# Nanomolar Antimalarial Agents against Chloroquine-Resistant Plasmodium falciparum from Medicinal Plants and Their Structure-Activity Relationships 

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#### Abstract

Inspired by the discovery of the antimalarial drug artemisinin from a traditional Chinese medicine (TCM), a natural product library of 44 lindenane-type sesquiterpenoids was assessed for activities against the Dd2 chloroquine-resistant strain of the malaria parasite Plasmodium falciparum. These compounds were mainly isolated from plants of the Chloranthus genus, many species of which are named "Sikuaiwa" in TCM and have long been used to treat malaria. The compounds consisted of 41 sesquiterpenoid dimers and three monomers, including the twelve new dimers $\mathbf{1}-\mathbf{1 2}$ isolated from C. fortunei. The results showed that 16 dimers exhibited potent antiplasmodial activities ( $<100 \mathrm{nM}$ ); in particular, compounds $\mathbf{1 , 1 4}$ and $\mathbf{1 9}$ exhibited low nanomolar activities with $\mathrm{IC}_{50}$ values ranging from 1 to 7 nM , which is comparable to the potency of artemisinin, and selectivity index values toward mammalian cells greater than 500 . A comprehensive structure-activity relationship (SAR) study clearly indicated that three functional groups are essential and two motifs can be modified.


## Graphical Abstract

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## Keywords

Antimalarial activity; Structure-activity relationship; Natural products; Lindenane-type sesquiterpenoids

Malaria is a global parasitic infectious disease caused by Plasmodium parasites, which are transmitted by the bite of female anopheles mosquitoes. The disease is widespread in tropical and subtropical regions, with the majority of deaths occurring in Africa and caused by $P$. falciparum, resulting in losses of US $\$ 12$ billion a year. ${ }^{1}$ Approximately 3.2 billion people remain at risk of contracting malaria, leading to an estimated 214 million malaria cases, and 438,000 deaths in 2014 alone. ${ }^{2}$ Drug resistance also poses a growing problem for malaria treatment, including the gold standard artemisinin combination therapies. ${ }^{3}$ Therefore, the development of new antimalarial agents to expand the repertoire of drugs suitable for use in combination therapies remains a priority to retain effective malaria control and elimination.

The Chloranthaceae family has three genera with a total of 16 species, including Chloranthus (thirteen sp.), Sarcandra (two sp.), and Hedyosmum (one sp.), which are mainly distributed in the southern area of China. ${ }^{4}$ Our ongoing studies on plants of the former two genera in the Chloranthaceae family have led to the isolation of a large number of sesquiterpenoids and sesquiterpenoid dimers with diverse bioactivities. ${ }^{5}$

Natural products have played a very important role in antimalarial drug development, and have led to the well-known antimalarial drugs quinine and artemisinin. ${ }^{6}$ The potent antimalarial artemisinin was developed from a traditional Chinese medicine, the aerial parts of Artemisia annua dubbed "Qinghao", which was used to treat malaria as early as 340 CE, as documented in the ancient Chinese pharmacopeia "Principal Prescription Emergency" in China's Eastern Jin Dynasty ${ }^{7}$ Interestingly, many Chloranthus species, called "Sikuaiwa" in TCM, were also used for the same purpose. ${ }^{8}$ A small library of structurally related compounds isolated from five Chloranthus species and one Sarcandra specie was thus subjected to antiplasmodial bioassay against the chloroquine-resistant Dd 2 strain of $P$. falciparum. These compounds included three lindenane-type sesquiterpenoid monomers and 41 dimers (Figures 1-3). Compounds $\mathbf{1} \mathbf{- 1 2}$ are new dimeric sesquiterpenoids isolated from the twigs of $C$. fortunei and reported in this study. The antiplasmodial screening revealed that several of the sesquiterpenoid dimers exhibited very potent antiplasmodial activities, and three compounds had potent activities similar to that of artemisinin. These results open
the way to further studies on this new class of promising antiplasmodial compounds that may act through a new mechanism of action because of their unique structural architectures. We herein present the isolation and structure elucidation of the new compounds, as well as the antiplasmodial activities and structure-activity relationships of all 44 compounds.

## RESULTS AND DISCUSSION

Compound $\mathbf{1}$ had a molecular formula of $\mathrm{C}_{41} \mathrm{H}_{46} \mathrm{O}_{14}$ as determined by the sodiated (+)HRESIMS ion at $m / z 785.2781[\mathrm{M}+\mathrm{Na}]^{+}$(calcd 785.2780) and ${ }^{13} \mathrm{C}$ NMR data. Analysis of the NMR data (Tables 1 and 4) suggested that $\mathbf{1}$ is a lindenane-type sesquiterpenoid dimer with the distinct features of two 1,2 -substituted cyclopropane rings, one $a, \beta$-unsaturated $-\gamma$ lactone, and a persubstituted double bond. Comprehensive analysis of the 1D and 2D NMR data further revealed that its structure is similar to that of chlorajaponol $\mathrm{G}^{9}$ except for the presence of a $\gamma$-formylsenecioate motif instead of the angeloate moiety of the latter. This conclusion was verified by HMBC correlations of $\mathrm{H}-4^{\prime \prime}\left(\delta_{\mathrm{H}} 9.67\right) / \mathrm{C}-2^{\prime \prime}$ and $\mathrm{C}-5^{\prime \prime}$ within the motif that was attached to $\mathrm{C}-15^{\prime}$ by the key HMBC correlation from $\mathrm{H}_{2}-15^{\prime}$ to $\mathrm{C}-1^{\prime \prime}$ ( $\delta_{\mathrm{C}} 165.5$ ) (Figure 4A). Additionally, a methyl succinoate group was attached to $\mathrm{C}-13^{\prime}$ as indicated by the key HMBC correlation of $\mathrm{H}_{2}-13^{\prime} / \mathrm{C}-9^{\prime \prime}\left(\delta_{\mathrm{C}} 172.1\right)$. The relative configuration of 1 was assigned primarily by its ROESY data (Figure 4B), in which the correlations of $\mathrm{H}-1 / \mathrm{H}-2 a, \mathrm{H}-1 / \mathrm{H}-3, \mathrm{H}-1 / \mathrm{H}-9, \mathrm{H}-1^{\prime} / \mathrm{H}-2^{\prime} a, \mathrm{H}-2^{\prime} a / \mathrm{H}-3^{\prime}, \mathrm{H}-3^{\prime} / \mathrm{H}-15^{\prime}$, and $\mathrm{H}-5^{\prime} / \mathrm{H}-15^{\prime}$ indicated that they are spatially close and assigned arbitrarily as $a$-oriented. Accordingly, the ROESY correlations of $\mathrm{H}-2 \beta / \mathrm{H}_{3}-14, \mathrm{H}_{3}-14 / \mathrm{H}-6, \mathrm{H}-2^{\prime} \beta / \mathrm{H}_{3}-14^{\prime}$, and $\mathrm{H}_{3}-14^{\prime} / \mathrm{H}-9^{\prime}$ revealed that they were $\beta$-oriented. The $\Delta^{2^{\prime \prime}}$ double bond was assigned $E$ by the ROESY correlation of $\mathrm{H}_{2}-4^{\prime \prime} / \mathrm{H}-2^{\prime \prime}$. Thus, the structure of fortunilide A (1) was depicted as shown.

The molecular formula of compound 2 was established as $\mathrm{C}_{41} \mathrm{H}_{48} \mathrm{O}_{14}$, by the sodium adduct (+)-HRESIMS ion at $m / z 787.2964[\mathrm{M}+\mathrm{Na}]^{+}$(calcd 787.2936) and the ${ }^{13} \mathrm{C}$ NMR data. The NMR data (Tables 1 and 4) of $\mathbf{2}$ showed many similarities to those of shizukaol F, ${ }^{10}$ suggesting that they are structural analogues. Comprehensive analysis of the spectroscopic data of 2 revealed that it shared the same dimeric sesquiterpenoid core with shizukaol F and the structural differences were the substituent groups at $\mathrm{C}-13^{\prime}$ and $\mathrm{C}-15^{\prime}$. Compound $\mathbf{2}$ was assigned as a methanolysis product of shizukaol F with the cleavage between $\mathrm{C}-13^{\prime}$ and $\mathrm{C}-9^{\prime \prime}$ in the 18-membered macrocyclic trilactone of the latter. This conclusion was confirmed by the HMBC correlations (Figure S13, Supporting Information) from H-4" ( $\delta_{\mathrm{H}}$ 4.62) to $\mathrm{C}-2^{\prime \prime}\left(\delta_{\mathrm{C}} 114.8\right)$ and $\mathrm{C}-6^{\prime \prime}\left(\delta_{\mathrm{C}} 171.8\right)$, from both $\mathrm{H}-7^{\prime \prime}\left(\delta_{\mathrm{H}} 2.67\right)$ and $\mathrm{H}-8^{\prime \prime}$ ( $\delta_{\mathrm{H}}$ 2.73 ) to $\mathrm{C}-6^{\prime \prime}$ and $\mathrm{C}-9$ " $\left(\delta_{\mathrm{C}} 173.0\right)$, and from $\mathrm{OCH}_{3}\left(\delta_{\mathrm{H}} 3.67\right)$ to $\mathrm{C}-9^{\prime \prime}$. The relative configuration of $\mathbf{2}$ was established by ROESY data (Figure S14, Supporting Information), in particular, the $\Delta^{2^{\prime \prime}}$ double bond was assigned as $E$-geometry by the key ROESY interaction between $\mathrm{H}-4^{\prime \prime}$ and $\mathrm{H}-2^{\prime \prime}$. Hence, the structure of $\mathbf{2}$, fortunilide B , was unequivocally characterized.

Compound 3 had the molecular formula $\mathrm{C}_{36} \mathrm{H}_{42} \mathrm{O}_{10}$ based on its ${ }^{13} \mathrm{C}$ NMR data and sodium adduct (+)-HRESIMS ion at $m / z 657.2672[\mathrm{M}+\mathrm{Na}]^{+}$(calcd 657.2670). Analysis of its NMR data (Tables 1 and 4) revealed that its structure is closely related to that of shizukaol $\mathrm{K}^{11}$ with the differences being the substituents at $\mathrm{C}-13^{\prime}$ and $\mathrm{C}-15^{\prime}$. The presence of a
senecioate moiety was identified by HMBC correlations within this motif, and its attachment to $\mathrm{C}-15^{\prime}$ was indicated by the key HMBC cross-peak from $\mathrm{H}_{2}-15^{\prime}$ ( $\delta_{\mathrm{H}} 4.83$ and 4.10 ) to $\mathrm{C}-1^{\prime \prime}$ ( $\delta_{\mathrm{C}} 166.8$ ) (Figure S 22 , Supporting Information). A hydroxy group was located at C-13' by the chemical shifts of $\mathrm{H}-13^{\prime}\left(\delta_{\mathrm{H}} 4.41\right.$ and 4.34 , each $1 \mathrm{H}, \mathrm{d}, J=13.6 \mathrm{~Hz}$ ) and $\mathrm{C}-13^{\prime}\left(\delta_{\mathrm{C}} 54.9\right)$. A ROESY experiment (Figure S23, Supporting Information) showed that the relative configuration of $\mathbf{3}$ was identical to that of shizukaol K . The structure of compound $\mathbf{3}$, fortunilide C , was thus established as shown.

