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Article

Novel Synthesis of Hydrazide-Hydrazone Derivatives and Their Utilization in the Synthesis of Coumarin, Pyridine, Thiazole and Thiophene Derivatives with Antitumor Activity

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Abstract: The reaction of cyanoacetyl hydrazine (1) with 3-acetylpyridine (2) gave the hydrazide-hydrazone derivative **3**. The latter compound undergoes a series of heterocyclization reactions to give new heterocyclic compounds. The antitumor evaluation of the newly synthesized products against three cancer cell lines, namely breast adenocarcinoma (MCF-7), non-small cell lung cancer (NCI-H460) and CNS cancer (SF-268) was performed. Most of the synthesized compounds showed high inhibitory effects.

Keywords: hydrazide-hydrazone; 3-acetylpyridine; thiophene; coumarin; antitumor activity

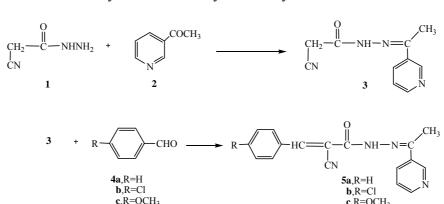
1. Introduction

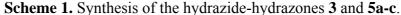
Hydrazines and their derivatives constitute an important class of compounds that has found wide utility in organic synthesis [1,2]. While hydrazines have traditionally been employed as reagents for the derivatization and characterization of carbonyl compounds, in recent years the N-N linkage has been used as a key structural motif in various bioactive agents. In particular, an increasing number of N-N bond-containing heterocycles and peptidomimetics have made their way into commercial applications as pharmaceutical and agricultural agents [3,4]. Recently, hydrazide-hydrazones have gained great

importance due to their diverse biological properties including antibacterial, antifungal, anticonvulsant, anti-inflammatory, antimalarial and antituberculosis activities [5-17]. With the aim of obtaining novel hydrazide-hydrazones with a wide spectrum of pharmaceutical applications, we report herein the synthesis of a series of hydrazide-hydrazones together with their use in a series of heterocyclic transformations and their evaluation as anti-tumor agents [18-21].

2. Results and Discussion

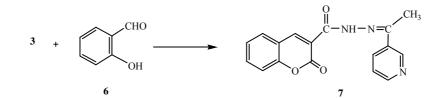
Recently, our research group became involved with a comprehensive program involving the synthesis of a series of hydrazide-hydrazone derivatives and their utilization in the synthesis of heterocyclic compounds with potential pharmaceutical and biological activities [22,23]. In continuation to this program, we report herein the reaction of cyanoacetylhydrazine (1) with 3-acetylpyridine (2) in 1,4-dioxane to form the hydrazide-hydrazone derivative **3**. The structure of compound **3** was confirmed based on its analytical and spectral data. Thus, the ¹H-NMR showed a singlet at δ 2.28 for the CH₃ group, a singlet at δ 4.26 for the CH₂ group, a multiplet at δ 7.43-8.99 for the pyridyl group and a singlet (D₂O exchangeable) at δ 10.81 for the NH group. Moreover, the ¹³C- NMR spectrum showed peaks at δ : 14.2 (CH₃), 28.9 (CH₂), 116.8 (CN), 122.1, 123.4, 133.7, 150.2, 151.0 (pyridine C), 168.1 (C=N), 169.8 (C=O). Further structure elucidation of compound **3** was obtained through the study of its reactivity towards chemical reagents. Thus, the reaction of **3** with either benzaldehyde (**4a**), 4-chlorobenzaldehyde (**4b**) or 4-methoxybenzaldehyde (**4c**) gave the corresponding benzal derivatives **5a-c**, respectively (Scheme 1).





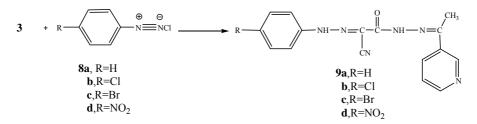
On the other hand, the reaction of compound 3 with salicylaldehyde (6) gave the coumarin derivative 7 (Scheme 2). Analytical and spectral data of the product are in agreement with the proposed structure (see Experimental section).

Scheme 2. Synthesis of 7.



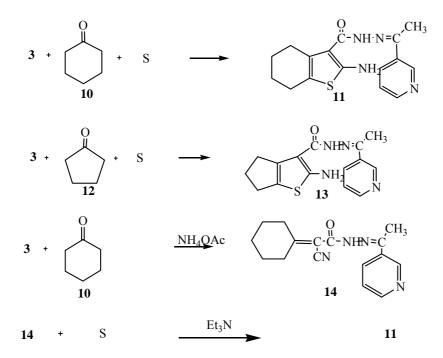
Next, we studied the reactivity of the active methylene group present in compound **3** towards diazonium salts. Thus, the reaction of **3** with either benzenediazonium chloride (**8a**), 4-chlorobenzenediazonium chloride (**8b**), 4-bromobenzenediazonium chloride (**8c**) or 4-nitrobenzenediazonium chloride (**8d**), gave the hydrazone derivatives **9a-d**, respectively (Scheme 3). Analytical and spectral data of the latter reaction products are all consistent with the proposed structures.

