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1	Long-term (2001-2012) observation of the modeled hygroscopic growth						
2 3	factor of remote marine TSP aerosols over the western North Pacific: impact of long-range transport of pollutants and their mixing states						
4							
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32 Abstract

In order to assess the seasonal and annual variability of long-range transported 33 anthropogenic pollutants from East Asia and their effect on the hygroscopicity and 34 precipitation process over the western North Pacific, we conducted a long-term calculation of 35 36 bulk hygroscopicity, g(90%)_{ZSR}, based on ZSR model using chemical composition data during 2001-2012 at Chichijima Island. We found sea-salts (Na⁺ and Cl⁻) are the major mass 37 fraction (65%) of total water-soluble matter followed by SO_4^{2-} (20%) and WSOM (6%). 38 Seasonal variation of g(90%)_{ZSR} showed high in summer to autumn and low in winter to 39 spring months, probably due to the influence of long-range transport of anthropogenic SO_4^{2-} . 40 41 dust, and organics from East Asia and their interaction with sea-salts by heterogeneous reactions. On the other hand, annual variations of $g(90\%)_{ZSR}$ showed a decrease from 2001 to 42 2006 and then an increase from 2007 to 2012. Interestingly, the annual variations of SO_4^{2-} 43 mass fractions showed an increase from 2001 to 2006 and then a decrease from 2007 to 2012, 44 demonstrating that SO_4^{2-} seriously suppress the hygroscopic growth of sea-salt particles over 45 the western North Pacific. This is further supported by the strong negative correlation 46 between SO_4^{2-} and $g(90\%)_{ZSR}$. Based on the MODIS satellite data, the present study 47 demonstrates that long-range transported anthropogenic pollutants from East Asia to the 48 49 North Pacific can act as efficient cloud condensation nuclei but significantly suppress the precipitation by reducing the size of cloud droplet over the western North Pacific. 50

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Keywords: Hygroscopicity, ZSR model, inorganic ions, organics, long-range transport,
 western North Pacific

55 **1. Introduction**

Particulate matter is microscopic solid or liquid suspended in the earth's atmosphere. 56 Although they represent a small portion of atmospheric mass, atmospheric particles largely 57 impact on climate and global biogeochemistry¹. Sea-salt particle is one of the most widely 58 59 distributed natural aerosols, which forms via the evaporation of sea-spray droplets produced by the bubble bursting of entrained air during whitecap formation². They are characterized as 60 non-light-absorbing, highly hygroscopic, and coarse particles. Due to its strong hygroscopic 61 62 nature, a sea salt particle can serve as efficient cloud condensation nuclei (CCN), altering cloud reflectivity, lifetime, and precipitation process³⁻⁵. Nonetheless, knowledge of 63 64 hygroscopicity of sea-salt aerosols and their impact on radiative forcing calculations are still 65 unclear⁶⁻⁷.

66 Atmospheric aerosols consist of organic and inorganic compounds with their internal 67 or external mixing. Hygroscopicity of aerosol particles is linked to the chemical composition and their mixing states. Cruz and Pandis⁸ reported that organic acids such as glutaric and 68 pinonic acids generally increase the water uptake capability of (NH₄)₂SO₄ but decrease that 69 70 of NaCl salts based on a volume fraction relative to that of inorganic salts. Choi and Chan⁹ 71 studied the effect of organic species (glycerol, succinic acid, malonic acid, citric acid, and 72 glutaric acid) on the hygroscopic behaviour of pure inorganic salts and found that the 73 presence of all these organics in the mixed particle reduces the water absorption of NaCl but 74 enhance that of (NH₄)₂SO₄ relative to that of the pure inorganic salts. Therefore, the 75 interaction between organic and inorganic species (or mixing state) in aerosol particles is 76 complex and this interaction acts either positively or negatively depending on the organic mass fractions and salt types^{8, 10}. 77

Zdanovskii-Stokes-Robinson (ZSR) relation¹¹ is the most common model, used to predict the hygroscopic growth of mixed aerosol particles using chemical composition data. Sjogren et al.¹² reported that the measured hygroscopic growth factors at the high-alpine site, Jungfraujoch, agreed well with predictions made by ZSR model. On the other hand, the hygroscopicity of ambient marine aerosols show a significant underestimation compared to the prediction estimated by numerical thermodynamic models¹³.

