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1 On the potential of lignin-containing cellulose nanofibrils (LCNFs): a review on

2 properties and applications

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15

16 Abstract:

This review outlines the present state and recent progress in the area of lignin-containing cellulose nanofibrils (LCNFs), an emerging family of green cellulose nanomaterials. Different types of LCNF raw materials are described, with main focus on wood-based raw materials, and the properties of the resulting LCNFs are compared. Common problems faced in industrial utilization of CNFs are discussed in the light of potential improvements from LCNFs, covering areas such as chemical and energy consumption, dewatering and redispersibility. Out of the potential applications, barrier films, emulsions and nanocomposites are considered.

24 Keywords: *lignin-containing nanocellulose, lignin, cellulose nanofibrils, nanofibrillated cellulose*

25 Introduction

During the last decades, the exponential growth of technology has allowed manipulating materials 26 at scales lower than 100 nm. At this scale, fundamental material properties show different 27 behaviors than those corresponding to the macroscale (Kamel 2007). Simultaneously, the 28 29 utilization of bio-based products, mainly cellulose, to replace petroleum-based materials has generated an enormous impact, which can be directly seen on the increasing body of research on 30 31 cellulosic nanomaterials, or nanocellulose, over the years. As is the case with other nanomaterials, also for nanocellulose, the interfacial interactions and chemical composition play a vital role in the 32 properties of the material and on their interactions with other components (Ratner et al. 2013). 33

Nanocellulose can be obtained from different raw materials utilizing diverse approaches. The two most common man-made types are cellulose nanocrystals (CNCs) – produced by acid hydrolysis – and cellulose nanofibrils (CNFs) – produced by mechanical disintegration of the respective lignocellulosic fibers. Since the production methods are different, the properties of the fibers, as well as their final application, will vary among these nanoparticles (Klemm et al. 2011; Moon et al. 2011; Lavoine et al. 2012). In this review, we focus on CNFs.

Over time, research on CNFs has generally centered on using fully bleached cellulose nanofibrils
(BCNFs). Nevertheless, during the last years, efforts have been made to incorporate lignincontaining cellulose nanofibrils (LCNFs) in different applications, like additives in papermaking
(Delgado-Aguilar et al. 2016); neat and composite LCNF films (Rojo et al. 2015; Horseman et al.

2017); or fillers and reinforcing agents in different matrices (Sun et al. 2014; Wang et al. 2014; 44 Ago et al. 2016; Ballner et al. 2016; Ferrer et al. 2016; Herzele et al. 2016; Visanko et al. 2017a). 45 This review aims to provide a complete overview of the current state of LCNF research; mainly 46 focused on the effect of chemical composition on their processing, properties, and applications. 47 The chemistry involved in pulping and bleaching processes is considered in an attempt to elucidate 48 their impact on the final nanocellulose properties. Additionally, the role of lignin and 49 hemicelluloses in LCNF fibers is carefully described. Finally, a summary of applications such as 50 51 barrier films, emulsions, and nanocomposites is presented.

52 Structure and chemistry of wood-based biomass

In native state. Wood forms a major part of the plant-based biomass in the world. Structurally, wood can be described as a complex three-dimensional nanocomposite, composed of aligned cellulose microfibrils that are embedded in a matrix of lignin and heteropolysaccharides (mainly hemicelluloses). As a simplification, cellulose microfibrils act as the major load-bearing component in wood, whereas the role of the surrounding matrix is to transfer mechanical stress across the structure, to control the content of water in the wood tissue and to protect the tree against pathogens and insects (Higuchi 1985; Hatfield and Vermerris 2001).

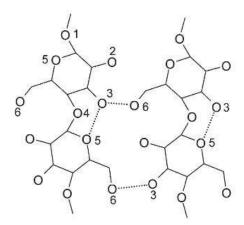
Despite ongoing controversies on the exact structure and composition of the native plant cell wall 60 (Keegstra 2010; Crestini et al. 2011; Agarwal et al. 2016), it is commonly accepted that the average 61 cellulose microfibril angle is different at different cell wall layers (primary, secondary and tertiary 62 cell wall layers) and that these microfibrils consist of alternating crystalline and less ordered 63 domains. Cellulose microfibrils are surrounded by a crosslinked lignin matrix (Fengel and 64 Wegener 1984; Kilpeläinen et al. 2007) that is covalently bound to surrounding carbohydrates 65 (Minor 1986; Tenkanen et al. 1999; Lawoko et al. 2005) in the form of the so-called lignin-66 carbohydrate complexes (LCC). For the purpose of this review, it suffices to say that wood species 67 are generally divided into softwoods (SW) and hardwoods (HW) that have differences both in 68 structural complexity and chemical composition; the latter been summarized in Table 1. 69

Table 1 Chemical compositions of softwood and hardwood as weight percentages of dry wood.
 Adapted from Sjöström and Westermark (1999) with permission of Springer Nature.

Wood type Cellulose Hemicelluloses Lignin Extractives

Softwood	37-43	20-30	25-33	2-5
Hardwood	39-45	17-35	20-25	2-4

Cellulose is the main component in wood tissue (~40% of dry weight). It is a semicrystalline, linear 72 homopolymer that consists of thousands of anhydro-D-glucopyranosyl units linked by β -(1-4)-73 glycosidic bonds; the degree of polymerization (DP) depending on the source and treatment history 74 75 of the cellulose, ranging between 15,000 in cotton cellulose and about 10,000 in native wood (Sjöström 1993). A number of different crystalline structures have been identified for cellulose 76 (O'Sullivan 1997), but native celluloses are of crystallinity type I (hydrogen bond structure of 77 cellulose I shown in Fig. 1). Furthermore, for plant-based materials, the dominant crystalline form 78 79 is cellulose IB that consists of two conformationally different chains in a monoclinic unit cell (Atalla and VanderHart 1984). In contrast, its allomorph Ia has one molecular chain in a triclinic 80 81 unit cell, found in certain algae and bacterial cellulose (Kontturi et al. 2006). In addition to the 82 crystalline parts, cellulose contains a significant portion of less ordered or 'amorphous' regions that are more accessible to water and chemically more reactive than the crystalline domains. As is 83 the case with the DP, also the crystallinity index (CrI) depends heavily on the raw material and its 84 chemical and mechanical processing (O'Sullivan 1997; Liitiä et al. 2003; Ago et al. 2004; Kontturi 85 86 et al. 2006; Karinkanta et al. 2013).

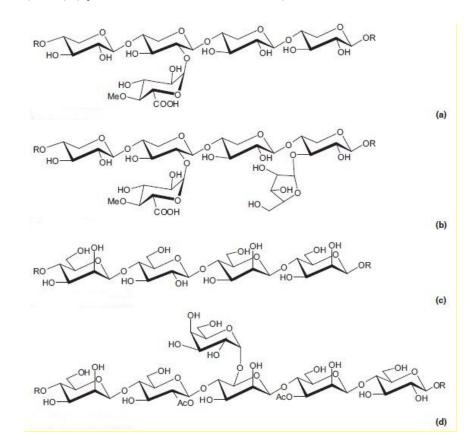


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Fig. 1 The supramolecular structure of native cellulose I. Reprinted from Kontturi et al. (2006)
with permission of Royal Society of Chemistry (Note: Hydrogen atoms are not indicated)

90 The next class of wood biopolymers to be discussed are hemicelluloses, amorphous 91 heteropolysaccharides that have structural resemblance to cellulose but have lower molecular

