

## Ionic responses of bean (*Phaseolus vulgaris* L.) plants under salinity stress and humic acid applications

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### Abstract

This study aimed to investigate the effects of different humic acid application methods (control, soil, foliar and soil + foliar) on chlorophyll content, dry matter weight of shoots and roots, concentrations of potassium (K), calcium (Ca) and sodium (Na), and K/Na and Ca/Na ratios of bean plants exposed to increasing salinity levels (0, 50, 100 and 150 mM). The effects of salt damage on shoots and roots of bean plants were significantly differed in humic acid application methods. Chlorophyll content decreased with the increase in salt doses at control and soil application of humic acid, while the decrease in chlorophyll content was lower in foliar application of humic acid. Shoot dry weight was not affected by humic acid applications, while root dry weight increased in soil + foliar application method. Soil + foliar humic acid application caused the highest shoot and root Na concentrations. Shoot Ca (2.61%) concentration in soil + foliar application was significantly higher compared to the other treatments, while the highest Ca concentration in roots (1.55%) was recorded in soil humic acid application method. The highest K concentration in roots was obtained in the control treatment (2.50%) followed by soil + foliar humic acid application (2.48%). The ratios of K/Na and Ca/Na in shoots decreased with the increase in salt application rates. The highest shoot K/Na (1456.1%) and Ca/Na (1274.1%) ratio in humic acid x salt interactions was found in soil application of humic acid without salt treatment. The root and shoot dry matter yield and K and Ca concentrations of the plants indicated that soil+foliar application method has a preventive effect for the plants against the 50 mM salt damage. The results showed that soil+foliar humic acid application in addition to the mineral fertilization required for beans can contribute to the growth and mineral nutrition of the plants under moderate salt stress (50 mM NaCl).

**Keywords:** bean; chlorophyll content; humic acid; NaCl; salinity; salt stress; soil conditioner

### Introduction

Soil salinity has a major impact on plant growth, however, excess salt is an abiotic stress factor that causes both osmotic and ionic stresses by inhibiting plant growth and mineral uptake (Benlloch *et al.*, 2005; Dhanapackiam and Ilyas, 2010). Direct impact of salinity is related to the osmotic stress and ion uptake of plants, while, indirect effect of salinity causes structural deteriorations in the plants. The indirect effects of salinity are synthesis of reactive oxygen species (ROS) that damage DNA, protein, chlorophyll and cell permeability, and causes inhibition of photosynthesis, metabolic toxicity, inhibition of potassium (K) uptake and ultimately cell deaths (Hong *et al.*, 2009). Applications of nitrogen (N), calcium (Ca), potassium (K)

fertilizers and humic substances to the growth environment have been recommended to alleviate the negative effect of soil salinity (Frechilla *et al.*, 2001; Walker and Bernal, 2008).

Humic substances contribute to the plant growth and mineral nutrient uptake by improving the soil structure (Mackowiak *et al.*, 2001; Quni *et al.*, 2014). Clay layers in soils are found as piled on top of each other. Salts cause a reduction in the negative electrical charge, while the positive charges on the edges of the clay minerals increase. The positive charges lead binding of clay layers to each other and soil compaction which inhibits the penetration of plant roots. Humic acid facilitates the penetration of water into clay particles by removing salts from the surfaces of clay particles due to the increase in the net negative charge which causes repulsion of clay particles each other and loosens the soil structure. In addition, the negatively charged carbon molecules (-COOH- carboxyl) of the humic acid are bound to the edges of positively charged clay particles and reduce the net negative charge on clay surfaces (Khaled and Fawy, 2011). This causes repelling of clay particles each other and leads to the loosening of soil.

Studies revealed that humic and fulvic acids stimulate plant growth by directly affecting the physiological functions of plants and indirectly influencing the plant growing environment. Previous studies showed that humic substances accelerate germination, improve the root and above-ground plant organs and increase the uptake of mineral elements (Mackowiak *et al.*, 2001; Nardi *et al.*, 2002; Chen *et al.*, 2004). The aforementioned positive effects led to the extensive use of humic substances by the growers. The aim of this study was to investigate the effects soil, foliar and soil + foliar applications of humic acid on growth and mineral nutrient content of bean (*Phaseolus vulgaris* L.) plant, which is salt sensitive plant (Lopez *et al.*, 2002), under increasing salt stress conditions.

