

Ortho-H₂ and the age of prestellar cores

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Received 29/04/2011; accepted ...

ABSTRACT

Prestellar cores form from the contraction of cold gas and dust material in dark clouds before they collapse to form protostars. Several concurrent theories exist to describe this contraction which are currently difficult to discriminate. One major difference is the time scale involved to form the prestellar cores: some theories advocate nearly free-fall speed via e.g. rapid turbulence decay while others can accommodate much longer periods to let the gas accumulate via e.g. ambipolar diffusion. To discriminate between these theories, measuring the age of prestellar cores could greatly help. However, no reliable clock currently exists. We present a simple chemical clock based on the regulation of the deuteration by the abundance of ortho-H₂ that slowly decays away from the ortho-para statistical ratio of 3 down to or less than 0.001. We use a chemical network fully coupled to a hydrodynamical model which follows the contraction of a cloud, starting from uniform density, and reaches a density profile typical of a prestellar core. We compute the N₂D⁺/N₂H⁺ ratio along the density profile. The disappearance of ortho-H₂ is tied to the duration of the contraction and the N₂D⁺/N₂H⁺ ratio increases in the wake of the ortho-H₂ abundance decrease. By adjusting the time of contraction, we obtain different deuteration profiles that we can compare to the observations. Our model can test fast contractions (from 10⁴ to 10⁶ cm⁻³ in ~0.5 My) and slow contraction (from 10⁴ to 10⁶ cm⁻³ in ~5 My). We have tested the sensitivity of the models to various initial conditions. The slow contraction deuteration profile is approximately insensitive to these variations while the fast contraction deuteration profile shows significant variations. We found that in all cases, the deuteration profile remains clearly distinguishable whether it comes from the fast collapse or the slow collapse. We also study the para-D₂H⁺/ortho-H₂D⁺ ratio and find that its variation is not monotonic and therefore not discriminant between models. Applying this model to L183 (= L134N), we find that **the N₂D⁺/N₂H⁺ ratio would be larger than unity for evolutionary timescales of a few megayears independently of other parameters such as, e.g., cosmic ray ionization rate or grain size (within reasonable ranges). A good fit to the observations is obtained for fast contraction only (≤ 0.7 My from the beginning of the contraction and ≤ 4 My from the birth of the molecular cloud based on the necessity to keep a high ortho-H₂ abundance when the contraction starts – ortho-H₂/para-H₂ ≥ 0.2 – to match the observations). This chemical clock therefore rules out slow contraction in L183 and steady state chemical models, as steady state is clearly not reached here. This clock should be applied to other cores to help discriminate between slow and fast contraction theories over a large sample of cases.**

Key words. ISM: abundances — ISM: clouds — evolution — ISM: molecules — astrochemistry — ISM: individual objects : L183

1. Introduction

Star formation is one of the fundamental problems astrophysics has been tackling since it exists. It has subdivided into 2 or even 3 branches by recognizing that low mass, intermediate mass and high mass star formations take different paths. Among the three, the low mass star formation process is the best understood, because there are many more low mass stars than high mass stars, therefore there are a number of low mass star forming regions close to us which can be studied in details. However the details of the first steps of this low mass star formation remain uncertain. What causes the contraction of the prestellar core and its subsequent collapse? Is contraction due to supersonic turbulence fast decay or is it due to slow ambipolar diffusion in a magnetically subcritical cloud? Which initial or evolving conditions set up the final mass of a star? How do multiple systems form? All these questions have been asked long ago and are still present. It is obviously difficult to measure constraining

parameters such as the magnetic field intensity, the cosmic ray ionization rate and its partly related dark cloud electronic abundance which would help to assess the importance of ambipolar diffusion (see however Crutcher et al. 2010). Because the different prestellar core making theories imply different time scales, by a factor up to 10, measuring the age of the cores could help to discriminate between slow and fast theories. Until now, such clocks have not been found. In principle, chemical clocks could be used because many species appear with different time scales: SO and NH₃ take at least 1 My to appear while CS peak abundance is reached 10 times faster (Roberts et al. 2004; Flower et al. 2006a), and the SO/CS ratio could be considered as such a clock. Nilsson et al. (2000) have shown however that such a ratio was depending on multiple factors, time evolution being only one of them, and other conditions such as the initial O/C⁺ ratio would strongly influence the result making the SO/CS ratio an unreliable clock if the other parameters cannot be estimated

with some accuracy. The existence of chemical clocks is not yet established (van Weeren et al. 2009).

Recently, we suggested two new clocks to measure the age of prestellar cores (Pagani et al. 2009, 2010b). One is linked to the direct detection and measurement of grown grains in dark clouds (the coreshine effect, Steinacker et al. 2010), which we have shown to be feasible in many clouds (Pagani et al. 2010b). These big grains have built up from interstellar standard grains and by applying a growth model such as those of Ormel et al. (2009), it should be possible to determine the age of the cloud. As a first example, we have made a simple application of the coreshine effect to study the age of clouds in the Gum/Vela region (Pagani et al. 2012). The other one is linked to chemistry but in a particular case where the chemistry is simple and the initial conditions known or of limited influence. The chemistry is the one which appears in depleted cores where all heavy species have disappeared, including CO (e.g. Caselli et al. 1999; Bergin et al. 2002; Bacmann et al. 2002; Tafalla et al. 2004; Pagani et al. 2005; Ford & Shirley 2011). These depleted cores are well-known for displaying large deuterium enhancements as the deuterium chemistry is greatly favored once CO has disappeared, as illustrated by the detection of H₂D⁺ itself (e.g. Caselli et al. 2003; Vastel et al. 2006; Caselli et al. 2008; Pagani et al. 2009, hereafter P09), D₂H⁺ (Vastel et al. 2004; Parise et al. 2011) or improbable triply deuterated species like ND₃ (Lis et al. 2002; Tak et al. 2002) and CD₃OH (Parise et al. 2004). This chemistry not only depends on the shrinkage of CO by probably 2 to 3 orders of magnitude but also on the disappearance of ortho-H₂. The importance of the ortho/para spin states of H₂ in the deuterium chemistry was first invoked by Pineau des Forêts et al. (1991) but considering only para-H₂D⁺. The role of o-H₂ on the ortho/para H₂D⁺ ratio was discussed in detail for the first time by Pagani et al. (1992) and was subsequently expanded to include D₂H⁺ and D₃⁺ by Flower et al. (2004, 2006b). In P09, we updated Flower et al. (2004) chemical model with the complete set of rates involving ortho and para states of H₂, H₃⁺ and their isotopologues, as estimated by Hugo et al. (2009). Sipilä et al. (2010) and Parise et al. (2011) have since developed similar models but the former focused their studies on the steady-state case while the latter concluded that the steady-state case gave a better fit than earlier times. In P09, we assumed full depletion of heavy species except N₂ and CO whose abundance was a free parameter of the model. We applied it to reproduce the abundance profile of ortho-H₂D⁺ and the N₂D⁺/N₂H⁺ ratio profile across the L183 (also known as L134N) main prestellar core that had been derived from the observations of these species (Pagani et al. 2007). We noted that the profile could be fitted but that the needed time to reach the observed ratio for each layer was negatively correlated with its density. As the evolution speed of chemistry depends on the collision rate between species which in turn depends on the density of the medium, the result could be explained by the density gradient along the core. Consequently, it appeared that the time discrepancy would be solved if the core had not always been as observed today but started with a lower and probably relatively uniform density and contracted to its present state. The central part that becomes the densest, would evolve faster and therefore further than the outer layers in the same time period but not as fast as if it had always been at the 10⁶ cm⁻³ level. The conclusion was that the model was giving a lower limit to the age of the contracting core and that this limit was set by the slowest evolving layer, i.e., the most external one, with the lowest density. This lower limit was 2 × 10⁵ years, therefore compatible with even a free-fall time (~0.5 My for a density of 10⁴ cm⁻³). Fast contraction could not be ruled out. Such a lower limit is

not particularly useful however and a model that includes both the dynamical evolution and the chemistry is needed to better determine the age of the prestellar core. We therefore used the chemical model of P09 in the 1 dimensional chemical and hydrodynamical model of Lesaffre et al. (2005) to try to follow the deuteration amplification through the density increase of a model core. We report here the general predictions of this new model and its application to the L183 case. Though no quantitative estimates were given, it must be noted that Caselli (2002) already showed that the N₂D⁺/N₂H⁺ ratio would increase with the age of the prestellar core and its collapse advancement.

In Sect. 2, we discuss the chemical clock scheme. In Sect. 3, we briefly present the model. In Sect. 4, we discuss the role of ortho-H₂, its abundance and evolution in the cloud prior to the prestellar core formation, and in Sect. 5, we show the predictions upon the N₂D⁺/N₂H⁺ ratio profile depending on different initial conditions or rates. In Sect. 6, we analyse the conditions which reproduce best the L183 observations and present our conclusions in Sect. 7. Three appendices are added, the first one (A) shows the impact of the new ortho-to-para H₂ conversion rate (presented in Sect. 4), the second one (B) presents a similar study to Sect. 5 but centred on the para-D₂H⁺/ortho-H₂D⁺ ratio, and the third one (C) compiles some useful chemical reaction rates.

2. The chemical clock

One of the main problems with chemical clocks, is our ignorance of the initial conditions, especially the abundance of the various species in the gas phase. Apart from H₂, HD (but see Bell et al. 2011, for a discussion on its fluctuations due to turbulence) and possibly CO, which seems to reach an abundance close to X[CO] = CO/H₂ ≈ 1–2 × 10⁻⁴ in most clouds if not depleted, all other species abundances are difficult to model without knowing the initial conditions (see Liszt 2009, for a recent discussion upon initial conditions). A second problem is to retrieve the distribution along the line of sight of the observed species to account for density and temperature variations, chemical fluctuations and possibly depletion effects. In this respect, line of sight-averaged depletion measurements of CO are merely a measurement of the size of the undepleted region compared to the size of the depleted core (as soon as CO abundance drops in the core, its contribution to the total CO line intensity becomes unmeasurable directly, as discussed in e.g. Pagani et al. 2005, 2010a) and the correlation between CO depletion and deuterium enrichment obtained with this simple method is not reliable. Here, we reduce the number of free parameters by considering the ratio of two species with common reactants and by considering the variation of this ratio along a prestellar core profile for which a 1D symmetry is a relatively good approximation. These two species, N₂H⁺ and N₂D⁺ (known as diazenylium and deuterated diazenylium), have N₂ as a common parent and their production and destruction follow similar routes. They are also confined to the depleted core of the cloud and therefore the physical conditions are relatively well-known and the chemistry is simple. There is no line of sight interference with other parts of the cloud as these species are readily destroyed by CO when the latter is abundant, and N₂ is also relatively slow to form, thus delaying its apparition in the low density parts of the cloud (Flower et al. 2006a). Therefore these species remain absent from the rest of the cloud (a similar work with DCO⁺ and HCO⁺ would be impossible because of the widespread presence of HCO⁺. See also Caselli 2002, for another discussion on the comparative benefits of N₂D⁺ versus DCO⁺). While it is easy to model a given abundance for any of these two species (N₂D⁺ and N₂H⁺), they both depend on

the unmeasurable abundance of N₂, hence their ratio is independent of this unknown quantity and remains a more constraining quantity to match. This ratio alone is however not sufficient to discriminate between different models. For each model, we can most often find a solution to match any single ratio. Because the density changes with time and because the chemistry evolution speed is dependent upon the actual density, the N₂D⁺/N₂H⁺ ratio profile along the density profile is sensitive to the history of the contraction of the cloud as explained in the Introduction and shown with the preliminary results of P09. Therefore the outcome depends on the collapse duration as long as the chemistry has not reached its steady state. Indeed P09 has shown that the N₂D⁺/N₂H⁺ steady state ratios are much larger than the observed ratios in the case of L183 (their Fig. 7) and therefore discrimination is possible.

Initial conditions are not a difficulty in this model apart from the initial H₂ OPR which is not known. The abundance of N₂ need not be strictly homogeneous as this should not change the diazenylium ratio and both diazenylium isotopologues are undetectable before the contraction starts because CO has not depleted yet (as noted above CO destroys efficiently both species. A spectacular such example is reported in Pagani et al. 2010a). Moreover, all the initial conditions are probably homogeneous enough in the region of interest at the beginning of the contraction when the density is relatively low and uniform. Thus, the variation of the diazenylium ratio profile should trace solely the chemical evolution from the beginning of the contraction.

Apart from density, the chemistry evolution speed itself depends on several conditions which we list below.

1. The H₂ OPR at the start. This could be the only problematic initial condition. The lower the ratio, the faster the deuteration of diazenylium occurs.
2. The grain size. The bigger the grains, the less numerous they are and the recombination of H⁺ (and somewhat H₃⁺) on negatively charged grains is less efficient in removing H⁺ from the gas phase. In turn, a larger abundance of H⁺ allows a faster conversion of ortho-H₂ into para-H₂ (Gerlich 1990) and an acceleration of the deuteration of diazenylium.
3. The cosmic ray ionization rate. The higher the rate, the higher the production of H⁺ and H₃⁺, both converters of ortho-H₂ in para-H₂. Globally, a higher ionization rate means a proportional acceleration of the chemistry.

All these parameters are tested and discussed below and we will show that, for any reasonable value, they do not change our conclusions.

Because the deuteration of diazenylium relies upon the abundance of the deuterated isotopologues of H₃⁺, it could be interesting to investigate also the abundance of these isotopologues. Due to the paucity of the observable transitions from the ground (one ground transition for ortho-H₂D⁺ and one ground transition for para-D₂H⁺) and the uncertainty on collisional coefficients (Hugo et al. 2009), their study is however much more difficult. For the sake of completeness, we present in Appendix B the results of our model for the para-D₂H⁺/ortho-H₂D⁺ ratio in a similar fashion as those of N₂D⁺/N₂H⁺ which we present in Sect. 5.

3. Numerical method

The code we use is described in details in Lesaffre et al. (2005). It solves for the dynamical evolution of hydrodynamics fully coupled to cooling, chemistry and simplified radiative transfer on grains. The system is one-dimensional with spherical geometry.

In the present study, it is discretized with 200 Lagrangian zones initially logarithmically spaced in radius.

The starting conditions of our simulations are a homogeneous sphere with uniform composition, size 0.1 pc (= 20,000 A.U.), and mass 1 M_⊙. As the system tries to relax to a stable Bonnor-Ebert configuration, the external pressure is continuously increased, which eventually leads to gravitational collapse. We stop the computation when the central H nuclei density reaches ~10⁶ cm⁻³, which corresponds to typical prestellar core central densities.

As the cloud first relaxes towards hydrostatic equilibrium, it experiences a few rebounds. On some occasions, these rebounds can trigger spherically convergent shocks which stall the code as they asymptotically reach the centre. In order to avoid this unwanted transient behaviour, we make use of an additional friction term in the velocity equation. The time scale for this friction is 10 My which is short enough to avoid the unwanted shocks but long enough so that it does not affect the collapse.

The chemical network used in Lesaffre et al. (2005) was mainly designed to account for the major coolants (CO, H₂O, OH, H₂) of the multiphase interstellar medium with a minimum number of reactions (120 reactions and 35 species). However, in the present study, we limit the computation to the early phases of the collapse which experience very low temperatures around 10 K. In these conditions, CO remains the only efficient coolant and we are thus free to reduce the cooling-motivated part of our network to the simple reactions of adsorption and desorption of the CO molecule. We then incorporated the full deuteration network of P09 and extended it to include reactions with atomic D which were neglected in P09, unlike all the other models derived from Walmsley et al. (2004), which do include these reactions. These reactions with D (such as H₃⁺ + D → H₂D⁺ + H) have a sizeable influence mostly at steady state when atomic deuterium becomes important. In P09, we concentrated on a 7K gas for which we computed or selected in tables all the rates we needed. As the temperature is self-consistently evaluated here (see below), we had to replace tabulated coefficients with Kooij/Arrhenius expressions ($k = \alpha(T/300)^\beta e^{-\gamma/T} \text{cm}^3 \text{s}^{-1}$ where T is the temperature in Kelvin and α, β , and γ the coefficients to be adjusted to fit the tabulated values) to be able to follow the evolution of the chemical rates with temperature. This concerns the whole Hugo reaction rates (Hugo et al. 2009, values are fitted only between 5 and 20 K in this paper) and the Dissociative Recombination rates from P09. They are given in Appendix C. CO is either a constant (free) parameter or its time-dependent depletion/desorption is taken into account (see Sect. 5.3.3; note that the N₂ abundance is still kept constant as an adjustable parameter, though). The resulting network has 35 independent species for about 400 chemical reactions. We homogenized the treatment of the grains throughout the code: they are now treated as single size grains with an average (adjustable) radius of 0.1 μm.

