# Medicinal Flowers．I．Aldose Reductase Inhibitors and Three New Eudesmane－Type Sesquiterpenes，Kikkanols A，B，and C，from the Flowers of Chrysanthemum indicum $\mathbf{L}$ ． 

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#### Abstract

The methanolic extract from the flowers of Chrysanthemum indicum L．，Chrysanthemi Indici Flos，was found to show inhibitory activity against rat lens aldose reductase．By bioassay－guided separation，the active components，such as flavone and flavone glycosides，were isolated from the extract together with three new eudes－ mane－type sesquiterpenes，kikkanols $A, B$ ，and $C$ ．The structures of kikkanols $A, B$ ，and $C$ were elucidated on the basis of chemical and physicochemical evidence，which included application of the modified Mosher＇s method．


Key words kikkanol；Chrysanthemum indicum；aldose reductase inhibitor；medicinal flower；eudesmane－type sesquiterpene； flavonoid

The flowers of Chrysanthemum indicum L．and related Chrysanthemum plants（Compositae）have been used as a Chinese natural medicine，Chrysanthemi Indici Flos［Yagik－ ka（野菊花）in Japanese］，which is prescribed for antiinflam－ matory，analgesic，and antipyretic purposes and the treatment of eye disease in Chinese traditional preparations．As chemi－ cal constituents of this plant，several bisabolane－and gua－ iane－type sesquiterpenes have been isolated from the flowers and aerial parts of European C．indicum，${ }^{1)}$ and the composi－ tion of the essential oil and isolation of a flavonoid have been also reported．${ }^{2)}$ In addition，several flavonoids and sesquiter－ penes have been isolated from the Chinese natural medi－ cine．${ }^{3)}$ However，the pharmacological activity and bioactive constituents of this natural medicine are left uncharacterized．

In the course of our studies on bioactive principles of nat－ ural medicines ${ }^{4}$ and medicinal foodstuffs，${ }^{5)}$ we have found that the methanolic extract from the flowers of C．indicum exhibited potent inhibitory activity against rat lens aldose re－ ductase．From the methanolic extract，active components such as flavone and flavone glycosides were isolated by bioassay－guided separation using aldose reductase inhibitory activity together with three new eudesmane－type sesquiter－ penes called kikkanols A（1），B（2），and C（3）．In this paper， we describe the characterization of the inhibitors from the flowers of C．indicum against rat lens aldose reductase and the structure elucidation of kikkanols（ $\mathbf{1} \mathbf{- 3}$ ）．

The flowers of Chinese C．indicum were extracted with methanol under reflux．Since the methanolic extract was found to exhibit inhibitory activity against rat lens aldose re－ ductase，the methanolic extract was partitioned in an ethyl acetate／water mixture to give an ethyl acetate－soluble portion and an aqueous phase．The aqueous phase was further ex－ tracted with 1－butanol to give a 1－butanol－soluble portion and a water－soluble portion．As is apparent from Table 1，the ethyl acetate－soluble and 1－butanol－soluble portions showed inhibitory activity $\left(\mathrm{IC}_{50} 1.3 \mu \mathrm{~g} / \mathrm{ml}\right.$ and $3.5 \mu \mathrm{~g} / \mathrm{ml}$ ，respec－ tively）．The ethyl acetate－soluble portion was subjected to sil－ ica gel and ODS column chromatography and finally HPLC to furnish kikkanols A（1，0．0008\％from the natural medi－ cine），B（ $\mathbf{2}, 0.0003 \%$ ），and C（3，0．0007\％），cis－spiroketal－

[^0]enolether polyyne ${ }^{6)}(4,0.079 \%)$ ，trans－spiroketalenolether polyyne ${ }^{6)}(5,0.015 \%)$ ，clovanediol ${ }^{7}$（ $6,0.0017 \%$ ），caryolane $1,9 \beta$－diol ${ }^{8)}$（7， $0.0007 \%$ ），oplopanone ${ }^{9)}$（8， $0.012 \%$ ），eu－ patilin $^{7}$（9，0．0008\％），and luteolin ${ }^{10)}(\mathbf{1 0}, 0.0016 \%)$ ．From the 1－butanol－soluble portion，luteolin 7－O－$\beta$－d－glucopyra－ noside ${ }^{10)}(11,0.13 \%)$ ，luteolin 7－O－$\beta$－D－glucopyranosiduronic $\operatorname{acid}^{11)}(\mathbf{1 2}, 0.048 \%)$ ，acacetin $7-O-\left(6^{\prime \prime}-\alpha\right.$－L－rhamnopyrano－ syl）－$\beta$－D－glucopyranoside ${ }^{12)}$（13， $0.014 \%$ ），and chlorogenic $\operatorname{acid}^{13)}(14,0.058 \%)$ were isolated by silica gel and ODS column chromatography．

Structures of Kikkanols A（1），B（2），and C（3） Kikkanol A（1）was isolated as a colorless needles of mp $165-167^{\circ} \mathrm{C}$ with positive optical rotation（ $[\alpha]_{\mathrm{D}}^{27}+57.2^{\circ}$ ）． The electron impact（EI）－MS of $\mathbf{1}$ showed a molecular ion $\left(\mathrm{M}^{+}\right)$peak at $m / z 254$ in addition to fragment ion peaks at $m / z 236\left(\mathrm{M}^{+}-\mathrm{H}_{2} \mathrm{O}\right)$ and $m / z 123$（base peak）．The molecular formula $\mathrm{C}_{15} \mathrm{H}_{26} \mathrm{O}_{3}$ of $\mathbf{1}$ was determined from the molecular ion peak observed in the EI－MS and by high－resolution MS measurement．The IR spectrum of $\mathbf{1}$ showed absorption bands ascribable to hydroxyl and olefin functions at 3532， 3456，1647，and $1073 \mathrm{~cm}^{-1}$ ．The ${ }^{1} \mathrm{H}-\mathrm{NMR}$（pyridine－$d_{5}$ ）and ${ }^{13} \mathrm{C}-$ NMR（Table 2 ）spectra of $\mathbf{1}$ showed signals assignable to a tertiary methyl $\left[\delta 1.66\left(\mathrm{~s}, 14-\mathrm{H}_{3}\right)\right]$ ，two secondary methyls ［ $\delta 1.02,1.03$（both d，$J=6.4 \mathrm{~Hz}, 12,13-\mathrm{H}_{3}$ ）］，two methines bearing a hydroxyl group［ $\delta 4.54(\mathrm{br} \mathrm{s}, 6-\mathrm{H}), 4.78(\mathrm{dd}, J=$ $6.4,10.1 \mathrm{~Hz}, 1-\mathrm{H}$ ）］，and an exo－methylene［ $\delta 5.17,5.84$（both $\left.\left.\mathrm{dd}, J=2.1,2.1 \mathrm{~Hz}, 15-\mathrm{H}_{2}\right)\right]$ together with four methylenes（ 2 ， $\left.3,8,9-\mathrm{H}_{2}\right)$ ，two methines（ $7,11-\mathrm{H}$ ），and three quaternary car－ bons（ $4,5,10-\mathrm{C}$ ）．