Anthropogenic emissions from East Asia have significantly increased over recent decades due to the rapid growth of the East Asian economies, which are implausible to decline in the next 20 years¹⁴. In addition, surface dust in spring is another distinct feature of the air quality over East Asia and the outflow regions^{15, 16}. These anthropogenic pollutants and dusts are transported from East Asia to the North Pacific by westerly winds^{17, 18} and perturb the remote marine background conditions and modify the physico-chemical
properties of sea-salt particles as well as ocean biogeochemistry by heterogeneous reactions¹⁹.
However, there is still knowledge gap on how long-range transported anthropogenic
pollutants affect the precipitation process over the western North Pacific.

93 Chichijima Island is a remote marine site in the western North Pacific and is 94 geographically located in the outflow region of Asian dusts and anthropogenic pollutants 95 from East Asia during winter and spring, whereas pristine marine air masses dominate over the island in summer and autumn²⁰. Therefore, this site is scientifically very important for 96 97 studying the long-range atmospheric transport of pollutants and mineral dusts and their influence on the precipitation process; however, studies on long-term observations of 98 transported aerosols over the western North Pacific are limited^{20, 21}. Moreover, there is no 99 study on the long-term observation of hygroscopic growth factors from the western North 100 101 Pacific. In this study, we investigate seasonal and annual variation of the bulk hygroscopic growth factor, g(90%)_{ZSR}, derived from ZSR model. We also discuss the impact of long-102 103 range atmospheric transport of anthropogenic pollutants and their mixing states on marine 104 bulk hygroscopicity and precipitation process over the western North Pacific during the study 105 period of 2001-2012.

106 The hygroscopic growth of fine and coarse particles should have different pictures 107 because of different chemical compositions and sizes, thus, their atmospheric importance is 108 dissimilar. Fine size particles are important for CCN formation, while coarse particles are 109 playing a major role on the radiative impacts of aerosol particles. However, physico-chemical 110 processes (coagulation and condensation as well as heterogeneous reactions) could make fine 111 particles to be larger particles in the water mediated atmosphere (under high RH conditions). 112 As a result, background conditions of the atmosphere may perturb, especially over the marine atmosphere. In the present study, the estimated hygroscopic growth factors from TSP 113 114 aerosols would better elucidate about the mixing state of the above mentioned aerosol 115 particles and their climatic effects.

116 **2. Methods**

117 **2.1. Aerosol sampling**

118 Aerosol (TSP) samples were collected on pre-combusted (450 °C, 3 hours) quartz 119 filter (20 x 25 cm, Pallflex 2500QAT-UP) at 5 m above the ground level of Satellite Tracking 120 Centre of Japan Aerospace Exploration Agency (JAXA, elevation 254 m) in Chichijima 121 Island ($27^{\circ}04'$ N; $142^{\circ}13'$ E) using a high volume air sampler with a flow rate of 1 m³ min⁻¹ 122 during 2001-2012²⁰. Aerosol samples were collected on a weekly basis, because of less local pollution at Chichijima Island. After sampling, filters were put in a clean glass jar with a
Teflon-lined screw cap and stored at -20 °C.

125 **2.2. Analysis of Chemical species**

126 For water-soluble inorganic species, a filter cut of 20 mm in diameter from each filter 127 was extracted with 10 mL of organic-free ultra pure water (>18.2 MQcm) under 128 ultrasonication and filtrated through disk filters (Millex-GV, 0.22 µm, Millipore). These 129 filtrated extracts were analyzed for inorganic species using an ion chromatography (761 130 Compact IC, Metrohm) and results were reported in elsewhere (Boreddy and Kawamura, 131 2015). For water-soluble organic carbon (WSOC), a punch of 20 mm was extracted with 20 132 mL of organic-free ultra pure water under ultrasonication and filtrated through a disk filters 133 (Millex-GV, 0.22 µm, Millipore). These extracts were analyzed for WSOC using Total Carbon Analyzer (TOC-Vcsh, Shimadzu, Japan)²². 134

