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## Microbial Proteases and Application as Laundry Detergent Additive

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**Abstract:** Proteases represents one of the major groups of industrial enzymes and a number of detergent stable proteases have been isolated and characterized because of its widespread use in detergents. It is worthwhile to screen microbes from new habitats for proteases with novel properties to meet the needs of rapidly growing detergent industry. High-alkaline serine proteases have been successfully applied as protein degrading components of detergent formulations and are subject to extensive protein engineering efforts to improve their stability and performance. Protein engineering has been extremely used to study the structure-function relationship in proteases and led to deeper understanding of the factors influencing the cleaning performance of detergent proteases. This study, discusses the types and sources of proteases with an overview on applications of proteases as laundry detergent additives and some advances in improving the stability and performance of detergent enzymes.

**Key words:** Protease classification, enzyme detergent formulation, detergent gels, protein engineering

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## INTRODUCTION

The estimated value of world wide use of industrial enzyme has increased from \$1 billion in 1995 to \$1.5 billion in 2000 (Kirk *et al.*, 2002). As per the forecast, the global demand for enzymes will rise 7% per annum through 2006 to \$6 billion in 2011 (McCoy, 2000). Proteases represent one of the major groups of industrial enzymes, because of their widespread use in detergents and dairy industry and industrial sales of protease are estimated at more than \$350 million annually (Kirk *et al.*, 2002). Proteases account for the 60-65% of the global industrial enzyme market and out of this 25% is constituted by alkaline proteases, 3% by trypsin, 10% by renin and 21% by the other proteases (Bhosale *et al.*, 1995; Rao *et al.*, 1998). Proteases show a vast diversity in their physio-chemical and catalytic properties and lot of literature is available on their biochemical and biotechnological aspects (Rao *et al.*, 1998; Saeki *et al.*, 2007). The proteases of industrial importance are obtained from animals, plants and microorganisms. The proteolytic enzymes hydrolyse the peptide links of proteins and peptides to form smaller subunits of amino acids and are produced both extracellularly as well as intracellularly (Gajju *et al.*, 1996; Kumar *et al.*, 2002). The proteases play an important role in a wide range of industrial processes *viz.*, baking, brewing, detergents, leather processing, pharmaceuticals, meat tenderization, cosmetics, peptide synthesis and medical diagnosis (Bhalla *et al.*, 1999; Kumar and Bhalla, 2005; Kumar *et al.*, 2003; Gupta *et al.*, 2002; Najafi *et al.*, 2005). The application of proteases in different industries is show in Table 1. After the food industry, proteases find its widespread application in laundry detergents (Maurer, 2004; Prakash *et al.*, 2005; Jaswal and Kocher, 2007;

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Table 1: Application of proteases in industry

Industry	Protease	Application
Baking	Neutral protease	Dough conditioner
Beverage	Papain	Chill proofing, removal of haze in beverages
Dairy	Fungal proteases, chymosin, other proteases	Replacement of calf rennet, whey protein processing, production of enzyme modified cheese (EMC)
Detergent	Alkaline protease, subtilisin	Laundry detergents for protein stain removal
Food processing	Several proteases	Modification of protein rich material i.e., soy protein or wheat gluten
Leather	Trypsin, other proteases	Bating of leather, dehairing of skins
Meat and fish	Papain, other proteases	Meat tenderization, recovery of protein from bones and fish waste
Medicine	Trypsin	Dead tissue removal, blood clot dissolution
Photography	Several proteases	Recovery of silver from used X-ray and photographic films
Sweetner	Thermolysin	Reverse hydrolysis in aspartame synthesis

Takimura *et al.*, 2007). The thermostability and activity at high pH are the characteristics that have made proteolytic enzymes an ideal candidate for laundry applications (Gupta *et al.*, 2002). Proteases are present in different brands of detergents for use in home and commercial establishments. It is important to understand the type and characteristics of protease before designing a new detergent enzyme for laundry formulation. This study gives an overview on properties and applications of proteases for laundry detergent formulations and some advances in improving the stability and performance of detergent enzymes.

