Involvement of Methyltransferase-Activating Protein and Methyltransferase 2 Isoenzyme II in Methylamine:Coenzyme M Methyltransferase Reactions in *Methanosarcina barkeri* Fusaro

ROEL W. WASSENAAR,* PIET J. H. DAAS, WIM J. GEERTS, JAN T. KELTJENS, AND CHRIS VAN DER DRIFT

Department of Microbiology, Faculty of Science, University of Nijmegen, Nijmegen, The Netherlands

Received 22 December 1995/Accepted 30 September 1996

The enzyme systems involved in the methyl group transfer from methanol and from tri- and dimethylamine to 2-mercaptoethanesulfonic acid (coenzyme M) were resolved from cell extracts of *Methanosarcina barkeri* Fusaro grown on methanol and trimethylamine, respectively. Resolution was accomplished by ammonium sulfate fractionation, anion-exchange chromatography, and fast protein liquid chromatography. The methyl group transfer reactions from tri- and dimethylamine, as well as the monomethylamine:coenzyme M methyl-transferase reaction, were strictly dependent on catalytic amounts of ATP and on a protein present in the 65% ammonium sulfate supernatant. The latter could be replaced by methyltransferase-activating protein isolated from methanol-grown cells of the organism. In addition, the tri- and dimethylamine:coenzyme M methyltransferase (MT₂), which is different from the analogous enzyme from methanol-grown *M. barkeri*. In this work, it is shown that the various methylamine:coenzyme M methyltransfer steps proceed in a fashion which is mechanistically similar to the methanol:coenzyme M methyl transfer, yet with the participation of specific corrinoid enzymes and a specific MT₂ isoenzyme.

Methanosarcina barkeri is a methanogenic archaeon which is able to grow on acetate and various one-carbon compounds such as CO₂, methanol, and mono-, di-, and trimethylamine. The reduction of methanol and methylamines to methane occurs via methyl-coenzyme M (CH₃-S-CoM), the substrate of the final step in methanogenesis in all methanogens studied so far (13, 14). Synthesis of CH₃-S-CoM from methanol and coenzyme M (2-mercaptoethanesulfonic acid [HS-CoM]) is catalyzed by the concerted action of two methyltransferases. Firstly, methanol:5-hydroxybenzimidazolylcobamide (B₁₂-HBI) methyltransferase (MT_1) binds the methyl group of the substrate to its corrinoid prosthetic group (20). Secondly, the methyl group is transferred to HS-CoM by Co-methyl-5-hydroxybenzimidazolylcobamide:HS-CoM methyltransferase (MT₂) (18). MT_1 is catalytically active only when the central cobalt atom of its corrinoid prosthetic group is present in the highly reduced Co(I) state (B₁₂-HBI_s) (21). Enzyme molecules that are inactivated because of oxidation of their corrinoid group, however, can be reactivated by use of a reducing system (hydrogen, hydrogenase, and ferredoxin), ATP, and an enzyme named methyltransferaseactivating protein (MAP) (3-5).

Naumann et al. (17) were the first to explore the methyl group transfer reactions from the three methylamines to HS-CoM in cell-free systems of *M. barkeri*. The ATP dependence of the reactions prompted the authors to propose that the conversions might take place in a way analogous to the synthesis of CH₃-S-CoM from methanol, though with the participation of differentially expressed methyltransferase systems. Subsequent investigations suggested that an isoenzyme of MT₂

that had previously been described, MT_2 isoenzyme II $[MT_2(II)]$ (8), plays a role in the reactions (1, 7, 22).

In this paper, we present evidence that MAP functions in the methyl group transfer reactions from tri-, di-, and monomethylamine and that the trimethylamine:HS-CoM and dimethylamine:HS-CoM methyltransferase reactions are carried out by specific corrinoid proteins. In addition, it is shown not only that MT₂(II) is involved in methyl group transfer from monomethylamine as published before (1) but also that it quite specifically mediates the conversions of tri- and dimethylamine.

MATERIALS AND METHODS

Culture methods and preparation of cell extract. Cells of *M. barkeri* Fusaro (DSM 804) were cultured in a 20-liter fermentor in a mineral medium containing either 250 mM methanol or 100 mM trimethylamine as the carbon and energy source (10, 11). Cells were harvested anaerobically at the end of growth and stored under N_2 at -80° C until use.

Cell extracts were prepared by suspending the wet cells (1:1, wt/vol) in 50 mM N-tris-(hydroxymethyl)methyl-2-aminoethanesulfonic acid (TES) buffer, pH 7.0, containing 15 mM MgCl₂, 1 mM dithiothreitol, RNase (10 $\mu g/ml$), and DNase (10 $\mu g/ml$) and subsequently passing the suspension through a French pressure cell at 138 MPa under continuous flushing with N₂. Anaerobic centrifugation at 13,200 \times g (20 min, 4°C) pelleted the cell debris and unbroken cells. The supernatant, referred to as cell extract, was collected in an anaerobic glove box (98% N₂/2% H₂), and stored under N₂ at $-80^{\circ}\mathrm{C}$. The protein concentration of the cell extract from methanol-grown cells was 18 mg/ml; the cell extract of trimethylamine-grown cells contained 46 mg of protein per ml.

Enzyme assays. Reaction mixtures were prepared in the glove box, and the reactions were performed in crimp-sealed 10-ml serum flasks (2). Unless stated otherwise, a typical reaction mixture (final volume, 200 μ l) contained 50 mM TES-K+ buffer (pH 7.0), 25 mM MgCl₂, 10 mM HS-CoM, 2 mM ATP, 1 mM 2-bromoethanesulfonic acid, 10 mM substrate (either methanol, monomethylamine, dimethylamine, or trimethylamine) and 40- μ l amounts of each of the different enzyme fractions involved.

After being subjected to pressure (100% H_2 [10⁵ Pa]), the vials were kept on ice. Fractions obtained from ammonium sulfate, column chromatographical, or fast protein liquid chromatography (FPLC) separations were tested for their stimulation of the HS-CoM conversion in 50-, 100-, and 100- μ l amounts, respectively, together with 10 to 15 μ l of crude extract. Here, cell extract was substituted for the other enzymic components (2, 18); the methyltransferase activity by the amount of extract alone was negligible. Reactions were started by putting the

^{*} Corresponding author. Mailing address: Department of Microbiology, Faculty of Science, University of Nijmegen, Toernooiveld 1, NL-6525 ED Nijmegen, The Netherlands. Phone: 31 24 3653218. Fax: 31 24 3553450. Electronic mail address: roelw@sci.kun.nl.

