Research Article

Kenaf Bast Fibers—Part II: Inorganic Nanoparticle Impregnation for Polymer Composites

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The objective of this study was to investigate an inorganic nanoparticle impregnation (INI) technique to improve the compatibility between kenaf bast fibers and polyolefin matrices. The Scanning Electron Microscopy (SEM) was used to examine the surface morphology of the INI-treated fibers showing that the CaCO₃ nanoparticle crystals grew onto the fiber surface. Energy-dispersive X-ray spectroscopy (EDS) was used to verify the CaCO₃ nanoparticle deposits on the fiber surface. The tension tests of the individual fiber were conducted, and the results showed that the tensile strength of the fibers increased significantly (more than 20%) after the INI treatments. Polymer composites were fabricated using the INI-treated fiber as reinforcement and polypropylene (PP) as the matrix. The results showed that the INI treatments improved the compatibility between kenaf fibers and PP matrix. The tensile strength of the composites reinforced with INI-treated fibers increased by 25.9% and 10.4%, respectively, compared to those reinforced with untreated kenaf fibers.

1. Introduction

Lignocellulosic fiber is a renewable and biodegradable natural polymer which has been used in a variety of applications, such as textile, pulp and paper, and so forth. In recent years, there has been an increasing interest in utilizing natural fibers to replace synthetic glass or carbon fibers to fabricate sheet molding compound (SMC) composites for automobile structural component design. Natural fibers are light weight, economical and environmentally friendly. However, in the processing of lignocellulosic fiber-reinforced polymer composites, the following issues should be addressed. The cell wall structure of natural fibers contains many micropores. If a chemical pulping process is used, additional micropores would be created since some of the lignin and hemicellulose of natural fibers are removed [1]. The presence of these micropores in the cell wall structure could cause manufacturing defects in composites, such as interfacial failure and air pockets. Compatibility between the fiber surface and polymer matrix has been a major issue for lignocellulosic fiber-reinforced polymer composites processing [2–5]. As it was reported in the first paper of this series [6] that the kenaf fibers retted using an alkaline solution by a hermetical process had poor compatibility with the polypropylene (PP) matrix resulting in delaminations of the composites. A common way to improve the compatibility between the cellulosic natural fibers and the polymer matrix is to use coupling agents [7]. However, the organic coupling agents are usually costly and also cause environmental concern. Previous study indicated that the deposited nanoparticles on the fiber surface served as nucleation sites to initiate the crystalline orientation of the molten polymer matrix [8]. Therefore, a proper process to introduce nanoparticles onto the fiber surface serving as attraction force manipulators to polyolefin matrixes has a potential to improve the crystalline formation in the polymer matrix. Since the mechanical behavior of a composite material strongly depends on the adhesion between the reinforced fibers and the matrix, the properties of the composites should be enhanced [9].

Directly impregnating the commercial nanoparticles into the micropore cell wall structure of the cellulosic nature fibers can be difficult and costly. For the inorganic nanoparticle impregnation (INI) process in this study, the primary salt, Na₂CO₃, and secondary ionic salt, CaCl₂, were impregnated into micropores of the fiber cell wall consecutively. The two chemicals react at certain temperature and pressure conditions, and the inorganic nanoparticles of CaCO₃ are formed. The chemical reaction is described as follows:

$$CaCl_2 \cdot H_2O + Na_2CO_3 \xrightarrow{H_2O} CaCO_3 + 2NaCl + 2H_2O.$$
 (1)

The INI technology has been used in pulp, paper, and surface coating industries [10]. The main purpose of INI in pulp and paper was to improve the printability of the papers. The inorganic nanoparticles loaded into cell wall and on the surface may provide strong static electric attractive forces to nonpolar polymer surface [11], and hence the compatibility between the fibers and the polymer matrix may be improved. These nanoparticle-impregnated fibers can be incorporated readily into the existing manufacturing process to make fiber sheets for composite products. The impregnation of the inorganic nanoparticles into the micropore structure of the fiber cell walls can reduce the microvoid volumes in the fiber, as well as reduce air bubble formation during the composite fabrication process. This process is low cost (using inexpensive ionic salts), has little environmental concern (weak chemical solutions are used and can be reused), and simple (the inorganic nanoparticles form directly). The products can be potentially used as vehicle components because of their light-weight, good mechanical properties and environmental friendliness.

In this study, the kenaf bast fibers retted from the hermetical alkaline process were further treated with INI processes. The objectives of this study were to optimize the INI process for the kenaf bast fibers and to evaluate the potentials on the property improvement for the kenaf fiber-PP composites.

