

Low to High Spin-State Transition Induced by Charge Ordering in Antiferromagnetic YBaCo₂O₅

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(Received 4 August 1999)

The oxygen-deficient double perovskite YBaCo₂O₅, containing corner-linked CoO₅ square pyramids as principal building units, undergoes a paramagnetic to antiferromagnetic spin ordering at 330 K. This is accompanied by a tetragonal to orthorhombic distortion. Below 220 K orbital ordering and long-range Co²⁺/Co³⁺ charge ordering occur as well as a change in the Co²⁺ spin state from low to high spin. This transition is shown to be very sensitive to the oxygen content of the sample. To our knowledge this is the first observation of a spin-state transition induced by long-range orbital and charge ordering.

PACS numbers: 75.50.Ee, 61.12.-q

The recent discovery of colossal magnetoresistance in manganese oxides has sparked a renaissance in the study of spin, charge, and orbital ordering in transition metal oxides. A related phenomenon, not observed in the manganates due to strong Hund's rule coupling, is a spin-state transition. These are observed in the isostructural cobaltates, such as LaCoO₃, which has a low spin ⁶t_{2g}⁰e_g magnetic ground state. On warming a spin-state transition to an intermediate ⁵t_{2g}¹e_g spin state is observed near 100 K [1,2]. At 500 K a second spin-state transition involving high and low spin Co³⁺ occurs. The specifics of these thermally activated low to high spin-state transitions are still controversial [3,4]. They have also been shown to prevail in dilute hole-doped systems such as La_{1-x}Sr_xCoO₃ [5]. We report here on structurally related LBaCo₂O_{5+x} materials, whose structures are derived from perovskites via ordering of the rare earth (*L*) and Ba cations into layers along *c* and removing oxygen exclusively from the *L* layer [6,7]. This creates an apically connected double layer of corner-sharing CoO₅ pyramids. For *x* > 0 the extra oxygen ions are incorporated into the *L* layer of LBaCo₂O_{5+x} to form disordered octahedra along the *c* axis. We have synthesized LBaCo₂O_{5.00} and studied the thermal evolution of its structure and properties using synchrotron x-ray [8] and neutron powder diffraction [9] as well as conductivity and magnetization measurements. This composition was selected since equal concentrations of Co²⁺ and Co³⁺ should maximize the Coulomb stabilization energy of the charge-ordered state. The samples were synthesized from nanoscale precursors obtained by liquid mixing of citrate melts [10]. They were single phase according to conventional powder x-ray diffraction measurements. High-resolution synchrotron x-ray powder diffraction at 300 K confirmed this for the *L* = Gd sample, whereas for *L* = Y approximately 10% of the same phase with slightly enhanced oxygen content was detected. This minority phase

is thought to be a result of surface oxidation during storage at room temperature. The results we report below clearly demonstrate the occurrence of orbital and charge ordering, accompanied by a low to high spin-state transition *upon cooling*.

Above 330 K YBaCo₂O₅ is paramagnetic and tetragonal (space group *P4/mmm*). At approximately 330 K an antiferromagnetic (AFM) phase transition occurs as indicated by a cusp in the magnetic susceptibility measurements (Fig. 1) and the appearance of magnetic superstructure reflections in neutron powder diffraction data. Synchrotron x-ray powder diffraction measurements show that this magnetic transition occurs simultaneously with a tetragonal-to-orthorhombic (*T-O*) structural phase transition. Furthermore, as mentioned above, the minority tetragonal phase persists below the *T-O* transition. The two-phase coexistence was confirmed by ultra-high-resolution diffraction measurements using a crystal analyzer. We stress that the observation of two phases in synchrotron x-ray powder diffraction experiments, with an order of magnitude better resolution than available with a standard laboratory x-ray diffractometer, can be traced to minute oxygen-content variations in the order of magnitude around *x* = 0.01 [11]. In the presence of strong lattice coupling these small compositional variations can lead to phase coexistence, which can be observed in high-resolution experiments. This behavior beautifully illustrates how sensitive phase transitions within these systems are to minute variations in the oxygen stoichiometry.

