# Medicinal Flowers. VI. ${ }^{1)}$ Absolute Stereostructures of Two New Flavanone Glycosides and a Phenylbutanoid Glycoside from the Flowers of Chrysanthemum indicum L.: Their Inhibitory Activities for Rat Lens Aldose Reductase 

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#### Abstract

Two new flavanone glycosides, (2S)- and (2R)-eriodictyol 7- $O$ - $\beta$-d-glucopyranosiduronic acids, and a new phenylbutanoid glycoside, ( $2 S, 3 S$ )-1-phenyl-2,3-butanediol 3- $O$ - $\beta$-d-glucopyranoside, were isolated from the flowers of Chrysanthemum indicum L. cultivated in China together with eight flavonoids. The absolute stereostructures of the new compounds were determined on the basis of chemical and physicochemical evidence. Both of the new flavanone glycosides were found to show inhibitory activity for rat lens aldose reductase.


Key words aldose reductase inhibitor; Chrysanthemum indicum; medicinal flower; (2S)-eriodictyol 7-O- $\beta$-d-glucopyranosiduronic acid; ( $2 R$ )-eriodictyol 7-O- $\beta$-d-glucopyranosiduronic acid; ( $2 S, 3 S$ )-1-phenyl-2,3-butanediol 3-O- $\beta$-d-glucopyranoside

The Compositae plant, the flowers of Chrysanthemum indicum L. (Chrysanthemi Indici Flos) have been prescribed for antiinflammation, analgesic, and antipyretic purposes and the treatment of eye disease in Chinese traditional preparations. In the course of our characterization studies on the bioactive constituents from medicinal foodstuffs ${ }^{2-9}$ ) and medicinal flowers, ${ }^{1,10-13)}$ we have reported the isolation and structural elucidation of three eudesmane-type sesquiterpenes, kikkanols $\mathrm{A}, \mathrm{B}$, and C , and five germacrane-type sesquiterpenes, kikkanols D, D monoacetate, E, F, and F monoacetate, from the methanolic extract of the flowers of $C$. indicum cultivated in China. ${ }^{10,11)}$ In addition, the methanolic extract and several constituents were found to show inhibitory activities against rat lens aldose reductase ${ }^{10)}$ and nitric oxide (NO) production in lipopolysaccharide (LPS)-activated mouse peritoneal macrophages. ${ }^{11)}$ As a continuing study on this natural medicine, we recently isolated two new flavanone glycosides, $(2 S)$ - and $(2 R)$-eriodictyol $7-O-\beta$-Dglucopyranosiduronic acids (1,2), and a new phenylbutanoid glycoside, ( $2 S, 3 S$ )-1-phenyl-2,3-butanediol 3-O- $\beta$-d-glucopyranoside (3). This paper deals with the isolation and absolute stereostructure elucidation of new constituents (1-3) from the flowers of $C$. indicum as well as the inhibitory activity of $\mathbf{1}$ and $\mathbf{2}$ for rat lens aldose reductase.

Isolation of Chemical Constituents from the Dried Flowers of C. indicum The methanolic extract from the flowers of C. indicum was partitioned into a mixture of ethyl acetate and water to furnish the ethyl acetate-soluble portion and an aqueous phase. The aqueous phase was further extracted with 1-butanol to give a 1-butanol-soluble portion and a water-soluble portion as described. ${ }^{10)}$ The 1-butanolsoluble portion was subjected to silica gel and octadecyl silica gel (ODS) column chromatography and finally HPLC to furnish three new constituents called (2S)-eriodictyol 7-O- $\beta$ -D-glucopyranosiduronic acid (1, $0.0027 \%$ from the natural medicine), ( $2 R$ )-eriodictyol 7-O- $\beta$-D-glucopyranosiduronic acid (2, 0.0023\%), and (2S,3S)-1-phenyl-2,3-butanediol 3-O-$\beta$-d-glucopyranoside (3, $0.019 \%$ ), and three known flavonoids, apigenin $7-O-\beta$-D-glucopyranoside ( $=$ apigetrin, 4, ${ }^{14}$ ) $0.014 \%$ ), diosmetin $7-O-\beta$-D-glucopyranoside ( $\mathbf{5},{ }^{14)} 0.086 \%$ ),
and quercetin 3,7-di- $O$ - $\beta$-D-glucopyranoside ( $6,{ }^{15}$ ) $0.0038 \%$ ), together with luteolin, ${ }^{10)}$ luteolin 7-O- $\beta$-D-glucopyranoside, ${ }^{10)}$ luteolin $7-O-\beta$-d-glucopyranosiduronic acid, ${ }^{10)}$ acacetin $7-O-$ ( $6^{\prime \prime}-\alpha$-L-rhamnopyranosyl)- $\beta$-d-glucopyranoside, ${ }^{10}$ ) and eupatilin. ${ }^{10}$

