

REVIEW

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Pathophysiology and clinical implications of the veno-arterial PCO_2 gap

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Abstract

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Introduction

The persisting high mortality of circulatory shock highlights the need to search for sensitive early biomarkers to assess tissue perfusion and cellular oxygenation, which could provide important prognostic information and help guide resuscitation efforts. Although blood lactate and venous oxygen saturation (SvO_2) are commonly used in this perspective, their usefulness remains hampered by several limitations. The veno-arterial difference in the partial pressure of carbon dioxide (Pv-aCO_2 gap) has been increasingly recognized as a reliable tool to evaluate tissue perfusion and as a marker of poor outcome during circulatory shock, and it should therefore be part of an integrated clinical evaluation. In this chapter, we present the physiological and pathophysiological determinants of the Pv-aCO_2 gap and review its implications in the clinical assessment of circulatory shock.

Physiological aspects of CO_2 production and transport

Under aerobic conditions, CO_2 is produced at the mitochondrial level as a by-product of substrate oxidation (pyruvate and citric acid cycle intermediates) (Fig. 1). The relationship between the amount of oxygen consumed

(VO_2) and CO_2 produced (VCO_2) during aerobic metabolism is termed the respiratory quotient ($\text{RQ} = \text{VCO}_2 / \text{VO}_2$), and differs according to the main type of oxidized substrate (glucose, $\text{RQ} = 1$; proteins, $\text{RQ} = 0.8$; lipids, $\text{RQ} = 0.7$). Under anaerobic conditions, protons (H^+) resulting from lactic acid production and ATP hydrolysis may generate CO_2 following buffering by bicarbonates (HCO_3^-), leading to the formation of so-called “anaerobic CO_2 ” [1]. Once formed, CO_2 diffuses within the surrounding environment and capillary blood, to be transported to the lungs for elimination. In blood, CO_2 transport is partitioned into three distinct fractions [2]:

1. Dissolved CO_2 fraction, which is in equilibrium with the partial pressure of CO_2 (PCO_2), according to Henry's law of gas solubility: $V_{\text{gas}} = S_{\text{gas}} \times (P_{\text{gas}} / P_{\text{atm}})$, where V_{gas} is the volume of dissolved gas (in ml/ml), S_{gas} is the Henry's constant of gas solubility (0.52 ml/ml for CO_2 at 37 °C), and P_{atm} the atmospheric pressure. Thus, in arterial blood with a PaCO_2 of 40 mmHg (at sea level, 37 °C), dissolved $\text{CO}_2 = [0.52 \times (40/760)] = 27$ ml/l, which is about 5% of the total CO_2 (note that, in mmol/l, Henry's constant for $\text{CO}_2 = 0.03$ mmol/l/mmHg; also note that the conversion factor from mmol to ml CO_2 is ~ 22.3).
2. Bicarbonate (HCO_3^-). CO_2 in blood readily diffuses within red blood cells (RBCs), where it combines with H_2O to form carbonic acid (H_2CO_3), a reac-

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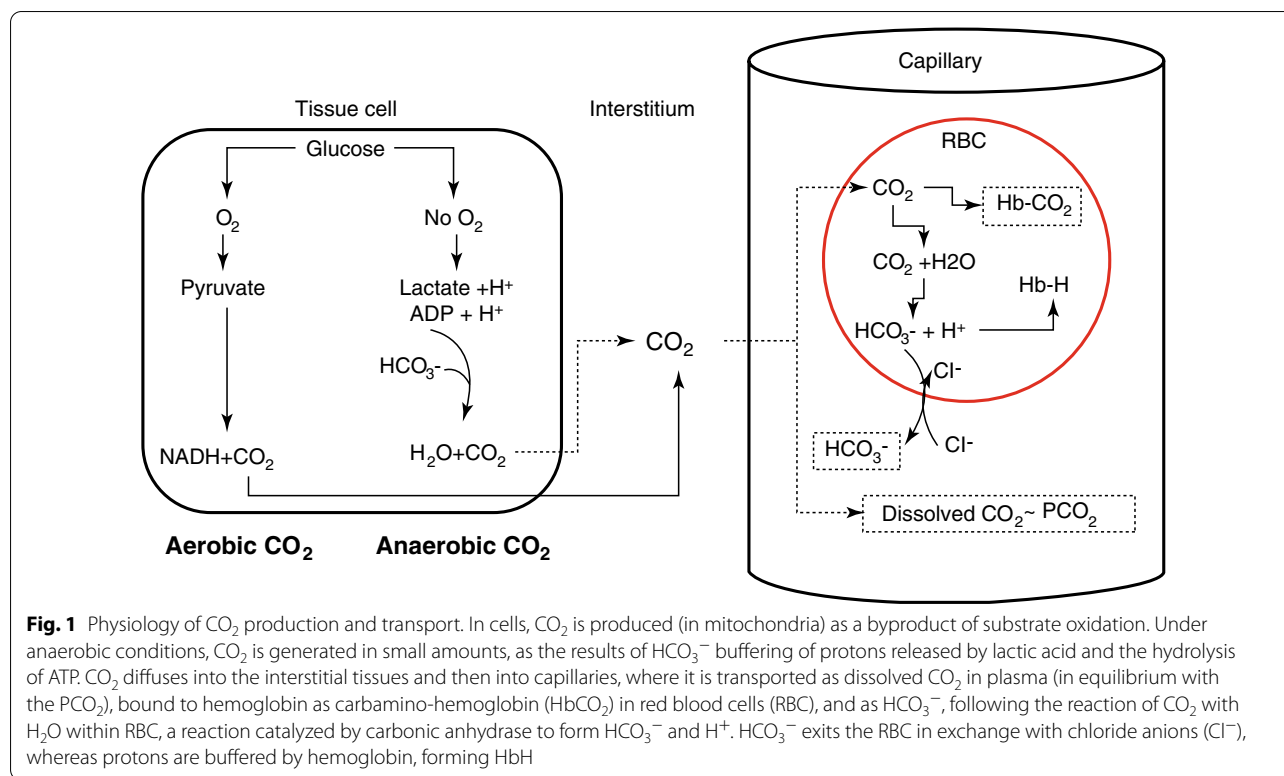


