






Nano-patterning on multilayer MoS₂ via block copolymer lithography for highly sensitive and responsive phototransistors

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Indirect bandgap of multilayer molybdenum disulfide has been recognized as a major hindrance to high responsivity of MoS₂ phototransistors. Here, to overcome this fundamental limitation, we propose a structural engineering of MoS₂ via nano-patterning using block copolymer lithography. The fabricated nanoporous MoS₂, consisting of periodic hexagonal arrays of hexagon nanoholes, includes abundant edges having a zigzag configuration of atomic columns with molybdenum and sulfur atoms. These exposed zigzag edges are responsible for multiple trap states in the bandgap region, as confirmed by photo-excited charge-collection spectroscopy measurements on multilayer nanoporous MoS₂ phototransistors, showing that in-gap states only near the valence band can result in a photogating effect. The effect of nano-patterning is therefore to significantly enhance the responsivity of multilayer nanoporous MoS₂ phototransistors, exhibiting an ultra-high photoresponsivity of 622.2 A W⁻¹. Our nano-patterning of MoS₂ for photosensing application paves a route to structural engineering of two-dimensional materials for highly sensitive and responsive optoelectronic devices.

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Due to the dramatic advance in complementary metal-oxide-semiconductor (CMOS) image sensors in the past decades, they are widely used in present-day smartphones and digital cameras^{1,2}. However, one of the most critical bottlenecks of CMOS image sensors is the size that is smaller than target objects, making one-to-one direct mapping between a sensor and an object impossible without using lenses. This leads to the limitation of their use for touch-based applications such as interactive display, fingerprint security, and human-robot interaction. In this regard, thin-film phototransistor-based image sensors can be a talented alternative as they can be easily scaled up with lower process costs, provide new functionality such as flexibility, and extend the boundaries of practical applications^{3–5}. Therefore, exploring channel materials of thin-film transistors (TFTs) that can bring high carrier mobility, high photoresponsivity, and mechanical flexibility is of great importance to make a new breakthrough for next-generation optoelectronic devices^{6–8}.

Two-dimensional (2D) layered nanomaterials, such as transition metal dichalcogenides (TMDs), can be promising contenders for the active material of thin-film phototransistor image sensors^{9–12} due to their excellent optical and electrical performances as well as mechanical stability^{13–15}. The optical characteristics of TMDs significantly vary with the thickness of the material^{13,16,17}. In general, direct bandgap monolayer TMDs exhibit high photoresponsivity, whereas multilayer TMDs show relatively low photoresponsivity due to their indirect bandgap^{8,18}. However, multilayer TMDs have been recognized as being more suitable for versatile photodetector applications than monolayer TMDs, with great benefits arising from the higher carrier density¹⁹ and the wider spectral response from near-infrared to ultraviolet (UV)¹⁸, as well as easier fabrication.

Recently, structural engineering of TMDs has drawn significant attention as it can alter the fundamental material properties²⁰. Several approaches to overcome the limitation of the low photoresponsivity of multilayer TMDs have been reported, which include stacking of heterostructures²¹, decoration with quantum dots^{22,23}, and introduction of different alignment directions²⁴. On the other hand, nano-patterning, a relatively new technique for structural engineering, can provide a novel route to tune the optical and electrical performances of 2D materials by directly modifying the electronic states. For example, graphene, which has a significant limitation for electronic applications due to its zero bandgap, can obtain a finite bandgap through a nanoporous patterning by inducing quantum confinement effect^{25–28}. In addition, Kim et al.²⁹ reported that nanopore-patterned multilayer molybdenum disulfide (MoS₂) presents a large photoluminescence emission peak similar to the direct bandgap monolayer MoS₂. However, experimental work has not been reported for the nanoporous MoS₂ device and its optoelectronic characteristics.

Here, we present the nano-patterning of multilayer MoS₂ to achieve high-performance multilayer MoS₂ phototransistors. The block copolymer (BCP), which consists of two homopolymers and forms periodic nanostructures through phase separation^{30–32}, was used to pattern multilayer MoS₂ into a nanoporous structure containing an array of periodic nanoholes. It should be noted that the photoresponsivity of our nanoporous MoS₂ phototransistors (622.2 A W⁻¹) is even higher than the direct bandgap monolayer MoS₂ counterparts reported in the literature under similar optical power densities and wavelengths of incident light^{33–35}. To investigate a fundamental origin of this ultra-high responsivity from the nanoporous MoS₂ phototransistors, we explored the exposed edge structure of the nanoporous MoS₂ by Raman spectroscopy, X-ray photoelectron spectroscopy (XPS), and scanning transmission electron microscopy (STEM). Photo-excited charge-collection spectroscopy (PECCS) measurement³⁶ was performed to develop

understanding of the newly generated trap states in the bandgap by adopting the nanohole structure on MoS₂. Finally, device simulation elucidated the mechanism of photoresponse in nanoporous MoS₂ phototransistors.