6938 WASSENAAR ET AL. J. BACTERIOL.

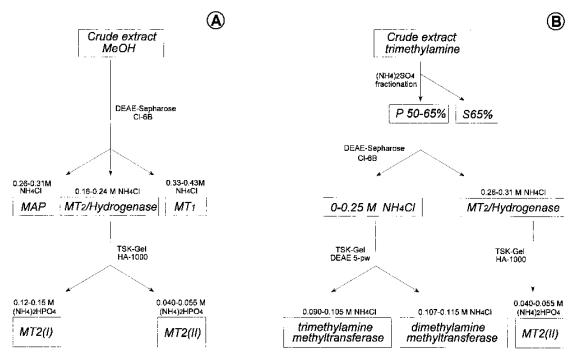


FIG. 1. Flowchart illustrating the resolution of the methanol (A)- and tri- and dimethylamine (B)-converting systems from *M. barkeri* Fusaro. Enzyme fractions used in the experiments are boxed. MeOH, methanol; P 50–65%, 50 to 65% ammonium sulfate pellet; S65%, 65% ammonium sulfate supernatant.

vials in a water bath at 37°C. After the appropriate incubation periods, 15-µl aliquots of the reaction mixture were withdrawn with a gas-tight syringe and added to 1 ml of Ellman's reagent. The level of activity of methyl group transfer from the substrate to HS-CoM was routinely assayed by measuring the rate of decrease in HS-CoM.

Hydrogenase and MT_2 activities were assayed qualitatively with microtiter plates placed inside the anaerobic glove box. Hydrogenase activity in the enzyme fractions was screened by adding 10 μ l of enzyme to 90 μ l of 1 mM benzyl viologen in 100 mM TES-K⁺ buffer (pH 7.0). MT_2 activity was detected by mixing 25 μ l of enzyme fraction with 75 μ l of a solution containing 2.5 mM methylcobalamin–5 mM HS-CoM in 100 mM TES-K⁺ buffer (pH 7.0). Active fractions showed a color change of the reaction mixture from red (methylcobalamin) to brown [cob(II)alamin] within a 30-min incubation at ambient temperature (15). The quantitative determination of MT_2 proceeded as described by Harms and Thauer (9).

Fractionation of cell extracts. Since several enzymes involved in the transfer of the methyl group from the substrate (either methanol or methylamines) to HS-CoM are oxygen labile, all handlings were performed in the anaerobic glove box.

A schematic representation of the resolution of the trimethylamine-, dimethylamine-, and methanol-converting systems is given in the flowchart shown in Fig. 1.

Methanol-grown cells. The cell extract (30 ml) of methanol-grown cells was applied to a column (12 by 3.0 cm) packed with DEAE Sepharose Cl-6B and equilibrated in buffer A (50 mM TES-K⁺ [pH 7.0], 15 mM MgCl₂, 1 mM dithiothreitol, 10% [vol/vol] ethylene glycol). After application of the sample, separation was performed at a flow rate of 2 ml/min by washing with 150 ml of buffer A, and this was followed by a 400-ml linear gradient of 0 to 0.6 M NH₄Cl in buffer A and a 100-ml wash with 0.6 M NH₄Cl in buffer A. The eluate was monitored at 280 nm, and 5-ml fractions were collected. MT2-hydrogenase was present in the fractions that eluted between 0.16 and 0.24 M NH₄Cl (i.e., that were eluted at a NH₄Cl concentration from 0.16 to 0.24 M). MAP and MT₁ were recovered in the fractions that were eluted between 0.26 to 0.31 and 0.33 to 0.43 M NH₄Cl, respectively. The three collected pools were separately desalted and thoroughly washed in buffer A on an Amicon YM3 ultrafiltration membrane (Amicon, Lexington, Mass.), adjusted to a volume of 10 ml, and stored at -20°C under N₂ until use. The MT₁ fraction contained 8 mg of protein per ml; the MT₂-hydrogenase fraction contained 10 mg of protein per ml, and the MAP fraction contained 9 mg of protein per ml. The MT2-hydrogenase fraction was purified further on a hydroxylapatite fast protein liquid chromatography (FPLC) column (7.5 by 0.75 cm) packed with TSK-Gel HA-1000. The column was equilibrated in buffer A lacking MgCl2. Separation was carried out at a flow rate of 0.45 ml/min by washing with 12 ml of buffer, and this was followed by a 60-ml linear gradient of 0 to 200 mM (NH₄)₂HPO₄ in buffer A without MgCl₂. The eluate was collected in 2-ml fractions. Two distinct MT2 peaks that eluted between 0.040 and 0.055 M (NH₄)₂HPO₄ and 0.120 and 0.150 M (NH₄)₂HPO₄ were obtained. The latter contained about 85% of the total MT₂ activity. Comparison of the separation characteristics of this fraction with those described by Grahame (8) shows it to contain the MT₂(I) isoenzyme involved in methanol conversion. The combined MT₂(I) fractions were desalted by ultrafiltration as described above, adjusted to a volume of 10 ml (0.7 mg of protein per ml), and stored under N₂ at $-20^{\circ}\mathrm{C}$.

Trimethylamine-grown cells. The cell extract of trimethylamine-grown M. barkeri (15 ml; 46 mg of protein per ml) was subjected to 50% ammonium sulfate fractionation by slowly adding an equal volume of neutralized saturated ammonium sulfate. After gentle stirring for an additional hour, the suspension was anaerobically centrifuged (18,000 \times g; 20 min; 4 °C). The supernatant was carefully decanted, adjusted to 65% ammonium sulfate saturation, and centrifuged as described above. The 50 to 65% ammonium sulfate pellet was resuspended in buffer A, thoroughly washed in the buffer on an Amicon YM3 ultrafiltration unit, and concentrated to a volume of 15 ml. Similarly, the 65% ammonium sulfate supernatant was washed, desalted, and concentrated to 15 ml (6 mg of protein per ml).

The 50 to 65% ammonium sulfate fraction was further resolved on a column (12 by 2.8 cm) packed with DEAE Sepharose Cl-6B equilibrated in buffer A. After application of the sample (15 ml; 30 mg of protein per ml), separation was carried out as described above. In this step, fractions that were eluted between 0 to 0.32 M NH₄Cl in the gradient step strongly stimulated the conversion of trimethylamine as well as of dimethylamine when assayed with crude cell extract. The fractions that eluted between 0.26 and 0.32 M NH₄Cl containing MT₂ and hydrogenase activity were combined and washed separately.

