## Iron-mediated Intermolecular $\boldsymbol{N}$-Group Transfer Chemistry with Olefinic Substrates

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## General Considerations

All manipulations of metal complexes were carried out in the absence of water and dioxygen using standard Schlenk techniques, or in an MBraun inert atmosphere drybox under a dinitrogen atmosphere. All glassware was oven dried for a minimum of 1 h and cooled in an evacuated antechamber prior to use in the drybox. Benzene and diethyl ether were dried and deoxygenated on a Glass Contour System (SG Water USA, Nashua, NH) and stored over $4 \AA$ molecular sieves (Strem) prior to use. Chloroform- $d$ was purchased from Cambridge Isotope Labs and used as received. Benzene- $d_{6}$ was purchased from Cambridge Isotope Labs and was degassed and stored over $4 \AA$ molecular sieves prior to use. All reagents, unless otherwise specified, were purchased from Aldrich and used as received. Cis- $\beta$-deuterostyrene was synthesized according to literature procedure, ${ }^{1}$ using Schwartz's reagent and freshly prepared $d_{1}$-phenylacetylene. Allylic amination substrates were distilled from $\mathrm{CaH}_{2}$ and deoxygenated prior to use in the amination reactions. Celite® 545 (J. T. Baker) was dried in a Schlenk flask for 24 h under dynamic vacuum while heating to at least $150{ }^{\circ} \mathrm{C}$ prior to use in a drybox. Silica gel 32-63 $\mu$ (AIC, Framingham, MA) was used as received.

## Characterization and Physical Measurements

${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}$, and were recorded on Varian Mercury 400 MHz or Varian Unity/Inova 500 MHz spectrometers. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR chemical shifts are reported relative to $\mathrm{SiMe}_{4}$ using the chemical shift of residual solvent peaks as reference. Gas chromatography/mass spectrometry (GC/MS) was performed on an Agilent GC/MS 5975 Turbo system. Elemental analyses were carried out by Complete Analysis Laboratories, Inc. (Parsippany, NJ).

## Catalysis and Competition Experiments

General Procedure for Amination and Aziridination Reactions. Under an inert $\mathrm{N}_{2}$ atmosphere, 1 -azidoadamantane ( $1-10$ equiv.) was added to a stirring solution of $\left({ }^{\mathrm{Ad}} \mathrm{L}_{\mathrm{Cl}_{2}}\right) \mathrm{FeCl}\left(\mathrm{OEt}_{2}\right)(\mathbf{1})(20 \mathrm{mg}, 0.028 \mathrm{mmol})$ or $\left({ }^{\mathrm{tBu}} \mathrm{L}_{\mathrm{Cl}_{2}}\right) \mathrm{FeCl}\left(\mathrm{OEt}_{2}\right)(\mathbf{2})(16 \mathrm{mg}, 0.028 \mathrm{mmol})$ in 1 mL of substrate in a 20 mL scintillation vial. The resultant inky, dark red solution was stirred for 12 hours at $25^{\circ} \mathrm{C}$. The dark mixture was flash chromatographed through a short pipette of triethylamine-treated silica gel (10:1 hexanes:EtOAc) to yield a brightly colored solution. The

1. L. T. Ball, G. C. Lloyd-Jones and C. A. Russell, Chem.-Eur. J., 2012, 18, 2931.
solvent was removed in vacuo to yield pure amine or aziridine product. The yields were determined by ${ }^{1} \mathrm{H}$ NMR via integration against ferrocene, averaging over three runs for each substrate. GC/MS yields for allylic amination reactions were also obtained after obtaining a calibration curve for the desired allylic amine products. Products were isolated for characterization via silica gel flash chromatography using 9:1 DCM:MeOH as eluent.

