; Pobedinskas, P.; Haenen, K.; Rios, A.; Sanchez-Rojas, J. L. *Colloids Surf., B* **2011**, *88*, 191.
- (339) Son, K. J.; Ahn, S. H.; Kim, J. H.; Koh, W. G. ACS Appl. Mater. Interfaces 2011, 3, 573.

(340) Lim, T. S.; Vedula, S. R.; Kausalya, P. J.; Hunziker, W.; Lim, C. T. *Langmuir* **2008**, *24*, 490.

- (341) Lee, H. U.; Shin, H. Y.; Lee, J. Y.; Song, Y. S.; Park, C.; Kim, S. W. J. Agric. Food Chem. **2010**, 58, 12096.
- (342) Zimnitsky, D.; Jiang, C.; Xu, J.; Lin, Z.; Zhang, L.; Tsukruk, V. V. *Langmuir* **2007**, 23, 10176.
- (343) Huy, T. Q.; Hanh, N. T.; Thuy, N. T.; Chung, P. V.; Nga, P. T.; Tuan, M. A. *Talanta* **2011**, *86*, 271.
- (344) Veerapandian, M.; Subbiah, R.; Lim, G. S.; Park, S. H.; Yun, K.; Lee, M. H. *Langmuir* **2011**, *27*, 8934.
- (345) Kim, J.-Y.; Ahn, J.-H.; Moon, D.-I.; Kim, S.; Park, T. J.; Lee, S. Y.; Choi, Y.-K. *BioNanoScience* **2012**, *2*, 35.
- (346) Kim, W. J.; Kim, S.; Lee, B. S.; Kim, A.; Ah, C. S.; Huh, C.; Sung, G. Y.; Yun, W. S. *Langmuir* **2009**, *25*, 11692.
- (347) Jonsson, C.; Aronsson, M.; Rundstrom, G.; Pettersson, C.; Mendel-Hartwig, I.; Bakker, J.; MacCraith, B.; Ohman, O.; Melin, J. *Eur. Cells Mater.* **2007**, *14*, 64.
- (348) Vashist, S. K. Protocol Exchange 2012, DOI: 10.1028/protex.2012.051.
- (349) Gubala, V.; Siegrist, J.; Monaghan, R.; O'Reilly, B.; Gandhiraman, R. P.; Daniels, S.; Williams, D. E.; Ducree, J. Anal. Chim. Acta 2013, 760, 75.
- (350) Brothier, F.; Pichon, V. Anal. Chim. Acta 2013, 792, 52.
- (351) Chepyala, R.; Panda, S. Appl. Surf. Sci. 2013, 271, 77.
- (352) Choi, E.; Choi, Y.; Nejad, Y. H. P.; Shin, K.; Park, J. Sens. Actuators, B: Chem. 2013, 180, 107.
- (353) Matatagui, D.; Fontecha, J.; Fernández, M. J.; Oliver, M. J.; Hernando-García, J.; Sánchez-Rojas, J. L.; Gràcia, I.; Cané, C.; Santos, J.
- P.; Horrillo, M. C. Sens. Actuators, B: Chem. 2013, 189, 123.
- (354) Gao, Z. D.; Guan, F. F.; Li, C. Y.; Liu, H. F.; Song, Y. Y. Biosens. Bioelectron. 2013, 41, 771.
- (355) Chang, A. Y.; Lu, M. S. Biosens. Bioelectron. 2013, 45, 6.
- (356) Zhang, B.; Morales, A. W.; Peterson, R.; Tang, L.; Ye, J. Y. *Biosens. Bioelectron.* **2014**, *58*, 107.
- (357) Sharma, V.; Kumar, A.; Ganguly, P.; Biradar, A. Appl. Phys. Lett. 2014, 104, 043705.
- (358) Şimşek, Ç. S.; Teke, M.; Sezgintürk, M. K. *Electroanalysis* **2014**, 26, 328.
- (359) Wang, L.; Wang, F.; Shang, L.; Zhu, C.; Ren, W.; Dong, S. *Talanta* **2010**, *82*, 113.
- (360) Torabi, S. F.; Khajeh, K.; Ghasempur, S.; Ghaemi, N.; Siadat, S. O. J. Biotechnol. **2007**, *131*, 111.
- (361) Tien, C. J.; Chiang, B. H. Process Biochem. 1999, 35, 377.
