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Origin of cross-chain geochemical variation in Quaternary lavas from the northern Izu arc: Using a quantitative mass balance approach to identify mantle sources and mantle wedge processes

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[1] We present major, trace element, and Pb-Sr-Nd-Hf isotope data for Quaternary basalt and basaltic andesite lavas from cross-chain volcanoes in the northern Izu (N-Izu) arc. Lavas from Izu-Oshima, Toshima, Udonejima, and Niijima islands show consistent chemical changes with depth to the Wadati-Benioff zone, from 120 km beneath Izu-Oshima to 180 km beneath Niijima. Lavas from Izu-Oshima at the volcanic front (VF) have elevated concentrations of large ion lithophile elements (LILEs), whereas rear-arc (RA) lavas are rich in light rare earth elements (LREEs) and high field strength elements (HFSEs). VF lavas also have more radiogenic Pb, Nd, Sr, and Hf isotopic compositions. We have used the Arc Basalt Simulator version 3 (ABS3) to examine the mass balance of slab dehydration and melting and slab fluid/melt-fluxed mantle melting and to quantitatively evaluate magma genesis beneath N-Izu. The results suggest that the slab-derived fluids/melts are derived from ~20% sediment and ~80% altered oceanic crust, the slab fluid is generated by slab dehydration for the VF magmas at 3.3–3.5 GPa/660°C–700°C, and slab melt for RA magmas is supplied at 3.4-4.4 GPa/830°C-890°C. The degree of fluxed melting of the mantle wedge varies between 17% and 28% (VF) and 6% and 22% (RA), with a slab flux fraction of 2%-4.5% (VF fluid) to 1%-1.5% (RA melt), and at melting depths 1–2.5 GPa (VF) and 2.4–2.8 GPa (RA). These conditions are consistent with a model whereby shallow, relatively low temperature slab fluids contribute to VF basalt genesis, whereas deeper and hotter slab melts control formation of RA basalts. The low-temperature slab dehydration is the cause of elevated Ba/Th in VF basalt due mainly to breakdown of lawsonite, whereas deeper breakdown of phengite by slab melting is the cause of elevated K and Rb in RA basalts. Melting in the garnet stability field, and at lower degrees of partial melting, is required for the elevated LILEs, LREEs, and HFSEs observed in the RA basalts. Less radiogenic Sr, Nd, Hf, and Pb in RA basalts are all attributable to lesser slab flux additions. The low H₂O predicted for RA basalt magmas ($\leq 1.5 \text{ wt }\%$) relative to that in VF basalt magmas ($\leq -8 \text{ wt }\%$) is also due to melt addition rather than fluid. All these conclusions are broadly consistent with existing models; however, in this study they are quantitatively confirmed by the geochemical mass balance deduced from petrological ABS3 model. Overall, the P-T-X(H₂O) structure of the slab and the mantle wedge exert the primary controls on arc basalt genesis.

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Geochemistry

Geophysics Geosystems

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1. Introduction

[2] Arc basalt compositions frequently exhibit spatial variations across and along arcs [*Elliott et al.*, 1997; *Woodhead et al.*, 1993]. Although the ultimate source of along arc variations may be difficult to constrain due to along arc variations in the composition of subducted material [*Carr et al.*, 2007; *Jicha et al.*, 2004; *Sadofsky et al.*, 2008; *Stern et al.*, 2003], slab age and geometry [*Kimura et al.*, 2005; *Yogodzinski et al.*, 2001], and upper plate contributions [*Kersting et al.*, 1996; *Kimura and Yoshida*, 2006], across arc variations are more broadly consistent and thus predictable. Studies on trace element and isotopic compositions of cross arc volcanic chains have the further advantage of minimizing the effects of source and slab variations, allowing models to focus on how slab processes vary with different depths within a single subduction zone. In most instances there are systematic increases in all incompatible elements, including large ion lithophile elements (LILEs), light rare earth elements (LREEs), high field strength elements (HFSEs), and alkalis from the volcanic front (VF) to the rear arc (RA), accompanied by systematic decreases in LILE/HFSE ratios [*Arculus and Johnson*, 1981; *Churikova et al.*, 2001; *Dickinson*, 1975; *Gill*, 1981; *Hatherton and Dickinson*, 1969; *Hochstaedter et al.*, 2001; *Kimura and Yoshida*, 2006; *Moriguti et al.*, Geochemistry Geophysics KIMURA ET AL.: CROSS-CHAIN MAGMA VARIATION IN N-IZU ARC 10.1029/2010GC003050

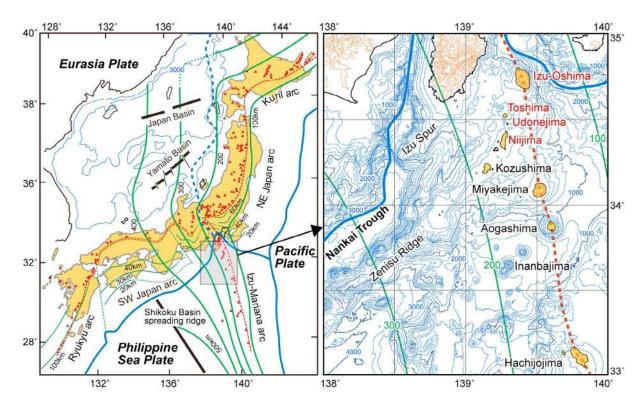


Figure 1. Location and tectonic setting of the northern Izu arc. (left) Thick blue line, trench; dotted thick blue line, North America and Eurasia Plate boundary; thick green line, depth contour of the Pacific Plate slab surface; thin green line, depth contour of the Philippine Sea Plate slab surface; thin dotted red line, volcanic front; thin blue line in the Japan Sea, depth contour; thick black line, spreading ridge; red triangle, Quaternary volcano. (right) Thick blue line, trench; thin blue line, depth contour; thick green line with number, depth contour of the Pacific Plate slab surface in km; brown thin line, contours on land; thick dotted red line, volcanic front.

2004; Pearce et al., 1995; Pearce and Parkinson, 1993; Pearce et al., 2005; Peate et al., 1997; Sakuyama and Nesbitt, 1986; Shibata and Nakamura, 1997; Stern et al., 2003; Tamura et al., 2007; Tatsumi and Eggins, 1995]. Isotopic compositions also typically vary, with the transition from VF to RA marked by decreasingly radiogenic Sr, Nd, Hf, and Pb, although there are exceptions to this [Hochstaedter et al., 2001; Shibata and Nakamura, 1997; Tollstrup et al., 2010].

[3] The northern Izu arc (N-Izu) is in many respects a typical oceanic arc, and exhibits clear across-arc geochemical variations [*Hochstaedter et al.*, 2001; *Ishikawa and Tera*, 1999; *Ishizuka et al.*, 2003, 2006; *Machida et al.*, 2008; *Stern et al.*, 2003; *Straub and Layne*, 2003; *Tamura et al.*, 2007; *Taylor and Nesbitt*, 1998; *Tollstrup et al.*, 2010]. The cross chain of interest for this study extends to the southwest from Izu-Oshima (Figure 1), the largest mafic active volcano in the Izu arc, and includes four mafic to intermediate volcanoes behind the VF, Toshima, Udonejima and Niijima. These include the only RA volcanoes built above sea level in the entire Izu-Bonin-Mariana (IBM) arc. The relatively robust magmatic nature of this cross chain may reflect extensional stresses on the bending NW Philippine Sea plate as it is subducted beneath central Japan.

[4] Several geochemical and isotopic studies have been carried out on Izu cross-chain volcanoes. On the basis of trace element and isotopic compositions, Taylor and Nesbitt [1998] inferred that there was less slab contribution in RA lavas relative to those of the VF. Hochstaedter et al. [2001] proposed a semidynamic model whereby RA slab melt addition causes fluxed melting in the RA mantle, with the depleted RA mantle then convecting to beneath the VF and melting again due to greater slab fluid addition to form VF basalt. Ishizuka et al. [2006] concluded that slab fluid affected the VF basalt source, whereas slab melt was more important for forming RA basalt magmas. Tamura et al. [2007] suggested phengite breakdown best explained high K contents in Izu RA basalts. Machida et al. [2008]

examined fluid-fluxed mantle melting and found differences in the slab flux addition between VF and RA. Control of magma compositions by variations in mantle source compositions has also been suggested by *Ishizuka et al.* [2003]. Recently, *Tollstrup et al.* [2010], in a study of lavas from the southern Izu arc lavas used inversion calculations to elucidate source mantle and slab flux chemistry and suggested that the source mantle composition was less heterogeneous than previously suggested, but that a transition from slab-derived aqueous fluids beneath the VF to supercritical fluids beneath the RA were the primary source of across arc variations.

Geochemistry

Geophysics Geosystems

[5] To date only qualitative or semiquantitative approaches have been applied in evaluating the contributions of different mantle/fluid/melt sources in the Izu arc, and fully integrated mass balance calculations of slab dehydration/melting and fluxed mantle melting within the mantle wedge have not been applied. Here we report a new geochemical data set including major and minor elements, trace elements determined by ICP-MS, and Sr-Nd-Hf-Pb isotope compositions on basalt and basaltic andesite lavas from the Izu-Oshima, Toshima, Udonejima, and Niijima chain (Figure 1). In concert with this data set, mass balance calculations have been performed using the Arc Basalt Simulator version 3 (ABS3), an algorithm developed for quantitative mass balance modeling of subduction related magmatism. ABS3 is an improved version of the ABS2 model that has been reported elsewhere [Kimura et al., 2009]. We use ABS3 modeling to infer intensive and extensive variables in the slab and constrain the mantle processes that lead to subduction zone basalt genesis for the Oshima-Niijima cross-chain magmatic system. This is the first application of the ABS (including ABS2 and ABS3) mass balance model to an oceanic arc where the effect of crustal contamination is believed to be minimal.

2. Geological Background and Samples

[6] The N-Izu volcanic chain is the result of subduction of the Pacific Plate beneath the Philippine Sea plate. The Pacific Plate adjacent to the trench was formed at 120–130 Ma [*Nakanishi et al.*, 1992] and the plate currently subducts at a rate of ~10 cm/ year [*Seno et al.*, 1993]. The subduction angle of the Pacific Plate (hereafter referred to as "slab") is steepened from 20° beneath the fore arc through 45° beneath the volcanic front to ~60° beneath the rear arc. The slab is traceable by seismic activity down to the 550 km mantle transition zone depth at a distance of ~400 km behind the trench [Seno et al., 1993; Stern et al., 2003]. This study focuses on a cross-arc island chain, located behind the VF volcano of Izu-Oshima (Figure 1). Along this chain the depths to the slab surface range from ~120 km at the VF (Izu-Oshima) to ~180 km (Toshima, Udonejima, Niijima; Figure 1). Izu arc volcanoes are built on ~25 km thick arc crust, which formed by magmatic additions in an oceanic environment over the past ~48 million years [Kodaira et al., 2007; Stern et al., 2003].

[7] Quaternary volcanoes of interest to this study belong to a late Cenozoic volcanic cross chain that developed on the NE part of the Zenisu Ridge [Ishizuka et al., 2003] (Figure 1). The ages of Zenisu Ridge volcanic rocks and the Quaternary Izu-Oshima to Niijima cross chain show younging from the RA to VF. The ages range from 6.4 \pm 0.12 Ma through 5.77 ± 0.05 Ma, 2.58 ± 0.06 Ma, 2.01 ± 0.05 Ma on the Zenisu Ridge, to historical and present-day eruptions at Quaternary RA and VF volcanoes [Ishizuka et al., 2006]. No evidence of volcanic activity is recognized in the Izu VF before 10 Ma [Isshiki, 1977, 1984, 1986]. This may reflect cessation of spreading in the Shikoku Basin back arc basin (BAB) ~15 Ma [Okino et al., 1994] leading to establishment of the present N-Izu volcanic arc [Ishizuka et al., 2003]. Beyond the study area, there has been rifting in the southern part of the Izu arc since ~5 Ma [Ishizuka et al., 2006], although systematic across arc geochemical variations appears to be disturbed perhaps due to upwelling of deep asthenosphere beneath the rift zone [Hochstaedter et al., 2001; Machida et al., 2008]. There is no evidence of rifting in the northernmost Izu arc, where our study is focused.