135 **2.3. The ZSR relation**

136 The Zdanovskii-Stokes-Robinson (ZSR) model¹¹, which relates the hygroscopicity of 137 mixture to that of the individual components at the same relative humidity (RH). The 138 hygroscopic growth factor of aerosol particle can be predicted from the growth factors of 139 individual components of the aerosol composition and their respective volume fractions (ϵ) 140 using ZSR relation as,

141
$$g(RH)_{ZSR} = \left(\sum_{i} \varepsilon_{i} g_{i} (RH)^{3}\right)^{\frac{1}{3}}$$
(1)

where $g(RH)_{ZSR}$ is the growth factor of the mixed particle, $g_i(RH)$ and ε_i are the growth factor of individual component and their respective volume fraction, respectively. The volume fractions are calculated by assuming that the dry aerosol is composed of sodium chloride, ammonium sulfate, ammonium chloride, sodium nitrate, sodium sulfate and water-soluble organic matter. Rest of the volume fractions is neglected because of its low abundance to WSM.

148 The ZSR model assumes that the particles are spherical and their mixing behaviour is 149 ideal and also hygroscopic growth of the organic and inorganic components is independent.

150 **3. Results and discussion**

151 **3.1 Validation of bulk g(90%)**_{ZSR}

In order to validate the retrieved growth factor, $g(90\%)_{ZSR}$, of TSP aerosols obtained from ZSR model, we compared $g(90\%)_{ZSR}$ with measured growth factor, $g(90\%)_{HTDMA}$, using HTDMA during 2001-2003 over the same sampling site⁷ as shown in Fig. 1. We found that $g(90\%)_{ZSR}$ are strongly correlated with $g(90\%)_{HTDMA}$ with a correlation coefficient (R²) of 156 0.80 and this correlation is statistically significant at the 99% confidence level (probability 157 value less than 0.001). It can be seen from Fig. 1 that the difference in standard deviation 158 (standard error) between the $g(90\%)_{ZSR}$ and $g(90\%)_{HTDMA}$ is 0.04 (0.006). We also found that 159 $g(90\%)_{ZSR}$ are overestimated by on average 15%, probably due to the presence of marine 160 organic compounds and the subsequent formation of less hygroscopic organic salts through 161 aqueous phase reactions⁷. As a result, $g(90\%)_{HTDMA}$ may be decreased due to the internal 162 mixing of chemical species and subsequent formation of less water-soluble salts.

163 Interestingly, this discrepancy in growth factor showed higher in spring followed by 164 winter and lowest in summer, indicating that interactions of water-soluble organics with dust particles (for example, formation of CaC₂O₄) can suppress the hygroscopic growth of 165 166 particles in the outflow regions of Asian dust in spring. This point will be discussed more in the following section 3.4. This is further supported by the previous studies^{7, 13, 23-26}, which 167 168 reported that ambient marine aerosol particles with less than 30% organic mass show a 15% 169 decrease in the hygroscopic growth factor. This perturbation adds significant uncertainty to 170 the radiative balance calculations. However, in the present study, we found that contribution 171 of WSOM to total water-soluble matter (WSM) is less than 30% (on average 6%). Therefore, 172 it is reasonable to use ZSR model to predict the hygroscopic growth factors over the western 173 North Pacific.

174 **3.2.** Temporal and seasonal trends in major mass fractions of WSM and g(90%)_{ZSR}

Fig. 2 presents the temporal variation of major chemical mass fractions (Cl⁻, Na⁺, 175 SO_4^{2-} , NO_3^{-} and WSOM) of WSM and the g(90%)_{ZSR} of TSP aerosols collected at Chichijima 176 Island in the western North Pacific during 2001-2012. Sea-salt components (i.e., Na⁺ and Cl⁻) 177 are found to compose a major mass fraction (range: 0.00-0.88 and 0.03-0.41, respectively), 178 whose abundances are on average more than three times higher than SO_4^{2-} (0.00-0.77), ten 179 times higher than WSOM (0.00-0.57), and sixteen times higher than NO_3^- (0.00-0.14) mass 180 181 fractions. The g(90%)_{ZSR} values ranged from 1.63 to 2.28 with an average of 2.07±0.09 182 during the study period of 2001-2012. Although there is no clear increasing or decreasing temporal trends in all the species during the study period, a similar variation can be seen 183 between the $g(90\%)_{ZSR}$ and sea-salt components, indicating that the hygroscopicity of 184 185 Chichijima aerosols is mainly controlled by sea salt particles.

