

1 **Nitrous oxide emission related to ammonia-oxidizing bacteria and mitigation options**
2 **from N fertilization in a tropical soil**

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16

17 **Abstract**

18 Nitrous oxide (N₂O) from nitrogen fertilizers applied to sugarcane has high environmental
19 impact on ethanol production. This study aimed to determine the main microbial processes
20 responsible for the N₂O emissions from soil fertilized with different N sources, to identify
21 options to mitigate N₂O emissions, and to determine the impacts of the N sources on the soil
22 microbiome. In a field experiment, nitrogen was applied as calcium nitrate, urea, urea with
23 dicyandiamide or 3,4 dimethylpyrazone phosphate nitrification inhibitors (NIs), and urea
24 coated with polymer and sulfur (PSCU). Urea caused the highest N₂O emissions (1.7% of N
25 applied) and PSCU did not reduce cumulative N₂O emissions compared to urea. NIs reduced
26 N₂O emissions (95%) compared to urea and had emissions comparable to those of the control
27 (no N). Similarly, calcium nitrate resulted in very low N₂O emissions. Interestingly, N₂O
28 emissions were significantly correlated only with bacterial *amoA*, but not with denitrification
29 gene (*nirK*, *nirS*, *nosZ*) abundances, suggesting that ammonia-oxidizing bacteria, via the
30 nitrification pathway, were the main contributors to N₂O emissions. Moreover, the treatments
31 had little effect on microbial composition or diversity. We suggest nitrate-based fertilizers or
32 the addition of NIs in NH₄⁺-N based fertilizers as viable options for reducing N₂O emissions
33 in tropical soils and lessening the environmental impact of biofuel produced from sugarcane.

34

35 **Introduction**

36 Agriculture is the main anthropogenic source of N₂O emissions, which are predicted to
37 increase as nitrogen fertilizer use increases worldwide to meet the global food demand¹.
38 Currently, N₂O emissions derived from N fertilizers account for up to 40% of total
39 greenhouse gases (GHG) emissions in ethanol production from sugarcane². High N₂O
40 emissions can negate the benefits of GHG reduction of biofuels used to replace fossil fuels³.

41 Emissions of N₂O from soils occur mainly through nitrification and denitrification
42 processes. These processes are carried out by autotrophic and heterotrophic microorganisms
43 belonging to Bacteria, Archaea and Fungi divisions⁴⁻⁶. Other N transformations such as
44 nitrifier denitrification, dissimilatory reduction of NO₃⁻ to NH₄⁺, chemo-denitrification and
45 co-denitrification may also produce N₂O. Despite considerable knowledge of the processes
46 evolving N₂O, the prevalence of these processes in tropical soils has only begun to be
47 addressed.

48 The denitrification process has been demonstrated to contribute more to N₂O emissions
49 than nitrification at soil moisture levels above 75% of the water-filled pore space (WFPS);
50 however, nitrification has been observed to be more prevalent in soil at 60% WFPS⁷. High
51 correlation between N₂O emissions and bacterial *amoA* and *nirK* abundances are observed⁸,
52 suggesting that both nitrification and denitrification and/or nitrifier denitrification processes
53 are responsible for N₂O emissions when cattle urine is applied to soils with 100 and 130% of
54 water-holding capacity.

55 In the central-west and southeast regions of Brazil, about 80% of the land area is cultivated
56 with sugarcane⁹. The dominant soils in these regions are Red Latosols (Hapludox), which are
57 highly weathered, deep and well-drained soils¹⁰. Here, we expected that denitrification would
58 be low because the optimal conditions are at least 60% WFPS. Though high levels of rainfall
59 and anaerobic conditions in soil micropores may increase the contribution of denitrification to

60 N₂O emissions¹¹, we predicted based on the high soil drainage that nitrification would be the
61 major pathway contributing to N₂O emissions. In this case NH₄⁺-based fertilizer would result
62 in higher N₂O emissions than those from NO₃⁻-based fertilizers in these soils. Up to date, this
63 process has not been shown for these types of soils grown with sugarcane.

64 The Intergovernmental Panel on Climate Change (IPCC) estimates that 1% of N applied is
65 emitted as N₂O as default value¹². However, in practice, different amounts of N₂O are emitted
66 depending on N fertilizers and soil types, and environmental conditions¹³⁻¹⁵. Therefore,
67 experiment-based nitrogen management is an important tool to decrease N₂O emissions and to
68 reduce the environmental impact of agricultural practices¹⁵.

69 Urea is the most widely used fertilizer in the world, and generally has been linked to higher
70 N₂O emissions compared with other N sources¹⁴. One way to reduce N₂O emissions is the
71 addition of specific nitrification inhibitors (NIs) such as dicyandiamide (DCD), 3,4
72 dimethylpyrazone phosphate (DMPP), nitrapyrin, and others with urea fertilization^{15,16}. These
73 nitrification inhibitors block the enzyme ammonia monooxygenase in the first step of
74 nitrification¹⁷. The gene encoding this enzyme is *amoA*, present in ammonia-oxidizing
75 bacteria (AOB) and archaea (AOA). Several studies indicate that DCD and DMPP reduce
76 AOB or AOA gene abundances, depending on which microorganism was prevalent^{8,18-20}.
77 DCD has been reported to reduce also the abundance of *nirK*, probably because AOB
78 abundances are correlated with *nirK* abundances, implying reduced nitrifier denitrification
79 abundances⁸.

80 To our knowledge, there are no studies identifying the main microbial processes, the effect
81 of different fertilizers on N₂O emissions, and the impact of different N fertilizers on the
82 microbial community in tropical soils grown with sugarcane. Therefore, the goals of this
83 study were to (i) determine the main microbial process responsible for the N₂O emissions, (ii)
84 evaluate the efficacy of enhanced-efficiency fertilizers, including nitrification inhibitors, in

85 reducing N₂O emissions, and (iii) determine the short-term effects of the fertilizer treatments
86 on bacterial community composition and diversity.