Absolute Stereostructures of (2S)- and (2R)-Eriodictyol 7-O- $\boldsymbol{\beta}$-d-Glucopyranosiduronic Acids (1, 2) (2S)-Eriodictyol $7-O-\beta$-D-glucopyranosiduronic acid (1) was isolated as a yellow powder with negative optical rotation $\left([\alpha]_{\mathrm{D}}^{26}-35.6^{\circ}\right)$. The positive-ion fast atom bombardment (FAB)-MS of 1 showed a quasimolecular ion peak at $m / z 487(\mathrm{M}+\mathrm{Na})^{+}$, while a quasimolecular ion peak was observed at $m / z 463$ $(\mathrm{M}-\mathrm{H})^{-}$together with a fragment ion peak at $\mathrm{m} / \mathrm{z} 287$ ( $\left.\mathrm{M}-\mathrm{C}_{6} \mathrm{H}_{9} \mathrm{O}_{6}\right)^{-}$in the negative-ion FAB-MS. The molecular formula $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{O}_{12}$ of $\mathbf{1}$ was determined from the quasimolecular ion peaks and by high-resolution MS measurement. The IR spectrum of $\mathbf{1}$ showed absorption bands at 3417, $1719,1655,1637$, and $1071 \mathrm{~cm}^{-1}$ ascribable to hydroxyl, carboxyl, chelated carbonyl, aromatic ring, and ether functions. In the UV spectrum of $\mathbf{1}$, absorption maxima were observed at $248(\log \varepsilon 4.21)$ and 326 (3.55) nm, suggestive of the flavanone structure. The ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right)$ and ${ }^{13} \mathrm{C}$ NMR (Table 1) spectra of $\mathbf{1}$, which was constructed on the basis of ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}^{-1} \mathrm{H}$ correlation spectroscopies ( $\mathrm{H}-\mathrm{H}$ and $\mathrm{C}-\mathrm{H}$ COSY), showed signals assignable to a dihydropyrone moiety in flavanone structure by a characteristic ABX type coupling pattern $\{[\delta 2.74(\mathrm{dd}, J=3.0,17.4 \mathrm{~Hz}), 3.25$ (dd, $J=12.5,17.4 \mathrm{~Hz}$ ), $\left.3-\mathrm{H}_{2}\right], 5.45$ (dd, $J=3.0,12.5 \mathrm{~Hz}, 2-$ $\mathrm{H})\}$, an anomeric proton [ $\delta 5.07\left(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 1^{\prime \prime}-\mathrm{H}\right)$ ], five aromatic protons [ $\delta 6.13,6.16$ (both d, $J=2.1 \mathrm{~Hz}, 6,8-\mathrm{H}$ ), $6.75\left(2 \mathrm{H}, \mathrm{brs}, 5^{\prime}, 6^{\prime}-\mathrm{H}\right), 6.89\left(\mathrm{~s}, 2^{\prime}-\mathrm{H}\right)$ ], and hydroxyl and chelated hydroxyl groups [ $\delta 8.96(2 \mathrm{H}, \mathrm{brs}), 12.01$ (brs), $-\mathrm{OH}]$.
(2R)-Eriodictyol 7-O- $\beta$-D-glucopyranosiduronic acid (2) was also isolated as a yellow powder with negative optical rotation $\left([\alpha]_{D}^{24}-54.5^{\circ}\right)$ and its IR, UV, MS, and NMR spectra were very similar to those of $\mathbf{1}$. Acid hydrolysis of $\mathbf{1}$ and $\mathbf{2}$ with $5 \%$ sulfuric acid $\left(\mathrm{H}_{2} \mathrm{SO}_{4}\right)$ furnished d-glucuronic acid, which was identified by GLC analysis of the trimethylsilyl (TMS) thiazolidine derivative. ${ }^{16)}$ Enzymatic hydrolysis of the


erindietyol (7)
(23,35)-1-phenyl-2,3-butantediol (8)
Chart 1
mixture of 1 and 2 with $\beta$-glucuronidase liberated eriodictyol (7). ${ }^{17)}$ In the heteronuclear multiple bond correlation (HMBC) experiment on $\mathbf{1}$ and 2, a long-range correlation was observed between the anomeric proton and 7 -carbon (Fig. 1), so that the connectivities between the aglycon and D-glucuronic acid moiety were elucidated.

The absolute stereostructures of $\mathbf{1}$ and $\mathbf{2}$ were determined on the basis of circular dichronic (CD) spectroscopic analysis. Thus, the CD spectra of $\mathbf{1}$ showed negative Cotton effects $[248(\Delta \varepsilon+0.24), 287(-3.59) \mathrm{nm}]$, which indicated the absolute configuration of the 2-position to be $S$ orientation, while positive Cotton effects [252(-1.00), $292(+3.66)]$ suggestive of $2 R$-orientation were observed at the CD spectra of 2. ${ }^{18-20)}$ On the basis of this evidence, the absolute stereostructures of $\mathbf{1}$ and $\mathbf{2}$ were elucidated as shown. The mixture of ( $2 S$ )- and ( $2 R$ )-eriodictyol 7-O- $\beta$-D-glucopyranosiduronic acids was previously isolated from the rhizome of a fern Davallia mariesii Moore, whereas the separation of these diastereomers was unsuccessful. ${ }^{21)}$

