



Methane emission estimates using chamber and tracer release experiments for a municipal waste water treatment plant

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Abstract. This study presents two methods for estimating methane emissions from a waste water treatment plant (WWTP) along with results from a measurement campaign at a WWTP in Valence, France. These methods, chamber measurements and tracer release, rely on Fourier transform infrared spectroscopy and cavity ring-down spectroscopy instruments. We show that the tracer release method is suitable for quantifying facility- and some process-scale emissions, while the chamber measurements provide insight into individual process emissions. Uncertainties for the two methods are described and discussed. Applying the methods to CH₄ emissions of the WWTP, we confirm that the open basins are not a major source of CH₄ on the WWTP (about 10 % of the total emissions), but that the pretreatment and sludge treatment are the main emitters. Overall, the waste water treatment plant is representative of an average French WWTP.

1 Introduction

Human activities cause greenhouse gas (GHG) emissions at a large scale, changing the atmospheric chemical composition by measurable and consequential amounts. Anthropogenic GHG emissions such as methane (CH₄) now represent a significant fraction of total greenhouse gas emissions

into the atmosphere. To better understand the anthropogenic sources of GHGs, with the goal of ultimately reducing these emissions, it is essential to accurately quantify the emissions at different spatial scales, from the country to the process scale, and to monitor the possible temporal variabilities. We can sort estimation methods into two groups depending on the type of measurement used: the top-down approach based on atmospheric measurements of GHGs at different scales (global, regional, local) and the bottom-up approach that uses activity data, emission factors and flux modeling to calculate emissions. Both approaches can be applied from the global to the process scale depending on the representativity of the measurements.

Methane is a potent anthropogenic greenhouse gas with a global warming potential 28 times as strong as that of CO₂ on a 100-year time horizon (Stocker et al., 2013). Primary sources of anthropogenic methane emissions are landfills, waste water treatment plants (WWTPs), rice paddies, ruminants and manure management, oil and gas production and transport activities. Combining the two approaches by using top-down measurements at all scales to validate or adjust benchmark bottom-up calculations and emission factors can help not only improve inventories by a more robust quantification but also provide valuable information for how to prioritize emission reduction activities.

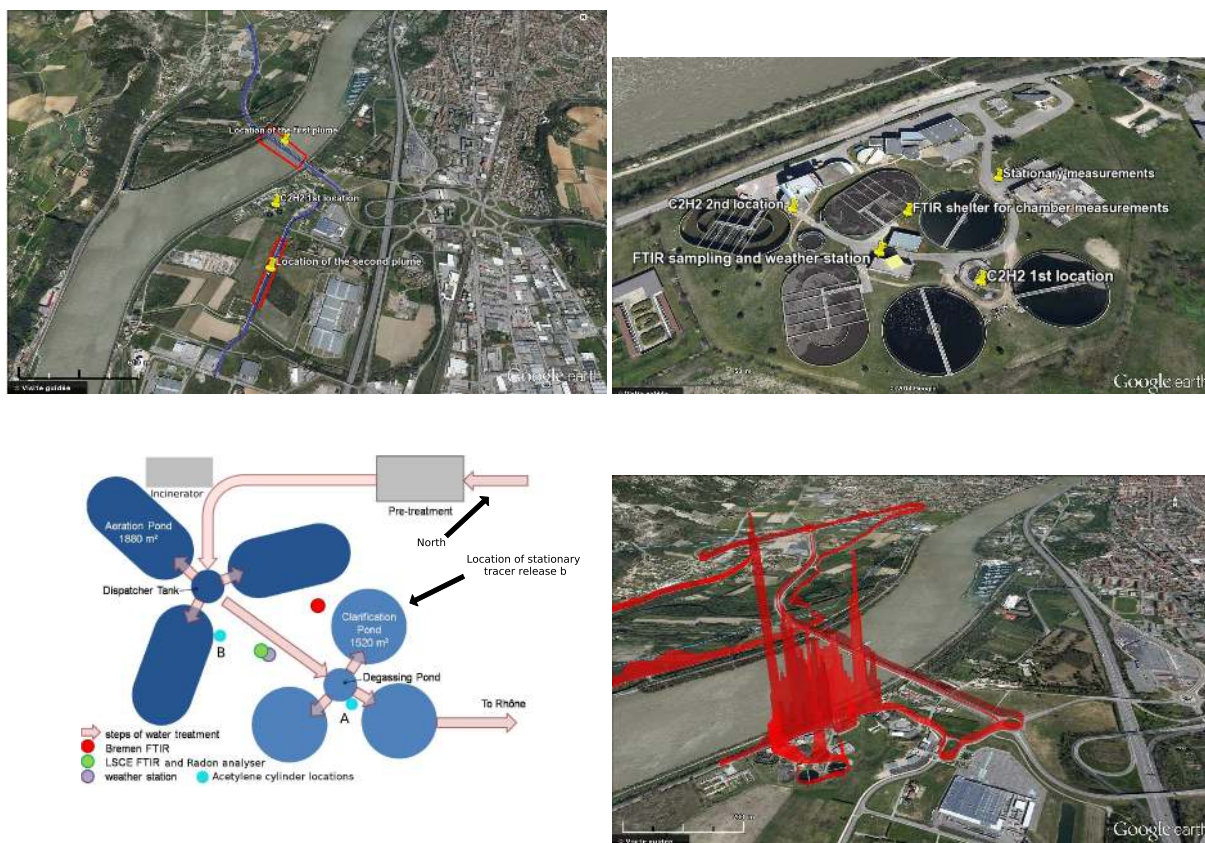


Figure 1. (a and b) Aerial view (Google Earth) of the WWTP. The blue lines show the driving paths during the tracer release experiment and the red rectangles show the location of the plumes. (c) Schematic view of the waste water treatment plant. (d) Aerial view of the WWTP with methane concentrations shown as red rectangles measured on 18 September with a northeast wind. The signals showed are above 1850 ppb. The highest signal near the incinerator is 10 ppm.

In France, methane emissions from waste management (waste water treatment and landfills) accounted for about 19 % of the total methane emissions in 2011 following the national inventory from CITEPA (CITEPA, 2013). Landfills are the largest emitter with 17 %, but waste water treatment plants still represent a non-negligible part (2 %). However, these values are estimated with 100 % uncertainty due to the difficulty in accurately estimating the biological demand of oxygen (BOD), quantity of CH₄ emitted by kg of BOD, fraction of treated incoming waste water and anoxic/oxic conditions, which are the parameters used by CITEPA to derive CH₄ emissions from WWTP (CITEPA, 2013). Several studies have been conducted in different countries to provide more accurate estimates of the emissions for WWTPs. Cakir and Stenstrom (2005) and El-Fadel and Massoud (2001) present estimations based on process modeling, but some studies such as Czepiel et al. (1993), Wang et al. (2011) and Daelman et al. (2012) calculate emissions using CH₄ measurements with mass budget. Finally, a recent study by Yoshida et al. (2014) used the tracer release method as described in this paper to estimate CH₄ and N₂O emissions

from a WWTP. In these papers, emissions vary from 0.011 to 1.3 kg yr⁻¹ per population equivalent depending on the WWTP design (e.g., depending on the use of aerobic or anaerobic processes, presence of a sludge digester) and the estimation method as the tracer release allows the capturing of leakage emissions that could be omitted by the other methods. For municipal WWTPs using activated sludge (aerobic) treatment, emissions still vary from 0.039 to 0.309 kg yr⁻¹ per population equivalent. This range of estimate shows that the WWTP CH₄ emissions depend on the design and the size of the WWTP. In France, according to the BDERU for 2008 (database for urban waste water, <http://www.statistiques.developpement-durable.gouv.fr/lessentiel/ar/306/1168/assainissement-traitement-collectif-eaux-usees.html>), there are about 18 600 WWTPs, half of which treat water for a fewer-than-500 population equivalent. However, the 6 % of WWTP with more than 10 000 population equivalent treat 80 % of the waste water. In this study, we focused on one of these medium-sized WWTPs that employs activated sludge treatment. We used two methods – chamber measurements and tracer release method with acetylene – that have been

rarely used on WWTPs to calculate GHG emissions at the process and the plant scale. We aimed not only to estimate the total emissions of the site but also to investigate individual processes and evaluate the missing elements between these two measurement scales. Another goal was to estimate the uncertainties for each method to provide a more robust emission estimation and be able to compare our results with other studies or inventories. An intensive measurement campaign was thus conducted at one of the WWTP of Valence, France, from 17 to 21 September 2012.