### TYPES OF MICROBIAL PROTEASE SYSTEMS

Microbial proteases are classified in to various groups, depending on whether they are active under acidic, neutral, or alkaline conditions and on the characteristics of the active site group of the enzyme (Garcia-Carreno, 1991). There are four classes of proteases with six families recognized by the International Union of Biochemistry *viz.*, serine I and II, cysteine (plants), cysteine (animals), aspartic and metalloproteases. The microbial proteases which cleave internal peptide bonds are subdivided into four groups according to their side chain specificity and on the functional group present at the active site (Rao *et al.*, 1998):

- Serine proteases (EC.3.4.21)
- Cysteine proteases (EC.3.4.22)
- Aspartic proteases (EC.3.4.23)
- Metalloproteases (EC.3.4.24)

The schematic division of proteases is given in Fig. 1. Only serine protease finds their application in detergent formulations. On the other hand, thiol proteases (e.g., papain) easily get oxidized by the bleaching agents and metalloproteases (e.g., thermolysin) lose their metal cofactors due to complexing with the water softening agents or hydroxyl ions.

Alkaline serine proteases are the most important group of enzymes exploited commercially. They have a nucleophilic serine residue in their active site and are also distinguished by having aspartate and histidine residues which, along with serine forms the catalytic triad (Maurer, 2004). They are generally active from neutral to alkaline pH, with optimum at pH 7-11. However, high optima with some protease have been reported and show broad substrate specificity and are generally of low molecular mass (18-35 kDA). Various types of serine proteases have been described in great detail earlier by Kalisz (1988), Garcia-Carreno (1991) and Morihara (1974). A number of thermophilic alkaline serine proteases have been reported from *Thermus* sp. Strain TOK3 (Saravani *et al.*, 1989), *Bacillus stearothermophilus* AP-4 (Dhandapani and Vijayaragavan, 1994), *Bacillus* sp. KSM-K16 (Kobayashi *et al.*, 1995), *Thermoactinomyces* sp. E79 (Lee *et al.*, 1996),

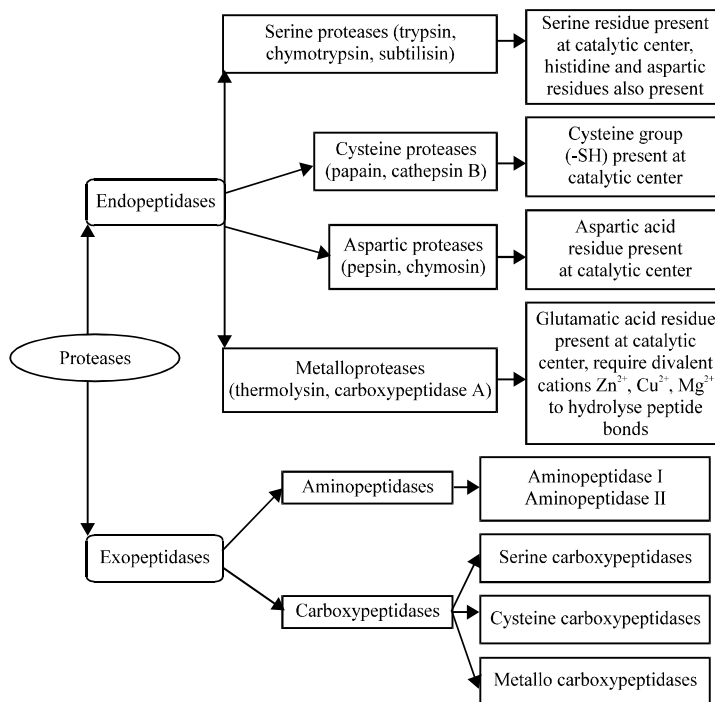


Fig. 1: Classification of proteases

*Bacillus cereus* (Prakash *et al.*, 2005) and *Bacillus circulans* (Jaswal and Kocher, 2007) and used in detergents.

