Lawrence Berkeley National Laboratory

Recent Work

Title

MEASUREMENT OF THE ROTATIONAL SPECTRUM OF HF+ BY LASER MAGNETIC RESONANCE

Permalink

https://escholarship.org/uc/item/5r91x18j

Author

Hovde, D.C.

Publication Date

1984



Lawrence Berkeley Laboratory

UNIVERSITY OF CALIFORNIA

Materials & Molecular Research Division

LAWRENCE BERKELEY LABORATORY

MAR 14 1984

Submitted to Molecular Physics

LIBRARY AND DOCUMENTS SECTION

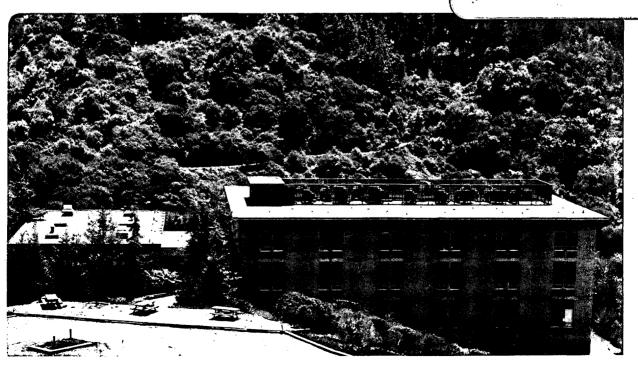
MEASUREMENT OF THE ROTATIONAL SPECTRUM OF HF+ BY LASER MAGNETIC RESONANCE

D.C. Hovde, E. Schäfer, S.E. Strahan, C.A. Ferrari, D. Ray, K.G. Lubic, and R.J. Saykally

January 1984

For Reference

Not to be taken from this room



Prepared for the U.S. Department of Energy under Contract DE-AC03-76SF00098

DISCLAIMER

This document was prepared as an account of work sponsored by the United States Government. While this document is believed to contain correct information, neither the United States Government nor any agency thereof, nor the Regents of the University of California, nor any of their employees, makes any warranty, express or implied, or assumes any legal responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by its trade name, trademark, manufacturer, or otherwise, does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof, or the Regents of the University of California. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof or the Regents of the University of California.

Measurement of the Rotational Spectrum of HF by Laser Magnetic Resonance

D. C. Hovde, E. Schäfer, S. E. Strahan, C. A. Ferrari, D. Ray, K. G. Lubic, ** and

R. J. Saykally

Department of Chemistry and Materials and Molecular Research Division,
Lawrence Berkeley Laboratory
University of California, Berkeley, CA 94720

Abstract

Rotational transitions in the $X^2\Pi$ state of the HF⁺ molecular ion were measured by laser magnetic resonance spectroscopy. Hyperfine splittings from both nuclei were resolved. An analysis of the $J=3/2 \rightarrow 5/2$ transitions in the $\Omega=3/2$ spin sublevel is presented.

* Current address: Vogelsbergstr. 20 D-6479 Ranstadt West Germany

** Current address: Herzberg Institute for Astrophysics

National Research Council

100 Sussex Drive

Ottawa ON, Canada KlA OR6

Submitted to: Molecular Physics

The first high-resolution study of fluoroniumy1 (HF^+), the simplest member of the hydrogen halide ion family, was reported in 1975 by Gewurtz, Lew and Flainek, who photographed the $\mathrm{A}^2\Sigma^+$ - $\mathrm{X}^2\Pi$ electronic emission bands from 385-483 nm in a low-pressure discharge with resolution sufficient to determine the principal electronic and structural parameters of the two states. However, the relatively large hyperfine interactions due to the fluorine nucleus were not resolved. Halide hyperfine structure, as studied recently by laser magnetic resonance rotational spectroscopy, has provided detailed information about the electronic structures of HCl^+ and HBr^+ . In the present work on HF^+ , we continue our study of this isovalent series of molecular ions. This case is of particular significance: with only nine electrons, HF^+ possesses a structure which can be directly compared with rigorous ab initio calculations. 5,6

By directly summing the contributions of both bound and continuum wavefunctions from the shallow $A^2\Sigma^+$ state, Hutson and Cooper have computed Λ -doubling parameters for the $X^2\Pi$ state with an accuracy comparable to that obtained from emission spectroscopy experiments. Unlike the case for isoelectronic OH, continuum wavefunctions make a large contribution to the ground state Λ -doubling in HF^+ , and such low-lying unbound levels are difficult to treat by conventional methods. By examining the spectrum of the $X^2\Pi$ state of HF^+ under the very high resolution available with LMR, we can rigorously assess the reliability of this new direct-sum method. Finally, despite the low cosmic abundance of fluorine, one model of the interstellar medium cites HF^+ as a possible interstellar maser; accurate rest frequencies obtained from these laboratory measurements will permit an astronomical search.