First, we present the details of the site under study, followed by the different emission estimation methods, measurement techniques and instruments employed during the experimental campaign. Finally, we present and discuss the results obtained for CH₄ from the process scale up to the site scale. All the emission estimates hereafter refer directly to CH₄, i.e., the notation kg of CH₄ day⁻¹ or kg of CH₄ yr⁻¹ per population equivalent is replaced by kg day⁻¹ or kg yr⁻¹ per population equivalent.

2 Description of the site

The WWTP is located in the southwest of the city of Valence, around 50 m east from the Rhône river, which flows in a north–south direction (see Fig. 1). Valence is located in the southeastern part of France, 500 km southeast of Paris, 100 km south of Lyon and 70 km southwest of Grenoble. The station is managed by Veolia France and treats the water for 150 000 inhabitant equivalents, which represents about 2800 m³ h⁻¹ with an exiting BOD of 35 kg m⁻³ (<http://www.valenceagglom.fr/stations-depuration>).

The water follows a several step treatment (see Fig. 1). After being filtered for solids, the water is filtered for sand particles (down to 200 microns in diameter) by sedimentation, and oil is removed by injection of air bubbles. The water is then distributed to three aeration basins (12 000 m³ each) via a dispatcher basin. In the aeration basins, air is periodically injected to help aerobic bacteria to digest the organic matter. The water and the sludge are sent to a degassing/dispatcher basin and then separated by sedimentation inside three clarification basins (6000 m³ each). The sludge from the different steps is collected and dried before being incinerated. The cleaned water from the overflow of the clarification basins is discharged into the Rhône river. During the campaign, one of the aeration basins was being cleaned, so only two were in use.

We anticipated the potential for methane release during all steps of the process. In the aeration basins, periods of aeration with aerobic reaction alternate with rests when anaerobic reactions can occur. Methane formed during these resting phases is then transported to the surface when aeration restarts and provokes a mixing of water. In the degassing basin, water is mixed and dissolved methane can be released. In the clarification basin, as there is a slow mixing, some de-

gassing could still be expected, with bacteria from the active sludge still producing methane. Finally, the sludge may still contain methane that could be emitted during centrifugation, storage and incineration. In addition, methane dissolved in the incoming water from the city will be released at the plant, starting from the first exposure to the atmosphere, and certainly during the aeration process. Figure 1d shows a qualitative image of the methane measured with the mobile instrument described in Sect. 4 around the site on 18 September with a southwest wind. We indeed see higher CH₄ concentrations on the site than outside with peaks for the degassing basin, the water pretreatment and the sludge incinerator.

3 Emission estimation methods

3.1 Chamber measurements on the basins

Depending on the basin areas under investigation, two different modes of chamber measurements were employed: (a) accumulation closed-chamber measurements (Frankignoulle, 1988) and (b) flow-through open-chamber measurements. The former mode was employed on the clarification basin (18 September) and on the aeration basin (19 September) outside of the aerated area of the basin, which had rather calm surfaces, and the latter on the aerated part of the aeration basin, where air is injected in the basin, resulting in a large air flux and turbulent surface (see Figs. 2 and 3).

Accumulation (closed-chamber) measurements

The chamber was closed against ambient air and the mass flux F is calculated from the linear increase of the measured gas mole fraction in the chamber with time (see Fig. 3a):

$$F = \frac{\Delta C}{\Delta t} \frac{pVM A_{\text{basin}}}{RTA}, \quad (1)$$

where $\frac{\Delta C}{\Delta t}$ is the fitted linear increase of the gas mole fraction in the chamber with time (mol mol⁻¹ s⁻¹), p is the pressure in the floating chamber (Pa), T the temperature (K), R the universal gas constant (8.314 m³ Pa K⁻¹ mol⁻¹), V represents the volume of the chamber (m³), A the water surface area enclosed by the chamber (m²), A_{basin} the area of the basin (m²) and M the molar mass of methane (g mol⁻¹). Fluxes were converted to the unit g day⁻¹. The chamber had a small vent hole (ca. 10 mm in diameter). When the chamber was first placed on the water, it was vented to the atmosphere to allow the chamber pressure equalize to atmospheric pressure. After about 20 s, the vent was closed. Ambient pressure was recorded at the weather station.

As for the errors, five main sources of uncertainty were considered. First, the error associated with the linear fit was taken into account and calculated as the coefficient of variation (CV). Secondly, the uncertainty associated with the volume of the chamber was considered. This uncertainty arises both from the initial measurement of the total volume of the

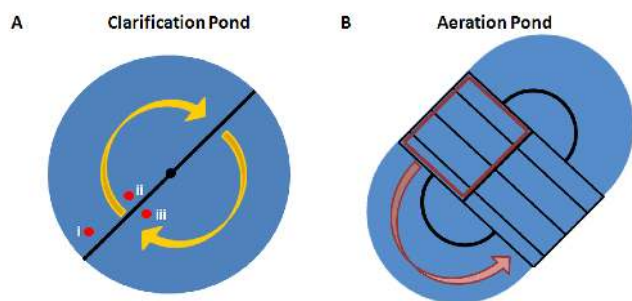


Figure 2. Schematic of the two basins that were measured with the floating chamber. **(a)** Clarification basin: the yellow arrow shows the direction in which the arm rotates. The red dots and symbols refer to the location of the chamber during runs 1, 2, 3 (symbol i), 4, 5, 7 (symbol ii) and 6, 8, 9 (symbol iii). **(b)** Aeration basin: the red rectangle denotes the aeration area, the arrow the water flow.

chamber and the uncertainty associated with the water level in the chamber. Because of the conic shape of the chamber, the uncertainty of the water level also affected the uncertainty of the water surface area enclosed by the chamber. Here it was assumed that the water level varied by 1 cm. The uncertainties associated with the pressure and temperature sensors were also considered in terms of the confidence interval provided by the manufacturer. The overall uncertainty was calculated for each run using propagation of uncertainties (Bevington and Robinson, 2003).

Flow-through (open-chamber) measurements

The chamber was modified for flow-through measurements with five small holes (ca. 10 mm in diameter) present in the top of the chamber to allow excess injected air to escape. During aeration times, the air in the chamber was replaced within a few minutes. Hence, the gas concentration in the chamber represented the concentration in the aeration air emitted from the basin once several mixing times in the chamber volume had occurred. Therefore, the mass flux of the emitted gas could be calculated by the amount of injected air, the gas concentration in the injected air and its integration over time (see Fig. 3b):

$$F = \sum_t (C_{\text{chamber}} - C_{\text{background}}) \frac{M}{V_m} \frac{dV_{\text{aeration}}}{dt} A_{\text{aeration}}, \quad (2)$$

where C_{chamber} is the gas mole fraction measured in the chamber (mol mol^{-1}), $C_{\text{background}}$ is the background gas mole fraction in the injected air, $\frac{dV_{\text{aeration}}}{dt}$ is the volume of air injected inside the basin per time ($\text{m}^3 \text{h}^{-1}$), M is the molar mass of CH₄ (g mol^{-1}), V_m is the molar volume of ideal gases ($\text{m}^3 \text{mol}^{-1}$) and A_{aeration} is the surface area of the aeration area in the aeration basin. The volume of air injected in the aeration basins was monitored with an Endress + Hauser AT70 flowmeter. The uncertainty given by the manufacturers is 2%. The air injected into the aeration basin was ambient