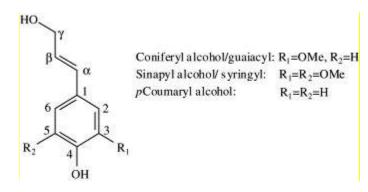
weight and often contain side chains (Fig. 2). Some of the hemicelluloses present in the cell wall 92 are covalently linked to lignin (Lawoko et al. 2005; Iversen and Wännström 2009; Brunow and 93 Lundquist 2010), and have a high affinity to cellulose (Eronen et al. 2011; Arola et al. 2013; 94 Villares et al. 2015). These properties enable hemicelluloses to act as compatibilizers between the 95 cellulose microfibrils and the lignin-rich matrix surrounding them. Moreover, hemicelluloses have 96 been proposed to direct the structural development of native plant cell wall during its biosynthesis 97 (Atalla et al. 1993; Martínez-Sanz et al. 2016). Perhaps related to the structural differences in SWs 98 and HW fibers, also their hemicellulose compositions are different. As a generalization, the 99 SWs typically consist of galactoglucomannans (15-20%) 100 hemicelluloses in and arabinoglucuronoxylans (5-10%), whereas HWs are rich in glucuronoxylans (15-30%) and 101 glucomannans (2-5%) (Sjöström and Westermark 1999). 102



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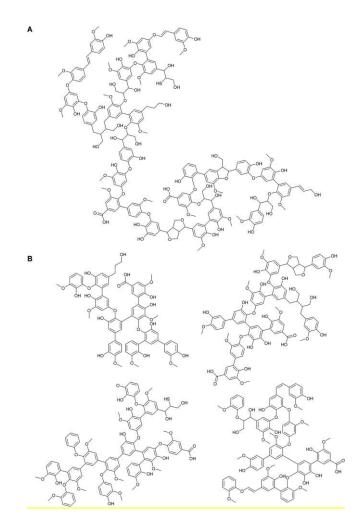
Fig 2. Molecular structures of hemicelluloses on HWs and SWs; a) Harwood xylan, b) Softwood xylan, c) Hardwood glucomannan, and d) Softwood glucomannan. Adapted from Sixta (2006)

Whereas cellulose and hemicelluloses are both carbohydrates, lignin is a complex amorphous 107 polymer that consists of phenyl propane units with one, two or three methoxy groups (Fig 3). It is 108 109 often referred to in plural form to accentuate the enormous variation in its structure, depending on where and how it is isolated (proposed structure of softwood kraft lignin shown in Fig 4). Due to 110 its complex network structure (Brunow et al. 1999; Hatfield and Vermerris 2001; Ralph et al. 111 2004) and covalent attachment to surrounding carbohydrates (Lawoko et al. 2005; Iversen and 112 Wännström 2009), the precise chemical structure and molecular weight of native lignin remains 113 unknown, despite of the number of extensive studies on the chemical composition of isolated 114 lignins (Froass et al. 1996; Ämmälahti et al. 1998; Capanema et al. 2001, 2004; Brunow 2005; 115 Balakshin et al. 2009; Brunow and Lundquist 2010; Crestini et al. 2011, 2017; Alekhina et al. 116 2015; Lupoi et al. 2015). In general, it can still be said that lignin is more hydrophobic than the 117 carbohydrates that form the majority of the plant cell wall. For this reason, it has a major role in 118 enabling water transport in living plants, as well as protecting them against microbial and fungal 119 attacks (Ruiz-Dueñas and Martínez 2009). 120



121

Fig 3. Three lignin precursors. Reprinted from Chakar and Ragauskas (2004) with permission from
 Elsevier.



124

Fig 4. Proposed constitutional structural schemes for softwood kraft lignin. (a) Acetone insoluble fraction (AIKL); (b) acetone soluble fraction (ASKL). Reprinted from Crestini et al. (2017) with permission of the Royal Chemical Society.

Extractives, the remaining class of wood components, are a chemically diverse class of low-128 molecular weight components that typically comprise about 2-5% of wood, but their amount 129 130 greatly varies, depending on the wood species and tissue in question (Fengel and Wegener 1984; Brunow et al. 1999). Their biological role is to provide microbial resistance to the wood tissue. 131 Details of their chemistry and properties are outside the scope of this review. For the purposes of 132 the present paper, it suffices to state that most extractives are lipophilic, and some possess 133 134 antioxidant or antimicrobial properties. Examples of common wood extractives include components such as fatty and resin acids and lignans (Willför et al. 2003a, b). Despite their low 135 content, extractives have a tendency to enrich on the fiber surfaces (Laine et al. 1994; Fardim and 136 Holmbor 2005), potentially impacting the surface chemistry of especially unbleached pulps. 137

Chiefly, it is important to bear in mind that as wood biomass is pulped, bleached, and processed 138 into cellulose nanofibrils (CNFs), it undergoes many structural and chemical changes. These will 139 140 be outlined briefly in the following sections to prepare the reader for the subsequent discussion of 141 lignin-containing pulps and their potential as a raw material for cellulose nanofibrils. For a more thorough overview on the subject of wood chemical composition, the reader is advised to consult 142 143 more extensive reviews and textbooks on the chemistry of wood and its individual components, such as the ones cited here (Fengel and Wegener 1984; Salmén and Olsson 1998; Sjöström and 144 Westermark 1999; Grabber et al. 2004; Ralph et al. 2004, 2007; Klemm et al. 2005; Glasser et al. 145 2012; Lupoi et al. 2015). 146

Mechanical pulping. Generally speaking, mechanical pulps have approximately the same 147 148 chemical composition as native wood. However, the combination of mechanical shear and elevated temperature (typically about 120 °C) at a high moisture content causes some changes in the 149 150 structure and chemistry of the material. Notably, part of the heteropolysaccharide fraction is degraded, dissolved and resorbed during mechanical pulping (Thornton et al. 1994). The elevated 151 152 temperature and high moisture content of the mechanical pulping process cause wood to soften (Salmén 1984; Blechschmidt et al. 1986), enabling the separation of individual fibers. 153 Furthermore, depolymerisation of the cell wall polymers occurs both by thermal and mechanical 154 means, resulting in the formation of reactive free radicals in the course of mechanical pulping (Hon 155 156 1979; Widsten et al. 2001). Widsten et al. (2001) reported that high-temperature mechanical pulping causes lignin depolymerisation and formation of phenoxy radicals. However, the presence 157 of phenoxy radicals may be a consequence of their high stability and not necessarily of their role 158 as the primarily formed radicals. The significance of radical formation in CNF manufacturing will 159 160 be addressed in a later section (Lignin as an antioxidant).

As a process, the major advantage of mechanical pulping is its high yield, which is typically above 80% of the original wood material. However, the high yield comes at the expense of using significant amounts of electricity and achieving only a limited degree of fiber-fiber bonding – a critical property in end uses such as papermaking. For this reason, different chemical treatments have been developed to ease the separation of wood fibers. These processes can be classified to chemi-mechanical and chemical pulping processes, both of which will be outlined briefly in the following sections. 168 **Chemi-mechanical pulping.** In addition to purely mechanical pulps, there is another class of so-169 called high yield pulps, namely chemi-mechanical pulps, where mechanical treatment is preceded 170 by a sulphite treatment in alkaline or neutral conditions (Annergren and Rydholm 1959; Gellerstedt 171 et al. 1976; Hanhikoski et al. 2016b). The lower energy consumption associated with chemi-172 mechanical pulping is most often attributed to sulphonation of lignin that allows a more extensive 173 swelling of the fibers by water, accompanied by more extensive softening of the material (Beatson 174 et al. 1984; Konn et al. 2002).

