Review Article

Iron assimilation and storage in prokaryotes

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Introduction

Iron is an essential element for most living organisms. Its roles in microbial physiology are numerous. Iron is a constituent of all haem enzymes, which include cytochromes and hydroperoxidases. The common type of ribonucleotide reductase contains iron, and non-iron nitrogenases require an iron protein in a complex for their activity (Robson *et al.*, 1986). However, exceptions do exist; for example, certain lactobacilli devoid of haem and containing a cobalt form of ribonucleotide reductase appear to have no iron requirement for their growth (Archibald, 1983).

Iron, the fourth most-represented element in the earth's crust, is abundant in the environment and should not be a limiting factor for bacterial growth. However, in the presence of oxygen and at a non-acidic pH, iron is particularly insoluble and tends to precipitate as ferric hydroxides. Therefore, bacteria have evolved various powerful systems to overcome this low solubility of external iron (Lankford, 1973). Besides its insolubility, another problem associated with the metabolism of iron resides in its ability to react with reduced forms of oxygen (hydrogen peroxide and superoxide), leading to the production of deleterious free radicals responsible for lipid peroxidation, as well as for alterations in protein and nucleic acids (Flitter et al., 1983). Therefore, in order to avoid such toxicity, iron homeostasis is strictly controlled and results from a co-ordinated integration of assimilation, utilization and storage of this element. Iron uptake is an obvious step to be regulated in response to variations in environmental iron concentration, and has been actively studied in the last decade. The first part of this paper is a non-exhaustive overview of iron assimilation and its control in prokaryotes. The storage of iron within bacteria is also important, particularly in properly tuning the flux of iron required by iron-proteins involved in metabolic reactions. The breakdown and turnover of iron-proteins could lead to an increase in the pool of intracellular free ionic iron that would need to be detoxified. On the other hand, stored iron could be used for the construction of active iron-containing enzymes during an increase in growth rate. It is widely accepted that the amount of an element 'must be regulated by controlled absorbtion' (Underwood, 1971) and thus could explain why much less attention has been accorded to the function of iron storage in bacteria. A review of iron storage in prokaryotes comprises the second part of this paper. Finally, an interesting situation concerning iron uptake and storage by eukaryotic organelles (mitochondria in all cells and plastids in plants), thought to have evolved from prokaryotic progenitors (Gray, 1989), is covered at the end of this review.

Iron assimilation by prokaryotes

Because a tremendous amount of information is available in the literature concerning this topic, a rapid overview will be presented prior to focussing on regulation at the molecular level.

At high levels of environmental iron (>10 μ M), lowaffinity systems are responsible for passive iron uptake and they are poorly understood. High-affinity transport systems operate when iron is limited in the environment and these systems have been extensively studied over the past 30 years. This interest comes probably in part because of the importance that iron acquisition by microbial pathogens from the host plays during infection (Byers, 1987; Crosa, 1989; Byers *et al.*, 1991). Among these high-affinity transport systems, often associated (but not always) with pathogenicity, siderophores (Lankford, 1973) represent low molecular mass carriers (400 to 1000 Da), marked by a very high affinity for ferric iron. Siderophores have highly divergent structures but can be classified into two main groups: phenolates, with

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enterobactin as a classic member, and hydroxamates, with aerobactin as an example. Siderophores are secreted into the environment of cells where they bind ferric iron. Ferrisiderophore assimilation is achieved by specific receptors at the cell surface associated with other periplasmic and cytoplasmic membrane proteins (Bagg & Neilands, 1987*a*); iron reaches the cytoplasm either by internalization of the ferrisiderophore itself or of the iron alone.

Besides these highly specific siderophore systems, some microbes have developed the ability to take up iron directly from naturally occuring iron-binding acids like pyruvate, malate, isocitrate and citrate (Archibald & De Voe, 1980), the latter being by far the most studied of these systems (see below). In mycobacteria, under irondeficient conditions, iron can be transported into cells as ferric salicylate (Messenger *et al.*, 1986); the role of this system, however, is restricted because it does not operate in the presence of phosphate (Ratledge *et al.*, 1974).

Siderophore-mediated iron assimilation

Iron starvation leads to the synthesis of native siderophores by bacteria and to the concomitant induction of the specific transport system responsible for ferrisiderophore reception and internalization. Beside specific transport systems, other systems for transport of exogenous siderophores can also be induced.

