

## Next-Generation Biomaterials for Bone-Tissue Regeneration: Mg-Alloys on the Move

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**Keywords:** Mg-alloys, Metal-matrix composites, Bone implants, Bio-mechanical properties, Corrosion rate, Biodegradation

**Abstract.** Disorders related to the bone health are becoming a significant concern due to subsequent rise in ageing human population. It is estimated that more than two million bone-surgeries are performed worldwide with an annual cost of \$2.5 billion. In order to replace damaged bone-tissues and restore their function, biomaterials consisting of stainless steels, cobalt-chromium and titanium alloys are implanted. However, these permanent (non-biodegradable) implants often lead to stress-shielding effects and ions release as they interact with the cells and fluids in the body. It is required to overcome these issues by improving the quality of implant materials and increasing their service life. Recently, research in biodegradable materials, consisting of magnesium alloys in particular, has received global attention owing to their biocompatibility and closer mechanical properties to the natural bone. However, due to their rapid corrosion rate in the body fluids, clinical applications of mg-alloys as viable bone-implants have been restricted. A number of mg-alloys have been tested since (both in-vivo and in-vitro) to optimize their biodegradation rate and corrosion properties. The present review summarizes the most recent developments in Mg-alloys designed with biodegradation tailored to the bone-cells growth and highlights the most successful ways to optimize their surface properties for optimum cell/material interaction.

### Introduction

Bone defects and associated health problems affect a significant number of human population throughout the world. It is estimated that 1.5 million people suffer from bone fracture every year [1]. Approximately one-third of these patients are hospitalized and subjected to surgeries for implant replacement incurring significant health care expenditure [2]. Due to their superior mechanical properties, metallic biomaterials have dominated the market associated with bone-implants because they provide sufficient load-bearing capacity to the implant. This is one of the reasons that first-generation biomedical implants made extensive use of stainless steels, cobalt-chromium and titanium alloys [3-4] due to their high demand for load-bearing applications [5-6]. Key mechanical properties of some of these biomedical alloys are shown in Table 1 [7-12].

These biomaterials were developed from 1960 to 1970 with the goal to achieve a suitable combination of physical properties to replace the effected tissues with minimal toxicity. It was not until 1980 that problems with stainless steel and titanium-based implants were surfaced in the form of stress-shielding phenomenon [13-14] and release of ions through corrosion and wear process, which could cause infections and diseases [15-16]. These permanent implants were essentially designed to be inert i.e. not reactive to the physiological conditions of the human body. In addition to metallic materials, ceramic-based biomaterials such as alumina, zirconia and porous ceramics were also investigated in the 1<sup>st</sup> generation biomaterials. However due to their brittleness and allergic effects, they found limited use [17]. Polymers such as polyurethane, acrylic resins, polypropylene were also investigated but they found limited applications as load-bearing implants. Amid of their wide scale applications, these biomaterials offered limited bioactivity which was strongly required for maximum tissue/cells interaction with the implant surface.

Table 1. Mechanical properties of typical biomedical alloys [7-12]

Material	Elastic Modulus [GPa]	Yield Strength [MPa]	Ultimate Strength [MPa]	Merits	Demerits
Stainless steel 316L	196–210	170–750	465–950	High strength and mechanical properties	Dissolves rapidly and also causes erosion of adjacent bones
Co-Cr-Mo	220–230	275–1585	600-1785	Corrosion resistance even in chloride environment. Excellent fatigue strength and wear resistance	Expensive. High elastic modulus (200-230 GPa causing stress yielding
Ti grade 4 (ASTM F67)	105-115	692-795	785-860	Highest tensile and yield strengths, highly integrated to the bone. Moderate elastic modulus	Low shear resistance. Expensive
Ti6Al4V	110	850–900	960–970		

The 2<sup>nd</sup> generation biomaterials consisting of bioactive constituents were developed from 1980 to 2000 with the ability to interact with biological environment in order to increase the cell/material response and bonding with the tissues. Most of the research in this era was focused on surface modifications of 1<sup>st</sup> generation biomaterials to increase bonding between the bone cells and the implant surface. Some of the new biomaterials in this class also had the ability to degrade while the new tissues regenerate such as calcium phosphate, bioactive glasses, glass ceramics and composites. However, the presence of silicon in bioactive glasses and glass ceramics led to weakening of bones causing osteogenesis [18].