Results

Nano-scale patterning on multilayer MoS₂. A nano-scale template based on the phase separation behavior of the BCP was utilized to fabricate the nanoporous structure on multilayer MoS₂ films. Figure 1a shows the fabrication processes to achieve nanoporous MoS₂. MoS₂ multilayer sheets were separated from bulk MoS₂ by a mechanical exfoliation and then located onto a Si/SiO₂ substrate. SiO₂ with a thickness of 10 nm was deposited on the MoS₂ surface by an electron-beam (e-beam) evaporator to lead to covalent bonds with the grafting layer of a random copolymer (RCP) as well as to prevent damage to MoS₂ during the etching process^{26,31,37}. The BCP was spin-coated on the RCP film and annealed at 230 °C to produce cylindrical phase separation of polymethyl methacrylate (PMMA) and polystyrene (PS). The cylindrical PMMA phase was selectively removed by shining UV light on the BCP film and acetic acid, resulting in the nanoporous template of PS. A field emission-scanning electron microscopy (FE-SEM) image of the template shows a uniform and periodic structure of circular nanoholes with an average diameter of 21.1 ± 1.1 nm (Supplementary Fig. 1). Moreover, the nanohole dimension could be controlled from 24.7 to 30.8 nm by simply changing the treatment time of oxygen (O₂) plasma reactive-ion etching (RIE), as confirmed by FE-SEM images (Supplementary Fig. 2a–d) and the linear relationship between the hole diameter and the treatment time (Supplementary Fig. 2e). Using the BCP as a template, the underlying SiO₂ was perforated by sulfur hexafluoride (SF₆) plasma RIE. Finally, uniform hexagon nanoholes were formed on the multilayer MoS₂ sheets through two consecutive steps: (i) dry etching using boron trichloride (BCl₃) plasma RIE and (ii) wet etching using potassium ferricyanide solution. After removing the remaining SiO₂ layer, the patterned surface of MoS₂ was observed by annular dark field (ADF)-STEM, as shown in Fig. 1b. The uniformity of the nanoholes was validated by calculating the hole diameters in a large-area ADF-STEM image (Supplementary Fig. 3a), in which the average hole diameter was turned out to be 26.2 ± 2.3 nm (Supplementary Fig. 3b). The substructure of the hole distribution was consistently characterized. The hexagonal nanoholes with an incircle radius (r_h) of ~12 nm were mostly distributed into hexagonal array with a secondary incircle radius (R_{h-h}) of ~40 nm (Fig. 1b). This crystallographic feature of the nanoporous MoS₂ film was corroborated by the fast Fourier transform (FFT) pattern analysis of the ADF-STEM image (Supplementary Fig. 3a), which is equivalent to electron diffraction of the corresponding region (Fig. 1c). The first diffraction ring shows the overall periodicity of the nanoholes, with a length repetition of ~40 nm, and the outmost faint diffraction ring shows the hexagonal shape of the nanoholes, with r_h of ~12 nm. The spotty diffraction pattern indicates that the nanoporous MoS₂ has a multi-domain hexagonal array of the hexagonal nanoholes. The ADF intensity profile across several hexagonal nanoholes (Supplementary Fig. 3c) supports this 2D length repetition in the nanoporous MoS₂. According to STEM-energy dispersive X-ray spectroscopy (EDX) elemental mapping of Mo K, S K, and Si K peaks on the nanoporous MoS₂ transferred to a copper grid (Fig. 1d), it is clarified that the chemical composition of the nanoporous structure is MoS₂. The thickness of the nanoporous MoS₂ nanosheet was estimated to be ~7.6 nm based on the cross-section ADF-STEM image (the bottom of Fig. 1d), which corresponds to 11 layers of MoS₂.

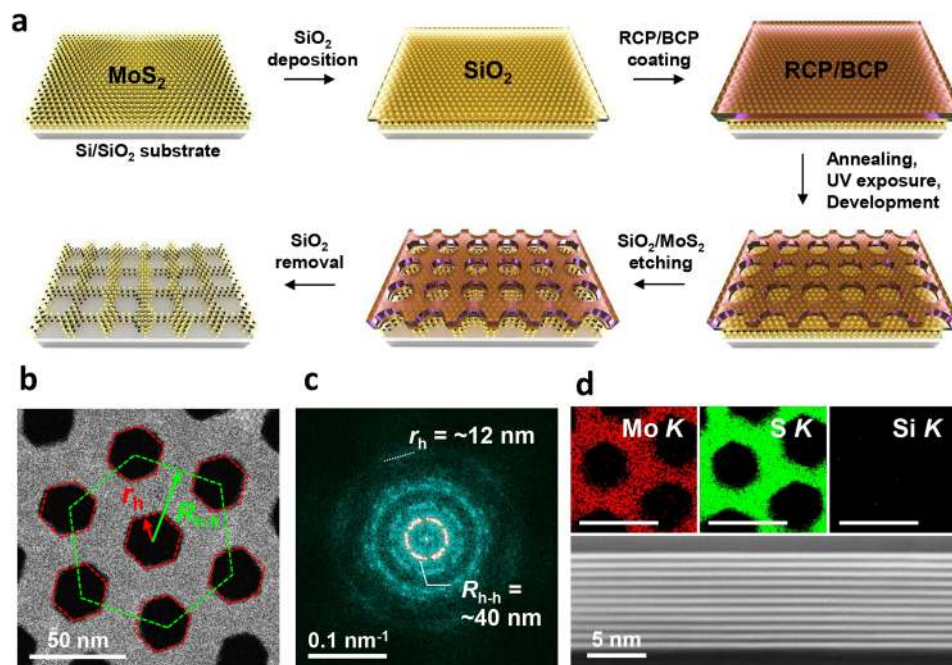


Fig. 1 Preparation and microscopic characterization of the nano-patterned MoS₂ multilayer. **a** A schematic illustration of nano-scale patterning process on multilayer MoS₂. **b** ADF-STEM image showing an array of nanoholes in the multilayer MoS₂. **c** FFT pattern representing a crystallographic hexagonal array of hexagon holes (from Supplementary Fig. 3a). **d** STEM-EDX elemental mapping (top panels) for Mo K, S K, and Si K peaks (scale bars of 50 nm) and cross-sectional view (bottom panel) of the multilayer nanoporous MoS₂.

Hexagonal nanohole edges in nanoporous MoS₂. In order to confirm the material characteristics of the multilayer nanoporous MoS₂ regarding the formation of nanoholes, Raman spectroscopy, and XPS were employed. In Fig. 2a, a characteristic Raman doublet of pristine MoS₂ is present at 383.7 cm⁻¹ for E_{2g}¹ (in-plane mode) and 407.5 cm⁻¹ for A_{1g} (out-of-plane mode), which corresponds to multilayer MoS₂³⁸. The peak positions were rarely changed in the nanoporous MoS₂, whereas the relative intensity of E_{2g}¹/A_{1g} was reduced from 0.824 for pristine MoS₂ to 0.604 for nanoporous MoS₂. This noticeable change is in good agreement with the increased number of edges in nanoporous MoS₂ because A_{1g} is preferentially excited for edge-terminated films compared to E_{2g}¹ due to their polarization dependence^{20,29,39,40}. A decrease in the peak intensity of E_{2g}¹/A_{1g} was consistently observed for nine different nanoporous MoS₂ films (inset of Fig. 2a), strongly supporting the reliability of the manufacturing process.