## Determination of Substituent Effects for Amination and Aziridination Reactions via

 Competition Experiments. Under an inert $\mathrm{N}_{2}$ atmosphere, 1 mL of an equimolar mixture of the two substrates was added to 1 -azidoadamantane ( $6.1 \mathrm{mg}, 0.028 \mathrm{mmol}, 1$ equiv) and $\left({ }^{\mathrm{Ad}} \mathrm{L}_{\mathrm{Cl} 2}\right) \mathrm{FeCl}\left(\mathrm{OEt}_{2}\right)(\mathbf{1})(10 \mathrm{mg}, 0.014 \mathrm{mmol}, 1$ equiv.) in a 20 mL scintillation vial. The resultant inky, dark red solution was stirred for 12 hours at $25{ }^{\circ} \mathrm{C}$. The dark mixture was flash chromatographed through a short pipette of silica gel $\left(20: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}: \mathrm{MeOH}\right)$ to yield a red solution. For the aziridination reactions, relative ratios were determined by ${ }^{1} \mathrm{H}$ NMR via integration of the benzylic resonance (dd, $\delta \sim 2.5-3 \mathrm{ppm}$ ). For the amination reactions, relative ratios were determined by GC/MS. Yields were averaged over three runs. Table S-1 lists values and parameters used to plot the data.Table S-1. Competition Experiment Details. Variation of $\log k_{\mathrm{R}}$ with $\sigma^{+}, \sigma_{\mathrm{mb}}, \sigma_{\mathrm{JJ}}{ }^{\bullet}$ scales for the aziridination and amination of para-substituted styrenes and toluenes.

| Substrate (para substituent) | Product Ratio | $\begin{gathered} \log k \\ \text { (exptl_avg) } \end{gathered}$ | $\begin{gathered} \log k_{\text {avg }} \\ \text { (std dev) } \end{gathered}$ | $\sigma^{+}$ | $\sigma_{\mathrm{mb}}$ | $\sigma_{\mathrm{JJ}}{ }^{\circ}$ | $\begin{gathered} \log k_{\mathrm{R}} \\ \left(\text { (calcd) }{ }^{b}\right. \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Aziridination |  |  |  |  |  |  |  |
| Cl | 1.15 | 0.06070 | $\begin{aligned} & \hline 0.0311 \\ & (0.014) \end{aligned}$ | 0.11 | 0.11 | 0.18 | 0.0324 |
|  | 1.09 | 0.03742 |  |  |  |  |  |
|  | 1.08 | 0.03342 |  |  |  |  |  |
| Br | 1.13 | 0.05308 | $\begin{aligned} & 0.0452 \\ & (0.015) \end{aligned}$ | 0.15 | 0.13 | 0.26 | 0.0459 |
|  | 1.07 | 0.02938 |  |  |  |  |  |
|  | 1.06 | 0.02531 |  |  |  |  |  |
| OMe | 1.21 | 0.08279 | $\begin{aligned} & \hline 0.0801 \\ & (0.004) \end{aligned}$ | -0.78 | -0.77 | 0.42 | 0.0793 |
|  | 1.19 | 0.07555 |  |  |  |  |  |
|  | 1.21 | 0.08278 |  |  |  |  |  |
| Amination |  |  |  |  |  |  |  |
| Cl | 1.09 | 0.03743 | $\begin{aligned} & 0.0559 \\ & (0.036) \end{aligned}$ | 0.11 | 0.11 | 0.18 | 0.0521 |
|  | 1.25 | 0.09691 |  |  |  |  |  |
|  | 1.08 | 0.03342 |  |  |  |  |  |
| Br | 1.17 | 0.06819 | $\begin{aligned} & \hline 0.0703 \\ & (0.022) \end{aligned}$ | 0.15 | 0.13 | 0.26 | 0.0725 |
|  | 1.12 | 0.04922 |  |  |  |  |  |
|  | 1.24 | 0.09342 |  |  |  |  |  |
| OMe | 1.09 | 0.03743 | $\begin{aligned} & \hline 0.0334 \\ & (0.007) \end{aligned}$ | -0.78 | -0.77 | 0.42 | 0.0333 |
|  | 1.09 | 0.03743 |  |  |  |  |  |
|  | 1.06 | 0.02531 |  |  |  |  |  |
| $t \mathrm{Bu}$ | 1.2 | 0.07918 | $\begin{aligned} & 0.0426 \\ & (0.032) \end{aligned}$ | -0.26 | -0.22 | 0.26 | 0.0427 |
|  | 1.05 | 0.02119 |  |  |  |  |  |
|  | 1.06 | 0.02531 |  |  |  |  |  |

