

# 1 **Impact of Microstructure on the Electron-hole Interaction in Metal Halide**

## 2 **Perovskites**

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25 **Abstract**

26 Despite the remarkable progress in the performance of devices based on the metal halide  
27 perovskite semiconductor family, there is still a lack of consensus on their fundamental  
28 photophysical properties. Here, using magneto-optical transmission spectroscopy we elucidate  
29 the impact of the microstructure on the Coulomb interaction between photo-created electron-  
30 hole pairs in methylammonium lead triiodide (MAPbI<sub>3</sub>) and the triple-cation lead mixed-  
31 halide composition, Cs<sub>0.05</sub>(MA<sub>0.17</sub>FA<sub>0.83</sub>)<sub>0.95</sub>Pb(I<sub>0.83</sub>Br<sub>0.17</sub>)<sub>3</sub> (Cs: Cesium, MA:  
32 methylammonium, FA: formamidinium) by investigating thin films with a wide range of grain  
33 sizes from tens of nanometers to microns. At low temperatures, in which thermal fluctuations  
34 of the interactions are frozen and the rotational disorder of the organic cation is negligible, the  
35 exciton binding energy and reduced effective mass of carriers remain effectively unchanged  
36 with grain size. We conclude that the microstructure plays a negligible role in the Coulomb  
37 interaction of the photo-created electron-hole pairs, in contrast to previous reports. This  
38 renewed understanding of the relationship between these fundamental electronic properties  
39 and the microstructure is critical for future fundamental studies and improving device design.

40 *Key words:* hybrid perovskites, microstructure, exciton binding energy, excitonic reduced  
41 effective mass, magneto-optical spectroscopy

42 **Broader context:**

43 Metal halide perovskites such as methylammonium lead iodide (CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>) and alloyed  
44 mixed-cation lead mixed-halide counterparts such as Cs<sub>0.05</sub>(MA<sub>0.17</sub>FA<sub>0.83</sub>)<sub>0.95</sub>Pb(I<sub>0.83</sub>Br<sub>0.17</sub>)<sub>3</sub> are  
45 creating enormous excitement for their use in high-performance solar cells with potential for  
46 exceptionally low capital-intensity for production. An important parameter that dictates the  
47 design of a solar cell is the strength of the Coulomb interaction between the photo-generated  
48 electron in the conduction band and the hole in the valence band (i.e. the exciton binding

49 energy). Recently, it has been suggested that the exciton binding energy of the perovskite  
50 material is influenced by the size of the grains of the perovskite film, and therefore different  
51 device design will be required depending on the grain size. Here, we perform magneto-optic  
52 measurements on films with different grain sizes to directly show that the microstructure has a  
53 negligible influence on the excitonic properties in the samples, and that excitons do not play a  
54 significant role in films with any of the polycrystalline morphologies studied. This means that  
55 excitons should not be factor for the design of optoelectronic devices based on polycrystalline  
56  $\text{CH}_3\text{NH}_3\text{PbI}_3$  and similar materials with different grain sizes, making future device  
57 architecture optimisation less constrained.

## 58 **Introduction**

59 Solar cells based on lead halide perovskites (with general chemical formula of  $\text{ABX}_3$ ; A =  
60  $\text{CH}_3\text{NH}_3^+$ ,  $\text{HC}(\text{NH}_2)_2^+$  or  $\text{Cs}^+$ ; B =  $\text{Pb}^{2+}$ ; X = I<sup>-</sup>, Br<sup>-</sup> or Cl<sup>-</sup>) have achieved astonishingly high  
61 power conversion efficiencies exceeding 22% in only seven years since they were first used as  
62 sensitizers in photovoltaic solar cells<sup>1-3</sup>. Since then, the remarkable properties of these  
63 crystalline semiconductors, such as the tunability of their optoelectronic properties through  
64 fabrication of layers with different morphologies, compositions and crystal qualities<sup>4-7</sup>, have  
65 been further exploited to develop a variety of promising optoelectronic devices<sup>8-11</sup>.

66 As for any emerging semiconductor material deployable in optoelectronic devices, the  
67 strength of the Coulomb interaction between the photo-generated electron in the conduction  
68 band and the hole in the valence band is of high importance. This was a focus of several  
69 studies on perovskites about two decades ago<sup>12-14</sup> and has recently regained attention<sup>15-18</sup>  
70 mainly due to the promising outlook of perovskites for solar photovoltaics<sup>10</sup>, light-emitting  
71 diodes<sup>11</sup> and lasing applications<sup>19</sup> with polycrystalline thin films and single crystals of  
72 superior quality than those synthesized two decades ago.

73 The latest studies have come to a consensus that the exciton binding energy – alternatively  
74 known as effective Rydberg ( $Ry^*$ ) in the Wannier-Mott exciton model – of well-studied  
75  $CH_3NH_3PbI_3$  and  $CH_3NH_3PbBr_3$  polycrystalline thin films is  $\sim 5$ - $12$  meV<sup>15, 17, 18</sup> and  $\sim 25$  meV  
76<sup>16, 18</sup> in the room temperature phase, respectively. These low values, with respect to thermal  
77 energy at room temperature ( $k_b \times T \approx 26$  meV;  $k_b$  is the Boltzmann constant and  $T$  is the  
78 temperature in Kelvin), are due to generally weak many-body effects (i.e. Coulombic  
79 interactions) in these semiconductor systems<sup>20</sup>. Nevertheless, there is ongoing debate  
80 concerning the possible tunability of this parameter with variations of the microstructure and  
81 hence crystal quality of the perovskite layer<sup>21, 22</sup>, being speculated to influence the photo-  
82 excited exciton binding energies.

