1	Published version at http://pubs.acs.org/doi/abs/10.1021/acs.est.5b04083
2	Linking in vitro effects and detected organic micropollutants in surface
3	water using mixture toxicity modeling
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# 27 Abstract

Surface water can contain countless organic micropollutants, and targeted chemical analysis alone 28 may only detect a small fraction of the chemicals present. Consequently, bioanalytical tools can be 29 applied complementary to chemical analysis to detect the effect of complex chemical mixtures. In 30 31 this study, bioassays indicative of activation of the aryl hydrocarbon receptor (AhR), activation of the pregnane X receptor (PXR), activation of the estrogen receptor (ER), adaptive stress responses 32 to oxidative stress (Nrf2), genotoxicity (p53) and inflammation (NF-κB) and fish embryo toxicity 33 34 were applied along with chemical analysis to water extracts from the Danube River. Mixture toxicity modeling was applied to determine the contribution of detected chemicals to the biological 35 36 effect. Effect concentrations for between 0 to 13 detected chemicals could be found in the literature 37 for the different bioassays. Detected chemicals explained less than 0.2% of the biological effect in the PXR activation, adaptive stress response and fish embryo toxicity assays, while five chemicals 38 39 explained up to 80% of ER activation and three chemicals explained up to 71% of AhR activation. 40 This study highlights the importance of fingerprinting the effects of detected chemicals.

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42 Keywords: bioassay; chemical analysis; bioanalytical equivalent concentration; mixture effect
43 modeling; surface water, concentration addition

## 44 Introduction

Human-impacted rivers can contain a complex mixture of micropollutants, such as pharmaceuticals, 45 pesticides and industrial compounds, as well as their transformation products.<sup>1,2</sup> The sources of 46 these contaminants can include both point sources, such as wastewater effluent discharge, and 47 diffuse sources, such as run-off from urban and agricultural areas.<sup>3</sup> Given the diversity of 48 49 micropollutants in water, targeted chemical analysis alone is insufficient to detect all chemicals present in the aquatic environment. Bioanalytical tools can be applied complementary to chemical 50 51 analysis as they can provide information about the biological effect of chemicals present in a sample and reveal the presence of active compounds not detected by targeted analysis. 52

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In vitro bioassays based on various cellular response pathways, including induction of xenobiotic 54 metabolism, receptor-mediated effects, adaptive stress responses and cytotoxicity, have been 55 applied to detect the presence of micropollutants in water samples.<sup>4-6</sup> While activation of these 56 57 endpoints does not necessarily translate into higher level effects, biological response at the cellular level is a key step in the adverse outcome pathway.<sup>7</sup> Further, bioassays indicative of xenobiotic 58 59 metabolism and repair and defense mechanisms can be applied as sensitive tools to detect the presence of micropollutants, as effects in these endpoints often occur at lower concentrations than 60 61 those causing cell death or damage.

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Bioanalytical tools also have the advantage that they can take into account mixture effects among chemicals, rather than focusing on individual chemicals. The mixture effects that occur among chemicals can be categorized as concentration addition or independent action for chemicals acting according to the same or a different mode of action, respectively, both of which assume no interaction among the mixture components, or synergism or antagonism, where the mixture components can interact.<sup>8</sup> For environmental samples, such as surface water, containing many chemicals at low concentrations, synergism is rare and instead concentration addition has been 70 suggested as a conservative approach to evaluate mixture toxicity of multicomponent mixtures not only for receptor-mediated effects,<sup>9</sup> but also adaptive stress responses<sup>10</sup> and cytotoxicity.<sup>11</sup> Mixtures 71 72 that act in a concentration additive manner can be described using the bioanalytical equivalent 73 concentration (BEQ) concept, which represents the concentration of a reference compound that 74 elicits an equivalent response in a particular assay as the sample and can be determined from both bioassavs and chemical analysis. By comparing the BEQ from bioanalysis (BEQ<sub>bio</sub>) and BEQ from 75 chemical analysis (BEQ<sub>chem</sub>) it is possible to determine the contribution of detected chemicals to the 76 biological effect.<sup>12</sup> This approach has been applied to a wide range of water types including surface 77 water,<sup>13,14</sup> wastewater,<sup>6,14,15</sup> recycled water<sup>16</sup> and swimming pool water.<sup>17</sup> 78

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80 In this study, a suite of bioanalytical tools was applied to water samples from the human-impacted Danube River. The BEQ concept was utilized as a simple mixture effect prediction model to 81 82 determine the contribution of detected chemicals to the biological effect. The battery of bioassays 83 follows previous recommendations on the selection of sensitive indicator bioassays that cover endpoints related to different stages of cellular toxicity pathways including induction of xenobiotic 84 85 metabolism and receptor-mediated effects representing important molecular initiating events, as well as adaptive stress responses and cytotoxicity or other apical endpoints.<sup>4</sup> Bioassays indicative of 86 specific modes of action, such as estrogenic activity, have previously shown that a small number of 87 chemicals often explain a high proportion of the biological effect in wastewater,<sup>14,16,18</sup> but less is 88 89 known about the explanatory power of known chemicals in other endpoints.

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Activation of the ligand-dependent transcription factor aryl hydrocarbon receptor (AhR) was assessed using the CAFLUX assay.<sup>19</sup> While most applications focus on dioxin-like compounds, which are unlikely to be found in the water phase due to their hydrophobicity, around 16% of the 320 environmental compounds examined by Martin et al.<sup>20</sup> were found to induce AhR-dependent gene expression. Activation of the pregnane X receptor (PXR), which is an important factor in

96 xenobiotic metabolism regulation, was assessed using the HG5LN-hPXR assay and 73% of chemicals studied in Martin et al.<sup>20</sup> activated PXR. Activation of the estrogen receptor (ER) was 97 98 assessed using the reporter gene MELN assay. A suite of bioassays indicative of adaptive stress 99 responses reacting to oxidative stress (ARE-bla), genotoxicity (p53RE-bla) and inflammation (NF-100  $\kappa$ B-*bla*) were also included. Adaptive stress response pathways are activated to restore the cell to homeostasis after damage.<sup>21</sup> The oxidative stress response is mediated by Nrf2 and the antioxidant 101 response element,<sup>22</sup> and 26% of the 1859 chemicals in the US EPA ToxCast database were active in 102 the ARE-bla assay.<sup>23</sup> The p53 response is activated after DNA damage, leading to either repair or 103 apoptosis,<sup>24</sup> and activation of p53 can indicate the presence of genotoxic carcinogens.<sup>25</sup> 104 105 Approximately 15% of chemicals in the ToxCast database were active in the p53RE-bla assay.<sup>23</sup> 106 The NF-kB pathway is an important driver of the inflammatory response and can target cytochrome P450s, cytokines and apoptosis regulators,<sup>21</sup> and 3% of chemicals in the ToxCast database induced 107 a response in the NF- $\kappa$ B-*bla* assay.<sup>23</sup> Finally, the fish embryo toxicity (FET) test using zebrafish 108 109 was applied complementary to the cell-based bioassays as it can provide information about the 110 organism-level response. Apical endpoints in this test include embryo coagulation and lack of heartbeat<sup>26</sup> and a recently published database containing 641 chemicals showed that 74% of reviewed 111 chemicals caused mortality in the FET test.<sup>27</sup> 112

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The current study aimed to assess what fraction of the biological effects of the cellular toxicity 114 pathway can be explained by the quantified chemicals with available effect data. We used samples 115 116 from a large water body with high dilutions and low levels of micropollutants stemming from very 117 diverse sources to test the hypothesis that in these water types there is typically not a dominant 118 chemical or chemical group but instead the effects are largely driven by the mixture effects of many 119 chemicals. Large volume solid phase extraction (LVSPE) water extracts from the Danube River were analyzed in the bioassays introduced above to determine BEQ<sub>bio</sub>. The effect analysis was 120 121 complemented with targeted chemical analysis of 272 water relevant chemicals, including pesticides, pharmaceuticals, artificial sweeteners, steroidal hormones and industrial compounds. The target list is by no means comprehensive, but is based on previous targeted and non-targeted analysis of the Danube River.<sup>28</sup> Effect concentrations for the individual detected chemicals were collected from the literature to determine BEQ<sub>chem</sub>. By comparing BEQ<sub>bio</sub> and BEQ<sub>chem</sub> it was possible to determine the extent to which the detected chemicals contributed to the mixture effect in each bioassay.