Fig. 1 Physiology of CO₂ production and transport. In cells, CO₂ is produced (in mitochondria) as a byproduct of substrate oxidation. Under anaerobic conditions, CO₂ is generated in small amounts, as the results of HCO₃⁻ buffering of protons released by lactic acid and the hydrolysis of ATP. CO₂ diffuses into the interstitial tissues and then into capillaries, where it is transported as dissolved CO₂ in plasma (in equilibrium with the PCO₂), bound to hemoglobin as carbamino-hemoglobin (HbCO₂) in red blood cells (RBC), and as HCO₃⁻, following the reaction of CO₂ with H₂O within RBC, a reaction catalyzed by carbonic anhydrase to form HCO₃⁻ and H⁺. HCO₃⁻ exits the RBC in exchange with chloride anions (Cl⁻), whereas protons are buffered by hemoglobin, forming HbH

tion catalyzed by the enzyme carbonic anhydrase. In turn, H₂CO₃ dissociates to form HCO₃⁻ and H⁺. While H⁺ is buffered by hemoglobin (formation of HbH), HCO₃⁻ exits the RBC in exchange for a chloride anion (Cl⁻) via a HCO₃⁻-Cl⁻ transporter (erythrocyte chloride shift or Hamburger effect). Thus, the HCO₃⁻ concentration increases in venous blood whereas the Cl⁻ concentration diminishes. CO₂ transport as HCO₃⁻ (RBC and plasma fraction) represents about 90% of the total CO₂ content in arterial blood (this proportion is lower in venous blood due to the Haldane effect). Taking into account a normal hematocrit of 0.45, the CO₂ content under the form of HCO₃⁻ (in whole blood) is ~435 ml/l.

3. Formation of carbamino compounds within hemoglobin: part of the CO₂ within the RBC combines with free amino (R-NH₂) groups within hemoglobin to form carbamino-hemoglobin (R-NH₂-CO₂). This reaction is enhanced when hemoglobin carries less oxygen, implying that more CO₂ is transported as (R-NH₂-CO₂) when the PO₂ decreases, which is the basis of the Haldane effect described below. CO₂ transport under the form of (R-NH₂-CO₂) represents about 5% of the total CO₂ content in arterial blood (~ 1.1 mmol/L ≈ 25 ml/l).

In summary, the total CO₂ content of blood under physiological conditions equals:

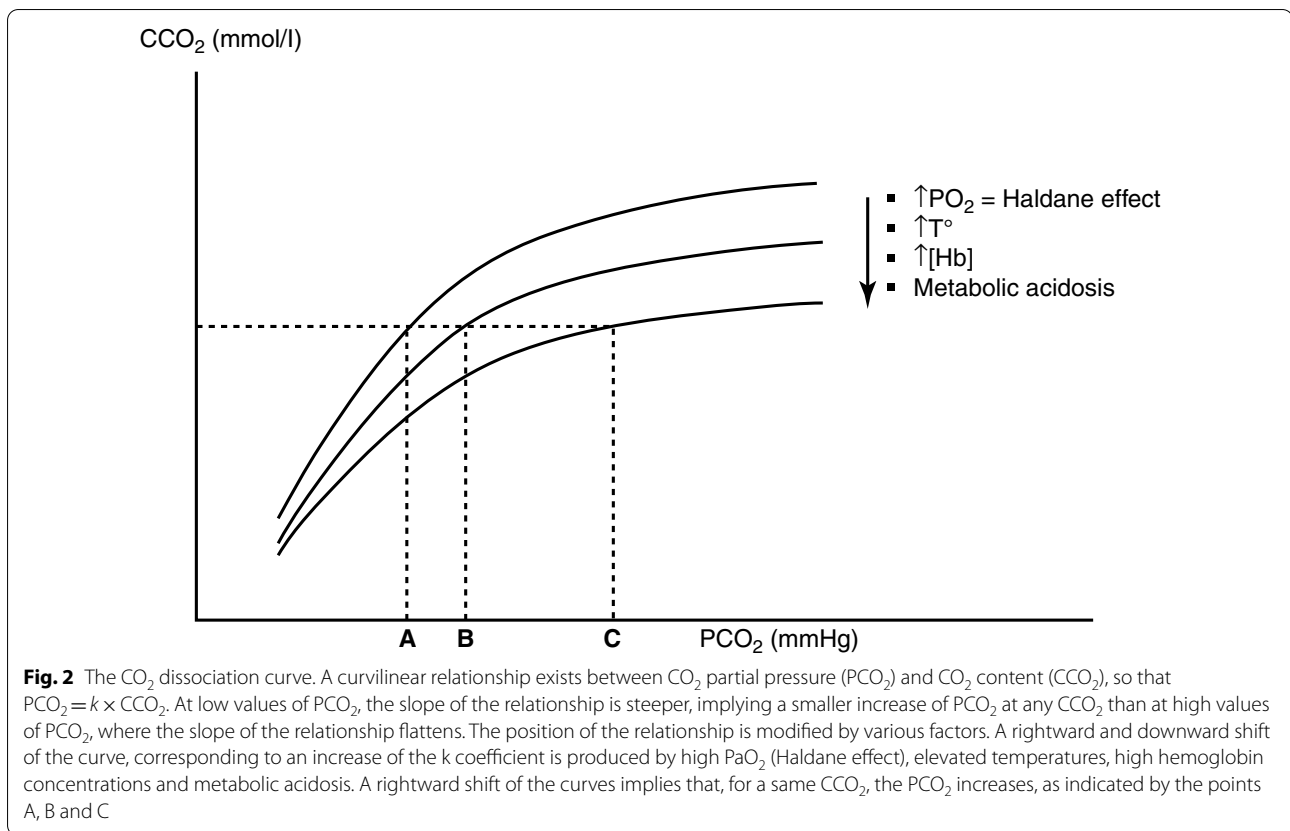
$$[\text{Dissolved CO}_2] + [\text{HCO}_3^-] + [\text{R} - \text{NH}_2 - \text{CO}_2]$$

which is ≈490 ml/l in arterial blood and ≈535 ml/l in mixed venous blood, hence a veno-arterial difference of approximately 45 ml/l. A more precise calculation of the CO₂ content of blood can be obtained by the Douglas equation, but this is too complex to be calculated at the bedside [3].

The CO₂ dissociation curve (PCO₂-CCO₂ relationship)

As is the case for oxygen, a relationship exists between the PCO₂ and the CO₂ content (CCO₂) of blood (Fig. 2). However, in contrast to the sigmoid shape of the O₂ dissociation curve, the CO₂ dissociation curve is slightly curvilinear, indicating a proportional increase in CCO₂ over a wide range of PCO₂. In the physiological range, the relationship between CCO₂ and PCO₂ can therefore be resolved by the equation:

$$\text{PCO}_2 = k \times \text{CCO}_2 \tag{1}$$



Important information provided by the PCO₂-CCO₂ relationship is the shift produced at different values of oxygen saturation of hemoglobin (HbO₂). Indeed, as hemoglobin gets saturated with O₂, it can carry less CO₂ as carbaminoHb, and inversely. This behavior is known as the Haldane effect, which implies that for a same PCO₂, CCO₂ is higher at lower HbO₂ saturation. In other words, this means that as the k constant in the relationship above decreases, the PCO₂-CCO₂ curve is shifted to the left. The consequence of this effect is that, in tissues, more CO₂ is loaded by Hb as it releases O₂, allowing PCO₂ to increase only moderately (from 40 to 46 mmHg), in spite of a marked increase in CCO₂ due to the tissue production of CO₂. Without the Haldane effect, the venous PCO₂ would increase significantly more for a similar increase in CO₂ content.