Absolute Stereostructure of (2S,3S)-1-Phenyl-2,3-butanediol 3- $\boldsymbol{O}$ - $\boldsymbol{\beta}$-d-Glucopyranoside (3) ( $2 S, 3 S$ )-1-Phenyl-2,3-butanediol 3-O- $\beta$-d-glucopyranoside (3) was isolated as a white powder with negative optical rotation $\left([\alpha]_{D}^{27}-21.9^{\circ}\right)$. The molecular formula $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{7}$ of $\mathbf{3}$ was determined from the quasimolecular ion peaks at $m / z 329(\mathrm{M}+\mathrm{H})^{+}$and $m / z 327(\mathrm{M}-\mathrm{H})^{-}$in the positive- and negative-ion FAB-MS and by high-resolution MS measurement. The IR spectrum of 3 showed absorption bands at $3417,1605,1507$, and $1075 \mathrm{~cm}^{-1}$ ascribable to hydroxyl, aromatic ring, and ether functions. Acetylation of 3 furnished the pentaacetate (3a). The ${ }^{1} \mathrm{H}$-NMR $\left(\mathrm{CDCl}_{3}\right)$ and ${ }^{13} \mathrm{C}$-NMR (Table 1) spectra of 3

Table 1. ${ }^{13} \mathrm{C}-\mathrm{NMR}$ Data for (2S)-Eriodictyol 7-O- $\beta$-d-Glucopyranosiduronic acid (1), (2R)-Eriodictyol 7-O- $\beta$-D-Glucopyranosiduronic acid (2), and ( $2 S, 3 S$ )-1-Phenyl-2,3-butanediol 3-O- $\beta$-d-Glucopyranoside (3), and Its Acetate (3a)

|  | $\mathbf{1}^{a)}$ |  | $\mathbf{2}^{a)}$ | $\mathbf{3}^{b)}$ | $\mathbf{3 a}^{b)}$ | $\mathbf{3 a}^{b)}$ |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: |
| C-1 |  |  | 39.1 | 35.8 | -OAc | 20.6 |
| C-2 | 78.6 | 78.6 | 76.5 | 71.5 | 20.6 |  |
| C-3 | 42.1 | 42.1 | 81.6 | 76.3 | 20.6 |  |
| C-4 | 197.0 | 197.0 | 14.3 | 17.5 | 20.7 |  |
| C-5 | 162.8 | 162.8 |  |  | 21.0 |  |
| C-6 | 96.4 | 96.4 |  |  | 169.3 |  |
| C-7 | 165.0 | 164.8 |  |  | 169.4 |  |
| C-8 | 95.3 | 95.4 |  |  | 170.3 |  |
| C-9 | 162.7 | 162.6 |  |  | 170.4 |  |
| C-10 | 103.3 | 103.3 |  |  | 170.6 |  |
| C-1' | 129.2 | 129.2 | 138.5 | 137.2 |  |  |
| C-2' | 114.4 | 114.3 | 129.5 | 129.2 |  |  |
| C-3' | 145.2 | 145.2 | 128.3 | 128.5 |  |  |
| C-4 | 145.7 | 145.7 | 126.3 | 126.7 |  |  |
| C-5' | 115.3 | 115.3 | 128.3 | 128.5 |  |  |
| C-6' | 117.9 | 117.8 | 129.5 | 129.2 |  |  |
| C-1" | 99.2 | 99.0 | 104.2 | 101.3 |  |  |
| C-2" | 72.8 | 72.8 | 73.7 | 76.5 |  |  |
| C-3" | 75.8 | 75.9 | 75.5 | 73.0 |  |  |
| C-4" | 71.4 | 71.4 | 69.2 | 68.5 |  |  |
| C-5" | 74.8 | 74.8 | 76.3 | 71.7 |  |  |
| C-6" | 170.3 | 170.2 | 61.2 | 62.0 |  |  |

a) Measured in DMSO- $\left.d_{6}, b\right) \mathrm{CDCl}_{3}$ at 125 MHz .
and 3a showed signals assignable to a secondary methyl $[\delta$ 3: $\left.1.24\left(\mathrm{~d}, J=4.0 \mathrm{~Hz}, 4-\mathrm{H}_{3}\right) ; 3 \mathrm{a}: 1.27\left(\mathrm{~d}, J=6.4 \mathrm{~Hz}, 4-\mathrm{H}_{3}\right)\right]$, a methylene $\left\{\delta\right.$ 3: $\left[2.59(\mathrm{br} \mathrm{s}), 2.79(\mathrm{br} \mathrm{d}, J=c a .11 \mathrm{~Hz}), 1-\mathrm{H}_{2}\right]$; 3a: $[2.76$ (dd, $J=8.8,14.0 \mathrm{~Hz}), 2.89(\mathrm{dd}, J=4.8,14.0 \mathrm{~Hz}), 1-$


Fig. 1


Fig. 2

Table 2. Inhibitory Activity of Constituents from C. indicum for Rat Lens Aldose Reductase

| Compound | $\mathrm{IC}_{50}(\mu \mathrm{~m})$ |
| :--- | :---: |
| $(2 S)$-Eriodictyol 7- $O$ - $\beta$-d-glucopyranosiduronic acid (1) | 2.1 |
| $(2 R)$-Eriodictyol 7- $O$ - $\beta$-d-glucopyranosiduronic acid (2) | 1.5 |
| $(2 S, 3 S)$-1-phenyl-2,3-butanediol 3- $O$ - $\beta$-d-glucopyranoside (3) | $>100(9.7)$ |
| 3a | $>100(16.8)$ |
| Apigenin 7- $O$ - $\beta$-D-glucopyranoside (4) | 23 |
| Diosmetin 7- $O-\beta$-d-glucopyranoside (5) | 23 |
| Quercetin 3,7-di- $O$ - $\beta$-d-glucopyranoside (6) | 84 |
| Eriodictyol (7) | 7.7 |
| (2S,3S)-1-Phenyl-2,3-butanediol (8) | $>100(0.4)$ |
| Epalrestat | 0.072 |