The plane structure of $\mathbf{1}$ was constructed on the basis of

Table 1．Inhibitory Activity of MeOH Extract，AcOEt－，1－BuOH－，and $\mathrm{H}_{2} \mathrm{O}$－Soluble Fractions from C．indicum on Rat Lens Aldose Reductase

|  | $\mathrm{IC}_{50}(\mu \mathrm{~g} / \mathrm{ml})$ |
| :--- | :---: |
| MeOH extract | 3.5 |
| AcOEt－soluble fraction | 1.3 |
| 1－BuOH－soluble fraction | 3.5 |
| $\mathrm{H}_{2} \mathrm{O}$－soluble fraction | 113 |



kikkanol A (1)

kikkanol B (2)
kikkanol C (3)

cis-spiroketalcnolether polyyne (4)
trons-spiroketalenolether polyyne (5)

clovanediol (6)

caryolane 1,9, 9 -diol (7)


орlopanome (8)

eupatilin (9)

nuteolin (10): $\mathrm{R}=\mathrm{H}$
11: R=Gle
12: R=GIcA


13


Gle: $\beta$-D-glucopyranosyl
GleA: $\beta$ - D -gltreopyranosiduronic acid
Chart 1
${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ correlation spectroscopy ( $\mathrm{H}-\mathrm{H}$ COSY) and heteronuclear multiple bond correlation (HMBC) experiments. Thus, the $\mathrm{H}-\mathrm{H}$ COSY experiment on $\mathbf{1}$ indicated the presence of two partial structures as shown in Fig. 1: from C-1-C-3, from C-6-C-9 and C-7-C-13. In the HMBC experiment, long-range correlations were observed between the following protons and carbons of $\mathbf{1}\left(4-\mathrm{C}\right.$ and $3,15-\mathrm{H}_{2} ; 5-\mathrm{C}$ and $6-\mathrm{H}$, $14-\mathrm{H}_{3} ; 10-\mathrm{C}$ and $\left.1-\mathrm{H}, 14-\mathrm{H}_{3}\right)$, so that the connectivities of the quaternary carbons in $\mathbf{1}$ were clarified. The above-mentioned evidence led us to confirm the skeleton of kikkanol A (1) to be 4 (15)-eudesmene 1,5,6-triol.
The ${ }^{1} \mathrm{H}$-NMR nuclear overhauser effect spectroscopy (NOESY) experiment on 1 showed NOE correlations between the signals of following proton pairs ( $1-\mathrm{H}$ and $3 \alpha, 9 \alpha-$ $\mathrm{H} ; 7-\mathrm{H}$ and $6,8 \alpha, 9 \alpha-\mathrm{H} ; 14-\mathrm{H}_{3}$ and $8 \beta, 9 \beta$-H), as depicted in Fig. 1. The coupling patterns of the 1 -proton (dd, $J=6.4$, 10.1 Hz ) and 6 -proton (brs) indicated the configurations of the 1 - and 6 -hydroxyl groups to be equatorial and axial, respectively. Furthermore, examination of the pyridine- $d_{5}$ induced solvent shift in ${ }^{1} \mathrm{H}$-NMR spectrum, ${ }^{14}$ ) showed that signals due to the $1-[\delta 3.96(\mathrm{dd}, J=5.3,11.6 \mathrm{~Hz})], 3 \alpha-[\delta 2.71$ $(\mathrm{m})], 6-[\delta 3.99(\mathrm{brs})]$, and $7-[\delta 1.67(\mathrm{~m})]$ protons in the $\mathrm{CD}_{3} \mathrm{OD}$ solution were markedly deshielded in the pyridine- $d_{5}$ solution at $\delta 4.78$ (dd, $J=6.4,10.1 \mathrm{~Hz}, 1-\mathrm{H}$ ), 3.21 (br ddd, $3 \alpha-\mathrm{H}$ ), 4.54 (br s, $6-\mathrm{H}$ ), and 2.25 (br ddd, 7-H). Thus, the



Fig. 1

Table 2. ${ }^{13} \mathrm{C}$-NMR Data on Kikkanols A (1), B (2), B 15-Pivalate (2a), and C (3)

| Position | $\mathbf{1}^{a)}$ | $\mathbf{2}^{b)}$ | $\mathbf{2 a}^{b)}$ | $\mathbf{3}^{b)}$ |
| :--- | ---: | ---: | ---: | ---: |
| C-1 | 73.0 | 75.6 | 75.8 | 25.8 |
| C-2 | 31.9 | 31.5 | 31.6 | 24.4 |
| C-3 | 31.8 | 122.7 | 124.9 | 150.4 |
| C-4 | 151.5 | 138.8 | 138.7 | 144.8 |
| C-5 | 77.0 | 135.5 | 131.5 | 35.3 |
| C-6 | 73.3 | 126.5 | 126.7 | 27.8 |
| C-7 | 44.6 | 42.4 | 42.5 | 42.2 |
| C-8 | 21.3 | 20.2 | 20.5 | 72.5 |
| C-9 | 32.0 | 33.7 | 33.7 | 78.1 |
| C-10 | 42.6 | 37.9 | 37.9 | 35.0 |
| $\mathrm{C}-11$ | 28.9 | 32.2 | 32.3 | 28.3 |
| $\mathrm{C}-12$ | $* 21.5$ | $* 19.6$ | $* 19.7$ | $* 21.1$ |
| $\mathrm{C}-13$ | 21.1 | $* 18.9$ | $* 19.1$ | $* 20.5$ |
| $\mathrm{C}-14$ | 14.0 | 16.3 | 16.3 | 20.8 |
| $\mathrm{C}-15$ | 108.9 | 64.2 | 65.0 | 194.2 |
| $\left(\mathrm{CH}_{3}\right)$ CO- |  |  | 27.0 |  |
| $\left(\mathrm{CH}_{3}\right) \mathbf{C O}-$ |  |  | 178.2 |  |

Measured in a) pyridine- $\left.d_{5}, b\right) \mathrm{CDCl}_{3}$ at 125 MHz . * Assignments may be interchangeable within the same column.
configuration of the 5 -hydroxyl group was shown to be $\alpha$ axial. On the basis of these findings, the relative stereostructure of $\mathbf{1}$ was elucidated.