Scheme 3. Synthesis of phenylhydrazo derivatives 9a-d.



Moreover, the reaction of compound **3** with cyclohexanone (**10**) and elemental sulfur in the presence of triethylamine was studied as an application of Gewald's thiophene synthesis. The reaction led to the formation of 4,5,6,7-tetrahydrobenzo[*b*]thiophene derivative **11**. On the other hand, the reaction of compound **3** with cyclopentanone (**12**) and sulfur gave the cyclopentene[*b*]thiophene derivative **13**. The structures of compounds **11** and **13** were based on analytical and spectral data (see Experimental section). Further confirmation of structure **11** was obtained through its synthesis via another reaction route. Thus, the reaction of compound **3** with cyclohexanone in the presence of ammonium acetate in an oil bath at 140 °C gave the Knoevenagel condensation product **14**. The latter reacted with elemental sulfur in the presence of triethylamine to produce the same tetrahydrobenzo[*b*]thiophene derivative **11**. It is convenient to notice that the yield (70%) for formation of compound **11** using the latter procedure was higher than in the synthetic pathway described before (62%) (Scheme 4).

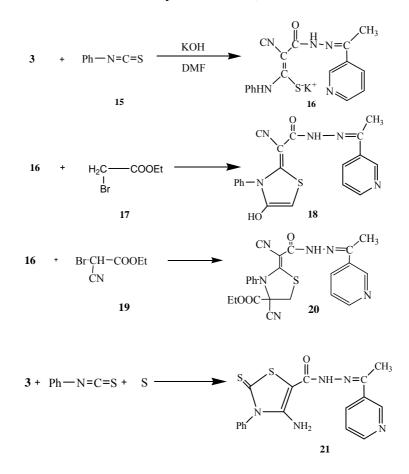
Scheme 4. Synthesis of 11, 13 and 14.



Next, we studied the reactivity of **3** towards phenyl isothiocyanate in basic dimethylformamide solution followed by heterocyclization with active methylene reagents like α -haloketones with the aim of synthesizing thiazole derivatives with potential antitumor activity. Thus, compound **3** reacted with phenylisothiocyanate (**15**) in DMF/KOH solution at room temperature to give the intermediate potassium sulphide salt **16**. Heterocyclization of **16** with α -haloketones like ethyl bromoacetate (**17**) gave the thiophene derivative **18**. The structure of **18** was confirmed based on analytical and spectral data. The ¹H-NMR showed a singlet at δ 2.22 for the CH₃ group, a singlet at δ 7.46 for the thiazole hydrogen, a multiplet at δ 7.54-8.95 for the pyridine H and C₆H₅ and two singlets at δ 8.80/10.13 for the NH and OH groups, respectively. Moreover, the ¹³C-NMR spectrum showed the presence of peaks at δ 14.3 (CH₃), 93.0, 101.2 (C=C), 115.8 (CN), 119.2, 120.5, 121.8, 122.3, 124.0, 124.8, 127.6, 130.5, 133.2, 137.0, 150.7, 152.2 (C₆H₅, thiazole, pyridine C), 168.4 (C=N), 170.0 (C=O).

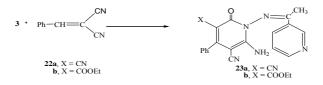
In a similar way, the reaction of **16** with ethyl bromocyanoacetate (**19**) gave the thiazole derivative **20**. Furthermore, compound **3** reacted with phenyl isothiocyanate and elemental sulfur in 1,4-dioxane containing triethylamine to give the thiazole derivative **21** (Scheme 5). The analytical and spectral data of **21** were in agreement with the proposed structure (see Experimental section).

Scheme 5. Synthesis of 18, 20 and 21



Next, we studied the reactivity of the hydrazide-hydrazone derivative **3** towards cinnamonitrile derivatives. Thus, the reaction of **3** with either 2-benzylidenemalononitrile (**22a**) or ethyl 2-cyano-3-phenylacrylate (**22b**) gave the pyridine derivatives **23a** and **23b**, respectively (Scheme 6).

Scheme 6. Synthesis of 23a,b.



Effect on the Growth of Human Tumor Cell Lines

The effect of compounds **3-25** on the *in vitro* growth of three human tumor cell lines representing different tumor types, namely, breast adenocarcinoma (MCF-7), non-small cell lung cancer (NCI-H460) and CNS cancer (SF-268) was evaluated after a continuous 48 h exposure. The results are summarized in Table 1. All of the tested compounds were able to inhibit the growth of the tested human tumor cell lines in a dose-dependant manner (data not shown). The results indicated in Table 1 revealed that compound **3** showed the highest inhibitory effect against all the three tumor cell lines. In addition compound **5c** showed the best inhibitory effect against CNS cancer (SF-268), while compounds **7** and **23a** showed high inhibitory effects against non-small cell lung cancer (NCI-H460) and breast adenocarcinoma (MCF-7), respectively. Compounds **5b**, **9a**, **23b** showed the lowest inhibitory effects. The rest of the compounds showed a moderate growth inhibitory effect. Comparing compound **5b** with **5c**, it is obvious that presence of the CH₃O group in **5c** resulted in a higher inhibitory effect than **5b** with the Cl group. Comparing the pyridine derivatives **23a** (with the cyano group) and **23b** (with the carboxyethyl group), the first has a greater inhibitory effect than the second towards the three cell lines.