At room temperature there is no evidence for any long-range charge ordering. However, upon cooling long-range charge ordering is detected below 200 K, by the appearance of the ($\frac{1}{2}$ 0 2), ($\frac{1}{2}$ 1 2), ($\frac{1}{2}$ 1 0), and ($\frac{1}{2}$ 1 1) superlattice reflections indicating a doubling of the unit cell along the *a* axis. These very weak superlattice reflections,

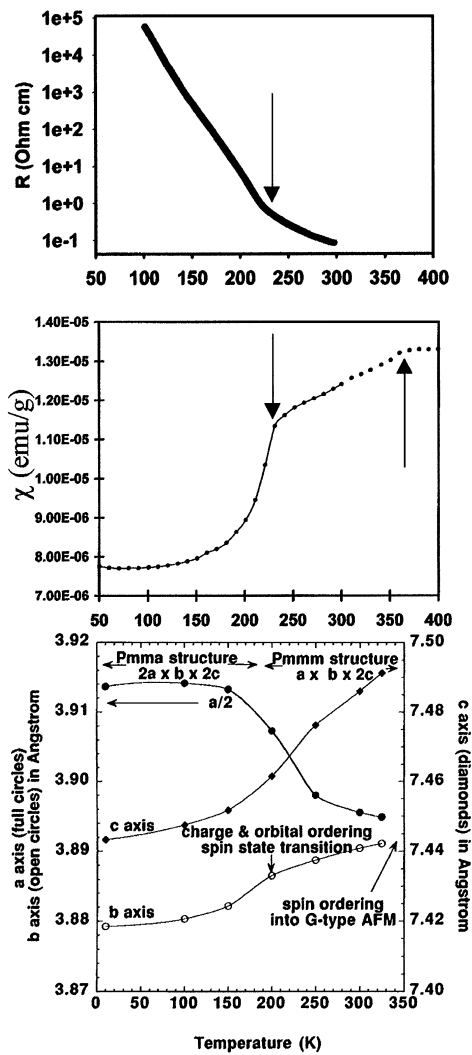


FIG. 1. Four-probe resistivity and magnetization data of YBaCo_2O_5 data indicate two phase transitions (arrows) near 350 and 220 K. In the lower panel the evolution of the lattice parameters as a function of temperature is shown. For details, see text.

which have intensities of $\sim 0.1\%$ of the strongest Bragg reflection were observed in both the position sensitive detector (PSD) and the Si(111) crystal analyzer modes. In order to explain the presence of these superlattice reflections we considered the following structural phase transformation mechanisms: (a) rotations of the square pyramidal units about [001], $\text{Co}^{2+}/\text{Co}^{3+}$ charge ordering into (b) a rocksalt arrangement, similar to that observed in YBaMn_2O_5 [12], (c) checkerboard ordering in the ab plane, inverted in the c direction, similar to $(\text{Nd}_{0.5}\text{Sr}_{0.5})\text{MnO}_3$ [11], (d) chains of Co^{2+} ions running parallel to [010] but alternating with Co^{3+} chains in the a and c directions (see Fig. 2). Table I plainly shows that only model (d) is consistent with the observed unit cell and space group. Figure 1 shows the evolution of the unit cell dimensions, magnetic susceptibility, and electrical resistivity with temperature. Near 220 K we

