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Perovskite Light-Emitting Diodes based on Solution-Processed, Self-Organised Multiple Quantum Wells

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Organometal halide perovskites can be processed from solutions at low temperatures to form crystalline direct-bandgap semiconductors with promising optoelectronic properties¹⁻⁵. However, the efficiency of their electroluminescence is limited by nonradiative recombination associated with defects and leakage current due to incomplete surface coverage⁶⁻⁹. Here we demonstrate a solution-processed perovskite light-emitting diode (LED) based on self-organised multiple quantum wells (MQWs) with excellent film morphology. The MQW-based LED exhibits a very high external quantum efficiency of up to 11.7%, good stability and exceptional high-power performance with an energy conversion efficiency of 5.5% at a current density of 100 mA cm⁻². This outstanding performance arises because the lower bandgap regions which generate electroluminescence are effectively confined by perovskite MQWs with higher energy gaps, giving very efficient radiative decay. Surprisingly, there is no evidence that the large interfacial areas between different bandgap regions cause luminescence quenching.

Recently, high-efficiency photovoltaic devices based on three-dimensional (3D) organometal halide perovskites, such as CH₃NH₃PbI₃ and NH₂CH=NH₂PbI₃ (FAPbI₃), have been demonstrated³⁻⁵. Hybrid 3D perovskites exhibit high photoluminescence quantum efficiencies (PLQEs) and good charge mobilities, making them also attractive for electroluminescence (EL) applications^{2,8,9}. Encouraging performance metrics of light-emitting diodes (LEDs) based on 3D perovskites, such as low turn-on voltages and external quantum efficiencies (EQEs) of up to 3.5% at high current densities, have been demonstrated⁹. However, the EL quantum efficiency is far behind the limit predicated by ~70% PLQE of the 3D perovskites, mainly due to the existence of current losses caused by incomplete surface coverage of the perovskite films and the fact that the high PLQE can only be obtained at high excitations^{8,9}. By using thick (>300 nm) perovskite films, Cho *et al.* obtained LEDs with over 8% EQE¹⁰. However, for this device, the turn-on voltage is high and the power efficiency is low, which may result from the thick perovskite layer used. In order to further enhance the performance of 3D perovskite-based LEDs, it is

essential to obtain perovskite thin films with both complete surface coverage and high PLQE⁸⁻¹⁰. Moreover, device stability, which was proven to be a vital issue in organic-inorganic halide perovskite-based photovoltaics¹¹, has not been addressed in perovskite LEDs.

The 3D perovskites are actually an extreme case of layered organometal halide perovskites with a general formula of $L_2(SMX_3)_{n-1}MX_4$, where M, X, L, and S are a divalent metal cation, a halide, and organic cations with long and short chains, respectively (Fig. 1a)¹²⁻¹⁴. Here n is the number of semiconducting MX_4 monolayer sheets within the two organic insulating layers (cation L), with $n=\infty$ corresponding to the structure of a 3D perovskite SMX_3 . With smaller numbers of MX_4 layers, quantum confinement effects, such as an increase in bandgap and exciton energy, become important^{6,15}. In consequence, the layered perovskites naturally form quantum-well structures. At the opposite extreme, when $n=1$, the layered perovskites form a monolayer structure of a two-dimensional (2D) perovskite L_2MX_4 . The 2D L_2MX_4 perovskites generally have good film-formation properties¹³. Nevertheless, the PLQEs of the 2D perovskites are rather low at room temperature, owing to fast exciton quenching rates^{6,7}. LEDs based on the 2D perovskites have been attempted, while the devices are either very low in efficiency or only operational at cryogenic temperatures¹⁶⁻¹⁸. Here we demonstrate very efficient (up to 11.7% EQE) and high-brightness EL achievable at room temperature by using solution-processed perovskite multiple quantum wells (MQWs) with an energy cascade, which can combine the advantages of 2D and 3D perovskites. We note, a relevant perovskite LED work¹⁹ which shows a peak EQE of 8.8% has been published online during the revision of this paper.

A precursor solution of 1-naphthylmethylamine iodide (NMAI), formamidinium iodide (FAI), and PbI_2 with a molar ratio of 2:1:2 dissolved in N,N-dimethylformamide (DMF) was used to deposit perovskite films (see Methods for details), which are abbreviated as NFPI₇ films below. Atomic force microscopy

(AFM) measurements show that the NFPI₇ film has a smooth and uniform surface coverage (Fig. 1b). The root-mean-square roughness of the NFPI₇ film, 2.6 nm, is comparable to that of the 2D (NMA)₂PbI₄ (n=1) perovskite film, 1.4 nm, and much lower than that of the 3D FAPbI₃ (n=∞) perovskite film, 18.8 nm (Supplementary Fig. S1).