87

88 **Results**

89 *Greenhouse gases emissions.*

90 Emissions of N₂O were low in the first 10 days after fertilizer application, with less than 10 g
91 ha⁻¹ day⁻¹ of N emitted (Fig. 1). High N₂O emission followed rain events coupled with high
92 soil inorganic N availability (Fig. 2). The UR treatment had the highest peak of N₂O emission,
93 on the 14th day, which corresponded to a loss rate of more than 200 g ha⁻¹ day⁻¹ of N. On the
94 29th day, another high emission peak of about 170 g ha⁻¹ day⁻¹ occurred. Between these peaks,
95 N₂O emissions were still relatively high ranging from 15 to 70 g ha⁻¹ day⁻¹ in the UR
96 treatment. The treatments UR+DCD, UR+DMPP, and calcium nitrate had smaller N₂O fluxes
97 than those of UR, showing emission levels similar to the control treatment (around 5 g ha⁻¹
98 day⁻¹ of N). Urea containing NIs (UR+DCD-R, UR+DMPP-R) had been reapplied in the same
99 plots in the previous two years¹⁶. Emissions in the plots with repeated application of NIs were
100 also low at ≤ 5 g ha⁻¹ day⁻¹ of N (Fig.1). The controlled release fertilizer PSCU treatment
101 showed lower N₂O emission (80 g ha⁻¹ day⁻¹) compared to the UR treatment on the 14th and
102 29th days, but was similar to UR treatment levels afterwards, until the 50th day. Between 70
103 and 120 days after fertilizer application, N₂O emissions were greater in the PSCU treatment
104 (between 10 – 20 g ha⁻¹ day⁻¹) compared to the other treatments (2 g ha⁻¹ day⁻¹) (Fig. 1).

105 Cumulated N₂O emissions in the control treatment were equivalent to 0.3 kg ha⁻¹ after 278
106 days. The UR treatment emitted more than 2.3 kg ha⁻¹ of N₂O-N, which corresponded to 1.7%
107 of total N applied. The UR+DCD, UR+DMPP treatments resulted in considerable reductions
108 in cumulated N₂O emissions compared to UR, with emissions that did not differ from those of
109 the control (Table 1). The reduction of N₂O emissions by addition of NIs to UR varied from

110 88 to 97% (95% in average). The PSCU treatment resulted in cumulative emissions similar to
111 those of UR. Calcium nitrate resulted in low N₂O emissions that did not differ from those of
112 the UR+DCD and UR+DMPP treatments or the control (Table 1).

113 Cumulative N₂O emissions data were fit with sigmoidal or exponential equations for the
114 control, UR and PSCU treatments. These models were not applied to the UR+DCD,
115 UR+DMPP, and calcium nitrate treatments because in these treatments the N₂O emissions did
116 not differ from the control. The N₂O emissions in the control treatment were low and whether
117 included or excluded did not affect the model equations. The UR treatment achieved 90% of
118 maximum N₂O loss 40 days after fertilizer application. The N₂O emission lasted longer with
119 the PSCU treatment than with UR; 90% of total N₂O emission was achieved 187 days after
120 fertilization with PSCU (Fig. 3).

121 The total CO₂ and CH₄ emissions were around 6 t ha⁻¹ and -600 g ha⁻¹, respectively. For
122 both CO₂ emissions and CH₄ consumption, no differences between the treatments were
123 observed (Table 1).

124

125 *Soil analysis.*

126 In the first soil sampling seven days after fertilizer application, the calcium nitrate treatment
127 showed N inorganic content (NH₄⁺ + NO₃⁻) around 600 mg kg⁻¹, which was lower than the
128 1000 mg kg⁻¹ of N in the 10 cm soil layer found in the UR treatment (Fig. 2). Afterwards the
129 N content in soil decreased exponentially; during this time, nitrification inhibitors in the
130 UR+DCD and UR+DMPP treatments maintained soil N mostly in the NH₄⁺ form
131 (Supplementary Fig. S1). At the 82th day, soil treated with PSCU showed N content higher
132 than other treatments, near 200 mg kg⁻¹ of N, as opposed to 100 mg kg⁻¹ from the UR
133 treatment.

134 The original soil pH (control treatment) was around 5.1 but increased to 8 in the UR,
135 UR+DCD and UR+DMPP treatments by seven days after fertilizer application, because of
136 urea hydrolysis. PSCU showed pH value 1.4 units lower (pH 6.6) but treatments with calcium
137 nitrate did not affect soil pH. In the 42th day, the soil pH of the UR+DCD and UR+DMPP
138 treatments had dropped to values around 7 (Fig. 2).

139

140 *Nitrogen cycle gene abundances.*

141 The abundances of N cycling genes related to N₂O emissions are depicted in Fig. 4 for one
142 timepoint sampling that featured high N₂O emissions: 16 days after fertilizer application,
143 corresponding to the second soil sampling. The qPCR results from all data sampling
144 timepoints are available in Supplementary Table S1. The abundance of *amoA* belonging to
145 ammonia-oxidizing archaea (AOA) was lower in treatments with N sources than in the
146 control plot, and did not show significant differences among the N sources across nearly all
147 data sampling points. The correlation between AOA *amoA* abundance and N₂O emissions was
148 negative (Supplementary Table S2). The gene abundance representing total archaea showed a
149 similar pattern as the AOA *amoA* abundance (Fig. 4 and Supplementary Table S1). On the
150 other hand, the *amoA* abundance of ammonia-oxidizing bacteria (AOB) was best correlated
151 with N₂O emissions, showing a coefficient (R^2) of 0.18 ($p \leq 0.05$) (Supplementary Table S2).
152 Over almost all data sampling points, AOB *amoA* abundances were higher in the UR
153 treatment than in other treatments, following the data from N₂O emissions (Fig. 4 and
154 Supplementary Table S1). For example, concurrent to high N₂O emissions at day 16 after
155 fertilizer application, the coefficient of correlation (R^2) between N₂O and AOB *amoA* was
156 0.53 (Supplementary Fig. S2).