After washing with buffer A, 3-ml aliquots of the 0 to 0.25 M NH₄Cl fraction (9 mg of protein per ml) obtained from DEAE Sepharose Cl-6B chromatography were further resolved on an FPLC ion-exchange column (7.5 by 0.75 cm) packed with TSK-Gel DEAE-5pw and equilibrated in buffer A. Proteins were separated at a flow rate of 0.75 ml/min by a 15-ml wash with buffer A followed by a linear 105-ml gradient of 60 to 150 mM NH₄Cl in buffer A to collect 1.5-ml fractions. When assayed with crude extract, the fractions that were eluted between 0.090 and 0.105 M NH₄Cl strongly stimulated the trimethylamine:HS-COM methyl group transfer reaction but not the dimethylamine or monomethylamine methyl group transfers. Likewise, the fractions eluting between 0.107 and 0.115 M NH₄Cl specifically enhanced the dimethylamine conversion. The trimethylamine (0.090 to 0.105 M NH₄Cl)- and dimethylamine (0.107 to 0.115 M NH₄Cl)-converting fractions were collected separately, washed, and concentrated to 3-ml volumes, containing 0.8 and 2.4 mg of protein per ml, respectively, as described above.

 MT_2 obtained from DEAE Sepharose Cl-6B chromatography was further purified on the TSK-Gel HA-1000 FPLC hydroxylapatite column by the procedure described above for the separation of MT_2 isoenzymes from methanol-grown cells. In this case, only one peak of MT_2 activity, which was eluted between

0.040 and 0.055 M (NH₄)₂HPO₄ and accounted for MT₂(II), was detected. The peak was pooled, concentrated, washed with buffer A, and adjusted to a final volume of 8 ml (1 mg of protein per ml).

Preincubation of fractionated enzymes with ATP. Inside the glove box, 200 nmol of ATP was added to the enzyme fraction to be tested, and this was followed by a 10-min incubation at ambient temperature. Subsequently, excess ATP was removed by adding 400 nmol of glucose, hexokinase (22 U), and myokinase (15 U) to the preincubation mixture. After another 10-min incubation, the preincubated fraction was assayed with all other components except for ATP in the standard methyltransferase reaction described above.

Analytical methods. The concentration of HS-CoM was determined by mixing 15-μl samples with 1 ml of 0.48 mM 2,2'-dinitro-5,5'-dithiobenzoic acid (Ellman's reagent [6]) in 150 mM Tris-Cl- buffer (pH 8.0) and immediately measuring the A_{412} . HS-CoM concentrations were quantified by comparison with a calibration curve made from a freshly prepared stock of the compound. Trimethylamine, dimethylamine, and monomethylamine concentrations were measured by gas chromatography. Samples (25 µl) of the reaction mixture were diluted 1:1 in 6 M KOH, and isopropylamine was added to a concentration of 6 mM as an internal standard. Aliquots (2 $\mu l)$ of the prepared sample were analyzed on a Pye-Unicam GCD gas chromatograph equipped with a Hewlett-Packard 3390 A integrator and a column packed with 60/80 Carbopack B-4% Carbowax (20 M)-0.8% KOH (Supelco, Bellefonte, Pa.). The sample was injected at a temperature of 150°C and separated at 90°C and a flow rate of 50 ml of N2 per min; detection was performed at 150°C with a flame ionization detector. The concentrations of the methylated amines were determined by comparing the detected peak areas with the calibration curves of the compounds

The fractions obtained by TSK-Gel DEAE-5pw FPLC that stimulated the trimethylamine or dimethylamine:HS-CoM methyltransferase reactions were analyzed for the presence of corrinoids. Protein samples and the hydroxy- B_{12} -HBI standard were diluted 1:1 in 0.04% (wt/vol) KCN, boiled for 5 min, and centrifuged for 10 min in an Eppendorf centrifuge to remove precipitated protein. Samples and the standard were subjected to high-performance liquid chromatography (HPLC) at 40°C on a Hewlett-Packard HP 1090 Series II HPLC equipped with an HP 1046A programmable UV-visible light detector and a 5-μm LiChrosorb RP-18 column (150 by 46 mm; Alltech Europe, Eke, Belgium). After injection of the sample (250 µl), separation was achieved at a flow rate of 0.8 ml/min by elution with a 5-min linear gradient of 0 to 25% (vol/vol) methanol followed by a 10-min linear gradient from 25 to 45% (vol/vol) methanol and a 5-min wash with 45% (vol/vol) methanol. The solvents were buffered in 25 mM sodium acetate (pH 6.0). The eluate was scanned at three pilot wavelengths, 260, 361 and 518 nm, and complete UV-visible light spectra (210 to 600 nm) of the recognized peaks were taken. Corrinoids were identified by their retention times and UV-visible light spectra. The retention time of cyano-B₁₂-HBI was 11.72

Protein concentrations were calculated with Bio-Rad dye reagent concentrate (Bio-Rad Laboratories GmbH, Munich, Germany), with bovine serum albumin used as a standard.

Analytical gel electrophoresis was performed with a Bio-Rad Mini Protean II assembly (Bio-Rad Laboratories) used according to the manual. Native polyacrylamide gel electrophoresis (PAGE) was performed on a 7% homogeneous slab gel, using urease ($M_{\rm r}$, 545,000 and 272,000), bovine serum albumin ($M_{\rm r}$, 132,000 and 66,000), chicken egg albumin ($M_{\rm r}$, 45,000), bovine carbonic anhydrase ($M_{\rm r}$, 29,000), and bovine α -lactalbumin ($M_{\rm r}$, 14,000) as molecular weight standards. Sodium dodecyl sulfate (SDS)-PAGE was carried out on a 10% polyacrylamide gel using a mixture of rabbit muscle phosphorylase b ($M_{\rm r}$, 97,000), bovine serum albumin ($M_{\rm r}$, 66,000), chicken egg albumin ($M_{\rm r}$, 45,000), bovine carbonic anhydrase ($M_{\rm r}$, 29,000), soybean trypsin inhibitor ($M_{\rm r}$, 21,000), and egg white lysozyme ($M_{\rm r}$, 14,000) for calibration. After electrophoresis, the polyacrylamide gels were stained with Coomassie brilliant blue R-250.

