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Photoionization dynamics of *cis*-dichloroethene from investigation of vibrationally resolved photoelectron spectra and angular distributions

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The influence of vibronic coupling on the outer valence ionic states of *cis*-dichloroethene has been investigated by recording photoelectron spectra over the excitation range 19–90 eV using plane polarized synchrotron radiation, for two polarization orientations. The photoelectron anisotropy parameters and electronic state branching ratios derived from these spectra have been compared to theoretical predictions obtained with the continuum multiple scattering approach. This comparison shows that the photoionization dynamics of the \tilde{A}^2B_2 , \tilde{B}^2A_1 , \tilde{C}^2A_2 , and \tilde{D}^2B_1 states, all of which are formed through the ejection of an electron from a nominally chlorine lone-pair orbital, exhibit distinct evidence of the Cooper minimum associated with the halogen atom. While retaining a high degree of atomic character, these orbital ionizations nevertheless display clear distinctions. Simulations, assuming the validity of the Born-Oppenheimer and the Franck-Condon approximations, of the $\tilde{X}^{2}B_{1}$, $\tilde{A}^{2}B_{2}$, and $\tilde{D}^{2}B_{1}$ state photoelectron bands have allowed some of the vibrational structure observed in the experimental spectra to be assigned. The simulations provide a very satisfactory interpretation for the $\tilde{X}^2 B_1$ state band but appear less successful for the $\tilde{A}^2 B_2$ and $\tilde{D}^2 B_1$ states, with irregularities appearing in both. The \tilde{B}^2A_1 and \tilde{C}^2A_2 state photoelectron bands exhibit very diffuse and erratic profiles that cannot be reproduced at this level. Photoelectron anisotropy parameters, β , have been evaluated as a function of binding energy across the studied photon energy range. There is a clear step change in the β values of the \tilde{A}^2B_2 band at the onset of the perturbed peak intensities, with β evidently adopting the value of the $\tilde{B}^2 A_1$ band β . The $\tilde{D}^2 B_1$ band β values also display an unexpected vibrational level dependence, contradicting Franck-Condon expectations. These various behaviors are inferred to be a consequence of vibronic coupling in this system. Published by AIP Publishing. https://doi.org/10.1063/1.5042216

I. INTRODUCTION

The three isomers (*cis*, *trans*, and *iso*) of dichloroethene ($C_2H_2Cl_2$) provide an ideal means of studying substitution effects on the valence shell electronic structure and photoionization dynamics. The present study concerns *cis*-1,2-dichloroethene and reveals that vibronic interactions between neighbouring ionic states^{1,2} not only influence photoelectron band envelopes and the associated vibrational structure but also influence the photoelectron angular distributions.

The photoelectron spectrum of the unsubstituted parent molecule (ethene, C₂H₄) was one of the first in which the influence of vibronic interactions on the electronic structure was studied in detail.^{3,4} A good summary of our current understanding of these effects on the ionic states of ethene has been given by Hazra and Nooijen.^{5,6} The theoretical work shows that although vibronic coupling occurs between the ground ionic state and the first excited ionic state, non-adiabatic effects in the ionic ground state are weak. The calculated \tilde{X} state photoelectron spectrum⁵ reproduced the vibrational structure observed experimentally.⁷ In contrast, the experimental photoelectron spectra for the \tilde{A} , \tilde{B} , and \tilde{C} states of ethene are difficult to interpret, exhibiting only a few broad peaks.⁷ These diffuse photoelectron bands, which are simulated satisfactorily in the vibronic coupling calculations,⁶ are the result of significant non-adiabatic effects in all three excited ionic states.

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It is now well established that vibronic interactions between ionic states can radically alter the vibrational structure in a photoelectron band from that expected from models employing the Born-Oppenheimer and Franck-Condon approximations.¹ Such models are based upon the concept that each electronic state has an associated isolated potential energy surface. Vibronic coupling, namely the interaction of two or more energetically close-lying electronic states through the nuclear motion, results in a breakdown of the Born-Oppenheimer approximation. The ensuing nonadiabatic effects, which account for the motion of the nuclei on more than one potential energy surface, can lead to changes in the vibrational structure. These can vary between slightly irregular vibrational energy spacings and/or peak intensitiesbut still recognisable vibrational progressions-in the case of weak vibronic coupling to a complete loss of any distinguishable vibrational pattern for strong vibronic coupling. In the latter case, vibronic coupling calculations predict numerous vibrational excitations, some of which involve nontotally symmetric vibrational modes. The summation of these excitations can result in a diffuse photoelectron band.