## Materials and Methods

### *Plant and humic acid materials*

A commercial dry bean variety called Yunus-90 was used as the plant material of the experiment. Humic acid material used in the study was a commercially available natural organic soil conditioner (TKI-Humas) containing humic and fulvic acids extracted from leonardit. The chemical properties of humic acid were given in Table 1. The humic+fulvic acids and organic matter contents of the humic material given in the Table 1 were determined by the producer company. Total humic and fulvic acid (%) were analysed by titration using TS 5869 ISO 5073 method (TSE, 2003). Organic matter content (%) was analysed by dry burning at 550 °C as specified in AOAC (Association of official Analytical Chemists) method of 967.05 (Horwitz and Latimer, 2007). Electrical conductivity (EC) and pH of humic material were carried out in a 1:10 (material:water) solution. Nitrogen content of humic material was determined using Kjeldahl method (Bremner and Mulvaney, 1983). Humic acid material was digested using a microwave oven (Mars-6, CEM, Microwave Technology Ltd., Matthews, N.C., USA). The concentrations of sodium (Na), K, phosphorus (P), Ca, magnesium (Mg), iron (Fe), copper (Cu), manganese (Mn) and zinc (Zn) in digested solution were measured using an inductively coupled plasma mass spectrometry (NexION 350 ICP- Mass Spectrometry; Perkin Elmer, USA).

### *Experimental soil*

The pot experiment was carried out in a plant growth chamber and surface soil collected from 0-30 cm depth of University Research fields was used as the potting medium. The particle size distribution of the experimental soil was determined using the hydrometer method, lime content was determined using the method described by Richards (1954), organic matter content was analysed using the modified Walkley-Black method (Nelson and Sommers, 1982), and soil reaction and EC of soil samples were determined in 1:2.5 soil:water mixture. Plant available phosphorus (P) concentration was determined by the method of Watanabe and Olsen (1965) using thermo-aquamete UV spectrophotometer. Available K was extracted with neutral 1M ammonium acetate and determined by using the BWB/XP2011 model flame photometer (Richards, 1954).

Available iron (Fe), copper (Cu), zinc (Zn) and manganese (Mn) were extracted with DTPA (0.005M DTPA + 0.01M CaCl<sub>2</sub> + 0.1M TEA, pH 7.3) and determined by an atomic adsorption spectrophotometer (Analytic-Jena novAA 350, Germany) (Lindsay and Norvell, 1978). The results of analysis indicated that experimental soil was slightly alkaline, non-saline and sandy-loamy textured. Organic matter, K, P, Mn, Zn, and Fe contents were low, while Cu and lime contents were sufficient (Table 2).

**Table 1.** Chemical properties of humic acid used in the study

Properties	Unit	Values	Properties	Unit	Values
pH	-	11.8	Ca	(mg kg <sup>-1</sup> )	5228
EC	(dS m <sup>-1</sup> )	5.6	Mg	(mg kg <sup>-1</sup> )	680
HA+FA	(%)	12.0	Fe	(mg kg <sup>-1</sup> )	550
Organic Matter	(%)	5.0	P	(mg kg <sup>-1</sup> )	72
K	(%)	1.60	Cu	(mg kg <sup>-1</sup> )	2.4
Na	(%)	0.49	Mn	(mg kg <sup>-1</sup> )	6.2
N	(%)	0.23	Zn	(mg kg <sup>-1</sup> )	2.4

HA: Humic acid, FA: Fulvic acid, OM: Organic matter

**Table 2.** Physical and chemical properties of soil used in the experiment

Properties	Unit	Values	Properties	Unit	Values
pH	-	7.67	Organic Matter	(%)	0.65
EC	(dS m <sup>-1</sup> )	1.20	K	(mg kg <sup>-1</sup> )	136
Lime	(%)	11.83	P	(mg kg <sup>-1</sup> )	0.49
Texture			Mn	(mg kg <sup>-1</sup> )	0.89
Clay	(%)	10.28	Cu	(mg kg <sup>-1</sup> )	1.10
Sand	(%)	67.45	Zn	(mg kg <sup>-1</sup> )	0.23
Silt	(%)	22.27	Fe	(mg kg <sup>-1</sup> )	1.20

#### *Basic fertilizer applications*

The soil was sieved through 4 mm sieve and the 4-L plastic pots were filled with 3 kg air dried and sieved soil. Before planting, 200 mg N, 125 mg K kg<sup>-1</sup> and 100 mg P kg<sup>-1</sup>, 5 mg Zn kg<sup>-1</sup> and 2.5 mg Fe kg<sup>-1</sup> were applied to each pot as basic fertilization and the fertilizers were mixed thoroughly with soil. The sources of plant nutrients were (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, KH<sub>2</sub>PO<sub>4</sub>, ZnSO<sub>4</sub>·7H<sub>2</sub>O and Fe-EDTA.

