

Inhibition of Kupffer cell functions modulates arsenic intoxication in Wistar rats

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Abstract. Study determined the influence of the inhibition of Kupffer cell functions by GdCl₃ in arsenic intoxication. Twenty-four Wistar rats weighing between 150 and 160 g were randomly assigned into four groups. Group 1 received sodium arsenite (1.5 mg/kg b.w.) once a day, Group 2 received GdCl₃ (2 mg/kg b.w.) once, 24 hours before commencing the arsenite (1.5 mg/kg b.w.) treatment. Group 3 received GdCl₃ (2 mg/kg b.w.) once and subsequently given distilled water. Group 4 received distilled water only. The treatments were daily by oral gavage and lasted for 28 days. Animals were euthanized 24 hours after the last treatment. Arsenic exposure elevated the activities of rat plasma AST, ALT, ALP and γ -GT, indicative of liver injury. Arsenic exposure in rat lowered GSH concentration but potentiated inflammation and oxidative stress evidenced in the raised levels of MPO, NO and MDA. Rats with arsenic exposure were predisposed to atherosclerosis, lowering the HDL-C but elevated the LDL-C concentration. The histopathological assessment showed degenerating cellular lesion caused by arsenic. However, the inhibition of Kupffer cell functions by GdCl₃ suppressed arsenic intoxication improving the liver function indices, oxidative stress status, lipid profile, neutrophilic inflammation and ultimately restored the cellular architecture. Data suggest that specific inhibition of Kupffer cells by GdCl₃ protected against arsenic intoxication.

Key words: Arsenic — Atherosclerosis — Hepatotoxicity — Kupffer cells — Metals

Abbreviations: AI, atherogenic index; ALP, alkaline phosphatase; ALT, alanine transaminase; AST, aspartate transaminase; CHOL, total cholesterol; GdCl₃, gadolinium chloride; GSH, reduced glutathione; γ -GT, γ glutamyl transferase; HDL-C, high density lipoprotein cholesterol; LDL-C, low density lipoprotein cholesterol; MDA, malondialdehyde; MPO, myeloperoxidase; NO, nitric oxide.

Introduction

Arsenic, as trivalent (As³⁺) and pentavalent salt (As⁵⁺), is released into the environment *via* agricultural, industrial and medical applications (Aliyu et al. 2012). Arsenic is a toxic metalloid and a common contaminant of drinking water. Arsenical exposure through drinking water is common and occurs in many areas of the world (Morakinyo et al. 2010).

Consumption of arsenic-contaminated water increases risk of cardiovascular disease, lung disease, hepatic disease, and cancer in millions of people worldwide (Okoji et al. 2002; Straub et al. 2008; Das et al. 2012; Kumar et al. 2014). Arsenic-contaminated drinking water is associated with several adverse health effects in many developing areas around the world, including China, India, Mexico, and Bangladesh (Aliyu et al., 2012; Das et al. 2012).

Arsenite intoxication limits ATP production (Hughes et al. 2011), while potentiating oxidative stress which readily leads to cellular dysfunction (Rana et al. 2011). While the mechanism of arsenic toxicity remains limited, current treatment and preventive measures for arsenic intoxica-

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tion are still elusive (Chowdhury et al. 2006). Therefore, identifying the cellular factors that contribute to modulate the liver response to arsenicals is central to understanding their toxicology. We reasoned that, identifying cellular factors involved in liver response to arsenic intoxication would further aid our understanding of its mode of toxic action as well as provide opportunity for the development of newer treatment strategy.