### Subtilisins

Subtilisins are a prototypical group of bacterial serine proteases used extensively in detergents. Subtilisins are defined by their catalytic mechanism as serine proteases. Their amino acid sequence and three dimensional structures can be clearly differentiated from other serine proteases, such as chymotrypsin, carboxypeptidase and peptidase A from *E. coli*. All the subtilisins used in detergents have a size of approximately 27 kDA. Alkaline proteases from *Bacillus amyloliquefaciens* and *Bacillus licheniformis* represents the lead molecules for subtilisin. Subtilisins are used in all types of laundering detergents and used to degrade proteinaceous stains (Ward, 1986). Three subtilisins of detergent importance are (i) Subtilisin Carlsberg produced by *B. licheniformis* (Barfoed, 1981; Geesecke *et al.*, 1991) and was discovered in 1947 by Linderstrom, Lang and Otteson at Carlsberg Laboratory (Rao *et al.*, 1998); (ii) Subtilisin bacterial protease nagase (BPN) from *Bacillus subtilis*, *B. stearothermophilus* and *B. amyloliquefaciens* and (iii) Subtilisin Novo produced from *B. subtilis*.

### Other Alkaline Proteases

In addition to the serine proteases a large number of uncharacterized proteases have been produced by alkaliphilic bacteria with a potential for its application in detergent formulations. Some of these includes *viz*, *Bacillus sphaericus* (Singh *et al.*, 2004), *Bacillus* sp. (Patel *et al.*, 2005), alkaliphilic bacteria (Dodia *et al.*, 2006), *Bacillus brevis* (Aftab *et al.*, 2006), *Bacillus halodurans* (Ibrahahim *et al.*, 2007), *Bacillus licheniformis* N-2 (Nadeem *et al.*, 2007) and *Bacillus* sp. (Chu, 2007).

## PROTEASES AS LAUNDRY DETERGENT ADDITIVE

Enzymes have been added to laundry detergents for over 50 years to facilitate the release of proteinaceous materials in stains such as those of milk and blood. The proteinaceous dirt coagulates on the fabric in the absence of proteinases as a result of washing conditions (Maurer, 2004). The enzyme remove not only the obvious stains, such as blood, but also other obvious materials including proteins from body secretion and food such as milk, egg, meat and fish. The suitability of an enzyme preparation for use in detergents depends on its compatibility with the detergents at a high temperature. An ideal detergent enzyme should be stable and active in the detergent solution and should have adequate temperature stability to be effective in a wide range of washing temperatures (Bhosale *et al.*, 1995).

The proteases are one of the standard ingredients of all kinds of detergents ranging from those used for household laundering to reagents used for cleaning contact lenses or dentures. The preparation of the first enzymatic detergent Brunus dates back to 1913 and consisted of sodium carbonate and a crude pancreatic extract. The first detergent containing the bacterial enzyme was introduced in 1956 under the trade name Bio-40. In 1960, Danish company Novo Industry A/S introduced Alcalase, produced by *Bacillus licheniformis* under commercial name Biotex. This was followed by Maxatase, detergent by Gist-Brocades (Rao *et al.*, 1998). Detergents such as Era Plus<sup>®</sup> (Procter and Gamble), Tide<sup>®</sup> (Colgate Palmolive) and Dynamo<sup>®</sup> (Procter and Gamble) contain proteolytic enzymes, the majority of which are produced by the members of the genus *Bacillus* (Samal *et al.*, 1989). During 1995, high cost of manufacturing and increasing pressure from manufacturers to reduce the raw material cost led to rationalization of detergent enzyme industries. Following this, Genencor International purchased the detergent enzyme business of Gist-Brocades and Solvay and Novo Nordisk acquired Showa-Denko detergent enzyme business. Today, Novozymes and Genencor International are the major suppliers of detergent enzymes, supplying up to 95% of the global market of proteases. All major subtilisins for detergents are produced by *Bacillus*, because these species are able to secrete large amounts of extracellular enzymes (Gupta *et al.*, 2002).

