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# Mobility of Au and related elements during the hydrothermal alteration of the oceanic crust: implications for the sources of metals in VMS deposits.

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## Abstract

Volcanogenic Massive Sulphide (VMS) deposits are commonly enriched in Cu, Zn and Pb and can also be variably enriched in Au, As, Sb, Se and Te. The behaviour of these elements during hydrothermal alteration of the oceanic crust is not well known. Ocean Drilling Program (ODP) Hole 1256D penetrates a complete in-situ section of the upper oceanic crust providing a unique sample suite to investigate the behaviour of metals during hydrothermal alteration. A representative suite of samples was analysed for Au, As, Sb, Se and Te using low detection limit methods, and a mass balance of metal mobility has been carried out through comparison with a fresh Mid-Oceanic Ridge Basalt (MORB) glass database. The mass balance shows that Au, As, Se, Sb, S, Cu, Zn and Pb are depleted in the sheeted dyke and plutonic complexes with mobilities of  $-46\pm 12\%$ ,  $-27\pm 5\%$ ,  $-2.5\pm 0.5\%$ ,  $-27\pm 6\%$ ,  $-8.4\pm 0.7\%$ ,  $-9.6\pm 1.6\%$ ,  $-7.9\pm 0.5\%$  and  $-44\pm 6\%$  respectively. Arsenic and Sb are enriched in the volcanic section due to seawater-derived fluid

22 circulation. Calculations suggest that large quantities of metal are mobilised from the oceanic  
23 crust but only a small proportion is eventually trapped as VMS mineralisation. The quantity of  
24 Au mobilised and the ratio Au to base metals are similar to that of mafic VMS and a ten times  
25 enrichment of Au would be needed to form a Au-rich VMS. The Cu-rich affinity of mafic VMS  
26 deposits could be explained by base metal fractionation both in the upper sheeted dykes and  
27 during VMS deposit formation.

28 Keywords: VMS deposit, Au-rich VMS, ODP Hole 1256D, hydrothermal alteration in the  
29 oceanic crust

## 30 **Introduction**

31 Volcanogenic Massive Sulphide (VMS) deposits are formed from hydrothermal  
32 mobilisation of metals in the oceanic crust (e.g. Schiffman et al. 1987; Richardson et al. 1987;  
33 Barrie and Hannington 1999). It is generally accepted that the metals enriched in these deposits,  
34 particularly those from ridge-related settings, were partly leached from deeper levels in the  
35 oceanic crust itself during hydrothermal alteration. For example, Ocean Drilling Program (ODP)  
36 drill cores from Holes 1256D and 504B in the Pacific Ocean show that S, Cu and Zn are depleted  
37 from the lower sheeted dykes and from the plutonic complex relative to the upper volcanic  
38 sections of the crust (Alt et al. 1989; Alt 1995; Teagle et al. 2006; Alt et al. 2010). Ophiolitic  
39 exposures of oceanic crust such as at Troodos in Cyprus also reveal large areas of hydrothermally  
40 altered rocks that are depleted in base metals (Richardson et al. 1987; Schiffman et al. 1987;  
41 Schiffman and Smith 1988; Jowitt et al. 2012). The areas of epidosite alteration that characterise  
42 the lower parts of the sheeted dyke complex at Troodos are systematically depleted in Cu, Zn, Ni  
43 and Mn relative to unaltered rocks (Jowitt et al. 2012). These depletions observed in the modern

44 day oceanic crust and in the ophiolitic section are considered to represent the source areas for the  
45 metals enriching the VMS deposits that occur in these regions (e.g. Alt et al. 1989; Jowitt et al.  
46 2012).

47 VMS deposits have high concentrations in the base metals Cu, Zn and Pb (e.g. Galley et  
48 al. 2007; Mudd et al. 2013) but can also be enriched in Au, Ag, As, Sb, Se, Te and Bi. Deposits  
49 enriched in the latter group of elements may be sub-classified depending on the degree of Au  
50 enrichment into auriferous, anomalous and Au-rich VMS deposits (Mercier-Langevin et al.  
51 2011), with the Au-rich deposits defined by a grade of more than 3.46 g/t Au and 31 t Au  
52 (Mercier-Langevin et al. 2011) or by  $\text{Au (ppm)} > \text{Cu+Zn+Pb (wt.\%)}$ ; e.g. Hannington et al. 1999;  
53 Huston 2000). They account for a significant part of global Au production (e.g. 13 % in Canada,  
54 80 % in Sweden; Mercier-Langevin et al. 2011). Despite the economic significance, the complex  
55 and variable mechanisms leading to the enrichment of Au, As, Sb, Se and Te (hereafter referred  
56 to as Au and related elements) are not well understood. The processes proposed to explain the  
57 enrichment in Au and related elements include 1) sub-seafloor boiling in a shallow water  
58 environment which changes the fluid chemistry causing enrichment of Au in a gas-rich fluid  
59 (Urabe et al. 1987; Huston and Large 1989; Butterfield et al. 1990; Poulsen and Hannington  
60 1996; Hannington et al. 1999), 2) formation in regions that contain high source area Au  
61 concentrations such as in back-arc settings (Huston 2000; Moss et al. 2001; Pitcairn 2011) or  
62 mantle plumes (Webber et al. 2013); and 3) input of Au and related elements from a magmatic  
63 source such as shallow sub-seafloor intrusions (e.g. Urabe et al. 1987; Stanton 1990, Sillitoe et al.  
64 1996).

65 Although mobility of base metals during alteration of the oceanic crust is relatively well  
66 documented, very little is known about the effects of hydrothermal alteration on Au and related

67 elements (Keays and Scott 1976; Nesbitt et al. 1987; Korobeynikov and Pertsev 1995). For  
68 example, Nesbitt et al. (1987) reported mobility of Au from the sheeted dykes during  
69 hydrothermal alteration of ODP Hole 504B in the Nazca plate. However, a similar study from  
70 ODP Hole 504B but using a different analytical method showed no such mobility (Korobeynikov  
71 and Pertsev 1995; Korobeynikov and Pertsev 1996). There have been no reports of the behaviour  
72 of As, Sb, Se and Te during hydrothermal alteration in the oceanic crust. Systematic investigation  
73 of the behaviour of Au and related elements during hydrothermal alteration of the oceanic crust  
74 would greatly improve our understanding of the source area processes that generate Au-bearing  
75 fluids in the VMS environment.

76 ODP Hole 1256D in the Cocos Plate, Pacific Ocean is an ideal location to investigate  
77 these processes. The Hole 1256D drill core is unique in being the only core to sample a complete  
78 section of oceanic crust down to the plutonic complex (Teagle et al. 2006; Wilson et al. 2006).  
79 The alteration in Hole 1256D resulting from fluid circulation is well described (Teagle et al.  
80 2006; Alt et al. 2010; Alt and Shanks 2011), providing a solid framework for investigation of the  
81 mobility of Au and related elements. In this study we investigate the behaviour of Au, As, Sb, Se,  
82 Te and S, as well as base metals Cu, Zn and Pb, during the hydrothermal fluid circulation in the  
83 oceanic crust sampled by Hole 1256D. We use low detection limit analytical methods for  
84 quantification of Au and related elements (Pitcairn et al. 2006a; 2006b). The objectives of the  
85 study are to quantify the mobility of Au and related elements and constrain their behaviour during  
86 the alteration of the oceanic crust. The results of the study provide significant insight into source  
87 areas for metals and the trapping efficiency of metals during formation of VMS deposits,  
88 particularly those that are Au-rich.

## 89 **Geological setting**

### 90 1. ODP Hole 1256D lithologic units:

91 ODP Hole 1256D is located in the Cocos Plate (6.736° N, 91.934° W), in a 15 Myr old  
92 crust that was generated at the East Pacific Rise during an episode of superfast spreading (~200  
93 mm/yr; Wilson et al. 2003). Basement oceanic crustal material was recovered during four drilling  
94 cruises: ODP Leg 206 and IODP Expeditions 309/312 and 335 (Wilson et al. 2003; Teagle et al.  
95 2006; Teagle and Harris 2011). Underlying 250 m of pelagic sediment, the oceanic crust at Site  
96 1256 can be divided into four main lithological units: the volcanic section, transitional zone,  
97 sheeted dyke complex and plutonic complex (Figure 1).

98 The volcanic section extends down to 1004 meters below seafloor (mbsf), and comprises an  
99 upper section of lava ponds and inflated flows that were formed during off-axis volcanic events,  
100 and a lower section of phyric to aphyric sheeted flows and generally aphyric massive units  
101 (Wilson et al. 2003; Teagle et al. 2006). Below the volcanic section, the transitional zone extends  
102 from 1004 mbsf down to 1061 mbsf and is mainly composed of aphyric sheeted flows with  
103 sulphide-mineralised breccias at 1028 mbsf (Teagle et al. 2006). The mineralised breccias at 1028  
104 mbsf are composed of angular aphyric cryptocrystalline basaltic clasts cemented by chalcedony,  
105 saponite, carbonate, albite, anhydrite and sulphides (Teagle et al. 2006). The sheeted dyke  
106 complex extends from 1061 mbsf to 1407 mbsf and comprises massive aphyric basalt (Teagle et  
107 al. 2006) with common sub-vertical intrusive contacts. In the lower part, the dykes have  
108 granoblastic textures formed by high-temperature recrystallization due to intrusion of the  
109 underlying gabbros (Teagle et al. 2006; Koepke et al. 2008). At the bottom of the hole two  
110 gabbro bodies intrude the sheeted dykes. The upper gabbro is characterised by gabbros, oxide  
111 gabbros, quartz-rich oxide diorites and small trondhjemite dykelets (Teagle et al. 2006; Koepke et

112 al. 2008). The lower gabbro body consists of medium-grained gabbro-gabbronorite with minor  
113 trondhjemite dykelets. The contacts between the gabbro bodies and the dykes are intrusive with  
114 dyke fragments occurring in the margin of the lower gabbro body (Teagle et al. 2006; Alt et al.  
115 2010).

## 116 2. Hydrothermal system:

117 The Hole 1256D oceanic crust preserves a complex history of hydrothermal fluid flow.  
118 Two main domains of fluid flow have been identified: an upper seawater-derived fluid domain  
119 and a lower hydrothermal fluid domain (Teagle et al. 2006, Alt et al. 2010). The seawater-derived  
120 domain occurs mostly in the volcanic section and the transitional zone and is characterised by  
121 alteration formed from circulation of low-temperature (<50°C-185°C; Fig.1) oxidised fluids (Alt  
122 et al. 2010; Coggon et al. 2010). The most common alteration style is relatively low intensity  
123 background alteration (2-20 % recrystallisation) where the primary mineralogy is partially  
124 replaced with saponite, celadonite, iron oxyhydroxides, chalcedony and minor pyrite, giving the  
125 rocks a more greyish colour (Wilson et al. 2003; Teagle et al. 2006; Alt et al. 2010). The intensity  
126 of alteration is controlled by the permeability which is a function of the lava morphology and the  
127 distribution of breccias and fractures (Teagle et al. 2006). Massive flows act as impermeable  
128 barriers and fluid flow is channelled along their margins (Harris et al. 2015). An example of the  
129 alteration caused by this channelled fluid flow occurs at 648 mbsf where an intensively altered 80  
130 cm interval referred to as the “red brick horizon” is recrystallised to celadonite, saponite, K-  
131 feldspar, chlorite and quartz (Harris et al. 2015). Veins with intensely altered black and brown  
132 alteration halos are common in the volcanic section. The black alteration halos, containing mainly  
133 celadonite, are suggested to have been formed by early low-temperature alteration under anoxic  
134 conditions (Teagle et al. 2006). The brown alteration halos, characterised by iron oxyhydroxides

135 and iron-rich saponite and commonly exhibiting a disseminated pyrite front external to the halo,  
136 are interpreted to have formed during the flow of cold oxidising seawater (Alt et al. 2010).

137         The hydrothermal fluid domain occurs in the sheeted dyke complex and the plutonic rocks  
138 and is characterised by alteration formed from circulation of high-temperature (300°C to >650°C;  
139 Fig. 1) reduced fluids (Alt et al. 2010). The background alteration of the rocks is pervasive and  
140 gives a light to dark green colour (Teagle et al. 2006). Secondary minerals include chlorite,  
141 actinolite, albite, titanite and pyrite, which correspond to sub-greenschist and greenschist facies  
142 conditions (Teagle et al. 2006; Alt et al. 2010). Below 1300 mbsf actinolite is more common than  
143 chlorite, indicating an increase in the temperature of alteration. The intensity of alteration tends to  
144 be higher in the vicinity of veins forming dark green halos. Additionally, veins of quartz, chlorite,  
145 epidote, pyrite, chalcopyrite, magnetite and rare sphalerite overprint the background alteration  
146 and have been interpreted to represent hydrothermal fluid precipitates (Alt et al. 2010). Fluid  
147 flow in the sheeted dyke complex is preferentially channelled along dyke margin (Harris 2015).  
148 Below 1348 mbsf the granoblastic dykes are recrystallised to secondary clinopyroxene,  
149 orthopyroxene, actinolitic hornblende, plagioclase, magnetite and ilmenite (Teagle et al. 2006;  
150 Koepke et al. 2008). Alteration of the gabbro is controlled by grain size, with coarse-grained  
151 rocks tending to be more intensely altered.

152         The upper seawater-derived and lower hydrothermal fluid domains overlap and interact in  
153 the lava-dyke transitional zone unit. Here, alteration assemblages preserve a very steep thermal  
154 step over a short vertical distance from lavas altered at low temperature alteration and rocks  
155 partially recrystallised to greenschist facies assemblages. The transitional zone is a region of  
156 interaction between down-welling seawater-like fluids and up-welling hydrothermal fluids; this  
157 mixing has resulted in the precipitation of anhydrite, pyrite, sphalerite, and minor chalcopyrite



158 veins. A 2.8 m-wide section of quartz-sulphide-mineralised hyaloclastic breccia, where the main  
159 sulphides are pyrite, sphalerite and minor chalcopyrite, occurs in this unit (Teagle et al. 2006; Alt  
160 et al. 2010).

## 161 **Sampling and analytical methods**

162 Rock chips and powdered rock samples from the Hole 1256D drill core were prepared at  
163 the National Oceanographic Centre Southampton (NOCS). A suite of 63 samples distributed  
164 along the whole length of the hole (from 261 mbsf to 1495 mbsf, Fig. 1), representative of the  
165 variation in lithology and the style and intensity of alteration, was selected. Gold analyses were  
166 carried out at Stockholm University using a Thermo XSeries 2 ICP-MS following the ultra-low  
167 detection limit method described in Pitcairn et al (2006a). The  $3\sigma$  method detection limit is 0.033  
168 ppb Au. Analytical precision for Au analyses at Stockholm University was controlled through  
169 analyses of CANMET reference material TDB1 and USGS reference materials WMS-1, CH-4  
170 and BIR-1 (Table 1). Arsenic, Sb, Se and Te analyses were also carried out at Stockholm  
171 University by Hydride Generation-Atomic Fluorescence Spectrometry (HG-AFS) using a PSA  
172 10.055 Millennium Excalibur instrument following the method described in Pitcairn et al.  
173 (2006b). Analyses were carried out on the same acid digests as those used for Au analyses. The  
174  $3\sigma$  instrumental detection limits are 0.043 ppb, 0.079 ppb, 0.038 ppb, and 0.22 ppb for As, Sb, Se  
175 and Te, respectively. Reference materials TDB-1, WMS-1 and CH-4, along with our internal  
176 reference sample BAS 206, were used to control analytical precision and accuracy (Table 1).