Kupffer cells are macrophage native to the liver and mediate hepatic response to chemical assault (Roberts et al. 2007). Generally, Kupffer cells play an important role in the normal physiology and homeostasis of the liver as well as participate in the acute and chronic responses of the liver to toxic compounds and have been shown to modulate hepatic injury (Yee et al. 2000; Roberts et al. 2007). Several studies have demonstrated the central role played by Kupffer cells in liver responses to the toxic action of several drugs (Yee et al. 2000; Ju et al. 2002; Ito et al. 2003; Prins et al. 2004; Yao et al. 2004; Kresse et al. 2005). Meanwhile, it has been demonstrated that pretreatment of mice with $GdCl_3$, a specific inhibitor of Kupffer cells, offered protection against liver injury (Jorgensen et al. 2001). Further, in a separate investigation, Ding et al. (2003) showed that the intraperitoneal administration of $GdCl_3$ to mice at doses between 10–20 mg/kg caused apoptosis of Kupffer cells and blocked the Kupffer cell effector function without necessarily damaging the liver. Currently, there is no empirical data on how Kupffer cells affect arsenic toxicity. Therefore, determining the role Kupffer cells play in arsenical intoxication may be essential to understanding the mode of the toxic action. To this end, this study evaluated the influence of the inhibition of Kupffer cell functions by $GdCl_3$ in experimental arsenic intoxication.

Materials and Methods

Chemicals reagents

All chemicals and reagents used were of analytical grade. Sodium arsenite ($NaAsO_2$) and $GdCl_3$ were products of Sigma Chemicals (St. Louis, MO, USA). Reagent assay kits for aspartate aminotransferase (AST), alanine aminotransferase (ALT), γ -glutamyl transferase (γ -GT), total cholesterol (CHOL), high-density lipoprotein (HDL), bilirubin, and albumin were products of Randox Laboratories (Crumlin, UK).

Experimental animals

Twenty-four male Wistar rats weighing between 150 and 160 g were purchased from the experimental animal farm at the Landmark University, Nigeria. The rats were housed

in plastic cages and kept in a well-ventilated room, with alternate light and dark cycle (12 h each). The animals were acclimatized for 14 days before the treatment commenced. The animals were fed standard rat chow and clean water *ad libitum*. Handling of animals was in accordance with relevant institutional and ethical guidelines as approved for scientific study.

Experimental design

The Wistar rats were randomly assigned into four groups. Details of groupings and treatments are as follows:

Group 1: Administered with arsenite ($NaAsO_2$) at 1.5 mg/kg body weight.

Group 2: Administered once with $GdCl_3$ at 2 mg/kg b.w., 24 h before starting treatment with arsenite ($NaAsO_2$) at 1.5 mg/kg b.w.

Group 3: Administered once with $GdCl_3$ at 2 mg/kg b.w. and subsequently received distilled water.

Group 4: Administered with distilled water and served as sham control.

The arsenite dose was premised on report by Zaki et al (2011). The arsenite dose 1.5 mg/kg body weight/day falls within the range of LD_{50} of a 70 kg b.w. human (1–4 mg/kg) and lesser than 1/25 of LD_{50} of rats (40 mg/kg). The treatments were daily by oral gavage and lasted for 28 days. Handling of animals was as approved by the Landmark University Research Ethics Committee and consistent with relevant guidelines on the care and use of laboratory animals (National Research Council 2011).

Necroscopy

Rats were sacrificed under mild chloroform anesthesia 24 h after the last treatments. Blood samples were collected into EDTA bottles, centrifuged at $5000 \times g$ for 5 minutes in a refrigerated centrifuge (Anke TDL-5000B, Shanghai, China) to obtain the plasma used for the biochemical determinations. The liver from each animal was excised into iced isotonic solution (pH 7.4) and weighed immediately. A fifth of the liver sample was homogenized in iced 0.25 M sucrose solution (1:5, w/v). The liver homogenates were used for the biochemical determinations. The remaining part of the liver was fixed in 10% buffered neutral formalin (BNF) and used for histopathology examinations.

Biochemical assays

The biochemical parameters were determined in rat plasma using a UV/Vis spectrophotometer (Jenway, Staffordshire, United Kingdom) where applicable. The concentration of rat plasma total protein (TP), aspartate aminotransferase (AST, EC: 2.6.1.1), alanine aminotransferase (ALT, EC

2.6.1.2), alkaline phosphatase (ALP, EC: 3.1.3.1), γ -glutamyl transferase (γ -GT, EC 2.3.2.2), bilirubin, albumin, lipid profile including total cholesterol (CHOL), high-density lipoprotein-cholesterol (HDL-C) were determined using Randox assay kits (Crumlin, UK). The low-density lipoprotein cholesterol (LDL-C) was estimated according to the Friedewald formula (Warnick et al. 1990). Total protein was determined using the Lowry's method (Lowry et al. 1951). Reduced glutathione level (GSH) was determined by the procedure described by Ellman (1959) and Akanji et al (2009). Malondialdehyde (MDA) was determined using the method described by Niehaus and Samuelson (1968). The concentration of nitric oxide as nitrite/nitrate level was determined using the Greiss method according to method described by Ekanem et al. (2009). The myeloperoxidase (MPO – EC 1.11.2.2) activity was determined as previously described (Adeyemi and Sulaiman 2014), according to the method by Eiserich et al. (1998).