83 It has been proposed that disordered organic cation dipole domains in small and defective  
84 crystals with a large portion of poly-crystallinity can create a large electrostatic potential  
85 fluctuation which causes a substantial screening of the photo-generated excitons<sup>22</sup>. This could  
86 lead to a very small value of  $Ry^*$  on the order of  $\sim 2$ - $5$  meV<sup>23, 24</sup> meaning that the material  
87 behaves as if in the non-excitonic regime. On the other hand, large and defect-free crystals can  
88 exhibit an order of magnitude greater values of  $Ry^*$ <sup>22, 25</sup>. As a result, one current hypothesis is  
89 that it is not possible to assign a universal value to the exciton binding energy of the  
90 commonly used  $CH_3NH_3PbI_3$  hybrid lead halide perovskites.

91 A key consequence of this hypothesis could be a change in the working regime (i.e. photo-  
92 excited carrier dissociation and transport) of solar cells fabricated from active layers with  
93 diverse microstructures. Knowledge of the value of the exciton binding energy is essential  
94 because it determines the nature of the majority of the photo-generated species (i.e. free  
95 carriers versus excitons) in a semiconductor. For devices comprised of materials with binding  
96 energies greater than the probability of thermal dissociation at the working temperature, there  
97 would need to be an additional mechanism for exciton dissociation into free carriers, such as

98 an additional heterojunction, through which they can readily contribute to the photocurrent <sup>26</sup>.  
99 In the case that the binding energies are so low that the fraction of photo-excited species that  
100 are excitons is negligible, the n- and p-type contacts in the heterojunction structure of the solar  
101 cell (cf. most perovskite solar cells) would simply act as charge-selective contacts for the  
102 photo-generated electrons and holes, respectively, facilitating simpler device designs.  
103 Altogether, these points highlight the need to fully understand the exciton-morphology  
104 relationship.

105 Here, we elucidate the influence of the microstructure, manipulated through the perovskite  
106 fabrication method, on the exciton binding energy  $Ry^*$  and the excitonic reduced mass ( $\mu$ ) for  
107 a variety of  $\text{CH}_3\text{NH}_3\text{PbI}_3$  crystal sizes covering the size range reported in devices to date. The  
108 impact of the local order and the degree of poly-crystallinity on electron-hole interaction is  
109 unraveled through a direct measurement of  $Ry^*$  and  $\mu$  for thin films with different grain sizes.  
110 These measurements are performed at low temperature (2 K) at which the impact of  $\text{CH}_3\text{NH}_3^+$   
111 thermal orientational dynamic rearrangement is mitigated. We provide strong evidence that  
112 the grain size has negligible influence on the exciton binding energy and effective mass in  
113  $\text{CH}_3\text{NH}_3\text{PbI}_3$  and, thus, does not affect the dissociation probability of the bulk excitons and  
114 their stability in the photo-excited charge carrier population.

## 115 **Results and Discussion**

### 116 *Sample Morphology*

117 A wide range of fabrication methods, from single-step <sup>27</sup> to sequential deposition <sup>28</sup> and  
118 solution-processed <sup>29</sup> to vapor-assisted <sup>30</sup> techniques, have been employed to fabricate the  
119 active layer of perovskite optoelectronic devices in the literature. Here, different deposition  
120 methods, which have been shown to give corresponding high-performance solar cells with  
121 efficiencies of >16 % <sup>6, 31, 32</sup>, are used to controllably tune the perovskite grain size in the

122 films. These samples with  $\text{CH}_3\text{NH}_3\text{PbI}_3$  chemical composition but fabricated via different  
123 methods with various resulting morphologies simulate the range of different degrees of  
124 structural disorder and poly-crystallinity in the  $\text{CH}_3\text{NH}_3\text{PbI}_3$  layer in high-performance  
125 perovskite devices fabricated to date <sup>22</sup>. The top-view scanning electron microscope (SEM)  
126 images of the samples are presented in **Figure 1A-D**. The morphologies vary from a planar  
127 polycrystalline film with grain sizes of  $772\pm 227$  nm <sup>32</sup> (**A**; large grain polycrystalline  
128 annotated **LPC**) and polycrystalline film with grain sizes of  $214\pm 57$  nm <sup>31</sup> (**B**; small grain  
129 polycrystalline annotated **SPC**), to small crystals of  $291\pm 64$  nm fabricated by a two-step  
130 dipping technique <sup>6</sup> (**C**; annotated **SC**) and also perovskite infiltrated into a  $\sim 2$   $\mu\text{m}$  thick  
131 mesoporous Aluminum oxide scaffold (mp- $\text{Al}_2\text{O}_3$ ) with grain sizes of  $< 50$  nm (**D**; annotated  
132 **MP**). As can be seen in the top-view SEM image of the **MP** sample, there is no visible  
133 capping layer formed on top of the mesoscopic phase and, therefore, a potential artefact from  
134 the larger grain sizes of a capping layer does not influence the related data analysis.

### 135 *Magneto-optical Transmission Measurements*

136 Studying the optical behavior of excitons under external fields has proven to be a powerful  
137 tool to reveal properties of these quasi-particles in semiconductors <sup>33</sup>. Here, we conducted  
138 optical transmission measurements under high magnetic fields ( $B$ ) up to 70T on the samples  
139 presented in **Figure 1**. Typical transmission spectra for the large grain polycrystalline thin  
140 film sample, showing a strong 1s excitonic transition, are given in **Figure 2A**. For  
141 unambiguous identification of the weaker optical transitions, we show in **Figure 2B** the ratio  
142 of the transmission spectrum at a specific high magnetic field over the transmission spectrum  
143 taken at zero field, illustrating the free-carrier transitions. In **Figure 2C** we show transmission  
144 measurements at several monochromatic probe energies under very high magnetic field  
145 ( $< 150\text{T}$ ) achieved using a short pulse technique. The related spectra for other morphologies  
146 are provided in the Supplementary Information, **Figures S1-S3**.

147 We note that a hydrogen-like excitonic behavior <sup>34</sup> can be assumed for CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> in the  
 148 low-temperature orthorhombic phase. The strong minimum in the transmission spectra  
 149 (**Figure 2A**), attributed to the 1s state of the hydrogen-like free exciton <sup>17</sup>, blue shifts with  
 150 increase of the magnetic field. The second minimum, located at slightly higher energies than  
 151 the 1s state, is attributed to the first excitonic excited state (labelled as 2s in **Figure 2A**) and  
 152 also exhibits a blue-shift with magnetic field.