In this article we report the first detection and a preliminary analysis of rotational transitions in HF+, with resolution of both F and H hyperfine splitting. Previous papers^{2,3} describe the details of our LMR spectrometer. Briefly, it consists of a 15-inch electromagnet and an optically pumped far-infrared laser. A beam splitter separates the intracavity sample cell from the laser gain region, and can be rotated to select either σ or π polarization of the laser electric field relative to the magnetic field. The vacuum for the sample region is provided by a Roots pump (Leybold-Heraus WA-250,4600 1/min) backed by a mechanical pump (Welch 1396F, 2800 1/min). The HF^+ was produced in the sample region by a 10 mA $_{\mathrm{rms}}^-$ A.C. discharge through helium containing a trace of hydrogen fluoride. Powered at 60 Hz by a Variac stepped up with neon sign transformer, the A.C. discharge system is similar to the D.C. discharge used to generate HCl+, but causes less plasma noise and degrades the electrodes more slowly. Helium (99.999%) was used without further purification. The HF (Matheson commercial grade) was condensed into a teflon trap at -78°C; volatile impurities were pumped away. This procedure removes the H_2 formed as HF oxidizes the walls of the lecture bottle. The purified HF was added directly from the trap at room temperature. Spectra were optimized with 0.4 Pa (3 mTorr) HF, and from 40 (0.3 Torr) to 267 (2 Torr) Pa He. The optimum signals were obtained with 90 Pa (700 mT) He (1 Pa = 7.6 mTorr).

Predictions of the $X^2\Pi$ rotational transition frequencies, made with the constants of Gewurtz et al., ¹ indicated suitable coincidences with four known far-infrared laser lines. Using the 122.5 μ m CH₂F₂ and 124.4 μ m CH₂DOH laser lines, we observed 20 LMR resonances due to the J = 3/2 \rightarrow 5/2, Ω = 3/2 rotational transition. With the 184.3 μ m CH₂F₂ and 186.0 μ m CH₃OH laser lines, the J = 1/2 \rightarrow 3/2, Ω = 1/2 rotational transition was measured.

Laser frequencies and magnetic field strengths are presented in Table 1. Chemical and spectroscopic evidence forms the basis for assigning the spectra to HF^+ . The LMR signals all depend in the same way on the amount of HF added to the discharge, and can not be duplicated using HCl instead; the peaks disappear when the HF is shut off or increased above 2 Pa (15 mTorr). Unpurified HF, which is contaminated by H_2 as noted above, produces weak signals. These observations are explained by the thermodynamics of proton transfer to form $\mathrm{H}_2\mathrm{F}^+{:}10{,}11$

$$AHF^{+} + HF \rightarrow H_{2}F^{+} + F$$
 $\Delta H = -155 \text{ kJ/mol}$
 $AHF^{+} + H_{2} \rightarrow H_{2}F^{+} + H$ $\Delta H = -290 \text{ kJ/mol}$.

Rates for these reactions are not available from the literature, but the analogous reactions for ${\rm HCl}^+$ are fast. 12,13 Laser pulling experiments, which measure the sign of dH/dv, confirm the position of zero-field predictions made from the optical data of Gewurtz et al. 1

The 122.5 μ m LMR spectrum shown in Figure 1, as well as the 124.4 μ m spectrum, exhibit the pattern expected for a doublet II state split by two inequivalent spin 1/2 nuclei. ¹⁴ The magnitude of the Λ -doubling agrees well with that observed in the optical data of Gewurtz et al. ¹ The fluorine hyperfine splitting is nearly as large as the Λ -doubling; HCl ⁺ and HBr ⁺ likewise show large hyperfine splitting. ^{3,4} Proton hyperfine structure appears as a small but clearly resolved doubling for HF ⁺, much as it does for isoelectronic OH. ¹⁵ By comparison, hydrogen hyperfine structure was not observed in the FIR LMR spectra of the heavier hydrogen halide ions, nor in their neutral analogs SH ¹⁶ and SeH. ¹⁷

The Ω = 1/2 state of HF⁺ is approximately diamagnetic, since the angular momentum coupling is very near Hund's Case (a); hence, a far-infrared laser

frequency must lie very close to a corresponding zero field transition for it to be tuned into resonance by available magnetic field strengths. Fortunately, two close coincidences exist for HF⁺ near the J = $1/2 \rightarrow 3/2$, $\Omega = 1/2$ transition. The 184.3 µm CH₂F₂ laser lies within 50 MHz of one component of the Λ -doublet, while the 186.0 µm CH₃OH laser is several hundred MHz from the other component.