The radiative transfer on the grains is done as in Lesaffre et al. (2005), and the temperature of the grains depends on the respective coupling between the grains, the local radiation field, and the gas. The core is supposed to be buried in a cloud with A_V = 10 mag attenuation from the surface. Although the UV field plays only a minor role in the heating of the core, this envelope will have re-processed the external radiation into longer wavelengths and we need to model the heating of the grains via this external radiation field. To this purpose we used the same approximations as in Zucconi et al. (2001). We fitted the following analytical expression to the calculated heating of grains due to

external radiation:

$$\Gamma_{\text{external radiation}} = 1.06 \times 10^{-24} G_0 \frac{0.5n_{\text{H}}}{1 + (A_{\text{V}}/4.9)^3} \text{ erg cm}^{-3} \text{ s}^{-1} \quad (1)$$

where A_{V} is the local value of the attenuation from the surface (which increases towards the centre from $A_{\text{V}} = 10$ mag at the outer boundary of the computational domain). We assume a standard external radiation field and set $G_0 = 1$. Compared to Lesaffre et al. (2005), we also slightly improved on the treatment of the dust temperature as we no longer assume thermal equilibrium but compute the time evolution of the dust temperature including the thermal inertia of the grains. However, because this inertia is very small, the results do not change and the only noticeable effect of this modification is to help improving on the numerical convergence. The temperature of the gas itself is controlled mainly by the balance between the grains heat exchange, the cosmic ray heating and the CO cooling. The gas temperature does not always follow the grain temperature, especially when the density is still low. This will be illustrated and discussed in Sect. 5.2.

4. The importance of the ortho-H₂ abundance at collapse start

Some chemical paths are sensitive to the presence of ortho-H₂. It is true for the first steps of ammonia formation (Le Boulrot 1991) and it is true for the first steps of the deuterium chemistry as first discussed by Pineau des Forêts et al. (1991), Pagani et al. (1992) and further developed by Walmsley et al. (2004), Flower et al. (2006b), and P09. It has however no role in the warm deuterium chemistry (based on CH₃⁺), which has been observed by Parise et al. (2009). The decay of ortho-H₂ is a slow process, because of its presumably large initial abundance. It is therefore the limiting factor in the deuteration process, in other words, the one that gives the age of the core. We showed in P09 that its abundance has to drop below ~1% that of para-H₂ to allow for large deuterium enrichments and the time it took to decay to that level was compatible with the free-fall time. However, the core probably does not start to form as soon as all atomic hydrogen has been converted to molecular hydrogen, i.e., as soon as the molecular cloud is considered to be formed, and even if molecular hydrogen has been produced with the largest possible OPR, namely 3, the question arises whether in the cloud, time has been long enough to significantly let ortho-H₂ decay or not before the prestellar core starts to form. Here we discuss the initial OPR when H₂ is formed on the grains and its decrease with time in the cloud prior to the core contraction.

4.1. The initial ortho-para H₂ ratio at formation time

The 4.5 eV energy release of the H₂ formation reaction is large enough to populate many levels of both ortho and para species: it is hence generally accepted, though not totally certain, that the OPR is 3:1 when H₂ forms on grains. A first laboratory confirmation has recently been brought by the experiment of Watanabe et al. (2010) conducted on amorphous solid water. The ortho-para conversion of H₂ can then occur in the gas phase and onto the surface of dust grains (Le Boulrot 2000). The conversion rate on a solid surface is however highly uncertain (Watanabe et al. 2010; Chehrouri et al. 2011)). Once released in the gas phase, H₂ tends to relax towards its thermal equilibrium state but never reaches it because at trace abundance levels, ortho-H₂ destruction is compensated by fresh H₂ formed onto grains from resid-

ual H and from destruction of ions such as H₃⁺ and H₂D⁺ that can release an ortho-H₂ (Le Boulrot 1991; Flower et al. 2006b). This equilibrium value is 3 to 4 orders of magnitude larger than the thermal equilibrium value at 10 K (10⁻³ – a few 10⁻⁵ compared to 10⁻⁷, Flower et al. 2006b, P09, Parise et al. 2011). It is thus necessary to model the chemical evolution of the H₂ OPR to understand the evolution of dark clouds. We therefore adopt a H₂ OPR of 3 at the formation time of the molecule and discuss now the conversion processes in the gas phase prior to the collapse start.

4.2. The new H₂ + H⁺ reaction rate

Radiative transitions between the ortho and para forms of H₂ are known to be very slow, with an interconversion lifetime greater than the age of the Universe for the J = 1–0 transition (Pachucki & Komasa 2008). Inelastic collisions are unable to change the spin state of H₂ either. Therefore only reactions with cations, mostly H⁺ and H₃⁺, can perform such an exchange. The H₂ + H₃⁺ reaction rate has been statistically estimated by Hugo et al. (2009) and recently, we performed an accurate fully-quantum calculation of the H₂ + H⁺ reaction rate (Honvault et al. 2011a,b, 2012). The Kooij/Arrhenius rate formula for the ortho-H₂(J=1) + H⁺ → para-H₂(J=0) + H⁺ reaction is $k(T) = 1.82 \times 10^{-10} (\frac{T}{300})^{0.129} e^{0.0214/T} \text{ cm}^3 \text{ s}^{-1}$ (from Honvault et al. 2011b)¹. At 10 K, the new rate for this reaction is 62% lower than the one previously estimated by Gerlich (1990), respectively 1.13×10^{-10} and $2.972 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$ but it is 43% lower than the one used even at 10 K in Le Boulrot (1991), Walmsley et al. (2004), Flower et al. (2006b), and P09, i.e., $2 \times 10^{-10} \text{ cm}^3 \text{ s}^{-1}$, which is the rate value which was valid only above 20 K in Gerlich (1990) equation 10. Both the H₂ + H₃⁺ and the H₂ + H⁺ reactions have rates an order of magnitude lower than the Langevin rate. As we will show below, in the depleted core, H⁺ can become the dominant cation and rule the decay of ortho-H₂.

4.3. The evolution of the H₂ OPR in the undepleted and low density phase of dark clouds

Though the timescale to convert the bulk of atomic hydrogen to molecular hydrogen is still a debated question, we can conventionally set the birth of the molecular cloud at the time when this conversion is almost complete. From that moment, ortho-H₂ is slowly changing to para-H₂ and the question arises as to how long this takes. To address this question, we have merged one of the recent Ohio State University (OSU) reaction files available (osu_01_2009) with our own deuterium chemistry reaction file and run the new file with our modified version of Nahoon (for the original version, see Wakelam et al. 2005), as described in P09, to introduce the ortho and para states of H₂, H₃⁺ and their related isotopologues and ions (H₂⁺, D₂, etc.). All important reactions involving H₂ or H₃⁺ as reactants in the OSU file have been adapted to the ortho and para states of these species but minor reactions producing H₂, such as CH⁺ + NH₂ → HCN⁺ + H₂ have only been considered to produce para-H₂ which is the most energetically favorable case. We checked that this approximation is not too strong and found that, after 10⁵ years of model evolution for example, 99.9% of the production reactions of para-H₂ are due to reactions of ortho-H₂ with H⁺ and H₃⁺. We run the model for a cloud with a density of $n(\text{H} + 2\text{H}_2) = 2 \times 10^4 \text{ cm}^{-3}$, 10 K, grain radius of 0.1 μm and the standard

¹ Note the erratum: the constant coefficient is $1.82 \times 10^{-10} \text{ cm}^{-3}$ instead of $6.51 \times 10^{-10} \text{ cm}^{-3}$ (Honvault et al. 2012)

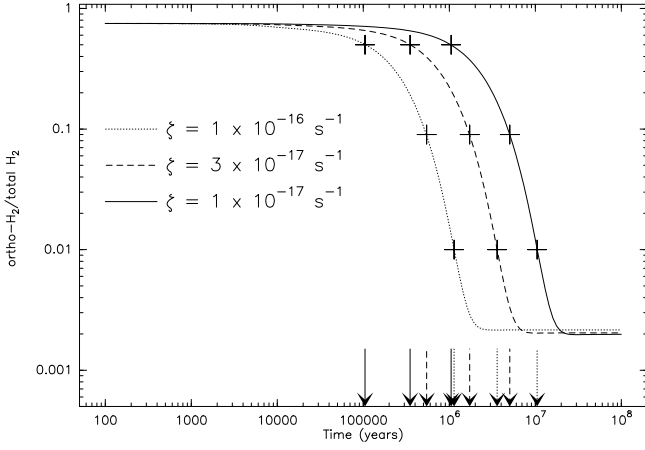


Fig. 1. Ortho-H₂ abundance variation with time for 3 different values of ζ , the cosmic ray ionization rate. Crosses and arrows indicate the time it takes to reach an OPR of 1 (full arrows), 0.1 (dashed arrows), and 0.01 (dotted arrows) for a starting value of 3. The calculations are done for a temperature of 10 K and a density of $n(\text{H} + 2\text{H}_2) = 2 \times 10^4 \text{ cm}^{-3}$

cosmic ray ionization rate, ζ of $1 \times 10^{-17} \text{ s}^{-1}$, which is in fact near the lower end of the range of estimated values in the literature (Padovani et al. 2009). We tried different C/O ratios from 0.4 to 1 and, as expected, found no major difference concerning the evolution of the H⁺ and H₃⁺ abundances and therefore a similar decay rate of the H₂ OPR despite the fact that the main species which react with these two cations are completely different whether the cloud is oxygen-rich or carbon-rich. In this molecular-undepleted model, the recombination of H⁺ and H₃⁺ on negatively charged grains remains a minor route to suppress these cations and therefore, grains with larger size as witnessed via the coreshine effect (Pagani et al. 2010b) have no influence on the result either. Larger cosmic ray ionization rates have a predictable effect: they increase the production of H⁺ and H₃⁺ and therefore accelerate the decay of ortho-H₂. Figure 1 shows the variation of ortho-H₂ abundance with time for 3 different values of ζ (1, 3, and $10 \times 10^{-17} \text{ s}^{-1}$). For the lowest of the 3, the OPR remains above 1 during 1 million years and drops below 1% in 10 million years. If we increase ζ by a factor of 10, the timescale drops also by a factor of 10. In P09, only an initial OPR of 3 at the beginning of the prestellar core phase was considered to model the appearance of deuterated species and the high OPR ratio was the limiting factor to the apparition of these species (see P09 their fig. 8). Any lower OPR, either because H₂ production on grains depart from the statistical ratio or because the cloud is old enough for a given cosmic ray ionization rate, would greatly accelerate the apparition of several deuterated species, especially DCO⁺, and they would reach unobserved high abundances all over the cloud, not only in the depleted cores (Pagani et al. 2011). OPR starting values below 0.1 are therefore excluded and for values close to 0.1, either clouds have a very short lifetime or cosmic ray ionization rate is lower than $1 \times 10^{-17} \text{ s}^{-1}$ which might be difficult to explain (Padovani et al. 2009). It seems therefore most probable that at the moment that dark clouds form and become molecular, the H₂ OPR is between >0.1 and 3. Note that in B68, Troscompt et al. (2009) constrain the H₂ OPR below 1, from H₂CO anomalous absorption observations.

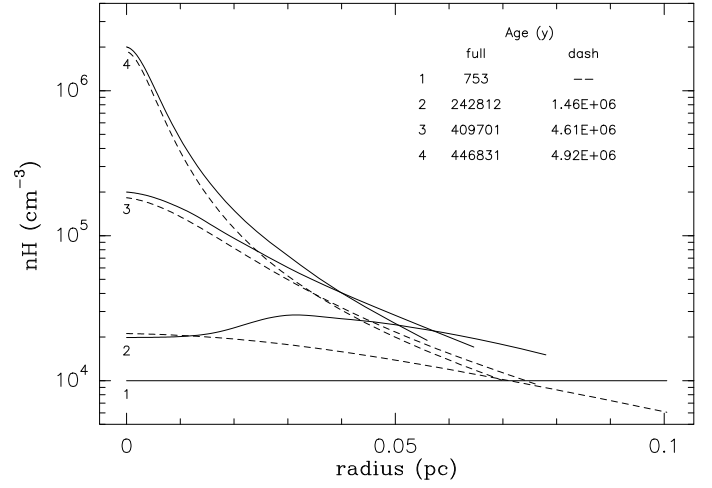


Fig. 2. Density profile evolution with a low, growing, external pressure (slow collapse, dashed lines) and with a high, constant, external pressure (fast collapse, full lines). The conditions after 700 yr of evolution are still identical and traced only once. The increasing ages follow the increasing central densities.

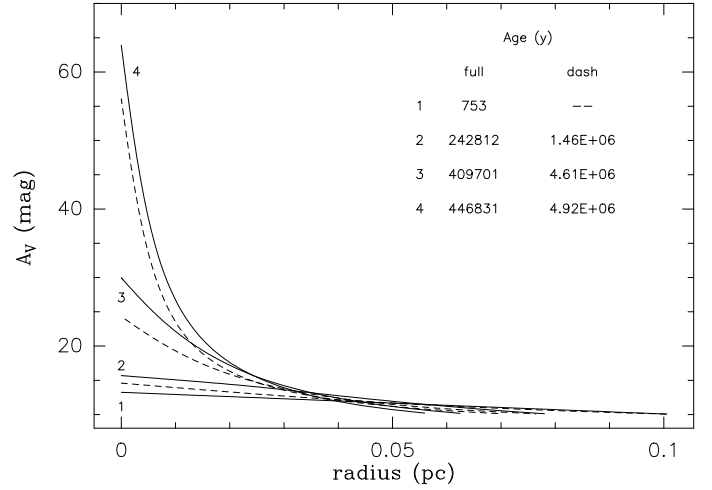


Fig. 3. Column density profile evolution with a low, growing, external pressure (slow collapse, dashed lines) and with a high, constant, external pressure (fast collapse, full lines). The conditions after 700 yr of evolution are still identical and traced only once. The increasing ages follow the increasing central column densities.

5. The model results

We present first the density profile evolution for a fast and a slow collapse with average conditions. This sets up a frame with which we will compare the N₂D⁺/N₂H⁺ ratio profile variation between the two types of models and subsequently the impact of the variations of different parameters on these profiles.

5.1. The density profile

The 1D model starts with a constant proton density set at 10^4 H cm^{-3} , a (depleted) CO abundance of $X[\text{CO}] = \text{CO}/\text{H}_2 = 2 \times 10^{-6}$ and a temperature of 10 K. All the hydrogen is molecular from the start and therefore the initial proton density is equivalent to a particle density of $5900 (\text{H}_2 + \text{He}) \text{ cm}^{-3}$. We apply either a low external pressure ($p_{\text{ext}}/k_B = 2.8 \times 10^4 \text{ K cm}^{-3}$, where k_B is the Boltzmann constant), increasing slowly with a growth time scale of 10^7 years (see Lesaffre et al. 2005, Sect 2.1) to

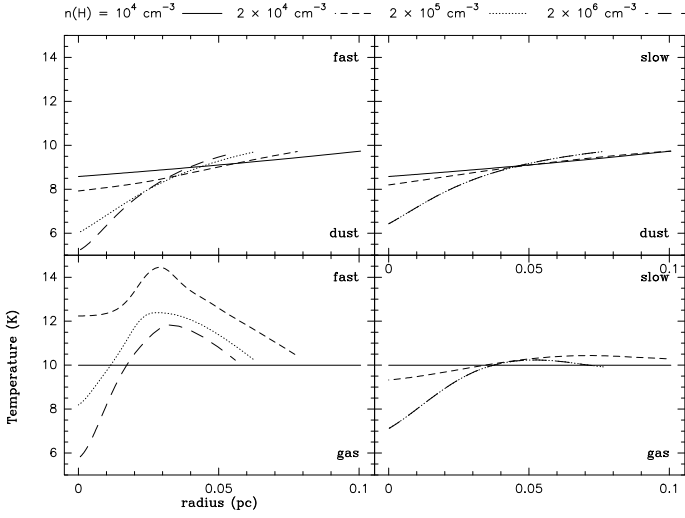


Fig. 4. Evolution of the dust (top row) and gas (bottom row) temperatures with time for the slow (right handside) and fast (left handside) models. The results are traced for the same density profiles as those of Fig. 2. The above mentioned values indicate the peak central densities.

induce a slow collapse, or an external pressure ~ 3 times as high ($p_{\text{ext}}/k_B = 8 \times 10^4 \text{ K cm}^{-3}$) but constant in time to induce a fast collapse, but low enough to avoid the apparition of a shock front. Depending on the parameters (e.g., the cooling of the cloud depends on the CO abundance), the fast collapse reaches a density of $\sim 10^6 \text{ cm}^{-3}$ in a few hundred thousand years, comparable to the free-fall time, while the slow collapse is typically 10 times slower (Fig. 2), taking ~ 5 My to reach the same density. This time scale is comparable to the time scale predicted for models of prestellar core formation by ambipolar diffusion (Tassis & Mouschovias 2007). Note that in the slow collapse model we start with an external pressure slightly below the internal pressure of the cloud, estimated at thermal balance between heating and cooling. Hence, the cloud starts with a small expansion before contracting. In Figure 2, we have selected 3 representative density profiles, which are quite similar for both models. The main difference is that during the first couple of million years, during the slow collapse, after the small expansion the cloud oscillates around a density of a few 10^4 cm^{-3} before starting to contract quasi-statically. The first slow collapse profile is traced just after the end of the oscillation, which occurs ~ 1 My after the start of the model (at 700 yr, the profile is identical for both types of collapse and traces the initial state, i.e. 10^4 cm^{-3}). Hereafter, we will call these models the fast model and the slow model. The time intervals to reach given central densities can change slightly between models when we vary the different input parameters to be tested but the change is on the order of 15% maximum except when the heating or the cooling are modified (e.g. CO abundance variation, because CO is the main coolant of the cloud and therefore its presence helps to accelerate the contraction, or ζ increase, which increases the heating input to the cloud), in which case we readjust the initial external pressure to return close to the 0.5 and 5 My duration of the models.