The lower bioactivity of the metals was enhanced by depositing bioactive coatings of a ceramic or through chemical modification of the surface by reacting with calcium phosphate. Biodegradable and bioactive polymers such as polyglycolide (PGA), polylactide (PLA), poly( $\epsilon$ -caprolactone) (PCL), polyhydroxybutyrate (PHB), polyorthoester, chitosan, poly(2-hydroxyethyl-methacrylate) (PHEMA) and other hydrogels were extensively studied [19] during this period. However, most of these biomaterials offered low mechanical strengths and high degradation rates and as such could not be applied successfully as load bearing implants.

The 3<sup>rd</sup> generation biomaterials consisted of materials with a combined effect of biodegradability and bioactivity. It involved improved design of materials that were biodegradable, biocompatible with inherent ability to degrade into non-toxic constituents. Among various biomaterials from this class, Mg showed higher biocompatibility due to its mechanical properties [20] matching to the cortical bone as shown in Table 2. This helped to reduce the stress-shielding effects as observed by the 1<sup>st</sup> and 2<sup>nd</sup> generation biomaterials. Moreover, Mg is essential to human metabolism and is naturally found in bone tissues whereas the excess is harmlessly excreted by the body. It has been reported that Mg-alloys produced no inflammatory reactions to the cells and tissues during and after its biodegradation [21].

**Mg-alloys for Biomedical Implants.** Unalloyed Mg in the as-cast condition has a very low strength, at just under 30 MPa, but a fast corrosion rate i.e. 2.89 mm/year in 0.9% NaCl solution [26]. Therefore, alloying additions have been required to improve the mechanical properties by strengthening through solid-solution, precipitation or grain-refinement. However most of the alloying elements have limited solubility in Mg therefore limiting their use to modify the

mechanical properties. Mostly investigated Mg-based alloys with biodegradability, such as Mg–Al-based, Mg–Zn-based and most Mg–rare earth (RE)-based alloys, have obvious precipitation hardening due to high solubility of the secondary element in Mg. Alloy systems such as Mg–Ca and Mg–Si did not support strengthening through heat treatment.

Table 2. Properties of Mg and natural bone [22-25]

Properties	Natural Bone	Mg
Density [g/cm <sup>3</sup> ]	1.8-2.1	1.74-2
Elastic Modulus [GPa]	3-20	41-45
Yield Strength [MPa]	130-180	65-100
Fracture toughness [MPa√m]	3-6	15-40

It has been observed that strengthening by grain refinement is much more effective in Mg-alloys compared to others. Mg when alloyed with Zinc or Zirconium has been observed to reduce/refine the grain size thus leading to improve the yield strength as well as impart corrosion resistance. Zinc is a nontoxic element that is essential for immune system [27] and thus adds to the biocompatibility of Mg. Similarly, calcium (Ca) is the most abundant mineral in the human body that is important for bone function. The solubility of Ca in Mg is about 1.34 wt. %, and under the equilibrium conditions, Ca contributes to solid solution strengthening, grain boundary strengthening, precipitation and grain-refinement [28].

Due to their optimum biodegradation in-vivo, Mg-Al-Zn based alloys have been among the most widely researched biomedical alloys for typical bone-implants (such as bone screws, plates) and other bone fixation implants [29]. Aluminum added to Mg could provide better corrosion resistance [30]. During the past decades, several new alloy systems such as Mg-Al-Ca [31], Mg-Re-Zn-Zr, Mg-Sc-Mn and Mg-Y-Re-Zr [32] were developed for biomedical applications. In addition, binary Mg-Ag alloys were also designed as implant materials to combine the favorable properties of Mg with the well-known antibacterial property of silver [32-33]. In spite of these alloys, the extensive applications of Mg-based implants have been limited by their high corrosion/degradation rates and consequent loss in mechanical integrity after implantation in the human body [20, 34-35]. It is therefore essential to improve the surface properties of potential Mg-alloys in order to realize their clinical application. Table 3 shows the mechanical and corrosion properties of a variety of Mg-alloys [36-38].