177 Sulphur, Cu, Zn, Pb, major and trace elements data are reported in Harris (2011). Samples  
178 were analysed for major and trace elements concentration by X-ray fluorescence spectrometry  
179 (XRF) using Phillips PW2404 automatic X-Ray Spectrometers at the University of Leicester

180 (ODP 206 samples) and at the University of Edinburgh (IODP Expedition 309/312 samples)  
181 following the methods of Harvey (1989) and Fitton et al. (1998), respectively (see Harris, 2011  
182 for details). Sulphur contents were determined using a LECO CS 225 CS-analyser at the  
183 University of Leicester. Lead analyses were carried out by ICP-MS using a Thermo Fisher X-  
184 series Mk II at the University of Southampton. Precision and accuracy were estimated through  
185 analyses of international reference materials of BIR-1, BHVO-1 and BCR-1 for Cu and Zn with  
186 values similar to those reported in Fitton et al. (1998); BAS ECRM 877-1 was used for S and  
187 BIR-1 and BHVO-2 for Pb (Table 1).

## 188 **Whole-rock results**

189 The distribution of Au and related elements in Hole 1256D is shown in Figures 2 and 3,  
190 the distribution of S and base metals in Figure 4, and all data is reported in Appendix 1. The  
191 sample populations have a positively skewed distribution due to a small number of samples with  
192 very high concentrations. These samples are mostly located in the transitional zone and  
193 correspond to sulphide-rich mineralisation formed from the mixing of rising hydrothermal fluids  
194 with cold seawater-derived fluids (Teagle et al. 2006; Alt et al. 2010; Harris et al. 2015). All  
195 elements show the highest concentrations in the transitional zone except for Cu, which has the  
196 highest concentrations in the upper sheeted dyke section. In groups of samples from different  
197 sections of the hole, the positively skewed distribution is highlighted by a discrepancy between  
198 the arithmetic mean and the median for each element. We report the range of values, the  
199 arithmetic mean (referred to below as the mean) and the median for all elements. The standard  
200 deviation is not reported when it is higher than the arithmetic mean. The median concentration of  
201 Au (0.30 ppb) in Hole 1256D is similar to the median Au value of fresh MORB glass estimated

202 by Webber et al. (2013; 0.34 ppb, n=22). Median concentrations of As, Sb, Se, S, Cu, Zn and Pb  
203 (74 ppb, 27 ppb, 200 ppb, 0.11%, 81 ppm, 96 ppm and 0.26 ppm, respectively) are also similar to  
204 that of average MORB (110 ppb, 14 ppb, 200 ppb, 0.11 %, 70 ppm, 80 ppm and 0.26 ppm,  
205 respectively; Arevalo and McDonough 2010). The median value of Te (23 ppb) is higher than the  
206 average fresh MORB value of 5 ppb Te estimated by Yi et al. (2000). In the volcanic section, Au,  
207 As, Sb, Se and S are affected by the variable style of low temperature alteration causing a large  
208 variation in concentrations. Tellurium, Cu, Zn and Pb are less strongly affected by low  
209 temperature alteration and show more homogeneous distributions (Fig 2 and 4). In the sheeted  
210 dyke and plutonic complexes, Au, As, Se, S, Cu show decreasing concentrations with depths  
211 from ~1100 mbsf and downward. Concentrations of Zn and Pb also decrease with depth, but from  
212 slightly deeper levels (~1200 mbsf), and Sb and Te show no systematic variation in content in  
213 these two units. Within the plutonic complex there is no apparent systematic difference in  
214 concentrations between the upper gabbro, the sheeted dyke screen and the lower gabbro for the  
215 elements investigated.

#### 216 Gold:

217 The Au concentrations in Hole 1256D range from 0.05 to 8.2 ppb with mean and median  
218 values of 0.49 ppb and 0.30 ppb respectively (Figs. 2 and 3; Appendix 1). In the volcanic section,  
219 the red brick horizon has the lowest Au concentrations (0.07 to 0.1 ppb) whereas the brown and  
220 black alteration halos and the breccia samples have relatively high Au concentrations (0.7 to 2.1  
221 ppb). Samples affected by the background alteration in the volcanic section have a relatively  
222 heterogeneous distribution with mean and median Au concentrations of  $0.47 \pm 0.45$  ppb and 0.35  
223 ppb respectively. The highest Au concentrations in sulphide mineralised breccia from the  
224 transitional zone are up to 8.2 ppb Au. Samples affected only by background alteration in the

225 sheeted dyke and the plutonic complexes have mean and median concentrations of  $0.20 \pm 0.12$  ppb  
226 and 0.20 ppb respectively, showing homogeneous distribution but decreasing content with depth  
227 (Figs. 2 and 3; Appendix 1).

228 Arsenic:

229 Arsenic concentrations in Hole 1256D range from 17 ppb to 18.2 ppm with mean and  
230 median values of 520 ppb and 74 ppb respectively. Samples affected by the background alteration  
231 in the volcanic section show homogeneous distribution with mean and median As concentrations  
232 of  $114 \pm 66$  ppb and 108 ppb respectively. Samples with specific alteration (e.g. black and brown  
233 halos) have higher As concentrations up to 432 ppm (Figs. 2 and 3; Appendix 1). In the  
234 transitional zone the sulphide-mineralised samples show up to 18.2 ppm As; samples from the  
235 transitional zone that are affected by the background alteration also have high As concentrations  
236 (up to 4.5 ppm). In the sheeted dyke and the plutonic complexes, samples affected only by  
237 background alteration have mean and median As concentrations of  $72 \pm 42$  ppb and 64 ppb  
238 respectively (Fig.2; Appendix 1); in the upper sheeted dykes sulphide-bearing samples in veins or  
239 breccia show the highest concentration in these units.

240 Antimony:

241 The Sb concentrations in Hole 1256D drill core range from 8.2 to 592 ppb with mean and  
242 median values of 45 ppb and 27 ppb, respectively. In the volcanic section, samples affected by  
243 the specific low temperature alterations have higher concentrations than background altered  
244 samples, which have an average concentration of  $29 \pm 12$  ppb (Figs. 2 and 3; Appendix 1).  
245 Antimony concentrations in the sheeted dyke and plutonic complexes show little variation with  
246 an average concentration of  $24 \pm 8$  ppb and no clear variation with depth.

247 Selenium:

248           The Se concentrations in Hole 1256D range from 3.9 ppb to 2.7 ppm with mean and  
249 median values of 315 ppb and 200 ppb, respectively. The red brick horizon at 648 mbsf has very  
250 low Se concentrations (8.6 to 30 ppb), whereas the brown alteration halos are relatively Se-rich  
251 (Figs. 2; Appendix 1). The samples with background alteration in the volcanic section show  
252 homogeneous distribution with mean  $217\pm 69$  ppb Se. Selenium concentrations in the mineralised  
253 breccias and sulphide-rich samples are up to 2.6 ppm. In the sheeted dyke and the plutonic  
254 complexes Se shows a wide concentration range, from 3.9 ppb up to 2.7 ppm, and a marked  
255 decreasing trend with depth. In the upper sheeted dykes high concentration samples are sulphide-  
256 bearing, similar to those of the transitional zone (Fig. 2).

257 Tellurium:

258           Tellurium concentrations in Hole 1256D range from 8.3 to 92 ppb with mean and median  
259 values of  $26\pm 14$  ppb and 23 ppb respectively (Figs. 2 and 3; Appendix 1). Unlike the other  
260 elements, Te has a very homogenous distribution with values varying by only one order of  
261 magnitude. In the volcanic section the background altered samples have a mean Te  
262 concentration of  $25\pm 8.6$  ppb (Figs. 2 and 3). In the sheeted dyke complex the mean Te  
263 concentrations is  $26\pm 14$  ppb, which is similar to the plutonic complex mean of  $24\pm 11$  ppb (Figs.  
264 2 and 3; Appendix 1).

265 Sulphur and base metals:

266           Sulphur and base metal concentrations for the Hole 1256D drill core have been previously  
267 investigated, but using a different sub-set of samples to this study (Alt et al. 2010; Alt and Shanks  
268 2011). Sulphur and base metal concentrations in Hole 1256D are shown in Figure 4. The median  
269 S concentration (0.11 wt.%) is similar to average fresh MORB of 0.11 % (Arevalo and

270 McDonough 2010) but higher than that mean measured by Alt and Shanks (2011) of  $528 \pm 342$   
271 ppm for the Hole 1256D volcanic section. In the volcanic section the S concentrations are  
272 strongly affected by the variable styles of alteration (Fig. 4, Alt and Shanks 2011) but background  
273 altered samples have a mean value of  $0.11 \pm 0.08$  wt.% S. Copper, Zn and Pb show little variation  
274 in the volcanic section, with mean values of  $81.1 \pm 14.3$  ppm,  $98.1 \pm 18.3$  ppm and  $0.37 \pm 0.20$  ppm,  
275 respectively, and are less affected by variable styles of alteration except in the red brick horizon,  
276 where they all show the lowest concentrations. In the sheeted dyke and plutonic complexes the S,  
277 Cu, Zn and Pb show wide ranges of concentrations and have median values of 0.12 wt. %, 76.9  
278 ppm, 99.4 ppm and 0.23 ppm, respectively (Fig. 4; Appendix 1). Similarly to Au, As and Se, the  
279 highest S, Cu, Zn and Pb concentrations are in brecciated or sulphide-bearing samples within  
280 these two units.

## 281 **Discussion**

### 282 Primary metal contents:

283 The distribution of Au and related elements in the different lithological units of the Hole  
284 1256D drill core suggests that they have been variably mobilised by hydrothermal alteration. To  
285 quantify the degree of mobility, the primary concentrations of Au and related elements prior to  
286 hydrothermal fluid circulation must be determined. A suite of nine least-altered samples from the  
287 volcanic section selected on criteria that include the preservation of spherical magmatic  
288 sulphides,  $K_2O$  concentrations  $\leq 0.11$  % and  $^{87}Sr/^{86}Sr$  close to basalt values (Harris et al. 2015),  
289 combined with fresh glass values from Geldmacher et al. (2013), provides an estimate of the  
290 primary metal contents of the Hole 1256D crust (Table 2). However, these values may not be a  
291 true indication of the primary metal contents because 1) these samples are still slightly altered,

292 which may have affected the original metal concentrations, and, 2) the primary metal contents of  
293 the rocks in the Hole 1256D crust may have varied, as it was generated by more than one  
294 magmatic episode and has undergone varying degrees of magmatic fractionation (Teagle et al.  
295 2006; Sano et al. 2011).

296 We assume that the intrusive gabbros in the plutonic section and the near off-axis lava  
297 flows that comprise the upper part of the volcanic section were generated from a melt source with  
298 a generally similar primary composition to that which produced the main volcanic and sheeted  
299 dyke sections of the Hole 1256D crust (Teagle et al. 2006). The main compositional variation in  
300 the Hole1256D oceanic crust is most likely to be due to magmatic fractionation (Teagle et al.  
301 2006; Sano et al. 2011). In order to assess the behaviour of Au and related elements during  
302 fractionation, metal concentrations from a large number of fresh glass samples from Pacific  
303 Ocean spreading ridges are used (Jenner and O'Neill 2012). Samples from the Jenner and O'Neill  
304 (2012) database are selected based on the following criteria: 1) samples are from Pacific Ocean  
305 spreading centres and, 2) samples with anomalously high  $K_2O$  concentrations that have most  
306 likely been affected by seawater alteration are not used. Immobile element concentrations of the  
307 selected samples are indicative of fresh MORB compositions that have not been affected by  
308 alteration (Pearce and Cann, 1973; Appendix 2). The behaviour of Au and related elements  
309 during fractionation can be observed by plotting the element of interest against Y, which is used  
310 as a proxy for magmatic fractionation (Figs. 5 and 6; Jowitt et al., 2012). We have chosen Y  
311 rather than Mg# or Ti because Y is magmatically incompatible, immobile during hydrothermal  
312 alteration (Staudigel 2003), and not influenced by Fe-Ti-oxide saturation during differentiation  
313 (Jowitt et al. 2012). A series of differentiation curves have been plotted using the following  
314 relationship from Jowitt et al. (2012):

315 
$$E_e = AY^B \quad [1]$$

316 where  $E_e$  is the estimated element concentration for a given Y concentration and A and B are  
317 regression coefficients (Table 3). Errors on differentiation curves are calculated using the Root  
318 Mean Square Deviation (RMSD, Table 3). In order to have differentiation curves specific to Hole  
319 1256D samples, the curves determined using Jenner and O'Neill (2012) database are centred on  
320 the primitive compositions estimated in Table 2 (Fig. 5 and 6, Table 3). All elements show strong  
321 trends with Y indicating that their concentrations are affected by magmatic fractionation (Figs. 5  
322 and 6). Due to its strong chalcophile nature, Te is a compatible element during differentiation (Yi  
323 et al. 2000; Patten et al. 2013); however, no database of fresh glass from Pacific Ocean spreading  
324 ridges are known to the authors, preventing the calculation of a differentiation curve for Te.

325 Mass balance:

326 Mass balance calculations are carried out from the differentiation curves using the method  
327 described by Jowitt et al. (2012):

328 
$$\% \text{ Change} = (E_s - E_e) / E_e * 100 \quad [2]$$

329 where  $E_s$  is the measured value from the samples and  $E_e$  the estimated primary crust composition  
330 from equation [1]. Mass change calculations are only relevant if Y concentrations have remained  
331 constant during alteration. Large additions or removal of mass during alteration could cause  
332 residual dilution or concentration of Y. Jowitt et al. (2012) used the mass variation of  $\text{Al}_2\text{O}_3$  as a  
333 proxy for bulk mass changes during alteration of the rock. Both  $\text{Al}_2\text{O}_3$  and Y are immobile during  
334 alteration, but they behave differently during magmatic differentiation as Y is incompatible  
335 whereas  $\text{Al}_2\text{O}_3$  is compatible. A discrepancy between the measured and estimated  $\text{Al}_2\text{O}_3$   
336 concentrations will therefore indicate whether there have been mass changes during alteration



337 (Jowitt et al. 2012). The estimated Al<sub>2</sub>O<sub>3</sub> concentration in Hole 1256D for a given Y value is  
338 estimated using the following equation:

$$339 \quad \text{Al}_2\text{O}_3(\text{wt. \%}) = 29.298 * Y^{-0.214} \quad [3]$$

340 determined using the Jenner and O'Neill (2012) database and the primary Al<sub>2</sub>O<sub>3</sub> concentration  
341 (13.8 wt.%) from the least altered samples (Table 2). The average mass variation of Al<sub>2</sub>O<sub>3</sub> in  
342 Hole 1256D samples calculated from equation [3] is -2.46 % (i.e. -0.34 wt.% Al<sub>2</sub>O<sub>3</sub>), which is  
343 similar to the RMSD of 2.31 % indicating that relatively little mass change affected the samples.  
344 The Y concentrations in Hole 1256D can thus be considered to have remained constant in the  
345 rock during alteration and to be a suitable proxy to determine primary crust composition through  
346 magmatic differentiation.

347 As no differentiation curve could be calculated for Te, the mass variation is carried out  
348 using the method described by Nesbitt (1979):

$$349 \quad \text{Te \% Change} = \left( \frac{\text{Te}_{\text{sample}}}{\text{Y}_{\text{sample}}} * \frac{\text{Y}_{\text{primary}}}{\text{Te}_{\text{primary}}} - 1 \right) * 100 \quad [4]$$

350 where Te<sub>sample</sub> and Y<sub>sample</sub> are the concentrations of Te and Y in the sample, respectively,  
351 and Te<sub>primary</sub> and Y<sub>primary</sub> are the concentrations of the primary crust estimated in Table 2. This  
352 mass calculation, assuming that Te concentration is constant in the primary crust, is an  
353 oversimplification. All elemental mass changes calculated are shown in Figures 7 and 8 and listed  
354 in Table 4 for each unit. The error propagation for the mass balance calculations is described in  
355 detail in Appendix 3. The main errors are caused by the heterogeneous distribution of the  
356 chemical elements in the rocks, which can be estimated from the relative standard deviation of  
357 preferred reference materials (Table 1) and the normalised RMSD (NRMSD) of the Jenner and  
358 O'Neill (2012) database (Table 3). Errors on mass balance calculations vary from 8.5 % for S up

359 to 25.0 % for Au. Tellurium has high error (44.6 %) due to the poor constraint of protolith  
360 concentrations.