2. Materials and Methods

2.1. Materials. The Kenaf stalks were obtained from MSU North Farm. After the separation of the kenaf core and bast, the bast was cut into 50.8 mm lengths and dried to a moisture content of 7.4% at 103°C. Sodium hydroxide (NaOH) solution (5%, w/v) prepared with NaOH beads (Lab grade, Thermo Fisher Scientific Inc.) and distilled water was used as digestion agent. Glacial acetic acid (17.4 N, Regent grade, Thermo Fisher Scientific Inc.) was used as a pH neutralizer. Sodium carbonate (Na₂CO₃) aqueous solutions (0.1 mol/L) and calcium chloride (CaCl₂) aqueous solutions (0.1 mol/L and 0.2 mol/L) were prepared, respectively, using distilled water. The sodium carbonate and calcium chloride were supplied by Fisher Scientific Inc. Polypropylene (PP) films (CO-EX Oriented Polypropylene), provided by Plastic Suppliers, Inc. Dallas, Tex, USA, and were used to fabricate kenaf fiber/PP composites.

2.2. INI Treatments. The flowchart on the treatment details is described in Figure 1 [5, 12, 13]. Kenaf bast was retted with a 5% NaOH solution (fiber: NaOH solution = 1:30, g/mL) in a hermetical reactor (Parr Instrument Co. 251 M) at 160°C. The retting process took one hour and was aided with a mechanical stirring. The autogenous vapor pressure was

0.60 MPa. After the retting process, the pH of the retting liquid and retted fibers was adjusted to 7.0 using acetic acid. The neutralized fibers were washed with water to remove chemicals from the fibers. These fibers were used as the control fibers. The control fibers and 0.1 mol/L Na₂CO₃ water solution (fiber: solution = 5:400, g/mL) were mixed in the hermetic reactor with mechanical stirring at 70°C with autogenous vapor pressure of 0.1 MPa for 30 minutes. The excessive primary ionic solution was removed from kenaf fibers by gravity. The secondary ionic solution was used to impregnate the fibers at three different temperatures and autogenous vapor pressures (100°C, 0.15 MPa; 130°C, 0.30 MPa, and 160°C, 0.70 MPa) for 15 minutes. The controlled molar ratios of Na₂CO₃ to CaCl₂ were 1:1 and 1:2 by the concentration of CaCl₂ solution, 0.1 mol/L or 0.2 mol/L. Three temperatures and the corresponding pressures provided a total of six INI treatment conditions. After the impregnation of the two ionic solutions, the primary ionic salt (Na₂CO₃) reacted with the secondary ionic salt (CaCl₂) in kenaf bast fibers to generate CaCO₃ nanoparticles in the micropore structure of the fiber cell wall, from which the nanoparticle crystals may grow onto the fiber surface. The impregnated fibers were washed to remove excess CaCO₃ particles and other ions on fiber surface.

2.3. Determination of $CaCO_3$ Loading. The percentage $CaCO_3$ loading in the fibers was calculated based on the difference in ash contents between the untreated fibers and INI-treated fibers. The ash content was determined by burning the material in a muffle furnace first at 400°C for 30 minutes, then at 850°C for 45 minutes.

2.4. Characterization of INI-Treated Fibers. Surface morphology, tensile properties of individual fibers, surface hardness, and elastic modulus of the fibers were examined in the same procedures described in the first paper of this series [6]. Elements determination was conducted using a Bruker Quantax 200 X Flash Energy-Dispersive X-ray Spectrometer (EDS) System (LN2-free high speed 30 mm² SDD Detector).

2.5. Composites Fabrication. The composites were fabricated with a sheet molding compound process. The control fibers and INI-treated kenaf fibers were dispersed in water by vigorous mechanical stirring. The fiber suspension was poured into a 355 mm \times 355 mm deckle box and then passed through a screen (mesh 35), on which the fiber sheets were formed as the water flowed down gravitationally. The fiber sheets were dried in an oven set at 80°C. The fiber sheets and PP films were cut into a dimension of 15.2 cm imes15.2 cm and laminated alternatively. The fiber to PP weight ratio in the composite panel was 50:50. The laminated mats were pressed at 200°C and 0.7 MPa for 2.5 minutes. The pressure was not released until the platen was cooled to room temperature. The kenaf fiber/PP panels were removed from the press and stored in a desiccator with the silica gel for two days before preparing the mechanical testing specimens. Three panels were fabricated for each formula. The density of all of the kenaf fiber/PP composites (including the control



FIGURE 1: Flowchart of INI treatment.

fibers reinforced and INI-treated fibers-reinforced composites) was 0.88 ± 0.03 g/cm³.