We study the optical properties of the NFPI₇ films. Because of the quantum confinement effects, the absorption and emission features of the layered perovskites depend on the value of n ^{6,15}. Previous studies on PbI₄-based perovskites showed that the absorption peaks at ~2.4, ~2.2, ~2.0 and ~1.9 eV correspond to excitonic absorption of QWs with $n=1, 2, 3$ and 4 respectively^{6,20,21}. Our results confirm that the (NMA)₂PbI₄ perovskite film, i.e. the $n=1$ QW film, presents a strong peak at 2.43 eV (Supplementary Fig. S2). The absorption spectrum of the NFPI₇ film shows a strong exciton absorption peak at 2.18 eV (Fig. 1c), indicating that the major component of the perovskite film is (NMA)₂(FAPbI₃)PbI₄, i.e. the $n=2$ QWs. The shoulders at 2.43 and 1.95 eV suggest the existence of small fractions of $n=1$ and $n=4$ perovskite QWs respectively, while the absorption peak of the $n=3$ QWs at ~2.0 eV may be hidden by the absorption tail of the $n=2$ QWs. Optical features corresponding to QWs with higher n values are not evident in the absorption spectrum. They become visible in the PL spectrum, as shown by the dominating peak at 1.62 eV. This emission is ~40 nm blue-shifted compared to that of bulk 3D ($n=∞$) FAPbI₃ perovskite (1.54 eV, Supplementary Fig. S2). The PL spectrum on a semi-log scale reveals several weak emission peaks at 2.38, 2.14 and 1.91 eV. With the absorption and emission spectra plotted in the same panel (Fig. 1c), one can tell that absorption and emission peaks for $n \leq 4$ QWs correspond very well to each other. We point out that although $n=2$ QWs give the strongest absorption peak in the NFPI₇ films, emission from them is very weak.

The investigations above suggest that our NFPI₇ film is an ensemble of self-organised multiple QWs with different exciton energies. The absorption and PL measurements

clearly demonstrate energy transfer from QWs with large exciton energies to QWs with lower exciton energies. PL excitation measurements on the NFPI₇ film show that different emissions share a strong excitation peak at 2.18 eV (Fig. 1d), showing that there is energy transfer from n=2 QWs to the lower energy emissions. For the 765 nm/1.62 eV emission, the PL excitation spectrum also reveals weak contributions from QWs with n>2. The observation of weak emission from the n=1, n=2 and n=4 QWs demonstrates that the energy migration is not complete. The cascade energy transfer of photon excitation in the perovskite MQWs is illustrated in Fig. 1e.

Transient absorption (TA) measurement further shows that the energy transfer between the perovskite QWs is an ultrafast process (Fig. 1f). With 400 nm excitation, the excitons are primarily formed in the n=2 QWs. This manifests as a fast build-up of the photobleaching (PB) at the exciton absorption peak (2.18 eV) of the n=2 QWs. The evolution of the TA spectra shows that the excitons reach the large-n QWs (~1.64 eV) after some time delay. The PB of the large-n QWs resonance grows over time. Time traces at selected probe wavelengths are shown in Fig. 1g. The PB of n=2 QWs (~2.18 eV) shows an ultrafast dominant decay with a time constant of ~0.5 ps. This decay time is closely matched with the fast PB formation time of the large-n QWs (~1.64 eV). This result indicates that a substantial portion of the photo-generated excitons from n=2 QWs will be localised to large-n QWs within 0.5 ps. Subsequently, a slower decay of the PB of n=2 QWs also matches well with a relative slower formation of the large-n QWs PB. This relatively slower exciton localisation time is fitted to be ~50 ps.

The rapid energy transfer process in the perovskite MQWs can essentially avoid the exciton quenching effect presented in previously reported 2D perovskites^{6,7}. Figure 1h shows the light-intensity-dependent PLQE of the NFPI₇ film under a 445 nm cw laser excitation. The high PLQE of up to 60% suggests that the photogenerated excitons in low-n QWs can be energy transferred to the large-n QWs with superior emission properties, allowing efficient radiative recombination. Remarkably, the MQW films

present high PLQEs at an excitation as low as $\sim 0.3 \text{ mW cm}^{-2}$. This feature is very different from that of the bulk 3D perovskite films which have high PLQEs only at very high excitations, e.g. over 500 mW cm^{-2} (Supplementary Fig. S3 and ref. 2). High PLQEs of the perovskite emitters can be obtained when the nonradiative decay centres are filled by excitations⁹. In our MQW perovskite films, the exciton decay in large-*n* QWs is much slower (Supplementary Fig. S4) than the energy transfer process. As a result, most of the photo-generated excitons formed are concentrated to the small space of the large-*n* QWs, which can substantially increase the local exciton density in the emissive large-*n* QWs. Furthermore, the QW structure can effectively confine the excitons and reduce the exciton diffusion lengths, which can decrease the exciton-defect quenching probability.