Vibronic coupling has also been predicted to affect the photoionization dynamics^{8,9} as characterized by the photoelectron anisotropy parameter (β). Boron trifluoride provides a good example of such behavior.¹⁰ Franck-Condon forbidden excitation of a single quantum of the non-totally symmetric $v_3^+(e')$ mode in the $\tilde{E}^{-2}A_1'$ state photoionization can be ascribed to vibronic coupling between the \tilde{D} ²E' and the \tilde{E} ²A₁' ionic states. The anisotropy parameters for these perturbed $\tilde{E}^2 A_1'$ state excitations were similar to those of the \tilde{D} ²E' state and differed considerably from those of unperturbed $\tilde{E}^{2}A_{1}'$ excitations. Thus, the experimental results confirmed the predictions by Haller et al.⁸ that the anisotropy parameters associated with photoelectron peaks induced by vibronic coupling should reflect those of the state to which they are coupling and from which they derive their intensity.

In this paper, we present experimental results, including photoelectron spectra, electronic branching ratios, and anisotropy parameters, obtained for *cis*-dichloroethene. The vibrational structure is analyzed at the Franck-Condon level of approximation with the help of simulations of the vibrational envelope, and the photoionization dynamics are likewise modeled using electron multiple scattering calculations.¹¹ In a companion paper,¹² we give a more complete theoretical treatment of the vibronic interactions coupling the $\tilde{A} \, {}^{2}B_{2}$, $\tilde{B} \, {}^{2}A_{1}$ and $\tilde{C} \, {}^{2}A_{2}$ states and the $\tilde{D} \, {}^{2}B_{1}$ and $\tilde{E} \, {}^{2}B_{2}$ states that can be inferred from the present results and analysis. Consideration of the inner valence region is also deferred to that paper.

The outer valence shell orbital sequence of *cis*dichloroethene in its molecular ground state may be written as (using C_{2v} symmetry)

$$(9a_1)^2(8b_2)^2(2b_1)^2(2a_2)^2(10a_1)^2(9b_2)^2(3b_1)^2.$$

Mulliken atomic populations¹³ allowing the character of these molecular orbitals to be assessed are reproduced in Table I, and visual representations of these orbitals are given in Fig. 1. The outermost $3b_1$ orbital has a mixed carbon and chlorine character, but the next four lower occupied orbitals (9 b_2 ,

TABLE I. Mulliken atomic population in the outer valence molecular orbitals of *cis*-dichloroethene (units are electrons; sum over all atoms is 2) calculated at the HF/cc-pVTZ level.¹²

	$3b_1$	$9b_2$	$10a_1$	$2a_2$	$2b_1$	$8b_2$	$9a_1$	8 <i>a</i> 1
Atom	(π)	(σ_{ClLP})	(σ_{ClLP})	(π_{ClLP})	$(\pi_{Cl LP})$	(σ)	(σ)	(σ)
С	0.58	0.04	0.08	0.06	0.40	0.33	0.39	0.59
Н	0.01	0.02	0.05	0.0	0.0	0.15	0.0	0.28
Cl	0.41	0.94	0.87	0.94	0.59	0.52	0.61	0.13

 $10a_1$, $2a_2$, and $2b_1$) contain a significant Cl 3p content. These four orbitals comprise essentially two in-plane and two outof-plane chlorine lone-pairs, and for brevity we will henceforth use σ or π to identify, respectively, the in-plane or out-of-plane orientation of the halogen "p" orbitals. Photoelectron bands associated with nominally lone-pair orbitals are usually dominated by an intense adiabatic transition and have little accompanying vibrational structure.¹⁴ However, our experimental and simulated spectra show that this is not the case for these orbitals (9 b_2 , $10a_1$, $2a_2$, and $2b_1$) in *cis*-dichloroethene.

The high Cl 3p content in the lone-pairs will manifest itself in the photoelectron angular distributions and the photoionization partial cross sections associated with these molecular orbitals. Specifically, the Cooper minimum^{15,16} associated with a Cl 3p orbital¹⁷ will, in the atomic limit, result in a fairly isotropic photoelectron angular distribution and a reduction in the partial cross section in the photon energy range around 40 eV. Such effects have also been observed previously in chlorine containing molecules^{18–21} and can be used to assess the extent to which the atomic orbital is modified by the molecular environment.

Previous experimental investigations on *cis*-dichloroethene include photoelectron spectra recorded with HeI^{22–26} HeII,²⁶ Al K α ,²⁷ and synchrotron²⁸ radiation, mass analyzed threshold ionization (MATI)²⁹ and pulsed field ionization photoelectron (PFI-PE)³⁰ spectra of the ground ionic state, and electron momentum spectroscopy.³¹ Fragmentation studies have also been performed.^{32,33} Theoretical predictions



FIG. 1. The outer valence orbitals of cis-dichloroethene.

for the orbital binding energies^{25,26,34} and the valence shell photoelectron spectra^{26,34,35} have been obtained.