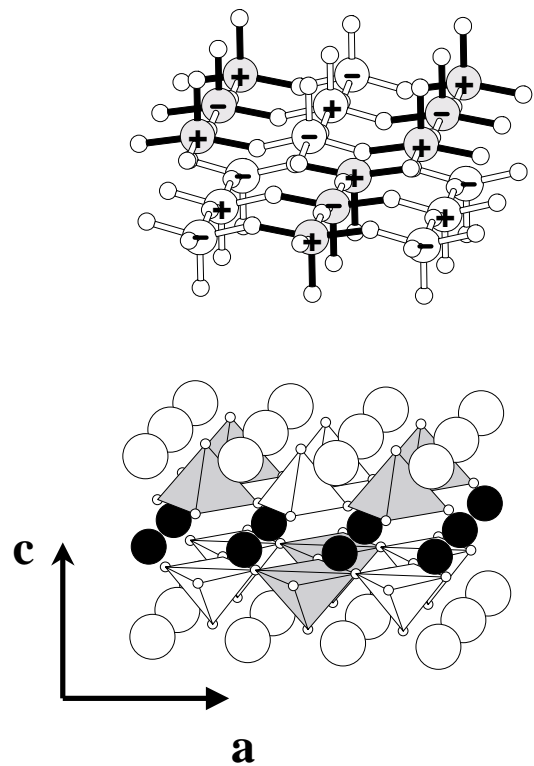


FIG. 2. The crystal and magnetic structure of YBaCo_2O_5 as determined at 10 K. The top panel shows the individual Co^{2+} (shaded spheres) and Co^{3+} (white spheres) ions and their bonds to oxygen (small white spheres). Ba and Y have been omitted for clarity. All bond lengths $> 2.05 \text{ \AA}$ are drawn in black. The plus and minus symbols denote the antiferromagnetic spin arrangement with the moments aligned parallel to the b axis. The bottom panel shows a polyhedral representation of the structure, highlighting the Co^{2+} and Co^{3+} square-pyramidal coordinations. Ba and Y ions are represented by white and black spheres, respectively.

see a pronounced upturn in the resistivity, indicative of electron localization, that would be associated with $\text{Co}^{2+}/\text{Co}^{3+}$, charge ordering. This is accompanied by a strong enhancement in the splitting of the a and b axes, which signifies the development of an orbitally ordered state. Using synchrotron x-ray data collected at low temperatures with a PSD, we refined a model of this charge-ordered state in order to confirm the charge-ordering phenomenon (see Table II and Fig. 2).

To investigate in more detail the evolution of the magnetic ordering and confirm our observation of charge ordering based on x-ray experiments, we carried out variable temperature neutron powder diffraction experiments. The results of the structural and magnetic refinements are listed in Table II. High-resolution neutron powder diffraction experiments at 400 K revealed a paramagnetic tetragonal phase. At 300 K a so-called G -type AFM structure [13] is found in which each cobalt ion couples antiferromagnetically to its six neighboring cobalt ions (see Fig. 2). The spins were located in the ab plane with

TABLE I. Structural models for low-temperature YBaCo₂O₅.^a

| Model ^b | Space group | Unit cell |
|--------------------|---------------|------------------------|
| a | <i>P4/nbm</i> | $\sqrt{2}a \times c$ |
| b | <i>P4/nmm</i> | $\sqrt{2}a \times c$ |
| c | <i>P4/mmm</i> | $\sqrt{2}a \times c$ |
| d | <i>Pmma</i> | $2a \times b \times c$ |

^aDerived using the program ISOTROPY (written by H. T. Stokes and D. M. Hatch, Department of Physics, Brigham Young University).

^bFor description of each model, see text.

a moment of 2.10(1) μ_B per cobalt ion. This corresponds quite well to the spin-only value expected for a mixed valence compound containing equal concentrations of low spin (LS) Co²⁺ (⁶*t*_{2g}¹*e*_g, *S* = $\frac{1}{2}$, $\mu_{SO} = 1.73\mu_B$ [1]) and intermediate spin (IS) Co³⁺ (⁵*t*_{2g}¹*e*_g, *S* = 1, $\mu_{SO} = 2.85\mu_B$). The very small orthorhombic splitting prevented the determination of the orientation of the moments within the *ab* plane.

The refined structural parameters obtained for the low temperature x-ray (50 K) and neutron (25 K) data are in good agreement. Bond valence calculations [14] yield valence sums of 2.69 for Co(1) and 2.02 for Co(2). These valences and the bond distances confirm the long-range charge ordering of Co²⁺/Co³⁺. However, the charge-

