

Measurements of the total ion flux from vacuum arc cathode spots

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Abstract – The ion flux from vacuum arc cathode spots was measured in two vacuum arc systems. The first was a vacuum arc ion source which was modified allowing us to collect ions from arc plasma streaming through an anode mesh. The second discharge system essentially consisted of a cathode placed near the center of a spherically shaped mesh anode. In both systems, the ion current streaming through the mesh was measured by a biased collector. The mesh anodes had geometric transmittances of 60% and 72%, respectively, which were taken into account as correction factors. The ion current from different cathode materials was measured for 50-500 A of arc current. The ion current normalized by the arc current was found to depend on the cathode material, with values in the range from 5% to 19%. The normalized ion current was generally greater for elements of low cohesive energy. The ion erosion rates were determined from values of ion current and ion charge states, which were previously measured in the same ion source. The absolute ion erosion rates ranged from 16-173 $\mu\text{g}/\text{C}$.

KEYWORDS: Cathodic vacuum arcs, ion flux, ion current, ion erosion rates

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I. INTRODUCTION

Cathodic arc discharge phenomena have been investigated for over 200 years [1-4]. Many investigators have shown that the basic processes of plasma formation of cathode material occur at non-stationary cathode spots [5-9]. Cathode spots are highly dynamic, self-organized structures. For rough orientation, the following characteristic parameters are usually given: ignition and explosive time 1-10 ns, spot “lifetime” 10-100 ns, spot size $\sim 1 \mu\text{m}$, current density $\sim 10^{12} \text{ A/m}^2$, and plasma density before expansion at least 10^{26} m^{-3} . The main difficulties of arc spot research are associated with these extreme parameters, and therefore a number of phenomena are still the subject of research. Recent research points to the fundamentally fractal character of electron and plasma emission sites (see companion paper in this issue [10], and [11]), which puts the above mentioned characteristic parameters in a different light: they may reflect—at least to some degree—the limits of measuring techniques rather than parameters of truly elementary processes.

Measurements of the total ion flux generated in the vacuum arc plasma are important for fundamental and practical reasons. First, such data are critical to theories of vacuum arc discharges. The ion erosion rate, γ_i , is an important parameter that can be determined by ion flux measurements. The rate γ_i is commonly expressed in $\mu\text{g/C}$ and defined as the cathode mass that is phase-transformed from solid to plasma ions, per electric charge transferred,

$$\gamma_i = \Delta m_i / \int I_{arc} dt . \quad (1)$$

Second, ion flux data have immediate relevance for vacuum arc ion sources and cathodic arc deposition equipment. For ion sources, one seeks to know the maximum value of ion current that can be extracted from the vacuum arc plasma. For deposition via energetic condensation of arc plasma, the ion flux is the main parameter determining the deposition rate.

The ion erosion rate can be determined based on measurements of the total ion current, I_i , the arc current, I_{arc} , and mean ion charge state $\bar{Q} = \sum_{Q=1}^{Q_{max}} Q n_Q / \sum_{Q=1}^{Q_{max}} n_Q$, as follows:

$$\gamma_i = \frac{I_i}{I_{arc}} \cdot \frac{M_i}{e \bar{Q}} = \alpha_i \cdot \frac{M_i}{e \bar{Q}} \quad (2)$$

where n_Q is the density of ions of charge state Q , with $Q = 1, 2, \dots, Q_{max}$, M_i is the atomic mass of the cathode material and e is elementary charge. In (2), the normalized ion current $\alpha_i = I_i / I_{arc}$ was introduced; (2) provides

an often-used relation between ion erosion rate and ion current.

Pioneering work in determining ion erosion rates was done by Plyutto and co-workers [12] and a few years later by Kimblin [13, 14]. Kimblin used an arc system with pulse duration in the range 0.1-4.5 s and arc current of 50-1000 A. He determined the ion erosion rates by two methods: one is based on measuring ion currents to ion collectors, and the other is based on weighing the cathode before and after its use. The methods resulted in data for α_i and γ_i , which are related by equation (2). Using published data for mean ion charge states [12, 15], Kimblin [13, 14] realized that the results by these two methods are not consistent unless a significant fraction of mass loss is neutral. Kimblin reconciled the results assuming that evaporation of atoms must have occurred. With this assumption he could determine the fractional ionization of eroded cathode material, finding 15% for cadmium at the low end and 90% for molybdenum at the high end. Referring to work by Udriș [16], Kimblin [13] correctly noted that vacuum arc erosion also includes macroparticles, and therefore his values “should be regarded as lower limits to the fractional ionization.”