- (362) Vinoba, M.; Bhagiyalakshmi, M.; Jeong, S. K.; Yoon, Y. I.; Nam, S. C. Colloids Surf., B **2012**, 90, 91.
- (363) Alptekin, Ö.; Tükel, S. S.; Yıldırım, D.; Alagöz, D. J. Mol. Catal. B: Enzym. 2009, 58, 124.
- (364) Hartono, S. B.; Qiao, S. Z.; Liu, J.; Jack, K.; Ladewig, B. P.; Hao, Z.; Lu, G. Q. M. J. Phys. Chem. C **2010**, *114*, 8353.
- (365) Yadav, R.; Wanjari, S.; Prabhu, C.; Kumar, V.; Labhsetwar, N.; Satyanarayanan, T.; Kotwal, S.; Rayalu, S. *Energy Fuels* **2010**, *24*, 6198.
- (366) Lomako, O. V.; Menyailova, I. I.; Nakhapetyan, L. A.; Nikitin, Y.; Kiselev, A. V. *Enzyme Microb. Technol.* **1982**, *4*, 89.
- (367) Rogalski, J.; Szczodrak, J.; Dawidowicz, A.; Ilczuk, Z.; Leonowicz, A. *Enzyme Microb. Technol.* **1985**, 7, 395.
- (368) Wojtas-Wasilewska, M.; Luterek, J.; Rogalski, J. *Phytochemistry* **1988**, 27, 2731.
- (369) Sarkar, J. M.; Leonowicz, A.; Bollag, J.-M. Soil Biol. Biochem. 1989, 21, 223.
- (370) Janowski, F.; Fischer, G.; Urbaniak, W.; Foltynowicz, Z.; Marciniec, B. J. Chem. Technol. Biotechnol. **1991**, *51*, 263.
- (371) Emneus, J.; Gorton, L. Anal. Chim. Acta 1993, 276, 303.

- (372) Jen, J.-F.; Zen, J.-H.; Cheng, F.-C.; Yang, G.-Y. Anal. Chim. Acta 1994, 292, 23.
- (373) Rogalski, J.; Joźwik, E.; Hattaka, A.; Leonowicz, A. J. Mol. Catal. A: Chem. **1995**, 95, 99.
- (374) Rogalski, J.; Szczodrak, J.; Pleszczyńska, M.; Fiedurek, J. J. Mol. Catal. B: Enzym. **1997**, 3, 271.
- (375) Licklider, L.; Kuhr, W. G. Anal. Chem. 1998, 70, 1902.
- (376) Rogalski, J.; Dawidowicz, A.; Jóźwik, E.; Leonowicz, A. J. Mol. Catal. B: Enzym. **1999**, *6*, 29.
- (377) Kharitonov, A. B.; Zayats, M.; Lichtenstein, A.; Katz, E.; Willner, I. Sens. Actuators, B: Chem. 2000, 70, 222.
- (378) Ida, J.; Matsuyama, T.; Yamamoto, H. *J. Electrost.* **2000**, *49*, 71. (379) Ferreira, L.; Ramos, M. A.; Dordick, J. S.; Gil, M. H. *J. Mol. Catal. B: Enzym.* **2003**, *21*, 189.
- (380) Shioji, S.; Hanada, M.; Hayashi, Y.; Tokami, K.; Yamamoto, H. Adv. Powder Technol. **2003**, *14*, 231.
- (381) Chong, A. S. M.; Zhao, X. S. Catal. Today 2004, 93-95, 293.
- (382) Chong, A. S. M.; Zhao, X. S. Appl. Surf. Sci. 2004, 237, 398.
- (383) de Lathouder, K. M.; Bakker, J.; Kreutzer, M. T.; Kapteijn, F.; Moulijn, J. A.; Wallin, S. A. *Chem. Eng. Sci.* **2004**, *59*, 5027.
- (384) Tukel, S. S.; Alptekin, O. Process Biochem. 2004, 39, 2149.
- (385) Phadtare, S.; Vinod, V. P.; Mukhopadhyay, K.; Kumar, A.; Rao, M.; Chaudhari, R. V.; Sastry, M. *Biotechnol. Bioeng.* **2004**, *85*, 629.
- (386) Minier, M.; Salmain, M.; Yacoubi, N.; Barbes, L.; Methivier, C.; Zanna, S.; Pradier, C. M. *Langmuir* **2005**, *21*, 5957.
- (387) Longo, L.; Vasapollo, G.; Guascito, M. R.; Malitesta, C. Anal. Bioanal. Chem. 2006, 385, 146.
