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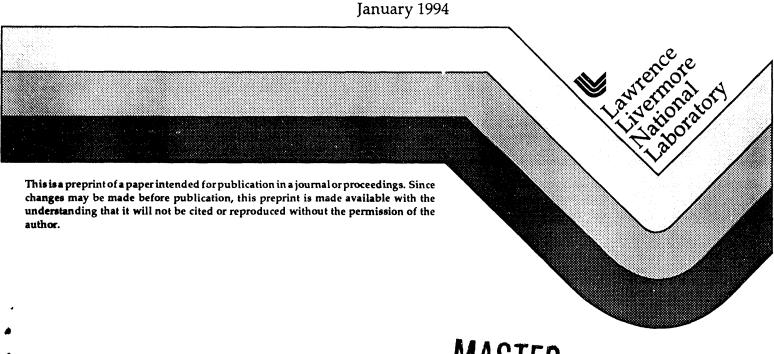
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<u>ABSTRACT</u>

Thin bismuth strontium calcium copper oxide (BSCCO) films and BSCCO/insulator/BSCCO trilayers have been prepared on SrTiO3 and MgO substrates by evaporation from elemental sources in an ozone atmosphere. Accurate control of the stoichiometry is achieved through monitoring of the atomic fluxes by use of *in situ* atomic absorption spectroscopy, as well as by reflection high-energy electron diffraction (RHEED). Nevertheless, nanometer-scale second-phase precipitates are sometimes observed. These defects and the flat regions around them have been probed by a variety of microanalytical techniques, including Rutherford backscattering spectroscopy (RBS), particle-induced x-ray emission (PIXE), atomic force microscopy (AFM), microscopic secondary ion mass spectroscopy and transmission electron microscopy (TEM).

INTRODUCTION

The ideal properties of high temperature superconducting oxide materials make them appropriate for many applications in both high current and electronic applications. These include current-carrying tapes for high-field electromagnets, electronic devices for microwave applications, and tunnel junction devices for sensors and electronics. Applications of films of high temperature superconductors require varying degrees of approximation to ideal characteristics. Current-carrying films require some limitations on texture. Films to make SQUIDs for sensors must be well-oriented but can tolerate significant levels of local defects. However, films for some types of electronic devices must approach the order of perfection of macroscopic single-crystalline structures. Films with such a high level of order can be fabricated by epitaxial deposition of the superconducting material onto an appropriate substrate with strict controls on the deposition rates and environment.

In this paper we report on attempts to better understand the details of the deposition process, the resulting microstructure and device characteristics. We will demonstrate the high level of order of films deposited by high-vacuum evaporation at Varian. Preliminary observations about the most common defects will be discussed. In the Varian Reactive Chemical Epitaxy (RCE) process fine control over the deposition rate is achieved by *in situ* RHEED monitoring of the film surface during deposition[1-4], supplemented by atomic adsorption measurements of the flux.[5] Material is deposited in molecular layers so that atomistic control of the composition of each unit cell is achieved and complex structures can be formed. This procedure results in films of high quality over significant portions of the deposition area. When grown as multilayer composites, these films can be patterned into high-quality Josephson junctions with c-axis transport[6].

Film quality

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Fabrication of superconducting devices requires the deposition of epitaxial heterostructures. Such devices require growth of an insulating layer on a superconducting base and then subsequent growth of superconducting overlayers with good epitaxial matching at the interfaces. A typical epitaxial superlattice structure is Bi₂Sr₂Cu₁O₆ called (2201), and Bi₂Sr₂Ca₁Cu₂O₈, called (2212), layers deposited alternately. Other useful heterostructures are the analogously named (2278) and (0278) layered between (2201) or (2212). (2278), 8 Cu-O layers, and (0278), one Bi-O layer separating the cells, are of special interest as they can be doped to control the transport behavior in a superconductor-insulator-superconductor heterostructure[7,8]. The types of defects that typically occur in these heterostructures include problems with matching at the interfaces between different.

In a single-phase film the crystalline order and layer homogeneity can be investigated using probes such as x-ray diffraction (NRD) or Ruthertord back scattering spectroscopy (RBS) that sample a significant fraction of the film — These techniques give limited information about interfacial quality — Microscopic techniques are required to address interfacial ordering and coherency issues. In order to investigate the level of perfection in these films, we have performed transmission electron microscopy (TEM), scanning tunneling microscopy (STM), and atomic force microscopy (AFM) on films containing layers of (2201), layers of (2212), superlattices of (2201)/(2212) and layers of the large unit cell (2278) all epitaxially grown on SrTiO₃.

