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IRRADIATION-INDUCED PHASE TRANSFORMATIONS IN ZIRCONIUM ALLOYS*
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L. M. Howe¹, D. Phillips¹, A. T. Motta², and P. R. Okamoto³

¹AECL Research, Reactor Materials Research Branch, Chalk River
Laboratories, Chalk River, Ontario, Canada, K0J 1J0

²Department of Nuclear Engineering, Pennsylvania State University,
University Park, Pa 16802

³Materials Science Division, Argonne National Laboratory, Argonne, IL
60439

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L.M. Howe¹, D. Phillips¹, A.T. Motta² and P.R. Okamoto³

¹Reactor Materials Research Branch, AECL Research, Chalk River Laboratories, Chalk River, Ontario, Canada K0J 1J0

²Department of Nuclear Engineering, The Pennsylvania State University, University Park, PA 16802, USA

³Materials Science Division, Argonne National Laboratory, Argonne, IL 60439, USA

Ion and electron irradiations were used to follow the irradiation-induced crystalline-to-amorphous transformation in Zr_3Fe , $ZrFe_2$, $Zr(Cr,Fe)_2$, and $ZrCr_2$, as well as in $Zr(Cr,Fe)_2$ and $Zr_2(Ni,Fe)$ precipitates in Zircaloy-4.

⁴⁰Ar and ²⁰⁹Bi ion irradiations of Zr_3Fe were performed at 35-725 K using 15-1500 keV ions. The role of the deposited-energy density θ_v in the collision cascade on the nature of the damaged regions in individual cascades was investigated. Amorphization kinetics of Zr_3Fe during in situ electron irradiation were also determined. The electron fluence required for amorphization increased exponentially with temperature and the critical temperature for amorphization was ~ 220 K, compared with 575-625 K for ion irradiation. The difference between the heavy ion and electron irradiation results is attributed to the fact that ion irradiation produces displacement cascades, while electron irradiation produces isolated Frenkel pairs. The dependence of the damage production on the incident electron energy was determined for Zr_3Fe and the results could be analyzed in terms of a composite displacement cross section dominated at high energies by displacements of Zr and Fe atoms, by displacements of Fe atoms at intermediate energies and of secondary displacements of lattice atoms by recoil impurities at low energies.

An investigation was initiated on $ZrFe_2$, $Zr(Cr,Fe)_2$ and $ZrCr_2$ to study the role of variation of stoichiometry and presence of lattice defects on irradiation-induced amorphization. The irradiation-induced amorphization of the intermetallic precipitates $Zr(Cr,Fe)_2$ and $Zr_2(Ni,Fe)$ in Zircaloy-4 was also studied during in situ 350 keV ⁴⁰Ar ion bombardments. The amorphization morphology was shown to be homogeneous. These results are discussed in the context of previous experimental results of neutron and electron irradiations and likely amorphization mechanisms are proposed.

INTRODUCTION

The irradiation behaviour of the various intermetallic compounds of zirconium containing Fe, Cr, Ni, Sn and Si, which are found in the fuel cladding and pressure tubes (e.g. Zircaloy-2, Zircaloy-4 and Zr-2.5Nb alloys) in water-cooled nuclear reactors, is of technological interest. An irradiation-induced crystalline-to-amorphous transformation has been observed for $Zr(Cr,Fe)_2$ and $Zr_2(Ni,Fe)$ precipitates in Zircaloy-2 and Zircaloy-4; see refs. [1-4] for example. Depending upon the precipitate, temperature of irradiation and nature of the bombarding projectile, the transformation could occur with or without a concurrent preferential depletion of Fe from the precipitates. The intermetallic compounds Zr_3Fe [5-8], Zr_2Ni [9,10], $ZrNi$ [10] and $ZrNi_3$ [10] can also be rendered amorphous by irradiation. Irradiation-induced redistribution of Fe has also been observed in neutron-irradiated zirconium alloys [2,11,12] and is associated with accelerated irradiation growth. Consequently, it is of considerable interest to obtain more detailed information on the mass transport of Fe in irradiated Zr alloys, including the irradiation behaviour of various intermetallic compounds (Zr_3Fe , Zr_2Fe , $ZrFe_2$, $ZrCr_2$, $Zr(Cr,Fe)_2$ and $Zr_2(Ni,Fe)$).

In this investigation, results are presented on the role of the deposited-energy density $\bar{\theta}_v$ on the nature of the damaged regions in individual cascades produced by ion bombardment of Zr_3Fe . A detailed comparison is also made between irradiation-induced-amorphization of Zr_3Fe during electron irradiation [8] and under ion bombardments [5-7]. The dependence of the damage production on the incident electron energy in Zr_3Fe has also been determined. Preliminary results are also discussed for the amorphization of $ZrFe_2$, $Zr(Cr,Fe)_2$ and $ZrCr_2$ by electron irradiation. The results of a recent investigation [13] on amorphization of $Zr(Cr,Fe)_2$ and $Zr_2(Ni,Fe)$ precipitates are discussed in the

context of previous experimental results of neutron and electron irradiations and likely amorphization mechanisms are proposed.