153 The Landau levels, the free carrier states with quantum numbers  $N=0,1,2,\dots$  formed by  
 154 quantization of the particle's motion in the plane perpendicular to the magnetic field <sup>33</sup>, can be  
 155 observed at higher energies in the strong magnetic field regime in which  $\hbar\omega_c \gg Ry^*$  (  
 156  $\hbar\omega_c = \hbar eB/\mu$ ), where  $\hbar$  and  $e$  are Planck's constant and the electron charge, respectively,  $\omega_c$  is  
 157 the cyclotron frequency, and  $\mu^{-1} = m_e^{-1} + m_h^{-1}$  is the reduced mass of the exciton ( $m_e$  and  $m_h$  are  
 158 electron and hole masses, respectively). The free carrier Landau level transitions are dominant  
 159 at high magnetic fields with transition energy described by  $E(B) = E_g + (N + 1/2)\hbar\omega_c$  <sup>16, 17</sup>  
 160 where  $E_g$  is the bandgap. We can then fit the measured optical transitions between Landau  
 161 levels to extract the reduced mass,  $\mu$ , independent of the exciton binding energy. The value of  
 162 the reduced mass thus obtained is used as a fixed parameter in the fit of the hydrogen-like  
 163 excitonic transitions to extract the exciton binding energy, based on a model which describes  
 164 the hydrogen atoms in high magnetic field <sup>35</sup>. In this model, a dimensionless parameter  $\gamma$   
 165 which is defined as  $\gamma = \hbar\omega_c/2Ry^*$  is used. Makado and McGill calculated the transition  
 166 energies of excitonic states  $E_n(\gamma)$  as a function of quantum number  $n$  up to 4 and  $\gamma$  ranging  
 167 from 0 to 10 <sup>35</sup>. Given values of the magnetic field  $B$  and the reduced mass  $\mu$ , the binding  
 168 energy  $Ry^*$  is the only fitting parameter for the full set of hydrogenic transitions.

169 The complete set of these excitonic and free carrier transition energies for the **LPC** sample is  
170 plotted in the fan diagram in **Figure 2D**. The observation of many Landau levels in the  
171 magnetic field <150T allows precise determination of the reduced mass. Knowledge of the  
172 reduced mass in turn imposes strong constraints on fitting the excitonic binding energy. An  
173 additional influential constraint on the value of the binding energy is set if the higher  
174 excitonic transition resonances such as the 2s state are resolved, as for the **LPC** morphology.  
175 The observation of a well-resolved 2s state at intermediate magnetic field fixes the transition  
176 energy of 2s at zero magnetic field using the model described earlier. The energy separation  
177 of 1s-2s at zero magnetic field follows the series of 3D hydrogen-like energy states,  
178  $E_n = E_g - Ry^*/n^2$  where  $E_n$  is the excitonic transition energy at zero magnetic field. This  
179 gives a strong constraint on the exciton binding energy.

180 The fitting to the hydrogen-like excitonic (dashed lines) and free carrier (solid lines)  
181 transitions, as explained in detail in refs. <sup>16, 17, 35</sup>, is presented in **Figure 2D** and the  $Ry^*$  and  $\mu$   
182 values for the **LPC** sample are calculated to be  $16\pm 1$  meV and  $0.102\pm 0.002 m_0$ , respectively.  
183 For the morphologies other than **LPC** (i.e. **SPC**, **SC** and **MP**), the identification of free carrier  
184 transitions at high energies allows us to obtain the values of the the reduced masses. The  
185 extracted  $\mu$  values are presented in **Table 1** and are very close to the value originally reported  
186 by Miyata *et al.* ( $\mu = 0.104\pm 0.003 m_0$ ) <sup>17</sup>. Importantly, we find that the microstructure has  
187 negligible influence on the reduced mass. This indicates that the polaron coupling constant,  
188 incorporating the effective longitudinal optical (LO) phonon energy and the optical and static  
189 dielectric constants of the medium, which couples to the band-edge bare effective masses to  
190 form the polaronic masses <sup>18, 36</sup>, is only minimally influenced by the variation in  
191 microstructure.



192 In the absence of a clear observation of the 2s excitonic transitions for the **SPC**, **SC** and **MP**  
193 samples, the global fit to the full excitonic and free-carrier transitions was employed and the  
194 resulting unchanged  $Ry^*$  values are presented in **Table 1**. To visualize the negligible change in  
195 the optical response under external magnetic fields of the  $\text{CH}_3\text{NH}_3\text{PbI}_3$  films with different  
196 morphology, the overlaid fan diagrams of the complete set of excitonic and free carrier optical  
197 transitions after shifting by the corresponding bandgap offsets are presented in **Figure 3(A)**.  
198 Here we clearly show that these  $\text{CH}_3\text{NH}_3\text{PbI}_3$  samples, although composed of dramatically  
199 different grain sizes and morphologies, exhibit similar magneto-optical behaviour.  
200 Furthermore, the XRD patterns (**Figure S4**) show that the  $\text{CH}_3\text{NH}_3\text{PbI}_3$  **SPC** and **SC** thin  
201 films have some  $\text{PbI}_2$  impurities, indicating that the presence of small amounts of  $\text{PbI}_2$  also  
202 does not affect the excitonic properties of perovskites. We also note that, although the changes  
203 in the grain size and morphology do not influence the low-temperature excitonic properties of  
204 perovskites, they can affect the optoelectronic response of these materials including their PL  
205 behaviour and the fraction of retained tetragonal phase within the orthorhombic phase upon  
206 lowering the temperature <sup>37-39</sup>.