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# 129 Experimental

130 Sampling

Sampling occurred during the 3<sup>rd</sup> Joint Danube Survey (JDS3)<sup>29</sup> between August and September 131 2013 using LVSPE (Maxx GmbH, Rangendingen, Germany).<sup>30</sup> The sampling locations, which 132 included both the Danube River and its tributaries, are shown in Table S1 and Figure S1 of the 133 134 Supporting Information (SI), along with detailed information about sample enrichment and extraction. Briefly, up to 500 L of water was passed through a stainless steel chamber containing 135 neutral sorbent Chromabond® HR-X, anionic exchanger Chromabond® HR-XAW and cationic 136 exchanger Chromabond® HR-XCW (Macherey-Nagel, Dueren, Germany). After extraction, each 137 138 solid phase was freeze-dried, then extracted with solvents and the eluates were combined. The 139 sample aliquots were reduced to dryness via rotary and nitrogen evaporation prior to shipping, then 140 re-suspended in either DMSO or methanol, depending on the assay.

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## 142 Chemical Analysis

Target screening analysis of the JDS sample extracts for 264 chemicals was performed by liquid chromatography-high resolution mass spectrometry (LC-HRMS) using an Agilent 1200 LC coupled to a Thermo LTQ Orbitrap XL. Analysis was run in both positive and negative mode electrospray ionization. For further details see Hug et al.<sup>31</sup> Eight steroidal hormones and industrial phenolic compounds were analyzed by LC-MS/MS using an Agilent 1260 LC coupled to a ABSciex QTrap 148 6500 instrument operated in negative mode electrospray ionization. Further details are provided in 149 Section S2 and Table S2. A full list of analyzed chemicals is provided in Table S3, along with 150 method detection limits (MDL) for each chemical in units of nanogram per liter. For mixture 151 modeling the detected chemical concentration was converted to molar units.

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## 153 Bioanalysis

154 Information about the studied bioassays and the derivation of effect concentrations can be found in 155 Table 1 and Section S3. The data was expressed in units of relative enrichment factor (REF), which takes into account sample enrichment by LVSPE and dilution in the assays. The data was expressed 156 157 as concentration causing 10% effect (EC<sub>10</sub>), effect concentration causing an induction ratio of 1.5 158 (EC<sub>IR1.5</sub>) or concentration causing 50% mortality (LC<sub>50</sub>). Linear concentration-effect curves were used to determine EC<sub>10</sub> and EC<sub>IR1.5</sub>, while LC<sub>50</sub> was evaluated using log-logistic concentration-159 effect curves.<sup>4</sup> Cytotoxicity was assessed in parallel for the AhR, ER, oxidative stress response, p53 160 161 response and NF-kB assays and cell viability EC<sub>10</sub> values were derived from log-logistic concentration-effect curves.<sup>32</sup> 162

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## 164 Bioanalytical Equivalent Concentrations

165 The LC and EC values were converted to  $BEQ_{bio}$  using Equation 1, with the  $LC_{50}$  or  $EC_{10}$  or  $EC_{IR1.5}$ 166 value of the reference compound (ref) and the matching  $LC_{50}$  or  $EC_{10}$  or  $EC_{IR1.5}$  value of the extract 167 only.

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$$BEQ_{bio} = \frac{LC_{50} \text{ (ref)}}{LC_{50} \text{ (extract)}} \text{ or } \frac{EC_{10} \text{ (ref)}}{EC_{10} \text{ (extract)}} \text{ or } \frac{EC_{IR1.5} \text{ (ref)}}{EC_{IR1.5} \text{ (extract)}}$$
(1)

172 The BEQ concept has been typically applied to log-logistic concentration-effect curves; however, 173 for many environmental samples, linear concentration-effect curves may be more suitable for data evaluation. This is because environmental samples may only induce low effects and to obtain the 174 50% effect concentration one would have to either use an unfeasibly high enrichment factor or 175 176 extrapolate the data. For linear concentration-effect curves to remain valid they should reach no more than 20 to 30% effect or have an induction ratio (IR) no greater than 5 to ensure that they 177 remain in the linear range of the curve. Linear concentration-effect curves have previously been 178 179 shown to be a robust data evaluation method for environmental samples, individual chemicals and chemical mixtures.<sup>10</sup> It must be stressed that the BEQ concept is only valid if the slopes of the 180 sample and reference compound are parallel in log-logistic concentration-effect curves.<sup>33</sup> However, 181 182 parallel slopes are not a requirement for linear concentration-effect curves with a common intercept at the effect axis as the BEQ is the ratio between concentrations at a given effect level and is 183 184 therefore independent of the effect level. The EC value from a linear concentration-effect curve was 185 calculated using Equation 2 using the example of  $EC_{10}$ , but the same equation is applicable for EC<sub>20</sub>, for example, with 20% used instead of 10% or EC<sub>IR1.5</sub> with an IR of 1.5 as the effect 186 benchmark. BEQ can then be calculated using Equation 3 and while this example is for  $EC_{10}$ , this 187 188 ratio is constant across the entire linear concentration-effect curve range.

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$$EC_{10} = \frac{10\%}{slope}$$

(2)

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191

$$BEQ_{bio} = \frac{EC_{10} (ref)}{EC_{10} (extract)} = \frac{10\%}{slope (ref)} \cdot \frac{slope (extract)}{10\%} = \frac{slope (extract)}{slope (ref)}$$
192
(3)

To calculate  $BEQ_{chem}$  it was first necessary to determine the relative effect potency (REP<sub>i</sub>) of the detected chemicals (i). As the EC values for the detected chemicals were generally provided as  $EC_{50}$ values in the literature, it was necessary to use  $EC_{50}$  values derived from log-logistic concentrationeffect curves for the AhR, PXR and ER assays. REP<sub>i</sub> was calculated using Equation 4, with the  $LC_{50}$  or  $EC_{50}$  or  $EC_{IR1.5}$  value of the reference compound and the matching  $LC_{50}$  or  $EC_{50}$  or  $EC_{IR1.5}$ value of detected chemical i.