TEM images were obtained in a view transverse to the film surface so that each of the separate layers could be individually seen. Plan-view images were obtained on the surface of the film with AFM and STM – Samples for AFM and STM are measured as received. For TEM destructive sample preparation was required. Samples were thinned by mechanical polishing and ion milling to less than 50 nm in thickness and images were collected on a 400 keV, JEOL 400EX transmission electron microscope.

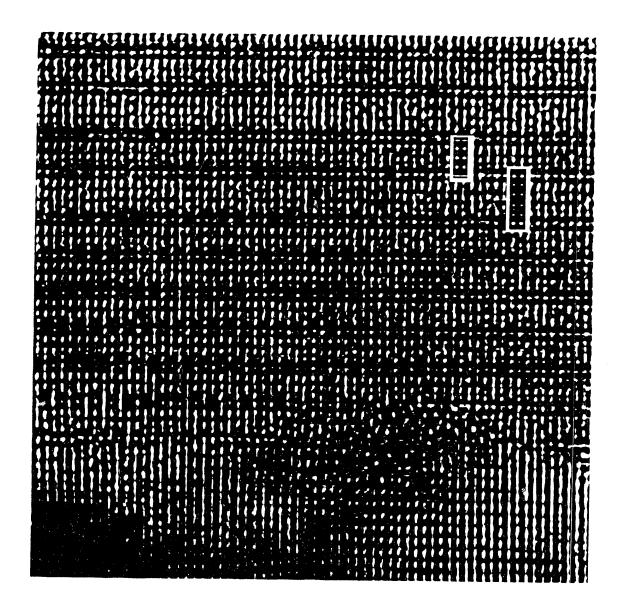


Figure 1

I ransmission electron microscope image of a (2212)/(2201) superlattice grown on the (001) face of a St 1(03 substrate

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Figure 2.

Transmission electron microscope image of a BSCCO heterostructure including a metastable (2278) layer, which contains 8 copper oxide layers. Typical lattice defects which can be identified in this picture include antiphase boundaries, incomplete bismuth oxide layers and a-b twin boundaries. The twin boundary is marked by a dashed line, while the antiphase region is indicated by an arrow.

The ordered epitaxual growth of the (2212)/(2201) superlattice is displayed in figure 1 — The alternation of the Bi-O and Cu-O layers in the unit cell and the variation in the 1.54 nm layer spacing of (2212) and the 1.23 nm layer spacing of (2201) is evident. Areas enclosed by the boxes are calculated TEM images for (2201) and (2212). Several features typical of successful film growth are evident in this picture. The surface roughness of the SrTiO3 substrate is good on a microscopic length scale and the first BSCCO layers have good coherence and epitaxial matching. The epitaxy spans the imperfections in the substrate seen midway in the frame. There is a strong tendency for the BSCCO system to match the exposed substrate lattice and to construct single continuous layers as the film thickness increases. The crystalline perfection of the (2201)/(2212) superlattice demonstrates the delicate control over deposition that can be obtained in the BSCCO system using RCE.

A second, more dramatic example of heteroepitaxy is seen in figure 2. In this case the intended structure is (2212) /(2278)/(2212). This structure is the basis for fabrication of Josephson junctions with novel, controlled properties. The tunneling characteristics of junctions with c-axis transport fabricated from this and closely related trilayer structures have been reported in [6]. Good boundary matching is seen at the interface of the barrier layer and the superconductor, but there are several imperfections visible in the figure 2. Above the barrier layer there is a break in the continuity of the double BiO layer separating two (2212) layers and an anti-phase boundary defect indicated by the arrow. At the break in the Bi-O layers is the abutment of a new set of phases with the layer spacing of (2223) and (3312), i.e. an extra Bi-O layer on one side and an extra Cu-O layer on the other. The anti phase boundary in nearby layers extends over several unit cells for which the continuity of the Bi-O layers is completely interrupted. Below the barrier layer there is a twin boundary separating regions of orientation of the **a** and **b** axes as indicated by the dotted line and within the minority twin region is a second anti-phase boundary.

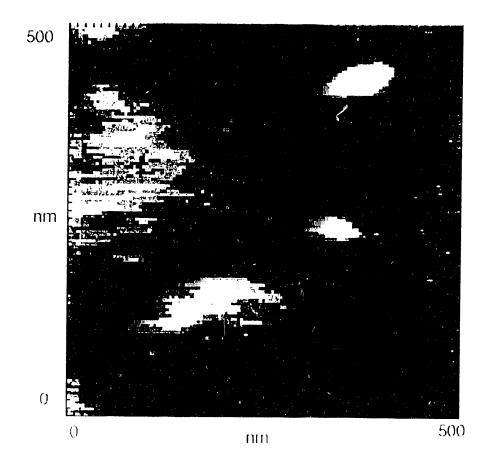


Figure 3

Atomic force microscope image of the defect-free surface of the BSCCO heterostructure pictured in figures 1 and 2. The full vertical range is 7.7 nm -1 lat regions of about 100 nm in diameter and step heights of 1 to 2 nm are seen.