The absolute configuration of $\mathbf{1}$ was determined by application of the modified Mosher's method ${ }^{15)}$ for the 1-mono$(R)$ - and (S)-2-methoxy-2-trifluorophenylacetate (MTPA esters, 1a and 1b), which were prepared by selective esterification of the 1-equatorial hydroxyl group in 1 with $(R)$ - and (S)-2-methoxy-2-trifluoromethylphenylacetic acid in the presence of 1-ethyl 3-(3-dimethylaminopropyl)-carbodiimide• $\mathrm{HCl}(\mathrm{EDC} \cdot \mathrm{HCl})$ and 4-dimethylaminopyridine (4-DMAP). Here again, the relative stereostructures of the $1-, 6-$, and $7-$ positions in 1 were confirmed by examination of the ${ }^{1} \mathrm{H}$ NMR data of $\mathbf{1 a}$ and $\mathbf{1 b}$. Thus, the ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ spectra of 1a and 1b showed signals assignable to the 1-axial, 6equatorial, and 7 -axial protons [1a: $\delta 5.46$ (dd, $J=4.6$, $11.7 \mathrm{~Hz}, 1-\mathrm{H}$ ), 3.99 (d, $J=2.3 \mathrm{~Hz}, 6-\mathrm{H}$ ), 1.36 (dddd, $J=1.6$, $2.3,4.4,12.7 \mathrm{~Hz}, 7-\mathrm{H})$; 1b: $\delta 5.46$ (dd, $J=4.9,11.6 \mathrm{~Hz}, 1-\mathrm{H})$, 3.99 (d, $J=3.1 \mathrm{~Hz}, 6-\mathrm{H}), 1.36$ (dddd, $J=1.9,3.1,4.9$, $12.1 \mathrm{~Hz}, 7-\mathrm{H})$ ]. As shown in Fig. 1, the signals due to protons attached to the 2,3 , and 15 -carbons in the ( $S$ )-MTPA ester (1b) were observed at a lower field compared with those of the $(R)$-MTPA ester (1a) [ $\Delta \delta$ : positive], while the signals due to protons on the 8,9 , and 12 -carbons in $\mathbf{1 b}$ were observed at higher fields compared with those of 1a [ $\Delta \delta$ : negative]. Thus, the absolute configuration at the 1-carbon in $\mathbf{1}$ has been shown to be $R$ and the absolute stereostructure of kikkanol A (1) has been determined as shown.

Kikkanol B (2) was isolated as a colorless oil with negative optical rotation $\left([\alpha]_{\mathrm{D}}^{22}-28.0^{\circ}\right)$. The molecular formula $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{2}$ of 2 has been determined for the molecular ion peak at $m / z 236\left(\mathrm{M}^{+}\right)$in the EI-MS of 2 and by high-resolution MS measurement. The IR spectrum of $\mathbf{2}$ showed absorption bands ascribable to hydroxyl and olefin functions at 3460,1653 , and $1060 \mathrm{~cm}^{-1}$, while its UV spectrum indicated the presence of a hetero-annular diene chromophore from a characteristic triplet with absorption maxima at $238(\log \varepsilon$ 3.93), 271 (3.05), and 336 (3.05) nm. The ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ and ${ }^{13} \mathrm{C}$-NMR (Table 2) spectra of $\mathbf{2}$ showed signals assigna-

Fig. 2

ble to a tertiary methyl $\left[\delta 0.92\left(\mathrm{~s}, 14-\mathrm{H}_{3}\right)\right.$ ], two secondary methyls [ $\delta 0.89,0.93$ (both d, $J=6.7 \mathrm{~Hz}, 12,13-\mathrm{H}_{3}$ )], a hydroxymethyl [ $\delta 4.20,4.29\left(\mathrm{ABq}, J=13.2 \mathrm{~Hz}, 15-\mathrm{H}_{2}\right)$ ], a methine bearing a hydroxyl group [ $\delta 3.54(\mathrm{dd}, J=5.8,10.1 \mathrm{~Hz}$, $1-\mathrm{H})$ ], and two olefin [ $\delta 5.66(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 6-\mathrm{H}), 5.69$ (br d, $3-\mathrm{H})$ ] together with three methylenes, two methines, and three quaternary carbons. The connectivities of the ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ and the quaternary carbons in $\mathbf{2}$ was clarified by $\mathrm{H}-\mathrm{H}$ COSY and HMBC experiments as shown in Fig. 2. Thus, long-range correlations were observed between the following protons and carbons of $2\left(4-\mathrm{C}\right.$ and $3-\mathrm{H}, 15-\mathrm{H}_{2} ; 5-\mathrm{C}$ and $6-\mathrm{H}, 15-\mathrm{H}_{2}$; $10-\mathrm{C}$ and $1-\mathrm{H}, 9-\mathrm{H}_{2}, 14-\mathrm{H}_{3}$ ) in the HMBC experiment. On the basis of this evidence, the plane structure of 2 was characterized to be $3,5(6)$-eudesmadien- 1,15 -diol.

Next, the relative stereostructure of 2 was elucidated by a NOESY experiment, which showed NOE correlations between the following proton pairs (1-H and $2 \alpha, 9 \alpha-\mathrm{H} ; 7-\mathrm{H}$ and $8 \alpha, 9 \alpha-\mathrm{H} ; 14-\mathrm{H}_{3}$ and $2 \beta, 8 \beta-\mathrm{H}$ ), as depicted in Fig. 2. In order to characterize the absolute stereostructure of $\mathbf{2}$, the 15 pivaloyl ester (2a), which was obtained by selective esterification of the 15-primary hydroxyl group in $\mathbf{2}$ with pivaloyl chloride, was subjected to the modified Mosher's method. As shown in Fig. 2, the signals due to protons attached to the 2, 3, and 15-carbons in the 1-mono-(S)-MTPA ester (2c) were observed at higher fields compared with those of the 1-mono- $(R)$-MTPA ( $\mathbf{2 b}$ ) [ $\Delta \delta$ : positive], while the signals due to protons of $8,9,11$, and 12 -carbons in $\mathbf{2 c}$ were observed at lower fields compared with those of $\mathbf{2 b}$ [ $\Delta \delta$ : negative]. Consequently, the absolute configuration at the 1-position of 2 has been elucidated as $R$ and the absolute stereostructure of kikkanol B (2) was determined to be as shown.