#### *Experimental design and humic acid applications*

The experimental layout was randomized plots with three replications and the treatments were control (basic fertilizer application), soil humic acid, foliar humic acid and soil + foliar humic acid applications. The soil and soil + foliar applications of humic acid per pot containing basic fertilizer and soil mixture was carried out by using 750 mg L<sup>-1</sup> stock solution prepared from humic acid material. Humic acid applied was thoroughly mixed with the soil before planting and the pots were incubated for two weeks.

#### *Plant growth conditions*

Four bean seeds, soaked in deionized water for eight hours, were sown in each pot. The pots were placed in the plant growth room with the temperature set at 25/16 °C (day/night), the relative humidity at 70% and the light regime of 16/8 hours (day/night) at 10 Klux. Germinated seeds were grown under the aforementioned conditions until the harvest. The seedlings with one to two sets of leaves were thinned to two per pot 10 days after the planting. The plants were irrigated with deionized water until the salt applications started. Application of salt solutions (0: control, 50: low, 100: moderate and 150: high mM NaCl) started when the true leaves developed (25 days after sowing), and continued until the harvest.

Foliar and soil + foliar humic acid applications started when the plants reached 5-6 leaf period (35 days after sowing). Humic acid solution of 0.2% was sprayed to the shoots and the upper and lower parts of the leaves by a hand sprayer. The same foliar application was repeated seven days after the first application.

#### *Chlorophyll measurement*

Chlorophyll contents in the fully expanded top three leaves were determined by a SPAD meter (SPAD-502-meter; Minolta Camera Co., Osaka, Japan) before harvesting.

#### *Plant harvest and dry matter*

Bean plants in all pots were harvested in the pre-flowering period from 1 cm above the soil surface 50 days after the sowing. The harvested roots and above-ground parts were washed under tap water and then with deionized water. The harvested plants were dried at 65 °C for 48 hours until reaching to a constant weight and dry matter yield (g pot<sup>-1</sup>) of root and shoot (including leaves) was determined.

#### *Determination of Na, K and Ca*

The dried root and green samples were ground and wet digested in H<sub>2</sub>O:HNO<sub>3</sub> (3:2) acid mixture in a closed-vessels microwave oven (CEM Mars6) to determine the mineral contents. Sodium and K concentrations of plant samples were measured in a flame photometer (BWB / XP2011), and Ca concentration was measured in an atomic absorption spectrophotometer (Analytic-Jena novAA 350, Germany). The accuracy of the analysis performed was checked by analysing the element contents of a reference tomato leaf sample (from National Institute of Standards and Technology, Gaithersburg, MD, USA) using the same analytical method.

#### *Statistical analysis*

The data obtained was subjected to the variance analysis (ANOVA) according to randomized plots experimental design using General Linear Model with the Statview (SAS Institute) software. When a statistical difference was detected for the overall ANOVA ( $p < 0.05$ ), Tukey's test at  $p < 0.05$  level of significance was applied to the mean values of the treatments.

## **Results**

#### *Chlorophyll content*

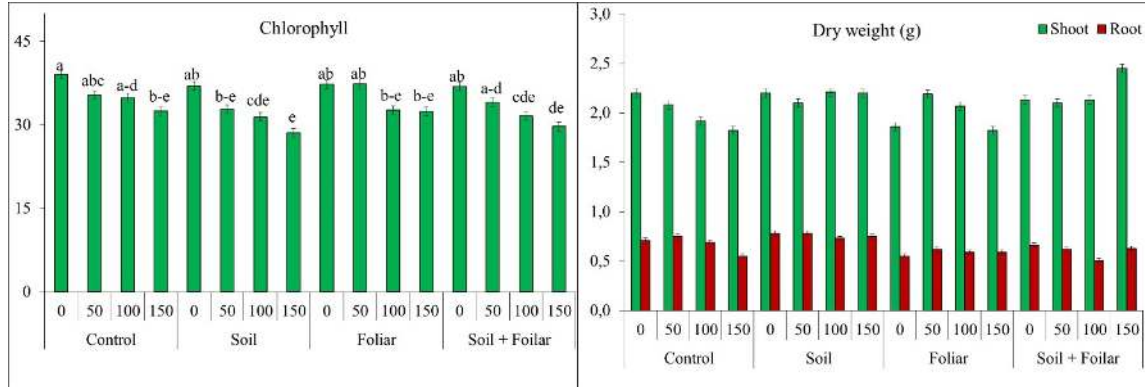
The effect of humic acid and salt applications on chlorophyll content of bean plants was shown in Figure 1. The effect of salt stress and humic acid applications on the chlorophyll content of plants was statistically significant, while the effect of humic acid x salt interaction was not significant. The highest chlorophyll content was obtained in the control group, while the lowest chlorophyll content among the humic acid application methods was recorded in soil humic acid application. The chlorophyll content decreased with the increased salt application rates to bean plants. The lowest chlorophyll content in the soil humic acid application was obtained in high NaCl (150 mM) treatment. The chlorophyll content in foliar humic acid treatment at 150 mM NaCl application was lower compared to the other humic acid treatments, while the plants had the highest chlorophyll content in low (50 mM NaCl) salt application rate.