5.2. Grain and gas temperatures

In the early phases, the gas thermal balance is dominated by CO cooling and photo-electric heating. **The cosmic ray heating throughout the whole cloud amounts to a less important contribution.** On the other hand, the grain temperature is con-

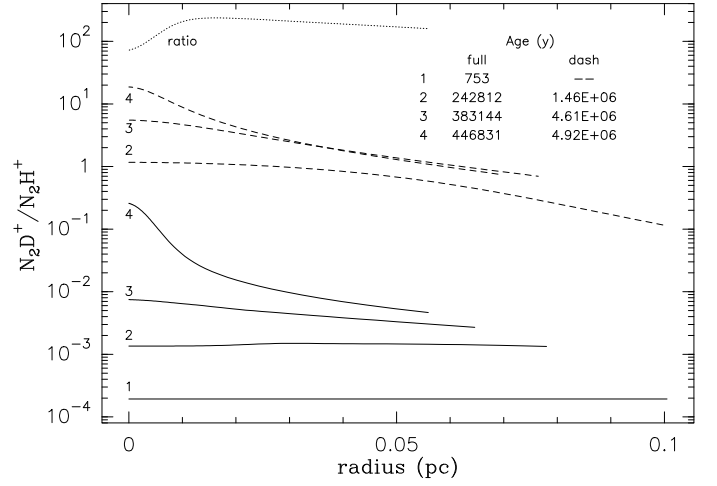


Fig. 5. Evolution of the $\text{N}_2\text{D}^+/\text{N}_2\text{H}^+$ ratio with time for the slow (dashed lines) and fast (full lines) models. The dotted line traces the slow/fast ratio for the last epoch (corresponding to a peak density of $2 \times 10^6 \text{ cm}^{-3}$). The results are traced for the same density profiles as those of Fig. 2

trolled by the radiative balance between the external radiative heating and the grain emissivity. Both temperatures (grain and gas) are then independent from each other. This remains true in the envelope throughout all runs. However, when the core is dense enough to be shielded from external photons, the grain heating gets dominated by the collisional exchange with the gas and the compressional heating becomes the main source of heat for the gas. At that point, the compressional heating is transferred to the grains via collisions and is then evacuated by the radiation off the grains. As the density increases, the grain and gas temperatures become closer and closer to the cosmic microwave background radiation temperature (2.725 K). Figure 4 shows the evolution of the gas and dust temperatures during the contraction. The bump at radius 0.03 pc in the gas temperature profile of the first snapshot (dashed line, $2 \times 10^4 \text{ cm}^{-3}$ case) of the fast case is due to the compressional heating of a pressure wave which is still propagating inward at this time while the density is still too low to couple efficiently the gas and the dust. Due to the spherical convergence towards the centre, this bump later steepens to reach 23 K at its maximum when it hits the centre. However, this temperature rise lasts only for a short time (the central region, < 0.01 pc, remains above 15 K for 4×10^4 years only) and is not seen to affect significantly the chemical composition of the gas. The pressure wave dies out shortly after it moves back outwards. When the CO depletion is treated self-consistently, the pressure bump effect is even more attenuated, because the enhanced CO cooling forces temperature to a lower level as the pressure wave passes before CO becomes strongly depleted. No surge is seen in the slow case.

5.3. Model predictions

We will first compare the deuteration profile for the slow and fast models in between them, and then present their sensitivity to different parameters for each of them separately by comparison with the first two (fast and slow) models taken as references.

5.3.1. The reference models

Figure 5 shows the $\text{N}_2\text{D}^+/\text{N}_2\text{H}^+$ ratio for both models with $X[\text{CO}] = 2 \times 10^{-6}$ (equivalent to a depletion factor between 50

and 100 for a standard undepleted CO abundance between 1 and 2×10^{-4} , Dickman 1978; Pineda et al. 2010. **This value is close to typical depletion values found in the literature, such as in Tafalla et al. 2002 or Pagani et al. 2005**, uniform and constant from the start, $X[\text{N}_2] = 3 \times 10^{-7}$, uniform and constant from the start (**this value is representative of the values we found in P09**. As its abundance has a negligible role in the model, N₂ will always be kept constant and uniform), a grain radius of 0.1 μm , a H₂ OPR of 3 and a cosmic ray ionization rate of $1 \times 10^{-17} \text{ s}^{-1}$. The dust to gas mass ratio is 0.01 and the dust density is 3 g cm^{-3} , the values usually adopted in chemical models. There is a large difference between the two models. The slow model has reached steady state in the outer layers before the density profile has reached its 10^6 cm^{-3} peak density and the difference in the ratio profile between slow and fast models is variable, ranging from 70 to 200 for the final density profile. As expected the models can help to discriminate between slow and fast collapse theories with an unambiguous N₂D⁺/N₂H⁺ ratio profile variation.

This pair of models will be the reference for further comparisons hereunder. For each model, slow or fast, we will compare the new model with the one presented here in the next figures (new model = full lines, reference model = dashed line except where stated differently). To keep the figures readable, we will compare separately the fast and slow models but plot them next to each other with identical scales.

5.3.2. The temperature sensitivity

During the contraction of the cloud the gas and dust temperatures evolve as a result of imbalance between heating and cooling processes (see Sect. 3 & 5.2). The gas temperature influences those rates in the chemical network which are temperature dependent. Because the gas temperature profile self-consistently computed by the model differs from the gas temperature profile constrained by our N₂H⁺ observations and radiative transfer modeling of L183 (Pagani et al. 2007, though L183 might not be a reference case in terms of temperature profile by itself), we have tested what would happen if we keep the chemical network temperature constant at low values (between 7 and 15 K). Figure 6 shows that the N₂D⁺/N₂H⁺ ratio is slightly flatter with fixed temperatures as a function of radius for the fast case, and that for the slow case, it is basically unchanged with a slightly different equilibrium value towards the centre, which can be linked to the final temperature (similar examples are given in Parise et al. 2011). As we mention in Sect. 5.2, the pressure wave has also little effect on the chemistry, past the transient event in the fast case. Overall the variation is negligible and the actual temperature profile is not a major source of uncertainty on the diazenylium deuteration enrichment estimate.

5.3.3. The CO abundance impact

Constant CO abundance The impact of CO abundance on the diazenylium ratio is displayed in Fig. 7. From the reference models, we have either decreased the CO abundance (or increased the depletion) by a factor of 2 ($X[\text{CO}] = 1 \times 10^{-6}$, depletion factor between 100 and 200) or increased it by a factor of 5 ($X[\text{CO}] = 1 \times 10^{-5}$, depletion factor between 10 and 20), to cover the usual range of local depletion factors found in the literature (e.g. Caselli et al. 1999; Bergin et al. 2002; Bacmann et al. 2002; Tafalla et al. 2004; Pagani et al. 2005). The fast model gives a stronger deuteration level (gain $\approx 2-4$) when reaching the peak

density if CO is depleted further or a much lower deuteration (drop up to 30) if CO is less depleted. This is due to a higher fraction of H₂D⁺ being diverted to react with CO instead of HD (and therefore a diminished production of the D₂H⁺ and D₃⁺ deuteration vectors). The effect is therefore cumulative which explains the amplification between the CO increase and the spectacular deuteration profile drop. The slow model shows little sensitivity to the decrease of CO abundance but the factor 5 increase has a more dramatic effect. The final deuteration profile drops by up to a factor of 8. This is one of the two cases where the N₂D⁺/N₂H⁺ ratio remains below 5 in the slow model, which could marginally fit some observations, especially if the peak density is still low. However the slow model leaves ample time for CO to deplete by more than the factor 10-20 used here and therefore this situation might be uncommon. With a freeze-out timescale $\tau_{\text{fo}} = 5 \times 10^9 / n_{\text{H}_2}$ years (Bergin & Tafalla 2007), a depletion factor of 20 is reached in 1.5 My for a density as low as $1 \times 10^4 \text{ cm}^{-3}$.

Dynamic CO depletion We now follow the dynamical depletion of CO during the contraction **starting from a CO/H₂ relative abundance of 1×10^{-4}** (Fig. 8). If the CO depletion mechanism is relatively well-understood, its desorption is much less certain (e.g. Öberg et al. 2005; Bisschop et al. 2006; Roberts et al. 2007; Öberg et al. 2009). Therefore it did not seem secure to force desorption evolution in the model in this present study especially as a full model would request to include all species not depleted at the start with a large chemical network to follow their overall disappearance along with CO in the contracting and depleting core, which is beyond the scope of the present model. Even limiting the depleting species to CO and N₂ is a problem since their very different depletion profile behaviour is not explained by recent models and laboratory work, (see Öberg et al. 2005; Bisschop et al. 2006; Pagani et al. 2012). However, the slow disappearance of CO must slow down the deuteration of N₂H⁺ and we need to check the importance of the effect to evaluate its impact on the model behaviour. We have therefore added in the chemical model two rates, one to describe the sticking probability of CO on the grains upon collision (which we set to 1, following the laboratory work of Bisschop et al. 2006, which gives a lower limit of 0.87 ± 0.05), and one to describe the CO desorption upon the cosmic ray impact on the dust grains. More details are given in Lesaffre et al. (2005, their Sect. 3.1.2). The CO abundance evolution is displayed in the two upper boxes of Fig. 8. In the center, the depletion can reach a factor of ~ 500 but in the outer parts of the prestellar core, the depletion is less than a factor of 2.5 in the fast case (after 0.5 My) and about a factor of 5 in the slow case (after a few My). Such slopes are possibly seen in a few Taurus cases (Tafalla et al. 2002, 2004) but do not seem to fit the model in B68 (Bergin et al. 2006), our own observations of L183 (Pagani et al. 2007), which show a strong CO depletion all over the prestellar core, or the L1506C case (Pagani et al. 2010a), which shows a steep drop in CO abundance even for relatively low density and extinction. Therefore this dynamical CO model may represent some real cases but not all of them. CO being more abundant by more than an order of magnitude in the outer parts compared to the reference model, its destruction of H₃⁺ and H₂D⁺ prevents an efficient deuteration of N₂H⁺ to occur. In the fast case, the deuteration is 2 to 14 times lower for the final density profile. In the slow case, though the production of D₂H⁺ and D₃⁺ are slowed down by the higher CO abundance, they eventually build up enough to almost produce the same level of deuteration. For the final density profile, the

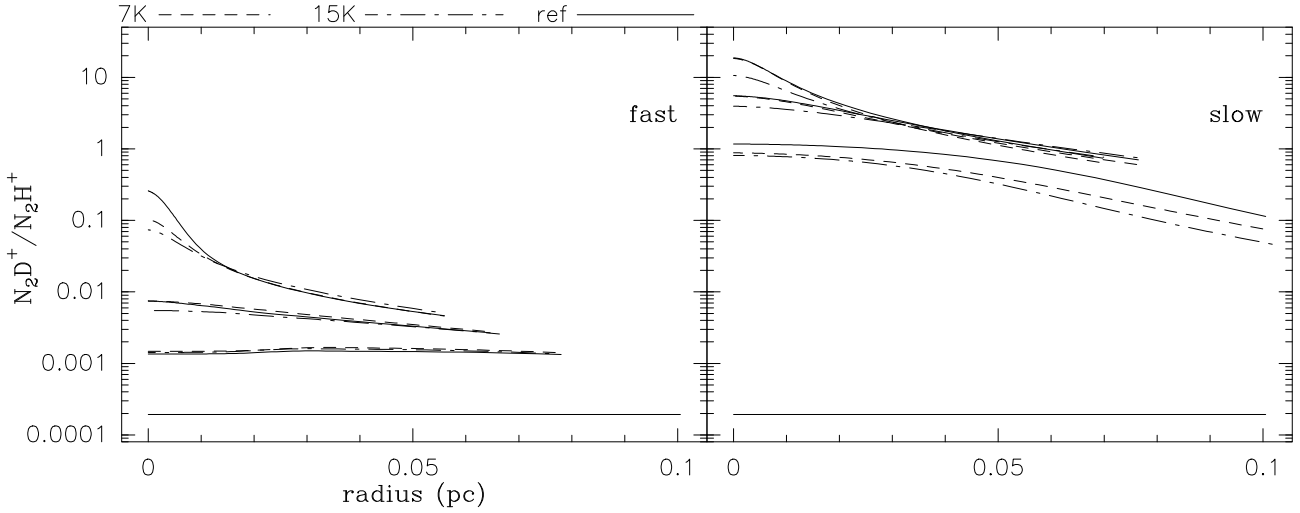


Fig. 6. Evolution of the N_2D^+/N_2H^+ ratio with temperature for the slow (right) and fast (left) models. In the reference model, the temperature of the chemical network is the temperature of the gas. The two other cases are for constant temperature imposed to the chemical network. To better see the differences, the reference model has been drawn with the full line mode (from respectively the slow and fast cases displayed in Fig. 5).

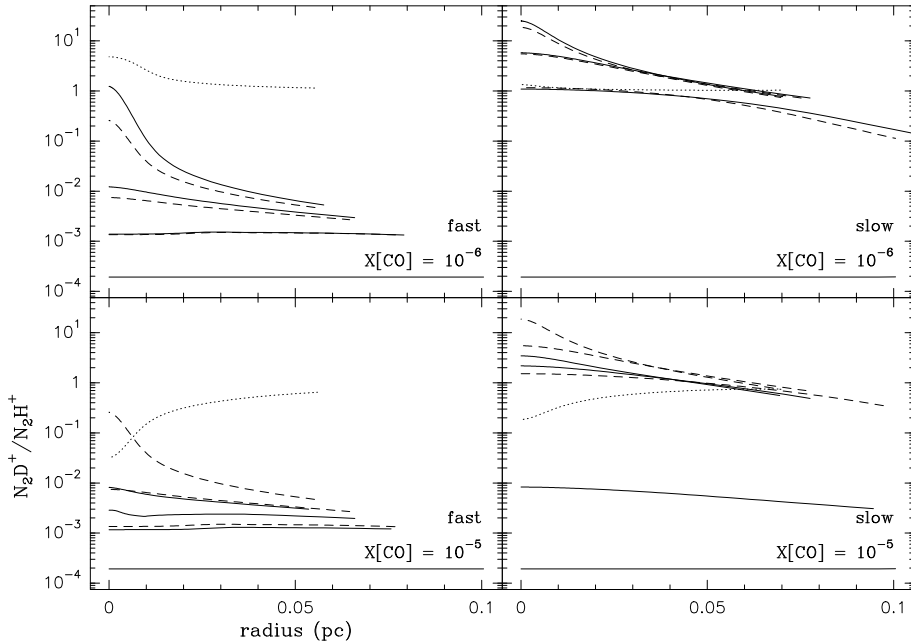


Fig. 7. Evolution of the N_2D^+/N_2H^+ ratio with time for the slow (right) and fast (left) models as a function of CO depletion (depletion multiplied by 2 in the top row, $X[CO] = 1 \times 10^{-6}$, divided by 5 in the bottom row, $X[CO] = 1 \times 10^{-5}$). The dashed lines trace the reference results (from respectively the slow and fast cases displayed in Fig. 5). The dotted line traces the new versus reference model ratio for the last epoch (corresponding to a peak density of $2 \times 10^6 \text{ cm}^{-3}$) in each frame. The results are traced for the same density profiles as those of Fig. 2.

CO abundance is lower than in the reference case in the central 0.015 pc, and hence we find a higher level of deuteration. The N_2H^+ and N_2D^+ abundances are however somewhat reduced in the outer parts of the core because of the higher abundance of CO (but since N_2 chemistry is not dealt with, this has little implications).

5.3.4. The N_2 abundance impact

As expected, we found no influence whatsoever on the diazenylium ratio when varying the N_2 abundance from 1×10^{-9} to 5×10^{-7} . The main impact is to change the diazenylium isotope abundances.

5.3.5. The H_2 OPR impact

The reference models both started with a H_2 OPR of 3. Though we do not expect OPR ratios lower than ~ 0.1 typically at the

start of the contraction (see Sect. 4.3), we have tested the effect of different initial OPR values from 1 down to 0.003 (Fig. 9). To avoid crowding in Fig. 9, we display only the two highest density profiles (peak densities of 10^5 and 10^6 cm^{-3}) in two different frames for 5 OPR values (0.003, 0.01, 0.1, 1 and 3). The slow model is totally insensitive to initial OPR variations which indicates that the cloud is old enough to have already converted most of its ortho- H_2 to para- H_2 , bringing maximum deuteration capability even for high initial OPRs. The slow model is evolving with its chemistry in a steady-state situation before density becomes really high. For the fast model, the variation is more dramatic. For $OPR \leq 0.01$, the model is close to chemical steady-state and hardly discernable from the slow model (observational uncertainties would be much larger than their differences). This is not surprising as the main slow evolving chemical species is the ortho- H_2 , everything else adjusting rapidly to its new value. Therefore if the model starts at such low OPRs, full deuteration is expected from the start as shown in P09 (and DCO^+ would be detectable all over the cloud as discussed in Pagani et al. 2011).

