IN VIVO CORROSION OF COBALT-CHROMIUM AND TITANIUM WEAR PARTICLES

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Tissue reaction to wear particles from metal implants may play a major role in the aseptic loosening of implants. We used electron microprobe elemental analysis to determine the chemical composition of wear particles embedded in the soft tissues around hip and knee implants from 11 patients at revision surgery for aseptic loosening. The implants were made of cobalt-chromium-molybdenum alloy or titanium-aluminium-vanadium alloy.

Histological examination showed a widespread giant-cell reaction to the particles. Elemental analysis showed that the chemical composition of the particles was different from that of the implanted alloys: cobalt and titanium were reduced, often down to zero, whereas chromium and aluminium persisted.

Our findings indicate that corrosion is continually changing the shape, size and chemical composition of the implanted alloy. This may alter the biochemical environment of the tissue surrounding an implant to favour bone resorption.

J Bone Joint Surg [Br] 1995;77-B:962-6. Received 9 March 1995; Accepted after revision 30 May 1995 Aseptic loosening of orthopaedic implants is frequently attributed to tissue reaction to wear debris (Willert and Semlitsch 1977; Goldring et al 1983; Agins et al 1988; Lombardi et al 1989; Galante et al 1991; Haynes et al 1993). Although the tissue response to metal wear particles has been well described the chemical composition of the particles has seldom been studied (Vernon-Roberts and Freeman 1977; Maloney et al 1983; Howie and Vernon-Roberts 1988; Howie 1990; Lee et al 1992; Haynes et al 1993; Jiranek et al 1993). It is often assumed that wear particles maintain the volume and the composition of the implanted alloy. Chemical analysis of tissues containing metal debris has usually used techniques such as atomic absorption spectroscopy or mass spectrometry which reveal only the bulk composition of particles (Woodman et al 1984; Agins et al 1988; Sunderman et al 1989; Case et al 1994). Techniques capable of determining individual particle chemistry, such as electron microprobe elemental analysis, have seldom been employed (Agins et al 1988; Case et al 1994; Shanbhag et al 1994).

There is indirect evidence, however, that metal particles undergo chemical changes in vivo. Swanson, Freeman and Heath (1973), after laboratory wear studies, used atomic absorption spectroscopy to detect ionic cobalt in the lubrication medium and suggested that wear particles may undergo corrosion in vivo. Systemic elevation of trace metal in patients with failed implants also indicates that particles undergo corrosion in vivo (Sunderman et al 1989).

Corrosion alters the shape, size and chemical composition which influence the type and extent of tissue response. Individual metals liberated from the alloy undergo different forms of oxidation or bind with different proteins (Connors, Skilleter and Brown 1990).

We report the chemical composition of wear particles embedded in soft tissue around failed hip and knee implants. Histological examination and polarised light and electron microscopy were used to locate metal wear particles and energy-dispersive electron microprobe analysis (EDX) was used for their elemental analysis.

MATERIALS AND METHODS

Soft tissues were obtained from failed hip and knee implants from 11 patients undergoing revision surgery for aseptic loosening. Two specimens were from hip implants

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©1995 British Editorial Society of Bone and Joint Surgery 0301-620X/95/61095 \$2.00

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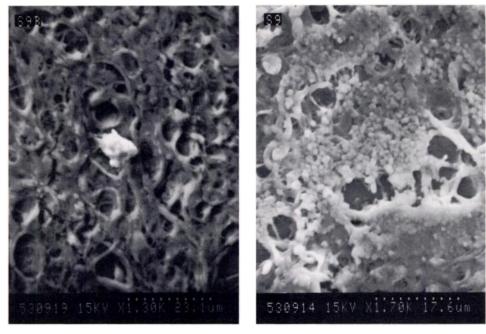


Fig. 1a Fig. 1b

Scanning electron micrographs showing metal deposits in soft tissue. Figure 1a - A newly worn particle ($\times 650$). Figure 1b - An older particle (granulation structure; $\times 850$).

with cobalt-chromium-molybdenum alloy femoral components and cups: these had been in place for 25 and 19 years. One specimen was from a hip implant with a cobalt-chromium-molybdenum femoral component and a polyethylene cup which had been in place for seven years. Two specimens were from knee implants with cobalt-chromium-molybdenum alloy femoral components and polyethylene tibial inserts backed with the same alloy: these had been in place for two and three years. Six specimens were from hips with titanium-aluminium-vanadium femoral components and polyethylene cups: they had been in place for three to nine years. All the implants had been cemented in position.