In alkaline conditions, a significant portion of heteropolysaccharides gets dissolved. This effect is 175 much less pronounced in near-neutral sulphite pulping but the degree of delignification remains 176 lower unless the reaction is catalysed with e.g. anthraquinone (Annergren et al. 1961; Gellerstedt 177 178 et al. 1976; Konn et al. 2002; Hanhikoski et al. 2016b). From the point of view of nanofibrillation, it is important to note that both -a high degree of lignin sulphonation and the preservation of wood 179 180 heteropolysaccharides – are expected to promote fiber swelling, thus facilitating the production of cellulose nanofibrils. Lahtinen et al. (2014) observed improved nanofibrillation of chemi-181 182 mechanical pulps in comparison to the corresponding mechanical pulps. More recently, Hanhikoski et al. (2016a) found that near-neutral sodium sulphite pulps with yields 58-64% can 183 be efficiently fibrillated to LCNFs, which was interpreted as an indication of high hemicellulose 184 content promoting fibrillation, provided that the remaining lignin was sufficiently degraded. 185

Chemical pulping. A major difference of chemical pulps to mechanical and chemi-mechanical pulps is that their yield is only about 40-50%. Unbleached chemical pulps contain 2-5% residual lignin, and fully bleached chemical pulps contain only trace amounts of it (<0.1%) (Sjöström and Westermark 1999; Chakar and Ragauskas 2004). Furthermore, the DP of their polysaccharides is lower than in native wood – typically 2,000-4,000 for cellulose and 50-100 for hemicelluloses (Sjöström and Westermark 1999). Examples of typical polysaccharide compositions in chemical pulps are given in Table 2.</p>

Table 2 Polysaccharide compositions of common bleached chemical pulps as weight percentages
 of dry pulp. Adapted from Sjöström and Westermark (1999) with permission of Springer Nature

Pulp type	Cellulose	Glucomannan	Xylan
Pine kraft	81.3	8	10.7

Birch kraft	67.5	1.9	30.6
Spruce sulphite	82.9	8.8	8.3

Kraft pulping, the most common industrially used chemical pulping method, is chosen as an example to demonstrate the chemical changes that occur in wood upon chemical treatment. During kraft pulping, wood chips are subjected to alkaline conditions (an aqueous solution of NaOH and Na₂S) at an elevated temperature (170 °C), causing lignin to break down into smaller, watersoluble molecules (Chakar and Ragauskas 2004). Towards the end of the cooking, the selectivity of the process decreases, and degradation of carbohydrates, especially hemicelluloses, takes place.

In comparison to the extensively fractured soluble lignin, the residual lignin that remains in the 201 pulp has a much higher molecular weight and contains more carbohydrates, as can be expected 202 from the presence of the aforementioned lignin-carbohydrate complexes. It has been reported for 203 204 softwood kraft pulps that some of the LCCs are broken by kraft pulping, whereas some remain intact even after oxygen delignification (Tenkanen et al. 1999; Lawoko et al. 2004, 2005). 205 Moreover, studies have shown that the lignin fraction that is attached to glucomannan undergoes 206 condensation and is thus of higher molecular weight than the more extensively degraded lignin 207 208 fraction that is attached to xylan (Lawoko et al. 2005).

209 **Bleaching.** A detailed description of pulp bleaching chemistry is outside the scope of this review, but it is worth mentioning that before commercial use, both mechanical and chemical pulps are 210 211 typically bleached to reach a higher brightness. This is especially important for kraft pulps that undergo significant colour formation in the alkaline conditions of the process (Falkehag et al. 1966; 212 Ziobro 1990). For the purpose of this review, it is sufficient to say that pulping strategies are 213 generally based on two main principles: either removing lignin or changing the chemical structure 214 215 of lignin and carbohydrates to remove their colour-causing structures, referred to as chromophores (Agarwal and Atalla 1994; Vuorinen et al. 2004; Rosenau et al. 2007; Jääskeläinen et al. 2009). 216

For understanding the chemical changes taking place during the bleaching of chemical pulps, let us consider a SW kraft pulp that is bleached by an oxidative chemical, such as ozone, oxygen, or chlorine dioxide. In this situation, the residual lignin undergoes the following changes (Lachenal et al. 1995):

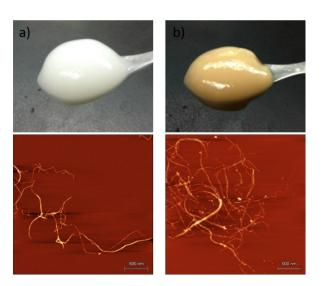
- The content of aliphatic and phenolic hydroxyls decreases slightly;
- The carboxylic acid content increases, except for ozone;
- Methoxy groups are removed;
- Some C=O is formed during oxygen and ClO₂ treatments.

225 It should be stressed that the aforementioned changes only describe the structure of the residual lignin fraction, not the carbohydrates that form nearly 100% of bleached chemical pulps. Upon 226 bleaching, most of the cellulose is retained, but some hemicelluloses are lost together with the 227 residual lignin (Annergren and Rydholm 1959; Annergren et al. 1961; Hanhikoski et al. 2016b). 228 The changes that take place in hemicelluloses during bleaching include degradation and oxidation 229 (formation of carbonyl and carboxylic groups (Jääskeläinen et al. 2000). More detailed information 230 on the chemistry of different bleaching stages is available in the articles and textbooks referenced 231 herein (Lachenal et al. 1995; Dence 1996; Dence and Reeve 1996; Vuorinen et al. 1999; Potthast 232 et al. 2009). Overall, it is important to remember that, depending on the pulping and bleaching 233 234 processes used, the pulps will have different chemical compositions but also different surface charges and charge densities (the latter are predominantly caused by the presence of sulphonic and 235 carboxylic acid groups), both of which influence the ease of pulp fibrillation. These issues will be 236 237 addressed in more detail in section The importance of hemicelluloses.

238 Cellulose nanofibrils – state of the art and bottlenecks

The pioneers in the production of cellulose nanofibrils (CNFs) were Turbak et al. (1982) and 239 240 Herrick et al. (1983) who investigated how to produce microfibrillated cellulose (MFC) by homogenizing cellulose pulp suspensions under pressure during the early 80's. They discovered 241 that after intensive mechanical disintegration, cellulose pulps result in colloidal fibrous 242 suspensions at a very low solid content (typically ~2% wt.). The gel-like structure (Fig 5) and the 243 enhanced hydrogen bonding capability of CNFs confer it great potential for being incorporated in 244 formulations for thickeners and emulsifiers for food, cosmetics, and paints (Turbak et al. 1982; 245 Herrick et al. 1983; Lavoine et al. 2012). Furthermore, CNFs present high aspect ratio, low density 246 and thermal expansion, high strength modulus and stiffness (Hsieh et al. 2008; Eichhorn et al. 247 2010; Siró and Plackett 2010), as well as remarkable film-formability when dried (Spence et al. 248

2011a); properties that make them interesting in terms of producing high-strength composites andbarrier materials from renewable resources.



251

Fig 5. Gel-like structure and Atomic Force Microscopy image showing the morphology of a)bleached CNFs, and b) unbleached CNFs.

Despite the versatile application potential of CNFs, the original efforts of Turbak and Herrick were 254 initially abandoned due to the inefficiency of the process; large amounts of energy were needed 255 for the refining and beating of the cellulose pulps, resulting in rather low yields of fibril 256 suspensions. To address this issue, a number of different types of mechanical treatments, including 257 homogenization (Nakagaito and Yano 2004; Spence et al. 2011b; Moser et al. 2015), grinding 258 (Taniguchi and Okamura 1998; Iwamoto et al. 2007; Spence et al. 2011a), ultrasonication (Zhao 259 et al. 2007; Cheng et al. 2009; Chen et al. 2011), and cryocrushing (Dufresne et al. 1997; 260 Chakraborty et al. 2005), have since been developed to produce CNFs. The two most common 261 262 methods are (1) homogenization, also referred to as microfluidization, where the cellulose suspension is forced to pass through a small orifice which causes the fracture of the fibers into 263 smaller fragments (Turbak et al. 1982), and (2) micro-grinding, where the cellulose is beaten 264 between two ceramic disks, defibrillating the fibers and reducing their size (Iwamoto et al. 2005; 265 266 Stelte and Sanadi 2009). These methods can be used individually or in combination with one another (Iwamoto et al. 2005; Stelte and Sanadi 2009) but both require several passes through the 267 268 system, contributing to a high energy consumption. Consequently, the main objective of CNF

269 manufacturing has been to reduce the energy consumption with different pre-treatments – these
270 can again be classified as chemical and/or enzymatic pre-treatments.