Materials. HS-CoM, TES, 2-bromoethanesulfonic acid, and methylcobalamin were purchased from Sigma Chemical Co. (St. Louis, Mo.). Hydroxy-B₁₂-HBI was purified as described by Daas et al. (2). Dithiothreitol was from Serva Feinbiochemika (Heidelberg, Germany). ATP, hexokinase, and myokinase were purchased from Boehringer (Mannheim, Germany). FPLC and liquid chromatography were performed with a Perkin-Elmer series 410 LC BIO Pump equipped with a Perkin-Elmer LC 90 BIO spectrophotometric UV detector. DEAE Sepharose Cl-6B was from Pharmacia LKB Biotechnology A.B. (Uppsala, Sweden). TSK-Gel DEAE-5pw and TSK-Gel HA-1000 FPLC columns were purchased from TOSO-HAAS (Stuttgart, Germany). Gases were supplied by Hoek-Loos (Schiedam, The Netherlands). To remove traces of oxygen, H₂-containing gases were passed over a BASF RO-20 catalyst at room temperature, and N₂ was passed over a prereduced BASF R3-11 catalyst at 150°C. The catalysts were a gift of BASF Aktiengesellschaft (Ludwigshafen, Germany).

RESULTS

Resolution of the enzyme system involved in the conversion of methanol. Cell extracts prepared from methanol-grown cells of *M. barkeri* Fusaro readily converted methanol and HS-CoM into CH₃-S-CoM. By the procedure developed by Daas et al.

(2), the enzyme system involved in CH₃-S-CoM synthesis from *M. barkeri* MS could be resolved into three enzymic components, notably, MT₁, MT₂-hydrogenase, and MAP. Application of the procedure to the Fusaro strain of the organism yielded an identical separation into the three components, and as shown below, only the combination of these was able to reconstitute the methanol:HS-CoM methyltransferase reaction.

The MT_2 -hydrogenase-containing fraction was further resolved by the method described by Grahame (8). By this method, two distinct MT_2 activities were obtained, with the major peak eluting between 0.12 and 0.15 M $(NH_4)_2HPO_4$. Since $MT_2(I)$ is known to elute from hydroxylapatite at about this concentration (8), the MT_2 activity in this peak was attributed to $MT_2(I)$.

Resolution of the enzyme system involved in the conversion of methylamines. Cell extracts of methanol-grown M. barkeri were incapable of the conversion of tri-, di-, or monomethylamine. The capability to use methylated amines as a substrate for growth was achieved only via an intermediary shift from methanol (250 mM) to acetate (120 mM) as the carbon and energy source. Acetate-adapted cells acquired the potential to utilize tri-, di-, and monomethylamine after subsequent transfers to media containing trimethylamine and progressively decreasing amounts of acetate. Once adapted to the methylamine, cell extracts of the organism grown on the substrate were capable of tri-, di-, and monomethylamine conversion with HS-CoM as the methyl group acceptor but lacked the activity of methanol:HS-CoM methyl transfer. In the presence of 2.25 mg of protein and 1 mM 2-bromoethanesulfonate, which is an inhibitor of the terminal methane-forming reaction, the rates of HS-CoM conversion from tri-, di-, and monomethylamine amounted to 89, 80, and 44 nmol/min · mg of protein, respectively. Reactions were strictly dependent on the presence of ATP (optimal concentration, 1 to 2 mM). However, even with an ATP concentration as low as 0.1 mM (20 nmol in the assay), HS-CoM (2,000 nmol) was completely converted to CH₃-S-CoM. This indicates that the compound is required only in catalytic amounts.

Our first approach to resolve the enzyme systems involved in the conversion of the three methylamines consisted of an ammonium sulfate fractionation of the cell extracts. It appeared that the enzyme fraction that was precipitated between 50 and 65% ammonium sulfate saturation was able to catalyze the conversion of tri-, di-, and monomethylamine (Fig. 2). However, the initial rates were low and the reactions took off after lagging. The lag period was relieved by the addition of carefully desalted 65% ammonium sulfate supernatant. In the presence of 25 μ l of the supernatant, the reactions started immediately and the reaction rates compared well with those measured with crude extracts. The enzyme fraction that was pelleted at 50% ammonium sulfate saturation (200 μ g of protein in the assay) did not show such an effect. The 65% ammonium sulfate supernatant alone was not capable of conversion of the methylamines.

The 50 to 65% ammonium sulfate fraction was subsequently subjected to DEAE Sepharose Cl-6B chromatography. When the column fractions were tested with a minimal amount of cell extract (10 to 15 μl), which by itself did not sustain the reaction, two fractions that strongly enhanced the conversion of triand dimethylamine were found. The fraction that eluted between 0.26 and 0.32 M NH₄Cl coincided with MT₂. This fraction also contained part of the viologen-reducing hydrogenase activity. The other was eluted as a broad band between 0 and 0.25 M NH₄Cl.

The fraction containing MT₂ was separated further on a hydroxylapatite FPLC column packed with TSK-Gel HA-1000

6940 WASSENAAR ET AL. J. BACTERIOL.

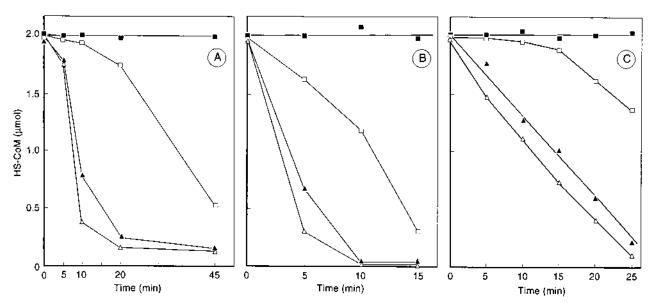


FIG. 2. The methyl group transfer reactions from trimethylamine (A), dimethylamine (B), and monomethylamine (C) to HS-CoM, catalyzed by crude cell extracts and by ammonium sulfate-fractionated extracts from *M. barkeri* Fusaro. The assays were performed as described in Materials and Methods in the presence of 10 mM trimethylamine (A), 10 mM dimethylamine (B), or 10 mM monomethylamine (C) together with 40 μ l of cell extract (\triangle), 40 μ l of the 50 to 65% ammonium sulfate pellet (\square), 40 μ l of 65% ammonium sulfate supernatant (\blacksquare), and 40 μ l each of the ammonium sulfate pellet and supernatant fractions (\triangle).

as described above, yielding only one peak of MT_2 activity. By comparison with the chromatographical properties described by Grahame (8), we concluded that this should represent $MT_2(\Pi)$.

