Methane in the Baltic and North Seas and a reassessment of the marine emissions of methane

H. W. Bange, U. H. Bartell, S. Rapsomanikis, and M. O. Andreae

Biogeochemistry Department, Max Planck Institute for Chemistry, Mainz, Germany

Abstract. During three measurement campaigns on the Baltic and North Seas, atmospheric and dissolved methane was determined with an automated gas chromatographic system. Area-weighted mean saturation values in the sea surface waters were $113 \pm 5\%$ and $395 \pm 82\%$ (Baltic Sea, February and July 1992) and $126 \pm 8\%$ (south central North Sea, September 1992). On the bases of our data and a compilation of literature data the global oceanic emissions of methane were reassessed by introducing a concept of regional gas transfer coefficients. Our estimates computed with two different air-sea exchange models lie in the range of 11-18 Tg CH₄ yr⁻¹. Despite the fact that shelf areas and estuaries only represent a small part of the world's ocean they contribute about 75% to the global oceanic emissions. We applied a simple, coupled, three-layer model to evaluate the time dependent variation of the oceanic flux to the atmosphere. The model calculations indicate that even with increasing tropospheric methane concentration, the ocean will remain a source of atmospheric methane.

Introduction

Methane is an atmospheric trace gas that contributes about 15% to the greenhouse effect [Houghton et al., 1990] and plays an important role in tropospheric and stratospheric chemistry [Cicerone and Oremland, 1988]. Today, the global tropospheric concentration of methane is increasing at an annual rate of about 0.7-1% [Steele et al., 1987, 1992; Khalil and Rasmussen, 1990]. On the basis of estimates of the sources and sinks of methane, predictions of further impacts on the world's climate can be made. Therefore it is necessary to evaluate the contributions of natural and anthropogenic sources to the global budget of methane. The world's oceans, as a natural source of methane, play only a modest role in the global methane budget. Oceanic emissions account for about 2% of all natural and anthropogenic sources [Cicerone and Oremland, 1988]. However, an accurate assessment of the oceanic source is still missing because of the paucity of available data [Ehhalt, 1974; Ehhalt and Schmidt, 1978; Cicerone and Oremland, 1988; Lambert and Schmidt, 1993].

In this paper we present recent measurements of atmospheric and dissolved methane from different marine environments. Data sets from two cruises on the Baltic Sea and one campaign on the North Sea, together with data from the literature, are the basis for a reassessment of the release of methane from the ocean to the atmosphere. Using a simple three-layer model, we evaluate the time dependent variation of the flux across the ocean-atmosphere interface due to increasing tropospheric methane concentrations.

Copyright 1994 by the American Geophysical Union.

Paper number 94GB02181. 0886-6236/94/94GB-02181\$10.00

Study Areas and Cruise Tracks

Baltic Sea

On basis of the hydrographic and bathymetric conditions the Baltic Sea can be divided into a series of basins (e.g., Belt Sea, Bornholm Sea, Gotland Sea, and Bothnian Sea) [*Ehlin*, 1981; *Omstedt*, 1990]. These basins (except the shallow transition basins Kattegat, Belt Sea, and Arkona Sea) are characterized by a permanent salinity stratification [*Kullenberg*, 1981]. The freshwater supply from the rivers and the discontinuously inflowing saline North Sea water result in a brackish surface layer and more saline deep and bottom water masses. The two Baltic cruises took place in February 1992 and July/August 1992 on the German research vessel R/V *Alexander von Humboldt*. On the first cruise we occupied 63 stations in the southern Belt Sea, Arkona Sea, Bornholm Sea, and Gotland Sea. The second cruise covered 23 selected stations with the same coordinates as those occupied during the first cruise (Figure 1a).

North Sea

The North Sea forms part of the European shelf and can be classified as a coastal sea. In contrast to the Baltic, the North Sea is characterized by the great influence of tidal motions on the ecological conditions. Owing to the tides, pronounced horizontal and vertical exchange and transport effects occur. The salinity distribution is dominated by water masses entering from the North Atlantic and the British Channel [*Otto et al.*, 1990]. Our measurements were made during a campaign on the research platform R/P *Nordsee* (September 1992) located in the southern part of the central North Sea (54°42' N, 7°10' E) (Figure 1b).

Methods

For the determination of atmospheric and dissolved methane we used a gas chromatographic (GC) system which was operated

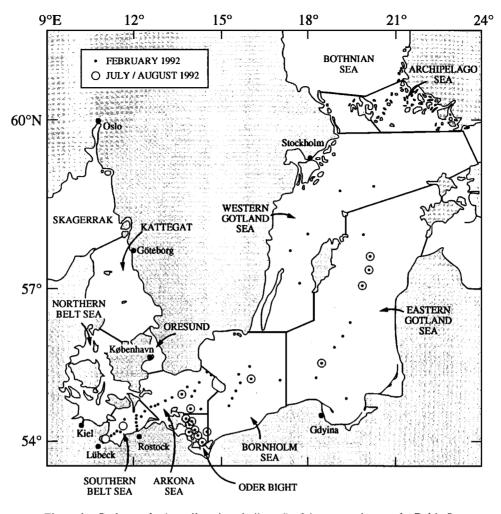


Figure 1a. Cruise tracks (not all stations indicated) of the two cruises on the Baltic Sea.

semiautomatically during the first cruise and fully automatically during the following cruises. Air was pumped continuously (1-2 L min⁻¹) through 9.6 mm (OD) (6.4 mm (ID)) polypropylene tubing from the ship's mast into the lab. The maximum length of the tubing was 90 m (R/P Nordsee), resulting in a maximal flushing time of about 3 min. The location of the air inlet was chosen to avoid contamination from the exhaust of the ship's engine. Seawater was pumped continuously (15 L min⁻¹ Baltic Sea, 30 L min⁻¹ North Sea) from a depth of 3-5 m into the lab using the ship's water pumps. By means of a seawater-air equilibrator developed by R.F. Weiss (Scripps Institution of Oceanography, La Jolla, California) the surface seawater was equilibrated with sample air (for details, see Butler et al. [1988]). Temperature differences between the seawater and the water in the equilibrator due to heating from the pumps were observed during the three expeditions. The differences ranged from 2 ± 0.1 °C (winter cruise, Baltic Sea) to 0.8 ± 0.1 °C (summer cruise, Baltic Sea) and 0.5 ± 0.1 °C (North Sea). Therefore calculated saturation values were corrected for this temperature difference. Possibly occurring methane production within the equilibrator was only observed during the second Baltic Sea cruise and was eliminated by cleaning when necessary.

The setup of our system consisted of four sample gas streams (ambient air, equilibrator headspace, and two standards) which could be selected for analysis with a system of three magnetic

valves. Before the analysis the sample gas streams were dried with SicapentTM (phosphorus pentoxide drying agent with indicator, E. Merck, Darmstadt, Germany). After the sample stream passed the moisture trap a 2-mL thermostated sample loop was flushed for 5 min with a sample flow of 25 mL min⁻¹ and then the sample was injected. We used helium (55 mL min⁻¹) as carrier gas and hydrogen (30 mL min⁻¹) and synthetic air (250 mL min⁻¹) as flame gases. For the gas chromatographic separation we used a packed column (1.80 m x 3.2 mm, stainless steel) filled with washed molecular sieve 5A (mesh 80/100, Alltech Associates, Inc.). The GC oven was operated isothermally (60°C), and the heated zone of the flame ionization detector was held at a temperature of 250°C. Two sets of standard gas mixtures were used for calibration. On the first cruise we used a mixture of methane in synthetic air with methane mixing ratios of 1.80 and 5.96 parts per million by volume $(ppmv) \pm 5\%$. The second set had methane mixing ratios of 1.73 and 3.02 ppmv $\pm 2\%$ in synthetic air and was calibrated by the manufacturer (DEUSTE Steininger GmbH, Mühlhausen, Germany) against standards of the U.S. National Institute of Standards and Technology (Gaithersburg, Maryland). The average absolute error of a single measurement was ± 0.1 ppmv.

During the Baltic Sea cruises the measurements of atmospheric and dissolved methane were performed when the ship was on station (time resolution 10 min on the first cruise and

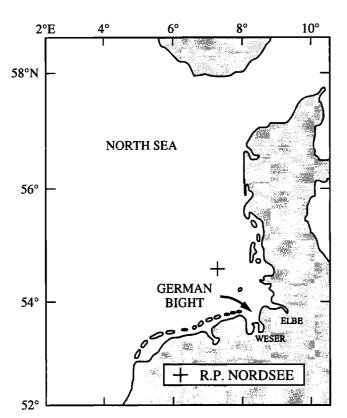


Figure 1b. Location of the research platform R/P *Nordsee* in the south central North Sea.

25 min on the second). The number of measurements during a single station varied, depending on the time spent at the position. Measurements of the standard gas mixtures were performed during the time when the ship moved to the next station. A continuous working mode was applied on the R/P *Nordsee*. During this campaign a series of measurements of atmospheric and dissolved methane followed by standards was repeated every 125 min.