Of the chemical methods, carboxymethylation (Wågberg et al. 1987, 2008) and TEMPO (2,2,6,6-271 tetramethylpiperidine-1-yl)oxyl radical) mediated oxidation (Saito et al. 2006, 2007) are the most 272 widely used. Both methods increase the fiber electrical charges, thus increasing the repulsion 273 between the individual fibrils and facilitating the fiber disintegration. Moreover, during the last 274 275 years, alternative chemical pre-treatments have been developed in order to obtain CNFs with improved properties at reduced cost, in an effort to reduce the energy input while avoiding 276 expensive reagents. As an example of such processes, recyclable organic acids, such as oxalic and 277 maleic acid, have been used for simultaneous production of nanofibrillated and nanocrystalline 278 279 cellulose, while recovering hydrolysed sugars (Chen et al. 2016; Bian et al. 2017a, b, c; Wang et al. 2017). 280

The enzymatic pre-treatments, on the other hand, employ different types of cellulases to weaken the fiber structure, thereby decreasing the amount of mechanical energy required for efficient nanofibrillation (Henriksson et al. 2007; Pääkko et al. 2007). Two main categories have been identified for cellulases: cellobiohydrolases that can cleave highly crystalline cellulose, and endoglucanases that tend to require a certain level of disorder in order to degrade cellulose (Missoum et al. 2013). As a result, these enzyme types have synergistic effects and work best when used together.

288 Despite these efforts, a few factors still limit CNF usage in an industrial scale. Such obstacles include the costs of enzymes and chemical reagents (especially TEMPO), difficulties in CNF 289 290 dewatering and redispersibility, and the incompatibility of CNFs with commercial polymers that limits the mechanical performance of CNF-reinforced composites. Presently, nearly all CNF 291 292 grades are produced from fully bleached chemical pulps that contain only trace amounts of residual 293 lignin. The aim of the present review, however, is to discuss the benefits and limitations of utilizing 294 cheaper, less extensively bleached lignin-containing pulp varieties in the production of 295 lignocellulosic nanofibers and assess whether they could solve some of the aforementioned problems related to fully bleached CNFs. 296

297 Lignin-containing cellulose nanofibrils (LCNFs)

In this review, we use the term broadly and refer to any cellulosic fibrils of <100 nm in width and 298 a lignin content of >1% as lignin-containing cellulose nanofibrils (LCNFs). Over the past decade, 299 300 several efforts have been made to extract LCNFs from different raw materials. Sources like empty palm fruit bunches (Ago et al. 2016), wheat straw (Sánchez et al. 2016; Espinosa et al. 2017), 301 banana, jute, pineapple (Abraham et al. 2011), triticale straw (Tarrés et al. 2017), sunflower stalks 302 (Ewulonu et al. 2019), bamboo chips (Lu et al. 2018) and even bark (Chen et al. 2019) are some 303 of the recently investigated examples. Additionally, many research groups have been working on 304 the isolation, characterization, and application of LCNFs from wood sources. Wang et al. (2012) 305 produced two LCNF samples from kraft wood pulps by using an initial acid hydrolysis following 306 by homogenization at high pressure. Samples containing 5 and 10% of lignin were obtained and 307 characterized. Herrera et al. (2018) successfully isolated LCNFs with 23% lignin content from 308 309 Eucalyptus pulp through catalysed chemical oxidation followed by high-pressure microfluidization. A method using organic acids followed by mechanical treatment was developed 310 for the integrated production of lignin containing nanocrystals (LCNCs) and LCNFs from 311 unbleached mixed hardwood pulp (mainly birch and maple). This methodology allows for the 312 313 recovery of the organic acids and reduces the overall energy consumption of the process (Bian et al. 2017b). LCNFs with as high as 18.5% lignin content was obtained via this procedure. More 314 315 recently, Wen et al. (2019) isolated and characterized LCNFs containing 15.5, 18.6, and 23.15% lignin from a poplar high-yield pulp via TEMPO-mediated oxidation followed by homogenization. 316

Generally, LCNFs are produced from unbleached cellulose pulps with different lignin contents. 317 However, it has been demonstrated that LCNFs can also be directly isolated from raw wood 318 microparticles. In a recent study, nanofibers were produced using alkali-treated poplar wood 319 powder followed by controlled delignification steps in order to obtain samples with 22.1, 14.1, 8.2, 320 2.0, 0.4, and 0.2% residual lignin. Subsequent mechanical grinding was utilized to produce 321 nanofibers (Chen et al. 2018). Yousefi et al. (2018), on the other hand, isolated nanofibers 322 containing 30% lignin by mechanical grinding of *Paulownia Fortunei* wood without utilizing any 323 chemical pre-treatment. 324

Regardless of the raw material, the role of lignin in the nanocellulose suspension needs to be clarified in order to better understand the behaviour of the resulting suspensions in further applications. More details about the properties conferred by the lignin present on LCNFs will bediscussed in the following sections.

329 Lignin – a friend or a foe?

330 Lignin can have two, seemingly contradictory effects on pulp fibrillation, based on what kind of pulps are utilized as the starting material. On the one hand, it can hinder fibrillation, as has been 331 demonstrated in the case of mechanical pulps (Lahtinen et al. 2014). On the other hand, residual 332 lignin can even significantly lower the energy consumption of fibrillation in the case of chemical 333 334 pulps (Spence et al. 2011b), serving as an example on how lignin structure affects its function. Moreover, the presence of residual lignin in chemical pulps has been reported to result in the 335 formation of finer CNFs at comparable energy consumption levels (Solala et al. 2012; Rojo et al. 336 2015). 337

Assuming a complex, network-like structure for native lignin, it seems probable that the initial 338 crosslinked structure prevents efficient fibrillation of the highest-lignin pulp grades by 'locking' 339 340 the individual microfibrils together (Lahtinen et al. 2014; Hanhikoski et al. 2016a). The hydrophobic character of lignin may also play an important role in preventing fiber swelling and 341 fibrillation, as will be discussed in the next section. In contrast, the residual lignin present in 342 chemical pulps is significantly degraded and present in much lower quantities, therefore less able 343 to prevent fiber swelling and fibrillation. We have previously proposed (Ferrer et al. 2012; Solala 344 et al. 2012) that the ease of fibrillation observed in lignin-containing chemical pulps is due to lignin 345 346 acting as an antioxidant, preventing broken covalent bonds from being formed again. This hypothesis is discussed in more detail in section Lignin as an antioxidant. 347

348 Lignin hydrophobicity and thermoplasticity

It has been extensively reported on the literature that lignin is covalently linked to cellulose and hemicelluloses within the wood structure (Sjöström 1993; Tenkanen et al. 1999; Lawoko et al. 2003, 2005; Balakshin et al. 2009, 2011; Iversen and Wännström 2009). It is also generally accepted that lignin presents more hydrophobic character than carbohydrate polymers (Abe et al. 2010; Laurichesse and Avérous 2014). However, different processing methods and chemical compositions of raw material result in different lignin surface free energies and thereby also 355 different hydrophobicities. Understandably, processes that cleave the β -aryl ether linkages or 356 introduce electrical charges in lignin increase its hydrophilicity.