The curvilinearity of the CO₂ dissociation curve indicates that CCO₂ increases more steeply at low values of PCO₂ and is more flat at high PCO₂ values. It is also noticeable that the curve can be displaced by a certain number of factors: In conditions of metabolic acidosis, the reduction in HCO₃⁻ due to H⁺ buffering reduces the formation of carbamino (R-NH₂-CO₂) compounds inside hemoglobin [4]. As a result, for a given CCO₂, the PCO₂ must increase, which means an increase in the k

constant, and a rightward shift of the relationship. The opposite occurs under conditions of metabolic alkalosis. Other factors influencing the curve are the hematocrit and temperature. At increasing hematocrit, there is a decrease in plasma space with a reduction of HCO₃⁻ and a decrease in CO₂ content at any value of PCO₂, with a shift to the right of the curve. At increasing temperatures, the reduced CO₂ solubility also shifts the relationship to the right [4]. These considerations imply, therefore, that PvCO₂ may vary at constant total venous CCO₂ according to the particular conditions (HbO₂ saturation [i.e., the Haldane effect], arterial pH, temperature and hematocrit).

The Pv-aCO₂ gap: pathophysiology and clinical implications

As discussed earlier, the CCO₂ in the venous side of the circulation is determined by the aerobic production of CO₂ in tissues, influenced by the metabolic rate and the respiratory quotient, and may also increase via non-aerobic production of CO₂. The generation of CO₂ de facto increases the CCO₂ on the venous side of the circulation, implying an obligatory difference between arterial and venous CCO₂, termed the veno-arterial

difference in CCO_2 , or veno-arterial CCO_2 gap: $va-CCO_2$ gap = (venous - arterial) CCO_2 [1].

The tissue VCO_2 does not accumulate under normal conditions, being washed out by the blood flowing across the tissue and eliminated by the lungs. Accordingly, any reduction in tissue blood flow (stagnant condition) will result in an accumulation of tissue CO_2 , implying an increase in the $va-CCO_2$ gap, in accordance with Fick's principle:

$$VCO_{2\text{tissue}} = [(\text{Blood flow}_{\text{tissue}} \times (va - CCO_2 \text{ gap}_{\text{tissue}}))]$$

At the systemic level, the relationship is:

$$VCO_2 = [(\text{Cardiac output} \times (va - CCO_2 \text{ gap}))]$$

According to the equation ($PCO_2 = k \times CCO_2$), the Fick equation for CO_2 can be rewritten as:

$$k \times VCO_2 = [\text{Cardiac output} \times (Pv - PaCO_2)]$$

and

$$(Pv - PaCO_2) = [(k \times VCO_2) / \text{Cardiac output}]$$

Therefore, the $Pv-aCO_2$ gap represents a very good surrogate indicator of the adequacy of cardiac output and tissue perfusion under a given condition of CO_2 production. The normal $Pv-aCO_2$ gap is comprised between 2 and 6 mmHg [5], and many studies assessing $Pv-aCO_2$ gap in clinical conditions used a cut-off value of 6 mmHg above which the gap is considered abnormally elevated. Although the venous PCO_2 should ideally be obtained in a mixed venous blood sampling, good agreement between central and mixed venous PCO_2 values has been reported [6]. Therefore, both central and mixed venous PCO_2 can be used for the calculation of the $va-CO_2$ gap, as long as the variables are not interchanged during treatment in a given patient.

The inverse relationship between cardiac output and the $Pv-aCO_2$ gap

The inverse relationship between cardiac output and the $Pv-aCO_2$ gap (Fig. 3) has been repeatedly demonstrated in both experimental [7] and clinical [8] settings. It is noteworthy that this relationship is not linear, but curvilinear (Fig. 3). At very low cardiac output, the ($Pv-aCO_2$ gap) indeed increases more rapidly. This large increase in $Pv-aCO_2$ gap is primarily due to the flattened relation between CCO_2 and PCO_2 at high values of CCO_2 in conditions of tissue hypercarbia [5], and this is further magnified if tissue metabolic acidosis develops, due to the rightward shift of the PCO_2-CCO_2 relationship in acidic conditions (increased k coefficient, see above). Also, venous accumulation of CO_2 will increase as a consequence of low pulmonary perfusion and CO_2 elimination,

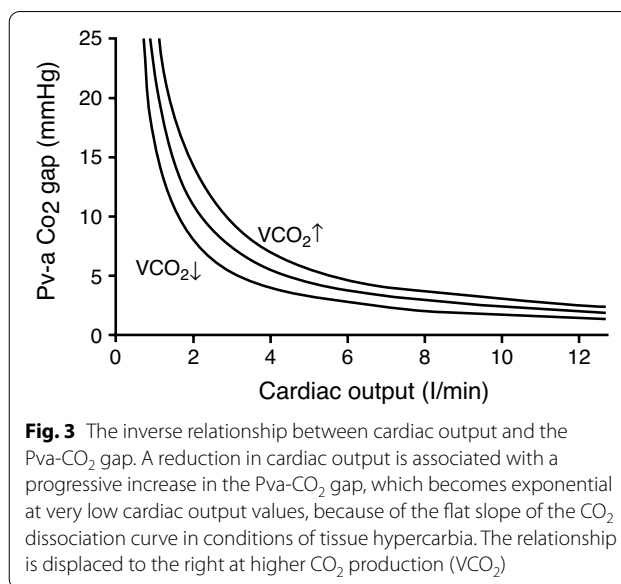


Fig. 3 The inverse relationship between cardiac output and the $Pv-aCO_2$ gap. A reduction in cardiac output is associated with a progressive increase in the $Pv-aCO_2$ gap, which becomes exponential at very low cardiac output values, because of the flat slope of the CO_2 dissociation curve in conditions of tissue hypercarbia. The relationship is displaced to the right at higher CO_2 production (VCO_2)