2.6. Composites Tensile Properties Testing. Tensile properties of the INI-treated fiber/PP composites and control fiber (untreated fiber)/PP composites were tested with Instron 5869 (load cell 50 kN). The crosshead extensions were used as the specimen deformations. Composites samples were kept in desiccators for one week before tensile testing. The procedures referred to ASTM 1037. The crosshead speed during the tension testing was 2.5 mm/min. Nine replicates of each composite formulation were tested. Multiple comparison of the results was conducted with Fisher's Least Square method at $\alpha = 0.05$ using SAS 9.2 software (SAS Institute Inc. NC, USA). The fracture surfaces of the samples were observed using scanning electron microscopy (SEM, Zeiss Supra TM 40).

3. Results and Discussion

3.1. CaCO3 Loading. Table 1 shows the loading percentage of CaCO₃ in the fibers. Each specimen was examined twice, and the errors were less than 0.1%. The fibers treated at 130°C (Na₂CO₃ to CaCl₂ = 1 : 1, mol : mol) had the highest CaCO₃ loading. Although the differences in the loading percentages of CaCO₃ among the impregnation variables were no more than 1%, a small increase in CaCO₃ nanoparticle loading could result in a significant change in surface characteristics of the fibers since the particles impregnated in were in nano scales and with a high specific surface area. A small amount of CaCO₃ nanoparticles may give a large surface area, which may have a significant impact on the interfacial compatibility between the fiber and the polymer matrix.

3.2. Surface Morphology and Element Determination. Figure 2 shows the SEM images of the fiber surfaces treated with INI. As the temperature increased, the inorganic nanoparticle size increased, indicating that temperature played an important role in the formation of CaCO₃ crystals. The sizes of the CaCO₃ nanoparticles in the fibers treated at 100°C and 130°C were smaller than 90 nm. However, some CaCO₃ particles generated at 160°C (Na₂CO₃ : CaCl₂ = 1 : 1, mol : mol) grew to a diameter over nanoscale.

INI treatment conditions				
$Na_2CO_3: CaCl_2$ (mol:mol)	CaCO ₃ loading (%)			
1:1	1.91			
1:2	2.30			
1:1	2.86			
1:2	2.57			
1:1	2.43			
1:2	2.21			
	$\begin{array}{c} \text{nt conditions} \\ \text{Na}_2\text{CO}_3:\text{CaCl}_2 \\ (\text{mol}:\text{mol}) \\ \hline 1:1 \\ 1:2 \\ 1:1 \\ 1:2 \\ 1:1 \\ 1:2 \\ 1:1 \\ 1:2 \end{array}$			

The EDS spectrum and elements maps are shown in Figures 3 and 4. In the spectrum, calcium (Ca), carbon (C), and oxygen (O) were detected indicating that the CaCO₃ nanoparticles exist in the fibers. Calcium may exist as calcium ion (Ca^{2+}) with Cl⁻ ion or in CaCO₃ molecule. Since, neither sodium (Na) nor chlorine (Cl) was detected from the EDX. Therefore, Calcium can only exist in CaCO₃. This evidence testified that the reaction between Na₂CO₃ and CaCl₂ occurred during the INI process and the CaCO₃ was synthesized. The byproduct, NaCl, would have been washed off after the INI treatment. The peaks for gold (Au) and palladium (Pd) came from the specimens coating treatment for the SEM and EDS.

3.3. Surface Hardness and Elastic Modulus. Table 2 shows the surface hardness and elastic modulus of the INI-treated fibers.

The fibers treated at 130° C (0.30 MPa) yielded a 52.6– 76.9% increase in modulus and a 47.0–59.1% increase in hardness. The improvement of the hardness and modulus properties was found for the other INI treated fibers also compared with the untreated fiber. The micropores in the cell wall of the lignocellulosic fibers were filled up by the CaCO₃ particles, which enhanced the stress transfer between cellulose fibrils resulting a higher hardness and modulus. A higher variation in the hardness and modulus properties was found in Table 2. This may be because that some indentation tests are on the CaCO₃, while others on the fibers. In addition, the component variation at different locations of

TABLE 1: CaCO₃ loading percentages of the INI-treated fibers.