Saturation values were calculated as the ratio of dissolved methane to the expected equilibrium value derived from the ambient air concentration by applying the equation of Wiesenburg and Guinasso [1979]. Fluxes of methane across the seawater-atmosphere interface were computed with the equations of Liss and Merlivat [1986]. The transfer coefficient was corrected according to the Schmidt number Sc of methane (Sc = kinematic viscosity of seawater/diffusion coefficient ofmethane) at the actual salinity and temperature of the seawater. Diffusion coefficients were obtained from the expression of Jähne et al. [1987], and the kinematic viscosity was calculated with the equations given by Siedler and Peters [1986]. All meteorological and oceanographic data were obtained from the ships' records and conductivity-temperature-depth (CTD) measurements performed during the cruises. The wind speed, recorded at different heights during the three expeditions, was normalized to a 10-m height by using a neutral drag coefficient [Garratt, 1977].

Results

During both Baltic Sea cruises the surface waters were found to be supersaturated with respect to the ambient atmospheric concentrations of methane (Figures 2a and 2b). The atmospheric mixing ratios during both cruises were almost uniform and ranged from 1.58 to 2.68 ppmv (average 1.89 ± 0.16 ppmv, winter cruise) and 1.72 to 2.22 ppmv (average 1.89 ± 0.09 , summer cruise). An overview of the average saturations measured on the two cruises is given in Figure 3. The comparison of the two data sets shows a strong seasonal variation. During winter conditions (February 1992) the areaweighted average saturation was $113 \pm 5\%$, which is much lower than the area-weighted average saturation of $395 \pm 82\%$ found under summer conditions (July/August 1992) (Table 1). A more detailed description of the distribution of the methane saturation is obtained by looking at the saturations observed in the different basins of the Baltic Sea (Table 2). The mean saturation values were generally higher in summer compared with winter, but the differences were not uniform. In the southern Belt Sea we found the strongest differences, whereas the saturation values of the eastern Gotland Sea show only a modest variation. A slight trend of decreasing saturation values from the southern Belt Sea $(162 \pm 41\%)$ to the western Gotland Sea $(104 \pm 2\%)$ was observed in winter. This trend was more pronounced during the summer cruise. During the summer expedition the saturation values ranged from $3760 \pm 3520\%$ (southern Belt Sea) to $160 \pm 24\%$ (eastern Gotland Sea). The data from the southern Belt Sea generally show great variability compared with data from the other basins. However, measurements at stations in the southern Belt Sea or Arkona Sea, which are expected to have coastal influences, showed significantly higher values during both cruises. The Oder Bight, a part of the southern Arkona Sea and influenced by the plume of the river Oder, is a good example of a coastal area. We observed in this region a clear relationship between salinity and methane water concentrations (Figures 4a and 4b). During both the winter and the summer cruise the concentrations of dissolved methane in the Oder Bight decreased with increasing salinity.

For each station we calculated flux densities based on the actual saturation value and the actual wind speed normalized to 10 m height. These flux densities computed with the equations of *Liss and Merlivat* [1986] ranged during the winter cruise from $0.11 \pm 0.28 \ 10^{-3} \ \text{nmol m}^{-2} \ \text{s}^{-1}$ to $0.17 \pm 0.03 \ \text{nmol m}^{-2} \ \text{s}^{-1}$. For the summer expedition we calculated flux densities of $1.17 \pm 0.23 \ 10^{-3} \ \text{nmol m}^{-2} \ \text{s}^{-1}$ to $13.9 \pm 2.3 \ \text{nmol m}^{-2} \ \text{s}^{-1}$ (Table 1).

In September 1992 we measured atmospheric and dissolved methane in the southern central North Sea at one station during 14 days, which provided a time series (Figure 5a). The atmospheric mixing ratio ranged from 1.70 to 2.46 ppmv, with an average value of 1.92 ± 0.10 ppmv. Except for three single measurements, the surface water was supersaturated with methane during the campaign. During the first 3 days the saturation shows a slight decreasing trend from 140% to 115%, then for the next 5 days the values increase up to 140%. During the following 5 days the saturation values slightly decrease again to saturation values of about 115% (Figure 5b). Since the saturation values showed no significant trend, we averaged them over the measurement period, $126 \pm 8\%$. The calculated flux densities were in the range of $-6.6 \pm 8.0 \ 10^{-3} \text{ nmol m}^{-2} \text{ s}^{-1}$ to 0.12 ± 0.03 nmol m⁻² s⁻¹. The average flux density was $0.025 \pm$ $0.023 \text{ nmol m}^{-2} \text{ s}^{-1}$ (Table 1).

Discussion

Methanogenesis (anoxic microbial fermentation of organic matter) is the sole known pathway for the formation of biogenic

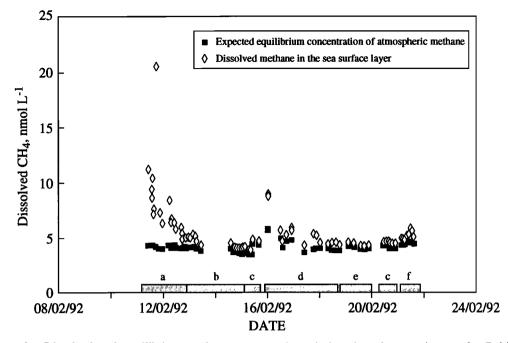


Figure 2a. Dissolved and equilibrium methane concentrations during the winter cruise on the Baltic Sea (equilibrium values calculated after *Wiesenburg and Guinasso* [1979]). Abbreviations are a, southern Belt Sea; b, Arkona Sea; c, Bornholm Sea; d = eastern Gotland Sea; e, western Gotland Sea; and f, Oder Bight.

methane in marine sediments and anoxic marine basins [Cicerone and Oremland, 1988]. Besides this biogenic methane, seepages of thermogenic methane, which is mainly a product of petroleum formation, occurs in shelf areas [Hovland et al., 1993]. The presence of methane in surface waters of estuarine, deltaic, and shallow coastal waters may be predominantly due to the production of methane in sediments enriched in organic material, riverine sources, or shallow submarine sources [Butler et al., 1987; de Angelis and Lilley, 1987; Ward et al., 1989; Cynar and Yayanos, 1992; Hovland et al., 1993; Jones and Amador, 1993]. The observation of methane supersaturation in the surface layer of the open ocean, away from sedimentary sources, raises some questions which remain unanswered. The reported concentrations of dissolved methane in the surface layer cannot be explained by the normal methanogenesis mechanism, which requires strictly anoxic and highly reducing conditions, or by physical transport

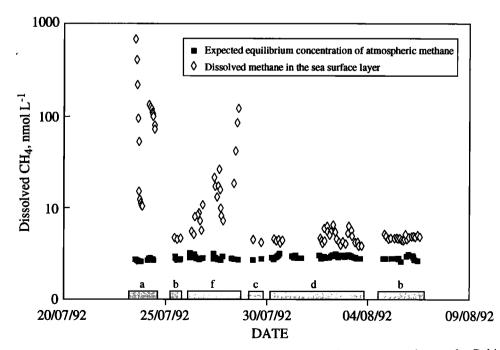


Figure 2b. Dissolved and equilibrium methane concentrations during the summer cruise on the Baltic Sea (equilibrium values calculated after *Wiesenburg and Guinasso* [1979]). Abbreviations are a, southern Belt Sea; b, Arkona Sea; c, Bornholm Sea; d = eastern Gotland Sea; e, western Gotland Sea; and f, Oder Bight.

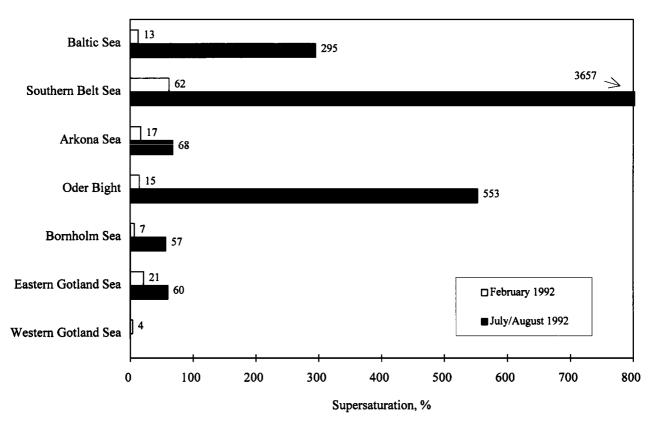


Figure 3. Overview of the mean supersaturations in various basins of the Baltic Sea.

processes. Therefore an additional in situ mode of methane production is required [Scranton and Brewer, 1977; Scranton and Farrington, 1977; Owens et al., 1991]. Various mechanisms have been discussed, e.g., formation in anoxic microniches (i.e., in decaying organic particles, faecal pellets, or zooplankton guts) or production in association with phytoplankton [Scranton and Brewer, 1977; Traganza et al., 1979; Burke et al., 1983; Conrad and Seiler, 1988; Bianchi et al., 1992; Karl and Tilbrook, 1994]. Production of methane in marine environments by a nonbiological mechanism, e.g., photoproduction, has not been observed [*Wilson et al.*, 1970].

Methane saturation data from other shelf areas comparable with the Baltic Sea are rare (Table 3). Nevertheless, measurements in coastal regions, e.g., the southern California Bight [*Cynar and Yayanos*, 1992] or the estuarine-deltaic system of the Orinoco River/Gulf of Paria [*Jones and Amador*, 1993] show also significant seasonal variations.