The 0 to 0.25 M NH₄Cl fraction from DEAE Sepharose Cl-6B that stimulated both the tri- and dimethylamine conversion was purified further on an FPLC ion-exchange column packed with TSK-Gel DEAE-5pw. The fraction that was eluted between 0.090 and 0.105 M NH₄Cl strongly stimulated the trimethylamine:HS-CoM methyltransferase reaction, but not the dimethylamine or monomethylamine:HS-CoM methyltransferase reaction, when assayed with limited amounts of crude extract. Native PAGE indicated that the fraction essentially contained only two proteins in about equal amounts, showing apparent molecular masses of 270 and 300 kDa (Fig. 3). HPLC analysis after cyanolysis of the protein fraction un-

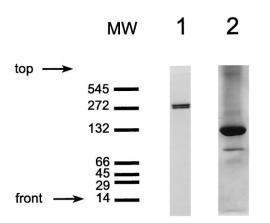


FIG. 3. Native PAGE of trimethylamine methyltransferase (lane 1; 0.090 to 0.105 M NH $_4$ Cl fraction from TSK-Gel DEAE-5pw chromatography; 3 μ g of protein) and dimethylamine methyltransferase (lane 2; 0.107 to 0.115 M NH $_4$ Cl fraction; 11 μ g of protein). The numbers at the left are the molecular weights (in thousands) of the protein standards.

equivocally demonstrated that it harbored cyano-B₁₂-HBI (Fig. 4B). A second fraction that was eluted from the TSK-Gel DEAE-5pw FPLC column between 0.107 and 0.115 M NH₄Cl specifically stimulated dimethylamine conversion. HPLC analysis of the cyanolyzed sample revealed the presence of two chromophoric compounds, notably, cyano-B₁₂-HBI and factor F_{430} (Fig. 4C). Upon native gel electrophoresis of the fraction, two predominant protein bands could be detected: a major band, showing an apparent M_r of 130,000, and a minor band, with a molecular mass of about 100,000 kDa (Fig. 3). Denaturing SDS-PAGE indicated the former to be composed of three subunits (66, 42, and 31 kDa). These data are fully consistent with the presence of factor F₄₃₀-containing CH₃-S-CoM reductase (12). A more careful analysis of the patterns of the enzymes that were eluted from the TSK-Gel DEAE-5pw FPLC column, however, showed that the dimethylamine-converting activity always coincided with the 100-kDa band, which consequently represents dimethylamine methyltransferase.

Reconstitution of the trimethylamine, dimethylamine, and methanol:HS-CoM methyl transfer reactions and evidence for the role of specific MT₂ isoenzymes in the processes. Reconstitution experiments established that trimethylamine conversion could be brought about only by the combination of three fractions obtained during purification of extracts prepared from trimethylamine-grown cells: (i) the 0.090 to 0.105 M NH₄Cl pool from TSK-Gel DEAE-5pw chromatography, (ii) the MT₂(II) pool from TSK-Gel HA-1000 chromatography, and (iii) the 65% ammonium sulfate supernatant (Fig. 5A). A similar combination of fractions also actively catalyzed the methyl group transfer from dimethylamine to HS-CoM if fraction i above was replaced by the 0.107 to 0.115 M NH₄Cl fraction from TSK-Gel DEAE-5pw (Fig. 5B). Likewise, the methanol:HS-CoM methyltransferase reaction could be reconstituted by three protein fractions isolated from methanolgrown cells, notably, MT₁, MT₂(I), and a protein fraction harboring MAP (Fig. 5C).

The role of the MT_2 isoenzymes in the reactions proved to be quite specific. $MT_2(II)$ isolated from trimethylamine-grown

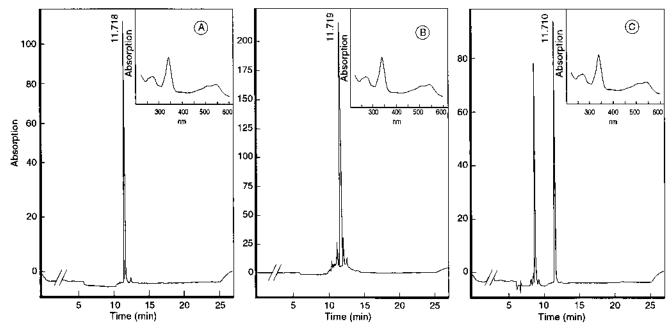


FIG. 4. HPLC analysis of corrinoids. Corrinoid extraction and analysis took place as described in Materials and Methods. The figure shows the HPLC profiles of the cyano-B₁₂-HBI standard (A), trimethylamine methyltransferase (0.090 to 0.105 M NH₄Cl fraction from TSK-Gel DEAE-5pw chromatography) (B), and dimethylamine methyltransferase (0.107 to 0.115 M NH₄Cl fraction from the TSK-Gel DEAE-5pw column) (C). In the insets, the UV-visible light spectra (200 to 600 nm) of the peaks eluting at 11.72 min are shown. Spectral analysis of the peak in panel C having a retention time of 8.91 min demonstrated that it represents factor F₄₃₀.

cells was active only in the conversion of trimethylamine (Fig. 5A) and dimethylamine (Fig. 5B), but it was inactive in the methanol:HS-CoM methyltransferase reaction (Fig. 5C). Conversely, $MT_2(I)$ from the methanol-grown cells showed activity in the methanol, but not in the tri- and dimethylamine, methyl

transfer steps. The combination of the two MT₂ isoenzymes did not further stimulate the reactions, but rather it could have had some inhibitory effect (Fig. 5A and B).