EXPERIMENTAL DETAILS

The Zr_3Fe and $ZrFe_2$ materials used in the investigation were obtained by prolonged heat treatment of Zr-20.0, Zr-25.0 and Zr-75.0 at % Fe alloys that had been prepared by arc melting. $ZrCr_2$ and $Zr(Cr,Fe)_2$ were made in a similar fashion. The irradiation behaviour of $Zr(Cr,Fe)_2$ and $Zr_2(Ni,Fe)$ precipitates in Zircaloy-4 was also investigated. Zr_3Fe has an orthorhombic crystal structure, $ZrFe_2$, $Zr(Fe,Cr)_2$ and $ZrCr_2$ have the $MgZn_2$ Laves phase structure and $Zr_2(Ni,Fe)$ has the C16 body-centered tetragonal structure. Discs 0.30 cm in diameter were produced from all of the above alloys by a combination of sparkcutting, sparkplaning, mechanical polishing and electropolishing operations.

The samples were irradiated at 35-725 K with ^{40}Ar or ^{209}Bi ions in either the Chalk River high-voltage mass separator or the Chalk River 70 kV isotope separator. They were surrounded by a cryoshield operating at ~ 20 K, which maintained ultrahigh vacuum conditions around the target during the ion implantation. For studies of the nature of the damage produced in individual cascades (or early stages of cascade overlap) monoenergetic (15-350 keV) ^{209}Bi ion implants were performed at 35-40 K to fluences of 1.0×10^{11} - 7.5×10^{11} ions cm^{-2} . Investigations of the irradiation-induced crystalline-to-amorphous transformation were performed at 35-725 K using 1.5, 1.0 and 0.5 MeV ^{40}Ar or ^{209}Bi ions, successively. Weighted ion fluences were used for the three different energies of the ion beams. The irradiation damage that was produced during the 1.5, 1.0 and 0.5 MeV ion bombardments was distributed over depths of ~ 1 and $0.3 \mu m$ for ^{40}Ar and ^{209}Bi ions, respectively.

Transmission-electron-microscopy (TEM) examinations were performed at 295 K in a Philips CM30 electron microscope operating at 300 kV. Analytical-electron-microscopy (AEM) was also performed from selected areas of the irradiated specimens using the CM30 coupled to a Link X-ray analyzing system.

For obtaining kinetic information on the irradiation-induced amorphization of the intermetallic compounds, in situ irradiations and observations were made using the dual-ion-beam HVEM facility at Argonne National Laboratory (ANL). Irradiations were performed at 23-650 K using either 200-900 keV electrons or 350 keV ^{40}Ar ions.

RESULTS AND DISCUSSION

1. Collision Cascades in Zr_3Fe

It will be useful to discuss the experimental results in terms of the average deposited-energy density $\bar{\theta}_v$ in the cascade [14-18]. $\bar{\theta}_v$ is a measure of the average amount of energy expended in nuclear collision events, that is deposited per atom in the volume occupied by an individual collision cascade.

For monatomic ion implants, $\bar{\theta}_v$ will be defined as follows [14-18]:

$$\bar{\theta}_v = 0.2 \nu(E)/N_v V_R \quad (1)$$

where $\nu(E)$ is the portion of the monatomic ion energy that is lost in nuclear collision events, N_v is the number of lattice atoms contained within a spheroid whose axes are determined by the longitudinal $\langle \Delta X^2 \rangle^{1/2}$ and transverse $\langle Y^2 \rangle^{1/2}$ straggling components of the statistical damage distribution [19,20] and V_R is the volume ratio defining the fraction of the statistical volume filled on average by an

individual cascade. Only $\sim 20\%$ of the total $v(E)$ will be contained within the statistical volume, hence the factor 0.2 in eq. (1). $v(E)$, $\langle \Delta X^2 \rangle^{1/2}$, $\langle Y^2 \rangle^{1/2}$ and N_v were determined using the analytical theory outlined in ref. [19,20]. Based upon Monte Carlo damage simulations, Walker and Thompson [14] determined that $V_R = 0.52$ for ^{209}Bi ions in Ge. The atomic charge Z and atomic weight of Ge are 32 and 72.60, respectively, compared with the average values of Z and A for Zr_3Fe of 36.50 and 82.37. Hence a value for V_R of 0.52 will also be used for ^{209}Bi ions in Zr_3Fe . For 15, 30, 45, 60, 90, 120, 250, 350, 500, 1000 and 1500 keV ^{209}Bi ion implants in Zr_3Fe , the corresponding $\bar{\theta}_v$ values are 4.04, 1.96, 1.29, 0.94, 0.60, 0.42, 0.17, 0.11, 0.07, 0.03 and 0.01, eV/atom, respectively.

Using various operating reflections, the damaged regions produced by ^{209}Bi ion bombardment of Zr_3Fe were imaged under dynamical conditions in the electron microscope, as shown in figure 1a and 1b. Detailed bright and dark-field analysis indicated that the damaged regions were not dislocation loops, stacking-fault tetrahedra or either vacancy or interstitial clusters having an appreciable strain field. Instead, they appeared to be regions in the crystal exhibiting strong structure-factor contrast and consequently probably have a significantly different structure than the crystalline matrix. This would be consistent with the damaged regions being essentially amorphous, as also indicated by TEM investigations on ion bombarded Si, [15-18, 21-24], Ge [17,25,26] and NiAl₃ [27]. Damaged regions having similar features have also been seen in Zr_3Fe bombarded with 0.5-1.5 MeV ^{40}Ar ions [6], including in situ bombardments and observations using 350 keV ^{40}Ar ions, [28].