157 The denitrification genes *nirS* and *nosZ* as well as the 16S rRNA gene of total bacteria did
158 not show differences in abundance between treatments over almost all data samplings. The

159 *nirK* occurring in both ammonia-oxidizing and denitrification microorganisms had a negative
160 correlation with N₂O emissions, while the total bacteria abundance resulted in a positive
161 correlation with this emission (Supplementary Table S2). In some data samplings, the
162 abundance of the *nosZ* was higher in treatment with N than the control treatment, but no
163 differences in *nosZ* abundance were observed between the N sources treatments (Fig. 4 and
164 Supplementary Table S1).

165 A good fit, with an R² of 0.47, was obtained by stepwise regression model relating N₂O
166 emissions to environmental variables, including the AOB *amoA* abundance, rain amount
167 accumulated one week before GHGs measurement, NH₄⁺-N and NO₃-N contents, total
168 bacteria abundance, pH, and CO₂ emission (Supplementary Table S3). Removing the
169 treatments without nitrification inhibitors, NH₄⁺-N content in soil was correlated with N₂O,
170 while NO₃⁻-N was not (Supplementary Table S2 and S3).

171

172 *Bacterial community composition and diversity.*

173 Because the AOB *AmoA* abundance was correlated with N₂O emissions, we sequenced the
174 16S rRNA genes from our samples to ascertain the effect of the treatments on the entire
175 microbial (bacterial and archaeal) community. After processing the 16S rRNA amplicon
176 sequences, the 177 samples (8 treatments x 8 timepoints x 3 replicates, excluding
177 undersampled samples and outliers) contained between 2,000 and 56,638 sequences, with a
178 total of 3,607,143 sequences distributed into 9,267 Operational Taxonomic Units (OTUs).
179 Rarefaction curves indicated that most of the community diversity was captured with our
180 sequencing depth (Supplementary Fig. S3). The top nine bacterial phyla across the samples
181 were *Acidobacteria*, *Actinobacteria*, *Bacteroidetes*, *Firmicutes*, *Gemmatimonadetes*,
182 *Proteobacteria* and *Verrucomicrobia* (Supplementary Table S4). The top phyla that differed
183 between the treatments within at least one timepoint were *Firmicutes*, *Bacteroidetes*,

184 *Nitrospira*, *Proteobacteria*, *Verrucomicrobia* and *Acidobacteria*. Shannon diversity indices of
185 the bacterial communities ranged between 5.3 and 6.3 in treatments over all time points, and
186 were significantly different between treatments only for days 27 (Control *versus* UR) and 82
187 (PSCU *versus* UR+DMPP; Table 2).

188 Based on phylum-level relative abundances, the samples were significantly grouped by
189 treatment on days 7, 27, 35, 42, 85 and 158 (permutation test, $P < 0.100$; Table 2) with a
190 range of observed values between 0.83 and 0.44. At day 7, three separate clusters were
191 formed respectively with the Control and calcium nitrate treatments, the PSCU treatment and
192 the other treatments (Supplementary Table S5). However, this clustering pattern was not
193 observed during the remaining other days, indicating no differences in the bacterial
194 communities between treatments. When bacterial community compositions were compared at
195 the taxonomical level of genera, the samples were significantly grouped by treatment and the
196 grouping was characterized by low observation values for all days except day 16 ($P=0.119$;
197 Table 2 and Supplementary Table S5). The sample treatment grouping pattern from day 7 was
198 seen in the genus-level comparisons (Supplementary Table S5).

199

200 **Discussion**

201 Nitrous oxide emissions from the urea treatment were higher than the emissions found in two
202 previous sugarcane cycles in this experimental area¹⁶. Here, the emission factor was 1.7% of
203 N applied, which is greater than the emission factors found in other studies of sugarcane soils
204 in Brazil, around 0.7 – 1% of N applied as urea^{13,16,21}. The sugarcane plant phenology may
205 give insight into higher emissions. The N fertilizer treatments were applied 20 days after the
206 previous sugarcane ratoon was harvested. At the time of fertilization, the soil was relatively
207 dry – below 30% of the WFPS (Fig. 1) and sugarcane plants were still beginning to sprout. At
208 this stage the root system was being reformed and nutrient uptake was slow, as the high soil N

209 concentration indicated (Fig. 2). This probably led to the greater N₂O emissions than
210 expected. The high peaks of N₂O emission occurred after two high rain events in the first 35
211 days (total 65 mm and 90 mm) on a mostly dry soil (15 – 20% WFPS) but with high air
212 temperatures as shown in Fig. 1. Moreover, high correlation between N₂O emission and
213 accumulated rain in the week was found (Supplementary Table S2). Thus, the climatic
214 conditions in this season contributed to the high N₂O emission values.

215 A strong reduction in N₂O emissions due to the addition of nitrification inhibitors to urea
216 was found, as well as a lack of beneficial effects of the controlled release fertilizer PSCU,
217 supporting similar observations in the same area¹⁶. The N₂O emissions from PSCU were
218 lower than those from UR in the first 30 days after N application, as expected from a slow-
219 release fertilizer (Fig. 2). There was a dry spell from mid-January to the end of March (Fig.
220 1), which may have slowed down the release of N from PSCU. Subsequent release of N from
221 the PSCU pellets likely led to the observed increase in N₂O emissions (Fig. 1). In this way,
222 the N₂O emission from PSCU had lower peaks than those of UR but lasted longer (Fig. 3).
223 Thus, in the end of the experiment cumulated N₂O of UR and PSCU emissions were similar,
224 suggesting that PSCU is not an environmentally friendly N source during one cycle of
225 sugarcane.

226 In the present study, the calcium nitrate treatment showed very small N₂O emissions that
227 were similar to those of the control plots or plots with urea and nitrification inhibitors (NIs).
228 In the present study intensive GHG measurements under field condition were performed over
229 a whole yearly cycle of sugarcane. We maintain that this is the first field study demonstrating
230 much lower N₂O emissions of a nitrate-N source in comparison to high emissions with urea-N
231 or NH₄⁺-N sources; the reduction in N₂O emissions were 98% when compared to urea. This
232 reduction in emissions might be attributed to the high drainage capacity of the soil of the
233 present study, which was classified as Typic Hapludox²² or Red Latosol¹⁰. Water

234 accumulation does not tend to occur in these soil profiles, and consequently, favourable
235 conditions for denitrifiers are avoided.