[8] For this study we have sampled Quaternary basalt to basaltic andesite lavas and tuffs from Izu-Oshima, Toshima, Udonejima, and Niijima islands (Figure 1). Izu-Oshima, Toshima, and Udonejima consist largely of basalt to basaltic andesite lavas [Isshiki, 1977, 1984, 1986]. Niijima consists largely of dacite to rhyolite lava domes and pyroclastics, although a mafic tuff ring volcano surrounding the Wakago Bay erupted basaltic andesite scoria [Isshiki, 1986]. Overall, the basalts to basaltic andesites used in this study were all erupted in the late Quaternary, and thus represent a relatively recent cross-chain sample set. The basalt to basaltic andesite lavas from Izu-Oshima are all younger than 20 ka [Isshiki, 1984]. Most lavas are historical and the latest eruption occurred in 1986 [Nakano and Yamamoto, 1991]. Toshima is located immediately on the RA side of Izu-Oshima. This island also consists of



lavas and pyroclastics of basalt to basaltic andesite composition. Radiocarbon dating of soils developed on volcanic surfaces suggest ages of about 7 ka [*Isshiki*, 1977]. Udonejima is an isolated and unpopulated island which consists of alternating lavas and pyroclastics with numerous feeder dikes. Udonejima volcano is dissected, suggesting a somewhat older age, although there are no radiometric age determinations. The estimated age of the volcano is a few tens of thousand years based on its morphology [*Isshiki*, 1986]. The Wakago tuff ring deposits on Niijima are younger than 3 ka [*Isshiki*, 1986].

Geochemistry

Geophysics Geosystems

[9] We report data for 58 samples, including 20 from Izu-Oshima, 19 from Toshima, 14 from Udonejima, and 5 mafic scoria samples from Niijima. All 58 samples were analyzed for major and trace elements by X-ray fluorescence spectrometer (XRF; Data Set S1).¹ Selected basalts to basaltic andesites were further analyzed by ICP-MS for trace elements and Sr, Nd, Pb, and Hf isotopes (Table 1). All the samples are plagioclase-rich and pyroxenes are ubiquitous with occasional small olivines contained in basaltic samples. All the samples are fresh and no clay minerals are observed.

3. Analytical Methods

[10] Lava samples were broken into small chips a few millimeters in diameter using an iron pestle and soaked in distilled water at ~60°C on a hot plate overnight. Supernatant water was exchanged and the same procedure repeated seven times. The rinsed chips were oven dried at 110°C overnight and pulverized by an agate mortar for 30 min. Powdered samples were heated at 1000°C for 2 h and fused with alkali fluxes to form low-dilution factor (sample: flux ratio of 1:2) glass discs. The glass discs were analyzed with a RIX 2000 XRF spectrometer (Rigaku) at Shimane University, Japan for 10 major elements and Ba, Cr, Ga, Ni, Pb, Rb, Sr, V, Y, and Zr [*Kimura and Yamada*, 1996] (see Data Set S1).

[11] Sample powders of 0.1000 g were digested in HF-HClO₄ in tightly sealed Teflon® jars overnight and evaporated to dryness. The dried samples were redigested with a mixed acid and diluted 1000x to 100 g stock solutions. The stock solutions were further diluted to 50000x and used for a standard

addition method ICP-MS analyses using a modified VG-PQ3 (Thermo Fisher Scientific) quadrupole ICP-MS at Shimane University [*Kimura and Yoshida*, 2006; *Kimura et al.*, 1995]. Standard basalt JB-2 (Izu-Oshima Miharayama basalt lava provided by the Geological Society of Japan) was also analyzed. Analytical results and comparison of our result is listed in Table 1.

[12] Sr and Nd were separated using a conventional cation resin separation for Sr and REE [lizumi et al., 1994, 1995]. Nd was further separated from REE separates using HDEHP columns [Richard et al., 1976]. Separated Sr and Nd were analyzed with a Finnigan MAT262 (Thermo Fisher Scientific) thermal ionization mass spectrometer (TIMS) at Shimane University [lizumi et al., 1994, 1995]. Pb was separated by single step miniature anion exchange columns and analyzed using a VG Plasma 54 (Thermo Fisher Scientific) multicollector ICP-MS (MC-ICP-MS) at Shimane University [Kimura et al., 2006]. Hf was separated by a two-step column method [Blichert-Toft et al., 1997] and analyzed with the same MC-ICP-MS. Standard materials of NIST SRM987 (Sr), La Jolla (Nd), JB-1a (Hf) and JB-2 (Pb) were also analyzed and the results listed in Table 1. All data for the standard samples agree well with reported values for these isotopic standards [Hanyu et al., 2005; Iizumi et al., 1994. 1995; Ishizuka et al., 2003] (see Table 1).

4. Geochemical Variations of N-Izu Quaternary Lavas

4.1. Major Elements

[13] Our major element XRF analytical results are similar to those reported previously (Figure 2). Almost all the N-Izu lavas plot as subalkali basalt suite on the total alkalis versus silica diagram (Figure 2), although two Udonejima basalt samples lie on the subalkaline and alkaline boundary [LeMaitre et al., 1989]. The Izu-Oshima samples are extremely low in total alkalis but some Toshima samples plot closer to the Izu-Oshima field. Major element composition of the Izu-Oshima lavas is identical with those from N-Izu volcanic front [Ishizuka et al., 2006]. Apart from some exceptions, total alkali contents tend to increase from Izu-Oshima to Toshima, Udonejima and Niijima corresponding to the increasing depth of the Wadati-Benioff zone (120-180 km). The difference between RA basalts is subtle, and the data cluster into VF (Izu-Oshima) and RA (Toshima, Udonejima, and Niijima) fields (Figure 2).

¹Auxiliary material data sets are available at ftp://ftp.agu.org/ apend/gc/2010gc003050. Other auxiliary material files are in the HTML.