## Characterization Data


$N$-benzyladamantan-1-amine ${ }^{2}$

$\boldsymbol{N}$-(4-(tert-butyl)benzyl)adamantan-1-amine: ${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta / \mathrm{ppm} 7.34$ (d, $J=$ $8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.29(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 3.73(\mathrm{~s}, 3 \mathrm{H}), 2.10(\mathrm{~m}, 3 \mathrm{H}), 1.77-1.59(\mathrm{~m}, 12 \mathrm{H}), 1.30(\mathrm{~m}$, 9H). ${ }^{13} \mathbf{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 149.6,138.3,128.0,125.3,51.1,44.7,42.7,36.7,34.4$, 31.4, 29.6. IR (thin film) $\mathrm{n}_{\max }=2966,2908,2852,1718,1516,1454,1362,1311,1265,1136$, $1097 \mathrm{~cm}^{-1}$. HRMS $\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ Calc. $298.2564\left[\mathrm{C}_{21} \mathrm{H}_{31} \mathrm{~N}+\mathrm{H}\right]^{+}$, Found $298.2571[\mathrm{M}+\mathrm{H}]^{+}$.

$\boldsymbol{N}$-(4-chlorobenzyl)adamantan-1-amine: ${ }^{\mathbf{1}} \mathbf{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 7.32-7.21$ (m, 4H), $3.72(\mathrm{~s}, 3 \mathrm{H}), 2.11-2.05(\mathrm{~m}, 3 \mathrm{H}), 1.71-1.56(\mathrm{~m}, 12 \mathrm{H}){ }^{13} \mathbf{C}$ NMR ( $\left.125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 140.0$, $132.3,129.6,128.4,51.0,44.4,42.7,36.6,31.4,29.7$. IR (thin film) $\mathrm{n}_{\max }=2931,2852,1723$, 1493, 1451, 1359, 1312, 1265, 1137, 1097. HRMS (ESI $\left.{ }^{+}\right) m / z$ Calc. $276.1549\left[\mathrm{C}_{17} \mathrm{H}_{22} \mathrm{ClN}+\mathrm{H}\right]^{+}$, Found $276.1544[\mathrm{M}+\mathrm{H}]^{+}$.

$\boldsymbol{N}$-(4-bromobenzyl)adamantan-1-amine: ${ }^{\mathbf{1}} \mathbf{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 7.30(\mathrm{~d}, J=8.6 \mathrm{~Hz}$, $2 \mathrm{H}), 6.84(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.76(\mathrm{~s}, 3 \mathrm{H}), 3.71(\mathrm{~s}, 3 \mathrm{H}), 2.08(\mathrm{~m}, 3 \mathrm{H}), 1.75(\mathrm{~m}, 6 \mathrm{H}), 1.65(\mathrm{q}, J=$ $12.1 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathbf{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta: 158.7,131.8,130.0,113.8,55.2,53.4,44.3,42.0$,
2. For physical and spectroscopic characterization data, see: S. Calet, F. Urso and H. Alper, J. Am. Chem. Soc., 1989, 111, 931.
36.5, 31.6, 29.5. IR (thin film) $\mathrm{n}_{\max }=2960,2932,2912,1720,1613,1513,1458,1252,1248$, 1137, $1037 \mathrm{~cm}^{-1}$. HRMS (ESI $) m / z$ Calc. $272.2014\left[\mathrm{C}_{18} \mathrm{H}_{25} \mathrm{NO}+\mathrm{H}\right]^{+}$, Found $272.2005[\mathrm{M}+\mathrm{H}]^{+}$.

$N$-(4-methoxybenzyl)adamantan-1-amine: ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 7.42(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, $2 \mathrm{H}), 7.23(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.71(\mathrm{~s}, 2 \mathrm{H}), 2.09(\mathrm{~m}, 3 \mathrm{H}), 1.73-1.63(\mathrm{~m}, 12 \mathrm{H}) .{ }^{13} \mathbf{C}$ NMR (100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta / \mathrm{ppm} 140.8,131.6,130.2,120.7,53.7,44.7,43.0,36.8,29.9$. IR (thin film) $\mathrm{n}_{\max }$ $=2911,2851,1702,1591,1488,1450,1358,1098,1070 \mathrm{~cm}^{-1}$. HRMS $\left(\right.$ ESI $\left.^{+}\right) \mathrm{m} / \mathrm{z}$ Calc. $320.0954\left[\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{NBr}+\mathrm{H}\right]^{+}$, Found $320.0968[\mathrm{M}+\mathrm{H}]^{+}$.