236 Under controlled conditions with ^{15}N -labeled sources, denitrification was more important
237 at the high soil moisture (75% WFPS), while N_2O emissions with NH_4^+ fertilizers were two
238 times higher than with NO_3^- fertilization at 60% WFPS⁷. In our study WFPS reached a
239 maximum of 40% (Fig. 1), which is more favourable for nitrification. The O_2 concentrations
240 were likely not low enough to favour the denitrification process in relation to nitrification.

241 An alternate explanation for the low N_2O emissions in the calcium nitrate treatment could
242 be NO_3^- leaching. Indeed, the N concentration in the 0-10 cm soil layer of the calcium nitrate
243 treatment was lower than that observed with the other N sources (Fig. 2). However, with only
244 70 mm of rain on a dry soil in 15 days, nitrate is unlikely to have moved beyond 30 cm.
245 Moreover studies with ^{15}N labelling showed little NO_3^- leaching in sugarcane fields in
246 Brazil^{23,24}. Therefore, NO_3^- leaching was not expected to explain the small amount of N_2O
247 emission found with calcium nitrate in the present study. However, further studies should
248 include NO_3^- leaching measurements to confirm the present data.

249 Another aspect that may have contributed to the small N_2O emission in the calcium nitrate
250 treatment was the relatively low organic carbon content in the soil, approximately 1%¹⁶.
251 Sugarcane trash and vinasse have been reported to increase N_2O emissions²⁵, especially under
252 high soil moisture conditions²⁵. Here, we did not include in our treatments C sources such as
253 vinasse, filter cake or sugarcane trash, common sugarcane residues or by-products. These
254 residues can favour not only denitrifiers but also nitrifiers and other microorganisms related to
255 the N cycle²⁶. Application of exclusively NO_3^- -N sources with the addition of C sources, as
256 commonly applied during sugarcane production, may result in N_2O emissions different from
257 those observed here and deserves further attention.

258 Smaller N₂O emissions from calcium nitrate as compared to UR or NH₄-based fertilizer
259 have been previously reported^{27,28}. However, in one study, N₂O emissions observed with all
260 the studied N sources were low (around 0.5% of the N applied), which makes it difficult to
261 compare the treatments²⁸. In a field grown with maize in Brazil, no differences were reported
262 between UR and calcium nitrate in N₂O emissions, but the emission factor was 0.2% of the N
263 applied¹¹. In our study, the N₂O emissions from UR were high at 1.7 % of N applied; the low
264 N₂O emission under calcium nitrate occurred concurrently to high N₂O emissions from UR
265 treatment, which highlights the relevance of the present study.

266 If soil moisture conditions are favourable to denitrification, nitrate-based N fertilizers may
267 produce higher N₂O emissions than urea or ammonium fertilizers. That is the case of the
268 study conducted in a Gleysol soil in which the WFPS was above 60% during most of the
269 experimental period²⁹.

270 Based on Between-Class ordinations of the 16S rRNA compositional data as well as the
271 total 16S rRNA gene copy numbers, the bacterial community appeared to be more affected by
272 sampling day than by treatment. This suggests overall a minimal impact of the treatments on
273 bacterial community composition and diversity. Though further work should examine the
274 long-term impacts, there appears to be a low short-term impact of NIs on the bacterial
275 community. Culturing or shotgun metagenomic and metatranscriptomic techniques may
276 provide future avenues to illuminate the activity of specific nitrifiers under the environmental
277 conditions in this study and to enhance predictions of N₂O emissions due to nitrification in
278 tropical soils.

279 Archaeal *amoA* abundances were highest in the control treatment compared to the
280 treatments with any N sources. Elevated ammonia concentrations and higher soil pH are
281 suggested to favour bacteria compared to archaea^{8,18,30}. Interestingly, the plots with calcium
282 nitrate also showed a reduction in archaeal *amoA* abundance compared to the control. This

283 may reflect the accumulated effect of ammonium nitrate applied in the two previous cycles as
284 this was the N source previously used in a separate study¹⁶.

285 Significant correlations between N₂O emissions and *amoA* abundances were found for
286 ammonia-oxidizing bacteria (AOB), indicating that in our study, N₂O emissions occurred via
287 nitrification. During the first month after UR application, a peak in the AOB *amoA*
288 abundances was observed. Concurrently, the nitrate content in soil increased whereas
289 ammonium decreased, soil pH decreased from 8 to around 6.3, and the soil temperature was
290 25°C (Supplementary Figure S5). Thus, the soil and climatic conditions were favourable to
291 AOB and N₂O emissions from nitrification.

292 Apart from identifying nitrification as the likely source of N₂O emissions here, the data
293 also suggest that denitrification was very low. No differences among treatments nor
294 significant correlations between N₂O emissions and the genes encoding denitrification process
295 as *nirK*, *nirS* and *nosZ* abundances were observed. Further, the model that best estimated N₂O
296 emissions included bacterial *amoA* abundances and N present in the NH₄⁺ form but not in
297 NO₃⁻. Thus, our gene abundance data supported the results of the low N₂O emission data
298 obtained from application of nitrate as the N source.

299 Significant correlation between AOB *amoA* abundance and N₂O emissions was also
300 shown. Under controlled conditions, Venterea et al.³¹ found a high correlation of N₂O
301 emissions from urea and NO₂⁻-N content in soil resulting from increased bacterial *amoA*
302 abundance with no increase in the abundance of the *nxr* gene, which encodes for nitrite
303 oxidation. The authors discussed that N₂O emissions occurred more during nitrification than
304 denitrification, similar to the results found here. Dicyandiamide (DCD) application with cattle
305 urine effectively inhibited the growth of AOB and reduced N₂O emissions as well as the
306 numbers of the *nirK* gene, which encodes for a nitrate reductase⁸. Since DCD did not affect
307 the abundance of other denitrification genes, the authors concluded that AOB, including