- (388) Bai, Y. X.; Li, Y. F.; Yang, Y.; Yi, L. X. J. Biotechnol. 2006, 125, 574. (389) Brandi, P.; D'Annibale, A.; Galli, C.; Gentili, P.; Pontes, A. S. N. J. Mol. Catal. B: Enzym. 2006, 41, 61.
- (390) Bai, Y.-X.; Li, Y.-F.; Yang, Y.; Yi, L.-X. Process Biochem. 2006, 41, 770.
- (391) David, A. E.; Wang, N. S.; Yang, V. C.; Yang, A. J. J. Biotechnol. 2006, 125, 395.
- (392) Kim, J.; Jia, H.; Lee, C.-W.; Chung, S.-W.; Kwak, J. H.; Shin, Y.; Dohnalkova, A.; Kim, B.-G.; Wang, P.; Grate, J. W. *Enzyme Microb. Technol.* **2006**, *39*, 474.
- (393) Gómez, J. L.; Bódalo, A.; Gómez, E.; Bastida, J.; Hidalgo, A. M.; Gómez, M. *Enzyme Microb. Technol.* **2006**, 39, 1016.
- (394) Oliveira, E. M.; Beyer, S.; Heinze, J. Bioelectrochemistry 2007, 71, 186.
- (395) Libertino, S.; Fichera, M.; Aiello, V.; Statello, G.; Fiorenza, P.; Sinatra, F. *Microelectron. Eng.* **2007**, *84*, 468.
- (396) Libertino, S.; Scandurra, A.; Aiello, V.; Giannazzo, F.; Sinatra, F.; Renis, M.; Fichera, M. *Appl. Surf. Sci.* **200**7, 253, 9116.
- (397) Wang, A.; Zhou, C.; Wang, H.; Shen, S.; Xue, J.; Ouyang, P. Chin. J. Chem. Eng. **2007**, 15, 788.
- (398) Kahraman, M.; Bayramoglu, G.; Kayamanapohan, N.; Gungor, A. *Food Chem.* **2007**, *104*, 1385.
- (399) Shah, P.; Sridevi, N.; Prabhune, A.; Ramaswamy, V. *Microporous Mesoporous Mater.* **2008**, *116*, 157.
- (400) de Lathouder, K. M.; van Benthem, D. T. J.; Wallin, S. A.; Mateo, C.; Lafuente, R. F.; Guisan, J. M.; Kapteijn, F.; Moulijn, J. A. *J. Mol. Catal. B: Enzym.* **2008**, *50*, 20.
- (401) Kim, M. I.; Kim, J.; Lee, J.; Shin, S.; Na, H. B.; Hyeon, T.; Park, H. G.; Chang, H. N. *Microporous Mesoporous Mater.* **2008**, *111*, 18.
- (402) Fernandez, R. E.; Bhattacharya, E.; Chadha, A. Appl. Surf. Sci. 2008, 254, 4512.
- (403) Ma, J.; Liang, Z.; Qiao, X.; Deng, Q.; Tao, D.; Zhang, L.; Zhang, Y. Anal. Chem. **2008**, 80, 2949.
- (404) Manyar, H. G.; Gianotti, E.; Sakamoto, Y.; Terasaki, O.; Coluccia, S.; Tumbiolo, S. J. Phys. Chem. C **2008**, *112*, 18110.
- (405) Zimmermann, J.; Rabe, M.; Verdes, D.; Seeger, S. Langmuir 2008, 24, 1053.
- (406) Libertino, S.; Giannazzo, F.; Aiello, V.; Scandurra, A.; Sinatra, F.; Renis, M.; Fichera, M. *Langmuir* **2008**, *24*, 1965.
- (407) Pita, M.; Kramer, M.; Zhou, J.; Poghossian, A.; Schoning, M. J.; Fernandez, V. M.; Katz, E. *ACS Nano* **2008**, *2*, 2160.

- (408) Lee, C. C.; Chiang, H. P.; Li, K. L.; Ko, F. H.; Su, C. Y.; Yang, Y. S. Anal. Chem. **2009**, *81*, 2737.
- (409) Kannan, K.; Jasra, R. V. J. Mol. Catal. B: Enzym. 2009, 56, 34.

(410) Tanvir, S.; Pantigny, J.; Boulnois, P.; Pulvin, S. J. Membr. Sci. 2009, 329, 85.