Determination of the methanogenesis and methane oxidation

	Table 1. Summar	of Results	Obtained From	Various Cruises
--	-----------------	------------	---------------	-----------------

	Average Atmospheric Mixing Ratio, ppmv			Mean Saturation, %	Number of Stations	Flux Density, nmol m ⁻² s ⁻¹		
			_	Minimum	Maximum			
Baltic Sea I				_				
Feb. 1992	1.89 ± 0.16	113 ± 5^{a}	63	$0.11 \pm 0.28 \ 10^{-3}$	0.168 ± 0.025			
Baltic Sea II								
July 1992	1.89 ± 0.09	395 ± 82^{a}	23	$1.17 \pm 0.23 \ 10^{-3}$	13.89 ± 2.99			
North Sea								
Sept. 1992	1.92 ± 0.10	126 ± 8	117 ⁶	-6.6 ± 8.0 10 ⁻³	0.117 ± 0.032			

Flux densities F are from the air-sea exchange model of Liss and Merlivat [1986] as follows: F = methane transfer coefficient $T_{CH4} \propto \Delta C$; $T_{CH4} = 0.17$ wind speed at 10-m height $u_{10} (600/Sc)^{2/3}$, $u_{10} \leq 3.6$ m s⁻¹; $T_{CH4} = (2.85 u_{10} - 9.65) (600/Sc)^{1/2}$, $3.6 < u_{10} \leq 13$ m s⁻¹; $T_{CH4} = (5.9 u_{10} - 49.3) (600/Sc)^{1/2}$, $u_{10} > 13$ m s⁻¹.

* Values are area weighted.

^b Value indicates number of measurements at the fixed station.

	Southern Belt Sea		Arko	ona Sea	Oder Bight		
	Winter	Summer	Winter	Summer	Winter	Summer	
- Average, ± 1σ	162 ± 40	3757 ± 3524	117±6	168 ± 10	115 ± 7	653 ± 943	
Maximum	262 ± 105 (497 ± 110)	15,040 ± 789	128 ± 8	190 ± 12	127 ± 8	4526 ± 273	
Mininum	120 ± 8	412 ± 119	110±9	151 ± 9	106 ± 7	162 ± 9	
Number of Stations	17	2	12	2	9	14	
	Bornholm Sea		Eastern Gotland Sea		Western Gotland Sea		
	Winter	Summer	Winter	Summer	Winter		
Average, $\pm 1\sigma$	107 ± 3	157	121 ± 14	160 ± 24	104 ± 2		
Maximum	112 ± 9	165 ± 11	156 ± 7	215 ± 116	. 10'	7 ± 7	
Minimum	101 ± 7	148 ± 10	108 ± 8	131 ± 8	10	1 ± 8	
Number of Stations	10	1	10	4		5	

Table 2. Methane Saturations of Various Basins of the Baltic Sea in Winter and Summer

All values are in percent. Parentheses indicate extreme maximum.

rates in the sediments of the brackish water ecosystem at the east coast of the Island of Rügen (southern Arkona Sea) showed that methanogenesis rates are relatively high. However, 2-68% of the produced methane is oxidized within the sediments and the methanogenesis rates depend strongly on the availability of

organic matter [Berger and Heyer, 1990; Heyer et al., 1990]. Nevertheless, the observation of high summer saturations in the shallow southern Belt Sea (depths < 25 m) and the Oder Bight (depths < 20 m) might be caused by sedimentary production of methane and/or riverine inputs. On the last day of the

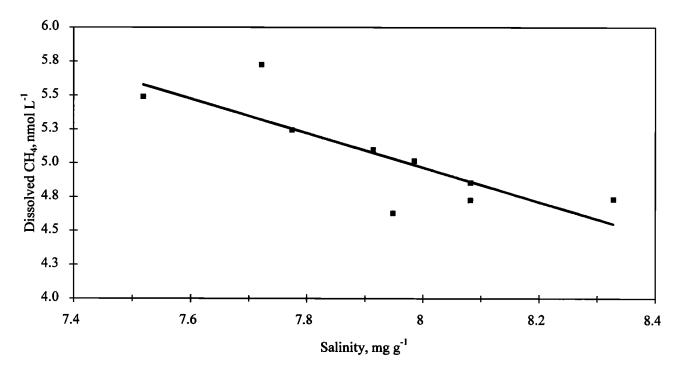


Figure 4a. Dissolved methane in the Oder Bight in February 1992; correlation coefficient $r^2 = 0.6497$ (number of samples n = 9).

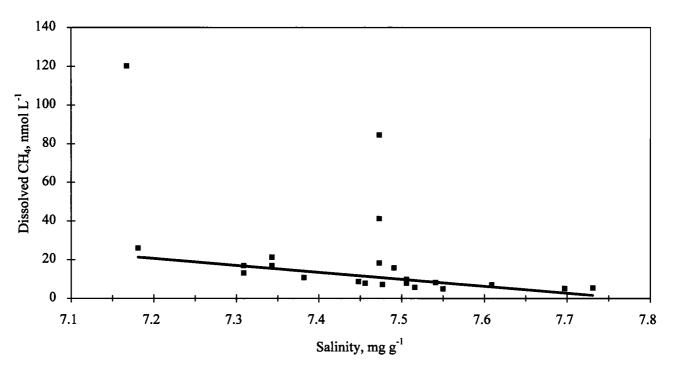


Figure 4b. Dissolved methane in the Oder Bight in July 1992; correlation coefficient $r^2 = 0.6868$ (number of samples n = 18).

measurement program in the Oder Bight the wind speed increased from low (1.5-7.0 m s⁻¹) to high speeds with peaks of 15.7 m s⁻¹. At the same time, we observed at a station, which had been sampled 2 days before during low wind speeds, a considerable increase of the dissolved methane from 8.16 ± 3.10 nmol L⁻¹ to 120.2 ± 3.16 nmol L⁻¹. This might be explained by enhanced release of methane from disturbed sediments to the well-mixed water column (water depth at this station, 11 m) or advection of methane-enriched water from a different source region. This temporary change of the saturation is reflected by the scatter of the data in Figure 4b.

Methane seeps as another possible source of dissolved methane are documented for the Kattegat and Skagerrak region [Hovland et al., 1993] and are not known for the measurement area. The slightly lower supersaturation observed in the deeper and stratified northern basins might be explained by horizontal

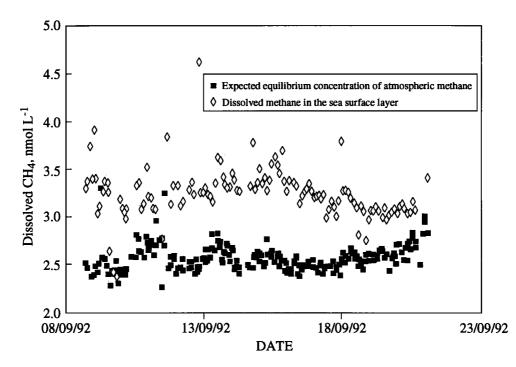


Figure 5a. Dissolved and equilibrium methane concentrations in south central North Sea (equilibrium values calculated after *Wiesenburg and Guinasso* [1979]).

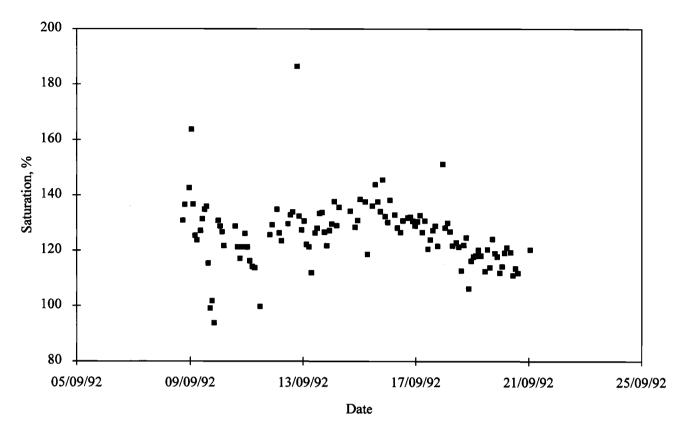


Figure 5b. Saturation of methane at the R/P Nordsee in the south central North Sea.

flows of surface waters from the well-mixed shallow southern basins (southern Belt Sea, Arkona Sea) to northern basins (Bornholm Sea, Gotland Sea) in conjunction with additional riverine sources.

The saturation values observed in the south central North Sea are comparable with observations of Conrad and Seiler [1988] in the southern North Sea and the British Channel in January 1979 and October 1980. Measurements from Scranton and McShane [1991] in the southern bight of the North Sea and the plume of the Rhine and Scheldt Rivers in March 1989 ranged from 95-130% (offshore) up to 12,000% (near shore) saturation. In fall the water column around the location of the R/P Nordsee (water depth 30 m) is typically well-mixed and influenced by water masses coming from the German Bight [Otto et al., 1990]. Salinity measurements during the campaign confirm this hydrographic situation. Therefore the observed saturations which were most of the time significantly higher than 100% could result from riverine sources (rivers Elbe and Weser) or release from sediments. Active methane seepages reported for the northern region of the North Sea [Hovland et al., 1993] do not influence the south central North Sea. Methane releases from oil and gas rigs which are located in the southern North Sea seem to be negligible [Scranton and McShane, 1991] and probably do not influence the waters around the R/P Nordsee.

The flux densities calculated for the three expeditions (Table 1) are in good agreement with fluxes given in the literature (Table 3). A compilation of methane measurements in marine environments (only measurements in the surface layer are considered) shows that the occurrence of methane supersaturation is not limited to biologically productive areas (i.e., estuaries, coastal zones, and upwelling areas) (Table 3).