GI50 (µM)			
Compound	MCF-7	NCI-H460	SF-268
3	0.1 ± 0.009	0.2 ± 0.001	0.6 ± 0.001
5a	20.0 ± 0.2	26.6 ± 1.4	38.4 ± 0.6
5b	60.6 ± 16.9	38.9 ± 10.8	28.8 ± 8.6
5c	2.0 ± 0.2	3.0 ± 1.6	0.07 ± 0.001
7	66.8 ± 12	10 ± 6.2	36.8 ± 3.0
9a	74.7 ± 17.5	48.2 ± 12.8	62.0 ± 9.01
9b	20 ± 0.4	20.3 ± 0.8	22.2 ± 0.8
9c	28.9 ± 0.9	40.6 ± 1.8	54.8 ± 0.8
11	30 ± 0.6	17.3 ± 1.4	22.3 ± 1.5
13	36.0 ± 1.8	44.0 ± 0.8	20.5 ± 1.1
14	50.1 ± 0.7	23.2 ± 4.8	18.4 ± 1.8
17	22 ± 0.4	20.3 ± 0.8	30 ± 0.8
19	22.0 ± 0.2	24.1 ± 0.8	38.4 ± 0.6
21	35.4 ± 10.2	24.1 ± 0.8	18.9 ± 6.8
23a	11.9 ± 0.5	14.1 ± 0.6	20.3 ± 0.5
23b	70.9 ± 0.9	40.6 ± 1.8	60.8 ± 0.8

Table 1. Effect of the newly synthesized product on the growth of three human tumor cell.

Results are given in concentrations that were able to cause 50 % of cell growth inhibition (GI₅₀). after a continuous exposure of 48 h and show means \pm SEM of three-independent experiments performed in duplicate. Doxorubicin was used as positive control, GI50: MCF-7 = 42.8 \pm 8.2 nM; NCI-H460 = 94.0 \pm 8.7 nM, and SF-280 = 94.0 \pm 7.0 nM.

3. Experimental

3.1. General

Melting points were determined on an Electrothermal melting point apparatus (Electrothermal 9100) and are uncorrected. IR spectra were recorded for KBr discs on a Pye Unicam SP-1000 spectrophotometer. ¹H-NMR and ¹³C-NMR spectra were measured on a Varian EM-390-200 MHz in CD₃SOCD₃ as solvent using TMS as internal standard, and chemical shifts are expressed as δ . Analytical data were obtained from the Microanalytical Data Unit at Cairo University, Giza, Egypt. Antitumor evaluation for the newly synthesized products were performed by a research group at the National Research Center and the National Cancer Institute at Cairo University.

2-*Cyano-N'-(1-(pyridine-3-yl)ethylidene)acetohydrazide* (**3**). To a solution of cyanoacetylhydrazine (**2**, 0.99 g, 0.01 mol) in 1,4-dioxane (20 mL), 3-acetylpyridine (1.21 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 2 h then poured onto a beaker containing an ice/water mixture. The formed solid product was collected by filtration and dried to give white crystals (from ethanol). Yield: 1.50 g, 74%, m.p. 203-205 °C; IR (KBr) ν/cm^{-1} : 3500-3400 (NH), 3066 (CH-aromatic), 2885 (CH₃), 2200 (CN), 1680 (C=O); MS *m/z* (%) 202 [M⁺, 8%]; ¹H-NMR δ : 2.28 (s, 3H, CH₃), 4.26 (s, 2H, CH₂), 7.43-8.99 (m, 4H, pyridine-H),10.81(s, 1H, NH); ¹³C-NMR δ : 14.2 (CH₃), 28.9 (CH₂), 116.8 (CN), 122.1, 123.4, 133.7, 150.2, 151.0 (pyridine C), 168.1 (C=N), 169.8 (C=O); Anal. Calcd. for C₁₀H₁₀N₄O (202.21): C, 59.40; H, 4.98; N, 27.71%. Found: C, 59.72; H, 5.20; N, 28.01%.

3.2. General Procedure for the Synthesis of **5a**, **5b** or **5c**

To a solution of **3** (2.02 g, 0.01 mol) in 1,4-dioxane (20 mL), either benzaldehyde (1.06 g, 0.01 mol), p-chlorobenzaldehyde(1.12 g, 0.01 mol) or p-methoxybenzaldehyde (1.08 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 6 h then poured onto a beaker containing an ice/water mixture. The formed solid product was collected by filtration and dried.