The HeI excited photoelectron spectra of *cis*-dichloroethene reveal that some of the outer valence bands exhibit vibrational progressions, although the rather low resolution employed in these studies limits the observable structure. Assignments were proposed through comparison of vibrational energies measured in the ionic states with known molecular ground state values. Such procedures often provide a satisfactory overall interpretation of the vibrational structure associated with isolated electronic states, but, as will be shown, more sophisticated methods are required for perturbed states.¹²

II. EXPERIMENTAL APPARATUS AND PROCEDURE

Experimental measurements were made using a VG Scienta R4000 hemispherical electron energy analyser mounted on the soft X-ray undulator-based PLÉIADES beamline at the SOLEIL synchrotron radiation facility and followed procedures which we have previously reported.^{36,37} Further details concerning our use of this beamline and endstation are provided as the supplementary material.

Under the conditions employed here, the anticipated optical resolution varied between 0.9 meV at hv = 20 eV and 5.0 meV at hv = 90 eV, while the electron spectrometer resolution was predicted to be 5 meV. Translational Doppler broadening also contributes to the overall observed peak width.³⁸ Such contributions amount to 4.1 and 11.4 meV for electrons ejected from *cis*-dichloroethene with kinetic energies of 10.3 or 80.3 eV (corresponding to the formation of the \tilde{X} ²B₁ state in the vibrationally unexcited level using photon energies of 20 or 90 eV, respectively).

Dichloroethene is a liquid with a significant vapour pressure at room temperature. The vapour was introduced into the experimental chamber, without heating, after removing dissolved air and volatile impurities.

The normalised photoelectron angular distribution, $I(\theta)$, can be written as

$$I(\theta) = 1 + \beta P_2(\cos \theta),$$

where P_2 is the second Legendre polynomial, and θ is the angle between the electron emission and the polarization vector. We extract the anisotropy parameter, β , from spectra recorded with polarizations set perpendicular and parallel to the spectrometer axis, as described in the supplementary material. Also described there is the method to simulate so-called "magic angle" spectra having relative intensities that are independent of any angular variations (β parameters).

III. COMPUTATIONAL DETAILS

Geometry optimization, and thence vertical ionization energies and vibrational parameters, were calculated using standard methods, as noted below and implemented in the Gaussian09 package.³⁹ Harmonic normal mode analyses for the ground state neutral and ion were obtained using density functional theory (DFT) calculations with the B3LYP functional and cc-pVTZ basis. For the electronically excited ion states, time-dependent (TD) DFT calculations were made using the same functional and basis. Franck-Condon simulations of the vibrational structuring of the various photoelectron bands, including Duschinsky rotation of the modes between ground and ion states, were then performed using the FC-Lab II package.⁴⁰ To facilitate visual comparison with the experiment, the calculated harmonic frequencies were scaled by a factor of 0.97 in accordance with common practice.⁴¹ To provide a more realistic appearance, but not necessarily an exact match to the experimental band profiles, the resulting stick spectra were broadened by convolution to an empirically chosen width of 75 cm⁻¹ (~10 meV FWHM).

Calculations of photoionization cross sections and photoelectron anisotropy parameters were made using the CMS-X α method, following procedures which we have described previously.¹¹ Here, we adopt the same approach and parameter choices for defining a model potential in which the electron dynamics can be investigated as in recent applications to dichlorobenzene isomers.¹⁹⁻²¹ Starting from atomic coordinates derived from a B3LYP/cc-pVTZ geometry optimization, the potential was constructed as overlapping spherical regions situated on the atomic centres in which the exchange contribution to an effective one-electron potential is represented using the X α local density approximation. Solutions are obtained as expansions in a symmetry adapted basis of spherical harmonic functions: for the neutral ground state, these expansions are truncated at l_{max} values of 6 (outer sphere), 4 (non-hydrogenic atoms), and 2 (H atoms); for the continuum electron, these values are increased to 18, 12, and 9, respectively. We have systematically checked for adequate convergence of the calculated photoionization matrix elements, and hence cross section and anisotropy parameters, against variation of these l_{max} choices. The results presented here are sensitive to neither the choice of the equilibrium geometry calculation nor the sphere radii.

IV. RESULTS AND DISCUSSION

Figure 2 shows an overview of the outer valence shell photoelectron spectrum of *cis*-dichloroethene obtained at a photon energy of 20.5 eV. The calculations reported by Trofimov *et al.*¹² find that at binding energies below 14.58 eV the single particle model of ionization⁴² is adequate; the \tilde{F}^2A_1 band may, in contrast, contain contributions from satellite states. Consequently, the lowest six bands, \tilde{X} – \tilde{E} , will receive most attention in the following discussion.