further widening the gap [9]. In contrast, the increase in $Pv-aCO_2$ in very low flow states with conditions of VO_2 -oxygen delivery (DO_2) dependence will be attenuated by the mandatory reduction in aerobic VCO_2 . Such a decrease in VCO_2 results in a leftward shift of the cardiac output/ $Pv-aCO_2$ gap relationship, as shown in Fig. 3 [5].

$Pv-aCO_2$ gap and tissue dysoxia

In addition to tracking changes in cardiac output and tissue perfusion, the $Pv-aCO_2$ gap can increase through an augmentation of VCO_2 [8]. Under aerobic conditions, that is in the absence of any clinical sign of shock or increased blood lactate, such an increase reflects an increased metabolic demand or an increase in RQ (glucidic diet), or both. Physiologically, an increased metabolic rate is generally coupled with an increase in cardiac output, but such adaptation may not occur in critically ill patients with inadequate cardiovascular reserves, which may result in an increased $Pv-aCO_2$ gap. Interventions should here be targeted first to reduce the metabolic demand. Persistence of an increased $Pv-aCO_2$ gap should not necessarily prompt therapies to increase cardiac output, given the risk associated with deliberate increase in cardiac output in the absence of tissue dysoxia [10]. However, it is noteworthy that an increased $Pv-aCO_2$ gap immediately after surgery in high risk patients, independent of their hemodynamic condition, SvO_2 and lactate, has been associated with significantly more complications [11]. This suggests that a high $Pv-aCO_2$ gap could track insufficient resuscitation and might represent a goal

for hemodynamic optimization in such patients, but this issue is controversial and remains to be proven [9].

Under *anaerobic* conditions, the question as to whether the Pv-aCO₂ gap can be used as a marker of tissue dysoxia, by detecting increased anaerobic VCO₂ from H⁺ buffering, has attracted much attention. An advantage of Pv-aCO₂ gap in this sense would be its ability to rapidly track changes in CO₂ formation, hence providing sensitive, rapid and continuous detection of ongoing anaerobiosis. This would contrast from usual markers of tissue dysoxia, such as SvO₂ or lactate. Indeed, SvO₂ can be unreliable in conditions of reduced oxygen extraction and hyperdynamic circulation (sepsis) [12]. The disadvantage of lactate is its lack of specificity as a marker of dysoxia (type A vs type B hyperlactatemia), and its relatively slow clearance kinetics dependent on liver perfusion and function [13], which limits its utility to rapidly track changes in tissue oxygenation [9].

The Pv-aCO₂ gap in stagnant dysoxia

In essence, tissue dysoxia is classically attributed to stagnant, hypoxic, anemic and cytopathic mechanisms. As a sensitive marker of reduced cardiac output, an increased Pv-aCO₂ gap is a reliable indicator of stagnant dysoxia. Importantly, the major gap noted under very low flow conditions (see earlier) has been associated with a global reduction in VCO₂ (VO₂-DO₂ dependence), implying that any increase in anaerobic VCO₂ could not offset the depressed aerobic VCO₂ [7]. Therefore, the increased Pv-aCO₂ gap depends entirely on the stagnant accumulation of tissue CO₂, but not on increased anaerobic VCO₂ in low flow conditions [1, 14].

The Pv-aCO₂ gap in hypoxic or anemic dysoxia

To address the role of the Pv-aCO₂ gap to detect hypoxic dysoxia, Vallet et al. reduced DO₂ below the critical threshold in an isolated dog hindlimb model, by reducing blood flow or by decreasing PO₂ [15]. Both conditions similarly reduced VO₂ and O₂ extraction, but the Pv-aCO₂ gap increased exclusively in the ischemic, but not hypoxic condition, implying that stagnant, but not hypoxic dysoxia was the responsible mechanism [15]. Comparable results were obtained by Nevière et al. in the intestinal mucosa of pigs, following the systemic reduction in DO₂ to similar levels either by reduction of cardiac output or arterial PO₂ [16]. With respect to anemic dysoxia, similar conclusions were obtained in sheep hemorrhage models, in which no increase in Pv-aCO₂ gap was detected under conditions of VO₂/DO₂ dependency due to reduced hemoglobin concentration [17], unless there was a concomitant reduction in cardiac output [18].

Hence, significant hypoxic or anemic dysoxia occurs in the absence of any Pv-aCO₂ gap increase.