Specimens for histology and polarised light microscopy were processed by routine methods and paraffin sections were stained with haematoxylin and eosin. For scanning electron microscopy 10 µm thick paraffin sections were mounted on perspex blocks, dewaxed, coated with carbon and examined at an acceleration voltage of 15 kV. The secondary electron mode was used to examine the morphology of the particles. The backscatter electron mode was used for localisation and the X-ray analyser for EDX of particles. The specimens were of uniform thickness and it was ensured that they were flat. Specimen tilt (45°), working distance (15 mm) and X-ray detector position were constant. In all cases, X-ray spectra were obtained for 200 seconds. For transmission electron microscopy the specimens were fixed in 2% neutral (pH 7) glutaraldehyde and embedded in resin. Ultrathin sections were examined by transmission electron microscopy with scanning transmission (STEM) and X-ray analysis facilities. Heavy metal stains routinely used in transmission electron microscopy were omitted.

RESULTS

The specimens were grey to black. Those contaminated with titanium-aluminium-vanadium alloy particles were more heavily stained than those contaminated with cobalt-chromium alloy particles. Metal and polyethylene wear particles were identified microscopically using polarised light and both were generally engulfed by foreign-body giant cells. Scanning (Fig. 1) and transmission (Fig. 2) electron microscopy showed that wear particles of cobalt-chromium alloy were spherical and ranged from submicron to 4 μm in diameter whereas those of titanium alloy were spherical or rod-like, and ranged from submicron to 9 μm in length.

A typical alloy contains about 57% cobalt, 37% chromium and less than 5% molybdenum with traces of impurities, but EDX showed that the wear particles contained varying quantities of cobalt and chromium. The proportion of cobalt ranged from that close to the implant alloy down to zero (Figs 3 to 5). This pattern was found to be independent of implantation time and wear particles with greatly reduced cobalt were regarded as having been in the tissue for longer than those with proportions similar to the implant alloy. Cobalt was not detected in isolation. In contrast, chromium was regularly seen in isolation in particles that were both free and within cells (Figs 2 and 6).

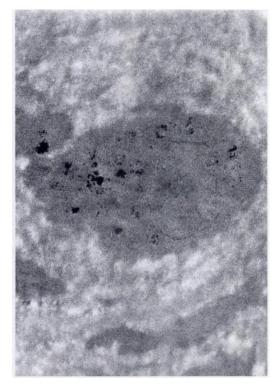
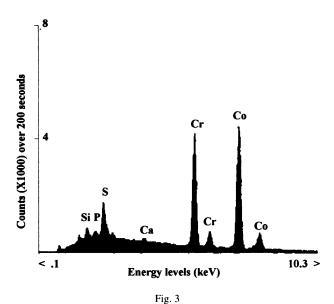


Fig. 2

Transmission electron micrograph showing intercellular metal deposits (\times 2700, section not stained).



X-ray spectrum of a newly worn particle, as in Figure 1a, showing the proportions of cobalt and chromium close to those of the implant alloy.

We were not able to detect molybdenum because its peaks coincide with those of sulphur, which as expected was always present in these biological specimens. With the 15 kV excitation energy used, the L lines of molybdenum (Mo La = 2.29, Lb = 0.54) coincide with the K lines of sulphur (S Ka = 2.31; Kb = 0.54).

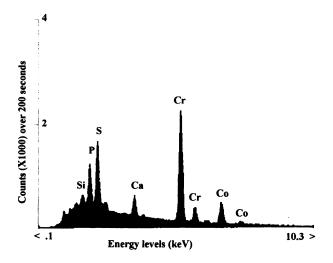


Fig. 4

X-ray spectrum showing reduced cobalt in an older wear particle.