Kikkanol C (3) was also isolated as a colorless oil with positive optical rotation $\left([\alpha]_{D}^{22}+7.7^{\circ}\right)$. Here again, the molecular formula $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{3}$ of $\mathbf{3}$ was determined from the EI-MS $\left[m / z 252\left(\mathrm{M}^{+}\right)\right]$and by high-resolution MS measurement. In the UV spectrum of $\mathbf{3}$, an absorption maximum was observed at $234 \mathrm{~nm}(\log \varepsilon 3.97)$, suggestive of an $\alpha, \beta$-unsaturated carbonyl function. The IR spectrum of $\mathbf{3}$ showed absorption


Fig. 3

Table 3. Inhibitory Activity of Constituents from C. indicum on Rat Lens Aldose Reductase

| Compounds | $\mathrm{IC}_{50}(\mu \mathrm{~m})$ |
| :--- | :---: |
| Kikkanol A (1) | $>100(3.3)$ |
| Kikkanol C (3) | $>100(9.4)$ |
| cis-Spiroketalnenolether polyyne (4) | $>100(24.2)$ |
| trans-Spiroketalnenolether polyyne (5) | $>100(21.1)$ |
| Clovanediol (6) | 96 |
| Caryolane 1,9 $\beta$-diol (7) | 45 |
| Oplopanone (8) | $>100(24.2)$ |
| Eupatilin (9) | 25 |
| Luteolin (10) | 0.45 |
| Luteolin 7- $O$ - $\beta$-D-glucopyranoside (11) | 0.99 |
| Luteolin 7- $\beta$ - $\beta$-D-glucopyranosiduronic acid (12) | 3.1 |
| Acacetin 7- $O$-(6"- $\alpha$-L-rhamnopyranosyl)- $\beta$ - | 4.7 |
| D-glucopyranoside (13) | 1.8 |
| Chlorogenic acid (14) | 0.072 |
| Epalrestat |  |

bands assignable to hydroxyl and $\alpha, \beta$-unsaturated carbonyl functions at $3456,1669,1634$, and $1024 \mathrm{~cm}^{-1}$. The ${ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(\mathrm{CDCl}_{3}\right)$ and ${ }^{13} \mathrm{C}$-NMR (Table 2) spectra of $\mathbf{3}$ showed signals due to a tertiary methyl $\left[\delta 0.89\left(\mathrm{~s}, 14-\mathrm{H}_{3}\right)\right]$, two secondary methyls [ $\delta 0.95,0.96$ (both d, $J=6.1 \mathrm{~Hz}, 12,13-\mathrm{H}_{3}$ )], two methine bearing a hydroxyl group [ $\delta 3.49$ (d, $J=2.7 \mathrm{~Hz}, 9-$ H), 4.09 (br d, $8-\mathrm{H}$ )], an olefin [ $\delta 6.70$ (dd, $J=3.4,3.7 \mathrm{~Hz}, 3-$ $\mathrm{H})$ ], and an aldehyde [ $\delta 9.42(\mathrm{~s}, 15-\mathrm{H})$ ] together with three methylenes, two methines, and two quaternary carbons. The 8,9-dihydroxy-3-eudesmen-15-al structure of 3 was elucidated on the basis of $\mathrm{H}-\mathrm{H}$ COSY and HMBC experiments as shown in Fig. 3. Furthermore, the relative stereostructure of 3 was clarified by a NOESY experiment, which showed NOE correlations between the following proton pairs ( $1 \alpha-\mathrm{H}$ and 5 $\mathrm{H} ; 1 \beta-\mathrm{H}$ and $14-\mathrm{H}_{3} ; 5-\mathrm{H}$ and $6 \alpha-\mathrm{H} ; 6 \beta-\mathrm{H}$ and $7-\mathrm{H} ; 7-\mathrm{H}$ and $8-\mathrm{H} ; 8-\mathrm{H}$ and $9-\mathrm{H} ; 9-\mathrm{H}$ and $14-\mathrm{H}_{3}$ ). This evidence allowed us to confirm the stereostructure of kikkanol C (3). ${ }^{16)}$

Inhibitory Effect of Constituents from Flowers of $C$. indicum on Rat Lens Aldose Reductase Aldose reductase as a key enzyme in the polyol pathway is reported to catalyze the reduction of glucose to sorbitol. Sorbitol does not readily diffuse across cell membranes, and the intracellular accumulation of sorbitol has been implicated in the chronic complications of diabetes such as cataract. Since the flowers of $C$. indicum has been proved useful for the treatment of eye disease in Chinese traditional medicine, we examined the inhibitory activity of extracts (vide ante) and the components isolated from the active fractions on rat lens aldose reductase. As shown in Table 3, a flavone, luteolin (10), three
flavone glycosides ( $\mathbf{1 1} \mathbf{- 1 3}$ ), and chlorogenic acid (14) were potent inhibitories of rat lens aldose reductase, but their activity was weaker than that of a commercial synthetic aldose reductase inhibitor, epalrestat. ${ }^{17)}$ Another flavone, eupatilin $(9)$, and two sesquiterpenes $(6,7)$ exhibited less activity than luteolin (10) and flavone glycosides (11-13). Other sesquiterpenes $(\mathbf{1}, \mathbf{3}, \mathbf{8})$ and two polyacetylenes $(\mathbf{4}, \mathbf{5})$ exhibited little activity. Taking this data into account, these results indicate that flavonoids and chlorogenic acid are active principles of this natural medicine.

## Experimental

The following instruments were used to obtain physical data: melting points, Yanagimoto micro-melting point apparatus MP-500D (values are uncorrected); specific rotations, Horiba SEPA-300 digital polarimeter ( $l=$ 5 cm ); UV spectra, Shimadzu UV-1200 spectrometer; IR spectra, Shimadzu FTIR-8100 spectrometer; EI-MS and high-resolution MS, JEOL JMS-GCMATE mass spectrometer; ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectra, JNM-LA500 ( 500 MHz ) spectrometer; ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectra, JNM-LA500 $(125 \mathrm{MHz})$ spectrometer with tetramethylsilane as an internal standard.
The following experimental conditions were used for chromatography: or-dinary-phase silica gel column chromatography, Silica gel BW-200 (Fuji Silysia Chemical, Ltd., 150-350 mesh); reversed-phase silica gel column chromatography, Chromatorex ODS DM1020T (Fuji Silysia Chemical, Ltd., 100-200 mesh); TLC, pre-coated TLC plates with Silica gel $60 \mathrm{~F}_{254}$ (Merck, 0.25 mm ) (ordinary phase) and Silica gel RP-18 60F 254 (Merck, 0.25 mm ) (reversed phase); reversed-phase HPTLC, pre-coated TLC plates with Silica gel RP-18 60 $\mathrm{WF}_{254 \mathrm{~S}}$ (Merck, 0.25 mm ); detection was achieved by spraying with $1 \% \mathrm{Ce}\left(\mathrm{SO}_{4}\right)_{2}-10 \%$ aqueous $\mathrm{H}_{2} \mathrm{SO}_{4}$ and heating.

