

Supporting information

For

**Luminescent and Proton Conducting
Lanthanide Coordination Networks Based
on a Zwitterionic Tripodal Triphosphonate**

*Montse Bazaga-García,[†] Giasemi K. Angeli,[‡] Konstantinos E. Papathanasiou,[‡] Inés R.
Salcedo,[†] Pascual Olivera-Pastor,[†] Enrique R. Losilla,[†] Duane Choquesillo-Lazarte,[§]
Gary B. Hix,[‡] Aurelio Cabeza,^{†,*} and Konstantinos D. Demadis,^{*‡}*

[‡]Crystal Engineering, Growth and Design Laboratory, Department of Chemistry,
University of Crete, Voutes Campus, Crete, GR-71003, Greece

[†]Universidad de Málaga, Departamento de Química Inorgánica, Campus Teatinos s/n
29071-Málaga, Spain

[§]Laboratorio de Estudios Cristalográficos, IACT-CSIC, Granada, Spain

[‡]School of Science and Technology, Nottingham Trent University, Clifton Lane,
Nottingham, NG11 8NS, United Kingdom

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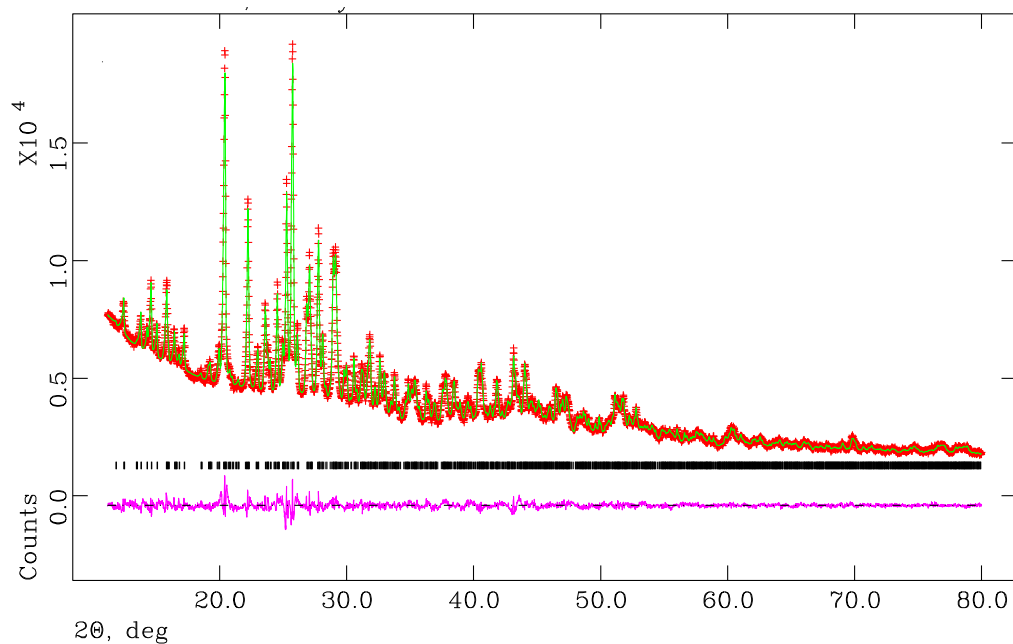


Figure S1. Rietveld plot for $[\text{Ce}_2(\text{H}_3\text{NMP})_2(\text{H}_2\text{O})_4]\cdot 4.5\text{H}_2\text{O}$. Experimental data (red scatters); the best-fit profile (upper trace) and the difference pattern (lower trace) are drawn as solid green and magenta lines, respectively. Black vertical bars indicate the angular positions of the allowed Bragg reflections.

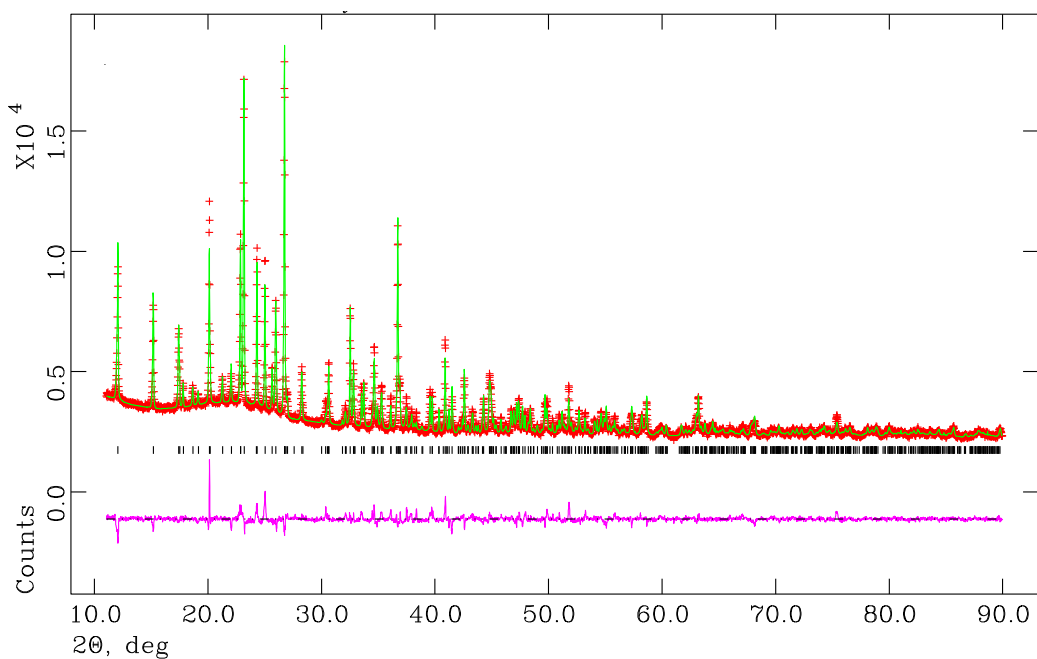


Figure S2. Rietveld plot for $[\text{Sm}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl}\cdot 2\text{H}_2\text{O}$. Experimental data (red scatters); the best-fit profile (upper trace) and the difference pattern (lower trace) are drawn as solid green and magenta lines, respectively. Black vertical bars indicate the angular positions of the allowed Bragg reflections.

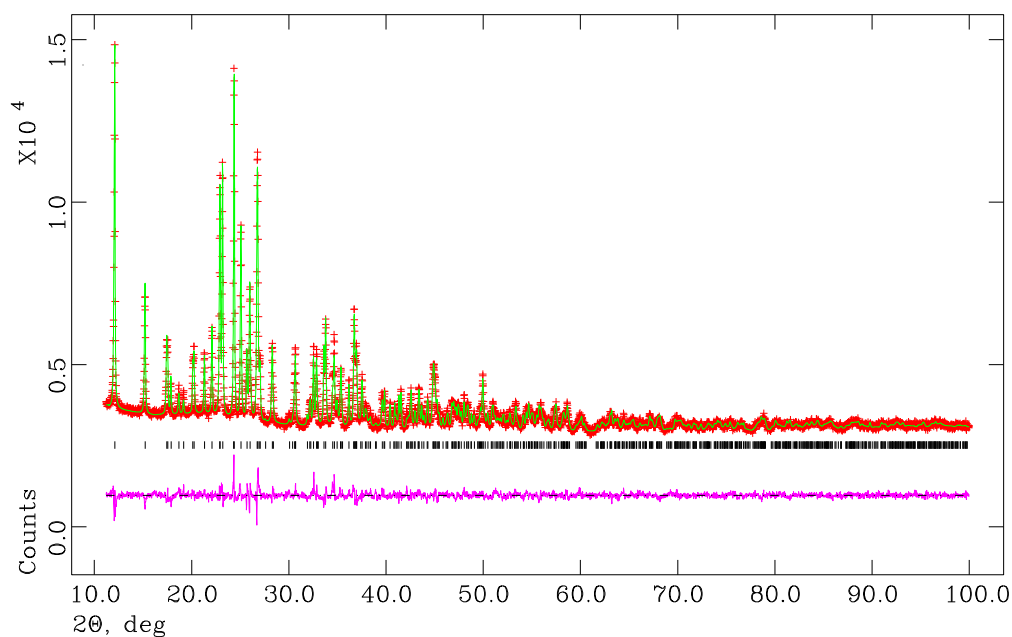


Figure S3. Rietveld plot for $[\text{Eu}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl}\cdot 2\text{H}_2\text{O}$ Experimental data (red scatters); the best-fit profile (upper trace) and the difference pattern (lower trace) are drawn as solid green and magenta lines, respectively. Black vertical bars indicate the angular positions of the allowed Bragg reflections.

