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Optical properties of Mg-doped VO₂: Absorption measurements and hybrid functional calculations

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Mg-doped VO₂ thin films with thermochromic properties were made by reactive DC magnetron co-sputtering onto heated substrates, and spectral absorption was recorded at room temperature in the $0.5 < \hbar \omega < 3.5$ eV energy range. Clear evidence was found for a widening of the main band gap from 1.67 to 2.32 eV as the Mg/(V + Mg) atomic ratio went from zero to 0.19, thereby significantly lowering the luminous absorption. This technologically important effect could be reconciled with spin-polarized density functional theory calculations using the Heyd-Scuseria-Ernzerhof [Heyd *et al.*, J. Chem. Phys. **118**, 8207 (2003); *ibid.* **124**, 219906 (2006)] hybrid functional. Specifically, the calculated luminous absorptance decreased when the Mg/(V + Mg) ratio was increased from 0.125 to 0.250. © 2012 American Institute of Physics. [http://dx.doi.org/10.1063/1.4766167]

This letter reports experimental and theoretical data on Mg-doped VO₂ thin films. The main constituent (VO_2) is a well known thermochromic material capable of reversible transition between a solar-transparent, semiconducting, and monoclinic state below a "critical" temperature τ_c of ~68 °C and a near-infrared reflecting, metallic-like and tetragonal state above τ_{c} .^{1,2} The associated modulation of solar energy transmittance makes VO₂-based thin films a very promising candidate for window coatings for energy efficient buildings.3-6 The dopant (Mg) serves to widen the main band gap of VO₂ and boosts luminous transmittance,⁷ which otherwise tends to be undesirably low and constitutes a major obstacle for practical implementation of thermochromic fenestration.⁶ Mg doping also decreases τ_c and brings it closer to room temperature. We present spectral measurements of optical absorption in the $0.5 < \hbar\omega < 3.5 \text{ eV}$ range of photon energy and find that these data can be reconciled with hybrid functional calculations.

Mg-doped VO_2 thin films were made by reactive DC magnetron sputtering from 5-cm-diameter targets of V (99.5%) and Mg (99.9%) onto glass and carbon plates concurrently. The substrates were put on a rotating holder located 13 cm below the targets and were kept at \sim 450 °C depositions. The deposition during chamber was evacuated to 6.3×10^{-7} mbar before the substrates were heated. 80 ml/min of Ar and 5 ml/min of O₂ (both 99.997%) were then introduced via mass-flow-controlled inlets. A total pressure of ~9.2 mTorr was maintained during film fabrication, and sputter powers were 172 W and 0-40 W for the V and Mg targets, respectively. Films were grown at ~ 0.056 nm/s to thicknesses d of 56 to 67 nm. We recorded d on a Bruker DektakXT profilometer, and supporting data were found by modeling of optical measurements. Elemental compositions were obtained from Rutherford backscattering spectrometry (RBS) applied to films on carbon. Mg contents were inferred from simulations based on the SIMNRA program,⁸ which executes iterative least-square fits to experimental spectra. Mg/(V + Mg) $\equiv z$ atomic ratios lay in the $0 \le z < 0.19$ interval.

Spectral normal transmittance *T* and near-normal reflectance *R* were recorded in the $0.5 < \hbar\omega < 3.5$ eV range for the semiconducting phase of the films, deposited onto glass, by use of a single-beam spectrophotometer for absolute measurements.⁹ The glass is transparent, and the absorption coefficient α was obtained from Beer-Lambert's Law, i.e., $\alpha d = \ln[T/(1 - R)]$. Optical band gaps (denoted E_g) were determined—using standard procedures¹⁰—from $(\alpha\hbar\omega)^m = A(\hbar\omega - E_g)$, where *A* is a constant. The exponent *m* depends on the nature of the optical transition and is 1/2, 1/3, 2, and 2/3 for indirect-allowed, indirect-forbidden, direct-allowed, and direct-forbidden optical transitions, respectively. Linear extrapolation of $(\alpha\hbar\omega)^m$ vs $\hbar\omega$ near the band gap gave E_g as the intercept with the $\alpha = 0$ axis.