Values in parentheses represent the inhibition (\%) at $100 \mu_{\mathrm{M}}$.
$\mathrm{H}_{2}$ ] $\}$, two methines bearing an oxygen function [ $\delta$ 3: 3.55 (br s, 3-H), 3.68 (br d, $J=c a .11 \mathrm{~Hz}, 2-\mathrm{H}$ ); 3a: 3.83 (dd, $J=$ $4.9,6.4 \mathrm{~Hz}, 3-\mathrm{H}), 5.06$ (ddd, $J=4.8,4.9,8.8 \mathrm{~Hz}, 2-\mathrm{H})$ ], an anomeric proton [ $\delta$ 3: 4.44 (br d, $J=c a .7 \mathrm{~Hz}, 1^{\prime \prime}-\mathrm{H}$ ); 3a: 4.65 (d, $J=7.9 \mathrm{~Hz}, 1^{\prime \prime}-\mathrm{H}$ )], and aromatic protons $[\delta 3: 7.14-7.25$ $(5 \mathrm{H}, \mathrm{m})$; 3a: $7.16-7.31(5 \mathrm{H}, \mathrm{m})$ ]. Acid hydrolysis of $\mathbf{3}$ with $5 \% \mathrm{H}_{2} \mathrm{SO}_{4}$ furnished D-glucose, which was identified by GLC analysis. ${ }^{16)}$ Enzymatic hydrolysis of $\mathbf{3}$ with $\beta$-glucosidase yielded $(2 S, 3 S)$-1-phenyl-2,3-butanediol (8). ${ }^{22)}$ In the HMBC experiment of $\mathbf{3}$, long-range correlation was observed between the 1 "-proton and 3 -carbon. Consequently, the absolute stereostructure of $\mathbf{3}$ was clarified as shown.

Inhibitory Activity of $\mathbf{1 - 6}$ for Rat Lens Aldose Reductase Aldose reductase as a key enzyme in the polyol pathway is reported to catalyze the reduction of glucose to sorbitol. Sorbitol does not readily diffuse across cell membranes, and the intracellular accumulation of sorbitol has been implicated in the chronic complications of diabetes such as cataract. Previously, the methanolic extract of the flowers of C. indicum and isolated flavonoids exhibited potent inhibitory activity against rat lens aldose reductase. ${ }^{10)}$ In the present study, inhibitory effects of additional isolated compounds ( $\mathbf{1 - 6}$ ) and their derivatives were examined as shown in Table 2. Among them, $\mathbf{1}\left(\mathrm{IC}_{50}=2.1 \mu \mathrm{M}\right)$ and $2(1.5$ $\mu_{\mathrm{M}}$ ) showed the potent inhibitory activity for rat lens aldose
reductase. However, the inhibitory activities of $\mathbf{1}$ and $\mathbf{2}$ were weaker than those of luteolin $(0.45 \mu \mathrm{~m})$ and luteolin $7-O-\beta$-Dglucopyranoside $(0.99 \mu \mathrm{M})$, which were also isolated from the flowers of $C$. indicum. ${ }^{10)}$

## Experimental

The following instruments were used to obtain physical data: specific rotations, Horiba SEPA-300 digital polarimeter ( $l=5 \mathrm{~cm}$ ); UV spectra, Shimadzu UV-1600 spectrometer; CD spectra, JASCO J-720WI spectropolarimeter; IR spectra, Shimadzu FTIR-8100 spectrometer; MS and high-resolution MS, JEOL JMS-GCMATE mass spectrometer; ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectra, JEOL JNM-LA500 $(500 \mathrm{MHz})$ spectrometer, ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectra, JEOL JNMLA500 ( 125 MHz ) spectrometer with tetramethylsilane as an internal standard; HPLC detector, Shimadzu SPD-10A UV-VIS detector (254 nm).

The following experimental conditions were used for chromatography: ordinary-phase column chromatography; Silica gel BW-200 (Fuji Silysia Chemical, Ltd., 150-350 mesh), reversed-phase silica gel column chromatography; Chromatorex ODS DM1020T (Fuji Silysia Chemical, Ltd., 100-200 mesh). Packed column for HPLC: YMC-Pack ODS-A ( $10 \times$ 250 mm , i.d. and $20 \times 250 \mathrm{~mm}$, i.d.). TLC, pre-coated TLC plates with Silica gel $60 \mathrm{~F}_{254}$ (Merck, 0.25 mm ) (normal-phase) and Silica gel RP-18 $\mathrm{F}_{254 \mathrm{~s}}$ (Merck, 0.25 mm ) (reversed-phase); HPTLC, pre-coated TLC plates with Silica gel RP-18 $\mathrm{WF}_{254 \mathrm{~S}}$ (Merck, 0.25 mm ) (reversed-phase). Detection was done by spraying with $1 \% \mathrm{Ce}\left(\mathrm{SO}_{4}\right)_{2}-10 \%$ aqueous $\mathrm{H}_{2} \mathrm{SO}_{4}$ followed by heating.