At high deposited-energy densities ($\bar{\theta}_v > \sim 2.0$ eV/atom), which corresponds to fairly low Bi ion energies (15-30 keV), the visible damage produced in a single cascade consisted essentially of a single isolated damaged region (figure 1a). With decreasing values of $\bar{\theta}_v$ (i.e. increasing ion implant energies), there was an increasing tendency for multiple damaged regions to form within a main

cascade (figure 1b). Average values of the diameter of the damaged regions present in the electron micrographs were obtained and these were correlated with a particular cascade, as determined from the longitudinal $\langle \Delta X^2 \rangle^{1/2}$ and transverse $\langle Y^2 \rangle^{1/2}$ straggling components of the statistical volume. The initial results were given in ref. [7] and even more detailed measurements have now been performed. These measurements were performed on Zr_3Fe bombarded with 15-350 keV ^{209}Bi ions and at fairly low ion fluences ($1.0 \times 10^{11} - 2.5 \times 10^{11}$ ions cm^{-2}) in order to avoid appreciable cascade overlap.

The number of multiple damaged regions (subcascades) formed within an individual collision cascade and the average total volume occupied by the subcascades within a cascade were determined from the detailed measurements performed on the visible damaged regions. As shown in figure 2, the average number n_{sc} of multiple damaged regions increased steadily from 1.1 at 15 keV to 3.1 at 350 keV. This is in good agreement with the results obtained for Si [18] and Ge [17] bombarded with various ions. Also shown in figure 2 is the fraction F of the theoretical volume (determined from the analytical theory outlined in ref. [19,20], as discussed previously) that was occupied on average by the visible multiple damaged regions comprising a cascade. The results show that F decreases quite rapidly with increasing ion energy (decreasing $\bar{\theta}_v$ values). Similar behaviour was found in ion implanted Ge and Si [17,18]. The Ge [17] results are also plotted in figure 2 for comparison with the Zr_3Fe results. For 10, 15, 30, 45, 60, 90 and 120 keV ^{209}Bi ion implants in Ge, the corresponding $\bar{\theta}_v$ values are 4.10, 2.75, 1.28, 0.83, 0.61, 0.38 and 0.27 eV/atom, respectively.

The above results for Zr_3Fe , as well as those for Si and Ge, [17,18] emphasize the important distinction between very high deposited-energy density cascades ($\bar{\theta}_v > \sim 2$ eV/atom) and those of considerably lower deposited-energy density ($\bar{\theta}_v < \sim 0.5$ eV/atom). In the former case, appreciable fractions of the theoretical collision cascade volume (as defined above) have been rendered

amorphous. This behaviour is certainly consistent with the operation of an energy spike [15-18] for high deposited-energy density cascades as it would explain why amorphous regions could extend over the whole of the theoretical collision cascade volume and even beyond it. In Ge, for example, $F = 0.93$ and 1.18 for 15 keV Bi ($\bar{\theta}_v = 2.75$ eV/atom) and 10 keV Bi ($\bar{\theta}_v = 4.10$ eV/atom) implants, respectively [17]; see figure 2. In the lower deposited-energy density cascades, there are localized regions (i.e. subcascades) that are essentially amorphous but these are obviously surrounded by areas (which are still within the single main cascade) having a high density of point defects and small defect clusters.

2. Irradiation-Induced Crystalline-to-Amorphous Transformation in Zr_3Fe

Investigations have revealed that Zr_3Fe can be completely amorphized during irradiation with ^{40}Ar and ^{209}Bi ions [5-7] and electrons [8]. An example of this behaviour is shown in figure 1c and 1d for ^{209}Bi ion bombardments. The criteria used for determining that an amorphous phase had been formed were the interruption of all bend contours in the TEM images of the irradiated areas and the disappearance of all crystalline spots from the diffraction patterns. A detailed study of the temperature dependence of the dose-to-amorphization has now been completed for 0.5 - 1.5 MeV ^{40}Ar ion irradiations and the results are shown in figure 3. It can be seen that in the region 160 - 220 K, there is a step in the dose-to-amorphization curve, followed by a plateau region extending up to ~ 450 K. Preliminary results indicate that the magnitude of this step is less for 0.5 - 1.5 MeV ^{209}Bi ion irradiations than for the ^{40}Ar ion irradiations. This would suggest that the magnitude decreases with increasing $\bar{\theta}_v$ values, which is consistent with the details of the damaged regions depending upon $\bar{\theta}_v$, as discussed above. In the temperature range 160 - 280 K, there was a pronounced dose-rate (flux) effect [6]. The fraction of the irradiated volume that had been rendered amorphous at a given ion

fluence increased noticeably as the ion flux was increased from 0.9×10^{11} to 14.7×10^{11} ions $\text{cm}^{-2}\text{s}^{-1}$. For example, irradiation of Zr_3Fe at 180 K with ^{40}Ar ions (1.5, 1.0 and 0.5 MeV) to the same total ion fluence of 1.0×10^{15} ions cm^{-2} resulted in the production of amorphous regions throughout a heavily-damaged crystalline matrix at an ion flux of 0.9×10^{11} ions $\text{cm}^{-2}\text{s}^{-1}$, strained crystalline regions in an amorphous matrix at an ion flux of 2.9×10^{11} ions $\text{cm}^{-2}\text{s}^{-1}$, and complete amorphization at an ion flux of 14.7×10^{11} ions $\text{cm}^{-2}\text{s}^{-1}$.