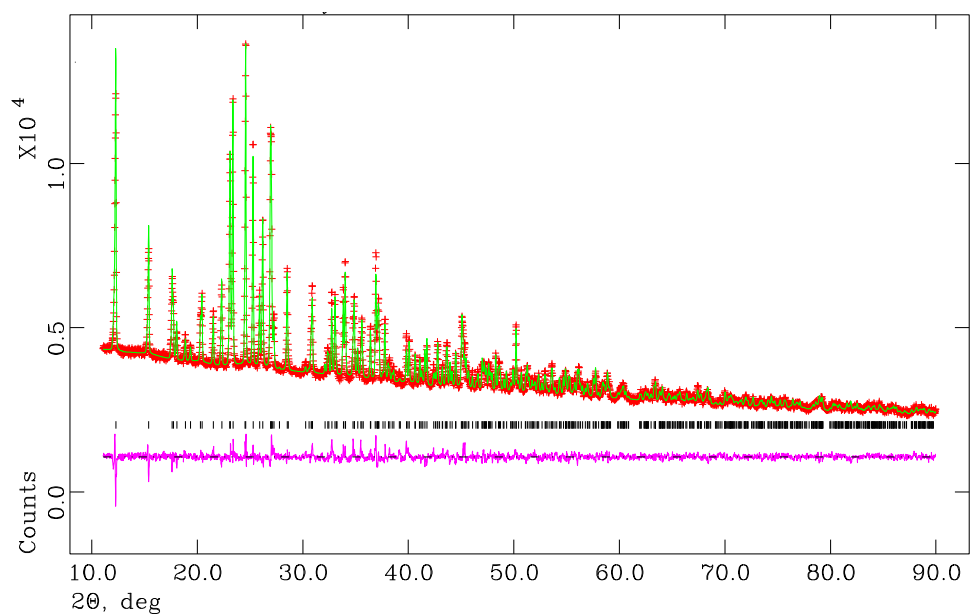


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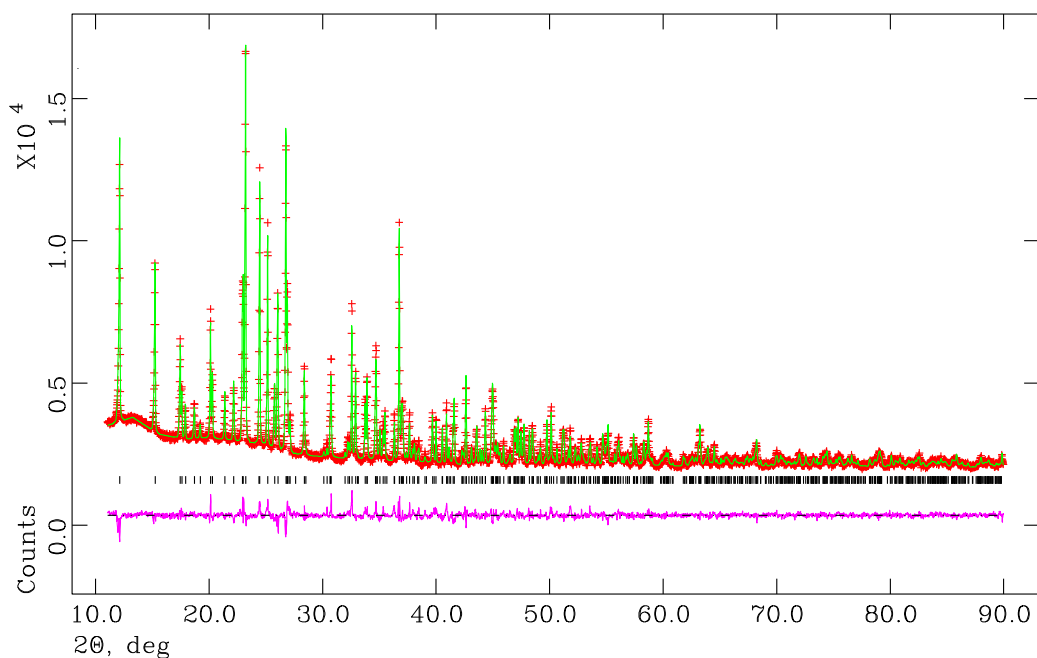


Figure S5. Rietveld plot for $[\text{Tb}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl}\cdot 2\text{H}_2\text{O}$. Experimental data (red scatters); the best-fit profile (upper trace) and the difference pattern (lower trace) are drawn as solid green and magenta lines, respectively. Black vertical bars indicate the angular positions of the allowed Bragg reflections.

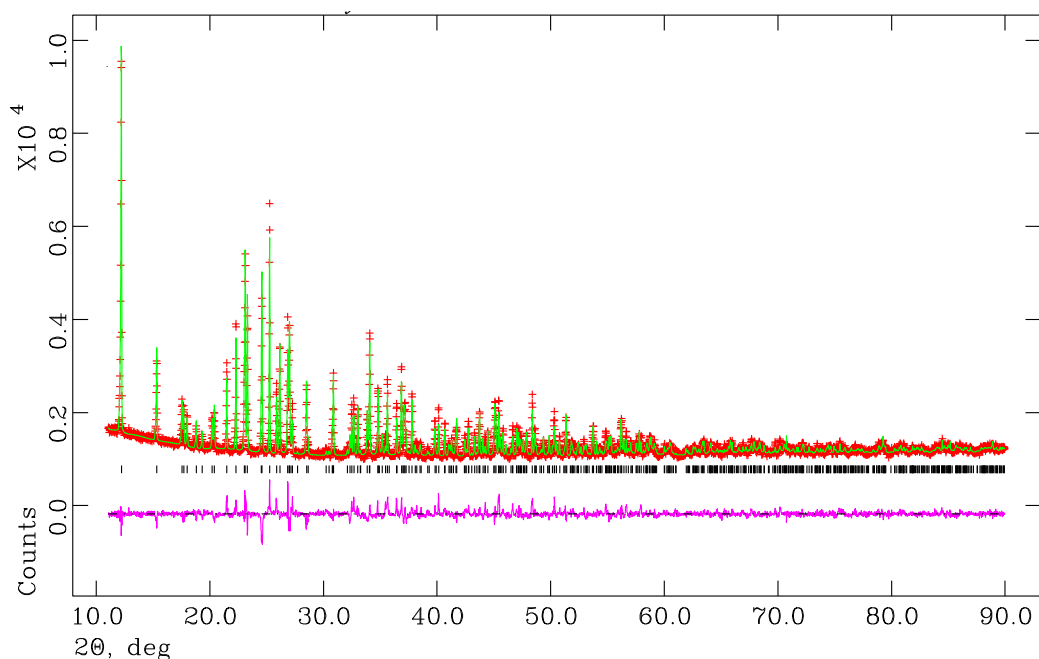


Figure S6. Rietveld plot for $[\text{Dy}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl}\cdot 2\text{H}_2\text{O}$. Experimental data (red scatters); the best-fit profile (upper trace) and the difference pattern (lower trace) are drawn as solid green and magenta lines, respectively. Black vertical bars indicate the angular positions of the allowed Bragg reflections.

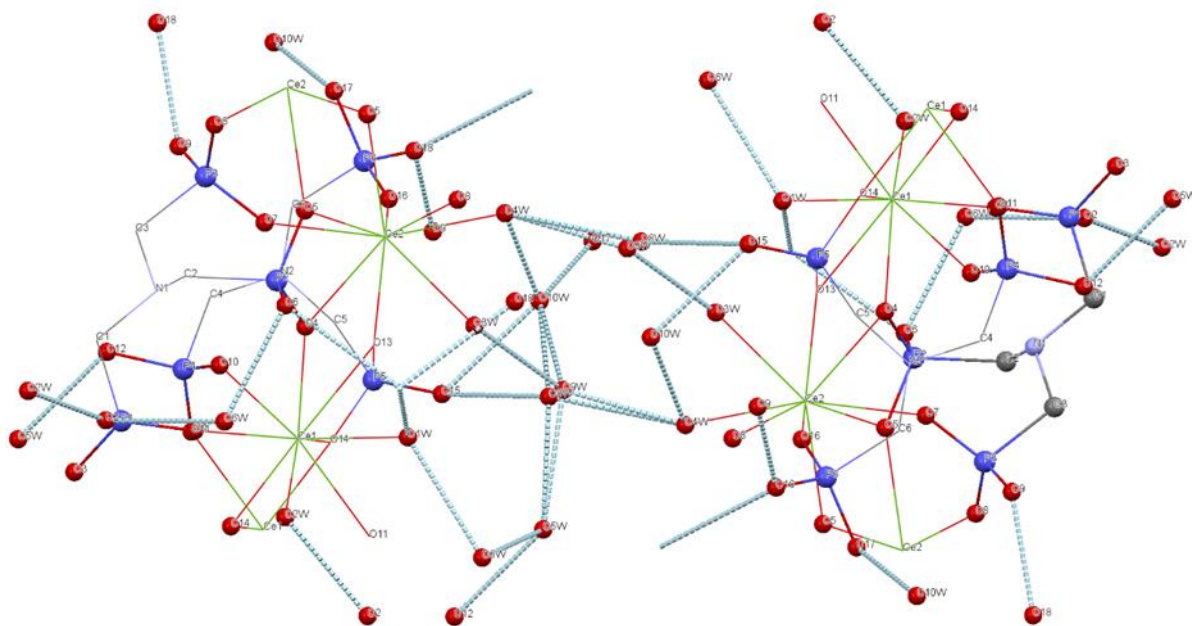


Figure S7. Extended hydrogen bonding interactions observed in the crystal structure of $[\text{Ce}_2(\text{H}_3\text{NMP})_2(\text{H}_2\text{O})_4] \cdot 4.5\text{H}_2\text{O}$.