The stoichiometry of the trimethylamine and dimethylamine conversions catalyzed by the respective resolved enzyme sys-

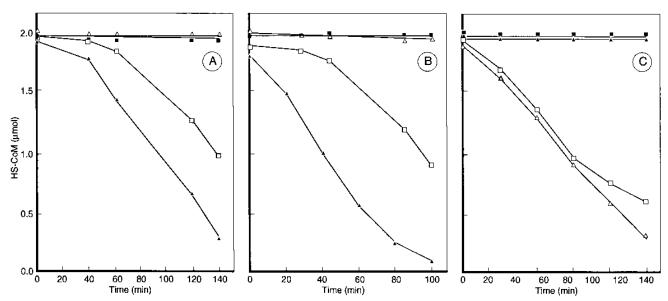


FIG. 5. Reconstitution of the enzyme systems and evidence for the role of specific MT_2 isoenzymes involved in the methyl group transfer reactions from trimethylamine (A), dimethylamine (B), and methanol (C) to HS-CoM. The assays were performed as described in Materials and Methods in the presence of 10 mM trimethylamine (A), 10 mM dimethylamine (B), or 10 mM methanol (C) and 30- μ l amounts each of the following enzymic components: 65% ammonium sulfate supernatant (A and B) and either trimethylamine methyltransferase (A) or dimethylamine methyltransferase (B) or the MAP plus MT_1 fractions from methanol-grown cells (\blacksquare). The assay mixtures were supplemented with 30 μ l (each) of $MT_2(II)$ (\triangle), $MT_2(I)$ (\triangle), or both $MT_2(II)$ and $MT_2(I)$ (\square), isolated from trimethylamine- and methanol-grown cells, respectively. No reactions were observed when either the methyltransferase or the 65% ammonium sulfate supernatant was omitted (A and B) or in the absence of MT_1 or MAP (C). Protein concentrations of the enzyme fractions used were as specified in the Materials and Methods; enzyme denotations are explained in Fig. 1.

6942 WASSENAAR ET AL. J. BACTERIOL.

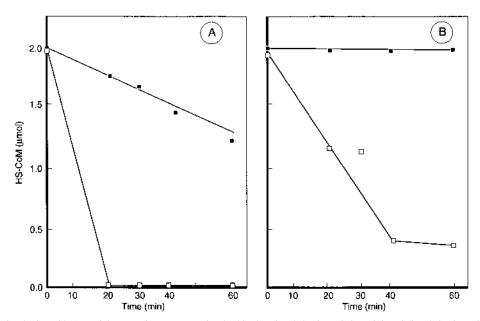


FIG. 6. Effect of preincubation with ATP of the enzymic components involved in the trimethylamine:HS-CoM (A) and dimethylamine:HS-CoM (B) methyltransferase reactions. Preincubations with ATP and the subsequent assays without ATP were performed as described in Materials and Methods in the presence of 10 mM trimethylamine (A) or 10 mM dimethylamine (B) and 40- μ l amounts of the enzymic components. In the curves with solid squares, the $0 \text{ to } 0.25 \text{ M NH}_4\text{Cl}$ and MT_2 -hydrogenase fractions from DEAE-Sepharose Cl-6B chromatography were preincubated with ATP. In the curves with open squares, ATP-preincubated 65% ammonium sulfate supernatant was used. No reactions were observed when ATP pretreated with glucose-hexokinase-myokinase was added to the preincubations or to the assays.

tems was determined by simultaneously measuring the concentrations of HS-CoM, trimethylamine, dimethylamine, and monomethylamine during the incubation periods. In the trimethylamine-converting reaction, the consumption of 1 mol of trimethylamine paralleled the production of 1 mol of dimethylamine and the consumption of 1 mol of HS-CoM. Similarly, the consumption of 1 mol of dimethylamine-converting system is accompanied by the simultaneous consumption of 1 mol of HS-CoM and the production of 1 mol of monomethylamine. These data are in agreement with the transfer of one methyl group from trimethylamine or dimethylamine by the respective systems to HS-CoM, resulting in the simultaneous production of dimethylamine or monomethylamine, respectively.

Evidence for the role of MAP. The methyl group transfers from tri-, di-, and monomethylamine catalyzed by the crude and resolved systems always required the presence of catalytic amounts of ATP. In addition, in the resolved systems, a component(s) present in thoroughly washed 65% ammonium sulfate supernatant was required for activity (Fig. 1 and 4). Above, it was mentioned that the 50 to 65% ammonium sulfate pellet of crude cell extracts could be separated into two parts, the 0 to 0.25 M eluate and the MT₂-hydrogenase fraction, by DEAE Sepharose Cl-6B chromatography. The combination of the two fractions was preincubated with ATP, and excess ATP was subsequently removed by a glucose-hexokinase-myokinase trap. The same was done with the 65% ammonium sulfate supernatant. After this step, reaction mixtures were prepared both from the ATP-preincubated protein fraction and from the nonpreincubated counterpart, and the trimethylamine and dimethylamine conversions were measured without the addition of ATP to the mixture. It appeared that ATP was no longer necessary when ATP-preincubated 65% ammonium sulfate supernatant was added to the assays (Fig. 6). The residual activity in the trimethylamine:HS-CoM methyltransferase reaction with the combined ATP-preincubated Sepharose fractions is likely to be the result of some contamination of the active component in the rather crude enzyme pool. Treatment of the supernatant with air (overnight at 4°C) completely destroyed its activity. In contrast, a similar treatment of the 50 to 65% ammonium sulfate pellet or its subsequently purified enzymic constituents caused at most a short lag in the tri- or dimethylamine methyltransferase reactions, when assayed with anaerobically stored 65% ammonium sulfate supernatant. The role of the oxygen-labile component present in the latter is highly reminiscent of that of MAP in the methanol: HS-CoM methyltransferase reaction (2, 3, 5). Indeed, the MAP fraction isolated from methanol-grown cells was able to substitute for the ammonium sulfate supernatant in the conversion of trimethylamine (Fig. 7A) and in dimethylamine methyl transfer (Fig. 7B). Conversely, the 65% ammonium sulfate supernatant was able to substitute for the MAP fraction in the reconstituted methanol:HS-CoM methyltransferase system (Fig. 7C).

DISCUSSION

CH₃-S-CoM synthesis from methanol in *M. barkeri* is catalyzed by a complex system in which no fewer than five proteins participate (2, 4, 8, 14, 17–22) (Fig. 8A). The system comprises two distinct methyltransferases: MT₁, which accepts the methyl group from the substrate and binds it to the corrinoid prosthetic group, and MT₂, which mediates in the methyl group transfer between MT₁ and HS-CoM. The other three enzymic components, hydrogenase, ferredoxin, and MAP, function in the reactivation of oxidatively inactivated MT₁. Hydrogen, hydrogenase, and ferredoxin, which enhances the rate of reactivation but which is not absolutely required, deliver the reducing equivalents to convert the corrinoid cobalt to the active Co(I) state (2). MAP and ATP play a role in facilitating this thermodynamically unfavorable reduction. In the process, MAP is phosphorylated by ATP, and phosphorylated MAP