Where no saturation value was reported, we estimated this value from the seawater concentration using (if available) the sea surface temperature and salinity and assuming an annual tropospheric growth rate of 0.8%. Saturation equilibrium or undersaturation have been observed only in a few cases. Undersaturation might be the result of seasonal effects due to changes of hydrographic parameters, especially the sea surface temperature.

There is a clear trend toward higher supersaturation ratios in coastal and highly productive environments. The data listed in Table 3 indicate that most parts of the ocean act as a source for methane. One has to be aware, however, that there is a tendency to make oceanographic measurements during the spring, summer, and fall seasons. Seasonal differences can cause considerable variation in the distribution of methane in biologically productive regions (e.g., Baltic Sea), so that the present data set is likely to be biased toward the higher saturation ratios typical of the warmer seasons.

On the bases of our own results and the literature data in Table 3 we reassessed the oceanic flux of methane as the sum of regional fluxes from the different oceanic provinces of the world's oceans. Using the bathymetric classification of oceanic provinces by *Menard and Smith* [1966], regional annual mean methane transfer coefficients were calculated. These coefficients were derived from matrices of mean annual CO₂ transfer coefficients computed with the National Center for Atmospheric Research Community Climate Model 1 (NCAR CCM1) (D. J. Erickson, private communication, 1994). Two parameterizations of the air-sea exchange were applied, the recently published stability dependent model of *Erickson* [1993] (hereinafter referred to as E 93) and the widely used relationship of *Liss and*

	Saturation, %	Flux Density, nmol m ⁻² s ⁻¹	Number of Stations	Date	Reference
		Es	tuaries		
Lake Nitinat	1900		1	Nov. 1967	Lamontagne et al. [1973]
Chesapeake Bay	1100 - 1700		6	June 1968	Swinnerton et al. [1969]
Saanich Inlet	1300		- 1	June 1978	Lilley et al. [1982]
Saanich Inlet	1300		1	July, Aug. 1978	Bullister et al. [1982]
Alsea Bay	300 - 29000	2.09	7	1979 - 1982	de Angelis and Lilley [1987]
Yaquina Bay	300 - 11600	(0.03 - 15.2)	9		8 71 1
Salmon Bay	5200 - 12400	()	1		
Yaquina Bay	100 - 5500 ^a	1.41 - 3.28	10	1983 - 1984	Butler et al. [1987]
Saanich Inlet	1200	0.03	1	1983, Aug. 1986	Ward et al. [1989]
Pettaquamscutt River estuary	8000 - 11000	4.70 - 49.19 ^b	8	Aug. 1989, Aug.	Scranton et al. [1993]
ettuquumseutt reiver estuary	0000 11000	0.217 - 3.906°	Ŭ	1990, May, June, Aug. 1991	
Hudson River estuary	2500 - 7200	3.54 - 4.48	12	March, Aug. 1991	de Angelis and Scranton [1993]
Amvrakikos Bay, Greece	522	0.167	cont.	July 1993	H. W. Bange et al. $(1994)^d$
Univiaritos Day, Greece	146 - 974	0.107	cont.	July 1775	
	140 - 974	Coasta	and Shelf		
ted Sea	170	Considi	l	June 1967	Lamontagne et al. [1973]
Black Sea	180		1	May 1967	Lumonugne et ut. [1975]
Southern Beaufort Sea	100 - 790 ^a		18	Aug. 1974	Macdonald [1976]
Soutient Beautort Sea	85 - 260 ^a		25	Aug. 1974 Aug. 1975	Mucuonata [1976]
Northwestern Gulf of Mexico	120 - 23900		25 50	1975 - 1976	Brooks et al. [1981]
Western Mediterranean	260		1	July 1978	Traganza et al. [1979]
			2	Nov., Dec. 1984	Belviso et al. [1987]
Philippine Sea	$120,790^{a}$			NOV., DEC. 1984	Beiviso et al. [1987]
North Molucca Sea	380 ^a		1		
Celebes Basin	700, 3100 ^a		2	1 1 1007	
Gulf of Aden	140 - 170 ^a		4	July 1987	Jean-Baptiste et al. [1990]
Western Sagami Bay, Japan	150 ^a		1	Feb. 1988	Gamo et al. [1988]
Black Sea	200 - 500	0.31	1	July 1988	Reeburgh et al. [1991]
Southern North Sea	95 - 12000	0.07 - 700	75	March 1989	Scranton and McShane [1991]
NW Mediterranean Sea	477		1	Oct. 1989	Lambert and Schmidt [1993]
	181 - 2625			May 1990	
Southern California Bight	85 - 4600 ^a		70	Nov. 1989	Cynar and Yayanos [1992]
	125 - 42000 ^a			March 1990	
Southern California Bight	130 - 430 ^a	0.04, 1.7	1	Jan. 1990	Ward [1992]
	150ª			July 1990	
Sea of Okhotsk, ice covered	≤ 7 5 00	1.39		March 1991,	Lammers [1993]
ice free	≥ 1000			summer 1992	
		Oligotrophic a	and Transition	al	
Vestern tropical North	130 (120 - 150)		25	May 1970	Lamontagne et al. [1973]
tlantic	130 (120 - 150)		50	June 1970	
astern tropical North Pacific	140 (120 - 150)		70	Aug. 1971	
lorwegian Greenland Sea	130 (120 - 170)		23	Aug. 1971	
Freenland Ice Pack	130 (120 - 130)		6	May 1971	
argasso Sea	140 (130 - 140)		3		
Caribbean Sea	150 (120 - 160)		4		
ariaco Trench	160		1		
ulf of Mexico	130		1	May 1967	
fediterranean Sea	130		1	May 1967	
astern Mediterranean Sea				-	
tlantic Ocean, Sargasso Sea	100		6	June 1968	Swinnerton et al. [1969]
outh Pacific, Antarctica	100		16	Feb. 1971	Williams and Bainbridge [1973]
orth, South Pacific	130	≤ 0.0 8	10	Nov., Dec. 1972	Lamontagne et al. [1974]
intarctica, ice covered	80			, — -	0 L J
	148 - 231	0.03	8	Feb. 1975	Scranton and Brewer [1977]
Western subtropical North-					
Atlantic	100 100		0	Mer. 1070	Dunka at al [1002]
tlantic astern tropical North Pacific	100 - 180	0.01	9 aant	May 1979	Burke et al. [1983]
Atlantic Eastern tropical North Pacific	100 - 180 110 ± 5	0.01	9 cont.	Jan., Feb. 1979,	Burke et al. [1983] Conrad and Seiler [1988]
-		0.01			

Table 3. Compilation of Methane Measurements in Surface Waters of Various Marine Environments

Table 3. (continued)

	Saturation, %	Flux Density, nmol m ⁻² s ⁻¹	Number of Stations	Date	Reference
Mid-Atlantic Ridge 12N-15N	100 - 130 ^a		3	Feb., March 1985	Charlou et al. [1988]
Loihi Seamount, Hawaii	80 ^a		1	Sept. 1985	Gamo et al. [1987]
Cariaco Basin	104	0.003	1	Jan., Feb. 1986	Ward et al. [1987]
East Pacific Rise 11N-13N	120, 110 ^ª		2	Dec. 1986	Charlou et al. [1991]
Northeast Atlantic 47N,20W	125 ± 9	0.020	cont.	April, May 1992	S. Rapsomanikis et al. (unpublished data 1992)
Aegean Sea	176 ± 16	0.021	cont.	July 1993	H. W. Bange et al. (1994) ^d
Northwestern Levantine Basin	118 ± 7	0.035		•	,
Eastern Ionian Sea	135 ± 24	0.040			
		Upv	velling		
Arabian Sea	170		1	June 1967	Lamontagne et al. [1973]
Walvis Bay	200 - 30000		9	Dec. 1975 - Jan. 1976	Scranton and Farrington [1977]
Subtropical North Atlantic	287		1	July 1978	Traganza et al. [1979]
Equatorial Pacific	110-120		cont.	July 1979 - June 1980	R. F. Weiss (personal communication, 1981)
Equatorial Pacific	105 ± 2	0.005	cont.	Feb., March 1990	Bates et al. [1993]
Arabian Sea	192 (157 - 286)	0.05 - 0.16	13	Sept., Oct. 1986	Owens et al. [1991]
Peruvian Coast	86 - 438	- 0.006 - 0.097	7	March, April 1992	Lammers and Suess [1994]

Numbers in parentheses indicate ranges. Cont. indicates continous measurements.

^a Values are extrapolated; for details see text.

^b Values are measured.

^c Values are calculated.

^d Reference is The Aegean Sea as a source of atmospheric nitrous oxide and methane, submitted to Marine Chemistry, 1994.

Merlivat [1986] (hereinafter referred to as LM 86). The resulting matrices were plotted and analyzed graphically to derive the mean CO_2 regional transfer coefficients. These values were corrected according to the Schmidt numbers *Sc* of methane and carbon dioxide at a salinity of 35‰

$$T_{CH4} = T_{CO2} (Sc_{CO2} / Sc_{CH4})^{0.5}$$

where T is the transfer coefficient.