Extraction and Isolation Dried flowers of C. indicum ( 5.8 kg , cultivated in China and purchased from Koshiro Co. Ltd., Osaka) were finely minced and extracted with methanol under reflux. Evaporation of the solvent under reduced pressure gave the MeOH extract $(1650 \mathrm{~g})$. The MeOH extract $(1600 \mathrm{~g})$ was partitioned in a $\mathrm{AcOEt}-\mathrm{H}_{2} \mathrm{O}(1: 1)$ mixture, and the aqueous layer was further extracted with $1-\mathrm{BuOH}$. Removal of the solvent under reduced pressure from the AcOEt-, $1-\mathrm{BuOH}-$, and water-soluble portions yielded $438 \mathrm{~g}, 226 \mathrm{~g}$, and 900 g of residues, respectively.

The AcOEt-soluble portion $(150 \mathrm{~g})$ was subjected to ordinary-phase silica gel column chromatography $\left[3.0 \mathrm{~kg}, n\right.$-hexane- $\mathrm{AcOEt} \rightarrow \mathrm{CHCl}_{3}-$ $\mathrm{MeOH} \rightarrow \mathrm{CHCl}_{3}-\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ ] to afford twelve fractions [fr. $1(14.2 \mathrm{~g})$, fr. 2 $(10.7 \mathrm{~g})$, fr. $3(5.6 \mathrm{~g})$, fr. $4(5.6 \mathrm{~g})$, fr. $5(10.8 \mathrm{~g})$, fr. $6(7.3 \mathrm{~g})$, fr. $7(8.1 \mathrm{~g})$, fr. 8 $(3.7 \mathrm{~g})$, fr. $9(3.4 \mathrm{~g})$, fr. $10(6.3 \mathrm{~g})$, fr. $11(56.4 \mathrm{~g})$, fr. $12(17.9 \mathrm{~g})$ ]. Fraction 3 $(5.6 \mathrm{~g})$ was further subjected to reversed-phase silica gel column chromatography to furnish cis-spiroketalenolether polyyne (4, 1.5 g ). Fraction 5 $(10.8 \mathrm{~g})$ was separated by reversed-phase silica gel column chromatography and, finally, HPLC [YMC-pack ODS-A, $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ ] to give trans-spiroketalenolether polyyne $(\mathbf{5}, 285 \mathrm{mg})$. Fraction $8(3.7 \mathrm{~g})$ was purified by reversedphase silica gel column chromatography and finally HPLC [YMC-pack ODS-A, $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ ] to furnish kikkanols A $(\mathbf{1}, 15 \mathrm{mg}), \mathrm{B}(\mathbf{2}, 6 \mathrm{mg})$, and C $(\mathbf{3}, 13 \mathrm{mg})$, oplopanone $(\mathbf{8}, 228 \mathrm{mg})$ and eupatilin $(9,15 \mathrm{mg})$. Fraction 10 $(6.3 \mathrm{~g})$ was subjected to reversed-phase silica gel column chromatography and finally HPLC to furnish clovanediol $(6,32 \mathrm{mg})$, caryolane $1,9 \beta$-diol ( 7 , $14 \mathrm{mg})$, and luteolin ( $\mathbf{1 0}, 32 \mathrm{mg}$ ). The $1-\mathrm{BuOH}$-soluble portion $(182 \mathrm{~g})$ was also subjected to ordinary-phase silica gel column chromatography $\left[\mathrm{CHCl}_{3}-\right.$ $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ ] to afford seven fractions [fr. $1(7.5 \mathrm{~g})$, fr. $2(38.6 \mathrm{~g})$, fr. 3 $(21.9 \mathrm{~g})$, fr. $4(28.2 \mathrm{~g})$, fr. $5(14.2 \mathrm{~g})$, fr. $6(22.2 \mathrm{~g})$, fr. $7(32.3 \mathrm{~g})]$. Fraction 3 $(21.9 \mathrm{~g})$ was further subjected to reversed-phase silica gel column chromatography to furnish luteolin 7-O- $\beta$-d-glucopyranoside ( $\mathbf{1 1}, 5.9 \mathrm{~g}$ ), chlorogenic acid $(14,610 \mathrm{mg})$, and acacetin $7-O-\left(6^{\prime \prime}-\alpha\right.$-L-rhamnopyranosyl)- $\beta$-dglucopyranoside ( $\mathbf{1 3}, 800 \mathrm{mg}$ ). Fraction $5(14.2 \mathrm{~g})$ was purified by reversedphase silica gel column chromatography to furnish luteolin $7-O-\beta$-d-glucopyranosiduronic acid $(\mathbf{1 2}, 2.2 \mathrm{~g})$. These constituents were identified by comparison of their physical data with those of authentic samples $(\mathbf{6}, \mathbf{9}, \mathbf{1 3})$ or with reported values. ${ }^{6,8-12)}$
Kikkanol A (1): Colorless fine crystals from $n$-hexane-AcOEt, mp $165-$ $167^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}^{27}+57.2^{\circ}(c=2.1, \mathrm{MeOH})$. High-resolution EI-MS: Calcd for $\mathrm{C}_{15} \mathrm{H}_{26} \mathrm{O}_{3}\left(\mathrm{M}^{+}\right): 254.1882$. Found: 254.1899. IR (film): 3532, 3456, 2958, $1647,1073 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\right.$ pyridine- $d_{5}$ ) $\delta: 1.02,1.03$ ( 3 H each, both d, $J=$ $6.4 \mathrm{~Hz}, 12$ and $\left.13-\mathrm{H}_{3}\right), 1.66\left(3 \mathrm{H}, \mathrm{s}, 14-\mathrm{H}_{3}\right), 1.84(1 \mathrm{H}, \mathrm{m}, 8 \alpha-\mathrm{H}), 1.96(1 \mathrm{H}$, $\mathrm{m}, 8 \beta-\mathrm{H}), 1.98(1 \mathrm{H}, \mathrm{dq}, J=3.7,6.4 \mathrm{~Hz}, 11-\mathrm{H}), 2.11\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right), 2.25(1 \mathrm{H}$, br ddd, $7-\mathrm{H}), 2.26(1 \mathrm{H}, \mathrm{m}, 9 \beta-\mathrm{H}), 2.33(1 \mathrm{H}$, ddd-like, $3 \beta-\mathrm{H}), 2.34(1 \mathrm{H}, \mathrm{m}$, $9 \alpha-\mathrm{H}), 3.21(1 \mathrm{H}$, br ddd, $3 \alpha-\mathrm{H}), 4.54(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 6-\mathrm{H}), 4.78(1 \mathrm{H}, \mathrm{dd}, J=6.4$,
$10.1 \mathrm{~Hz}, 1-\mathrm{H}), 5.17,5.84\left(1 \mathrm{H}\right.$ each, both dd, $\left.J=2.1,2.1 \mathrm{~Hz}, 15-\mathrm{H}_{2}\right)$; $\left(\mathrm{CD}_{3} \mathrm{OD}\right) \delta: 0.95\left(3 \mathrm{H}, \mathrm{s}, 14-\mathrm{H}_{3}\right), 0.95,0.97(3 \mathrm{H}$ each, both d, $J=6.2 \mathrm{~Hz}, 12$ and $\left.13-\mathrm{H}_{3}\right), 1.56,1.58\left(1 \mathrm{H}\right.$ each, both $\left.\mathrm{m}, 8-\mathrm{H}_{2}\right), 1.61,1.63(1 \mathrm{H}$ each, both $\left.\mathrm{m}, 9-\mathrm{H}_{2}\right), 1.65(1 \mathrm{H}, \mathrm{m}, 11-\mathrm{H}), 1.67(1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}), 1.68\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right), 2.11$ $(1 \mathrm{H}, \mathrm{m}, 3 \beta-\mathrm{H}), 2.71(1 \mathrm{H}, \mathrm{m}, 3 \alpha-\mathrm{H}), 3.96(1 \mathrm{H}, \mathrm{dd}, J=5.3,11.6 \mathrm{~Hz}, 1-\mathrm{H})$, $3.99(1 \mathrm{H}$, br s, $6-\mathrm{H}), 4.91,5.26\left(1 \mathrm{H}\right.$ each, both dd-like, $\left.15-\mathrm{H}_{2}\right) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ (pyridine- $d_{5}$ ) $\delta_{\mathrm{c}}$ : given in Table 2. EI-MS $m / z(\%): 254\left(\mathrm{M}^{+}, 2\right), 236\left(\mathrm{M}^{+}-\right.$ $\left.\mathrm{H}_{2} \mathrm{O}, 6\right), 123$ (100).