FIGURE 2: SEM images of INI-treated fibers. (a) 100° C (Na₂CO₃: CaCl₂ = 1 : 1, mol : mol), (b) 100° C (Na₂CO₃ : CaCl₂ = 1 : 2, mol : mol), (c) 130° C (Na₂CO₃ : CaCl₂ = 1 : 1, mol : mol), (d) 130° C (Na₂CO₃ : CaCl₂ = 1 : 2, mol : mol) at 1 : 2, (e) 160° C (Na₂CO₃ : CaCl₂ = 1 : 1, mol : mol), (d) 130° C (Na₂CO₃ : CaCl₂ = 1 : 2, mol : mol) at 1 : 2, (e) 160° C (Na₂CO₃ : CaCl₂ = 1 : 1, mol : mol), (d) 130° C (Na₂CO₃ : CaCl₂ = 1 : 2, mol : mol). The white arrows designated the CaCO₃ particles.

INI treatment conditions		Surface hardness (MPa)			Elastic modulus (GPa)		
Temperature (°C)	Na_2CO_3 : CaCl ₂ , mol: mol	Mean	Stdev.	LSD	Mean	Stdev.	LSD
	Untreated fiber	287.91	113.75	А	4.64	1.73	А
100	1:1	264.83	26.36	А	4.88	0.43	А
100	1:2	272.31	51.35	А	5.40	0.68	А
130	1:1	423.25	47.89	В	7.08	0.78	В
130	1:2	458.19	112.81	В	8.21	1.03	В
160	1:1	213.26	87.56	А	4.30	1.00	А
160	1:2	295.57	42.39	А	5.28	0.50	А

TABLE 2: Surface hardness and elastic modulus of the INI treated fibers.

Stdev.: means standard deviation. Average of four samples. Means with the different letter are significantly different at $\alpha = 0.05$.



FIGURE 3: EDS spectrum of kenaf fiber treated with INI process at 100° C, Na₂CO₃: CaCl₂ = 1:2, mol:mol. The white circle designated the point that was analyzed.

the fiber itself would also cause the variation in hardness and modulus properties.

3.4. Tensile Properties of Individual Fibers. The tensile properties of the individual fibers are shown in Table 3.

Table 3 shows the tensile property comparison for the INI-treated and -untreated fibers. The results showed that the INI treatments improved the tensile properties. While the improvement in tensile modulus did not show statistically different ($\alpha = 0.05$), the tensile strength of the fibers increased significantly after the INI treatments. A 27% increase in tensile strength (from 810 MPa to 1,032 MPa) was obtained for the fiber treated at 160°C (Na₂CO₃: CaCl₂ = 1:2, mol:mol). The improvement in the tensile strength should be attributed to the impregnation of CaCO₃ particles in fiber cell wall, which increases the density of the fibers and





FIGURE 4: EDS elements mapping of kenaf fiber treated with INI process at 100°C (Na₂CO₃ : CaCl₂ = 1 : 2, mol : mol). Left top: calcium (Ca). Right top: oxygen (O). Left bottom: carbon (C). Right bottom: SEM image.

reduces the defect for the fiber by filling up the micropores. The CaCO₃ particles in the fiber also helped for the stress transfer between the cellulose fibrils, yielding the overall tensile strength improvement.

3.5. Tensile Properties of INI Treated Fiber/PP Composites. The tensile strength and tensile modulus of INI treated fiber/PP composites are shown in Table 4.

For most of the treatment conditions, the PP composites reinforced with the INI-treated fibers showed improvement in both tensile strength and tensile modulus. Except for the fibers treated at 100°C (Na₂CO₃:CaCl₂ = 1:1, mol:mol), significant improvement in tensile strength was found for all other combinations. The INI treatment condition at $(100^{\circ}C, Na_2CO_3: CaCl_2 = 1:2, mol: mol)$ showed the best reinforcing effects compared to other conditions, with about 10.4% increase in tensile strength. No significant difference in tensile strength was found among the six INI treatment conditions.

The improvement in tensile modulus of the composites does correlate with the INI treatment conditions. INI treatment significantly increased tensile modulus for all combinations. The composites reinforced with the fibers treated at 160° C (Na₂CO₃: CaCl₂ = 1:1, mol: mol) gave the

TABLE 3: Tensile properties of individual fibers.

INI treatment conditions	Modulus (GPa)	LSD Test	Strength (MPa)	LSD test
Untreated fiber	13.5	А	810	А
100°C, 1:2	14.7	А	1001	В
160°C, 1:2	14.6	А	1032	В

1:2 means the molar ratio of Na₂CO₃ to CaCl₂. Average of 30 samples. Means with the different letter are significantly different at $\alpha = 0.05$.

highest tensile modulus, which was 25.9% higher than those reinforced with untreated fibers.