1-(adamantan-1-yl)-2-phenylaziridine: ${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta / \mathrm{ppm} 7.34-7.30$ (m, 2 H), $7.28(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.22-7.18(\mathrm{~m}, 1 \mathrm{H}), 2.81(\mathrm{dd}, J=3.2,6.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.10$ (br. s., 3 H), $2.07(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.72-1.66(\mathrm{~m}, 3 \mathrm{H}), 1.64(\mathrm{~m}, 6 \mathrm{H}), 1.62(\mathrm{~d}, J=3.2 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta / \mathrm{ppm} 141.5,128.1,126.7,126.5,52.8,40.4,39.8,31.7,29.5,28.4$. HRMS (ESI $\left.{ }^{+}\right) \mathrm{m} / \mathrm{z}$ Calc. $254.1909\left[\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{~N}+\mathrm{H}\right]^{+}$, Found $254.1940[\mathrm{M}+\mathrm{H}]^{+}$.


1-(adamantan-1-yl)-3-deutero-2-phenylaziridine: ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 7.33-7.25$ (m, 4 H ), $7.23-7.17$ (m, 1 H ), 2.81 (d, $J=6.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.09 (br. s., 3 H ), 2.06 (d, $J=6.4 \mathrm{~Hz}, 1$ $\mathrm{H}), 1.72-1.66(\mathrm{~m}, 3 \mathrm{H}), 1.62$ (br. s, 6 H ).


1-(adamantan-1-yl)-2-(4-chlorophenyl)aziridine: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta / \mathrm{ppm} 7.24$ (m, $4 \mathrm{H}), 2.76(\mathrm{dd}, J=6.3,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.12-2.01(\mathrm{~m}, 5 \mathrm{H}), 1.79-1.47(\mathrm{~m}, 12 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (125 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta / \mathrm{ppm} 140.4,132.3,128.4,128.2,53.1,40.6,37.0,34.9,31.8,29.7$. IR (thin film) $\mathrm{n}_{\max }=2955,2852,1722,1492,1456,1356,1311,1265,1136,1076 \mathrm{~cm}^{-1}$. HRMS $\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ Calc. $288.1514\left[\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{NCl}+\mathrm{H}\right]^{+}$, Found $288.1508[\mathrm{M}+\mathrm{H}]^{+}$.


1-(adamantan-1-yl)-2-(4-bromophenyl)aziridine: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta / \mathrm{ppm} 7.38(\mathrm{~d}$, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.18(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.74(\mathrm{dd}, J=6.4,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.11-2.02(\mathrm{~m}, 5 \mathrm{H})$, $1.76-1.37(\mathrm{~m}, 12 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta / \mathrm{ppm} 144.7,140.9,131.4,128.6,53.1,45.7$, 40.6, 37.0, 31.8, 29.7. IR (thin film) $\mathrm{n}_{\max }=2910,2899,1488,1450,1358,1289,1136,1012 \mathrm{~cm}^{-}$ ${ }^{1}$. HRMS $\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ Calc. $332.1008\left[\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{NBr}+\mathrm{H}\right]^{+}$, Found $332.1000[\mathrm{M}+\mathrm{H}]^{+}$.


1-(adamantan-1-yl)-2-(4-methoxyphenyl)aziridine: ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.29$ (d, $J$ $=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.89(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.84(\mathrm{~s}, 3 \mathrm{H}), 2.17-2.13(\mathrm{~m}, 5 \mathrm{H}), 1.71-1.30(\mathrm{~m}, 12 \mathrm{H})$. ${ }^{13} \mathbf{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta / \mathrm{ppm} 164.3,131.4,127.8,113.7,55.4,52.8,40.5,37.0,31.8$, 29.9. IR (thin film) $=2959,2913,2855,1613,1513,1300,1248,1036 \mathrm{~cm}^{-1} . \mathbf{H R M S}^{\left(\mathrm{ESI}^{+}\right)} \mathrm{m} / \mathrm{z}$