An analysis of the Ω = 3/2 transitions was carried out with the high precision $^2\Pi$ Zeeman Hamiltonian of Veseth, 18,19 evaluated in a Case C_{α} basis set, and appropriate for a molecule with one nuclear spin. This Hamiltonian has been incorporated into a computer program capable of including both zero field and magnetic resonance spectra in a single weighted least squares fit. Because the program currently is limited to one nuclear spin, the hydrogen hyperfine structure was averaged out and only the much larger fluorine hyperfine pattern was included in this preliminary analysis. The results of the eight parameter fit using both LMR data and combination differences from the emission spectrum are presented in Table We have determined the Case(a)fluorine hyperfine constants h (h = a + a $lar{1}{2}(b+c))$ and b, where a, b, and c are the usual Frosch and Foley parameters. Our analysis also refines the values of the other molecular parameters, particularly the Λ -doubling constants p and q. Hutson and Cooper 7 calculated the values p = 7630 MHz and q = -1210 MHz, in excellent agreement with the results of our determination. (We have converted their values using the relationship $p = p_{HC}/2 + q_{HC}$, $q = q_{HC}$).

Acknowledgments

This work was supported by the Director, Office of Energy Research, Office of Basic Energy Sciences, Chemical Sciences Division of the U. S. Department of Energy under Contract No. DE-ACO3-76SF00098. Equipment was provided by the National Science Foundation (Grant CHE82-07307). CAF thanks the Brazilian government for a CNPq grant. ES acknowledges a fellowship from NATO through the German Academic Exchange Service.

References

- 1. Gewurtz, S., Lew, H., and Flainek, P., 1975, Can. J. Phys., 53, 1097.
- Lubic, K. G., 1982, Ph.D. Dissertation, University of California, Berkeley.
- 3. Ray, D., Lubic, K. G., and Saykally, R. J., 1982, Mol. Phys., 46, 217.
- 4. Saykally, R. J., and Evenson, K. M., 1979, Phys. Rev. Lett., 43, 515.
- 5. Rosmus, P., and Meyer, W., 1977, J. Chem. Phys., 66, 13.
- 6. Bondybey, V., Pearson, P. K., and Schaefer, H. F. III, 1972, J. Chem. Phys., 57, 1123.
- 7. Hutson, J. M., and Cooper, D. L., 1981, J. Chem. Phys., 75, 4502.
- 8. Wilson, I. D. L., 1978, J. Mol. Spectrosc., 70, 394.
- 9. Gold, E., Hammersley, R. E., and Richards, W. G., 1980, Proc. R. Soc. Lond. A, 373, 269.
- 10. Koshchel, D., editor, 1982, Gmelin Handbook of Inorganic Chemistry, 8th Ed., Fluorine Supplement, vol. 3 (Springer Verlag).
- 11. JANAF Thermochemical Tables, 2nd Ed., 1971, U. S. Dept. of Commerce,
 National Bureau of Standards NSRDS-NBS37.
- 12. Fehsenfeld, F. C., and Ferguson, E. E., 1974, J. Chem. Phys. 60, 5132.
- 13. Field, F. H., and Lampe, F. W., 1958, J. Am. Chem. Soc., 80, 5583.
- 14. Evenson, K. M., Saykally, R. J., Jennings, D. A., Curl, R. F. Jr., and Brown, J. M., 1980, Chemical and Biochemical Applications of Lasers, Vol. 5, edited by C. B. Moore (Academic).
- 15. Brown, J. M., Kerr, C. M. L., Wayne, F. D., Evenson, K. M., and Radford, H. E., 1981, J. Mol. Spectrosc., 86, 544.
- 16. Davies, P. B., Handy, B. J., Murray Lloyd, E. K., and Russell, D. K., 1978, Mol. Phys., 36, 1005.
- 17. Davies, P. B., Handy, B. J., Murray Lloyd, E. K., and Russell, D. K., 1978, J. Chem. Phys. 68, 3377.

U

- Veseth, L., 1976, J. Mol. Spectrosc., <u>59</u>, 51; Veseth, L., 1976,
 J. Mol. Spectrosc., <u>63</u>, 180; Veseth, L., 1979, J. Mol. Spectrosc.,
 <u>77</u>, 195.
- 19. Meerts, W. L., and Veseth, L., 1980, J. Mol. Spectrosc., $\underline{82}$, 202.