Table 3. Mechanical properties of Mg-alloys and their corrosion rates

Alloy	Condition	YS [MPa]	UTS [MPa]	Elong. [%]	In-vitro Corrosion Rate [mm/year]	Corrosion Medium	Ref.
Mg-1Ca	As-cast	40	71.38	1.87	12.56	SBF	[36]
Mg-1Sn	As-cast	79	194	20	2.54	SBF	[37]
Mg-1Ag	As-cast	23.5	116.2	13.2	8.12	SBF	[37]
Mg-1Mn	As-cast	28.5	86.3	7.5	2.46	SBF	[37]
Mg-1Zn	As-cast	25.5	134	18.2	1.52	SBF	[37]
Mg-1Zr	As-cast	67.5	172	27	2.20	SBF	[37]
Mg-2Sr	As-rolled	147.3	213.3	3.15	0.87	Hanks	[38]
Mg-6Zn	As-extruded	169.5	279.5	18.8	0.16	SBF	[37]

**Improving Surface Properties of Mg-alloys.** Irrespective of the advantages associated with Mg-Al-Zn, Mg-Zn-Zr and Mg-Ca based alloys, their degradation rates were still higher than required for orthopedic implants. In order to decrease the degradation/corrosion rate of Mg-alloys two strategies have been mostly investigated:

1. Altering the composition and microstructure, including the grain size and texture of the base material, not only through alloying but also through the development of optimized manufacturing methods and the availability of suitable raw materials.

2. Carry out surface treatments or form coatings which produce protective ceramic, polymer or composite layers [39-45].

Both of these options have offered limited solutions as selecting the type and concentration of the alloying elements to achieve suitable mechanical properties with the optimised degradation profile has not been commercially successful. Clinicians stress the relevance of toxicity assessment that must be considered during material design. On the other hand, the surface treatments of Mg-alloys have produced appreciable initial results [46]. Bioactive coatings consisting of hydroxyapatite (HA) and its derivatives have shown promising effect to reduce the degradation rate of Mg and improve the host response for bone tissue regeneration. However, in many cases satisfactory results were not achieved, typically due to crack formation or poorly controlled adjustment of the specific calcium phosphate phases.

**Mg-alloys Composites.** Following the discussion in earlier section, a third solution has been gradually setting its pace worldwide that is related to the development of Mg-alloy composites. In this context more research is required in designing and fabricating innovative Mg-alloy composites with reinforcements, which are more biocompatible and corrosion resistant. Recently, Mg-based metal matrix composites (MMCs) with bioactive reinforcements, such as calcium phosphate (CaP), is an interesting development [47]. However, further investigation would be required to analyze the effect of CaP and HA derivatives on the degradation rates of Mg-alloy and understanding their structure, property relationship to fabricate commercially viable bone-implants [48]. The reason to using HA as a reinforcement in Mg-alloys is related to its chemical resemblance to the mineral part of the natural bone.

Sun et al. [25] investigated the development of HA/Mg–Zn–Zr nanocomposite structure to improve the properties of these particular Mg-alloys. A number of composite samples with varying HA concentrations were compared through immersion tests in the simulated body fluid (SBF) as shown in Table 4.

Table 4. Properties of Mg-Zn-Zr based nano-composites [25]

Composite	0.2% Yield Strength [MPa]	Tensile Strength [MPa]	Elongation [%]	Corrosion Potential $E_{corr}$ [V]	Charge transfer resistance, $R_t$ [Ohm*cm <sup>2</sup> ]
Mg–3Zn–0.5Zr (C0)	238	275	14.3	-1.88	2.22
Mg–3Zn–0.5Zr–0.5HA (C1)	255	281	15.4	-1.68	4.2
Mg–3Zn–0.5Zr–1HA (C2)	256	285	18.6	-1.62	8
Mg–3Zn–0.5Zr–1.5HA (C3)	275	302	20.9	-1.65	5.36

It can be observed that composite C2 offered higher corrosion resistance while the composite C3 offered better tensile and yield strength. The mechanical properties of the composite with 1.5wt% HA exhibited higher ultimate tensile strength, yield strength and elongation, which are significantly greater than those of monolithic Mg–3Zn–0.5Zr alloy by 9.8%, 15.5% and 46.2%, respectively. A similar increase in the yield strength and tensile strength is also found in C1 and C2 composite samples.