The desirable film morphology and optical properties of the NFPI₇ film encourage us to fabricate LEDs based on perovskite MQWs. Figure 2a shows the cross-sectional view of a device with a multi-layer structure of indium tin oxide (ITO)/polyethylenimine ethoxylated (PEIE)-modified zinc oxide (ZnO, $\sim 20 \text{ nm}$)/perovskite MQWs ($\sim 30 \text{ nm}$)/poly(9,9-dioctyl-fluorene-co-N-(4-butylphenyl)diphenylamine) (TFB, $\sim 40 \text{ nm}$)/molybdenum oxide (MoO_x, $\sim 7 \text{ nm}$)/gold (Au, $\sim 60 \text{ nm}$). All layers, apart from the MoO_x/Au electrode, were deposited from solutions. The flat-band energy levels of the layers are indicated in Fig. 2b, and the energy level values for NFPI₇ film were estimated by UPS (Supplementary Fig. S5) and optical measurements. The PEIE-modified ZnO and TFB layers are employed as electron-transporting interlayers and hole-transporting interlayers respectively. The MoO_x/Au bilayers and ITO are selected as top electrodes and bottom electrodes respectively.

A cross-sectional sample of the NFPI₇ device was analysed by an aberration-corrected scanning transmission electron microscope (STEM). A low-magnification high-angle annular dark-field (HAADF) image (Fig. 2a) illustrates the multiple layers of materials with distinct contrast. The $\sim 30 \text{ nm}$ thick perovskite layer is smooth and

pin-hole free based on through STEM inspections, agreeing well with the AFM results that the NFPI₇ film has a low surface roughness. Zoomed-in observations indicate that the perovskite layer exhibits contrasts, with the majority of brighter regions close to the perovskite/TFB interface. We propose that the contrasts in the perovskite layer reflect the spatial distributions of QWs with different *n* values. Perovskite QWs with larger *n* values accommodate more PbI₄ units per volume, thereby result in a higher intensity of large-angle elastic scattering and in consequence appear brighter in the HAADF imaging. In other words, the majority of large-*n* QWs are located in the regions close to the perovskite/TFB interface. This deduction is supported by both energy-dispersive X-ray spectroscopy (EDX) elemental mapping and high-resolution transmission electron microscopy (HRTEM) results (Figs. 2c, d). EDX maps show stronger signals of both lead and iodide in the brighter regions. Distribution of lead and iodide across the perovskite layer suggests that the large-*n* QWs and low-*n* QWs locate at the interface of perovskite/TFB and perovskite/ZnO respectively. The large-*n* QWs region is much narrower than the low-*n* QWs region which presumably is the *n*=2 QWs, and a gradually graded or mixed QWs region is in between. HRTEM and the corresponding Fast Fourier Transformation (FFT) analyses (Fig. 2d and Supplementary Fig. S6) show that the perovskite crystals close to the perovskite/TFB interface share the same cubic structure with the 3D α -FAPbI₃²².

Figure 3a shows the EL spectra of devices based on NFPI₇ MQW films (see Supplementary Fig. S7 for characterization setup). The EL emission peak at near-infrared (NIR), 786 nm/1.58 eV, is ~20 nm blue-shifted compared with the EL from bulk 3D FAPbI₃ perovskites (Supplementary Fig. S8). The shapes of the EL spectra do not change at different bias voltages (Supplementary Fig. S9). The NFPI₇-MQW LED turns on at a low voltage of 1.5 V (Fig. 3b). The current density and radiance rise rapidly by several orders of magnitude after turn-on, implying excellent charge injection and transport in the MQW LEDs. The EQE of the LED reaches 9.6% at 2.3 V with a current density of 15 mA cm⁻² (Fig. 3c). A high radiance of 55 W sr⁻¹ m⁻² is achieved at 3.6 V. The angular emission intensity of the device

follows a Lambertian profile (Supplementary Fig. S10). The statistics from 67 devices present an average EQE of 7.0% with a relative standard deviation of 18% (Fig. 3d), demonstrating good reproducibility of the NFPI₇-MQW LEDs.