As shown in figure 3, the critical temperature for amorphization of Zr_3Fe by ^{40}Ar ion irradiation was 570-600 K (i.e. Zr_3Fe could not be rendered amorphous during irradiation above this temperature), and preliminary investigations indicate that it is slightly higher (~ 625 K) for ^{209}Bi ion irradiation. A critical temperature for amorphization (i.e., the temperature above which amorphization can no longer be achieved) of 570-625 K for the ion irradiations is consistent with the observation that the discrete damaged regions produced during ^{40}Ar ion irradiations anneal out at ~ 400 -500 K [6]. Also, in Zr_3Fe that had been rendered completely amorphous during ion bombardment, small crystals were observed to form in the amorphous matrix during 0.5 h anneals at 500 - 600 K and extensive growth of crystalline regions was observed during a 3 h anneal at 623 K [5]. Neither the irradiation-induced crystalline-to-amorphous transformation nor its reversal during post irradiation annealing produced any change in the chemical composition of the matrix, i.e. it remained Zr_3Fe [5].

Irradiations of Zr_3Fe with 0.9 MeV electrons were performed at 28-220 K in the high-voltage electron microscope at ANL. By measuring the onset, spread and final size of the amorphous region, factoring in the Gaussian distribution of the beam, a kinetic description of amorphization in terms of dose, dose rate and temperature was obtained [8]. The dose required to reach a given radius for the amorphous region (i.e. the dose-to-amorphization for a given dose rate) is shown as a function of irradiation

temperature in figure 3. The critical temperature for amorphization by electron irradiation was ~ 220 K, compared with 570-600 K for ^{40}Ar ion irradiation, as shown in figure 3. The dose-to-amorphization increased exponentially with temperature. Results also indicated that the rate of growth of the amorphous region under the electron beam decreased with increasing temperature and the dose-to-amorphization decreased with increasing dose rate.

The results of the ion and electron irradiations can be understood in the framework of damage accumulation under irradiation. Damage produced by irradiation is opposed by thermal annealing. The balance between the rate of damage production and the ability of the material to repair itself dictates the response of the material to irradiation. At low temperatures, damage produced by irradiation remains "frozen in", with little annealing taking place. In this temperature regime, the dose-to-amorphization is almost independent of temperature: the damage accumulates in the lattice unchecked by thermal annealing. At a certain level of irradiation damage in Zr_3Fe , there appears to be a driving force to exchange the imperfect form of long-range order resulting from irradiation for a local configuration of short-range order, where the bonding restrictions of chemical species and directionality are followed more closely. The difference between the ion and electron irradiations is due to the fact that ion irradiation produces displacement cascades, while electron irradiation produces isolated Frenkel pairs. Within the cascades, the density of damage is quite high and amorphization can then occur either directly in the cascade or upon cascade overlap. As the irradiation temperature increases, annealing processes are activated that offset the production of damage. When annealing reduces the effective damage rate without overwhelming it, a "step" appears in the dose-to-amorphization versus temperature curve, increasing the dose-to-amorphization at that temperature. The temperature at which the activated annealing process completely overwhelms damage production is the critical temperature, and amorphization is no longer possible above that temperature. It follows

then that a lower damage rate is offset at a lower temperature than a higher damage rate. This implies different critical temperatures for different damage rates.

The increase in dose-to-amorphization between 150 and 200 K under electron irradiation corresponds well to the change in amorphization regime occurring between 110 and 160 K under ^{40}Ar ion irradiation. At 110 K and below, the dose-to-amorphization during ^{40}Ar ion irradiation is smaller than at 160 K. Below 160 K, the regions in between cascades do not get annealed before the next cascade hits. At 160 K and above, some type of defect becomes mobile. The resulting defect annihilation produces a less damaged outer region of the cascade, leaving a smaller heavily damaged core, thus requiring more extensive coverage of the material for complete amorphization.

Above ~ 400 K, different defects become mobile in Zr_3Fe than in the 160-220 K range. Eventually, there is sufficient annealing occurring in the damage cascades that amorphization can no longer occur even under ion irradiations. For ^{40}Ar ion irradiation of Zr_3Fe this stage is reached at 570-600 K, and for ^{209}Bi ion irradiations of Zr_3Fe it is slightly higher (~ 625 K). From investigations of amorphization in CuTi , Koike et al. [29] also concluded that for similar fluxes (dose rates) of bombarding projectiles, the critical temperature for amorphization increased with the mass of the incident projectile (i.e. with increasing average deposited-energy density in the collision cascade).