2-*Cyano-3-phenyl-N'-(1-(pyridin-3-yl)ethylidene)acrylohydrazide* (**5a**). White crystals (from acetone). Yield: 1.32 g, 45.4%, m.p. 115-118 °C; IR (KBr) ν/cm^{-1} : 3460-3380 (NH), 3050 (CH-aromatic), 2890 (CH₃), 2220 (CN), 1683 (C=O), 1635 (C=N); MS *m/z* (%) 290 [M⁺, 39.7%];¹H-NMR δ : 2.28 (s, 3H, CH₃), 3.31 (s, 1H, CH), 7.30-7.91 (m, 9H, C₆H₅, pyridine-H), 8.71 (s, 1H, NH); ¹³C-NMR δ : 14.0 (CH₃), 98.3 (<u>C</u>=CH), 148.0 (C=<u>CH</u>), 116.8 (CN), 124.3, 125.7, 15.9, 126.2, 128.0, 129.4, 136.5, 150.2, 151.6 (benzene, pyridine C), 165.3 (C=N), 170.0 (C=O); Anal. Calcd. for C₁₇H₁₄N₄O (290.32): C, 70.33; H, 4.86; N, 19.30%. Found: C, 70.01; H, 4.92; N, 19.48%.

3-(4-Chlorophenyl)-2-cyano-N'-(1-(pyridin-3-yl)-ethylidene)acrylo-hydrazide (**5b**). Pale yellow crystals (from acetone). Yield: 1.95 g, 60.1%, m.p. 100 °C; IR (KBr) ν/cm^{-1} : 3460-3380 (NH), 3050 (CH-aromatic), 2890 (CH₃), 2200 (CN), 1683 (C=O), 1623 (C=N); MS *m*/*z* (%) 324 [M⁺, 5.17%]; ¹H-NMR δ : 2.27 (s, 3H, CH₃), 3.38 (s, 1H, CH), 7.17-7.94 (m, 8H, C₆H₄, pyridine-H), 10.00 (s, 1H, NH); ¹³C-NMR δ : 14.1 (CH₃), 98.0 (<u>C</u>=CH), 147.8 (C=<u>CH</u>), 116.9 (CN), 122.8, 124.6, 15.9, 126.0, 127.1, 128.9, 136.5, 150.2, 151.6 (benzene, pyridine C), 165.3 (C=N), 170.0 (C=O); Anal. Calcd. for C₁₇H₁₃ClN₄O (324.76): C, 62.87; H, 4.03; N, 17.25%. Found: C, 63.01; H, 4.26; N, 17.32%.

2-*Cyano-3-(4-methoxyphenyl)-N'-(1-(pyridin-3-yl)ethylidene)acrylo-hydrazide* (**5c**). Yellow crystals (from acetone). Yield: 2.40 g, 75%, m.p. 130-133 °C; IR (KBr) ν/cm^{-1} : 3460-3370 (NH) , 3100 (CH-aromatic), 2890 (CH₃), 2200 (CN), 1683 (C=O), 1640 (C=N); MS *m/z* (%) 320 [M⁺, 44.28%]; ¹H-NMR δ : 2.49 (s, 3H, CH₃), 3.31(s, 1H, CH), 3.82 (s, 1H, CH₃), 7.03-7.82 (m, 8H, C₆H₄, pyridine-H), 8.62 (s, 1H, NH); ¹³C-NMR δ : 14.4, 54.6 (2 CH₃), 99.9. (<u>C</u>=CH), 148.3 (C=<u>CH</u>), 116.3 (CN), 122.3, 124.6, 125.3, 125.9, 127.8, 128.4, 135.3, 150.0, 151.9 (benzene, pyridine C), 165.6 (C=N), 168.9 (C=O); Anal. Calcd. for C₁₈H₁₆N₄O₂ (320.35): C, 67.49; H, 5.03; N, 17.49%. Found: C, 67.55; H, 4.89; N, 17.31%.

2-*Oxo-N'-(1-(pyridin-3-yl)ethylidene)-2H-chromene-3-carbohydrazide* (**7**). To a solution of **3** (2.02 g, 0.01 mol) in 1,4-dioxane (20 mL), salicylaldehyde (1.22 g,0.01 mol) was added. The reaction mixture was heated under reflux for 3 hours then poured onto a beaker containing an ice/water mixture. The formed solid product was collected by filtration and dried to give white crystals (from ethanol). Yield: 2.10 g, 68%, m.p. 235-238 °C; IR (KBr) ν/cm^{-1} : 3450-3360 (NH), 3200 (CH-aromatic), 2890 (CH₃), 2200 (CN), 1670 (C=O), 1650 (C=N); MS *m/z* (%) 307 [M⁺, 2.5%]; ¹H-NMR δ 2.51(s, 3H, CH₃), 7.30 (s, 1H, coumarin H-4), 7.48-9.33 (m, C₆H₄, pyridine-H), 13.69 (s, 1H, NH); ¹³C-NMR δ : 13.8 (CH₃), 121.8, 122.6, 122.9, 124.5, 127.1, 128.0, 130.9, 131.6, 150.2, 152.3 (C₆H₄, pyridine C), 159.9 (coumarin C=O), 167.9 (C=N), 169.8 (C=O); Anal. Calcd. for C₁₇H₁₃N₃O₃ (307.30): C, 66.44; H, 4.26; N, 13.67%. Found: C, 66.38; H, 4.35; N, 13.92%.