The Pv-aCO₂ gap in cytopathic dysoxia

An acquired intrinsic abnormality of tissue O₂ extraction and cellular O₂ utilization, primarily related to mitochondrial impairment, defines the concept of cytopathic hypoxia, and the resulting cellular bioenergetic failure could represent an important mechanism of organ dysfunction in sepsis [19]. Mitochondrial defects have been demonstrated in several tissues obtained from animals in various models of sepsis, and limited data also exist on altered mitochondrial metabolism in human biopsy samples or circulating blood cells [20]. The detection of cytopathic hypoxia, however, is still not feasible at the bedside, although new techniques such as the measurement of mitochondrial O₂ tension using protoporphyrin IX-Triplet State Lifetime Technique (PpIX-TSLT) are currently being developed [21]. Furthermore, impaired O₂ extraction in sepsis does not necessary imply cytopathic hypoxia, as it may be related to impaired microcirculation.

Theoretically, the increased anaerobic CO₂ generation in conditions of cytopathic hypoxia could result in increased anaerobic VCO₂ leading to an increased Pv-aCO₂ gap. This assumption has been evaluated in a porcine model of high dose metformin intoxication, which induces mitochondrial defects comparable to cyanide poisoning [22]. As expected, treated pigs exhibited reduced VO₂ and marked lactic acidosis, in spite of preserved systemic DO₂. However, although VCO₂ decreased less than VO₂, suggesting some anaerobic VCO₂, no significant increase in Pv-aCO₂ gap was noted. In a human case report of massive metformin intoxication, Waldauf et al. also reported no elevation in Pv-aCO₂ gap despite major lactic acidosis and reduced aerobic VO₂, as detected by increased SvO₂ [23]. Therefore, although data are very limited, cytopathic dysoxia related to impaired mitochondrial respiration appears not to widen the Pv-aCO₂ gap.

The Pv-aCO₂ gap in sepsis

Ongoing tissue dysoxia with persistent lactic acidosis is a hallmark of sepsis, and associated with a poor prognosis. Although a hyperdynamic circulation is characteristic of sepsis, many septic patients may have a cardiac output that is insufficient to meet metabolic demands, because of persistent hypovolemia or concomitant myocardial dysfunction. An increased Pv-aCO₂ gap has been reported in patients with lower cardiac output in sepsis, consistent with the ability of the Pv-aCO₂ gap to detect stagnant dysoxia, also in the context of sepsis [24]. In

such conditions, an increase in cardiac output correlates with a parallel decrease in Pv-aCO₂ gap [25]. Importantly, as reported by Vallee et al. [26], the Pv-aCO₂ gap is able to detect persistently low cardiac output even in patients with a normal SvO₂. Such a high Pv-aCO₂ gap during the early resuscitation of septic shock has been correlated with more organ dysfunction and worse outcomes [27].

Many septic patients display persistent lactic acidosis in spite of an elevated cardiac output and normal or even increased SvO₂. This implies that mechanisms unrelated to macrohemodynamics sustain tissue dysoxia in this setting, i.e., a loss of so-called hemodynamic coherence, with significant negative impact on outcome [28]. Impaired microcirculatory perfusion is indeed a prototypical perturbation in experimental [29] and human sepsis [30], which may impair tissue oxygenation. Such microcirculatory derangements result in tissue CO₂ accumulation, which can be tracked, for example, by sublingual capnometry, as shown by Creteur et al. [31]. Accordingly, in a prospective observational study including 75 patients with septic shock, Ospina-Tascon et al. found a significant correlation between Pv-aCO₂ gap and microcirculatory alterations. These were independent of systemic hemodynamic status and persisted even after correction for the Haldane effect [32], indicating that the Pv-aCO₂ gap may be a useful tool to assess impaired microcirculation in sepsis [33]. Furthermore, Creteur et al. reported that increasing cardiac output with dobutamine in patients with impaired microcirculation resulted in a decreased regional PCO₂ gap (sublingual and gastric mucosal) that was associated with a significant increase in well-perfused capillaries [31].

In summary, an elevated (>6 mmHg) Pv-aCO₂ gap in sepsis detects stagnant dysoxia, whether related to a low cardiac output or a derangement in microcirculatory blood flow, and this holds true even in the presence of a normal or elevated SvO₂. As such, a high Pv-aCO₂ gap might prompt a trial to improve tissue blood flow by increasing cardiac output [34].

Finally, many septic patients with an elevated cardiac output exhibit a normal Pv-aCO₂ gap, resulting from elevated CO₂ washout by increased tissue blood flow. Many of these patients still display signs of ongoing dysoxia with lactic acidosis and organ dysfunction. Whether this pattern reflects cytopathic dysoxia or regional microcirculatory alterations not tracked by Pv-aCO₂ gap elevation remains to be established.