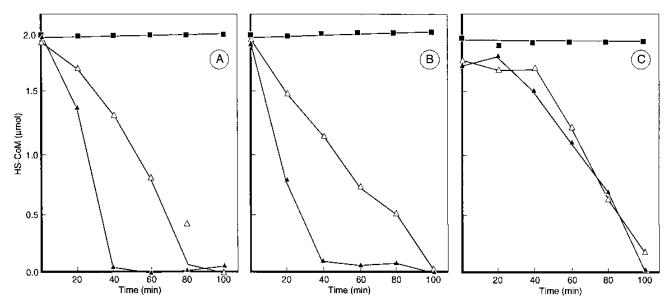


FIG. 7. Capability of the 65% ammonium sulfate supernatant and MAP fractions to substitute in the methyl group transfers from trimethylamine (A), dimethylamine (B), and methanol (C) to HS-CoM. Reactions were performed as described in Materials and Methods in the presence of 10 mM trimethylamine (A), 10 mM dimethylamine (B), or 10 mM methanol (C). The reactions indicated by solid squares contained 40-µl amounts each of MT_2 -hydrogenase and 0 to 0.25 M NH_4 Cl pool (A and B) or MT_2 -hydrogenase and MT_1 (C) obtained by DEAE-Sepharose Cl-6B chromatography of extracts from trimethylamine- or methanol-grown cells, respectively. The reaction mixtures were supplemented with either 40 μ l of 65% ammonium sulfate supernatant (Δ) or 40 μ l of MAP fraction (Δ) derived from trimethylamine- or methanol-grown cells.

induces a conformational change of MT_1 that makes the reduction feasible (2, 3, 5).

The explorative study with crude cell extracts of M. barkeri by Naumann et al. (17) indicated that the conversion of trimethylamine occurs by three consecutive demethylation reactions, each proceeding in a way resembling the methanol:HS-CoM methyl transfer. The work by Burke and Krzycki (1) proved this to be true for the monomethylamine:HS-CoM methyl transfer step (Fig. 8B). In that study (1), a 29-kDa corrinoid protein was methylated by the substrate; the MT₂(II) isoenzyme subsequently transferred this methyl group to HS-CoM. Apart from the two purified components, crude extract had to be added to allow the reactions to proceed, indicating that the process involves more components. The present study establishes and refines the analogy with the methanol:HS-CoM methyltransferase reaction as to the conversions of methylamines. The enzyme systems involved in the demethylation of trimethylamine and dimethylamine could each be separated into three fractions: (i) a specific corrinoid enzyme, (ii) the MT₂(II) fraction from hydroxylapatite FPLC, and (iii) the 65% ammonium sulfate supernatant. In the trimethylamine-converting system, the corrinoid protein was present in the 0.090 to 0.104 M NH₄Cl fraction eluted from the TSK-Gel DEAE-5pw FPLC column. Native PAGE demonstrated the presence of only two protein bands, showing apparent molecular masses of 270 and 300 kDa (Fig. 3). Subsequent resolution of the two proteins indicated that both catalyzed the trimethylamine:HS-CoM methyl transfer (data not shown). Therefore, the proteins may represent either two different isoenzymes or posttranscriptionally modified forms of a common enzyme. The corrinoid protein required for the dimethylamine conversion was recovered in the 0.107 to 0.115 M NH₄Cl eluate from the column described above. Native PAGE revealed the presence of two predominant proteins in this fraction. The major protein was a 130-kDa CH₃-S-CoM reductase which was inactive in dimethylamine conversion. Dimethylamine-converting activity consistently coincided with the presence of the minor

protein band present in the preparation, showing an apparent $M_{\rm r}$ of 100,000. This protein should harbor the B₁₂-HBI found after HPLC analysis of the protein fraction.

From the reconstitution experiments, it can be seen that the trimethylamine:HS-CoM methyltransferase reaction is strictly dependent on the presence of $MT_2(II)$ in the reaction mixture. No evidence for a stimulatory role of $MT_2(I)$ on the reaction was found (Fig. 5). This partly contradicts the results reported by Ferguson et al. (7), who observed that the trimethylamine: HS-CoM methyl transfer in M. barkeri MS was stimulated by both $MT_2(II)$ and $MT_2(II)$. The specific activity of the latter in our hands may be related to a difference in the bacterial strains used or to the fact that our experiments were performed with a more completely resolved enzyme system.

Furthermore, the trimethylamine and dimethylamine conversion in reconstituted systems needed a component which is present in the 65% ammonium sulfate supernatant. The component reduced the lag phase of the trimethylamine-to-HS-CoM and dimethylamine-to-HS-CoM methyltransferase reactions (Fig. 2). Preincubations showed this component to be the primary site of the action of ATP (Fig. 6). Moreover, the 65% ammonium sulfate supernatant could replace the MAP fraction in the methanol conversion. This demonstrates that the component present in the supernatant is MAP or a MAP-like protein. The requirement for the supernatant in the monomethylamine conversion (Fig. 2C) shows that MAP plays an essential role in that reaction as well.

In conclusion, in order to convert its methylated substrates (methanol, methylamines) into CH₃-S-CoM, *M. barkeri* has developed a common strategy taking advantage of the supreme catalytic properties of the corrinoid in methyl group transfer (16) (Fig. 8). In order to keep the group activated, the organism employs a sophisticated ATP-dependent activation system featuring MAP. Yet, a number of substrate-specific variations with a common theme are found. First of all, each specific methyl group transfer reaction requires its own corrinoid protein. Furthermore, two MT₂ isoenzymes for *M. barkeri* have

6944 WASSENAAR ET AL. J. BACTERIOL

(B)

(CH3)NH2
$$\longrightarrow$$
 MAP ATP

H⁺ \longrightarrow CO(I) \longrightarrow CH3-S-CoM + H⁺

CH3-Co(III) \longrightarrow HS-CoM

NH₃

FIG. 8. Analogy between the corrinoid enzyme-dependent methyl group transfers from methanol (A) and tri-, di-, and monomethylamine (B) to HS-CoM. Abbreviations: *MTt*, trimethylamine:B₁₂-HBI methyltransferase; *MTd*, dimethylamine:B₁₂-HBI methyltransferase; and *MTm*, monomethylamine:B₁₂-HBI methyltransferase. The role of MAP in the reductive activation of corrinoid proteins involved in the conversion of the methylamines (Fig. 7B) is proposed to be the same as that in the methanol methyl group transfer (Fig. 7A) but is represented in a more simplified way for the sake of clarity of the scheme.

been characterized (8, 22). One [MT₂(I)] functions in the methanol conversion, while the other [MT₂(II)] was demonstrated to act in monomethylamine methyl transfer (1, 22). The findings presented in this paper show that the tri- and dimethylamine:HS-CoM methyl transfer reactions are strictly dependent on $MT_2(II)$ (Fig. 5). This demonstrates that $MT_2(II)$ is specifically involved in methanogenesis from di- and trimethylamine, making it the specific MT_2 isoenzyme for all methylated amines in M. barkeri Fusaro.