#### *Dry weight*

Humic acid and salt applications did not have a significant effect on the shoot dry weight, while the root dry weight was significantly ( $p < 0.01$ ) affected only by the humic acid application methods. The highest mean shoot dry weight in humic acid application methods was obtained in soil + foliar application (2.20 g) (Figure 2), while the highest mean shoot dry weight in NaCl treatments was recorded in 50 mM NaCl application

(2.12 g). The highest mean root dry weight in humic acid applications was obtained in soil application (0.76 g) and the highest mean root dry weight in salt treatments was in 50 mM NaCl application (0.69 g) (Figure 1).

The highest shoot dry weight (2.45 g) in humic acid x salt interactions was obtained in 150 mM salt and soil + foliar humic acid treatment (Figure 2). The highest root dry weight in humic acid x salt interactions was obtained in the soil application of humic acid with 0 and 50 mM NaCl (0.78 g for two doses) treatments.



**Figure 1.** Changes in chlorophyll and dry matter weight of beans in humic acid and salt applications

*Changes in Na concentration of plants*

The effects of salt and humic acid applications and humic acid x salt interactions on shoot and root Na concentration of bean plants were statistically significant ( $p < 0.01$ ). The highest Na concentration in shoot was determined in control plants. The highest mean Na concentration in humic acid application methods was obtained in soil + foliar humic acid application (0.021%), In salt treatments, the highest Na concentration was recorded in 150 mM NaCl treatment (0.030%). The highest shoot Na concentration (0.039%) in humic acid x salt interactions was obtained in 150 mM NaCl application and control plants (Figure 3).

The highest mean root Na concentration (0.962%) in the mean humic acid application methods was obtained in soil + foliar humic acid application. The highest Na concentration (0.919%) in the salt treatments was obtained in 150 mM NaCl application. The highest root Na concentration (1.468%) in humic acid x salt interactions was obtained in soil + foliar humic acid with 150 mM NaCl treatment.