Extraction and Isolation Dried flowers of C. indicum L. ( 5.8 kg , cultivated in China and purchased from Koshiro Co., Ltd., Osaka) were finely cut and extracted with methanol under reflux. Evaporation of the solvent under reduced pressure gave the MeOH extract $(1650 \mathrm{~g}, 28.4 \%)$. The MeOH extract $(1600 \mathrm{~g})$ was partitioned in an $\mathrm{AcOEt}-\mathrm{H}_{2} \mathrm{O}(1: 1)$ mixture, and the aqueous layer was further extracted with $1-\mathrm{BuOH}$. Removal of the solvent under reduced pressure from the AcOEt- and $1-\mathrm{BuOH}$-soluble portions yielded $438 \mathrm{~g}(7.8 \%)$ and $226 \mathrm{~g}(4.0 \%)$ of residues, respectively. The 1BuOH -soluble portion ( 182 g ) was subjected to ordinary-phase silica gel column chromatography $\left[3.0 \mathrm{~kg}, \mathrm{CHCl}_{3}-\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O} \quad(10: 3: 1\right.$, lower layer $\rightarrow 6: 4: 1) \rightarrow \mathrm{MeOH}]$ to give seven fractions [Fr. $1(7.5 \mathrm{~g})$, Fr. $2(38.6 \mathrm{~g})$, Fr. $3(21.9$ g), Fr. $4(28.2$ g), Fr. $5(14.2$ g), Fr. $6(22.2$ g $)$, Fr. $7(32.3$ g)], as reported previously. ${ }^{10)}$ Fraction $2(7.0 \mathrm{~g})$ was subjected to reversed-phase silica gel column chromatography $\left[140 \mathrm{~g}, \mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}(30: 70 \rightarrow 50: 50, \mathrm{v} / \mathrm{v}) \rightarrow\right.$ MeOH ] and finally HPLC [YMC-Pack ODS-A, $20 \times 250 \mathrm{~mm}$, i.d., $\mathrm{MeOH}-$ $\mathrm{H}_{2} \mathrm{O}(40: 60$ or $50: 50, \mathrm{v} / \mathrm{v})$ ] to give ( $2 S, 3 S$ )-1-phenyl-2,3-butanediol 3- $-\beta-$ D-glucopyranoside ( $\mathbf{3}, 153 \mathrm{mg}, 0.019 \%$ ), apigenin $7-O-\beta$-d-glucopyranoside $(4,113 \mathrm{mg}, 0.014 \%)$ and diosmetin $7-O-\beta$-D-glucopyranoside $(5,693 \mathrm{mg}$, $0.086 \%)$. Fraction $6(20.0 \mathrm{~g})$ was also subjected to reversed-phase silica gel column chromatography $\left[600 \mathrm{~g}, \mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}(30: 70) \rightarrow \mathrm{MeOH}\right]$ and finally HPLC [YMC-Pack ODS-A, $20 \times 250 \mathrm{~mm}$, i.d., $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}(30: 70$, v/v)] to furnish the mixture fraction of $(2 S)$ - and ( $2 R$ )-eriodyctyol 7-O- $\beta$-d-glucopyranosiduronic acids ( 205 mg ) and quercetin 3,7 -di- $O-\beta$-d-glucopyranoside ( $6,156 \mathrm{mg}, 0.0038 \%$ ). The above mixture $(20 \mathrm{mg})$ was further purified by HPLC [YMC-Pack ODS-A, $10 \times 250 \mathrm{~mm}$, i.d., isopropanol- $\mathrm{H}_{2} \mathrm{O}-$ $\mathrm{AcOH}(10: 90: 1, \mathrm{v} / \mathrm{v})]$ to yield $(2 S)$-eriodictyol $7-O-\beta$-D-glucopyranosiduronic acid ( $\mathbf{1}, 11 \mathrm{mg}, 0.0027 \%$ ) and ( $2 R$ )-eriodictyol 7-O- $\beta$-d-glucopyranosiduronic acid (2, $9 \mathrm{mg}, 0.0023 \%$ ). The known constituents were identified by comparison of their physical data (IR, UV, and ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$ NMR) with those of authentic samples $(\mathbf{4}, \mathbf{5})^{14)}$ or with reported values. ${ }^{15)}$
(2S)-Eriodictyol 7-O- $\beta$-d-Glucopyranosiduronic Acid (1): A yellow powder, $[\alpha]_{\mathrm{D}}^{26}-35.6^{\circ}(c=0.20, \mathrm{MeOH})$. High-resolution positive-ion FAB-MS: Calcd for $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{O}_{12} \mathrm{Na}(\mathrm{M}+\mathrm{Na})^{+}$: 487.0852. Found: 487.0867. CD [MeOH, $\mathrm{nm}(\Delta \varepsilon)]: 248(+0.24), 287(-3.59), 340(+0.24)$. UV [MeOH, nm $(\log \varepsilon)$ ]: 284 (4.21), 326 (3.55). IR (KBr): 3417, 1719, 1655, 1637, 1458, $1071 \mathrm{~cm}^{-1}$. $\left.{ }^{1} \mathrm{H}-\mathrm{NMR}(500 \mathrm{MHz} \text {, DMSO-d })_{6}\right) \delta:[2.74(1 \mathrm{H}, \mathrm{dd}, J=3.0,17.4$ $\left.