In this section, siderophore-mediated uptake systems of Escherichia coli will be reviewed, prior to a brief look at this process in other bacteria. Siderophore biosynthesis requires the expression of different genes, often organized in an operon. Enterobactin synthesis needs the products of genes entA,B,C to produce 2,3-hydroxybenzoic acid from chorismic acid, and the products of genes entD,E,F,G to catalyse production of one enterobactin molecule from three molecules of 2,3-dihydrobenzoic acid and L-serine (reviewed in Crosa, 1989). All these genes are located on the chromosome. In the case of aerobactin, the pColV-K 30 plasmid [Gibson & Magrath, 1969; reviewed in Crosa (1989) and in Bagg & Neilands (1987a)] harbours genes in an operon responsible for the synthesis of this hydroxamate siderophore. The genes iucA,B,C,D encode, respectively, synthetase, acetylase, synthetase and oxygenase. In addition, the gene iutA, also belonging to the aerobactin operon, codes for the specific receptor of ferrienterobactin. Instead of being plasmid-encoded, the aerobactin system is encoded on the chromosome in some invasive E. coli (Valvano & Crosa, 1984, 1988; Valvano et al., 1986). The assimilation of all phenolate and hydroxamate siderophores in E. coli requires two proteins, TonB and ExbB, thought to reside within the cytoplasmic membrane (Postle & Good, 1983). Their necessity for iron uptake has been made evident by mutations in tonB (Hantke & Braun, 1978) and exbB (Eick-Helmerich et al., 1987; Hantke & Zimmermann, 1981). Genetic data have also shown that the TonB protein interacts with outer-membrane receptors (Hantke & Braun, 1978; Braun et al., 1987). These outer-membrane receptors confer specificity to the iron transport systems. In other words, for each ferrisiderophore, a specific receptor is found in the outer membrane: ferrichrome is recognized by FhuA, coprogen and ferric rhodotorulate by FhuE, ferric aerobactin by IutA and ferric enterobactin by FepA (reviewed by Bagg & Neilands, 1987a). Then, a set of proteins are required to define a permease responsible for the translocation and internalization of the ferrisiderophore. Two groups of proteins are involved. The first is composed of the proteins FhuC,D,B, and this system is responsible for translocation accross the inner membrane of all ferric hydroxamates. These proteins share all the characteristics of periplasmic transport systems (Ames, 1986; Braun et al., 1983; Burkhardt & Braun, 1987; Coulton et al., 1987; Fecker & Braun, 1983; Köster & Braun, 1989). The FhuD protein is probably the ferrihydroxamate-binding protein. FhuC faces the cytosol and shares a strong sequence homology with ATPbinding proteins (Burkhardt & Braun, 1987; Coulton et al., 1987) suggesting its involvement in energy coupling. FhuB is the transmembrane component responsible for the translocation of ferrihydroxamates across the membrane.

The second protein system to be described is responsible for ferric-phenolate siderophore uptake (reviewed in Earhart, 1987). It also has similarities with periplasmic permease (Ames, 1986) and requires FepB, a periplasmic protein which binds phenolate siderophores (Pierce *et al.*, 1983; Pierce & Earhart, 1986; Elkins & Earhart, 1989), FepC, an inner-membrane ATP-binding protein (Pierce & Earhart, 1986; Ozenberger *et al.*, 1987; Shea & McIntosh, 1991) and two hydrophobic proteins, FepD and FepG, encoded in the same operon as *fepC* (*fepDGC*) (Chenault & Earhart, 1991; Shea & McIntosh, 1991).

Siderophore production for iron assimilation is not a trait restricted to *E. coli*. Many other bacteria use this method for acquisition of iron. The more well-defined system concerns essentially bacteria pathogenic for plants and animals. Siderophore production by *Pseudomonas* strains has been widely studied: two molecules, pseudobactin and pyochelin are synthesized (reviewed in Crosa, 1989). The ferric-pseudobactin receptor, PupA, from *Pseudomonas putida* WCS 358, has been characterized recently (Bitter *et al.*, 1991). Mycobacteria, including *Mycobacterium leprae*, are able to produce myobactin, a lipid-soluble siderophore (Hall & Ratledge, 1984) and exochelin, which is found in the culture medium of