A. Vibrational analysis of the photoelectron bands

Most of the photoelectron bands due to the outer valence orbitals display vibrational structure. Our calculated vibrational frequencies for the neutral molecule and the \tilde{X} ²B₁, \tilde{A} ²B₂ and \tilde{D} ²B₁ cation states are listed in the supplementary material (Tables S-I and S-II). In discussing the vibrational excitations accompanying ionization from the vibrationally unexcited level in the neutral ground state, the vibrational transition is denoted as n^k (or for hot bands, n_l^k) where nis the number of the vibrational mode, and k is the number of vibrational quanta in the final state (and l is the number of vibrational quanta in the initial state—if not 0). The



FIG. 2. The outer valence shell photoelectron spectrum of *cis*-dichloroethene recorded at a photon energy of 20.5 eV. States are labeled as being due to ionization from a single molecular orbital. Also marked along the bottom axis are the calculated OVGF/cc-pVQZ vertical ionization energies.

vibrational modes are numbered as recommended by Herzberg.⁴³ This convention can, however, create some apparent inconsistencies between the mode labeling in the neutral ground state and the cation structures, particularly for the \tilde{D} state ion. This, we calculate, distorts to a double minimum, non-planar (C₂) equilibrium geometry. In the lowered C₂ point group, vibrational modes that were of a_1 and a_2 symmetry in the C_{2v} neutral molecule descend to *a* symmetry and so are merged in a common, reordered list according to the Herzberg prescription, with an analogous adjustment for the b_1 and b_2 symmetry modes. Should the number *n* consequently designate quite dissimilar vibrations in the initial and final states, to avoid ambiguity when discussing hot band excitations we will explicitly designate the neutral ground state as \bar{n}_l , with the subscript *l* indicating the number of vibrational quanta in the initial *neutral* mode \bar{n} . Elsewhere v'' and v^+ are used to label vibrational levels in the neutral and ionic states, respectively.

1. The $\tilde{X}^2 B_1$ state

The vibrational analysis of the $(3b_1)^{-1} \tilde{X} {}^2B_1$ state photoelectron band (Fig. 3) appears relatively straightforward, and our 0 K Franck-Condon simulations are in good agreement with the experimental spectrum. Figure 4 shows an expanded view of the photoelectron peak arising mainly from the adiabatic (0–0) transition to the $\tilde{X} {}^2B_1$ state. Although this peak is fairly symmetric, it displays shoulders on both the low and the high binding energy sides, with a small but distinct additional feature at an energy (9.57 eV) falling clearly below the adiabatic transition energy. It is readily deduced that much of the additional experimental structure in this region can be attributed to hot bands originating from thermally populated, low frequency vibrational levels in the molecular ground state.

A hot band simulation in the 0-0 origin region, calculated using harmonic frequencies as described previously for bromobenzene,³⁷ is shown in Fig. 4. As well as transitions originating from the vibrationless neutral level, transitions originating from the $v_4'' = 1$, $v_5'' = 1$, $v_7'' = 1$, $v_8'' = 1$,



FIG. 3. Experimental \tilde{X} ²B₁ state photoelectron band of *cis*-dichloroethene (recorded with hv = 20.5 eV) and a cold (0 K) Franck-Condon simulation using B3LYP/cc-pVTZ calculated vibrational modes, with harmonic frequencies scaled by a factor of 0.985. The calculated stick spectrum is convoluted with a shaping function to provide more realistic visual comparison. Also shown is a second simulated spectrum that includes hot bands arising from an assumed 300 K thermal population of the lower frequency neutral ground state modes.

 $v_{12}'' = 1$, $v_5'' = 2$, and $v_7'' = 2$ levels are included. Transitions from other vibrational levels were ignored because their initial populations were predicted to be negligible. The stick heights (proportional to the photoelectron intensity) are given by the Franck-Condon factor for the specific transition multiplied by the appropriate Boltzmann population at 300 K. These simulations indicate that additional features observed in this origin region arise principally from sequence bands having $\Delta v = 0$, particularly those involving the v_5'' or v_7'' modes.⁴⁴



FIG. 4. Expanded vibrational origin region of the \tilde{X} ²B₁ photoelectron band of *cis*-dichloroethene (Fig. 2) comparing the experimental spectrum (recorded with hv = 20.5 eV) and B3LYP/cc-pVTZ Franck-Condon simulations. The latter are shown both as a stick spectrum and convoluted with Gaussian functions having either 13 meV or 15 meV FWHM for more realistic comparisons. The stick spectrum is colour coded to indicate those transitions having a common lower level, and the assignments for the most prominent peaks (similarly colour coded) are included in the figure. For clarity, small vertical offsets are applied between the curves showing experimental and simulated spectra.