## ENZYME DETERGENT GRANULATION AND FORMULATIONS

The use of enzymes in detergent formulations is now common in developed countries, with over half of all detergents presently available containing enzymes. However, in today's dynamic market, the most successful detergent brands combine proteases with other enzymes to deliver outstanding cleaning performance. In spite of the fact, that the detergent industry is one of the largest market for enzymes, details of the enzymes used and the ways in which they are used, have rarely been published by Graham *et al.* (2001). Other enzymes used in detergents include  $\alpha$ -amylases, cellulases and lipases (Gerhartz, 1990). A number of subtilisin variants are used in detergent worldwide with different trade names (Table 2).

The rapid growth of detergents was temporarily set back in the early 1970s, when workers in detergent factories developed allergies to the enzyme preparations. Enzyme manufacturers solved this problem by developing dust free protease formulations. Today, the enzyme preparations are marketed either as a stabilized enzyme solution or as encapsulated and coated granulates. The liquid preparations normally have a reduced water content and contain significant amounts of 1,2-propanediol. Granulation processes makes the use of extrusion, high shear mixing and fluidized beds (Becker *et al.*, 1997). In developing dust-free granulates (about 0.5 mm in diameter), the enzyme is incorporated into an inner core, containing inorganic salts (e.g., NaCl) and sugars as preservative, bound with reinforcing, fibers of carboxymethylcellulose or similar protective colloid. This core is coated with inert waxy materials

Table 2: Some commercial protease detergents

Trade name	Producer	Microbial strain
Alcalase	Novozymes, Denmark	<i>B. licheniformis</i>
Savinase	Novozymes, Denmark	<i>B. amyloliquefaciens</i>
Esperase	Novozymes, Denmark	<i>B. licheniformis</i> , <i>B. halodurans</i>
Everlase	Novozymes, Denmark	<i>B. clausii</i>
Durazyme	Novozymes, Denmark	<i>Bacillus</i> sp.
Kannase	Novozymes, Denmark	<i>B. clausii</i>
KAP	Kao Corporation, Japan	<i>B. alkalophilus</i>
FNA	Genencor International USA	<i>B. amyloliquefaciens</i>
Purafect	Genencor International USA	<i>B. lentus</i>
Purafect OXP	Genencor International USA	<i>B. lentus</i>
Properase	Genencor International USA	<i>B. alkalophilus</i> PB92
BLAP S	Henkel, Germany	<i>B. lentus</i>
BLAP X	Henkel, Germany	<i>B. lentus</i>
Bioprase	Nagase Biochemicals, Japan	<i>B. subtilis</i>
Godo-Bap	Godo Shusei, Japan	<i>B. licheniformis</i>
Wuxi	Wuxi Synder Bioproducts, China	<i>Bacillus</i> sp.
Protosol	Advance Biochemicals, India	<i>Bacillus</i> sp.

Table 3: Composition of a powder enzyme detergent and liquid detergent

Ingredients	Powder (% wt.)	Liquid (% wt.)
Anionic surfactants	2-10	8-10
Nonionic surfactants	0.5-6	18-20
Soap	1-5	7-12
Sequestrants	30-50	-
Ethylendiaminetetraacetic acid	-	1
Bleach+activator	20-30	-
Triethanolamine	-	7-13
Ethanol	-	4-6
Enzymes	1-5	1-5
Perfume	0.2	0.4
Optical brighteners	0.4-0.8	0.2
Sodium sulfate	To make 100%	-
Water	-	To make 100%

made from paraffin oil or polyethylene glycol plus various hydrophilic binders, which later disperse in the wash. This combination of materials both prevents dust formation and protects the enzymes against damage by other detergent components during the storage (Chaplin and Bucke, 1990). After release from granulated form enzyme must withstand anionic and non-ionic detergents, soaps, oxidants such as sodium perborate which generate hydrogen peroxide, optical brighteners and various less-reactive materials, all at pH values between 8.0 and 10.5 (Table 3).