ordering topology is not of the NaCl- or *G*-type as one might expect from purely electrostatic considerations. Instead we have alternating long and short bonds (Co²⁺-O-Co³⁺) along the doubled *a* axis and chains of either Co²⁺-O-Co²⁺ or Co³⁺-O-Co³⁺ linkages running parallel to *b* (see Fig. 2). The magnetic structure remains *G*-type when charge ordered. However, the magnitude of the spin states is altered because of the *ordered* distribution of Co²⁺ and Co³⁺ ions, which necessarily leads to different magnetic moments. The magnetic structure now contains antiferromagnetic Co²⁺-O-Co²⁺ and Co³⁺-O-Co³⁺ chains with the moments aligned parallel to *b*. The Co³⁺ and Co²⁺ ions have refined moments of 2.7(1) μ_B and 4.2(1) μ_B , respectively. These values are in good agreement with a charge-ordered state that contains IS Co³⁺ (⁵*t*_{2g}¹*e*_g, *S* = 1, $\mu_{SO} = 2.85\mu_B$) and HS (high spin) Co²⁺ (⁵*t*_{2g}²*e*_g, *S* = $\frac{3}{2}$, $\mu_{SO} = 3.87\mu_B$). However, it is conceivable that the observed moments are reduced due to the presence of domains and a model based upon HS Co³⁺ (⁴*t*_{2g}²*e*_g, *S* = 2, $\mu_{SO} = 4.89\mu_B$) and HS Co²⁺ (⁵*t*_{2g}²*e*_g, *S* = $\frac{3}{2}$, $\mu_{SO} = 3.87\mu_B$) may apply, or as argued in the case of LaCoO₃, the ground state could have a strongly mixed character [15]. In any case the experimental data clearly indicate that YBaCo₂O₅ undergoes a spin-state transition upon cooling which differs from the one observed in LaCoO₃, and the hole-doped

TABLE II. Crystallographic parameters of YBaCo₂O₅ obtained from Rietveld refinements.

| Temp. (K) radiation | 400 Neutron | 300 Neutron | 25 Neutron | 50 X ray ^a |
|-------------------------------------|---------------------------------------|--|---|--|
| <i>R</i> _{wp} ^b | 6.32% | 5.78% | 5.96% | 10.8% |
| Space group | <i>P4/mmm</i> | <i>Pmmm</i> | <i>Pmma</i> | <i>Pmma</i> |
| <i>a</i> (Å) | 3.890 69(2) | 3.892 19(4) | 7.8235(1) | 7.832 41(6) |
| <i>b</i> (Å) | | 3.884 98(4) | 3.8746(1) | 3.879 68(3) |
| <i>c</i> (Å) | 7.4994(1) | 7.4808(1) | 7.4367(2) | 7.443 94(5) |
| Atomic positions | | | | |
| Ba | 0, 0, 0 | $\frac{1}{2}, \frac{1}{2}, 0$ | 0, 0, 0 | 0, 0, 0 |
| Y | 0, 0, $\frac{1}{2}$ | $\frac{1}{2}, \frac{1}{2}, \frac{1}{2}$ | 0, 0, $\frac{1}{2}$ | 0, 0, $\frac{1}{2}$ |
| Co(1) | $\frac{1}{2}, \frac{1}{2}, 0.2593(4)$ | 0, 0, 0.2585(4) | $\frac{1}{4}, \frac{1}{2}, 0.271(2)$ | $\frac{1}{4}, \frac{1}{2}, 0.2687(5)$ |
| Co(2) | ... | ... | $\frac{1}{4}, \frac{1}{2}, 0.755(2)$ | $\frac{1}{4}, \frac{1}{2}, 0.7509(5)$ |
| O(1) | $\frac{1}{2}, \frac{1}{2}, 0$ | 0, 0, 0 | $\frac{1}{4}, \frac{1}{2}, -0.006(1)$ | $\frac{1}{4}, \frac{1}{2}, 0.000(4)$ |
| O(2) | 0, $\frac{1}{2}, 0.3122(1)$ | 0, $\frac{1}{2}, 0.3140(9)$ | $\frac{1}{4}, 0, -0.317(1)$ | $\frac{1}{4}, 0, -0.309(3)$ |
| O(3) | ... | $\frac{1}{2}, 0, 0.3096(9)$ | $\frac{1}{4}, 0, 0.307(1)$ | $\frac{1}{4}, 0, 0.310(3)$ |
| O(4) | ... | ... | -0.0153(5), $\frac{1}{2}, 0.3105(5)$ | -0.014(2), $\frac{1}{2}, 0.3135(9)$ |
| Co(1)-O Dist. | 1 × 1.944(3) 4 × 1.9853(7) | 1 × 1.934(3) 2 × 1.986(2) 2 × 1.983(1) | 1 × 2.06(2) 2 × 1.955(2) 2 × 2.096(5) | 1 × 2.00(3) 2 × 1.964(4) 2 × 2.08(2) |
| Co(2)-O Dist. | | | 1 × 1.82(2) 2 × 2.004(4) 2 × 1.894(5) | 1 × 1.87(3) 2 × 1.991(6) 2 × 1.93(2) |