Kikkanol B (2): Colorless oil, $[\alpha]_{\mathrm{D}}^{22}-28.0^{\circ}\left(c=0.1, \mathrm{CHCl}_{3}\right)$. High-resolution EI-MS: Calcd for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{2}\left(\mathrm{M}^{+}\right): 236.1776$. Found: 236.1774. UV $(\mathrm{MeOH}, \mathrm{nm}, \log \varepsilon) 238$ (3.93), 271 (3.05), 336 (3.05). IR (film) 3460, 2962, 1653, $1060 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.89,0.93$ (3H each, both d, $J=$ $6.7 \mathrm{~Hz}, 12$ and $\left.13-\mathrm{H}_{3}\right), 0.92\left(3 \mathrm{H}, \mathrm{s}, 14-\mathrm{H}_{3}\right), 1.33(1 \mathrm{H}, \mathrm{m}, 9 \alpha-\mathrm{H}), 1.43(1 \mathrm{H}$, dddd, $J=3.4,11.0,13.7,16.4 \mathrm{~Hz}, 8 \beta-\mathrm{H}), 1.68(1 \mathrm{H}, \mathrm{m}, 8 \alpha-\mathrm{H}), 1.71(1 \mathrm{H}, \mathrm{m}$, $11-\mathrm{H}), 1.90(1 \mathrm{H}, \mathrm{ddd}, J=3.4,3.4,12.5 \mathrm{~Hz}, 9 \beta-\mathrm{H}), 2.11(1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}), 2.18$ $(1 \mathrm{H}, \mathrm{m}, 2 \beta-\mathrm{H}), 2.43(1 \mathrm{H}, \mathrm{ddd}, J=5.5,5.8,13.6 \mathrm{~Hz}, 2 \alpha-\mathrm{H}), 3.54(1 \mathrm{H}, \mathrm{dd}$, $J=5.8,10.1 \mathrm{~Hz}, 1-\mathrm{H}), 4.20,4.29\left(2 \mathrm{H}, \mathrm{ABq}, J=13.2 \mathrm{~Hz}, 15-\mathrm{H}_{2}\right), 5.66(1 \mathrm{H}, \mathrm{d}$, $J=2.1 \mathrm{~Hz}, 6-\mathrm{H}), 5.69(1 \mathrm{H}$, br d, $3-\mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta_{\mathrm{c}}$ : given in Table 2. EI-MS $m / z(\%): 236\left(\mathrm{M}^{+}, 34\right), 218\left(\mathrm{M}^{+}-\mathrm{H}_{2} \mathrm{O}, 79\right), 193$ (100).

Kikkanol C (3): Colorless oil, $[a]_{\mathrm{D}}^{22}+7.7^{\circ}\left(c=0.1, \mathrm{CHCl}_{3}\right)$. High-resolution EI-MS: Calcd for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{3}\left(\mathrm{M}^{+}\right): 252.1725$. Found: 252.1727. UV ( $\mathrm{MeOH}, \mathrm{nm}, \log \varepsilon$ ): 234 (3.97). IR (film): 3456, 2939, 1669, 1636, $1024 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.89\left(3 \mathrm{H}, \mathrm{s}, 14-\mathrm{H}_{3}\right), 0.95,0.96(3 \mathrm{H}$ each, both d, $J=6.1 \mathrm{~Hz}, 12$ and $\left.13-\mathrm{H}_{3}\right), 1.04(1 \mathrm{H}$, ddd, $J=5.8,12.5,12.8 \mathrm{~Hz}, 6 \beta$ H), $1.15(1 \mathrm{H}, \mathrm{ddd}, J=3.3,5.8,13.7 \mathrm{~Hz}, 1 \alpha-\mathrm{H}), 1.50(1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}), 1.53(1 \mathrm{H}$, $\mathrm{m}, 11-\mathrm{H}), 1.79(1 \mathrm{H}$, ddd, $J=3.3,3.7,12.8 \mathrm{~Hz}, 6 \alpha-\mathrm{H}), 2.40\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right)$, $2.50(1 \mathrm{H}, \mathrm{m}, 1 \beta-\mathrm{H}), 2.52(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 3.49(1 \mathrm{H}, \mathrm{d}, J=2.7 \mathrm{~Hz}, 9-\mathrm{H}), 4.09$ $(1 \mathrm{H}, \mathrm{br} \mathrm{d}, 8-\mathrm{H}), 6.70(1 \mathrm{H}, \mathrm{dd}, J=3.4,3.7 \mathrm{~Hz}, 3-\mathrm{H}), 9.42(1 \mathrm{H}, \mathrm{s}, 15-\mathrm{H}) .{ }^{13} \mathrm{C}-$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta_{\mathrm{c}}$ : given in Table 2. EI-MS: $m / z(\%): 252\left(\mathrm{M}^{+}, 6\right), 234$ $\left(\mathrm{M}^{+}-\mathrm{H}_{2} \mathrm{O}, 19\right), 166$ (100).