Considering *distributions* of ion charge states [17], one can quickly realize that the degree of ionization must be very high for practically all cathode elements. For example, the fraction of neutrals in cadmium arc plasmas was calculated to only 2.3%, and to $4.1 \times 10^{-4}\%$ for molybdenum [18].

The discrepancy between the ion collection and weighing methods clearly points to the contribution of macroparticles to cathode erosion. It has been shown that macroparticle mass losses are indeed important [19, 20]. Experimental results by Daalder [19] and others (e.g. Table I in [20]) show that the mass of cathode material removed by the macroparticles flux is about the same or greater than the mass of cathode material leaving in form of ions.

Besides ion current measurements, Daalder [19, 21] also used the weighing method to determine the total erosion rate of several cathode materials. From the analysis of the erosion rate as a function of charge transferred by the vacuum arc, Daalder concluded that the total erosion rate approaches the ion erosion rate if the charge transferred by the arc is small. In other words, losses by macroparticles are less important if $\int I_{arc} dt$ is kept small. Based on the assumption that the ratio of ion to arc current equals 0.1, Daalder [21] calculated the ion erosion rate for fourteen cathode materials using values of mean ion charge states published in [12, 15].

Mesyats has shown that the basic processes in cathode spots of vacuum arcs are similar to processes in vacuum breakdown and vacuum sparks: all can be based on the “ecton” concept [8, 22]. An ecton can be seen as

the minimum number of electrons involved in a single explosive electron emission process [23, 24], hence the ecton is like a “quantum of discharge” for vacuum discharges. This model is based on the concept that it is impossible for a cathode spot to exist with emission of electrons of less than one ecton. Emission processes are thought of as a sequence of ectons. The ecton concept explains a number of basic characteristics of the vacuum arc. Mesyats and Barengol’ts [25] presented a calculation of ion erosion rates based on the ecton theory. Using only parameters of the cathode material and mean ion charge states from different publications as input parameters, and assuming $\alpha_i = 0.1$, they found good agreement between their results and experimental erosion rates by Daalder [21]. The assumption of $\alpha_i = 0.1$ is prevalent in the ecton approach as well as in many other works on vacuum arcs although there are indications that noticeable deviation may occur, for example for carbon arcs at high current [26].

In the present work, ion currents are measured in two systems in order to have more and better data for γ_i and α_i . The first system was a vacuum arc ion source modified for total ion current measurements. This system was selected because it was previously used for measurements of ion charge state distributions [17, 27]. Combining measurements obtained in one system allows us to produce ion erosion data in a coherent manner. The second vacuum arc system had the same cathode unit as first one, but its mesh anode offered spherical geometry providing high transmission independent of ion flow direction. In the measurements of the ion erosion rate γ_i and normalized ion current α_i , much attention was paid to geometry effects. This was necessary to determine correction factors as accurately as possible, thereby accounting for ions collected by the anode and thus not recorded by the designated, biased ion collectors.

Cathode spots occurred on a cathode surface that is cleaned by the arc as it operates, and hence all results apply to type 2 arc spots, where non-metallic plasma contributions are negligible [27].

II. EXPERIMENTAL DETAILS AND RESULTS

A. Ion source setup

The first round of experiments was performed using the vacuum arc ion source “Mevva V” at Lawrence Berkeley National Laboratory. The source has been described elsewhere [17]; here we describe only the procedure for the ion erosion rate experiments. As already mentioned, one reason for choosing the ion source

facility was that we can directly integrate erosion rate data with other data obtained at the same facility, such as ion charge state distributions [17], electron temperature [18], directed ion velocity [28], and burning voltage [29].

The design of the ion source “Mevva V” was modified as shown in Fig. 1(a). Cathode spots of the vacuum arc were ignited on the front surface of the rod cathode “C,” which had a diameter of 6.25 mm. The original annular anode was replaced by a semi-transparent anode made from fine stainless steel mesh. The mesh was spherical in shape with mesh openings of 0.8 mm x 0.8 mm and a geometric transparency of 60%. The anode mesh “AM” was electrically insulated from the original ion source anode and the extraction system.