^aRefinement based on x-ray data. Data consist of 6451 points ranging from 5.5° to 70° in steps of 0.01° in 2θ. Wavelength used was 0.798 91 Å determined by a calibration with a CeO₂ standard.

^b $R_{wp} = \left\{ \sum w_i [(y_i)_{obs} - (y_i)_{calc}]^2 / \sum w_i [(y_i)_{obs}]^2 \right\}^{1/2}$; *y*_{*i*} is the background-corrected intensity at point *i*. The weights *w*_{*i*} are derived purely from counting statistics. The summation is taken over all points that contribute to Bragg intensities.

systems $\text{La}_{1-x}\text{Sr}_x\text{CoO}_3$, in three important aspects: (i) the spin-state transition occurs within a long-range AFM ordered magnetic structure, (ii) it is driven by long-range charge ordering, and (iii) the spin state changes from low spin to high spin *upon cooling* [16].

We point out that neutron powder diffraction measurements on similar earlier samples produced results different from those reported above. The pronounced splitting of the a and b axes at low temperatures was not observed, suggesting the absence of long-range charge and orbital ordering. Furthermore, the refined magnetic moment at 12 K was determined to be $2.16(2)\mu_B$ per cobalt, very similar to the value obtained at 300 K, thus indicating that no spin-state transition occurred. The absence of both charge ordering and spin-state transition can be attributed to an increased oxygen content of these samples. In our initial experiments we did not take sufficient care to prevent room temperature oxidation of the material. The subsequent change in the $\text{Co}^{2+}:\text{Co}^{3+}$ ratio destabilizes the charge-ordered state and the spin-state transition as well. This extreme sensitivity of the charge-order-induced spin-state transition to the oxygen content helps explain the above mentioned phase coexistence occurring at T_N . It appears plausible that certain regions of the sample, most likely the surface, can have an oxygen stoichiometry slightly in excess of 5.00. These regions would remain in the tetragonal state without undergoing charge ordering and the subsequent spin-state transition, as observed in bulk samples with defined oxygen stoichiometries greater than 5.00. This is corroborated by the refined magnetic moment of $2.5(1)\mu_B$ for the mixed valent cobalt in the tetragonal minority tetragonal phase at 25 K.

In summary, we have demonstrated that the low temperature structural phase transition in YBaCo_2O_5 is a spin-state transition driven by a charge and orbital ordering which occurs in a G -type AFM structure. The charge ordering topology of Co^{2+} and Co^{3+} is distinct from that observed in the isostructural manganates where $T_{CO} \geq T_N$ [12]. Furthermore, it has been shown that the incorporation of excess oxygen into the rare earth layer destroys the charge-ordering and spin-state transition, thus confirming the strong coupling between the two. The change from a LS to a HS state upon cooling, as well as the presence of AFM spin ordering, is distinctly different from the behavior observed in LaCoO_3 and $\text{La}_{1-x}\text{Sr}_x\text{CoO}_3$. To our knowledge, YBaCo_2O_5 is the first material to exhibit a spin-state transition in combination with long-range charge, spin, and orbital ordering.

The authors thank D. E. Cox, M. Strongin (BNL), and B. J. Kennedy (Sydney University) for stimulating discussions. T. V. thanks the Australian Nuclear Science & Technology Organization and, in particular, the members of the Neutron Scattering Group at Lucas Heights for their

hospitality during various visits. Research at the National Synchrotron Light Source at Brookhaven National Laboratory was carried out under Contract No. DE-AC02-98CH10886, Division of Materials Sciences, U.S. Department of Energy.

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