The Franck-Condon stick spectrum in Fig. 4 has been convoluted with a Gaussian function of either 13 or 15 meV (FWHM) to achieve a more realistic appearance. At this photon energy, 20.5 eV, the theoretically estimated broadening of the photoelectron peak in cis-dichloroethene would be expected to have contributions of 4 meV from the translational Doppler effect, 5 meV from the spectrometer bandpass, and 1 meV from the monochromator bandpass, giving an overall estimated resolution of ~6.5 meV. Under these same experimental conditions, we recorded a width of 8.8 meV for the peak due to the $(4p)^{-1}$ ²P_{3/2} state in the photoelectron spectrum of krypton.³⁷ The 13–15 meV width chosen for the convolution of the stick spectrum exceeds these estimates of the instrumental resolution, which may also suggest that some broadening results from the (unresolved) rotational envelope and possibly also from a rotational Doppler effect.45

Significantly, after convolution with these shaping functions, the maximum of the resulting composite peak is found to be shifted from the adiabatic origin by 0.7 meV toward higher energy. An adiabatic ionization energy of 9.6578 eV was determined in the MATI experiment.²⁹ Therefore, we have calibrated our binding energy scale by setting the maximum of the first peak in our measured \tilde{X} ²B₁ state photoelectron band to 9.6585 eV (this calibration has been applied to all spectra presented here).

The 300 K hot band simulation is shown spanning the whole \tilde{X} ²B₁ state photoelectron band in Fig. 3, where it can be seen to further improve the already good agreement with the experimental spectrum. Guided by the full simulation, the principal structure across the complete \tilde{X} ²B₁ photoelectron band can be assigned to six progressions involving the v₂⁺, v₃⁺, and v₄⁺ modes, either alone or in combination with each other, and these are indicated in Fig. 5 (experimental binding energies and assignments are collected and listed in Table S-III of the supplementary material). We obtain energies of 178, 150, and 100 meV for the v₂⁺, v₃⁺, and v₄⁺ modes, respectively. These values have been obtained simply as the difference between the binding energy, quoted with an



FIG. 5. The *cis*-dichloroethene \tilde{X} ²B₁ photoelectron band with vibrational assignments of transitions from the vibrationless neutral ground state shown. Some identified hot band assignments are also shown (see Fig. 4).

uncertainty of ± 1 meV, of the photoelectron peak due to the 2^1 , 3^1 , or 4^1 transition, and that of the peak due to the adiabatic transition. All our reported vibrational energies have been estimated in this manner. These values are consistent with those of 177.1, 148.3, and 100.3 meV, respectively, determined in the MATI study.²⁹ In the neutral ground state, $\tilde{X}^{1}A_{1}$, the corresponding vibrational energies are 196.8, 146.2, and 88.2 meV.⁴⁴ In addition to the main progressions, our hot band simulations suggest that the feature observed at 9.64 eV can be attributed primarily to transition 7_1^1 with weaker contributions from 5_1^0 and 5_2^1 . Similarly, a feature on the opposite side of the adiabatic peak, at 9.681 eV, is attributed to a group of close transitions 5_n^{n+1} , where n = 0,1,2. A further weak feature at 9.729 eV is ascribed in part to the transition 7^2 . Transitions involving the v_7 mode, which is of a_2 symmetry, were also observed in the MATI experiment.²⁹ Finally, a predicted transition $7_1^1 4^1$ at 9.74 eV may just be discerned in Fig. 5 as a weak shoulder on the side of the second principal peak at 9.76 eV.

2. The $\tilde{A}^2 B_2$, $\tilde{B}^2 A_1$ and $\tilde{C}^2 A_2$ states

Figure 6 shows the $\tilde{A}^{-2}B_2$ state photoelectron band, together with a 0 K Franck-Condon simulation. The simulations predict that the vibrational structure arises mainly from three progressions, all involving the v₅⁺ mode, with two of the progressions also involving the excitation of an additional one or two quanta of the v₄⁺ mode. Binding energies of the principal vibrational peaks seen in the experimental spectrum are collected in Table S-IV (supplementary material). A vibrational energy of ~27 meV for the v₅⁺ mode can be derived from the experimental spectrum, in good agreement with the calculated value of 26.9 meV (see Table S-I of the supplementary material).

In the preparation of Fig. 6, the simulated spectrum energy scale has been aligned such that the peak due, principally, to the



FIG. 6. Experimental photoelectron spectrum of *cis*-dichloroethene in the \tilde{A}^2B_2 band region, (recorded with hv = 20.5 eV) and a cold (0 K) Franck-Condon simulation using B3LYP/cc-pVTZ calculated vibrational modes, with harmonic frequencies scaled by a factor of 0.968. The principal structure is identified as three long progressions in the v_5^+ mode, building on zero (magenta), one (olive), or two (navy) quanta of the v_4^+ mode. The calculated stick spectrum is convoluted with a shaping function to provide more realistic visual comparison.

 5^{29} excitation coincides with the most intense peak occurring at 11.706 eV in the experimental spectrum. As our simulations place the 5^{29} excitation 0.754 eV above the adiabatic transition, this suggests an adiabatic ionization energy of 10.95 eV. However, this is effectively a long extrapolation back down the ladder of harmonic oscillator levels, and so its reliability may be questioned. In our companion paper,¹² we obtain by calculation an alternative estimate of 11.43 eV for the \tilde{A} ²B₂ state adiabatic ionization energy.