The molecular formula, $\mathrm{C}_{41} \mathrm{H}_{48} \mathrm{O}_{16}$, of compound 4 was determined by its sodium adduct (+)-HRESIMS ion at $m / z 819.2851[\mathrm{M}+\mathrm{Na}]^{+}$(calcd 819.2835). Detailed analysis of its NMR data (Tables 1 and 4) revealed it to be a methanolysis product of spicachlorantin $D^{12}$ with the cleavage between $\mathrm{C}-13^{\prime}$ and $\mathrm{C}-9^{\prime \prime}$ in the 18 -membered macrocyclic trilactone of the latter. This conclusion was confirmed by the HMBC correlations (Figure S31, Supporting Information) from $\mathrm{H}-4^{\prime \prime}\left(\delta_{\mathrm{H}} 4.66\right)$ to $\mathrm{C}-2^{\prime \prime}\left(\delta_{\mathrm{C}} 115.2\right)$ and $\mathrm{C}-6^{\prime \prime}\left(\delta_{\mathrm{C}} 171.9\right)$, from both $\mathrm{H}-7^{\prime \prime}\left(\delta_{\mathrm{H}} 2.67\right)$ and $\mathrm{H}-8^{\prime}\left(\delta_{\mathrm{H}} 2.73\right)$ to $\mathrm{C}-6^{\prime \prime}$ and $\mathrm{C}-9^{\prime \prime}$ ( $\delta_{\mathrm{C}} 173.0$ ), and from $\mathrm{OCH}_{3}\left(\delta_{\mathrm{H}} 3.69\right)$ to $\mathrm{C}-9$ ". In addition, a characteristic singlet signal at $\delta_{\mathrm{H}} 8.64(\mathrm{OOH})$ and a quaternary carbon resonance at $\delta_{\mathrm{C}} 90.8$ indicated the presence of a hydroperoxy group at $\mathrm{C}-4 .{ }^{12}$ The relative configuration of 4 was identical to that of spicachlorantin D as indicated by its ROESY data (Figure S32, Supporting Information). The structure of 4, fortunilide D, was thereby determined as shown.

Compound 5 had the same molecular formula $\mathrm{C}_{41} \mathrm{H}_{48} \mathrm{O}_{16}$ as $\mathbf{4}$, as determined by the (+)HRESIMS ion at $m / z 819.2842[\mathrm{M}+\mathrm{Na}]^{+}$(calcd 819.2835) and ${ }^{13} \mathrm{C}$ NMR data. Analysis of the NMR data (Tables 2 and 4) revealed that $\mathbf{4}$ and 5 were isomers, with differing substituents at $\mathrm{C}-13^{\prime}$ and $\mathrm{C}-15^{\prime}$. A methyl succinoate and a $\gamma$-hydroxysenecioate moieties were attached to $\mathrm{C}-13^{\prime}$ and $\mathrm{C}-15^{\prime}$ of $\mathbf{5}$ as shown by the key HMBC correlations of $\mathrm{H}_{2}-13^{\prime}$ / $\mathrm{C}-9^{\prime \prime}$ and $\mathrm{H}_{2}-15^{\prime} / \mathrm{C}-1^{\prime \prime}$ (Figure S 40 , Supporting Information), respectively. The $\Delta^{2^{\prime \prime}}$ double bond was assigned as $E$ by the ROESY interaction between $\mathrm{H}_{2}-4^{\prime \prime}$ and $\mathrm{H}-2^{\prime \prime}$ (Figure S 41 , Supporting Information). Thus, the structure of fortunilide E (5) was unequivocally characterized as shown.

Compound 6 had the molecular formula $\mathrm{C}_{31} \mathrm{H}_{36} \mathrm{O}_{10}$, based on its (+)-HRESIMS ion at $\mathrm{m} / \mathrm{z}$ $591.2211[\mathrm{M}+\mathrm{Na}]^{+}$(calcd 591.2201) and ${ }^{13} \mathrm{C}$ NMR data. Comparison of its NMR data (Tables 2 and 4) with those of spicachlorantin $\mathrm{J}^{13}$ revealed that their structures are closely related and both of them possessed a hydroperoxy moiety at C-4. The only variation was the presence of a hydroxymethyl group at $\mathrm{C}-4^{\prime}$ in 6 instead of the exocyclic $\Delta^{4^{\prime}\left(15^{\prime}\right)}$ double bond in the latter, as supported by the HMBC correlations from $\mathrm{H}_{2}-15^{\prime}\left(\delta_{\mathrm{H}} 3.57\right.$ and 3.35) to $\mathrm{C}-3^{\prime}, \mathrm{C}-4^{\prime}$, and $\mathrm{C}-5^{\prime}$ (Figure S49, Supporting Information). The structure of $\mathbf{6}$, fortunilide F, was finally confirmed by the HMBC and NOESY spectra (Figure S50, Supporting Information).

Compound 7 had the molecular formula $\mathrm{C}_{40} \mathrm{H}_{44} \mathrm{O}_{13}$ as determined by the mass of the sodiated molecular ion at $\mathrm{m} / \mathrm{z} 755.2683[\mathrm{M}+\mathrm{Na}]^{+}$(calcd 755.2674) in the (+)-HRESIMS and the ${ }^{13} \mathrm{C}$ NMR data. The NMR data (Tables 2 and 4) of 7 showed strong similarities to those of henriol $\mathrm{C}^{14}$ except for the changes regarding the location of $\mathrm{CH}_{3}-5^{\prime \prime}$, which was attached to $\mathrm{C}-3^{\prime \prime}$ in the lactone bridge of 7 instead of $\mathrm{C}-2^{\prime \prime}$ in that of henriol C , as deduced
by the key HMBC correlation from H-5" to C-4". The chemical shift of $\mathrm{H}_{2}-4^{\prime \prime}$ ( $\delta_{\mathrm{H}} 4.71$ ) suggested an $E$-geometry ${ }^{15}$ for the $\Delta^{2^{\prime \prime}}$ double bond in the macrolide motif. The structure of 7, fortunilide G, was further verified by HMBC and ROESY spectra (Figures S58 and S59, Supporting Information).

Compound 8, named fortunilide H , was assigned the molecular formula $\mathrm{C}_{40} \mathrm{H}_{44} \mathrm{O}_{14}$ on the basis of its (+)-HRESIMS ion at $m / z 771.2625$ (calcd 771.2623) and its ${ }^{13} \mathrm{C}$ NMR data. Analysis of the NMR data (Tables 2 and 4) showed many similarities with those of chloramultiol $\mathrm{F},{ }^{16}$ with a large conjugated system of two additional double bonds and one $a, \beta$-unsaturated- $\gamma$-lactone. The main differences were the presence of methyl succinoate and $\gamma$-hydroxysenecioate residues at $\mathrm{C}-13^{\prime}$ and $\mathrm{C}-15^{\prime}$, respectively, instead of the 18 membered macrocyclic trilactone ring of chloramultiol F . The locations of these groups were assigned by the HMBC correlations of $\mathrm{H}_{2}-13^{\prime} / \mathrm{C}-9^{\prime \prime}$ and $\mathrm{H}_{2}-15^{\prime} / \mathrm{C}-1^{\prime \prime}$ (Figure S67, Supporting Information). In addition, a hydroxy group was placed at C-8 by the chemical shift ( $\delta_{\mathrm{C}} 104.7$ ) instead of the methoxy group in chloramultiol F , which was assigned as $a$ oriented by the pyridine-induced solvent shifts for $\mathrm{H}-5^{\prime}(\Delta \delta=-0.63) .{ }^{17}$ The $\Delta^{2^{\prime \prime}}$ double bond was assigned as E-geometry by the key ROESY correlation of $\mathrm{H}_{2}-4^{\prime \prime} / \mathrm{C}-2^{\prime \prime}$ (Figure S68, Supporting Information). The structure of $\mathbf{8}$ was further verified by 2D NMR spectra.

Compound 9 had the molecular formula $\mathrm{C}_{40} \mathrm{H}_{46} \mathrm{O}_{16}$, as determined by its (+)-HRESIMS ion peak at $m / z 805.2659[\mathrm{M}+\mathrm{Na}]^{+}$(calcd 805.2678) and ${ }^{13} \mathrm{C}$ NMR data. The NMR data (Tables 3 and 4) of $\mathbf{9}$ showed many similarities to those of $\mathbf{8}$, suggestive of structural analogues. In particular, an oxygenated quaternary carbon signal ( $\delta_{\mathrm{C}} 89.9, \mathrm{C}-4$ ) and a methylene carbon signal ( $\delta_{\mathrm{C}} 36.1, \mathrm{C}-15$ ) were observed for 9 in place of the carbon signals of an $\Delta^{4(15)}$ double bond for 8 , indicating the presence of a hydroperoxy moiety at C-4. Two groups of a methyl succinoate and a $\gamma$-hydroxysenecioate were assigned to $\mathrm{C}-13^{\prime}$ and $\mathrm{C}-15^{\prime}$ respectively by the HMBC correlations of $\mathrm{H}_{2}-13^{\prime} / \mathrm{C}-9^{\prime \prime}$ and $\mathrm{H}_{2}-15^{\prime} / \mathrm{C}-1^{\prime \prime}$ (Figure S76, Supporting Information). $8-\mathrm{OH}$ was assigned as $a$-oriented by the pyridine-induced solvent shifts for $\mathrm{H}-5^{\prime} \quad(\Delta \delta=-0.74) .{ }^{17}$ The structure of fortunilide I (9) finally confirmed by HMBC and ROESY data (Figure S77, Supporting Information), was elucidated as shown.