Figure 2b shows XPS spectra of Mo 3d and S 2p core levels of pristine MoS₂ and nanoporous MoS₂. To compensate for sample charging, all the XPS spectra were calibrated by C 1s peak located at 284.5 eV. Most intense doublets of the Mo 3d and S 2p were observed at 229.15 eV (Mo⁴⁺ 3d_{5/2}) and 161.96 eV (S²⁻ 2p_{3/2}), respectively, in both pristine and nanoporous MoS₂. However, these two doublets in the nanoporous MoS₂ had wider full-width at half-maximum (FWHM) than pristine MoS₂. Therefore, we deconvoluted the Mo 3d peak into three component peaks corresponding to intrinsic MoS₂ (*i*-MoS₂), substoichiometric MoS₂ (*s*-MoS₂), and molybdenum oxide (MoO₃) bonding⁴¹⁻⁴⁴. The peak of *s*-MoS₂ is attributed to the electronic structure different from stoichiometric MoS₂ (ratio of S/Mo = 2) because there are fewer S atoms around Mo atoms at exposed edge sites⁴⁴. The doublet of *s*-MoS₂ at 228.55 eV (Mo⁴⁺ 3d_{5/2}) and 231.68 eV (Mo⁴⁺ 3d_{3/2}), respectively, was observed in both the pristine and the nanoporous MoS₂, whereas the atomic ratio of *s*-MoS₂ increased from 6.32% in pristine MoS₂ to 14.41% in nanoporous MoS₂ (Supplementary Table 1). In addition, as evidenced by a decrease of the ratio of S²⁻ 2p to Mo⁴⁺ 3d (*i*-MoS₂ + *s*-MoS₂)

from 2.00 to 1.89 after nanoporous patterning, the degree of the edge exposure introduced by formation of nanoholes on multilayer MoS₂ can be quantitatively estimated.

The MoO₃ may be derived from oxidation of surface and etched region of MoS₂ when exposed to air. An atomic ratio of Mo⁶⁺ 3d (MoO₃), located at 232.16 eV (Mo⁶⁺ 3d_{5/2}) and 235.29 eV (Mo⁶⁺ 3d_{3/2}), rarely increased from the pristine (10.90%) to the nanoporous MoS₂ (12.11%) due to the protective effect of the SiO₂ layer on the MoS₂ film. In addition, the absence of Cl 2p, B 1s, N 1s peaks (Supplementary Fig. 4a-c) verifies the protective function of the SiO₂ layer, which helps to prevent chemical contaminations during the dry etching with BCl₃ plasma RIE and wet etching with potassium ferricyanide solution.

To determine the atomic configuration of the hexagonal nanohole edges in the nanoporous MoS₂, the atomic structures of multilayer MoS₂ showing a stacking sequence as well as the edge of the nanohole should be scrutinized. Figure 2c presents a typical hexagonal shape of the nanohole in the nanoporous MoS₂. The inset in Fig. 2c shows the selected area diffraction pattern (SADP) for the MoS₂ multilayer region, suggesting a 2H-MoS₂ structure (S.G. = P6₃/mmc) with an AA' stacking configuration in which atomic columns along the [001] orientation are mixed with Mo and S atoms. To identify the edge structure of the hexagonal hole, atomic-scale ADF-STEM observation of sides and apexes of the hexagonal hole was performed, as shown in Fig. 2d-h. According to atomic configuration of the top MoS₂ layer inserted in each figure, the nanohole edges mostly have a zigzag configuration of atomic columns except for a few local disordered regions.

It is worth noting that the nanohole shape was modified when further etching was carried out on the nanohole. The nanohole structure was circular when the MoS₂ was dry etched with BCl₃ plasma RIE (Supplementary Fig. 5a). When the MoS₂ was dipped in potassium ferricyanide solution, the circular-shaped nanoholes became hexagonal structure (Supplementary Fig. 5b, c). The wet etching process changed not only the shape of the holes, but also the atomic configuration of nanohole edges. While MoS₂ has two

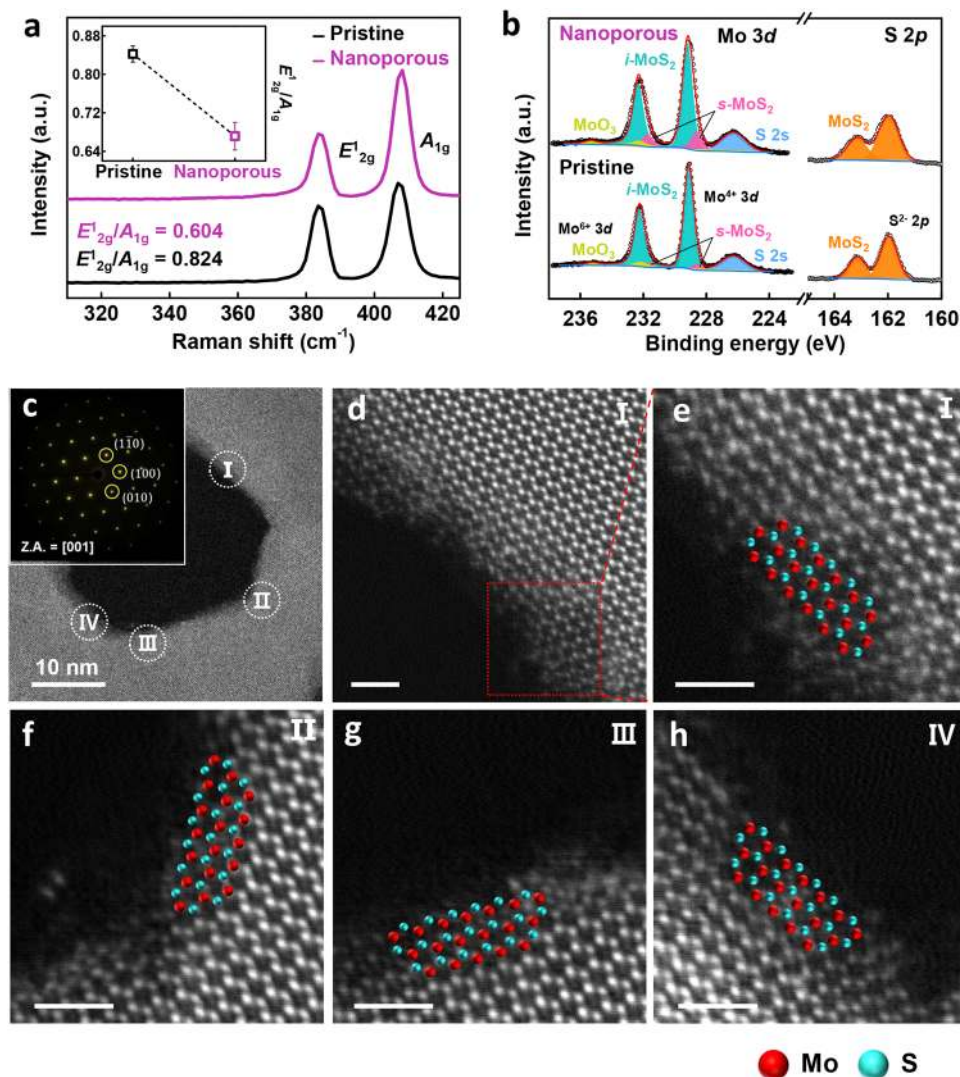


Fig. 2 Edge structure of the hexagonal nanohole in the multilayer nanoporous MoS₂. **a** Raman and **b** XPS spectra of multilayer pristine and nanoporous MoS₂, respectively. **c**, ADF-STEM image of a typical hexagonal nanohole in the nanoporous MoS₂ (inset is SADP exhibiting a typical 2H-MoS₂ structure). **d-h** Atomic-scale ADF-STEM observations of the four regions denoted by I, II, III, and IV in **c** (scale bars of 1 nm).