Both collector electrodes “COL1” and “COL2” were negatively biased up to $U_b = -200$ V with respect to the mesh anode “AM.” Jointly they served as a large collector to all ions coming through the anode mesh. The standard electrical circuit for the arc supply of the ion source was used. An eight-stage pulse-forming-network provided arc pulses of 250 μ s duration. The arc current was 100 A for most experiments. The base pressure of the cryogenically pumped system was about $5 \cdot 10^{-5}$ Pa. The measurements included monitoring arc current and ion current to the collector electrodes. The signals from broad-band current transformers (Pearson™ coils) were recorded by a digital oscilloscope (Tektronix TDS 744). Ion erosion rates for several cathodes were derived from the recorded current data. Preliminary results were reported in the Proceedings of the ISDEIV in Yalta. A comparison with literature data showed that the normalized ion currents and ion erosion rates for some well-investigated material, such as copper, were smaller than expected. A detailed analysis of experiment conditions and inspection of electrodes indicated that, given the cathode position in the ion source, those ions streaming under a large angle to cathode surface normal were deposited on the holder of the mesh anode and therefore were not registered at ion collector. This finding called for improved experiment.

B. Improved vacuum arc setup

The shortcoming was corrected by using a second, similar, but improved setup as shown in Fig. 1(b). The main differences were in the cathode position, mesh shape, and mesh transmittance. The cathode surface was now positioned exactly in the center of a semi-sphere, stainless steel mesh anode, whose radius of curvature was 109 mm. Each opening of the mesh was 2 mm x 2 mm, and the geometric transmittance was 72%. Using this geometry, ions would “see” the same mesh regardless of their flow direction. The other conditions were comparable to the conditions of the previous “Mevva V” experiment. The mesh anode “AM” was surrounded by

a stainless steel collector, which is labeled “COL” in Fig. 1(b). The collector was negatively biased up to 100 V with respect to the mesh anode. Ion saturation was clearly reached when the negative bias approached 75 V. During an arc pulse, bias of -100 V dropped less than 5 V at maximum ion current. The system rarely experienced “breakdown,” i.e., plasma formation by arc spots on the negatively biased collector. When breakdown occurred, data were of course discarded.

A seven-stage pulse-forming-network provided arc pulses of 500 μ s duration, adjustable amplitude up to 500 A, with a repetition rate of several pulses per second. The base pressure of the experimental system was about 10^{-4} Pa. The measurements included again monitoring the arc current and ion current to the collector electrode by a digital oscilloscope (Tektronix TDS 224). An example of experimental data is shown in Fig. 2.

Figure 3 shows the dependence of the (geometry-corrected) ion current as a function of vacuum arc current. It is satisfactory to see the expected linear relation. The results indicate that γ_i and α_i do not depend on the vacuum arc current for the current range investigated. The results of the ion erosion rate experiments are presented in Table I for sixteen different cathode materials that were investigated.

III. DISCUSSION

From Table I we can see that the results are in reasonable agreement with Kimblin’s experimental data [13, 14]. The approximate independence of arc current is consistent with findings that increasing the arc current leads to an increase in the *number* of emission centers rather than changing the character of these centers.

From the data presented in Table I it can be seen that the normalized ion current α_i is not as independent of material as claimed in [13] and often used in other work. The values of α_i range from 5% (W) to 19% (C). For estimates one could use an average α_i -value of 8% keeping in mind that the actual value may be different by more than 50%.

The ion erosion rates γ_i in Table I assigned to Daalder [21] were slightly corrected by using the more recent mean charge states published by Brown [17]. One should note that Kimblin found a linear relationship between ion flux and arc current for DC arcs up to 3000 A. Focusing on carbon, Schülke et al. [26] determined a slight increase of α_i by 1.2% per 100 A in the range 100 A to 1200 A. This could be in part due to a slight increase of the mean ion charge state with increasing current [30].

The ion erosion rates estimated by our measurements are lower than Mesyats’ [8] and Daalder’s [21].