| Table 1. | Major, 1 | iviajor, mace, and isotopic compositions of in-izu basan ravas | Ardmost | and man | | | | g | | | | | | | | | | | |
|---------------|----------|--|---------|---------|--------|--------|--------|--------|-------|-------|-------|--------|--------|--------|--------|--------|--------|----------------------|-----------|
| Element | IO-5 | IO-6 | IO-19 | JB-2 | TS-5 | TS-10 | TS-12 | TS-13 | UJ-2 | UJ-8 | 6-IU | UJ-10a | UJ-10b | NJ-1 | NJ-2 | NJ-5 | NJ-6a | Standard | Reference |
| Major (wt %) | | | | | | | | | | | | | | | | | | | |
| SiO_2 | 50.00 | 51.52 | 51.90 | 53.25 | 51.21 | 51.04 | 51.37 | 51.47 | 50.83 | | 50.32 | 50.66 | 50.79 | 51.29 | 51.03 | 51.18 | 49.61 | 52.93 ^b | |
| TiO_2 | 0.57 | 0.68 | 1.20 | 1.19 | 0.85 | 0.91 | 0.85 | 0.84 | 0.89 | | 0.94 | 0.96 | 0.74 | 1.11 | 1.13 | 1.13 | 1.11 | 1.18 ^b | |
| Al_2O_3 | 18.97 | 17.22 | 14.65 | 14.64 | 17.86 | 18.13 | 17.40 | 17.37 | 16.73 | | 16.92 | 17.18 | 15.48 | 16.62 | 16.42 | 16.39 | 16.79 | 14.55 ^b | |
| Fe_2O_3 | 10.30 | 11.58 | 15.33 | 14.25 | 10.44 | 10.73 | 10.53 | 10.56 | 10.57 | | 10.67 | 11.01 | 10.61 | 14.10 | 14.14 | 14.17 | 14.53 | 14.22 ^b | |
| MnO | 0.15 | 0.17 | 0.19 | 0.22 | 0.15 | 0.16 | 0.15 | 0.15 | 0.15 | | 0.15 | 0.16 | 0.15 | 0.20 | 0.20 | 0.20 | 0.20 | 0.22^{b} | |
| MgO | 5.52 | 5.99 | 4.74 | 4.62 | 6.03 | 5.75 | 6.58 | 6.60 | 7.34 | | 7.42 | 7.61 | 9.91 | 5.03 | 4.91 | 4.86 | 5.43 | 4.59 ^b | |
| CaO | 12.02 | 11.46 | 9.43 | 9.82 | 10.68 | 10.56 | 10.14 | 10.09 | 10.03 | | 9.65 | 9.94 | 9.97 | 9.49 | 9.50 | 9.46 | 9.95 | 9.76^{b} | |
| Na_2O | 1.19 | 1.29 | 1.85 | 2.04 | 2.24 | 2.24 | 2.25 | 2.28 | 2.55 | | 2.87 | 1.79 | 1.68 | 2.35 | 2.42 | 2.47 | 2.23 | 2.03^{b} | |
| K_2O | 0.09 | 0.22 | 0.39 | 0.42 | 0.39 | 0.35 | 0.49 | 0.52 | 0.40 | | 0.50 | 0.28 | 0.32 | 0.49 | 0.45 | 0.45 | 0.33 | 0.42^{b} | |
| P_2O_5 | 0.02 | 0.04 | 0.09 | 0.10 | 0.16 | 0.17 | 0.17 | 0.18 | 0.18 | | 0.20 | 0.21 | 0.13 | 0.13 | 0.14 | 0.14 | 0.12 | 0.10^{b} | |
| Total | 98.85 | 100.17 | 77.66 | 100.55 | 100.00 | 100.03 | 99.92 | 100.05 | 69.66 | | 99.65 | 99.80 | 99.78 | 100.81 | 100.34 | 100.45 | 100.31 | 100.00^{b} | |
| Ba | 94 | 141 | 222 | 222 | 145 | 128 | 147 | 157 | 66 | | 106 | 106 | 110 | 106 | 116 | 121 | 100 | I | |
| Ċ | 52 | 63 | 45 | 28 | 91 | 71 | 143 | 125 | 179 | | 211 | 207 | 184 | 25 | 19 | 17 | 25 | I | |
| Ga | 14.9 | 14.3 | 14.8 | 17.0 | 18.7 | 19.4 | 18.0 | 18.0 | 14.1 | 15.6 | 13.1 | 16.0 | 15.2 | 17.1 | 17.6 | 17.5 | 17.2 | I | |
| ï | 26 | 23 | 17 | 17 | 35 | 30 | 85 | 71 | 88 | | 105 | 103 | 102 | 6 | 8 | 6 | 11 | Ι | |
| Sr | 178 | 169 | 178 | 178 | 378 | 382 | 395 | 393 | 462 | | 489 | 494 | 482 | 271 | 270 | 269 | 270 | I | |
| ^ | 337 | 387 | 612 | 575 | 285 | 297 | 268 | 272 | 235 | | 209 | 231 | 239 | 399 | 399 | 393 | 400 | I | |
| Y | 11.7 | 13.7 | 20.9 | 24.9 | 16.0 | 16.6 | 16.7 | 16.8 | 16.8 | | 18.1 | 17.8 | 17.1 | 19.1 | 18.7 | 19.1 | 18.0 | I | |
| Zr | 25.4 | 31.4 | 51.9 | 51.2 | 55.2 | 59.4 | 62.1 | 61.3 | 63.0 | | 64.4 | 64.9 | 63.9 | 58.1 | 58.2 | 58.2 | 52.9 | I | |
| Trace (ppm) | | | | | | | | | | | | | | | | | | | |
| Li | 7.63 | 4.94 | 19.24 | 8.87 | 5.86 | 3.58 | 3.98 | 4.26 | | | | 3.28 | 8.85 | 4.14 | 4.21 | 4.96 | 5.97 | 7.86 ^b | |
| Be | 0.138 | 0.133 | 0.337 | 0.238 | 0.524 | 0.397 | 0.395 | 0.437 | | | | 0.406 | 0.427 | 0.398 | 0.481 | 0.433 | 0.360 | 0.247 ^b | |
| Rb | 1.83 | 3.08 | 6.29 | 5.88 | 4.14 | 4.62 | 6.62 | 7.02 | | | | 3.10 | 5.12 | 6.35 | 6.42 | 6.27 | 5.79 | 6.19 ^b | |
| Y | 11.8 | 12.9 | 22.4 | 21.5 | 15.4 | 16.2 | 15.7 | 15.5 | | | | 16.4 | 16.8 | 19.0 | 18.1 | 18.9 | 17.1 | 20.9^{b} | |
| Zr | 18.8 | 21.9 | 51.2 | 44.5 | 72.3 | 51.5 | 55.5 | 53.2 | | | | 50.9 | 54.9 | 51.2 | 56.5 | 48.8 | 66.7 | 47.0 ^b | |
| Рb | 0.101 | 0.152 | 0.531 | 0.434 | 1.962 | 1.283 | 1.411 | 1.395 | | | | 1.642 | 1.591 | 1.253 | 1.141 | 1.171 | 1.035 | 0.467^{b} | |
| Sb | 0.077 | 0.196 | 0.304 | 0.217 | 0.054 | 0.039 | 0.042 | 0.041 | | | | 0.038 | 0.061 | 0.046 | 0.036 | 0.036 | 0.046 | $0.231^{\rm b}$ | |
| \mathbf{Cs} | 0.217 | 0.497 | 0.736 | 0.728 | 0.537 | 0.249 | 0.370 | 0.397 | | | | 0.131 | 0.094 | 0.366 | 0.361 | 0.361 | 0.309 | 0.771^{b} | |
| La | 0.661 | 1.210 | 2.565 | 2.393 | 8.564 | 6.715 | 6.563 | 6.849 | | | | 6.472 | 6.128 | 4.784 | 4.264 | 4.838 | 3.863 | 2.188 ^b | |
| Ce | 2.05 | 3.22 | 6.81 | 6.61 | 19.87 | 15.83 | 15.99 | 16.09 | | | | 15.41 | 14.95 | 11.36 | 10.82 | 11.85 | 9.61 | 6.42 ^b | |
| Pr | 0.379 | 0.590 | 1.158 | 1.120 | 2.662 | 2.259 | 2.212 | 2.264 | | | | 2.258 | 2.153 | 1.695 | 1.667 | 1.746 | 1.480 | 1.095 ^b | |
| PN | 2.202 | 3.234 | 6.224 | 5.953 | 11.850 | 10.730 | 10.300 | 10.450 | | | | 10.630 | 10.020 | 8.357 | 8.173 | 8.351 | 7.341 | 6.065 ^b | |
| Sm | 0.858 | 1.207 | 2.097 | 2.135 | 2.799 | 2.717 | 2.594 | 2.599 | | | | 2.724 | 2.684 | 2.441 | 2.440 | 2.459 | 2.248 | 2.149 ^b | |
| Eu | 0.386 | 0.497 | 0.773 | 0.806 | 0.954 | 0.957 | 0.897 | 0.909 | | | | 0.984 | 0.980 | 0.956 | 0.952 | 0.962 | 0.871 | 0.818^{b} | |
| Gd | 1.345 | 1.764 | 2.998 | 3.218 | 3.005 | 2.986 | 2.856 | 2.836 | | | | 3.020 | 3.037 | 3.042 | 3.052 | 3.088 | 2.906 | 3.016^{b} | |
| Tb | 0.254 | 0.318 | 0.520 | 0.563 | 0.457 | 0.467 | 0.452 | 0.451 | | | | 0.485 | 0.491 | 0.516 | 0.512 | 0.522 | 0.454 | 0.560^{b} | |
| Dy | 1.729 | 2.196 | 3.506 | 3.922 | 2.883 | 3.029 | 2.928 | 2.859 | | | | 3.141 | 3.183 | 3.427 | 3.428 | 3.432 | 3.099 | 3.856 ^b | |
| Но | 0.381 | 0.488 | 0.795 | 0.849 | 0.594 | 0.621 | 0.601 | 0.590 | | | | 0.642 | 0.652 | 0.721 | 0.718 | 0.722 | 0.653 | 0.829^{b} | |
| Er | 1.063 | 1.383 | 2.126 | 2.355 | 1.606 | 1.695 | 1.630 | 1.598 | | | | 1.729 | 1.772 | 1.975 | 1.963 | 2.002 | 1.777 | 2.336^{b} | |
| Tm | 0.169 | 0.218 | 0.330 | 0.372 | 0.254 | 0.261 | 0.249 | 0.246 | | | | 0.266 | 0.277 | 0.305 | 0.311 | 0.312 | 0.281 | 0.364^{b} | |
| Yb | 1.139 | 1.450 | 2.206 | 2.462 | 1.724 | 1.708 | 1.646 | 1.628 | 1.598 | 1.712 | 1.776 | 1.752 | 1.813 | 2.057 | 2.033 | 2.063 | 1.899 | 2.440 ^b | |
| Lu | 0.170 | 0.228 | 0.349 | 0.379 | 0.265 | 0.260 | 0.252 | 0.250 | | | | 0.268 | 0.274 | 0.311 | 0.314 | 0.315 | 0.284 | $0.378^{\rm b}$ | |

Table 1. Major, Trace, and Isotopic Compositions of N-Izu Basalt Lavas^a



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|-------------|-------|
| | 10 10 |
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| (continued) | u CI |
| Table 1. | ī |

Geochemistry Geophysics Geosystems

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| Element | IO-5 | IO-6 | IO-19 | JB-2 | TS-5 | TS-10 | TS-12 | TS-13 | UJ-2 | NJ-8 | 01-9 | UJ-10a | UJ-10b | NJ-1 | NJ-2 | NJ-5 | NJ-6a | Standard | Reference |
|--|---------------------------|---|--------------------------------|------------------------------------|-------------------------|--------------------|--|-------------------------------|------------|-------------|------------|--------------|------------|-------------|-------------|--------------|------------|-----------------------|-----------------------|
| Hf | 0.507 | 0.757 | 1.304 | 1.438 | 1.894 | 1.513 | 1.556 | 1.515 | 1.375 | 1.429 | 1.564 | 1.471 | 1.557 | 1.377 | 1.404 | 1.387 | 1.213 | 1.434 ^b | |
| Та | 0.0026 | 0.0107 | 0.0280 | 0.0318 | 0.1479 | 0.0855 | 0.0952 | 0.0911 | 0.0973 | 0.1062 | 0.1154 | 0.1108 | 0.1074 | 0.0938 | 0.0803 (| 0.0903 | 0.0649 (| 0.0349 ^b | |
| Π | 0.012 | 0.025 | 0.026 | 0.038 | 0.075 | 0.021 | 0.023 | 0.063 | 0.006 | 0.014 | 0.004 | 0.007 | 0.025 | 0.058 | 0.057 | 0.054 (| 0.049 (| 0.037 ^b | |
| Pb | 1.44 | 2.32 | 3.42 | 5.35 | 2.93 | 2.38 | 3.24 | 3.48 | 1.80 | 2.33 | 2.63 | 2.39 | 2.27 | 2.15 | 2.34 | 2.07 | 3.85 | 5.64 ^b | |
| Th | 0.087 | 0.165 | 0.343 | 0.252 | 1.482 | 0.797 | 0.813 | 0.980 | 0.528 | 0.516 | 0.689 | 0.638 | 0.684 | 0.751 | 0.596 (| 0.765 (| 0.493 (| 0.247 ^b | |
| U | 0.042 | 0.082 | 0.148 | 0.156 | 0.337 | 0.169 | 0.224 | 0.218 | 0.131 | 0.157 | 0.142 | 0.163 | 0.184 | 0.183 | 0.178 (| 0.183 (| 0.169 (| 0.150^{b} | |
| Isotopes ⁸⁷ Sr/ ⁸⁶ Sr | 0.703673 | 0.703673 0.703689 0.703636 0.703676 0.703428 0.703418 | 0.703636 | 0.703676 | 0.703428 | | 0.703401 | 0.703399 | 0.703247 | 0.703244 | 0.703273 | 0.703283 | 0.703272 | 0.703294 | 0.703351 | 0.703306 | 0.703390 | 0.710248° | 0.710250° |
| 2SE | 0.000015 | 0.000015 0.000017 | 0.000017 | 0.000015 | 0.000015 0.000015 | | 0.000015 | 0.000014 | 0.000015 | 0.000015 | 0.000015 | 0.000015 | 0.000015 | 0.000017 | 0.000015 0 | 0.000015 0 | 0.000015 (| 0.000013 | |
| 143Nd/144Nd | 0.513086 | 0.513086 0.513064 0.513100 | 0.513100 | 0.513076 | 0.513105 0.513006 | 0.513006 | 0.513053 | 0.512997 | 0.513026 | 0.513101 | 0.513084 | 0.513080 | 0.513088 | 0.513144 | 0.513026 (| 0.513062 (| 0.513096 (| 0.511867 ^d | 0.511848 ^d |
| 2SE | 0.000014 | 0.000014 0.000014 | 0.000014 | 0.000013 | 0.000016 0.000 | 015 | 0.000011 | 0.000014 | 0.000014 | 0.000012 | 0.000013 | 0.000012 | 0.000014 | 0.000012 | 0.000011 (| 0.000014 | 0.000013 (| 0.000013 | |
| 176 Hf/ 177 Hf | 0.283300 | Ι | I | 0.283282 | 0.283304 | I | I | I | 0.283219 | I | 0.283219 | | J | I | 0.283223 - | | - | 0.282997° | 0.282990° |
| 2SE | 0.000018 | I | I | 0.000019 | 0.000040 | I | I | I | 0.000014 | I | 0.000014 | | I | I | 0.000014 | | - | 0.000007 | |
| $^{206} Pb/^{204} Pb$ | 18.4018 | 18.3968 | 18.4194 | 18.3409 | 18.2692 | 18.2576 | 18.2526 | 18.2385 | 18.2094 | 18.2103 | 18.2168 | 18.1845 | 18.1566 | 18.3016 | 18.2858 | 18.3072 | 18.1643 | 18.3409 ^b | 18.3390 ^b |
| 2SE | 0.0015 | 0.0015 | 0.0011 | 0.0013 | 0.0011 | 0.0009 | 0.0011 | 0.0011 | 0.0011 | 0.0010 | 0.0011 | 0.0010 | 0.0013 | 0.0011 | 0.0010 | 0.0010 | 0.0008 (| 0.0013 | |
| $^{207}\text{Pb}/^{204}\text{Pb}$ | 15.5552 | 15.5517 | 15.5532 | 15.5653 | 15.5332 | 15.5334 | 15.5346 | 15.5030 | 15.5388 | 15.5312 | 15.5203 | 15.5410 | 15.5382 | 15.5374 | 15.5335 | 15.5324 | 15.5624 | 15.5653 ^b | 15.5610 ^b |
| 2SE | 0.0013 | 0.0014 | 0.0011 | 0.0012 | 0.0010 | 0.0008 | 0.0010 | 0.0011 | 0.0010 | 0.0010 | 0.0009 | 0.0010 | 0.0011 | 0.0009 | 0.0010 | 0.0009 | 0.0010 (| 0.0012 | |
| $^{208}\text{Pb}/^{204}\text{Pb}$ | 38.3161 | 38.3093 | 38.3505 | 38.2677 | 38.2146 | 38.1957 | 38.1990 | 38.1962 | 38.1473 | 38.1464 | 38.1276 | 38.1169 | 38.0683 | 38.2257 | 38.2217 | 38.2307 | 38.1040 | 38.2677 ^b | 38.2730 ^b |
| 2SE | 0.0033 | 0.0039 | 0.0034 | 0.0031 | 0.0029 | 0.0037 | 0.0030 | 0.0030 | 0.0034 | 0.0032 | 0.0024 | 0.0028 | 0.0030 | 0.0024 | 0.0031 | 0.0026 | 0.0039 (| 0.0031 | |
| ^a Isotope ratio reference values are from <i>lizumi et al.</i> [1994] for SRM981 Sr, <i>liizumi et al.</i> [1995] for LaJolla Nd, <i>Hanyu et al.</i> [2005] for JB-1b Hf, and <i>Ishizuka et al.</i> [2003] for JB-2 Pb. Error for analyzed standards are given as 2 standard errors of n = 3 for Sr and Nd and n = 1 for Hf and Pb. | tio referen given as 2 | ce values a 2 standard | tre from <i>Ii</i> errors of 1 | <i>zumi et al.</i> $n = 3$ for 5 | [1994] for Sr and Nd | - SRM981 and n = 1 | A981 Sr, <i>liizumi et al</i> $n = 1$ for Hf and Pb. | <i>i et al.</i> [19 1d Pb. | 95] for La | Jolla Nd, J | Hanyu et a | ıl. [2005] f | or JB-1b l | Hf, and Isl | iizuka et a | l. [2003] fi | or JB-2 Pt | . Error for | analyzed |
| ^b IR-2 hasalt standard rock material nrovided by the Geological Survey of Ianan | ilt standar | d rock mai | terial nrov | rided hv th | he Geologia | ral Survey | r of Janan | | | | | | | | | | | | |