Preparation of the ( $R$ )-MTPA Ester (1a) and the ( $\boldsymbol{S}$ )-MTPA Ester (1b) from Kikkanol A (1) A solution of $1(2.0 \mathrm{mg}, 7.9 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(1.0 \mathrm{ml})$ was treated with $(R)$-MTPA $(9.2 \mathrm{mg}, 39.4 \mu \mathrm{~mol})$ in the presence of EDC $\cdot \mathrm{HCl}(7.6 \mathrm{mg}, 39.4 \mu \mathrm{~mol})$ and $4-\mathrm{DMAP}(2.9 \mathrm{mg}, 23.6 \mu \mathrm{~mol})$ and the mixture was heated under reflux for 6 h . After cooling, it was poured into ice-water and the whole was extracted with AcOEt. The AcOEt extract was successively washed with $5 \%$ aqueous HCl , saturated aqueous $\mathrm{NaHCO}_{3}$, and brine, then dried over $\mathrm{MgSO}_{4}$ powder and filtered. Removal of the solvent from the filtrate under reduced pressure furnished a residue, which was purified by silica gel column chromatography $(0.2 \mathrm{~g}, n$-hexane-AcOEt $=5: 1)$ to give $\mathbf{1 a}(1.7 \mathrm{mg})$ and $\mathbf{1}(0.9 \mathrm{mg})$. Through a similar procedure, $\mathbf{1 b}(1.2 \mathrm{mg})$ was prepared from $1(2.0 \mathrm{mg})$ using $(S)$-MTPA $(9.2 \mathrm{mg}), \mathrm{EDC} \cdot \mathrm{HCl}(7.6 \mathrm{mg})$ and 4-DMAP ( 2.9 mg ).

1a: Colorless oil, ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.97,0.98$ (3H each, both d, $J=$ $6.6 \mathrm{~Hz}, 12$ and $\left.13-\mathrm{H}_{3}\right), 1.01\left(3 \mathrm{H}, \mathrm{s}, 14-\mathrm{H}_{3}\right), 1.36(1 \mathrm{H}$, dddd, $J=1.6,2.3,4.4$, $12.7 \mathrm{~Hz}, 7-\mathrm{H}), 1.43(1 \mathrm{H}, \mathrm{m}, 8-\mathrm{H}), 1.56(1 \mathrm{H}, \mathrm{m}, 11-\mathrm{H}), 1.65(1 \mathrm{H}, \mathrm{m}, 8-\mathrm{H})$, $1.66\left(2 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}_{2}\right), 1.71(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 1.99(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 2.25(1 \mathrm{H}$, ddd, $J=4.5,5.1,13.4 \mathrm{~Hz}, 3-\mathrm{H}), 2.75(1 \mathrm{H}$, ddd-like, $3-\mathrm{H}), 3.52(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.99$ $(1 \mathrm{H}, \mathrm{d}, J=2.3 \mathrm{~Hz}, 6-\mathrm{H}), 5.04,5.30\left(1 \mathrm{H}\right.$ each, both s, $\left.15-\mathrm{H}_{2}\right), 5.46(1 \mathrm{H}$, dd, $J=4.6,11.7 \mathrm{~Hz}, 1-\mathrm{H}), 7.39-7.54(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph})$.

1b: Colorless oil, ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.96,0.97$ (3H each, both d, $J=$ $6.4 \mathrm{~Hz}, 12$ and $\left.13-\mathrm{H}_{3}\right), 1.01\left(3 \mathrm{H}, \mathrm{s}, 14-\mathrm{H}_{3}\right), 1.27(1 \mathrm{H}$, dddd, $J=3.6,4.9,7.3$, $14.1 \mathrm{~Hz}, 8-\mathrm{H}), 1.36(1 \mathrm{H}$, dddd, $J=1.9,3.1,4.9,12.1 \mathrm{~Hz}, 7-\mathrm{H}), 1.55(1 \mathrm{H}, \mathrm{m}$, $8-\mathrm{H}), 1.56(1 \mathrm{H}, \mathrm{m}, 11-\mathrm{H}), 1.62(2 \mathrm{H}, \mathrm{m}, 9-\mathrm{H} 2), 1.82(1 \mathrm{H}$, dddd, $J=5.8,8.5$, $11.6,13.7 \mathrm{~Hz}, 2-\mathrm{H}), 2.02(1 \mathrm{H}$, dddd, $J=4.9,6.4,7.3,13.7 \mathrm{~Hz}, 2-\mathrm{H}), 2.28$ $(1 \mathrm{H}, \mathrm{ddd}, J=5.8,7.3,14.4 \mathrm{~Hz}, 3-\mathrm{H}), 2.77(1 \mathrm{H}, \mathrm{ddd}, J=6.4,8.5,14.4 \mathrm{~Hz}, 3-$ H), $3.52(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.99(1 \mathrm{H}, \mathrm{d}, J=3.1 \mathrm{~Hz}, 6-\mathrm{H}), 5.05,5.31(1 \mathrm{H}$ each, both s, $\left.15-\mathrm{H}_{2}\right), 5.46(1 \mathrm{H}, \mathrm{dd}, J=4.9,11.6 \mathrm{~Hz}, 1-\mathrm{H}), 7.38-7.53(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph})$.

Treatment of 2 with Pivaloyl Chloride and Pyridine A solution of 2 $(2.1 \mathrm{mg}, 8.9 \mu \mathrm{~mol})$ in pyridine $(0.5 \mathrm{ml})$ was treated with pivaloyl chloride $(30 \mu \mathrm{l})$ and the mixture was stirred at room temperature for 2 h . The reaction mixture was poured into ice-water and the whole was extracted with AcOEt. The AcOEt extract was treated in the usual manner to give a residue, which was purified by silica gel column chromatography $(0.5 \mathrm{~g}, n$-hexane-AcOEt $=$ $5: 1$ ) to furnish $\mathbf{2 a}(2.2 \mathrm{mg})$.
2a: Colorless oil, ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.89,0.93$ (3H each, both d, $J=$ $6.7 \mathrm{~Hz}, 12$ and $\left.13-\mathrm{H}_{3}\right), 0.91\left(3 \mathrm{H}, \mathrm{s}, 14-\mathrm{H}_{3}\right), 1.19\left(9 \mathrm{H}, \mathrm{s}\right.$, pivaloyl $\left.-\mathrm{CH}_{3}\right), 1.30$ $(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 1.42(1 \mathrm{H}, \mathrm{m}, 8-\mathrm{H}), 1.65(2 \mathrm{H}, \mathrm{m}, 8$ and $11-\mathrm{H}), 1.89(1 \mathrm{H}$, ddd, $J=3.1,3.3,12.2 \mathrm{~Hz}, 9-\mathrm{H}), 2.06(1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}), 2.18(1 \mathrm{H}, \mathrm{ddd}, J=1.9,10.1$, $18.6 \mathrm{~Hz}, 2-\mathrm{H}), 2.43(1 \mathrm{H}$, ddd, $J=5.8,6.1,18.6 \mathrm{~Hz}, 2-\mathrm{H}), 3.53(1 \mathrm{H}, \mathrm{dd}, J=$ $6.1,10.1 \mathrm{~Hz}, 1-\mathrm{H}), 4.63,4.71\left(2 \mathrm{H}, \mathrm{ABq}, J=12.5 \mathrm{~Hz}, 15-\mathrm{H}_{2}\right), 5.58(1 \mathrm{H}, \mathrm{d}$, $J=1.8 \mathrm{~Hz}, 6-\mathrm{H}), 5.69(1 \mathrm{H}, \mathrm{dd}, J=1.9,5.8 \mathrm{~Hz}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta_{\mathrm{c}}$ : given in Table 2. EI-MS $m / z(\%): 320\left(\mathrm{M}^{+}, 3\right), 302\left(\mathrm{M}^{+}-\mathrm{H}_{2} \mathrm{O}, 1\right), 218$ (100).