Compound $\mathbf{1 0}$, named fortunilide J , was obtained as a white amorphous powder. A molecular formula of $\mathrm{C}_{35} \mathrm{H}_{40} \mathrm{O}_{11}$ was assigned for $\mathbf{1 0}$ with 16 DBEs on the basis of its (-)HRESIMS ion at $m / z 635.2475[\mathrm{M}-\mathrm{H}]^{-}$(calcd 635.2498) as well as its ${ }^{13} \mathrm{C}$ NMR data. Analysis of the NMR data (Tables 3 and 4) suggested that 10 is a lindenane-type sesquiterpenoid dimer with the distinct features of two 1,2-substituted propane rings, two $a, \beta$-unsaturated- $\gamma$-lactone groups, a trisubstituted double bond and a $\gamma$-hydroxysenecioate residue. Comprehensive investigation of the spectroscopic data including 2D NMR revealed that it is structurally related to henriol $B,{ }^{14}$ and the differences were the presence of a $\Delta^{4(15)}$ double bond instead of the $\Delta^{5}$ double bond of the latter, and the $\gamma$-hydroxysenecioate residue in place of the senecioate residue of henriol B. The different position of the double bond was confirmed by the key HMBC correlations (Figure 5 A ) from $\mathrm{H}-15\left(\delta_{\mathrm{H}} 5.82\right)$ to $\mathrm{C}-3$ and $\mathrm{C}-8^{\prime}$ ( $\delta_{\mathrm{C}} 92.0$ ), and from $\mathrm{H}-6$ to $\mathrm{C}-4\left(\delta_{\mathrm{C}} 146.1\right)$. In addition, the $\gamma$-hydroxysenecioate group was attached to $\mathrm{C}-15^{\prime}$ by the HMBC correlations of $\mathrm{H}_{2}-15^{\prime} / \mathrm{C}-1^{\prime \prime}\left(\delta_{\mathrm{C}} 168.3\right)$. The relative configuration of $\mathbf{1 0}$ was verified mainly by its ROESY data (Figure 5B), in which ROESY correlations of $\mathrm{H}-1 / \mathrm{H}-2 a, \mathrm{H}-1 / \mathrm{H}-3, \mathrm{H}-1 / \mathrm{H}-9, \mathrm{H}^{\prime} 1^{\prime} / \mathrm{H}-2^{\prime} a, \mathrm{H}-2^{\prime} a / \mathrm{H}-3^{\prime}, \mathrm{H}-3^{\prime} / \mathrm{H}-15^{\prime}$, and
$\mathrm{H}-5^{\prime} / \mathrm{H}-15^{\prime}$ showed that these protons were co-facial and were assigned arbitrarily as $a$ oriented. In consequence, the ROESY correlations of $\mathrm{H}-2 \beta / \mathrm{H}_{3}-14, \mathrm{H}_{3}-14 / \mathrm{H}-5, \mathrm{H}_{3}-14 / \mathrm{H}-6$, $\mathrm{H}-2^{\prime} \beta / \mathrm{H}_{3}-14^{\prime}$, and $\mathrm{H}_{3}-14^{\prime} / \mathrm{H}-9^{\prime}$ indicated that they were $\beta$-oriented The $\Delta^{2^{\prime \prime}}$ double bond was assigned as $E$ by the ROESY correlation of $\mathrm{H}_{2}-4^{\prime \prime} / \mathrm{H}-2^{\prime \prime}$. The $8-\mathrm{OH}$ was assigned as $a$ by the significant pyridine-induced solvent shift ${ }^{17}$ for $\mathrm{H}-5^{\prime}(\Delta \delta=-0.81)$.

Compound 11 was obtained as a colorless gum and was assigned the molecular formula $\mathrm{C}_{36} \mathrm{H}_{40} \mathrm{O}_{9}$ by the (-)-HRESIMS ion peak at $\mathrm{m} / \mathrm{z} 615.2588[\mathrm{M}-\mathrm{H}]^{-}$(calcd for 615.2600 ), requiring 17 double bond equivalents (DBE). The IR spectrum revealed the presence of hydroxy ( $3455 \mathrm{~cm}^{-1}$ ), carbonyl ( $1761 \mathrm{~cm}^{-1}$ ), and olefinic ( $1656 \mathrm{~cm}^{-1}$ ) functionalities. Initial analysis of the NMR data (Tables 3 and 4) of $\mathbf{1 1}$ suggested it was a lindenane-type sesquiterpenoid dimer possessing the rare sarcanolide A skeleton. ${ }^{18}$ Comprehensive analysis of the 1 D and 2D NMR spectra of $\mathbf{1 1}$ further revealed that the only structural difference between it and sarcanolide A was the existence of $\Delta^{4}$ double bond in place of the 4,5-diol of the latter. This was verified by the multiple HMBC correlations (Figure 6) of H-15/C-4 ( $\delta_{\mathrm{C}}$ 151.3 ) and $\mathrm{C}-5$ ( $\delta_{\mathrm{C}} 135.4$ ), and $\mathrm{H}-14 / \mathrm{C}-5$. The unique $\beta$-directed 3-methylenedihydrofuran-2(3H)-one motif was deduced by the key HMBC correlations from $\mathrm{H}-13^{\prime}\left(\delta_{\mathrm{H}} 5.52\right.$ and 6.20$)$ to $\mathrm{C}-7^{\prime}, \mathrm{C}-11^{\prime}\left(\delta_{\mathrm{C}} 145.9\right)$, and $\mathrm{C}-12^{\prime}\left(\delta_{\mathrm{C}} 168.4\right)$. The tigloyl unit was assigned to $\mathrm{C}-15^{\prime}$ by the key HMBC correlation from $\mathrm{H}-15^{\prime}$ to $\mathrm{C}-1^{\prime \prime}$ ( $\delta_{\mathrm{C}} 167.9$ ). The relative configuration of $\mathbf{1}$ was elucidated by its ROESY data (Figure 6), which is similar to that of sarcanolide A, except for that $\mathrm{CH}_{3}-13$ was assigned as $a$-configured on the basis of the strong correlation between $\mathrm{H}_{3}-13$ and $\mathrm{H}-6^{\prime} a$. The structure of compound $\mathbf{1 1}$ (fortunilide K) was hence characterized as shown.

Fortunilide $\mathrm{L}\left(\mathbf{1 2 )}\right.$ had the molecular formula $\mathrm{C}_{41} \mathrm{H}_{46} \mathrm{O}_{13}$ as determined by its (+)HRESIMS ion at $m / z 769.2830[\mathrm{M}+\mathrm{Na}]^{+}$(calcd 769.2831), requiring 19 DBEs. Detailed analysis of NMR data (Tables 3 and 4) of $\mathbf{1 2}$ showed that it shared a common dimeric sesquiterpenoid core with 11, and the only structural difference was in the substituent at C-15'. HMBC correlations (Figure 6) within the substituent confirmed it to be an ( $E$ )-4-((4-methoxy-4-oxobutanoyl)oxy)-2-methylbut-2-enoyloxy group, and it was assigned to C-15' by the HMBC correlations from $\mathrm{H}-15^{\prime}$ to $\mathrm{C}-1^{\prime \prime}$. The $E$-configuration of the $\Delta^{2^{\prime \prime}}$ double bond was assigned based on the ROESY cross-peak between $\mathrm{H}_{2}-4^{\prime \prime}$ and $\mathrm{H}-2^{\prime \prime}$ (Figure S105, Supporting Information). The structure of $\mathbf{1 2}$ was thus established as depicted.

The 15 known lindenane-type sesquiterpenoid dimers, sarglabolide I, ${ }^{19}$ sarglabolide J, ${ }^{19}$ shizukaol F, ${ }^{10}$ shizukaol K, ${ }^{11}$ shizukaol I, ${ }^{10}$ shizukaol C, ${ }^{20}$ shizukaol M, ${ }^{11}$ chlorajaponilide $\mathrm{C},{ }^{21}$ chlorahololides $\mathrm{D},{ }^{5 \mathrm{j}}$ spicachlorantin $\mathrm{D},{ }^{12}$ chloramultilide $\mathrm{C},{ }^{22}$ shizukanolide $\mathrm{F},{ }^{23}$ chloranthalactone $\mathrm{C},{ }^{24}$ and isoshizukanolide, ${ }^{25}$ and sarcandrolide $\mathrm{J}^{5 \mathrm{a}}$ were also isolated and identified by spectroscopic data and analogy with the reported data in literature.

## Antiplasmodial Activity and Structure-Activity Relationships

A total of 44 structurally related compounds obtained from our previous studies were assessed for potential antiplasmodial activity. Twelve new compounds fortunilides A-L (112), as well as 10 known analogues sarglabolide $I(13)$, sarglabolide $J$ (14), shizukaol $K$ (15), shizukaol I (16), shizukaol C (17), schizukaol M (18), chlorahololide D (20), shizukanolide

F (42), chloranthalactone C (43), and isoshizukanolide (44) were isolated currently from $C$. fortunei. Chlorahololide A (33), ${ }^{15 \mathrm{~b}}$ chlorahololide $\mathrm{C}(\mathbf{3 4}),{ }^{17 \mathrm{~b}}$ and chlorahololide $\mathrm{E}(\mathbf{3 8}){ }^{17 \mathrm{~b}}$ were obtained from C. holosteigius. Chloramultilide A (35), ${ }^{26}$ chloramultilide D (37), ${ }^{22}$ chloramultilide $\mathrm{C}(\mathbf{3 9}),{ }^{22}$ and chloramultilide $\mathrm{B}(\mathbf{4 1})^{22}$ came from C. multisachys. Chlorajaponilide $\mathrm{C}(\mathbf{1 9}),{ }^{21}$ shizukaol $\mathrm{N}(\mathbf{2 1}),{ }^{11}$ shizukaol E (25), ${ }^{10}$ shizukaol D (26), ${ }^{20}$ shizukaol F (28), ${ }^{10}$ shizukaol G(29), ${ }^{10}$ shizukaol B (30), ${ }^{20}$ spicachlorantin D (31), ${ }^{27}$ shizukaol A (32), ${ }^{28}$ and spicachlorantin $\mathrm{B}(\mathbf{4 0})^{29}$ were isolated from the plants C. serratus and C. spicatus. Sarcandrolide B(22), ${ }^{5 \mathrm{~b}}$ sarcandrolide A (23), ${ }^{5 \mathrm{~b}}$ sarcandrolide J (24), ${ }^{5 \mathrm{a}}$ sarcandrolide $\mathrm{E}(\mathbf{2 7}),{ }^{5 \mathrm{~b}}$ and sarcandrolide $\mathrm{D}(\mathbf{3 6})^{5 \mathrm{~b}}$ were isolated from S. glabra.

The antiplasmodial activities of all the compounds were tested using a SYBR-Green assay against $P$. falciparum strain Dd2 (chloroquine-resistant) with artemisinin as the positive control as described previously. ${ }^{30,31}$ Sixteen compounds exhibited $\mathrm{IC}_{50}$ values below 100 nM , eight compounds showed mid to high nanomolar $\mathrm{IC}_{50}$ values ( $100-860 \mathrm{nM}$ ), eight compounds were in the micromolar range, and the others were inactive at $25 \mu \mathrm{M}$, which was the highest concentration tested (Table 5). Fifteen of the compounds that exhibited $\mathrm{IC}_{50}$ values below 100 nM were tested for mammalian cytotoxicity toward normal embryonic lung tissue (WI-38 cell line). Nine of the 15 compounds presented a selectivity index (SI) value of $\geq 100$ (Table 6). Compounds $\mathbf{1}, \mathbf{1 4}$, and $\mathbf{1 9}$ had $\mathrm{IC}_{50}$ values of $5.2 \pm 0.6$, and 7.2 $\pm 1.3$, and $1.1 \pm 0.2 \mathrm{nM}$, respectively, and thus have similar potencies to the positive control artemisinin $\left(\mathrm{IC}_{50}=4.0 \pm 4.2 \mathrm{nM}\right)$. Moreover, these compounds showed a SI value of 1,700 , 561 and 4,900 respectively (Table 6) supporting their selective activity against the malaria parasite.