Possible reasons for these differences may lie in the details and methodology of data interpretation. Although much attention was paid to use accurate correction factors, a systematic error in our measurement could be due to overestimating the actual transmittance of the anode mesh. However, it is more likely that the contributions of macroparticles were not accurately accounted for in experiments that utilized the weighing method [8, 21].

The ion current and cohesive energy for different cathode material are presented in Fig. 4. From this picture is clear that the ion current and cohesive energy are in opposing phase, as one would expect from the “Cohesive Energy Rule” for vacuum arc discharges [31]. Figure 4 demonstrates that, as a rule, a cathode material of greater cohesive energy (i.e., the energy needed to remove an atom from the solid to infinity) will produce less ions for a given discharge current. The physical interpretation is clear: Materials of greater cohesive energy require by definition more energy for the cathode phase transformation leading to metal plasma, and thus an equal amount of energy produces more (or less) plasma for materials of lower (or higher, respectively) cohesive energy. One should note that other factors affect plasma formation too, and therefore the Cohesive Energy Rule is a valuable empirical rule without having the merit of a physical law.

IV. SUMMARY

The ion erosion rates for sixteen cathode elements have been measured based on ion current measurements. It was found that the normalized ion current is not the “universal” constant of 0.1, as often claimed and used, but slightly depending on the material. The range of normalized ion current α_i is 5% – 19%. Elements with high cohesive energy tend to have smaller ion erosion rate α_i . This is in accordance with the empirical Cohesive Energy Rule, which was originally formulated for ion charge state distributions and arc voltage. Using known mean ion charge state data, material-dependent ion erosion rates in the range 16-173 $\mu\text{g}/\text{C}$ have been determined for arc currents up to 500 A.

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FIGURE CAPTIONS

Fig. 1 Experimental setups: Modified vacuum arc ion source “Mevva V” (a), top, and discharge system with spherical anode (b), bottom. “C” is the cathode rod, “TR” is the trigger, “AM” is the anode mesh, “COL” is the ion collector. For the “Mevva V” experiment, the two connected collector electrodes “COL1” and COL2” were originally part of the ion extraction system that was used for ion charge state measurements, which is the reason for the somewhat unusual geometry of the ion collectors.

Fig. 2. Examples of original data for arc current pulse, and carbon and copper ion currents.

Fig.3. Ion current, corrected for limited transmission by the mesh anode, as a function of arc current for different cathode materials.

Fig.4. Normalized ion current, α_i , and cohesive energy of different cathode materials. The lines are to guide the eye.