We consider that the superior EL performance in the MQW LEDs is due to the confinement of injected charges in the large-*n* QWs which have excellent emission properties as shown by the PL measurements. In EL device operation, electrons and holes are injected into the perovskite layer and accumulated at the large-*n* QWs because of the cascade energy structure of the perovskite MQW films. In other words, the injected charges can directly recombine at the very thin layer of large-*n* QWs. This is consistent with our finding that the weak emissions at higher energies observed in the PL spectrum are absent in the EL measurements and the EL emission peak is slightly red-shifted compared to the main PL peak (Fig. 3a and Fig. 1c). As shown by the PL investigations (Fig. 1h), the very thin layer of large-*n* QWs is highly emissive even at low excitations, which results in the high EL efficiency in our MQW perovskite LEDs. In order to fabricate high-efficiency perovskite LEDs, it is a challenge to obtain a large-*n* emissive thin layer with good coverage sandwiched between two non-quenching contacts with appropriate energy levels, while our MQW structure provides a simple alternative solution to it.

One of the advantages for perovskite materials is that the optoelectronics properties can be easily tuned by adjusting the compositions. In perovskite solar cells, it has been shown that the device performance can be optimised by using mixed halogens which improves crystallinity of the perovskite film²³. We show that it is possible to further enhance the efficiency of MQW LEDs by adopting a similar strategy (Supplementary Table S1). For example, we used a precursor solution of NMAI: FABr: PbI₂ with a molar ratio of 2:1:2 in DMF to deposit new perovskite MQW films (abbreviated as NFPI₆B below). This NFPI₆B film has a PLQE of 67% (Supplementary Fig. S3), higher than that of the NFPI₇ film. The EL emission peak of LEDs based on the NFPI₆B film is shifted to 763 nm (Fig. 3a). The champion

NFPI₆B LED turns on at 1.3 V. The EQE reaches 11.7% at 2.6 V with a current density of 38 mA cm⁻². A high radiance of ~82 W sr⁻¹ m⁻² is achieved at 3.6 V. These results represent the highest EQE for perovskite-based LEDs⁹. Remarkably, the device is highly efficient at large current densities. The energy conversion efficiency (wall-plug efficiency) of our NFPI₆B MQW LED is 5.5% at 100 mA cm⁻², which is more efficient than the state-of-the-art vacuum-deposited organic LEDs at this high current density (Supplementary Table S2)^{24–26}. The EQE histogram for 70 NFPI₆B-MQW LEDs shows an average peak EQE of 8.8% and a small relative standard deviation of 11% (Fig. 3d). Owing to the good film uniformity, the size of the perovskite MQW LEDs can be easily scaled up by adopting the same device structure and fabrication process. Figure 3a also shows uniform and bright EL emission from a ~64 mm² NFPI₆B-MQW device with a peak EQE of 7.5%.

We can tune the EL across a wide range of the spectrum by further changing the halide compositions in precursor solutions. For example, the perovskite MQW LEDs can achieve narrow emission peaks at 736, 685, 664, 611 or 518 nm, when the MQW films are produced by using precursor solutions with various ratios of Br to I (Fig. 4a). The various combinations of precursor solutions to produce the NFPI_yB_{7-y} (y represents the fraction of I) films are provided in Supplementary Table S1. Figure 4b demonstrates bright and uniform emission from the green and red MQW LEDs, respectively. The symmetric and narrow emission peaks at 518 nm and 685 nm correspond to Commission Internationale de l'Eclairage (CIE) colour coordinates of (0.11, 0.74) and (0.72, 0.28) respectively, which are ideal for display applications. Supplementary Table S1 also summarises the device performance of different-colour MQW LEDs with the same device structure. The performance of the visible LEDs is inferior to that of the NIR devices, which is likely due to either inefficient charge injection to the emitters with larger band gap or lower PL efficiencies. By further optimising the device structure and film processing, it is possible to enhance the MQW LED device performance in the visible range.

In addition to the remarkably high efficiency and readily colour tunability, the devices based on perovskite MQWs also demonstrate reasonably good stability. As shown in Fig. 4c, the EQE of the NFPI₆B devices dropped to half of the initial value after ~2 h under a constant current density of 10 mA cm⁻². This result shows that the stability of perovskite MQW LEDs is significantly improved (by two orders of magnitude) compared with LEDs based on the 3D perovskites produced in our laboratory, which can only survive ~1 minute under similar conditions (Supplementary Fig. S8). We believe both the high power conversion efficiency of the LED and the stability of the perovskites QW films²¹ contribute to the improved device lifetime.

The use of semiconductors with different bandgaps to engineer high performance optoelectronic devices has of course been very effectively developed in III-V semiconductors. This allows control of charge recombination sites and luminescence wavelength, but it requires very close control of lattice parameter matching to avoid electronic defects that cause trapping and quenching. Our findings reveal that these mixed energy gap perovskite structures show no evidence for higher electronic defect densities than single gap materials, in spite of the very high interfacial areas that must be present. This fact provides real promise for the further development of this class of materials, particularly for light emission.