186 Interestingly, anti-correlations were found between the $g(90\%)_{ZSR}$ and WSOM, $SO_4^{2^-}$, 187 and NO_3^- . These results indicate that anthropogenic pollutants, water-soluble organic matter 188 and their mixing state can alter the hygroscopicity of sea-salt particles through the heterogeneous reactions, in particular, during the period of continental outflow over thewestern North Pacific. These points will be discussed in more detail in the following sections.

The seasonal variations in g(90%)_{ZSR} and major chemical mass fractions of total 191 192 WSM are shown in Fig. 3a-f during 2001-2012. The vertical hinges represent data points 193 from lower to upper quartile (i.e., 25th and 75th percentiles). The whiskers represent data 194 points from the 5th to 95th percentiles. We found a clear seasonal variation in all the 195 chemical mass fractions, which are clearly reflected in the $g(90\%)_{ZSR}$. These seasonal 196 differences are verified by *t*-test. These results show that these differences are statistically 197 significant with two tailed p values of less than 0.001. As illustrated in Fig. 3a, monthly means of g(90%)_{ZSR} are high in autumn followed by summer and scarce in spring and winter 198 199 months, probably due to the atmospheric processing associated with dechlorination that may 200 decrease the growth factor over the western North Pacific, especially during the period of 201 continental outflow.

202 It is noteworthy that the similar seasonal pattern can be seen in the mass fractions of sea-salt particles (see Fig. 3b and c). On the other hand, SO_4^{2-} and NO_3^{-} mass fractions are 203 maximized in spring followed by winter, probably due to the long-range atmospheric 204 205 transport of anthropogenic pollutants from East Asia and minimized in summer and autumn 206 months (see Fig. 3d and e). Mass fractions of WSOM are higher in summer followed by 207 spring and lower in autumn (see Fig. 3f). We again found an opposite trend between the mass fractions of Cl⁻ and SO₄²⁻ and similar trend between the g(90%) and Cl⁻ mass fraction, 208 demonstrating that SO_4^{2-} seriously suppresses the hygroscopic growth of sea-salt particles 209 along with WSOM via heterogeneous reactions over the western North Pacific. This point 210 211 will be discussed in more details in the following sections.

212 **3.3.** Annual trends in major mass fractions of WSM and g(90%)_{ZSR}

Fig. 4a-d presents the annual variation of $g(90\%)_{ZSR}$ and mass fractions of selected 213 214 components for different seasons during the period of 2001-2012. It can be clearly seen from Fig. 4a that g(90%)_{ZSR} showed a clear annual trend with a decrease from 2001 to 2006 and 215 216 then an increase from 2007 to 2012 in winter and autumn. In spring, we couldn't find any clear annual trend, probably due to the influence of organo-metallic interaction during long-217 range transport of Asian dusts⁷. On the other hand, the annual variation of mass fractions of 218 SO_4^{2-} showed an increase from 2001 to 2006 and a decrease from 2007-2012 (p<0.005). Lu et 219 al.²⁷ reported that the emission rate of SO₂ over East Asia slowed down around 2005 and 220 221 began to decrease after 2006. The reduction in SO₂ emissions in China was mainly due to the 222 wide usage of Flue-Gas Desulfurization (FGD) devices in power plants. Similar annual pattern can also be seen in the mass fractions of NO₃⁻. Mass fractions of WSOM showed a clear annual trend (p<0.005) in winter with an increase from 2001 to 2006 and a decrease from 2007-2012 (see Fig. 4d). Based on all these results, we demonstrate that although WSOM often acts to decrease the hygroscopicity over the western North Pacific, anthropogenic SO_4^{2-} and NO_3^{-} that are long-range transported from East Asia also suppress the hygroscopic growth of sea-salt particles during winter and spring months.