To assess the structure-activity relationships of this new class of antiplasmodial compounds, we first compared the structures of the most active compounds, 1, 2, 7, 14, 17-21, and 28$\mathbf{3 0}$ with those inactive and/or less potent compounds, 10, 13, 24, 25, 27, 32-44 (Figures 1-3) in Table 5. All the active compounds against $P$. falciparum are lindenane-type sesquiterpenoid dimers, and the three lindenane-type sesquiterpenoid monomers tested (4244) were inactive. The most active dimers, e.g. compounds $\mathbf{1}, \mathbf{2}, \mathbf{1 4}, \mathbf{1 7}, \mathbf{1 9}, \mathbf{2 0}$, and $\mathbf{2 8}-\mathbf{3 0}$, featured the common motifs of a conjugated system of methyl ( $Z$ )-5-hydroxy-4-oxopent-2enoate, a $\Delta^{4}$ double bond, and a hydroxy group at $\mathrm{C}-4^{\prime}$ as marked in red. All the compounds without methyl ( $Z$ )-5-hydroxy-4-oxopent-2-enoate and/or $\Delta^{4}$ double bond motifs showed marginal activities or were inactive. Compounds $\mathbf{2 0}$ and $\mathbf{2 7}$ showed very similar structures except for the facts that $\mathbf{2 0}(13 \pm 3 \mathrm{nM})$ had a $\Delta^{4}$ double bond, while $27(>25 \mu \mathrm{M})$ possessed a 4-OH and $\Delta^{5}$ double bond, suggesting that a $\Delta^{4}$ double bond is necessary for the activity, and this was supported by the very similar observations between compounds $\mathbf{3 0}(27 \pm 3 \mathrm{nM})$ and $35(>25 \mu \mathrm{M})$. Moreover, a similar observation between compounds 19 and 5 was observed where the absence of the $\Delta^{4}$ double bond in compound 5 reduced the antiplasmodial activity about 4-fold; however, the toxicity toward mammalian cells was also tremendously reduced since no toxicity was observed for compound 5 at $100 \mu \mathrm{M}$ (Table 6). Interestingly, the absence of the $\Delta^{4}$ double bond in compound 4 reduced the antiplasmodial activity by less than 1 -fold as compared with the structurally similar counterpart compound 2; however, the toxicity toward mammalian cells was also increased by 6 -fold (Table 6). The absence of the $4^{\prime}-\mathrm{OH}$ renders the compounds less active as compared with structurally
similar counterparts, e.g. the compound pairs $\mathbf{2 1}(100 \pm 10 \mathrm{nM}) / \mathbf{2 6}$ (without $4^{\prime}-\mathrm{OH}, 580$ $\pm 90 \mathrm{nM}$ ), and $\mathbf{1 3}(4.6 \pm 0.2 \mu \mathrm{M}) / 24$ (without $4^{\prime}-\mathrm{OH}, 11.4 \pm 1.6 \mu \mathrm{M}$ ), and this was consistent with the fact that compounds $\mathbf{2 5}$ and $\mathbf{3 2}$ without $4^{\prime}-\mathrm{OH}$ groups also showed $P$. falciparum growth inhibition in the micromolar range. The rare dimeric compounds $\mathbf{1 1}$ and $\mathbf{1 2}$ without the $(Z)$-configured $\Delta^{7(11)}$ double bond showed attenuated antiplasmodial activities with $\mathrm{IC}_{50}$ values of $4.7 \pm 0.5 \mu \mathrm{M}$ and $99 \pm 18 \mathrm{nM}$, respectively, as compared with their structurally similar analogues $\mathbf{2}, \mathbf{1 7}$, and $\mathbf{2 0}$. In addition, when the methyl ( $Z$ )-5-hydroxy-4-oxopent-2enoate motif of compound $28(11 \pm 1 \mathrm{nM})$ changed to the $E$-geometry of $7(48 \pm 12 \mathrm{nM})$, the antiplasmodial activity was reduced about 4-fold, suggesting that the motif with a $Z$ geometry seemed more favorable for the antiplasmodial activity. However, both compounds showed SI values below 30 where the $E$-geometry also decreased the mammalian cytotoxicity by 5 -fold (Table 6). Furthermore, compound $\mathbf{8}$ (198 $\pm 22 \mathrm{nM})$ had a very similar structure with that of the most potent antiplasmodial compound $\mathbf{1 9}(1.1 \pm 0.2 \mathrm{nM})$ except for the absence of the $\Delta^{4}$ double bond (replaced by $\Delta^{4(15)}$ and $\Delta^{5}$ double bonds) and the methyl ( $Z$ )-5-hydroxy-4-oxopent-2-enoate motif (ring closed to a five-membered $a, \beta$-unsaturated semiketal lactone), suggesting that these two groups are very important for the antiplasmodial activity observed with compound 19.

Interestingly, the presence of two ester chains $\mathrm{R}^{1}$ and $\mathrm{R}^{2}$ at $\mathrm{C}-13^{\prime}$ and $\mathrm{C}-15^{\prime}$, respectively, dramatically improves the antiplasmodial activities as observed for the low nanomolar antiplasmodial compounds $\mathbf{1}, \mathbf{1 4}$, and $\mathbf{1 9}$. When the $\mathrm{R}^{1}$ ester group was absent, for example compounds $\mathbf{2}, \mathbf{1 7}, \mathbf{1 8}$, and 21, the antiplasmodial potencies were reduced considerably. In addition, when the two ester chains formed an 18 -membered macrocyclic trilactone ring, e.g. compounds 7, 28-30, the antiplasmodial potencies were slightly reduced as compared with the ring-opened analogues. For the compounds possessing the 18 -membered macrocyclic trilactone rings, the antiplasmodial activities of compounds with an $E$-acyloxy-3-methylbut-2-enoate ( $\mathbf{2 8}$ and $\mathbf{4 1}$ ) were better than those with an $E$-acyloxy-2-methylbut-2enoate ( $\mathbf{3 0}$ and $\mathbf{3 9}$ ); however, those with those with an $E$-acyloxy-2-methylbut-2-enoate showed lower mammalian cytotoxicity and better SI values. The presence of a $7^{\prime \prime} a-\mathrm{OH}$ at the 18 -membered macrocyclic trilactone ring $(29,13 \pm 1 \mathrm{nM})$ also increased the potency as compared with compound $\mathbf{3 0}(27 \pm 3 \mathrm{nM})$, however, also increased mammalian cytotoxicity by $\sim 10$-fold (Table 6). It is evident that chain lengths, geometry of the double bonds, and oxidation patterns of the $\mathrm{R}^{1}$ and $\mathrm{R}^{2}$ motifs influence the $P$. falciparum growth inhibition activities significantly, indicating that the $\mathrm{R}^{1}$ and $\mathrm{R}^{2}$ motifs can be modified to improve the antiplasmodial activities of this compound class.

Compounds $\mathbf{4}, \mathbf{5}, \mathbf{9}$, and $\mathbf{3 1}$ with a hydroperoxy group at $\mathrm{C}-4(4-\mathrm{OOH})$ and $\Delta^{5}$ double bond showed remarkable antiplasmodial activities with $\mathrm{IC}_{50}$ values ranging from the low nanomolar to the low micromolar range (Table 5). These potencies are much stronger than those predicted by the previous SAR results, with inactive analogues 27, 33-35 bearing a 4OH and $\Delta^{5}$ double bond (Figures 1 and 3 ), suggesting that the $4-\mathrm{OOH}$ is the crucial moiety for this antiplasmodial compound subclass that likely operates by a different mechanism and/or has a different SAR than the other active dimers. On the other hand, the absence of the $4^{\prime}-\mathrm{OH}$ group and the two acyl groups at $\mathrm{C}-13^{\prime}$ and $\mathrm{C}-15^{\prime}$ reduced antiplasmodial potency dramatically. Thus compound $6(5.3 \pm 2 \mu \mathrm{M})$ is much less active than compound 4
and 5 , indicating that the $4^{\prime}-\mathrm{OH}$ and the two ester chains $\mathrm{R}^{1}$ and $\mathrm{R}^{2}$ at $\mathrm{C}-13^{\prime}$ and $\mathrm{C}-15^{\prime}$ are also important motifs for the antiplasmodial activities for this compound subclass.

Taken together, the above analysis has outlined a clear SAR for the tested compounds as follows (Figure 7). (1) All the active compounds are the Diels-Alder adducts of lindenanetype sesquiterpenoid dimers, while the monomers were inactive. (2) The most potent dimers feature common motifs of a $\Delta^{4}$ double bond, a $4^{\prime}-\mathrm{OH}$, and a conjugated system of methyl (Z)-5-hydroxy-4-oxopent-2-enoate. (3) The presence of two ester chains $\mathrm{R}^{1}$ and $\mathrm{R}^{2}$ at C - $13^{\prime}$ and $\mathrm{C}-15^{\prime}$, respectively, dramatically affects the antiplasmodial activities, suggesting that these groups can be modified to improve the antiplasmodial potency of this class of compounds. (4) For the $4-\mathrm{OOH}$ compound subclass, the $4-\mathrm{OOH}$ moiety is the crucial motif for the antiplasmodial activity, which likely operates by a different mechanism and/or has a different SAR than the other active dimers.

In conclusion, a new class of potent antiplasmodial agents against the chloroquine-resistant strain of $P$. falciparum with low mammalian cytotoxicity was discovered from the plants of Chloranthus genus known as "sikuaiwa", which have long been used in traditional Chinese medicine to treat malaria. Among the active compounds, three of them presented low nanomolar $\mathrm{IC}_{50}$ values antiplasmodial similar to artemisinin and a selectivity index value of 2500. A SAR study of this new class has also been performed and clearly indicated that two motifs in this class can been modified to enhance the antiplasmodial potency. Therefore, lindenane sesquiterpnoid dimers are new class of promising antiplasmodial agents, and likely act through a new mode-of-action because of their unique scaffold that is different from all currently known antiplasmodial agents. Further investigation of this new antiplasmodial class is thus warranted and is currently underway.

## EXPERIMENTAL SECTION

General Experimental Procedures-Optical rotations were recorded on an Autopol VI polarimeter. The UV data were obtained by using a Shimadzu UV-2550 spectrophotometer. The IR spectra were acquired on a Thermo IS5 spectrometer with KBr disks. The NMR spectra were run on a Bruker AM-500 spectrometer with TMS as internal standard. The ESIMS and HRESIMS were obtained on a Brucker Daltonics Esquire 3000 plus and a Waters-Micromass Q-TQF Ultima Global mass spectrometer, respectively. Semipreparative HPLC was performed on a Waters 1525 binary pump system with a Waters 2489 detector ( 210 nm ) using a YMC-Pack ODS-A $(250 \times 10 \mathrm{~mm}, S-5 \mu \mathrm{~m}$ ). Silica gel (200-300 mesh, Qingdao Haiyang Chemical Co., Ltd), $\mathrm{C}_{18}$ reversed-phase (RP-18) silica gel (20-45 $\mu \mathrm{M}$, Fuji Silysia Chemical LTD), CHP20P MCI gel (75-150 $\mu \mathrm{m}$, Mitsubishi Chemical Corporation), D101-macroporous absorption resin (Shanghai Hualing Resin Co., Ltd), and Sephadex LH-20 gel (Amersham Biosciences) were used for column chromatography (CC). Precoated silica gel $\mathrm{GF}_{254}$ plates (Qingdao Haiyang Chemical Co., Ltd.) were used for TLC monitors. All the solvents used for CC were of analytical grade (Shanghai Chemical Reagents Co., Ltd.), and the solvents used for HPLC were of HPLC grade (J \& K Scientific Ltd.).