207 It was recently shown that when replacing  $\text{CH}_3\text{NH}_3^+$  by a larger monovalent organic cation,  
208 formamidinium ( $\text{HC}(\text{NH}_2)_2^+$ ), the excitonic characteristics of the corresponding perovskite,  
209  $Ry^*$  and  $\mu$ , decreased by less than 15 % <sup>16</sup>. The larger steric size of formamidinium compared  
210 to methylammonium results in a larger equatorial bond angle (lead-iodide-lead) and, thus, a  
211 narrower optical bandgap <sup>40</sup>. The variation of the crystal size would result in a modified  
212 interplay between the organic and the inorganic moieties in the perovskite structure, which in  
213 turn changes the strain induced on the crystal structure and thus, impacts the bandgap. The  
214 negligible impact on the excitonic characteristics of the degree of dipolar cation order and  
215 structural arrangement induced by the variation of perovskite microstructure might be  
216 analogous to the small impact of cation substitution on the excitonic properties, with the lead-

217 halogen sub-unit remaining unchanged in both cases. This indeed demonstrates the critical  
218 role of the electronic properties of the inorganic cage in the determination of the excitonic  
219 binding energy and reduced effective mass of perovskites. Moreover, this important  
220 observation agrees with the recent work by Perez-Osorio *et al.* in which the largest  
221 contribution to the static dielectric constant of CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> was demonstrated to originate  
222 from the stretching and rocking vibrational modes of the lead-halogen system rather than the  
223 internal modes of the organic cation <sup>41</sup>.

224 Using the values obtained for the effective dielectric constant ( $\epsilon = \sqrt{Ry^H \times \mu / Ry^* \times m_0}$ ,  
225 where  $Ry^H = 13.6$  eV is the Rydberg constant of the hydrogen atom) and the reduced mass, the  
226 Bohr radius of the bound electron-hole system can be calculated. This is similar for all the  
227 CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> perovskite morphologies and is about 5 nm in the low temperature phase. The  
228 Bohr radius is substantially smaller even than the perovskite crystal size of the **MP** sample  
229 and it is consistent with the absence of any signature of a quantum confinement effect on the  
230 optical transition of the corresponding excitons in the mesoscopic phase.

231 The correct interpretation of the optical absorption spectrum for the exciton binding energy  
232 derivation in the room temperature tetragonal phase of CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> (> 160 K) becomes more  
233 difficult than in the orthorhombic phase as the large spectral broadening at higher  
234 temperatures in the tetragonal phase can potentially mask critical optical transition features  
235 even under high magnetic fields <sup>17</sup>. The increased broadening is most likely due to a  
236 thermally-induced increase in the orientational disorder and rotational dynamics of the  
237 unlocked organic cation above the structural phase transition <sup>42</sup>. The excitonic binding energy  
238 has been reported in the literature <sup>17, 18, 24, 43</sup> to decrease upon orthorhombic-to-tetragonal  
239 phase transition with a likely reason to be the slight increase in the dielectric constant of the

240 medium, which is in turn linked to the increase in the effective phonon energy coupling to the  
241 carriers.

242 We argue that the change in the binding energy at the transition to the tetragonal phase for all  
243 the morphologies investigated in this study would be similar. This can be partly understood by  
244 the dominant influence of thermal disorder on the electrostatic fluctuations at room  
245 temperature over the degree of poly-crystallinity and molecular order which are dominant at  
246 lower temperatures<sup>22</sup>; the latter properties we have already shown above to have insignificant  
247 impact on the electron-hole interaction. This is further supported by a recent theoretical study  
248 by Motta *et al.* in which the impact of the degree of molecular dipole orientational disorder –  
249 which could be induced by the grain size variation – on the binding energy in the high  
250 temperature phase of CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> was found to be only ~10 %<sup>44</sup>.

251 To generalise our findings for CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> to other perovskite compositions, we repeat the  
252 measurements on a state-of-the-art triple-cation lead mixed-halide perovskite,  
253 Cs<sub>0.05</sub>(MA<sub>0.17</sub>FA<sub>0.83</sub>)<sub>0.95</sub>Pb(I<sub>0.83</sub>Br<sub>0.17</sub>)<sub>3</sub><sup>2</sup>. We consider two significantly different morphologies  
254 of this perovskite composition, namely a planar polycrystalline thin film and a film with the  
255 perovskite infiltrated into a thick mp-Al<sub>2</sub>O<sub>3</sub> scaffold (corresponding SEM images are shown  
256 in **Figure S5**). We show in **Figure S6** that both morphologies exhibit similar magneto-optical  
257 responses and, thus, similar fundamental excitonic properties, with the binding energy and  
258 reduced mass of 13±2 meV and 0.096±0.008 *m*<sub>0</sub>, respectively (see **Table S1**). These results  
259 allow us to extend our conclusions to the wider alloyed perovskite family, providing strong  
260 evidence that the microstructure has a negligible impact on the excitonic properties in metal  
261 halide perovskites.

262 Recently, Petrozza and co-workers used temperature-dependent time-resolved transient  
263 absorption (TA) spectroscopy to track the fate of excitons following photo-excitation and to