starved Myobacterium smegmatis (Macham & Ratledge, 1975; Ratledge, 1984). Most of the mesophilic Aeromonas species produce amonabactin and/or enterobactin (Barghouti et al., 1989). Recently, an amonabactin biosynthetic gene (amoA) from Aeromonas hydrophila, has been cloned and sequenced (Barghouti et al., 1991), revealing its homology with E. coli entC. Ferrioxamines, a group of hydroxamate siderophores, are produced by members of the genera *Erwinia* and *Hafnia*. They are also taken up efficiently by bacteria which not produce them, such as E. coli, Yersinia enterocolitica and Serratia marscesens. FoxA, a receptor protein for ferrioxamine, has recently been characterized in Y. enterocolitica (Bäumler & Hantke, 1992b) and a lipoprotein, PCP_{YE}, which facilitates uptake of this siderophore, has been cloned and sequenced (Bäumler & Hantke, 1992a). This protein is homologous to PCP_{Hi}, a lipoprotein of Haemophilus influenzae (Deich et al., 1988).

The marine fish pathogen Vibrio anguillarum has its virulence mediated by the pJM1 plasmid (Crosa, 1980, and reviewed in Crosa, 1989). The siderophore produced by this bacterium is known as anguibactin and resembles pyochelin. The molecular characterization of the iron transport system encoded by pJM1 has recently been reported (Köster *et al.*, 1991). It shares remarkable homology with other iron transport systems described in members of the *Enterobacteriaceae*. However, besides the probable existence of a Fur-like negative repressor (see below) in *V. anguillarum*, a *trans*-acting factor, Taf, is also required for the transport system found on pJM1 (Tolmasky & Crosa, 1984; Tolmasky *et al.*, 1988).

Phytopathogenic bacteria also produce siderophores for iron assimilation (Loper & Buyer, 1991). Although their role in virulence is largely unknown, there is one case, *Erwinia chrysanthemi*, for which production of the catechol siderophore chrysobactin (Persmark *et al.*, 1989) contributes to systemic virulence (Enard *et al.*, 1988). *Erwinia carotovora* subsp. *carotovora* W3C105 has recently been shown to produce aerobactin and a ferric aerobactin receptor encoded by a chromosomal fragment; however, a functional aerobactin acquisition system is not necessary for the pathogenicity of this bacterium (Ishimaru & Loper, 1992).

Iron-dicitrate transport system

Besides siderophore-mediated iron assimilation, organic acids are also able to bind external iron and facilitate its assimilation by bacteria. Of these organic acids, citratedependent iron uptake by *E. coli* K 12 has received much attention over the last 10 years (Hussein *et al.*, 1981). The *fec* operon is responsible for iron-dicitrate uptake. It has been cloned and sequenced (Staudenmaier *et al.*, 1989).

The FecA protein is the outer-membrane receptor. FecB is a protein found in the periplasmic space, while FecC and D are very hydrophilic proteins found in the cytoplasmic membrane. FecE is also a cytoplasmicmembrane protein, sharing two regions of sequence homology with nucleotide-binding proteins. The TonB protein is also required for transport of iron(III)-dicitrate (Zimmermann et al., 1984). Therefore, it appears that the organization of the iron(III)-dicitrate transport system in E. coli is very reminiscent of that described above for ferrisiderophore uptake. However, the iron-dicitrate transport system has the pecularity of being induced by both iron and citrate (Hussein et al., 1981; Zimmermann et al., 1984). A large amount of iron in the medium suppresses the system, because it is under control of the Fur repressor (see below), like the ferrisiderophore systems. The regulatory mechanisms of the iron-dicitrate system have recently been reviewed (Braun & Hantke, 1991). Iron(III)-dicitrate is the inducer and does not need to be internalized to switch-on the fec operon. It just requires to be transported across the outer membrane via the action of the FecA and TonB proteins. The inducer could bind to a transmembrane signalling protein, which could be the product of fecR encoded upstream of the *fec* operon. An activator, encoded by fecI, also found upstream of fec, could be bound to this transmembrane protein in the uninduced state, and be released by iron(III)-dicitrate, leading to its binding onto the operator found upstream of *fecA*, and consequently to initiation of transcription of the *fec* operon.