Plant Material—Twigs of C. fortunei were collected in June of 2013 in Guilin city, Guangxi Province, China, and were authenticated by Professor Shao-Qing Tang of Guangxi Normal University. A voucher specimen has been deposited in Shanghai Institute of Materia Medica, Chinese Academy of Sciences (accession no: CHF-2011-1Y).

Extraction and Isolation—Dried powder of C. fortunei ( 5 kg ) was extracted with $95 \%$ EtOH at room temperature to give a crude extract $(520 \mathrm{~g})$, which was then partitioned between EtOAc and $\mathrm{H}_{2} \mathrm{O}$. The EtOAc soluble fraction (230 g) was subjected to CC (D101macroporous absorption resin) eluted with $30 \%, 50 \%, 80 \%$ and $95 \% \mathrm{MeOH}$ in $\mathrm{H}_{2} \mathrm{O}$ to give four fractions $1-4$, respectively. Fraction $2(120 \mathrm{~g})$ was separated by an MCI gel column $\left(\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, 4: 6\right.$ to $\left.9: 1\right)$ to afford three fractions A-C, and fraction $3(100 \mathrm{~g})$ was treated similarly to afford two fractions D and E. Fraction A (15g) was chromatographed over a silica gel column and eluted with petroleum ether-acetone (from 50:1 to 1:5) in gradient to afford seven subfractions A1-A7. Fraction A2 $(600 \mathrm{mg})$ was purified by semipreparative HPLC ( $50 \% \mathrm{CH}_{3} \mathrm{CN}$ in $\mathrm{H}_{2} \mathrm{O}, 3 \mathrm{~mL} / \mathrm{min}$ ) to yield shizukanolide $\mathrm{F}(4.7 \mathrm{mg})$, chloranthalactone C ( 5.7 mg ), and isoshizukanolide ( 3.7 mg ). Fraction A6 (1.3 g) was chromatographed over a column of Sephadex LH-20 to yield three major parts, and each of them was purified by semipreparative HPLC ( $65 \% \mathrm{CH}_{3} \mathrm{OH}$ in $\mathrm{H}_{2} \mathrm{O}, 3 \mathrm{~mL} / \mathrm{min}$ ) to give shizukaol F ( 10 mg ), shizukaol K ( 8.3 mg ), chlorajaponilide C ( 11 mg ), and chlorahololide D ( 6.5 mg ). Fraction A7 ( 6.1 g ) was further separated on a column of reversed phase $\mathrm{C}_{18}$ silica gel (30-80\% MeOH in $\mathrm{H}_{2} \mathrm{O}$ ) to yield three major components ( $\mathrm{A} 7 \mathrm{a}-\mathrm{A} 7 \mathrm{c}$ ). Component A7a ( 3.5 g ) was fractioned on a silica gel column $\left(\mathrm{CHCl}_{3}-\mathrm{MeOH}, 500: 1\right.$ to $\left.50: 1\right)$ to give four major parts A7a1-A7a4. Fraction A7a1 ( 80 mg ) was purified by semipreparative HPLC ( $55 \% \mathrm{CH}_{3} \mathrm{CN}$ in $\mathrm{H}_{2} \mathrm{O}, 3 \mathrm{~mL} / \mathrm{min}$ ) to yield compounds $\mathbf{1 1}(9.5 \mathrm{mg})$. In the same ways, fraction A7a2 ( 400 mg ) gave $\mathbf{1}(2.2 \mathrm{mg})$ and shizukaol I ( 3.7 mg ); fraction A7a3 (1.5 g) yielded compounds $2(44 \mathrm{mg}), 4(4.7 \mathrm{mg}), 5(43 \mathrm{mg}), 7(22 \mathrm{mg})$, and spicachlorantin D (3 mg ). Fraction A7a4 ( 900 mg ) was chromatographed over a column of Sephadex LH-20 to yield three major parts, and each of them was purified by semipreparative HPLC ( $40 \%$ $\mathrm{CH}_{3} \mathrm{CN}$ in $\mathrm{H}_{2} \mathrm{O}, 3 \mathrm{~mL} / \mathrm{min}$ ) to give $\mathbf{8}(3.0 \mathrm{mg}), \mathbf{1 0}(8.3 \mathrm{mg})$, sarglabolide I ( 5.8 mg ), chloramultilide $\mathrm{C}(3.3 \mathrm{mg})$, and sarcandrolide $\mathrm{J}(6.8 \mathrm{mg})$. By the similar separating procedures used in fraction A, fraction $\mathrm{D}(52 \mathrm{~g})$ was successively subjected to silica gel column (petroleum ether-acetone: 10:1 to 1:5), Sephadex LH-20 (eluted with EtOH), reversed phase $\mathrm{C}_{18}$ silica gel column ( $\mathrm{MeOH}, 30-80 \%$ ), silica gel column $\left(\mathrm{CHCl}_{3}-\mathrm{MeOH}\right.$, 100:1 to 10:1), and finally purified by semipreparative HPLC to yield $\mathbf{3}(49 \mathrm{mg}), \mathbf{6}(23 \mathrm{mg})$, $\mathbf{9}(4.0 \mathrm{mg}), \mathbf{1 2}(28 \mathrm{mg})$, shizukaol C ( 23 mg ), shizukaol M (13 mg), and sarglabolide J (23 $\mathrm{mg})$.

Fortunilide A (1): white, amorphous powder; $[a]^{24}{ }_{\mathrm{D}}-14.8$ (c $\left.0.2, \mathrm{MeOH}\right) ; \mathrm{UV}(\mathrm{MeOH})$ $\lambda_{\text {max }}(\log \varepsilon) 222(4.40) \mathrm{nm}$; IR (KBr) $\nu_{\text {max }} 3446,2926,1733,1632,1438,1384,1223,1159$, $1037,991 \mathrm{~cm}^{-1 ;}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CDCl}_{3}\right)$, see Table 1 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$, see Table 4; (+)-ESIMS m/z 785.3 [M + Na] ${ }^{+}$; (-)-ESIMS m/z761.9 [M - H] ${ }^{-}$; (+)-HRESIMS m/z $785.2781\left[\mathrm{M}+\mathrm{Na}+^{+}\right.$(calcd for $\mathrm{C}_{41} \mathrm{H}_{46} \mathrm{O}_{14} \mathrm{Na}, 785.2780$ ).

Fortunilide B (2): white, amorphous powder; [ $a]^{24} \mathrm{D}^{-115.1}$ (c 1.0, MeOH); UV(MeOH)
$\lambda_{\text {max }}(\log \varepsilon) 219(4.49) \mathrm{nm}$; IR (KBr) $\nu_{\max } 3458,2952,1736,1604,1438,1376,1225,1157$,

1086, $991 \mathrm{~cm}^{-1 ;}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CDCl}_{3}\right)$, see Table 1 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$, see Table 4; (+)-ESIMS m/z $787.2[\mathrm{M}+\mathrm{Na}]^{+} ;(-)-E S I M S ~ m / z 763.2[\mathrm{M}-\mathrm{H}]^{-} ;(+)-H R E S I M S ~ m / z$ $787.2964[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{41} \mathrm{H}_{48} \mathrm{O}_{14} \mathrm{Na}, 787.2936$ ).

Fortunilide C (3): white, amorphous powder; $[a]^{24}{ }_{\mathrm{D}}-152.6$ (c 1.0, MeOH ); UV(MeOH) $\lambda_{\max }(\log \varepsilon) 219(4.50) \mathrm{nm}$; IR (KBr) $\nu_{\max } 3465,2949,1739,1601,1437,1383,1228$, 1144, $991 \mathrm{~cm}^{-1 ;}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CDCl}_{3}\right)$, see Table 1 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$, see Table 4; (+)-ESIMS m/z $657.2\left[\mathrm{M}+\mathrm{Na}^{+} ;(-)\right.$-ESIMS $m / z 633.3[\mathrm{M}-\mathrm{H}]^{-} ;(+)-H R E S I M S ~ m / z$ $657.2672[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{36} \mathrm{H}_{42} \mathrm{O}_{10} \mathrm{Na}, 657.2670$ ).

Fortunilide D (4): white, amorphous powder; $[a]^{24}{ }_{\mathrm{D}}-79.5(c 0.5, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH})$ $\lambda_{\text {max }}(\log \varepsilon) 219$ (4.47) nm; IR (KBr) $\nu_{\text {max }} 3454,2952,1735,1438,1277,1223,1153,976$ $\mathrm{cm}^{-1}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CDCl}_{3}\right)$, see Table 1 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$, see Table 4; (+)-ESIMS $m / z 819.2[\mathrm{M}+\mathrm{Na}]^{+} ;(-)$-ESIMS m/z $795.3[\mathrm{M}-\mathrm{H}]^{-}$; (+)-HRESIMS m/z 819.2851 [M + $\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{41} \mathrm{H}_{48} \mathrm{O}_{16} \mathrm{Na}, 819.2835$ ).

Fortunilide E (5): white, amorphous powder; $[a]^{24}{ }_{\mathrm{D}}-94.2$ (c $\left.0.9, \mathrm{MeOH}\right) ; \mathrm{UV}(\mathrm{MeOH})$ $\lambda_{\max }(\log \varepsilon) 219(4.44) \mathrm{nm}$; IR (KBr) $\nu_{\text {max }} 3442,2955,1736,1439,1280,1216,1151,976$ $\mathrm{cm}^{-1 ;}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CDCl}_{3}\right)$, see Table 2 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$, see Table $4 ;(+)$-ESIMS $m / z 778.9\left[\mathrm{M}-\mathrm{H}_{2} \mathrm{O}+\mathrm{H}\right]^{+} ;(+)$-HRESIMS $\mathrm{m} / \mathrm{z} 819.2840[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\left.\mathrm{C}_{41} \mathrm{H}_{48} \mathrm{O}_{16} \mathrm{Na}, 819.2835\right)$.

Fortunilide F (6): white, amorphous powder $[a]^{24}{ }_{\mathrm{D}}-130.2(c 0.8, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH})$ $\lambda_{\text {max }}(\log \varepsilon) 227$ (4.24) nm; IR (KBr) $\nu_{\text {max }} 3436,2925,2878,1735,1677,1434,1383$, $1109,971 \mathrm{~cm}^{-1 ;}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CD}_{3} \mathrm{OD}\right)$, see Table 2 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right)$, see Table 4; (+)-ESIMS m/z $569.3[\mathrm{M}+\mathrm{H}]^{+}, 1159.6[2 \mathrm{M}+\mathrm{Na}]^{+} ;(+)-H R E S I M S ~ m / z 591.2211[\mathrm{M}+$ $\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{31} \mathrm{H}_{36} \mathrm{O}_{10} \mathrm{Na}, 591.2201$ ).

Fortunilide G (7): white, amorphous powder; $[a]^{24}{ }_{\mathrm{D}}-49.8$ (c 0.7, MeOH); UV (MeOH) $\lambda_{\text {max }}(\log \varepsilon) 213$ (4.37) nm; IR (KBr) $\nu_{\text {max }} 3371,2941,2855,1751,1658,1438,1223$, $1158,998,871 \mathrm{~cm}^{-1 ;}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CDCl}_{3}\right)$, see Table 2 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$, see Table 4; (+)-ESIMS m/z 755.3 [M + Na] ${ }^{+} ;(-)$-ESIMS $m / z 731.7[\mathrm{M}-\mathrm{H}]^{-} ;(+)-H R E S I M S ~ m / z$ $755.2683[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\left.\mathrm{C}_{40} \mathrm{H}_{44} \mathrm{O}_{13} \mathrm{Na}, 755.2674\right)$.