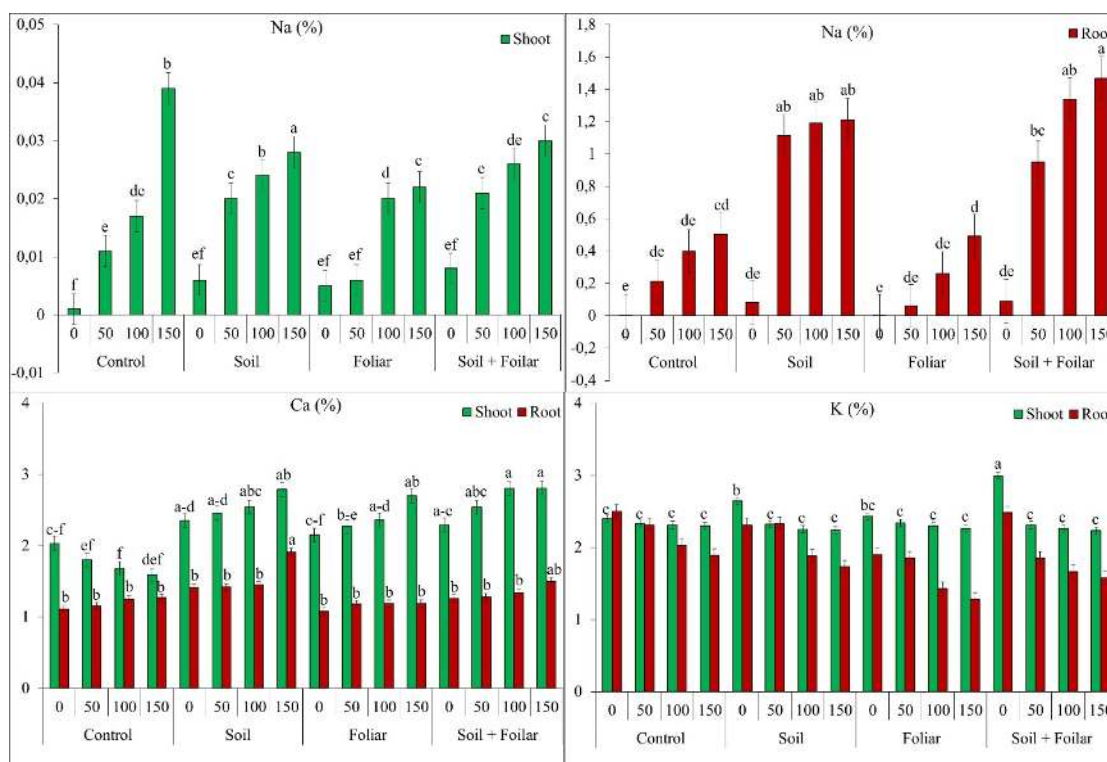
*Changes in Ca concentration of plants*

Salt and humic acid applications had a significant ( $p < 0.01$ ) effect on the shoot and root Ca concentration of plants. The effect of humic acid x salt interaction on shoot Ca concentration was statistically significant ( $p < 0.5$ ), while the effect on root Ca concentration was insignificant. The highest shoot Ca concentration in humic acid applications was obtained in soil + foliar applications (2.61%), followed by soil (2.53%), foliar (2.37%) and control (1.88%) applications (Figure 3). The highest shoot Ca concentration (2.58%) under increasing NaCl application rates was obtained in 150 mM salt treatment. The highest shoot Ca concentration (2.81%) in humic acid x salt interactions was recorded in soil + foliar humic acid x 150mM NaCl treatment (Figure 3). The highest root Ca concentration in humic acid application methods was obtained from soil humic acid application (1.55%), in salt treatments from 150 mM NaCl (1.47%) dose, and in humic acid x salt interaction from foliar humic acid x 100 mM and 150 mM NaCl (1.90%) treatments.



**Figure 2.** Visual effects of humic acid application methods under salinity stress on bean (C: Control, S: Soil, L: Leaf)





**Figure 3.** Sodium (Na), potassium (K) and calcium (Ca) concentrations in shoots and roots of bean plants in humic acid and salt applications

*Changes in K concentration of plants*

Humic acid ( $p < 0.05$ ) and salt ( $p < 0.01$ ) applications and humic acid x salt interactions ( $p < 0.01$ ) had a significant effect on the shoot K concentration of the bean plants. The effect of humic acid and salt on K concentration of roots was also statistically significant ( $p < 0.01$ ). The highest mean shoot K concentration in humic acid applications was obtained in soil + foliar humic acid application. The ranking of K concentration in humic acid applications where NaCl was not applied was soil + foliar > soil > foliar > control, respectively. The highest mean shoot K concentration (2.62%) in salt treatments was obtained in the control plants, and the K concentration of shoots decreased with the increase in the salt application doses (Figure 3).

The highest root K concentration in humic acid applications without salt treatment was obtained in control (2.50%) followed by soil+foliar humic acid application (2.48%). The highest root K concentration (2.33%) in humic acid x salt interaction was recorded in soil humic acid + 50 mM NaCl application. The lowest root K concentration was obtained in the foliar application of humic acid.

*K/Na and Ca/Na ratios in bean grown under salt stress and humic acid applications*

Humic acid application methods did not have a significant effect on shoot K/Na and Ca/Na ratios (Table 3). However, the effect of salt doses and humic acid x salt interaction was statistically significant ( $p < 0.01$ ). The K/Na and Ca/Na ratios of bean plants changed depending on the humic acid application method. The K/Na and Ca/Na ratios of the shoots in humic acid applications were ranked as soil > foliar > soil + foliar > control, respectively. The K/Na and Ca/Na ratios of shoots decreased as the salt application rates increased. The highest shoot K/Na (1456.1%) and Ca/Na ratio (1274.1%) in humic acid x salt interactions were recorded in soil humic acid application without salt treatment. The lowest shoot K/Na and Ca/Na ratios were obtained in 150 mM NaCl treatment of soil humic acid application.

The effects of salt and humic acid applications and humic acid x salt interaction on the K/Na and Ca/Na ratios of the roots were significant (Table 3). Similar to the shoots, the K/Na and Ca/Na ratios of the roots were decreased with increase in the salt application rate. However, the effect of humic acid application methods on root K/Na and Ca/Na ratio was not similar to the effect on the shoot. The highest root K/Na and Ca/Na ratios, except for control plants, were recorded in the foliar humic acid application (4276.1% for K/Na and 2466.0% for Ca/Na), while the lowest ratios were obtained in soil + foliar application (8.1% for K/Na and 4.5% for Ca/Na). The highest root K/Na and Ca/Na ratios in the humic acid x salt interactions were obtained from the plants grown in the salt-free environment of the foliar humic acid application.