361 Metal mobility in deep hydrothermal alteration zones:

362 In the sheeted dyke and plutonic complexes, where hydrothermal fluids are hot (>300 °C;  
363 Alt et al. 2010), almost all of the investigated elements have significantly lower concentrations  
364 than primary crustal compositions, implying that they have been mobilised during hydrothermal  
365 alteration (Figs. 7 and 8). In the sheeted dyke complex, sulphide-rich and brecciated samples,  
366 present mostly in the upper sheeted dyke section represent local metal precipitation from the  
367 hydrothermal fluids and do not represent the background altered samples. Metal depletions are  
368 therefore calculated only from median values of background altered samples (Table 4). Although  
369 few sulphide-rich samples are present in the plutonic complex, they are interpreted to be  
370 associated with hydrothermally altered rocks and the median values of all samples are used for  
371 metal depletion calculations. Gold and Pb are the most depleted elements in the sheeted dyke and  
372 plutonic complexes ( $-46\pm 11.5\%$  and  $-44.4\pm 6.6\%$ , respectively) followed by As and Se ( $-$   
373  $27.2\pm 5.4\%$  and  $-26.7\pm 5.7\%$ , respectively) and Cu, S, Zn and Sb ( $-9.6\pm 1.9\%$ ,  $-8.4\pm 0.9\%$ ,  $-$   
374  $7.9\pm 0.9$  and  $-2.5\pm 0.5\%$ , respectively); Te is the only element showing enrichment ( $13.7\pm 5.7\%$ ;  
375 Table 5). This variation in depletion implies different solubility in the hydrothermal fluids in the  
376 following order: Au~Pb>As~Se>S~Cu~Zn>Sb>Te, with Au being the most soluble and Te the  
377 least. The Te enrichment could be an artefact of the mass balance calculation as an appropriate  
378 differentiation curve could not be determined. The mass balance calculation for S is an  
379 underestimate of the S mobilised by hydrothermal fluid circulation, as the whole rock S data  
380 comprise S from anhydrite and from pyrite that formed by seawater sulphate reduction.

381 The range of Au, As and Sb concentrations in the sheeted dyke and plutonic complexes is  
382 considerably lower than in the volcanic section, indicating that the deep high-temperature  
383 hydrothermal alteration in the lower crust effectively leaches the rocks of these elements,  
384 resulting in a more homogeneous distribution (Fig. 3). The depletion of Au is in the same order of  
385 magnitude as that from Hole 504B, where a ~60 % loss of Au from greenschist altered rocks is  
386 estimated (Nesbitt et al., 1987). Samples with high sulphide content either in the form of veins,  
387 breccias or disseminated grains are enriched in all elements relative to the background altered  
388 samples (Table 4), suggesting that sulphides play an important role on their distribution. Overall,  
389 Au, As, Se, S, Cu, and Sb (and to a lesser extent Pb) show similar depletion patterns through the  
390 sheeted dyke and the plutonic complexes being mobilised from below ~1100-1150 mbsf. Zinc,  
391 however, shows a strikingly different depletion pattern. Although Zn is depleted overall in the  
392 two units, it shows enrichment in the upper sheeted dykes and depletion in the lower sheeted  
393 dykes and the plutonic complex. In the upper sheeted dykes both background altered and  
394 sulphide-rich samples are enriched in Zn, implying that silicate phases and/or oxides also host Zn  
395 (Doe, 1994). Sphalerite is a relatively common sulphide mineral in the upper sheeted dyke veins  
396 but the data indicate that silicates such as chlorite, which is the most abundant secondary mineral  
397 in the sheeted dykes, also host Zn. This is in agreement with previous work from the Troodos  
398 ophiolite, which shows that in rocks altered by hydrothermal fluids, Zn can be hosted by Mg-  
399 bearing minerals such as chlorite (Jowitt et al. 2012). Lead shows a generally similar depletion  
400 pattern to Au, As, Se, S, Cu and Sb in the sheeted dyke and plutonic complexes, but it also shows  
401 some enrichment in the upper sheeted dyke section (Fig. 8), suggesting some similarity in  
402 behaviour with Zn. The enrichment of Zn in the upper sheeted dyke section implies that metal  
403 fractionation occurs during metal transportation from the deep crust towards the seafloor, which  
404 will affect the metal composition of the hydrothermal fluids from which VMS deposits are

405 generated. Mafic VMS deposits commonly have a Cu/Zn ratio of around 1.8 (e.g. Barrie and  
406 Hannington 1999), which is different from the ratio in fresh MORB source rock (fresh MORB  
407 Cu/Zn=0.9; Arevalo and McDonough 2010). This may be partly due to preferential trapping of  
408 Zn relative to Cu in the upper sheeted dyke section during hydrothermal fluid circulation.

409 Metal mobility in shallow seawater-dominated alteration zones:

410 Alteration in the volcanic section is controlled by circulation of low temperature  
411 seawater-derived fluids that cause both the background alteration and the brown and black halos.  
412 Alteration halos, veins and breccias make up ~6.5 % of the Hole 1256D drill core (Alt et al.  
413 2010) and background altered samples the remaining 93.5 %. In the samples showing background  
414 alteration, As, Sb and Se are enriched relative to the protolith concentrations (20 %, 19.9 % and  
415 13.8 %, respectively) whereas S is depleted (-10.5 %); the other elements do not show significant  
416 variation (<±10 %, Table 4). Alt and Shanks (2011) suggested that S loss from the volcanic  
417 section was due to degassing at eruption. Alteration halos and brecciated samples show strong  
418 enrichment in Au, As and Sb and depletion in Pb (Table 4). Taking into account the relative  
419 proportions of background alteration and vein halos, the mass variation in the volcanic section as  
420 a whole shows that As, Sb and Se are enriched (30.4 %, 24.5 % and 12.7 %, respectively),  
421 whereas other elements do not show significant variation (<±10 %, Table 4).

422 In the volcanic section, As and Sb show some correlation with K<sub>2</sub>O and Rb (Fig. 9),  
423 indicating that they were added by seawater-derived fluid circulation. Specific alteration such as  
424 brown and black alteration halos are strongly enriched in As and Sb (Fig. 9, Table 4), implying  
425 that these elements are easily added in rocks that are more intensely altered by seawater-derived  
426 fluids. Mass variations in the red brick horizon are not calculated, but the elemental  
427 concentrations suggest that Au, Se, S, Cu, Zn and Pb were removed whereas Sb and most likely

428 As were added (Fig. 2 and 4). The red brick horizon is suggested to be a typical example of a  
429 pathway for channelled low temperature (~60 °C, Alt et al. 2010) hydrothermal fluids in the  
430 volcanic section (e.g. Harris et al. 2015). The low concentrations of Au, Se, S, Cu, Zn and Pb in  
431 this horizon suggest that there was little precipitation of metals from the hydrothermal fluids  
432 during their circulation through the volcanic section.

433         The black and brown alteration halos are relatively scarce in the Hole 1256D crust and  
434 therefore do not contribute greatly to the mass balance. Seawater-derived fluid alteration is  
435 considered to be limited in Hole 1256D crust primarily due to fast sediment blanketing and the  
436 presence of a massive lava flow in the upper volcanic section, which sealed the oceanic crust  
437 from late seawater infiltration (Alt et al, 2010; Harris et al. 2015). Brown and black alteration  
438 halos are much more common in Hole 504B (Wilson et al. 2003) and it is likely that the volcanic  
439 section there is more strongly enriched in As and also in Sb. Elements such as As and Sb are  
440 amongst those shown to be released from subducting oceanic crust into the overlying mantle  
441 wedge during subduction (Hattori et al. 2005; Hattori and Guillot 2003). Sediments overlying the  
442 subducting slab may provide the bulk of the As and Sb released in the mantle wedge element but  
443 the volcanic section could be an important alternative source.

444         Gold is enriched in brecciated, brown altered and black altered samples but does not  
445 correlate with K<sub>2</sub>O or Rb, suggesting that Au is not enriched by seawater-derived fluid circulation  
446 in the volcanic section. Nesbitt et al. (1987) showed that in Hole 504B volcanic section Au was  
447 not enriched but that redistribution did occur.

448 Metal behaviour in the transitional zone:

449           The transitional zone, where the seawater-derived fluid and hydrothermal fluid domains  
450 meet in Hole 1256D, is characterised by zones of sulphide mineralisation (Teagle et al. 2006).  
451 Relative to the primary crust, the sulphide-mineralised samples are strongly enriched in all the  
452 elements investigated in this study, with the exception of Cu (Figs. 7 and 8 and Table 4). Whereas  
453 Hole 504B hosts a quartz-epidote-pyrite-chalcopyrite mineralised stockwork (e.g. Alt et al.,  
454 1996), the Hole 1256D mineralized breccia at 1028 mbsf appears to be a relatively low  
455 temperature feature (Teagle et al. 2006; Harris et al. 2015), which may have prevented extensive  
456 Cu precipitation. Copper mineralisation in Hole 1256D occurs at deeper levels in the sheeted  
457 dykes (1090-1115 mbsf) as chalcopyrite-rich veinlets and breccias (e.g. Harris et al. 2015). In the  
458 transitional zone, some of the elements, including As, Sb, Se and Pb, are also enriched in the  
459 samples with background alteration (Figs. 7 and 8, Table 4), indicating that these elements might  
460 also be incorporated into silicate or oxide minerals. The source of the metals locally enriched in  
461 the transitional zone is interpreted to be the depletion zones observed in the underlying sheeted  
462 dyke complex and plutonic complex. It is suggested that the mass of metals precipitated in the  
463 transitional zone from the hydrothermal fluids represents only a fraction of the total quantity  
464 present in the fluids; the majority of the metals mobilised are believed to be transported to the  
465 seafloor (Nesbitt et al. 1987; Hannington et al. 1990; Alt et al. 2010). In other localities, however,  
466 the transitional zone represents an important metal sink for the metals transported by  
467 hydrothermal fluid (Coogan and Dosso 2012; Hannington 2013).

468 Implications for VMS formation

469           Mid-ocean ridge (MOR) VMS deposits are classified as “mafic deposits” (Barrie and  
470 Hannington 1999). They have been reported to occur along all mid-ocean spreading ridges so far

471 investigated, including the East Pacific Rise, the Mid-Atlantic Ridge (e.g. TAG hydrothermal  
472 field) and the Juan de Fuca Ridge (Herzig and Hannington 1995; Beaulieu 2010). They are  
473 typically smaller in size than other classes of VMS deposits, ranging from  $\leq 1$  to 5 Mt (Herzig and  
474 Hannington 1995) with an average size of 2.8 Mt (Barrie and Hannington 1999). They occur at  
475 water depths of  $>1600$  m where boiling is prevented by the hydrostatic pressure, and comprise a  
476 stockwork zone overlain by zoned massive and semi-massive sulphides (Hannington et al. 1998).  
477 The deposits are on average Cu-rich and Pb-Zn-poor compared to other VMS types (e.g. Barrie  
478 and Hannington 1999; Table 6). Barrie and Hannington (1999) reported an average mafic VMS  
479 composition of 2 % Cu, 1.1 % Zn, 0.1 % Pb and 1.7 ppm Au.

480         The mass balance reported above indicates that significant amounts of Au, As, Sb, Se, S,  
481 Cu, Zn and Pb are mobilised from the sheeted dyke and plutonic complexes to the seafloor. Some  
482 of these metals are trapped in the mineralisation at the transitional zone or in the upper sheeted  
483 dyke section but the majority were most likely transported towards the seafloor. The quantity of  
484 metal mobilised from the oceanic crust at ODP 1256D is compared with two different modern-  
485 day seafloor VMS systems: the TAG active mound from the Mid-Atlantic Ridge (MAR) and the  
486 axial graben VMS deposits at  $12^{\circ}50'N$  on the East Pacific Rise (EPR). The TAG active mound is  
487 probably the most extensively investigated modern-day seafloor VMS deposit and is  
488 characteristic of a slow-spreading ridge system (e.g., Humphris et al. 1995; Teagle et al. 1998).  
489 The axial graben VMS deposits at  $12^{\circ}50'N$  EPR are a less extensively investigated VMS system  
490 but are characteristic of a fast-spreading ridge system (e.g. Hekinian and Fouquet 1985) which is  
491 relevant to the tectonic settings of OPD Hole 1256D.

492         The main ore zone from the 3.8 Mt TAG active mound contains 42.1 % S as sulphide  
493 (Hannington et al. 1998) and hosts approximately 1.9 t Au, 16.3 t As, 7.6 t Sb, 53.2 t Se, 79.8 Kt

494 Cu, 22.8 Kt Zn and 274 t Pb (Table 6). To form a hypothetical VMS deposit of similar size and  
495 composition from the metals mobilised in the Hole 1256D crust, the sizes of reaction zones  
496 required to supply the Au, As, Sb, Se, Cu, Zn and Pb would be 3.3 km<sup>3</sup>, 2.4 km<sup>3</sup>, 4.1 km<sup>3</sup>, 0.4  
497 km<sup>3</sup>, 3.0 km<sup>3</sup>, 0.9 km<sup>3</sup> and 0.6 km<sup>3</sup> respectively (Fig. 10). Sulphide minerals at the TAG active  
498 mound have a <sup>34</sup>S/<sup>32</sup>S ratio of 6-8 ‰ implying that part of the precipitated S is from hydrothermal  
499 fluid dominated by MORB S and the other is from reduced seawater sulphate (Gemmell and  
500 Sharpe 1998). Assuming that two thirds of the S has a MORB origin, the reaction zone required  
501 to supply the S to the TAG active mound would be 2.4 km<sup>3</sup>. These source area size calculations  
502 are based on the assumptions that 1) there was no major magmatic input of metals in the  
503 hydrothermal fluids, and 2) all of the metals mobilised at depth were transported to the site of  
504 deposition. The TAG hydrothermal field formed from many episodes of focused hydrothermal  
505 activity over a long period of time (e.g. 20 000 years, Peterson et al. 2000) fed by deep faults in a  
506 slow-spreading environment (e.g. McCaig et al., 2007). In contrast, the intense magmatic activity  
507 at fast-spreading ridges generally prevents long-lived hydrothermal activity, resulting in the  
508 formation of numerous but low tonnage VMS deposits (<3000 t, Hannington et al. 2011).

509         At the East Pacific Rise (EPR) near 12°50'N, Hekinian and Fouquet (1985) reported the  
510 presence of 80 small VMS deposits within the ridge axial graben occurring as conical shaped  
511 edifices averaging 20 000 tons (Table 6) and characteristic of fast-spreading ridge VMS deposits.  
512 Assuming hypothetical axial graben VMS deposits similar to the ones occurring at 12°50'N EPR  
513 (0.02 Mt at 34.3 % S, Table 6), the source areas required to mobilise metals from the oceanic  
514 crust at Hole 1256D would, for Au, As, Se, Cu, Zn and Pb, be 0.002 km<sup>3</sup>, 0.028 km<sup>3</sup>, 0.006 km<sup>3</sup>,  
515 0.017 km<sup>3</sup>, 0.002 km<sup>3</sup> and 0.008 km<sup>3</sup>, respectively (Fig. 10). Assuming a similar behaviour for S  
516 as in the TAG active mound, the reaction zone for S would be 0.016 km<sup>3</sup>. These reaction zones



517 are two to three orders of magnitude smaller than the ones required to form a TAG-style deposit,  
518 indicating that a relatively small volume of source rock is required to mobilise sufficient metals  
519 to form fast-spreading ridge VMS deposits. These differences in element behaviour during VMS  
520 formation suggest that although source rock composition controls the hydrothermal fluid  
521 composition and thus the type of VMS deposit generated (i.e. alteration of MORB form mafic  
522 type VMS; e.g. Franklin et al. 1981), the processes that affect the fraction of metals available for  
523 VMS formation, such as metal loss during transport, focus discharge, precipitation efficiency and  
524 zone refining, exert key control on the actual bulk metal composition of the VMS deposit.