308 nitrifier denitrifiers containing *nirK*, were the main contributors to N₂O emissions⁸. In the
309 present study no evidences relating *nirK* and N₂O emissions was found, but the nitrifier
310 denitrification process could have great contribution to N₂O released due the presence of the
311 gene *norB* in AOB^{4,5,32}. Besides nitrifier denitrification, nitrous oxide could be emitted
312 during oxidation of hydroxylamine by ammonia-oxidizing bacteria³³, heterotrophic
313 nitrifiers^{34,35}, and/or abiotic chemodenitrification³⁶. Abiotic N₂O emissions also occur due
314 nitrite reduction by organic and inorganic compounds as amine, copper and iron^{4,37}. Others
315 processes that could be involved in N₂O emissions are abiotic or biotic co-denitrification, by
316 archaea, bacteria or fungi. In co-denitrification, a reducing compound as NO⁻, NO₂⁻ or NO₃⁻
317 combined with organic N, hydroxylamine or ammonium generates N₂O emissions in oxic and
318 anoxic conditions^{6,36}. More studies targeting these reactions can pin down the relative
319 contribution of factors explaining N₂O fluxes from nitrification. The present study showed
320 high N₂O losses from urea, but very small from a nitrate fertilizer source and nitrification was
321 the most relevant microbial process associated with such losses, which has not been reported
322 in soil with sugarcane. The relationship between N₂O emissions and bacterial *amoA*
323 abundances may, therefore be a useful indicator for N management strategies to mitigate N₂O
324 emissions in tropical soils. Other classes of soils and N sources are necessary to confirm our
325 data.

326

327 **Methods**

328 *Experimental set up.*

329 The present experiment was carried out in the 2013/14 season, corresponding to the third
330 ratoon cycle of sugarcane, the variety SP791011, in the experimental area of the Agronomic
331 Institute in Campinas, Brazil (22°52'15" S, 47°04'57" W). The soil in the area was classified
332 as Typic Hapludox or Red Latosol^{10,22}. The same experiment was carried out during the

333 seasons of 2011/12 and 2012/13¹⁶. However, in the 2013/14 season an extra treatment with
334 calcium nitrate was included to consider N₂O emissions due to nitrification or denitrification
335 processes. Here, soil samples were collected in order to associate greenhouse gases (GHG)
336 emissions with the microbial processes that were involved. The treatments were: 1) Control
337 plot without N fertilization (control); 2) urea (UR); 3) UR + DCD; 4) UR + DMPP; 5)
338 Polymer and Sulphur Coated Urea (PSCU); 6) UR + DCD-R; 7) UR + DMPP-R; 8) Calcium
339 Nitrate. R stands for reapplication of inhibitors in the same plots during the previous two
340 cycles of the experiment. The fertilizers were applied on 19 December 2013, 20 days after the
341 harvest of the previous cycle. Phosphorus and potassium were concurrently applied to all
342 plots at rates of 20 and 100 kg ha⁻¹ of P and K, respectively.

343 Nitrogen was applied at a rate of 120 kg ha⁻¹; the nitrification inhibitor DCD (Sigma
344 Aldrich) was added in a dose of 5% DCD-N in relation to urea-N whereas DMPP (powder
345 form) was added as 1% DMPP (w/w) to urea-N; PSCU was produced by Produquímica
346 (Produquímica Ltda, Brazil) and calcium nitrate by Yara (Yara International ASA). Fertilizers
347 were incorporated at a 5 cm soil depth to avoid NH₃ volatilization from urea and the effect of
348 NIs on this N loss³⁸. The fertilizers were applied on either side of the plant row, 10 cm away
349 from the recently harvested sugarcane plants. On one side of the plant row the greenhouse
350 gases were measured; on the opposite side of the same plant, soil for chemical and molecular
351 microbial analyses was collected.

352 Sugarcane yields were not measured in this study because the amount of N lost as N₂O is
353 generally much too low to affect yields. Furthermore, the plot size necessary to evaluate
354 yields usually exceeds 100 m². Because our focus was on GHGs emissions, which are
355 dependent on localized soil conditions, small plots were chosen. In our study, large plots were
356 not only unnecessary but would contribute to noise in the gas flux data.

357

358 *Greenhouse gases analysis.*

359 Greenhouse gases were collected using static chambers¹⁶. Chambers were fixed in the soil 5
360 cm deep along two 25-m long rows of sugarcane. In total, 32 chambers were used, with four
361 replicates per treatment, in a completely randomised design. Gases were sampled in the
362 morning and three times per week during the first three months after fertilizer application,
363 then biweekly as previously done¹⁶. In each sampling date, gas samples were taken at three
364 time intervals: 1, 15, and 30 minutes.

365 After sampling, the gases were immediately stored in pre-evacuated Extainers vials (Labco
366 Limited, Ceredigion, United Kingdom) and analysed in a Shimadzu gas chromatograph (GC-
367 2014). Cumulated gas emissions were calculated by linear interpolation between gas
368 samplings periods. Details of the procedures used for gas analysis and calculations are
369 described elsewhere^{16,21}.

370

371 *Soil chemical analysis.*

372 Soil samples (0-10 cm depth) were collected more intensively in the first two months after
373 fertilizer application, a period corresponding to higher N₂O emissions. Using an auger, three
374 subsamples were collected as a composite sample per experimental plot. In total, eight soil
375 sampling campaigns were collected at 7, 16, 18, 27, 35, 42, 82 and 158 days after fertilizer
376 application. The soil samples were stored in plastic bags at -20°C. Gravimetric moisture after
377 constant weight was attained at 105 °C. The water-filled pore space (WFPS) was calculated
378 considering soil bulk density and porosity determined at the beginning of the experiment. Soil
379 pH was measured in CaCl₂ (0.0125mol L⁻¹) and NH₄⁺-N and NO₃⁻-N contents were
380 determined by steam distillation after soil extraction in 2 mol L⁻¹ KCl solution³⁹.