Histopathological examination

The rat liver was fixed in 10% buffered neutral formalin immediately following excision from animals. Fixed tissues were subsequently processed for histopathological examinations as previously described (Adeyemi and Sulaiman 2012). Capture and scoring for morphological changes were done by a pathologist blind to the treatments, at the University of Ilorin Teaching Hospital, Ilorin, Nigeria. The photomicrographs were captured at $\times 100$ using the software "Presto! Image Folio package".

Data analysis

Data were analyzed using the one-way ANOVA (GraphPad Software Inc., San Diego, CA) and presented as the mean \pm standard error of mean (SEM). Differences among the group means were determined by the Tukey's test. Mean values at $p < 0.05$ were considered to be significant.

Results

Arsenite exposure had no appreciable effect on rat average weights when compared to the other treatment groups (Fig. 1). However, the rat liver weights as well as the liver-to-body weight ratio were significantly reduced by arsenite.

Furthermore, oral exposure of rats to arsenite did not alter the concentration of plasma total protein. In contrast, the rat plasma albumin concentration was appreciably decreased relative to the other treatment groups (Fig. 2). Arsenic treatment in rats led to a significant elevation of plasma bilirubin concentration relative to the other treatments.

In comparison to the other treatment groups, the oral exposure of rats to arsenic for 28 days suggest impairment to liver function as evidenced by increased ($p < 0.05$) activities of plasma AST, ALT, ALP and γ -GT (Fig. 3). Also, arsenic exposure elevated rat plasma and hepatic MDA concentration but depleted the GSH levels when compared to the other treatments (Fig. 4). Likewise, arsenic exposure increased the levels of MPO and NO relative to the other treatment groups (Fig. 5). The rat plasma lipid profile was not spared, as arsenic exposure elevated the concentration of CHOL, LDL-C but reduced the HDL-C level, thereby increasing the atherogenic index (AI). The negative alterations of plasma lipid profile by arsenic were absent in the other treatment groups (Fig. 6).

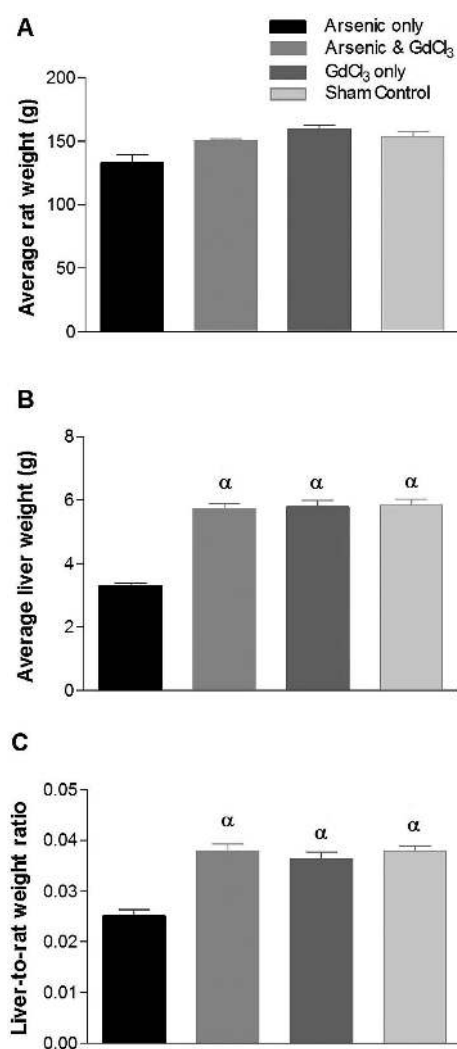


Figure 1. Effect of oral exposure of rats to arsenite and/or gadolinium chloride (GdCl₃) on average rat weight (A), rat liver weight (B) and liver-to-body weight ratio (C). Data are presented as mean value \pm standard error of mean (SEM), $n = 6$. α is significant at $p < 0.001$ relative to arsenic only.