Figure 1(a) shows $(\alpha \hbar \omega)^{1/2}$ vs $\hbar \omega$ for films with different Mg contents and presuming indirect-allowed transitions. Extrapolations yield two band gaps for pure VO₂, E_{g1} and E_{g2} , at 1.67 eV and ~0.5 eV, respectively, which are in line with well established data.¹¹ It is essential to note that E_{g1} shifts towards higher energy for increasing doping levels, specifically from 1.67 eV to 2.32 eV as z goes from zero to 0.190. For z = 0.116, which corresponds to about one of eight V atoms substituted by Mg, one finds $E_{g1} \approx 1.95$ eV. The strong displacement of E_{g1} leads to significantly decreased luminous absorptance in thin films, as inferred from the bell-shaped shaded region in Fig. 1 illustrating the relative luminous efficiency of the eye.¹²

The observation of two band gaps can be rationalized from Goodenough's seminal work on the electronic structure

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FIG. 1. (a) Experimental and (b) calculated $(\alpha \hbar \omega)^{1/2}$ vs photon energy $\hbar \omega$, where α is absorption coefficient, for Mg-doped VO₂ with the shown Mg/(V + Mg) atomic ratios. The shaded bell-shaped region signifies relative luminous efficiency of the eye (Ref. 12).

of VO₂,¹³ which is illustrated in Fig. 2. The crystal-fieldinduced division of the V3*d* t_{2g} band into $d_{||}$ and π^* sub-bands, and further dimerization of V atoms in the otherwise tetragonal lattice, makes the $d_{||}$ band split into two parts. The lower $d_{||}$ band is completely filled and the π^* band is empty. E_{g1} is associated with the band gap between O2*p* and π^* , and E_{g2} is associated with the band gap between the lower $d_{||}$ and π^* . This compelling assignment of the band gaps cannot explain the effect of Mg, however, and hence we turn now to calculations.

Electronic and optical properties of the Mg-doped monoclinic M1 phase of VO_2 were studied using the density functional theory (DFT) plane wave basis-set code package



VASP.¹⁴ Core electrons were treated with the projector augmented wave (PAW) method.¹⁵ We used the Perdew-Burke-Ernzerhof¹⁶ (PBE) version of the generalized gradient approximation functional and the Heyd-Scuseria-Ernzerhof¹⁷ (HSE) hybrid functional and performed spin-polarized calculations. The energy cut-off for basis-set expansion was put at 500 eV, and the Gaussian broadening width for electronic smearing was 0.05 eV. Geometries were relaxed until residual atomic forces were <0.1 eV/nm, at which point the total energies were calculated to a convergence threshold of 10^{-5} eV per cell for both functionals.

The crystallographic unit cell for the M1 phase comprises four V and eight O atoms¹⁸ and was used in calculations for pure VO₂. To simulate 25% doping, one V atom was replaced by Mg in this cell, while 12.5% doping was simulated by creation of a $2 \times 1 \times 1$ supercell wherein one V atom was replaced by Mg. Gamma-centered *k*-point grids¹⁹ of $2 \times 4 \times 4$ (for the supercell) and $4 \times 6 \times 6$ (for the unit cell) were used. The crystal structure—i.e., cell parameters and atomic positions—was relaxed in each case. For pure VO₂, this procedure yielded lattice parameters agreeing with experimental data¹⁸ to within 0%–3% for different cell axes, always exceeding experimental values. Mg concentrations below 12.5% would require a supercell at least twice as large, which was deemed too demanding for HSE functional calculations.