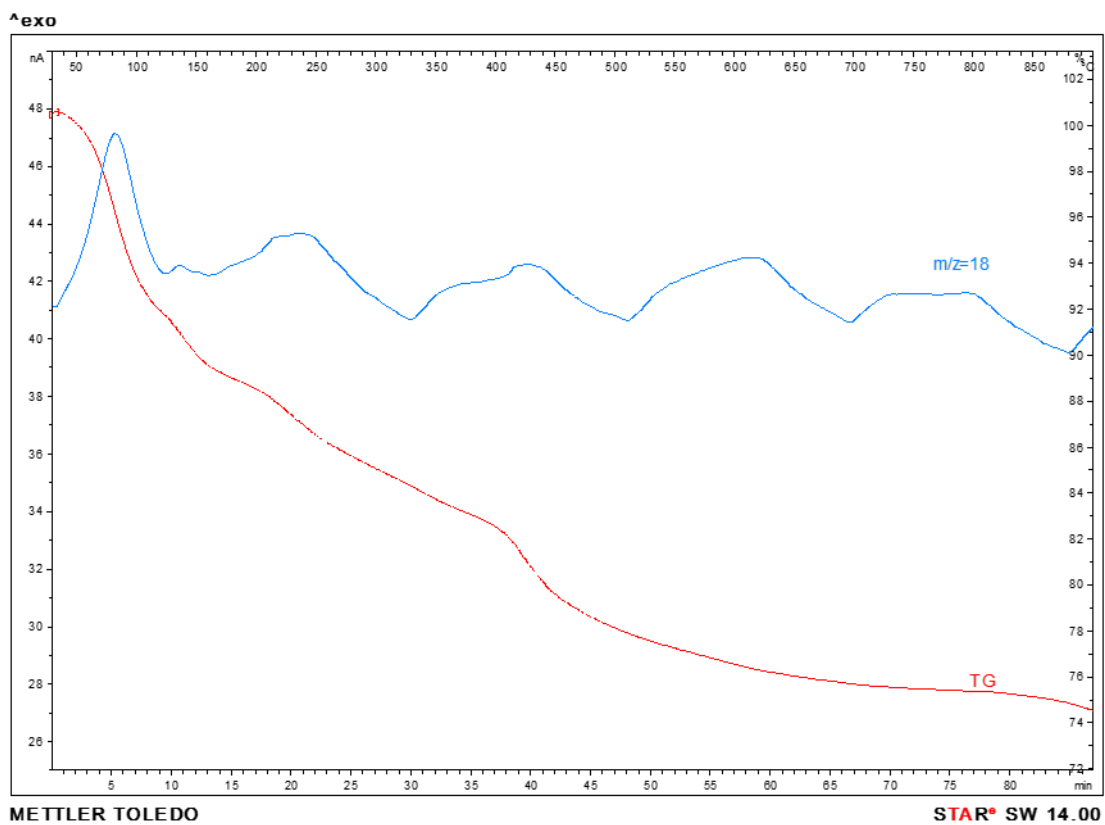


Figure S8. TG-MS curves for $[\text{Gd}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl}\cdot 2\text{H}_2\text{O}$ and H_2O ($m/z = 18$).

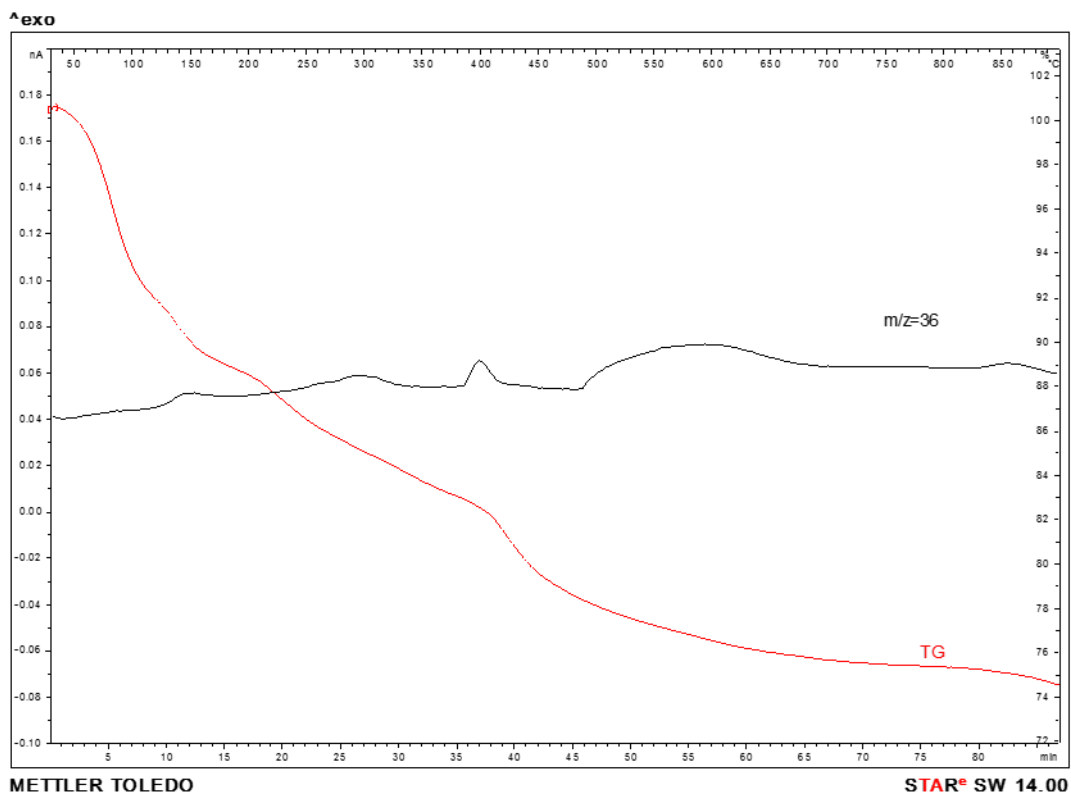


Figure S9. TG-MS curves for $[\text{Gd}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl}\cdot 2\text{H}_2\text{O}$ and HCl ($m/z = 36$).

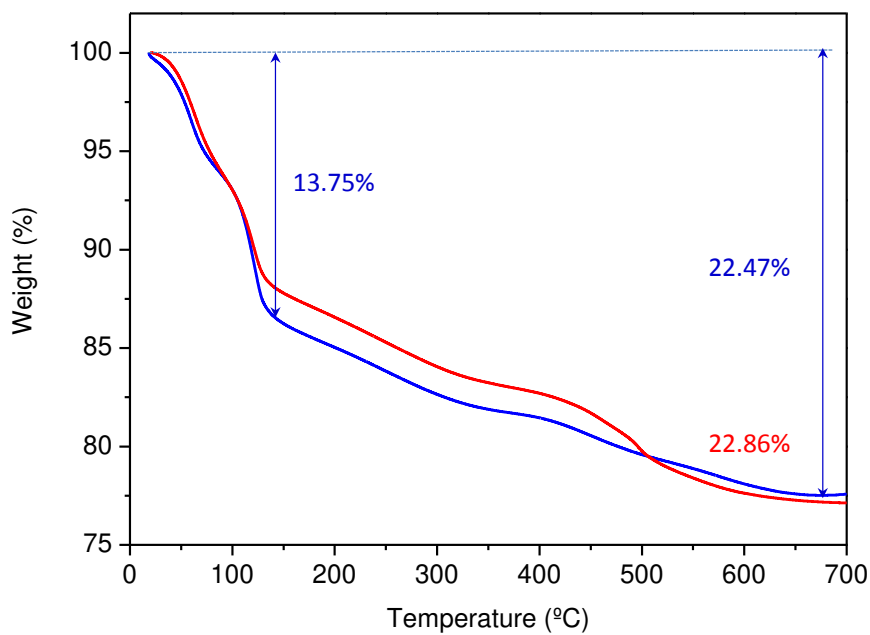


Figure S10. TGA curves for $[\text{Ce}_2(\text{H}_3\text{NM})_2(\text{H}_2\text{O})_4]\cdot 4.5\text{H}_2\text{O}$ as-synthesized (blue) and post-impedance analysis (red).