The Schmidt numbers for CO_2 were calculated with the equation given by *Erickson* [1993]. Annual mean sea surface temperatures for each oceanic province were derived from monthly data fields which were used as input for the calculations of the transfer coefficients with the NCAR CCM1 (D. J. Erickson, personal communication, 1994). The resulting methane transfer coefficients T_{CH4} and sea surface temperatures are listed in Table 4. Since the spatial resolution of the NCAR CCM1 was 2.8° x 2.8° latitude-longitude (i.e., 8192 grid points), some of the oceanic provinces were characterized by less than 10 grid points. Therefore the mean values of these provinces (e.g., Gulf of California, Baltic Sea, Red Sea etc.) have to be regarded as uncertain. Additional uncertainties in the wind and temperature fields and the applied equations used in the CCM1 introduce an error for the transfer coefficient of at least 50% [*Erickson*, 1988].

Regional methane fluxes S_o across the air-sea interface were calculated with the equation

$$S_o = A T_{CH_4} (SR C_a - C_a)$$

where T_{CH4} is the methane transfer coefficient (listed in Table 4),

SR is the saturation ratio (SR = saturation/100), A is the area of the oceanic province, and C_a is the equilibrium seawater concentration for a tropospheric mixing ratio of 1.7 ppmv calculated with the equation of *Wiesenburg and Guinasso* [1979] with a salinity of 35‰ (except for the estuary province where we used 18‰) and the annual sea surface temperature of the oceanic province.

Following the concept of the regional transfer coefficients, we tried to compute the corresponding regional methane saturation values. Again we used the classification of Menard and Smith [1966], but in this case we additionally distinguished between the shelf and the open ocean region of an oceanic province. (Menard and Smith [1966] defined the shelf region as the whole region from the shoreline to the base of the steep continental slope.) The methane saturation values were taken from the overview in Table 3. If saturation values were reported for an open ocean or a shelf region, then we calculated the mean saturation value applying the following procedure: when different types of saturation values (i.e., mean values, ranges, and single values) existed, we decided to give the reported mean values high weight and the ranges low weight. In the cases of reported ranges we calculated the mean by using the maximum and the minimum values when more than one range was given for the same region. When only single values for a region were available, we took the average. Oceanic regions represented only by three single values or less were not considered. Table 5 gives an overview of the flux densities and the resulting sea-to-air fluxes from the various oceanic provinces. Because saturation data were only available for 70% of the world's ocean area, we extrapolated the missing fluxes using area-weighted saturation values for the open ocean

	SST ± 1σ, K	$T_{CH4} \pm 1\sigma^{a},$ 10 ⁻⁵ m s ⁻¹	$T_{CH4} \pm 1\sigma^{b},$ 10 ⁻⁵ m s ⁻¹
Pacific Ocean	 287.8 ± 1.5	6.2 ± 4.0	3.9 ± 1.7
Asiatic Mediterranean	300.4 ± 0.3	3.6 ± 1.6	2.5 ± 1.3
Bering Sea	276.7 ± 4.0	7.2 ± 1.7	3.9 ± 0.7
Sea of Okhotsk	272.1 ± 11.6	5.0 ± 1.7	3.0 ± 0.8
Yellow and East China Sea	290.6 ± 7.0	5.7 ± 2.5	3.8 ± 1.4
Sea of Japan	283.3 ± 8.7	4.9 ± 1.6	3.4 ± 0.9
Gulf of California ^c	293.1	3.0	2.4
Atlantic Ocean	283.8 ± 1.4	6.2 ± 3.4	3.8 ± 1.5
American Mediterranean	298.0 ± 1.3	4.8 ± 1.7	3.7 ± 1.2
European Mediterranean	292.6 ± 5.6	3.3 ± 1.0	2.3 ± 0.8
Black Sea ^c	285.2	2.6	2.0
Baltic Sea ^c	277.4	2.6	2.3
North Sea	282.7 ± 4.1	3.9 ± 0.7	3.0 ± 0.4
Indian Ocean	283.0 ± 3.3	7.1 ± 4.3	4.1 ± 1.8
Arabian Sea	300.2 ± 0.9	3.7 ± 1.2	2.7 ± 1.1
Red Sea [°]	299.3	2.5	1.6
Persian Gulf ^e	299.0	2.4	1.2
Arctic Ocean	252.6 ± 15.1	0.97 ± 0.44	0.77 ± 0.35
Arctic Mediterranean	253.2 ± 17.4	0.91 ± 0.33	0.76 ± 0.29
Global Mean ^d	286.2	6.0	3.7
		(5.6) ^e	(3.3) ^e

Table 4. Regional Sea Surface Temperatures (SST) and Methane Transfer Coefficients T_{CH4}

^a Source is Erickson [1993].

^b Source is *Liss and Merlivat* [1986].

° Mean values are derived from less than 10 grid points.

^d Values are area weighted.

^e Global area weighted means are computed by Erickson [1993] and corrected for a sea surface temperature of 286.2 K.

(120%) and the shelf regions (395%). Introducing area-weighted mean transfer coefficients of 6.2 10^{-5} m s⁻¹ (E 93) and 3.7 10^{-5} m s⁻¹ (LM 86), we estimated the oceanic fluxes for the missing open ocean and shelf regions.

In Table 5 we show separately the estuarine region which represents an important part of the oceanic methane source. The sum of the regional fluxes yielded global oceanic fluxes of 17.8 Tg CH₄ yr⁻¹ (E 93) and 10.9 Tg CH₄ yr⁻¹ (LM 86) (Table 5). The discrepancy between the two values shows the great uncertainties which are introduced by different air-sea exchange models. Transfer coefficients derived from ¹⁴C measurements and dualtracer experiments indicated that calculations based on the approach of Liss and Merlivat [1986] seem to underestimate the gas transfer coefficient [Wanninkhof et al., 1993]. Thus our calculations suggest that the global oceanic methane flux lies in the range from 11 to 18 Tg CH₄ yr⁻¹ considering the result of the Liss and Merlivat [1986] model as a lower limit and the result of the Erickson model [1993] as a upper limit. This is still in reasonable agreement with the widely used estimation of Cicerone and Oremland [1988] (mean, 10 Tg CH₄ yr⁻¹; range, 5-20 Tg CH_4 yr⁻¹) which is based mainly on the calculations of Ehhalt [1974] and an estimation published by Khalil and Rasmussen [1983] (13 Tg CH₄ yr⁻¹). A recently published flux estimation of 10 Tg CH₄/a [Lambert and Schmidt, 1993] is identical to the mean value of Cicerone and Oremland [1988]. Despite the fact that shelf regions (including the estuaries) occupy only about 16% of the world's ocean area they contribute approximately 75% to the global oceanic methane emissions. Nevertheless, estimations of the oceanic flux of methane are generally suffering from the poor database and the neglect of the seasonal variability of the methane saturation in the seawater surface layer. A further uncertainty is caused by the fact that about 5% of the world's ocean (e.g., Arctic Ocean, Arctic Mediterranean, Sea of Okhotsk, Bering Sea, Baltic Sea, etc.) are temporarily covered by ice, and therefore no exchange of methane between the water and the atmosphere occurs during this period.

Model calculations

Despite the fact that the marine methane emissions play only a modest role in the global budget the following question arises: How does the oceanic flux respond to an increasing tropospheric methane mixing ratio? From ice core measurements it is known that since the beginning of the industrialization (approximately 150 years ago), the tropospheric methane mixing ratio has increased from 0.8 ppmv to 1.7 ppmv [Houghton et al., 1990]. Reliable measurements of dissolved methane are available only since the late 1960s (see Table 3). Moreover, no continuous measurements of dissolved methane in representative areas of the world's ocean are available, so that no conclusions for the temporal development of the oceanic flux can be drawn.

Our aim was to model the time dependent variation of the flux density across the ocean-atmosphere interface over a time of 100 years to show the effect of increasing tropospheric methane on the marine methane emissions. The model consists of the following three coupled reservoirs (Figure 6): (1) mixed surface layer of the ocean,

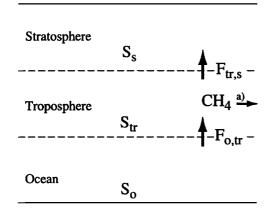
			Erickso	n [1993]	Liss and Merlivat [1986]	
Oceanic Region	Area, Average Saturation, 10 ³ km ² %	Flux Density, nmol m ⁻² s ⁻¹	Oceanic Flux, Tg CH ₄ yr ⁻¹	Flux Density, nmol m ⁻² s ⁻¹	Oceanic Flux, Tg CH ₄ yr ⁻¹	
Pacific Ocean	154,942	116	0.023	1.79	0.014	1.12
(Pacific Ocean)	(11,299)	(240)	(0.200)	(1.14)	(0.125)	(0.71)
(Sea of Okhotsk)	(1254)	(2739)	(4.563)	(2.89)	(2.733)	(1.73)
Atlantic Ocean	73,899	124	0.037	1.40	0.023	0.87
American Mediterranean	2468	145	0.040	0.05	0.031	0.04
European Mediterranean	1045	185	0.059	0.03	0.041	0.02
(European Mediterranean)	(1465)	(266)	(0.116)	(0.09)	(0.080)	(0.06)
Black Sea	245	316	0.138	0.02	0.107	0.01
(Baltic Sea)	(382)	(254)	(0.121)	(0.02)	(0.105)	(0.02)
(North Sea)	(575)	(126)	(0.026)	(0.01)	(0.020)	(0.01)
Arabian Sea	1560	192	0.061	0.05	0.045	0.04
(Arctic Ocean)	(5874)	(259)	(0.118)	(0.35)	(0.093)	(0.28)
Σ Open Ocean	234,159			3.3		2.1
Σ (Shelf Regions)	(20,849)			(4.5)		(2.8)
Open Ocean	72,450	120 ^ª	0.029	1.08 ^b	0.018	0.65 ^c
(Shelf Regions)	(34,527)	(395) ^a	(0.434)	(7.57) [♭]	(0.261)	(4.54)°
(Estuaries)	(1400)	(1230)	(1.876)	(1.32) ^b	(1.126)	(0.80)°
Σ Open Ocean	306,609			4.4		2.8
Σ (Shelf Regions)	(56,776)			(13.4)		(8.1)
Total	363,385			17.8		10.9

Table 5. CH₄ Saturations and Fluxes for Various Oceanic Provinces

Shelf regions are given in parentheses. All "area" data are from Menard and Smith [1966] except North Sea, Otto et al. [1990]; Arabian Sea, Owens et al. [1991]; and estuaries, Woodwell [1980]. The Σ indicates sum. Flux densities F and oceanic fluxes S_o are calculated as follows: $F = T_{CH4} \times (SR \times C_a - C_a)$, where T_{CH4} is gas transfer coefficient; SR, saturation ratio (saturation/100); and C_a , equilibrium water concentration based on the tropospheric methane mixing ratio (1.7 parts per million by volume); $S_o = F \times A$, where A is area of the oceanic province.