229 3.4. A relation between g(90%)_{ZSR} and chemical mass fractions of WSM

230 Chemical compositions and their mixing states are crucial for modelling studies on the hygroscopic²⁸ and optical properties²⁹ of aerosol particles, which provide the information 231 about their chemical aging³⁰. Fig. 5 shows the scatter plots between the $g(90\%)_{ZSR}$ and mass 232 233 fractions of different chemical species and/or components for different seasons during 2001-234 2012. Their corresponding Pearson correlation coefficient (R) values are reported in Table 1. 235 We found an excellent positive correlation (range: 0.87 to 0.92) between Cl⁻ mass fractions and g(90%)_{ZSR} for all seasons, demonstrating that the growth factors at Chichijima Island are 236 mainly controlled by sea-salt particles as shown in Fig. 5a. Mass fractions of Mg²⁺ show 237 good correlations with $g(90\%)_{ZSR}$ in winter (0.56), followed by summer (0.44) and spring 238 (0.34), but weakly correlated in autumn (0.18) (Fig. 5h). These results indicate that Mg^{2+} 239 also can partially contribute to the hygroscopicity because Mg²⁺ largely comes from the 240 ocean rather than continental sources²⁰. We also found a strong positive correlation (0.73 to 241 0.86) between Cl⁻/Na⁺ mass fraction ratio and $g(90\%)_{ZSR}$ in all seasons (see Fig. 5j). These 242 243 results demonstrate that the hygroscopicity at Chichijima Island is mainly controlled by the 244 sea-salt particles and atmospheric processing associated with chloride depletion that may suppress the hygroscopic growth of marine aerosol particles. 245

In contrast, SO_4^{2-} and WSOM showed strong negative correlations (-0.70 to -0.90 and 246 -0.6 to 0.82) with $g(90\%)_{ZSR}$, especially during the period with an influence of continental 247 outflow from East Asia (winter and spring). On the other hand, mass fractions of NH_4^+ and 248 NO_3^- also negatively correlated with g(90%)_{ZSR}. These results demonstrate that mixing of 249 anthropogenic pollutants (such as SO₂ and NO_x) with inorganic or organic species that can 250 251 convert the high hygroscopic salts (NaCl) into low hygroscopic salts ((NH₄)₂SO₄, Na₂SO₄, 252 NaNO₃, etc.) by heterogeneous reactions (see reactions R1 to R6) and other physical 253 mechanisms such as coagulation and condensation during long-range atmospheric transport. 254 Moreover, in spring, Asian dusts can interact with organics and leads to the formation of less hygroscopic salts such as calcium oxalate⁷ or ammonium oxalate. 255

This is further supported by the negative correlation (0.25) between $g(90\%)_{ZSR}$ and 256 Ca^{2+} during the spring. We also found good correlation between NH_4^+ and oxalic acid during 257 winter (R=0.75; p<0.0001) and spring (R=0.54; p<0.001), suggesting the formation of 258 259 ammonium oxalate over the western North Pacific. This correlation is more significant in 260 winter than spring, probably due to the long-range atmospheric transport of biomass burning aerosols from East Asia. Reid et al.^{31, 32} documented that NH4⁺ was significantly correlated 261 with organic species, such as oxalic acid rather than SO_4^{2-} in the regional haze dominated by 262 biomass burning smoke in Brazil. Our recent study³⁰ also supported the formation of 263 264 ammonium oxalate in biomass burning aerosols collected at Tanzania, East Africa. Hence, it is likely that biomass burning may be a significant source of formation of ammonium oxalate. 265 266 This result is further supported by significant negative correlations between $g(90\%)_{ZSR}$ and oxalic acid during winter (-0.63) and spring (-0.43) (see Table 1). 267

Ma and He³³ reported that formation of calcium nitrate will also lower the 268 hygroscopicity of particles by mixing with oxalic acid (see reaction R6). The internal mixing 269 270 of oxalic acid with sea-salt particles also leads to lower the hygroscopicity during the longrange atmospheric transport³⁴. This kind of mixing in the dust particles is very important for 271 explaining the mass transfer process in the atmosphere³⁵ and direct and indirect climate 272 forcing of dust particles^{36, 37}. Therefore, it is important to note that the declined $g(90\%)_{ZSR}$ in 273 winter and spring was probably due to the formation of less hygroscopic particles or salts 274 275 through heterogeneous reactions that play a significant role in aged mineral dust particles and 276 anthropogenic pollutants, resulting in a decrease of the hygroscopic growth of marine aerosol 277 particles. The possible acid displacement reactions, which occur over the western North 278 Pacific, are given below.