^bJB-2, basaft standard rock material provided by the Geological Survey of Japan. ^cSRM987, Sr carbonate standard provided from the National Institute of Standards and Technology. ^dLaJolla, Nd isotope standard provided from the Woods Hole Oceanographic Institution. ^eJB-1b, basalt standard rock material provided by the Geological Survey of Japan.

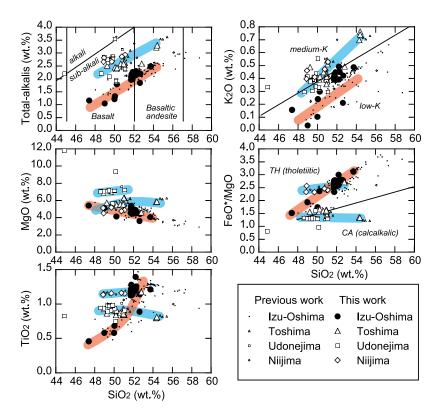


Figure 2. Major element compositions of N-Izu basalt to basaltic andesite lavas. Thick red line, volcanic front trend; thick blue line, rear-arc trends. Discrimination lines are from *LeMaitre et al.* [1989] for the total alkalis–silica (TAS) diagram, *Peccerillo and Taylor* [1976] for the K-silica diagram, and *Miyashiro* [1974] for the FeO*/MgO-silica diagram.

[14] A similar pattern is evident from the K_2O versus silica diagram. Izu-Oshima lavas are distinctively low in K_2O , and all samples but one plot in the low-K field [*Peccerillo and Taylor*, 1976]. In contrast, the RA lavas plot around the low-K and medium-K discrimination line or in the medium-K field.

Geochemistry

Geophysics Geosystems

[15] The MgO contents of the lavas range from 3.8 to 9.5 wt % with relatively high MgO in RA lavas at given silica content (Figure 2). Because of the presence of a small number of secondary veins, we believe that one low-silica lava sample with high MgO (~12 wt %) from Udonejima appears to be affected by alteration (although obvious veins were removed prior to analysis). The sample has been eliminated from further discussion.

[16] The FeO/MgO diagram reveals that all Izu-Oshima and Niijima lavas have tholeiitic affinities, whereas the Udonejima and Toshima lavas are calc-alkalic. Although the Niijima lava is rich in FeO*, the FeO*/MgO ratio is constant suggesting an atypical tholeiitic trend. The TiO₂ contents of the lavas increase with increasing silica for Izu-Oshima tholeiitic lavas, whereas no increases are shown by the RA lava suites (Figure 2). The Niijima lavas again do not behave like typical tholeiites, which show TiO_2 -silica covariation but instead, have their highest TiO_2 content in lavas with the lowest silica content. The TiO_2 contents of the most mafic lavas increase from Izu-Oshima through Toshima and Udonejima to Niijima, similar to the variations in total alkali contents. This indicates that Izu-Oshima lavas derive from primitive magmas resulting from either higher degrees of melting (*F*) than RA lavas, and/or that Izu-Oshima lavas are derived from a more depleted mantle source.

4.2. Trace Elements

[17] Trace element abundances and variation are shown in normalized multielement plots (Figure 3). The data are from selected less differentiated basalts from each volcanic island together with two basaltic andesites (JB-2 and lava flow near JB-2 sample site) from Izu-Oshima for comparison (Table 1).

[18] The Izu-Oshima VF basalts are extremely low in incompatible element abundances, including K_2O , P_2O_5 , and TiO₂ (Figure 2), and are also very low in REEs, with extremely depleted LREE and

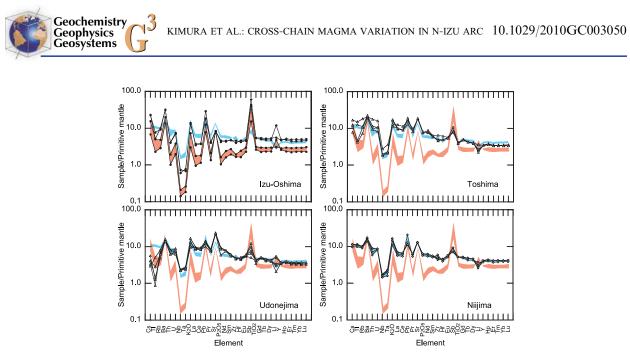


Figure 3. Normalized multielement plot of representative basalt lavas from N-Izu. Normalization values are from *Sun and McDonough* [1989]. Red and blue shaded areas are references for VF and RA basalts. Two elevated values in Izu-Oshima are from JB-2 andesite and a sample from the same lava flow.

HFSE. Positive spikes are prominent in Cs, Ba, U, Pb, Sr, and Sb. Li also shows a slight positive anomaly. The behavior of the REE and the trace HFS elements is similar to that of the major HFS elements (TiO₂ and P₂O₅). These characteristics are ubiquitous in the late Cenozoic VF lavas from elsewhere along the Izu arc [*Ishizuka et al.*, 2006; *Tamura et al.*, 2007].

[19] In contrast to VF basalts, the RA basalts are relatively enriched in incompatible elements. RA basalts show elevated REE with fractionated LREE and HFSE, with the HFSEs strongly depleted relative to LREE and LILEs. Some samples have Cs and Tl abundances identical to those in VF basalts [Tamura et al., 2007]. Positive spikes in Ba, K₂O, Pb. Sr. and Sb are all similar between RA and VF lavas; however, the increased concentrations of these elements are less prominent in RA lavas than in VF lavas. Although depletions in Tl and Rb are evident in some Toshima and all Udonejima basalts, other incompatible element characteristics are similar between Toshima, Udonejima, and Niijima basalts. Thus, although there are some internal variations within RA and VF lavas, the differences between these two groups are consistently much more significant. In the following discussion the N-Izu basalts are classified into VF and RA groups as a basis for comparison.

4.3. Incompatible Trace Element Behavior

[20] Incompatible trace element ratios are useful in characterizing the sources that contribute to basaltic

magmas. In subduction zone settings, Ba/Th and Pb/Ce are believed to be indicators of slab-derived fluid addition to the mantle peridotite basalt source because Ba and Pb are more fluid mobile whereas Th and Ce are relatively compatible in melts but not in hydrous fluids [Pearce et al., 2005; Peate et al., 1997]. Sb behaves similarly to Ba and Pb during slab dehydration [Bebout et al., 1999], such that Sb/Gd is also useful. Figure 4 plots [Ba/Th]n versus [Pb/Ce]n and [Sb/Gd]n versus [Pb/Ce]n ([X/ YIn indicates primitive mantle normalized element ratios where [X/Y]n = 1 using the primitive mantle (PM) estimate of Sun and McDonough [1989]). In both plots, the RA basalts plot close to PM but still have somewhat elevated Ba/Th, Pb/Ce and Sb/Gd. In contrast, the VF basalts have very high ratios suggesting significant slab fluid addition.

[21] The [Nb/Yb]n, [La/Yb]n, and [Nb/Zr]n ratios record differences in the amount of source depletion, degree of mantle melting and/or the contribution of residual garnet in the basalt source mantle [*Pearce et al.*, 2005; *Peate et al.*, 1997]. [Nb/Yb]n and [Nb/Zr]n are significantly lower than unity, indicating strong depletion in incompatible HFSE in the mantle source compared to PM. In contrast, [La/Yb]n is less than unity in the VF but greater than unity in the RA. There are near-linear correlations between [La/Yb]n and other HFS element ratios increasing in the order of VF, Niijima, and Udonejima and Toshima in the RA. This correlation and trend might be achieved by different degrees of partial melting of compositionally similar

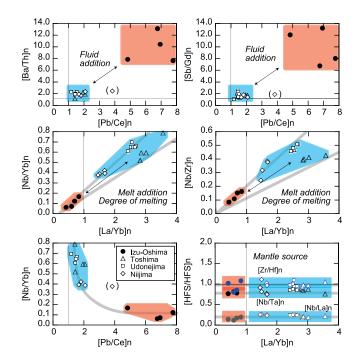


Figure 4. Incompatible trace element ratios of basalt lavas from the N-Izu arc. Trace element concentrations are normalized to primitive mantle values [*Sun and McDonough*, 1989]. An outlier from Niijima basalts is bracketed. Red and blue fields are for VF and RA basalts. Thin lines are primitive mantle ratios. Note that PM values for Nb/Zr and NbYb are unity and off scale.

sources or by mixing between low La/Nb and high La/Nb melts.

Geochemistry

Geophysics Geosystems

[22] A comparison between HFSE ratios and trace element indices that are sensitive to fluid inputs (e.g., [Nb/Yb]n versus [Pb/Ce]n) also suggests a continuum between VF and RA processes with the results fitting a hyperbolic trend (Figure 4). This relation may be produced by fluid addition from the slab to the VF basalt source (increasing Pb/Ce), whereas melt addition occurs within the RA basalt source, as suggested by previous workers [*Ishizuka et al.*, 2006; *Tamura et al.*, 2007; *Tollstrup et al.*, 2010; *Turner et al.*, 1997].

[23] Finally, unlike other incompatible element ratios, the HFSE ratios ([Zr/Hf]n, [Nb/Ta]n, [Nb/La]n) versus [La/Yb]n show no consistent change across the rear arc although [La/Yb]n shows considerable change (Figure 4). This may indicate a similar mantle source composition beneath RA and VF, coupled with differences in the degree of partial melting (greater for VF and less for RA). If slab melt addition is the cause elevated [Nb/Yb]n in the RA basalts, the HFSE ratios probably cannot remain unchanged as these ratios would differ between AOC, SED, and PM [*Green*, 1995]. We evaluate this problem further in sections 5.2.6 and 5.4.1.