### DETERGENT GELS

Laundering is a worldwide common need, which has to be satisfied from view point of health, hygiene-care and life style. Cleaning habits are changing day by day with the development of new technologies, new soil substrate systems and diversity of water. The rapid growth and diversity of detergent market requires continuous consideration of new materials to meet the demand of new, improved and modified products. Detergent gels are exciting new type of detergent formulations used for fabric cleaning with advantages *viz*, 100% water solubility, preventing fibers weakening and depolarization, eco-friendly, reduced environmental impacts, easy and small dosing with high cleaning performance (Rachna and Tyagi, 2006). A typical detergent gel consists of all or some of the components shown in Table 4 in addition to the special ingredients designed for specific application.

Table 4: Common constituents of detergent gels

Ingredients	Example
Surfactants	Alkyl sulphate, alkyl ethoxy sulphate
Builders	Sodium & potassium hydrophosphates, silicates, carbonates, sodium aluminosilicates (zeolite), citrates
Optical brightners	Colourless dyes emitting bluish light
Enzymes	Protease, amylase, lipase, cellulase
Bleaches	Sodium hypochlorite
Antimicrobial agents	Alcohol, sodium hypochlorite, iodine, pine oil and phenolic compounds
Fragrances	Citronella and lemon grass oil
Processing aids	Clays, polymers, sodium silicate, sodium sulfate
Preservatives	Butylated hydroxytoluene, EDTA, glutraldehyde
Solvents	Alcohol, acetone (depending on organic compounds)
Acids	Phosphoric acid, hydrochloric acid, citric acid
Alkalis	Sodium hydroxide, sodium metasilicate, sodium carbonate, sodium bicarbonate

Reference (Rachna and Tyagi, 2006)

### SEARCH FOR NOVEL DETERGENT PROTEASES

Until a few years ago new enzymes for the detergent industry were isolated in classical way by screening microorganisms in nature. Industrial strain improvement programs using classical microbiological methods have been carried out over many years and have resulted in the development of several highly productive strains (Maurer, 2004). This practice is still used today for screening but new technologies *viz.* genetic engineering and protein engineering are changing the way new enzymes are developed. Genetic engineering can help to increase production yield and thereby open up new possibilities. Enzymes from genetically engineered microorganisms are already in the market and the number will increase significantly in the future and this development will give both more stable and better performing enzymes in the future. Protein-engineered enzymes entered the market at the beginning of the 1990s and established themselves as benchmarks in several applications (Maurer, 2004). Kobayash *et al.* (1995) have isolated an alkalophilic *Bacillus* sp. KSM-K16 that produces an alkaline protease with properties that fulfill the essential requirements for use in both powder and liquid detergents. The structural gene for a thermostable protease from *B. stearothermophilus* was cloned in a plasmid pTB90 and was expressed in both *B. stearothermophilus* and *B. subtilis*. Recombinant plasmid produced 15 fold increase in protease production compared to the wild type strain (Fujii *et al.*, 1983).

All detergent proteases currently used in the market are serine proteases produced by *Bacillus* strains. *B. licheniformis* is one of the predominant sources of alkaline proteases (Gupta *et al.*, 2002). The fungal alkaline proteases are advantageous due to the ease of downstream processing to prepare a microbe free enzyme. An alkaline protease from *Conidiobolus coronatus* was found to be compatible with the commercial detergents used in India (Phadataré *et al.*, 1993) and retained 43% of its activity at 50°C for 56 min in the presence of Ca<sup>2+</sup> (25 mM) and glycine (1M) (Bhosale *et al.*, 1995). Other detergent proteases have been isolated from *Conidiobolus coronatus* (Bhosale *et al.*, 1995; Phadataré *et al.*, 1993), *B. licheniformis* (Ward, 1983), *B. stearothermophilus* (Dhandapani and Vijayaragavan, 1994; Cowan *et al.*, 1985; Coolbear *et al.*, 1991), *B. thermoruber* (Tomaschova *et al.*, 1998), *Thermoactinomyces* sp. HS682 (Tsuchiya *et al.*, 1992) and *Bacillus* sp. APR-4 (Kumar and Bhalla, 2004a, b).