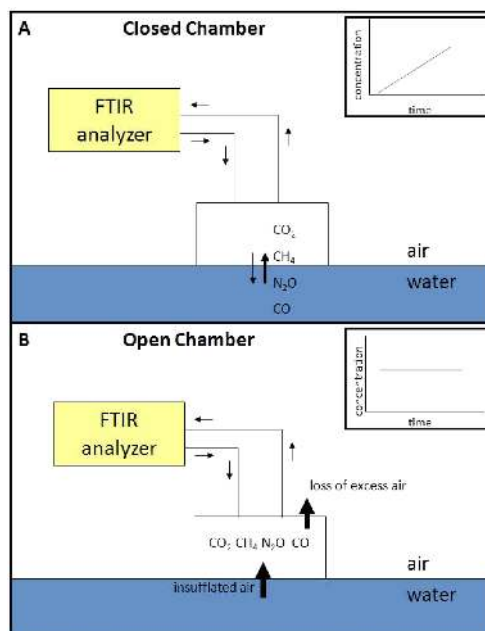


Figure 3. Schematic showing different modes of chamber deployment. **(a)** Conventional floating chamber used on a calm surface (accumulation closed-chamber measurements). The schematic concentration vs. time points out how the gas accumulates in the chamber over time (in case of a positive net flux from water to air). This increase is linearly approximated and from the slope, the flux is calculated. **(b)** Flow-through open chamber: the excess air escapes and the concentration measured in the chamber relates directly to the concentration in the emitted air. Thus, here we refer to the concentration reached in one time interval.

air. Note that multiplication with A_{aeration} contains the assumption that air is injected homogeneously in the basin. As the air is released from approximately evenly spaced diffusers at the bottom of the basin, we think that this assumption is warranted. The uncertainty was then calculated with error propagation, taking into account both the uncertainty of the injected air volume (2%), the uncertainty of the background CH₄ concentration and the error of the CH₄ measurement.

3.2 Tracer release method

The tracer release method consists of releasing a tracer gas (here C₂H₂) at a known rate from a location which is collocated with the unknown emission of a trace gas to be determined, here CH₄. This method has often been used in previous studies to determine CH₄ from landfills and more recently WWTPs (Czepiel et al., 1996; Galle et al., 2001; Spokas et al., 2006; Fredenslund et al., 2010; Mønster et al., 2014b; Yoshida et al., 2014). Concentrations of the tracer as well as the gas of interest are measured using a mobile instrument downwind in the co-propagating plumes. The ratio of the area of the two plume signals is proportional to the emission rate. Thus, knowing the emission rate of the released gas

Table 1. Instruments used during the campaign and their specifications.

Instrument	Integration time used in the study	Species	Uncertainty for species of interest
FTIR LSCE	1 min/30 min	CO ₂ , CH ₄ , N ₂ O, CO and $\delta^{13}\text{C}$	< 0.1 % (CH ₄)
FTIR Bremen	5 min	CO ₂ , CH ₄ , N ₂ O, CO and $\delta^{13}\text{C}$	< 0.1 % (CH ₄)
CRDS	1 s/1 min	CH ₄ , CO ₂ , C ₂ H ₂ , H ₂ O	< 0.1 % (CH ₄), < 5 % (C ₂ H ₂)
Weather station	1 min	Wind speed, wind direction, temperature, relative humidity and atmospheric pressure	3 %, 3°, 0.3 °C, 3, 0.05 %

and the concentrations of both gases, we could calculate the emission rate of the gases of interest:

$$F_{\text{CH}_4} = F_{\text{C}_2\text{H}_2} \frac{A_{\text{CH}_4}}{A_{\text{C}_2\text{H}_2}} \frac{M_{\text{CH}_4}}{M_{\text{C}_2\text{H}_2}}, \quad (3)$$

where F_{CH_4} are the emissions of CH₄ (kg h⁻¹), $F_{\text{C}_2\text{H}_2}$ are the known emissions of C₂H₂ (kg h⁻¹), $\frac{A_{\text{CH}_4}}{A_{\text{C}_2\text{H}_2}}$ is the ratio of the areas under the signals of CH₄ and C₂H₂ once the background subtracted and $\frac{M_{\text{CH}_4}}{M_{\text{C}_2\text{H}_2}}$ is the ratio of the molar masses of CH₄ and C₂H₂. For stationary experiments, Eq. (3) was modified such that the slope of the CH₄ vs. C₂H₂ linear regression was used to calculate the unknown flux instead of the area under the signals. Indeed, in this case, as there was no crossing of the plumes, there is no area under the signals to integrate but instead a mixed signal varying with the wind direction.

In this method, the uncertainties arise then from the concentration measurements, the tracer flux and the collocation of the plumes. CH₄ and C₂H₂ concentration errors are less than 0.1 and 5 %, respectively, for 1 s average. Once the gas cylinder is installed and regulated, the flow of the tracer gas is steady and well known, and this error depends on the precision, the reproducibility (given by the manufacturer) and the reading error. The precision is defined by the maximum value that the flowmeter can read and is here below 2 % on 1507 L h⁻¹ (Mønster et al., 2014a). The reproducibility on the read flow is 0.5 % and the reading error is estimated as a quarter the size of the float, i.e., 1 mm. Thus the maximum total uncertainty on the C₂H₂ flow is 0.5 kg day⁻¹ with the precision being the major factor.

The main uncertainties come from the imperfect collocation of the plumes and from the analysis of the plumes, especially the background determination for CH₄ and the calculation of the areas, as the signal/noise ratio is not very high in this study. Indeed, even while driving several hundreds of meters out of the plumes, in the “WWTP-free” air, the background for CH₄ was still highly variable from one crossing to the other. To address this issue, the background for each CH₄ plume was calculated using a linear regression between

the first and last point of the peak instead of removing an average background value for the whole event. Once this background was subtracted, the ratio of the areas was calculated. C₂H₂ background values were almost 0, so no background was subtracted. We used 1 s averaged data. Indeed, the more data points are used, the better the resolution of the signal is, allowing for a finer area estimation, which is the observation we are looking to extract. The autocorrelation of the errors on the 1 s data is taken into account in the global error which is the aggregation of the different errors. To estimate the non-collocation error, we ran one experiment with the C₂H₂ cylinder at a different location; however, due to the small amplitudes of the signal as well as the CH₄ high noise, these data could not be used quantitatively. To reduce this error as much as possible, we drove far enough away as was convenient with the existing roads (500 m to 1 km away) to consider the two signals collocated. The goal is also to position the cylinder such as it is neither downwind nor upwind of the CH₄ source to minimize dispersion discrepancies. We also discuss this assumption qualitatively in Sect. 5.

4 Instruments and setup

During this 1-week campaign, two Fourier transform infrared (FTIR) analyzers measuring CO₂, CH₄, N₂O, CO and $\delta^{13}\text{C}$ in CO₂ (Ecotech and University of Wollongong, Australia), one cavity ring-down spectroscopy (CRDS) instrument (custom prototype, Picarro Inc., Santa Clara) measuring CH₄, CO₂ and H₂O or C₂H₂, CH₄ and H₂O and a weather station were installed to measure GHG concentrations and/or estimate CH₄ emissions (see Fig. 1 and Table 1). Here, we focused only on CH₄ and C₂H₂ concentrations even though the instruments measured more species. One of the FTIR analyzers was used to measure samples from the basins with the chamber technique described above and the second mostly sampled air at the same location as the weather station but performed some measurements above the basins as well. These ambient air measurement gave a general picture of the conditions during the campaign and the concentration variability. The CRDS instrument, used for the tracer release

method, was installed in a car along with a real-time GPS device and was thus mobile except for a one-night-long comparison with the FTIR. The instrumental techniques and the setup of the instruments during the campaign are detailed hereafter.