The dielectric function $\varepsilon(\omega) \equiv \varepsilon_1(\omega) + i\varepsilon_2(\omega)$ fully describes the optical properties of a medium at all photon energies. The dielectric function for interband transitions was calculated in the momentum representation, which requires matrix elements between occupied and unoccupied eigenstates. The imaginary part of the dielectric function can be derived from^{20,21}

$$\varepsilon_{2}(\omega) = \frac{4\pi^{2}e^{2}}{\Omega} \lim_{q \to 0} \frac{1}{q^{2}} \sum_{c,v,\mathbf{k}} 2w_{\mathbf{k}}\delta(\epsilon_{c\mathbf{k}} - \epsilon_{v\mathbf{k}} - \omega) \\ \times \langle u_{c\mathbf{k}+e_{1}q} | u_{v\mathbf{k}} \rangle \langle u_{c\mathbf{k}+e_{2}q} | u_{v\mathbf{k}} \rangle^{*}, \qquad (1)$$

where *e* is elementary charge, Ω is volume of the Brillouin zone, indices *c* and *v* refer to conduction and valence band states, respectively, $w_{\mathbf{k}}$ is weight of the *k*-point vector, $\epsilon_{c\mathbf{k}}$ and $\epsilon_{v\mathbf{k}}$ are energy levels, $u_{v\mathbf{k}}$, $u_{c\mathbf{k}+e_1q}$, and $u_{c\mathbf{k}+e_2q}$ are cell periodic parts of the wavefunction at a given *k*-point, and e_1 and e_2 are components of the unit vector. A Kramers– Kroning transformation was used to get the real part ε_1 from ε_2 by

$$\varepsilon_1(\omega) = 1 + \frac{2}{\pi} P \int_0^\infty \frac{\varepsilon_2(\omega')\omega'}{{\omega'}^2 - \omega^2 + i\eta} d\omega', \qquad (2)$$

where *P* signifies the principal value and the integral is evaluated in the limit where η approaches zero. The absorption coefficient was then derived from

$$\alpha(\omega) = 2(\omega/c) \{ [[\varepsilon_1^2(\omega) + \varepsilon_2^2(\omega)]^{1/2} - \varepsilon_1(\omega)]/2 \}^{1/2}.$$
 (3)

We first consider pure VO_2 whose electronic band structure was recently studied with the HSE functional;^{22,23} the

not aware of any experimental attempts to determine the structural modifications invoked in VO_2 by Mg doping.

z = 0.25. The overall agreement with Goodenough's band scheme,¹³ illustrated in Fig. 2, is apparent. The band gaps E_{g1} and E_{g2} , calculated from the eigenvalues, are noted in the figure. Specifically, our E_{g1} and E_{g2} values are 2.07 and 0.72 eV for z = 0.125 and 2.12 and 1.01 eV for z = 0.25, respectively. Band structure calculations give clear evidence

of indirect band gaps for z = 0.25 and pure VO₂.

We also studied the electronic properties of Mg-doped VO₂, and Fig. 4 shows density of states for z = 0.125 and

Summarizing, we investigated the optical properties of

Mg-doped VO₂ films experimentally and computationally.



FIG. 3. Optimized structure for Mg-doped VO₂ with Mg/(V + Mg) = 0.125, as obtained from calculations using the HSE functional. The figure shows the contents of the $2 \times 1 \times 1$ supercell plus some additional atoms from a neighboring supercell. V–V separations are given in nanometers.



FIG. 4. Total and partial density of states (DOS) for V and O calculated by use of the HSE functional for Mg-doped VO₂ with the Mg/(V + Mg) atomic ratio (a) 0.125 and (b) 0.25. Characters of the bands are noted, as are the band gaps. E_F denotes Fermi energy and f.u. means formula unit.

electronic and optical properties have also been calculated by several other methods.²⁴ Our results are almost identical to those of Eyert.²² Specifically, we find an indirect band gap of 1.38 eV corresponding to the difference between the maximum and the minimum of the bands in *k*-space. Figure 1(b) shows calculated spectral absorption coefficients for VO₂ as well as the doped systems. A comparison with the experimental data in Fig. 1(a) for z = 0.116 and z = 0.190—which are the compositions closest to those of the calculations shows clear *qualitative* correspondence with theory.