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Materials & Correspondence

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Author Contributions

J.W. had the idea for and designed the experiments. J.W. and W.H. supervised the work. L.C., R.G., N.W. and S.Z. carried out the device fabrication and characterizations. Y.M., Y.S. and Y.C. conducted the optical measurements. W.Z. managed to set up the testing systems and took part in the optical measurements. C.Y. and Y.C. synthesized the NMAI/NMABr and measured AFM. R.Y., Q.G., Y.K., M.Y., D.D. and L.Y. participated in device fabrication and characterizations. G.X. measured transient absorption. Y.L., Q.D., H.T, C.J., Y.J. and Y.W. carried out HRTEM and

STEM characterizations. J.W., N.W. and F.G. wrote the first draft of the manuscript. Y.J., R.H.F. and W.H. participated in data analysis and provided major revisions. All authors discussed the results and commented on the manuscript.

Competing financial interests

The authors declare no competing financial interests.

Figures:

Figure 1

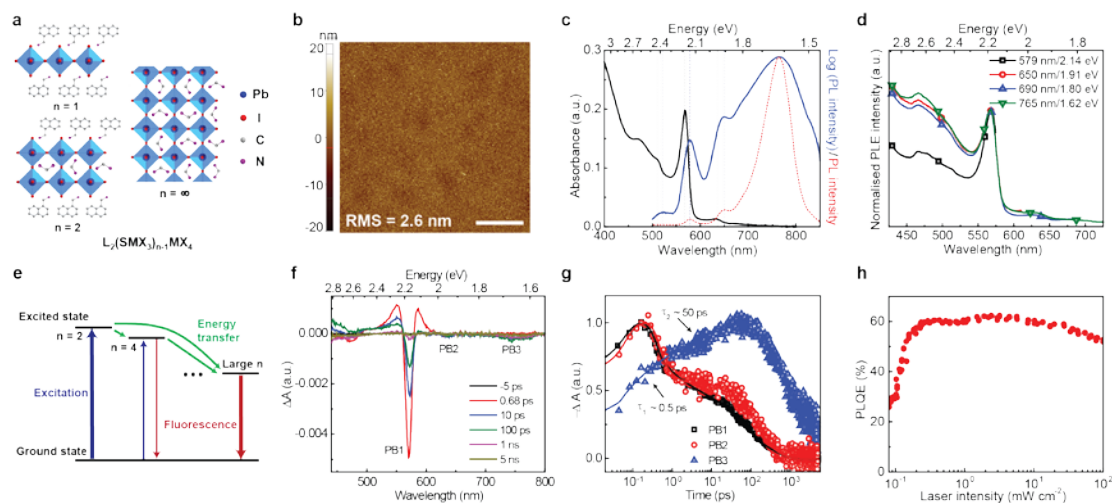


Figure 1. Perovskite MQW films. **a**, Schematic representation of the structures of the layered lead halide perovskites with $n=1$, $n=2$ and $n=\infty$. **b**, An AFM height image of the NFPI₇ MQW film (scale bar: 5 μm). **c**, Absorption and PL (445 nm excitation) spectra of the NFPI₇ MQW film deposited onto quartz substrates. The PL spectrum is plotted in log scale (blue curve) and linear scale (red curve). The three pairs of the dashed lines (from left to right) correspond to the absorption (black) and PL (red) peaks from the QWs with $n=1$, 2 and 4, respectively. **d**, PL excitation spectra of the NFPI₇ MQW film at various emission energies. a.u., arbitrary units. **e**, Schematic of cascade energy transfer in MQWs. Excitation energy is transferred downstream from lower- n QWs to larger- n QWs, and the emission is mainly from larger- n QWs. **f**, TA spectra of the MQWs at selected probe delay times. **g**, Normalised bleaching kinetics at 2.18 eV (black Square), 1.95 eV (red Circle) and 1.64 eV (blue Triangular) for the MQWs following excitation at 400 nm (1 KHz, 50 fs, $\sim 0.4 \mu\text{J cm}^{-2}$). **h**, Excitation-intensity-dependent PLQE of the NFPI₇ MQW film.