4.1 FTIR analyzers

An FTIR analyzer records a spectrum over a broad IR range (1800–5000 cm⁻¹), thereby offering the possibility of measuring a large number of species simultaneously. Spectra are stored and can be analyzed at a later date with a different method to get data with a higher accuracy or study new species. In the FTIR, the infrared signal passes first through a Michelson interferometer, then this modulated beam traverses the sample cell. The resulting time-modulated signal is then converted into an infrared spectrum through Fourier transform. The FTIR analyzer operated by the LSCE is a commercially available Ecotech instrument. The instrument operated by the Bremen University was built at the University of Wollongong, Australia. Both instruments are functionally identical. A detailed description is found in Griffith et al. (2012) and Hammer et al. (2012). Briefly, each of the two instruments consists of a commercially available FTIR interferometer (IRcube, Bruker Optics, Germany) with a 1 cm⁻¹ resolution coupled with a 3.5 L multi-pass cell with a 24 m optical path length (PA-24, InfraredAnalysis, Anaheim, USA). The cell and the interferometer are put together on an optical bench inside a temperature-controlled chamber. An in situ PT100 platinum resistance thermometer and a pressure sensor (HPM-760s, Teledyne Hastings, USA) are installed on the multi-pass cell. Nitrogen (grade 4.5) is used to purge the interferometer housing as well as the transfer optics between the cell and the interferometer. A drying system composed of a 24 in counter-flow Nafion dryer (Permapure, Toms River, NJ, USA) followed by a chemical dryer (Mg(ClO₄)₂) was located upstream from the cell.

During the campaign, both instruments were installed in small shelters without air conditioning. The first one, operated by LSCE, was installed to sample ambient air above the whole station for the majority of the time. During the last day, air was sampled above different basins. For this instrument, the pressure of the cell is controlled using a mass flow sensor mounted at the outlet of the cell, and the flow is controlled by another mass flow controller installed upstream from the drying system. Four calibration gases and a control gas were used regularly during the 5 days of the campaign for calibration (once a day, 45 min for each calibration gas) and quality control (every 3–4 h). During these 5 days, the temperature inside the shelter sometimes exceeded 30 °C. In order to keep the performances unchanged, the FTIR and the cell were kept at 32 °C instead of the typical 30 °C. However, the temperature variations in the shelter were leading to cell temperature fluctuations, and therefore the reproducibility error was higher than in the laboratory (0.01 vs. 0.005 %, respectively).

The main sampling inlet was installed on top of a building located between the clarification and the aeration basins at about 7 m a.g.l. Ambient air measurements took place from 17 September 17:00 to 20 September 2012 14:00 and then from 20 September 18:30 to 21 September 2012 05:00. During the afternoon of 20 September, measurements above the clarification, the aeration and the degassing basins were taken with the LSCE FTIR analyzer sampling inlet 50 cm above the basins to compare with the ambient sampling.

The second FTIR was operated by the Bremen University and was used to analyze samples from a floating chamber operated on the clarification and the aeration basins. Due to the complexity of moving the shelter or deploying longer lines, no other places, such as the pretreatment area, could be measured with the chamber. The chamber consisted of a large upside-down flower pot surrounded by a tractor tire inner tube, which served as a floating device. The edge of the flower pot was filled with water, so that the chamber was sealed with respect to the water surface. The edge extended 3 cm into the water. A 12 V computer fan inside the chamber ensured mixing of the air in the chamber. The volume of the chamber was 0.10 m³, and the surface area of the water in the chamber was 0.28 m². The chamber was connected to the FTIR in situ analyzer using PFA sampling lines and air from the chamber was circulated in a closed loop through the analyzer with a flow rate of 0.06 m³ h⁻¹. The data were calibrated using a suite of secondary standards measured once during the campaign with methane concentrations ranging from 1.8 to 40 ppm.

On the clarification basin, chamber placements aimed to capture spatial flux variations and covered three approximate positions as indicated in Fig. 2a. The clarification basin possessed a rotating arm, or mixer, that was used to gently stir the basin and encourage the drainage of benthic sludge towards and out of a central hole at the bottom of the basin. Whilst the mixer was on, the floating chamber was tethered to the rotating arm and moved very slowly with the arm. Consequently, whilst sampling, the chambers moved about one-half to one full rotation around the basin. The movement-induced turbulence was assumed to have a negligible effect on the flux, as the arm rotated at a slow rate, covering one rotation of 360° in approximately 30 min. Fluxes were calculated from the accumulation of methane in the chamber over time (closed chamber), as described above. On the aeration basin, two floating chamber measurements were conducted outside the aeration area. Focus was laid on measurements in the area where the aeration took place. Due to the high air flux in the aeration area, closed-chamber measurements were not suitable. Instead, we modified the setup and operated the chamber as open chamber over night (19–20 September), as detailed previously.

4.2 CRDS analyzer

For the mobile tracer release measurement, we used an acetylene/methane/carbon dioxide/water vapor analyzer based on cavity ring-down spectroscopy, an optical technology in which direct measurement of infrared absorption loss in a sample cell is used to quantify the mole fraction of the gas. This instrument (*S/N* DFADS2006, Picarro, Inc., Santa Clara, CA) is a custom analyzer based upon a standard C₂H₂/CH₄/H₂O model (G2203, Picarro, Inc., Santa Clara, CA) to which a high precision CO₂ measurement was added (Crosson, 2008). The inherent stability of the CRDS instrument allows it, when properly calibrated to traceable reference standards, to deliver accurate measurements that need very infrequent calibration relative to other CO₂ and CH₄ instrumentation. The overall measurement interval is just below 1 s (i.e., one to two measurements registered during 1 s). There are two modes of operation for this analyzer: a C₂H₂/CH₄/H₂O mode and a CO₂/CH₄/H₂O mode. The spectroscopy of CO₂, CH₄ and H₂O is identical to the algorithms that are used in several standard models from the same manufacturer (e.g., models G1301, G2301, G2401); the performance of these instruments for atmospheric measurements of CO₂, CH₄ and H₂O has been described in detail elsewhere (Crosson, 2008; Chen et al., 2010; Winderlich et al., 2010; Rella et al., 2013; Fang et al., 2013). The basic performance reported in these papers should be highly representative of the performance of this analyzer. For the C₂H₂/CH₄ mode, the performances are described in details in Mønster et al. (2014b). A series of laboratory tests was performed in order to establish the basic performance of the analyzer, consisting of continuous measurements on prepared gas mixtures. The uncertainty calculated from these tests is summarized in Table 1. The CH₄ measurements were calibrated in the field using the same calibration gases as the LSCE FTIR. The C₂H₂ measurement was not calibrated directly with a standard gas but using another instrument. For our purpose, the instrument was installed in the back of a car and was powered by the car battery. It was connected to a GPS mounted on the car roof and to internet via a 3G router. This allowed us to visualize in real time the location and intensity of the concentrations we were measuring and to ensure we are totally crossing the emission plumes. The air inlet is fixed on the GPS and its length is about 1 m long.

The scope of this campaign was to test the tracer release method to estimate the whole site emissions using mobile measurements. Three releases were performed in that manner. However, when the wind conditions were favorable, one stationary experiment, focusing on a single element of the site, was also performed. The typical transects for the mobile measurements as well as the location of the stationary experiment are indicated on Fig. 1. Mobile measurements occurred on 17 and 18 September while fixed measurements with the inlet next to the LSCE FTIR inlet were performed during the night of 19–20 September. During the first three

releases, a 0.05 m³ cylinder of C₂H₂ was situated next to the degassing basin. The flow is controlled with a glass tube flowmeter (Sho-Rate from Brooks) with a precision better than 5 % and a reproducibility of 0.5 %. During the first release episode, the wind was coming from the south. Using the nearby bridge above the Rhône (about 500 m away), we transected the plumes about 10 times. The plumes were located on the bridge and we drove before and after for at least the same distance as the length of the bridge to ensure that we were back on background levels. The flow on the flowmeter was fixed at 40 mm which translates to 10.3 kg day⁻¹. Later that day, a stationary experiment was performed to measure the emissions from the degassing basin with the car parked about 65 m away from it and the flowmeter was adjusted to a flow of 10.6 kg day⁻¹. The C₂H₂ gas cylinder was situated on the eastern edge of the degassing basin, about 7 m east of the center of the 5 m radius basin. On 19 September, the wind was stronger and coming from the north. The instruments were driven along the roads south of the station about 400 m away to cross the plumes. As for the previous day, we ensured that the plumes were fully crossed. The flow on the flowmeter was adjusted to 105 mm to compensate for the stronger dispersion which translates to 27.8 kg day⁻¹. Finally, a last experiment with the C₂H₂ cylinder close to the clarification basin to the station was conducted.