different edge types of zigzag and armchair^{45,46}, the zigzag configuration was found more dominant in the hexagonal nanoholes. Such transformation of edge configuration can be assigned to the higher thermodynamic stability of the zigzag configuration in comparison to that of the armchair configuration^{47,48}. It can be reasoned that the energetically less favorable armchair structure gradually changed to more favorable zigzag structure. An in-depth study of the dependence of the hole shape on the crystallographic orientation and surface energy of MoS₂ is still in progress, which is beyond the scope of this study. Given the AA' stacking configuration of the MoS₂ sample, the exposed atomic terminations at every edge are regarded as a mixture of Mo- and S-terminated zigzag structures layer by layer, which can be confirmed by the same intensity profiles of Mo *K* and S *K* peaks extracted from edge to edge across the hole (Supplementary Fig. 6c) in the STEM-EDX elemental mapping data (Supplementary Fig. 6a, b).

Electronic structural characterization of nano-patterned MoS₂.

As opposed to a pristine MoS₂, it is expected that the nanoporous MoS₂ has a different band structure due to the abundant broken

bonds in the MoS₂ lattice represented by the exposure of zigzag edges. In order to explore the electric states of the nanoporous MoS₂, PECCS measurement was conducted. The PECCS is a useful tool to probe the sub-states inside the bandgap by calculating a photoinduced threshold voltage shift (ΔV_{TH}) from phototransistors³⁶. Figure 3a, b show photo-excited transfer curves of the multilayer nanoporous MoS₂ TFTs under monochromatic light illumination with various wavelengths and photon energy dependence of ΔV_{TH} extracted from individual photo-excited transfer curves. In Fig. 3b, it was observed that ΔV_{TH} values of both multilayer devices, the pristine and nanoporous MoS₂ TFTs, gradually decreased as the photon energy of the illuminated monochromatic light increased up to the edge of the bandgap energy, and then increased rapidly above the bandgap energy. However, compared to the pristine MoS₂ TFT, it can be seen that the V_{TH} of the nanoporous MoS₂ TFT shifted significantly, implying remarkable interband charge transitions⁴⁹. The areal density of states (DOS) of the multilayer MoS₂ TFTs can be extracted by distinguishing the excited charge values from the incident photon energy from the ΔV_{TH} dependent curves. Figure 3c presents the experimentally obtained spectral DOS of both multilayer pristine MoS₂ and nanoporous MoS₂ TFTs. They

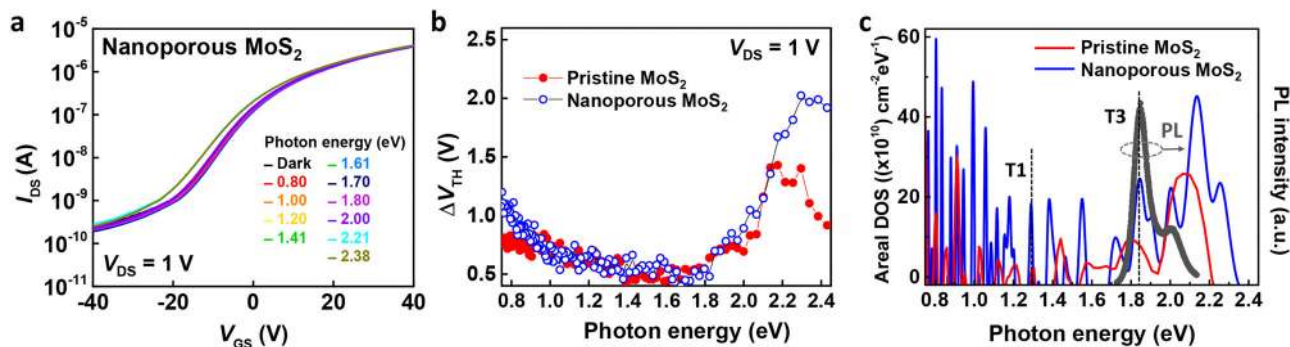


Fig. 3 Optoelectronic characteristics of MoS₂ TFTs. **a** Photo-excited transfer characteristics of the multilayer nanoporous MoS₂ TFT under monochromatic light illumination with various wavelengths. **b** Dependence of ΔV_{TH} on the incident photon energy, which was extracted from individual photo-excited transfer curves of the MoS₂ TFTs. **c** Photon energy-dependent areal DOS profiles of the multilayer pristine and nanoporous MoS₂ TFTs. The photoluminescence (PL) spectra (bold gray line) of the nanoporous MoS₂ is overlaid on the graph to compare the peak position observed by PECCS and the interband transitions measured by PL.

show the charge transition peaks corresponding to the energy states assigned as T1 (~1.30 eV, indirect bandgap) and T3 (~1.85 eV, direct bandgap)⁴⁹ matched with the photoluminescence spectra of MoS₂. More importantly, additional peaks were observed in the photon energy region lower than the bandgap energy, which correspond to the interband charge transitions. In particular, the experimentally observed interband states in the nanoporous MoS₂ were overwhelmingly more than for the pristine MoS₂, which seems to have been caused by the abundant exposure of edge atoms in nanoporous structure. Here, note that the T3 transition peaks of the pristine and nanoporous MoS₂ were not perfectly matched at the same energy due to the thickness difference between the two MoS₂ flakes, rather than the formation of nanoporous structure.