Figure 2

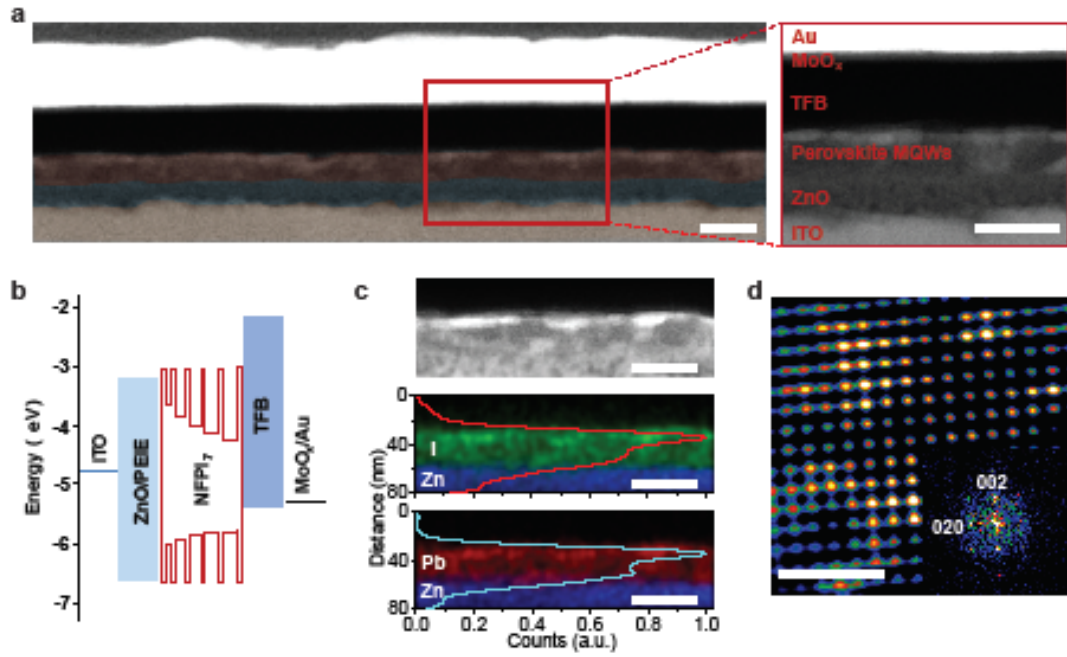


Figure 2. Device structure of the NFPI₇ perovskite MQW LED. **a**, A cross-sectional scanning transmission electron microscopy (STEM) image (scale bar: 50 nm) showing the device architecture: ITO/PEIE modified ZnO (~20 nm)/NFPI₇ (~30 nm)/TFB (~40 nm)/MoO_x (~7 nm)/Au. The different layers in the left panel have been tinted. **b**, Flat-band energy level diagram. Except NFPI₇, energy level values of other layers were taken from the literature⁹. **c**, EDX mapping. Color-mixed EDX mapping images (scale bar: 50 nm) present the element distribution of lead (Red), iodide (Green) and zinc (Blue). Normalised EDX counts distribution of lead and iodide across the perovskite layer are also presented. **d**, Fast-Fourier-transform-filtered HRTEM image (scale bar: 2 nm) and FFT analyses (inset) of a perovskite crystal close to the TFB/perovskite interface show that this crystal has a cubic structure similar to that of the 3D α -FAPbI₃²².