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$$REP_{i} = \frac{LC_{50} \text{ (ref)}}{LC_{50} \text{ (i)}} \text{ or } \frac{EC_{50} \text{ (ref)}}{EC_{50} \text{ (i)}} \text{ or } \frac{EC_{IR1.5} \text{ (ref)}}{EC_{IR1.5} \text{ (i)}}$$

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202 (4)

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All LC values for the detected chemicals in the FET test were collected from Scholz et al.<sup>27</sup>, while 204 205 the EC values were collected from the peer-reviewed literature (AhR, PXR, ER assays) or the ToxCast database (oxidative stress response, p53 response, NF-κB assays),<sup>23</sup> which includes over 206 207 1800 compounds in over 800 different assays. All ToxCast data was re-evaluated to determine EC<sub>IR1.5</sub> using linear concentration-effect curves. As each chemical in the ToxCast database was run 208 209 multiple times, it was possible to determine the mean EC<sub>IR1.5</sub> value and the associated standard deviation. BEQ<sub>chem</sub> was calculated for each JDS sample using REP<sub>i</sub> and the detected concentration 210 (M) (Equation 5). The variability associated with BEQ<sub>chem</sub> for the chemicals present in the ToxCast 211 212 database was assessed using error propagation. EC and LC values collected from the literature 213 generally did not include standard deviation, so it was not possible to determine the variability associated with BEQ<sub>chem</sub> for the AhR, PXR, ER and FET assays. 214

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$$BEQ_{chem} = \sum_{i=1}^{n} REP_i \cdot C_i$$

216

(5)

# 217 **Results and Discussion**

#### 218 Chemical Analysis

Of the 272 analyzed chemicals, 94 were detected at least once in the 22 JDS samples. The number 219 220 of chemicals detected at each site ranged from 20 to 64. The sum of the molar concentration and 221 number of detected chemicals at each site are shown in Figure 1A, with the concentrations in pM for each of the detected chemicals at the different sampling sites shown in Table S4. The most 222 frequently detected chemicals were the artificial sweetener acesulfame, the industrial compounds 223 224 triphenylphosphine oxide and 2-benzothiazolesulfonic acid and the antimicrobial sulfamethoxazole, 225 which were present at detectable levels at all studied sites. In all but one tributary other common 226 wastewater micropollutants, including carbamazepine and its transformation products, the corrosion 227 inhibitors benzotriazole and methylbenzotriazole, the artificial sweeteners cyclamate and sucralose and several herbicides and transformation products (metolachlor, isoproturon, atrazine, and 228 229 terbuthylazine-2-hydroxy), were detected. The antidiabetic pharmaceutical metformin was found at 230 the highest concentrations, with concentrations up to 7.6 nM. Overall, chemical contamination was 231 relatively low, with none of the detected chemicals exceeding the Water Framework Directive environmental quality standards.<sup>34</sup> 232

233

#### 234 Bioanalysis

The EC and LC values for the different JDS water samples are shown in Figure 1B and Table S5, with the concentration-effect curves for all assays shown in Figure S2. The assays indicative of activation of ER, activation of PXR, activation of AhR and NF- $\kappa$ B response tended to be the most responsive, followed by the oxidative stress response. The p53 response occurred at higher effect concentrations. The least responsive assay was the FET test, which required a REF of 100 to 500 for 50% mortality, or a REF of 50 to 300 for 10% mortality.

While most samples did have a response in the assays, the effects were relatively low, with the EC values for the oxidative stress and AhR assays similar to previously benchmarked EC values for surface water.<sup>35</sup> Kittinger et al.<sup>36</sup> also only detected minimal effects in Danube River samples when assessing genotoxicity. Further, ER activation, when expressed as BEQ<sub>bio</sub> (0.02-1.1 pM), was lower than generally observed in wastewater effluent,<sup>37</sup> due to dilution in the river, though one contaminated site, JDS 41 (BEQ<sub>bio</sub> 4.7 pM), was identified by this assay.

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249 JDS 64 had the lowest sum chemical concentration, and this corresponded to no effect at the 250 maximum REF for the oxidative stress response, p53 response and NF-KB assays, and only minimal 251 effects at high concentrations in the other assays. Cell viability was assessed in parallel for most 252 cell-based assays and in most cases there was negligible cytotoxicity in the studied concentration 253 range. However, cytotoxicity did mask other endpoints manifestation at high REFs in some samples 254 for the AhR (JDS 41 and 63), ER (JDS 35, 55, 57, 59, 63 and 67), oxidative stress response (JDS 55 255 and 67), p53 response (JDS 41, 55, 57 and 63), and NF-kB (JDS 36 and 41) assays. Hence, it was not possible to derive EC values for induction for these particular samples, but  $EC_{10}$  values for 256 cytotoxicity were calculated and are included in Table S5. JDS 41, which had the highest effect in 257 258 the ER activation and oxidative stress response assays and was cytotoxic in several other assays, 259 was the most polluted site with the highest amount of total detected chemical concentration. 260 Overall, there was no significant relationship between effect and sum detected chemicals at each 261 site for the different assays.

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For mixture modeling, the EC and LC values were converted to BEQ<sub>bio</sub> using the respective reference compounds for each assay (Table 2, Table S5). While the EC or LC values give an indication of the sensitivity of the assay, BEQ<sub>bio</sub> converts the effect into the concentration of reference compound that would elicit the same response as the sample mixture. Further, the BEQ concept simplifies mixture toxicity modeling.<sup>10,11</sup>

#### 269 Bioanalytical equivalent concentration from chemical analysis

Prior to calculating BEQ<sub>chem</sub>, the published literature and ToxCast database were searched for EC or LC values for the detected chemicals. For each assay, between 0 and 13 literature EC or LC values could be found for the 94 detected chemicals (Table S6). Using the literature EC or LC values for each chemical and the EC or LC value of the assay reference compound, the REP<sub>i</sub> was calculated using Equation 4 (Table 3).

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 $LC_{50}$  values at 48 h exposure for the FET test were collected from Scholz et al.<sup>27</sup>, with REP<sub>i</sub> 276 277 calculated using the mean 3,4-dichloroaniline  $LC_{50}$  value from the same study. Thirteen EC values 278 were collected from the literature for the PXR assay, while six EC values were available for the ER assay. Although no EC values were available for the detected chemicals in the AhR CAFLUX 279 280 assay, EC values were available for three of the detected chemicals (terbuthylazine, carbaryl, daidzein) in the mouse AhR CALUX assay. While these assays focus on the same endpoint 281 282 (reporter gene expression), they utilize different animal cell lines (rat hepatoma versus mouse hepatoma) and previous work has shown species-specific differences in responsiveness to some 283 AhR ligands.<sup>38</sup> To account for differences in sensitivity between the mouse and rat AhR models, 284 TCDD EC values were also collected from each study to calculate REP<sub>i</sub>, rather than using the 285 286 TCDD EC value from the current study.

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EC values for the oxidative stress response, p53 response and NF- $\kappa$ B assays were collected from the ToxCast database.<sup>23</sup> A total of 486, 278 and 62 chemicals in the ToxCast database were active in the oxidative stress response, p53 response and NF- $\kappa$ B assays, respectively (Figure 2). Of the 94 chemicals detected in the JDS samples, 49 of these were also included in the ToxCast database. However, many of the detected compounds were not active in the assays, with 13 compounds active in the oxidative stress response assay, 4 compounds active in the p53 response assay and none active in the NF- $\kappa$ B assay. EC<sub>IR1.5</sub> values for the detected chemicals in the oxidative stress response and p53 response assays were calculated from raw emission data available in the ToxCast MySQL database. To derive REP<sub>i</sub> for the detected chemicals, experimental EC<sub>IR1.5</sub> values for reference compounds tBHQ and mitomycin were used.

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299 Using REP<sub>i</sub> and the detected chemical concentration, BEQ<sub>chem</sub> was calculated using Equation 5 for 300 each water sample (Table 2). BEQ<sub>chem</sub> could not be calculated for some samples for the AhR and 301 ER bioassays as none of the chemicals with literature EC values were detected in the samples. 302 Further, it was not possible to derive BEQ<sub>chem</sub> values for the NF-KB assay as none of the detected 303 chemicals were active, despite the NF-kB assay being one of the more responsive for the JDS 304 samples (Figure 1B). Only 3% of the 1859 chemicals in the ToxCast database were active in the 305 NF-kB assay, compared to 26 and 15% of chemicals in oxidative stress response and p53 response 306 assays, respectively (Figure 2). The NF- $\kappa$ B assay has been used for water quality monitoring in only one study<sup>4</sup> and it is still unclear what types of water-based pollutants induce a response in this 307 308 assay.