Fortunilide $\mathrm{H}(\mathbf{8})$ : white, amorphous powder; $[a]^{24} \mathrm{D}-45.1(c 0.8, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH})$ $\lambda_{\text {max }}(\log \varepsilon) 215(4.46) \mathrm{nm}$; IR (KBr) $\nu_{\text {max }} 3432,2923,1761,1655,1387,1223,1153,1075$ $\mathrm{cm}^{-1 ;}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CDCl}_{3}\right)$, see Table 2 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$, see Table $4 ;(+)$-ESIMS $m / z 749.3[\mathrm{M}+\mathrm{H}]^{+} ;(+)-$HRESIMS $m / z 771.2625[\mathrm{M}+\mathrm{Na}]^{+}\left(\right.$calcd for $\mathrm{C}_{40} \mathrm{H}_{44} \mathrm{O}_{14} \mathrm{Na}$, 771.2623).

Fortunilide I (9): white, amorphous powder; $[a]^{24}{ }_{\mathrm{D}}+6.0(c 0.4, \mathrm{MeOH})$; $\mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }$ $(\log \varepsilon) 221$ (4.42) nm; IR (KBr) $\nu_{\text {max }} 3433,2930,1751,1441,1220,1153,1081,1012,970$ $\mathrm{cm}^{-1 ;}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CD}_{3} \mathrm{OD}\right)$, see Table 3 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right)$, see Table $4 ;(+)$-ESIMS $\mathrm{m} / \mathrm{z} 800.3\left[\mathrm{M}-\mathrm{H}_{2} \mathrm{O}+\mathrm{H}\right]^{+}$; (+)-HRESIMS m/z $805.2659[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{40} \mathrm{H}_{46} \mathrm{O}_{16} \mathrm{Na}, 805.2678$ ).

Fortunilide J (10): white, amorphous powder; $[a]^{24}{ }_{\mathrm{D}}+107.9(c 0.3, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH})$
$\lambda_{\max }(\log \varepsilon) 208(4.56) \mathrm{nm}$; IR (KBr) $\nu_{\text {max }} 3432,2925,1730,1660,1448,1384,1226$, 1145, $959 \mathrm{~cm}^{-1 ;}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CD}_{3} \mathrm{OD}\right)$, see Table 3 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right)$, see Table 4; (-)-ESIMS m/z $635.6[\mathrm{M}-\mathrm{H}]^{-} ;(-)$-HRESIMS $m / z 635.2475[\mathrm{M}-\mathrm{H}]^{-}$(calcd for $\left.\mathrm{C}_{35} \mathrm{H}_{39} \mathrm{O}_{11}, 635.2498\right)$.

Fortunilide $\mathrm{K}(11)$ : colorless gum; $[a]^{24}{ }_{\mathrm{D}}-47.2(c 1.0, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon)$ 215 (4.37) nm; IR (KBr) $\nu_{\max } 3455,2926,1761,1656,1444,1377,1252,1133,1092,951$ $\mathrm{cm}^{-1}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CDCl}_{3}\right)$, see Table 3 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$, see Table $4 ;(+)$-ESIMS $m / z 617.4[\mathrm{M}+\mathrm{H}]^{+} ;(-)-E S I M S ~ m / z 615.7[\mathrm{M}-\mathrm{H}]^{-} ;(-)-H R E S I M S ~ m / z 615.2588[\mathrm{M}-$ $\mathrm{H}]^{-}$(calcd for $\mathrm{C}_{36} \mathrm{H}_{39} \mathrm{O}_{9}, 615.2600$ ).

Fortunilide L(12): colorless gum; $[a]^{24}{ }_{\mathrm{D}}-26.9(c 0.9, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon)$ 215 (4.53) nm; IR (KBr) $\nu_{\max } 3465,2952,1737,1617,1438,1384,1242,1041,968$ $\mathrm{cm}^{-1}{ }^{1} \mathrm{H}$ NMR data $\left(\mathrm{CDCl}_{3}\right)$, see Table 3 and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$, see Table $4 ;(+)$-ESIMS $m / z 747.5[\mathrm{M}+\mathrm{H}]^{+}, m / z 769.3[\mathrm{M}+\mathrm{Na}]^{+} ;(-)-$ESIMS $m / z 745.8[\mathrm{M}-\mathrm{H}]^{-} ;(+)-$HRESIMS $m / z 769.2830[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{41} \mathrm{H}_{46} \mathrm{O}_{13} \mathrm{Na}, 769.2831$ ).

## Antiplasmodial Evaluation of Compounds 1-44

P. falciparum in vitro growth inhibition assay—Dose-dependent growth inhibition against $P$. falciparum strain Dd2 (chloraquine-resistant) was measured in a 72 h growth assay in the presence of inhibitor as described previously. ${ }^{30-31}$ Artemisinin was used as a positive control. Parasite growth was normalized to untreated controls in the presence of DMSO. Ring stage parasite cultures ( $100 \mu \mathrm{~L}$ per well, with $1 \%$ hematocrit and $1 \%$ parasitaemia) were grown for 72 h in the presence of increasing concentrations of the inhibitor in a $5.05 \%$ $\mathrm{CO}_{2}, 4.93 \% \mathrm{O}_{2}$ and $90.2 \% \mathrm{~N}_{2}$ gas mixture at $37{ }^{\circ} \mathrm{C}$. After 72 h in culture, parasite viability was determined by DNA quantitation using SYBR Green I as described previously. The half-maximum inhibitory concentration $\left(\mathrm{IC}_{50}\right)$ values were calculated with Kaleida Graph using nonlinear regression curve fitting, and the reported values represent averages of at least three independent experiments performed in triplicates with standard deviations.

## Mammalian Cytotoxicity Evaluation of Selected Compounds with Antimalarial Activity

WI-38 cell line in vitro growth inhibition assay-Compounds were evaluated for their cytotoxicity against normal cell line WI-38 (normal embryonic lung tissue). Briefly, 10,000 cells per well were plated in a clear-bottom 96 well plate. Cells were allowed to adhere and then the media was replaced with $100 \mu \mathrm{~L}$ of media containing varying amounts of the test inhibitor and incubated for 24 h . After the incubation time was completed, $10 \mu \mathrm{~L}$ of resazurin sodium salt (Sigma) at $0.125 \mathrm{mg} / \mathrm{mL}$ was added to each well and incubated for 2 h . Cell viability was determined by measuring the fluorescence at 585 nm after excitation at 540 nm . The $\mathrm{IC}_{50}$ values were calculated as described above, and the reported values represent averages of at least two independent.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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|  | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | $\mathrm{R}^{3}$ |
| :---: | :---: | :---: | :---: |
| 1 | F | B | OH |
| 2 | OH | J | OH |
| 3 | OH | A | OH |
| 13 | OH | OH | OH |
| 14 | G | E | OH |
| 15 | OAc | A | OH |
| 16 | OH | C | OH |
| 17 | OH | D | ОН |
| 18 | H | C | OH |
| 19 | F | C | OH |
| 20 | OAc | D | OH |
| 21 | OH | OAc | OH |
| 22 | OH | E | OH |
| 23 | H | D | OH |
| 24 | OH | OH | H |
| 25 | H | OAc | H |
| 26 | OH | OAc | H |


|  | $R^{1}$ | $R^{2}$ | $R^{3}$ | $R^{4}$ |
| :--- | :--- | :--- | :--- | :--- |
| 4 | $O H$ | $J$ | $O H$ | $O O H$ |
| 5 | $F$ | $C$ | $O H$ | $O O H$ |
| 6 | $H$ | $O H$ | $O H$ | $O O H$ |
| 27 | $O A C$ | $D$ | $O H$ | $O H$ |


17 OH D OH






Figure 1.
The tested compounds 1-6, and 13-27.

$7 \quad \mathrm{CH}_{3} \quad \mathrm{CO}_{2} \mathrm{CH}_{3}$
$28 \mathrm{CO}_{2} \mathrm{CH}_{3} \mathrm{CH}_{3}$



$11 \mathrm{R}=\mathrm{H}$

8



Figure 2.
The tested compounds 7-12, and 28.


$33 \mathrm{R}=\mathrm{H}$
$34 \mathrm{R}=\mathrm{OH}$


38


35

$39 \mathrm{R}=\mathrm{H}$
$40 \mathrm{R}=\mathrm{CH}_{3}$

$42 \mathrm{R}_{1}=\mathrm{OH} ; \mathrm{R}_{2}=\mathrm{OH}$
44

Figure 3.
The tested compounds 29-44.


Figure 4.
(A) HMBC and (B) ROESY correlations of 1.


Figure 5.
(A) HMBC and (B) ROESY correlations of $\mathbf{1 0}$.


Figure 6.
Selected HMBC, COSY, and ROESY correlations of $\mathbf{1 1}$ and $\mathbf{1 2}$.


Figure 7.
Brief SAR of antiplasmodial compounds.

Table 1
${ }^{1} \mathrm{H}$ NMR Data of Compounds $1-4\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$