Sunil et al. [49] investigated the fabrication of Mg-based MMCs for degradable implants with nano-HA powder. The purpose was to develop a lamellar structured Mg-composite prepared with 8, 10 and 15 wt.% nanosize HA reinforcement that were fabricated by ball milling and spark plasma sintering. After milling the Mg particles plastically deformed into thin flakes covered with HA particles. Corrosion resistance was found to be higher for Mg–10HA composite compared to other samples. Fracture toughness was higher for pure Mg and Mg–8HA samples but decreased with further increase in HA. Therefore, Mg–8 wt.%HA and Mg–10 wt.%HA were found to be promising composites for implants as they exhibited optimum corrosion resistance and mechanical behavior.

Fig. 1 presents the potention-dynamic polarization curves of the samples that shows an increase in corrosion resistance as the HA content was increased up to 10%. The corresponding electrochemical parameters are tabulated in Table 5 [49].

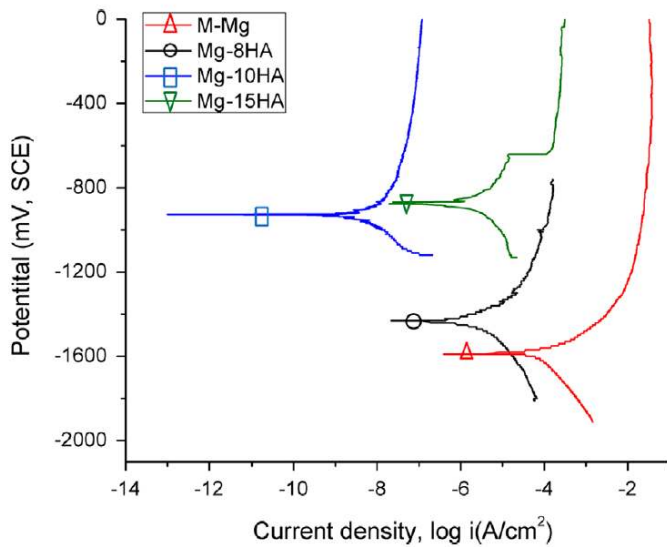


Fig. 1. Potention-dynamic polarization curves [49] (with permission from Elsevier)

Table 5. Electrochemical parameters obtained from potention-dynamic polarization plots [49]

Sample	$E_{\text{corr}}$ [V]	$I_{\text{corr}}$ [ $\text{A}/\text{cm}^2$ ]
M-Mg	$-1.586 \pm 0.017$	$20.987 \pm 0.45 \times 10^{-3}$
Mg-8HA	$-1.434 \pm 0.051$	$4.387 \pm 0.31 \times 10^{-3}$
Mg-10HA	$-0.939 \pm 0.039$	$2.67 \pm 0.27 \times 10^{-4}$
Mg-15HA	$-0.872 \pm 0.042$	$3.397 \pm 0.63 \times 10^{-3}$

The microstructural properties have a direct influence on the surface properties such as corrosion rate and host response to the bone cells. Fig. 2 shows the surface morphologies of the samples after electrochemical test observed using SEM. It is observed that each composite suffered from localized, inter-lamellar and pitting corrosion. Among all the samples, moderate inter-lamellar corrosion and lower pitting was observed for Mg-10HA composite.

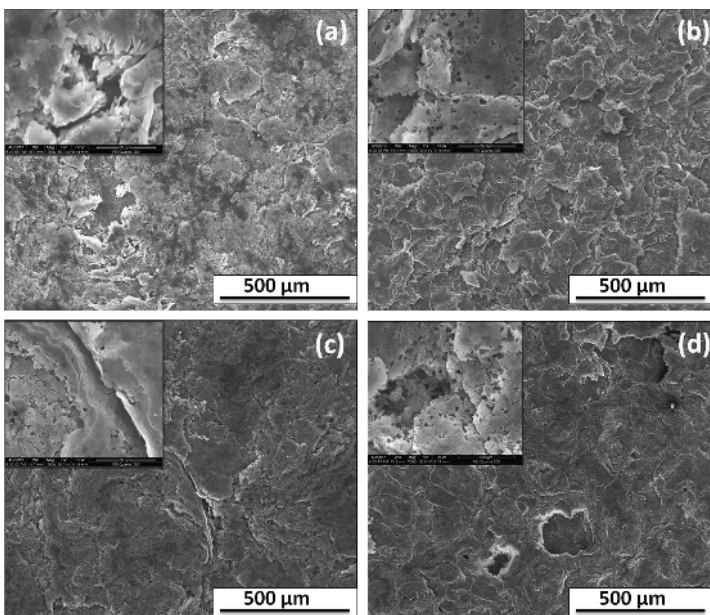


Fig. 2. Surface morphologies of the samples after electrochemical test observed using SEM at 50x magnification: (a) M-Mg, (b) Mg-8HA, (c) Mg-10HA and (d) Mg-15HA (insets at 1000x) [49] (with permission from Elsevier)