The typical stress-strain curves of the INI-treated fiber/PP composites are shown in Figure 5. Since the load taken up by the fibers decreases as the strain increases [14], the tensile behaviors of the composites drive to plastic deformation, and the slopes of the stress-strain curves become smaller as the strain increases. The dislocation of INI-treated kenaf fibers in the composites may occur under a higher load than untreated fibers do because CaCO₃ particles modified the compatibility between kenaf fiber and PP matrix. Therefore, the initial slopes of the stress-strain curves are higher for the INI-treated fiber/PP composites than that for the untreated fiber/PP composites, and the elongations of the



FIGURE 5: Stress-strain curves of the INI-treated fiber/PP composites. The labels mean the INI treatments conditions for the kenaf fibers. Untreated fiber is the control fiber that has been retted at 160°C but not treated with INI processes.

TABLE 4: Tensile strength and tensile modulus of INI treated fiber/PP composites.

	Tensile Strength (MPa)			Tensile Modulus (GPa)		
INI treatments	Mean	Stdev.	LSD Test	Mean	Stdev.	LSD Test
Untreated	46.77	3.43	А	1.70	0.20	А
100°C, 1:1	49.23	3.77	AB	1.88	0.11	В
100°C, 1:2	51.65	2.57	В	1.96	0.15	BC
130°C, 1:1	50.20	3.30	В	1.93	0.22	BCD
130°C, 1:2	51.61	2.51	В	2.05	0.11	DE
160°C, 1:1	51.41	4.33	В	2.14	0.22	Е
160°C, 1:2	50.94	4.31	В	1.99	0.16	BE

1:1 and 1:2 means the molar ratio of Na₂CO₃ to CaCl₂. Average of 27 samples. Means with the same letter are not significantly different at α = 0.05. Stdev: standard deviation of the mean.

INI-treated fiber/PP composites were higher than that of the untreated fiber/PP composites.

The mechanical property improvement for the kenaf fiber/PP composites should be attributed to the improvement in interfacial bonding between the fiber and PP matrix. Figure 6 shows the SEM images for the fracture surfaces of fiber/PP composites with both the INI-treated fibers (two treatment conditions) and -untreated fibers. Fiber pullout is clearly shown at the fracture surface for the composites with untreated fibers (Figure 6(a)) indicating poor interfacial compatibility between untreated fibers and PP. However, for the composites with INI-treated fibers, more simultaneous failure was observed. The interfacial compatibility between the INI-treated fiber and PP matrix was much improved compared to the control samples, indicating stronger adhesion between the fiber and the PP matrix. It may be reasoned that the inorganic nanoparticles deposited on the fiber surface served as the nucleation sites to initiate the crystalline



(c)

 $10 \, \mu m$

FIGURE 6: SEM images of the fracture surfaces of kenaf fiber/PP composites. (a) PP composites reinforced with untreated fibers; (b) PP composites reinforced with INI-treated fibers (INI treatment conditions were 100° C, Na_2CO_3 : $CaCl_2 = 1:2$, mol:mol), and (c) PP composites reinforced with INI-treated fibers (INI treatment conditions were 160° C, $Na_2CO_3: CaCl_2 = 1:1$, mol:mol).

formation of the semicrystalline polymer matrix around the fibers.

4. Conclusions

Inorganic nanoparticles (CaCO₃) were successfully impregnated into kenaf bast fibers. From the SEM images, the nanoparticle deposition is clearly observed on the fiber surfaces. INI treatment (at 130°C) increased the surface hardness of the fiber by 52.6–76.9% and the elastic modulus by 47.0–59.1%. The fibers treated at 160°C with INI processes yielded a 27% improvement in tensile strength. The impregnated inorganic nanoparticles improved the compatibility between the fibers and the PP matrix, resulting in an increase in tensile modulus and tensile strength of the kenaf fiber/PP composites. Based on the current lab condition, the INI treatment conditions including (1) 130° C, 1:1 (Na₂CO₃:CaCl₂, mol:mol), (2) 160° C, 1:1 (Na₂CO₃:CaCl₂, mol:mol), and (3) 160° C, 1:2 (Na₂CO₃:CaCl₂, mol:mol) produced the optimal kenaf fibers that had the best reinforcement effects for PP. The PP composites reinforced with the INI-treated (160° C, 1:1) fibers showed a 10.4% improvement in tensile strength and 25.9% in tensile modulus compared to those reinforced with untreated fibers.

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