The histopathological assessment of rat liver sections showed inimical cellular lesions including congestion and inflammation caused by arsenic exposure (Fig. 7). However, pre-treating rats with $GdCl_3$ before daily exposure to arsenic improved the cellular alterations. In the other groups that received $GdCl_3$ alone and/or distilled water only, the liver cellular architecture was intact.

Discussion

Arsenic is a toxic metalloid (Aliyu et al. 2012; Das et al. 2012). However, the knowledge of the mechanism of arsenic toxicity

remains limited. We reasoned that, identifying the cellular factors that contribute to modulate the response to arsenical toxicity would provide for a better understanding of the mode of toxicity. Kupffer cells have been shown to modulate hepatic injury as well as play key role in the mediation of

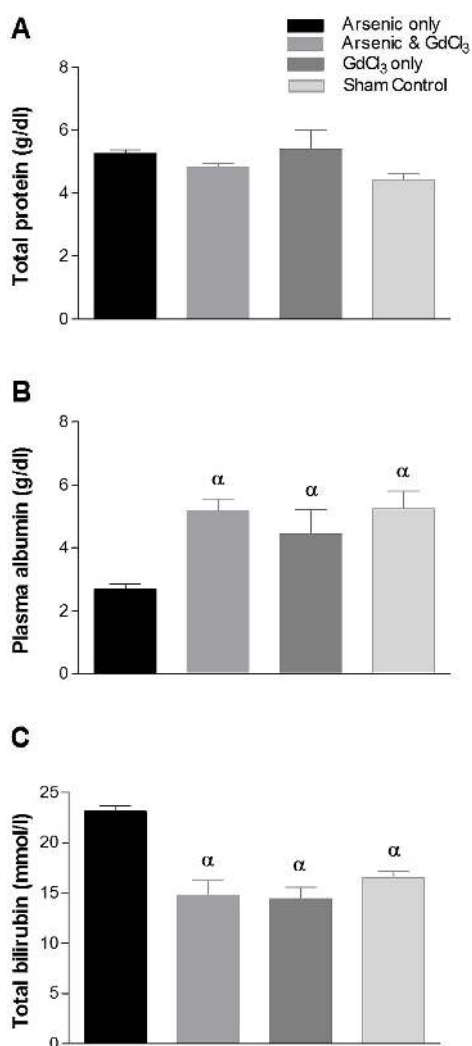


Figure 2. Effect of oral exposure of rats to arsenite and/or gadolinium chloride ($GdCl_3$) on rat plasma protein concentration (A), rat plasma albumin (B) and rat plasma bilirubin (C). Data are presented as mean value \pm SEM, $n = 6$. α is significant at $p < 0.001$ relative to arsenic only.