Use of the Pv-aCO₂ gap as a prognostic tool

In sepsis, evidence exists that a Pv-aCO₂ gap >6 mmHg, even after normalization of blood lactate, is predictive of poor outcomes [35–37], which has been highlighted in a recent systematic review of 12 observational studies

[38]. Whether this holds true for a broader population of critically ill patients with circulatory shock has been questioned in a recent meta-analysis of 21 studies with a total of 2155 patients from medical, surgical and cardiovascular ICUs [37]. Overall, a high Pv-aCO₂ gap was associated with higher lactate levels, lower cardiac output and central venous oxygen saturation (ScvO₂), and was significantly correlated with mortality. The latter was however restricted to medical and surgical patients, with no association found for cardiac surgery patients. Since the meta-analysis included only two studies in cardiac surgery, this negative result should be interpreted with caution. Three recent retrospective studies not included in the meta-analysis [39–41] indeed reported a negative impact of high postoperative Pv-aCO₂ gap on major complications and mortality after cardiac surgery, although with limited diagnostic performance [41].

Future studies are needed to refine the value of the Pv-aCO₂ gap as a prognostic biomarker in cardiac surgery patients, taking into account the low mortality (3.4%) in this population [42].

Pitfalls in the interpretation of the Pv-aCO₂ gap

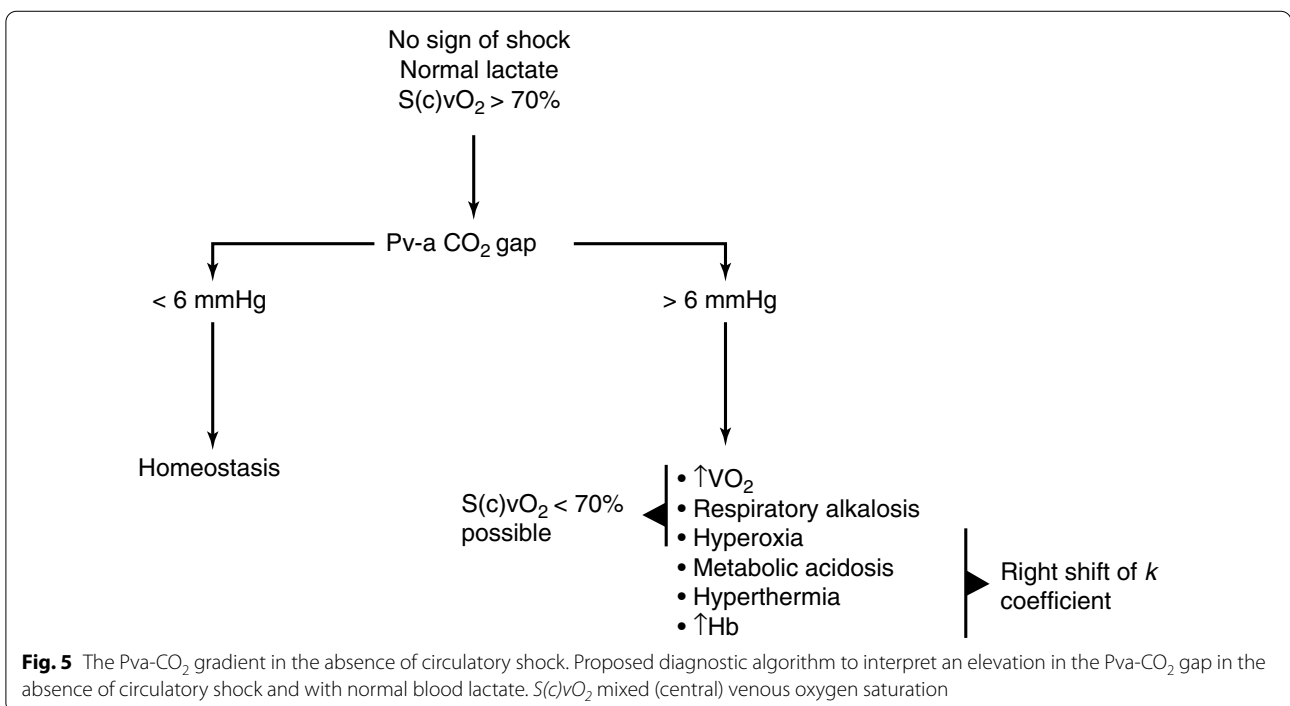
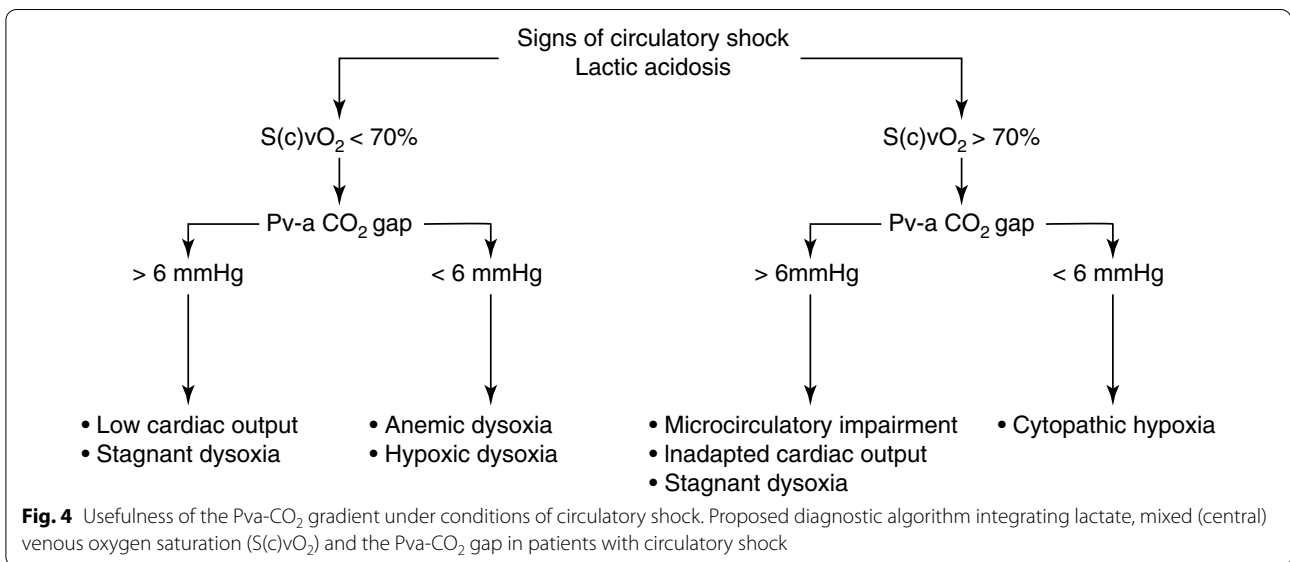
As already mentioned, several factors may influence the position of the PCO₂-CCO₂ relationship by influencing the *k* factor of proportionality between both variables (see Fig. 2), which must be taken into account for a proper interpretation of the Pv-aCO₂ gap. These include the oxygen saturation of hemoglobin (Haldane effect), metabolic shifts of pH, temperature and hemoglobin concentration. In addition, it is essential to consider possible sources of errors in the measurement of PCO₂, including contamination of the samples with fluid or air bubbles, and insufficient precision of the gas analyzer. When comparing successive determinations of Pv-aCO₂ gap, it is therefore recommended to consider only variations of at least ± 2 mmHg as real changes [43].