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# 310 What percent of biological effect can be explained by chemical analysis?

The comparison between BEQ<sub>bio</sub> and BEQ<sub>chem</sub> for each assay is shown in Table 2, while the contribution of the individual detected chemicals to the biological effect is shown in Figure 3. For some JDS samples it was not possible to determine the contribution of detected chemicals to the biological effect, and this was attributed to either cytotoxicity masking manifestation of other endpoints, no effect at the maximum REF or the active chemicals being below the MDL.

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The BEQ<sub>chem</sub> for AhR activation was calculated using only three chemicals, but they explained between 3 and 71% of the biological effect (Figure 3A). The effect was mostly driven by the phytoestrogen daidzein, which has previously been shown to be a weak AhR activator in mouse cells, but not in human cells,<sup>39</sup> and the herbicide terbuthylazine. The insecticide carbaryl only explained 0.5% of the effect in JDS 67. Similarly, the BEQ<sub>chem</sub> for activation of ER could explain up to 80% of the effect, with the hormone estrone and the phytoestrogen genistein contributing significantly. Estrogenic effects in wastewater are often explained by the presence of potent natural and synthetic estrogenic hormones, such as 17β-estradiol and 17α-ethinylestradiol,<sup>14,18</sup> but these compounds were below the detection limit in the current study. Previous studies have also attributed ER activation in river water to genistein.<sup>40</sup>

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328 In contrast, the detected chemicals could explain less than 0.2% of the biological effect in the 329 adaptive stress response assays, PXR assay and the FET test (Figure 3). A number of studies have 330 also shown similarly low contributions of detected chemicals to the oxidative stress response in a range of water types including wastewater and pool water.<sup>10,17,41</sup> It has also been demonstrated that 331 332 detected chemicals in surface water and wastewater can only explain a small fraction of PXR activity.<sup>15</sup> However, for this receptor, the use of the concentration addition model may be limited as 333 it has been recently demonstrated that, due to a large ligand-binding pocket, PXR can stably bind 334 binary mixtures of certain weakly active chemicals, which leads to synergistic activation of target 335 genes.<sup>42</sup> The BEQ<sub>chem</sub> vs BEQ<sub>bio</sub> comparison for water samples has not been conducted previously 336 for the p53 response or FET assays, thus it was not possible to compare our results with the 337 338 literature. The herbicide metolachlor mostly contributed to the effect in the PXR assay. Genistein 339 dominated the contribution of quantified chemicals to the biological effect for the oxidative stress 340 response. While carbaryl and the disinfectant chlorophene were detected in the JDS samples and are 341 active in the oxidative stress response assay, they occurred only in samples where cytotoxicity masked induction and could not be used to explain the biological effect. Genistein and the industrial 342 343 compound 2,4-dinitrophenol mostly contributed to the p53 response in samples collected from 344 Austria to Serbia (JDS 8 to 39), while the fungicide carbendazim dominated the effect further 345 downstream. Finally, 2,4-dinitrophenol and genistein together contributed up to 0.08% of the effect explained by quantified chemicals in the FET assay, though antimicrobial triclosan alone could
explain up to 0.15% of the biological effect in JDS 59 (Figure 3).

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349 The small contribution of detected chemicals to the biological response in the adaptive stress 350 response and PXR assays is not surprising since many compounds can activate these endpoints, as discussed earlier. Further, 471 out of 641 or 74%, of reviewed compounds in Scholz et al.<sup>27</sup> had a 351 response in the FET test. Consequently, many compounds are active in these assays and while 352 353 comparability may have been improved with more literature EC and LC values for the detected compounds, it is unlikely to have a significant influence on the comparison. To illustrate this point, 354 355 if we assume that 74% of chemicals should have an effect in the FET test, then 70 of the 94 356 detected chemicals could be active in this assay. However, published LC values were only available for 12 of the detected chemicals. If we simply extrapolate the effect explained by the detected 357 358 chemicals with available  $LC_{50}$  values in each sample to all 70 detected chemicals, without 359 considering differences in potencies, we can still only explain up to 1.6% of the effect. This 360 example does not take into consideration differences in mode of action or chemical potency, but simply aims to illustrate the potential for many compounds to contribute to effect in apical 361 362 endpoints.

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## 364 *Limitations and outlook*

There are some limitations associated with the current study. Primarily, improved understanding of the contribution of the detected chemicals to the biological effect is hampered by the lack of REP<sub>i</sub> values for detected chemicals. Out of the 94 detected chemicals in the JDS samples, between 0 and 13 corresponding EC or LC values could be found for the different assays. While the US EPA ToxCast program provides EC values for a large number of compounds, many of these are not typical water pollutants and only 52% of chemical detected in the JDS samples were present in the ToxCast database. Many of the detected chemicals in the ToxCast database were not active in the 372 adaptive stress response assays, but such information is not readily available for the other studied assays. However, this information is important as it makes a difference to the effect balance if a 373 chemical's contribution is zero or if it is unknown. Consequently, fingerprinting the biological and 374 375 toxicological effect(s) of commonly detected water pollutants is recommended to help fill in the 376 knowledge gap. Further, the available literature data stems from a number of different sources and it is possible that the experimental protocols for the same assay may differ slightly, leading to 377 378 potential differences in sensitivity or reproducibility. This limitation could be overcome by 379 improved standardization of bioassays.

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381 A specific limitation associated with the AhR assay is that the EC values available in the literature 382 are based on the mouse AhR model, while BEQ<sub>bio</sub> is based on the rat AhR model. Hence, potential differences in species sensitivity may be a source of variability for the comparison of BEQ<sub>bio</sub> and 383 384 BEQ<sub>chem</sub>. A further limitation with using literature EC and LC values is that the error associated with the value is often not provided. This was the case for the AhR, PXR, ER and FET assays and 385 386 consequently it was not possible to calculate the error associated with the BEQ<sub>chem</sub> values. It was 387 possible to calculate the error associated with BEQ<sub>chem</sub> for the oxidative stress response and p53 388 response assays as the EC<sub>IR1.5</sub> values used to derive REP<sub>i</sub> were re-evaluated from a series of 389 replicate experiments from the ToxCast MySQL database (standard deviations associated with 390 BEQ<sub>chem</sub> are provided in Table S7).

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This study demonstrated the applicability of the BEQ concept to assess the contribution of detected chemicals to the biological effect of chemical mixtures present in the Danube River. As the detected chemicals could not explain a significant proportion of the effect, particularly in the adaptive stress response, PXR and FET assays, this supports the application of bioanalytical tools complementary to chemical analysis for water quality monitoring. Further, as targeted chemical analysis was applied, we cannot exclude the fact that we may not have targeted the most relevant chemicals. 398 Consequently, further identification using tools such as effect-directed analysis may provide 399 improved understanding about chemical stressors in the Danube River.