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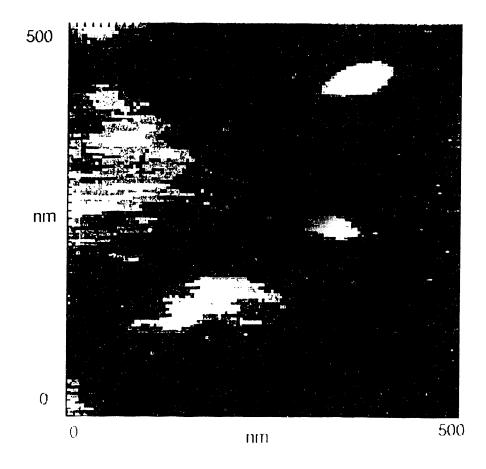


Figure 3.

Atomic force microscope image of the defect-free surface of the BSCCO heterostructure pictured in figures 1 and 2. The full vertical range is 7.7 nm Flat regions of about 100 nm in diameter and step heights of 1 to 2 nm are seen.

These defects are minor deviations from perfect order in this epitaxial thin film. They occur with varying frequency throughout the volume of the films that have so far been studied. The spatial distribution of the interruptions to continuity in the Bi-O planes can be determined from examination of TEM images taken from several parts of the film. The scale of the roughness in the surface of the film can be independently determined by AFM. An AFM image of the smooth surface of the same film as used to obtain the TEM images is seen in figure 3. In this image we see surface roughness with a lateral scale of roughly 100 nm and a vertical scale of 7.7 nm. The roughness is not randomly distributed but appears in steps having a typical step size of about 1.5 nm. This corresponds closely with the c axis layer spacing of 1.54 nm for the terminating Bi(2212) phase. This is a surface with a high degree of smoothness. In some parts of the sample there are surface asperities distributed with a spacing of microns or larger. The roughness variations in the vertical direction in figure 3 are typical of those found in defect-free regions on all of the samples measured and the scale of the in plane variations are consistent with those taken from TEM.

Microscopic roughness in the substrate surface is accommodated by the strong tendency toward long in-plane coherence in BSCCO. However, large-scale substrate roughness can serve to initiate breaks in layers. For instance, in the YBa2Cu3O7 system edge junctions are formed at high steps patterned in the substrates. We have investigated the roughness of randomly selected SrTiO3 substrates. Both optical interferometery and AFM determinations of the substrate surface were performed. In optical interferometry the root-mean-square of the substrate roughness was typically 0.2 nm. In AFM similar roughness was observed. With both techniques the BSCCO film surface was found to be at least as smooth as the substrate in most regions.

A low density of larger size defects can be observed on the surface of nearly all films made by the RCE process. In good films these features only have a significant density at the edges of the substrate water. However, in films deposited under non-optimal conditions high densities of defects have been observed on the surface. The largest of these have micron scale dimensions and can be seen in an optical microscope. A high density of these surface defects is correlated with degraded Josephson junction performance. At present the source and composition of these defects is not well understood. There may be several defect types resulting from separate initiation processes in the film deposition.

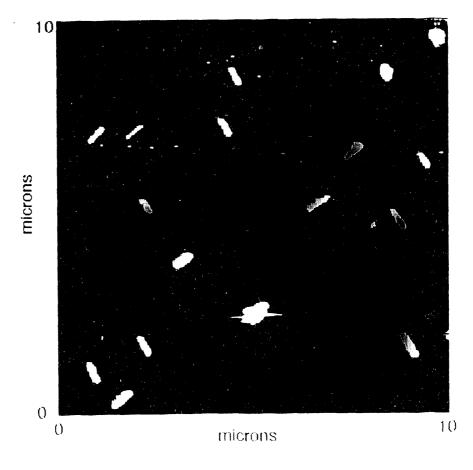


Figure 4.

Atomic force microscopy image of defect -rich corner of a BSCCO superlattice. Large irregular regions and smaller oriented defects both protrude from the smooth sample surface.

In order to understand the morphology of the defects and the conditions that lead to their growth, we have performed both microscopic and macroscopic analysis of the more highly defected films. The size and shape of the defects have been imaged using both AFM and STM. Microscopic images of elemental composition in a defected region have been imaged using an ion microscope, while the macroscopic stoichiometry has been determined using RBS.