## 1-butyl-2-phenylaziridine ${ }^{3}$



## 1,2-diphenylaziridine ${ }^{4}$



## 1-(4-(tert-butyl)phenyl)-2-phenylaziridine


$N$-(hex-2-en-1-yl)adamantan-1-amine: ${ }^{\mathbf{1}} \mathbf{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) \delta$ (mixture of $E$ and $Z$ isomers): 6.26-6.34 (m, 0.3 H), 6.12-6.21 (m, 0.7 H), 5.58-5.72 (m, 2 H), $3.46(\mathrm{~d}, J=7.0 \mathrm{~Hz}$, $0.6 \mathrm{H}), 3.30(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 1.4 \mathrm{H}), 2.07-2.14(\mathrm{~m}, 6 \mathrm{H}), 1.97-2.05(\mathrm{~m}, 2 \mathrm{H}), 1.84-1.90(\mathrm{~m}, 3 \mathrm{H})$,
3. For physical and spectroscopic characterization, see: S. Fantauzzi, E. Gallo, A. Caselli, C. Piangiolino, F. Ragaini and S. Cenini, Eur. J. Org. Chem., 2007, 36, 6053.
4. For physical and spectroscopic characterization, see: W. Chamchaang and A. R. Pinhas, J. Org. Chem., 1990, 55, 2943.
1.34-1.49 (m, 8 H ), $0.87-0.93(\mathrm{~m}, 3 \mathrm{H}) .{ }^{13} \mathbf{C} \mathbf{N M R}\left(125 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) \delta(Z$-olefin in parentheses): 137.7(135.2), 122.8(122.5), 57.3, 42.4, 36.7(37.2), 35.8, 34.7, 29.5, 22.4(22.7), 13.9. HRMS $\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ Calc. $234.4017\left[\mathrm{C}_{16} \mathrm{H}_{27} \mathrm{~N}+\mathrm{H}\right]^{+}$, Found $234.4012[\mathrm{M}+\mathrm{H}]^{+}$.

$\boldsymbol{N}$-(octa-2,7-dien-1-yl)adamantan-1-amine: ${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta: 5.72$ (ddt, $J=16.7$, $9.9,6.7 \mathrm{~Hz}, 2 \mathrm{H}), 5.61-5.49(\mathrm{~m}, 1 \mathrm{H}), 5.03-4.90(\mathrm{~m}, 2 \mathrm{H}), 3.26(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 0.3 \mathrm{H}), 3.17(\mathrm{~d}, J$ $=6.8 \mathrm{~Hz}, 1.7 \mathrm{H}), 1.99-1.85(\mathrm{~m}, 7 \mathrm{H}), 1.62(b r . \mathrm{s}, 6 \mathrm{H}), 1.55-1.43(\mathrm{~m}, 6 \mathrm{H}), 1.38(\mathrm{p}, J=7.5 \mathrm{~Hz}, 2$ H). ${ }^{13} \mathbf{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta: 138.9,131.5,130.2,114.8,51.3,43.1,42.7,36.9,33.6,32.2$, 30.0, 29.0. HRMS $\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ Calc. $260.4392\left[\mathrm{C}_{18} \mathrm{H}_{29} \mathrm{~N}+\mathrm{H}\right]^{+}$, Found $260.4388[\mathrm{M}+\mathrm{H}]^{+}$.

$N$-(cyclohex-2-en-1-yl)adamantan-1-amine: ${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta: 6.21$ (d, $J=10.0$ $\mathrm{Hz}, 1 \mathrm{H}), 5.81-5.87(\mathrm{~m}, 1 \mathrm{H}), 3.48-3.56(\mathrm{~m}, 1 \mathrm{H}), 2.32-2.45$ (m, 2 H ), 2.13 (br. s., 6 H ), 1.90 (br. s., 5 H$), 1.79-1.87(\mathrm{~m}, 1 \mathrm{H}), 1.68-1.77(\mathrm{~m}, 1 \mathrm{H}), 1.31-1.56(\mathrm{~m}, 6 \mathrm{H}) .{ }^{13} \mathbf{C}$ NMR ( 125 MHz , $\mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta: 132.1,126.8,59.1,49.2,39.1,36.2,30.0,29.3,24.9,21.1 . \mathrm{HRMS}^{2}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ Calc. $234.4017\left[\mathrm{C}_{15} \mathrm{H}_{23} \mathrm{~N}+\mathrm{H}\right]^{+}$, Found $234.4018[\mathrm{M}+\mathrm{H}]^{+}$.