Multilayer nanoporous MoS₂ phototransistors. In general, in-gap states can directly influence the optoelectronic properties of materials. It has been reported that the in-gap states induced by interstitial atoms in multilayer molybdenum diselenide (MoSe₂) TFTs can result in significant photogating effect and high photoresponsivity under illumination⁴⁹. In this study, we explored the characteristics of the nanoporous MoS₂ phototransistors. Figure 4a shows a SEM image of a nanoporous MoS₂ TFT, in which source and drain electrodes were positioned on the nanoporous MoS₂ with a channel area of 34.28 μm^2 . The transfer characteristics of the nanoporous MoS₂ TFT were measured under the illumination of different incident power densities (P_{inc}) with light wavelength (λ_{ex}) of 405 nm and a drain voltage (V_{DS}) of 1 V (Fig. 4b). The same measurement has been carried out for a pristine multilayer MoS₂ TFT with a channel area of 56.28 μm^2 (Fig. 4c) for comparison. Although a significant increase in OFF-state current under illumination was commonly observed in both devices (due to the conductivity increase by the photoinduced excess carriers; i.e., the photoconductive effect), only the nanoporous MoS₂ phototransistor showed a negative V_{TH} shift (i.e., the photogating effect). Consequently, the photocurrent ($I_{ph} = I_{illumination} - I_{dark}$) of the nanoporous MoS₂ TFT presented a remarkable increase in the ON-state (i.e., at high gate voltages) (Supplementary Fig. 7a), unlike the pristine MoS₂ TFT (Supplementary Fig. 7b). We also extracted photoresponsivity ($R = I_{ph}/P_{inc}$) of both devices as a function of P_{inc} (Fig. 4d) (the photoresponsive characteristics of nanoporous MoS₂ under illumination of $\lambda_{ex} = 638$ nm are shown in Supplementary Fig. 8). The nanoporous MoS₂ photodetector exhibited a photoresponsivity of 622.2 A W^{-1} (at $P_{inc} = 0.8$ mW cm^{-2} and $\lambda_{ex} = 405$ nm), that is ~1240 times greater than the pristine MoS₂ phototransistor. To

confirm the reproducibility and reliability of the nanoporous MoS₂ phototransistors, we measured the photoresponsivity of five more multilayer nanoporous MoS₂ phototransistors with different channel areas. Supplementary Figure 9 shows average photoresponsivity and standard error of six nanoporous MoS₂ phototransistors, presenting consistent photoresponsivity with a very small standard error. Our result was compared with several papers that report the enhancement of photoresponsivity of the multilayer MoS₂ by applying various methods. Supplementary Figure 10 shows the relationship between photoresponsivity and P_{inc} extracted from various papers. The trend line (blue dotted line) of photoresponsivity of the nanoporous MoS₂ phototransistor is higher than most other devices except one having a heterostructure with direct bandgap monolayer MoS₂. The photodetection characteristics were also evaluated by specific detectivity ($D^* = R\sqrt{A}/2qI_{dark}$) and sensitivity ($S = I_{ph}/I_{dark}$), where A is the illuminated area and q is elementary charge. Significant increases in detectivity and sensitivity have been observed for the nanoporous MoS₂ phototransistor, as compared with the pristine MoS₂ counterpart (Fig. 4e, f). In addition, time-traced photoresponsive characteristic of the nanoporous MoS₂ phototransistor was measured under repetitive illumination with λ_{ex} of 405 nm and P_{inc} of 25 mW cm^{-2} (Supplementary Fig. 11). Rising and decay time were approximately 1.01 s and 4.02 s, respectively.

To develop further insight into the nanoporous MoS₂ phototransistor exhibiting excellent optical properties, quantum transport simulations were performed considering the in-gap states that can trap electrons or holes (see Methods section for details of the simulation approach)⁵⁰. As demonstrated by the PECCS measurements, the nanoporous MoS₂ contained multiple states within the bandgap (Fig. 3c). However, only certain states could be major contributors to the ultra-high photoresponsivity. To test this hypothesis, we simulated drain current (I_{DS}) vs gate voltage (V_{GS}) behaviors of the nanoporous MoS₂ phototransistor in the dark and under illumination ($P_{inc} = 417$ mW cm^{-2} and $\lambda_{ex} = 405$ nm), assuming a single trap state (trap concentration, $P_t = 5 \times 10^{25}$ m^{-3}) near conduction band (E_C) and valence band (E_V) in Fig. 5a, b, respectively. A notable observation from the simulated I_{DS} - V_{GS} curves is that, with the in-gap state near E_V , a significant ΔV_{TH} can be achieved (Fig. 5b), while this phenomenon did not appear with the trap state near E_C (Fig. 5a). When V_{GS} is high, the trap state near E_C is fully filled with electrons both in the dark and under illumination, due to the large number of electron carriers in E_C . In contrast, while the trap state near E_V is fully occupied by electrons in the dark, it can be partially filled by holes under illumination due to the increased hole