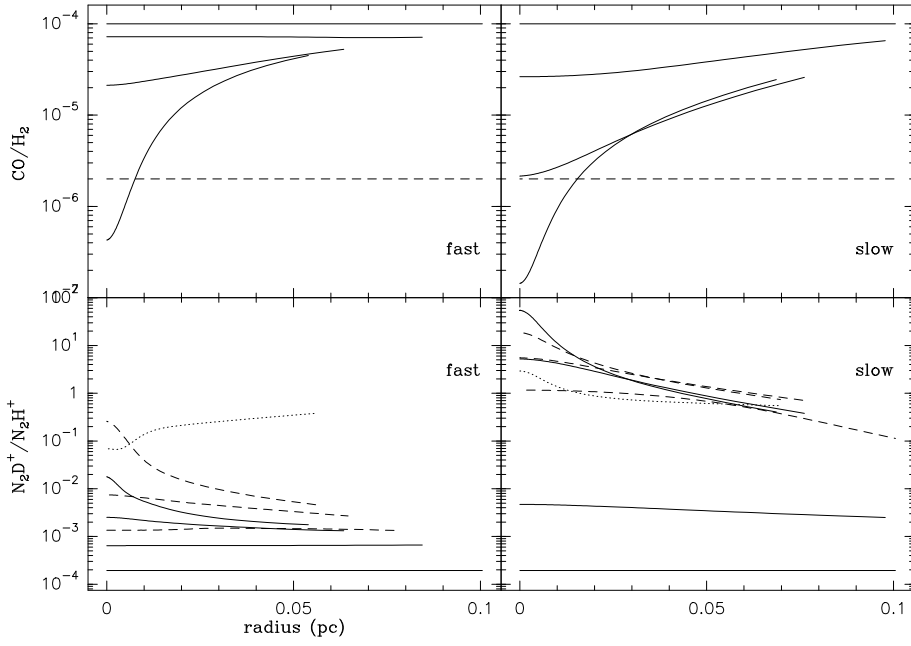


Fig. 8. Evolution of the N_2D^+/N_2H^+ ratio when CO depletion is varied compared to constant depletion, for the slow (right) and fast (left) models. CO variation is shown in the upper graphics, constant depleted CO appears as long dashed lines. In the bottom row, the dashed lines trace the reference results (from Fig. 5) and the dotted line traces the new versus reference model ratio for the last epoch (corresponding to a peak density of $2 \times 10^6 \text{ cm}^{-3}$) in each frame.

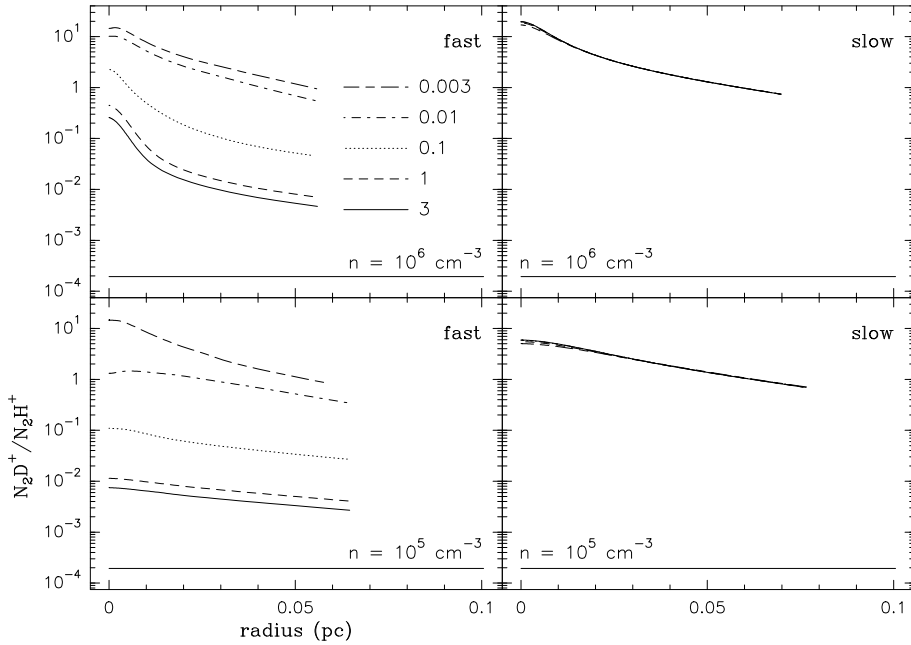


Fig. 9. Evolution of the N_2D^+/N_2H^+ ratio with the H₂ OPR values at the beginning of core contraction for the slow (right) and fast (left) models. The bottom row corresponds to the density profile with peak density of 10^5 cm^{-3} and the top row to the profile with peak density of $2 \times 10^6 \text{ cm}^{-3}$ (from Fig. 2). Initial OPR values are listed in the top left image. The full lines (OPR = 3) also trace the reference results (from Fig. 5).

5.3.6. The cosmic ray impact

The cosmic ray ionization rate is not well-known and seems to vary from place to place or from the diffuse medium to the dense medium (Padovani et al. 2009). As mentioned above (Sect. 4.3), a cosmic ray ionization rate below $1 \times 10^{-17} \text{ s}^{-1}$ seems improbable and increasing the rate will accelerate the decay of ortho-H₂. We have tested one case with a lower rate though ($3 \times 10^{-18} \text{ s}^{-1}$), and two cases with a higher rate ($3 \times 10^{-17} \text{ s}^{-1}$ and $1 \times 10^{-16} \text{ s}^{-1}$, Fig. 10). Decreasing the rate slows down the chemistry as fewer ions are produced while the chemistry is ion-driven. It is so much slowed that deuteration drops by a factor as high as 100. This is also the only situation for which the slow case does not reach equilibrium before the density has neared $1 \times 10^5 \text{ cm}^{-3}$. However, once the equilibrium has been reached, the deuteration of diazenylium is twice as high as in the standard case because of the lack of electrons to destroy the various deuterated ions. Increasing the cosmic ray ionization rate by as little

as a factor of 3 has also a major impact on the chemical evolution of the cloud. The ortho-H₂ abundance drops faster and the N_2D^+/N_2H^+ ratio increases faster for the fast model. The slow model is again evolving in steady state mode but the equilibrium settles lower than in the reference case. This is due to the higher electron abundance, directly proportional to the cosmic ray ionization rate, which increases the recombination efficiency of the cations and therefore prevents these cations to proceed further along the deuteration path. When increasing the cosmic ray ionization rate up to $1 \times 10^{-16} \text{ s}^{-1}$, the electron density increases further and therefore the deuterium profiles become even lower. In such cases, for moderate densities, the slow model deuteration level may become compatible with some observations.

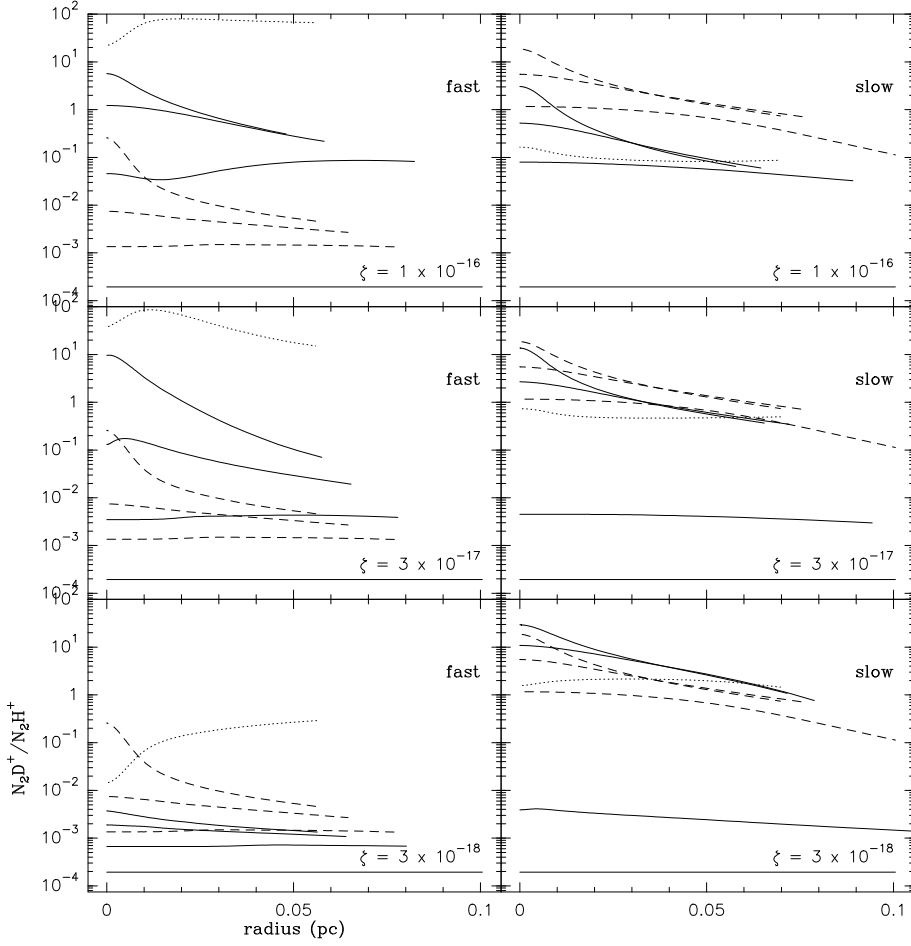


Fig. 10. Evolution of the N_2D^+/N_2H^+ ratio when the cosmic ray ionization rate is divided by 3 (bottom row), multiplied by 3 (middle row) and by 10 (top row) for the slow (right) and fast (left) models. The dotted line traces the new versus reference model ratio for the last epoch (corresponding to a peak density of $2 \times 10^6 \text{ cm}^{-3}$) in each frame. The dashed lines trace the reference results (from Fig. 5).

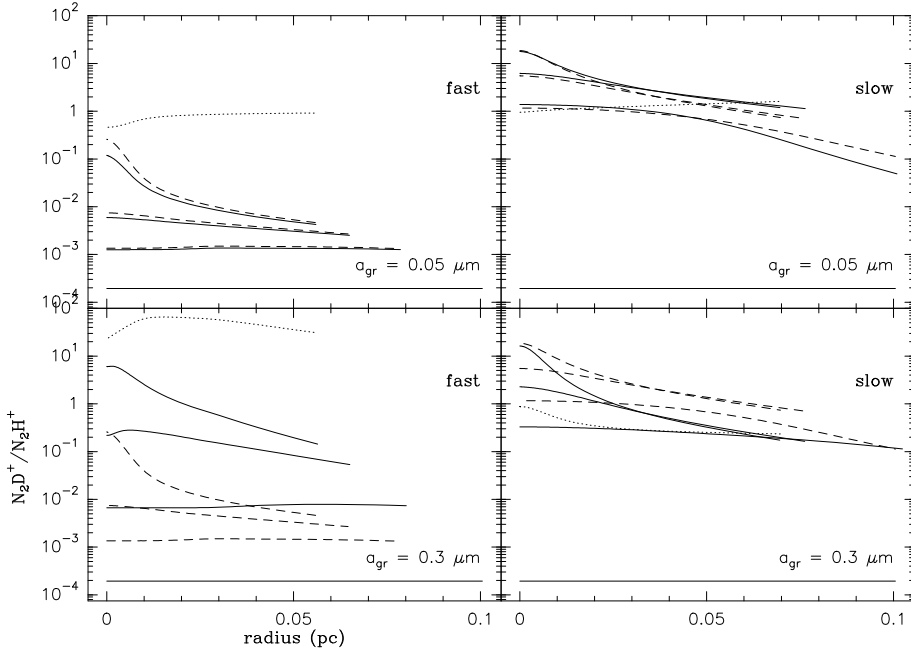


Fig. 11. Evolution of the N_2D^+/N_2H^+ ratio when the grain size is multiplied by 3 (bottom row) or divided by 2 (top row) for the slow (right) and fast (left) models. The dashed lines trace the reference results (from Fig. 5). The dotted line traces the new versus reference model ratio for the last epoch (corresponding to a peak density of $2 \times 10^6 \text{ cm}^{-3}$) in each frame.

5.3.7. The grain size impact

Grain size in dark clouds is an open subject which might improve with the combination of 3D radiative transfer dust models and diffraction observations from optical to mid-infrared (Steinacker et al. 2010; Pagani et al. 2010b; Juvela et al. 2012). A standard grain size of $0.1 \mu\text{m}$ is usually assumed in chemical models

but this is a strong approximation which might not be valid. In typical models, $\sim 90\%$ of the grains are negatively charged regardless of their size and following Draine & Sutin (1987), the grains are in principle able to carry 1 electron only. Therefore the average size of the grains determines the density number of charged grains and their ability to neutralize H^+ ions. For molec-

ular rich models (i.e. models without depletion) the recombination on grains is negligible, and the grain size does not matter. However, here recombination has a major impact. Therefore the question is what size of grains should we use in prestellar cores? Following our recent coreshine results, 50% of a large sample of prestellar cores have large grains (Pagani et al. 2010b) but on the other hand, Ormel et al. (2009) show that the grain size distribution shows a double peak profile (before it reaches a steady state between coagulation and fragmentation), with one peak of growing size grains and one peak of constant small size grains of comparable importance. While the growing size grains fraction can explain the coreshine effect, the small size grains will maintain a large number of grains and therefore of negatively charged grains. We have therefore explored the behaviour of our model with grains of 0.05 μm and 0.3 μm radius.

Increasing the grain size increases the abundance of H⁺, as less negatively charged grains are available for recombination with H⁺. In turn, the ortho-to-para H₂ conversion efficiency increases and deuteration efficiency follows. Like in the previous case, the fast case deuteration efficiency is higher (Fig. 11). However, the lower density number of grains also absorb less electrons and the number of free electrons remains higher, which brings us back to the previous situation too. Higher cosmic ray ionization rate or bigger grains have the same kind of influence on the deuteration efficiency in depleted cores. If we decrease the grain size, smaller grains are more numerous, and will interact more often with H⁺ and the ortho-to-para H₂ conversion efficiency will diminish. Opposite effects are seen as expected, however the effect is very small.

5.3.8. Slow versus fast models: a summary

Overall, the slow model has a relatively weak dependence upon all the possible initial conditions we have tested while the fast model is sensitive to them but remains always distinct in its predictions from the slow model. Therefore, if the observations are found to fit best one of these models (without much possibilities of adjustment for the slow model, with some flexibility for the fast one), we will be able to tell whether the deuteration of N₂H⁺ is compatible with rapid or slow collapse, independently of the initial conditions. Only two cases could prevent us to conclude: if the initial H₂ OPR is already below 1–10% before the cloud starts to collapse, or if the cosmic ray ionization rate is high. But that might have other consequences which can be tested like the presence of DCO⁺ all over the cloud (Pagani et al. 2011).

For the sake of completeness, we also compared the deuteration evolution using the old ortho-to-para conversion rate from Gerlich (1990) to our new one (see Sect. 4.2). Overall, the deuteration efficiency has dropped by a factor of three with the new, slower rate in the fast model while our new rate has no influence in the slow model case as with most of the other parameters (see Appendix A).

6. The L183 case: discussion

We now compare these models to our L183 observations in the same manner as in P09 but with a 1D evolving model instead of a juxtaposition of layers of different and constant densities. We first searched the closest density profile (Fig. 12). The fit is not very good which does not come as a surprise as the collapse model displays a density slope close to $\rho \propto r^{-2}$ from $r = 0.003\text{pc}$ ($= 600\text{A.U.}$) outward, while we showed that our data are compatible with $\rho \propto r^{-1}$ (Pagani et al. 2004) up to $0.02\text{ pc} = 4000\text{ A.U.}$, a feature seemingly difficult to explain apart from some

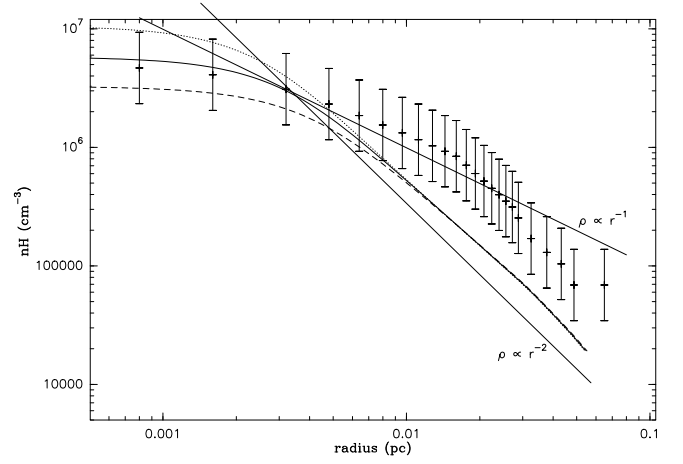


Fig. 12. Comparison between models and the L183 observed density profile (from Pagani et al. 2007). The error bar represents a variation by a factor of 2, not a real error estimate. The dashed, full and dotted lines are the three successive steps from one of the models. The x axis is cylindrical radius.