525         The sizes of the reaction zones required to form VMS deposits are most likely highly  
526 variable as they are dependent on multiple parameters including the depth, shape and temperature  
527 of the magmatic heat source (Morton and Sleep 1985; Barrie and Hannington 1999; Teagle et al.  
528 2003), the porosity of the crust (Barrie and Hannington 1999), the geometry of the circulation cell  
529 (e.g. Johnson et al. 2010) and the presence of major faults (McCaig et al. 2007). Ohmoto (1996)  
530 estimated a hydrothermal cell size of 40 km<sup>3</sup> to generate Kuroko-type VMS deposits but these  
531 deposits form in an arc environment where hydrothermal fluid circulation is most likely enhanced  
532 by abundant shallow level intrusions. Using seismic survey data, Tolstoy et al. (2008) identified  
533 hydrothermal circulation cell size of ~5 km<sup>3</sup> at near 9°50'N on the fast spreading EPR (500 m  
534 width, 1 km depth and ~10 km length along the ridge axis). A 5 km<sup>3</sup> reaction cell in the Hole  
535 1256D crust (3.9 km<sup>3</sup> of sheeted dyke complex and 1.1 km<sup>3</sup> of plutonic complex) would mobilise  
536 a total of 2.9 t Au, 339 t As, 9 t Sb, 741 t Se, 1.45 Mt S, 0.13 Mt Cu, 0.11 Mt Zn and 2270 t Pb  
537 (Table 5).

538         The quantity of metals mobilised from a 5 km<sup>3</sup> reaction cell in Hole 1256D would be  
539 enough to form at least ~180 axial graben VMS deposits similar to the ones occurring at 12°50'N

540 EPR. However, only a small proportion of the mobilised metals are trapped as VMS deposits.  
541 The metal budget from the source area to the deposit can be affected by many different processes.  
542 For example, metal loss can occur either during migration of hydrothermal fluids from the source  
543 areas toward the seafloor (e.g. Hannington 2013), by unfocused discharge of the hydrothermal  
544 fluids on the seafloor, by inefficient metal trapping mechanisms during precipitation or by  
545 extensive zone refining of VMS deposits (e.g. Petersen et al. 2000). The addition of metals to the  
546 hydrothermal fluids can also occur, such as by addition of metal-rich magmatic fluids (e.g.  
547 Stanton 1990, Ohmoto 1996). Calculation of the percentages of trapped metal required to form  
548 the TAG active mound and the axial graben VMS deposits at 12°50'N EPR from a 5 km<sup>3</sup>  
549 reaction cell, as well as the percentage of lost metals, are shown in Figure 11. Significant  
550 differences in trapping efficiencies for the different metals in both tectonic settings can be  
551 observed (over one order of magnitude maximum). Although Cu and Zn are mobilised in similar  
552 quantities from the lower crust (-9.6 % and -7.8 % respectively, Table 5), it appears that Cu is  
553 preferentially trapped during VMS formation probably due to higher precipitation efficiency or to  
554 preferential fractionation during zone refining. This suggests that precipitation efficiency and/or  
555 zone refining might have an important effect on the Cu-rich affinity of the mafic type VMS.

556         There is a striking difference in metal trapping mechanism efficiencies during VMS  
557 formation between the TAG active mound and the axial graben VMS deposits at 12°50'N EPR.  
558 In the latter, which are representative of fast-spreading ridges, the trapping efficiencies are less  
559 than 1 %, suggesting that of the significant quantity of metals mobilised, only a small portion is  
560 trapped as small VMS deposits with the majority being lost. In the TAG active mound the  
561 trapping efficiencies from Figure 11 are relatively high (from 7 % to 82 %). However, the greater  
562 longevity of slow-spreading ridge hydrothermal cell systems (e.g. 20 000 years; Petersen et al.

563 2000) most likely indicates more extensive and efficient metal mobilisation from the deeper  
564 crust. The trapping efficiencies estimated for the TAG active mound are therefore likely to be an  
565 overestimate.

566 During VMS formation mobilised metals that are not trapped as VMS mineralisation are  
567 vented into seawater. Vented metals can either form particles in plume and sediment as ochres,  
568 umbers and metalliferous sediments, or dissolve in seawater (e.g. James and Elderfield 1996). No  
569 VMS deposit or fossil hydrothermal vent has been observed in the vicinity of Hole 1256D,  
570 although some seamounts are present in the vicinity (Teagle et al. 2006). This suggests that the  
571 metals were likely either to be trapped in some form of deposit hidden under the blanketing  
572 sediment, or lost by diffuse unfocussed discharge. The main implication for VMS formation from  
573 the calculation of the metals mobilised from the oceanic crust at ODP Hole 1256D is that the  
574 mobilisation of large masses of metals from alteration zones at deeper levels in the oceanic crust  
575 does not always lead to the formation of a VMS deposit, and that metal trapping processes are  
576 perhaps more important parameters in the formation of VMS deposit.

#### 577 Implications for Au-rich VMS formation

578 Gold-rich VMS deposits can be defined by either a Au grade  $>3.46$  ppm and a Au  
579 tonnage  $\geq 31$  t (Mercier-Langevin et al. 2011) or by a Au to base metals ratio over unity (Au in  
580 ppm and base metals in per cent; e.g. Poulsen and Hannington 1996). Assuming a  $5 \text{ km}^3$  reaction  
581 zone, the quantity of Au mobilised from the oceanic crust at ODP Hole 1256D is 2.9 t, which is  
582 over an order of magnitude less than that required to form a Au-rich VMS deposit ( $\geq 31$  t Au).  
583 The Au grade of mafic-VMS deposits is also lower than that observed in Au-rich VMS deposits  
584 by around an order of magnitude (Table 6), although small deposits such as those in the axial  
585 graben at  $12^\circ 50' \text{N}$  EPR could have the Au grades of Au-rich deposits if the trapping mechanisms

586 for Au increased in efficiency from 0.034 % to 2.40 %. The Au to base metals ratio of the metals  
587 mobilised at ODP Hole 1256D (0.12) is similar to that of the TAG active mound (0.21,  
588 Hannington et al. 1998) and in the same order of magnitude as average mafic VMS deposits  
589 (0.53, Barrie and Hannington 1999). The ratio is one order of magnitude higher than the axial  
590 graben VMS deposits at 12°50'N EPR (0.02, Hekinian and Fouquet 1985; Fouquet et al.1988),  
591 possibly due to the low trapping efficiency of Au during VMS formation as suggested in Figure  
592 11. Whichever classification for Au-rich VMS deposits is used, it appears that Au needs to be  
593 enriched by one order of magnitude relative to the other elements for formation of a Au-rich  
594 VMS deposit. More effective leaching of Au from the source area, more efficient trapping  
595 mechanisms or magmatic input of Au would be required to form a Au-rich VMS deposit from the  
596 ODP Hole 1256D crust. The settings where abundant Au-rich VMS deposits occur are  
597 characterised by Au-rich source rock (e.g. back-arc basins; Moss et al. 2001; Pitcairn 2011), by  
598 abundant shallow level magmatic intrusions that may provide Au-rich magmatic volatiles (e.g.  
599 Stanton 1990, Ohmoto 1996), by shallow seawater which allows sub-seafloor boiling and  
600 enrichment of Au into a volatile phase (Urabe et al. 1987; Butterfield et al. 1990), or by a  
601 combination of the above. Mafic VMS deposits associated with mid-oceanic spreading ridges are  
602 unlikely to be Au-rich due to the low Au content of source rocks and the water depth which  
603 inhibits sub-seafloor boiling (Herzig and Hannington 1995).

## 604 **Conclusions**

605 Hydrothermal fluid circulation has a profound effect on the bulk chemistry of the oceanic  
606 crust and can lead to the formation of VMS deposits. Analyses, using low detection limit  
607 methods, of deep oceanic crust material from ODP Hole 1256D enable us to quantify the

608 mobility of Au, As, Sb, Se, Te along with S, Cu, Zn and Pb in the oceanic crust. This process is  
609 summarised in Figure 12. The major outcomes of this study are:

- 610 • Gold, Pb, As and Se are efficiently mobilised from the sheeted dyke complex and the  
611 plutonic complex by the hydrothermal fluid circulation (-46 %, -44.1 %, -27.2 % and -  
612 26.7 %, depletions respectively) followed by Cu, S, Zn and Sb (-9.6 %, -8.4 %, -7.9 %  
613 and -2.5 %, depletions respectively). Te appears to be less strongly affected by  
614 hydrothermal fluid circulation. The investigated elements, with the exception of Zn, show  
615 similar depletion patterns being removed from ~1100-1150 mbsf and downward. Zinc is  
616 enriched in the upper sheeted dyke section and is mobilised from ~1300 mbsf. Lead  
617 shows enrichment in the upper sheeted dykes similar to Zn.
- 618 • All elements except Cu are strongly enriched in the sulphide-rich samples of the  
619 transitional zone. Arsenic, Sb, Se and Pb are also enriched in background altered basalts  
620 from the transition zone, suggesting possible enrichment in silicate or oxide phases. Cu is  
621 enriched locally in sulphide-rich samples from the upper sheeted dykes.
- 622 • Arsenic and Sb are taken up by the volcanic section in background-altered samples during  
623 seawater-derived fluid circulation (30.4 % and 24.5 %, respectively). Such enrichments  
624 can have implications for mantle wedge contamination during later subduction. In oceanic  
625 crust more open to seawater circulation, substantial As and Sb enrichment in the volcanic  
626 section could take place and have more impact on mantle wedge contamination.
- 627 • Precipitation of the mobilised metals during fluid migration from the lower crust towards  
628 the seafloor occurs in the upper sheeted dykes (Zn and Pb) and in the transitional zone.  
629 The total mass of metals precipitated is, however, a minor proportion to those produced at

630 depth, and the vast majority of mobilised metals are therefore transported towards the  
631 seafloor.

632 • Quantity and ratios of metals mobilised from the sheeted dyke and plutonic complexes at  
633 ODP Hole 1256D are similar to mafic type VMS deposits. Source areas and percentage of  
634 trapped metals are estimated for the formation of the TAG active mound and the axial  
635 graben VMS deposits at 12°50'N on the EPR. It is suggested that in fast-spreading ridge  
636 systems metal trapping mechanisms are low ( $\leq 1\%$ ) during VMS formation, and that only  
637 a portion of the quantity of metals mobilised from the oceanic crust is trapped as VMS  
638 deposit. This emphasises that important metal mobilisation does not necessarily lead to  
639 VMS formation and that trapping processes such as focus flow, precipitation efficiency  
640 and zone refining are the key processes for VMS formation.

641 • The Cu-affinity of mafic VMS deposits could be enhanced by two processes: preferential  
642 precipitation of Zn, and possibly Pb, in the upper sheeted dykes relative to Cu during  
643 hydrothermal fluid migration towards the seafloor, and better trapping mechanisms for Cu  
644 relative to Zn and Pb during VMS formation.

645 • The mass of Au mobilised and the Au to base metals ratio of the metals mobilised from  
646 the oceanic crust at ODP 1256D are not sufficient to form Au-rich VMS deposits. Both  
647 the quantity of Au and an enrichment of Au relative to base metals would have to increase  
648 tenfold in order to form Au-rich VMS deposits. This could be achieved through more  
649 efficient metal mobilisation from the crust, higher source rock Au content, sub-seafloor  
650 boiling causing higher precipitation efficiency, zone refining or by magmatic input in the  
651 hydrothermal fluids.

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663 Science and Technology.

## 664 **Appendices**

### 665 Appendix 1& 2:

666           See tables Appendix 1 and Appendix 2

### 667 Appendix 3: Error propagation

668           The mass balance calculations for all elements except Te are determined using the elemental  
669 analyses from Appendix 1 and the differentiation curves calculated from Jenner and O'Neill  
670 (2012) database in Appendix 2. The errors associated with the mass balance calculations are  
671 controlled by the heterogeneous distribution of metals in the samples analysed and the error  
672 margin on the differentiation curves, and errors need to be propagated. The errors on the





693 concentration calculated in Table 2. Propagated mass balance errors are shown in Appendix 3  
694 table.

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## 899 **Figure captions**

900 Figure 1: Lithostratigraphy at ODP Hole 1256D with sample locations and Mg#, Y and  
901 K<sub>2</sub>O as indicators of the degree of differentiation and seawater alteration. Major secondary  
902 minerals and temperature are also shown. Modified from Teagle et al. (2006) and Alt et al.  
903 (2010).

904 Figure 2: Concentrations of Au, As, Sb, Se and Te in drill core samples from ODP Hole  
905 1256D. The type of alteration is indicated on the figure. The Hole 1256D median value and  
906 MORB average value are also shown. Median MORB value for Au is from Webber et al. (2013),  
907 average MORB values for As, Sb and Se are from Arevalo and McDonough (2010) and average  
908 MORB value for Te is from Yi et al. (2000).

909 Figure 3: Box plot of the Au, As, Sb, Se and Te content of all the samples in the four  
910 major lithological units in Hole 1256D. The boxes represent 95 % of the data, bars within the box  
911 are the medians, dash lines are the average values and external lines are the extreme values.

912 Figure 4: Concentrations of S, Cu, Zn and Pb in drill core samples from ODP Hole  
913 1256D. The type of alteration is indicated on the figure. The Hole 1256D median value and  
914 MORB average value are also shown. Average MORB values are from Arevalo and McDonough  
915 (2010).

916 Figure 5: Au and related element concentrations plotted versus Y concentrations. Light  
917 grey crosses correspond to Pacific Ocean MORB fresh glass samples from the Jenner and O'Neill  
918 (2012) database, with the exception of Te which represents the Mid-Atlantic Ridge database from  
919 (Yi et al. 2000). Light grey differentiation curves are calculated from the Jenner and O'Neill  
920 (2012) database whilst the red curves correspond to the Hole 1256D corrected differentiation  
921 curves. The style of alteration is indicated on the figure. Plain red squares represent the average  
922 composition of least altered samples. Arrows are data plotting outside the axes. See text for  
923 calculations.

924 Figure 6: Sulphur, Cu, Zn and Pb concentrations plotted versus Y concentrations. Light  
925 grey crosses correspond to Pacific Ocean MORB fresh glass samples from the Jenner and O'Neill  
926 (2012) database. Light grey differentiation curves are those calculated from the Jenner and  
927 O'Neill (2012) database whilst the red curves correspond to Hole 1256D corrected differentiation  
928 curves. The style of alteration is indicated on the figure. Plain red squares represent the average  
929 composition of least altered samples. Arrows are data plotting outside the axes. See text for  
930 calculations.

931 Figure 7: Mass variation of Au and related elements in ODP Hole 1256D drill core  
932 samples relative to depth in the section. The type of alteration is indicated on the figure.

933 Figure 8: Mass variation of S, Cu, Zn and Pb in ODP Hole 1256D drill core samples  
934 relative to depth in the section. The type of alteration is indicated on the figure.

935 Figure 9: Plots of As, Sb, K<sub>2</sub>O and Rb in the volcanic section. The type of alteration is  
936 indicated on the figure. Dashed equation lines correspond to background altered samples and  
937 plain ones to all the samples.

938           Figure 10: Hypothetical reaction zones in Hole 1256D crust from which sufficient metals  
939 are mobilised to form a) a deposit of the size and composition of the TAG active mound and b)  
940 deposits of the size and composition of the axial graben VMS deposits at 12°50'N EPR. Neither  
941 the Sb and Te reaction zones for the axial graben VMS deposits at 12°50'N EPR nor the Te  
942 reaction zone for the TAG active mound can be calculated. The graphic representations of the  
943 reaction zones are characterised by  $x=1/2y=z$ .