Figure 3

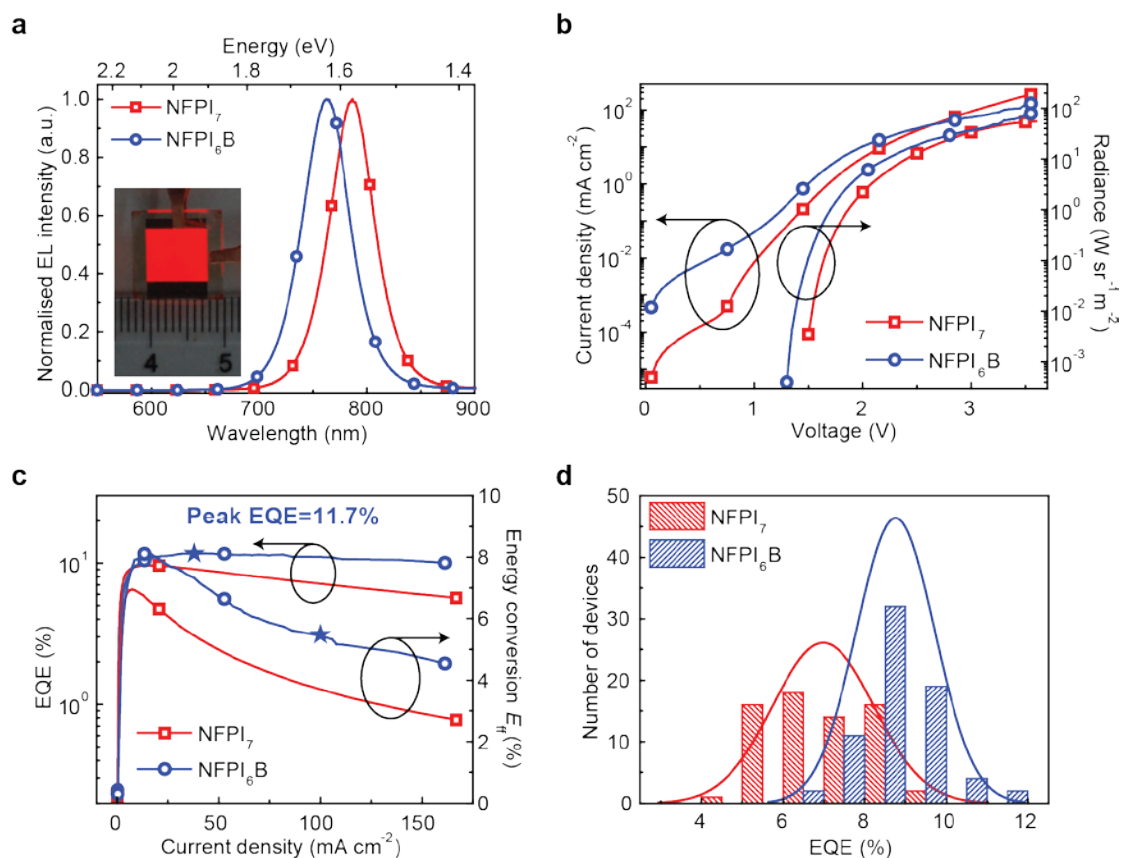
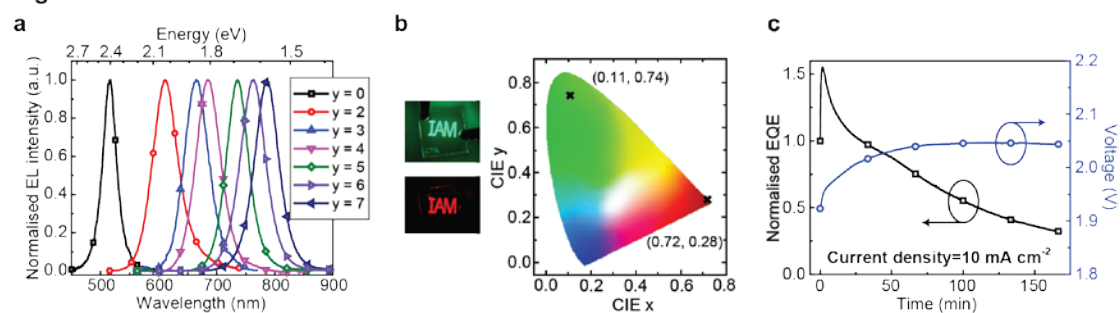


Figure 3. Optoelectronic characteristics of the NFPI₇ and NFPI₆B perovskite MQW LEDs. **a**, EL spectra of the devices under 3 V. Inset: a photograph of a 64-mm² NFPI₆B device. **b**, Dependence of current density and radiance on the driving voltage. Radiance of 55 W sr⁻¹ m⁻² and 82 W sr⁻¹ m⁻² are obtained under 3.6 V for NFPI₇ and NFPI₆B devices, respectively. **c**, EQE and energy conversion efficiency versus current density. For the NFPI₆B MQW LED, a peak EQE of 11.7% is achieved at a current density of 38 mA cm⁻² and a energy conversion efficiency of 5.5% is obtained at a current density of 100 mA cm⁻². **d**, Histograms of peak EQEs measured from 67 NFPI₇ and 70 NFPI₆B devices.

Figure 4**Figure 4. EL color tunability and device stability of the perovskite MQW LEDs.**

a, EL spectra of MQW LEDs based on the NFPI_yB_{7-y} films, $y = 0, 2, 3, 4, 5, 6$ and 7 , respectively, which are processed using various combinations of precursor solutions of NMAI (NMABr), FAI (FABr) and PbI₂ (PbBr₂) (refer to Supplementary Table S1 for details). **b**, Photographs of green and red MQW LEDs with the logo of the Institute of Advanced Materials (IAM), and the corresponding CIE coordinates. **c**, Stability data for a NFPI₇ EL device tested at a constant current density of 10 mA cm^{-2} .

Methods

Synthesis and materials preparation. NFPI₇ precursor solutions were prepared by dissolving NMAI, FAI and PbI₂ with a molar ratio of 2:1:2 in DMF (10 wt.%) and stirred at 60 °C for 2 h in a nitrogen-filled glovebox. The NMAI was synthesized by adding 4.34 g hydroiodic acid (45 wt.% in water) to a stirring solution of 1-naphthalenemethylamine (12.72 mmol) in tetrahydrofuran (THF, 50 mL) at 0 °C for 2 h. The solution was then evaporated at 50 °C to obtain the NMAI precipitate, which was washed three times with THF:CH₂Cl₂ (3:1) mixture and then dried under vacuum. NMABr was synthesized by using similar method with the hydroiodic acid substituted by the hydrobromic acid. NFPI_yB_{7-y} precursor solutions were prepared by altering composition of NMAI (NMABr), FAI (FABr), and PbI₂ (PbBr₂) with a molar ratio of 2:1:2 in DMF (10 wt.%), except for NFPI₂B₅ from a precursor solution with a molar ratio of 2:1:1:1 in DMF (10 wt.%) of NMABr:FABr:PbBr₂:PbI₂.