REFERENCES

 Burke, S. A., and J. A. Krzycki. 1995. Involvement of the "A" isozyme of methyltransferase II and the 29-kilodalton corrinoid protein in methanogen-

- esis from monomethylamine. J. Bacteriol. 177:4410-4416.
- Daas, P. J. H., K. A. A. Gerrits, J. T. Keltjens, C. van der Drift, and G. D. Vogels. 1993. Involvement of an activation protein in the methanol:2-mercaptoethanesulfonic acid methyltransferase reaction in *Methanosarcina barkeri*. J. Bacteriol. 175:1278–1283.
- Daas, P. J. H., W. R. Hagen, J. T. Keltjens, C. van der Drift, and G. D. Vogels. 1996. Activation mechanism of methanol:5-hydroxybenzimidazolyl-cobamide methyltransferase from *Methanosarcina barkeri*. J. Biol. Chem. 271:22346–22351
- Daas, P. J. H., W. R. Hagen, J. T. Keltjens, and G. D. Vogels. 1994. Characterization and determination of the redox properties of the 2[4Fe-4S] ferredoxin from *Methanosarcina barkeri* strain MS. FEBS Lett. 356:342–344.
- Daas, P. J. H., R. W. Wassenaar, P. Willemsen, R. J. Theunissen, J. T. Keltjens, C. van der Drift, and G. D. Vogels. 1996. Purification and properties of an enzyme involved in the ATP-dependent activation of the methanol: 2-mercaptoethanesulfonic acid methyltransferase reaction in *Methanosarcina barkeri*. J. Biol. Chem. 271:22339–22345.
- Ellman, G. L. 1958. A colorimetric method for determining low concentrations of mercaptans. Arch. Biochem. Biophys. 74:443

 –450.
- Ferguson, D. J., J. A. Krzycki, and D. A. Grahame. 1996. Specific roles of methylcobamide:coenzyme M methyltransferase in metabolism of methanol and methylamines in *Methanosarcina barkeri*. J. Biol. Chem. 271:5189–5194.
- Grahame, D. A. 1989. Different isoenzymes of methylcobalamin:2-mercaptoethanesulfonic acid methyltransferase predominate in methanol-versus acetate-grown *Methanosarcina barkeri*. J. Biol. Chem. 264:12890–12894.
- Harms, U., and R. K. Thauer. 1996. Methylcobalamin:coenzyme M methyltransferase isoenzymes MtaA and MtbA from *Methanosarcina barkeri*. Cloning, sequencing and differential transcription of the encoding genes, and functional overexpression of the *mtaA* gene in *Escherichia coli*. Eur. J. Biochem. 235:653–659.
- Hippe, H., D. Caspari, K. Fiebig, and G. Gottschalk. 1979. Utilization of trimethylamine and other N-methyl compounds for growth and methane formation by Methanosarcina barkeri. Proc. Natl. Acad. Sci. USA 76:494–498.
- Hutten, T. J., M. H. de Jong, B. P. H. Peeters, C. van der Drift, and G. D. Vogels. 1981. Coenzyme M derivates and their effects on methane formation from carbon dioxide and methanol by cell extracts of *Methanosarcina barkeri*. J. Bacteriol. 145:27–34.
- Jablonski, P. E., and J. G. Ferry. 1991. Purification and properties of methyl coenzyme M methylreductase from acetate-grown *Methanosarcina ther-mophila*. J. Bacteriol. 173:2481–2487.
- Keltjens, J. T., and C. van der Drift. 1986. Electron transfer reactions in methanogens. FEMS Microbiol. Rev. 39:259–303.
- Keltjens, J. T., and G. D. Vogels. 1993. Conversion of methanol and methylamines to methane and carbon dioxide, p. 253–303. *In J. G. Ferry* (ed.), Methanogenesis. Chapman & Hall, New York.
- 15. Kengen, S. W. M., P. J. H. Daas, E. F. G. Duits, J. T. Keltjens, C. van der Drift, and G. D. Vogels. 1992. Isolation of a 5-hydroxybenzimidazolylcobamide containing enzyme involved in the methyltetrahydromethanopterin: CoM methyltransferase reaction in Methanobacterium thermoautotrophicum. Biochem. Biophys. Acta 1118:249–260.
- Matthews, R. G., R. V. Banerjee, and S. W. Ragsdale. 1990. Cobamidedependent methyl transferases. Biofactors 2:147–152.
- Naumann, E., K. Fahlbusch, and G. Gottschalk. 1984. Presence of a trimethylamine:HS-coenzyme M methyltransferase in *Methanosarcina barkeri*. Arch. Microbiol. 138:79–83.
- van der Meijden, P., H. J. Heythuysen, A. Pouwels, F. Houwen, C. van der Drift, and G. D. Vogels. 1983. Methyltransferases involved in methanol conversion by *Methanosarcina barkeri*. Arch. Microbiol. 134:238–242.
- van der Meijden, P., H. J. Heythuysen, H. T. Sliepenbeek, F. P. Houwen, C. van der Drift, and G. D. Vogels. 1983. Activation and inactivation of methanol:2-mercaptoethanesulfonic acid methyltransferase from *Methanosarcina barkeri*. J. Bacteriol. 153:6–11.
- van der Meijden, P., B. W. te Brömmelstroet, C. M. Poirot, C. van der Drift, and G. D. Vogels. 1984. Purification and properties of methanol:5-hydroxybenzimidazolylcobamide methyltransferase from *Methanosarcina barkeri*. J. Bacteriol. 160:629–635.
- van der Meijden, P., C. van der Lest, C. van der Drift, and G. D. Vogels. 1984. Reductive activation of methanol:5-hydroxybenzimidazolylcobamide methyltransferase of *Methanosarcina barkeri*. Biochem. Biophys. Res. Commun. 118:760–766.
- Yelliseev, A., P. Gärtner, U. Harms, D. Linder, and R. K. Thauer. 1993. Function of methylcobalamin:coenzyme M methyltransferase isoenzyme II in *Methanosarcina barkeri*. Arch. Microbiol. 159:530–536.