E. coli is not the only bacterium that uses citrate as a shuttle for iron assimilation. Mycobacterium smegmatis (Messenger & Ratledge, 1982), Neisseria meningitidis (Archibald & De Voe, 1980) and Pseudomonas aeruginosa (Harding & Royt, 1990) can also do so. However, P. aeruginosa also uses citrate as a carbon source while in E. coli and N. meningitidis, dissociation of the metal and ligand occurs within the cell envelope and only iron enters the cell.

Other systems for iron assimilation

The utilization of siderophores and organic acids as iron chelators for assimilation of this element are not the only iron uptake mechanisms used by prokaryotes. Certain pathogenic bacteria, such as *Neisseria gonorrhoeae* and *Neisseria meningitidis*, possess receptor proteins for lactoferrin or transferrin (iron transporter proteins in eukaryotes) allowing the use of their iron (Mickelsen & Sparling, 1981). In the serum of vertebrates, free haemoglobin and haem are not abundant because they are complexed by haptoglobin and haemopexin. Certain haemolytic bacteria are able to acquire iron from these complexes (Francis et al., 1985; Dyer et al., 1987, Stull,

1987; Zakaria-Meehan et al., 1988). Haemophilus influenzae, an important human pathogen, is also dependent on an iron supply for virulence, but does not produce siderophores under iron-deficient conditions, nor is it able to use iron from siderophores produced by other bacteria (Morton & Williams, 1990; Pidcock et al., 1988; Williams et al., 1990). Instead, H. influenzae uses haem, complexed to albumin, haemopexin and haptoglobin, and iron from ferric-transferrin and ferric-lactoferrin as iron sources. These iron complexes bind to outermembrane proteins (Morton & Williams, 1990; Schryvers, 1989; Hanson & Hansen, 1991). Furthermore, two iron-repressed periplasmic proteins have been identified recently (Harkness et al., 1992). One of them is homologous to Fbp, the major iron-binding protein of N. gonorrhoeae and N. meningitidis. Motile aeromonads capable of infecting fishes also use haem present in the host as an iron source (Massad et al., 1992). Aeromonas species produce various cytolytic toxins, some of which are important in virulence (Chakraborty et al., 1987; Howard & Buckley, 1986)), and this could be a way to access iron in haem by disrupting the cells (Byers et al., 1991). Recently, in Serratia marcescens, a system was reported for iron assimilation which requires neither siderophore nor receptor protein production and which is independent of the TonB/ExbB functions (Zimmermann et al., 1989).

Fur: a key regulatory protein in the control of iron uptake

The high-affinity iron transport systems described above are generally activated under iron starvation conditions, as proposed many years ago (Garibaldi & Neilands, 1956). The evidence for an iron-binding repressor, inhibiting siderophore synthesis at high intracellular iron concentration, comes from a mutant of Salmonella typhimurium which constitutively overexpresses ironregulated outer-membrane proteins (Ernst et al., 1978). This mutation, termed fur (ferric uptake regulation), was then obtained and mapped from E. coli, and the corresponding gene was cloned and sequenced (Bagg & Neilands, 1985; Hantke, 1981, 1984; Schaffer et al., 1985). The Fur protein is unusually rich in histidine and has a molecular mass of 17 kDa; it lacks significant homology to any known DNA-binding proteins. A Fur overproducer plasmid has been constructed, allowing a large amount of pure Fur protein to be obtained (Wee et al., 1988); its DNA-binding properties have been characterized (De Lorenzo et al., 1987) and an 'iron box' defined (Bagg & Neilands, 1987a). Furthermore, it has been shown, using the operator/promoter of the aerobactin operon, and an in vitro transcription/translation system, that Fur acts as a classical negative repressor which blocks the aerobactin promoter with Fe(II), or some other metals, as a co-repressor (Bagg & Neilands, 1987b). Fur has also been postulated as a positive regulator in the case of expression control of *ird* (Hantke, 1987), of the iron-dependent superoxide dismutase encoded by *sodB* (Niederhoffer *et al.*, 1989) and of the catabolic pathways allowing growth on succinate, fumarate and acetate (Hantke, 1987). Regulatory studies have also shown that the *fur* gene of *E. coli* could have its expression controlled by its own product in a feedback mechanism, as well as by the cAMP-CAP (catabolite-activator protein) system (De Lorenzo *et al.*, 1988).