| No. | 1 | 2 | 3 | 4 |
| :---: | :---: | :---: | :---: | :---: |
|  | (mult., $J$ in Hz) | (mult., $J$ in Hz ) | (mult., $J$ in Hz) | (mult., $J$ in Hz) |
| 1 | 2.05, ddd (8.3, 5.7, 4.2) | 2.05, ddd (8.1, 5.7, 4.3) | 2.08, ddd (8.3, 5.8, 4.2) | 2.02, ddd (8.5, 6.2, 4.3) |
| $2 a$ | 0.98, m | 0.98, m | 1.02, m | 0.95, m |
| $2 \beta$ | 0.27, m | 0.29, m | 0.33, m | 1.26, m |
| 3 | 1.83, m | 1.82, m | 1.87, m | 1.57, m |
| 6 | 3.93, d (3.7) | 3.89, d (3.9) | 3.93, d (3.9) |  |
| 9 | 3.95, s | 3.92, s | 3.97, s | 3.78, s |
| 13 | 1.87, s | 1.90, s | 1.94, s | 1.78, s |
| 14 | 1.00, s | 0.98, s | 1.02, s | 1.01, s |
| 15 | $\begin{aligned} & 2.75, \mathrm{~m} \\ & 2.60, \mathrm{~m} \end{aligned}$ | $\begin{aligned} & 2.78, \mathrm{~m} \\ & 2.55, \operatorname{ddd}(16.4,6.2,3.9) \end{aligned}$ | $\begin{aligned} & \text { 2.81, d (16.2) } \\ & \text { 2.58, ddd (16.2, 6.2, 3.9) } \end{aligned}$ | $\begin{aligned} & 3.03, \mathrm{dd}(14.2,7.1) \\ & 1.61, \mathrm{~m} \end{aligned}$ |
| $1^{\prime}$ | $1.60, \mathrm{~m}$ | 1.57, dt (8.4, 3.8) | $1.60, \mathrm{dt}(8.8,5.7)$ | 1.84, m |
| $2^{\prime} a$ | 0.74, m | 0.68 , m | 0.71, m | 0.60, m |
| $2^{\prime} \beta$ | 1.32, m | 1.23, m | 1.26, m | 1.18, m |
| $3^{\prime}$ | 1.45, m | 1.49 , ddd (9.0, 7.3, 3.8) | 1.52, m | 1.25, m |
| $5^{\prime}$ | 1.81, m | 1.82 , m | 1.87, dd (13.7, 6.1) | 1.66, dd (13.2, 6.5) |
| $6^{\prime} a$ | 2.52, dd (18.4, 5.8) | 2.28, dd (18.5, 6.1) | $2.28, \mathrm{dd}(18.4,6.1)$ | 2.29, dd (17.6, 6.5) |
| $6^{\prime} \beta$ | 2.77, m | 2.72, m | 2.72 , dd (18.4, 13.7) | 2.89 , dd (17.6, 13.2) |
| $9^{\prime}$ | 1.83, m | 1.89, m | 1.94, m | 2.56 , dd (10.2, 7.1) |
| $13^{\prime}$ | $\begin{aligned} & 4.97, \mathrm{~d}(13.4) \\ & 4.82, \mathrm{~d}(13.4) \end{aligned}$ | $\begin{aligned} & 4.37, \mathrm{~d}(13.6) \\ & 4.30, \mathrm{~d}(13.6) \end{aligned}$ | $\begin{aligned} & 4.41, \mathrm{~d}(13.6) \\ & 4.34, \mathrm{~d}(13.6) \end{aligned}$ | $\begin{aligned} & 4.44, \mathrm{~d}(13.8) \\ & 4.37, \mathrm{~d}(13.8) \end{aligned}$ |
| $14^{\prime}$ | 0.88, s | 0.85, s | 0.88, s | 0.98, s |
| $15^{\prime}$ | $\begin{aligned} & 4.30, \mathrm{~d}(11.8) \\ & 3.83, \mathrm{~d}(11.8) \end{aligned}$ | $\begin{aligned} & 4.07, \mathrm{~d}(11.5) \\ & 3.81, \mathrm{~d}(11.5) \end{aligned}$ | $\begin{aligned} & 4.83, \mathrm{~d}(11.5) \\ & 4.10, \mathrm{~d}(11.5) \end{aligned}$ | $\begin{aligned} & 4.03, \mathrm{~d}(11.4) \\ & 3.97, \mathrm{~d}(11.4) \end{aligned}$ |
| $2^{\prime \prime}$ | 6.65, q (1.5) | 5.89, s | 5.73, m | 5.89, d (1.4) |
| $4^{\prime \prime}$ | 9.67, s | 4.62, s | 1.94, s | 4.66, m |
| $5^{\prime \prime}$ | 2.19, d (1.5) | 2.13, s | 2.19, s | 2.16, s |
| $7{ }^{\prime \prime}$ |  | 2.67, m |  | 2.67, m |
| $8^{\prime \prime}$ |  | 2.73, m |  | 2.73, m |
| $12-\mathrm{OMe}$ | 3.68, s | 3.75, s | 3.65, s | 3.77, s |
| $6^{\prime \prime}-\mathrm{OMe}$ | 3.66, s |  |  |  |
| $9^{\prime \prime}-\mathrm{OMe}$ |  | 3.67, s |  | 3.69 , s |
| $4-\mathrm{OOH}$ |  |  |  | 8.64, s |

Table 2
${ }^{1} \mathrm{H}$ NMR Data of Compounds $5-8$ at 500 MHz

| No. | $5^{a}$ | $6^{\text {b }}$ | $7^{a}$ | $8^{a}$ |
| :---: | :---: | :---: | :---: | :---: |
|  | (mult., $J$ in Hz) | (mult., $J$ in Hz) | (mult., $J$ in Hz) | (mult., $J$ in Hz) |
| 1 | 1.98, m | 1.98, ddd (8.3, 6.3, 4.3) | 1.99, m | 2.09, m |
| $2 a$ | 0.95, m | 0.94, ddd (8.3, 8.3, 5.7) | 0.97, m | 1.07, m |
| $2 \beta$ | 1.24, m | 1.00, ddd (5.7, 4.3, 4.0) | 0.33, m | 0.63, m |
| 3 | 1.55, m | 1.86, ddd (8.3, 6.3, 4.0) | 1.80, m | 2.21, m |
| 6 |  |  | 4.65, br d |  |
| 9 | 3.85, s | 3.94, s | 3.77, s | 3.85, s |
| 13 | 1.78, s | 1.69 , s | 2.05, s | 1.75, s |
| 14 | 1.00, s | $1.05, \mathrm{~s}$ | 1.05, s | 0.81, s |
| 15 | $\begin{aligned} & 3.06, \operatorname{dd}(14.2,7.1) \\ & 1.65, \operatorname{dd}(14.2,10.2) \end{aligned}$ | $\begin{aligned} & 3.20, \operatorname{dd}(14.3,7.2) \\ & 1.70, \mathrm{dd}(14.3,10.1) \end{aligned}$ | $\begin{aligned} & 2.73, \mathrm{~m} \\ & 2.50, \mathrm{~m} \end{aligned}$ | 6.16, d (4.6) |
| $1^{\prime}$ | 1.87, m | 1.50 , ddd (8.9, 7.2, 4.0) | 1.58, m | 1.88, m |
| $2^{\prime} a$ | 0.60, m | 0.69 , m | 0.72 , m | 0.66, m |
| $2^{\prime} \beta$ | 1.22, m | 0.61, m | 1.32, m | 1.20, m |
| $3^{\prime}$ | 1.55, m | 1.32, m | 1.37, m | 1.71, m |
| $4^{\prime}$ |  | 1.38, m |  |  |
| $5^{\prime}$ | 1.76, m | 1.78 ddd (13.2, 11.0, 6.8) | 1.76, m | $2.26, \mathrm{dd}(12.9,6.1)$ |
| $6^{\prime} a$ | 2.34, dd (17.7, 6.9) | 2.34 , dd (17.6, 6.8) | 2.57, m | 2.37 , dd (18.5, 6.1) |
| $6^{\prime} \beta$ | 2.82 , dd (17.7, 13.2) | 2.66 , dd (17.6, 13.2) | 2.83, m | 2.82 , dd (18.5, 12.9) |
| $9^{\prime}$ | 2.58, dd (10.2, 7.1) | 2.63 , dd (10.1, 7.2) | 1.76, m | 2.65, m |
| $13^{\prime}$ | $\begin{aligned} & 5.06, \mathrm{dd}(13.7,2.0) \\ & 4.78, \mathrm{~d}(13.7) \end{aligned}$ | $\begin{aligned} & 4.29, \mathrm{~d}(13.3) \\ & 4.24, \mathrm{~d}(13.3) \end{aligned}$ | $\begin{aligned} & 4.98, \mathrm{~d}(12.1) \\ & 4.50, \mathrm{~d}(12.1) \end{aligned}$ | $\begin{aligned} & 5.12, \mathrm{dd}(13.2,1.4) \\ & 4.81, \mathrm{~d}(13.2) \end{aligned}$ |
| $14^{\prime}$ | 0.96, s | 0.85, s | 0.77, s | 0.98, s |
| $15^{\prime}$ | 4.02, s | $\begin{aligned} & 3.57, \operatorname{dd}(10.5,4.2) \\ & 3.35, \operatorname{dd}(10.5,6.2) \end{aligned}$ | $\begin{aligned} & 4.49, \mathrm{~d}(11.4) \\ & 3.65, \mathrm{~d}(11.4) \end{aligned}$ | $\begin{aligned} & 4.09, \mathrm{~d}(11.2) \\ & 3.87, \mathrm{~d}(11.2) \end{aligned}$ |
| $2^{\prime \prime}$ | 5.90, dd (2.8, 1.5) |  | 6.01, d (1.7) | 5.98, m |
| $4^{\prime \prime}$ | $4.16, \mathrm{~m}$ |  | $\begin{aligned} & 4.84, \mathrm{~d}(16.7) \\ & 4.59, \mathrm{~d}(16.7) \end{aligned}$ | $\begin{aligned} & 4.14, \mathrm{~d}(16.6) \\ & 4.09, \mathrm{~d}(16.6) \end{aligned}$ |
| 5" | 2.19, d (1.5) |  | 2.19, s | 2.05, s |
| $7 \prime$ | 2.64, m |  | 2.58, m | 2.50, m |
| $8^{\prime \prime}$ | 2.70, m |  | 2.84, m | 2.70, m |
| 12 -OMe | 3.77, s | 3.75, s | 3.77, s |  |
| $6^{\prime \prime}-\mathrm{OMe}$ | 3.67, s |  |  | 3.68, s |
| $4-\mathrm{OOH}$ | 8.53, s |  |  |  |