A study was also undertaken to determine the energy dependence of the dose-to-amorphization during electron irradiation of Zr_3Fe at 23-30 K. Zr_3Fe samples were irradiated in the HVEM at ANL at energies from 200-900 keV. Amorphization occurred at energies from 900 down to 250 keV. Three distinct regions were observed, as shown in figure 4. Between 900 and 700 keV, amorphization occurred at a constant low dose of $5 \times 10^{21} \text{ e cm}^{-2}$; a higher plateau at $10^{22} \text{ e cm}^{-2}$ was observed

between 600 and 400 keV, and finally there was a sharp increase in the dose-to-amorphization below 400 keV. At 250 keV the dose required to amorphize Zr_3Fe was an order of magnitude higher than at 900 keV.

A dependence of the dose-to-amorphization on the orientation of the sample relative to the electron beam was observed at low electron energies. This is apparent in figure 5a, where the beam was centered on a triple junction of three Zr_3Fe grains, and irradiated at 250 keV. The three grains were identified by both compositional and diffraction analyses as being orthorhombic Zr_3Fe . One grain has turned completely amorphous while the other grains show no signs whatsoever of amorphization (as evidenced by diffraction analysis) nor of any form of damage upon post-irradiation examination. In the data shown in figure 4, the experimental point for the 250 keV irradiation was labelled "restrictive amorphization" because of the above behaviour. The dependence of amorphization on sample orientation can also be seen in figure 5b from an irradiation performed at 300 keV: the amorphous spot does not exactly follow the shape of the beam, but contains "wisps" of amorphous material (shown arrowed) in directions that follow the bend contours of the thin sample. Such orientation dependence disappears above 400 keV. An orientation dependence of the dose-to-amorphization under electron irradiation has been observed by Mori et. al. in Zr_3Al [30], but at higher electron energies. This orientation dependence can be qualitatively explained if the damage-producing process operates preferentially along particular lattice directions, that is, if the displacement energy is orientation dependent.

The results of the energy dependence experiments can be analyzed in terms of a composite displacement cross section dominated at high energies (700-900 keV) by displacements of Zr and Fe atoms, by displacement of Fe atoms at intermediate energies (400-600 keV) and of secondary

displacements of lattice atoms by recoil impurities at low energies (below 400 keV). One of the most predominant recoil impurities will be oxygen. Preliminary measurements using an elastic-recoil detection technique indicate oxygen levels of 1-3 at% in the bulk of the Zr_3Fe alloy, hence a value of 1 at% was used in the analysis discussed below. If the same amorphization mechanism is operative at all electron energies, then the total dose D (in dpa) required to produce amorphization in Zr_3Fe (considering both primary and secondary displacement mechanisms) should be independent of the electron energy. The displacement energies of the Zr and Fe atoms were adjusted to give the smallest variation in D from 250 to 950 keV. The set that gave the best results was $E_d^{Zr} = 24$ eV and $E_d^{Fe} = 14$ eV. In this analysis, which should be considered as being preliminary, the orientation dependence of the displacement and scattering cross sections was neglected, O was considered as the major impurity, secondary displacements in the Fe sublattice of Zr_3Fe were considered but those in the Zr sublattice were neglected, and the displacement cross sections used were those given by Owen [31]. A more detailed discussion of the energy dependence results will be given elsewhere [32]. The dpa values given in figure 3 were obtained using a value of $E_d = 25$ eV for Zr_3Fe for the conversion from electron and ion fluences to dpa.

3. Irradiation-Induced Amorphization of $ZrFe_2$, $ZrCr_2$, $Zr(Cr,Fe)_2$ and $Zr_2(Ni,Fe)$

Investigations have been initiated on various Zr-Fe-Cr intermetallic compounds to obtain information on the details of the damage structure in individual cascades, the critical temperature for amorphization by ions and electrons and the role of lattice defects, such as dislocations and stacking faults, and variation in stoichiometry in affecting the dose required to produce amorphization. For example, the critical temperature for amorphization of $ZrCr_2$, $Zr(Cr,Fe)_2$ and $ZrFe_2$ by 0.9 MeV electron irradiation was found to be 150-200 K, ~ 225 K and ~ 100 K, respectively. Since all of

these compounds have the same Laves phase structure, stoichiometry of the compound may play an important role. Also, in $ZrFe_2$, small grains that are heavily faulted amorphize more readily than larger grains that contain relatively few stacking faults. There is prior evidence from studies on other intermetallic compounds that preferential amorphization can occur at grain boundaries [33], dislocations [34], free surfaces [35] and anti-phase boundaries [36]. These are high energy regions and the local distortion of the lattice or high chemical energy gives an additional contribution to the free energy rise due to irradiation, thus enhancing the crystalline-to-amorphous transformation.

Studies were performed on thin foils of Zircaloy-4 irradiated at 300-650 K with 350 keV ^{40}Ar ions in the dual-ion-beam HVEM facility at ANL. The irradiation-induced amorphization of the intermetallic precipitates $Zr(Cr,Fe)_2$ and $Zr_2(Ni,Fe)$ was studied in situ. Details of this investigation are given elsewhere [13]. Here the results will be discussed in the context of previous experimental results of neutron and electron irradiations.