- (411) Wang, A.; Zhou, C.; Du, Z.; Liu, M.; Zhu, S.; Shen, S.; Ouyang, P. J. Biosci. Bioeng. **2009**, 107, 219.
- (412) Wang, Y.; Hasebe, Y. Talanta 2009, 79, 1135.
- (413) Upadhyay, S.; Rao, G. R.; Sharma, M. K.; Bhattacharya, B. K.; Rao, V. K.; Vijayaraghavan, R. *Biosens. Bioelectron.* **2009**, 25, 832.
- (414) Shi, B.; Wang, Y.; Guo, Y.; Wang, Y.; Wang, Y.; Guo, Y.; Zhang, Z.; Liu, X.; Lu, G. *Catal. Today* **2009**, *148*, 184.
- (415) Lee, Y.; Kim, J.; Kim, S.; Jang, W.-D.; Park, S.; Koh, W.-G. J. Mater. Chem. 2009, 19, 5643.
- (416) Arrabito, G.; Musumeci, C.; Aiello, V.; Libertino, S.; Compagnini, G.; Pignataro, B. *Langmuir* **2009**, *25*, 6312.
- (417) Goykhman, I.; Korbakov, N.; Bartic, C.; Borghs, G.; Spira, M. E.; Shappir, J.; Yitzchaik, S. J. Am. Chem. Soc. **2009**, 131, 4788.
- (418) Song, Y. S.; Lee, J. H.; Kang, S. W.; Kim, S. W. Food Chem. 2010, 123, 1.
- (419) Park, S. W.; Kim, Y. I.; Chung, K. H.; Hong, S. I.; Kim, S. W. React. Funct. Polym. 2002, 51, 79.
- (420) Kroll, S.; Treccani, L.; Rezwan, K.; Grathwohl, G. J. Membr. Sci. **2010**, 365, 447.
- (421) Bautista, L. F.; Morales, G.; Sanz, R. Bioresour. Technol. 2010, 101, 8541.
- (422) Pedrosa, V. A.; Paliwal, S.; Balasubramanian, S.; Nepal, D.; Davis,
- V.; Wild, J.; Ramanculov, E.; Simonian, A. Colloids Surf., B 2010, 77, 69. (423) Wang, M.; Qi, W.; Yu, Q.; Su, R.; He, Z. Biochem. Eng. J. 2010,
- 52, 168.
 (424) Shi, B.; Wang, Y.; Ren, J.; Liu, X.; Zhang, Y.; Guo, Y.; Guo, Y.; Lu, G. J. Mol. Catal. B: Enzym. 2010, 63, 50.
- (425) Zheng, X. T.; Yang, H. B.; Li, C. M. Anal. Chem. **2010**, 82, 5082.
- (426) Sun, L.; Ma, J.; Qiao, X.; Liang, Y.; Zhu, G.; Shan, Y.; Liang, Z.;
- Zhang, L.; Zhang, Y. Anal. Chem. 2010, 82, 2574.
- (427) Arrabito, G.; Pignataro, B. Anal. Chem. 2010, 82, 3104.
- (428) Areskogh, D.; Henriksson, G. Process Biochem. 2011, 46, 1071.
- (429) Alptekin, O.; Tukel, S. S.; Yildirim, D.; Alagoz, D. Enzyme Microb. Technol. 2011, 49, 547.
- (430) Wang, A.; Wang, M.; Wang, Q.; Chen, F.; Zhang, F.; Li, H.; Zeng, Z.; Xie, T. *Bioresour. Technol.* **2011**, *102*, 469.
- (431) Yilmaz, E.; Can, K.; Sezgin, M.; Yilmaz, M. Bioresour. Technol. 2011, 102, 499.
- (432) Meizler, A.; Roddick, F.; Porter, N. Chem. Eng. J. 2011, 172, 792. (433) Song, Y. S.; Lee, H. U.; Lee, J. H.; Park, C.; Kim, S. W. Process
- (124) Biochem. 2011, 46, 817.
- (434) Xu, Y.-Q.; Zhou, G.-W.; Wu, C.-C.; Li, T.-D.; Song, H.-B. Solid State Sci. 2011, 13, 867.
- (435) Wang, Y.; Hasebe, Y. J. Environ. Sci. (Beijing, China) 2011, 23, 1038.
- (436) Vinoba, M.; Lim, K. S.; Lee, S. H.; Jeong, S. K.; Alagar, M. Langmuir **2011**, 27, 6227.