The match between experimental and calculated spectral absorption is better for the Mg-doped samples than for pure VO₂, and one may ask why this is so. We speculate that a possible answer could be that electron correlation effects in VO₂ are reduced as a consequence of the Mg doping, thus allowing a more accurate description of electronic structure with DFT-based methods. Substitution of some V by Mg clearly influences the structure and leads to rearrangement of ionic positions, notably those of V. Thus, our calculations predict that the alternating V-V distances (0.315 and 0.252 nm, etc.) for pure VO₂ along the V chain in the *a* direction (rutile c axis) change for z = 0.125 and, moreover, become pair-wise similar (0.316 and 0.302 nm; 0.264 and 0.261 nm, etc.), as indicated in Fig. 3. It is interesting to note that even this modest Mg content affects the structure throughout the supercell. For z = 0.25, the V–V distances along the *a* direction become 0.302 and 0.290 nm. We are The qualitative agreement between the two sets of data gives strong evidence that the band gap widening—and concomitantly the technically important lowering of the luminous absorptance—can be reconciled with hybrid functional calculations. This finding may also open avenues towards computation to discover alternative band gap widening dopants for thermochromic and nano-thermochromic²⁵ VO₂-based window coatings. Furthermore, our work may be relevant to other potential applications of VO₂-based materials, such as nanoelectronic switches²⁶ and microbolometers.²⁷

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- ¹F. J. Morin, Phys. Rev. Lett. **3**, 34 (1959).
- ²S. M. Babulanam, T. S. Eriksson, G. A. Niklasson, and C. G. Granqvist, Sol. Energy Mater. **16**, 347 (1987).
- ³G. B. Smith and C. G. Granqvist, *Green Nanotechnology: Solutions for Sustainability and Energy in the Built Environment* (CRC, Boca Raton, FL, 2010).
- ⁴M. Saeli, C. Piccirillo, I. P. Parkin, R. Binions, and I. Ridley, Energy Build. 42, 1666 (2010).
- ⁵Y. Gao, H. Luo, Z. Zhang, L. Kang, Z. Chen, J. Du, M. Kanehira, and C. Cao, Nano Energy 1, 221 (2012).

- ⁶S.-Y. Li, G. A. Niklasson, and C. G. Granqvist, Thin Solid Films **520**, 3823 (2012).
- ⁷N. R. Mlyuka, G. A. Niklasson, and C. G. Granqvist, Appl. Phys. Lett. **95**, 171909 (2009).
- ⁸M. Mayer, AIP Conf. Proc. **475**, 541 (1999).
- ⁹P. Nostell, A. Roos, and D. Rönnow, Rev. Sci. Instrum. 70, 2481 (1999).
- ¹⁰F. Wooten, *Optical Properties of Solids* (Academic, New York, 1972).
- ¹¹H. W. Verleur, A. S. Barker, Jr., and C. N. Berglund, Phys. Rev. 172, 788 (1968); H. S. Choi, J. S. Ahn, J. H. Jung, T. W. Noh, and D. H. Kim, Phys. Rev. B 54, 4621 (1996); K. Okazaki, S. Sugai, Y. Muraoka, and Z. Hiroi, ibid. 73, 165116 (2006); M. M. Qazilbash, K. S. Burch, D. Whisler, D. Shrekenhamer, B. G. Chae, H. T. Kim, and D. N. Basov, ibid. 74, 205118 (2006); J. B. Kana Kana, J. M. Ndjaka, C. Vignaud, A. Gibaud, and M. Maaza, Opt. Commun. 284, 807 (2011); W. W. Li, J. J. Zhu, X. F. Xu, K. Jiang, Z. G. Hu, M. Zhu, and J. H. Chu, J. Appl. Phys. 110, 013504 (2011). Specifically, E_{g1} is in excellent agreement with recent observations for VO2 films made by sputtering of vanadium followed by post oxidation [Z. Luo, Z. Wu, X. Xu, T. Wang, and Y. Jiang, J. Vac. Sci. Technol. A 28, 595 (2010); Chin. Phys. B 19, 106103 (2010)]; it is also consistent with band gaps for some sol-gel-prepared VO2 films [M. M. Qazilbash, A. A. Schafgans, K. S. Burch, S. J. Yun, B. G. Chae, B. J. Kim, H. T. Kim, and D. N. Basov, Phys. Rev. B 77, 115121 (2008)], bcc-type VO₂ nanorod arrays [Y. Q. Wang, Z. J. Zhang, Y. Zhu, Z. C. Li, R. Vajtai, L. J. Ci, and P. M. Ajayan, ACS Nano 2, 1492 (2008)], and over-stoichiometric vanadium surfaces [A. Z. Moshfegh and A. Ignatiev, Thin Solid Films 198, 251 (1991)]. Band gaps evaluated from $(\alpha\hbar\omega)^2$ vs $\hbar\omega$ —as would be adequate for direct-allowed transitions—may appear as much as $\sim 1 \text{ eV}$ wider than if the assessment is for indirect-allowed transitions.
- ¹²G. Wyszecki and W. S. Stiles, Color Science: Concepts and Methods, Quantitative Data and Formulae, 2nd ed. (Wiley, New York, 2000).