279

$$280 \qquad 2 \operatorname{NaCl} + \operatorname{H}_2 \operatorname{SO}_4 \to \operatorname{Na}_2 \operatorname{SO}_4 + \operatorname{HCl} \qquad (R1)$$

$$281 \qquad \text{NaCl} + \text{HNO}_3 \rightarrow \text{NaNO}_3 + \text{HCl} \qquad (R2)$$

282
$$NaCl + CH_3SO_3H \rightarrow CH_3SO_3Na + HCl$$
 (R3)

$$283 \qquad 2 \operatorname{NaCl} + \operatorname{H}_2C_2O_4 \to \operatorname{Na}_2C_2O_4 + 2 \operatorname{HCl} \qquad (R4)$$

284
$$NaCl + CH_3COOH \rightarrow CH_3COONa + HCl$$
 (R5)

$$285 \qquad Ca(NO_3)_2 + H_2C_2O_4 \rightarrow CaC_2O_4 + 2 \text{ HNO}_3 \quad (R6)$$

286

Previous studies have shown that more hygroscopic group of particles is often observed in remote marine environment⁶. Growth factors of this group (g(90%)>1.85) are larger than those of pure ammonium sulfate (g(90%)=1.70) and pure ammonium bisulphate (g(90%)=1.79). During the strong influence of continental outflow of air masses, an uptake of H₂SO₄ (or HNO₃) by sea-salt particles followed by the subsequent release of HCl, for example, converts some of NaCl (g(90%)=2.4) into sodium sulfate (g(90%)<1.8) (or sodium nitrate) by the above mentioned reactions (R1 or R2). The formation of sodium sulfate (or sodium nitrate) will result in a significant reduction in the growth factor of sea-salt particles and thus affect the CCN and precipitation process over the western North Pacific.

Fig. 6 presents the schematic diagram, which shows the changes in microphysical and chemical properties of sea-salt particles under the influence of two distinct air masses that aged over the western North Pacific. The observation site, Chichijima, is a remote Island in the western North Pacific located in the outflow region of Asian dust and anthropogenic pollutants from East Asia, especially from China. This sampling site exists in the boundary of the westerly (in winter and spring) and easterly (in summer and autumn) wind regimes.

302 During the long-range atmospheric transport of air masses by westerly winds, sea-salt 303 particles are seriously modified by internal mixing of anthropogenic pollutants and dust 304 particles through heterogeneous reactions as mentioned above (R1-R6). This mixing state, 305 together with long-range transport of pollutants from East Asia, can depress the water uptake 306 properties of sea-salts due to the formation of less hygroscopic particles and can increase the 307 number of smaller size CCN particles, leading to the formation of smaller size of cloud 308 droplets, an enhanced reflection of solar radiation (less transmission), and an increase in 309 cloud lifetime. Moreover, smaller size particles absorb more radiation and suppress the 310 precipitation process by cloud evaporation or simply cloud burning.

311 On the other hand, easterly (trade) winds with low wind speed carry pristine air 312 masses to the sampling site especially in summer and autumn (see Fig. 6). In this process, 313 sea-salt particles uptake more water acting as giant CCNs, which could be nucleated into 314 larger cloud particles. By coagulation and condensation, less absorptive and more transmitant 315 cloud particles grow into rain droplets hence enhancing precipitation process. Although the 316 present study focus on the hygroscopic growth factor of bulk aerosols, the implications of these results (such as a reduction in water uptake of sea-salt particles due to internal mixing 317 318 of sulfate/nitrate particles and organics) directly affect the CCN number concentrations by 319 decreasing the size of particles and hence increases the number of smaller sized particles due 320 to the continental influence (see Fig.6). However, measurements of particle number size 321 distributions are required to further confirm these results.