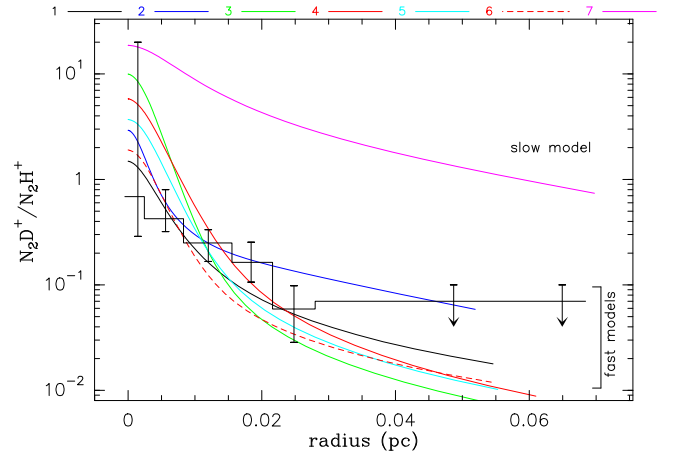


Fig. 13. Comparison between different models output for the peak density matching the observations and the N₂D⁺/N₂H⁺ local abundance ratio derived from the L183 observations (from Pagani et al. 2007). The error bars represent the maximum (resp. minimum) N₂D⁺ abundance profile compared to the minimum (resp. maximum) N₂H⁺ abundance profile as displayed in Fig. 4b of Pagani et al. (2007). The 7 models are listed in Table 1. The x axis is cylindrical radius.

possible logotropic models (see Pagani et al. 2004, for a discussion). The three successive steps of the model calculations traced in Fig. 12 show that only the very inner part of the density profile keeps increasing after a while. It is therefore difficult to fit our observed density profile precisely, even allowing for a factor of 2 variation in the density values and we have decided to take the profile which best fits the inner density part of our data, i.e. to stop the model when its peak density reaches $\sim 2 \times 10^6$ H₂ molecules cm⁻³. As discussed above (Sect. 5.1), the density profile remains basically the same in the density regime we are interested in here, whether coming from a slow collapse or from a fast collapse model.

Figure 13 shows the N₂D⁺/N₂H⁺ ratio for 7 different models compared to the ratio derived from the analysis of the observations (Pagani et al. 2007). Table 1 indicates the main differences between the models. Only the fast collapse models can pretend to match the observations. If the fast model in the standard case

Table 1. List of the seven models presented in Figs. 13 & 14.

model	CO/H ₂ abund.	grain size (μm)	collapse	ζ (s^{-1})	H ₂ OPR	T _{chem} ^a (K)	age ^b (My)
1	2×10^{-6}	0.2	fast	1×10^{-17}	3:1	7	0.46
2	calculated	0.3	fast	1×10^{-17}	3:1	free	0.48
3	5×10^{-7}	0.1	fast	1×10^{-17}	3:1	free	0.52
4	2×10^{-6}	0.1	fast ^c	1×10^{-17}	3:1	free	0.62
5	2×10^{-6}	0.1	fast	1.5×10^{-17}	3:1	free	0.46
6	2×10^{-6}	0.1	fast	1×10^{-17}	1:2	free	0.46
7	2×10^{-6}	0.1	slow	1×10^{-17}	3:1	free	4.90

Notes. (a) The chemical network temperature follows the gas kinetic temperature (free) or is kept constant to the given value. See Sect. 5.3.2. (b) The age is not an input parameter of the model but one of the results: the time it took to the model to reach the observed peak central density. (c) Though still a fast collapse case, the initial pressure has been reduced to $6 \times 10^4 \text{ K cm}^{-3}$, see Sect. 5.1

(0.1 μm size grains, $\zeta = 1 \times 10^{-17} \text{ s}^{-1}$, fixed CO depletion by a factor of 50, H₂ OPR = 3) produces not enough deuteration (not shown here), small changes of one of the parameters can bring a reasonable fit to the observations. Either a 50% increase in the cosmic ionization rate (model 5), a stronger depletion by a factor of 4 (model 3), a grain size increase by a factor of 2 (model 1), or a slight deceleration of the contraction speed (model 4, collapse time increased by 30%) are enough to get a profile comparable to the observed one. A lower H₂ OPR (0.5, model 6) similarly matches the observations. We tested the limits beyond which the model clearly does not fit the observations anymore and found that slowing down the collapse to make it last 0.8 My (collapse time increased by 60%) produces too much deuteration or that a starting H₂ OPR of 0.2 was only marginally compatible with the observations. With $\zeta = 1 \times 10^{-17} \text{ s}^{-1}$, it takes about 3.5 My to go from an OPR of 3 to an OPR of 0.2. This sets an upper limit to the cloud age since its formation (which we have defined as the moment where most of atomic H is converted into H₂) of $\sim 4 \text{ My}$, i.e., a limit even lower than what we obtained from considerations on the absence of DCO⁺ outside prestellar cores in dark clouds (Pagani et al. 2011). Another interesting model, is the model 2 which combines both variable depletion profile and big grains which could be closer to the actual situation than the other cases, though depletion is not yet measured precisely in the L183 core. Model 2 is the closest to the N₂D⁺/N₂H⁺ observed profile.

Because most slow models show little difference in between them when reaching densities above 10^6 cm^{-3} , we have shown only one typical case (model 7). Its predictions are far above the observations. In any case, it predicts a N₂D⁺/N₂H⁺ ratio above unity for most of the core and one order of magnitude too large at all radial distances. Therefore it seems impossible to adjust a slow collapse model to our observations. This also excludes fast collapse models for which the deuteration becomes too high, e.g. models where the initial H₂ OPR ratio is already below 0.1, or with cosmic ray ionization rates too high.

The fit is also relatively good with ortho-H₂D⁺ (Fig. 14). The average abundance is correct for the different fast models and even for the slow model but the profile is somewhat higher in the outer parts than observed. The best model (model 2) for the N₂D⁺/N₂H⁺ ratio is the worst one for the abundance of H₂D⁺. In the outer parts of the core, the models predict too much H₂D⁺. Despite this H₂D⁺ abundance which is too high beyond 0.03 pc, the Diazenylium fractionation remains slightly too low (Fig. 13). This contradiction is only apparent because the density being too low in the outer part of the model, the conversion of H₃⁺ to D₃⁺ is lagging behind and the chemical model still peaks on the production of H₂D⁺ more than on its conversion to D₂H⁺ and D₃⁺.

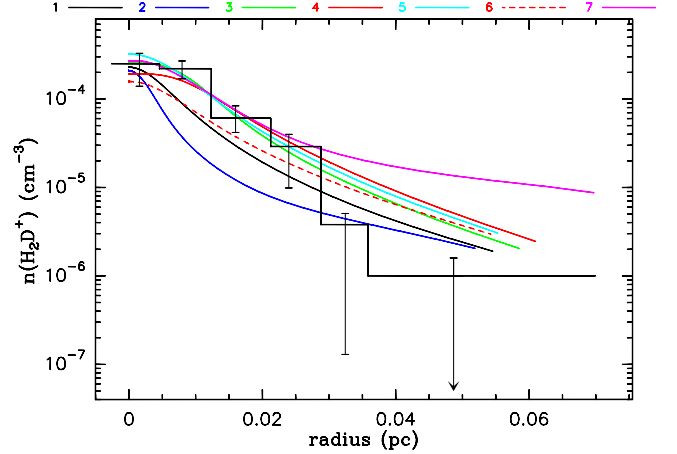


Fig. 14. Comparison between different models output for the peak density matching the observations and the H₂D⁺ local density derived from the observations. The error bars represent the range of possible values fitting the observations (Fig. 3 of P09). The 7 models are listed in Table 1.

This species (H₂D⁺) is not the only deuteration partner of N₂ and therefore its abundance is not forcibly linked to the diazenylium fractionation (see Sect. 3.3 in P09). This is why its similar abundance profile for the slow case is not linked to the too high diazenylium deuteration profile of Fig. 13. Therefore the ortho-H₂D⁺ abundance profile is not discriminant against slow or fast models.

7. Conclusions

To observe the diazenylium fractionation profile across prestellar cores is a possible way to establish the age of these cores. Deuteration enrichment is bound to increase in cold clouds, leading to a predictable diazenylium ratio profile. This should allow to estimate the age of all prestellar cores that undergo CO depletion.

L183 is compatible with a fast collapse model starting with a high H₂ OPR. An OPR ≥ 0.2 sets an upper limit to the beginning of the collapse of 3.5 My after the formation of the molecular cloud and the fast collapse itself lasts typically 0.5 My (and $\leq 0.7 \text{ My}$). Therefore, the deuterated profile of diazenylium in L183 sets an upper limit of 4 My to the age of the core since the formation of the cloud. This result is not compatible with the slowest of the ambipolar diffusion-based contraction models and confirms the results found by Lee & Myers (1999). It also confirms that steady-state chemistry is not reached in prestellar

cores and in particular in L183 and therefore steady-state chemical models should be avoided at this stage of star formation.

Acknowledgements. TGL would like to thank support from the Spanish Ministry of Science and Innovation under grants FIS2010-18132, FIS2011-29596-C02-01 and Consolider Ingenio 2010 CSD2009-00038. PL acknowledges support from the french embassy in the UK as he worked as overseas fellow in Churchill college, Cambridge when this work was initiated in 2009. We thank the referee, B. Parise for a careful reading of the manuscript and many constructive criticisms and discussions that helped to improve the paper.

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Appendix A: The new H₂ + H⁺ rate impact

It is interesting to check the importance of the new ortho-to-para conversion rate of H₂ on the deuteration efficiency as this conversion rate is the prime ruler among chemical reaction rates on the evolutionary time of the chemistry. Fig. A.1 shows the deceleration of the deuteration with the new rate (which is the reference case, i.e. the dashed lines) compared to the old rate (Gerlich 1990, which is the test case here, i.e. the full lines). The difference in deuteration is a factor of 3 for the fast case. There is however almost no difference for the slow case which has already reached steady state, even more so with the previous faster rate.

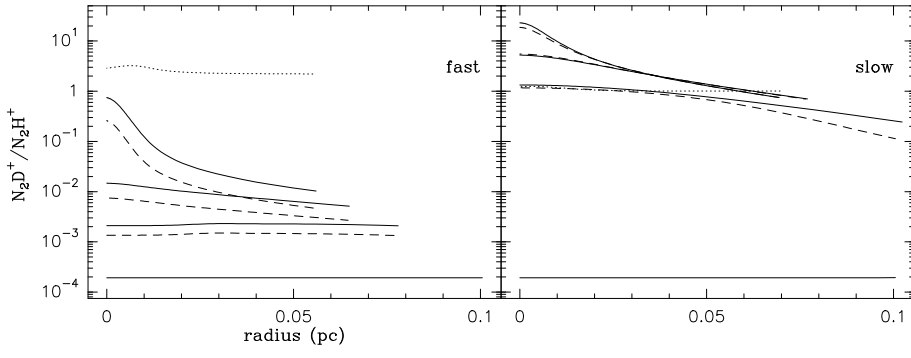


Fig. A.1. Evolution of the N_2D^+/N_2H^+ ratio with the old $H_2 + H^+$, for the slow (right) and fast (left) models. The dashed lines trace the reference results (from Fig. 5). The dotted line traces the new versus reference model ratio for the last epoch (corresponding to a peak density of $2 \times 10^6 \text{ cm}^{-3}$) in each frame.

Appendix B: The D₂H⁺/H₂D⁺ ratio

We study in this appendix the sensitivity of the para-D₂H⁺/ortho-H₂D⁺ (hereafter PDOH) ratio to the different parameter variations we have studied in the main text. The difficulty to observe those lines from the ground makes them of limited interest as a tool to study cold cores but are nevertheless interesting values when such observations have been achieved as these are basically the only H₃⁺ isotopologues observable from the ground. Their ortho and para counterparts are above 1 THz and difficult to detect due to the large energy difference between the upper and ground levels, which is incompatible with the low temperature necessary to have these species produced in large quantities. Their detection in absorption would require a strong background source, not available in all the presently known PSCs. Among the rare detections, an interesting mapping of D₂H⁺ has been reported by Parise et al. (2011).

The parameters are varied in the same order as in Sect. 5 for the N₂D⁺/N₂H⁺ modelling.

Comments on the different figures On the contrary to the N₂D⁺/N₂H⁺ ratio evolution with time, the PDOH ratio evolution is not forcibly monotonic because both species are transient from H₃⁺ to D₃⁺ and the two species are directly connected via chemical reactions (for N₂D⁺ and N₂H⁺, this is true only for reactions with H and D atoms which play a minor role). Indeed, in the reference case (Fig. B.1), this PDOH ratio first increases at all radial distances in a similar manner and then, in the slow case, drops in the inner 0.02 pc part with time. For the final density profile, the ratio is 5 to 300 times larger in the slow case than in the fast case. The latter has clearly not reached steady state.

The sensitivity to temperature is slightly larger than for the N₂D⁺/N₂H⁺ ratio in the slow case (compare Figs. B.2 & 6) in a non-obvious fashion. The sensitivity to CO abundance follows expectations (Fig. B.3). When CO abundance increases, it reacts with a larger quantity of H₂D⁺ ions, directly competing with HD to form HCO⁺ and DCO⁺ rather than D₂H⁺. It is also visible in the slow process when the CO abundance is increased.

The initial OPR ratio at collapse start has the same influence on the PDOH ratio as on the N₂D⁺/N₂H⁺ ratio (Figs. B.4 vs 9), i.e., as the initial OPR ratio is lowered, the chemistry nears the steady state more and more. The slow case is already in steady-state and does not show any change at all.

Figure B.5 (cosmic ray ionization rate dependence) shows a strongly non-linear behaviour along the core radius and when the cosmic ray ionization rate is increased. In the fast collapse

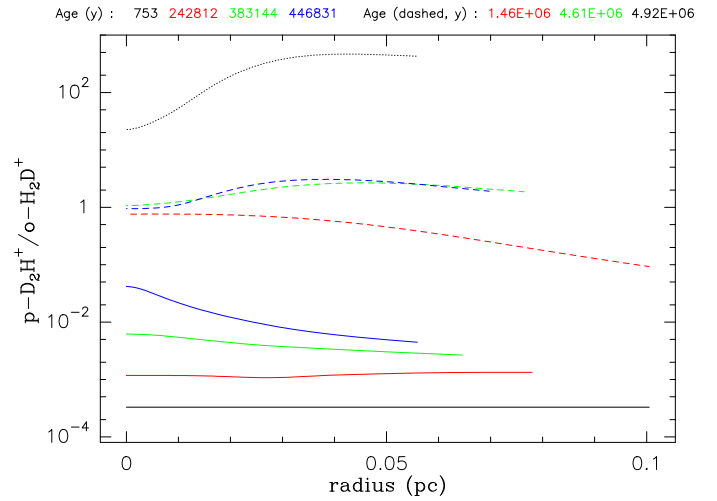


Fig. B.1. Evolution of the para-D₂H⁺/ortho-H₂D⁺ ratio with time for the slow (dashed lines) and fast (full lines) reference models. The dotted line traces the slow/fast ratio for the last epoch (corresponding to a peak density of $2 \times 10^6 \text{ cm}^{-3}$). The results are traced for the same density profiles as those of Fig. 2

case, for $\zeta = 1 \times 10^{-16} \text{ s}^{-1}$, steady state is reached when arriving at the highest density after passing by a higher ratio value in the center. The ratio has therefore not evolved monotonically.

For an average grain smaller than 0.1 μm , we find results similar to the N₂D⁺/N₂H⁺ ratio case (Figs. B.6 vs 11), with a small decrease of the deuteration efficiency for the fast case and almost no change for the slow case. For large grains, the fast case shows a strong amplification and for the slow case, the behaviour is non-linear, marking a small increase first and then a small decrease in the ratio.

The last two cases (sensitivity to the old H₂ + H⁺ reaction rate, Fig. B.7, and to variable CO depletion, Fig. B.8) show the same behaviours as those reported for the N₂D⁺/N₂H⁺ ratio cases (Figs. A.1 & 8).

Overall, in the fast collapse case, the PDOH ratio remains almost always below 1 while it varies from 1 in the centre to ~ 5 at distances around 0.03–0.05 pc away from it for the slow case but some ambiguity can appear in several cases. In both the fast and the slow cases, the variation of the PDOH ratio can be non-monotonic and therefore not a good constraint to differentiate between the models.