944           Figure 11: Mass of metals mobilised from a 5km<sup>3</sup> reaction zones and the efficiency of  
945 metal trapping mechanisms needed to form a) the TAG active mound and b) the axial graben  
946 VMS deposits at 12°50'N on the EPR. Percentage of metal lost is also shown along with some of  
947 the processes that contribute to metal loss during VMS formation.

948           Figure 12: Interpretation of the behaviour of Au, Au-related elements, S, Cu, Zn and Pb in  
949 the oceanic crust during hydrothermal fluid circulation at ODP Hole 1256D. Relative  
950 mobilisations of the different elements and hypothetical location of ODP Hole 1256D drill core  
951 are indicated.

Figure 1  
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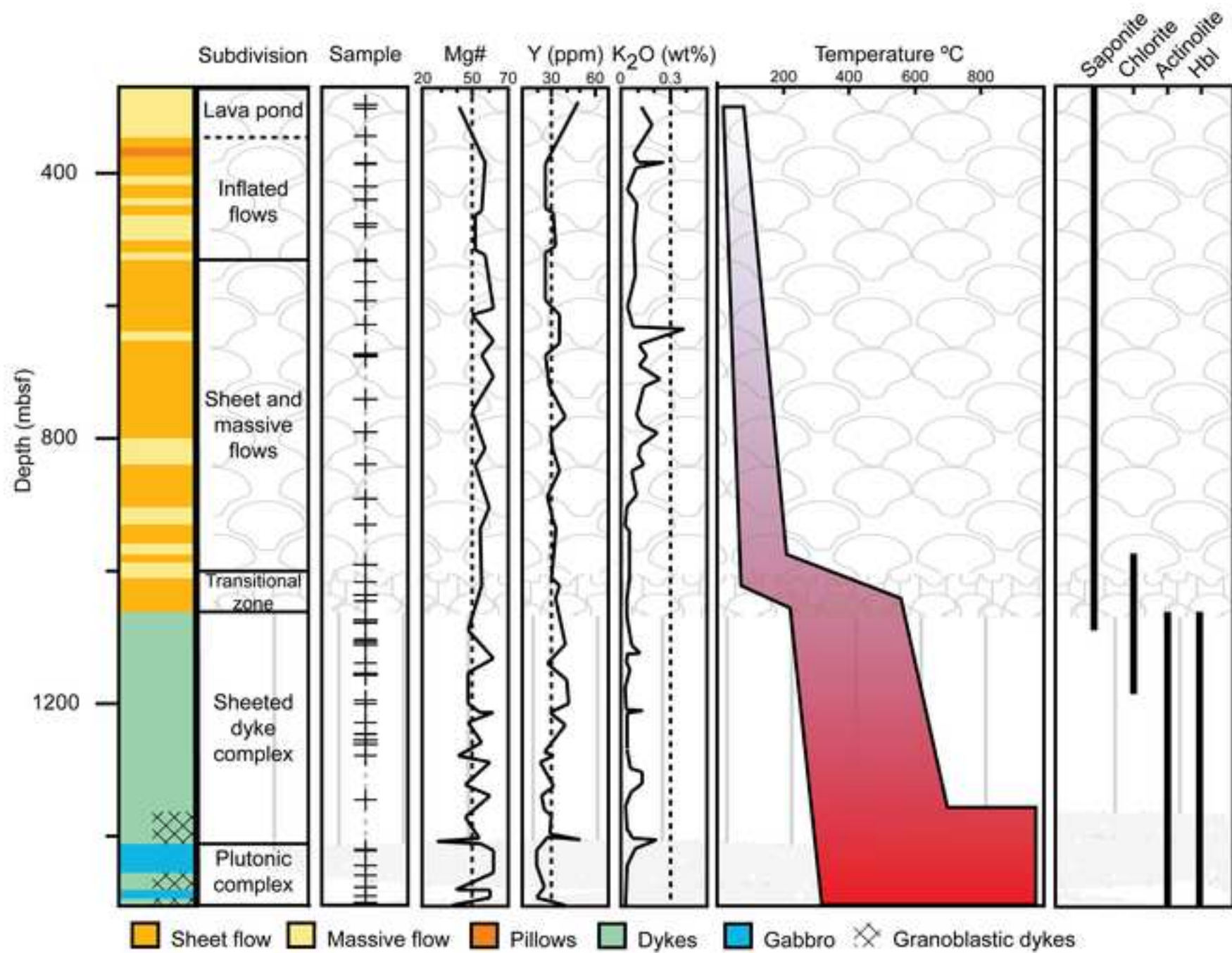


Figure 1

Figure 2  
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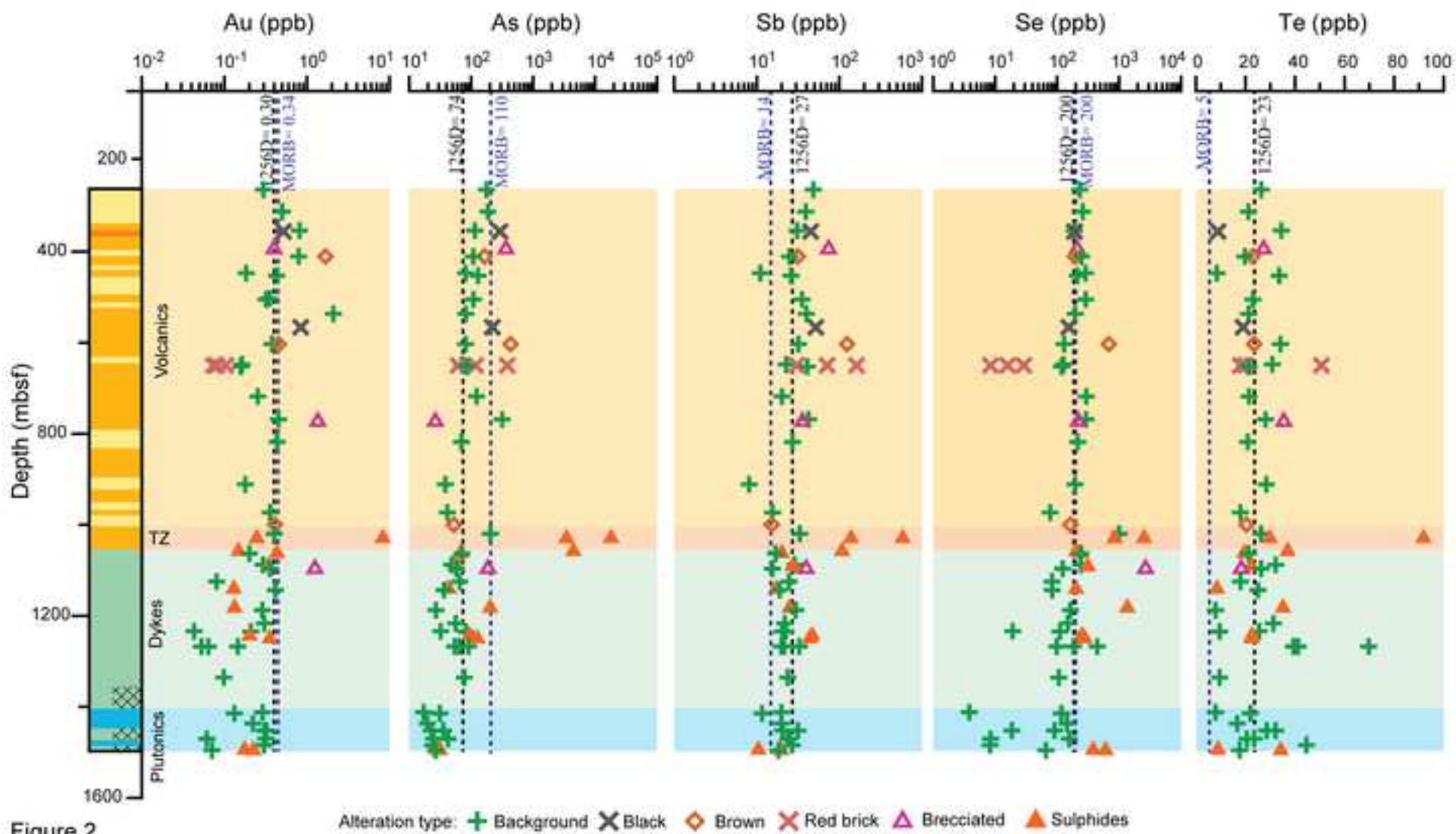


Figure 2

Figure 3  
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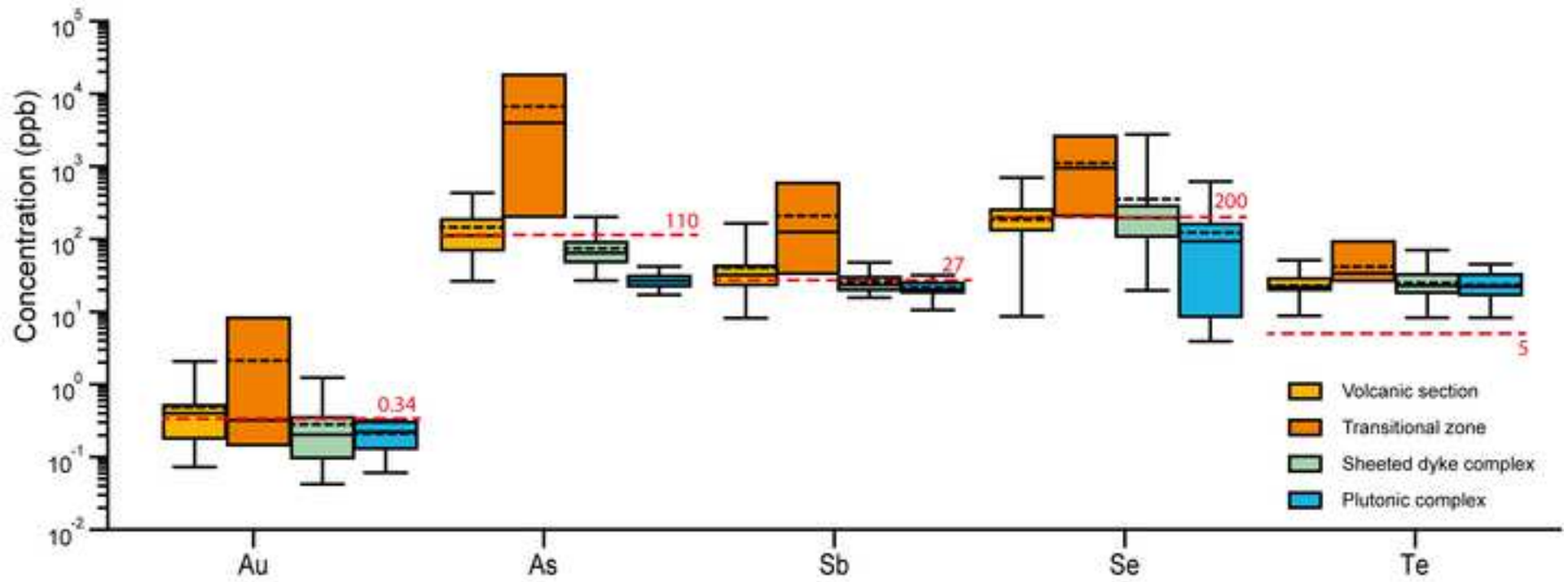


Figure 3



Figure 4  
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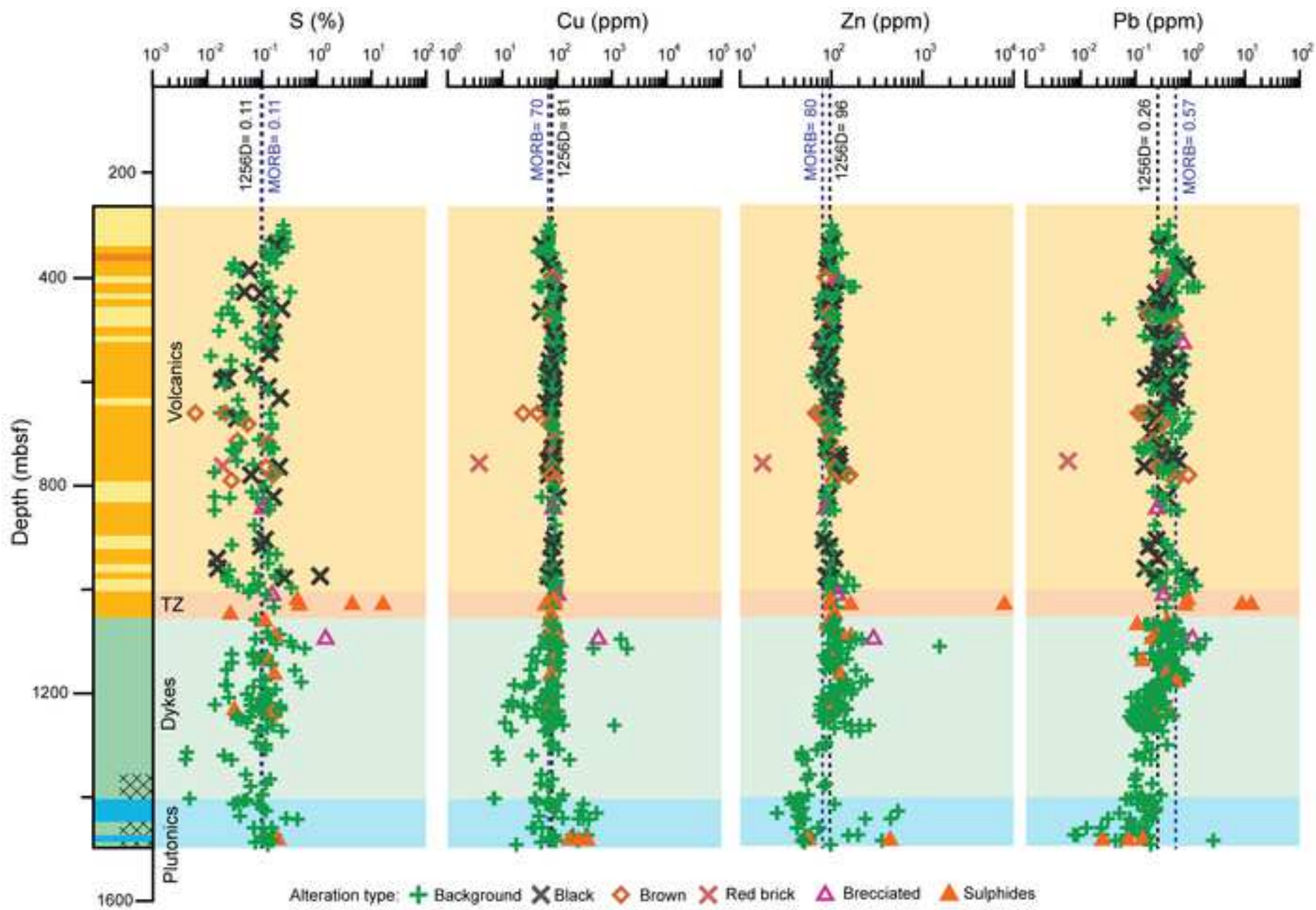