\mathrm{Hz}), 3.25(1 \mathrm{H}, \mathrm{dd}, J=12.5,17.4 \mathrm{~Hz}), 3-\mathrm{H}_{2}\right], 5.07\left(1 \mathrm{H}, \mathrm{d}, J=7.3 \mathrm{~Hz}, 1^{\prime \prime}-\mathrm{H}\right)$, $5.45(1 \mathrm{H}, \mathrm{dd}, J=3.0,12.5 \mathrm{~Hz}, 2-\mathrm{H}), 6.13,6.16(1 \mathrm{H}$ each, both d, $J=2.1 \mathrm{~Hz}$, $6,8-\mathrm{H}), 6.75\left(2 \mathrm{H}, \mathrm{brs}, 5^{\prime}, 6^{\prime}-\mathrm{H}\right), 6.89\left(1 \mathrm{H}, \mathrm{brs}, 2^{\prime}-\mathrm{H}\right),[8.96(2 \mathrm{H}, \mathrm{brs})$, $12.01(1 \mathrm{H}, \mathrm{brs}),-\mathrm{OH}] .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(125 \mathrm{MHz}\right.$, DMSO- $\left.d_{6}\right) \delta_{\mathrm{c}}$ : given in Table 1. Positive-ion FAB-MS: $m / z 487(\mathrm{M}+\mathrm{Na})^{+}$. Negative-ion FAB-MS: $m / z$ $463(\mathrm{M}-\mathrm{H})^{-}, 287\left(\mathrm{M}-\mathrm{C}_{6} \mathrm{H}_{9} \mathrm{O}_{6}\right)^{-}$.
(2R)-Eriodictyol 7-O- $\beta$-d-Glucopyranosiduronic Acid (2): A yellow powder, $[\alpha]_{\mathrm{D}}^{24}-54.5^{\circ}(c=0.10, \mathrm{MeOH})$. High-resolution positive-ion FAB-MS: Calcd for $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{O}_{12} \mathrm{Na}(\mathrm{M}+\mathrm{Na})^{+}$: 487.0852. Found: 487.0871. CD [MeOH, $\mathrm{nm}(\Delta \varepsilon)]: 252(-1.00), 292(+3.66), 334(-1.80)$. UV [MeOH, nm $(\log \varepsilon)]: 285$ (4.14), 332 (3.48). IR (KBr): 3436, 1719, 1655, 1638, 1458, $1069 \mathrm{~cm}^{-1}$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right) \delta:[2.75(1 \mathrm{H}, \mathrm{dd}, J=3.4$,
$\left.17.1 \mathrm{~Hz}), 3.25(1 \mathrm{H}, \mathrm{dd}, J=12.2,17.1 \mathrm{~Hz}), 3-\mathrm{H}_{2}\right], 5.10\left(1 \mathrm{H}, \mathrm{d}, J=7.6 \mathrm{~Hz}, 1^{\prime \prime}-\right.$ H), $5.44(1 \mathrm{H}, \mathrm{dd}, J=3.4,12.2 \mathrm{~Hz}, 2-\mathrm{H}), 6.13,6.18(1 \mathrm{H}$ each, both d, $J=2.1 \mathrm{~Hz}, 6,8-\mathrm{H}), 6.74\left(2 \mathrm{H}, \mathrm{brs}, 5^{\prime}, 6^{\prime}-\mathrm{H}\right), 6.90\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 2^{\prime}-\mathrm{H}\right),[8.95(2 \mathrm{H}$, $\mathrm{brs}), 12.00(1 \mathrm{H}, \mathrm{brs}),-\mathrm{OH}] .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(125 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right) \delta_{\mathrm{c}}$ : given in Table 1. Positive-ion FAB-MS: $m / z 487(\mathrm{M}+\mathrm{Na})^{+}$. Negative-ion FAB-MS: $m / z 463(\mathrm{M}-\mathrm{H})^{-} 287\left(\mathrm{M}-\mathrm{C}_{6} \mathrm{H}_{9} \mathrm{O}_{6}\right)^{-}$.
( $2 S, 3 S$ )-1-Phenyl-2,3-butanediol 3-O- $\beta$-D-Glucopyranoside (3): A white powder, $[\alpha]_{D}^{27}-21.9^{\circ}\left(c=2.00, \mathrm{CHCl}_{3}\right)$. High-resolution positive-ion FABMS: Calcd for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{7}(\mathrm{M}+\mathrm{H})^{+}: 329.1600$. Found: 329.1615. IR ( KBr ): $3417,1605,1507,1075 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 1.24(3 \mathrm{H}, \mathrm{d}$, $\left.J=4.0 \mathrm{~Hz}, 4-\mathrm{H}_{3}\right),\left[2.59(1 \mathrm{H}, \mathrm{br} \mathrm{s}), 2.79(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J=c a .11 \mathrm{~Hz}), 1-\mathrm{H}_{2}\right], 3.20$ $\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime \prime}-\mathrm{H}\right), 3.42\left(1 \mathrm{H}, \mathrm{m}, 2^{\prime \prime}-\mathrm{H}\right), 3.49\left(1 \mathrm{H}, \mathrm{m}, 5^{\prime \prime}-\mathrm{H}\right), 3.55\left(1 \mathrm{H}, \mathrm{m}, 4^{\prime \prime}-\mathrm{H}\right)$, $3.55(1 \mathrm{H}, \mathrm{brs}, 3-\mathrm{H}), 3.68(1 \mathrm{H}, \mathrm{brd}, J=c a .11 \mathrm{~Hz}, 2-\mathrm{H}), 3.73\left(1 \mathrm{H}, \mathrm{m}, 6^{\prime \prime}-\mathrm{H}_{2}\right)$, $4.44\left(1 \mathrm{H}, \mathrm{brd}, J=c a .7 \mathrm{~Hz}, 1^{\prime \prime}-\mathrm{H}\right), 7.14-7.25(5 \mathrm{H}, \mathrm{m}, \mathrm{ph}-\mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta_{\mathrm{c}}$ : given in Table 1. Positive-ion FAB-MS: m/z 329 $(\mathrm{M}+\mathrm{H})^{+}$. Negative-ion FAB-MS: $m / z 327(\mathrm{M}-\mathrm{H})^{-}$.