3.3. General Procedure for the Synthesis of 9a, 9b, 9c or 9d

To a cold solution of **3** (2.02 g, 0.01 mol) in ethanol (20 mL) containing sodium acetate (3.0 g) was added with continuous stirring either of the appropriate substituted benzenediazonium salt (0.01 mol) [prepared by adding sodium nitrite (1.6 g, 0.02 mol) in water (8 mL) to a cold solution of either of the appropriate substitute aniline **8a-d** in the appropriate amount of hydrochloric acid]. The reaction mixture was stirred for 2 h and the formed solid product, in each case, was collected by filtration.

2-*Cyano-2-(2-phenylhydrazinylidene)-N'-[1-(pyridin-4-yl)ethylidene]aceto-hydrazide* (**9a**). Orange crystals (from ethanol). Yield: 1.92 g, 63%, m.p. 158-161 °C; IR (KBr) ν/cm^{-1} : 3500-3400 (NH), 3050 (CH-aromatic), 2800 (CH₃), 2200 (CN), 1696 (C=O); MS *m/z* (%) 306 [M⁺, 93.51%]; ¹H-NMR δ 2.49 (s, 3H, CH₃), 7.11-8.22 (C₆H₅, pyridine H), 10.25, 12.02 (2s, 2H, 2NH); ¹³C-NMR δ : 14.1 (CH₃), 116.0 (CN), 115.9, 116.8, 118.6, 122.3, 123.9, 129.0, 142.8, 150.2, 152.8 (C₆H₅, pyridine C), 163.2, 166.7 (2 C=N), 168.8 (C=O); Anal. Calcd. for C₁₆H₁₄N₆O (306.32): C, 62.74; H, 4.61; N, 27.44%. Found: C, 62.90; H, 4.51; N, 27.52%.

2-[2-(4-Chlorophenyl)hydrazinylidene]-2-cyano-N'-[1-(pyridin-4-yl)ethylidene] acetohydrazide (**9b**). Orange crystals (from ethanol). Yield: 2.32 g, 68%, m.p. 174-177 °C; IR (KBr) ν/cm^{-1} : 3600-3500 (NH), 3100 (CH-aromatic), 2890 (CH₃), 2200 (CN), 1696 (C=O); MS *m/z* (%) 340 [M⁺, 6.17%]; ¹H-NMR δ 2.49 (s, 3H, CH₃), 6.95-7.89 (C₆H₄-pyridine H), 11.737, 12.75 (2s, 2H, 2NH); ¹³C-NMR δ : 14.0 (CH₃), 115.8 (CN), 116.0, 117.3, 118.9, 121.8, 124.5, 125.4, 143.5, 150.0, 151.8 (C₆H₅, pyridine C), 163.0, 166.9 (2 C=N), 168.9 (C=O); Anal. Calcd. for C₁₆H₁₃ClN₆O (340.77): C, 56.39; H, 3.85; N, 24.66%. Found: C, 56.60; H, 4.01; N, 24.90%. 2-[2-(4-Bromophenyl)hydrazinylidene]-2-cyano-N'-[1-(pyridin-4-yl)ethylidene] acetohydrazide (**9**c). Orange crystals (from ethanol). Yield: 2.62 g, 68%, m.p. 183-186 °C; IR (KBr) ν/cm^{-1} : 3550-3400 (NH), 3090 (CH-aromatic), 2890 (CH₃), 2200 (CN), 1685(C=O); MS *m/z* (%) 387 [M⁺, 19 %]; ¹H-NMR δ 2.49 (s, 3H, CH₃), 6.93-7.84 (C₆H₄, pyridine H), 9.50, 11.75 (2s, 2H, 2NH); ¹³C-NMR δ : 14.2 (CH₃), 116.6 (CN), 116.3, 118.0, 118.6, 122.8, 124.5, 128.0, 138.9, 150.2, 152.5 (C₆H₅, pyridine C), 163.0, 166.5 (2 C=N), 168.5 (C=O); Anal. Calcd. for C₁₆H₁₃BrN₆O (384): C, 49.89; H, 3.40; N, 21.82%. Found: C, 48.93; H, 3.62; N, 22.02%.