^a Value is area weighted extrapolation.

^b Value is calculated with global area weighted transfer coefficient $T_{CH4} = 6.2 \ 10^{-5} \text{ m s}^{-1}$ Erickson [1993]. ^c Value is calculated with global area weighted transfer coefficient $T_{CH4} = 3.7 \ 10^{-5} \text{ m s}^{-1}$ Liss and Merlivat [1986].



$$\frac{dC_o}{dt} = -\frac{F_{o,tr}}{H_o} + S_o$$
$$F_{o,tr} = T_{o,tr} (C_w - C_a),$$

(2) troposphere,

$$\frac{dC_{tr}}{dt} = \frac{F_{o,tr}}{H_o} - \frac{F_{tr,s}}{H_{tr}} + S_{tr} - K C_{(OH)} C_{tr},$$

and (3) stratosphere:

$$\frac{dC_s}{dt} = \frac{F_{tr,s}}{H_s} - S_s$$
$$F_{tr,s} = T_{tr,s} (C_{tr} - C_s)$$

Figure 6. Scheme of the three-layer model; $F_{o,tr}$ is flux density between the ocean o and troposphere tr; $F_{tr,s}$, flux density between the troposphere and stratosphere s; S_o , oceanic source; S_{tr} tropospheric (i.e., terrestrial) source; and S_{s} stratospheric sink. The label "a" indicates consuming reaction CH₄ + OH \rightarrow CH₃ + H₂O.

where o is ocean; tr, troposphere; s, stratosphere; w, dissolved methane; a, dissolved equilibrium methane calculated from the atmospheric value; C, methane concentration; H, height of the reservoir; $C_{(OH)}$, mean tropospheric OH concentration, and K, mean reaction constant for $CH_4 + OH \longrightarrow CH_3 + H_2O$.

Table 6. Parameters for the Three-Layer Model

	Model Input	Range
Height of ocean surface mixed layer ^a , m	70	± 25
Height of troposphere, m	11,000	-
Height of stratosphere, m	21,000	-
Oceanic source ^b , Tg CH ₄ yr ⁻¹ :	14	11 - 18
Ferrestrial source ^c , Tg CH ₄ yr ⁻¹	500	± 100
Stratospheric sink ^c , Tg CH ₄ yr ⁻¹	10	± 5
Preindustrial mixing ratio, ppmv	0.8 ^d	± 0.06 ^e
Reaction rate with OH ^f , 10 ⁻²¹ m ³ s ⁻¹ molecules ⁻¹	2.49	0.93 - 42.8
Globally averaged OH concentration ^g , 10 ¹¹ molecules m ⁻³	8.7	± 1.0
Fransfer coefficient ocean-troposphere ^h , 10 ⁻⁵ m s ⁻¹	5	3.7 - 6.0
Fransfer coefficient troposphere-stratosphere ⁱ , 10 ⁻⁶ m s ⁻¹	6	

^a Source is *Li et al.* [1984].

^b Source is this paper (see Table 5).

^c Source is Crutzen [1991].

^d Source is *Houghton et al.* [1990].

^e Source is Rasmussen and Khalil [1984].

^f Source is Vaghjiani and Ravishankara [1991].

⁸ Source is Prinn et al. [1992].

^h Source is this paper (see Table 4).

ⁱ This is a fitted parameter.

 $F_{o,tr}$ represents the flux density between the ocean and the troposphere, while $F_{tr,s}$ is the flux density between the troposphere and the stratosphere. Analogical to this, $T_{o,tr}$ stands for the gas transfer coefficient at the ocean/troposphere interface and $T_{tr,s}$ is an arbitrarily chosen model parameter (without physical meaning) at the troposphere/stratosphere interface. S_{tr} is the tropospheric source term which is equal to the sum of the terrestrial sources, and S_s represents the stratospheric sink.

We introduced the mean of the two reassessed global oceanic fluxes (14 Tg CH₄ yr⁻¹), and assumed it to be equal to the oceanic production term S_o . The in situ production or destruction (e.g., methane oxidation) rates for oceanic methane are not incorporated explicitly because these mechanisms are still unknown and poorly quantified. Methane oxidation is clearly too slow [Jones, 1991; Ward, 1992; Jones and Amador, 1993] to

Table 7. Summary	of	the 1	Model	Scenarios
------------------	----	-------	-------	-----------

Scenario	Modulation of Terrestrial Source Function S_{tr}	Resulting Annual Tropospheric Growth Rate	Trend of Resulting Dissolved Methane
Α	constant	none, 0 % yr ⁻¹	constant
В	~ t	linear, 0.7 % yr ⁻¹	constant
С	$\sim t + t^2$	accelerated	increasing
D	$\sim t - t^2$	decelerated	decreasing

The stratospheric sink and the oceanic source remain constant, t is time term.

cause a significant effect on the distribution in the surface layer of the most parts of the world's oceans. A constant oceanic methane source can reasonably be assumed because the production is controlled by ecological parameters, which are independent of atmospheric methane concentration, and because the ventilation to the atmosphere (timescale of days to weeks) dominates over the biological sinks (timescale of years) [Jones, 1991; Ward, 1992; Jones and Amador, 1993]. All input parameters for the model are summarized in Table 6.

The model starts with the preindustrial tropospheric mixing ratio of 0.8 ppmv and immediately includes all sources with a source strength of 341 Tg CH₄ yr⁻¹, which is arbitrarily chosen, but we have no information about the time variation of the sources during the last 100 years. Four scenarios reflecting different time dependencies of the increase of the tropospheric methane have been modeled. The various time dependencies of the tropospheric increase were obtained by modulating the terrestrial source function S_{tr} with a linear and a quadratic time term (Table 7). Scenario B was modulated to yield after 100 years a source strength of about 500 Tg CH₄ yr⁻¹, which is in good agreement with a budget calculation of Crutzen [1991]. Concentrations of dissolved methane were calculated from the tropospheric concentration value using the equation of Wiesenburg and Guinasso [1979] with a mean salinity (35‰) and the mean temperature of the world's ocean surface waters (Table 4). The reaction constant of the consuming reaction was averaged for the tropospheric temperature gradient.

The model results are illustrated in Figures 7a-7d, which show the tropospheric (Figure 7a) and dissolved methane (Figure 7b), and the resulting ocean-atmosphere flux densities (Figure 7c) and saturations (Figure 7d). The concentrations of dissolved methane in the mixed layer follow closely the trends of the tropospheric methane (Figure 7b). In the case of no increase or linear increase of the tropospheric methane (scenarios A and B) the flux densities reach a constant value (i.e., steady state conditions)

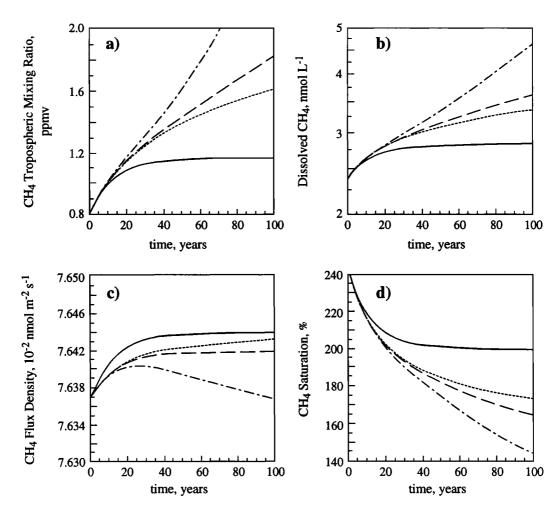


Figure 7. Results of the model calculations with time variation of (a) tropospheric methane, (b) dissolved methane concentration (logarithmic scale), (c) flux densities across the ocean-atmosphere interface, and (d) saturation. Solid line indicates scenario A (0% annual tropospheric growth rate); dashed line, scenario B (0.7% annual tropospheric growth rate); dashed-dotted line, scenario C (accelerated annual tropospheric growth rate); and dotted line, scenario D (decelerated annual tropospheric growth rate).

after approximately 60 years. In scenarios C and D the flux densities show linearly increasing (scenario C) or linearly decreasing (scenario D) trends after a period of approximately 40 years (Figure 7c). The differences between the flux densities F of the four scenarios are negligible and the resulting oceanic flux S_o $(S_o = F \times A, \text{ where } A \text{ is area of the world's ocean})$ is in the same range for all scenarios (about 14 Tg CH_4 yr⁻¹). The analyses of the trend of tropospheric methane during the last decade by Steele et al. [1992] and Dlugokencky et al. [1994] are indicating that scenarios C and D could be regarded as extreme cases which bracket the most realistic case (scenario B). On the basis of the model results we conclude that the world's oceans will continue to be a source of methane even with the observed annual tropospheric growth rates. This is illustrated by Figure 7d which shows the resulting trends of the methane saturations. Only with the assumption of an accelerated increase of tropospheric methane is a slight decrease in the flux density across the oceanatmosphere achieved (Figure 7c). Even in this case, however, the ocean remains a source of methane. To generate a negative flux density, an extreme (unrealistic) tropospheric accumulation of methane would be required.