The basic features of irradiation-induced amorphization of $Zr(Cr,Fe)_2$ and $Zr_2(Ni,Fe)$ precipitates in Zircaloy-4 may be summarized as follows:

- (a) Under electron irradiation, both types of precipitate amorphize if the irradiation temperature is lower than 300 K. Amorphization happens homogeneously and is thought to occur as a result of increased chemical disordering and a higher concentration of the less mobile point defect caused by the loss of the more mobile defect to the free surface [37].
- (b) Under low temperature neutron irradiation, amorphization occurs for both types of precipitates, through some combination of cascade chemical disordering and bulk point defect

concentration increase. Under high temperature neutron irradiation, the $Zr_2(Ni,Fe)$ precipitates remain crystalline and the $Zr(Cr,Fe)_2$ precipitates amorphize, while developing a duplex structure with amorphization starting at the precipitate-matrix interface and proceeding inwards [1-4]. It has been proposed [38] that in the latter case, amorphization occurs by a combination of less effective cascade disordering and a departure from exact stoichiometry at the precipitate-matrix interface caused by ballistic mixing.

- (c) Under ^{40}Ar ion irradiation [13], the $Zr(Cr,Fe)_2$ precipitates amorphize at temperatures up to 650 K while the $Zr(Ni,Fe)_2$ precipitates do so at least up to 600 K. Amorphization is homogeneous, and without associated changes in chemical composition. The amorphization mechanism is thought to be chemical disordering and possibly defect clustering induced by collision cascades. There is no need for a supplement from other mechanisms such as a departure from stoichiometry because, due to its higher damage rate, the damage produced by ion irradiation is more effective than that produced by neutron irradiation in causing amorphization.

CONCLUSIONS

Ion and electron irradiations have been used to obtain information on irradiation-induced crystalline-to-amorphous transformations in zirconium intermetallic compounds. In Zr_3Fe , investigations of the role of the deposited-energy density $\bar{\theta}_v$ in the collision cascade revealed that with decreasing $\bar{\theta}_v$ there was an increasing tendency for multiple damaged regions (subcascades) to form within a main cascade and the fraction of the theoretical collision cascade volume that was occupied by the visible damaged regions decreased rapidly. The critical temperature for amorphization of Zr_3Fe by electron

irradiation was ~ 220 K, compared with 570-600 K, for ^{40}Ar ion irradiation and this difference was attributed to the fact that ion irradiation produces displacement cascades, while electron irradiation produces isolated Frenkel pairs. The dependence of the damage production (dose-to-amorphization) on the incident electron energy was determined for Zr_3Fe and the results could be analyzed in terms of a composite displacement cross section dominated at high energies (700-900 keV) by displacements of Zr and Fe atoms, by displacements of Fe atoms at intermediate energies (400-600 keV) and of secondary displacements of lattice atoms by recoil impurities at low energies (below 400 keV). At low electron energies (250-350 keV), the dose-to-amorphization (in electrons cm^{-2}) of Zr_3Fe was dependent on the orientation of the sample relative to the electron beam. Studies on ZrFe_2 , $\text{Zr}(\text{Cr,Fe})_2$ and ZrCr_2 indicated that variations in stoichiometry and the presence of lattice defects may have an appreciable effect on the irradiation-induced amorphization process. Mechanisms were also proposed for the amorphization of $\text{Zr}(\text{Cr,Fe})_2$ and $\text{Zr}_2(\text{Ni,Fe})$ precipitates in Zircaloy-4 during electron, ion and neutron irradiation.

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FIGURE CAPTIONS

- Figure 1 Bright-field electron micrographs (a-c) and diffraction pattern (d) for Zr_3Fe bombarded with ^{209}Bi ions at the following energies and ion fluences (a) 30 keV; 2.5×10^{11} ions cm^{-2} , (b) 120 keV; 2.5×10^{11} ions cm^{-2} , (c) 0.5 - 1.5 MeV; 5.0×10^{13} ions cm^{-2} and (d) 0.5 - 1.5 MeV; 5.0×10^{13} ions cm^{-2} .
- Figure 2 The collision cascade parameters F and n_{ic} as a function of incident ion energy for ^{209}Bi ion irradiations of Zr_3Fe and Ge.
- Figure 3 Dose-to-amorphization, in displacements per atom (dpa), at various irradiation temperatures for 0.9 MeV electron and 0.5 - 1.5 MeV ^{40}Ar ion irradiations of Zr_3Fe .
- Figure 4 Dose-to-amorphization for Zr_3Fe at 25-30 K as a function of the incident electron energy.
- Figure 5 (a) Bright-field electron micrographs and associated diffraction patterns showing the strong orientation dependence of the damage production for Zr_3Fe irradiated with 250 keV electrons at 23-30 K. The irradiation was performed at a triple junction and only one of the Zr_3Fe grains at this junction has been rendered amorphous by the irradiation. The electron fluence was 6.8×10^{22} e cm^{-2} .
- (b) Bright-field electron micrograph showing indications of orientation dependence of damage production for Zr_3Fe irradiated at 23-30 K with 300 keV electrons to a fluence of 5.2×10^{22} e cm^{-2} .