While the $\tilde{A} {}^{2}B_{2}$ state band (Fig. 6) seems to exhibit fairly regular vibrational structure in the low binding energy region up to ~11.76 eV, there is a clear perturbation above this energy that is not seen in the Franck-Condon model. Beyond this, the experimental spectrum starts to merge with the adjacent $\tilde{B} {}^{2}A_{1}$ state band.

Inspection of the complex \tilde{B}^2A_1 and \tilde{C}^2A_2 state photoelectron bands (Fig. 2) shows neither has recognizable regular vibrational structure, and Franck-Condon simulations performed as previously offer no meaningful comparisons. According to our calculations,¹² the binding energy region between 11.2 and 13.0 eV encompasses only the \tilde{A}^2B_2 , \tilde{B}^2A_1 and \tilde{C}^2A_2 states and is devoid of satellites. We may thus infer the possibility of vibronic interstate interactions in the region above ~11.76 eV, but the photoelectron anisotropy parameters measured across this binding energy region will provide further insight (*vide infra*).

The influence of vibronic coupling, which we therefore infer from our experimental results, has been examined theoretically in a companion paper.¹² These calculations predict conical intersections between the potential energy surfaces associated with the \tilde{A} ²B₂ and \tilde{B} ²A₁ states and those associated with the \tilde{B} ²A₁ and \tilde{C} ²A₂ states, respectively, at 11.87 and 12.39 eV. These results go some way to corroborate that the low binding energy region of the \tilde{A} ²B₂ state photoelectron band might be interpreted within the adiabatic approximation. In contrast, the bands corresponding to the \tilde{B} ²A₁ and \tilde{C} ²A₂ states should be strongly affected by non-adiabatic dynamics, and this leads to the observed highly complex vibronic structure.

3. The $\tilde{D}^2 B_1$ and $\tilde{E}^2 B_2$ states

The $\tilde{D}^{2}B_{1}$ state photoelectron band shows rather distinct vibrational structure in the low binding energy region but, as seen in Fig. 2, this merges with the essentially featureless $\tilde{E}^{2}B_{2}$ state band to higher energy, losing such definition. Vibronic coupling calculations¹² predict that the \tilde{D} ²B₁ and $\tilde{E}^{2}B_{2}$ states lie virtually on top of each other with a very low point of conical intersection. Hence, complicated vibrational profiles might be expected for these two states. Nevertheless, it again appears worthwhile attempting to interpret some of the observed structure using Franck-Condon simulations for the \tilde{D} ²B₁ state. Calculations reveal that the \tilde{D} ²B₁ state cation adopts a twisted non-planar equilibrium geometry (C₂ symmetry); the corresponding harmonic vibrational frequencies appear in Table S-II (supplementary material). Both cc-pVDZ and cc-pVTZ basis sets were used for calculations, the former results being used here as providing a very marginally better fit with the experimental spectrum. A 300 K hot band spectrum was computed, as for the \tilde{X} band,

by the inclusion of transitions originating from the thermal single quantum excitation of v_5'' , v_6'' , v_7'' , v_{11}'' , and v_{12}'' and double quanta excitation of v_5'' and v_6'' . Transitions from other vibrational levels were ignored because their initial populations were predicted to be negligible. The \tilde{D} ²B₁ band simulation is compared with the experimental spectrum in Fig. 7.

The simulations provide a reasonably good description of the overall shape of the observed photoelectron band, although some perturbed intensities and, to a lesser extent, energy shifts are evident. An interpretation of the vibrational structure based simply on the measured binding energies suggests that the experimental spectrum can be associated with three progressions. A comparison with the calculated vibrational energies (Table S-II of the supplementary material) indicates that two of these progressions should be assigned as 5^n and 6^15^n . This results in energies of 46 and 73 meV for the v_6^+ and v_5^+ modes, respectively, in good agreement with the B3LYP/cc-pVDZ calculated values of 48.0 and 73.4 meV. The assignment of the third progression is less certain but the most likely assignment seems to be $6^{2}5^{n}$. The binding energies of the structure belonging to these three progressions are given in Table S-IV (supplementary material), where the progressions are labeled P2–P4. In addition to these three progressions, at energies below that due to the adiabatic transition, peaks occurring at 13.496 and 13.537 eV probably arise as hot band transitions from v_4'' and v_7'' (note that because of the differing electronic state symmetries these neutral molecule vibrational modes are not the same as v_4^+ and v_7^+ in the $\tilde{D}^2 B_1$ state ion). An assignment of 7^1 for the small peak observed at 13.610 eV is suggested by the simulation results, despite a small energy mismatch.