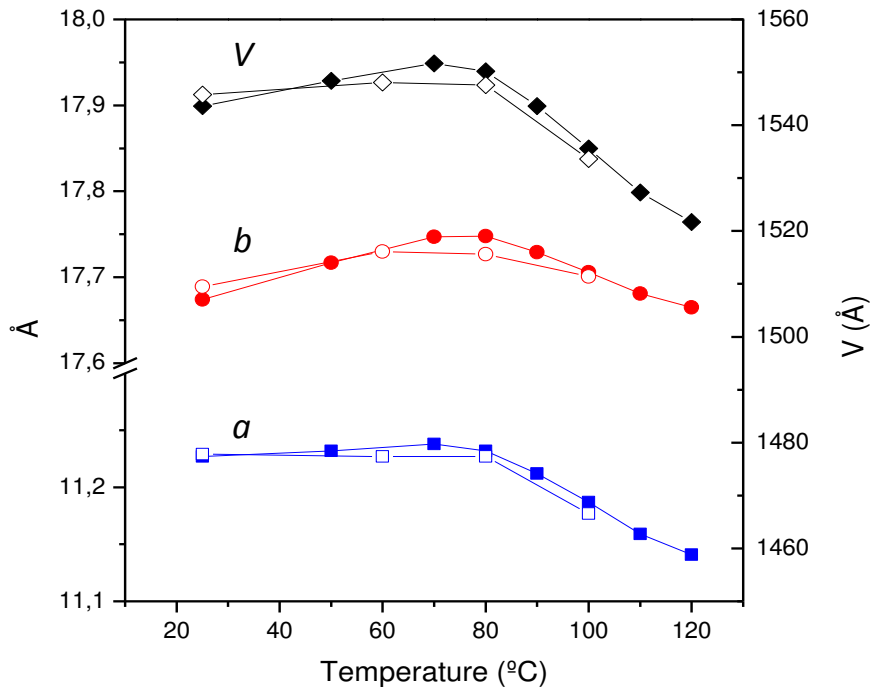
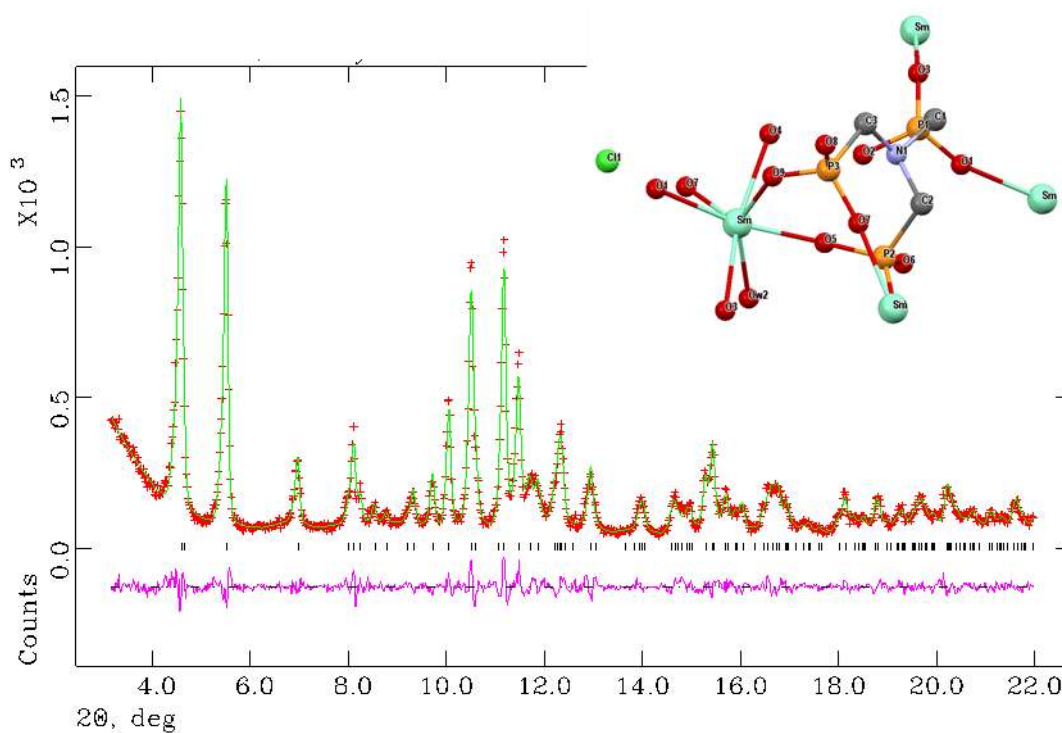


Figure S11. Variation of unit cell parameters and volume with the temperature for $[\text{Sm}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl}\cdot 2\text{H}_2\text{O}$ (open and filled symbols denote decreasing and increasing temperatures, respectively).



Figures S12. Rietveld plot of $[\text{Sm}(\text{H}_4\text{NMP})(\text{H}_2\text{O})]\text{Cl}$ heated at 120 °C and 10% RH. The powder pattern was collected using Mo $K_{\alpha 1}$ radiation ($\lambda = 0.7093 \text{ \AA}$). The inset shows the coordination environment around Sm^{3+} in $[\text{Sm}(\text{H}_4\text{NMP})(\text{H}_2\text{O})]\text{Cl}$. Selected crystallographic data: $a = 11.142(2) \text{ \AA}$; $b = 17.653(4) \text{ \AA}$; $c = 8.610(1) \text{ \AA}$; $\beta = 116.135(8)^\circ$; $V = 1520.4(7) \text{ \AA}^3$; s.g. $C c$; $R_{\text{WP}} = 0.0872$; $R_{\text{P}} = 0.0678$; $R_{\text{F}} = 0.0151$.

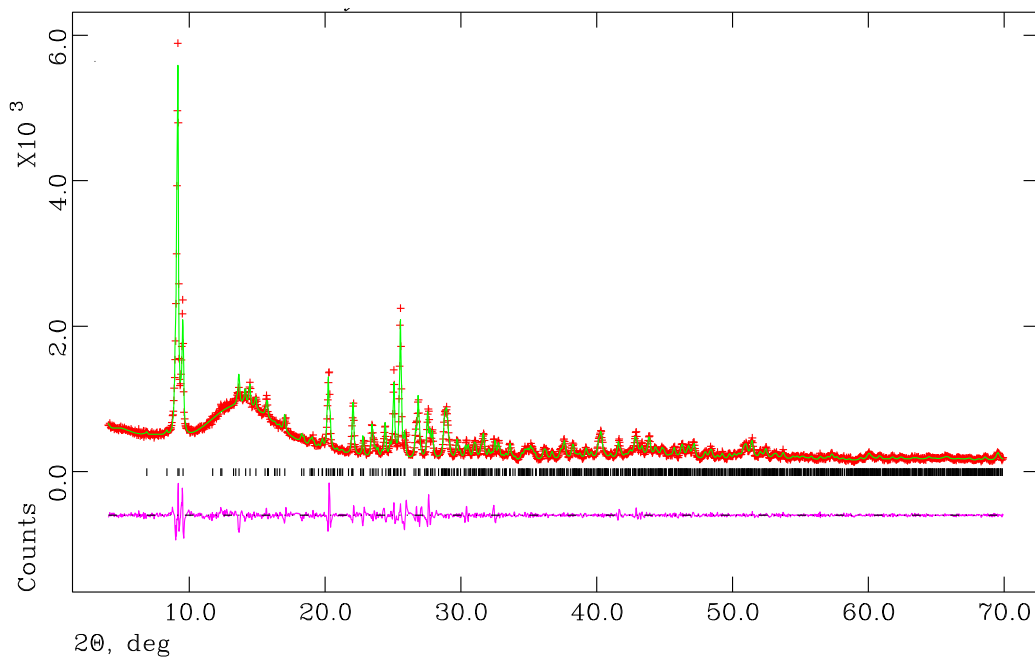


Figure S13. Le Bail fit for $[\text{La}_2(\text{H}_3\text{NMP})_2(\text{H}_2\text{O})_4] \cdot 4.5\text{H}_2\text{O}$ obtained by exposing $[\text{La}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl} \cdot 2\text{H}_2\text{O}$ at 80 °C and 95% RH for 30 h.