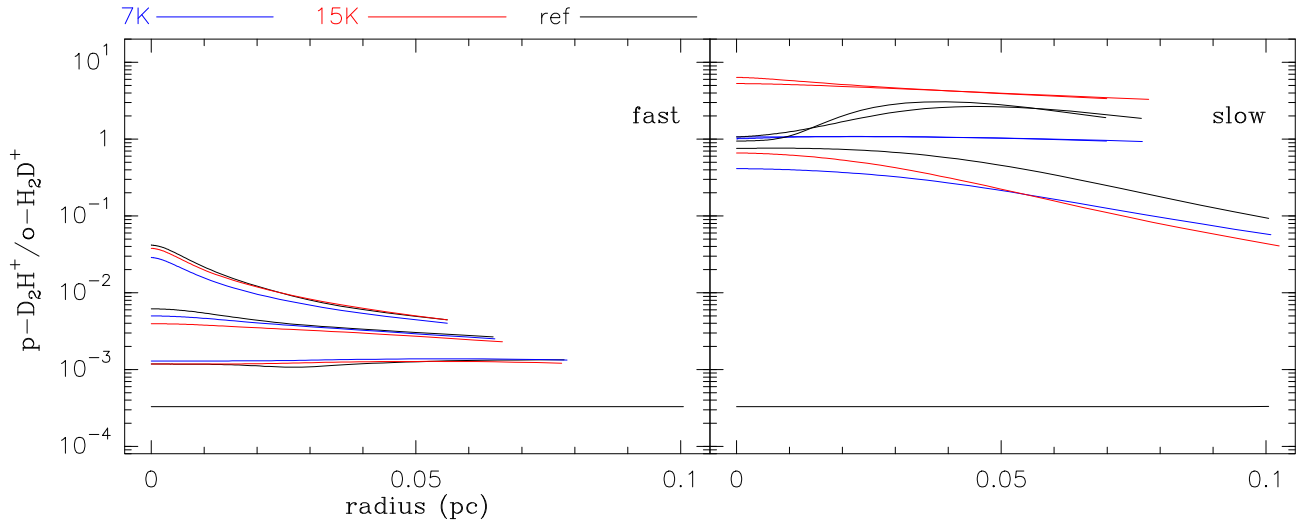


Fig. B.2. Evolution of the para-D₂H⁺/ortho-H₂D⁺ ratio with temperature for the slow (right) and fast (left) models. In the reference model, the temperature of the chemical network is the temperature of the gas. The two other cases are for constant temperature imposed to the chemical network. To better see the differences, the reference model has been drawn with the full line mode (from respectively the slow and fast cases displayed in Fig. B.1). The results are traced for the same density profiles as those of Fig. 2.

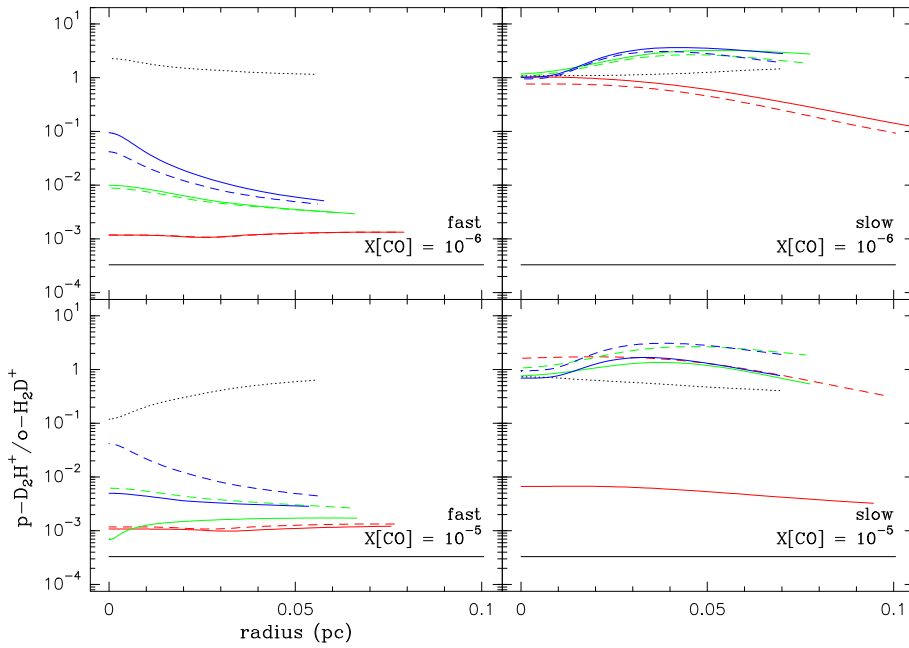


Fig. B.3. Evolution of the para-D₂H⁺/ortho-H₂D⁺ ratio with time for the slow (right) and fast (left) models as a function of CO depletion (depletion multiplied by 2 in the top row, divided by 5 in the bottom row). The dashed lines trace the reference results (from respectively the slow and fast cases displayed in Fig. B.1). The dotted line traces the new versus reference model ratio for the last epoch (corresponding to a peak density of $2 \times 10^6 \text{ cm}^{-3}$) in each frame. The results are traced for the same density profiles as those of Fig. 2.

Appendix C: Kooij/Arrhenius expression of the Hugo H₃⁺ + H₂ isotopologues reaction rates and of the H₃⁺ isotopologues dissociative recombination rates²

Hugo et al. (2009) fit their H₃⁺ + H₂ isotopologues reactions rates tables with the Arrhenius formula ($K = \alpha e^{-\beta/T} \text{ cm}^3 \text{ s}^{-1}$). However, they have validated their fits only in the 5–20 K temperature range. By employing the more complete Kooij/Arrhenius formula ($K = \alpha(T/300\text{K})^\beta e^{-\gamma/T} \text{ cm}^3 \text{ s}^{-1}$) and adjusting the coefficients by hand, we can provide a fit extending up to 50 K, the maximum temperature given in Hugo et al. (2009) tables, to a sufficient precision ($< \pm 10\%$). To compute

² If authors want to use these parameterized rates, they should still quote the original works by Hugo et al. (2009) and P09 for the two sets, respectively.

the Kooij/Arrhenius coefficients, we have loaded the tabulated values (between 5 and 50 K), and fitted by eye the first two terms of the formula (α, β) after multiplying the rate by $e^{\gamma/T}$ for the endothermic reactions. We have minimized the errors to keep them to a few % only over all the temperature range. Figs. C.1–C.3 show two examples of a fit. Figs. C.1–C.2 show the fitting of an endothermic reaction. At very low temperatures, the endothermic reaction coefficients become low enough to drop below 10^{-20} and are zeroed in Hugo et al. (2009) reaction files. These values therefore remain zeroed when corrected for the endothermicity (Fig. C.1) and cannot be recovered and used for the fit. The fit extrapolation is however good enough and its possible error is largely attenuated by the strong endothermic correction (Fig. C.2) which makes the rate drop quickly. In some cases, the Kooij formula cannot fit the data. Fig. C.3 shows such a case for an exothermic reaction. The fit is minimized so that the maximum error be the same in the middle and at the two

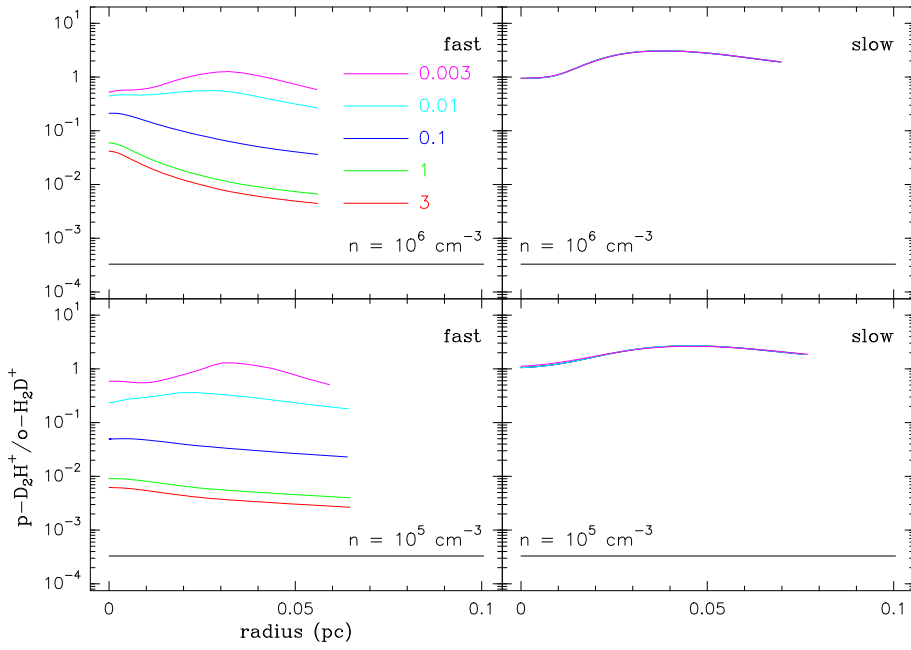


Fig. B.4. Evolution of the para-D₂H⁺/ortho-H₂D⁺ ratio with the H₂ OPR for the slow (right) and fast (left) models. The bottom row corresponds to the density profile with peak density of 10⁵ cm⁻³ and the top row to the profile with peak density of 2 × 10⁶ cm⁻³ (from Fig. 2). OPR values are listed in the top left image. The dashed lines trace the reference results (from Fig. B.1).

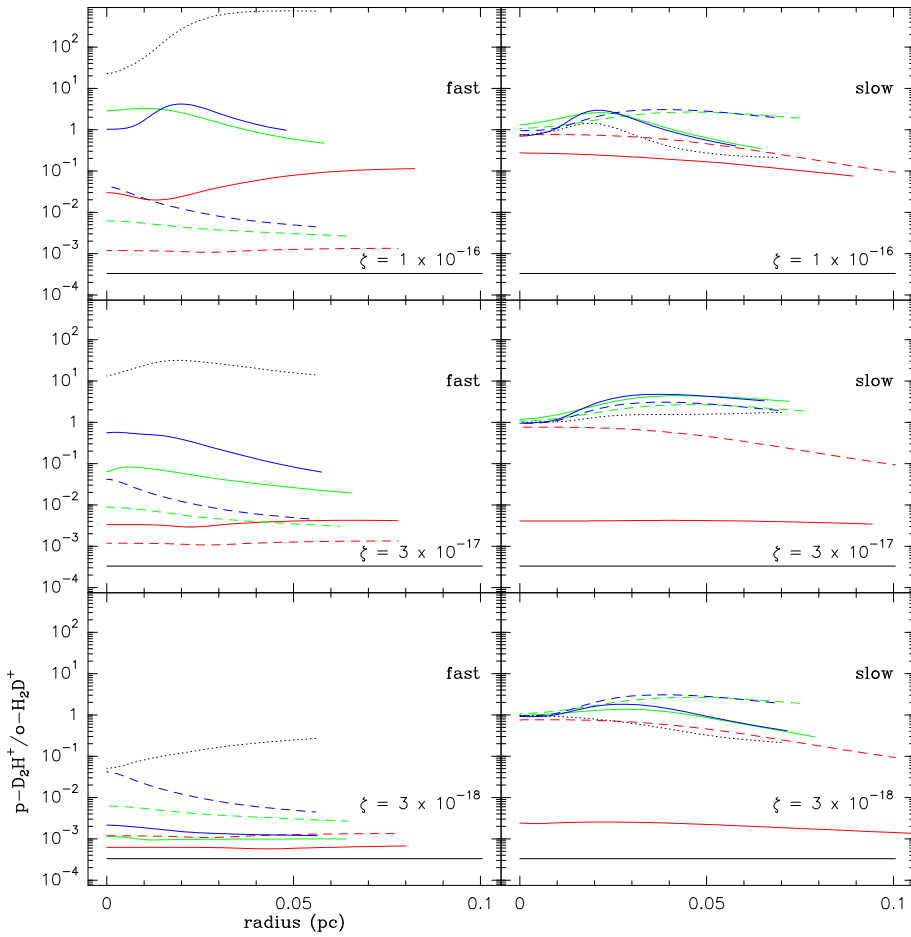


Fig. B.5. Evolution of the para-D₂H⁺/ortho-H₂D⁺ ratio when the cosmic ray ionization rate is multiplied by 3 (bottom row) and by 10 (top row) for the slow (right) and fast (left) models. The dotted line traces the new versus reference model ratio for the last epoch (corresponding to a peak density of 2 × 10⁶ cm⁻³) in each frame. The dashed lines trace the reference results (from Fig. B.1).

ends of the slope. The error remains lower than $\pm 10\%$ for all rates within the 5-50K range compared to the tabulated values. The Kooij/Arrhenius coefficients are given in Table C.1.

In Table C.2, we present the Kooij/Arrhenius coefficients of the dissociative recombination rates of the H₃⁺ isotopologues from P09.

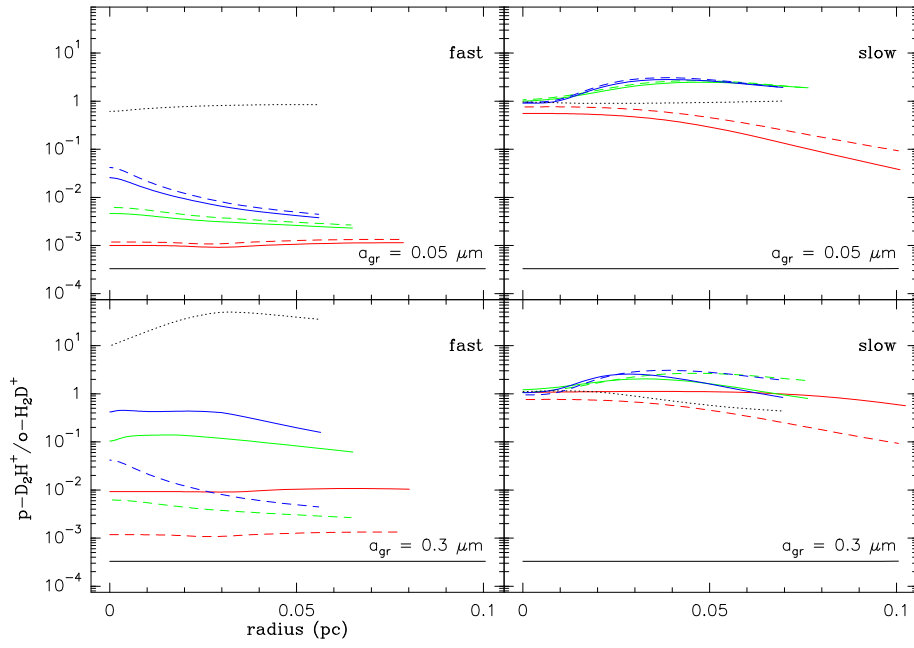


Fig. B.6. Evolution of the para-D₂H⁺/ortho-H₂D⁺ ratio when the grain size is multiplied by 3 (bottom row) or divided by 2 (top row) for the slow (right) and fast (left) models. The dashed lines trace the reference results (from Fig. B.1). The dotted line traces the new versus reference model ratio for the last epoch (corresponding to a peak density of $2 \times 10^6 \text{ cm}^{-3}$) in each frame.

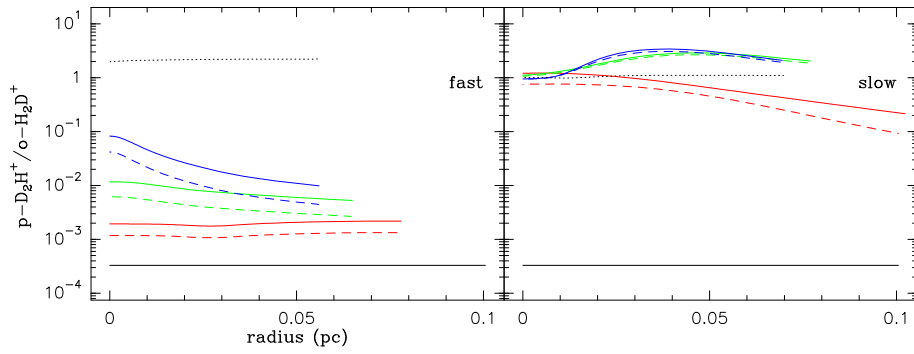


Fig. B.7. Evolution of the para-D₂H⁺/ortho-H₂D⁺ ratio with the old H₂ + H⁺, for the slow (right) and fast (left) models. The dashed lines trace the reference results (from Fig. B.1). The dotted line traces the new versus reference model ratio for the last epoch (corresponding to a peak density of $2 \times 10^6 \text{ cm}^{-3}$) in each frame.

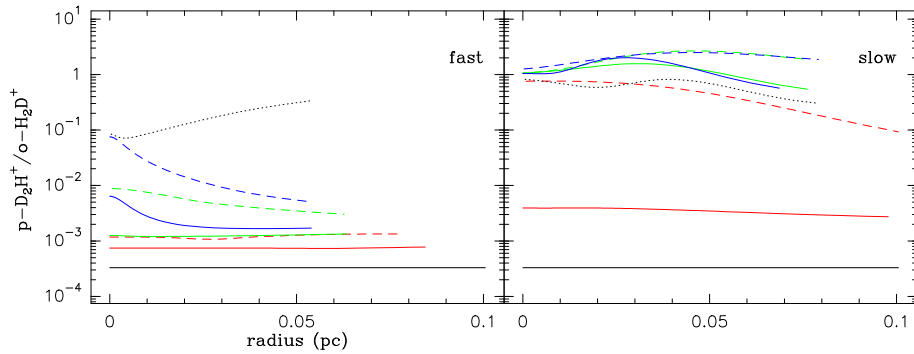


Fig. B.8. Evolution of the para-D₂H⁺/ortho-H₂D⁺ ratio when CO depletion is varying compared to constant depletion, for the slow (right) and fast (left) models. The dashed lines trace the reference results (from Fig. B.1) and the dotted line traces the new versus reference model ratio for the last epoch (corresponding to a peak density of $2 \times 10^6 \text{ cm}^{-3}$) in each frame.

Table C.1. Kooij/Arrhenius coefficients of the H₃⁺ + H₂ isotopologues reactions.