4.4. Isotopic Variations

[24] There are also distinct differences between the VF and RA lavas in terms of their radiogenic isotope compositions (Figure 5). In general Izu VF basalts typically show more radiogenic ⁸⁷Sr/⁸⁶Sr than in RA basalts [Hochstaedter et al., 2001; Ishizuka et al., 2003, 2006; Taylor and Nesbitt, 1998; Tollstrup et al., 2010]. The same is true in the N-Izu (Figure 5), where the Izu-Oshima VF lavas have elevated ⁸⁷Sr/⁸⁶Sr at a given ¹⁴³Nd/¹⁴⁴Nd compared to RA basalts. However, the subparallel fields, defined by the VF and RA basalts, are less prominent in our study area than in other parts of the N-Izu arc, even though the N-Izu VF basalt field almost exactly overlaps that of other Izu VF basalts [Ishizuka et al., 2006; Tamura et al., 2007]. In detail, the N-Izu RA basalts also have a slightly enriched Sr isotopic composition compared to other Izu RA basalts and define a field intermediate between VF and RA basalts.

[25] The Pb isotopic results show similar features to those evident from Sr, with more radiogenic Pb isotopic compositions found in VF basalts relative to those in RA basalt. Overall, the Izu and N-Izu basalts plot on a narrow linear array suggesting near binary mixing of Pb isotope components. Elevated fluid mobile element concentrations (including Pb

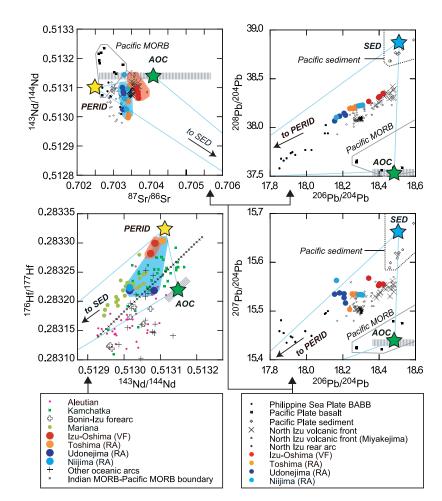


Figure 5. Sr-Nd, Pb-Pb, and Nd-Hf isotope plots of basalt lavas from N-Izu. Thick dotted gray line, isotopic variation of the Pacific MORB (Geochemistry of rocks of the oceanic and the continents (GEOROC), Max-Planck-Institut für Chemie, http://georoc.mpch-mainz.gwdg.de/georoc/). Stars are from the isotopic ratios used for ABS3 calculations as shown in Table 2. Thin blue lines connect sediment (SED), altered oceanic crust (AOC), and peridotite (PERID) source components.

and Sr) in VF basalts are consistent with the inferred strong slab fluid contribution to the N-Izu VF basalts, because these should carry radiogenic Pb and Sr [*Hochstaedter et al.*, 2001; *Ishizuka et al.*, 2003, 2006; *Taylor and Nesbitt*, 1998; *Tollstrup et al.*, 2010]. The linear Pb isotope relationship extends to the back arc Shikoku Basin basalt (Philippine Sea plate BABB in Figure 5), suggesting slab flux addition to a mantle source that is compositionally restricted. It is interesting that the Izu arc basalts have such uniform Pb isotopic compositions over a 1000 km range, considering that the age of the subducting Pacific Plate changes significantly over this distance (120–150 Ma).

Geochemistry

Geophysics Geosystems

[26] For Nd isotopes, along-arc variation of the mantle source has been suggested [*Hochstaedter et al.*, 2001; *Ishizuka et al.*, 2003, 2006; *Taylor and Nesbitt*, 1998]; however, increasing data from

the Izu-Bonin arc gave rise to the suggestion that these variations relate to local variety of the mantle source [Tollstrup et al., 2010]. To investigate mantle heterogeneity across the arc, we examined Nd-Hf isotopic compositions. Except for one Udonejima sample, all analyzed samples fall in the Indian MORB mantle domain [Woodhead et al., 2001] forming a trend that is slightly oblique to the Indian MORB and Pacific MORB mantle domain boundary. This is consistent with what is seen for Pb isotopic compositions, which also plot in the field for Indian Ocean basalts. The N-Izu Nd-Hf trend overlaps with the field of Mariana arc basalts rather than Bonin-Izu fore arc (Figure 5). This strongly suggests that the mantle wedge beneath the N-Izu largely has Indian MORB-like mantle affinities, with some heterogeneity (as shown by the Udonejima sample). The VF lavas are the most radiogenic, whereas RA basalts scatter considerably. The larger



Hf isotope variation in Mariana arc lavas has been interpreted as being due to contributions from slab melt to the mantle source [*Tollstrup and Gill*, 2005] and for RA in the southern Izu arc from "supercritical melt" [*Tollstrup et al.*, 2010]. The latter would explain the larger Hf-Nd isotopic variations seen in N-Izu RA basalts. However, as changes in element abundances and isotope ratios for Nd and Hf are subtle, more data are required to address this for N-Izu, and this is beyond the scope of this paper.

Geochemistry

Geophysics Geosystems

5. Origin of the Across Arc Chemical Variations

5.1. Arc Basalt Simulator Version 3 Model

[27] Kimura et al. [2009] developed a quantitative forward mass balance model for studying arc basalt genesis that explicitly addresses (1) slab dehydration during prograde metamorphism in the subducting plate, (2) fluid-mantle reaction at the base of the mantle wedge and in the high-temperature asthenospheric mantle wedge, and (3) P-T-XH₂O parameterized fluid-fluxed mantle melting. Arc Basalt Simulator version 2 (ABS2) is an Excel spreadsheet based forward model of coupled slab dehydrationfluid fluxed mantle melting [Kimura et al., 2009]. The model allows the user to input three source compositions: (1) subducted sediment (SED), (2) altered oceanic crust (AOC), and (3) choice of variously depleted MORB mantle (PERID). However, in order to study the origin of the N-Izu basalts, slab melting and melt-fluxed mantle melting processes need to be simulated together with the slab dehydration and fluid-fluxed melting functions already available with ABS2. We have modified the ABS2 and developed ABS3 for these purposes.

[28] Briefly, the ABS3 model includes new features that simulate slab melting and melt fluxed mantle melting (see Figure 1 in Text S1). The slab melting is based on incorporation of the available experimental data for AOC and SED at slab conditions between 0 and 6 GPa, and these data are incorporated into petrogenetic phase diagrams of slab AOC and SED (see Figure 2 in Text S1). For the appropriate slab melting conditions, melt compositions are calculated using the degree of melting and residual mineralogy of the slab SED/AOC. The H₂O content in the slab melts is calculated based on modes of consumed hydrous silicate minerals.

[29] Melt fluxed mantle melting is then simulated based on pMELTS calculations of mantle compo-

sitions variously modified by addition of the slab melt by up to 30 wt %. The effect of H_2O on melting is based on the formulation of *Katz et al.* [2003] assuming that H_2O in the slab melt controls the degree of melting of the fluxed mantle (see Text S1).

[30] Because of the limited availability of experimental data sets for carbonaceous, chert, and pelagic sediments, the SED component used in ABS3 is limited to a bulk sediment rather than distinct treatment on the various sediment components as done in ABS2. The ABS3 slab SED petrogenetic model for both subsolidus and supersolidus is calculated and formulated for the terrigeneous sediment composition studied by Hacker [2008]. This composition is close to that of the average upper continental crust composition [Taylor and McLennan, 1985]. At present this limitation is unavoidable due to the limited experimental results available for various SED components at wide P-T range. Slab dehydration and fluid-fluxed mantle melting modes are otherwise completely identical to the ABS2 model.

[31] Full details of the modification and calculation scheme of ABS3 are given in Text S1. A program code built on a Microsoft based Excel spreadsheet is available by downloading Software S1. Readers should also refer to *Kimura et al.* [2009] for the concept of ABS2, which is similar in many respects to ABS3. The advantages of the ABS3 model are its fully integrated petrologic treatment of slab fluid dehydration, slab melting, and fluid/melt-fluxed mantle melting with fully consistent incompatible element mass balance including isotope systems. An automated fitting function allows the best fit conditions to be found, including intensive and extensive parameters of slab-mantle processes (see Text S1 and Software S1 and discussions below).

5.2. ABS3 Examination of N-Izu Basalts

[32] In the following we describe the practical application of ABS3 to the N-Izu basalts.

5.2.1. Primary Arc Basalt Compositions

[33] An important constraint for ABS3 calculations is the primary arc basalt composition. The incompatible trace element abundances in the primary basalt can be estimated by various schemes. One example is an olivine maximum fractionation model, whereby equilibrium olivine is added until the basalt composition is in equilibrium with mantle peridotite [*Takahashi*, 1986]. *Kimura et al.* [2006] and *Machida et al.* [2008] have applied this method by

| Element | Izu-Oshima | Toshima | Niijima | X(AOC) | X(SED) | X(PERID) |
|--|------------|---------|---------|---------|---------|----------|
| Rb | 1.47 | 3.40 | 4.63 | 20.0 | 25.1 | 0.018 |
| Ba | 75.5 | 118 | 79.7 | 15.6 | 244 | 0.203 |
| Th | 0.070 | 0.948 | 0.394 | 0.173 | 8.00 | 0.0026 |
| U | 0.033 | 0.215 | 0.135 | 0.390 | 1.19 | 0.0010 |
| Nb | 0.081 | 1.26 | 0.828 | 2.89 | 3.87 | 0.057 |
| Та | 0.0062 | 0.095 | 0.052 | 0.210 | 0.239 | 0.0039 |
| Κ | 604 | 2051 | 2185 | 5147 | 7286 | 30.5 |
| La | 0.529 | 5.48 | 3.09 | 3.40 | 27.3 | 0.087 |
| Ce | 1.64 | 12.7 | 7.69 | 11.4 | 31.3 | 0.303 |
| Pb | 1.15 | 0.960 | 0.670 | 0.430 | 6.82 | 0.011 |
| Pr | 0.303 | 1.51 | 0.980 | 2.06 | 5.37 | 0.068 |
| Sr | 142 | 242 | 216 | 110 | 161 | 5.17 |
| Nd | 1.76 | 7.58 | 5.87 | 11.3 | 28.7 | 0.417 |
| Sm | 0.686 | 1.79 | 1.80 | 3.95 | 6.35 | 0.184 |
| Zr | 15.0 | 46.3 | 35.8 | 112 | 48.1 | 4.22 |
| Hf | 0.405 | 1.21 | 0.970 | 3.07 | 0.878 | 0.134 |
| Eu | 0.309 | 0.610 | 0.697 | 1.34 | 1.46 | 0.079 |
| Gd | 1.08 | 1.92 | 2.32 | 5.55 | 5.46 | 0.301 |
| Tb | 0.203 | 0.292 | 0.363 | 1.01 | 0.880 | 0.060 |
| Dy | 1.38 | 1.85 | 2.48 | 6.56 | 5.26 | 0.441 |
| Y | 9.42 | 9.88 | 13.7 | 40.7 | 36.8 | 2.78 |
| Но | 0.305 | 0.380 | 0.523 | 1.43 | 1.10 | 0.100 |
| Er | 0.850 | 1.03 | 1.42 | 4.09 | 3.15 | 0.306 |
| Tm | 0.135 | 0.163 | 0.225 | 0.617 | 0.461 | 0.049 |
| Yb | 0.911 | 1.10 | 1.52 | 4.02 | 2.90 | 0.331 |
| Lu | 0.136 | 0.169 | 0.227 | 0.636 | 0.415 | 0.052 |
| Isotopes | | | | | | |
| Isotopes ⁸⁷ Sr/ ⁸⁶ Sr | 0.70367 | 0.70343 | 0.70339 | 0.70400 | 0.71055 | 0.70240 |
| ¹⁴³ Nd/ ¹⁴⁴ Nd | 0.51309 | 0.51310 | 0.51310 | 0.51310 | 0.51235 | 0.51310 |
| ²⁰⁶ Pb/ ²⁰⁴ Pb | 18.401 | 18.271 | 18.165 | 18.400 | 18.450 | 17.500 |
| ²⁰⁷ Pb/ ²⁰⁴ Pb | 15.556 | 15.535 | 15.563 | 15.430 | 15.683 | 15.300 |
| ²⁰⁸ Pb/ ²⁰⁴ Pb | 38.316 | 38.215 | 38.104 | 37.600 | 38.955 | 37.600 |

 Table 2.
 Primitive Basalt Compositions of Izu-Oshima, Toshima, and Niijima and Subducted Sediment, AOC Compositions, and Mantle Peridotite Compositions Used for ABS3 Calculations^a

^aUnit is ppm.