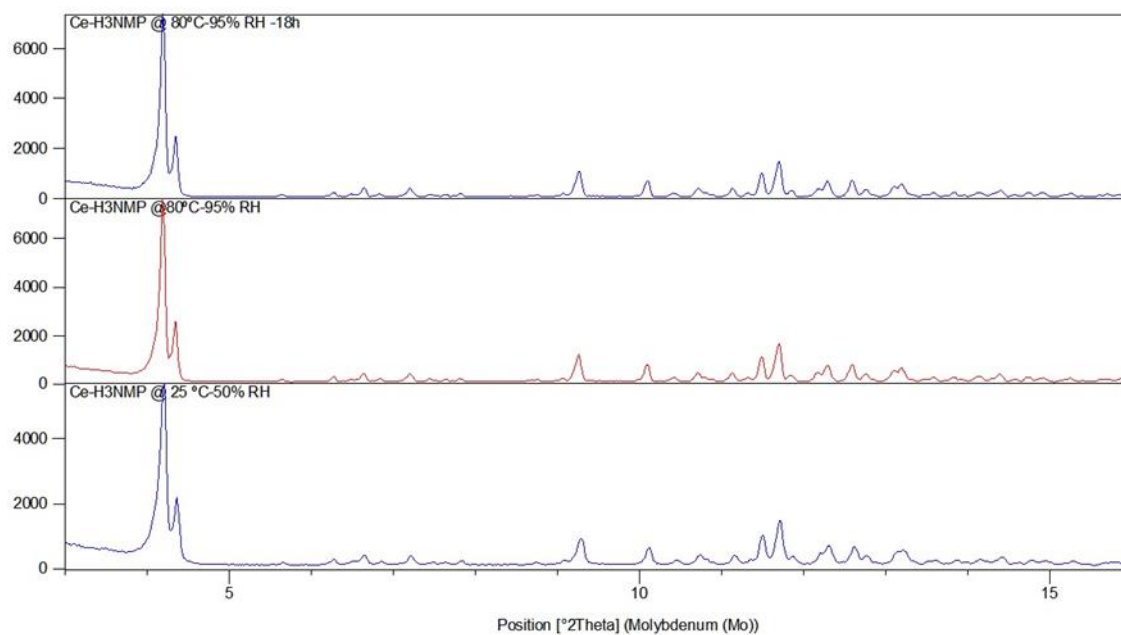


Figure S14. Powder XRD patterns for $[\text{Ce}_2(\text{H}_3\text{NMP})_2(\text{H}_2\text{O})_4]\cdot 4.5\text{H}_2\text{O}$ at: (a) 25 °C and 50% RH; (b) 80 °C and 95% RH after 1 h; and (c) 80 °C and 95 % RH after 18 h.

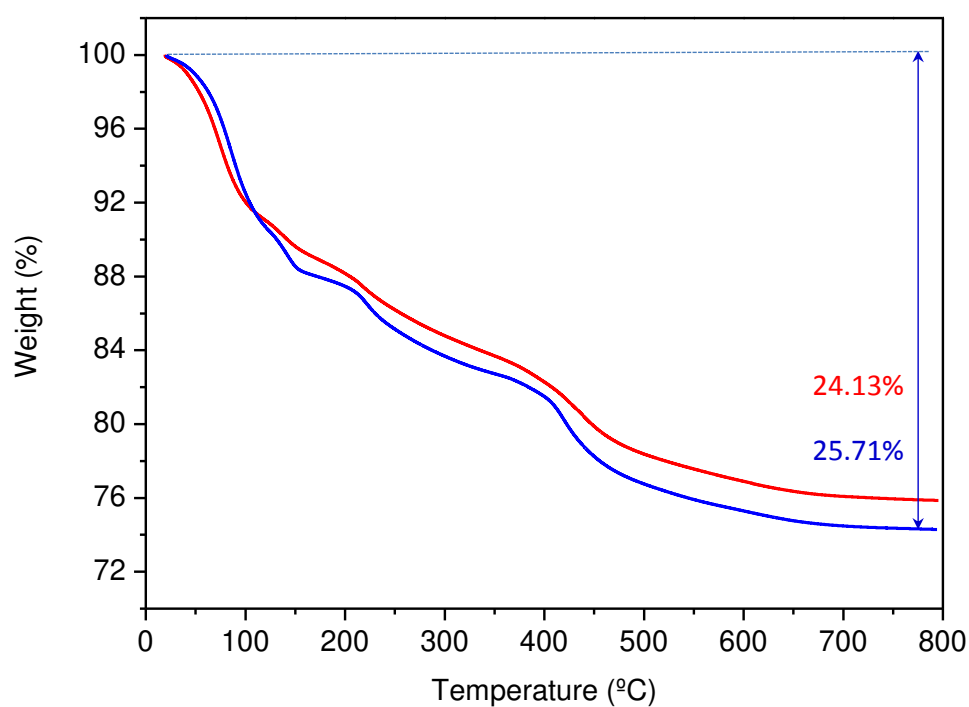


Figure S15. TGA curves of $[\text{Gd}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl}\cdot 2\text{H}_2\text{O}$ as-synthesized (blue) and post-impedance (red).

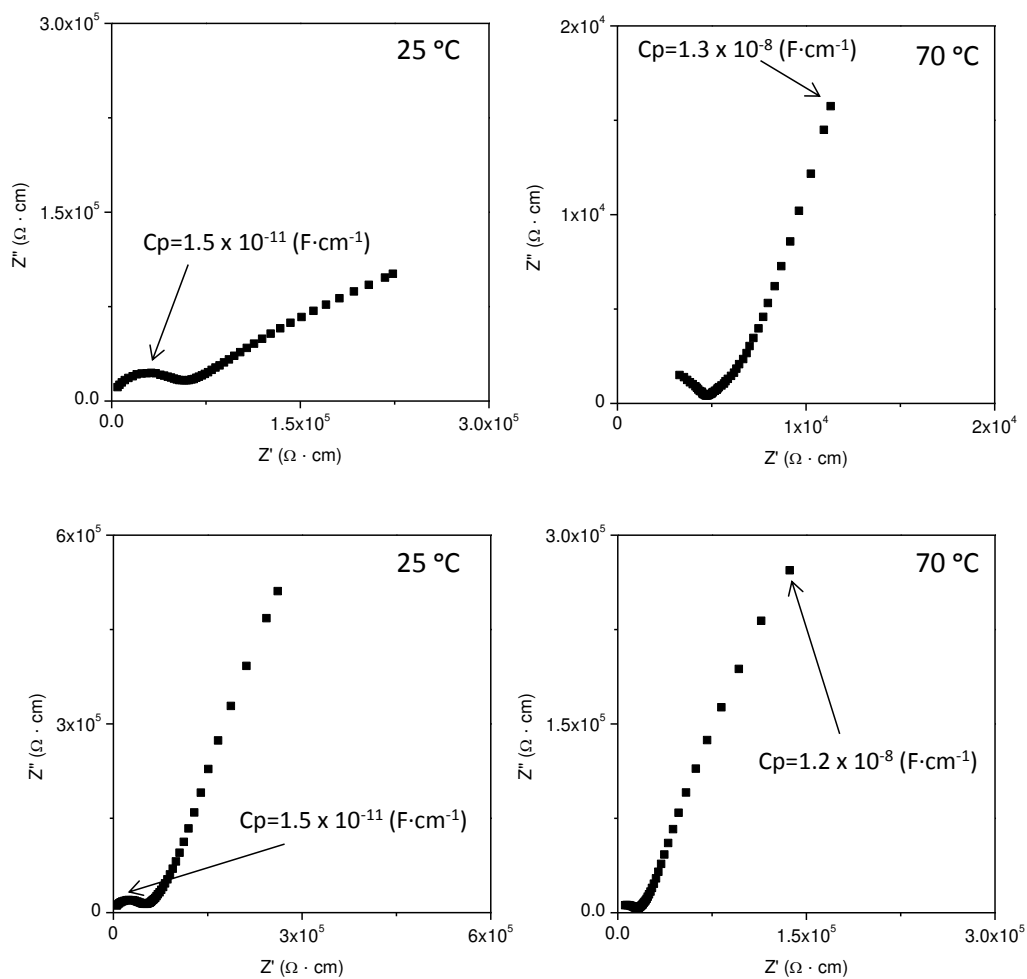


Figure S16. Complex impedance plane plots at 25 and 70 °C (95% RH) for (a) $[\text{Gd}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl} \cdot 2\text{H}_2\text{O}$ and (b) $[\text{Ce}_2(\text{H}_3\text{NMP})_2(\text{H}_2\text{O})_4] \cdot 4.5\text{H}_2\text{O}$.

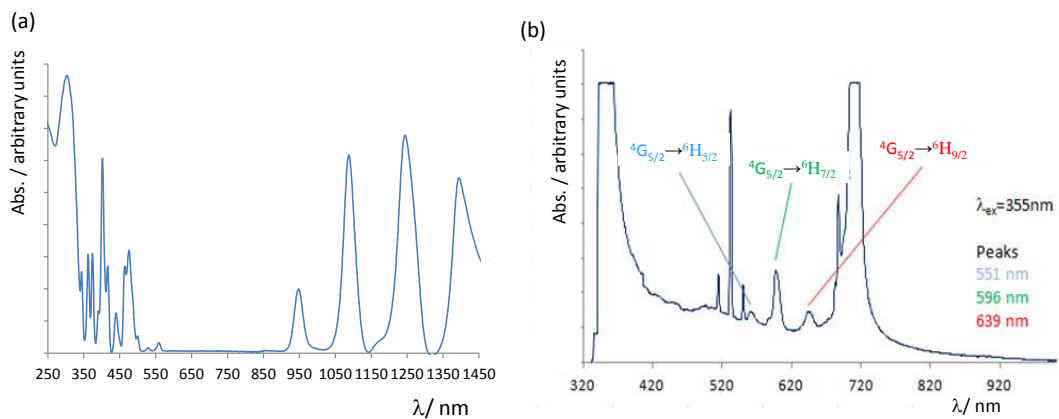


Figure S17. (a) UV-visible absorption and (b) emission spectra for **Sm-H₄NMPCI**.