**Table 3.** K/Na and Ca/Na ratios in shoots and roots of bean plant in relation to salt and humic acid (HA) application methods

		Shoot		Root	
NaCl	HA ap. met.	K/Na	Ca/Na	K/Na	Ca/Na
Control					
	0	371.4b	314.1b	22911.2	10115.6
	50	213.8b	165.6b	10.9	5.5
	100	140.0b	102.1b	7.6	3.2
	150	60.8b	52.8b	3.9	2.6
Soil					
	0	1456.2a	1274.1a	29.8	17.7
	50	27.9b	29.4b	2.1	1.3
	100	23.6b	26.6b	1.6	1.2
	150	16.3b	20.3b	1.4	1.6
Foliar					
	0	512.8b	453.5b	17058.2	9831.1
	50	371.1b	360.7b	38.0	25.6
	100	112.1b	115.2b	5.6	4.8
	150	102.7b	122.7b	2.6	2.5
Soil+foliar					
	0	383.9b	294.0b	27.0	13.9
	50	212.8b	234.7b	2.9	2.2
	100	143.2b	177.9b	1.2	1.0
	150	92.7b	116.8b	1.1	0.9
	Control	196.5	158.7	5733.4a	2531.7a
	Soil	381.0	337.6	8.7b	5.5c
	Foliar	274.7	263.0	4276.1a	2466.0b
	Soil+foliar	208.2	205.8	8.1b	4.5c
	0	681.1a	583.9a	10006.6a	4994.6a
	50	206.4b	197.6b	13.5b	8.7b
	100	104.7c	105.5b	4.0b	2.6b
	150	68.1d	78.2b	2.3b	1.9b
	HA	1.29 ns	1.54 ns	74.75**	218.83**
	NaCl	14.5**	14.15**	215.42**	656.92**
	HA x NaCl	4.01**	4.70**	74.51**	217.80**

ns: not significant \*\* significant at  $p < 0.01$  level, \* significant at  $p < 0.05$



## Discussion

Total chlorophyll content of leaves was reported decreasing under especially the salt stress conditions (Hong *et al.*, 2010; Gulmezoglu *et al.*, 2016). The decrease in chlorophyll content under salt stress conditions has been associated with the increase in enzyme activity that causes chlorophyll degradation and ion accumulation in leaves as well as salinity-related osmotic stress (Molassiotis *et al.*, 2006; Aydin *et al.*, 2012). Meganid *et al.* (2015) reported that humic acid application in salinity stress causes statistically significant increases in chlorophyll content of bean seedlings. Others have also indicated that humic acid increased the leaf chlorophyll content in soil applications (Karakurt *et al.*, 2009; Meganid *et al.*, 2015).

The chlorophyll content of plants in the different humic acid application methods under salt stress conditions was listed as foliar > soil + foliar > soil, respectively (Figure 1). The effects of foliar humic acid applications on chlorophyll content of bean plants have not been studied. Foliar application of humic acid most likely stimulates the chlorophyll synthesis of bean leaves or delays the salt stress damage. The increase in the leaf chlorophyll content of different plants by the foliar humic substance application has been reported by various studies (Ferrara and Brunetti, 2010; Gao *et al.*, 2020). The results of this study revealed that foliar application of humic acid may have a chlorophyll content increasing the effect which increases the resistance of plants in abiotic stress conditions such as salinity.

Humic acid applications increased the plant biomass, and the effect of humic acid on the increase in root growth was higher than the above-ground components. This result has also been indicated in some of the previous studies (Mackowiak *et al.*, 2001; Nardi *et al.*, 2002). In contrast, Olfati *et al.* (2010) reported that humic substances did not affect the amount of leaves, roots and fruit dry matter yield of the cucumber. The highest shoot dry weight of bean plants was obtained in soil + foliar humic acid application with an increase rate of 18% compared to control plants. The highest root dry weight was recorded in soil humic acid application with an increase rate of 11% compared to the control. Similar to our findings, many studies reported that humic substances promote the plant growth (Nardi *et al.*, 2009; Trevisan *et al.*, 2010; Canellas and Olivares, 2014). Rose *et al.* (2014) reported that humic acid application increased the shoot and root dry weight of different plant species by approximately 22%.

The highest shoot dry weight in humic acid x salt interactions was obtained in 150 mM salt and soil+ foliar humic acid treatment, while the highest root dry weight in humic acid x salt interactions was obtained in the soil application of humic acid with 0 and 50 mM NaCl treatments. These results revealed that humic acid was more efficient in reducing the negative impact of salt stress in aboveground components than that in the roots. Quni *et al.* (2014) stated that humic acid has a direct effect on the germination of seeds, the development of roots and shoots, and an indirect effect on physical, chemical and microbiological properties of soils. The positive effect of humic acid on root growth has been associated with having a large surface area for nutrient uptake and repairing salt tolerance by reducing cell membrane damage against salt stress (Yoon-Ha *et al.*, 2012; Daur and Bakhshwain, 2013).