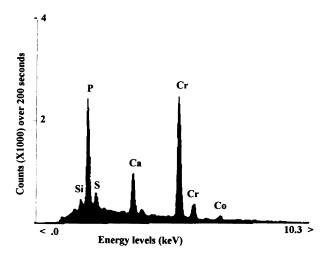
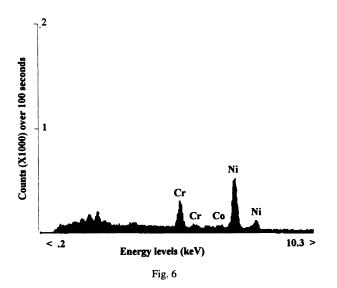


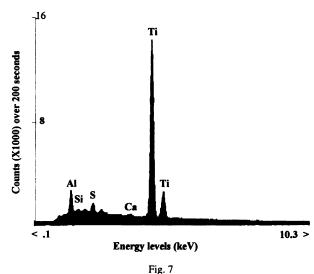
Fig. 5

X-ray spectrum showing markedly reduced cobalt in a considerably older wear particle, as in Figure 1b.

A similar pattern was found in specimens containing titanium-aluminium-vanadium alloy wear particles. The amount of titanium varied from close to that of the implant alloy (about 90% of total alloy) to just detectable. In contrast, the aluminium peak was often close to that of the implant alloy or greater (Figs 7 to 9). In general, titanium and aluminium were not detected alone. The titanium alloys also contain vanadium (4% or less) but vanadium is difficult to detect because its energy peaks overlap those of titanium (Ti Ka = 4.51; Kb = 0.27; V Ka = 4.95; Kb = 0.25).

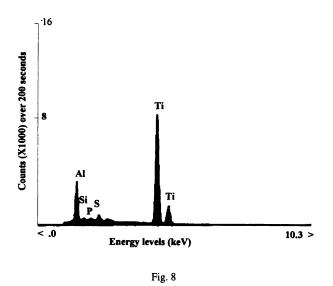
All the elements present could be accounted for. They were either all constituents of the implant alloy, bone

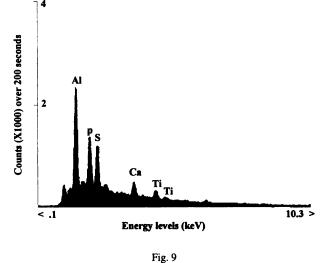




X-ray spectrum showing that the metal deposits as shown in Figure 2, are compounds of chromium alone, the nickel peak being from the specimen grid.

X-ray spectrum showing the proportions of aluminium and titanium in a newly worn particle.





X-ray spectrum showing less titanium in an older wear particle than that seen in Figure 7.

X-ray spectrum showing markedly reduced titanium in a considerably older wear particle.

cement (barium and zirconium), or tissue (calcium and phosphorus) or were likely to have been introduced from stainless-steel surgical instruments (iron, nickel and chromium).

DISCUSSION

The corrosion of wear particles in vivo has important clinical implications. It changes the shape and size of the particles which affects the tissue reaction thus favouring bone resorption (Maloney et al 1983; Howie 1990; Lee et al 1992; Shanbhag et al 1994).

Cobalt and titanium are more soluble and are therefore

removed faster than aluminium and chromium, which persist. Case et al (1994) found nearly five times more cobalt but only twice as much chromium in the draining lymph nodes than in the synovium of patients who had failed cobalt-chromium implants. Agins et al (1988) concluded that the wear particles have the same elemental composition as the implant alloy but their results clearly showed that the chemical composition of titanium-aluminium-vanadium wear particles changed in vivo. Different elements and different compounds of the same element evoke different tissue reactions (Rae 1981; Maloney et al 1983). Haynes et al (1993) attributed toxicity of cobalt to its high solubility and showed that cobalt-

chromium alloy particles are more toxic than those from titanium alloys.

We thank Dr Alice Warley, Physiology Department, UMDS, St Thomas' Hospital for her help with the STEM analysis and Mr B. L. Hinves, FRCS, Conquest Hospital, Hastings, for the supply of specimens and his encouragement.

No benefits in any form have been received or will be received from a commercial party related directly or indirectly to the subject of this article.

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