Preparation of the $(R)$-MTPA Ester (2b) and the $(S)$-MTPA Ester (2c) from Kikkanol B (2) A solution of 2a ( $1.0 \mathrm{mg}, 3.1 \mu \mathrm{~mol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(0.5 \mathrm{ml})$ was treated with $(R)$-MTPA $(7.5 \mathrm{mg}, 32.0 \mu \mathrm{~mol})$ in the presence of $\mathrm{EDC} \cdot \mathrm{HCl}(6.1 \mathrm{mg}, 32.0 \mu \mathrm{~mol})$ and 4-DMAP $(2.3 \mathrm{mg}, 19.2 \mu \mathrm{~mol})$ and the mixture was stirred at room temperature for 12 h . The reaction mixture was poured into ice-water and the whole was extracted with AcOEt. Work-up of the AcOEt extract as described above gave a product which was purified by silica gel column chromatography ( 0.5 g , $n$-hexane $-\mathrm{AcOEt}=5: 1$ ) to yield $\mathbf{2 b}(1.1 \mathrm{mg}) . \mathbf{2 c}(1.0 \mathrm{mg})$ was obtained from $\mathbf{2 a}(1.0 \mathrm{mg})$ using $(S)$-MTPA $(7.5 \mathrm{mg}), \mathrm{EDC} \cdot \mathrm{HCl}(6.1 \mathrm{mg})$ and 4-DMAP $(2.7 \mathrm{mg})$ by the same procedure.

2b: Colorless oil, ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.89,0.91(3 \mathrm{H}$ each, both d, $J=$ $7.3 \mathrm{~Hz}, 12$ and $\left.13-\mathrm{H}_{3}\right), 0.92\left(3 \mathrm{H}, \mathrm{s}, 14-\mathrm{H}_{3}\right), 1.19\left(9 \mathrm{H}, \mathrm{s}\right.$, pivaloyl- $\left.\mathrm{CH}_{3}\right), 1.35$ $(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 1.40(1 \mathrm{H}, \mathrm{m}, 8-\mathrm{H}), 1.67(1 \mathrm{H}, \mathrm{m}, 11-\mathrm{H}), 1.74(1 \mathrm{H}, \mathrm{m}, 8-\mathrm{H})$, $2.02(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 2.05(1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}), 2.34(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 2.63(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H})$, $3.53(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 4.64,4.71\left(2 \mathrm{H}, \mathrm{ABq}, J=12.8 \mathrm{~Hz}, 15-\mathrm{H}_{2}\right), 4.94(1 \mathrm{H}, \mathrm{dd}$, $J=5.8,10.7 \mathrm{~Hz}, 1-\mathrm{H}), 5.65(1 \mathrm{H}, \mathrm{d}, J=2.7 \mathrm{~Hz}, 6-\mathrm{H}), 5.68(1 \mathrm{H}$, dd-like, 3-H), $7.41-7.55(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph})$.

2c: Colorless oil, ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.87,0.91$ (3H each, both d, $J=$ $7.4 \mathrm{~Hz}, 12$ and $\left.13-\mathrm{H}_{3}\right), 0.88\left(3 \mathrm{H}, \mathrm{s}, 14-\mathrm{H}_{3}\right), 1.19\left(9 \mathrm{H}\right.$, s, pivaloyl- $\left.\mathrm{CH}_{3}\right), 1.32$ $(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 1.35(1 \mathrm{H}, \mathrm{m}, 8-\mathrm{H}), 1.64(1 \mathrm{H}, \mathrm{m}, 11-\mathrm{H}), 1.72(1 \mathrm{H}, \mathrm{m}, 8-\mathrm{H})$, $1.99(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 2.05(1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}), 2.37(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 2.67(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H})$, $3.54(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 4.65,4.71\left(2 \mathrm{H}, \mathrm{ABq}, J=13.2 \mathrm{~Hz}, 15-\mathrm{H}_{2}\right), 4.95(1 \mathrm{H}, \mathrm{dd}$, $J=5.8,10.3 \mathrm{~Hz}, 1-\mathrm{H}), 5.65(1 \mathrm{H}, \mathrm{d}, J=2.1 \mathrm{~Hz}, 6-\mathrm{H}), 5.70(1 \mathrm{H}$, dd-like, $3-\mathrm{H})$, $7.40-7.55$ ( $5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}$ ).

Bioassay Aldose reductase activity was assayed by the method described in a previous paper. ${ }^{4 a)}$ The supernatant fluid of rat lens homogenate was used as the crude enzyme. The incubation mixture contained 135 mm Na , K-phosphate buffer ( pH 7.0 ), $100 \mathrm{~mm} \mathrm{Li}_{2} \mathrm{SO}_{4}, 0.03 \mathrm{~mm} \mathrm{NADPH}, 1 \mathrm{~mm}$ DL-glyceraldehyde as a substrate, and $100 \mu \mathrm{l}$ of enzyme fraction, with or without $25 \mu 1$ of sample solution, in a total volume of 0.5 ml . The reaction was initiated by the addition of NADPH at $30^{\circ} \mathrm{C}$. After 30 min , the reaction was stopped by the addition of $150 \mu \mathrm{l} 0.5 \mathrm{~N} \mathrm{HCl}$. Then, 0.5 ml 6 N NaOH containing 10 mm imidazole was added, and the solution was heated at $60^{\circ} \mathrm{C}$ for 10 min to convert NADP to a fluorescent product. Fluorescence was measured using a spectrofluorometer (Type 650-10, Hitachi, Japan) at an excitation wavelength of 360 nm and an emission wavelength of 460 nm .

## References and Notes

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