The main difference suggested by our simulations to the interpretation given above is the participation of the v_4^+ mode, which can be categorised as a twisting torsional mode around



FIG. 7. Experimental photoelectron spectrum of *cis*-dichloroethene in the \tilde{D}^2B_1 band region (recorded with hv = 20.5 eV) and a Franck-Condon simulation using calculated B3LYP/cc-pVDZ harmonic frequencies, scaled by a factor 0.972. The convoluted stick spectrum (curve) includes calculated 300 K hot bands, but for clarity only the positions of transitions from the vibrationless ground state are marked. Three progressions, P2–P4, determined from the experiment are indicated and are listed in the supplementary material, Table S-IV.

the C-C axis. Our calculations indicate that the largest contribution to the peak observed at 13.677 eV corresponds to the 4¹ excitation, with the 6² excitation playing only a minor role. According to the Franck-Condon predictions, most excitations correspond to combination bands involving the v₄⁺, v₅⁺, and v₆⁺ modes, and these combination bands do not form regular progressions. Adding to the difficulty in assigning the vibrational structure is the fact that the sum of the energies for the v₄⁺ and v₆⁺ modes is almost equal to twice the energy of the v₅⁺ mode (Table S-II of the supplementary material). It should also be borne in mind, again, that any vibronic coupling between the \tilde{D} ²B₁ and \tilde{E} ²B₂ states may modify the vibrational energies considerably.

4. The F ²A₁ state

The photoelectron band associated with the $\tilde{F}^2 A_1$ state (Fig. 8) displays structure which can be assigned to two progressions, each involving a mode with an energy of ~75 meV, with one of the progressions having an additional, single excitation of a mode with an energy of ~25 meV. The observation of a mode with an energy of 75 meV, a value which is similar to that corresponding to the v₅⁺ mode in the $\tilde{D}^2 B_1$ state, suggests that the $\tilde{F}^2 A_1$ state cation may also possess C₂ symmetry. Unfortunately, we have been unable to calculate the $\tilde{F}^2 A_1$ state geometry and hence determine the energies of the vibrational modes. Therefore, two unassigned progressions, labeled P5 and P6, are shown in Fig. 8 and listed in Table S-IV (supplementary material).

B. Photoionization dynamics of the outer valence states

Photoelectron anisotropy parameters and branching ratios for the outer valence states are plotted in Figs. 9 and 10, respectively. These measurements were made by repeating the recording of the photoelectron spectra, for both horizontal and vertical linear polarization, at selected photon energies across the extended range 19–90 eV. The β parameters and branching ratios attributed to each orbital ionization were obtained by integrating over the binding



FIG. 8. Experimental $\tilde{F}^2 A_1$ photoelectron band (recorded with hv = 20.5 eV) with two identified progressions, P5 and P6.



FIG. 9. Experimental and theoretical (CMS– $X\alpha$) outer valence electronic state branching ratios for *cis*-dichloroethene. Experimental data are sampled over the binding energy ranges specified in Table II. Error bars representing the statistical uncertainty only are plotted but are usually smaller than the plotting symbol.

energy regions specified in Table II, with error bars estimated from the Poisson count statistics of the summed count.

The branching ratios (Fig. 9) appear to divide into two distinct behaviors, displaying either a minimum or a maximum around 40 eV. Of course, it should be borne in mind that as the branching ratios are normalized to unity, the minima occurring in the branching ratios for the $9b_2$, $10a_1$, and $2a_2$ orbitals will necessarily give rise to maxima in those of the other orbitals. In general, a very satisfactory agreement is obtained between the experimental and theoretical results shown in Fig. 9. (The structure in the calculated ratios below 20 eV can be discounted as largely spurious due to the staggered threshold values and the rapidly varying cross sections in this region.) The most significant discrepancy occurs for the branching ratios of the $\tilde{D}^{2}B_{1}$ and $\tilde{E}^{2}B_{2}$ states, where the measured branching ratio for the $\tilde{D}^2 B_1$ state is larger than predicted (especially around 40 eV) with that for the $\tilde{E}^{2}B_{2}$ state displaying the opposite behavior. Two factors may account for this discrepancy. The first is associated with the procedure used to analyze the experimental data, where the photoelectron intensity corresponding to a specific state is simply set by the binding energy region (Table II). This procedure makes no allowance for overlapping bands, which is likely to occur for the $\tilde{D}^2 B_1$ and $\tilde{E}^2 B_2$ states. The second factor concerns vibronic coupling where the contributions from the individual states are mixed and separation is invalid.



FIG. 10. Experimental and calculated photoelectron anisotropy parameters, β , for the outer valence states of *cis*-dichloroethene. Experimental data are sampled over the binding energy ranges specified in Table II, while the $\tilde{A} \,^2B_2$ state band data include a supplementary analysis made on a reduced size window region, 11.78 eV–11.85 eV. Error bars on the experimental results represent the statistical uncertainty only but usually are smaller than the plotting symbol. The CMS-X α calculations are plotted with an offset of +3 eV to compensate the overestimated attraction of the X α potential.