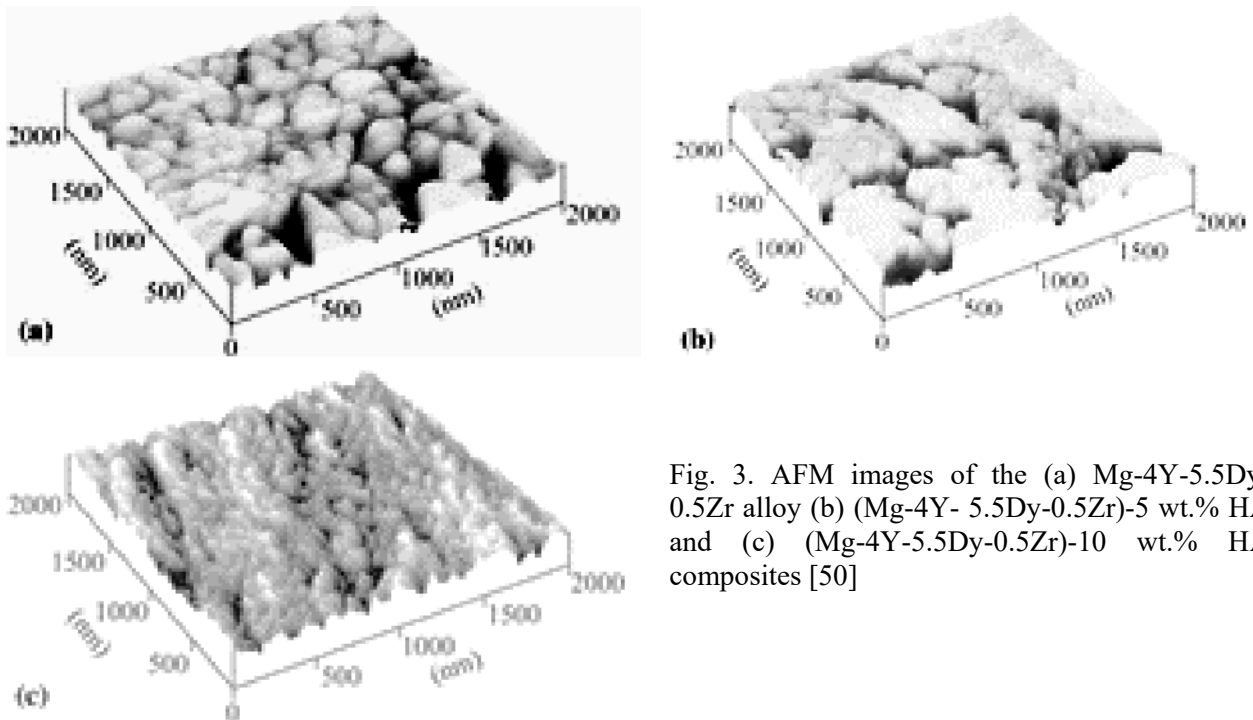


Fig. 3. AFM images of the (a) Mg-4Y-5.5Dy-0.5Zr alloy (b) (Mg-4Y- 5.5Dy-0.5Zr)-5 wt.% HA and (c) (Mg-4Y-5.5Dy-0.5Zr)-10 wt.% HA composites [50]

Kowalski et al. [50] studied the influence of the chemical composition on the microstructure, mechanical and corrosion properties of mechanically alloyed and sintered (Mg-4Y-5.5Dy-0.5Zr)-x wt.% HA composites. Fig. 3 shows the AFM images of (a) Mg-4Y-5.5Dy-0.5Zr alloy (b) (Mg-4Y-5.5Dy-0.5Zr)-5 wt.%HA and (c) (Mg-4Y-5.5Dy-0.5Zr)-10wt.%HA composites. The size distribution of particles was 35 to 245 nm for Mg-4Y-5.5Dy-0.5Zr alloy. The average size of the (Mg-4Y-5.5Dy- 0.5Zr)-5 wt. % HA composite particles is 95 nm. This value is bigger than the average size of the (Mg-4Y-5.5Dy- 0.5Zr)-10 wt.%HA composite particles i.e. 55 nm. Chemical modification of the Mg-4Y-5.5Dy-0.5Zr alloy by HA resulted in fine size particles.