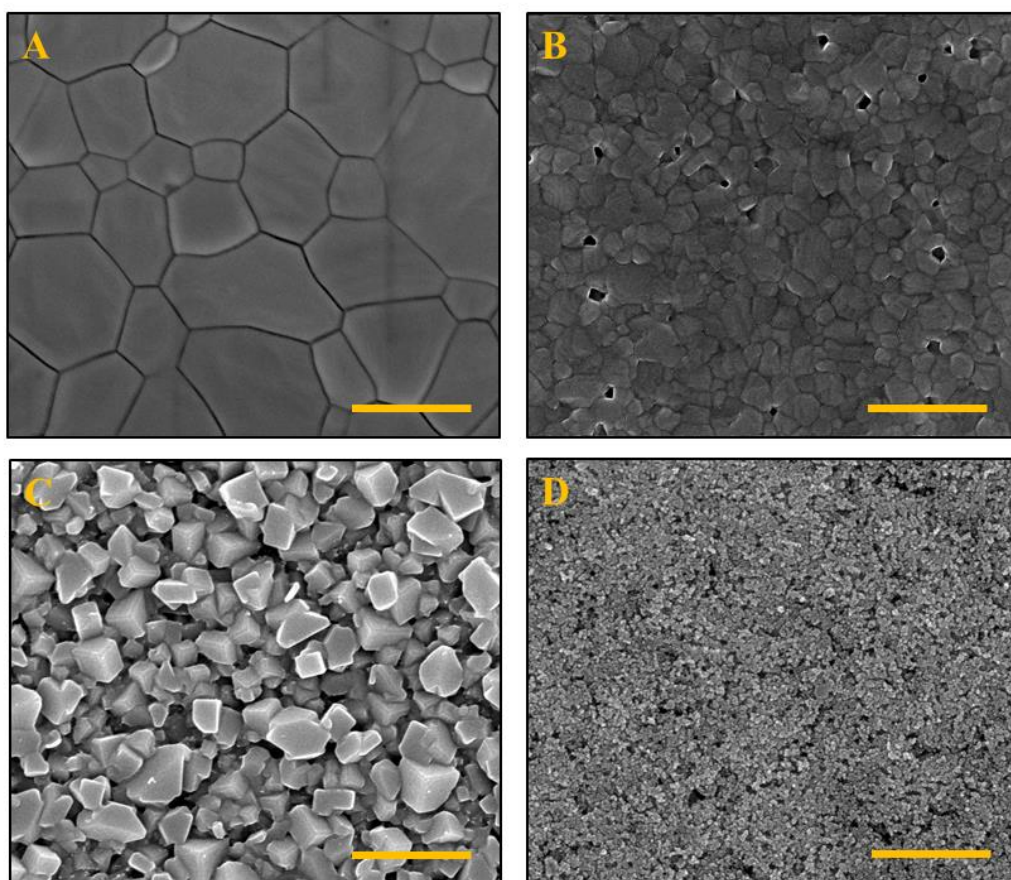
264 understand the relationship between the excitonic properties and film morphology <sup>22</sup>. The  
265 conclusion from this study was that the exciton binding energy and, therefore, exciton  
266 stability can vary depending on the perovskite film microstructure in the tetragonal phase  
267 which is most critical to solar cell operation. Their interpretation was based on the emergence  
268 and evolution of the negative feature located at the high-energy end of the photobleach  
269 signature in the transmission-difference spectrum, which was attributed to self-  
270 renormalization of the exciton energy due to a change in its concentration upon decrease in  
271 temperature and/or grain size variation. We note that the negative feature developing in the  
272 high-energy tail of the TA spectra over time has been proposed to be primarily a consequence  
273 of a bandgap renormalization effect (bandgap narrowing) and of a free carrier absorption,  
274 among other possible mechanisms <sup>45, 46</sup>. The spectral TA analysis used in ref. <sup>22</sup>, which is an  
275 indirect technique for characterising excitonic properties, can also be influenced by  
276 recombination mechanisms which are strongly influenced by the perovskite crystal  
277 morphology. Furthermore, we note from a previous report for CH<sub>3</sub>NH<sub>3</sub>PbBr<sub>3</sub> perovskite that  
278 while  $R_y^*$  probably remains unchanged by increasing the temperature from 2 K to room  
279 temperature, the increase in the broadening of the linear absorption spectra lessens the 1s  
280 transition strength <sup>18, 47</sup>. Although this spectral broadening increase is mainly attributed to  
281 thermally-activated mechanisms, a change in the intrinsic broadening of perovskites with  
282 different microstructures, due for instance to the difference in their inherent structural  
283 disorder, can similarly influence the spectral shape while  $R_y^*$  remains unchanged. This makes  
284 unambiguous interpretation of the linear and transient absorption spectra difficult when  
285 relying on the detection and the strength of the excitonic feature – at the absorption edge – as  
286 these techniques are indirect approaches for  $R_y^*$  determination compared to the direct  
287 approach employed in this study (see **Figure S7** for the zero-field transmission spectra of the  
288 four different CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> morphologies). We note that it is possible there are also transient

289 changes in excitonic populations and properties (e.g. due to dynamic photo-excited changes in  
290 dielectric constant) that will be averaged out in our steady-state measurements.