Table 1 Observed Laser Magnetic Resonance Transitions for V = 0 $\text{X}^2\Pi$ HF^+

Parit	y ^a M _J "	M _J '	M _I b	Flux (Gauss)	obs calc (MHz)
122.5 μm	CH_2F_2 (v =	2447969.	O MHz) S	$\lambda = 3/2, J =$	$3/2 \rightarrow 5/2$
-	-3/2	-1/2	1/2	10263.0	-4.1
-	-3/2	-1/2	-1/2	11315.9	-0.1
+	-3/2	-1/2	1/2	12112.0	3.6
+	-3/2	-1/2	-1/2	13182.4	0.3
.	-1/2	1/2	1/2	17511.3	14.7
-	-3/2	-3/2	1/2	17047.2	-1.1
124.4 µm	CH ₂ DOH (v	= 2409293	.0 MHz)	$\Omega = 3/2, J =$: 3/2 → 5/2
+	3/2	1/2	1/2	11747.0	4.6
+	3/2	1/2	-1/2	12967.0	-10.6
-				13635.5	
-	3/2	1/2	-1/2	14863.0	8.4
184.3 µm	CH_2F_2 (v =	1626602.	6 MHz) Ω	t = 1/2, J =	$1/2 \rightarrow 3/2$
	1/2	3/2	-1/2	1991.2	*
-	1/2	3/2	1/2	14161.6	*
_	-1/2	1/2	-1/2	3931.9	*
:	1/2	1/2	-1/2	3665.8	*
186.0 μ m CH_3OH ($\nu = 1611421.9 \text{ MHz}$) $\Omega = 1/2$, $J = 1/2 \rightarrow 3/2$					
+	-1/2	1/2	-1/2	1711.8	*
+	1/2	1/2	-1/2	1749.7	*

^{*} Data not included in analysis.

^a Parity of lower state.

^b All transitions are $\Delta M_{I} = 0$. M_{I} here refers to the fluorine nucleus.

For fitted lines, flux is averaged ~28G proton hyperfine doublet. Measurement precision is approximately ± 0.5 G.

Table 2

Molecular Parameters for $V = 0 X^2 \Pi HF^+ (MHz)^*$

 $B_0 = 513070.(10)$

 $D_0 = 66.23(27)$

A = -8736727(400)

 $A_{D} = -847(14)$

P = 7741(19)

q = -1214.9(18)

h = 3353(12)

b = 1384(130)

Fixed Molecular g-factors

 $g_{\ell} = 1.00232$

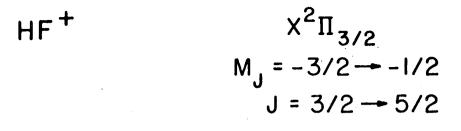
 $g_r = 0.00052$

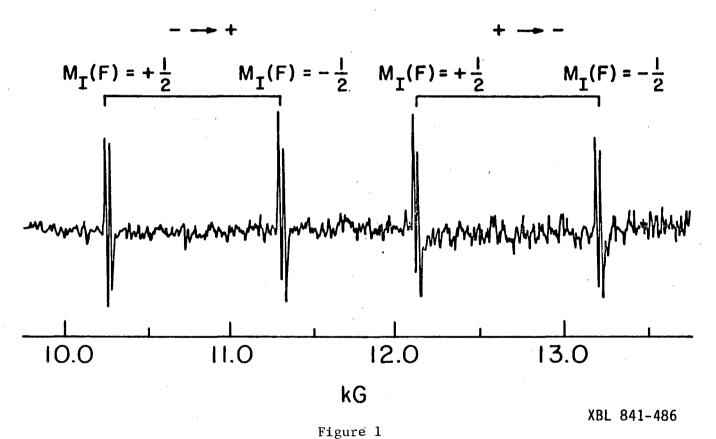
 $g_s = 2.00232$

Numbers in parentheses represent one standard deviation.

Figure Caption

Figure 1. A portion of the ${\rm HF}^+$ spectrum recorded with the 122.5 ${\rm \mu m}$ ${\rm CH_2F_2}$ laser. ${\rm HF}^+$ was generated in a 60 Hz discharge operating with 500 V and 9 mA (RMS values) in 94 Pa of helium containing a trace of HF. The data was obtained in a single scan with a time constant of 0.1 sec.





This report was done with support from the Department of Energy. Any conclusions or opinions expressed in this report represent solely those of the author(s) and not necessarily those of The Regents of the University of California, the Lawrence Berkeley Laboratory or the Department of Energy.

Reference to a company or product name does not imply approval or recommendation of the product by the University of California or the U.S. Department of Energy to the exclusion of others that may be suitable.



سند مسألات

TECHNICAL INFORMATION DEPARTMENT LAWRENCE BERKELEY LABORATORY UNIVERSITY OF CALIFORNIA BERKELEY, CALIFORNIA 94720