Table 3
${ }^{1} \mathrm{H}$ NMR Data of 9-12 at 500 MHz

| No. | $9^{a}$ | $10^{a}$ | $11^{b}$ | $12^{b}$ |
| :---: | :---: | :---: | :---: | :---: |
|  | (mult, $J$ in Hz) | (mult., $J$ in Hz ) | $\delta_{\mathbf{H}}($ mult, $J$ in Hz$)$ | $\mathcal{\delta}_{\mathbf{H}}($ mult, $J$ in Hz$)$ |
| 1 | 1.91, ddd (8.7, 6.4, 4.7) | 1.59, ddd (8.1, 5.8, 4.0) | 2.18, dt (8.6, 4.9) | 2.18, dt (8.6, 4.9) |
| $2 a$ | 0.93 , ddd (8.7, 5.7, 5.7) | 0.58, m | 1.17, m | 1.18, m |
| $2 \beta$ | 1.12, m | 0.80, ddd (8.7, 8.1, 5.1) | 0.62, m | 0.62, m |
| 3 | 1.82, ddd (8.5, 6.4, 3.8) | 2.13 , ddd (8.7, 5.8, 3.4) | 1.96, ddd (2, 5.5, 3.1) | 1.97, ddd (8.2, 5.4, 3.1) |
| 5 |  | $2.20, \mathrm{~m}$ |  |  |
| 6 |  | 2.86, m |  |  |
| 9 | 3.78, s | 3.97, s | 3.94, s | 3.99, s |
| 13 | 1.56, s | 1.84, s | 1.35, s | 1.38, s |
| 14 | 0.77, s | 1.07, s | 1.16, s | 1.16, s |
| 15 | $\begin{aligned} & 3.22, \operatorname{dd}(14.0,6.9) \\ & 1.73, \operatorname{dd}(14.0,10.6) \end{aligned}$ | 5.82, s | $\begin{aligned} & 2.30, \mathrm{dd}(18.0,7.1) \\ & 2.96, \mathrm{dd}(18.0,9.5) \end{aligned}$ | $\begin{aligned} & 2.29, \operatorname{dd}(18.0,7.2) \\ & 2.95, \operatorname{dd}(18.0,9.5) \end{aligned}$ |
| $1^{\prime}$ | 1.73, m | 1.93, m | 1.66, m | 1.65, m |
| $2^{\prime} a$ | 0.64 , ddd (8.9, 5.3, 5.3) | 0.63, m | 0.64, m | 0.64, m |
| $2^{\prime} \beta$ | 1.22, m | 1.20, m | 1.20, m | 1.20, m |
| $3^{\prime}$ | 1.73, m | 1.38 , ddd (8.9, 7.3, 3.5) | 1.66, m | 1.66, m |
| $5^{\prime}$ | 2.36, dd (12.7, 6.8) | 2.49 , dd (13.5, 5.6) | 1.65, m | 1.58, dd (14.6, 2.9) |
| $6^{\prime} a$ | 2.46 , dd (17.9, 6.8) | 2.37, dd (18.7, 5.6) | 2.45, d (13.0) | 2.54 , dd (14.6, 14.1) |
| $6^{\prime} \beta$ | 2.91, dd (17.9, 12.7) | 2.87, dd (18.7, 13.5) | 1.69 , m | 1.73, dd (14.1, 2.9) |
| $9^{\prime}$ | 2.71, dd (10.6, 6.8) | 2.21, m | 2.64, dd (9.5 7.1) | 2.64, dd (9.5 7.2) |
| $13^{\prime}$ | $\begin{aligned} & \text { 4.87, d (13.3) } \\ & 4.84, \mathrm{dd}(13.3,1.4) \end{aligned}$ | $\begin{aligned} & \text { 4.37, d (13.2) } \\ & 4.31, \operatorname{dd}(13.2,1.3) \end{aligned}$ | $\begin{aligned} & 5.52, \mathrm{brs} \\ & 6.20, \mathrm{brs} \end{aligned}$ | 5.51, brs 6.19, brs |
| $14^{\prime}$ | 1.02 , s | 0.91, s | 0.94, s | 0.94, s |
| $15^{\prime}$ | $\begin{aligned} & 4.04, \mathrm{~d}(11.0) \\ & 4.01, \mathrm{~d}(11.0) \end{aligned}$ | $\begin{aligned} & 4.41, \mathrm{~d}(11.2) \\ & 3.90, \mathrm{~d}(11.2) \end{aligned}$ | $\begin{aligned} & 4.23, \mathrm{~d}(11.1) \\ & 4.18, \mathrm{~d}(11.1) \end{aligned}$ | $\begin{aligned} & 4.30, \mathrm{~d}(11.1) \\ & 4.08, \mathrm{~d}(11.1) \end{aligned}$ |
| $2^{\prime \prime}$ | 6.07, m | 5.99, q (1.5) | 6.86, m | 5.91, q (1.5) |
| $4^{\prime \prime}$ | 4.10, m | 4.08, m | 1.81, brs | 4.62, m |
| 5" | 2.12, d (1.3) | 2.07, (br s) | 1.82, brs | 2.13, d (1.5) |
| $7{ }^{\prime \prime}$ | 2.57, m |  |  | 2.68, m |
| $8^{\prime \prime}$ | 2.67, m |  |  | 2.75, m |
| 12-OMe |  |  | 3.44, s | 3.43, s |
| $6^{\prime \prime}-\mathrm{OMe}$ | 3.69 , s |  |  |  |
| $9{ }^{\prime \prime}$ - OMe |  |  |  | 3.69, s |

${ }^{a}$ Data were measured in CD3OD and
$b_{\text {data were measured in } \mathrm{CDCl}_{3} \text {. }}$




| No. | $\mathbf{1}^{\boldsymbol{a}}$ | $\mathbf{2}^{\boldsymbol{a}}$ | $\mathbf{3}^{\boldsymbol{a}}$ | $\mathbf{4}^{\boldsymbol{a}}$ | $\mathbf{5}^{\boldsymbol{a}}$ | $\mathbf{6}^{\boldsymbol{b}}$ | $\mathbf{7}^{\boldsymbol{a}}$ | $\mathbf{8}^{\boldsymbol{a}}$ | $\mathbf{9}^{\boldsymbol{b}}$ | $\mathbf{1 0}^{\boldsymbol{b}}$ | $\mathbf{1 1}^{\boldsymbol{a}}$ | $\mathbf{1 2}^{\boldsymbol{a}}$ |
| :--- | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: | :--- |
| $14^{\prime}$ | 26.6 | 26.4 | 26.4 | 24.4 | 24.3 | 23.1 | 26.2 | 25.3 | 24.5 | 25.8 | 24.0 | 24.0 |
| $15^{\prime}$ | 72.4 | 70.5 | 70.1 | 69.8 | 69.1 | 64.1 | 72.2 | 68.3 | 69.3 | 72.4 | 69.5 | 70.4 |
| $1^{\prime \prime}$ | 165.5 | 166.1 | 166.8 | 166.0 | 166.3 |  | 166.2 | 167.0 | 168.4 | 168.3 | 167.9 | 166.0 |
| $2^{\prime \prime}$ | 135.4 | 114.9 | 115.4 | 115.2 | 114.5 |  | 113.0 | 113.4 | 114.4 | 114.1 | 128.2 | 114.3 |
| $3^{\prime \prime}$ | 151.2 | 153.4 | 158.8 | 153.9 | 160.9 |  | 153.9 | 159.1 | 159.9 | 160.1 | 138.7 | 154.2 |
| $4^{\prime \prime}$ | 195.1 | 67.4 | 27.6 | 67.9 | 67.1 |  | 66.2 | 67.2 | 67.2 | 67.3 | 14.7 | 67.3 |
| $5^{\prime \prime}$ | 11.0 | 15.9 | 20.6 | 16.1 | 15.8 |  | 15.9 | 16.0 | 16.0 | 15.8 | 12.3 | 16.0 |
| $6^{\prime \prime}$ | 173.2 | 171.8 |  | 171.9 | 173.3 |  | 172.2 | 174.8 | 174.8 |  |  | 171.7 |
| $7^{\prime \prime}$ | 28.7 | 28.8 |  | 28.9 | 28.6 |  | 29.5 | 28.5 | 173.9 |  |  | 28.9 |
| $8^{\prime \prime}$ | 28.7 | 29.0 |  | 29.1 | 28.7 |  | 29.2 | 28.5 | 29.5 |  |  | 29.0 |
| $9^{\prime \prime}$ | 172.1 | 173.0 |  | 173.0 | 172.1 |  | 172.9 | 171.9 | 29.5 |  |  | 172.9 |
| 12-OMe $^{\prime \prime}$ | 52.8 | 52.9 | 52.9 | 53.0 | 53.1 | 53.1 | 53.0 |  |  |  | 52.2 | 52.1 |
| $6^{\prime \prime}$-OMe | 52.2 |  |  |  | 52.3 |  |  | 53.0 | 52.6 |  |  |  |
| $9^{\prime \prime}$-OMe |  | 52.1 |  | 52.2 |  |  |  |  |  |  |  | 52.1 |
| a Data were measured in CDCl3 and |  |  |  |  |  |  |  |  |  |  |  |  |
| $\boldsymbol{l}_{\text {data were measured in CD3OD. }}$ |  |  |  |  |  |  |  |  |  |  |  |  |

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Table 5
P. falciparum growth inhibition for compounds $\mathbf{1 - 4 4}$

| Comps | $\mathrm{IC}_{50} \pm \mathrm{SD}(\mathrm{nM})$ | Comps | $\mathbf{I C}_{50} \pm \mathbf{S D}(\mathbf{n M})$ |
| :---: | :---: | :---: | :---: |
| 1 | $5.2 \pm 0.6$ | 23 | $320 \pm 130$ |
| 2 | $19 \pm 8$ | 24 | $11,400 \pm 1,600$ |
| 3 | $211 \pm 56$ | 25 | 1,800 $\pm 400$ |
| 4 | $30 \pm 8$ | 26 | $580 \pm 90$ |
| 5 | $43 \pm 3$ | 27 | IA |
| 6 | $5,300 \pm 2,000$ | $28$ | $11 \pm 1$ |
| 7 | $46 \pm 3$ | 29 | $13 \pm 1$ |
| 8 | $198 \pm 22$ | 30 | $27 \pm 3$ |
| 9 | $94 \pm 30$ | 31 | $474 \pm 12$ |
| $10$ | 9,900 $\pm 2,700$ | $32$ | $1,500 \pm 300$ |
| 11 | $4,700 \pm 500$ | $33$ | IA |
| 12 | $99 \pm 18$ | 34 | IA |
| 13 | $4,600 \pm 200$ | 35 | IA |
| 14 | $7.2 \pm 1.3$ | 36 | IA |
| 15 | $860 \pm 89$ | 37 | IA |
| 16 | $111 \pm 12$ | 38 | IA |
| 17 | $21 \pm 9$ | 39 | IA |
| 18 | $96 \pm 37$ | 40 | IA |
| 19 | $1.1 \pm 0.2$ | 41 | $7,100 \pm 1,000$ |
| 20 | $13 \pm 3$ | 42 | IA |
| 21 | $100 \pm 10$ | 43 | IA |
| 22 | $265 \pm 5$ | 44 | IA |

IA represents inactive $($ IC50 $>25 \mathrm{mM})$.
Artemisinin $(\mathrm{IC} 50=4.0 \pm 4.2 \mathrm{nM})$ was used as the positive control.

Table 6
Mammalian cytotoxic activity (WI-38 cell line) of 15 most active natural compounds ( $\mathrm{IC}_{50} \leq 100 \mathrm{nM}$ )

| Compound | WI- $\mathbf{3 8} \mathbf{I C}_{\mathbf{5 0}}(\boldsymbol{\mu M})$ | P. falciparum $\mathbf{I C}_{\mathbf{5 0}}(\boldsymbol{\mu M})$ | SI |
| :--- | :--- | :--- | :--- |
| $\mathbf{1}$ | 8.84 | 0.0052 | 1700 |
| $\mathbf{2}$ | 3.09 | 0.019 | 163 |
| $\mathbf{4}$ | 0.53 | 0.030 | 18 |
| $\mathbf{5}$ | $>100$ | 0.043 | Not cytotoxic |
| $\mathbf{7}$ | 1.24 | 0.046 | 27 |
| $\mathbf{1 2}$ | 15.5 | 0.099 | 157 |
| $\mathbf{1 4}$ | 4.04 | 0.0072 | 561 |
| $\mathbf{1 7}$ | 0.77 | 0.021 | 37 |
| $\mathbf{1 8}$ | 4.45 | 0.096 | 46 |
| $\mathbf{1 9}$ | 5.39 | 0.0011 | 4900 |
| $\mathbf{2 0}$ | 0.16 | 0.013 | 12 |
| $\mathbf{2 1}$ | 10.04 | 0.100 | 100 |
| $\mathbf{2 8}$ | 0.23 | 0.011 | 21 |
| $\mathbf{2 9}$ | 1.74 | 0.013 | 134 |
| $\mathbf{3 0}$ | 16.7 | 0.027 | 619 |
| Adriamycin ${ }^{\text {a }}$ | 0.08 | - | - |
| adriamycin was used as the positive control |  |  |  |


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    The authors declare no competing financial interest.
    ASSOCIATED CONTENT
    Supporting Information. IR, ESIMS, HRESIMS, 1D and 2D NMR spectra of compounds 1-12, and the purities of compounds 1-44 are provided. This material is available free of charge via the Internet at http://pubs.acs.org.