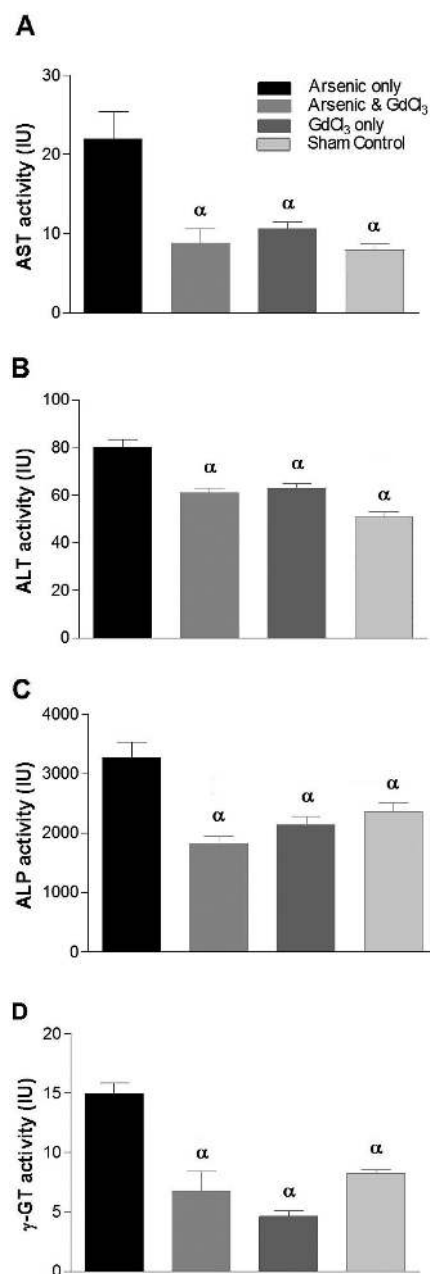


Figure 3. Effect of oral exposure of rats to arsenite and/or gadolinium chloride ($GdCl_3$) on rat plasma aspartate transaminase (AST) activity (A), rat plasma alanine transaminase (ALT) activity (B), rat plasma alkaline phosphatase (ALP) activity (C) and rat plasma γ -glutamyl transferase (γ -GT) activity (D). Data are presented as mean value \pm SEM, $n = 6$. α is significant at $p < 0.001$ relative to arsenic only.

metal toxicity (Roberts et al. 2007). Therefore, identifying the role Kupffer cells play in arsenic intoxication would aid understanding the mode of toxicity. The present study by

using aimed to determine how Kupffer cells affect arsenic intoxication by using $GdCl_3$ to inhibit Kupffer cell functions.

Arsenic exposure in rats significantly reduced the rat liver weight as well as the liver-to-body weight ratio. This may indicate ensuing adverse effect by arsenite as studies have associated changes to body or organ weights with chemical or drug intoxication (Adeyemi and Orekoya 2014; Sulaiman et al. 2015). This is consistent with other studies have demonstrated arsenic-induced changes to rat body or organ weights (Aliyu et al. 2012; Chandraayagam et al. 2013).

Further evidence for arsenite-induced liver injury is reflected by the negative alteration of the activities of rat plasma AST, ALT, ALP and γ -GT. Increased liver function indices are evidence of liver damage (Adeyemi and Akanji 2011a; Adeyemi et al. 2012). The arsenic-induced elevation of liver function indices is consistent with the potential to cause liver damage (Gbadegesin and Odunola 2010; Das et al. 2012; Kumar et al. 2014). More so, the arsenite-induced negative alteration of the plasma albumin and bilirubin concentration suggests liver injury or reduced functioning capacity.

The arsenic exposure also caused oxidative stress by potentiating lipid peroxidation while depleting the GSH concentrations. GSH plays key role protecting against metal toxicity (Swaran et al. 2005; Mishra et al. 2008) and its deple-