322 further MODIS satellite These results are supported by the aqua 323 (http://gdata1.sci.gsfc.nasa.gov) products and precipitation (downloaded from 324 http://www.jma.go.jp) at Chichijima Island in the western North Pacific during 2001-2012. Figs. 7a-b illustrate the seasonal variations of $g(90\%)_{ZSR}$ and precipitation at Chichijima 325 326 Island during 2001-2012. The precipitations at Chichijima clearly increased from winter to 327 autumn with lower precipitation values during westerly and higher values in trade wind 328 regimes. Fascinatingly, we found similar seasonal trend in $g(90\%)_{ZSR}$, although we found a 329 significant reduction in the growth factors during spring, which may be due to the influence 330 of Asian dusts. These results demonstrate that westerly winds transported anthropogenic 331 pollutants that suppress the hygroscopicity as well as precipitation over the western North 332 Pacific by enhancing the CCN, COD (Cloud Optical Depth, a proxy for cloud cover) and other aerosol products, such as AOD (Aerosol Optical Depth at 550 nm, a proxy for fine 333 334 mode aerosols), and particulate organic carbon concentrations, as shown in Figs. 7c-f, 335 respectively. All the satellite-derived products are abundant during winter and spring 336 (westerly) whereas scarce in summer and autumn (trade wind).

We also found a good opposite seasonal trend between the modeled $g(90\%)_{ZSR}$ and 337 338 particulate organic carbon concentrations, indicating that organics in aerosols suppress the 339 hygroscopic growth over the western North Pacific. Further, a robust opposite seasonal 340 variation was found between precipitation and particulate organic carbon concentrations, demonstrating that particulate organics repress the precipitation by boosting up the CCN 341 342 concentrations. Based on the above-mentioned results, it is reasonable to conclude that the 343 long-range atmospheric transport of pollutants from East Asia are significantly enhancing the 344 fine size CCN concentration, but seriously suppress the hygroscopic growth of aerosols 345 affecting the precipitation over the western North Pacific.

Previous studies suggested that polluted aerosols suppress deep convective 346 precipitation by decreasing cloud droplet size and delaying the onset of freezing³⁸⁻⁴⁰. In the 347 modeling study by Cui et al.⁴¹, the less precipitation is caused by drops evaporating more 348 rapidly in the high aerosol case⁴², which eventually reduces ice mass and hence precipitation. 349 In contrast, Koren et al.⁴³ found, based on satellite data analyses, that increases in aerosol 350 351 abundance correlated with a higher rate of rainfall in the tropics, subtropics and mid-latitudes. They also mentioned that the similar trends were seen across different locations and 352 353 environmental conditions, suggesting a link between increased aerosol levels and more intense rainfall. However, their observation period is limited to specific season (June-August 354 355 2007) and they didn't discuss the effect of long-range transport of pollutants and their relation 356 to precipitation. The present study should contribute to clarifying the above uncertainty, that is, anthropogenic chemical species in the midlatitudes can depress the rainfall intensity in theopen ocean.

Therefore, the above results highlight the complexity of the aerosol-cloudprecipitation system and demonstrate the sign for the global change in precipitation due to changes in the concentrations and composition of aerosols. It is important to note that microphysical processes can only change the temporal and spatial distribution of precipitation while the total amount of precipitation can only change if evaporation from the surface changes.

365 **3.5** Atmospheric implication and conclusions

The purpose of this study has been to observe the long-term (2001-2012) variations in 366 367 hygroscopic properties of TSP aerosols and their relation to the precipitation over the western 368 North Pacific. We found the bulk hygroscopicity of remote marine aerosols is mainly 369 controlled by sea-salt particles (NaCl) because of its high abundances in total WSM. 370 However, anthropogenic pollutants that are transported from East Asia are internally mixed 371 with sea-salt particles by heterogeneous reactions, leading to a suppression of the 372 hygroscopic growth thus affecting the CCN properties especially during winter and spring. 373 On the other hand, WSOM may often suppress the hygroscopic growth of remote marine 374 aerosols over the western North Pacific. We also found that there is a serious impact of Asian 375 dust and their interactions with organic/inorganic salts on hygroscopic growth.