Structural studies of Fur using NMR have demonstrated its folding (Saito & Williams, 1991; Saito *et al.*, 1991*a*, *b*). The structure/function relationship has also been recently adressed by a biochemical approach, showing the existence of two domains (Coy & Neilands, 1991): the N-terminal domain of the protein is responsible for DNA-binding and the C-terminal domain binds to metal ions.

The repression of siderophore production and assimilation by the Fur protein under iron-sufficient conditions seems to be a mechanism conserved in various bacterial species, including Yersinia pestis, Shigella dysenteriae, Vibrio cholerae and Corynebacterium diphtheriae (Staggs & Perry, 1991).

Iron deposition and storage in prokaryotes

After assimilation, iron can be used to supply metabolic requirements and/or stored internally in various forms. Mechanisms of iron deposition and storage, as well as functions played by the different compounds involved in these processes, have been investigated much less than iron assimilation, and are reviewed below.

Bacteriomagnetite

Amorphous colloidal species of iron hydroxides have been found associated with bacterial surfaces (Ferris et al., 1989; Ghiorse, 1984), but the most characterized metal deposition in prokaryotes is magnetite (Fe_3O_4), in the form of small membrane-bound magnetosomes (Blakemore, 1975, 1982; Frankel et al., 1979). The morphology of these magnetosomes has been studied mainly in Aquaspirillum magnetotacticum, where they are truncated octahedrons (Blakemore, 1982; Matsuda et al., 1983) with a thin dusting of amorphous iron on their surface (Mann et al., 1984). The magnetosomes are arranged in one or more lines along the cell axis, leading to alignment of the cell to the geomagnetic field (Blakemore, 1982). Each magnetite crystal is surrounded by a lipid bilayer membrane harbouring two unique proteins (Gorby et al., 1988).

Bacterioferritin

Because of the insolubility and toxicity of iron in the presence of oxygen, it is likely that all aerobic organisms have evolved a class of ubiquitous iron-storage proteins, the ferritins, able to sequester a few thousand iron atoms in their central cavity, in a soluble, non-toxic, bioavailable form (Grossman *et al.*, 1992).

Bacterioferritins have been purified and characterized from E. coli (Yariv et al., 1981), Azotobacter vinelandii (Stiefel & Watt, 1979), Azotobacter chroococcum (Chen & Crichton, 1982), Pseudomonas aeruginosa (Moore et al., 1986), Nitrobacter winogradskyi (Kurokawa et al., 1989) and the cyanobacterium Synechocystis PCC 6803 (Laulhère et al., 1992). They are multimeric (24-mers) proteins with subunits arranged in a 432 symmetry, resembling ferritin from eukaryotes (Smith et al., 1989; Harrison et al., 1991). Some important differences, however, do exist between eukaryotic and prokaryotic ferritins. Bacterioferritins contain haem residues (1 haem per 2 to 8 subunits) and in certain cases their identity with previously characterized cytochromes has been demonstrated [cytochrome b-1 in E. coli (Smith et al., 1988); cytochrome b-557.5 in A. vinelandii (Stiefel & Watt, 1979); cytochrome b-559 in N. winogradskyi (Kurokawa et al., 1989)]. The axial ligands of haem have recently been shown to be methionine residues in bacterioferritin from P. aeruginosa (Cheesman et al., 1990). Bacteroferritin iron-cores are amorphous due to the presence of a high amount of inorganic phosphate within them (Fe atoms: P atoms varying from 1/1 to 1/2), but this property depends upon the composition of the medium rather than upon the protein shell (Mann et al., 1986, 1987; Harrison et al., 1991). The complete amino acid sequences of E. coli and A. vinelandii bacterioferritin have been determined by cloning and sequencing the corresponding bfr gene (Andrews et al., 1989; Grossman et al., 1992). Partial amino acid sequences are also available for N. winogradskyi (Kurokawa et al., 1989) and Synechocystis PCC 6803 (Laulhère et al., 1992) bacterioferritins. These data and immunological results (Andrews et al., 1991) have pointed out the conservation of bacterioferritins among various bacterial species. However, they share a low sequence homology (between 24 and 29%) with eukaryotic ferritins, which could, nevertheless, be compatible with a common origin (Grossman et al., 1992). Only one type of subunit has been described for bacterioferritin, although a new gene has been characterized recently in E. coli, sharing a greater sequence homology with the eukaryotic Hferritin chain than the bfr gene (Izuhara et al., 1991); furthermore, in A. vinelandii, two non-identical subunits of bacterioferritin have been described (Harker & Wullstein, 1985). The question concerning bacterioferritin heteropolymers, as for ferritin in eukaryotes, is still open.