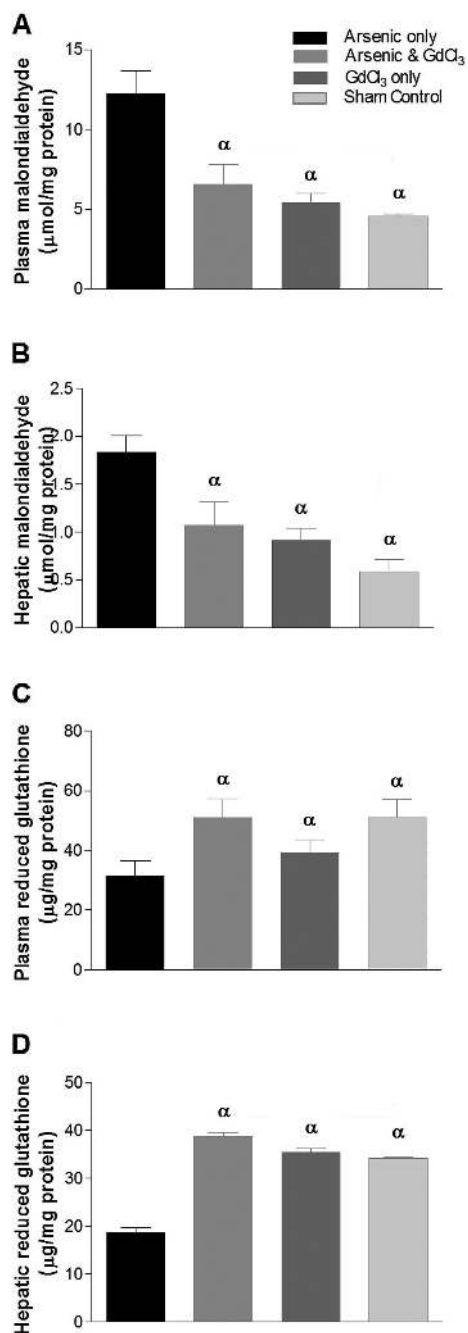


Figure 4. Effect of oral exposure of rats to arsenite and/or gadolinium chloride ($GdCl_3$) on rat plasma malondialdehyde (MDA) concentration (A), rat hepatic malondialdehyde (MDA) concentration (B), rat plasma reduced glutathione (GSH) concentration (C) and rat hepatic reduced glutathione (GSH) concentration (D). Data are presented as mean value \pm SEM, $n = 6$. α is significant at $p < 0.001$ relative to arsenic only.

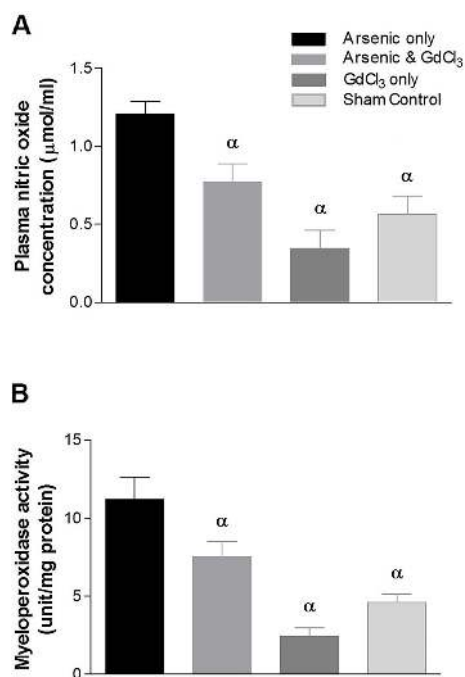


Figure 5. Effect of oral exposure of rats to arsenite and/or gadolinium chloride ($GdCl_3$) on rat plasma nitric oxide (NO) concentration measured as nitrate/nitrite level (A) and rat plasma myeloperoxidase (MPO) activity (B). Data are presented as mean value \pm SEM, $n = 6$. α is significant at $p < 0.001$ relative to arsenic only.

tion may signal oxidative stress (Akanji et al. 2009). In the present study, the use of GSH to detoxify arsenite may have led to the depletion and ensuing oxidative stress observed.