Geochemistry

Geophysics Geosystems

monitoring Ni and Mg # in olivine in the basalt until this approach yielded a melt in equilibrium with mantle olivine [Takahashi, 1986]. The same approach is applied to the most MgO-rich basalts (MgO > 6 wt %) from this study in order to minimize the effect of fractionation of plagioclase and clinopyroxenes, which are not directly accounted for by this correction. The estimated primitive basalt compositions are shown in Table 2. We also compared the calculation using the scheme of Lee et al. [2009]. Our estimates on the primary magma compositions are almost identical with the estimates by the Lee et al. [2009] olivine maximum fractionation model with H₂O content in the primary magmas that range between 3 and 6 wt % and with mantle olivine having Mg # = 91.

5.2.2. Source Material Compositions

[34] Subducting SED and AOC and wedge mantle compositions are required inputs for the ABS3

forward model. SED and AOC compositions for N-Izu subduction zone are available from ODP Site 1149 drill cores [*Hauff et al.*, 2003; *Kelley et al.*, 2003; *Plank et al.*, 2007; *Plank and Langmuir*, 1998] and are given in Table 2.

[35] The AOC/SED trace element and Sr-Nd-Pb isotopic data are set in the ABS3 DATA INPUT worksheet (Software S1). The isotopic composition of the mantle is taken from the Shikoku Basin basalts, whereas trace element composition is based on estimates of DMM (see Table 2). Three possible DMM models [Kimura and Yoshida, 2006; Salters and Stracke, 2004; Workman and Hart, 2005] are provided with ABS3. In this case we use the estimate of Workman and Hart [2005] for N-Izu, because this provided the best fit (see below). The degree of source DMM depletion prior to arc magma genesis can also be examined using a melt depletion parameter %MORBext (Text S1 and Software S1) [see also *Kimura et al.*, 2009], and is fully discussed in section 5.3.4.

Geochemistry Geophysics KIMURA ET AL.: CROSS-CHAIN MAGMA VARIATION IN N-IZU ARC 10.1029/2010GC003050

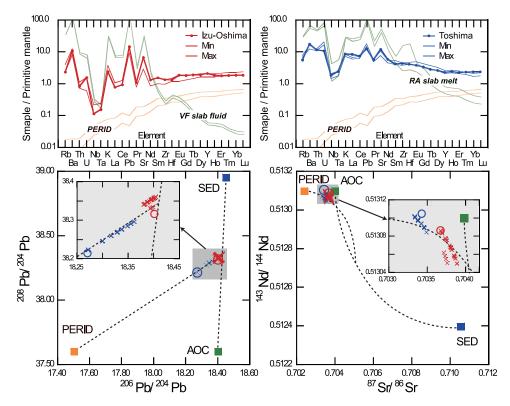


Figure 6. Fitting calculation results from Arc Basalt Simulator version 3 (ABS3) model calculations for Izu-Oshima (VF) and Toshima (RA) cross-chain basalts. (top) Thin lines delimit minimum and maximum values obtained from ABS3 calculations, solid line with circles denotes estimated primary basalt composition, green thin lines denote calculated melt/fluid compositions and their ranges, and thin orange lines denote calculated source PERID compositions and their ranges. (bottom left) Pb isotope calculation results (crosses) with primary basalt and source compositions (open circles). SED, sediment; PERID, peridotite; AOC, altered oceanic crust. (bottom right) Sr-Nd isotope systematics. Symbols are same as in Figure 6 (bottom left). Insets in Figure 6 (bottom) are close-ups of shaded areas.

5.2.3. Slab P-T Trajectory

[36] In addition, a slab P-T path necessary for the slab flux calculations was derived from a geodynamic model calculated specifically for N-Izu [*Syracuse et al.*, 2010]. ABS3 contains estimates of the slab P-T path for a number of subduction zones worldwide (Software S1).

5.2.4. Preliminary Mass Balance Fitting and Automated Fine Adjustment

[37] Once the various source compositions and slab P-T path were set, practical ABS3 analysis begins with manual determination of slab SED-AOC contributions to the slab flux (either fluid or melt) by monitoring mass balance calculation based on both trace element abundances and isotopic compositions (²⁰⁸Pb/²⁰⁴Pb- ²⁰⁶Pb/²⁰⁴Pb and ⁸⁷Sr/⁸⁶Sr-¹⁴³Nd/¹⁴⁴Nd systematics). Pb isotopes typically provide the most powerful tool for this process (see Figure 6). Slab SED liquid fraction in the slab flux (Fliq(SED) see Text S1 and Software S1 for a list of variables) can be determined by altering Fliq(SED) together with a variable corresponding to slab depth (Slab P(GPa)). The slab depth parameter determines at which depth the slab liquid is derived. Slab liquid compositions vary independently for slab SED and AOC along the slab P-T trajectory in ABS3. The amount of SED/AOC flux in the total slab flux is not constrained by the ABS3 model, so that slab depth and SED flux fraction determines where the slab flux is from and how much slab SED/AOC contributes to the flux.

[38] Mantle melting parameters are melting pressure (P(GPa)), melting temperature (T(C)), and slab liquid fraction added to the mantle (Fslb liq.%), all of which are required parameters for melting models [*Ozawa and Shimizu*, 1995], with petrological parameterization by *Katz et al.* [2003] and pMELTS melting algorithms [*Ghiorso et al.*, 2002] used for fluid/melt fluxed melting. All of above parameters (Fliq(SED) and Slab P(GPs) for slab flux parameters and P (GPa), T (C), and Fslb liq/% for mantle melting parameters) are adjusted iteratively.

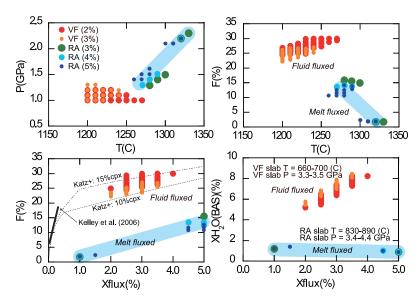


Figure 7. Intensive and extensive variables that reproduce the trace element and isotope compositions of the N-Izu cross-chain basalts. Note that percent values following VF and RA are mantle source depletions relative to the DMM of *Workman and Hart* [2005]. Thick line on the F(%)-Xflux(%) diagram shows the effect of H₂O on degree of melting from the model of *Kelley et al.* [2006]. Thin dash-dotted and dotted lines are the calculated results using the *Katz et al.* [2003] parameterization used in ABS3 for a source mantle with modal cpx 15% (dash-dotted line) and 10% (dotted line) and with the assumption that Xflux(%) = XH₂O(%) = H₂O_o (wt %) following the definition by *Kelley et al.* [2006]. The two models show a closer match at low H₂O fractions up to 0.3 wt % in the source. However, the effect cannot be extrapolated to higher H₂O fraction. Melt flux model does not follow the model because H₂O fractions in the calculated melts are considerably lower (XH₂O(%) \ll Xflux(%)) than those in the aqueous fluids.

[39] When a reasonable fit is achieved in terms of both the isotopic and trace element compositions of primary basalts (see Figure 6), this first-order adjustment allows approximation of the entire mass balance. Sr and Nd isotope ratios, in many cases, were in good agreement between model and natural basalt when reasonable fits for Pb isotopes and trace element abundances were achieved (see Figure 6). In the case of N-Izu, the average isotopic composition of ODP1149 SED source [Plank et al., 2007] predicted Sr and ²⁰⁶Pb/²⁰⁴Pb isotopic compositions in the model basalts that are systematically high, and Nd isotope ratios low. However, those compositions can be adjusted to provide adequate fits by using SED isotopic compositions that are within the range observed in ODP1149 SED [Plank et al., 2007; Plank and Langmuir, 1998].

Geochemistry

Geophysics Geosystems

[40] Based on this fit, the plausible ranges of the remaining parameters; slab dehydration/melting depth, open system melting pressure and temperature, and fraction of melt/fluid fluxing the mantle source ((Fliq(SED), Slab P(GPs), P (GPa), T (C), and Fslb liq/%) were explored by automated fitting using the macro function of the ABS3 Excel spread-sheet (Text S1 and Software S1). The source PERID depletion factor (%MORBext.) was also varied to test for possible source mantle heterogeneity (see details of these calculations by *Kimura et al.* [2009]).

5.2.5. Unused Parameters

[41] Variations in the liquid fraction from the hydrated mantle wedge (Flig(DMM), see Text S1 and Software S1) and the zone refining reactions between slab fluxes and mantle (n(PERID)) are modeled by ABS3 but did not prove effective in reproducing trace element abundances in the N-Izu basalts. A similar conclusion from ABS2, which explored NE Japan arc volcanic rocks was reported by Kimura et al. [2009]. Therefore, these parameters were not used in this study of N-Izu. Because of this and similar results in previous studies [e.g., Ayers, 1998], neither ABS2 nor ABS3 uses these parameters for automated fit calculations. However, users may manually alter parameters to improve fits for some specific cases, particularly where there are observed to be extremely high LILEs relative to other incompatible elements.

5.2.6. Role of Slab Melting

[42] The best fit results are shown graphically in Figure 6 and intensive parameters are shown in Figure 7. Overall, we found that slab fluids derived



from dehydration are necessary for the VF basalt formation, whereas slab melts are required to produce RA basalts. To best reproduce the trace element abundances in RA basalts the slab temperature beneath the RA needed to be increased about 20% over that predicted by geodynamic estimates [Syracuse et al., 2010]. Required slab P-T conditions are 3.3-3.5 GPa/660°C-700°C for VF and 3.4-4.4 GPa/830°C-890°C for RA (see Figure 7, top left). A similar effect (high-temperature slab) was found with ABS2 for NE Japan RA basalts [Kimura et al., 2009]. ABS2 did not deal with slab melting, therefore, the high-temperature flux for the NE Japan RA was considered to be a supercritical liquid that can carry more incompatible trace elements than a lower-temperature fluid [Kimura et al., 2009]. However, the use of supercritical liquid also produces unrealistically high estimates for H₂O, with the mass balance for NE Japan arc suggested up to 12 wt % H₂O in the RA basalts. This value appears too high for arc basalts compared to observed values [Kelley et al., 2006; Wallace, 2005]. The new ABS3 model deals with slab melting with the new phase diagrams (Text S1) and the H₂O content of N-Izu RA basalts up to $XH_2O(BAS) =$ 1 wt % (see Figure 7) can reasonably be modeled with the adjusted P-T path.

Geochemistry

Geophysics Geosystems

[43] Although our geochemical mass balance predicts slab melting, we note that geodynamic models do not predict slab melting in old-cold subduction zones like N-Izu [Syracuse et al., 2010]. Recent water saturated experimental data for slab SED and AOC, [e.g., Schmidt et al., 2004; Spandler et al., 2004], also indicate no slab melting at pressure range 3-6 GPa under slab temperature less than 800°C. The solidus temperatures of AOC and SED in the experiments (see Figure 2 in Text S1) are few tens to a hundred degrees higher than the N-Izu geotherm. However, considering that there are uncertainties associated with both geodynamic model and the experiments, we believe that slab melting remains a plausible solution. We examine this issue further below.