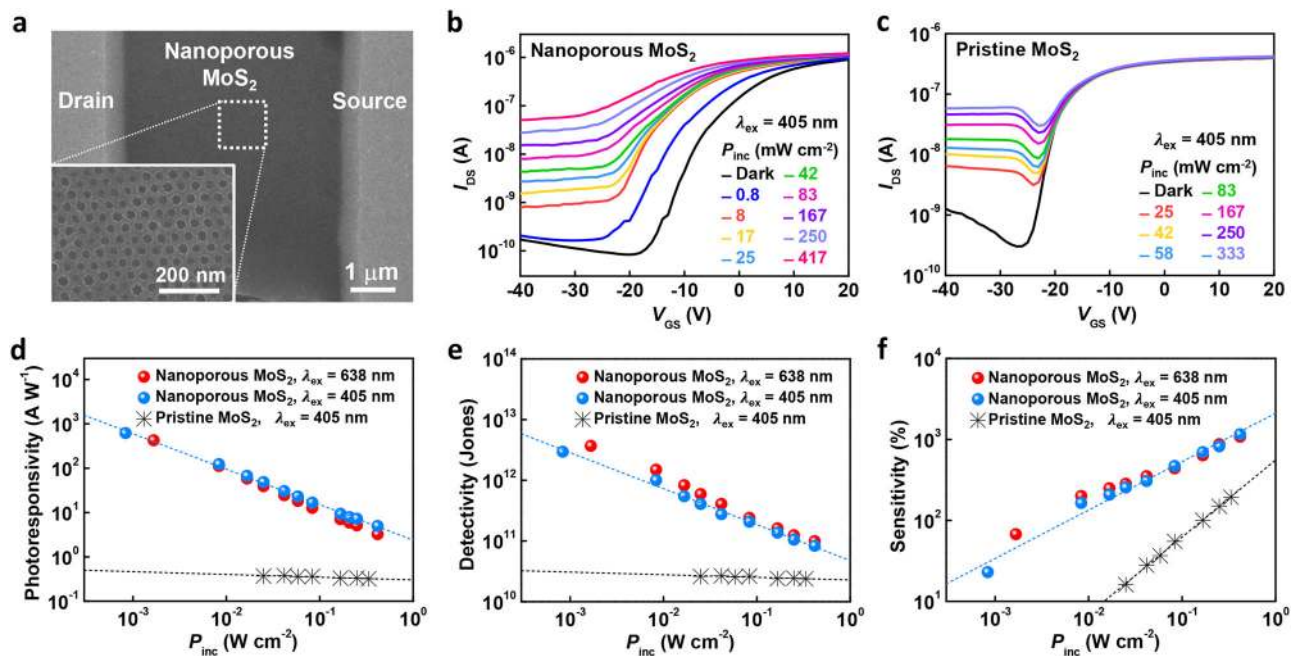


Fig. 4 Photoresponsive characteristics of the multilayer nanoporous MoS₂ TFTs. **a** SEM images of the nanoporous MoS₂ TFT. Transfer curves of **b** multilayer nanoporous MoS₂ and **c** pristine MoS₂ phototransistors in the dark and under illumination of λ_{ex} of 405 nm for various P_{inc} . **d** Calculated photoresponsivity, **e** specific detectivity, and **f** Sensitivity of the multilayer nanoporous and pristine MoS₂ phototransistors as a function of P_{inc} .

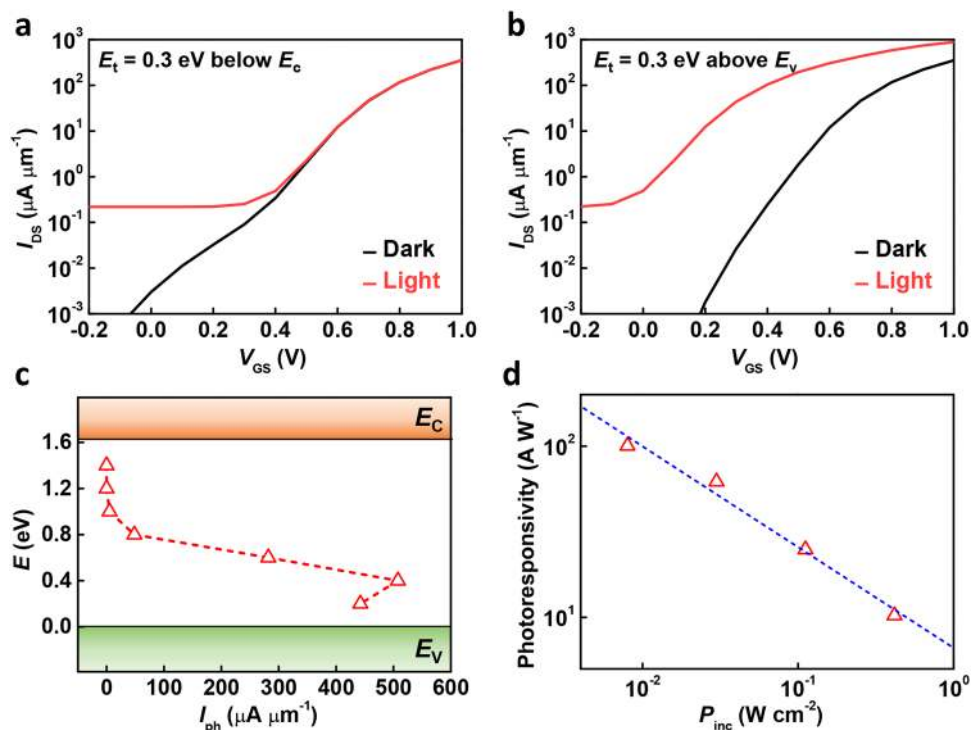


Fig. 5 Simulation of the photoresponsive behavior of the nanoporous MoS₂ phototransistor. Transfer characteristics of the nanoporous MoS₂ phototransistor assuming a single trap state (trap density of $P_t = 5 \times 10^{25} \text{ m}^{-3}$) near **a** E_c and **b** E_v in the dark and under illumination ($\lambda_{\text{ex}} = 405 \text{ nm}$, $P_{\text{inc}} = 417 \text{ mW cm}^{-2}$). **c** I_{ph} variation with a single trap state at different E_t within the bandgap (E_t from 0.2 eV to 1.4 eV above E_v) at $V_{\text{GS}} = 1 \text{ V}$. **d** Photoresponsivity as a function of P_{inc} at $\lambda_{\text{ex}} = 405 \text{ nm}$ with trap states at four different energy levels (the dashed line is a guide to the eye). $E_t = 0.3$ and 0.4 eV above (below) E_v (E_c). The total trap density is the same as in Fig. 5a-c (i.e., $1.25 \times 10^{25} \text{ m}^{-3}$).

concentration in the valence band with the generated excess holes, resulting in potential barrier lowering (i.e., the photogating effect). Our simulation results indicated that only the in-gap states near E_V can play an important role in the photogating effect, significantly increasing the electron injection from the source. It should be noted that the simulated $I_{DS}-V_{GS}$ characteristics shown in Fig. 5b manifest the same trend as the measured $I_{DS}-V_{GS}$ curves (Fig. 4b), which implies the presence of in-gap states near E_V in the nanoporous MoS₂ TFT.

To examine the impact of trap states at different energy levels (E_t), we have calculated I_{ph} by varying E_t from 0.2 eV to 1.4 eV above E_V in Fig. 5c, which reveals that I_{ph} can be significant only when the trap states are located near E_V ($E_t-E_V \leq 0.8$ eV). Finally, the photoresponsivity was plotted as a function of P_{inc} in Fig. 5d. To emulate the realistic material properties of the nanoporous MoS₂ including multiple trap states, we have included four trap states at $E_t = 0.3$ and 0.4 eV above E_V and below E_C with a trap concentration of $1.25 \times 10^{25} \text{ m}^{-3}$ each. The simulated photoresponsivity exhibits the same trend as the experimental results (Fig. 4d), indicating that our model could explain the underlying physics of the experimentally demonstrated nanoporous MoS₂ phototransistor.