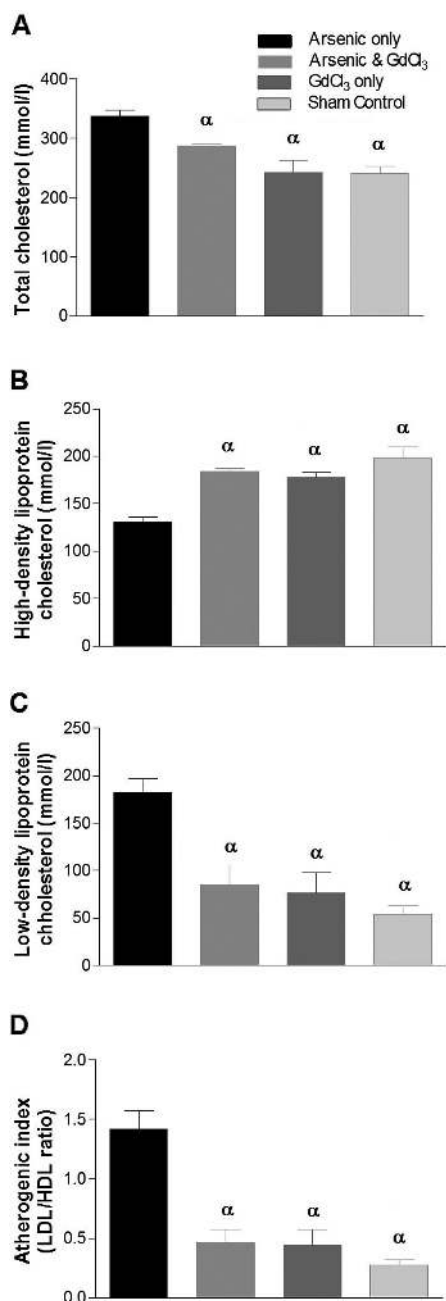


Figure 6. Effect of oral exposure of rats to arsenite and/or gadolinium chloride (GdCl₃) on rat plasma total cholesterol concentration (A), rat plasma high density lipoprotein-cholesterol (HDL-C) concentration (B), rat plasma low density lipoprotein-cholesterol (LDL-C) concentration (C), and rat atherogenic index (AI – LDL/HDL ratio) (D). Data are presented as mean value \pm SEM, $n = 6$. α is significant at $p < 0.001$ relative to arsenic only.

Several studies have implicated arsenic for lipid peroxidation, oxidative stress as well as liver damage (Zaki et al. 2011; Chandranayagam et al. 2013). Arsenic's reactivity toward sulfhydryl groups may affect several cellular functions. Studies have demonstrated that arsenite has affinity for thiol groups and could cause damage by promoting generation of reactive species (Shiobara et al. 2001; Rana et al. 2011).

The increased level of MPO and NO, consequent of arsenite intoxication may also have aided the lipid peroxidation as well as resultant oxidative cellular damage. Elevated production of NO can react with metals, reduced thiols, molecular oxygen and superoxide leading to adverse reactions (Bryan et al. 2005; Gladwin et al. 2005) leading to cellular damage (Adeyemi and Sulaiman 2014). Likewise, increased activity of MPO has been linked to oxidative cellular damage (Adeyemi and Akanji 2011b). MPO plays vital role in neutrophilic inflammation (Vita et al. 2004), and has been shown to potentiate atherosclerosis (Hazen et al. 2004). In like manner, the present study showed that arsenite exposure has capacity to predispose rats to cardiovascular disorder by increasing plasma total CHOL, LDL-C, while lowering the HDL-C levels which raised the atherogenic index. Elevated atherogenic index has been associated with the development of atherosclerosis (Olukanni et al. 2013; Adeyemi and Orekoya 2014). Our findings agree with studies that have linked arsenic toxicity to cardiovascular disorders (Straub et al. 2008; Das et al. 2012; Kumar et al. 2014).

Furthermore, the histopathological assessment of rat liver sections showed cellular lesion caused by arsenite exposure. This supports the biochemical evaluations. Arsenic intoxication has been shown to cause hepatic fibrosis and other degenerating cellular lesions (Datta et al. 2007; Rana et al. 2011; Zaki et al. 2011; Abnosi and Jafaru 2012) similar to the observation in the present study.

However, the pre-treatment of rats with GdCl₃ to inhibit the Kupffer cell functions before arsenite exposure suppressed and protected against arsenite intoxication by improving the liver function indices, the oxidative status, lipid profile and ultimately the cellular architecture. The Kupffer cells play central role in the hepatic response to toxicants or carcinogens. The activation of Kupffer cells either directly or indirectly promotes the release of inflammatory mediators as well as reactive oxygen species (ROS) which help modulate hepatic injury. In this study, the inhibition of Kupffer cell functions by GdCl₃ may have reduced the production of inflammatory mediators including NO and MPO thereby protecting against cellular inflammation and damage. NO and MPO under certain conditions are pro-inflammatory and the duo have been implicated for cellular damage (Adeyemi and Sulaiman 2014). This suggests that Kupffer cells are a target for arsenic toxicity. The arsenic exposure may have activated the Kupffer cells for protective response that develops to cause cellular damage with further stimulation.