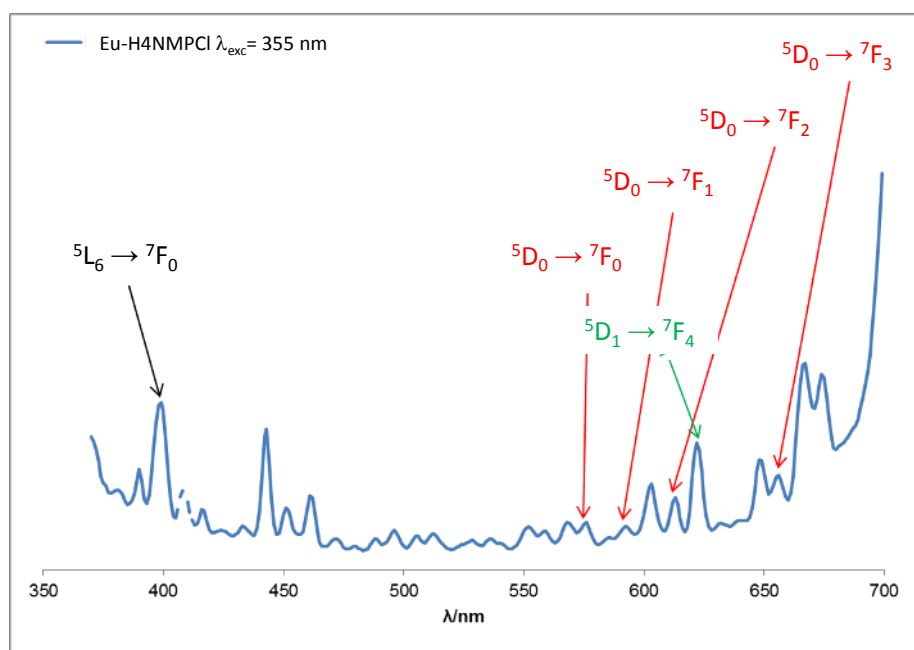


Figure S18. Emission spectrum for **Eu-H₄NMPCI**.

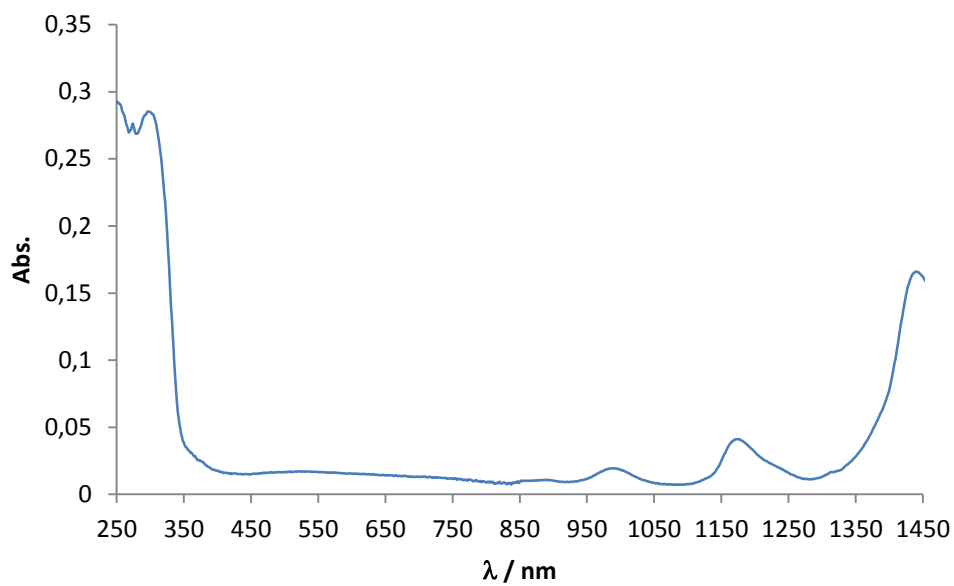


Figure S19. UV-visible absorption spectrum for **Gd-H₄NMPCl**.

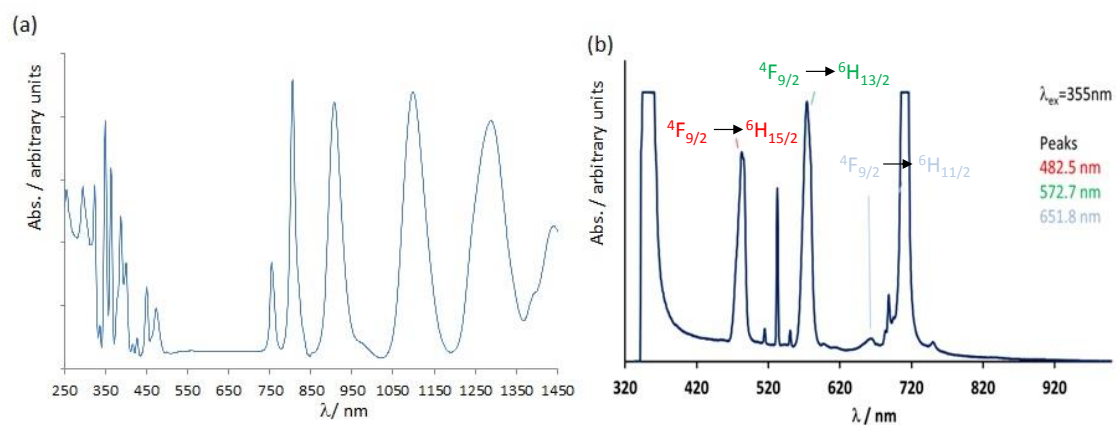


Figure S20. (a) UV-visible absorption and (b) emission spectra for **Dy-H₄NMPCl**.

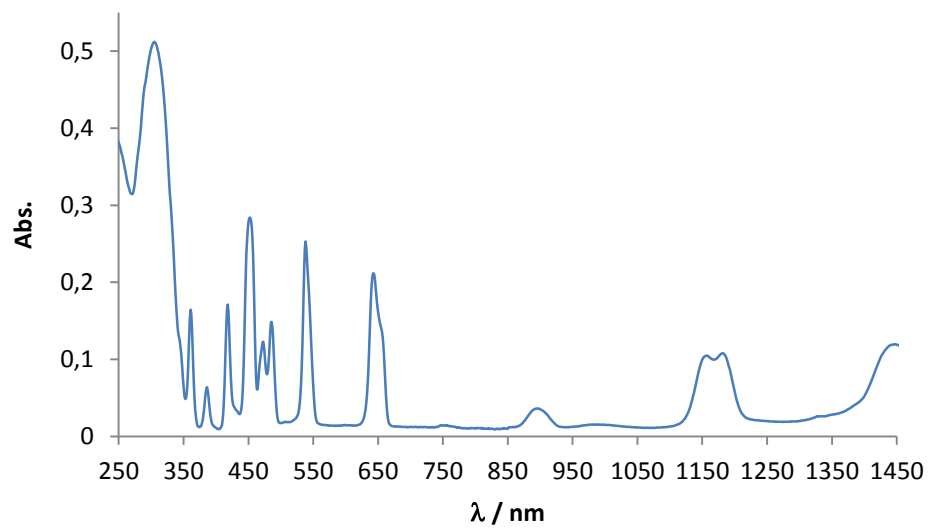


Figure S21. UV-visible absorption spectrum for **Ho-H₄NMPCl**.

Table S1. Selected bond distances (Å) for **Ln-H₄NMPCl**, (Ln= La, Pr, Sm, Eu, Gd, Tb, Dy and Ho) and **Ce-H₃NMP**.