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#### 401 Acknowledgements

402 This study is part of the SOLUTIONS project that is supported by the European Union Seventh Framework Programme (FP7-ENV-2013-two-stage Collaborative project) under grant agreement 403 404 number 603437. The study was also supported by the National Health and Medical Research 405 Council (NHMRC) - European Union Collaborative Research Grant (APP1074775), the Federal Ministry of Education and Research (BMBF) under grant agreement number 02WRS1282C (Tox-406 407 Box) and the International Commission for the Protection of the Danube (ICPDR). The samples were collected as part of the Joint Danube Survey 3 (JDS3), which was conducted by ICPDR. Nils 408 409 Klüver, UFZ, is thanked for useful discussions and provision of the FET data. Davne Filer, US 410 EPA, is thanked for assistance with the ToxCast database. Janet Tang, UQ, is thanked for 411 experimental assistance with the NF-kB assay. Margit Petre, Melis Muz, Riccardo Massai, Jörg 412 Ahlheim, all UFZ, Jaroslav Slobodnik, EI, and Peter Tarabek, MU, are thanked for logistical and 413 technical help.

414

415 Supporting Information: Further information on the LVSPE sampling, analyzed chemicals and
416 bioanalysis, as well as raw bioassay data, is provided.

417

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T742		Method	Positive reference	Maximum	Date and a the set of	EC or
FIIADOIII	ASSAY	reference	compound	REF	<b>D</b> аца еуацианоп шенноп	LC value
Activation of AhR	CAFLUX	Nagy et al. <sup>19</sup>	2,3,7,8-Tetrachlorodibenzo- <i>p</i> -dioxin (TCDD)	500	Linear concentration-effect curve	EC <sub>10</sub>
Activation of PXR	HG5LN-hPXR	Lemaire et al. <sup>43</sup> ; Creusot et al. <sup>15</sup>	SR 12813*	500	Linear concentration-effect curve	$EC_{10}$
Activation of ER	MELN	Balaguer et al. <sup>44</sup> ; Kinani et al. <sup>45</sup>	17β-Estradiol	500	Linear concentration-effect curve	EC <sub>10</sub>
Oxidative stress response	ARE-bla	Invitrogen <sup>46</sup>	tert-Butylhydroquinone (tBHQ)	500	Linear concentration-effect curve	EC <sub>IR1.5</sub>
p53 response	p53RE-bla	Neale et al. <sup>47</sup>	Mitomycin	500	Linear concentration-effect curve	EC <sub>IR1.5</sub>
NF-kB response	NF-кB- <i>bla</i>	Jin et al. <sup>48</sup>	Tumor necrosis factor alpha (TNFα)	250	Linear concentration-effect curve	EC <sub>IR1.5</sub>
Mortality	Fish embryo toxicity (FET)	OECD <sup>26</sup>	3,4-Dichloroaniline	1000	Log-logistic concentration-effect curve	$LC_{50}$

598 \*Tetraethyl 2-(3,5-di-tert-butyl-4-hydroxyphenyl)ethenyl-1,1-bisphosphonate

Table 2: BEQ<sub>bio</sub> and BEQ<sub>chem</sub> values for each sample in the different bioassays with the percentage of effect that can be explained by the detected 599