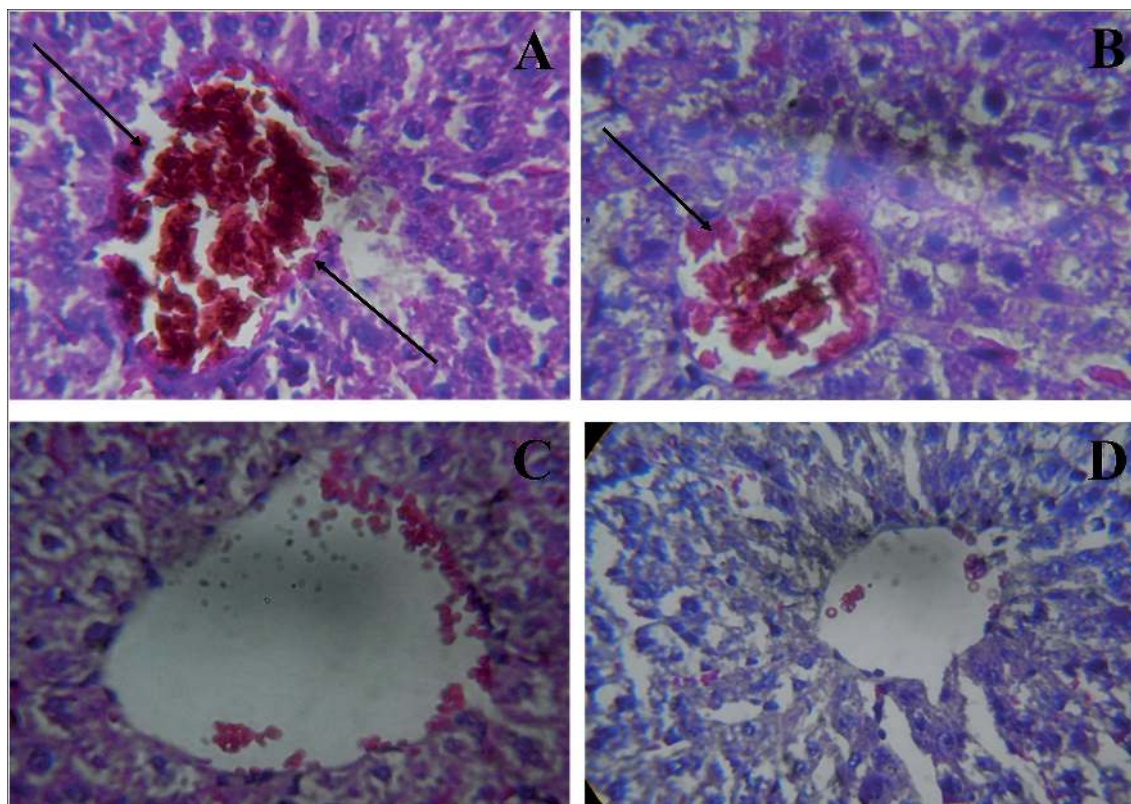


Figure 7. Photomicrographs of rat liver following exposure to arsenic and/or gadolinium chloride ($GdCl_3$). **A.** Arsenic treatment only – arrows showing distortion of the cellular architecture with a severe congestion of the central vein. **B.** Arsenic and gadolinium chloride treatment – arrows showing a slight distortion of the cellular architecture with a mild congestion of the central vein. **C.** Gadolinium chloride treatment only – showing a well preserved cellular morphology with a free central vein. **D.** Normal saline – showing a well preserved cellular architecture with a free central vein. H&E Stain ($\times 400$).

This is plausible considering that the inhibition of Kupffer cell functions by $GdCl_3$ significantly reduced and improved the indices of arsenite intoxication. Findings in the present study revealed consistence with Roberts et al (2007) which reported that $GdCl_3$ inhibition of Kupffer cells could mitigate chemical intoxication as well as resultant cellular damage.

Conclusion

Although Kupffer cells have dual role in the hepatic response to chemical toxicity, our data supports the complicity of Kupffer cells in arsenite intoxication. To our knowledge, this is the first study showing evidence that inhibiting Kupffer cell functions by using $GdCl_3$ offered protection against arsenite intoxication in rats. Data would contribute to our understanding of the mechanism of arsenite intoxication.

Acknowledgements. Authors acknowledge the laboratory staff at the Department of Biological Sciences for technical support.

Conflict of interest. Authors declare that there is no conflict of interest.

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Received: January 9, 2016

Final version accepted: September 6, 2016

First published online: February 2, 2017