Reaction							Rate ^a		
							α	β	γ
							(cm ³ s ⁻¹)		(K)
m-D ₃ ⁺	+	HD	→	p-D ₂ H ⁺	+	o-D ₂	1.15e-10	4.00e-02	2.06e+02
m-D ₃ ⁺	+	HD	→	o-D ₂ H ⁺	+	p-D ₂	9.62e-11	1.70e-01	2.41e+02
m-D ₃ ⁺	+	HD	→	o-D ₂ H ⁺	+	o-D ₂	2.32e-10	8.00e-02	1.56e+02
m-D ₃ ⁺	+	HD	→	o-D ₃ ⁺	+	HD	5.54e-10	3.00e-02	4.65e+01
m-D ₃ ⁺	+	o-D ₂	→	o-D ₃ ⁺	+	p-D ₂	1.57e-10	5.00e-02	1.32e+02
m-D ₃ ⁺	+	o-D ₂	→	o-D ₃ ⁺	+	o-D ₂	5.34e-10	9.00e-02	4.65e+01

Table C.1. continued.

Reaction						Rate ^a		
						α	β	γ
						(cm ³ s ⁻¹)		(K)
m-D ₃ ⁺	+	o-D ₂	→	m-D ₃ ⁺	+ p-D ₂	1.37e-10	1.70e-01	8.58e+01
m-D ₃ ⁺	+	o-H ₂	→	o-H ₂ D ⁺	+ o-D ₂	2.12e-10	1.00e-01	2.58e+02
m-D ₃ ⁺	+	o-H ₂	→	o-D ₂ H ⁺	+ HD	1.47e-09	0.0	6.43e+01
m-D ₃ ⁺	+	p-D ₂	→	p-D ₃ ⁺	+ o-D ₂	4.21e-11	2.50e-01	0.0
m-D ₃ ⁺	+	p-D ₂	→	o-D ₃ ⁺	+ p-D ₂	1.12e-10	0.0	4.65e+01
m-D ₃ ⁺	+	p-D ₂	→	o-D ₃ ⁺	+ o-D ₂	8.46e-10	1.20e-01	0.0
m-D ₃ ⁺	+	p-D ₂	→	m-D ₃ ⁺	+ o-D ₂	1.50e-10	0.0	0.0
o-D ₂ H ⁺	+	HD	→	p-H ₂ D ⁺	+ p-D ₂	8.80e-12	1.60e-01	1.94e+02
o-D ₂ H ⁺	+	HD	→	p-H ₂ D ⁺	+ o-D ₂	2.03e-11	2.50e-02	1.08e+02
o-D ₂ H ⁺	+	HD	→	o-H ₂ D ⁺	+ p-D ₂	4.05e-11	1.20e-01	2.80e+02
o-D ₂ H ⁺	+	HD	→	o-H ₂ D ⁺	+ o-D ₂	9.00e-11	1.00e-01	1.94e+02
o-D ₂ H ⁺	+	HD	→	p-D ₂ H ⁺	+ HD	2.53e-10	9.00e-02	5.02e+01
o-D ₂ H ⁺	+	HD	→	o-D ₃ ⁺	+ p-H ₂	7.69e-11	1.10e-01	0.0
o-D ₂ H ⁺	+	HD	→	o-D ₃ ⁺	+ o-H ₂	1.91e-10	1.20e-01	0.0
o-D ₂ H ⁺	+	HD	→	m-D ₃ ⁺	+ p-H ₂	1.04e-10	8.00e-02	0.0
o-D ₂ H ⁺	+	HD	→	m-D ₃ ⁺	+ o-H ₂	1.63e-10	7.00e-02	0.0
o-D ₂ H ⁺	+	o-D ₂	→	p-D ₂ H ⁺	+ p-D ₂	6.02e-11	4.00e-02	1.36e+02
o-D ₂ H ⁺	+	o-D ₂	→	p-D ₂ H ⁺	+ o-D ₂	7.86e-11	2.50e-02	5.02e+01
o-D ₂ H ⁺	+	o-D ₂	→	o-D ₂ H ⁺	+ p-D ₂	4.60e-11	0.0	8.58e+01
o-D ₂ H ⁺	+	o-D ₂	→	o-D ₃ ⁺	+ HD	5.50e-11	0.0	0.0
o-D ₂ H ⁺	+	o-D ₂	→	m-D ₃ ⁺	+ HD	5.85e-10	0.0	0.0
o-D ₂ H ⁺	+	o-H ₂	→	p-H ₃ ⁺	+ o-D ₂	5.59e-12	1.40e-01	1.70e+02
o-D ₂ H ⁺	+	o-H ₂	→	o-H ₃ ⁺	+ o-D ₂	1.41e-11	1.80e-01	2.02e+02
o-D ₂ H ⁺	+	o-H ₂	→	p-H ₂ D ⁺	+ HD	9.00e-11	0.0	1.66e+01
o-D ₂ H ⁺	+	o-H ₂	→	o-H ₂ D ⁺	+ HD	8.40e-10	8.00e-02	1.03e+02
o-D ₂ H ⁺	+	o-H ₂	→	o-D ₂ H ⁺	+ p-H ₂	3.31e-10	7.70e-02	0.0
o-D ₂ H ⁺	+	p-D ₂	→	p-D ₂ H ⁺	+ p-D ₂	3.75e-11	0.0	5.02e+01
o-D ₂ H ⁺	+	p-D ₂	→	p-D ₂ H ⁺	+ o-D ₂	1.56e-10	8.50e-02	0.0
o-D ₂ H ⁺	+	p-D ₂	→	o-D ₂ H ⁺	+ o-D ₂	1.16E-10	0.0	0.0
o-D ₂ H ⁺	+	p-D ₂	→	p-D ₃ ⁺	+ HD	5.05E-11	1.20e-01	0.0
o-D ₂ H ⁺	+	p-D ₂	→	o-D ₃ ⁺	+ HD	7.74E-10	8.00e-02	0.0
o-D ₂ H ⁺	+	p-D ₂	→	m-D ₃ ⁺	+ HD	2.22E-10	9.00e-02	0.0
o-D ₂ H ⁺	+	p-H ₂	→	p-H ₃ ⁺	+ o-D ₂	1.65E-11	1.50e-01	3.40e+02
o-D ₂ H ⁺	+	p-H ₂	→	p-H ₂ D ⁺	+ HD	3.43E-10	8.00e-02	1.87e+02
o-D ₂ H ⁺	+	p-H ₂	→	o-H ₂ D ⁺	+ HD	4.87E-10	1.30e-01	2.73e+02
o-D ₂ H ⁺	+	p-H ₂	→	o-D ₂ H ⁺	+ o-H ₂	3.73E-10	4.00e-02	1.70e+02
o-D ₃ ⁺	+	HD	→	p-D ₂ H ⁺	+ p-D ₂	1.53E-10	8.00e-02	2.45e+02
o-D ₃ ⁺	+	HD	→	p-D ₂ H ⁺	+ o-D ₂	1.69e-10	3.00e-02	1.59e+02
o-D ₃ ⁺	+	HD	→	o-D ₂ H ⁺	+ p-D ₂	1.38e-10	1.50e-01	1.95e+02
o-D ₃ ⁺	+	HD	→	o-D ₂ H ⁺	+ o-D ₂	9.96e-11	1.50e-01	1.09e+02
o-D ₃ ⁺	+	HD	→	p-D ₃ ⁺	+ HD	7.38e-11	8.00e-02	1.62e+01
o-D ₃ ⁺	+	HD	→	m-D ₃ ⁺	+ HD	2.35e-10	2.00e-02	0.0
o-D ₃ ⁺	+	o-D ₂	→	p-D ₃ ⁺	+ p-D ₂	2.05e-11	1.20e-01	1.02e+02
o-D ₃ ⁺	+	o-D ₂	→	p-D ₃ ⁺	+ o-D ₂	4.51e-11	1.80e-01	1.62e+01
o-D ₃ ⁺	+	o-D ₂	→	o-D ₃ ⁺	+ p-D ₂	3.03E-10	3.50e-02	8.58e+01
o-D ₃ ⁺	+	o-D ₂	→	m-D ₃ ⁺	+ p-D ₂	2.10e-10	1.60e-01	3.93e+01
o-D ₃ ⁺	+	o-D ₂	→	m-D ₃ ⁺	+ o-D ₂	1.99e-10	1.20e-01	0.0
o-D ₃ ⁺	+	o-H ₂	→	o-H ₂ D ⁺	+ p-D ₂	1.78e-10	1.50e-01	2.98e+02
o-D ₃ ⁺	+	o-H ₂	→	o-H ₂ D ⁺	+ o-D ₂	1.09e-10	8.00e-02	2.12e+02
o-D ₃ ⁺	+	o-H ₂	→	p-D ₂ H ⁺	+ HD	7.17e-10	1.20e-01	6.80e+01
o-D ₃ ⁺	+	o-H ₂	→	o-D ₂ H ⁺	+ HD	5.65e-10	1.20e-01	1.78e+01
o-H ₂ D ⁺	+	HD	→	p-H ₃ ⁺	+ p-D ₂	2.27e-12	8.00e-02	1.52e+02
o-H ₂ D ⁺	+	HD	→	p-H ₃ ⁺	+ o-D ₂	2.71e-12	1.50e-01	6.66e+01
o-H ₂ D ⁺	+	HD	→	o-H ₃ ⁺	+ p-D ₂	6.63e-12	6.00e-02	1.85e+02
o-H ₂ D ⁺	+	HD	→	o-H ₃ ⁺	+ o-D ₂	7.52e-12	1.60e-01	9.94e+01
o-H ₂ D ⁺	+	HD	→	p-H ₂ D ⁺	+ HD	9.15e-11	0.0	0.0
o-H ₂ D ⁺	+	HD	→	p-D ₂ H ⁺	+ p-H ₂	6.13e-11	0.0	0.0
o-H ₂ D ⁺	+	HD	→	p-D ₂ H ⁺	+ o-H ₂	2.55e-10	7.00e-02	0.0
o-H ₂ D ⁺	+	HD	→	o-D ₂ H ⁺	+ p-H ₂	1.16e-10	8.00e-02	0.0
o-H ₂ D ⁺	+	HD	→	o-D ₂ H ⁺	+ o-H ₂	4.72e-10	5.00e-02	0.0
o-H ₂ D ⁺	+	o-D ₂	→	o-H ₂ D ⁺	+ p-D ₂	4.91e-11	1.30e-01	8.58e+01
o-H ₂ D ⁺	+	o-D ₂	→	p-D ₂ H ⁺	+ HD	2.60e-10	3.00e-02	0.0
o-H ₂ D ⁺	+	o-D ₂	→	o-D ₂ H ⁺	+ HD	9.48e-10	4.00e-02	0.0
o-H ₂ D ⁺	+	o-D ₂	→	o-D ₃ ⁺	+ o-H ₂	6.34e-11	1.00e-01	0.0
o-H ₂ D ⁺	+	o-D ₂	→	m-D ₃ ⁺	+ o-H ₂	9.88e-11	2.00e-01	0.0

Table C.1. continued.

Reaction						Rate ^a		
						α	β	γ
						(cm ³ s ⁻¹)		(K)
o-H ₂ D ⁺	+	o-H ₂	→	p-H ₃ ⁺	+ HD	4.65e-11	2.50e-02	0.0
o-H ₂ D ⁺	+	o-H ₂	→	o-H ₃ ⁺	+ HD	1.21e-10	1.50e-01	8.20e+00
o-H ₂ D ⁺	+	o-H ₂	→	p-H ₂ D ⁺	+ p-H ₂	6.98e-11	8.00e-02	0.0
o-H ₂ D ⁺	+	o-H ₂	→	p-H ₂ D ⁺	+ o-H ₂	1.39e-10	8.00e-02	0.0
o-H ₂ D ⁺	+	o-H ₂	→	o-H ₂ D ⁺	+ p-H ₂	1.69e-10	1.00e-01	0.0
o-H ₂ D ⁺	+	p-D ₂	→	o-H ₂ D ⁺	+ o-D ₂	5.54e-11	1.00e-01	0.0
o-H ₂ D ⁺	+	p-D ₂	→	p-D ₂ H ⁺	+ HD	6.5e-10	0.0	0.0
o-H ₂ D ⁺	+	p-D ₂	→	o-D ₂ H ⁺	+ HD	5.01e-10	3.50e-02	0.0
o-H ₂ D ⁺	+	p-D ₂	→	p-D ₃ ⁺	+ o-H ₂	1.39e-11	4.00e-02	0.0
o-H ₂ D ⁺	+	p-D ₂	→	o-D ₃ ⁺	+ o-H ₂	1.30e-10	0.0	0.0
o-H ₂ D ⁺	+	p-H ₂	→	p-H ₃ ⁺	+ HD	1.06e-10	1.70e-01	1.45e+02
o-H ₂ D ⁺	+	p-H ₂	→	o-H ₃ ⁺	+ HD	9.37e-11	0.0	1.78e+02
o-H ₂ D ⁺	+	p-H ₂	→	p-H ₂ D ⁺	+ o-H ₂	4.96e-10	6.00e-02	8.36e+01
o-H ₂ D ⁺	+	p-H ₂	→	o-H ₂ D ⁺	+ o-H ₂	7.85e-10	1.40e-01	1.70e+02
o-H ₃ ⁺	+	HD	→	p-H ₃ ⁺	+ HD	3.52e-11	5.00e-02	0.0
o-H ₃ ⁺	+	HD	→	p-H ₂ D ⁺	+ o-H ₂	1.58e-10	3.50e-02	0.0
o-H ₃ ⁺	+	HD	→	o-H ₂ D ⁺	+ p-H ₂	1.85e-10	7.00e-02	0.0
o-H ₃ ⁺	+	HD	→	o-H ₂ D ⁺	+ o-H ₂	1.19e-09	3.00e-02	0.0
o-H ₃ ⁺	+	o-D ₂	→	o-H ₂ D ⁺	+ HD	1.12e-09	2.80e-01	0.0
o-H ₃ ⁺	+	o-D ₂	→	o-D ₂ H ⁺	+ o-H ₂	7.08e-10	1.40e-01	0.0
o-H ₃ ⁺	+	o-H ₂	→	p-H ₃ ⁺	+ p-H ₂	1.17e-10	3.00e-02	0.0
o-H ₃ ⁺	+	o-H ₂	→	p-H ₃ ⁺	+ o-H ₂	4.05e-10	0.0	0.0
o-H ₃ ⁺	+	o-H ₂	→	o-H ₃ ⁺	+ p-H ₂	9.58e-11	7.00e-03	0.0
o-H ₃ ⁺	+	p-D ₂	→	o-H ₂ D ⁺	+ HD	8.78e-10	3.00e-02	0.0
o-H ₃ ⁺	+	p-D ₂	→	p-D ₂ H ⁺	+ o-H ₂	6.86e-10	3.00e-02	0.0
o-H ₃ ⁺	+	p-H ₂	→	p-H ₃ ⁺	+ o-H ₂	9.78e-10	1.70e-01	1.37e+02
o-H ₃ ⁺	+	p-H ₂	→	o-H ₃ ⁺	+ o-H ₂	7.78e-10	5.00e-02	1.70e+02
p-D ₂ H ⁺	+	HD	→	p-H ₂ D ⁺	+ p-D ₂	1.91e-11	1.00e-01	1.44e+02
p-D ₂ H ⁺	+	HD	→	p-H ₂ D ⁺	+ o-D ₂	1.22e-11	0.0	5.76e+01
p-D ₂ H ⁺	+	HD	→	o-H ₂ D ⁺	+ p-D ₂	1.06e-10	1.50e-01	2.30e+02
p-D ₂ H ⁺	+	HD	→	o-H ₂ D ⁺	+ o-D ₂	5.21e-11	8.00e-02	1.44e+02
p-D ₂ H ⁺	+	HD	→	o-D ₂ H ⁺	+ HD	6.03e-10	2.30e-01	0.0
p-D ₂ H ⁺	+	HD	→	p-D ₃ ⁺	+ p-H ₂	9.50e-12	1.50e-01	0.0
p-D ₂ H ⁺	+	HD	→	p-D ₃ ⁺	+ o-H ₂	2.86e-11	1.20e-01	0.0
p-D ₂ H ⁺	+	HD	→	o-D ₃ ⁺	+ p-H ₂	7.82e-11	1.20e-01	0.0
p-D ₂ H ⁺	+	HD	→	o-D ₃ ⁺	+ o-H ₂	3.08e-10	3.00e-02	0.0
p-D ₂ H ⁺	+	o-D ₂	→	p-D ₂ H ⁺	+ p-D ₂	4.40e-11	6.00e-02	8.58e+01
p-D ₂ H ⁺	+	o-D ₂	→	o-D ₂ H ⁺	+ p-D ₂	1.12e-10	0.0	3.56e+01
p-D ₂ H ⁺	+	o-D ₂	→	o-D ₂ H ⁺	+ o-D ₂	1.86e-10	3.20e-01	0.0
p-D ₂ H ⁺	+	o-D ₂	→	p-D ₃ ⁺	+ HD	6.57e-11	3.00e-02	0.0
p-D ₂ H ⁺	+	o-D ₂	→	o-D ₃ ⁺	+ HD	6.09e-10	8.00e-02	0.0
p-D ₂ H ⁺	+	o-D ₂	→	m-D ₃ ⁺	+ HD	2.85e-10	0.0	0.0
p-D ₂ H ⁺	+	o-H ₂	→	p-H ₃ ⁺	+ p-D ₂	8.97e-12	8.00e-02	2.05e+02
p-D ₂ H ⁺	+	o-H ₂	→	o-H ₃ ⁺	+ p-D ₂	2.67e-11	5.00e-02	2.38e+02
p-D ₂ H ⁺	+	o-H ₂	→	p-H ₂ D ⁺	+ HD	1.37e-10	2.30e-01	0.0
p-D ₂ H ⁺	+	o-H ₂	→	o-H ₂ D ⁺	+ HD	7.80e-10	1.30e-01	5.28e+01
p-D ₂ H ⁺	+	o-H ₂	→	p-D ₂ H ⁺	+ p-H ₂	1.95e-10	1.30e-01	0.0
p-D ₂ H ⁺	+	p-D ₂	→	p-D ₂ H ⁺	+ o-D ₂	5.45e-11	2.00e-02	0.0
p-D ₂ H ⁺	+	p-D ₂	→	o-D ₂ H ⁺	+ p-D ₂	1.00e-10	3.50e-01	0.0
p-D ₂ H ⁺	+	p-D ₂	→	o-D ₂ H ⁺	+ o-D ₂	1.81e-10	1.10e-01	0.0
p-D ₂ H ⁺	+	p-D ₂	→	p-D ₃ ⁺	+ HD	7.11e-11	1.00e-01	0.0
p-D ₂ H ⁺	+	p-D ₂	→	o-D ₃ ⁺	+ HD	7.06e-10	5.00e-02	0.0
p-D ₂ H ⁺	+	p-H ₂	→	p-H ₃ ⁺	+ p-D ₂	2.80e-11	6.00e-02	3.75e+02
p-D ₂ H ⁺	+	p-H ₂	→	p-H ₂ D ⁺	+ HD	3.22e-10	0.0	1.36e+02
p-D ₂ H ⁺	+	p-H ₂	→	o-H ₂ D ⁺	+ HD	5.11e-10	1.70e-01	2.23e+02
p-D ₂ H ⁺	+	p-H ₂	→	p-D ₂ H ⁺	+ o-H ₂	6.08e-10	8.00e-02	1.70e+02
p-D ₃ ⁺	+	HD	→	p-D ₂ H ⁺	+ p-D ₂	3.22e-10	8.00e-02	2.29e+02
p-D ₃ ⁺	+	HD	→	p-D ₂ H ⁺	+ o-D ₂	2.46e-10	7.00e-02	1.43e+02
p-D ₃ ⁺	+	HD	→	o-D ₂ H ⁺	+ p-D ₂	1.76e-10	1.00e-01	1.79e+02
p-D ₃ ⁺	+	HD	→	o-D ₃ ⁺	+ HD	9.28e-10	5.00e-02	0.0
p-D ₃ ⁺	+	o-D ₂	→	o-D ₃ ⁺	+ p-D ₂	5.59e-10	0.0	6.96e+01
p-D ₃ ⁺	+	o-D ₂	→	o-D ₃ ⁺	+ o-D ₂	4.80e-10	1.00e-01	0.0
p-D ₃ ⁺	+	o-D ₂	→	m-D ₃ ⁺	+ p-D ₂	2.51e-10	1.70e-01	2.31e+01
p-D ₃ ⁺	+	o-H ₂	→	o-H ₂ D ⁺	+ p-D ₂	3.51e-10	1.40e-01	2.82e+02
p-D ₃ ⁺	+	o-H ₂	→	p-D ₂ H ⁺	+ HD	1.65e-09	2.00e-02	5.17e+01