These effects should be kept in mind when examining literature on the hygroscopicity of LCNF films or nanopapers. In literature, nanopaper water interactions are usually characterized by measuring parameters such as total bound water determined with differential scanning calorimetry (DSC), water retention value (WRV), film water absorption, water vapor transmission rate

361 (WVTR), and water contact angle (WCA) (Spence et al. 2010; Ferrer et al. 2012; Wang et al. 2012;

362 Horseman et al. 2017; Nair et al. 2017; Chen et al. 2018; Herrera et al. 2018; Lê et al. 2018).

WCAs are usually higher for LCNF nanopapers than the corresponding CNF nanopapers (Ferrer 363 364 et al. 2012; Rojo et al. 2015; Chen et al. 2018; Herrera et al. 2018; Lê et al. 2018), but other results are less consistent, showing for example higher water absorption and WVTR for LCNF films 365 (Spence et al. 2010; Ferrer et al. 2012). Ferrer et al. reported water absorption values (g/m^2) of 366 24.6±1.2, 23.1±1.2, and 17.1±1.2 for unbleached, oxygen delignified, and fully bleached CNF 367 samples, respectively. Additionally, for those samples, the WCA results were 60±6, 54±6, and 368 51±4. One might expect that the presence of lignin would automatically make the nanopapers more 369 hydrophobic, but lignin removal tends also to alter the carbohydrate chemistry, so that lignin-370 containing CNFs often contain more hemicelluloses than the corresponding fully bleached CNFs, 371 which will contribute to their hygroscopicity (Ferrer et al. 2012; Solala et al. 2012). Moreover, the 372 method of film preparation has a significant effect on properties such as porosity, which will in 373 turn affect the WVTR and specific surface area of nanopapers prepared from these nanofibrils. 374 There is a crucial difference in film casting and hot-pressing, for example – the first tends to form 375 more porous films from lignin-containing CNFs (Spence et al. 2010), whereas pressure filtration 376 377 followed by hot-pressing produces denser films when lignin is present (Rojo et al. 2015). This complex interdependency of film morphology, lignin content, hemicellulose content, and charge 378 379 density (Ferrer et al. 2012; Solala et al. 2012; Rojo et al. 2015; Herrera et al. 2018) makes direct comparisons between different studies challenging, but when these factors are controlled well, 380 381 lignin does indeed increase the hydrophobic nature of a cellulosic pulp and the CNFs and nanopapers derived from it (Rojo et al. 2015; Chen et al. 2018; Lê et al. 2018). 382

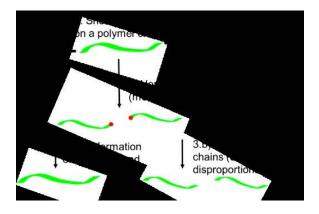
In addition to being hydrophobic, lignin softens at elevated temperatures, especially when 383 plasticized by water or some other component (Sakata and Senju 1975; Salmén 1984; Åkerholm 384 385 and Salmén 2004). It has been reported that at low water contents (<5%), lignin softening temperature is approximately 135 °C, but when the water content increases to 20-40%, the 386 softening temperature decreases to 80-90 °C (Salmén 1982, 1984). Similar behavior has been 387 reported also for hemicelluloses (Olsson and Salmén 2009). This property is well-known from the 388 manufacturing of thermomechanical pulps (Johansson et al. 1998; Gustafsson et al. 2003; Solala 389 et al. 2014), but lignin softening was recently utilized to produce nanofibers from ground wood 390 pulp without applying any chemical pre-treatment (Visanko et al. 2017a). Moreover, this softening 391 behavior has sparked research efforts to develop thermoplastic lignin-based polymers based on 392 these components to replace the presently ubiquitous oil-based polymers (Nägele et al. 2002; Cui 393 394 et al. 2013; Hilburg et al. 2014). As packaging and barrier films are among the most promising commercial applications for CNFs, the role of lignin hydrophobicity and thermoplasticity may 395 help improving CNF nanopaper moisture barrier properties, as we will discuss in section Barrier 396 films. 397

398 Lignin as an antioxidant

Mechanically treated polymers typically have a limiting DP, beyond which they will not degrade in the applied conditions (Glynn and van der Hoof 1973; Tomashevskii et al. 1975; Kondo et al. 2004). This levelling off behavior is often attributed to the chain length becoming too small to absorb mechanical energy. On the other hand, the radical content in mechanically stressed polymer samples typically decreases slightly after levelling off, probably due to the occurrence of recombination and disproportionation reactions that quench some of the formed radicals (Kondo et al. 2004).

Following this line of thought, the observed ease of fibrillation of lignin-containing chemical pulps has been hypothesized to be a result of lignin acting as an antioxidant owing to its polyaromatic resonant-stabilized structure (Dizhbite et al. 2004; Solala et al. 2012; Rojo et al. 2015). In other words, as mechanical treatments – be it by grinding, microfluidization or other means – produce free radicals (Sakaguchi and Sohma 1975; Hon 1979, 1983b, a; Solala et al. 2012, 2015), these radicals can get stabilized by the residual lignin present in the fibers due to the resonance stabilization enabled by the phenyl structures and other conjugated double bonds present in lignin.

This would lead to less favorable kinetics for the recombination reactions that would otherwise cause a partial reformation of the ruptured covalent bonds and thereby prevent effective nanofibrillation, as is illustrated schematically in Fig 6.



416

Fig. 6 Schematic presentation of mechanoradical formation when a polymer chain is subjected to a shear force (1). After the radicals are formed by homolytic chain scission (2), they typically react fast, either forming new covalent bonds (3.a) or by undergoing other reactions, e.g. disproportionation (3.b), that leave the fractured polymer chains separate. In the presence of antioxidants or radical scavengers, such as lignin, the probability of type (a) reactions decreases, enhancing the formation of separate polymer fragments

As mentioned earlier, there is enormous structural variation in different lignins, which inevitably affects their ability to stabilize free radicals (Barclay et al. 1997; Dizhbite et al. 2004; Ugartondo et al. 2008; Vinardell et al. 2008). These differences should be taken into account when evaluating the hypothesis of lignin acting as a radical scavenger. More specifically, the following trends have been found for lignin antioxidant properties (Barclay et al. 1997; Dizhbite et al. 2004; Hage et al. 2012; Ponomarenko et al. 2015; Sadeghifar and Argyropoulos 2015):

- Phenolic hydroxyls increase but aliphatic hydroxyls decrease antioxidant activity
- High molecular weight decreases the antioxidant activity (probably related to the relative number of phenolic hydroxyls)
- Conjugated C=C bonds in the propyl side chain increase the antioxidant activity
- Carbohydrate structures may decrease the antioxidant activity
- Oxygen-containing substituents in the side chains decrease the antioxidant activity

Information reported on the effect of methoxy groups is contradictory; some studies have associated especially ortho-positioned methoxy groups with enhanced antioxidant activity, but others have found that controlled methylation of the phenolic OH groups destroyed lignin's antioxidant ability (Barclay et al. 1997; Dizhbite et al. 2004; Sadeghifar and Argyropoulos 2015).

Given that the structure of lignin affects its radical scavenging ability significantly, it is important 439 to remember what is known about the structure of the lignin present in the pulps that are to be used 440 441 for CNF manufacturing. As an example, among the major structural differences between native and kraft lignin is that the number of phenolic hydroxyls increases significantly both in the 442 dissolved and in the residual lignin (Froass et al. 1996; Chakar and Ragauskas 2004), which would 443 be expected to enhance the radical scavenging ability. Another change that occurs is that some 444 445 methoxy groups are lost during kraft pulping (Froass et al. 1996), but since various research groups agree on the primary importance of phenolic hydroxyls (Barclay et al. 1997; Dizhbite et al. 2004; 446 447 Ponomarenko et al. 2015; Sadeghifar and Argyropoulos 2015), it seems reasonable to assume that the residual lignin in chemical pulps has radical scavenging properties, despite having 448 449 carbohydrate structures linked to it.

Indirect evidence for the antioxidant activity of residual lignin was also reported by Vänskä et al. 450 (2016), showing improved thermal stability of intensely refined softwood kraft pulp in the 451 presence of residual kraft lignin, as indicated by brightness and viscosity-based DP measurements. 452 453 High thermal stability has also been reported for high lignin content (20%) nanofibrils derived by sodium chlorite bleached pine bark (Nair and Yan 2015). On the other hand, no difference was 454 seen in the thermogravimetric analysis profile as a function of lignin content in organosolv-treated 455 456 pulps by Santucci et al. (2016). These inconsistencies highlight the importance of caution when 457 comparing results from different raw materials and processes. It may be speculated that 458 incorporation of a high phenolic OH lignin might be beneficial for the development of new 459 nanocellulose grades with high thermal stability, but more research is needed in this area before definite conclusions can be made. 460

461 The importance of hemicelluloses

Depending on the chosen raw material and utilized processing method, the fibers and nanofibrilsprepared from them may contain not only cellulose and lignin but also hemicelluloses. Thus, a

464 complete understanding about CNF behavior and its interactions should include the properties465 conferred not only by lignin but also by hemicelluloses.