Figure 4

Figure 5  
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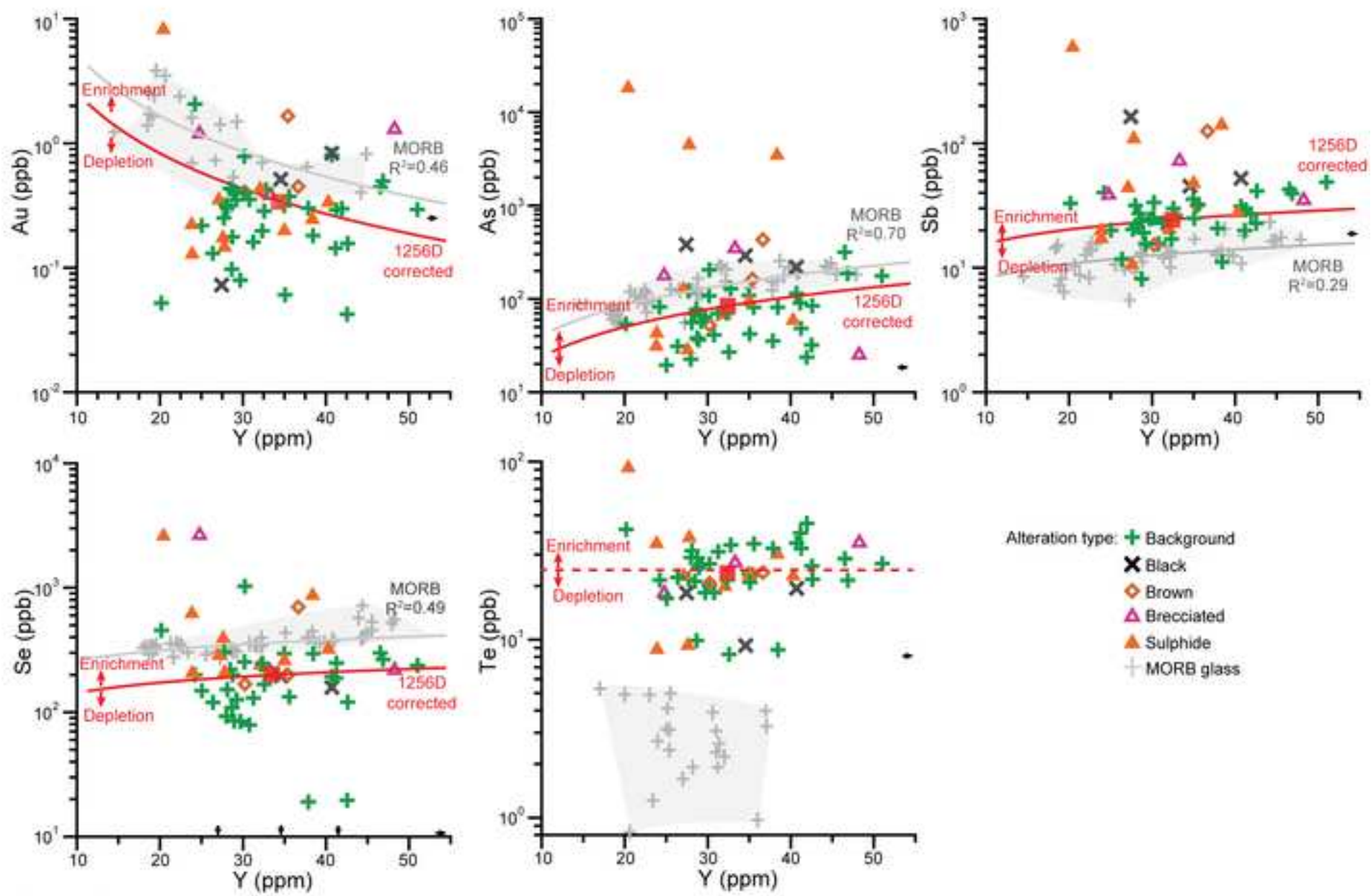


Figure 5

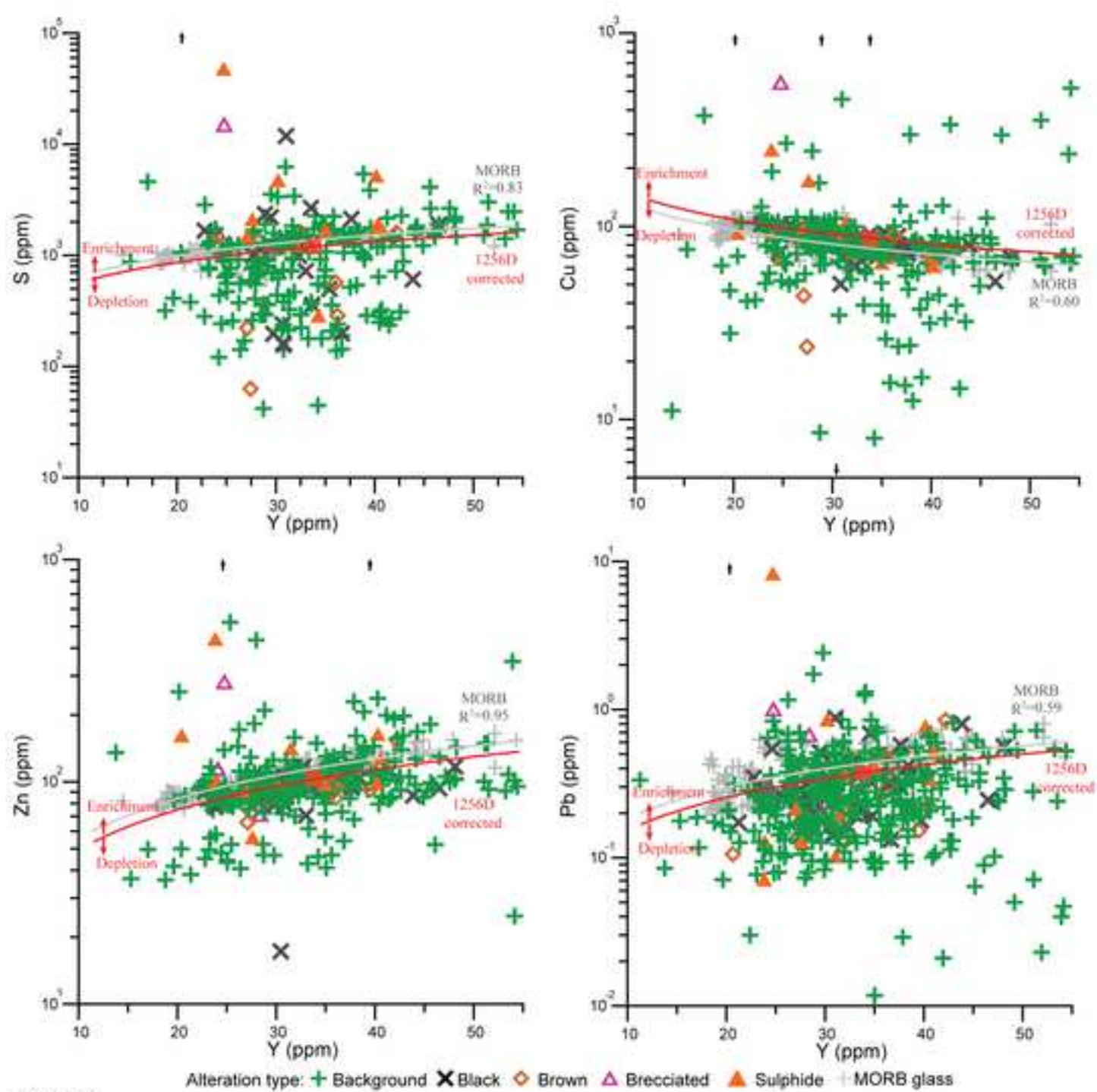


Figure 6

Figure 7  
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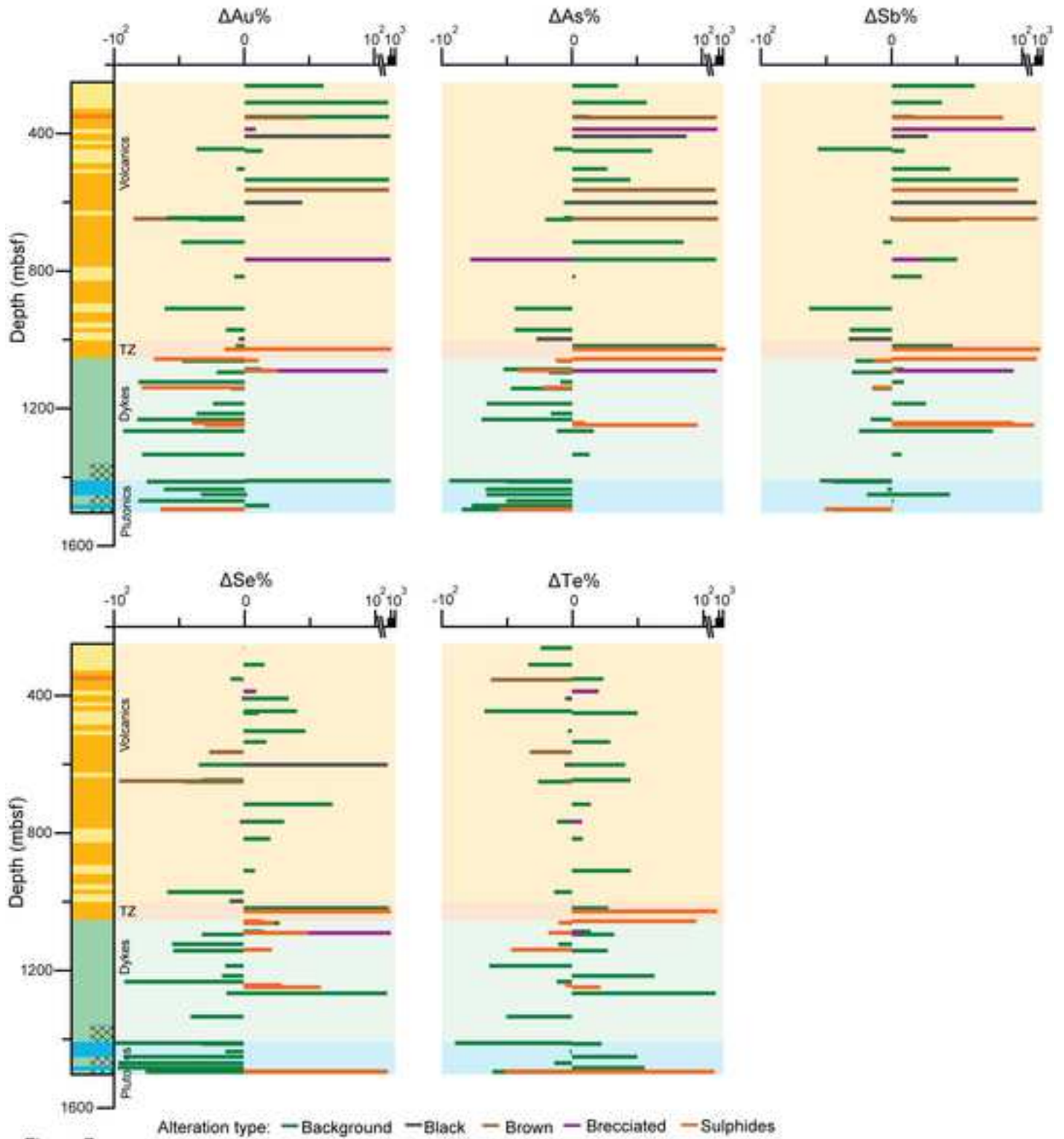


Figure 7

Figure 8  
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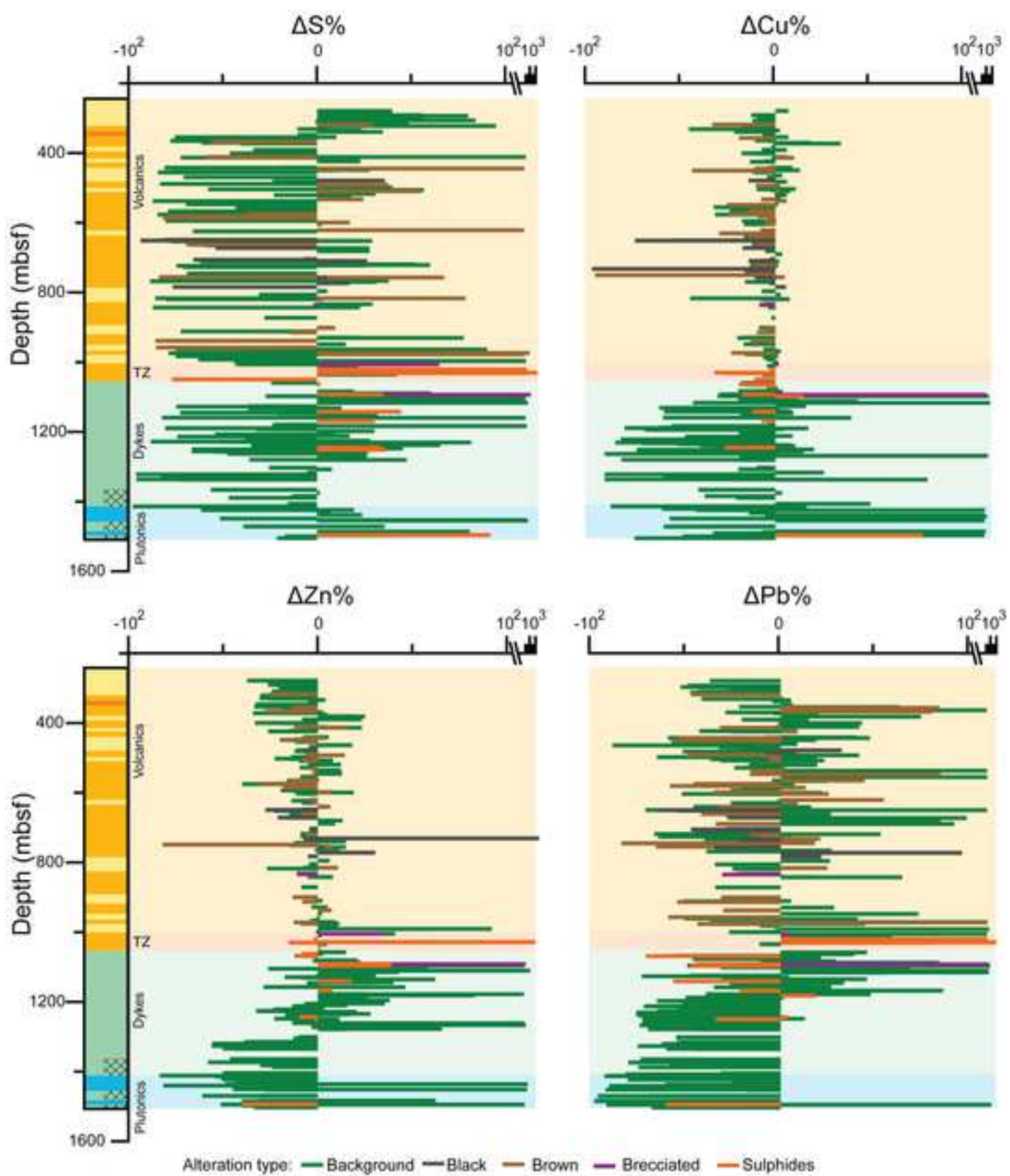