[La(H₄NMP)(H₂O)₂]Cl·2H₂O					
	Length (Å)		Length (Å)		Length (Å)
La1-O8	2.439(2)	P2-O10	1.497(3)	P4-O13	1.497(2)
La1-O20	2.458(2)	P2-O19	1.509(2)	P4-O11	1.497(2)
La1-O10	2.466(3)	P2-O24	1.589(2)	P4-O12	1.583(3)
La1-O19	2.470(2)	P2-C17	1.823(4)	P4-C18	1.839(3)
La1-O11	2.485(2)	P3-O20	1.496(2)	N9-C18	1.508(4)
La1-O13	2.487(2)	P3-O8	1.496(2)	N9-C17	1.508(4)
La1-O14	2.559(3)	P3-O15	1.570(3)	N9-C22	1.525(4)
La1-O6	2.649(3)	P3-C22	1.829(3)		
[Ce₂(H₃NMP)₂(H₂O)₄]·4.5H₂O					
	Length (Å)		Length (Å)		Length (Å)
Ce1-O1w	2.608(3)	P1-O1	1.526(4)	P5-O13	1.538(4)
Ce1-O2w	2.832(3)	P1-O2	1.533(4)	P5-O14	1.545(4)
Ce1-O1	2.306(1)	P1-O3	1.535(4)	P5-O15	1.539(4)
Ce1-O4	2.489(2)	P1-C1	1.800(4)	P5-C5	1.810(4)
Ce1-O10	2.469(2)	P2-O4	1.539(4)	P6-O16	1.545(4)
Ce1-O11	2.398(2)	P2-O5	1.543(4)	P6-O17	1.538(4)
Ce1-O13	2.943(2)	P2-O6	1.546(4)	P6-O18	1.542(4)
Ce1-O14	2.677(2)	P2-C2	1.808(4)	P6-C6	1.804(4)
Ce1-O14	2.545(1)	P3-O7	1.540(4)	N1-C1	1.507(4)
Ce2-O3w	2.484(3)	P3-O8	1.533(4)	N1-C2	1.494(4)
Ce2-O4w	2.427(3)	P3-O9	1.540(4)	N1-C3	1.507(4)
Ce2-O4	2.713(2)	P3-C3	1.807(4)	N2-C4	1.503(4)
Ce2-O5	2.729(2)	P4-O10	1.535(4)	N2-C5	1.501(5)
Ce2-O5	2.374(2)	P4-O11	1.535(4)	N2-C6	1.509(4)
Ce2-O7	2.497(2)	P4-O12	1.548(4)		
Ce2-O8	2.363(2)	P4-C4	1.805(4)		
Ce2-O13	2.404(1)				
Ce2-O16	2.440(1)				
[Pr(H₄NMP)(H₂O)₂]Cl·2H₂O					
	Length (Å)		Length (Å)		Length (Å)
Pr1-O8	2.395(8)	P2-O19	1.495(9)	P4-O13	1.484(9)
Pr1-O20	2.407(9)	P2-O10	1.506(11)	P4-O11	1.494(9)
Pr1-O10	2.426(1)	P2-O24	1.585(9)	P4-O12	1.584(9)
Pr1-O19	2.446(8)	P2-C17	1.838(13)	P4-C18	1.845(12)
Pr1-O11	2.450(8)	P3-O20	1.479(10)	N9-C17	1.497(15)
Pr1-O13	2.474(8)	P3-O8	1.488(9)	N9-C22	1.504(14)
Pr1-O14	2.548(1)	P3-O15	1.572(11)	N9-C18	1.500(14)
Pr1-O6	2.610(1)	P3-C22	1.850(12)		

Table S1(cont.). Selected bond distances (Å) for **Ln-H₄NMPCl**, (Ln= La, Pr, Sm, Eu, Gd, Tb, Dy and Ho) and **Ce-H₃NMP**.

[Sm(H₄NMP)(H₂O)₂]Cl·2H₂O					
	Length (Å)		Length (Å)		Length (Å)
Sm-O1	2.501(1)	P1-O1	1.536(5)	P3-O7	1.550(5)
Sm-O3	2.441(2)	P1-O2	1.531(5)	P3-O8	1.533(5)
Sm-O4	2.473(2)	P1-O3	1.541(5)	P3-O9	1.556(5)
Sm-O5	2.581(1)	P1-C1	1.803(5)	P3-C3	1.804(5)
Sm-O7	2.351(2)	P2-O4	1.535(5)	C1-N1	1.519(5)
Sm-O9	2.172(2)	P2-O5	1.539(5)	C2-N1	1.503(5)
Sm-Ow1	2.490(1)	P2-O6	1.532(5)	C3-N1	1.526(5)
Sm-Ow2	2.535(9)	P2-C2	1.804(5)		
[Eu(H₄NMP)(H₂O)₂]Cl·2H₂O					
	Length (Å)		Length (Å)		Length (Å)
Eu-O1	2.519(11)	P1-O1	1.528(4)	P3-O7	1.539(4)
Eu-O3	2.529(12)	P1-O2	1.537(4)	P3-O8	1.552(4)
Eu-O4	2.401(12)	P1-O3	1.541(4)	P3-O9	1.532(4)
Eu-O5	2.526(9)	P1-C1	1.809(4)	P3-C3	1.798(4)
Eu-O7	2.402(9)	P2-O4	1.534(1)	C1-N1	1.511(4)
Eu-O9	2.333(15)	P2-O5	1.536(4)	C2-N1	1.502(4)
Eu-Ow1	2.467(7)	P2-O6	1.538(4)	C3-N1	1.496(4)
Eu-Ow2	2.480(7)	P2-C2	1.808(4)		
[Gd(H₄NMP)(H₂O)₂]Cl·2H₂O					
	Length (Å)		Length (Å)		Length (Å)
Gd-O1	2.404(1)	P1-O1	1.533(4)	P3-O7	1.534(4)
Gd-O3	2.354(1)	P1-O2	1.539(4)	P3-O8	1.541(4)
Gd-O4	2.447(2)	P1-O3	1.532(4)	P3-O9	1.537(4)
Gd-O5	2.472(9)	P1-C1	1.812(4)	P3-C3	1.809(4)
Gd-O7	2.544(1)	P2-O4	1.532(4)	C1-N1	1.499(5)
Gd-O9	2.736(2)	P2-O5	1.536(4)	C2-N1	1.497(5)
Gd-Ow1	2.478(9)	P2-O6	1.541(4)	C3-N1	1.500(5)
Gd-Ow2	2.493(9)	P2-C2	1.808(4)		
[Tb(H₄NMP)(H₂O)₂]Cl·2H₂O					
	Length (Å)		Length (Å)		Length (Å)
Tb-O1	2.423(1)	P1-O1	1.530(4)	P3-O7	1.532(4)
Tb-O3	2.289(1)	P1-O2	1.532(4)	P3-O8	1.553(4)
Tb-O4	2.430(1)	P1-O3	1.535(4)	P3-O9	1.538(4)
Tb-O5	2.470(9)	P1-C1	1.810(4)	P3-C3	1.803(4)
Tb-O7	2.461(1)	P2-O4	1.540(4)	C1-N1	1.521(4)
Tb-O9	2.317(1)	P2-O5	1.535(4)	C2-N1	1.497(5)
Tb-Ow1	2.479(8)	P2-O6	1.540(4)	C3-N1	1.495(4)
Tb-Ow2	2.502(8)	P2-C2	1.806(4)		

Table S1 (cont.). Selected bond distances (Å) for **Ln-H₄NMPCl**, (Ln= La, Pr, Sm, Eu, Gd, Tb, Dy and Ho) and **Ce-H₃NMP**.

[Dy(H₄NMP)(H₂O)₂]Cl·2H₂O					
	Length (Å)		Length (Å)		Length (Å)
Dy-O1	2.418(2)	P1-O1	1.539(4)	P3-O7	1.536(4)
Dy-O3	2.389(2)	P1-O2	1.546(4)	P3-O8	1.548(4)
Dy-O4	2.332(2)	P1-O3	1.532(4)	P3-O9	1.536(4)
Dy-O5	2.407(1)	P1-C1	1.816(4)	P3-C3	1.803(4)
Dy-O7	2.353(2)	P2-O4	1.532(4)	C1-N1	1.509(4)
Dy-O9	2.321(2)	P2-O5	1.537(4)	C2-N1	1.499(5)
Dy-Ow1	2.440(9)	P2-O6	1.540(4)	C3-N1	1.488(4)
Dy-Ow2	2.459(9)	P2-C2	1.815(4)		
[Ho(H₄NMP)(H₂O)₂]Cl·2H₂O					
	Length (Å)		Length (Å)		Length (Å)
Ho01-O11	2.2875(2)	P1-O5	1.491(2)	P3-O11	1.493(2)
Ho01-O12	2.2967(2)	P1-O6	1.491(2)	P3-O12	1.494(2)
Ho01-O8	2.3143(2)	P1-O4	1.585(2)	P3-O10	1.570(1)
Ho01-O9	2.339(2)	P1-C1	1.841(2)	P3-C3	1.832(2)
Ho01-O6	2.3430(2)	P2-O9	1.496(2)	N1-C2	1.505(3)
Ho01-O5	2.3680(2)	P2-O8	1.510(1)	N1-C1	1.506(3)
Ho01-O2	2.3914(2)	P2-O7	1.584(2)	N1-C3	1.511(3)
Ho01-O1	2.5406(2)	P2-C2	1.817(3)		

Table S2. H-bond interactions (Å) for **[Ce₂(H₃NMP)₂(H₂O)₄]·4.5H₂O** (length < 3 Å).