There are few more reports related to Mg-alloy composite systems developed from different alloy composition as well as reinforcement materials. Table 6 summarizes the mechanical and corrosion properties of various types of Mg-MMCs in different conditions.

Table 6. Mechanical and corrosion properties of Mg-based MMCs

Material	Condition	UCS [MPa]	UTS [MPa]	$I_{\text{corr}}$ [A/cm <sup>2</sup> ]	Ref.
Mg-2Zn-0.5Ca/1 $\beta$ -TCP	Normal Casting of Mg-2Zn-0.5Ca then remelting to add TCP	-	-	$789.9 \pm 8.8$ CR*(0-36h) [mgcm <sup>-2</sup> h <sup>-1</sup> ]	[51]
$\beta$ -Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub> /Mg-Zn	PM + extrusion	-	-	$7 [\mu\text{A}\cdot\text{cm}^{-2}]$	[52]
Mg-Bredigite 40 vol%	PM + extrusion	-	190	-	[53]
Mg60	As-cast	580	-	-	[54]
Mg67	-	440	-	-	[54]
Mg60T40	-	800	-	-	[54]
Mg67T40	-	700	-	-	[54]
BG-5/Mg	PM	-	-	-	[55]
BG-10/Mg	-	-	-	-	[55]
BG-15/Mg	-	-	-	-	[55]
Mg-Mn-Zn-Zr	PM	-	-	$1.62 \times 10^{-4}$	[56-57]
Mg-Mn-Zn-Zr-5HA	-	-	-	$3.39 \times 10^{-4}$	[56-57]
Mg-Mn-Zn-Zr-5BG	-	-	-	$1.49 \times 10^{-4}$	[56-57]

Mg-Mn-Zn-Zr-5HA	-	-	-	$2.43 \times 10^{-4}$	[56-57]
Microcrystalline Mg:	-	-			
Mg-CS 10 wt %	-	178			
Mg-CS 20 wt %	-	235			
Mg-CS 30 wt %	-	232	-	$3.34 \times 10^{-4}$	[58]
Mg-CS 40 wt %	-	212			
Mg-CS 50 wt %	-	170			
Mg-5HAp	PM + extrusion	222	-	-	[59]
Mg-10HAp	PM + extrusion	219	-	-	[59]
Mg-15HAp	PM + extrusion	216	-	-	[59]
Mg-0.58(vol%)TiO <sub>2</sub>	-	285	128	-	[60]
Mg-0.97(vol%)TiO <sub>2</sub>	-	278.4	154	-	[60]
Mg-1.98(vol%)TiO <sub>2</sub>	-	297	165	-	[60]
Mg-2.5(vol%)TiO <sub>2</sub>	-	305.5	170	-	[60]
Mg-HA-TiO <sub>2</sub> -Mgo	-	253	-	255	[60]
AZ91-10FA	PM	-	-	$7.4 \times 10^{-5}$	[61]
AZ91-20FA	PM	-	-	$2.3 \times 10^{-6}$	[61]
AZ91-30FA	PM	-	-	$3.5 \times 10^{-7}$	[61]

## Conclusions

In recent years, biomaterials for orthopedic applications have evolved from an inert state to a moderate corrosion resistance as well as biodegradability. The possibility to control surface as well as corrosion properties at the micro/nano level constitutes one of the major breakthroughs in Mg-composites, because it opens a whole new range of strategies seeking the desired interaction with the biological environment. With advancements in new alloy designs and nanomaterials, it will be possible soon to realize Mg-alloys and composites as the emerging next-generation biomaterials for multifunctional bone implants with highly bioactive surface and antibacterial characteristics. Metals will keep moving on their journey to become more human friendly for a long lasting relationship with the natural bone and tissue regeneration. Mg-based composites will evolve more strongly with the advancements in manufacturing and nanotechnology enabling its clinical applications more successful.

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