4.3 Weather station

A weather station (WXT520, Vaisala) was installed next to the FTIR and radon analyzer inlets. Wind speed, wind direction, temperature, relative humidity and atmospheric pressure were measured every second and averaged every minute.

5 Results

In this section, we present first the weather conditions and the concentration measurements that allowed us to get a general picture of the site and to estimate our instruments comparability. Then, we show the results from the two methods to estimate CH₄ emissions.

5.1 Continuous ambient air measurements

The CH₄ concentrations from the LSCE FTIR analyzer, the wind speed, the wind direction and the temperature measured during the whole campaign (except the sampling above the basins) are shown in Fig. 4. Using the wind direction, we plotted the wind rose for the whole campaign. It can be seen from the wind rose and the time series that the wind varied between two major directions during the campaign: south-southwest (SSW) and northeast. During the first day, the wind was variable but came mainly from the SSW direction. On 17 September, the wind direction was the same and steadier. On 19 and 20 September, the wind direction was again more variable with northeast being the main direction.

Temperatures followed a typical daily pattern and varied between 10 and 24 °C.

CH₄ concentrations varied between 1900 and 3000 ppb. The gaps in the data correspond to calibration periods and sampling above the basins. The highest concentrations were observed on the first and last days matching stable air mass (almost no wind speed and quick changes in wind directions). We compared these data to a suburban site, Gif-sur-Yvette (about 50 km southwest of Paris), plotted in grey in the upper panel. We see that the concentrations measured there did not present peaks like measured at the WWTP. This supports the hypothesis of very local emissions from the plant elevating the measured concentrations up to 1000 ppb above what is observed in suburban area sites that are not located on a local source spot. On the concentration wind rose, we observed no preferential direction of higher concentrations. Note that for the direction with few data and low wind speed, we observed high concentrations that can be expected from a slower dispersion.

5.2 Instrument comparison

During the last night (20 to 21 September), the CRDS and the FTIR analyzers inlet lines were placed next to each other to sample the same air. The comparison of the two is shown in Fig. 5. Contrary to the FTIR analyzer, which was calibrated regularly during the entire campaign, the CRDS analyzer was calibrated only once before the in situ measurements. However, a good agreement was observed between the two instruments with a mean difference of 2.4 ± 3.9 ppb (SD). The WMO recommendation for laboratory intercomparison is < 2 ppb in background air (WMO, 2011). We can then reasonably expect that if we had calibrated the CRDS instrument more often, we would have reached the recommended goal even for polluted air masses. Indeed, more frequent calibrations would have helped to compensate the temperature and atmospheric pressure influences on the measurements. However, the best solution would be to have the instruments in insulated shelters. Moreover, in the case of the tracer release, no calibration is needed as the instrument is linear in the range of measured concentrations and we use differences to the background to infer the fluxes.

5.3 Process-scale measurements and fluxes

Clarification basin

During the floating chamber deployment (accumulation mode), the CH₄ concentration increased in the chamber over time. A total of eight chamber runs were made on the clarification basin. Where possible (see discussion below), the increase was approximated by linear least-square fitting and the fluxes were calculated. Only four out of eight floating chamber measurements on the clarification basin exhibited an approximately linear increase (chamber runs 2 (from minute

Table 2. Fluxes measured during chamber measurements on the clarification basin.

	Chamber run no.	Mixer	CH ₄ flux (mg min ⁻¹)
Diffusive emissions ^a	2	Off	2.2 ± 0.1
	3	Off	2.2 ± 0.1
	4	Off	3.1 ± 0.2
	7	On	7.6 ± 0.4
Erratic emissions ^b	1	Off	108.5 ± 6.4
	5	On	168.8 ± 9.9
	8	On	72.9 ± 4.3
	9	On	18.5 ± 1.1

^a The fluxes were calculated by a linear fit because diffusive flux could be assumed.

^b The observed concentration increase was not linear. The numbers are based on a linear fit of the steepest increase over a 10 min period. The fluxes calculated represent an upper limit.

7 on), 3, 4, 7; see Fig. 6b–d and f). The emissions calculated from these measurements averaged 3.8 mg min^{-1} (for the individual values see Table 2). The standard deviation, calculated to assess the spread of the individual measurements, was 2.6 mg min^{-1} . It is reasonable that upscaling to the whole basin introduced uncertainty when not all locations on the basin were covered by our measurements. The uncertainty in volume and area contributed to the squared total error by 52 and 48 %, respectively, for all four diffusive flux measurements. The uncertainties associated with CV, pressure and temperature were negligible. Based on our four measurements, we consider the obtained average of 3.8 mg min^{-1} or 5.4 g day^{-1} to give the order of magnitude of the diffusive exchange flux, which represents the lower limit of the total emissions from the clarification basin. For the other four measurements (see Fig. 6a, e, g and h), the increase cannot be linearly approximated. Due to the very sudden increase of the methane concentration in the chamber, we think that erratic methane emissions caused this nonlinearity, i.e., ebullition. Since such events might occur more frequently close to the rotating arm and the number of measurements is too small for estimating the frequency of such events, it is difficult to estimate the basin methane flux generated by erratic events. However, we can state that the highest average flux for these measurements over a 10 min period was 169 mg min^{-1} (chamber run 5, Fig. 6e).

Overall it can be stated that the fluxes of methane were higher when the mixer was on and the arm rotated. The rotating arm extended down through the water column and caused increased turbulence at the water–air interface, throughout the water column and within the methane-rich sediments. The increased turbulence combined with resultant release of methane from the sediments could very likely explain the elevated flux and the high variability of the fluxes whilst the mixer was on. Runs 1 and 5 both show a high CH₄ flux that differs remarkably from the other chamber runs. These runs