New interest in properties such as low-temperature performance and the complexity of the patent situation has led to renewed interest in screening for novel enzymes in nature. The development of new detergent compositions will also have an important impact on the development of new proteases and other types of detergent enzymes. The search for new proteases is, of course, not limited to subtilisins,

but is also directed at finding completely new protease backbones. Some interesting molecules have been identified, but none of them has as yet made it into a detergent product (Cherry and Fidantsef, 2003). Every year approximately ten new wild-type subtilisins are now being described in the scientific or patent literature. Interesting enzymes are still isolated by classical microbiological screening methods e.g., the oxidation-stable subtilisin (Saeki *et al.*, 2000, 2002). Graham *et al.* (2001) designed a laundry sheet containing a predetermined amount of a laundry detergent which permits the consumer to simply add the laundry sheet to the washing machine and can optionally include a fabric softener or other chemicals such as brighteners, oxidizing agents and provide an environmentally friendly detergent system to avoid the use of powdered detergents. The stabilization of proteases in liquid preparations is still a field for research (Russell and Britton, 2002). The major problem in aqueous environments is auto-proteolysis. Some general principles in formulating liquid detergents include the reduction of the free water concentration and the use of reversible inhibitors like borate or phenyl-boronic acid derivatives. Also, the composition and nature of the surfactants in the liquid detergent greatly influence the storage stability of the enzyme.

In addition to microbiological screening methods based on the cultivation of protease-producing microorganisms, the exploitation of genome programs and metagenomic screening methods have been established and enlarging the screening pool (Gupta *et al.*, 2002; Lorenz *et al.*, 2001).

### **ENGINEERING OF DETERGENT ENZYMES FOR IMPROVED STABILITY AND CATALYTIC BEHAVIOUR**

Protein engineering allows the introduction of pre-designed changes into the gene for synthesis of protein with an altered function that is desired for its application. Protein engineering techniques, where one or more amino acids in the protein are changed, enable manufacturer to improve enzyme for the detergent industry. Protein engineering creates possibilities for new as well as old enzymes. Most of the industrially important enzymes are now produced from genetically engineered microorganisms. This technique has made possible to selectively delete domains from multifunctional proteins and to examine the loss of the presumed function (Gupta *et al.*, 2002). Site directed mutation of active site amino acids has made it feasible to test their functional importance. This method has already proved its value in the study of pancreatic trypsin and carboxypeptidase and can be extended to many other proteases of industrial importance.

Protein engineering with subtilisins began in 1984 and all the amino acid positions have been modified either by site-directed mutagenesis based on rational design or, by different methods of random mutagenesis. Replacement of Met222 (adjacent to the active Ser221) with amino acid residues that are stable towards hydrogen peroxide has become an example of a rational approach to site-directed mutagenesis (Estell *et al.*, 1985). Dynamics play an important role in determining enzyme activity and specificity and it is therefore of interest to establish how local changes in internal mobility affect protein stability, specificity and performance. The dynamic properties of the 269 residue serine protease subtilisin PB92, secreted by *Bacillus lentus* and an engineered quadruple variant, DSAI, has improved washing performance. The loop regions involved in substrate binding become more structured in the engineered variant as compared with the two native proteases, exhibiting a relation between altered dynamics and performance (Mulder *et al.*, 1999).