2-*Cyano-2-[2-(4-nitrophenyl)hydrazinylidene]-N'-[1-(pyridin-4-yl)ethylidene]* acetohydrazide (**9d**). Orange crystals (from ethanol). Yield: 2.43 g, 69%, m.p. 162-165 °C; IR (KBr) ν/cm^{-1} : 3570-3400 (NH), 3100 (CH-aromatic), 2890 (CH₃), 2200 (CN), 1670 (C=O); MS *m/z* (%) 352 [M⁺, 20.2 %]; ¹H-NMR δ 2.49 (s, 3H, CH₃), 6.57-8.39 (C₆H₄, pyridine H), 10.48, 12.11 (2s, 2H, 2NH); ¹³C-NMR δ : 14.1 (CH₃), 116.0 (CN), 116.8, 116.9, 118.2, 123.8, 126.3, 128.6, 136.1, 151.0, 152.7 (C₆H₅, pyridine C), 163.1, 166.4 (2 C=N), 168.9 (C=O); Anal. Calcd. for C₁₆H₁₃N₇O₃ (351.32): C, 54.70; H, 3.73; N, 27.91%. Found: C, 54.85; H, 3.90; N, 29.11%.

2-Amino-4,5,6,7-tetrahydro-N'-(1-(pyridin-3-yl)ethylidene)benzo[b]thiophene-3-carbohydrazide (11). Method A: To a solution of **3** (2.02 g, 0.01 mol) in ethanol (40 mL) containing triethylamine (1 mL) and elemental sulfur (0.32 g, 0.01mol), cyclohexanone **10** (0.98 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 3 hours then poured onto a beaker containing an ice/water mixture. The formed solid product was collected by filtration and dried obtaining pale yellow crystals (from ethanol).

Method B: To a solution of compound **14** (2.82 g, 0.01 mol) in 1,4-dioxane (40 mL) containing triethylamine (0.5 mL), elemental sulfur (0.32 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 2 h then poured onto ice/water containing few drops of hydrochloric acid. The formed solid product was collected by filtration.

Yield: 1.95g, 62% (method A) and 2.20 g, 70% (method B), m.p. 112 °C; IR (KBr) ν/cm^{-1} : 3400-3300 (NH₂, NH), 3068 (CH-aromatic), 2886 (CH₃), 2250 (CN), 1690 (C=O), 1638 (C=C); MS *m/z* (%) 314 [M⁺, 2.19 %]; ¹H-NMR δ 2.29-2.31 (m, 8H, 4CH₂), 2.49 (s, 3H, CH₃), 4.25 (s, 2H, NH₂), 7.41-8.88 (m, 4H, pyridine-H), 10.76 (s, 1H, NH); ¹³C-NMR δ : 13.8 (CH₃), 22.8, 23.6, 24.1, 24.9 (4 CH₂), 116.5, 122.6, 124.8, 130.2, 150.6, 151.8 (thiophene, pyridine C), 163.6 (C=O), 170.0 (C=N); Anal. Calcd. for C₁₆H₁₈N₄OS (314.41): C, 61.12; H, 5.77; N, 17.82%. Found: C, 60.91; H, 6.01; N, 17.85%.

2-Amino-5,6-dihydro-N'-(1-(pyridin-3-yl)ethylidene)-4H-cyclopenta[b]thiophene-3-carbohydrazide

(13). To a solution of 3 (2.02 g, 0.01 mol) in ethanol (40 mL) containing triethylamine (1.0 mL) and elemental sulfur (0.32 g, 0.01 mol), cyclopentanone 12 (0.98 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 3 h then poured onto a beaker containing an ice/water mixture. The formed solid product was collected by filtration and dried obtaining pale yellow crystals (from ethanol).Yield: 1.82 g, 61%, m.p. 140-144 °C; IR (KBr) ν/cm^{-1} : 3450-3300 (NH₂, NH), 3080 (CH-aromatic), 2890 (CH₃), 2250 (CN), 1690 (C=O); MS *m/z* (%) 300 [M⁺, 0.34 %]; ¹H-NMR δ 2.28-2.35 (m, 6H, 3CH₂), 2.48 (s, 3H, CH₃), 4.25 (s, 2H, NH₂), 7.41-8.99 (m, 4H, pyridine-H),10.78 (s, 1H, NH); ¹³C-NMR δ : 14.0 (CH₃), 20.8, 24.6, 26.9 (3 CH₂), 116.9, 123.0, 124.6, 133.1, 150.1, 151.4 (thiophene,

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pyridine C), 163.4 (C=O), 169.8 (C=N); Anal. Calcd. for C₁₅H₁₆N₄OS (300.3): C, 59.98; H, 5.37; N, 18.65%. Found: C, 59.93; H, 5.39; N, 18.81%