5.3. Intensive and Extensive Parameters of Arc Basalt Genesis: A Forward Model Perspective

[44] The intensive and extensive parameters estimated from the ABS3 model provide insights into the origins of the observed across-arc geochemical variations, and several important conclusions have emerged for N-Izu. We discuss these and some of the advantages and limitations of various modeling approaches in more detail below.

5.3.1. Estimation of Degree of Partial Melting in Flux Melting

[45] Inverse calculation models often use relatively fluid immobile elements, such as HREE and HFSE, for estimation of degree of partial melting of the mantle source [*Kelley et al.*, 2006]. However, recent experiments suggest that HREE and HFSE are mobile even in supercritical liquid [*Kessel et al.*, 2005] similar to slab melts [*Rapp et al.*, 1999]. If this is the case then the problem of HREE/HFSE in the slab flux should be solved simultaneously with the mantle melting calculations. Our forward model does this using an open system melting calculation [*Zou*, 1998], maintaining the mass balance in the fluxed melting region.

5.3.2. Estimation of Slab Fluid/Melt Fraction

[46] Although trace element mass balance is maintained in the melting calculations, the effect of H₂O in the mantle melting also needs to be considered. The overall role of H₂O is constrained by independent data sets, such as H₂O in olivine melt inclusions and estimated degrees of mantle melting determined by trace element inversion calculations [Kelley et al., 2006; Stolper and Newman, 1994]. A comparison between such estimates and parameterization based on experimental data is given in Figure 7 (bottom left). In the low H₂O flux range (<~0.5 wt % fraction in the source), the empirical equation based on trace element inversion [Kelley et al., 2006] fits well to the experimental data [Katz et al., 2003]. However, a significant discrepancy occurs at higher degrees of fluxing. Such high degrees of slab flux (0.5 to >5 wt %) to the subarc mantle have been estimated based on trace element and isotope data [Ishizuka et al., 2006; Kimura et al., 2009; Tollstrup et al., 2010].

[47] We have dealt with the effect of H_2O in the melt/fluid-fluxed melting of the mantle peridotite [*Katz et al.*, 2003; *Kimura et al.*, 2009] in concert with the open system melting model for trace elements. ABS3 provides the petrological basis to solve the complex relationships involving the effects of variations in the degree of melting and amount of flux based on P(GPa), T(C), and Xflux(%). The weakness of this forward model is the requirement for explicitly defined slab flux and mantle compositions; however, we constrain these through model mantle and slab dehydration/melting calculations.

5.3.3. Estimation of Slab Flux Composition

Geochemistry

Geophysics Geosystems

[48] In previous studies the slab flux composition, either fluid or melt, has been examined indirectly based on the highly incompatible element pairs, such as Cs/Rb, and these geochemical characteristics have been ascribed to the role of certain slab minerals [*Tamura et al.*, 2007]. Inverse calculations have also been used to identify the chemical compositions of the slab flux [*Tollstrup et al.*, 2010].

[49] Our model uses petrological forward calculations for the given slab SED, AOC, PERID composition along the slab geotherm, and for the released fluids and melts from the breakdown of slab components. Progressive depletions in the trace elements and H₂O from the slab are calculated as a prograde metamorphic process. The mineralogy and chemistry of the downgoing slab and thus the chemistry of the released slab fluxes at designated depths are predictable from phase equilibria and trace element partitioning models (see Figures 2-6 in Text S1). In addition the slab P-T path is now constrained for 56 arcs worldwide [Syracuse et al., 2010]. Identification of the depth where the slab flux was produced can be determined from the slab flux composition required for the mantle melting mass balance.

5.3.4. Estimation of Source Mantle Depletion

[50] The degree of depletion of the source mantle can be examined simultaneously with the fluxed mantle melting model, and the effects of source depletion and degree of melting can be separated by inverse calculations using HREE and HFSE element abundances as shown by *Kelley et al.* [2006]. However, as stated above, the effect of slab H₂O on trace element behavior should be considered with such an approach. As we explicitly determine the slab flux composition, the degree of source mantle depletion from DMM can be directly calculated and the accuracy can be tested by forward fitting calculations.

5.3.5. Estimation of Mantle Potential Temperature

[51] Mantle potential temperature (Tp) can be estimated petrologically by a forward model using major element compositions in accumulated fractional melt for primary basalts (e.g., *Kelley et al.* [2006, equations 12 and 13] using Na(Fo90) and Fe(Fo90)). A previous study by *Tollstrup et al.* [2010] applied this method to southern Izu and estimated Tp = 1409° C- 1458° C. These are considerably higher than our estimates of Tp = 1189° C- 1267° C for VF and Tp = 1249° C- 1317° C for RA (Tp calculated from T(C) and P(GPa) in Figure 7). Large discrepancies are seen between the near dry Na(F090)-Fe(F090) relation and estimated Na (F090)-Fe(F090) relation from the southern Izu [see *Tollstrup et al.*, 2010, Figure 15], suggesting overestimate of Tp due to use of a dry melting model. For ABS3 we include explicit P-T-XH₂O relations along with the trace element mass balance. Therefore, our estimate is based on wet mantle melting, and results in a lower estimated Tp, which shows good agreement to wet melting Tp = 1155° C- 1320° C calculated following *Lee et al.* [2009].

[52] In the ABS3 model, the mass balance is examined by controlling limited numbers of intensive/extensive parameters. As a result ABS3 is an extremely rigid forward model with a robust petrological background.

5.4. Origin of Cross-Chain Geochemical Variation in N-Izu

[53] On the bases of the ABS3 mass balance model results, we explore in detail the origin of the cross-chain variation found in N-Izu arc.

5.4.1. Cross-Chain Variations in Intensive/ Extensive Petrogenetic Parameters

[54] Our model predicts that slab flux compositions for the VF and RA basalts differ significantly. For the VF, slab-derived fluids are produced at relatively low pressures (3.3-3.5 GPa), whereas the RA melt sources are deeper (3.4-4.4 GPa, see Figure 7). Some of the major compositional differences between VF and RA slab-derived fluxes relate to the stability of lawsonite in the slab. Lawsonite breaks down at a ~ 100 km depth (3–3.5 GPa) with the slab P-T path for N-Izu (see Figure 2 in Text S1). Lawsonite breakdown releases large amount of Pb, Sr, and Ba, contributing to the enrichment of these elements in the VF basalts. The shallow slab flux proposed by Pearce et al. [2005] for Mariana arc is primarily the signature of lawsonite breakdown predicted by ABS3.

[55] Deeper slab fluxes tend to have greater incompatible element abundances due to increases in partition coefficients of fluids or melt at higher temperatures [*Kessel et al.*, 2005; *Kimura et al.*, 2009]. As noted above the geodynamic estimates of slab temperature are not sufficiently high to reproduce slab flux compositions and needs to be



increased by ~20% (about 150°C) greater than predicted by the geodynamic model. With these higher temperatures, the ABS3 model predicts slab melts for RA basalt. This conclusion is consistent with the previous proposals based on incompatible element and isotope data [Hochstaedter et al., 2001; Ishizuka et al., 2003]. Moreover, melting of the slab decomposes mica minerals (phengite) in the slab (see Figure 2 in Text S1), resulting in release of K, Rb, and Cs. Combined with residual garnet and clinopyroxene in the slab, the geochemical characteristics of RA flux is derived from this breakdown. Additional fluid release also results from these dehydration melting reactions, but the amount is much less than released beneath the VF (<1.5 wt % in slab melt, see Figure 7). Our ABS3 model explicitly predicts the role of phengite breakdown in the slab, which was caused by slab melting.

Geochemistry

Geophysics Geosystems

[56] The degree of partial melting of the mantle PERID is predicted to be greater beneath VF (F =21%-30%) and smaller beneath the RA (F = 9–16) or 1%–1.5% in slab melt fluxed mantle melting, Figure 7), and the temperature of melting is higher in the mantle wedge beneath the RA (1260°C-1340°C) and lower in VF (1200°C-1270°C). Melting pressure should also be lower beneath the VF (1.0-1.3 GPa) compared to the RA (1.3-1.5 or)2.0–2.3 GPa, Figure 7). The lower melting degree with greater pressure allows residual garnet in the RA PERID source. Therefore, elevated LREE/ HFSEs observed in the RA lavas are primarily the result of combined lower melting degree and residual garnet in the source (see Software S1). The melting conditions for the RA lavas are apparently in two distinct regions (Figure 7); however, these are due to a strong statistical filter used in the fitting calculations. The fitting region lies between the two regions (shown by light blue thick line) indicating that there are less constraints on the melting conditions under the RA. Even so, the conditions for RA melting are clearly different from those for the VF lavas. The estimates on degree of mantle melting based on Kelley et al.'s [2006] inverse calculation method yield degree of melting for VF at $31 \pm 6\%$ and for RA $17 \pm 3\%$, and these are comparable to our maximum estimates. The Kelley et al. [2006] estimates are probably somewhat higher than ours as they do not account for mobility of HREE and HFSE in the slab fluxes added to the melting regions. Mantle melting temperatures (1200°C-1340°C) estimated by the ABS3 model are petrologically consistent with the pressure estimates and flux fractions (see section 5.3.3.). Therefore, the extremely high mantle temperatures suggested by *Tollstrup et al.* [2010] are not required to satisfy the trace element mass balance and the melting relationship, and the idea that unusually high mantle temperature is the source of deep slab melting [*Tollstrup et al.*, 2010] is not supported. High slab temperatures, as deduced by temperature-dependent olivine rheology [*Syracuse et al.*, 2010] and by our geochemical constraints are the alternative solution.

[57] The flux fraction is variable beneath the RA (1-1.5 to 4.5-5 wt % melt) and typically larger beneath the VF (2–4 wt % fluid, Figure 7). PERID melting degree is broadly proportional to the flux fraction, although the effect of fluxing is different between an aqueous fluid (VF) and melt (RA). This is due to different H₂O contents between fluid and melt. The greater H₂O addition beneath the VF is the prime control of mantle melting degree, and is responsible for the high VF magma production. This supports a previous inference derived from ABS1 and ABS2 calculations [Kimura and Stern, 2008; Kimura et al., 2009]. It is also consistent with the fact that fluid release by slab dehydration is at a maximum at ~3 GPa by breakdown of amphibole and lawsonite to form phengite quartz eclogite (amphibolite - eclogite transformation) (see Figure 2 in Text S1). The ABS3 model provides the petrological basis to examine the connection between focused high slab defluxing and high magma production rate. This has been predicted based on experimental data [Poli and Schmidt, 1995].

[58] An additional feature is that a more depleted mantle PERID than those of MORB source is required for both the VF and RA sources. This interpretation appears to be inconsistent with the fact that the Shikoku Back Arc Basin basalts are derived from a MORB source mantle [Hickey-Vargas, 1991, 1998; Hickey-Vargas et al., 1995; Machida et al., 2008]. However, extraction of MORB-type basalts to form BAB in the Miocene should have depleted the RA mantle source, and this may be the source that later melts to produce Quaternary RA and VF basalts. Progressive younging of volcanic activity from RA to VF after the Miocene [Ishizuka et al., 2006] may suggest inflow of the depleted deep and hot mantle beneath the N-Izu arc. Depletion of the PERID by the extraction of the RA melts is another option [Hochstaedter et al., 2001], but this does not account for mantle depletion in the RA source found in this study and Tollstrup et al. [2010]. Our model does not quantitatively evaluate *Hochstaedter et al.*'s [2001] semidynamic model and such a process makes interpreting cross-chain geochemical variations more complex [*Kimura and Stern*, 2008]. This may also lead to difficulties in interpreting the cause of the cross-chain K-h relationship [*Dickinson*, 1975; *Dickinson and Hatherton*, 1967; *Hatherton and Dickinson*, 1969]. A dynamic model to study this further is beyond the scope of this paper; however, ABS3 would potentially be useful for future studies aimed at constraining possible sources of mantle heterogeneity. The existence of variably depleted mantle sources has also been reported in the central to southern N-Izu [*Machida et al.*, 2008].