The highest Na concentration in the shoots and roots of the plants was obtained in the soil + foliar humic acid application. This may be attributed to Na content (0.11%) of the commercial product used as a source of humic acid (Table 2). The application of humic acid from the soil caused an increase in the Na concentration by 50% in the shoots and 220% in the roots compared to the shoots and roots of the control plants. The effect of other humic acid application methods indicated a lower Na concentration of the shoot than the control, while foliar humic acid application caused a lower Na concentration in the roots than the control. Akinci (2009) reported that the shoot and root Na concentrations in humic acid applied pod plants increased by 52 and 86% compared to the control, respectively. However, Taha and Osman (2018) stated that increasing doses of humic substance application reduced the Na concentration of bean leaves. In this study, the absence of salinity symptoms in bean plants and higher Na concentration in the plant roots than in the shoots indicated that humic acid application may prevent the salt damage in above ground plant organs.

The Na concentration in both shoots and roots of bean plants increased with the increasing rates of NaCl. The Na concentration of plants varies depending on the variety and genotypes. However, salt sensitive plants containing approximately 0.25% Na in the leaf dry matter easily show toxicity symptoms (Neumann *et al.*, 1988; Alam, 1999). The toxic effect of Na in the plants becomes more severe with the increase in Na uptake of plants. Excessive salt stress retards the root growth in plants, above ground plant organs are negatively affected and results in usually stunted growth. Soil salinity is an important factor for plant growth, while excess salt content in soil prevents plant growth through increased osmotic potential and toxic effects and also causes competition for nutrient uptake (Benlloch-Gonzalez *et al.*, 2005; Dhanapackiam and Ilyas, 2010). The increase in Na concentration in plant tissues decreases the water uptake of roots. The side roots of plants in salt stress are important for the storage of Na. The major damage associated with the accumulation of Na in plants is the strong inhibitory effect on Ca and K uptake by cells (Hadi and Karimi, 2012; Wang *et al.*, 2013).

Humic acid applications increased the Ca concentration of both shoots and roots as the salt application rate increased. The results revealed that humic acid applications reduced the negative impacts of salt stress, and caused an increase in the Ca concentration of the shoots. The Ca concentration in the shoots reached the highest value in soil + foliar humic acid application, while the highest root Ca concentration was obtained in soil humic acid application. The results on applications of humic acid to different plants (tomato, eggplant, wheat, etc.) showed that the Ca concentration in plants increases as the humic acid application rate increases (Dursun *et al.*, 1999; Rengrudkij and Partida, 2003; Celik *et al.*, 2012). Previous studies are in agreement with the results of this research. Application of humic acid to wheat plants increased the Ca uptake of plants in salty conditions (Asik *et al.*, 2009). Addition of humic substances to the soil increases the root growth and yield of plants. Nikbakht *et al.* (2008) indicated that the application of humic acid up to 1000 mg L<sup>-1</sup> increased Ca concentration in the leaf tissues of the gerbera plant, and the results were attributed to the hormone-like activities of humic acid (Serenella *et al.*, 2002; Zhang *et al.*, 2003), and the ability of humic acid to increase membrane permeability (Valdrighi *et al.*, 1996). This is related to the activities of both hydrophilic and hydrophobic surfaces of humic substances (Chen *et al.*, 2004).

Calcium is an essential mineral nutrient for maintaining a suitable K concentration in plant tissues (Ahanger *et al.*, 2017). Bacha *et al.* (2015) stated that Ca increases the K uptake of tomato plants, contributes to the maintenance of salt tolerance; thus, has beneficial effects on plant growth. The Na ions inhibit the uptake of Ca ions; therefore plants may die due to the Ca deficiency and Na toxicity in saline soils. The Ca content of plants is not as high as the K content. Calcium is a very important general signal carrier used by plants to transmit information in various cellular processes, and also inhibits Na entry into the cell (Gharsallah *et al.*, 2016). In this study, the increase in Ca uptake with the soil or soil + foliar application of humic acid decreased the Na damage and also affected the K uptake.