$\boldsymbol{N}$-(cyclooct-2-en-1-yl)adamantan-1-amine: ${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 5.50-5.58(\mathrm{~m}, 1 \mathrm{H})$, 5.30-5.38 (m, 1 H), 3.79-3.92 (m, 1 H), 2.13-2.24 (m, 1 H), 1.99-2.12 (m, 4 H$), 1.51-1.76(\mathrm{~m}$, $16 \mathrm{H}), 1.30-1.44(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathbf{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 138.3$, 127.1, 51.6, 47.3, 44.1, 39.0, 36.8, 29.9, 29.8, 27.5, 27.3, 24.9, 22.8. HRMS (ESI+) $\mathrm{m} / \mathrm{z}$ Calc. $260.2382\left[\mathrm{C}_{18} \mathrm{H}_{29} \mathrm{NO}+\mathrm{H}\right]^{+}$, Found $260.2373[\mathrm{M}+\mathrm{H}]^{+}$.

$\boldsymbol{N}$-cinnamyladamantan-1-amine: ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta / \mathrm{ppm} 7.23-7.15(\mathrm{~m}, 3 \mathrm{H}), 7.12$ (d, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.48-6.41(\mathrm{~m}, 1 \mathrm{H}), 6.29(\mathrm{~d}, J=13.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.35(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.97$ $(\mathrm{m}, 3 \mathrm{H}), 1.64(\mathrm{~m}, 6 \mathrm{H}), 1.55(\mathrm{~m}, 6 \mathrm{H}) .{ }^{\mathbf{1 3}} \mathbf{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta / \mathrm{ppm} 137.3,128.7,127.6$, 126.6, 52.0, 43.2, 42.5, 36.8, 29.9. HRMS (ESI $\left.{ }^{+}\right) \mathrm{m} / z$ Calc. $268.2065\left[\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{~N}+\mathrm{H}\right]^{+}$, Found $268.2069[\mathrm{M}+\mathrm{H}]^{+}$.

 bottom flask was charged with 2-tert-butylpyrrole ( $7.5 \mathrm{~g}, 60.9 \mathrm{mmol}$ ), 2,6-dichlorobenzaldehyde ( $5.3 \mathrm{~g}, 30.3 \mathrm{mmol}, 0.5$ equiv.) and 200 mL of dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. After stirring until all materials were dissolved, pyridinium $p$-toluenesulfonate ( $1.5 \mathrm{~g}, 5.98 \mathrm{mmol}, 0.1$ equiv.) was added. The reaction was refluxed at $40{ }^{\circ} \mathrm{C}$ for 12 h . The solution was concentrated in vacuo and filtered through a plug of silica gel in a medium porosity frit $(150 \mathrm{~mL})$ with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ to give an orange filtrate. Solvent was removed in vacuo affording 1,9-di(tert-butyl)-5-(2,6-dichloro)benzenedipyrromethane ( $12.2 \mathrm{~g}, 99 \%$ ) as an orange oil. ${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): 8.00 (br. s, 2 H ), $7.28-7.35(\mathrm{~m}, 2 \mathrm{H}), 7.11(\mathrm{~d}, J=0.76 \mathrm{~Hz}, 1 \mathrm{H}), 6.40(\mathrm{~s}, 1 \mathrm{H}), 5.82-5.91(\mathrm{~m}, 4 \mathrm{H}), 1.25(\mathrm{~s}, 18 \mathrm{H})$. The product ( $12.2 \mathrm{~g}, 30.4 \mathrm{mmol}$ ) was dissolved in $300 \mathrm{~mL} \mathrm{CH}_{2} \mathrm{Cl}_{2}$. The oxidant, 2,3-dichloro-5,6-dicyanoquinone (DDQ) ( $7.1 \mathrm{~g}, 31.3 \mathrm{mmol}, 1$ equiv.), was added to immediately give a dark red solution which was stirred overnight. The solution was concentrated in vacuo and residue was dissolved in ethyl acetate ( 350 mL ) and was washed with saturated aqueous sodium bicarbonate, water and brine, dried over magnesium sulfate, filtered and concentrated in vacuo to give a brown solid. The product was loaded onto an alumina plug and eluted with 10:1 hexanes:ethyl acetate to give a red filtrate. Solvent was removed in vacuo and resulting solid was
 ( $500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ): $\delta 13.35$ (br. s, 1 H ), 6.96 (dd, $J=8.01,0.76 \mathrm{~Hz}, 2 \mathrm{H}$ ), 6.55 (m, 1 H ), 6.47 (d, $J$ $=4.20 \mathrm{~Hz}, 2 \mathrm{H}), 6.24(\mathrm{~d}, J=4.20 \mathrm{~Hz}, 2 \mathrm{H}), 1.28(\mathrm{~s}, 18 \mathrm{H}) .{ }^{13} \mathbf{C} \mathbf{~ N M R}\left(125 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right): \delta 167.0$, 139.1, 136.5, 135.9, 132.8, 129.8, 127.6, 126.9, 115.2, 33.6, 29.8. HRMS (ESI ${ }^{+}$) m/z Calc. $401.1546\left[\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{Cl}_{2} \mathrm{~N}_{2}+\mathrm{H}\right]^{+}$, Found. $401.1562[\mathrm{M}+\mathrm{H}]^{+}$.
${ }^{\mathbf{t B u}} \mathbf{L}_{\mathrm{Cl}_{2}} \mathbf{F e C l}\left(\mathbf{O E t}_{2}\right)$ (2): Complex was prepared following literature precedent ${ }^{5}$ from the corresponding ligand ${ }^{\mathbf{E B u}} \mathbf{L}_{\mathbf{C l}_{\mathbf{2}}}(\mathbf{H}) .{ }^{\mathbf{1}} \mathbf{H}$ NMR: $\left(500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right): \delta 59.54$. (s), 21.42 (br. s), 10.65 (br. s), 6.82 (br. s), 1.61 (br. s), -5.68 (br. s). Elemental Anal. Calc. for $\mathrm{C}_{27} \mathrm{H}_{35} \mathrm{Cl}_{3} \mathrm{FeN}_{2} \mathrm{O}$ : C 57.32, H 6.24, N 4.95; Found: C 57.37, H 6.27, N 4.89. Single crystals of 2 were obtained by storing a concentrated diethyl ether solution at room temperature (Figure S-1).