Conclusion

Nano-scale patterning on multilayer MoS₂ was achieved by BCP lithography. A novel phototransistor based on the multilayer nanoporous MoS₂ channel exhibited significant enhancement in photoresponsivity, specific detectivity and photosensitivity, as compared with pristine multilayer MoS₂ phototransistors. A combination of Raman, XPS, and STEM-EDX measurements confirmed that the nanoporous structure, consisting of hexagonal arrays of hexagon holes, has abundant zigzag edges exposed with Mo- and S-termination. By means of PECCS experiment, it was shown that the nanoporous MoS₂ has high density of trap states into the bandgap region, and the device simulation revealed that the trap states existing near E_V are particularly crucial for the photogating effect and the overall optoelectronic performance of the nanoporous MoS₂ phototransistors. Hence, this work suggests using nano-patterning on multilayer MoS₂ for highly sensitive and responsive 2D material phototransistors.

Methods

Preparation of BCP nanotemplate. Sheets of multilayer MoS₂ were mechanically exfoliated from bulk, and then positioned on a Si/SiO₂ substrate. As a protective layer, SiO₂ with a thickness of 10 nm was deposited on the MoS₂ by e-beam evaporator. The entire substrate was then spin-coated with 1 wt% toluene solution of poly (styrene-*r*-methyl methacrylate) (Polymer source, $M_n = 8500$, $M_w/M_n = 1.45$) (P(S-*r*-MMA)) RCP at 3000 rpm. To stabilize the thin film of the RCP, the sample was annealed at 250 °C under vacuum condition for 2 h, followed by washing with toluene. The substrate was then spin-coated with 1 wt% toluene solution of poly (styrene-*b*-methyl methacrylate) (Polymer source, PS = 55,000, PMMA = 22,000, $M_w/M_n = 1.09$) (P(S-*b*-MMA)) BCP at 3000 rpm, and then annealed at 230 °C for 2 h. UV irradiation (VL-6.LC, Vilber lourmat) was performed for 30 min to selectively remove the PMMA from the BCP upon immersion in acetic acid for 20 min.

Fabrication of nanoporous MoS₂ TFTs. The BCP nanotemplate was used as a mask to make the nanoporous MoS₂. The O₂ plasma RIE process (50 W, 10 s, 10 sccm) was performed to etch down to the RCP layer and increased the hole size of the nanotemplate. The SF₆ plasma RIE (200 W, 15 s, 10 sccm) was utilized to imprint the nanoholes onto the SiO₂ layer. MoS₂ was then etched using BCl₃ plasma RIE (100 W, 10 sccm) to produce the nanoporous MoS₂ by varying the etching time depending on the MoS₂ thickness. Then, the MoS₂ was immersed in diluted potassium ferricyanide solution for 30 s, resulting in a change in the nanohole shape from circular to hexagonal. Finally, the SiO₂ layer was removed by immersing the device in buffered oxide etchant.

To fabricate the nanoporous MoS₂ TFTs, the pattern of source and drain electrodes was prepared using photolithography and Ti (20 nm) and Au (100 nm) were deposited. After removing unnecessary portions, the TFTs were annealed under vacuum condition at 200 °C for 2 h to enhance the electrical behaviors.

Characterizations. Raman measurement was carried out using a Micro-Raman spectrometer system (ALPHA300, WITec) with a green excitation laser. A XPS system (K-Alpha, Thermo Fisher Scientific) with monochromated Al K α radiation was used to investigate the chemical states and stoichiometry of the MoS₂ samples. The surface morphologies and atomic structures of the nanoporous MoS₂ samples were observed on an aberration-corrected STEM (JEM-ARM200CF, JEOL) in ADF imaging mode at accelerating voltages of 80 and 200 kV, respectively. The angle range of the ADF detector and probe-forming convergence angle were 45–180 and 23 mrad, respectively. Wiener filtering technique inserted in commercial software was used to decrease random noise background in the obtained atomic-scale ADF-STEM images (HREM Filter Pro, HREM research). Elemental mapping of the nanoporous MoS₂ was carried out by EDX (JED-2300T, JEOL) with a dual-type Si drift detector operating in the same ADF-STEM imaging mode. Each detector had an effective X-ray sensing area of 100 mm², providing a total collection efficiency of ~10% of the total generated X-ray signals (4 π sr). A semiconductor measurement system (4200-SCS, Keithley) equipped with a dark box was used for the electrical measurements. To measure the optoelectrical characteristics of the devices, laser radiation was perpendicularly incident on the MoS₂ channel area (MCLS1, Thorlabs) using single mode fiber optic patch cables (S405-HP for 405 nm and SM600 for 638 nm, Thorlabs).

Optoelectronic characterization. For PECCS measurements, the photo-excited transistor characterizations were carried out using an intense monochromatic light. The photon flux of almost $5 \times 10^{14} \text{ cm}^{-2} \text{ s}^{-1}$ was estimated from the optical power density of ~0.1 mW cm⁻². The PECCS system consists of a grating monochromator providing a spectral wavelength range from 300 to 1400 nm, a Hg (Xe) light source of 500 W, and an optical fiber that delivers light to the reactive MoS₂ surface of the devices. Electrical measurements were performed by electrical analyzer unit (4155C, Agilent Technologies). Photoluminescence measurements were performed by a spectrometer system (ALPHA300, WITec) with a 532 nm excitation laser.

Device simulation. Carrier transport through the nanoporous MoS₂ phototransistor was simulated using the non-equilibrium Green's function (NEGF) formalism within an effective mass approximation ($m^* = 0.45 m_0$, where m_0 is the free electron mass). For the nominal device, a 50-nm-long, 6.2-Å-thick MoS₂ channel and a 2.5-nm-thick SiO₂ gate dielectric were used in a bottom-gate Schottky barrier (SB) field-effect transistor (FET) structure with an SB height of 0.1 eV. A power supply voltage of $V_{DD} = 0.5$ V was applied. Here, we used smaller device dimensions to save simulation time; however, this will not affect the key underlying physics, and the conclusion from the simulation study will remain the same. The transport equation was iteratively solved with Poisson's equation until a self-consistent solution was achieved. Here, we first performed a ballistic transport simulation assuming ideal MoS₂ to identify the underlying physics of the photogating effect (Fig. 5a-c), and then the current degradation due to scattering in the actual nanoporous structure was taken into account by calibrating the simulation results against the experiments (Fig. 5d) using a fitting parameter (degradation factor of 8×10^{-6}).