381

382 *Real-time PCR analysis.*

383 Soil subsamples (20 g) were stored at -80°C for molecular analyses. Total soil DNA was
384 extracted from 0.25 g of soil using the Power Soil kit (Mobio, Carlsbad, CA USA) following
385 the manufacturer's instructions. The quantity and quality of DNA was measured by
386 NanoDrop ND-1000 spectrophotometer (NanoDrop Technologies, Montchanin, USA). The
387 DNA samples were diluted in water free of DNase and RNase (Sigma Aldrich) and the
388 abundance of the genes encoding for nitrification and for denitrification processes were
389 quantified by quantitative real-time PCR with a Qiagen Rotor-Gene Q6000 cycler
390 (RO212226). A reaction was performed in total volume of 12 µl, containing 6 µl Sybrgreen
391 Biotline SensiFAST SYBR non-rox mix, 0.5 µl of each primer (5 pmol) and 5 µl of DNA (3
392 ng). Exceptions were the reaction for the *nirK* amplification, for which the Sybrgreen Qiagen
393 Rotor-Gene SYBR Green PCR Kit was used, and the *nosZ* amplification, for which the
394 starting DNA concentration was 30 ng. Reactions were performed by a QIAgility robot
395 (003516).

396 The thermal conditions of each gene amplification are listed in Supplementary Table S5.
397 Acquisition was done at 72°C (cycle A) or 82-86°C (cycle B) to avoid primer dimers. Melt
398 curve analysis was done at 55-99°C to confirm specificity; the qPCR products were checked
399 by agarose gel electrophoresis to confirm the desired amplicon size. Plasmid DNA from
400 microorganisms containing the gene of interest or from environmental samples were used for
401 the standard curve and then cloned into vectors as described in Table S5. Normal PCR
402 reactions were carried out with similar thermal conditions as qPCR to confirm the fragment
403 size of interest, then cloned and transformed into JM109 High Efficiency Competent Cells
404 (Promega, In Vitro Technologies, Auckland, New Zealand). After overnight bacterial growth
405 in LB medium with ampicillin at 37°C, plasmids were extracted using the PureLin Quick
406 Plasmid Miniprep Kit (Life Technologies, Auckland, New Zealand). The quantity and quality
407 of plasmid DNA were checked by spectrophotometer (NanoDrop ND-1000 Technologies,

408 Montchanin, USA). Standard dilutions were obtained from 10 to 10⁸ copies/μl of each gene.
409 Each run included a DNA template, standard, and a no-DNA control – water free of DNase
410 and RNase (Sigma Aldrich) – done in duplicate. Reaction efficiency was 89-105% and R²
411 values ranged from 0.94 to 0.99.

412

413 *16S rRNA partial gene sequencing.*

414 To assess the impact of the treatments on the bacterial community, we sequenced the 16S
415 rRNA gene marker from total DNA extracted from the soil samples. The V4 region of the 16S
416 rRNA gene was amplified by using archaeal/bacterial primers 515F (5'-
417 GTGCCAGCMGCCGCGTAA-3') and 806R (5'-GGACTACVSGGGTATCTAAT-3'). The
418 samples were PCR-amplified using barcoded primers linked with the Ion adapter “A”
419 sequence (5'-CCATCTCATCCCTGCGTGTCTCCGACTCAG-3') and Ion adapter “P1”
420 sequence (5'-CCTCTCTATGGGCAGTCGGTGAT-3') to obtain a sequence of primer
421 composed for A-barcode-806R and P1-515F adapter and primers. The 16S rRNA gene
422 amplifications for library preparation were performed on the C1000 thermocycler (Biorad,
423 Hercules, CA, USA) with thermal conditions of 95°C-5 min.; 35x 95°C-30s, 53°C-30s, 72°C-
424 60s; 72°C-10 min. A reaction of 25 μl in total was done, including 2.5 μl of 10X PCR Buffer,
425 2.5 μl dNTPs (200 μM), 0.25 μl of each primer (0.1 pmol/μl), 0.2 μl of fast startExp-
426 Polymerase (0.056 U) and 1 μl of DNA (0.6 ng). The reactions were carried out in duplicate
427 and included a negative control. The amplicons were checked by gel electrophoresis. The
428 PCR products were purified by Agencourt AMPURE XP to remove primer dimers, then
429 quantified by Quant-iT PicoGreen and equimolar mixed for sequencing using the PGM Ion
430 Torrent (Life Technologies).

431

432 *16S rRNA amplicon sequences processing.*

433 MOTHUR Version 1.34.2 was used to process the 16S rRNA partial genes sequences,
434 implemented using a Snakemake workflow on a 32-node server running Linux Ubuntu 14.4⁴⁰.
435 Forward and reverse primer sequences were removed from each sample FASTQ file using
436 Flexbar version 2.5⁴¹. Reads were filtered based on sequence quality by running the Sickle
437 tool (minimum quality score 25, minimum length 150). Filtered reads were converted to
438 FASTA format and concatenated into a single file, then clustered into OTUs using the
439 UPARSE strategy of dereplication, sorting by abundance with at least two sequences and
440 clustering using the UCLUST smallmem algorithm⁴². These steps were performed with
441 VSEARCH version 1.0.10, which is an open-source and 64-bit multithreaded compatible
442 alternative to USEARCH. Chimeric sequences were detected using the UCHIME algorithm⁴³
443 implemented in VSEARCH. All reads before the dereplication step were mapped to OTUs
444 using the USEARCH_global method implemented in VSEARCH to create an OTU table and
445 then converted to the BIOM-Format 1.3.1⁴⁴. Last, taxonomic information for each OTU was
446 added to the BIOM file using the RDP Classifier version 2.10⁴⁵.

447

448 *Statistical analysis – gas fluxes and gene abundances.*

449 Daily GHG fluxes, cumulated emissions of N₂O, CO₂, CH₄ and gene abundance values were
450 checked for normal distribution of residues by Shapiro-Wilk test, and then submitted to
451 variance analysis (ANOVA) and the means compared by Tukey's test at $P \leq 0.05$. Soil pH
452 was transformed to H^+ : 10^{-pH} before statistical analysis. Linear correlations between N₂O
453 fluxes and environmental variables were evaluated at the 5% level of significance. Multiple
454 linear regressions, which were selected by the stepwise process at $p \leq 0.05$, were fitted
455 between N₂O fluxes and environmental variables. When necessary, the N₂O flux values were
456 $\log(x)$ transformed and rechecked to obtain a normal distribution of residues and variance

457 stability⁴⁶. The calculations were performed with the SISVAR statistical software⁴⁷ and
458 graphics plotted using Sigma Plot⁴⁸.