Dramatically different defect patterns may sometimes be found in widely separated regions on a single wafer. This suggests that the initiation of defect growth is not random and may be the result of some specific stimulus. A typical AFM image of the corner of a highly defected film is shown in figure 4. In this image we see a series of micron-sized, rectangular asperities aligned with the underlying axes of the (2212) film. There are also some roughly circular asperities seen as well. The height of the rectangular defects is 100 nm and of the circular defects is 20 nm. These features are spaced at separation of about 5 microns. In between the asperities are flat regions similar to those imaged in figure 3. It appears that these defects rise up from an otherwise smooth and well ordered film.

The features seen in the defect distribution in figure 4 are typical of those from all the wafers we have scanned. The lateral scale of the features is typically one micron or less. There have been occasional defects larger than a square micron in area. The variation in height and shape can be more extreme. One film had asperities extending as much as 1000 nm above the film surface along with defects of only a few nm high. While the area of the defects is uniform, the height and shape of the asperities can vary greatly. The regions between defect structures are smooth at the BSCCO unit cell level.

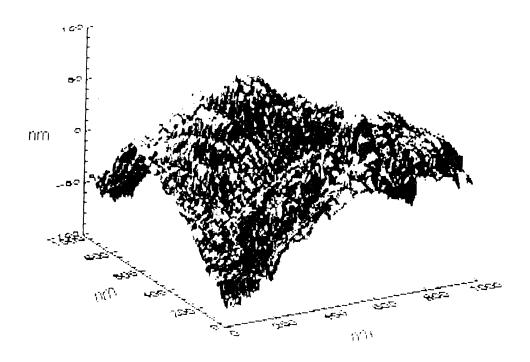


Figure 5

Scanning transmission microscope unage taken on the site of a protrucing defect on the surface of a BSCCO film Oriented corrugations can be seen in these defects Images taken with the AFM only show the shape of the surface features and leave several important questions unresolved For instance, the defects may be precipitate phases segregated from the film in order to preserve the overall stoichiometry. In this case the chemical composition and electrical transport of the defects may be different from the surrounding (2212) film. STM probing of the defects can in principal determine the resistivity of the defects and highlight differences between the defects and the rest of the film. Figure 5 shows an STM image of a defect similar to that imaged by AFM in figure 4. Preliminary I-V characteristics taken in air showed that in general the defects were more conductive than the flat areas of the film. However possible surface contamination prevents more definitive conclusions. The structure of the defect imaged in figure 5 is typical of the asperities which are laterally oriented with respect to the (2212) film. There are corrugations on the surface of the defect suggestive of steps or layers running perpendicular to the film surface and over the top of the defect. This corrugation is not seen in the film surface and is evidence that the defect is not simply a cluster of BSCCO material. The defects pictured in figure 5 appear to have a crystal structure with a strong symmetry plane in the vertical direction. STM results have only been obtained for oriented defects of this type. Other types of defects may show different structure and electrical properties.

To obtain information about the chemical structure and the thickness of the surface asperities, we performed both SIMS and RBS over areas of low and high defect concentration. A Cameca IMS 3f ion microscope was used to obtain the SIMS images. With this ion microscope, images can be formed using the secondary ion flux with lateral spatial resolution as fine as one micron. Depth profiles which track all of the major constituents and several suspected contaminates can be performed on 25-micron-wide areas.

The ion microscope was operated in two data collection modes. First ion fluxes of the expected elemental constituents were obtained in several widely separated spots on the film in order to search for local deviations from the expected stoichiometry. Scratches resulting from handling were clearly seen as regions of low copper and bismuth concentration and high strontium and titanium concentrations. Overlaying particles of foreign material were observed as a low concentration of all of the film constituents and a high concentration of unrelated elements such as sodium. Although these extraneous features could be resolved, no strong deviations from the 2212 stoichiometry were observed in these images. Large local stoichiometry variations would be expected if the asperities were precipitates of a second phase. The second SIMS imaging mode is to scan the probing ion beam in order to obtain a separate depth-dependent image profile for each of the BSCCO constituents. Defects that were significantly thicker than the film or those having a different ion sputtering rate would clearly show up in these scans. In spite of the high defect density in the region chosen for these measurements, there was no statistically significant feature seen that could be interpreted as a locally non-stoichiometric asperity. Therefore, one can conclude that the asperities have a stoichiometry similar to the (2212) film.