Regarding CNF preparation and properties, hemicelluloses have been found to impede the 466 agglomeration of the nanofibrils. This effect is partly electrostatic in nature, as many 467 hemicelluloses possess negative charges, generating repulsion between them (Arola et al. 2013). 468 On the other hand, the side chains present in some hemicelluloses contribute to steric repulsion 469 470 between nanofibrils, thereby also reducing their tendency to aggregate (Hubbe and Rojas 2008; Tenhunen et al. 2014). As a result of these interactions, the presence of hemicelluloses promotes 471 pulp fibrillation (Duchesne et al. 2001; Hult et al. 2001; Iwamoto et al. 2008; Tarrés et al. 2017) 472 and colloidal stability of CNF suspensions (Tenhunen et al. 2014), as will be discussed in the 473 474 following sections.

475 Fibrillation

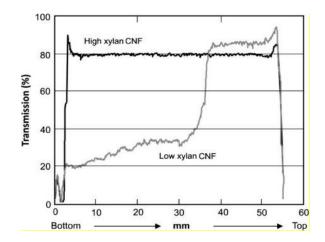
As mentioned before, hemicelluloses enhance pulp fibrillation. This behavior has been attributed 476 477 to their ability to inhibit fibril coalescence, resulting in an open, porous fiber structure (Duchesne et al. 2001; Hult et al. 2001; Pönni et al. 2012). Both cellulose and hemicelluloses readily swell in 478 479 moist environments but, due to their lower molecular weight and often higher degree of branching, hemicelluloses have higher accessibility to water than cellulose (Yang et al. 2013). It should be 480 mentioned, however, that it is not only the hemicellulose content that determines the swelling 481 behavior of a fiber material, but also the spatial distribution of cellulose and hemicelluloses within 482 the fibrils is relevant (Tenhunen et al. 2014; Kulasinski et al. 2015). The aforementioned fiber 483 swelling leads to its softening through the breakage of hydrogen bonds, contributing to efficient 484 fibrillation (Åkerholm and Salmén 2004; Kulasinski et al. 2014, 2015). Naturally, after breaking 485 the initial fiber structure into smaller fibrils and fibril bundles, reattachment of these fragments 486 487 needs to be prevented in order to obtain individual nanosized fibrils. This means that producing high-quality CNFs requires sufficient colloidal stability, which we will address next. 488

489 *Colloidal stability*

490 Colloidal stability, i.e. the ability of a particle suspension to resist agglomeration and/or 491 sedimentation by remaining in an equilibrium, plays a vital role in determining many of the

properties of CNF materials – from specific energy consumption (SEC) during fibrillation to 492 rheological properties and potential redispersibility after drying. As cellulose has strong affinity 493 494 towards itself, in practice it is necessary to prevent its characteristic tendency for fibril agglomeration. One way to counteract such fibril coalescence is to introduce electrical charges in 495 the pulp material, such as in the case of TEMPO-mediated oxidation (Saito et al. 2007; Isogai et 496 al. 2011a, b) or carboxymethylation (Rácz and Borsa 1997; Wågberg et al. 2008; Siró et al. 2011), 497 in which an increase in negative charge leads to electrostatic repulsion between the fibrils at pHs 498 where the carboxyl groups are in their dissociated state. Importantly, electrostatic repulsion is 499 greatly affected by the ionic strength of the system, meaning that it can be hindered simply by 500 adding salt into the suspension (Fall et al. 2011; Junka et al. 2013). This is not the case for steric 501 stabilization, however, in which bulky molecular chains prevent particle aggregation, regardless 502 of the pH or ionic strength of the system. Nevertheless, if the bulky chains possess electrical 503 charges, such as in the case of xylan, the conformation of the polymers depends on changes in pH 504 and ionic strength (Tenhunen et al. 2014). This will in turn affect the extent of steric repulsion. 505

506 As mentioned, hemicelluloses can contribute to colloidal stability of CNFs by both electric and steric mechanisms, depending on their structure (Hubbe and Rojas 2008). Xylans, for instance, 507 have carboxyl groups that give them a net negative charge, as well as side chains that contribute 508 to steric repulsion, and both properties influence the colloidal stability of CNFs (Tenhunen et al. 509 510 2014) (an example of this given in Fig.7). Similarly, galactoglucomannans can sterically stabilize pitch particles in thermomechanical pulping (TMP) process waters (Hannuksela et al. 2003; 511 Tammelin et al. 2007). Recently, a number of researchers have employed the ability of 512 hemicelluloses to adsorb tightly on cellulose (Tammelin et al. 2009; Eronen et al. 2011; Littunen 513 et al. 2015) for producing core-shell structured CNFs, aiming at achieving a better control over 514 their nanostructure (Prakobna et al. 2015; Tanaka et al. 2016). 515



516

Fig. 7 Light transmission profiles of high xylan content CNFs and low xylan content CNFs.
Reprinted from Tenhunen et al. (2014) with permission of Elsevier

519 It is evident that controlling the hemicellulose type and content by careful selection of raw materials, pulping conditions, and other processing methods is vital for controlling the properties 520 521 of the resulting CNFs. As an example of this, LCNFs derived from SO₂-ethanol-water (SEW) 522 pulps (Iakovlev et al. 2010; Iakovlev and Heiningen 2012; Yamamoto et al. 2014) and neutral sulphite pulps (Hanhikoski et al. 2016b, a) with comparable lignin contents appear to have very 523 different water retention properties. Although for methodological differences it is not possible to 524 quantitatively compare the water retentions of these LCNFs, the SEW-LCNFs displayed effortless 525 dewatering (Rojo et al. 2015), whereas the NS-LCNFs had markedly high water retention values 526 (in the range of 400 g/g) even after a minimal mechanical treatment (Hanhikoski et al. 2016a). 527 This is most probably due to the difference in their hemicellulose contents (Chakar and Ragauskas 528 2004; Rojo et al. 2015). One of the challenges in comparing the properties of different CNFs is 529 that there still are no established standardized tests that would allow a clear comparison between 530 different CNF grades. However, reported results indicate that by selecting raw materials with 531 varying (high or low) hemicellulose content for CNFs production, different performance of CNFs 532 will be achieved. We thus recommend a complete lignin and carbohydrate analysis to be run from 533 the fiber material processed to LCNFs to allow meaningful comparisons between LCNF grades. 534

535 Practical considerations for LCNFs

536 Consumption of chemicals and energy

As explained before, the required energy consumption in the production of CNFs is rather high: to 537 obtain cellulose nanofibers applying only mechanical treatments, SECs of ~30,000 kWh/ton are 538 539 necessary (Klemm et al. 2011; Moser et al. 2016). In contrast, by utilizing the TEMPO-mediated oxidation, the SEC can be lowered below 7 MJ/kg, corresponding to 1,900 kWh/ton, while 540 obtaining a uniform transparent gel with fibril width of 3-4 nm and a length of a few micrometers 541 542 (Isogai et al. 2011a). TEMPO-CNFs are of very high quality but due to chemical costs, still too expensive for bulk industry use (Delgado-Aguilar et al. 2015). As an alternative for TEMPO-543 mediated oxidation, Tejado et al. (2012) decreased the energy consumption for the preparation of 544 CNFs by periodate oxidation followed by a chlorite oxidation. Their results showed that cellulose 545 nanofibrils can be liberated from the original cellulose fiber structure at a constant yield with a 546 SEC of 1,250 kWh/ton when the carboxylate content is 2.5 mmol/g. They also reported that after 547 increasing the carboxylate content to 3.5 mmol/g, the energy necessary to obtain the same 548 fibrillation level decreased even further, introducing a less energy-intensive alternative for CNF 549 production. As promising as this approach is, it has, to the best of our knowledge, been thus far 550 551 utilized only for bleached pulps. Other affordable approaches have been sought from enzymatic 552 pre-treatments, but although enzymatically pre-treated CNFs can be prepared from bleached pulp using only 350 kWh/ton (Lindström 2016), the presence of lignin hinders the efficacy of enzymatic 553 554 hydrolysis (Hoeger et al. 2012, 2013), limiting the use only to bleached pulp varieties. With the interest of producing nanofibrils from fibers with higher yields, efforts have been made to 555 556 investigate the fibrillation of lignin-containing fibers.