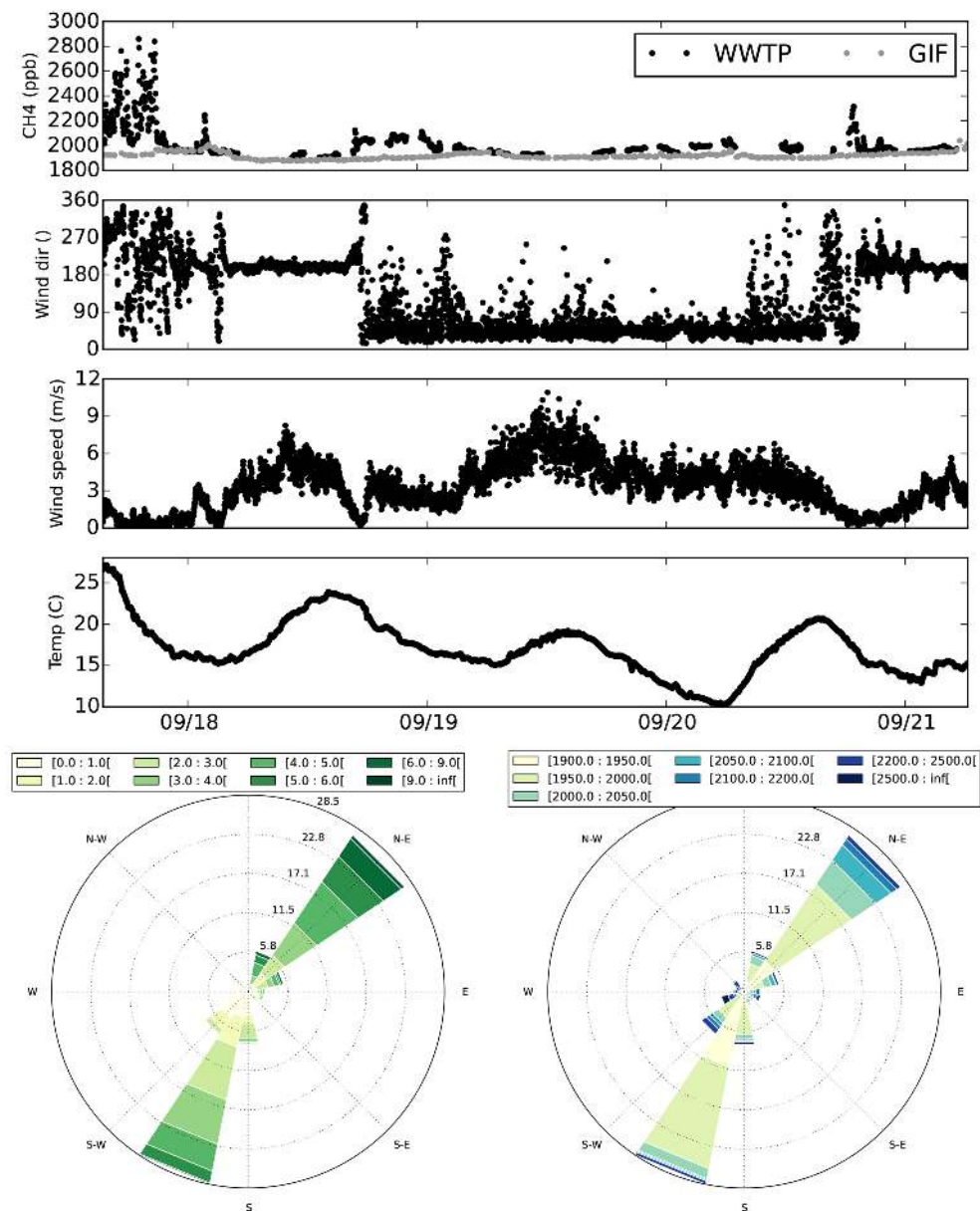


Figure 4. Upper panel: CH₄ concentrations from the LSCE FTIR analyzer at Valence and from the suburban station of Gif-sur-Yvette, wind speed, wind direction and temperature during the campaign. Lower panel: wind roses of the wind speed (m s⁻¹) and CH₄ concentration (ppb) during the campaign.

were the first measurements conducted after the mixer was turned off (run 1) and on (run 5). The switching of the mixer on and off may have momentarily increased ebullition, resulting in the nonlinear and rapid increase of the concentration in the chamber (see Fig. 6a and e). Repeated measurements at different locations in the basin and under different conditions (mixer on/off) could further reveal the actual pattern of the fluxes from the clarification basin.

Considering the lower limit (diffusive flux) of the observed fluxes, we can state that the emissions from the clarification basin due to diffusive emissions are about 5.4 ± 3.0 g day⁻¹.

In addition to the diffusive emissions, we observed erratic methane emissions, most likely due to bubbles, which would explain the very sudden increase to very high methane concentrations. Within the short time of measurements on the basin (1 day), it was not possible to do a systematic study of the methane emissions due to these erratic events. Therefore, here we can only provide an approximate estimate for erratic fluxes from the basin. We choose this approximate estimate in a way that it expresses the maximum erratic flux that we can consider possible according to our measurements. For this, we took the highest of the four erratic fluxes we mea-

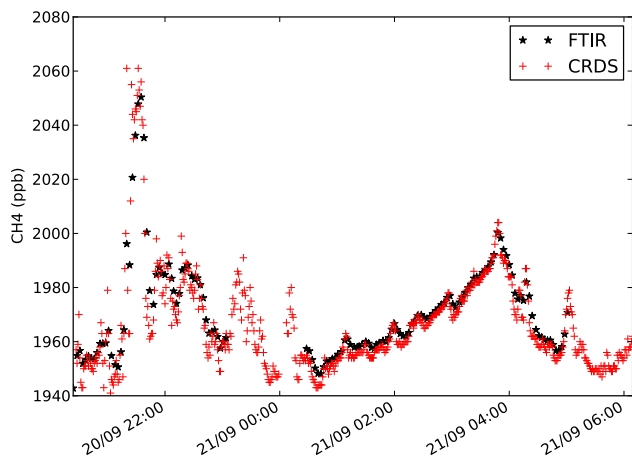


Figure 5. Comparison of CH₄ concentrations from the LSCE FTIR and CRDS analyzers during the night of 20 September.

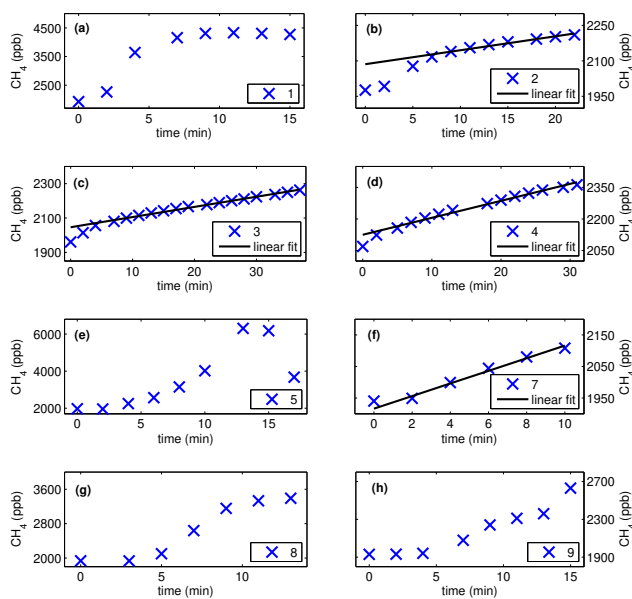


Figure 6. Floating chamber experiments conducted on the clarification basin when the mixer was off (a–d) and on (e–h), respectively.

sured and assumed that this flux, measured over 10 min, occurred for 24 h over the entire area of the basin. In that case, the emissions would sum up to 243 g day⁻¹.

Aeration basin

The fluxes from outside of the aeration area and their uncertainties were derived in the same way as for the diffusive emissions from the clarification basin. We calculated a mean flux of 36 ± 2 g day⁻¹ (38 ± 2 and 34 ± 2 g day⁻¹). This is more than 6 times higher than the diffusive flux measured on the clarification basin. We have no measurement for non-linear fluxes on the aeration basin, therefore, the value given here (36 g day⁻¹) is a conservative estimate including the dif-

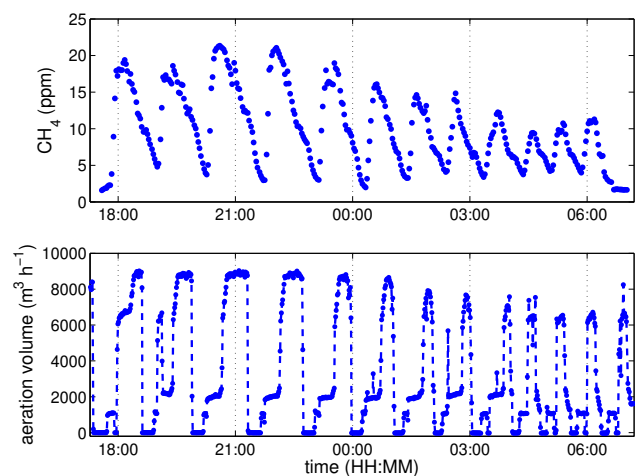


Figure 7. Measurements in the aeration area of the basin. Upper panel: methane concentration vs. time. Lower panel: respective volume of injected air during the same time period.

usive flux only and would represent the lower limit of the total flux (diffusive + erratic) from this area. The fluxes are very different where the aeration takes place. There, the floating chamber was operated in flow-through mode over night. Figure 7 shows the CH₄ mixing ratios in the chamber (upper panel) and the amount of injected air (lower panel). It can be seen that when the aeration starts, the methane concentration rises up to a maximum and already decreases before the aeration stops. We think that during the phases when no air and thus no oxygen is injected, there is a buildup of methane in the basin. Once the aeration starts, the methane is emitted from the basin with the aeration air.