The function of bacterioferritin in vivo remains obscure. A bfr mutation was found not to affect growth of E. coli (as reported by S. C. Andrews, Conference communication), and the amount of bacterioferritin subunit appears to be unchanged in Synechocystis PCC 6803 whatever the iron status of the cells, even under iron starvation conditions (Laulhère et al., 1992). This raises the question of whether or not bacterioferritin synthesis responds to excess iron. A mechanism for the control of E. coli BFR synthesis involving anti-sense RNA expression regulated by the Fur protein has been proposed (Andrews et al., 1989). A structure reminiscent of the iron responsive element (IRE), which is involved in the translational regulation of ferritin synthesis in response to iron in eukaryotic cells, has also been described in the case of A. vinelandii bfr (Grossman et al., 1992). No experimental data supporting these regulatory hypotheses have been reported.

Other compounds participating in iron storage

In E. coli bacterioferritin iron represents no more than 1% of the total cellular iron and the bulk of iron in the iron-rich cells is in the form of aggregates (Bauminger et al., 1980), which could be associated with a novel form of iron-protein (Matzanke et al., 1989). More recently, Laulhère et al. (1992) have shown that the bulk of iron in Synechocystis PCC 6803 is associated with low molecular mass (<10 kDa) compounds of unknown composition, which fulfil the storage function. These authors hypothesized that the abundance of saturable low molecular mass molecules, which could vary according to the cellular iron content (Bauminger et al., 1980), may control the fate of newly imported iron. In the case of an iron overload, the small molecule storage pool would take care of the excess iron and bacterioferritin would hardly be involved. On the other hand, in the case of iron deficiency, a rapid overflow of iron from the saturable low molecular mass molecules is conceivable and bacterioferritin would buffer this release of iron prior to its use for metabolic requirements.

Iron uptake and storage by organelles of eukaryotic cells

Mitochondria present in all eukaryotic cells and plastids present in plant cells have evolved from prokaryotic progenitors that invaded primitive eukaryotic cells (Gray, 1989). For this reason, it is relevant to report in this review our knowledge of iron uptake and storage by these organelles. The means by which iron is taken up by mitochondria is not well documented and no information is available for iron assimilation by plastids. Iron ferritin can be taken up in low amounts by mitochondria from hepatocytes (Sibille *et al.*, 1989) but the major pathway by which iron is assimilated by mitochondria involves mainly an AMP-Fe complex and, to a lower extent, ATP-Fe (Weaver & Pollack, 1990). Two kinds of mitochondrial receptors are involved: one that accepts iron from ATP with low efficiency and one from AMP with high efficiency.

No reports have appeared concerning iron-storage molecules in mitochondria. In contrast, it is well established that in plants, ferritins are localized in chloroplasts and non-green plastids, although they are encoded by nuclear genes (Seckback, 1982; Lescure *et al.*, 1991). Interestingly, it has been reported that plant ferritins have an amino acid sequence closely related to animal ferritin sequences and not to bacterioferritin sequences, including those of cyanobacteria, thought to be the ancestors of plastids; relics of a bacterioferritin gene are still present on the chloroplast genome, however (Laulhère *et al.*, 1992; Ragland *et al.*, 1990).

Conclusion

Iron assimilation is well documented in various model systems, particularly $E. \, coli$. It is clear that the knowledge gained from these systems will help to understand, and eventually to control, iron uptake by pathogenic bacteria, in order to have new tools to treat important diseases. The function and regulation in bacteria of the expression of the iron-storage protein ferritin is unclear, and deserves further study. Iron assimilation by eukaryotic organelles of prokaryotic type (i.e. mitochondria and plastids) is also an important area that has not been intensively scrutinized. The study of iron assimilation and storage in prokaryotes has provided much information related to a wide range of scientific activities, ranging from molecular genetics to pathology and agronomy, and more remains to be discovered.

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