600 chemicals

Ð	Acti	Activation of AhR	hR	Activ	Activation of PXR	XR	Acti	Activation of ER	R	Oxidat	Oxidative stress response	sponse	1	p53 response	a)	Fish e	Fish embryo toxicity	city
	$BEQ_{bio}$	$BEQ_{chem}$	%	$BEQ_{bio}$	$BEQ_{chem}$	%	$BEQ_{bio}$	$BEQ_{chem}$	%	$BEQ_{bio}$	$BEQ_{chem}$	0/ 2000	$BEQ_{bio}$	$BEQ_{chem}$	%	$BEQ_{bio}$	$BEQ_{chem}$	%
	(W)	(W)	effect	(W)	(W)	effect	(W)	(W)	effect	(W)	(W)	% ellect	(W)	(W)	effect	(W)	(W)	effect
×	4.40 $ imes 10^{-14}$	$1.94 \times 10^{-15}$	4.4%	$3.43 \times 10^{-9}$	$6.51 \times 10^{-13}$	0.02%	$9.61 \times 10^{-14}$	$\frac{1.79}{\times 10^{-14}}$	19%	$3.11 \times 10^{-8}$	$1.10 \times 10^{-11}$	0.04%	$1.51 \times 10^{-10}$	$\begin{array}{c} 4.00 \\ \times 10^{-14} \end{array}$	0.03%	$6.74 \\ \times 10^{-8}$	$2.66 \times 10^{-11}$	0.04%
22	$3.28 \times 10^{-14}$	<mdl< th=""><th>ı</th><th><math display="block">\begin{array}{c} 4.98 \\ \times 10^{-10} \end{array}</math></th><th><math>3.65 \times 10^{-13}</math></th><th>0.07%</th><th><math>3.14 \times 10^{-13}</math></th><th><math display="block">\begin{array}{c} 1.45 \\ \times 10^{-14} \end{array}</math></th><th>4.6%</th><th><math display="block">\begin{array}{c} 2.75 \\ \times 10^{-8} \end{array}</math></th><th><math>1.37 \times 10^{-12}</math></th><th>0.01%</th><th><math>2.03 \times 10^{-10}</math></th><th><math display="block">4.53 \\ \times 10^{-15}</math></th><th>0.002%</th><th><math>5.88 \times 10^{-8}</math></th><th><math display="block">6.48 \\ \times 10^{-12}</math></th><th>0.01%</th></mdl<>	ı	$\begin{array}{c} 4.98 \\ \times 10^{-10} \end{array}$	$3.65 \times 10^{-13}$	0.07%	$3.14 \times 10^{-13}$	$\begin{array}{c} 1.45 \\ \times 10^{-14} \end{array}$	4.6%	$\begin{array}{c} 2.75 \\ \times 10^{-8} \end{array}$	$1.37 \times 10^{-12}$	0.01%	$2.03 \times 10^{-10}$	$4.53 \\ \times 10^{-15}$	0.002%	$5.88 \times 10^{-8}$	$6.48 \\ \times 10^{-12}$	0.01%
27	$2.65 \times 10^{-14}$	$1.44 \times 10^{-15}$	5.4%	$2.40 \times 10^{-9}$	$4.94 \times 10^{-13}$	0.02%	$5.67 \times 10^{-13}$	$\begin{array}{c} 1.04 \\ \times 10^{-14} \end{array}$	1.8%	$3.59 \times 10^{-8}$	$1.77 \times 10^{-11}$	0.05%	$2.46 \times 10^{-10}$	$6.39 \times 10^{-14}$	0.03%	$5.07$ $ imes 10^{-8}$	$3.57 \times 10^{-11}$	0.07%
29	5.37 $ imes 10^{-14}$	<mdl< th=""><th>,</th><th><math>2.13 \times 10^{-9}</math></th><th><math>4.31 \times 10^{-13}</math></th><th>0.02%</th><th>7.44 ×10<sup>-13</sup></th><th><math>3.96 \times 10^{-14}</math></th><th>5.3%</th><th><math>3.40 \times 10^{-8}</math></th><th><math display="block">\begin{array}{c} 1.45 \\ \times 10^{-11} \end{array}</math></th><th>0.04%</th><th><math display="block">\begin{array}{c} 4.04 \\ \times 10^{-10} \end{array}</math></th><th><math>7.74 \times 10^{-14}</math></th><th>0.02%</th><th><math display="block">\begin{array}{c} 4.96 \\ \times 10^{-8} \end{array}</math></th><th><math>5.04 \times 10^{-11}</math></th><th>0.1%</th></mdl<>	,	$2.13 \times 10^{-9}$	$4.31 \times 10^{-13}$	0.02%	7.44 ×10 <sup>-13</sup>	$3.96 \times 10^{-14}$	5.3%	$3.40 \times 10^{-8}$	$\begin{array}{c} 1.45 \\ \times 10^{-11} \end{array}$	0.04%	$\begin{array}{c} 4.04 \\ \times 10^{-10} \end{array}$	$7.74 \times 10^{-14}$	0.02%	$\begin{array}{c} 4.96 \\ \times 10^{-8} \end{array}$	$5.04 \times 10^{-11}$	0.1%
30	$1.01 \times 10^{-14}$	<mdl< th=""><th></th><th><math>3.45 \times 10^{-10}</math></th><th><math>3.42 \times 10^{-13}</math></th><th>0.10%</th><th><math>1.03 \times 10^{-12}</math></th><th><math display="block">\begin{array}{c} 4.32 \\ \times 10^{-14} \end{array}</math></th><th>4.2%</th><th><math>2.05 \times 10^{-8}</math></th><th><math>2.39 \times 10^{-11}</math></th><th>0.1%</th><th><math>2.59 \times 10^{-10}</math></th><th><math>3.88 \times 10^{-14}</math></th><th>0.01%</th><th><math>9.20 \times 10^{-8}</math></th><th><math>2.32 \times 10^{-11}</math></th><th>0.03%</th></mdl<>		$3.45 \times 10^{-10}$	$3.42 \times 10^{-13}$	0.10%	$1.03 \times 10^{-12}$	$\begin{array}{c} 4.32 \\ \times 10^{-14} \end{array}$	4.2%	$2.05 \times 10^{-8}$	$2.39 \times 10^{-11}$	0.1%	$2.59 \times 10^{-10}$	$3.88 \times 10^{-14}$	0.01%	$9.20 \times 10^{-8}$	$2.32 \times 10^{-11}$	0.03%
32	$3.50 \times 10^{-14}$	$6.38 \\ \times 10^{-15}$	18%	<3.17× 10 <sup>-11</sup>	$4.74 \times 10^{-13}$	ı	$1.82 \times 10^{-13}$	$1.11 \times 10^{-13}$	61%	$5.93 \times 10^{-8}$	$\frac{1.27}{\times 10^{-10}}$	0.2%	$3.47$ $ imes 10^{-10}$	$2.54 \times 10^{-13}$	0.07%	$2.11 \times 10^{-7}$	$\begin{array}{c} 1.31 \\ \times 10^{-10} \end{array}$	0.06%
33	$\begin{array}{c} 4.05 \\ \times 10^{-14} \end{array}$	$\begin{array}{c} 1.34 \\ \times 10^{-15} \end{array}$	3.3%	7.72 ×10 <sup>-10</sup>	$4.82 \times 10^{-13}$	0.06%	$1.98 \times 10^{-13}$	$6.00 \\ \times 10^{-14}$	30%	$3.62 \times 10^{-8}$	$5.19 \times 10^{-11}$	0.1%	$1.89 \times 10^{-10}$	$1.31 \times 10^{-13}$	0.07%	$7.43 \times 10^{-8}$	$6.90 \times 10^{-11}$	0.09%
35	$3.36 \times 10^{-14}$	$\begin{array}{c} 8.55 \\ \times 10^{-15} \end{array}$	25%	$4.61 \times 10^{-9}$	6.45 ×10 <sup>-13</sup>	0.01%	Cytotox	$3.07 \times 10^{-14}$	ı	$9.58 \times 10^{-8}$	1.31 ×10 <sup>-11</sup>	0.01%	$\begin{array}{c} 4.04 \\ \times 10^{-10} \end{array}$	3.69 ×10 <sup>-14</sup>	0.01%	$7.08 \times 10^{-8}$	2.47 ×10 <sup>-11</sup>	0.03%
36	$2.94 \times 10^{-14}$	<wdl< th=""><th>ı</th><th>5.24 <math> imes 10^{-10}</math></th><th><math>3.57 \times 10^{-13}</math></th><th>0.07%</th><th><math>3.47 \times 10^{-13}</math></th><th><math>2.37 \times 10^{-14}</math></th><th>6.8%</th><th><math>3.60 \times 10^{-8}</math></th><th>1.69 ×10<sup>-11</sup></th><th>0.05%</th><th><math display="block">\begin{array}{c} 1.87 \\ \times 10^{-10} \end{array}</math></th><th>4.75 ×10<sup>-14</sup></th><th>0.03%</th><th><math>3.54</math> <math> imes 10^{-8}</math></th><th><math>2.57 \times 10^{-11}</math></th><th>0.07%</th></wdl<>	ı	5.24 $ imes 10^{-10}$	$3.57 \times 10^{-13}$	0.07%	$3.47 \times 10^{-13}$	$2.37 \times 10^{-14}$	6.8%	$3.60 \times 10^{-8}$	1.69 ×10 <sup>-11</sup>	0.05%	$\begin{array}{c} 1.87 \\ \times 10^{-10} \end{array}$	4.75 ×10 <sup>-14</sup>	0.03%	$3.54$ $ imes 10^{-8}$	$2.57 \times 10^{-11}$	0.07%
37	$<1.25$ $\times10^{-14}$	<mdl< th=""><th>ı</th><th><math>1.53 \times 10^{-9}</math></th><th><math>3.19 \times 10^{-13}</math></th><th>0.02%</th><th>2.25 ×10<sup>-13</sup></th><th><math>1.68 \times 10^{-14}</math></th><th>7.4%</th><th><math>2.79 \times 10^{-8}</math></th><th><math display="block">\begin{array}{c} 2.28 \\ \times 10^{-12} \end{array}</math></th><th>0.01%</th><th><math>5.58 \times 10^{-10}</math></th><th>2.45 ×10<sup>-14</sup></th><th>0.004%</th><th><math display="block">\begin{array}{c} 4.80 \\ \times 10^{-8} \end{array}</math></th><th>1.37 ×10<sup>-11</sup></th><th>0.03%</th></mdl<>	ı	$1.53 \times 10^{-9}$	$3.19 \times 10^{-13}$	0.02%	2.25 ×10 <sup>-13</sup>	$1.68 \times 10^{-14}$	7.4%	$2.79 \times 10^{-8}$	$\begin{array}{c} 2.28 \\ \times 10^{-12} \end{array}$	0.01%	$5.58 \times 10^{-10}$	2.45 ×10 <sup>-14</sup>	0.004%	$\begin{array}{c} 4.80 \\ \times 10^{-8} \end{array}$	1.37 ×10 <sup>-11</sup>	0.03%
39	$2.02  imes 10^{-14}$	$1.02 \\ \times 10^{-14}$	50%	$2.84 \times 10^{-9}$	$6.90 \times 10^{-13}$	0.02%	$1.15 \times 10^{-13}$	$\begin{array}{c} 4.63 \\ \times 10^{-14} \end{array}$	40%	$2.63 \times 10^{-8}$	$2.03 \times 10^{-11}$	0.08%	$2.63 \times 10^{-10}$	$6.67$ $ imes 10^{-14}$	0.03%	$4.90 \times 10^{-8}$	$3.99 \times 10^{-11}$	0.08%