(NMA)₂PbI₄ precursor solutions were prepared by dissolving NMAI and PbI₂ with molar ratio of 1:1 in DMF (10 wt.%) and stirred at 60 °C for 2 h in a glovebox.

Colloidal ZnO nanocrystals were synthesized by a solution-precipitation process and the details can be found in the literature⁹.

Device fabrication. Solutions of ZnO nanocrystals were spin-coated onto the ITO-coated glass substrates at 4,000 rpm for 45 s and annealed in air at 150 °C. Then the substrates were transferred into a glovebox. Next, a solution of PEIE in 2-methoxyethanol (0.4 wt.%) was spin-coated onto the ZnO films at a speed of 5,000 rpm. The substrates were rinsed twice with DMF, leaving ultrathin layers of PEIE on top of the ZnO films. The ultrathin layer of PEIE can significantly decrease the work function of ZnO²⁷. The perovskite films were prepared by spin coating the precursor solution onto the PEIE treated ZnO films, followed by annealing on a hot plate at 100 °C. The TFB layers were deposited from an m-xylene solution (8 mg mL⁻¹) at 2,000 rpm. Finally, the MoO_x/Au electrode was deposited using a thermal evaporation system through a shadow mask under a base pressure of ~6×10⁻⁷ Torr. The device area was 3 mm² as defined by the overlapping area of the ITO films and top electrodes.

Characterization. All perovskite LED device characterizations were carried out at room temperature in a nitrogen-filled glovebox. A Keithley 2400 source meter and a fiber integration sphere (FOIS-1) couple with a QE65 Pro spectrometer was used for the measurements²⁸. The LED devices are tested on top of the integration sphere and only forward light emission can be collected, which is consistent with the standard OLED characterization method²⁹. Figure S9 shows the layout of our device characterization setup. The devices were swept from zero bias to forward bias. The time evolution of EQEs and voltages were measured using the same testing system. The angular dependence of emission intensity was measured by using a Thorlabs PDA100A detector with a fixed distance of 200 mm away from the EL device. To verify the accuracy of the measurement setup, we have cross-checked the performance of two sets of perovskite LEDs prepared under identical conditions but separately measured at Nanjing Tech University and University of Cambridge. The results obtained in the two different labs are in good agreement.

AFM images were collected in non-contact mode (Park XE7). The STEM images and EDX elemental mapping of the perovskite films were made on a FEI Titan G2 80-200 ChemiSTEM operated at 200 keV. An element determination with outstanding sensitivity can be provided by ChemiSTEM EDX system. The HRTEM images were collected on a Tecnai G2 F20 microscope operated at 200 keV. The cross-sectional samples were prepared by using dual beam focused-ion-beam equipment (FEI Quata 3D FEG).

UV-vis absorbance spectra were recorded on a UV-vis spectrophotometer with an integrating sphere (Cary 5000, Agilent). PL spectra of the perovskite films were measured at room temperature using a fluorescent spectrophotometer (F-4600, HITACHI) with a 200 W Xe lamp as an excitation source. The excitation spectra and time resolved fluorescence spectra of the perovskite films were obtained by using an Edinburgh Instruments (FLS920) spectrometer. For the time-resolved PL measurements, the perovskite films were excited from the quartz glass substrate side by a 405 nm pulsed diode laser (EPL-405) with a fluence of $\sim 4 \text{ nJ cm}^{-2}$. A three-step technique was used to obtain PLQE of perovskite films by combination of a 445 nm

CW laser, optical fiber, spectrometer and integrating sphere³⁰.

For transient absorption measurements, the laser source was a Coherent LibraTM regenerative amplifier (50 fs, 1 KHz, 800 nm) seeded by a Coherent VitesseTM oscillator (50 fs, 80 MHz). 800 nm wavelength laser pulses were from the regenerative amplifier while 400 nm wavelength laser pulses were obtained with a BBO doubling crystal. Broad band femtosecond transient absorption spectra of the perovskite films were taken using the Ultrafast System HELIOSTM transient absorption spectrometer. The broadband probe pulses (420-800 nm) were generated by focusing a small portion ($\sim 10 \mu\text{J}$) of the fundamental 800 nm laser pulses into a 2 mm sapphire plate. The thickness of the perovskite films is ~ 30 nm (same as in EL devices) for the above optical characterizations.

UPS spectra were collected on a Thermo ESCALAB-250Xi spectrometer with an applied bias of -10 V. He I UV radiation source (21.2 eV) was used. The overall resolution of the instrument is 0.1 eV.

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