2-*Cyano-2-cyclohexylidene-N'-(1-(pyridin-3-yl)ethylidene)acetohydrazide* (14). Equimolar amounts of compound **3** (2.02 g, 0.01 mol) and cyclohexanone **10** (0.98 g, 0.01 mol) were heated in an oil bath at 140 °C for 1 h in presence of ammonium acetate. After cooling the reaction mixture, it was heated in ethanol, then poured into ice/water mixture and the formed solid product was collected by filtration and dried to give pale yellow crystals (from ethanol). Yield: 1.52 g, 54%, m.p. 145-146 °C; IR (KBr) ν/cm^{-1} : 3355-3370 (NH), 3067 (CH-aromatic), 2930 (CH₃), 2200 (CN), 1675 (C=O); MS *m/z* (%) 282 [M⁺, 0.40 %]; ¹H-NMR δ 2.26-2.34 (m, 10H, 5CH₂), 2.31(s, 3H, CH₃), 7.41-8.99 (m, 4H, pyridine-H), 10.90 (s, 1H, NH); ¹³C-NMR δ : 13.9 (CH₃), 26.8, 27.4, 26.9, 27.0 (cyclohexane C), 93.0 (C=C), 116.8 (CN), 123.4, 126.8, 126.4, 150.5, 151.2 (pyridine C), 168.2 (C=N), 177.3 (C=O); Anal. Calcd. for C₁₆H₁₈N₄O (282.34): C, 68.06; H, 6.43; N, 19.84%. Found: C, 67.85; H, 6.31; N, 20.22%

3.4. General Procedure for the Synthesis of 18 and 20

Compound 3 (2.02 g, 0.01 mol) is dissolved in ethanol and a few sodium hydroxide pellets were added. Phenylisothiocyanate (15, 1.35, 0.01 mol) is then added and the solution is covered and left standing overnight. Equimolar amounts of either ethyl 2-bromoacetate (17) or ethyl 2-bromo-2-cyanoacetate (19) are stirred in the following day, and the solution is covered for another night, after which the reaction mixture is poured onto ice and the precipitated solid is filtered off.

2-(4-Hydroxy-3-phenylthiazol-2(3H)-ylidene)-2-isocyano-N'-(1-(pyridin-3-yl)ethylidene) acetohydrazide (18). Orange crystals (from ethanol). Yield: 2.43 g, 64%, m.p. 125-127 °C; IR (KBr) v/cm⁻¹: 3500-3370 (NH), 3400 (OH), 3100 (CH-aromatic), 2900 (CH₃), 2189 (CN), 1675 (C=O); MS m/z (%) 377 [M⁺, 7.9 %]; ¹H-NMR δ 2.22 (s, 3H, CH₃), 7.46 (thiazole H-5),7.54-8.95 (C₆H₅-pyridine H), 8.80 (NH), 10.13 (OH); ¹³C-NMR δ : 14.3 (CH₃), 93.0, 101.2 (C=C), 115.8 (CN), 119.2, 120.5, 121.8, 122.3, 124.0, 124.8, 127.6, 130.5, 133.2, 137.0, 150.7, 152.2 (C₆H₅, thiazole, pyridine C), 168.4 (C=N), 170.0 (C=O); Anal. Cald. for C₁₉H₁₅N₅O₂S (377.42): C, 60.46; H, 4.01; N, 18.56; S, 8.50. Found: C, 60.50; H, 4.01; N, 18.55; S, 8.48%.

(2Z)-Ethyl-2-((1-(pyridin-3-yl)ethylideneaminocarbamoyl)(cyano)methylene)-4-cyano-3-phenylthiazolidine -4-carboxylate (**20**). Orange crystals (from ethanol).Yield: 3.22 g, 70%, m.p. 125-127 °C; IR (KBr) ν/cm^{-1} : 3577-3370 (NH), 3067 (CH-aromatic), 2930 (CH₃), 2200 (CN), 1675 (C=O); MS *m/z* (%) 461.1 [M⁺, 25.2 %]; ¹H-NMR δ 1.56 (t, 3H, *J* = 7.02 Hz, CH₃), 2.26 (s, 3H, CH₃), 4.25 (q, 2H, *J* = 7.02 Hz, CH₂), 6.47 (s, 2H, thiazole, CH₂), 7.08-8.12 (m, 9H, C₆H₅, pyridine-H), 10.72 (s, 1H, NH); ¹³C-NMR δ : 13.9, 14.5 (2 CH₃), 40.2 (thiazole CH₂), 58.9 (ester CH₂), 93.3, 101.6 (C=C), 116.0, 116.7 (2 CN), 119.2, 120.3, 121.2, 121.8, 122.0, 124.7, 125.3, 133.5, 133.9, 150.4, 152.1 (C₆H₅, thiazole, pyridine C), 160.2, 164.5 (2 C=O), 168.9 (C=N); Anal. Calcd. for C₂₃H₂₀N₆O₃S (460.51): C, 59.99; H, 4.38; N, 18.25; S, 6.96%. Found: C, 60.11; H, 4.42; N, 18.13; S, 7.26%.