41	Cytotox	$1.91 \times 10^{-14}$	ı	1.91 ×10 <sup>-9</sup>	$4.84 \times 10^{-13}$	0.03%	$4.56 \\ \times 10^{-12}$	$1.43 \times 10^{-13}$	3.1%	$1.58$ $\times 10^{-7}$	$1.25 \times 10^{-11}$	0.01%	Cytotox	$1.18 \times 10^{-13}$	ı	7.22 $\times 10^{-8}$	$\frac{1.48}{\times 10^{-11}}$	0.02%
44	$1.85 \times 10^{-14}$	$1.08 \times 10^{-14}$	59%	$3.22 \times 10^{-9}$	6.86 ×10 <sup>-13</sup>	0.02%	$4.47$ $\times 10^{-13}$	3.06 ×10 <sup>-14</sup>	6.8%	$5.20  imes 10^{-8}$	2.30 ×10 <sup>-11</sup>	0.04%	$7.02 \times 10^{-10}$	$1.57 \times 10^{-13}$	0.02%	$5.93 \times 10^{-8}$	3.58 ×10 <sup>-11</sup>	0.06%
53	$2.59  imes 10^{-14}$	1.48 ×10 <sup>-15</sup>	5.7%	$1.28$ $ imes 10^{-9}$	$4.75 \times 10^{-13}$	0.04%	$2.73 \times 10^{-13}$	$3.68 \times 10^{-14}$	14%	$5.69 \times 10^{-8}$	$3.67 \times 10^{-12}$	0.01%	<1.01 $\times 10^{-10}$	$2.91 \times 10^{-14}$	ı	$4.98 \times 10^{-8}$	$\frac{1.83}{\times 10^{-11}}$	0.04%
55	$2.99 \times 10^{-14}$	$\begin{array}{c} 1.48 \\ \times 10^{-15} \end{array}$	5.0%	$6.08 \\ \times 10^{-10}$	5.16 ×10 <sup>-13</sup>	0.08%	Cytotox	<th></th> <th>Cytotox</th> <th><math>2.82 \times 10^{-12}</math></th> <th>·</th> <th>Cytotox</th> <th><math>1.36 \times 10^{-13}</math></th> <th>ı</th> <th><math>5.32</math> <math> imes 10^{-8}</math></th> <th><math>1.85 \times 10^{-11}</math></th> <th>0.03%</th>		Cytotox	$2.82 \times 10^{-12}$	·	Cytotox	$1.36 \times 10^{-13}$	ı	$5.32$ $ imes 10^{-8}$	$1.85 \times 10^{-11}$	0.03%
57	$\begin{array}{c} 1.71 \\ \times 10^{-14} \end{array}$	1.22 ×10 <sup>-15</sup>	7.1%	$1.88 \times 10^{-9}$	$3.99 \times 10^{-13}$	0.02%	Cytotox	$\begin{array}{c} 1.22 \\ \times 10^{-14} \end{array}$		$4.07 \times 10^{-8}$	$2.45$ $\times 10^{-12}$	0.01%	Cytotox	$1.06 \times 10^{-13}$	ı	$5.10$ $ imes 10^{-8}$	$\begin{array}{c} 1.58 \\ \times 10^{-11} \end{array}$	0.03%
59	$2.96 \times 10^{-14}$	$2.05 \times 10^{-14}$	69%	$2.21$ $ imes 10^{-9}$	$6.94 \times 10^{-13}$	0.03%	Cytotox	$\begin{array}{c} 4.53 \\ \times 10^{-14} \end{array}$	·	$5.85$ $ imes 10^{-8}$	2.38 ×10 <sup>-11</sup>	0.04%	$2.45$ $ imes 10^{-10}$	$1.63 \times 10^{-13}$	0.07%	$4.94$ $ imes 10^{-8}$	9.91 ×10 <sup>-11</sup>	0.20%
60	$\begin{array}{c} 1.69 \\ \times 10^{-14} \end{array}$	$\begin{array}{c} 1.20 \\ \times 10^{-14} \end{array}$	71%	<3.17 ×10 <sup>-11</sup>	$4.42 \times 10^{-13}$	ı	$3.77 \times 10^{-13}$	$\frac{1.19}{\times 10^{-15}}$	0.31%	$5.50$ $ imes 10^{-8}$	$2.89 \times 10^{-12}$	0.01%	$2.23$ $ imes 10^{-10}$	$1.04 \times 10^{-13}$	0.05%	$\begin{array}{c} 4.52 \\ \times 10^{-8} \end{array}$	$7.85$ $\times 10^{-12}$	0.02%
63	Cytotox	$\begin{array}{c} 1.25 \\ \times 10^{-14} \end{array}$		2.41 $\times 10^{-9}$	$1.15 \times 10^{-12}$	0.05%	Cytotox	$\begin{array}{c} 4.14 \\ \times 10^{-14} \end{array}$		$8.11 \times 10^{-8}$	$\begin{array}{c} 8.74 \\ \times 10^{-12} \end{array}$	0.01%	Cytotox	2.16 ×10 <sup>-13</sup>	ı	$1.35$ $\times 10^{-7}$	5.18 ×10 <sup>-11</sup>	0.04%
64	$5.87$ $ imes 10^{-15}$	<mdl< th=""><th>,</th><th><math display="block">\begin{array}{c} 1.25 \\ \times 10^{-10} \end{array}</math></th><th>7.21 ×10<sup>-14</sup></th><th>0.06%</th><th><math display="block">\frac{1.69}{\times 10^{-14}}</math></th><th><math display="block">\begin{array}{c} 1.36 \\ \times 10^{-14} \end{array}</math></th><th>80%</th><th><math>&lt;4.99</math> <math>\times 10^{-9}</math></th><th><math>1.99 \times 10^{-12}</math></th><th>ı</th><th><math>&lt;\!\!1.01</math> <math>\times 10^{-10}</math></th><th><math display="block">\begin{array}{c} 1.88 \\ \times 10^{-14} \end{array}</math></th><th>ı</th><th><math>4.49</math> <math> imes 10^{-8}</math></th><th><math display="block">\begin{array}{c} 1.80 \\ \times 10^{-11} \end{array}</math></th><th>0.04%</th></mdl<>	,	$\begin{array}{c} 1.25 \\ \times 10^{-10} \end{array}$	7.21 ×10 <sup>-14</sup>	0.06%	$\frac{1.69}{\times 10^{-14}}$	$\begin{array}{c} 1.36 \\ \times 10^{-14} \end{array}$	80%	$<4.99$ $\times 10^{-9}$	$1.99 \times 10^{-12}$	ı	$<\!\!1.01$ $\times 10^{-10}$	$\begin{array}{c} 1.88 \\ \times 10^{-14} \end{array}$	ı	$4.49$ $ imes 10^{-8}$	$\begin{array}{c} 1.80 \\ \times 10^{-11} \end{array}$	0.04%
65	$3.33 \times 10^{-14}$	7.48 ×10 <sup>-15</sup>	22%	$9.97 \times 10^{-10}$	$6.28 \times 10^{-13}$	%90.0	$6.15 \times 10^{-13}$	$\begin{array}{c} 4.76 \\ \times 10^{-14} \end{array}$	7.7%	$6.53 \times 10^{-8}$	$4.20 \times 10^{-12}$	0.01%	$3.69 \times 10^{-10}$	1.29 ×10 <sup>-13</sup>	0.03%	$5.85$ $ imes 10^{-8}$	$\begin{array}{c} 1.78 \\ \times 10^{-11} \end{array}$	0.03%
67	$1.14 \times 10^{-14}$	$\begin{array}{c} 2.20 \\ \times 10^{-15} \end{array}$	19%	$8.43 \times 10^{-10}$	5.74 ×10 <sup>-13</sup>	0.07%	Cytotox	$3.12 \times 10^{-14}$	ŀ	Cytotox	2.91 ×10 <sup>-11</sup>	·	$4.52$ $\times 10^{-10}$	$1.42 \times 10^{-13}$	0.03%	$9.00 \times 10^{-8}$	$2.49 \times 10^{-11}$	0.03%
601	MDL: n	nethod de	tection 1	imit; cyto	otox: indu	iction wa	MDL: method detection limit; cytotox: induction was masked by	by cytotoxicity	xicity									