Our prediction of continued oceanic emissions of CH_4 is in contrast to the hypothesis of some authors [*Cicerone and Oremland*, 1988; *Lambert and Schmidt*, 1993], who suggested a significantly decreasing oceanic flux and postulated that the ocean will become a sink for methane in conjunction with the increase of tropospheric methane mixing ratios. This discrepancy is due to a difference in assumptions. *Lambert and Schmidt* [1993] assumed that the concentration of dissolved methane is fixed at the present-day value, which would require an increasing internal sink function or decreasing production. Since there is no evidence that these processes are controlled by dissolved methane concentrations, we feel that a constant net production in the surface ocean and thus a constant emission is a more realistic assumption.

Conclusions

The Baltic Sea and the North Sea are sources of methane to the atmosphere. We observed a strong seasonal variation of the methane saturation in the Baltic Sea. The Oder Bight as a typical shallow coastal region showed higher saturation values due to the Oder River plume and sedimentary production than offshore regions of the Baltic Sea. Occasional events, e.g., high wind speeds, could cause a considerable change in the methane saturation in the Oder Bight. On the basis of a compilation of methane measurements from different oceanic regions our reassessment of the global oceanic emissions, applying two different air-sea exchange models, yielded a low limit of 11 Tg CH₄ yr⁻¹ and a high limit of 18 Tg CH₄ yr⁻¹. About 75% of the global oceanic flux was attributed to shelf regions (e.g., Baltic Sea, North Sea, etc.). Model calculations with a simple, coupled three-layer model showed that even with increasing tropospheric concentrations of methane, a net sea-to-air flux of methane to the atmosphere will remain at a level close to the present source strength.

Acknowledgments. We thank the captains and crews of the R/V Alexander von Humboldt and R/P Nordsee for their helpful assistance at sea and G. Schebeske for his technical support. We thank D. J. Erickson (National Center for Atmospheric Research, Boulder, Colorado) for providing the matrices of the transfer coefficients and the sea surface temperatures and H. Gimm for plotting them. We appreciate the support of D. Nehring (Institute for Baltic Sea Research, Warnemünde, Germany) who enabled us to participate on the Baltic Sea expeditions. The campaign on the R/P Nordsee was part of the EUROTRAC subproject ASE. This research was supported by the Max Planck Society.

References

- Bates, T. S., K. C. Kelly, and J. E. Johnson, Concentrations and fluxes of dissolved biogenic gases (DMS, CH₄, CO, CO₂) in the Equatorial Pacific during the SAGA 3 experiment, J. Geophys. Res., 98, 16,969-16,977, 1993.
- Belviso, S., P. Jean-Baptiste, B. C. Nguyen, L. Merlivat, and L. Labeyrie, Deep methane maxima and ³He anomalies across the Pacific entrance to the Celebes Basin, *Geochim. Cosmochim. Acta*, 51, 2673-2680, 1987.
- Berger, U., and J. Heyer, Distribution and activity of methanogenic bacteria in a brackish water ecosystem, *Limnologica*, 20, 141-144, 1990.
- Bianchi, M., D. Marty, J.-L. Teyssie, and S. W. Fowler, Strictly aerobic and anaerobic bacteria associated with sinking particulate matter and zooplankton fecal pellets, *Mar. Ecol. Prog. Ser.*, 88, 55-60, 1992.
- Brooks, J. M., D. F. Reid, and B. B. Bernard, Methane in the upper water column of the northwestern Gulf of Mexico, J. Geophys. Res., 86, 11,029-11,040, 1981.
- Bullister, J. L., N. L. Guinasso, and D. R. Schink, Dissolved hydrogen, carbon monoxide, and methane at the CEPEX site, J. Geophys. Res., 87, 2022-2034, 1982.
- Burke Jr., R. A., D. F. Reid, J. M. Brooks, and D. M. Lavoie, Upper water column methane geochemistry in the eastern tropical North Pacific, *Limnol. Oceanogr.*, 28, 19-32, 1983.
- Butler, J. H., R. D. Jones, J. H. Garber, and L. I. Gordon, Seasonal distribution and turnover of reduced trace gases and hydroxylamine in Yaquina Bay, Oregon, *Geochim. Cosmochim. Acta*, 51, 697-706, 1987.
- Butler, J. H., J. H. Elkins, C. M. Brunson, K. B. Egan, T. M. Thompson, T. J. Conway, and B. D. Hall, Trace gases in and over the West Pacific and East Indian Oceans during the El Niño-Southern Oscillation event of 1987, *Data Rep. ERL ARL-16*, Air Resour. Lab., Natl. Oceanic and Atmos. Admin., Silver Spring, Md., 1988.
- Charlou, J. L., L. Dmitriev, H. Bougault, and H. D. Needham, Hydrothermal CH₄ between 12°N and 15°N over the Mid-Atlantic Ridge, *Deep- Sea Res.*, *Part A*, 35, 121-131, 1988.
- Charlou, J. L., H. Bougault, P. Appriou, P. Jean-Baptiste, J.

Etoubleau, and A. Birolleau, Water column anomalies associated with hydrothermal activity between $11^{\circ}40'$ and $13^{\circ}N$ on the East Pacific Rise: Discrepancies between tracers, *Deep- Sea Res., Part A, 38,* 569-596, 1991.

- Cicerone, R. J., and R. S. Oremland, Biogeochemical aspects of atmospheric methane, *Global Biogeochem. Cycles*, 2, 299-327, 1988.
- Conrad, R., and W. Seiler, Methane and hydrogen in seawater (Atlantic Ocean), *Deep- Sea Res., Part A*, 35, 1903-1917, 1988.
- Crutzen, P. J., Methane's sinks and sources, *Nature*, 350, 380-381, 1991.
- Cynar, F. J., and A. A. Yayanos, The distribution of methane in the upper waters of the Southern California Bight, J. Geophys. Res, 97, 11,269-11,285, 1992.
- de Angelis, M. A., and M. D. Lilley, Methane in the surface waters of Oregon estuaries and rivers, *Limnol. Oceanogr.*, 32, 716-722, 1987.
- de Angelis, M. A., and M. I. Scranton, Fate of methane in the Hudson River and Estuary, *Global Biogeochem. Cycles*, 7, 509-523, 1993.
- Dlugokencky, E. J., K. A. Masaire, P. M. Lang, P. P. Tans, L. P. Steele, and E. G. Nisbet, A dramatic decrease in the growth rate of atmospheric methane in the northern hemisphere during 1992, *Geophys. Res. Lett.*, 21, 45-48, 1994.
- Ehhalt, D. H., The atmospheric cycle of methane, *Tellus*, 26, 58-70, 1974.
- Ehhalt, D. H., and U. Schmidt, Sources and sinks of atmospheric methane, Pure Appl. Geophys., 116, 452-464, 1978.
- Ehlin, U., Hydrology of the Baltic Sea, in *The Baltic Sea*, edited by A. Voipio, pp. 123-134, Elsevier Science, New York, 1981.
- Erickson III, D. J., Simulation of the global air-sea transfer velocity of helium, *Geophys. Res. Lett.*, 15, 1495-1498, 1988.
- Erickson III, D. J., A stability dependent theory for air-sea gas exchange, J. Geophys. Res., 98, 8471-8488, 1993.
- Gamo, T., J.-I. Ishibashi, H. Sakai, and B. Tilbrook, Methane anomalies in seawater above the Loihi submarine summit area, Hawaii, *Geochim. Cosmochim. Acta*, 51, 2857-2864, 1987.
- Gamo, T., J.-I. Ishibashi, K. Shitshima, M. Kinoshita, M. Watanabe, E. Nakayama, Y. Sohrin, E.-S. Kim, T. Masuzawa, and K. Fujioka, Anomalies of bottom CH₄ and trace metal concentrations associated with high heat flow at the Calyptogena community off Hatsu-shima Island, Sagami Bay, Japan: A preliminary report of Tansei Maru KT-88-1 cruise Leg-1, *Geochem. J.*, 22, 215-230, 1988.
- Garratt, J. R., Review of drag coefficients over oceans and continents, *Mon. Weather Rev.*, 105, 915-929, 1977.
- Heyer, J., U. Berger, and R. Suckow, Methanogenesis in different parts of a brackish water ecosystem, *Limnologica*, 20, 135-139, 1990.
- Houghton, J. T., G. J. Jenkins, and J. J. Ephraums (Eds.), Climate Change: The IPCC Scientific Assessment, 365 pp., Cambridge University Press, New York, 1990.
- Hovland, M., A. G. Judd, and R. A. Burke Jr., The global flux of methane from shallow submarine sediments, *Chemosphere*, 26, 559-578, 1993.
- Jähne, B., G. Heinz, and W. Dietrich, Measurements of the diffusion coefficients of sparingly soluble gases in water, J. Geophys. Res., 92, 10,767-10,776, 1987.
- Jean-Baptiste, P., S. Belviso, G. Alaux, B. C. Nguyen, and N. Mihalopoulos, ³He and methane in the Gulf of Aden, *Geochim. Cosmochim. Acta*, 54, 111-116, 1990.
- Jones, R. D., Carbon monoxide and methane distribution and consumption in the photic zone of the Sargasso Sea, *Deep-Sea Res., Part A, 38*, 625-635, 1991.
- Jones, R. D., and J. A. Amador, Methane and carbon monoxide production, oxidation, and turnover times in the Caribbean Sea as influenced by the Orinoco River, J. Geophys. Res., 98, 2353-2359, 1993.
- Karl, D. M., and B. D., Tilbrook, Production and transport of

methane in oceanic particulate organic matter, *Nature*, *368*, 732-734, 1994.