- (437) Wang, C.; Zhou, G.; Xu, Y.; Chen, J. J. Phys. Chem. C 2011, 115, 22191.
- (438) Guyomard-Lack, A.; Delorme, N.; Moreau, C.; Bardeau, J. F.; Cathala, B. *Langmuir* **2011**, *27*, 7629.
- (439) Zhao, G.; Wang, J.; Li, Y.; Huang, H.; Chen, X. *Biochem. Eng. J.* **2012**, *68*, 159.
- (440) Lee, H. U.; Park, C.; Kim, S. W. Process Biochem. **2012**, 47, 1282.
- (441) Albertini, A. V. P.; Cadena, P. G.; Silva, J. L.; Nascimento, G. A.; Reis, A. L. S.; Freire, V. N.; Santos, R. P.; Martins, J. L.; Cavada, B. S.;
- Neto, P. J. R.; Pimentel, M. C. B.; Martínez, C. R.; Porto, A. L. F.; Lima Filho, J. L. *Chem. Eng. J.* **2012**, *187*, 341.
- (442) Deep, A.; Tiwari, U.; Kumar, P.; Mishra, V.; Jain, S. C.; Singh, N.; Kapur, P.; Bharadwaj, L. M. *Biosens. Bioelectron.* **2012**, *33*, 190.
- (443) Ma, Y.-X.; Li, Y.-F.; Zhao, G.-H.; Yang, L.-Q.; Wang, J.-Z.; Shan, X.; Yan, X. *Carbon* **2012**, *50*, 2976.
- (444) Song, Y. S.; Shin, H. Y.; Lee, J. Y.; Park, C.; Kim, S. W. Food Chem. 2012, 133, 611.

- (445) Ida, J.; Matsuyama, T.; Yamamoto, H. *Biochem. Eng. J.* **2000**, *5*, 179.
- (446) You, Q.; Yin, X.; Gu, X.; Xu, H.; Sang, L. *Bioprocess Biosyst. Eng.* 2011, 34, 757.
- (447) Wang, X. R.; Zhang, S. P.; Wang, P. Biotechnol. Lett. 2011, 33, 1831.
- (448) Yokoyama, K.; Tamiya, E.; Karube, I. *Electroanalysis* **1991**, *3*, 469.
- (449) Murakami, Y.; Suda, M.; Yokoyama, K.; Takeuchi, T.; Tamiya, E.; Karube, I. *Microchem. J.* **1994**, *49*, 319.
- (450) Wilson, N. G.; McCreedy, T.; Greenway, G. M. Analyst 2000, 125, 237.
- (451) Huang, J.; Zhao, R.; Wang, H.; Zhao, W.; Ding, L. Biotechnol. Lett. 2010, 32, 817.
- (452) Li, W.; Yuan, R.; Chai, Y.; Zhou, L.; Chen, S.; Li, N. J. Biochem. Biophys. Methods 2008, 70, 830.
- (453) Kim, D. N.; Lee, Y.; Koh, W.-G. Sens. Actuators, B: Chem. 2009, 137, 305.
- (454) Cho, S.-H.; Shim, J.; Yun, S.-H.; Moon, S.-H. Appl. Catal., A 2008, 337, 66.
- (455) Cabana, H.; Alexandre, C.; Agathos, S. N.; Jones, J. P. Bioresour. Technol. 2009, 100, 3447.
- (456) Rekuc, A.; Bryjak, J.; Szymanska, K.; Jarzebski, A. B. *Bioresour*. *Technol.* **2010**, *101*, 2076.
- (457) Zeng, L.; Luo, K.; Gong, Y. J. Mol. Catal. B: Enzym. 2006, 38, 24.
- (458) Huang, J.; Liu, Y.; Wang, X. J. Mol. Catal. B: Enzym. 2008, 55, 49.
- (459) Abdullah, A. Z.; Sulaiman, N. S.; Kamaruddin, A. H. Biochem. Eng. J. 2009, 44, 263.
- (460) Huang, J.; Liu, Y.; Wang, X. J. Mol. Catal. B: Enzym. 2009, 57, 10. (461) Chaubey, A.; Parshad, R.; Taneja, S. C.; Qazi, G. N. Process Biochem. 2009, 44, 154.
- (462) Hu, B.; Pan, J.; Yu, H.-L.; Liu, J.-W.; Xu, J.-H. Process Biochem. **2009**, 44, 1019.