		Activation of PXR	YK	Activation of ER	ER	Oxidative stress response	SS	p53 response	ISC	Fish embryo toxicity	oxicity
	°*	13		9		13		4		12	
	2,3,7,8- Tetrachlorodibenzo- <i>p</i> - dioxin (TCDD)	SR 12813†		17β-Estradiol	ol	tert-Butylhydroquinone (tBHQ)	inone	Mitomycin	.9	3,4-Dichloroaniline	aniline
Ι	al $REP_i$	Chemical	$REP_i$	Chemical	$REP_i$	$Chemical^k$	$REP_i$	Chemical <sup>k</sup>	$REP_i$	Chemical <sup>1</sup>	$REP_i$
Daidzei Terbuthyls	$^{1a}$ 3.33 $\times 10^{-6}$	Bezafibrate <sup>d</sup>	$6.21 \\ \times 10^{-4}$	Benzophenone- 3 <sup>g</sup>	7.04 $\times 10^{-7}$	2,4- Dinitrophenol	0.06	2,4- Dinitrophenol	$1.09 \times 10^{-3}$	2,4- Dinitrophenol	0.49
Terbuthyl		Bisphenol $A^d$	$4.63 \times 10^{-3}$	Bisphenol A <sup>e</sup>	4.49 ×10 <sup>-5</sup>	2-Phenylphenol	0.08	Carbendazim	$7.35 \times 10^{-3}$	Atrazine	0.04
	$1.36$ rise $1.36$ $\times 10^{-4}$	Carbamazepine <sup>d</sup>	4.61 $\times 10^{-4}$	Bisphenol Sh	1.18 ×10 <sup>-6</sup>	Bisphenol A	0.07	Diclofenac	$1.02 \times 10^{-3}$	Caffeine	$3.41 \times 10^{-3}$
		Diclofenac <sup>d</sup>	$1.02 \times 10^{-3}$	Daidzein	6.47 ×10 <sup>-5</sup>	Carbaryl	1.33	Genistein	$1.26 \times 10^{-3}$	Carbamazepine	0.05
		Diuron <sup>e</sup>	$2.61 \times 10^{-3}$	Estrone	0.02	Chlorophene	0.11			Carbaryl	0.18
		Erythromycin <sup>e</sup>	$2.92 \times 10^{-5}$	Genistein	5.35 ×10 <sup>-4</sup>	Daidzein	0.07			Chlorotoluron	0.45
		Estrone <sup>e</sup>	$3.92 \times 10^{-3}$			Diclofenac	0.06			Diclofenac	0.13
		Isoproturon <sup>e</sup>	$1.67 \times 10^{-3}$			Genistein	0.74			Genistein	0.62
		K et oprofen <sup>d</sup>	3.59			Metolachlor	0.35			Metoprolol	1.22

Table 3: The relative effect potencies (REP<sub>i</sub>) for the detected chemicals in the different assays.

$\times 10^{-3}$	0.22	0.07	10.53		
	p-Nitrophenol	Salicylic acid	Triclosan		
	Perfluoroheptanoic 0.03 acid	Tri(butoxyethyl) 0.14 phosphate	Triclosan 0.18	Triethyl citrate 0.04	
$\times 10^{-4}$	Metolachlor <sup>f</sup> 0.10	$Prometryn^{d} \frac{8.71}{\times 10^{-3}}$	Terbuthylazine <sup>d</sup> $\frac{3.76}{\times 10^{-3}}$	Triclosan <sup>d</sup> 0.02	*REP <sub>i</sub> data from AhR CALUX using mouse hepatoma cell line
					603

<sup>a</sup>Denison et al.<sup>49</sup>; <sup>b</sup>Long et al.<sup>50</sup>; <sup>c</sup>Ghisari et al.<sup>51</sup>; <sup>d</sup>Creusot et al.<sup>15</sup>; <sup>e</sup>Creusot<sup>52</sup>; <sup>f</sup>Lemaire et al.<sup>43</sup>; <sup>g</sup>Molina-Molina et al.<sup>53</sup>; <sup>h</sup>Molina-Molina et al.<sup>54</sup>; <sup>i</sup>current study; <sup>j</sup>Pillon et al.<sup>55</sup>; <sup>k</sup>US EPA ToxCast database<sup>23</sup>; <sup>1</sup>Scholz et al.<sup>27</sup>

Tetraethyl 2-(3,5-di-tert-butyl-4-hydroxyphenyl)ethenyl-1,1-bisphosphonate

604

607	List of	<b>Figures</b>
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608

Figure 1: A) The sum molar concentration of chemicals detected at each JDS site (black bars),
along with the number of chemicals detected at each site (red circles) and B) LC or EC values for
all samples in units of relative enrichment factor (REF).

612

613 **Figure 2**: Overview of the active and inactive detected chemicals present in the ToxCast database

614 in the oxidative stress response (ARE, red), p53 response (blue) and NF-κB response (green)

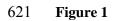
615 assays.

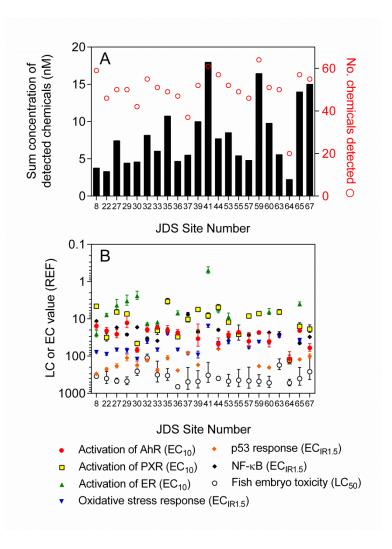
616

617 **Figure 3**: Percent of the biological effect explained by individual detected chemicals for A)

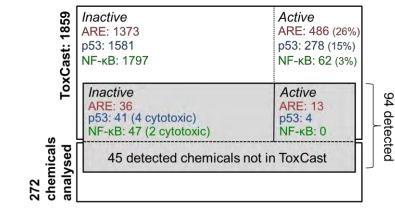
618 activation of AhR, B) activation of PXR, C) activation of ER, D) oxidative stress response, E) p53

619 response and F) fish embryo toxicity (FET)

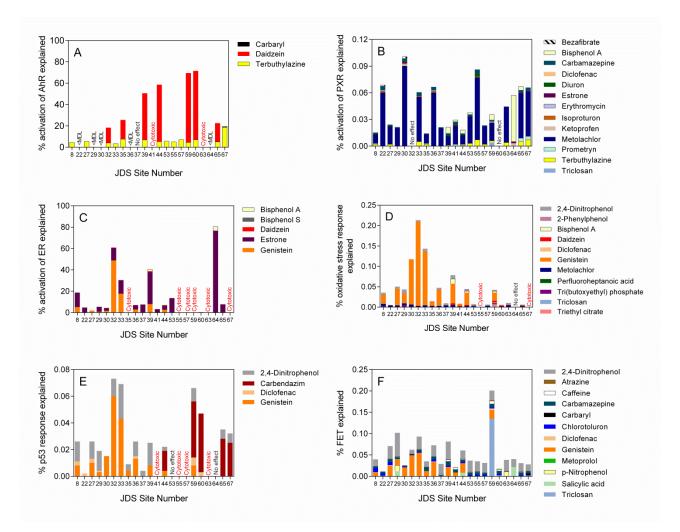












# 627 TOC Artwork