Substitutions at nearly every position in the mature 275 amino acid BPN0 subtilisin (*Bacillus* Protease Novo type, subtilisin from *B. amyloliquefaciens*) had been claimed in patents. The BPN0 subtilisin is generally considered to be the lead molecule for subtilisin modifications and mutations in other subtilisins often refer to the homologous position in this lead molecule. There are some excellent general reviews on the protein engineering of subtilisin, as well as articles on more specific detergent applications (Bryan, 2000; Bott, 1997). The three-dimensional structures of engineered variants of



*Bacillus lentus* subtilisin have been engineered to exhibit altered flexibility and increased activity of RSYSA and DSAI were determined by X-ray crystallography. In addition to identifying changes in atomic position a method for identifying protein segments with altered flexibility is reported. This method reveals changes in main-chain mobility in both variants. Residues 125-127 have increased mobility in the RSYSA variant while residues 100-104 have decreased mobility in the DSAI variant. These segments are located at the substrate-binding site and changes in their mobility relate to the observed changes in proteolytic activity and the effects of altered crystal lattice flexibility becomes apparent when these variants were compared with the native enzyme (Graycar *et al.*, 1999).

Gene shuffling is another engineering approach performed with subtilisins. Interesting results from the shuffling of 26 protease genes have been described for properties such as activity in organic solvents, temperature stability and activity at high or low pH (Ness *et al.*, 1999, 2002; Wintrode *et al.*, 2000). A highly flexible region for a psychrophilic enzyme, TA39 subtilisin (S39), was transferred in silico to the mesophilic subtilisin, savinase (EC 3.4.21.62), from *Bacillus lentus (clausii)*. The engineered hybrid and savinase were initially investigated by molecular dynamic simulations at 300K to show binding region and global flexibility. By site-directed modifications, the region was transferred to the binding region of savinase, thus a savinase-S39 hybrid, named H5, was constructed. The designed hybrid showed the same temperature optimum and pH profile as savinase. The H5 hybrid showed increased activity at low temperature, increased binding region and global flexibility, as investigated by molecular dynamic simulations and global destabilization from differential scanning calorimetric measurements (Tindbaek *et al.*, 2004).

Legendre *et al.* (2000) used subtilisin 309 to illustrate the potential of phage display method for the selection of enzymes with desired properties, modifying its substrate specificity with respect to the amino acid at the P4 site of the substrate. As all mutations affecting the specificity of subtilisins may also influence the autoproteolytic processing of the proenzyme to the mature form and the engineering of the proregion or its uncoupling from this biosynthesis step becomes relevant (Takagi and Takahashi, 2003; Almog *et al.*, 1998). Choi *et al.* (1999) have reported cloning of a gene encoding a serine type protease from an extreme thermophilic *Aquifex pyrophilus*, using a sequence tag containing the consensus sequence of protease as a probe. Sequence analysis showed an open reading frame of 619 amino acid residues with three residues forming the catalytic site of this serine protease. Protein protease inhibitors could potentially be used to stabilize proteases in commercial liquid laundry detergents and many inhibitors are susceptible to hydrolysis inflicted by the protease. *Streptomyces* subtilisin inhibitor (SSI) has been engineered to resist proteolysis by adding an inter-chain disulfide bond and removing a subtilisin cleavage site at leucine 63. These stabilizing changes were combined with changes to optimize the affinity for subtilisin, the resulting inhibitor provided complete protease stability for at least 5 months at 31°C in a subtilisin-containing liquid laundry detergent (Gunz *et al.*, 2004). The success of all above approaches has to be evaluated in market for its effective use.

## CONCLUSION

The use of enzymes in detergents can be extended both geographically and numerically. They have not found widespread use in developing countries which are often hot and dusty thus making frequent washing of clothes necessary. The development of subtilisins as typical detergent proteases has employed all the tools of enzyme technology, resulting in a constant flow of new and improved enzymes. A number of protease molecules have been identified and characterized and are in clear opposition to the number of molecules that are entering the market. Liquid detergents have to be formulated around the needs of the enzymes they contain, optimizing ways to stabilize and inhibit them reversibly. Finally, the search for next-generation novel detergent proteases shall be based on new backbones different from subtilisins.

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