The night measurements cover approximately 13 h and are therefore believed to offer a reasonably good temporal coverage for upscaling. We calculated the CH₄ emissions with Eq. (2), with a background concentration of 1973 ppb, which is the average ambient CH₄ measured with the LSCE FTIR during the night from 19 to 20 September. Accordingly, 553 ± 17 g day⁻¹ was emitted from the aeration basin (with the uncertainty of the background CH₄ concentration, taken as the largest deviation from the mean, equal to 2.5 %).

It can further be seen from Fig. 7 that the methane concentration maxima are lower during the late night than in the evening. In fact, an overall decrease of the maxima can be observed, along with shorter periods of non-aeration. Figure 8 indicates a linear correlation between the length of the non-aeration period and the methane maximum that is observed during the subsequent chamber measurement (correlation coefficient $R = 0.86$). This supports our hypothesis that methane production occurs during non-aeration times, which is, in turn, responsible for the high methane emitted in the subsequent aeration phase.

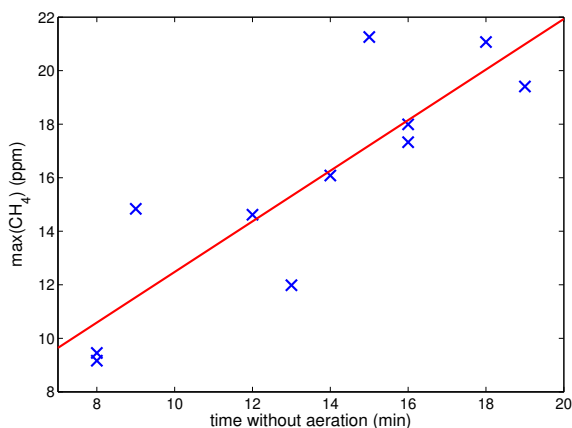


Figure 8. Methane maxima reached during the night chamber measurement vs. time without aeration. Blue are measurements; red is the linear fit.

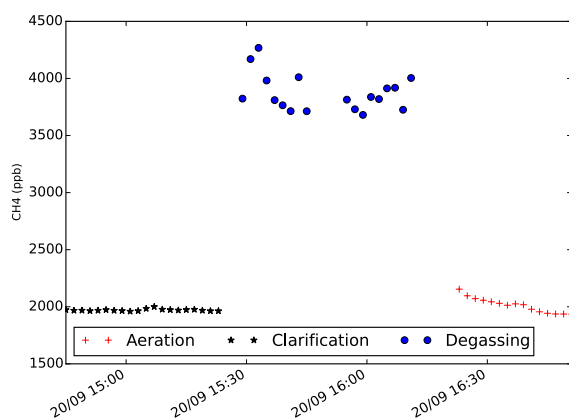


Figure 9. CH₄ concentrations measured by the LSCE FTIR analyzer over the different basins.

Degassing basin

It was not possible to measure the small degassing basin that feeds the clarification basins using the floating chamber method due to the obstructed access to this basin. However, measurements above the clarification, aeration and degassing basins were performed with the LSCE FTIR analyzer sampling inlet 50 cm above the basins (see Fig. 9). For the aeration and the clarification basins, the concentrations at the time of measurement were close to the concentrations measured for the whole station. For the degassing basin, elevated concentrations of CH₄ are measured up to 4300 ppb. The mobile CRDS CH₄/C₂H₂ instrument was also used to quantify the emissions from this source. On 18 September, with winds originating from the SSW, driving both immediately upwind and downwind of this basin (within 10 m) and the nearby clarification basin, a clear and distinct plume from this basin was identified. No significant emissions were observed from any of the clarification basins consistent with

the floating chamber measurements. The measurement vehicle was parked at a distance about 9 times greater than the separation of the C₂H₂ and CH₄ sources. We expect the plumes should be reasonably mixed at this distance, especially given the strong afternoon turbulent mixing of the atmosphere when these measurements were made (see Fig. 4). Under these well-mixed conditions, the static plume correlation method can be employed to estimate the emissions of CH₄. About half an hour of CH₄ and C₂H₂ measurements were made at this location, with the wind wafting the plumes back and forth across the measurement location. The winds came reliably from the SSW during this time, meaning that the measurements were not polluted by methane from the aeration basins or incineration building. The time series of C₂H₂ and CH₄ are shown in Fig. 10b. The signals are clearly correlated. We plotted methane as function of C₂H₂ and fitted the resulting distribution with a linear function. The fit has a slope of 0.244 ppb_{CH₄} ppb⁻¹_{C₂H₂}, with an R^2 of 0.62. Given the release rate of 10.6 kg day⁻¹ for C₂H₂, we found that the methane emissions from the degassing basin were 1.13 ± 0.5 kg day⁻¹. Given the wind direction, this emission number could include emissions from one or more of the clarification basins. However, whilst the floating chamber measurements showed that maximum emissions from the clarification basins were comparable (0.8 kg day⁻¹), on average they were about one-quarter of this amount. As this figure includes erratic fluxes, for which a conservative upper limit was given, the true fluxes are likely to be much lower. For example, if only diffusive emissions were included, then the flux would be smaller than 0.025 kg d⁻¹ per basin. This compares to 1.135 kg d⁻¹ from the degassing basin. Conclusively, the emissions from the clarification basin contribute only very little to the emissions from aquatic surfaces in the WWTP.

5.4 Plant scale

Figure 10a and c present measurements from CH₄ and C₂H₂ during the two successful tracer release episodes using dynamic measurements. We see that the acetylene baseline is very close to 0 and stable, while the CH₄ baseline varies between the releases and during them. The elevation of the signal above the background is on average 15–20 ppb for CH₄ and between 2 and 16 ppb for C₂H₂ which is 5 to 20 times lower than for the static measurements reflecting the distance to the sources compared to the static measurements. The ratio for each numbered peak was calculated and the results and their uncertainties are summed up in Table 3. We observe a large variability (approximately 35 %) between the plumes but with a consistent average between the two mobile release episodes. The average value over the 2 days is 34.2 ± 11.6 kg day⁻¹ or 83 ± 28 g yr⁻¹ per inhabitant. The errors here represent the SD of the measurements and therefore also include the emission variability.

Table 3. CH₄ emissions (kg day⁻¹) for the whole station using the tracer release method.

North wind		South wind	
Transect	CH ₄ emissions (kg day ⁻¹)	Transect	CH ₄ emissions
1	19.1 ± 0.5	1	28.7 ± 0.5
2	21.0 ± 0.5	2	49.6 ± 0.5
3	48.1 ± 0.5	3	30.0 ± 0.5
4	42.1 ± 0.5	4	25.0 ± 0.5
5	47.3 ± 0.5	5	18.6 ± 0.5
6	35.9 ± 0.5	6	47.0 ± 0.5
7	46.3 ± 0.5	7	31.2 ± 0.5
8	22.9 ± 0.5	8	
Average	35.3 ± 12.5	Average	32.9 ± 11.35

6 Discussion

6.1 Uncertainties

In this paper, we used two methods to estimate emissions with associated uncertainties. These uncertainties and the parameters they arise from are summarized in Table 4. Depending on the methods, the uncertainties range from 5 to 60%. However, in most cases there are several parameters that can be determined more accurately to reduce these uncertainties. In Table 4 the parameters in bold are the parameters with the higher uncertainty.