A number of studies have demonstrated that unbleached pulps tend to be easier to fibrillate than 557 fully bleached pulps. Spence et al. (2011b) reported SECs in the range of 420 kWh/ton for LCNFs 558 559 with an estimated specific surface area (SSA) of $\sim 80 \text{ m}^2/\text{g}$. Generally, the achieved SSAs were higher for unbleached fibrils than corresponding fully bleached fibrils at comparable specific 560 561 energy consumptions. Similarly, Lahtinen et al. (2014) reported that unbleached softwood kraft pulp achieved a higher level of fibrillation after utilizing the same specific energy consumption 562 than fully bleached kraft pulps. Also Solala et al. (2012) reported a more thorough fibrillation of 563 unbleached of hardwood kraft pulp in comparison with its fully bleached counterpart. Since 564 unbleached pulps render higher yields of the initial lignocellulosic material and require lower 565 dosages of chemicals to be produced in the first place, LCNFs derived from them might be a viable 566

option for bulk uses where their inherent brown colour does not present problems (Solala et al. 2012; Delgado-Aguilar et al. 2016). In terms of particle size, Rojo et al. (2015) reported softwood LCNF fibers containing 0, 2, 4, and 14% lignin with diameters of 44 ± 3 , 25 ± 1 , 20 ± 2 , and $16 \pm$ 2 nm respectively. Also these results positively correlate with the ease of fibrillation due to increases in lignin and hemicellulose contents.

Generally, it should be noted that the comparison of individual studies is complicated because of a lack of consistent, standardized vocabulary in the field; for instance, the exact meaning of the term 'nanocellulose' varies in the literature. Moreover, the reported values are often not comparable to one another, as different analytical techniques are favored by different research groups. Recently, this issue was addressed by Foster et al. (2018) on their review that discusses recommended analytical protocols in this research area.

578 Dewatering and redispersibility

Typically, CNF suspensions contain 98-99% water and only 1-2% of the cellulosic nanomaterial. 579 580 Even when higher consistencies can be achieved, they tend to be in the range of 5%, meaning that transportation costs for CNF gels are very high in relation to the transported dry mass. Water 581 removal from CNFs is energy intensive, which is not ideal on itself, but more crucially, it is 582 accompanied by irreversible fibril coalescence, often referred to as hornification by pulp and paper 583 scientists. The extent of coalescence upon drying can be limited by solvent exchanges (Henriksson 584 et al. 2008) or freeze-drying (Lovikka et al. 2016), but these methods are tedious and time-585 586 consuming, rendering them impractical in an industrial scale. Alternatively, the surface chemistry of the CNFs may be modified to allow redispersibility after drying by introducing electrostatic 587 repulsion, like in the case of carboxymethylated CNFs (Eyholzer et al. 2010), or by using steric 588 stabilization from pectin or other components (Hietala et al. 2017). Recently, Visanko et al. (2017) 589 590 reported having made redispersable LCNF nanopapers from spruce ground wood pulp (lignin content 27.4%) when dried from ethanol, which can probably be attributed to the lower density 591 and interfibrillar contact in these nanopapers. Importantly, similar behavior was not seen in fully 592 bleached CNFs (Visanko et al. 2017a), suggesting that at least some lignin-containing nanofibril 593 grades have unique properties that could promote their use in an industrial scale. 594

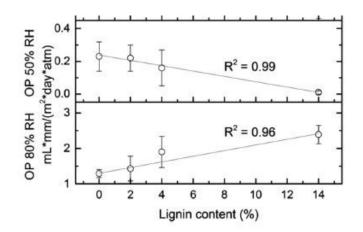
595 **Barrier films**

One of the frequently proposed uses for CNFs has been to use it in barrier applications, for example 596 597 in food packaging (Lavoine et al. 2012). With a combination of moisture and pressure, CNFs can 598 be turned into dense films, or so-called nanopapers, with porosities in the range of 10-20% (Sehaqui et al. 2010; Österberg et al. 2013). In dry state, the average pore size in such films is 599 typically around 1-10 nm (Henriksson et al. 2008; Lavoine et al. 2012) but the film structure 600 changes in moist conditions, causing a decline in its barrier properties for oxygen and water vapor 601 (Aulin et al. 2010). One working hypothesis has been that lignin-containing films should have 602 better barrier properties, as they are inherently less hygroscopic. However, as discussed earlier, 603 this depends heavily on the method of film formation. If films are produced at temperatures below 604 the softening temperature of lignin, efficient H-bonding between nanofibrils may be compromised, 605 leading to increased pore formation and poor barrier properties (Spence et al. 2010; Santucci et al. 606 2016). On the other hand, if hot-pressed in moist conditions, the lignin in the LCNF films will 607 soften and fill voids as schematically illustrated in Fig. 8. Rojo et al. (2015) demonstrated an 608 improvement on the barrier functionality of the films in terms of oxygen permeability as the lignin 609 content of the samples increases when measuring at 50% relative humidity. At 80% relative 610 611 humidity, the results showed a slight increase which was not significant when compared with the improvement at 50% relative humidity (Fig 9). Moreover, lignin has been shown to decrease water 612 613 uptake and increase water contact angle (Wang et al. 2012; Rojo et al. 2015; Herrera et al. 2018), as long as control over the hemicellulose content is maintained (Ferrer et al. 2012). 614



615

- **Fig. 8** Proposed model to describe the location of the lignin within LCNF suspension after filtering
- 617 (left) and LCNF nanopaper after pressing (right) Reprinted and adapted from Rojo et al. (2015)
- 618 with permission of Royal Society of Chemistry.



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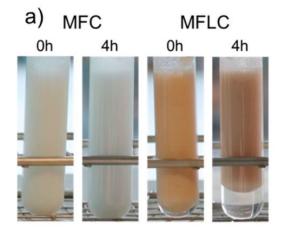
Fig. 9 Oxygen permeability of hot-pressed nanopapers with different lignin contents at 50 and
80% relative humidity. Reprinted and adapted from Rojo et al. (2015) with permission of Royal
Society of Chemistry.

In summary, considering just the lignin content of a potential CNF material is not sufficient; it is vital also to control the content of heteropolysaccharides and electrical charges, as both significantly affect the interactions between the material and water. In addition to the chemical composition of LCNF nanopapers, also their morphology in terms of porosity and pore size is of central importance when analysing nanopapers for their barrier properties.