Figure 8

Figure 9  
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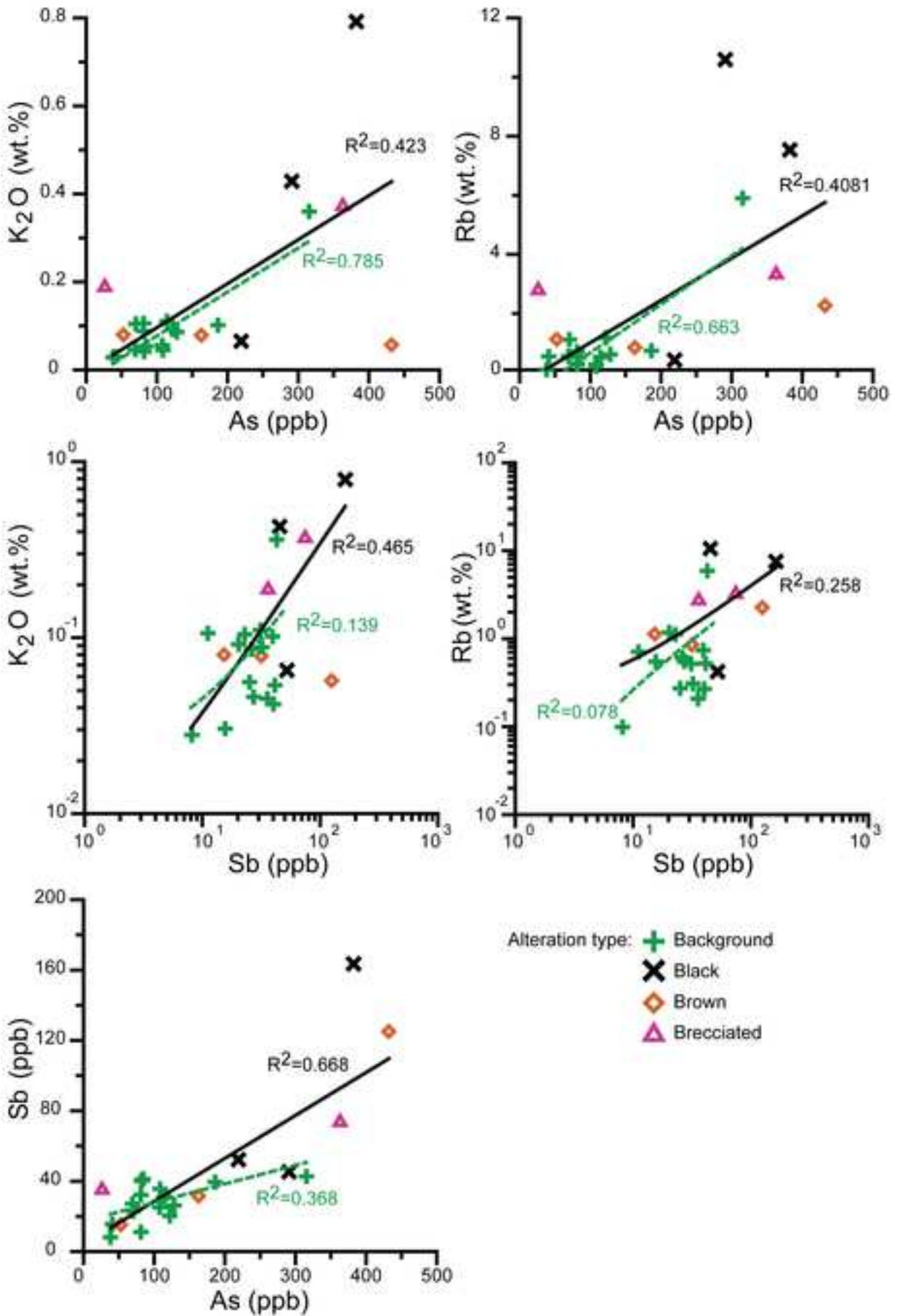


Figure 9

Figure 10  
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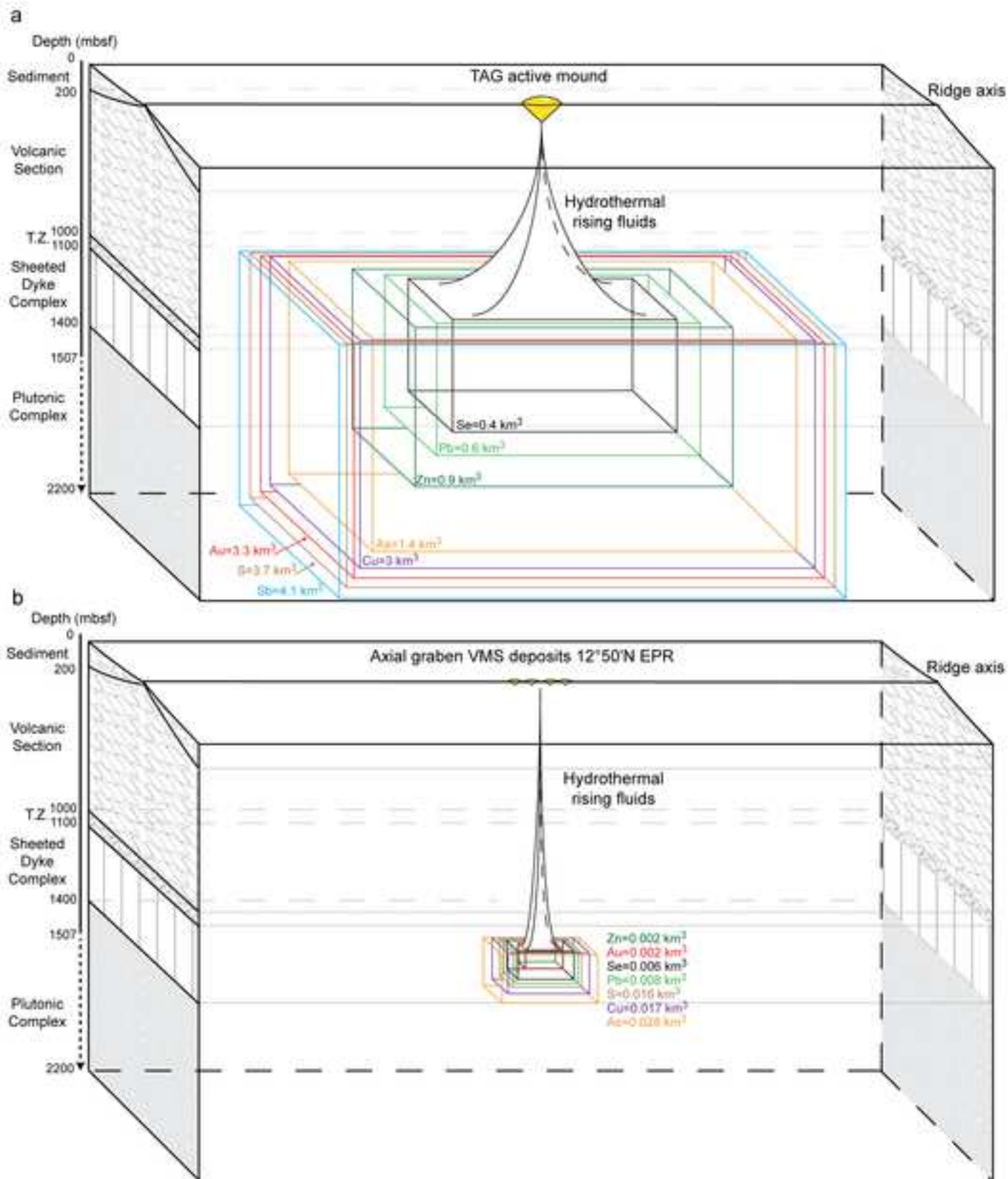


Figure 10

Figure 11  
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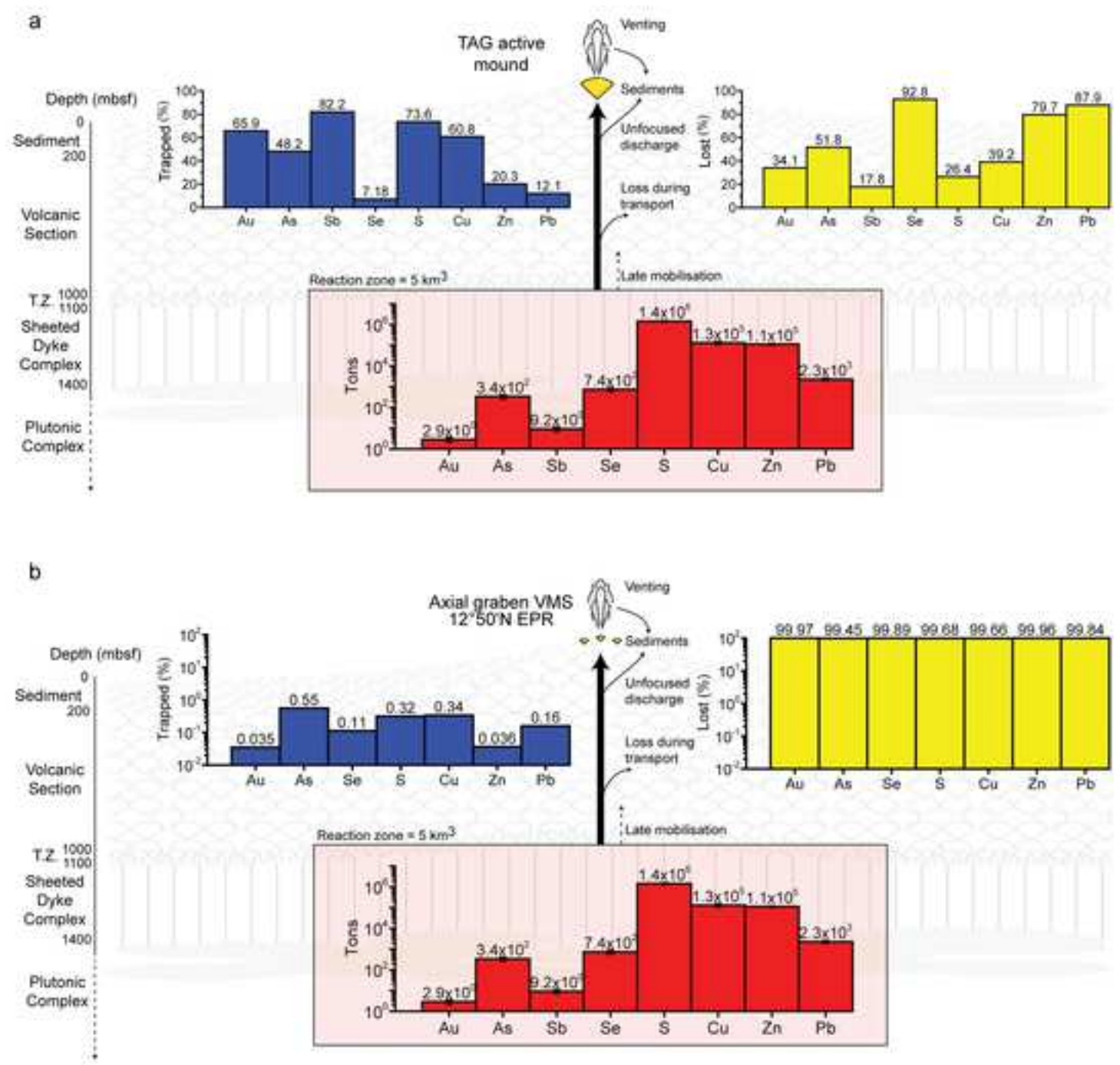


Figure 11



Figure 12  
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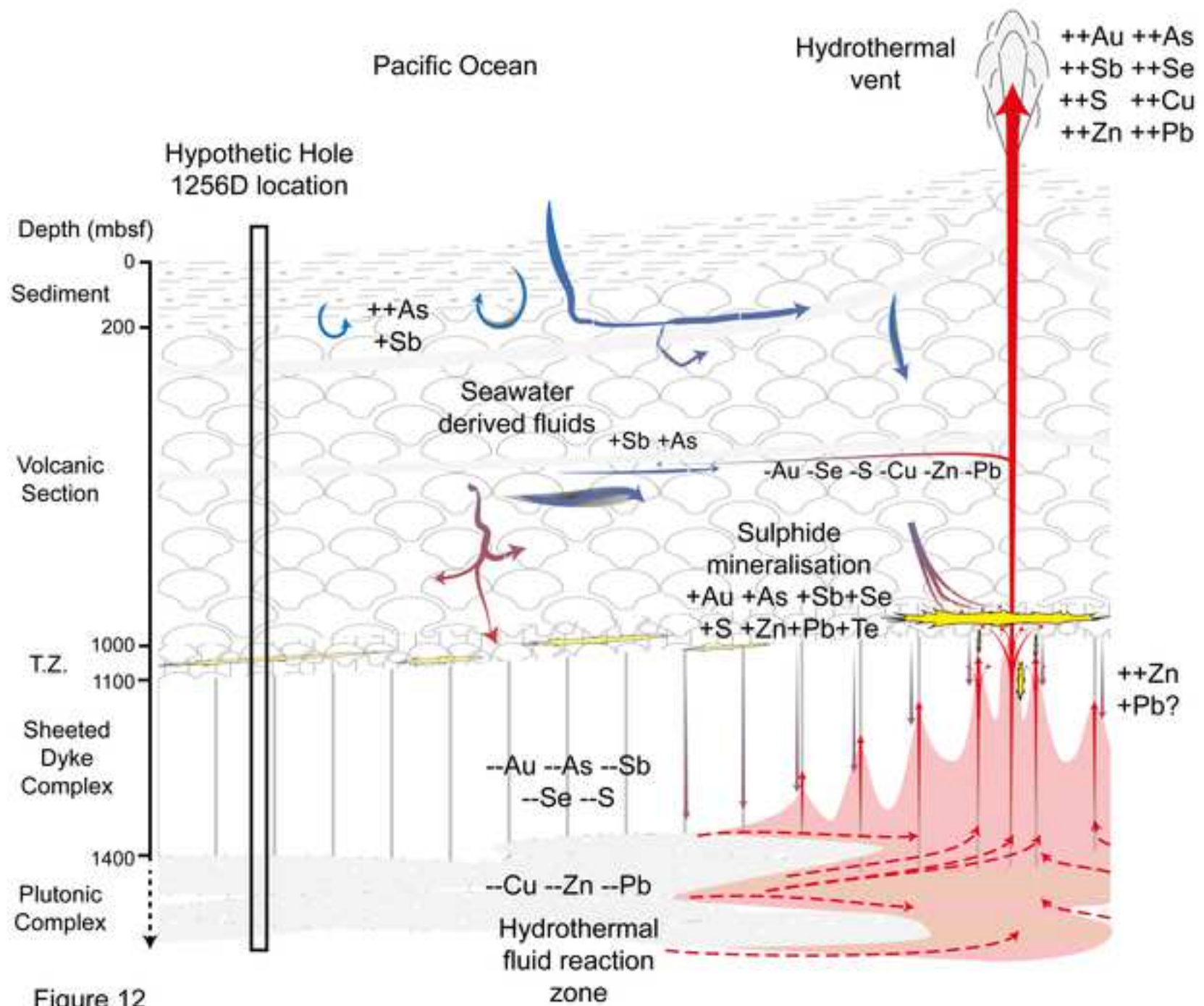


Figure 12

Table 1

Au and related elements											
Element	Reference Material	n	Average (ppm)	$\sigma$ (ppm)	RSD	Published values (ppm)	$\sigma$ (ppm)	RSD	Recovery rate	References Method	References
Au	TDB-1	9	6.37	0.51	7.9	6.3	1.34	21.3	101.1	ICP-MS	Bédard and Barnes, 2002a
	WMS-1	7	208.9	65.9	31.5	246	44	17.9	84.9	INAA	Constantin, 2009
	CH-4	7	632.3	145.7	23.0	713	62	8.7	88.7	INAA	Constantin, 2009
	BIR-1	5	2.21	0.37	16.7	2.53	0.21	8.3	87.4	INAA	Constantin, 2009
	BAS 206*	6	0.24	0.15	62.5						
As	TDB-1	15	2.08	0.37	17.8	2.2	0.3	13.6	94.5	INAA	Bédard and Barnes, 2002b
	WMS-1	10	23.6	2.83	12.0	19.1	0.7	3.7	123.6	INAA	Bédard and Barnes, 2002b
	CH-4	15	7.57	0.74	9.8	8.8	0.5	5.7	86.0		Leaver and Shalley (2010)**
	BAS 206*	6	450.7	83	18.4						
Sb	TDB-1	16	0.75	0.31	41.3	0.72	0.03	4.2	104.2	INAA	Bédard and Barnes, 2002b
	WMS-1	12	7.21	0.89	12.4	6.92	1	14.5	104.2		Leaver and Shalley (2007)**
	CH-4	9	0.79	0.39	49.4	0.77	0.4	51.9	102.6		Leaver and Shalley (2010)**
	BAS 206*	4	37.9	14.1	37.3						
Se	TDB-1	14	0.41	0.1	24.4	0.7			58.6		Leaver and Shalley (1994)**
	WMS-1	5	75.38	11.68	15.5	87			86.6		Leaver and Shalley (2007)**
	CH-4	12	1.94	0.14	7.2	2.1	0.2	9.5	92.4		Leaver and Shalley (2010)
	BAS 206*	5	256.9	18.3	7.1						
Te	TDB-1	7	0.198	0.04	20.5	0.2			99.0		Leaver and Shalley (1994)**
	WMS-1	7	2.29	0.69	30.1						
	CH-4	7	0.37	0.11	29.7						
	BAS 206*	5	53.5	32.9	61.6						
Sulphur and base metals***											
Element	Reference Material	n	Accuracy (%)	Precision (%)	Element	Reference Material	n	Accuracy (%)	Precision (%)		
S	BAS ECRM 877-1	60	10.11	5.05	Zn	BIR-1	7	-2.32	0.51		
						BHVO-1	7	-0.96	0.48		
Cu	BIR-1	7	4.56	0.45	Pb	BIR-1	7	10.87	3.45		
	BHVO-1	7	-1.87	0.30		BHVO-2	7	-0.79	7.57		
	BCR-1	7	-0.97	0.53							