## 291 **Conclusions**

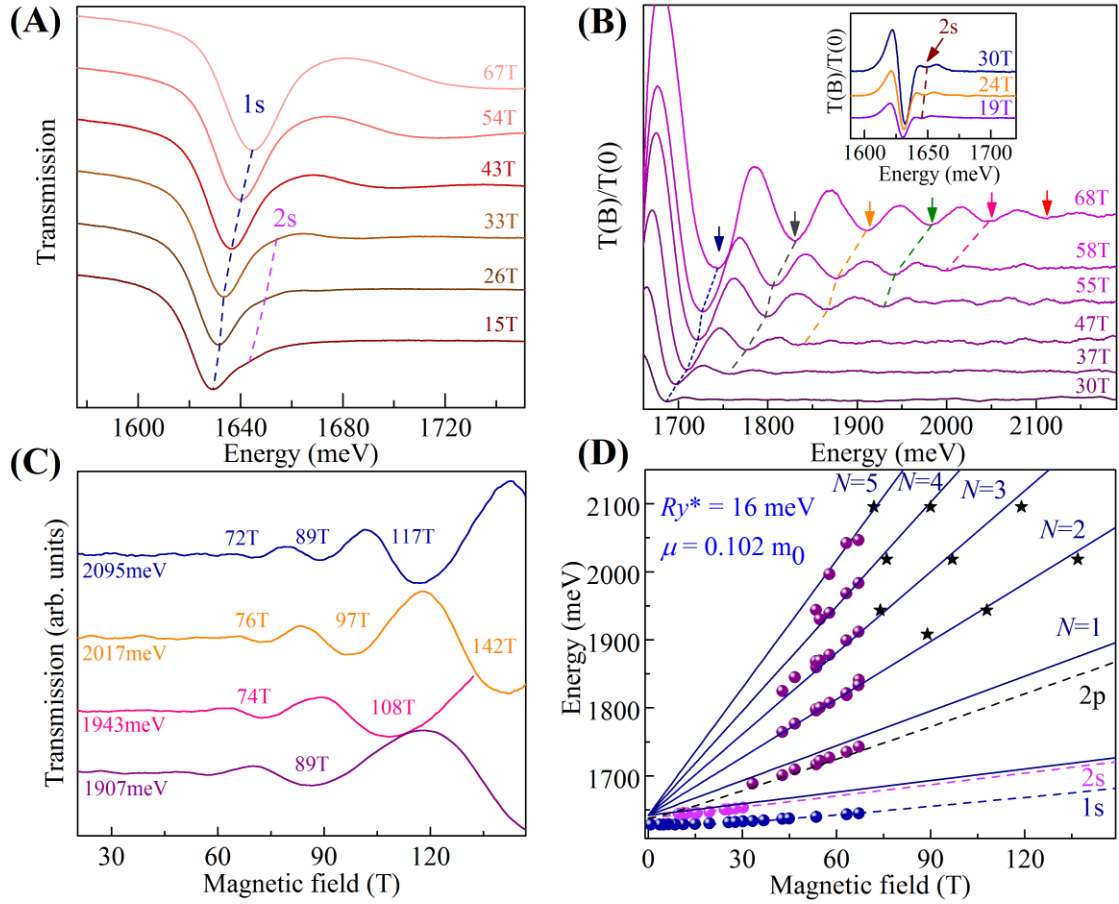
292 We investigated the relationship between the excitonic properties and microstructure of the  
293  $\text{CH}_3\text{NH}_3\text{PbI}_3$  and  $\text{Cs}_{0.05}(\text{MA}_{0.17}\text{FA}_{0.83})_{0.95}\text{Pb}(\text{I}_{0.83}\text{Br}_{0.17})_3$  perovskites using magneto-optical  
294 spectroscopy measurements. We find that the microstructure does not influence the excitonic  
295 binding energy of  $\text{CH}_3\text{NH}_3\text{PbI}_3$  or the broader range of mixed-cation and mixed-halide  
296 alloyed compositions. Thus, we conclude that excitons truly play a negligible role in the  
297 operation of mixed organic-inorganic metal halide-based perovskites regardless of the thin  
298 film deposition technique and final morphology, and that excitons need not be a factor for the  
299 design of optoelectronic devices based on polycrystalline  $\text{CH}_3\text{NH}_3\text{PbI}_3$  and similar materials  
300 with different grain sizes. The universal values of  $\sim 16$  meV and  $0.102\text{-}0.109 m_0$  for the  
301 exciton binding energy and reduced effective mass in the low-temperature phase of  
302  $\text{CH}_3\text{NH}_3\text{PbI}_3$  can be used as guidelines for interpretation of the spectroscopic data such as the  
303 complex spectra of transient absorption measurements. It also implies that the electronic  
304 structure of the inorganic cage –  $(\text{PbI}_3)^-$  in the case of  $\text{CH}_3\text{NH}_3\text{PbI}_3$  – is likely to have the  
305 greatest contribution to the excitonic properties of organic-inorganic lead halide perovskites  
306 rather than the degree of poly-crystallinity and the order of dipolar organic cation domains.

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310 **Figure 1 Morphology of  $\text{CH}_3\text{NH}_3\text{PbI}_3$  samples.** Top-view SEM images of  $\text{CH}_3\text{NH}_3\text{PbI}_3$  layer with various  
 311 morphologies fabricated on glass substrates. **A)** Polycrystalline thin film with grain sizes  $772\pm 227$  nm (LPC). **B)**  
 312 Polycrystalline thin film with grain sizes  $214\pm 57$  nm (SPC). **C)** Small crystals of  $\text{CH}_3\text{NH}_3\text{PbI}_3$  fabricated by the  
 313 two-step solution-processed dipping technique, with resulting crystal sizes  $291\pm 64$  nm (SC). **D)**  $\text{CH}_3\text{NH}_3\text{PbI}_3$   
 314 infiltrated into a thick mesoporous- $\text{Al}_2\text{O}_3$  scaffold with grain sizes  $<50$  nm (MP). Scale bars are  $1\ \mu\text{m}$ .



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**Figure 2 Magneto-optical transmission spectrum and fan chart measured at 2 K.** A) Sequence of typical

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optical transmission spectra of the **LPC** sample measured at the indicated magnetic fields which show the 1s and

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2s excitonic transitions with increased transmission. B) Sequences of the ratios of the transmission spectra in

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magnetic field  $T(B)$  to that measured at zero field  $T(0)$ . The resonant absorption features of the free carrier

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Landau levels correspond to minima. C) Typical results of the low temperature monochromatic transmission as a

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function of magnetic field obtained by the short pulse technique. D) Fan chart of  $\text{CH}_3\text{NH}_3\text{PbI}_3$  for the large grain

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polycrystalline sample. The circular data points are from long pulse field measurements and star symbol points