Geochemistry

Geophysics Geosystems

[59] Further studies of fluid immobile isotopes, such as Nd and Hf, will be useful to identify the relationship between source depletion and role of slab melt flux [Ishizuka et al., 2006]. In fact, Tollstrup et al. [2010] examined the HFSE behavior along with Hf isotopes and concluded that slab meltderived HFSE contributed to the greater Nd and Hf variation found in RA basalts. With the present ABS3 model, evaluation of HFSE behavior is a problem because HFSE behavior in the slab flux is largely controlled by minor Ti-bearing phases or zircon, and the stability relations of these phases is poorly constrained by Perple X phase model that we use. Consumption of these phases during melting may account for Hf/Hf* variations (Hf depletion or enrichment factor relative to Nd and Sm in a normalized plot) [Tollstrup et al., 2010; Tollstrup and Gill, 2005]. This question is left open for further development of the ABS model that includes the effect of those minor mineral phases once improved experimental constraints are available.

[60] Finally, the estimated H₂O contents in primary magmas (fraction of fluid flux in terms of open system melting [Ozawa and Shimizu, 1995]) are greater in the VF (5-8 wt %) and less in the RA (1-1.5 wt % irrespective of large variation in the)melt flux due to the counterbalancing effect of melting degree, Figure 7). Although H_2O contents of N-Izu olivine melt inclusions are not yet known, there is evidence that the VF basalts are more fluidrich than in the RA basalts. This is also consistent with the general water contents measured in VF and RA settings worldwide [Wallace, 2005; Kelley et al., 2006]. Water in the N-Izu VF basalt magmas has been estimated to be 3-6 wt % based on melt plagioclase element partitioning [Hamada and Fujii, 2007]. The minimum H_2O contents in the primary melt estimated by ABS3 was 5 wt %, and thus is broadly consistent with the petrological estimates. Estimates of primary magmatic H₂O contents based in ABS3 are somewhat higher in the VF basalts than measured water contents in Izu tephras and the Mariana fore-arc basalts [*Elliott et al.*, 1997; *Straub and Layne*, 2003], although these could have been affected by degassing. The general trend, of increasing H₂O from VF to BAB, is consistent with the previous studies [*Kelley et al.*, 2006]. Water contents in RA basalts from the southern Izu have water contents from 0.5 to 2.5 wt % (A. J. R. Kent, unpublished data, 2009), which is also consistent with our general predictions. Constrain on magmatic water contents in N-Izu basalts from olivine-hosted melt inclusions or other sources would provide an important further test of our model.

5.4.2. P-T Structure of the Slab and the Mantle Wedge

[61] We can also use our ABS3 results to reconstruct the P-T structure in the slab and the mantle wedge (Figure 8). From the compositions of the VF and RA basalts, we can derive several control points in the slab dehydration and mantle melting regions. We suggest that the slab-mantle P-T conditions from bottom to top are as follows: Slab dehydration is estimated to occur at 3.0–3.3 GPa/660°C– 700°C beneath the VF and slab melting occurs 3.6–4.4 GPa/830°C–890°C beneath the RA. Mantle melting conditions of the VF basalt range from 1.3 to 1.0 GPa/1200°C–1270°C and 2.3–2.0 GPa/ 1340°C–1300°C for RA basalts.

[62] As discussed above, the RA magmas sample only the shallowest portion of the mantle that melts. Deeper RA mantle should also melt because the wedge core temperature may exceed 1400°C [Syracuse et al., 2010]. The Shikoku back arc basin basalts are key to further constraining the rear-arc geotherm, The Shikoku Basin was actively spreading until 15 Ma and the youngest ocean basement immediately behind the N-Izu rear arc is about 28 Ma, forming when the BAB first opened [Okino et al., 1994]. Therefore, the 1000°C isotherm depth should follow the oceanic lithosphere cooling rate to lie ~30 km deep beneath N-Izu rear arc (Figure 8). It is possible with this situation that the 1300°C-1400°C geotherm is located ~100 km deep beneath the rear arc (see Figure 8). In this case, the melting column may extend down to 150-170 km deep (4–5 GPa) in the RA mantle (see melting columns in red in Figure 8) according to the estimate using P-T-XH₂O parameterization of Katz et al. [2003].

[63] Based on the P-T control points deduced from the ABS3 model and other constraints (see Figure 8),

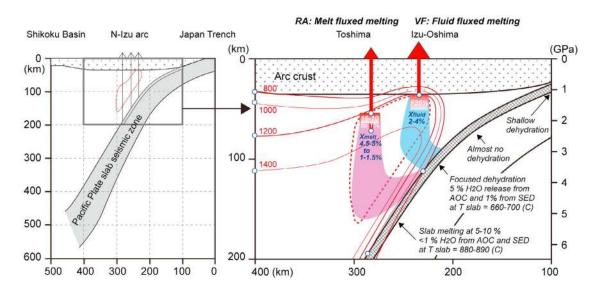


Figure 8. Schematic cross section of N-Izu volcanic arc. (left) Distribution of seismic zone beneath Izu-Oshima to Niijima cross chain [*Stern et al.*, 2003]. Izu-Oshima volcano is located above the 120 km depth contour of the Wadati-Beiioff zone, whereas Toshima and Niijima are situated on about 180 km depth contour. (right) Distribution of slab fluid path (shaded cold color regions) and mantle melting region (red gradational color regions) with estimated isotherm in the mantle wedge. The isotherms are constrained by the rear-arc geotherm (intermediate between spreading ridge and average ocean geotherm because the Shikoku Basin has been extinct since 15 Ma); temperature of magma segregation depths; and slab surface temperatures estimated for Izu-Oshima, Toshima, and Niijima basalt sources. P-T control points are given by open circles. Slab dehydration focuses on two regions, one beneath the fore arc (~1 GPa) and the other at about 120 km depth (~3 GPa) with the given slab surface temperature (SST) trajectory (for SST see details in section 5.4.2). Characteristics of this model are (1) fluids liberated from the slab rise up to the mantle melting region and (2) mantle melting takes place almost parallel to the slab surface, controlled by 1200°C and 1400°C isotherms. The deeper limit of the melting region is controlled by P-T-XH₂O relation [*Katz et al.*, 2003] at labout 2.7 GPa/~1400°C for VF and 3.5 GPa/~1500°C for RA. The parallel melting zone developed between 1 and 3.5 GPa in the mantle wedge is detected seismologically in NE Japan arc [*Hasegawa et al.*, 1991; *Nakajima and* Hasegawa, 2003; *Nakajima et al.*, 2005]. Such structure is thus common beneath basaltic arcs.

there are significant variation in magma production conditions beneath the magmatic arc. First, the slab dehydration is focused at ~3 GPa and this results in greater production of basalt magmas beneath the VF. The melting column would lie \sim 35–80 km (2.5– 1.0 GPa) deep with a temperature range of 1200°C-1400°C. Second, the rear-arc slab melting occurs at 140 km (3.5 GPa) and 890°C. The melting column may range from >170 km deep (5GPa) and 1400°C to ~80 km deep (2.5 GPa) and 1200°C. The mantle melting volume is continuous and forms a region in the mantle wedge which is subparallel to the slab surface (see dotted oval areas in Figure 8). A similar mantle wedge structure has been inferred in the NE Japan arc, based on seismic tomography [Hasegawa et al., 1991; Nakajima et al., 2005] that shows a slab parallel low-Vp and low-Vs zone is found in the middle of the mantle wedge. The shallowest part of the melting region is about 1 GPa and can involve the deepest arc crust [Hasegawa et al., 1991;

Geochemistry

Geophysics Geosystems

Kimura et al., 2002; *Nakajima et al.*, 2005; *Tatsumi et al.*, 2008], similar to estimates from our ABS3 modeling. Although the geophysical methods estimate mantle wedge temperatures [*Nakajima and Hasegawa*, 2003] that are ~200°C lower than the ABS3 estimates, the overall P-T structure in the wedge mantle is similar between the two approaches.

[64] It is also worth noting that if XH_2O is zero in such a mantle wedge P-T structure, melt can only be generated beneath the VF at a depth of 50–65 km (1.5–2.5 GPa) with <3% partial melting, and no melt is generated in the RA environment according to the *Katz et al.* [2003] model. This emphasizes the critical role of slab-derived H₂O in generating a large (10%–25%) amount of melting. The role of water is critical even for trace amounts, such as if 0.1% water is added, the mantle can generate >3% more melt due to fluid fluxed melting. The presence or absence of H₂O is thus also critical in determining the width of the volcanic arc and its magma production.



[65] The P-T structure constrained by the ABS3 model is broadly consistent with that inferred from seismological constraints. Improving the model requires further development and integration of the phase petrology, element partitioning, and dynamics of the fluid and the solid flows in the slab and the mantle wedge.

6. Conclusions

Geochemistry

Geophysics Geosystems

[66] The intensive and extensive variables in the slab and in the mantle that determine the geochemical variations observed in N-Izu cross-chain volcanism are summarized as follows.

6.1. Cross-Chain Variations in Slab Fluids

[67] The slab mineralogy changes along the slab P-T path. Shallow slab fluid release is focused at ~ 3 GPa, corresponding to the breakdown of amphibole and lawsonite. This shallow fluid dehydration possesses elevated LILEs such as Ba, K, Pb, and Sr primarily due to the lawsonite breakdown. In contrast, deeper slab melt is controlled by phengite-quartz-eclogite mineralogy and has more incompatible elements entering into the melt due to the dependence on partition coefficients. The resultant deep melts are LILE- and LREE-rich and relatively elevated in HFSE and HREE. The shallow and deep fluids contribute to mantle melting and basalt genesis in the VF and RA, respectively.

6.2. Cross-Chain Variations in Fluid Flux and Mantle Melting Degree and Depth

[68] About twice as much flux is added to the mantle beneath the VF than beneath the RA, and the greater addition of fluid, thus H₂O, in addition to lower pressures promotes more extensive mantle melting. The RA mantle experiences lower melt flux, consistent with melting of dryer phengite quartz eclogite in the deeper subducted slab. Diminished fluid flux along with high melting pressure suppresses the degree of melting of the RA mantle, resulting in elevated incompatible element abundances in RA basalts. Such a wedge mantle structure is consistent with seismic observations in other arcs, such as NE Japan [Hasegawa et al., 1991; Nakajima et al., 2005]. These melting depths and degrees were obtained independently from petrologic constraints by the ABS3 model using incompatible trace elements, but are quite consistent with crosschain models deduced from major element petrology [Kimura and Yoshida, 2006; Tatsumi et al., 1983].

6.3. Cross-Chain Variations in Radiogenic Isotopes

[69] Radiogenic Pb and Sr are dominant in the VF mantle source due to high fluid fraction and the greater abundances of these elements in the slab fluid. In contrast, radiogenic Nd and Hf in VF basalts reflect smaller contributions of these fluid immobile elements to the mantle source due to very low abundances of Nd and Hf in the same fluid. Less radiogenic Sr and Pb in RA basalts reflect the greater contribution of mantle sources due to lower degrees of partial melting and relatively high addition of deep melts into the basalt melts.

6.4. Fluid Flux, Mantle Melting, and Volatiles in Cross-Chain Basalts

[70] The fluid flux fraction in the mantle appears to be greater in the VF. The degree of mantle melting is roughly controlled by the flux. The H₂O contents estimated for primary basalt are consistent with geohygrometry (4–6 wt % in VF) [*Hamada and Fujii*, 2007].

[71] We have shown that the origins of cross-chain geochemical variation are largely controlled by slab and mantle wedge conditions, which in turn are petrologically controlled by the P-T structure in the slab and mantle wedge.

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