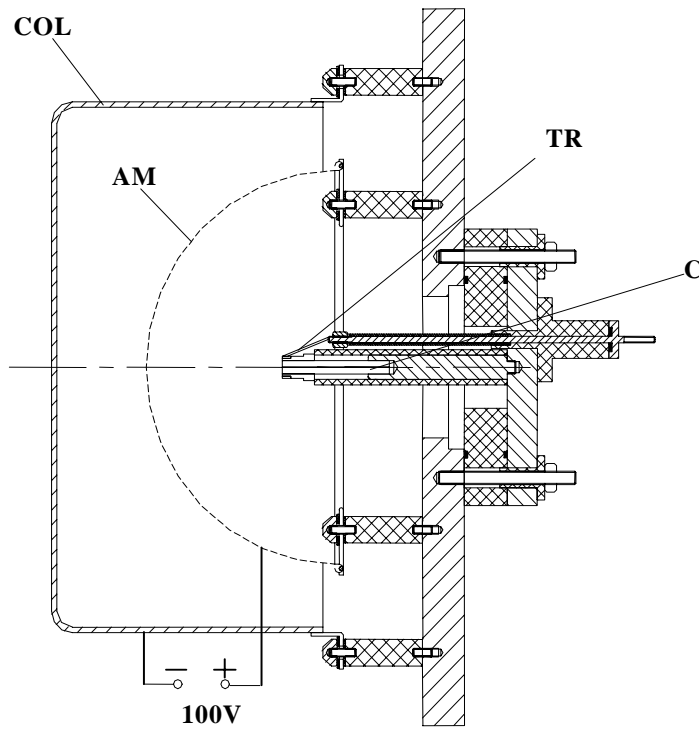
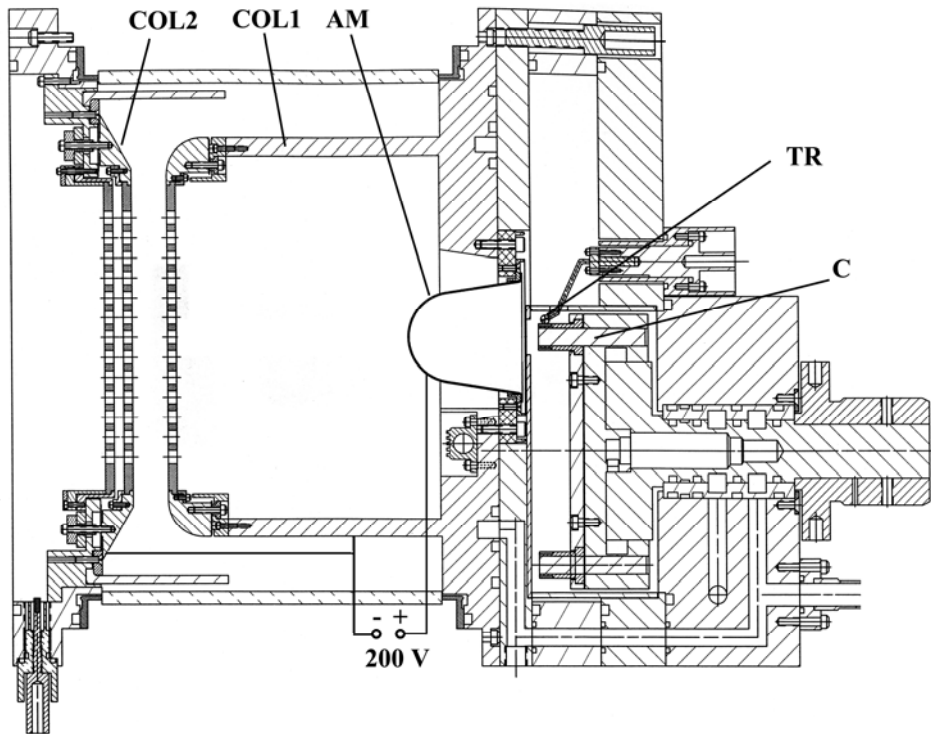


Fig. 1 (a) top, and (b) bottom.

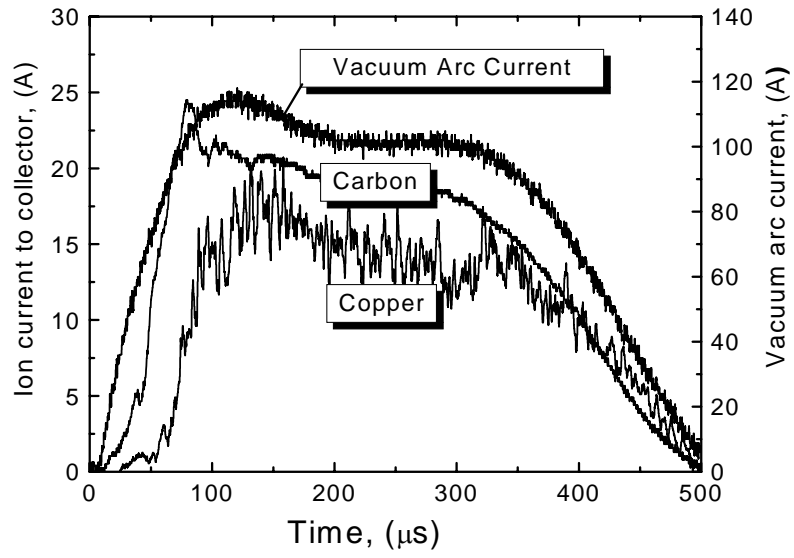


Fig. 2.