459 Cumulative N₂O emissions as a function of time were fitted by sigmoidal or exponential
460 equations, for which the sigmoid equation was:

$$461 \quad N_2O = \frac{a}{1+e^{-\frac{t-t_0}{b}}} \quad (1)$$

462 where N_2O is the cumulative N₂O-N emission, g ha⁻¹, t is the time in days after fertilizer
463 application, and a , t_0 and b are equation parameters, where a is the maximum loss and t_0 is the
464 time in which 50% of maximum loss occurs.

465 The exponential rise to the maximum model had the following equation:

$$466 \quad N_2O = a(1 - e^{-bt}) \quad (2)$$

467 where N_2O is the cumulative N₂O-N emission, g ha⁻¹, t is the time in days after fertilizer
468 application, and a and b are equation parameters, where a is the maximum loss and b is the
469 rate of rise.

470

471 *Statistical analysis – 16S rRNA amplicon sequence data.*

472 The 16S rRNA samples were analysed to compare bacterial community alpha diversity and
473 composition across treatments and time point (days 7, 16, 18, 27, 35, 42, 82 and 158). The
474 BIOM files were handled with the “phyloseq” package⁴⁹ in R⁵⁰. Rarefaction curves were
475 generated to ensure adequate sequencing depth across samples. Discarding undersequenced
476 samples, the minimum sample size was 2000. For alpha diversity analyses, the 16S rRNA
477 samples were rarefied to 2000 sequences using the “vegan” package⁵¹. Renyi diversities at
478 alpha level 1, corresponding to the Shannon diversity index were kept (“BiodiversityR” R
479 package). The Shannon diversity data was furthermore subjected to Kruskal-Wallis tests
480 among treatments and the Kruskal-Wallis multiple comparison test between treatments using
481 the “pgirmess” R package.

482 Comparisons of bacterial community compositions were evaluated using the Statistical
483 Analysis of Metagenomic Profiles (STAMP) software⁵². The top nine Bacterial phyla based
484 on relative abundances across all samples were compared among and between treatments for
485 each time point. The unclassified sequences were removed prior to analysis. The ANOVA
486 statistical and Tukey-Kramer post-hoc tests (CI 95%) were applied using the Benjamini
487 Hochberg multiple test correction. To explore beta diversity (treatment differences) of the
488 bacterial communities, Between-Class Analysis (BCA) of the non-rarefied 16S rRNA
489 samples grouped by treatment was performed using the “ade4” R package⁵³. First, with
490 unclassified sequences removed, correspondence analyses of the compositional data
491 agglomerated at the rank of Phylum and Genus were conducted, followed by BCA. Further,
492 the BCA groups for the phyla and genera analyses were tested using the Monte-Carlo
493 permutation method with 999 repetitions.

494

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496

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658

659 **Author contribution statement**

660 J.R.S., J.B.C., H.C. and E.E.K. designed research; J.R.S. and K.S.L. conducted the
661 experiment; J.R.S., N.A.C., A.M.K., A.P., H.J.L. and E.E.K. conducted the qPCR and the
662 sequencing analyses; J.R.S. and N.A.C. performed the statistical analyses; J.R.S., N.A.C.,
663 H.J.L., H.C. and E.E.K. wrote the paper. All authors reviewed the manuscript.

664

665 **Competing interest statement**

666 The authors declare no conflict of interest.

667

668 **Accession codes**

669 European Nucleotide Archive study accession number PRJEB13027.

670

671

672 **Table 1.** Cumulated nitrous oxide, carbon dioxide and methane emissions from Red Latosol
 673 soil during 278 days after application of urea with or without nitrification inhibitors (DCD
 674 and DMPP), polymer sulphur coated urea (PSCU) and calcium nitrate applied to sugarcane.

675

Treatments	N ₂ O-N				Reduction (%)	CO ₂ -C		CH ₄ -C	
	g ha ⁻¹	log*		% N applied†		kg ha ⁻¹ *	ns	g ha ⁻¹ *	ns
Control	286	2.4	C	-	-	5835	ns	-598	ns
UR	2301	3.4	A	1.68	-	5933		-633	
UR+DCD-R	531	2.7	B	0.20	88	5883		-612	
UR+DMPP-R	350	2.5	C	0.05	97	5871		-532	
PSCU	2165	3.3	A	1.57	7	5912		-648	
UR+DCD	410	2.6	Bc	0.10	94	5859		-633	
UR+DMPP	353	2.5	C	0.06	97	5897		-656	
Calcium nitrate	329	2.5	C	0.04	98	5973		-600	
<i>P</i> value		<0.00001						0.9769	0.9328

676 * Tukey test, $p \leq 0.05$; ns: no significant; N₂O-N: g ha⁻¹ transformed in log(X) † Results from treatment without
 677 N were subtracted for this calculation. - R means reapplication of inhibitors in the same plot in the two preceding
 678 years. Different characteristics in the column of N₂O mean significant differences ($p < 0.05$) between the values.
 679

680 **Table 2.** Shannon indices and Between-Class Analysis (BCA) ordinations of the bacterial
 681 communities present in the Red Latosol soil under treatments with urea with or without
 682 nitrification inhibitors (DCD and DMPP); polymer sulphur coated urea (PSCU) or calcium
 683 nitrate applied to sugarcane.

684

Treatments	16S rRNA gene diversity (Shannon index)															
	7 DAF*		16		18		27		35		42		85		158	
Control	6.2	ns	6.1	ns	6.0	ns	6.2	a	6.1	ns	6.2	ns	6.0	ab	6.1	ns
UR	5.8		6.0		6.0		5.3	b	5.7		5.8		5.6	ab	6.0	
UR+DCD-R†	5.9		6.2		5.8		6.0	ab	6.0		6.1		5.8	ab	6.2	
UR+DMPP-R	5.6		6.1		5.9		5.8	ab	6.1		6.2		6.1	ab	6.2	
PSCU	5.9		5.9		5.8		5.8	ab	5.1		5.2		5.4	a	5.8	
UR+DCD	5.7		6.1		5.7		5.8	ab	5.9		6.0		6.0	ab	6.2	
UR+DMPP	5.7		6.0		5.9		5.8	ab	5.5		6.1		6.1	b	6.1	
Calcium nitrate	6.3		6.0		5.7		5.9	ab	5.9		6.2		6.0	ab	6.2	

685 *DAF: Days after fertilizer application. Means in column followed by same letter did not differ; ns: no
 686 significant. † R means reapplication of inhibitors in same plots.