- (463) Mitchell, S.; Bonilla, A.; Pérez-Ramírez, J. Mater. Chem. Phys. 2011, 127, 278.
- (464) Ranjbakhsh, E.; Bordbar, A. K.; Abbasi, M.; Khosropour, A. R.; Shams, E. *Chem. Eng. J.* **2012**, *179*, 272.
- (465) Malpass, C. A.; Millsap, K. W.; Sidhu, H.; Gower, L. B. J. Biomed. Mater. Res. 2002, 63, 822.
- (466) Baur, B.; Howgate, J.; von Ribbeck, H. G.; Gawlina, Y.; Bandalo, V.; Steinhoff, G.; Stutzmann, M.; Eickhoff, M. *Appl. Phys. Lett.* **2006**, *89*, 183901.
- (467) Li, B.; Zhang, J.; Dai, F.; Xia, W. Carbohydr. Polym. 2012, 88, 206.
 (468) Park, S.; Lee, Y.; Kim, D. N.; Park, S.; Jang, E.; Koh, W.-G. React. Funct. Polym. 2009, 69, 293.
- (469) Cordeiro, A. L.; Lenk, T.; Werner, C. J. Biotechnol. 2011, 154, 216.
- (470) Miyazaki, M.; Kaneno, J.; Uehara, M.; Fujii, M.; Shimizu, H.; Maeda, H. *Chem. Commun.* **2003**, 648.
- (471) Puleo, D. A. J. Biomed. Mater. Res. 1995, 29, 951.
- (472) Li, Y.; Yan, B.; Deng, C.; Yu, W.; Xu, X.; Yang, P.; Zhang, X. Proteomics **200**7, 7, 2330.
- (473) Zhao, J.; Wu, D.; Zhi, J. Bioelectrochemistry 2009, 75, 44.
- (474) Liu, T.; Wang, S.; Chen, G. Talanta 2009, 77, 1767.
- (475) Vashist, S. K.; Zheng, D.; Sheu, F. S.; Al-Rubeaan, K. A. A mediator-less electrochemical glucose sensing procedure employing the leach-proof covalent binding of an enzyme(s) to electrodes and products thereof. WO 2013165318 A1, November 7, 2013.
- (476) Kjellander, M.; Mazari, A. M.; Boman, M.; Mannervik, B.; Johansson, G. *Anal. Biochem.* **2014**, 446, 59.
- (477) Yu, D.; Ma, Y.; Xue, S. J.; Jiang, L.; Shi, J. *LWT-Food Sci. Technol.* **2013**, *50*, 519.
- (478) Kopp, W.; da Costa, T. P.; Pereira, S. C.; Jafelicci, M., Jr.; Giordano, R. C.; Marques, R. F. C.; Araújo-Moreira, F. M.; Giordano, R. L. C. Process Biochem. **2014**, *49*, 38.
- (479) Aissaoui, N.; Landoulsi, J.; Bergaoui, L.; Boujday, S.; Lambert, J. F. *Enzyme Microb. Technol.* **2013**, *52*, 336.
- (480) Anuar, S. T.; Zhao, Y.-Y.; Mugo, S. M.; Curtis, J. M. J. Mol. Catal. B: Enzym. **2013**, 92, 62.

- (481) Bayramoglu, G.; Akbulut, A.; Arica, M. Y. J. Hazard. Mater. 2013, 244, 528.
- (482) Chauhan, N.; Narang, J.; Sunny; Pundir, C. S. Enzyme Microb. Technol. 2013, 52, 265.
- (483) Gupta, K.; Jana, A. K.; Kumar, S.; Maiti, M. J. Mol. Catal. B: Enzym. 2013, 98, 30.
- (484) Liu, T.; Zhao, Y.; Wang, X.; Li, X.; Yan, Y. Bioresour. Technol. 2013, 132, 99.
- (485) Min, W.; Cui, S.; Wang, W.; Chen, J.; Hu, Z. Anal. Biochem. 2013, 438, 32.
- (486) Raghavendra, T.; Basak, A.; Manocha, L. M.; Shah, A. R.; Madamwar, D. Bioresour. Technol. 2013, 140, 103.
- (487) Bhange, P.; Sridevi, N.; Bhange, D. S.; Prabhune, A.; Ramaswamy, V. Int. J. Biol. Macromol. 2014, 63, 218.
- (488) Hou, J.; Dong, G.; Ye, Y.; Chen, V. J. Membr. Sci. 2014, 452, 229.