- ¹³J. B. Goodenough, J. Solid State Chem. 3, 490 (1971).
- ¹⁴G. Kresse and J. Hafner, Phys. Rev. B 48, 13115 (1993); G. Kresse and J. Furthmüller, *ibid.* 54, 11169 (1996); G. Kresse and D. Joubert, *ibid.* 59, 1758 (1999).
- ¹⁵P. E. Blöchl, Phys. Rev. B 50, 17953 (1994).
- ¹⁶J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. 77, 3865 (1996); erratum 78, 1396 (1997).
- ¹⁷J. Heyd, G. E. Scuseria, and M. Ernzerhof, J. Chem. Phys. **118**, 8207 (2003); erratum **124**, 219906 (2006).
- ¹⁸G. Andersson, Acta Chem. Scand. **10**, 623 (1956).
- ¹⁹H. J. Monkhorst and J. D. Pack, Phys. Rev. B 13, 5188 (1976).
- ²⁰R. Ahuja, S. Auluck, J. M. Wills, M. Alouani, B. Johansson, and O. Eriksson, Phys. Rev. B 55, 4999 (1997).
- ²¹M. Gajdoš, K. Hummer, G. Kresse, J. Furthmüller, and F. Bechstedt, Phys. Rev. B 73, 045112 (2006).
- ²²V. Eyert, Phys. Rev. Lett. **107**, 016401 (2011).
- ²³F. Iori, M. Gatti, and A. Rubio, Phys. Rev. B 85, 115129 (2012).
- ²⁴A. Liebsch, H. Ishida, and G. Biehlmayer, Phys. Rev. B **71**, 085109 (2005); M. Gatti, F. Bruneval, V. Olevano, and L. Reining, Phys. Rev. Lett. **99**, 266402 (2007); J. M. Tomczak and S. Biermann, Phys. Rev. B **80**, 085117 (2009); Phys. Status Solidi B **246**, 1996 (2009); Europhys. Lett. **86**, 37004 (2009); R. Su, J. He, and Y.-J. Guo, Acta Phys. Sin. **60**, 107101 (2011); A. S. Belozerov, M. A. Korotin, V. I. Anisimov, and A. I. Poteryaev, Phys. Rev. B **85**, 045109 (2012).
- ²⁵S.-Y. Li, G. A. Niklasson, and C. G. Granqvist, J. Appl. Phys. **108**, 063525 (2010); **109**, 113515 (2011); Appl. Phys. Lett. **99**, 131907 (2011).
- ²⁶M. Nakano, K. Shibuya, D. Okuyama, T. Hatano, S. Ono, M. Kawasaki, Y. Iwasa, and Y. Tokura, Nature **487**, 459 (2012).
- ²⁷L. A. L. de Almeida, G. S. Deep, A. M. N. Lima, I. A. Khrebtov, V. G. Malyarov, and H. Neff, Appl. Phys. Lett. 85, 3605 (2004).