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693 **Figure legends**

694 **Figure 1.** Rainfall, air temperature, water-filled pore space (WFPS) and nitrous oxide fluxes
695 from Control (No N), urea (UR) with or without nitrification inhibitors (DCD and DMPP),
696 polymer sulphur coated urea (PSCU) and calcium nitrate applied to sugarcane. R:
697 reapplication of inhibitors in the same plots during the previous two cycles of the experiment.
698 N fertilizers were applied on 13 December 2013.

699

700 **Figure 2.** Soil concentration of $\text{NH}_4^+\text{-N} + \text{NO}_3^-\text{-N}$ and pH from Control (No N), urea (UR)
701 with or without nitrification inhibitors (DCD and DMPP), polymer sulphur coated urea
702 (PSCU) and calcium nitrate applied to sugarcane. R: reapplication of inhibitors in the same
703 plots during the previous two cycles of the experiment.

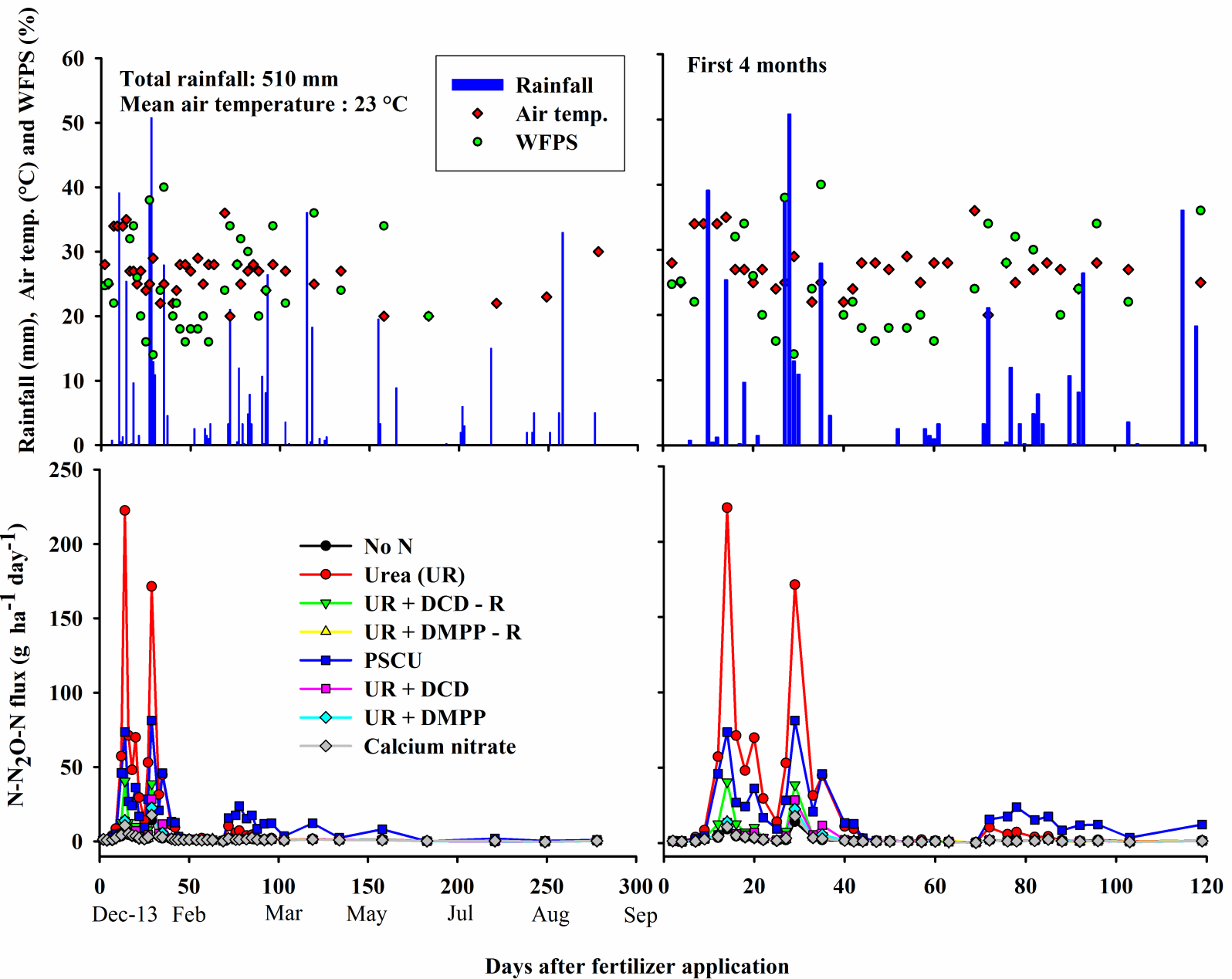
704

705 **Figure 3.** Cumulative N_2O emission (red dots) and sigmoidal or exponential equations fitted
706 (black lines) to data of urea (UR) and polymer sulphur coated urea (PSCU) applied to
707 sugarcane. (b) net UR and PSCU N_2O emissions calculated by subtracting data of the control
708 treatment.

709

710 **Figure 4.** Nitrous oxide fluxes, nitrogen cycle genes (*amoA* bacteria, *amoA* archaea, *nirK*,
711 *nirS*, *nosZ*) abundances and total bacteria and total archaea abundances in the Red Latosol soil
712 16 days after fertilizer application of urea with or without nitrification inhibitors (DCD and
713 DMPP); polymer sulphur coated urea (PSCU) or calcium nitrate applied to sugarcane. R:
714 reapplication of inhibitors in the same plots during the previous two cycles of the experiment.

715



Days after fertilizer application