Vector	Length (Å)	Vector	Length (Å)	Vector	Length (Å)
O1...O4	2.950(29)	O8...Ow3	2.609(33)	Ow1...Ow3	2.97(4)
O1...O10	2.831(28)	O8...Ow4	2.362(27)	Ow1...Ow6	2.50(4)
O1...Ow2	2.742(33)	O9...O18	2.290(24)	Ow1...Ow7	2.98(4)
O2...Ow2	2.671(33)	O10...O14	2.878(31)	Ow3...Ow4	2.85(4)
O2...Ow6	2.847(35)	O11...O14	2.874(32)	Ow3...Ow8	3.012(29)
O3...O15	2.478(25)	O12...Ow5	2.633(33)	Ow3...Ow9	2.724(30)
O4...O7	2.972(31)	O13...Ow3	2.882(32)	Ow4...Ow9	2.847(25)
O5...Ow4	2.922(31)	O13...O16	2.804(26)	Ow5...Ow6	2.69(4)
O5...O5	2.913(34)	O15...Ow8	2.836(19)	Ow5...Ow8	2.756(30)
O5...O8	2.758(32)	O15...Ow10	2.628(23)	Ow5...Ow9	2.777(30)
O6...O17	2.330(23)	O16...Ow4	2.797(32)	Ow8...Ow10	2.47261(6)
O6...Ow6	2.74(4)	O17...Ow10	2.841(20)	Ow9...Ow10	2.87273(8)
O6...Ow7	2.786(34)	O18...Ow7	2.92(4)		
O7...O10	2.963(27)				
O7...O16	3.004(25)				

Table S3. Absorption and emission bands observed for $[\text{Pr}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl}\cdot 2\text{H}_2\text{O}$.

<i>Absorption</i>		<i>Luminescence ($\lambda_{exc} = 355 \text{ nm}$)</i>	
Transition	Wavelength (nm)	Transition	Wavelength (nm)
$\leftarrow {}^3\text{H}_4$		${}^3\text{P}_0 \rightarrow$	
${}^3\text{F}_2$	1945		
${}^3\text{F}_3, {}^3\text{F}_4$	1559		
${}^1\text{G}_4$	1026		
${}^1\text{D}_2$	592	${}^3\text{H}_6$	621
${}^3\text{P}_0$	481	${}^3\text{H}_5$	535
${}^3\text{P}_1, {}^1\text{I}_6$	469	${}^3\text{H}_4$	481
${}^3\text{P}_2$	444		

Table S4. Absorption and emission bands observed for [Sm(H₄NMP)(H₂O)₂]Cl·2H₂O.

<i>Absorption</i>		<i>Luminescence ($\lambda_{exc}=355\text{ nm}$)</i>	
Transition	Wavelength	Transition	Wavelength
$\leftarrow {}^6\text{H}_{5/2}$	(nm)	${}^4\text{G}_{5/2} \rightarrow$	(nm)
${}^6\text{H}_{15/2}, {}^6\text{F}_{3/2}, {}^6\text{F}_{1/2}$	1500		
${}^6\text{F}_{5/2}$	1395		
${}^6\text{F}_{7/2}$	1244		
${}^6\text{F}_{9/2}$	1087		
${}^6\text{F}_{11/2}$	948	${}^6\text{H}_{9/2}$	644
${}^4\text{G}_{7/2}, {}^4\text{I}_{9/2}, {}^4\text{M}_{15/2},$	476	${}^6\text{H}_{7/2}$	598
${}^4\text{I}_{11/2}, {}^4\text{I}_{13/2}$	464	${}^6\text{H}_{5/2}$	562
${}^4\text{F}_{5/2}, {}^4\text{M}_{17/2}, {}^4\text{G}_{9/2}, {}^4\text{I}_{15/2}$	440		
${}^6\text{P}_{5/2}$	417		
${}^4\text{P}_{5/2}, {}^4\text{F}_{7/2}, {}^4\text{F}_{9/2}, {}^4\text{K}_{11/2}, {}^4\text{M}_{21/2}, {}^4\text{L}_{15/2},$	402		
${}^4\text{G}_{11/2}$	393		
${}^4\text{L}_{13/2},$	374		
${}^4\text{D}_{1/2}, {}^6\text{P}_{7/2}, {}^4\text{L}_{17/2}, {}^4\text{K}_{13/2}, {}^4\text{F}_{9/2}$	362		
${}^4\text{D}_{3/2}, {}^4\text{D}_{5/2}, {}^6\text{P}_{5/2}, {}^4\text{H}_{7/2}$	344		
${}^4\text{K}_{15/2}, {}^4\text{H}_{9/2}, {}^4\text{D}_{7/2}$			

Table S5. Absorption and Emission bands observed for $[\text{Eu}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl}\cdot 2\text{H}_2\text{O}$.

<i>Absorption</i>		<i>Luminescence ($\lambda_{exc} = 355 \text{ nm}$)</i>	
Transition	Wavelength (nm)	Transition	Wavelength (nm)
$\leftarrow {}^7\text{F}_0$		${}^5\text{D}_0 \rightarrow$	
${}^5\text{D}_0, {}^5\text{D}_1, {}^5\text{D}_2$	579		
	526		
	464		
${}^5\text{L}_6$	393	${}^7\text{F}_3$	652
${}^5\text{G}_2, {}^5\text{G}_4, {}^5\text{G}_6$	376	${}^7\text{F}_2$	611
${}^5\text{D}_4$	362	${}^7\text{F}_1$	592
		${}^7\text{F}_0$	579
		${}^5\text{D}_1 \rightarrow {}^7\text{F}_4$	624
		${}^5\text{L}_6 \rightarrow {}^7\text{F}_0$	400

Table S6. Absorption and emission bands observed for [Tb(H₄NMP)(H₂O)₂]Cl·2H₂O.

<i>Absorption</i>		<i>Luminescence ($\lambda_{exc} = 355 \text{ nm}$)</i>	
Transition	Wavelength (nm)	Transition	Wavelength (nm)
$\leftarrow {}^7F_6$		${}^5G_4 \rightarrow$	
7F_3			
${}^7F_0, {}^7F_1, {}^7F_2$	1941	7F_3	620
	1446	7F_4	583
5D_4	487	7F_5	546
	379	7F_6	489
${}^5G_6, {}^5D_3, {}^5L_{10}$	371		
${}^5G_5, {}^5G_2, {}^5G_4, {}^5L_9$	352		
${}^5D_2, {}^5L_8, {}^5L_7, {}^5G_3$	342		
${}^5H_7, {}^5D_0, {}^5D_1$	318		

Table S7. Absorption and emission bands observed for [Dy(H₄NMP)(H₂O)₂]Cl·2H₂O.

<i>Absorption</i>		<i>Luminescence ($\lambda_{ex}=355\text{ nm}$)</i>	
Transition	Wavelength	Transition	Wavelength
$\leftarrow {}^6\text{H}_{15/2}$	(nm)	${}^4\text{F}_{9/2}\rightarrow$	(nm)
${}^6\text{H}_{11/2}$	1715		
${}^6\text{H}_{9/2}, {}^6\text{F}_{11/2}$	1289		
${}^6\text{H}_{7/2}, {}^6\text{F}_{9/2}$	1099		
${}^6\text{H}_{5/2}, {}^6\text{F}_{7/2}$	907		
${}^6\text{F}_{5/2}$	806	${}^6\text{H}_{11/2}$	651
${}^6\text{F}_{3/2}$	755	${}^6\text{H}_{13/2}$	573
${}^4\text{F}_{9/2}$	474	${}^6\text{H}_{15/2}$	482
${}^4\text{I}_{15/2}$	451		
${}^4\text{G}_{11/2}$	427		
${}^4\text{M}_{21/2}, {}^4\text{I}_{13/2}, {}^4\text{F}_{7/2}, {}^4\text{K}_{17/2}, {}^4\text{M}_{19/2}$	388		
${}^4\text{I}_{11/2}, {}^6\text{P}_{5/2}, {}^6\text{P}_{3/2}$	364		
${}^4\text{M}_{15/2}, {}^6\text{P}_{9/2}$	350		
${}^4\text{G}_{9/2}, {}^4\text{F}_{5/2}, {}^4\text{I}_{9/2}$	337		
${}^4\text{K}_{15/2}, {}^4\text{L}_{19/2}, {}^6\text{P}_{3/2}, {}^4\text{M}_{17/2}, {}^4\text{G}_{7/2}$	324		

Table S8. Absorption and emission bands observed for $[\text{Ho}(\text{H}_4\text{NMP})(\text{H}_2\text{O})_2]\text{Cl}\cdot 2\text{H}_2\text{O}$.

Transition	Wavelength (nm)
$\leftarrow {}^5\text{I}_8$	
${}^5\text{I}_7$	1950
${}^5\text{I}_6$	1157
${}^5\text{I}_5$	896
${}^5\text{F}_5$	656,643*
${}^5\text{F}_4, {}^5\text{S}_2$	538
${}^5\text{F}_3$	486
${}^5\text{F}_2$	473
${}^5\text{F}_1$	452
${}^5\text{G}_5$	418
${}^5\text{G}_4$	386
${}^3\text{H}_6$	361
${}^5\text{G}_3,$	345
${}^3\text{F}_4, {}^3\text{K}_6$	332
${}^3\text{M}_{10}$	305

*Shoulder