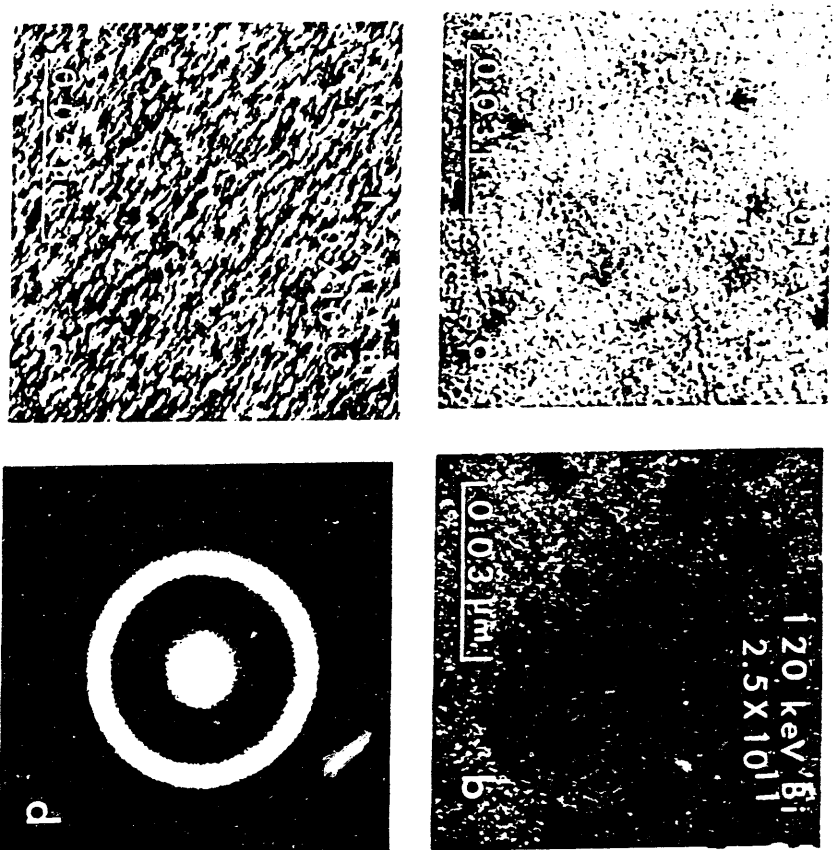


Figure 1

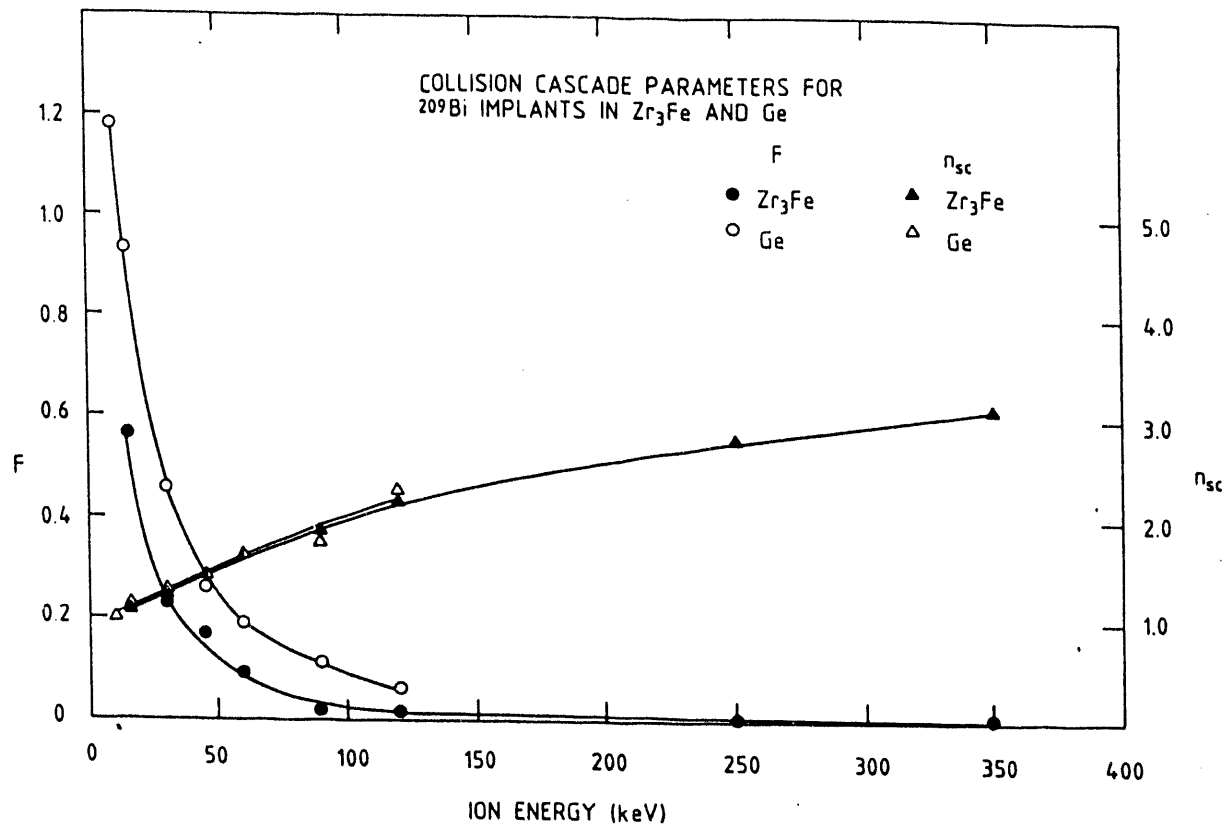


Figure 2

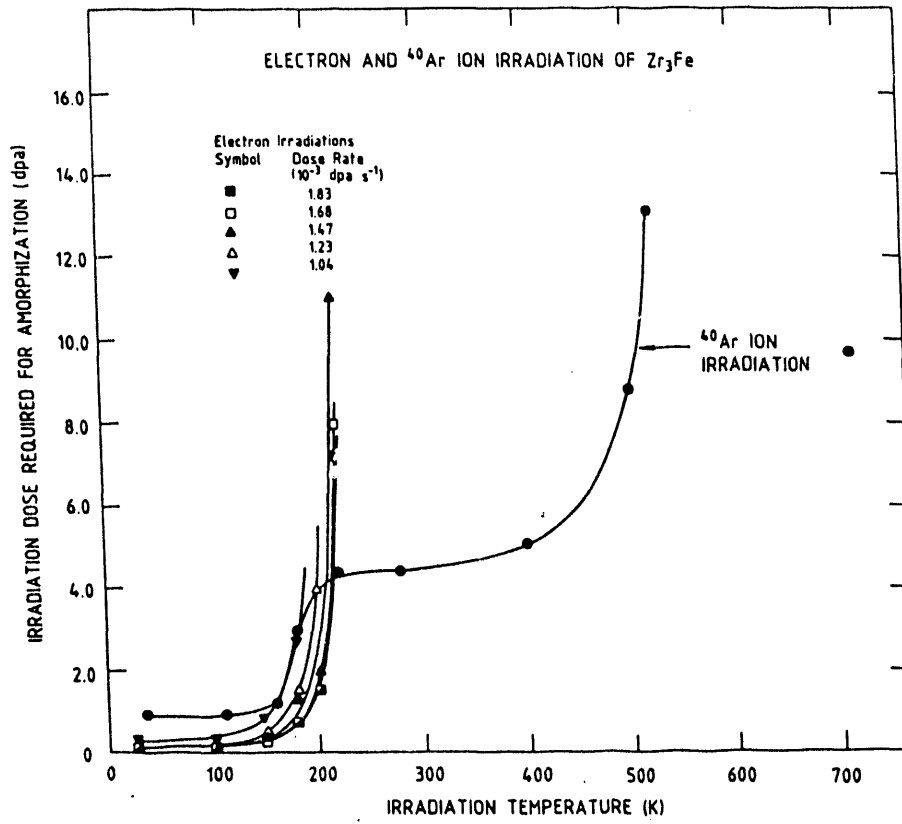