628 Emulsions

629 About a decade ago, Andresen and Stenius (2007) reported that hydrophobized microfibrillated cellulose can be used as a stabilizer in Pickering emulsions. Since then, a number of papers have 630 been published on the subject (Xhanari et al. 2011; Cunha et al. 2014; Gestranius et al. 2017). In 631 order to efficiently stabilize emulsions, particles should show a certain extent of amphiphilicity 632 (Kalashnikova et al. 2013). To achieve this, previous studies have commonly utilized synthetic 633 hydrophobization routes (Andresen and Stenius 2007; Xhanari et al. 2011; Cunha et al. 2014), but 634 635 similar effects can also be reached by utilizing LCNFs (Fig 10) (Ballner et al. 2016; Yan et al. 2016) or lignin particles (Nypelö et al. 2015; Li et al. 2016). Although the exact chemical nature 636 637 of residual lignin can be difficult to control, there are obvious benefits to avoiding the multistep process of first removing all lignin through sequential bleaching and then derivatizing the nearly 638 639 pure cellulose with synthetic hydrophobic moieties. We therefore foresee benefits in exploration of lignin-containing nanocelluloses for emulsion stabilization, especially in applications where in-640 641 situ polymerization is feasible, as this would solve many of the problems presently associated with

- 642 CNFs use in composite reinforcement. In fact, Yan et al. (2016), Ballner et al. (2016) and Gindl-
- 643 Altmutter et al. (2015) have successfully applied this principle to produce composites of LCNFs
- and synthetic polymers via emulsion polymerization, as discussed in the following section.



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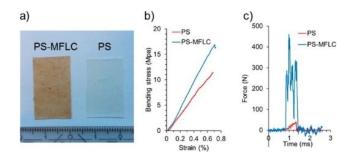
Fig. 10 Styrene-in-water emulsions stabilized using nanofibrillated cellulose (MFC) and lignincontaining nanofibrillated cellulose (MFLC), at 0 and 4 hours after after mixing. Readapted and reprinted from Ballner et al. (2016) with permission of the American Chemical Society.

649 *Nanocomposites*

The utilization of cellulose nanofibrils from bleached fibers has been widely studied as a reinforcing agent in composite materials, prepared by techniques such as solvent casting (Wågberg et al. 1987; Leitner et al. 2007) or emulsion polymerization (Nikfarjam et al. 2015). Due to the high aspect ratio and moderately low thermal stability of CNFs, melt extrusion is usually not a practical method to produce such composites.

It is well known that the reinforcing capacity of cellulose nanoparticles combined with different matrices occurs due to strong interactions between the fibrils, allowing to improve the mechanical performance of the composites made thereof (Siqueira et al. 2010; Kargarzadeh et al. 2018). This strong network of nanofibers can be considered the result of a percolation mechanism which assumes that above certain concentration of the fibers, the mechanical properties of the composites will be improved (Boufi et al. 2014) due to increased hydrogen bonding between them (Dubief et al. 1999).

In addition to CNFs, the incorporation of LCNFs to different composite materials has become more common, employing a number of matrix materials, such as polylactic acid (Sun et al. 2014; Wang et al. 2014), starch (Ago et al. 2016), polypropylene (Ferrer et al. 2016), polycaprolactone (Herzele et al. 2016), polystyrene (Ballner et al. 2016), and polyurethane (Visanko et al. 2017b). Ballner et al. (2016) utilized in-situ polymerization of styrene in water stabilized by LCNFs, followed by hot-pressing, and obtained composites with increased bending stress and Charpy impact bending strength in comparison to pure PS (Fig. 11). These results are an indication of the potential of LCNFs to be used in composite reinforcement.



670

Fig. 11 The visual appearance of hot-pressed PS and PS-MFLC films (a) and results of mechanical
characterization by static three-point bending (b) and Charpy impact bending (c). Reprinted from
Ballner et al. (2016) with permission of Royal Society of Chemistry

Similarly, Nair et al. (2017) reported that the tensile strength of LCNF/epoxy composites, with 674 675 23% of lignin content, increased from 65 \pm 6 MPa to 134 \pm 10 MPa. This increment was attributed to the increased compatibility between the fibrils and the epoxy matrix caused by the presence of 676 residual lignin. Following this trend, Chen et al. (2019) incorporated LCNFs as a reinforcement 677 material in polymeric diphenylmethane diisocyanate (pMDI) resins, commonly used for wood 678 679 composite manufacturing. Their results showed positive effects due to the increase in bond line between wood and adhesive at different LCNF loadings. On the other hand, Diop et al. (2017) 680 681 utilized LCNFs as an adhesive in fiberboards, reporting that by adding 20% LCNFs, the modulus of elasticity (MOE) was 300 MPa higher than the standard requirement of 1241 MPa. Regarding 682 683 the required modulus of rupture (MOR), by incorporating 20% LCNFs, they reached a MOR value of 12.1 MPa which was close to the target value for commercial panels of 12.4 MPa. Moreover, 684 even if the vast majority of studies have utilized oil-based, non-biodegradable polymers, there are 685 indications that LCNFs can be successfully combined with biobased, compostable polymer 686 687 matrices. As an example of this, the addition of LCNFs to polylactic acid was studied by Wang et 688 al. (2014) and Sun et al. (2014) who analysed the mechanical properties of films made thereof,

agreeing that the addition of LCNFs improved the mechanicals properties, such as tensile strengthand elongation at break, when compared with pure PLA films.

Despite these promising findings, the reports on LCNF-reinforced composites are scattered, occasionally showing moderate if any improvement in mechanical performance, often reporting increases only in Young's modulus (Iwamoto et al. 2014; Ferrer et al. 2016; Horseman et al. 2017; Kim et al. 2017). This highlights the complexity of the interplay between properties such as fibril size and shape, their distribution in the matrix, adhesion between different components, matrix crystallinity, etc. that influence the performance of composite materials.

Regarding the present state of the art, the use of LCNF composite materials holds great potential, as the LCNFs are inherently more compatible with hydrophobic matrices than the analogous bleached CNF, allowing facile mixing of the fibrils and the matrix. Combined with the potential of cost and environmental impact savings through lower SEC and need of chemical processing in comparison to bleached CNFs, we believe it is an area worth exploring even further to develop bio-based alternatives for oil-derived synthetic polymer materials.

703 Conclusions

Pulp and paper industry has traditionally viewed lignin as something that needs to be removed in order to produce high-quality paper. Although there are many similarities between pulping and papermaking and the production of cellulose nanofibrils, the end uses and therefore also the desired functionalities are partly different. For this reason, we promote a change of paradigm in which lignin is seen as a tool that can be used to tailor the properties of CNFs rather than a problematic component that is always detrimental for high-quality products.

For this to happen, it is of uttermost importance to consider the amount and structure of the lignin in question as this will affect the ease of fibrillation, the development of interfibrillar H-bonding as well as the water interactions of the material. Presently, the body of literature on LCNFs remains rather scattered, including a wide range of raw materials and mechanical, chemical or enzymatic processing methods. For this reason, comparisons between individual studies are not always straightforward. In general it can be stated that taking into account the high yield, low chemical consumption and manageable mechanical energy consumption of LCNFs, they can be viewed as an economically and ecologically viable family of materials with a number of potentialapplications.

In order to produce extensively fibrillated LCNFs, it is necessary to first break the network structure of native lignin. In practise, this means including a moderate pre-delignification step prior to fibrillation. Additionally, the presence of hemicelluloses is often linked to the presence of lignin; and both can facilitate fibrillation.

While lignin is believed to possess antioxidant properties, stabilizing mechanically or thermally
 formed free radicals, conclusive evidence of how this mechanism affects the fibrillation process is
 still needed.

Interactions between water and LCNFs are complex and depend on a number of factors, including lignin and hemicellulose contents, electrical charge density, and sample morphology. Lignin may be used as a natural hydrophobizing agent in CNFs, which could be beneficial for example in composite reinforcement and Pickering emulsions.

730 Main nomenclature

731 LCNFs: lignin-containing cellulose nanofibrils

732 BCNFs: bleached cellulose nanofibrils

733 CNFs: cellulose nanofibrils

734 CNCs: cellulose nanocrystals

735 MFC: microfibrillated cellulose

736 SW: softwood

737 HW: hardwood

738 LCCs: lignin-carbohydrate complexes

739 DP: degree of polymerization

740 CrI: crystallinity index

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- 741 DSC: differential scanning calorimetry
- 742 WRV: water retention value
- 743 WVTR: water vapour transmission rate
- 744 WCA: water contact angle
- 745 MOE: modulus of elasticity
- 746 MOR: modulus of rupture

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