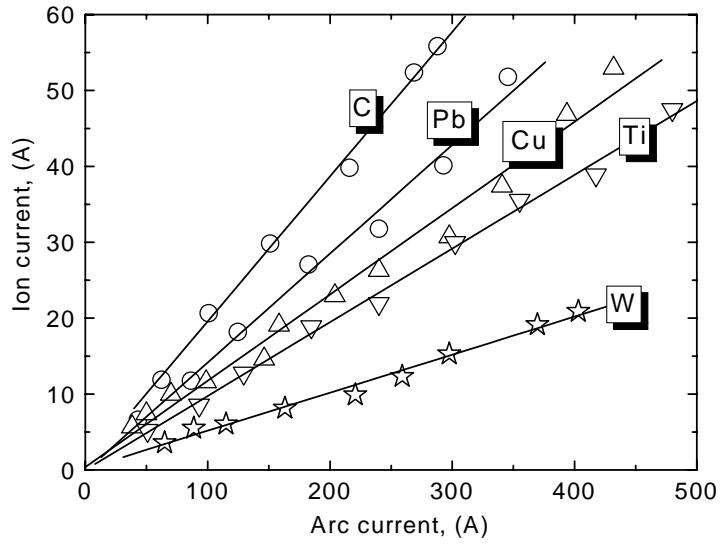


Fig. 3.

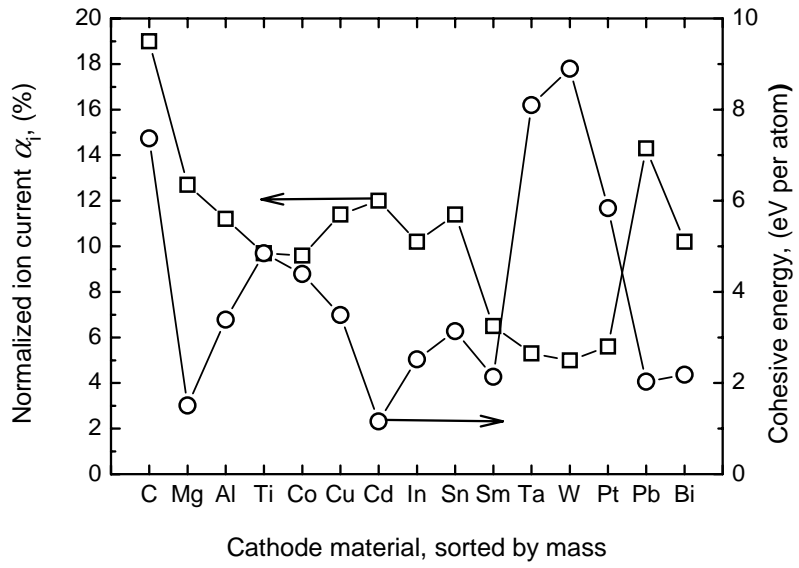


Fig.4.

TABLE I

NORMALIZED ION CURRENT AND ION EROSION RATES, AND COMPARISON WITH PUBLISHED DATA.

| Cathode material | This Experiment | | Ref. [13] | Ref. [8] | Ref. [21] |
|------------------|-----------------|-------------------------------------|----------------|-------------------------------------|-------------------------------------|
| | α_i , % | γ_i , $\mu\text{g}/\text{C}$ | α_i , % | γ_i , $\mu\text{g}/\text{C}$ | γ_i , $\mu\text{g}/\text{C}$ |
| C | 19 | 23.8 | 10 | 13-17 | ... |
| Mg | 12.7 | 18.8 | ... | 19-25 | 15 |
| Al | 11.2 | 15.9 | ... | 22-25 | 15 |
| Ti | 9.7 | 22.4 | 8.0 | ... | ... |
| Co | 9.6 | 30.4 | 8.0 | ... | ... |
| Cu | 11.4 | 33.4 | ... | 35-39 | ... |
| Zr | 10.5 | 36.3 | ... | ... | ... |
| Cd | 12 | 94.6 | 8.0 | 128-130 | 79.1 |
| In | 10.2 | 80.5 | ... | ... | ... |
| Sn | 11.4 | 83.1 | ... | ... | 72.8 |
| Sm | 6.5 | 46.1 | ... | ... | ... |
| Ta | 5.3 | 31.2 | ... | ... | 59 |
| W | 5 | 27.1 | 7.0 | 62-90 | 57 |
| Pt | 5.6 | 50.6 | ... | ... | ... |
| Pb | 14.3 | 172.8 | ... | ... | 120.8 |
| Bi | 10.2 | 171.5 | ... | ... | 168 |



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