Two additional confounders in the interpretation of the Pv-aCO₂ gap require some discussion. The first is hyperoxia. It has been observed that, in patients with circulatory shock, ventilation at 100% inspired oxygen fraction (FiO₂) for 5 min increased venous PCO₂, and hence the Pv-aCO₂ gap, independent of changes in the hemodynamic status [44]. While this observation may be explained by a lower CO₂ affinity of hemoglobin due to elevated venous PO₂ (Haldane effect) [44], it may also reflect some impairment in microcirculatory blood flow, owing to the vasoconstrictive effects of hyperoxia [45]. The second confounder is acute hyperventilation with respiratory alkalosis. For example, as shown by Mallat et al. in 18 stable septic shock patients [46], an acute decrease in arterial PCO₂ from 44 to 34 mmHg produced by transient hyperventilation

(30 min) induced a significant increase in PCO_2 gap (absolute 2.2 mmHg, relative + 48.5%). Possible mechanisms include, first, increased aerobic production of CO_2 due to stimulated aerobic glycolysis under conditions of cellular alkalosis, and second, a reduction in microcirculatory blood flow due to the acute drop of CO_2 . Thus, both acute hyperoxia and hypocapnia may be important confounders in the interpretation of an increased Pv-a CO_2 gap, which must be taken into account by the clinician.

Conclusion

The Pv-a CO_2 gap is a reliable indicator of impaired tissue perfusion, whether the result of a global reduction in cardiac output or to microcirculatory abnormalities, but it does not track tissue dysoxia, unless related to a stagnant mechanism. Being easily accessible and readily available, the Pva- CO_2 gap should be included in the integrated evaluation of the patient in circulatory shock. Several diagnostic algorithms incorporating Pva- CO_2 gradients have been proposed, such as those presented in Figs. 4



and 5. It remains to be established whether the Pva-CO₂ gap should be part of a resuscitation bundle protocol, and whether therapies aimed at normalizing an increased Pva-CO₂ gap could improve the dismal prognosis of circulatory shock.

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Authors' contributions

ZL performed the literature review drew the figures and drafted the manuscript. AS critically reviewed the manuscript. LL critically reviewed the manuscript. All authors read and approved the final manuscript.

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Competing interests

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