Table C.1. continued.

Reaction						Rate ^a			
						α	β	γ	
						(cm ³ s ⁻¹)		(K)	
p-D ₃ ⁺	+	p-D ₂	→	o-D ₃ ⁺	+	p-D ₂	5.83e-10	6.00e-02	0.0
p-D ₃ ⁺	+	p-D ₂	→	o-D ₃ ⁺	+	o-D ₂	4.98e-10	3.00e-02	0.0
p-D ₃ ⁺	+	p-H ₂	→	p-H ₂ D ⁺	+	p-D ₂	2.60e-10	2.50e-01	3.65e+02
p-D ₃ ⁺	+	p-H ₂	→	p-D ₂ H ⁺	+	HD	1.88e-09	7.00e-02	2.22e+02
p-H ₂ D ⁺	+	HD	→	p-H ₃ ⁺	+	p-D ₂	7.10e-12	7.00e-02	2.39e+02
p-H ₂ D ⁺	+	HD	→	p-H ₃ ⁺	+	o-D ₂	8.53e-12	1.40e-01	1.53e+02
p-H ₂ D ⁺	+	HD	→	o-H ₂ D ⁺	+	HD	3.75e-10	1.40e-01	8.64e+01
p-H ₂ D ⁺	+	HD	→	p-D ₂ H ⁺	+	p-H ₂	2.57e-10	1.50e-01	0.0
p-H ₂ D ⁺	+	HD	→	p-D ₂ H ⁺	+	o-H ₂	1.60e-10	6.00e-02	3.36e+01
p-H ₂ D ⁺	+	HD	→	o-D ₂ H ⁺	+	p-H ₂	6.32e-10	3.00e-02	0.0
p-H ₂ D ⁺	+	HD	→	o-D ₂ H ⁺	+	o-H ₂	2.10e-10	0.0	0.0
p-H ₂ D ⁺	+	o-D ₂	→	p-H ₂ D ⁺	+	p-D ₂	2.83e-11	1.00e-01	8.58e+01
p-H ₂ D ⁺	+	o-D ₂	→	p-D ₂ H ⁺	+	HD	2.31e-10	5.00e-02	0.0
p-H ₂ D ⁺	+	o-D ₂	→	o-D ₂ H ⁺	+	HD	9.88e-10	1.00e-01	0.0
p-H ₂ D ⁺	+	o-D ₂	→	o-D ₃ ⁺	+	p-H ₂	1.19e-10	8.00e-02	0.0
p-H ₂ D ⁺	+	o-D ₂	→	m-D ₃ ⁺	+	p-H ₂	1.34e-10	1.70e-01	0.0
p-H ₂ D ⁺	+	o-H ₂	→	p-H ₃ ⁺	+	HD	9.98e-11	2.00e-01	6.18e+01
p-H ₂ D ⁺	+	o-H ₂	→	o-H ₃ ⁺	+	HD	9.18e-11	1.00e-02	9.46e+01
p-H ₂ D ⁺	+	o-H ₂	→	o-H ₂ D ⁺	+	p-H ₂	1.06e-09	5.00e-02	0.0
p-H ₂ D ⁺	+	o-H ₂	→	o-H ₂ D ⁺	+	o-H ₂	7.81e-10	1.40e-01	8.64e+01
p-H ₂ D ⁺	+	p-D ₂	→	p-H ₂ D ⁺	+	o-D ₂	3.71e-11	1.10e-01	0.0
p-H ₂ D ⁺	+	p-D ₂	→	p-D ₂ H ⁺	+	HD	6.92e-10	0.0	0.0
p-H ₂ D ⁺	+	p-D ₂	→	o-D ₂ H ⁺	+	HD	5.82e-10	5.00e-02	0.0
p-H ₂ D ⁺	+	p-D ₂	→	p-D ₃ ⁺	+	p-H ₂	1.18e-11	1.30e-01	0.0
p-H ₂ D ⁺	+	p-D ₂	→	o-D ₃ ⁺	+	p-H ₂	9.82e-11	1.40e-01	0.0
p-H ₂ D ⁺	+	p-H ₂	→	p-H ₃ ⁺	+	HD	1.92e-10	2.00e-01	2.32e+02
p-H ₂ D ⁺	+	p-H ₂	→	o-H ₂ D ⁺	+	o-H ₂	1.05e-09	1.00e-02	2.56e+02
p-H ₃ ⁺	+	HD	→	o-H ₃ ⁺	+	HD	5.50e-11	3.00e-02	3.29e+01
p-H ₃ ⁺	+	HD	→	p-H ₂ D ⁺	+	p-H ₂	1.91e-10	1.60e-01	0.0
p-H ₃ ⁺	+	HD	→	p-H ₂ D ⁺	+	o-H ₂	3.88e-10	3.00e-02	0.0
p-H ₃ ⁺	+	HD	→	o-H ₂ D ⁺	+	p-H ₂	4.23e-10	1.40e-01	0.0
p-H ₃ ⁺	+	HD	→	o-H ₂ D ⁺	+	o-H ₂	6.98e-10	9.00e-02	2.46e+01
p-H ₃ ⁺	+	o-D ₂	→	p-H ₂ D ⁺	+	HD	4.33e-10	1.20e-01	0.0
p-H ₃ ⁺	+	o-D ₂	→	o-H ₂ D ⁺	+	HD	3.56e-10	1.10e-01	0.0
p-H ₃ ⁺	+	o-D ₂	→	o-D ₂ H ⁺	+	p-H ₂	3.56e-10	1.00e-01	0.0
p-H ₃ ⁺	+	o-D ₂	→	o-D ₂ H ⁺	+	o-H ₂	4.65e-10	4.00e-02	0.0
p-H ₃ ⁺	+	o-H ₂	→	p-H ₃ ⁺	+	p-H ₂	2.21e-10	1.10e-01	0.0
p-H ₃ ⁺	+	o-H ₂	→	o-H ₃ ⁺	+	p-H ₂	2.21e-10	1.50e-01	0.0
p-H ₃ ⁺	+	o-H ₂	→	o-H ₃ ⁺	+	o-H ₂	6.97e-10	5.00e-02	3.29e+01
p-H ₃ ⁺	+	p-D ₂	→	p-H ₂ D ⁺	+	HD	3.65e-10	0.0	0.0
p-H ₃ ⁺	+	p-D ₂	→	o-H ₂ D ⁺	+	HD	4.92e-10	1.00e-02	0.0
p-H ₃ ⁺	+	p-D ₂	→	p-D ₂ H ⁺	+	p-H ₂	2.81e-10	2.50e-02	0.0
p-H ₃ ⁺	+	p-D ₂	→	p-D ₂ H ⁺	+	o-H ₂	4.46e-10	5.00e-02	0.0
p-H ₃ ⁺	+	p-H ₂	→	p-H ₃ ⁺	+	o-H ₂	7.47e-10	1.30e-01	1.70e+02
p-H ₃ ⁺	+	p-H ₂	→	o-H ₃ ⁺	+	o-H ₂	5.55e-10	9.00e-02	2.03e+02

Notes. Tabulated rates from (Hugo et al. 2009)^(a) The fit is valid in the 5-50 K range.

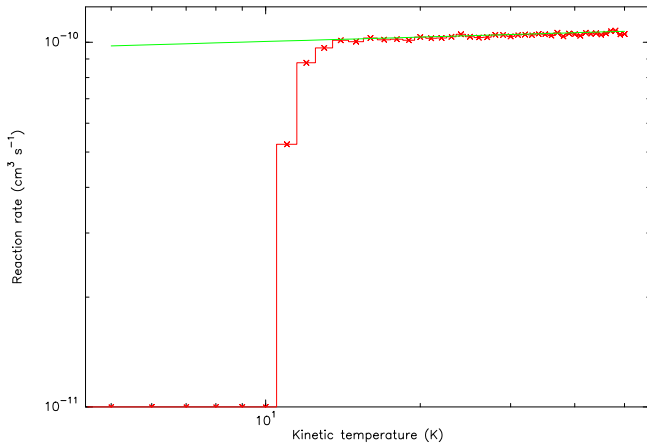


Fig. C.1. Fit for the reaction $m\text{-D}_3^+ + \text{HD} \rightarrow p\text{-D}_2\text{H}^+ + o\text{-D}_2$. The reaction being endothermic by 206 K, we have multiplied the coefficients by $e^{206/T_{\text{kin}}}$. For the lowest temperatures the original reaction rates were zeroed and cannot be retrieved. The first non-zeroed rates are not fitted because the endothermic correction will be dominant (see Fig. C.2). Data are in red, fit in green. The fit gives the first two coefficients, α and β .

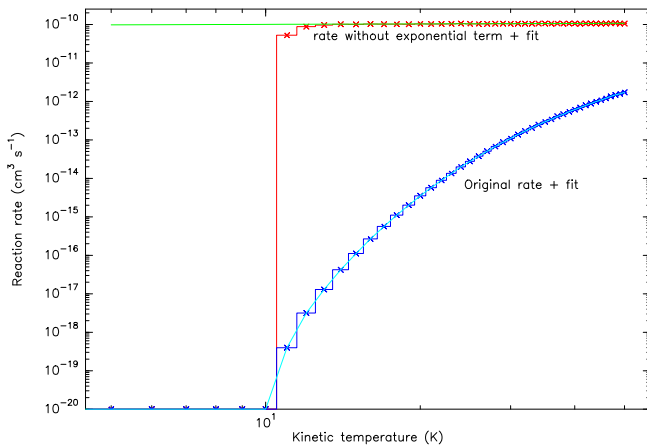


Fig. C.2. Same as Fig. C.1 but with a different Y-scale. The original rate is plotted in dark blue and the fit in cyan is the green fit equation with the endothermic correction ($e^{-206/T_{\text{kin}}}$) reintroduced. The fit is good even for the departing first non-zeroed rates at low T_{kin} .

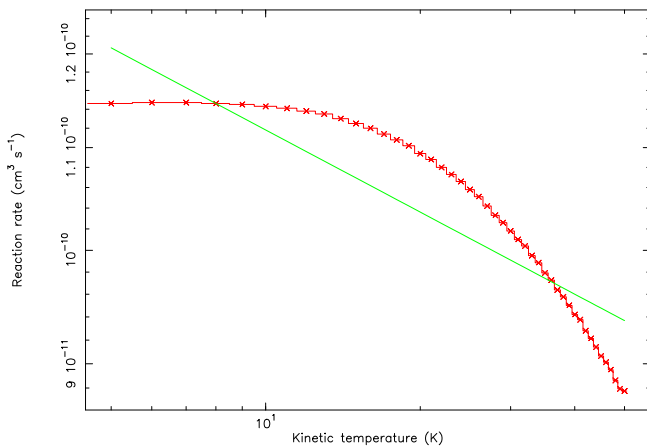


Fig. C.3. Fit for the reaction $o\text{-D}_2\text{H}^+ + \text{HD} \rightarrow o\text{-D}_3^+ + p\text{-H}_2$. This exothermic reaction is representative of a number of reactions which cannot be fitted correctly. In this case we tried to minimize the error at both ends of the fit and in the middle. Notice the small linear Y scale, though. Data are in red, fit in green.

Table C.2. Kooij/Arrhenius coefficients of the dissociative recombination rates of the H₃⁺ isotopologues from P09

Reaction							Rate	
							α	β
							(cm ³ s ⁻¹)	
p-H ₃ ⁺	+	E	→	H	+	H + H	4.96E-08	0.64
p-H ₃ ⁺	+	E	→	p-H ₂	+	H	1.10E-08	0.64
p-H ₃ ⁺	+	E	→	o-H ₂	+	H	0.55E-08	0.64
o-H ₃ ⁺	+	E	→	H	+	H + H	3.13E-08	0.046
o-H ₃ ⁺	+	E	→	o-H ₂	+	H	1.04E-08	0.046
p-H ₂ D ⁺	+	E	→	p-H ₂	+	D	1.70E-09	0.65
p-H ₂ D ⁺	+	E	→	HD	+	H	3.80E-09	0.65
p-H ₂ D ⁺	+	E	→	H	+	H + D	1.75E-08	0.65
o-H ₂ D ⁺	+	E	→	o-H ₂	+	D	2.20E-09	0.65
o-H ₂ D ⁺	+	E	→	HD	+	H	5.50E-09	0.65
o-H ₂ D ⁺	+	E	→	H	+	H + D	2.30E-08	0.65
p-D ₂ H ⁺	+	E	→	HD	+	D	1.17E-09	0.52
p-D ₂ H ⁺	+	E	→	D	+	D + H	7.00E-09	0.52
p-D ₂ H ⁺	+	E	→	p-D ₂	+	H	9.10E-10	0.52
o-D ₂ H ⁺	+	E	→	D	+	D + H	6.17E-09	0.52
o-D ₂ H ⁺	+	E	→	HD	+	D	1.04E-09	0.52
o-D ₂ H ⁺	+	E	→	o-D ₂	+	H	8.02E-10	0.52
m-D ₃ ⁺	+	E	→	p-D ₂	+	D	6.75E-09	0.65
m-D ₃ ⁺	+	E	→	D	+	D + D	2.02E-08	0.65
o-D ₃ ⁺	+	E	→	D	+	D + D	2.02E-08	0.65
o-D ₃ ⁺	+	E	→	o-D ₂	+	D	6.75E-09	0.65
p-D ₃ ⁺	+	E	→	p-D ₂	+	D	3.37E-09	0.65
p-D ₃ ⁺	+	E	→	o-D ₂	+	D	3.37E-09	0.65
p-D ₃ ⁺	+	E	→	D	+	D + D	2.02E-08	0.65