Acid Hydrolysis of 1-3 A solution of 1-3 (2 mg each) in 5\% aqueous $\mathrm{H}_{2} \mathrm{SO}_{4}-1,4$-dioxane ( $0.5 \mathrm{ml}, 1: 1, \mathrm{v} / \mathrm{v}$ ) was heated under reflux for 1 h . After cooling, the reaction mixture was neutralized with Amberlite IRA-400 $\left(\mathrm{OH}^{-}\right.$ form) and the residue was removed by filtration. After removal of the solvent from the filtrate in vacuo, the residue was transferred to a Sep-Pak C18 cartridge with $\mathrm{H}_{2} \mathrm{O}$ and MeOH . The $\mathrm{H}_{2} \mathrm{O}$ eluate was concentrated and the residue was treated with l-cysteine methyl ester hydrochloride ( 4 mg ) in pyridine $(0.5 \mathrm{ml})$ at $60^{\circ} \mathrm{C}$ for 1 h . After reaction, the solution was treated with $\mathrm{N}, \mathrm{O}$-bis(trimethylsilyl)trifluoroacetamide $(0.2 \mathrm{ml})$ at $60^{\circ} \mathrm{C}$ for 1 h . The supernatant was then subjected to GLC analysis to identify the derivatives of D-glucuronic acid (i) and D-glucose (ii). GLC conditions: column, Supeluco $\mathrm{STB}^{\mathrm{TM}}-1,30 \mathrm{~m} \times 0.25 \mathrm{~mm}$ (i.d.) capillary column; injector temperature, $230^{\circ} \mathrm{C}$, detector temperature, $230^{\circ} \mathrm{C}$; column temperature, $230^{\circ} \mathrm{C}$; He flow rate, $15 \mathrm{ml} / \mathrm{min}$; $t_{\mathrm{R}}, \mathbf{i}, 26.5 \mathrm{~min}$ (from $\mathbf{1}$ and 2); ii, 24.4 min (from 3).