Trap model. A modified Hornbeck-Haynes model of $\Delta\sigma = q\mu_n\Delta n_{PC} + q\mu_p\Delta p_{PC}$ was used to calculate the increase in conductivity regarding to illumination (photoconductive (PC) effect), where μ_n and μ_p are the mobilities of electrons and holes, and Δn_{PC} and Δp_{PC} are electron and hole excess densities in the conduction and the valence band, respectively. Measured carrier mobility of $2.68 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ was used for both μ_n and μ_p . Considering charge neutrality, $\Delta n_{PC} = g\tau_{r,PC}$ and $\Delta p_{PC} = \Delta n_{PC} + n_{t,PC}$ were used with the trap state near E_C , while $\Delta p_{PC} = g\tau_{r,PC}$ and $\Delta n_{PC} = \Delta p_{PC} + p_{t,PC}$ with the trap state near E_V , assuming that the fermi level is located at mid-gap⁵¹. $n_{t,PC}$ and $p_{t,PC}$ are the trapped electron and hole densities, which can be calculated as $n_{t,PC} = \frac{P_t dg\tau_{r,PC}}{g\tau_{r,PC} + P_t d\tau_{tn}/\tau_{gn}}$ and $p_{t,PC} = \frac{P_t dg\tau_{r,PC}}{g\tau_{r,PC} + P_t d\tau_{tp}/\tau_{gp}}$, where P_t is the total trap concentration⁵¹. g ($= \eta P_{abs}/h\nu$) is the generation rate of excess carriers, d is the thickness of the nanoporous MoS₂, and $\tau_{r,PC}$ is the carrier recombination lifetime for the photoconductive effect. η and $h\nu$ are internal quantum efficiency and single photon energy, respectively. The value of η was assumed to be 0.5 for λ_{ex} of 405 nm. The absorbed power density (P_{abs}) is calculated by $P_{inc} \left(1 - \frac{e^{-a^{\perp}d} + e^{-a^{\parallel}d}}{2}\right)$, where a^{\perp} and a^{\parallel} are the absorption coefficients in the vertical ($a^{\perp} = 18.9 \times 10^4 \text{ cm}^{-1}$ at $\lambda_{ex} = 405 \text{ nm}$)⁵² and lateral ($a^{\parallel} = 102.5 \times 10^4 \text{ cm}^{-1}$ at $\lambda_{ex} = 405 \text{ nm}$) directions⁵³, respectively. $\tau_{t,n/p}$ and $\tau_{g,n/p}$ are the trapping and detrapping times of electrons/holes, respectively, and the $\frac{\tau_{tn}}{\tau_{gn}}$ and $\frac{\tau_{tp}}{\tau_{gp}}$ ratios can be obtained by $1/\left(\frac{P_t}{N_C} \exp \frac{\Delta E_{t,C}}{kT}\right)$ and $1/\left(\frac{P_t}{N_V} \exp \frac{\Delta E_{t,V}}{kT}\right)$ ⁵¹, where N_C and N_V are effective density of states at E_C and E_V , respectively. $\Delta E_{t,C}$ and $\Delta E_{t,V}$ are defined as $E_C - E_t$ and $E_t - E_V$, respectively. The recombination lifetime ($\tau_{r,PC} = 100$ ps) was adopted from a previous report⁵⁴. Finally, photoconductive current ($I_{ph,PC}$) was obtained by $I_{ph,PC} = \frac{1}{L} V_{DS} \Delta\sigma$. The photogating (PG) effect due to barrier lowering can be simulated by considering trapped holes (i.e., total trap concentration - electrons in the trap state) as $p_{t,PG} = P_t(1 - f)$, where f is the probability of electron occupation at a given trap state, which can be calculated by $f = \frac{v_{th}\sigma_n n_C + e_p}{v_{th}\sigma_n n_C + e_n + v_{th}\sigma_p p_V + e_p}$ ⁵⁴. n_C and p_V are concentration of electrons in the conduction band and concentration of holes in the valence band, respectively, and v_{th} is thermal velocity. e_n and e_p are the probabilities of

electron and hole emission from the trap state, respectively, determined by $e_n = v_{th} \sigma_n n_0 \exp\left\{\frac{E_C - E_{F0}}{kT}\right\}$ and $e_p = v_{th} \sigma_p p_0 \exp\left\{\frac{E_{F0} - E_C}{kT}\right\}$, where σ_n and σ_p are capture cross-sections for electrons and holes, respectively⁵⁴. n_0 and p_0 are the concentration of free electrons in the conduction band and the concentration of free holes in the valence band, respectively, under equilibrium, which are defined as $n_0 = N_C \ln(1 + \exp\{(E_{F0} - E_C)/kT\})$ and $p_0 = N_V \ln(1 + \exp\{(E_V - E_{F0})/kT\})$, where E_{F0} is the fermi level under thermal equilibrium assumed to be at mid-gap in this study. For n-type transistors, the electron concentration (n_C) is obtained by $n_C = n_{inj} + \Delta n_{ph}$, where n_{inj} is the injected electron concentration obtained from the NEGF simulation. Under illumination, the excess electron concentration becomes a non-zero value, which can be modeled as $\Delta n_{ph} = g \tau_{r,PG}/d$, where the recombination lifetime for the photogating effect ($\tau_{r,PG}$) was assumed to be 1 μ s. The hole concentration (p_V) is mainly determined by the excess hole concentration (Δp_{ph}), which is the same as Δn_{ph} . For a given trap state, f is a strong function of $\frac{e_n}{\sigma_n}$, which was assumed to be a logistic function of energy in a logarithmic scale with maximum and minimum values of 10^3 and 10^{-3} at E_C and E_V edges, respectively. Since the trap sites are negatively charged in the presence of electrons, the concentration of trapped electrons ($n_{t,PG} = P_t - p_{t,PG}$) was considered in Poisson's equation as follows:

$$\nabla^2 V = -\frac{q}{\epsilon} \{N_D - N_A + p_V - (n_C + n_{t,PG})\}.$$

Data availability

The data presented in this study are available from the corresponding author upon reasonable request.

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Author contributions

S.K. and Y.Y. conceived the project. H.P., Y.J.K. and S.K. fabricated the nanoporous MoS₂ structure using BCP lithography and measured the electrical properties of the nanoporous MoS₂ phototransistors. J.L. and H.P. performed PECCS measurements on nanoporous MoS₂ phototransistors and analyzed optoelectronic characterization of MoS₂ nanosheet. G.H., A.A.M., J.L. and Y.Y. performed theoretical simulations of the nanoporous MoS₂ phototransistor behavior. Y.-H.K. and Y.-M.K. contributed to structural and chemical characterizations of the nanoporous MoS₂ using STEM. All authors wrote and contributed to the manuscript.

Competing interests

S.K. is an editorial board member for Communications Materials and was not involved in the editorial review, or the decision to publish, this article. All authors declare that there are no competing interests.

Additional information

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