Figure S-1. Solid-state structure for $\mathbf{2}$ with the thermal ellipsoids set at $50 \%$ probability level ( Fe orange, N blue, O red, C gray, H white).
5. E. T. Hennessy and T. A. Betley, Science, 2013, 340, 591.

Table S-2. X-ray crystallographic experimental details for 2.

|  | 2 |
| :---: | :---: |
| Formula | $\mathrm{C}_{27} \mathrm{H}_{35} \mathrm{FeN}_{2} \mathrm{O}$ |
| FW | 565.77 |
| Crystal System | Orthorhombic |
| Space Group (Z) | Pbca (8) |
| a (A) | 15.980(3) |
| b (A) | 14.393(2) |
| c (A) | 24.368(4) |
| $\boldsymbol{\alpha}\left({ }^{\circ}\right)$ | 90.00 |
| $\beta{ }^{\circ}{ }^{\circ}$ | 90.00 |
| $\gamma\left({ }^{\circ}\right)$ | 90.00 |
| Volume ( $\mathrm{A}^{\mathbf{3}}$ ) | 5604.7(16) |
| Calc. $\rho\left(\mathrm{g} / \mathrm{cm}^{3}\right)$ | 1.341 |
| $\mu\left(\mathrm{mm}^{-1}\right)$ | 0.846 |
| Reflections | 5371 |
| Compl. (to 20) | 99.4 |
| GOF on $\mathrm{F}^{2}$ | 0.984 |
| R1, wR2 ${ }^{\text {c }}$ | 0.0594 |
| $[\mathrm{I}>2 \boldsymbol{\sigma}(\mathrm{I})$ ] | 0.1665 |

${ }^{\mathrm{a}} \lambda=0.71073 \AA ;{ }^{\mathrm{b}} \mathrm{T}=100(2) \mathrm{K} ;{ }^{\mathrm{c}} \mathrm{R} 1=\Sigma| | F_{o}\left|-\left|F_{\mathrm{c}}\right|\right| \Sigma\left|F_{o}\right|, \mathrm{wR} 2=\left\{\Sigma\left[w\left(F_{o}{ }^{2}-F_{c}{ }^{2}\right)^{2}\right] / \Sigma\left[w\left(F_{o}{ }^{2}\right)^{2}\right]\right\}^{1 / 2}$