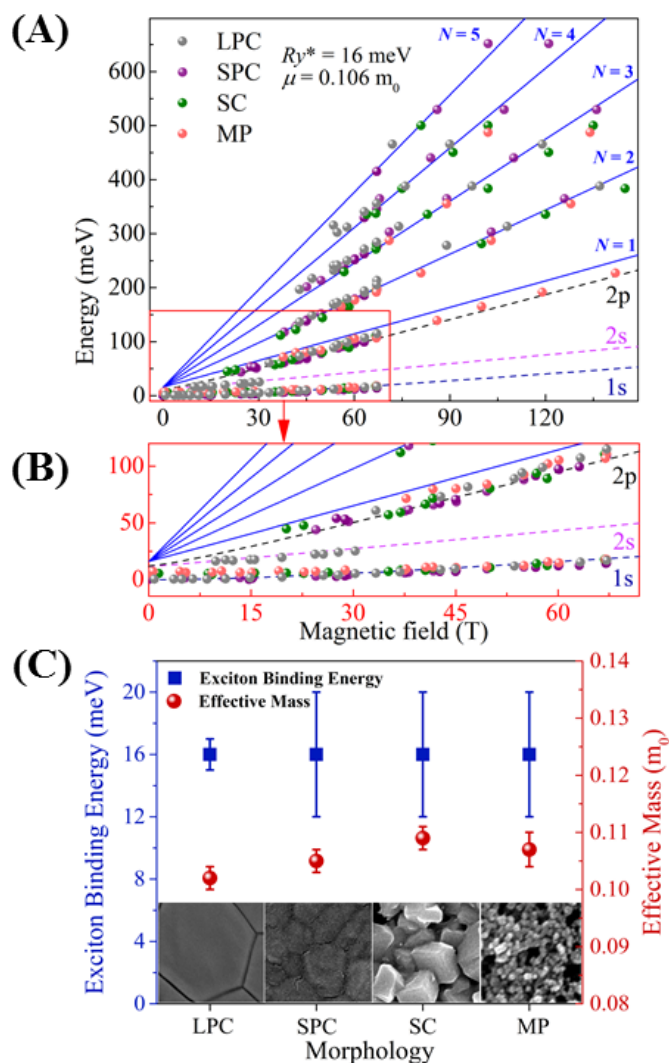
323

are collected by short pulse mega-gauss measurements. The solid lines and the dashed lines are the fits to the set

324

of Landau levels and the excitonic transitions, respectively.

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327 **Figure 3 Comparison of the excitonic properties of the four different  $\text{CH}_3\text{NH}_3\text{PbI}_3$  morphologies.** A)  
 328 Overlaid fan chart of the LPC, SPC, SC and MP samples. The solid lines and the dashed lines are the fits to the  
 329 set of Landau levels and the excitonic transitions, respectively. B) The zoomed-in plot of the excitonic transitions  
 330 region of the fan chart in panel (A). C) The bottom panel illustrates the change in the  $Ry^*$  and  $\mu$  for the various  
 331 morphologies.

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336 **Table 1** Summary of the parameters of the fits to the full Landau fan chart for the four different morphologies of  
 337  $\text{CH}_3\text{NH}_3\text{PbI}_3$  in the low temperature (2 K), orthorhombic phases. The figures in the brackets are the error  
 338 estimates of the parameters.

	<b>Size Distribution (nm)</b>	$E_g$ (meV)	$\mu$ ( $m_0$ )	$Ry^*$ (meV)
<b>Large Polycrystalline</b>	772(227)	1642(2)	0.102(0.002)	16(1)
<b>Small Polycrystalline</b>	214(57)	1643(2)	0.105(0.002)	16(4)
<b>Small Crystal</b>	291(64)	1639(2)	0.109(0.002)	16(4)
<b>Mesoporous</b>	<50	1638(2)	0.107(0.003)	16(4)

339

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356 **Author Contributions**

357 AMS conceived the idea and designed the experiments with SDS and PP. TY, RB, AP and  
358 MA-J fabricated the samples. AMS, SDS and MA-J performed the SEM imaging and XRD  
359 measurements. The magneto-optic experiments were performed, under the supervision of PP,  
360 by ZY, AM, AS, KG, JU and NZ, and the data analyzed by ZY. AH-B and MAG supervised  
361 AMS, Y-BC supervised AP, OP supervised AM, SDS supervised MA-J, SDS and VB  
362 supervised RB. AMS prepared the first draft of the manuscript with technical and editorial  
363 inputs from SDS, PP and RJN. All authors contributed to the final version of the manuscript.

364 **Methods**

365 **CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> Perovskite Sample preparation**

366 *Large grain polycrystalline film (LPC)*

367 LPC perovskite films were prepared using a methylammonium iodide (CH<sub>3</sub>NH<sub>3</sub>I) and lead  
368 acetate Pb(Ac)<sub>2</sub> · 3H<sub>2</sub>O precursor mixture<sup>32</sup>. To generate the perovskite solution, CH<sub>3</sub>NH<sub>3</sub>I  
369 (Dyesol) and Pb(Ac)<sub>2</sub> · 3H<sub>2</sub>O were dissolved in anhydrous N,N-dimethylformamide at a 3:1  
370 molar ratio with final concentration of 30 wt%, and the stabilizer hypophosphorous acid  
371 (HPA) was added at a molar ratio of 7.5% with respect to Pb(Ac)<sub>2</sub> · 3H<sub>2</sub>O. Pb(Ac)<sub>2</sub> · 3H<sub>2</sub>O  
372 (316512) and HPA (214906) were purchased from Sigma Aldrich. Microscope slides and  
373 coverslips were washed sequentially with soap (2% vol. Hellmanex in water), de-ionized  
374 water, isopropanol, acetone and finally treated under oxygen plasma for 10 min. The  
375 precursor solution was spin-coated at 2,000 r.p.m. for 45 s in a nitrogen-filled glovebox, and  
376 the substrates were then dried at room temperature for 10 min before annealing at 100 °C for 5  
377 min. The samples were then stored in a nitrogen-filled glovebox until used.

378 *Small grain polycrystalline film (SPC)*

379 Soda-lime glass substrates 1 mm thick were purchased from J. Melvin Freed. The substrates  
380 were cleaned ultrasonically using soap solution (Helmanex), water, ethanol and iso-propanol  
381 for 15 minutes, each.

382 PbI<sub>2</sub> (99.99%) was purchased from Alfa Aesar and used without further purification. MAI  
383 was either synthesised in-house using a common method or else purchased from Dyesol. All  
384 the film preparation work was carried out under nitrogen atmosphere.

385 CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> films were prepared by the “gas-assisted” method of Huang *et al.*<sup>31</sup>. A 25μL 45  
386 wt% CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> DMF solution, prepared from PbI<sub>2</sub> and CH<sub>3</sub>NH<sub>3</sub>I in a molar ratio of 1:1,

387 was spread on the substrate, on a spin-coater. An 80 L/min stream of nitrogen gas was blown  
388 over the film during spinning at 6500 rpm, 2 s after the spinning commenced. The films were  
389 annealed at 100°C on a hotplate for 10 min and then cooled to room temperature.