4-Amino-2,3-dihydro-3-phenyl-N'-(1-(pyridin-3-yl)ethylidene)-2-thioxothiazole-5-carbohydrazide (21). To a solution of 3 (2.02 g, 0.01 mol) in ethanol (40 mL) containing triethylamine (1.0 mL) and elemental sulfur (0.32 g, 0.01 mol), phenylisothiocyanate (15, 1.35 g, 0.01 mol) was added. The

reaction mixture was heated under reflux for 3 h then poured onto a beaker containing an ice/water mixture. The formed solid product was collected by filtration and dried obtaining yellow crystals (from ethanol). Yield: 2.46 g, 67%, m.p. 164-167 °C; IR (KBr) ν/cm^{-1} : 3465-3300 (NH₂, NH), 3166 (CH-aromatic), 2980 (CH₃), 1680 (C=O), 1658 (C=N), 1466 (C=C), 1241(C=S); MS *m/z* (%) 369.9 [M⁺, 13.27%]; ¹H-NMR δ 2.33 (s, 3H, CH₃), 3.31(s, 2H, NH₂),7.28-7.38 (m, C₆H₄-pyridine H), 10.67 (s, 1H, NH); ¹³C-NMR δ : 14.2 (CH₃), 120.3, 122.5, 124.1, 127.9, 128.3, 130.1, 133.4, 138.9, 150.0, 152.3 (C₆H₅, thiazole, pyridine C), 168.2 (C=N), 170.2 (C=O), 180.3 (C=S); Anal. Calcd. for C₁₇H₁₅N₅OS₂ (369.46): C, 55.26; H, 4.09; N, 18.96; S, 17.36 %. Found: C, 55.40; H, 4.31; N, 19.15; S, 17.60%.

3.5. General Procedure for the Synthesis of 23a or 23b

To a solution of **3** (2.02 g, 0.01 mol) in 1,4-dioxane (20 mL) either 2-benzylidenemalononitrile (**22a**, 1.54 g, 0.01 mol) or ethyl 2-cyano-3-phenylacrylate (**22b**, 2.01 g, 0.01 mol) was added. The reaction mixture was heated under reflux for 3 h then poured onto a beaker containing an ice/water mixture. The formed solid product was collected by filtration and dried.

1-(1-Phenylethylideneamino)-6-amino-1,2-dihydro-2-hydroxy-4-phenylpyridine-3,5-dicarbonitrile (**23a**). White crystals (from ethanol). Yield: 2.40 g, 68%, m.p. >300 °C; IR (KBr) ν/cm^{-1} : 3458-3328 (NH₂, NH), 3215 (CH-aromatic), 2890 (CH₃), 2192, 2225 (2 CN), 1688 (C=O), 1640 (C=N); MS *m/z* (%) 355 [M⁺, 4.9 %]; ¹H-NMR δ 2.301 (s, 3H,CH₃), 3.56 (s, 2H, NH₂), 7.46-9.22 (m, C₆H₅, pyridine H); ¹³C-NMR δ: 18.9 (CH₃), 116.9, 118.0 (2 CN), 110.2, 118.9, 120.6, 128.4, 137.2, 150.2, 152.4 (two pyridine C), 168.9 (C=N), 172.3 (C=O); Anal. Calcd. for C₂₀H₁₄N₆O (354.36): C, 67.79; H, 3.98; N, 23.72%. Found: C, 67.80; H, 4.00; N, 23.71%.

Ethyl-1-(1-phenylethylideneamino)-6-amino-5-cyano-1,2-dihydro-2-hydroxy-4-phenylpyridine-3-

carboxylate (**23b**). White crystals (from ethanol).Yield: 1.96 g, 49%, m.p. 255-259 °C; IR (KBr) ν/cm^{-1} : 3464-3339 (NH₂), 3200 (CH-aromatic), 2890 (CH₃), 2180 (CN), 1680 (C=O), 1640 (C=N); MS *m*/*z* (%) 401 [M⁺, 2.8 %]; ¹H-NMR δ : 1.65 (t, 3H, *J* = 6.83 Hz, CH₃), 2.31 (s, 3H, CH₃), 3.56 (s, 2H, NH₂), 4.18 (q, 2H, *J* = 6.83 Hz, CH₂), 7.27-8.79 (m, C₆H₅, pyridine H); ¹³C-NMR δ : 13.7, 19.0 (2 CH₃), 59.8 (CH₂), 116.9 (CN), 108.0, 117.4, 120.3, 124.6, 125.1, 127.0, 132.0, 138.3, 150.2, 152.4 (C₆H₅, two pyridine C), 167.3 (C=N), 170.1, 172.6 (2C=O); Anal. Calcd. for C₂₂H₁₉N₅O₃ (401.42): C, 65.83; H, 4.77; N, 17.45%. Found: C, 66.04; H, 4.83; N, 17.61%.

4. Conclusions

In this work, cyanoacetylhydrazine (1) reacted with 3-acetylpyridine (2) to afford the hydrazidehydrazone derivative **3**. The latter was reacted with different reagents to give coumarin, pyridine, thiazole and thiophene derivatives. The antitumor evaluations of the newly synthesized products were carried out, showing that both the hydrazide-hydrazone derivative **3** and the benzylidene derivative **5**c have the highest inhibitory effects.

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