Enzymatic Hydrolysis of 1 and 2 A solution of the mixture fraction of $\mathbf{1}$ and $2(10.0 \mathrm{mg}, 0.022 \mathrm{mmol})$ in 0.2 m acetate buffer ( $\mathrm{pH} 5.0,5.0 \mathrm{ml}$ ) was treated with $\beta$-glucuronidase ( 1.0 ml , Sigma) and the solution was stirred at $37^{\circ} \mathrm{C}$ for 14 h . After treatment of the reaction mixture with EtOH , the mixture was evaporated to dryness under reduced pressure and the residue was purified by ordinary-phase silica gel column chromatography $\left[1.0 \mathrm{~g}, \mathrm{CHCl}_{3}-\right.$ $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ (10:3:1, lower layer)] to give eriodictyol (7, $5.4 \mathrm{mg}, 87 \%$ ).

Acetylation of 3 A solution of $3(5.0 \mathrm{mg}, 0.015 \mathrm{mmol})$ in pyridine $(1.0 \mathrm{ml})$ was treated with $\mathrm{Ac}_{2} \mathrm{O}(0.8 \mathrm{ml})$ and the mixture was stirred at room temperature for 24 h . The reaction mixture was poured into ice-water and the whole was extracted with AcOEt. The AcOEt extract was successively washed with $5 \%$ aqueous HCl , saturated aqueous $\mathrm{NaHCO}_{3}$, and brine, then dried over $\mathrm{MgSO}_{4}$ powder and filtered. Removal of the solvent from the filtrate under reduced pressure furnished a residue, which was purified by ordi-nary-phase silica gel column chromatography $[0.5 \mathrm{~g}, n$-hexane-AcOEt (1:1)] to furnish the pentaacetate ( $\mathbf{3 a}, 8.3 \mathrm{mg}$, quant.).
3a: A white powder, $[\alpha]_{D}^{27}-7.8^{\circ}\left(c=0.10, \mathrm{CHCl}_{3}\right)$. High-resolution posi-tive-ion FAB-MS: Calcd for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{12} \mathrm{Na}(\mathrm{M}+\mathrm{Na})^{+}$: 561.1948. Found: 561.1963. IR (KBr): 1755, 1456, 1375, 1229, $1038 \mathrm{~cm}^{-1} .{ }^{1} \mathrm{H}-\mathrm{NMR}(500$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta: 1.27\left(3 \mathrm{H}, \mathrm{d}, J=6.4 \mathrm{~Hz}, 4-\mathrm{H}_{3}\right), 1.98,2.01,2.03,2.03,2.08$ (3H each, all s, Ac-H3), [2.76 (1H, dd, $J=8.8,14.0 \mathrm{~Hz}), 2.89(1 \mathrm{H}, \mathrm{dd}, J=4.8$, $\left.14.0 \mathrm{~Hz}), 1-\mathrm{H}_{2}\right], 3.69\left(1 \mathrm{H}, \mathrm{m}, 5^{\prime \prime}-\mathrm{H}\right), 3.83(1 \mathrm{H}, \mathrm{dd}, J=4.9,6.4 \mathrm{~Hz}, 3-\mathrm{H})$, [4.14 (1H, dd, $J=2.5,12.2 \mathrm{~Hz}), 4.23(1 \mathrm{H}, \mathrm{dd}, J=4.9,12.2 \mathrm{~Hz}), 6^{\prime \prime}-\mathrm{H}_{2}$ ], 4.65 $\left(1 \mathrm{H}, \mathrm{d}, J=7.9 \mathrm{~Hz}, 1^{\prime \prime}-\mathrm{H}\right), 5.04\left(1 \mathrm{H}, \mathrm{dd}, J=7.9,9.8 \mathrm{~Hz}, 2^{\prime \prime}-\mathrm{H}\right), 5.06(1 \mathrm{H}, \mathrm{ddd}$, $J=4.8,4.9,8.8 \mathrm{~Hz}, 2-\mathrm{H}), 5.08\left(1 \mathrm{H}\right.$, dd like, $\left.4^{\prime \prime}-\mathrm{H}\right), 5.20(1 \mathrm{H}, \mathrm{dd}, J=9.2$, $\left.9.8 \mathrm{~Hz}, 3^{\prime \prime}-\mathrm{H}\right), 7.16-7.31(5 \mathrm{H}, \mathrm{m}, \mathrm{ph}-\mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta_{\mathrm{c}}$ given in Table 1. Positive-ion FAB-MS: $m / z 561(\mathrm{M}+\mathrm{Na})^{+}$. Negative-ion FAB-MS: $m / z 537$ (M-H) ${ }^{-}$

Enzymatic Hydrolysis of 3 A solution of $\mathbf{3}(20.0 \mathrm{mg}, 0.061 \mathrm{mmol})$ in 0.2 m acetate buffer ( $\mathrm{pH} 5.0,5.0 \mathrm{ml}$ ) was treated with $\beta$-glucosidase ( 20.0 mg , Sigma) and the solution was stirred at $37^{\circ} \mathrm{C}$ for 24 h . After treatment of the reaction mixture with EtOH , the mixture was evaporated to dryness under reduced pressure and the residue was purified by ordinary-phase silica
gel column chromatography $\left[1.0 \mathrm{~g}, \mathrm{CHCl}_{3}-\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}(10: 3: 1\right.$, lower layer)] to give ( $2 S, 3 S$ )-1-phenyl-2,3-butanediol ( $\mathbf{8}, 9.8 \mathrm{mg}, 97 \%$ ). Compound $\mathbf{8}$ was identified by comparison of the physical data (IR, $[\alpha]_{\mathrm{D}}$, and ${ }^{1} \mathrm{H}-$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ ) with those of reported values. ${ }^{22)}$

Bioassay Aldose reductase activity was assayed by the method described in a previous paper. ${ }^{10,20)}$ The supernatant fluid of rat lens homogenate was used as the crude enzyme. The incubation mixture contained 135 mm Na , K-phosphate buffer ( pH 7.0 ), $100 \mathrm{~mm} \mathrm{Li}_{2} \mathrm{SO}_{4}, 0.03 \mathrm{~mm} \mathrm{NADPH}, 1 \mathrm{~mm}$ dL-glyceraldehyde as a substrate, and $100 \mu \mathrm{l}$ of enzyme fraction, with or without $25 \mu \mathrm{l}$ of sample solution, in a total volume of 0.5 ml . The reaction was initiated by the addition of NADPH at $30^{\circ} \mathrm{C}$. After 30 min , the reaction was stopped by the addition of $150 \mu \mathrm{l} 0.5 \mathrm{~m} \mathrm{HCl}$. Then, 0.5 ml 6 m NaOH containing 10 mm imidazole was added, and the solution was heated at $60^{\circ} \mathrm{C}$ for 10 min to convert NADP to a fluorescent product. Fluorescence was measured using a fluorophotometer (Luminescence Spectrometer LS50B, Perkin Elmer, England) at an excitation wavelength of 360 nm and an emission wavelength of 460 nm .

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