The oriented defects are usually confined to one corner of the wafers, which often have low defect concentrations overall. Both microscopic elemental analysis using the ion microscope and macroscopic analysis using RBS were performed at several locations on these wafers, including the high-defect corner and the center region which is nearly defect-free. Significant variations in the elemental concentrations were found in RBS and reproduced in the ion microscope. The strong correspondence of the microscopic and macroscopic measurements suggests that elemental concentrations are varying on a length scale of millimeters rather than microns. In the RBS data shown in table 1, both changes in the overall sample thickness and the locally determined stoichiometry were seen for copper and bismuth Statistically useful results were not obtained for calcium, strontium and oxygen in the film as their peaks were obscured by scattering from the strontium titanate substrate. The 10% variation in the elemental concentrations in this film is large compared to other films with fewer defects. Particle-induced x-ray emission (PIXE) spectra were also taken on these same films for comparison to the RBS and SIMS. This particular sample contained films of (2201) and (2212). The expected elemental concentrations are given in table 1. The measured elemental concentrations were nominally as expected for bismuth and copper-rich by all three techniques. In spite of being copper-rich, large areas of the sample were flat like the area shown in Figure 3.

Position	Defect Type	Cu concentration $(10^{16} \text{ cm}^{-2})$	Bi concentration (10 ¹⁶ cm ⁻²)
Corner	High-density, oriented	6.10	6.12
Center	Low-density, not oriented	6.10	5.53
Off-center	Low-density, not oriented	6.10	5.98
Calculated	None	5.10	5.79

Table 1

Elemental concentrations determined by RBS for a film nominally composed of 10 layers of (2201) and 32 layers of (2212). Data are shown for both defected and defect-free regions.

DISCUSSION

Epitaxial thin films grown by the RCE technique are fabricated with precise control over the layering and composition. **TEM shows that the bismuth-oxygen planes are coherent over distances of 100 nm**. At the interface with the substrate, this long **coherence** length allows the first (2212) layers to conceal the small-scale surface roughness of the substrate. The factors **controlling the lateral coherence** length of the (2212) film have not yet been identified. Two considerations may play a major **role: control of the stoichiometry and the kinetics of the chemical assembly during deposition**. The in-plane and c-axis diffusion **constants are determined by the surface potential that a newly deposited atom sees**. For instance, these potentials control whether growth occurs by cluster formation or layer-by-layer. Some control over the kinetics of growth can be accomplished by **variation of the deposition forms clusters which then reconstruct upon the addition of the Cu overlayer**. This **reconstruction can be avoided** by simultaneous deposition of the Ca and Cu overlayer. The fact that the first few (2212) layers are typically flatter than the substrate surface suggests that the kinetic factors strongly support the formation of continuous, homogenous layers.

The second important factor is control of the stoichiometry during deposition. According to RBS measurements, stoichiometry is presently controlled to within about 2%. However, larger inhomogeneities in composition can appear in the films if the substrate temperature is non-uniform. We have seen large variations on the mm scale of elemental concentrations in films that have a low concentration of micron-scale defects on the film surface. Some of the variation from perfect stoichiometry may alternately be accommodated by the more extended imperfections observed in TEM, such as incomplete layers of the superconducting phases.

Larger asperities are also observed the surfaces of these heterostructures. Some of these asperities rise above the film surface as much as 1 micron, but they are more typically 10 to 100 nm high. Areal dimensions of one square micron are most common, but appear to be an upper bound. The nucleation source of these defects remains undetermined. The defects measured so far have chemical composition similar to the surrounding film. Some of the asperities have corrugated surfaces, with layer

planes apparently growing up perpendicular to the plane of the film. a-axis growth may be the source of these features. Further measurements in progress are underway to confirm this hypothesis.

We can draw some preliminary conclusions from the variety of techniques we have used to characterize the BSCCO films and trilayers. First, control of both the stoichiometry and the kinetic variables of the deposition are important in producing thin films free from large inhomogeneities. The accommodation of excess copper into a smooth (2212) film described suggests that more than a stoichiometric imbalance is required for the growth of large defects. Perhaps there is a competition between the growth of additional BSCCO planes and the nucleation of three-dimensional, non-wetting asperities. The importance of nucleation processes in the growth of large defects is underscored by the high quality of most of the BSCCO material. The substrate smoothness and strong tendency of the system to assemble in islands which encompass small irregularities seems to rule out substrate roughness as a dominant source of nucleation sites. However, substrate features such as scratches or dirt may serve as nucleation sites for the large defects. Several important issues remain, such as whether surface asperities extend deep into the (2212) films, and whether the crystalline orientation of defects varies from the nominal orientation. The formation of imperfections in the BSCCO system is unusual because of the anisotropic crystal structure and the many layered phases which can be formed. The insights gained from the study of BSCCO may help us understand the mechanisms of growth in other anisotropic crystals.

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