Table 1. Reference materials and internal standards used for whole rock analyses. \*(ppb).\*\*not certified values.\*\*\* See Harris (2011) for details

Sample	Depth (mbsf)	Y	K2O	S	Zn	Cu	Pb	Au	As	Sb	Se	Te
		ppm	%	%	ppm	ppm	ppm	ppb	ppb	ppb	ppb	ppb
1256D 22R-3 13-20	408.7	30.18	0.06	0.34	96.69	90.56	0.48	<b>0.78</b>	<b>107.76</b>	<b>25.14</b>	<b>253.11</b>	<b>20.08</b>
1256D 27R-1 5- 12	445.5	38.45	0.11	0.16	97.84	85.29	0.37	<b>0.18</b>	<b>81.21</b>	<b>11.14</b>	<b>295.10</b>	<b>8.76</b>
1256D 28R-1 0- 8	450.9	32.78	0.09		97.60	83.80	0.22	<b>0.42</b>	<b>128.18</b>	<b>26.16</b>	<b>217.67</b>	<b>33.96</b>
1256D 35R-2 14-20	490.4	35.41	0.05	0.15	103.2	82.3	0.54					
1256D 37R-3 24-30	503.3	34.97	0.04	0.19	95.01	83.93		<b>0.32</b>	<b>108.49</b>	<b>35.64</b>	<b>295.91</b>	<b>23.37</b>
1256D 52R-1 65-75	601.4	35.56	0.09		102.90	71.50	0.18	<b>0.37</b>	<b>81.22</b>	<b>32.21</b>	<b>132.85</b>	<b>34.51</b>
1256D 57R-4 118-125	650.8	42.62	0.05	0.03	91.35	76.24	0.44	<b>0.16</b>	<b>84.47</b>	<b>41.28</b>	<b>120.55</b>	<b>21.78</b>
1256D 99R2 101-120	909.9	28.74	0.03	0.09	92.56	88.28	0.16	<b>0.18</b>	<b>38.11</b>	<b>8.15</b>	<b>199.98</b>	<b>28.74</b>
1256D 110R2 76-88	971.9	30.77	0.03	0.02	96.59	77.27	0.22	<b>0.35</b>	<b>41.04</b>	<b>15.62</b>	<b>78.94</b>	<b>18.32</b>
Glass values from Geldmacher et al. (2013)												
1256D 14R-1 139-142	360.9				100	79.1	0.75					
1256D 17R-1 65-68	374.2				103	89.4	0.35					
1256D 18R-1 81-84	378.8				110	92.3	0.33					
1256D 20R-1 33-37	387.7				117	85.5	0.38					
1256D 21R-1 116-119	398.0				175	48.2	1.07					
1256D 23R-2 16-20	411.9				112	92.8	0.3					
1256D 30R-1 44-59	461.8				95.9	105	0.03					
1256D 38R-1 121-124	505.5				92.4	106	0.36					
1256D 40R-1 31-36	517.8				87.1	104	0.31					
1256D 43R-1 9- 12	534.0				91.6	104	0.28					
1256D 51R-2 60-64	598.2				119	79.4	0.43					
1256D 62R-1 10-12	687.3				97.9	98.9	0.42					
Average primary crust composition		<b>34.39</b>	<b>0.06</b>	<b>0.12*</b>	<b>103.55</b>	<b>86.85</b>	<b>0.38</b>	<b>0.35</b>	<b>83.81</b>	<b>24.42</b>	<b>199.26</b>	<b>23.69</b>
$\sigma$		<b>4.36</b>	<b>0.03</b>	<b>0.2*</b>	<b>18.42</b>	<b>13.43</b>	<b>0.22</b>	<b>0.20</b>	<b>31.81</b>	<b>11.91</b>	<b>81.81</b>	<b>8.59</b>

Table 2. Average primitive crust composition calculated from least altered samples.  $\sigma$  = relative standard deviation. \* preferred S magmatic value from Alt and Shanks (2011).

Table 3

Element	MORB					Hole 1256D corrected				
	A	B	R <sup>2</sup>	RMSD	NRMSD	A	B	R <sup>2</sup>	RMSD	NRMSD
Au	210.86	-1.678	0.46	0.84	23.7	105.81	-1.678	0.46	0.84	23.73
As	3.423	1.121	0.70	31.45	15.6	1.590	1.121	0.70	31.45	15.63
Sb	2.1798	0.5102	0.29	3.10	17.4	4.02	0.5102	0.29	3.10	17.36
Se	79.94	0.447	0.49	65.13	14.5	40.99	0.447	0.49	65.13	14.48
S	153.84	0.616	0.83	97.68	6.9	135.66	0.616	0.83	97.68	6.89
Cu	334.8	-0.416	0.60	11.16	16.6	378.3	-0.416	0.60	11.16	16.65
Zn	13.81	0.603	0.95	6.24	6.4	12.28	0.603	0.95	6.24	6.41
Pb	0.0341	0.720	0.59	0.10	13.7	0.0298	0.720	0.59	0.10	13.71

Table 3. Regression coefficients and correlation coefficients used for equation 1. RMSD= root mean square of deviations. RMSD in ppb for Au, As, Sb and Se and in ppm for S, Cu, Zn and Pb. NRMSD= normalised RMSD in perc cent. MORB values are determined from Jenner and O'Neill (2012) database.

Table 4

			$\Delta$ Au %	$\Delta$ As %	$\Delta$ Sb%	$\Delta$ Se %	$\Delta$ Te %	$\Delta$ S %	$\Delta$ Cu %	$\Delta$ Zn %	$\Delta$ Pb %	
<b>Volcanic section</b>	Background altered samples	<b>median</b>	<b>3.7</b>	<b>20.0</b>	<b>19.9</b>	<b>13.8</b>	<b>2.7</b>	<b>-10.5</b>	<b>-4.7</b>	<b>-4.5</b>	<b>-4.6</b>	
		average	37.1	25.3	17.1	7.3	4.8	-12.6	-5.6	-4.0	2.8	
		min	-61.7	-44.5	-63.4	-58.4	-66.9	-88.8	-46.0	-39.4	-89.1	
		max	239.1	168.2	96.9	68.1	50.3	223.9	34.8	94.7	278.2	
		n	16	16	16	16	16	107	140	140	141	
	Specific alteration (brown, black and breccia)	<b>median</b>	<b>47.0</b>	<b>180.8</b>	<b>90.8</b>	<b>-2.4</b>	<b>-4.0</b>	<b>1.7</b>	<b>-9.4</b>	<b>-5.5</b>	<b>-16.3</b>	
		average	150.4	194.9	182.2	14.1	-9.7	20.4	-13.8	797.7	-10.7	
		min	-85.4	-78.6	-33.2	-95.2	-61.1	-94.0	-97.5	-82.1	-84.3	
		max	571.7	487.3	651.5	241.7	20.7	953.5	9.2	46616.1	153.4	
		n	8	8	8	8	8	35	59	59	69	
	<b>Median volcanic section</b>	<b>93.5% bck; 6.5% sp. alt.</b>	<b>6.6</b>	<b>30.4</b>	<b>24.5</b>	<b>12.7</b>	<b>2.2</b>	<b>-9.7</b>	<b>-5.0</b>	<b>-4.6</b>	<b>-5.3</b>	
	<b>Transitional zone</b>	Background altered samples	<b>median</b>	<b>-38.4</b>	<b>3426.7</b>	<b>222.2</b>	<b>230.7</b>	<b>61.7</b>	<b>22.1</b>	<b>-6.8</b>	<b>6.3</b>	<b>45.7</b>
average			-38.4	3426.7	222.2	230.7	61.7	13.4	-9.8	8.6	31.3	
min			-70.0	181.5	46.5	15.2	28.0	-24.4	-18.1	3.2	-11.2	
max			-6.8	6672.0	397.9	446.3	95.4	42.4	-4.5	16.4	59.3	
n			2	2	2	2	2	3	3	3	3	
Mineralised samples		<b>median</b>	<b>453.7</b>	<b>21170.2</b>	<b>1752.1</b>	<b>902.2</b>	<b>285.0</b>	<b>308.2</b>	<b>-14.2</b>	<b>2.8</b>	<b>1380.9</b>	
		average	453.7	21170.2	1752.1	902.2	285.0	4791.8	-17.4	1796.9	1851.5	
		min	-15.6	3559.9	443.1	315.7	14.1	-77.2	-32.4	-14.7	79.1	
		max	922.9	38780.6	3061.0	1488.6	555.8	18893.3	-3.2	8888.2	4565.1	
		n	2	2	2	2	2	6	6	6	5	
<b>Sheeted dyke complex</b>		Background altered samples	<b>median</b>	<b>-44.4</b>	<b>-16.6</b>	<b>0.00</b>	<b>-16.6</b>	<b>14.6</b>	<b>-14.7</b>	<b>-10.6</b>	<b>1.19</b>	<b>-35.3</b>
			average	-46.4	-24.7	1.2	-7.5	21.8	1.9	22.3	18.6	-21.9
	min		-93.6	-70.0	-30.7	-91.1	-63.2	-96.3	-90.8	-57.8	-80.7	
	max		12.8	16.5	77.3	190.1	199.4	455.1	2046.9	1222.5	426.8	
	n		11	11	11	11	11	70	124	124	175	
	Sulphide-rich	<b>median</b>	<b>-9.99</b>	<b>-1.5</b>	<b>48.30</b>	<b>39.40</b>	<b>-7.52</b>	<b>35.29</b>	<b>-7.43</b>	<b>8.94</b>	<b>-28.08</b>	
		average	-0.8	41.7	43.9	279.3	-8.0	230.0	44.1	34.9	-1.4	
		min	-79.4	-41.3	-15.2	21.8	-46.5	1.8	-26.8	-10.4	-71.5	
		max	110.0	220.9	103.3	1492.2	22.0	1426.8	469.1	231.5	241.2	
		n	6	6	6	6	6	8	9	9	10	
	<b>Plutonic complex</b>	All samples	<b>median</b>	<b>-51.7</b>	<b>-64.1</b>	<b>-11.4</b>	<b>-61.9</b>	<b>10.6</b>	<b>13.1</b>	<b>-6.3</b>	<b>-39.5</b>	<b>-75.7</b>
			average	6.4	-65.4	-17.7	-16.8	4.7	28.4	75.9	18.4	-54.1
min			-81.9	-94.6	-55.2	-98.8	-89.3	-98.0	-87.7	-83.7	-98.8	
max			466.4	-44.2	44.5	266.9	110.5	490.0	602.2	506.4	616.6	
n			10	10	10	10	10	22	34	34	48	

Table 4. Mass variations calculated for Au and related elements for the different lithological units. Gold, Se, S, Cu, Zn and Pb mass variations are calculated using the Jowitt et al. (2012) method whereas As, Sb and Te mass variations are calculated using the Nesbitt (1979) method. Average volcanic section calculated assuming 93.5% background altered samples and 6.5% samples being affected by specific alterations. Sulphide-rich samples correspond to samples hosting sulphide veins or breccias.

Table 5

	Au	As	Sb	Se	Te	S	Cu	Zn	Pb
$\Delta$ %	-46.0	-27.2	-2.5	-26.7	13.7	-8.4	-9.6	-7.9	-44.4
$\sigma$ (%)	11.5	5.4	0.5	5.7	5.7	0.7	1.6	0.5	6.3
$\Delta$ ppm	-1.9E-04	-0.02	-6.2E-04	-0.05	3.2E-03	-97.52	-8.83	-7.58	-0.15
$\sigma$ (ppm)	4.9E-05	0.004	1.33E-04	0.01	1.4E-03	8.33	1.47	0.49	0.02
<b>Mass (t)</b>	<b>2.9</b>	<b>339</b>	<b>9.2</b>	<b>741</b>	<b>48</b>	<b>1.4E+06</b>	<b>1.3E+05</b>	<b>1.1E+05</b>	<b>2.3E+03</b>
$\sigma$ (t)	0.7	66.7	2.0	157	20.1	1.2E+05	2.2E+04	7.2E+03	321

Table 5. Variations of Au and related elements mobilised by hydrothermal fluid circulation in the sheeted dyke complex and the plutonic complex. Median values of background altered samples from the sheeted dyke and plutonic complexes are used for the calculation. Mass calculation is done using a hydrothermal cell volume of 5 km<sup>3</sup> (3.9 km<sup>3</sup> of sheeted dyke complex and 1.1 km<sup>3</sup> of plutonic complex) and a density of 2.97 g.cm<sup>-3</sup> for the two units (Teagle et al. 2006).

Table 6

Deposit	Location	Tonnage	S	Cu	Zn	Pb	Au	As	Sb	Se	Te	References
		Mt	%	%	%	ppm	ppm	ppm	ppm	ppm	ppm	
TAG active mound*	MAR	3.8	42.1	2.1	0.6	72	0.5	43	2	14		Hannington et al. (1998); Petersen et al. (2000)
Galapagos Fossil Hydrothermal Field Zone A	GSC	1.5	45.7	4.7	0.8	100	0.08	136	5.6	498	2.4	Perfit et al. (1999)
Galapagos Fossil Hydrothermal Field Zone B	GSC		26.2	3.4	2.9	500	0.39	125	4	233	1.2	Perfit et al. (1999)
Bent Hill	JFR	9	32.1	0.48	4.8	477	0.27	78	22	43		Goodfellow and Franklin (1993)
Hydrothermal Field at 12°50'N EPR**	EPR	0.52	34.3	2.21	0.21	180	0.05	94		42		Hekinian and Fouquet (1985); Fouquet et al. (1988)
Large VMS at 12°43'N EPR***	EPR		36.4	0.765	2.33	277	0.19	130		52		Fouquet et al. (1996)
Average mafic VMS		1-5		4.3	11.7	0.2	1.2					Herzig and Hannington (1995)
Average mafic VMS (late phanerozoic)		3.3		2	1.1	100	1.7					Barrie and Hannington (1999)

Table 6. Examples of mafic VMS deposit composition in base metals, Au and related elements. MAR= Mid-Atlantic Ridge; GSC= Galapagos spreading centre; JFR= Juan de Fuca Ridge; Troodos= Troodos ophiolite. \*S as sulphide from Hannington et al. (1998) and metal concentrations from Petersen et al. (2000). \*\* Median values of all data from Henkinian and Fouquet (1985) and Fouquet (1988). \*\*\*Tonnage calculated assuming a cone of 200 m in diameter and 70 m of height (Fouquet et al. 1996) and a density of 3 g.cm<sup>-3</sup> (Henkinian and Fouquet 1985).

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