#### 390 ***Small crystal sample (SC)***

391 Soda-lime glass substrates treatment and precursor solution (PbI<sub>2</sub> and MAI) preparation were  
392 similar to that of the small grain polycrystalline film.

393 CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> films were prepared by the 2-step method of D’Innocenzo *et al.*<sup>6</sup>. The clean  
394 glass substrates were heated on a hotplate at 70° C. A PbI<sub>2</sub> solution in DMF (462 mg/mL) was  
395 warmed to 70°C and then spin-coated at 2000 rpm for 60 s. The PbI<sub>2</sub> films were subsequently  
396 annealed at 70°C for 30 min. CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub> films with small crystal size were prepared by  
397 dipping the PbI<sub>2</sub>-coated substrates in a 0.044 M CH<sub>3</sub>NH<sub>3</sub>I solution in isopropanol at 25°C for  
398 2 minutes. Samples were finally rinsed in anhydrous IPA to remove unreacted CH<sub>3</sub>NH<sub>3</sub>I.

#### 399 ***Mesoporous Al<sub>2</sub>O<sub>3</sub> (MP)***

400 Unless otherwise specified, all materials were purchased from either Alfa Aesar or Sigma-  
401 Aldrich and used as received. The synthesis of CH<sub>3</sub>NH<sub>3</sub>I was performed using the same  
402 previously reported method<sup>31</sup>. The Al<sub>2</sub>O<sub>3</sub> (Aldrich, 642991) nanoparticle paste used for  
403 characterization measurements was formed by suspending ~ 45 nm particles in a solution of  
404 ethanol, terpineol and ethyl cellulose.

405 Glass substrates were cleaned by sonication in ethanol for 10 mins. The mesoporous Al<sub>2</sub>O<sub>3</sub>  
406 layer was formed by drop coating approximately 30 μL/cm<sup>2</sup> of the nanoparticle solution on  
407 the glass substrates, which were then annealed at 500 °C for 30 mins. The perovskite  
408 precursor solution was formed by mixing stoichiometric amounts of CH<sub>3</sub>NH<sub>3</sub>I and PbI<sub>2</sub> in a  
409 combination of dimethyl sulfoxide (DMSO) and *N*-methyl-2-pyrrolidone (NMP) (7:3 vol) to

410 form a (46 wt%) concentration solution. Approximately 30  $\mu\text{L}/\text{cm}^2$  of the perovskite  
411 precursor solution was applied to the surface of the substrate prior to spin-coating. The films  
412 were spin-coated using a two-stage process: 1000 rpm for 5 s using an acceleration of 200  
413 rpm/s, then 6000 rpm for 50 s using an acceleration of 6000 rpm/s. A nitrogen gas flow was  
414 introduced after 20 s of the second spin-coating step and sustained for a further 20 s. The  
415 perovskite films were then annealed on a hot-plate at 100 °C for a duration of 10 mins.

#### 416 **Mixed-cation Mixed-halide Perovskite Sample preparation**

417 The organic cations were purchased from Dyesol; the lead compounds from TCI;  
418 Dimethylformamide (DMF), Dimethyl Sulfoxide (DMSO) and cesium iodide from Sigma  
419 Aldrich. The mixed cation lead mixed halide perovskite  
420 ( $\text{Cs}_{0.05}(\text{MA}_{0.17}\text{FA}_{0.83})_{0.95}\text{Pb}(\text{I}_{0.83}\text{Br}_{0.17})_3$ ) precursor solutions were prepared by dissolving 1 M  
421 FAI, 1.1 M  $\text{PbI}_2$ , 0.2 M MABr, 0.2 M  $\text{PbBr}_2$  in a 4:1 (v:v) mixture of anhydrous DMF:DMSO  
422 and 1.5 M stock solution of CsI in DMSO was added to the above solution in a 5:95  
423 (CsI:mixture) volume ratio. The perovskite solutions were then spin coated in a two-step  
424 program at 1000 and 6000 rpm for 10 and 30 s respectively. During the second step, 50  $\mu\text{l}$  of  
425 chlorobenzene was dripped on the spinning substrate 10 s prior to the end. The perovskite film  
426 was annealed at 100° C for 30 min.

#### 427 **SEM imaging**

428 The LPC samples were imaged using a Zeiss Merlin high-resolution scanning electron  
429 microscope with a beam energy of 3 kV. The SPC, SC and MP samples were imaged using a  
430 FEI Nova NanoSEM 450 scanning electron microscope unit with a beam energy of 15 kV.

#### 431 **Magneto-optical measurement**

432 The magneto transmission measurements were performed combining long pulse magnetic  
433 field measurements and 150 T short duration pulsed magnets.

434 For the long pulse measurements, the typical duration of the pulse is around 100 ms with the  
435 magnetic field up to 66 T. For these measurements, the sample was placed in a liquid helium  
436 cryostat. White light from a halogen lamp was used as the excitation source. The light emitted  
437 from the lamp was coupled in a 200  $\mu\text{m}$ -diameter multimode fiber, used to illuminate sample.  
438 The transmitted light was collected by a 400  $\mu\text{m}$ -diameter multimode fiber and guided to a  
439 spectrometer equipped with a liquid nitrogen cooled CCD camera. The typical exposure time  
440 was  $<3$  ms, which ensured that the transmission spectra were acquired at an essentially  
441 constant magnetic field value.

442 For very high magnetic field measurements ( $B < 150$  T), pulse having a typical duration  $< 10$   
443  $\mu\text{s}$  were generated by a single turn coil system with a bore diameter of 10 mm. A non-  
444 conducting helium-flow cryostat was located in the single turn coil. The sample was kept at a  
445 temperature of  $\sim 5$  K. Magneto-transmission measurements were conducted by using tunable  
446 Optical Parametric Oscillator pumped by a Ti:sapphire laser as the light source. A fast (100  
447 MHz) silicon detector and a high speed digital oscilloscope was used for detection.

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