Although there is no clear decadal trend in $g(90\%)_{ZSR}$, we found a systematic seasonal 376 377 variation with high values in autumn and summer and low values in spring and winter. In winter and autumn, the annual variation of $g(90\%)_{ZSR}$ decreased from 2001 to 2007 and then 378 increased continuously toward 2012. At the same time, SO_4^{2-} mass fractions increased from 379 2001 to 2006 and then decreased to 2012 probably due to the decrease in SO₂ emissions in 380 381 China after 2006. These results demonstrate that anthropogenic sulfate seriously suppress the 382 hygroscopic growth of marine aerosol particles through heterogeneous reactions over the 383 western North Pacific, especially during long-range atmospheric transport. This is further 384 supported by the regression analyses between $g(90\%)_{ZSR}$ and mass fractions of WSM during 385 the study period. The present study also demonstrates that long-range atmospheric transport 386 of pollutants from East Asia can enhance the CCN activity but suppress the precipitation by reducing the droplet size over the western North Pacific. However, size-segregated 387 388 measurements may provide a different picture of hygroscopic properties, depending on the 389 particle size, for variable chemical compositions.

In this study, we found that these modified sea salts have profound consequences on their evolving physicochemical properties especially on hygroscopic behaviour by increasing cloud droplet number concentration and reducing the cloud droplet size and thus reducing precipitation process. Therefore, in the marine atmosphere, the above consequences should add significant uncertainty to the climatic and radiative models.

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Table 1. Pearson correlation coefficient (R) values between $g(90\%)_{ZSR}$ and chemical mass fractions for different seasons during 2001-2012 at Chichijima Island over the western North Pacific.

Species	g(90%) _{ZSR}					
	Winter (n=132)	Spring (n=138)	Summer (n=132)	Autumn (n=143)		
Cl	0.92^{*}	0.90^{*}	0.87^{*}	0.88^*		
NO ₃ ⁻	-0.58*	-0.41*	-0.34*	-0.51*		
SO_4^{2-}	-0.88*	-0.82*	-0.70*	-0.90*		
Na^+	0.77^{*}	0.74^{*}	0.70^{*}	0.66^{*}		
$\mathrm{NH_4}^+$	-0.7*	-0.71*	-0.58*	-0.57*		
K^+	-0.49*	-0.33*	-0.39*	-0.44*		
Ca^{2+}	-0.15 [#]	-0.25#	- 0.19 [#]	-0.23#		
Mg^{2+}	0.56^{*}	0.34*	0.44^{*}	$0.18^{\#}$		
WSOM	-0.82*	-0.60*	-0.67*	-0.53*		
Oxalic acid	-0.64*	-0.43*	-0.39*	-0.57*		
Cl ⁻ /Na ⁺	0.86^{*}	0.73^{*}	0.77^*	0.78^{*}		

*Correlation is significant at less than 0.001 levels (2-tailed) #Correlation is significant at 0.05 levels (2-tailed)



Fig. 1. Validation of modeled $g(90\%)_{ZSR}$ with measured $g(90\%)_{HTDMA}$ at Chichijima Island in the western North Pacific during 2001-2003.



Fig. 2. Temporal variations in the $g(90\%)_{ZSR}$ and mass fractions of water-soluble matter at Chichijima Island in the western North Pacific during 2001-2012. Mass fractions are calculated using the data of ions and WSOC are taken from Boreddy and Kawamura (2015).







Fig. 4. Annual variations in the g(90%)_{ZSR} and mass fractions of WSM for different seasons. The data of ions and WSOC are from Boreddy and Kawamura (2015). Mass fractions are calculated using the data of ions and WSOC are taken from Boreddy and Kawamura (2015).



Fig. 5. Scatter plot, showing relation of the $g(90\%)_{ZSR}$ with (a-i) mass fractions of WSM, (j) Cl/Na⁺ 7 mass fraction ratio, and (k) C₂ di (μ gm⁻³) over the western North Pacific during 2001-2012. Mass 7 fractions are calculated using the data of ions and WSOC are taken from Boreddy and Kawamura 7 (2015).

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Fig. 6. Schematic diagram, showing changes in physico-chemical properties of sea-salt particles under the influence of two distinct aged air masses over the western North Pacific.



Fig. 7. Box and whisker plots, showing seasonal variations in (a) precipitation, (b) $g(90\%)_{ZSR}$, (c-f) MODIS Aqua satellite products of CCN number concentrations, cloud optical depth, aerosol optical depth, and particulate organic carbon, respectively for the region [140-145°E; 25-30° N] over the western North Pacific during the period of 2001-2012.