In this study, the increase in shoot K concentration with the humic acid applications was higher than the root K concentration, whereas under salt stress conditions, soil + foliar humic acid interaction increased the shoot K concentration. The soil + foliar application of humic acid had a positive effect on K concentrations of shoots and roots due to the high K content of the humic acid applied or higher dissolution rate of humic acid in saline conditions. The carboxyl, phenol and hydroxyl groups of humic substances facilitate the K uptake of plants (Wang and Huang, 2001). In addition, Alam (1999) reported that humic acid increases the permeability of electrolytes from biomembranes; thus, the K uptake of plants increases. Similarly, Nikbakht *et al.* (2008) stated that humic substances cause an increase in K content of the plant leaves. In addition, sufficient K nutrition due to the humic acid application has been reported promoting plant growth by maintaining ion homeostatism and osmotic balance under salt stress (Shabala and Cuin, 2008). Therefore, the K was reported reducing Na transfer from roots to shoots in low salt stress, decreasing Na accumulation in the leaves and promoting osmotic regulation (Chakraborty *et al.*, 2016; Gharsallah *et al.*, 2016). The decrease in root and shoot K concentrations with the increase of the salt rate may be related to the increase in Na content of the soil with the increase in NaCl doses as well as to the inhibition of K uptake.

Humic acid applications did not have a statistically significant effect on shoot K/Na and Ca/Na ratios. However, except the control treatment, the highest K/Na and Ca/Na ratios in shoots was obtained in soil application of humic acid, whereas in roots was recorded in foliar application. High K/Na and Ca/Na ratios in plant tissues are extremely important for the growth and metabolism processes of plants grown under saline conditions (Wang *et al.*, 2015). The decrease in shoot Ca/Na and K/Na ratio with the foliar and soil + foliar humic acid application methods under 150 mM NaCl treatment were less than the control and soil humic acid application. The results revealed that foliar and foliar + soil humic acid application methods are effective in reducing salt damage. Similar results were reported by other researchers (Aydin *et al.*, 2012; Taha and Osman, 2018).

The K/Na and Ca/Na ratios of shoots were higher than the K/Na and Ca/Na ratio of the roots. The results were related to the higher shoot K and Ca concentrations than the root, and to the higher root Na concentration than the Na concentration of shoot. The first plant organ that contact with salt is the roots. Therefore, the root Na concentration was higher than the Na concentration of shoot. The Na uptake of plants increased and Ca and K uptake decreased with increasing NaCl application rate. Therefore, the ratios of Ca/Na and K/Na were decreased (Tamam *et al.*, 2008). Acosta-Motos *et al.* (2017) suggested that the low permeability of the plant stem cell keeps the salts away from the aboveground plant parts. Stem cells of plants that can tolerate high salt levels have as high permeability as possible. In addition, plants can prevent themselves from salt damage by pumping Na ions to outside the cells as a salt tolerant mechanism. The Na influx across the cytoplasm through the Na pump and constraining the tolerable quantities of Na ions help plants tolerate the excess salt (Assaha *et al.*, 2017). Yasar *et al.* (2008) observed that the tolerance and adaptation of bean genotypes to salt can also be improved by reducing the transfer of Na ions from plant roots to the shoots and leaves. The researchers reported that the accumulation of K was high in the plant organs with low Na accumulations and vice versa. The results of Yasar *et al.* (2008) are in good agreement with our findings. High uptake of NaCl causes an increase in Na concentration of the cells while a decrease in Ca and K concentrations (Parida and Das, 2005). Saneoka *et al.* (2001) also reported that salt prevents accumulation of K and Ca in the root and shoot, NaCl has a negative effect on the transport of K and Ca to the leaves and causes deficiency of these elements. The results of root and shoot K/Na and Ca/Na ratios, the indicators of harmful impacts of salt on plant growth, showed that the soil application of humic acid is significantly effective in reducing the salt damage in the shoots of bean plants, while, the foliar application is significant to reduce the salt damage in the roots. The humic substances cause increase in the surface area of roots and the number of root hairs; thus the uptake of nutrients such as K and Ca are increase (Marschner, 2002).

## Conclusions

The results of this study showed that plant response to root and shoot salt damage varies depending on different humic acid application methods. Soil application of humic acid increased the shoot and root dry matter yield of the plant. The application of humic acid from soil + foliar was effective in K and Ca uptake of plants. The results revealed that aforementioned humic acid application methods have a protective effect against the salt damage occurring at 50 mM NaCl dose. The application of humic acid from the soil and foliar, along with the mineral fertilization required for beans, will significantly contribute to the plant growth. The findings obtained in plants grown under controlled conditions should be compared and verified in the field studies.

### Authors' Contributions

Conceptualization, N.G.; methodology, N. G.; software, N.G.; validation, N.G. and E.I.; formal analysis, N.G.; investigation, N.G. and E.I.; resources, N.G.; data curation, N.G.; writing-original draft preparation, N.G. and E.I.; writing-review and editing, N.G.; visualization, N.G.; supervision, N.G.; project administration, N.G. All authors have read and agreed to the published version of the manuscript. All authors read and approved the final manuscript.

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### Conflict of Interests

The authors declare that there are no conflicts of interest related to this article.

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