In the case of the closed chamber, the water area enclosed by the chamber and the air volume in the chamber are the parameters associated with the strongest uncertainties. They eventually depend on the uncertainty of the water level. Consequently, a more accurate measurement of the water level in the chamber and a minimization of its variation should be aimed at if lowering of the total uncertainty is desired. The error associated with the water surface area can be fully eliminated by choosing a box over a conic chamber. With a box, the variation of the water level would not affect the surface area across which the exchange takes place.

For the open-chamber measurements, the uncertainty comes mostly from the injected air flow measurement and is related to the WWTP equipment (in this case the measurement uncertainty is 2%).

For the tracer release method, the largest uncertainties come from the collocation assumption of the signals and the baseline estimates. These uncertainties can be reduced by lengthening the period of “clean” air measurement between each plume crossing and by ensuring that the signals are correlated. Moreover, controlled release exercises as done by Mønster et al. (2014b) can help quantify the non-collocation error. In our study, when the acetylene cylinder was located near the degassing basin, the two signals were slightly shifted in time (not shown) depending on the direction we were driving. This shows that this location was not optimal for sampling the methane emissions from the station when still close to it (about 500 m). Judging by the horizontal displacement

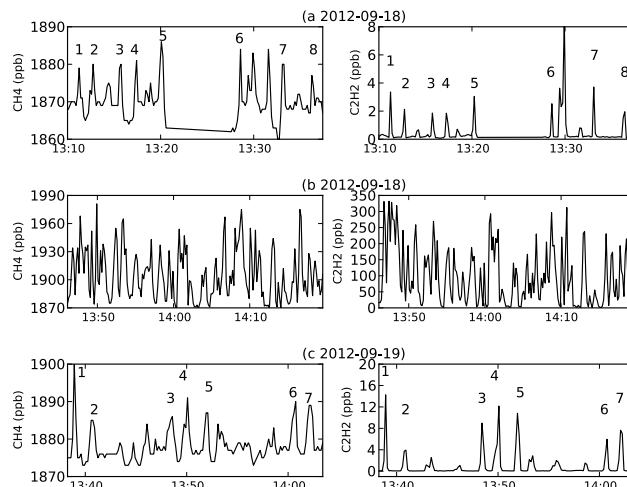


Figure 10. Concentrations of CH₄ and C₂H₂ during the three tracer release episodes. Episode (a): estimation of the whole plant emission on 18 September with a south wind; the C₂H₂ cylinder is located in A (see Fig. 1). Episode (b): estimation of the degassing basin emissions on 18 September with a south wind; the C₂H₂ cylinder is located in A. Episode (c): estimation of the whole plant emissions on 19 September with a north wind; the C₂H₂ cylinder is located in A. The numbers indicate the signals (peaks) that are used to calculate the CH₄ emissions.

of the plumes with respect to each other, and the direction of the wind, there was another methane source west of the degassing basin. When the cylinder was moved to the clarification basin, we observed an opposite horizontal displacement, which indicates that the cylinder was now located west to the axis of the methane plume propagating in the direction of the wind. It seems then that the optimal location would have been near the LSCE FTIR sampling lines. This means a displacement of about 50 m, which translates into an underestimation of 10–15% on the flux (Mønster et al., 2014b). This would translate into an average flux of about 38.9 kg day⁻¹, which lies within our uncertainty estimate (11.6 kg day⁻¹).

In summary, there is potential to reduce the uncertainty of each method, which should be considered when aiming for more robust WWTP emission estimates. Moreover, longer measurement campaigns over different times of the year would also allow us to catch the variability of the emissions of the site. Finally, if aiming for a general estimates, several WWTPs have to be investigated.

6.2 First insights into the Valence WWTP methane emissions

The results from the chamber and the tracer release measurements are summarized in Table 5. If we add the maximum contribution from the different basins, the emissions are approximately 3.1 kg day⁻¹, i.e., 8% of the total emissions observed from the facility using the tracer dilution method. We confirm then that the main source of emissions from the

Table 4. Summary of the uncertainties resulting from the different methods (in percent). The parameters with the higher uncertainty are in bold.

	Closed chamber	Open chamber	Tracer release
Uncertainties	6–60	3	15
Parameters	mixing ratio, chamber surface area and volume , pressure, temperature, linear fit	mixing ratio, air flow	mixing ratios, acetylene flux, correlation, baseline estimates

Table 5. Summary of the results from the process to the site scale. The given uncertainties were determined in different ways. Refer to the main text for details.

	Clarification basins (3)	Aeration basins (2) non-aeration area	Aeration basins (2) aeration area	Degassing basin (1)	WWTP
CH ₄ emissions (kg day ⁻¹)	0.02 ± 0.01	0.07 ± 0.01	1.11 ± 0.03	1.13 ± 0.24	34.2 ± 11.6
	+ emissions by erratic events (at most 0.77 ± 0.05)	+ emissions by erratic events (not determined)			

plant is not these basins but is located elsewhere, as shown in Fig. 1d.

In other studies presented in Daelman et al. (2012), the emissions from municipal waste water treatment plants using activated sludge treatment, such as the Valence plant, varied from 39 to 306 g yr⁻¹ per inhabitant. The higher limit was found for a plant using a sludge digester producing biogas. This unit was found to emit three-quarters of the total emissions of the plant, leaving approximately 77 g yr⁻¹ per inhabitant emitted by the other processes. The Valence plant estimate agrees with this last value (83 g yr⁻¹ per inhabitant).

We can also compare this estimate with inventories estimates. The European Database for Global Atmospheric Research (EDGAR, Olivier et al. (1996), which provides gridded maps, and the CITEPA, which is responsible for the French inventory, use the IPCC methodology to estimate CH₄ emission factors from WWTPs (see Eq. 4).

$$EF = \text{BOD} \times 365 \cdot B_o \times \sum_x (WS_x \cdot MCF_x), \quad (4)$$

where B_o is the maximum CH₄ production capacity, WS_x is the percentage of a certain process used in the WWTP and MCF_x is the conversion rate of this process. According to the hypothesis, the conversion rate for a WWTP like Valence (aerobic treatment) should be at the maximum 0.1 with a maximum of 0.4 if not well managed (IPCC chapter 6 table 6.3, http://www.ipcc-nggip.iges.or.jp/public/2006gl/pdf/5_Volume5/V5_6_Ch6_Wastewater.pdf). The B_o is usually estimated to be 0.6 kg_{CH₄} kg⁻¹_{BOD}. CITEPA estimates for France an average emission factor of 74 g yr⁻¹ per inhabitant, which is very close to the Valence estimate. Using the data from the WWTP, we can calculate the conversion rate and compare it to the awaited value. Here, we find a conver-

sion rate of 0.07, which is in the expected range. From these first measurements, it seems then that the Valence WWTP is an average French WWTP in terms of CH₄ emissions.

7 Conclusions

We measured CH₄ at one of the waste water treatment plants in the city of Valence, France. Two instruments – FTIR and CRDS – combined with different methods – floating chamber and tracer release – were used. They allowed us to span several scales from the individual process to the site.

The duration of the campaign, 4 days only, was too short to accurately quantify the emissions and sample the site variability. However, we have shown that these methods are suitable to evaluate emissions at these different scales and that they complement each other. The estimated uncertainty for any of the methods is under 60 % and could in most cases be reduced by more experiments (e.g., controlled release for the tracer release method) or by a more precise measurement of the experiment apparatus (e.g., the area of the chamber in contact with the water). From a qualitative point of view, the emissions from the waste water plant are representative of an average French WWTP. The estimates on three structures from the plant, the aeration, clarification and degassing basins, show that even though these are the largest open structures on site, they are not the main emitters of methane on the plant. Concentration measurements seem to indicate that the incinerator building and the pretreatment could be the main sources. Finally, these estimates are in the same range of values as found in the literature.

This study demonstrates the use of new techniques, FTIR and CRDS analyzers, to estimate small-scale emissions and

help improve emission factors for bottom-up inventories. Longer periods of measurements are, however, necessary to be able to sample statistically significant numbers of events and get more accurate emission estimates.

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