Figure 3

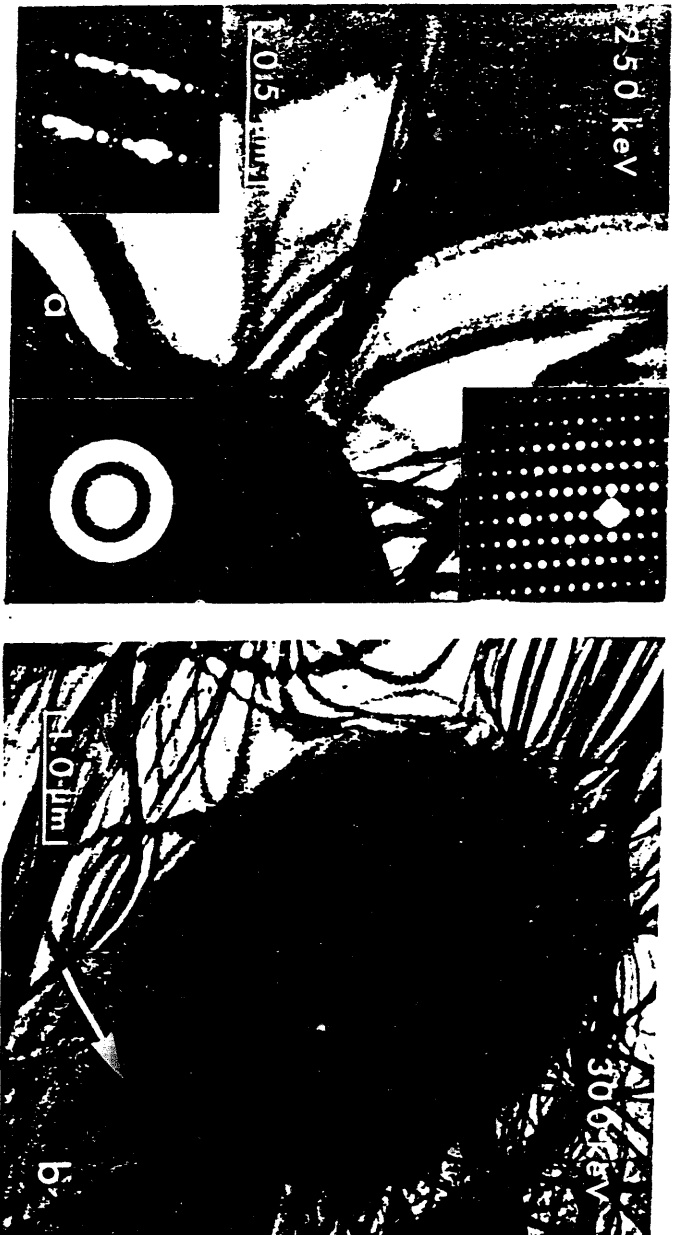


Figure 5

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