- Khalil, M. A. K., and R. A. Rasmussen, Sources, sinks, and seasonal cycles of atmospheric methane, J. Geophys. Res., 88, 5131-5144, 1983.
- Khalil, M. A. K., and R. A. Rasmussen, Atmospheric methane: Recent global trends, *Environ. Sci. Technol.*, 24, 549-553, 1990.
- Kullenberg, G., Physical oceanography, in *The Baltic Sea*, edited by A. Voipio, pp. 135-182, Elsevier Science, New York, 1981.
- Lambert, G., and S. Schmidt, Reevaluation of the oceanic flux of methane: uncertainties and long term variation, *Chemosphere*, 26, 579-589, 1993.
- Lammers, S., Methane in the water column off Sakhalin, paper presented at First Workshop on the Russian-German Cooperation in the Sea of Okhotsk-Kuril Island Arc System, sponsored by P. P. Shirshov Institute of Oceanology, Moscow, 1993.
- Lammers, S., and E. Suess, An improved head-space analysis method for methane in seawater, *Mar. Chem.*, 47, 115-125, 1994.
- Lamontagne, R. A., J. W. Swinnerton, V. J. Linnebom, and W. D. Smith, Methane concentrations in various marine environments, J. Geophys. Res., 78, 5317-5325, 1973.
- Lamontagne, R. A., J. W. Swinnerton, and V. J. Linnenbom, C₁-C₄ hydrocarbons in the North and South Pacific, *Tellus*, 26, 71-77, 1974.
- Lavoie, D. M., D. F. Reid, I. P. DePalma, and J. D. Hayes, Chemical, biological, and physical measurements from the Mediterranean Sea, summer 1980, *Tech. Note 138*, Naval Ocean Res. and Dev. Activity, NSTL Station, Miss., 1982.
- Li, Y.-H., T.-H. Peng, W. S. Broecker, and H. G. Östlund, The average vertical mixing coefficient for the oceanic thermocline, *Tellus, Ser. B.*, 36B, 212-217, 1984.
- Lilley, M. D., J. A. Baross, and L. I. Gordon, Dissolved hydrogen and methane in Saanich Inlet, British Columbia, *Deep- Sea Res.*, *Part A*, 29, 1471-1484, 1982.
- Liss, P. S., and L. Merlivat, Air-sea exchange rates: Introduction and synthesis, in *The Role of Air-Sea Exchange in Geochemical Cycling*, edited by P. Buat-Ménard, pp. 113-127, D. Reidel, New York, 1986.
- Macdonald, R. W., Distribution of low-weight hydrocarbons in the southern Beaufort Sea, *Environ. Sci. Technol.*, 10, 1241-1246, 1976.
- Menard, H. W., and S. M. Smith, Hypsometry of the ocean basin provinces, J. Geophys. Res., 71, 4305-4325, 1966.
- Omstedt, A., Modelling the Baltic Sea as thirteen sub-basins with vertical resolution, *Tellus*, Ser. A, 42A, 286-301, 1990.
- Otto, L., J. T. F. Zimmermann, G. K. F. Furnes, M. Mork, R. Saetre, and G. Becker, Review of the physical oceanography of the North Sea, Neth. J. Sea Res., 26, 161-238, 1990.
- Owens, N. J. P., C. S. Law, R. F. C. Mantoura, P. H. Burkill, and C. A. Llewellyn, Methane flux to the atmosphere from the Arabian Sea, *Nature*, 354, 293-296, 1991.
- Prinn, R., et al., Global average concentration and trend for hydroxyl radicals deduced from ALE/GAGE trichloroethane (methyl chloroform) data for 1978-1990, J. Geophys. Res., 97, 2445-2461, 1992.
- Rasmussen, R. A., and M. A. K. Khalil, Atmospheric methane in the recent and ancient atmospheres: Concentrations, trends, and interhemispheric gradient, J. Geophys. Res., 89, 11,599-11,605, 1984.
- Reeburgh, W. S., B. B. Ward, S. C. Whalen, K. A. Sandbeck, K. A. Kilpatrick, and L. J. Kerkhof, Black Sea methane geochemistry, *Deep- Sea Res., Part A*, 38, S1189-1210, 1991.

Scranton, M. I., and P. G. Brewer, Occurrence of methane in the

near-surface waters of the western subtropical North-Atlantic, *Deep- Sea Res.*, 24, 127-138, 1977.

- Scranton, M. I., and J. W. Farrington, Methane production in the waters of Walvis Bay, J. Geophys. Res., 82, 4947-4952, 1977.
- Scranton, M. I., and K. McShane, Methane fluxes in the southern North Sea: The role of European rivers, *Cont. Shelf Res.*, 11, 37-52, 1991.
- Scranton, M. I., P. Crill, M. de Angelis, P. L. Donaghay, and J. M. Sieburth, The importance of episodic events in contolling the flux of methane from an anoxic basin, *Global Biogeochem. Cycles*, 7, 491-507, 1993.
- Siedler G., and H. Peters, Properties of seawater, in Oceanography, Landolt-Börnstein, New Ser., Group V, vol. 3a, edited by J. Sündermann, pp. 233-264, Springer-Verlag, New York, 1986.
- Steele, L. P., P. J. Fraser, R. A. Rasmussen, M. A. K. Khalil, T. J. Conway, A. J. Crawford, R. H. Gammon, K. A. Masarie, and K. W. Thoning, The global distribution of methane in the troposphere, *J. Atmos. Chem.*, 5, 125-171, 1987.
- Steele, L. P., E. J. Dlugokencky, P. M. Lang, P. P. Tans, R. C. Martin, and K. A. Masarie, Slowing down of the global accumulation of atmospheric methane during the 1980s, *Nature*, 358, 313-316, 1992.
- Swinnerton, J. W., V. J. Linnenbom, and C. H. Cheek, Distribution of methane and carbon monoxide between the atmosphere and natural waters, *Environ. Sci. Technol.*, 9, 836-838, 1969.
- Traganza, E. D., J. W. Swinnerton, and C. H. Cheek, Methane supersaturation and ATP-zooplankton blooms in near-surface waters of the Western Mediterranean and the subtropical North Atlantic Ocean, *Deep- Sea Res., Part A*, 26, 1237-1245, 1979.
- Vaghjiani, G. L., and A. R. Ravishankara, New measurement of the rate coefficient for the reaction of OH with methane, *Nature*, 350, 406-409, 1991.
- Wanninkhof, R., W. Asher, R. Weppering, H. Chen, P. Schlosser, C. Langdon, and R. Sambrotto, Gas transfer experiments on Georges Bank using two volatile deliberate tracers, J. Geophys. Res., 98, 20,237-20,248, 1993.
- Ward, B. B., The subsurface methane maximum in the Southern California Bight, Cont. Shelf Res., 12, 735-752, 1992.
- Ward, B. B., K. A. Kilpatrick, P. C. Novelli, and M. I. Scranton, Methane oxidation and methane fluxes in the ocean surface layer and deep anoxic waters, *Nature*, 327, 226-229, 1987.
- Ward, B. B., K. A. Kilpatrick, A. E. Wopat, E. C. Minnich, and M. E. Lidstrom, Methane oxidation in Saanich Inlet during summer stratification, *Cont. Shelf Res.*, 9, 65-75, 1989.
- Wiesenburg, D. A., and N. L. Guinasso Jr., Equilibrium solubilities of methane, carbon monoxide, hydrogen in water and seawater, J. Chem. Eng. Data, 24, 356-360, 1979.
- Williams, R. T., and A. E. Bainbridge, Dissolved CO, CH₄, and H₂ in the Southern Ocean, J. Geophys. Res., 78, 2691-2694, 1973.
- Wilson, D. F., J. W. Swinnerton, and R. A. Lamontagne, Production of carbon monoxide and gaseous hydrocarbons in seawater: Relation to dissolved organic carbon, *Science*, 168, 1577-1579, 1970.
- Woodwell, G. M., Aquatic systems as part of the biosphere, in *Fundamentals of Aquatic Ecosystems*, edited by R. S. K. Barnes and K. H. Mann, Blackwell Scientific, Boston, Mass., 1980.

M. O. Andreae, H. W. Bange, U. H. Bartell, S. Rapsomanikis, Biogeochemistry Department, Max Planck Institute for Chemistry, P. O. Box 3060, D-55020 Mainz, Germany.

(Received September 9, 1993; revised August 15, 1994; accepted August 19, 1994.)