The agreement exhibited between the experimental and calculated β anisotropy parameters (Fig. 10) is also very satisfactory. Notably, the $\tilde{A} {}^{2}B_{2}$, $\tilde{B} {}^{2}A_{1}$, $\tilde{C} {}^{2}A_{2}$, and $\tilde{D} {}^{2}B_{1}$ states are predicted to have a pronounced dip at photon energies of 40–50 eV (electron kinetic energies of ~30 eV). This dip is due to the Cooper minimum associated with the atomic Cl 3p orbital. Such behavior is well understood for atomic photoionization and particularly for p-type orbitals which are relevant to the present study.¹⁷ Electric dipole selection rules restrict the orbital angular momentum of a photoelectron ejected from an

TABLE II. Energy regions used in the analysis of the photoelectron spectra recorded with plane polarized synchrotron radiation.

Orbital	Binding energy region (eV)		
3 <i>b</i> ₁ (π)	9.5–10.8		
$9b_2 (\sigma_{\text{Cl LP}})$	11.2–11.9		
$10a_1 (\sigma_{\text{Cl LP}})$	11.9–12.3		
$2a_2 (\pi_{\text{Cl LP}})$	12.3–13.1		
$2b_1 (\pi_{Cl LP})$	13.4–14.1		
8 <i>b</i> ₂ (σ)	14.1–15.0		
9 <i>a</i> ₁ (σ)	15.3–16.2		

initial orbital l to $l \pm 1$, and theoretical work has shown^{15,16} that the radial matrix element for the dominant $l \rightarrow l + 1$ transition (in our case the $p \rightarrow d$ transition) can change sign, provided that the initial wave function contains a node. Thus, at the energy coinciding with the change in sign, the photoionization partial cross section should exhibit a minimum (often referred to as a Cooper minimum). Moreover, at this energy, the photoelectron anisotropy parameter depends solely on the contribution from the l - 1 channel (s-type continuum in our case). For atomic photoionization from a p-orbital, the l - 1 channel possesses an isotropic angular distribution corresponding to $\beta = 0$. Thus, in the vicinity of the Cooper minimum associated with the Cl 3p orbital, both the anisotropy parameter and the photoionization partial cross section (branching ratio) should exhibit a characteristic and easily recognisable behavior. The degree to which these atomic properties are observable in the photoionization dynamics of molecular orbitals depends upon the extent to which the molecular orbitals retain their atomic properties.

We now consider the energy dependence of the β anisotropy parameters for the outer valence orbitals in relation to their predicted Mulliken atomic populations (Table I and Fig. 1). The experimental β parameter for the outermost $3b_1$ orbital, with the smallest predicted Cl 3p content, shows little evidence of the Cooper minimum. Instead, it exhibits a behavior typical of that observed^{18–21,37,46} and calculated^{19–21,37} for a π -type orbital, namely a rapid rise from a low value at threshold to reach a high plateau value above $h\nu \approx 50$ eV.

The influence of the Cooper minimum is, in contrast, clearly evident in the measured and calculated anisotropy parameters for the $9b_2$, $10a_1$, and $2a_2$ orbitals. Indeed, the deep minimum at a photon energy of ~40 eV is almost as pronounced as that calculated for the atomic Cl 3p orbital.¹⁷ The Mulliken atomic populations for these three orbitals indicate that they may be considered as essentially Cl 3p lone-pairs and that they retain to a large degree their atomic properties. The next orbital, $2b_1$, has a mixed chlorine and carbon character (Table I). The calculated anisotropy parameter for this orbital shows a reduced dip in the β -value in the region encompassing the Cooper minimum but also a greater deviation between theory and experiment. As already discussed, this discrepancy may be due, at least in part, to the data analysis procedure and/or to vibronic coupling. The experimental and calculated β parameters for the $8b_2$ and $9a_1$ orbitals exhibit weak minima in the vicinity of the Cl Cooper minimum, in accord with the predicted small chlorine content of these two orbitals.

Figures 11–13 provide examples to show variations in the β parameter measurements derived from the polarization dependent spectra across, respectively, the \tilde{X} , \tilde{A}/\tilde{B} , and \tilde{D}/\tilde{E} band regions. For the \tilde{X} band recorded at a photon energy of 25 eV (Fig. 11), the β parameter remains virtually constant across all the vibrational peaks in the photoelectron spectrum, and this observation is repeated at all other photon energies. This behavior is consistent with expectations from the Franck-Condon Principle, in which electronic and vibrational motions are considered fully decoupled. Although we do not show examples here, the $\tilde{C} \, {}^2A_2$ and $\tilde{F} \, {}^2A_1$ states have



FIG. 11. β parameter measurement across the \tilde{X} ²B₁ photoelectron band, recorded with 25 eV photon energy. The experimental "magic angle" photoelectron spectrum (PES) is plotted on the same energy scale for direct comparison with the measured β . Also included for comparison is the 300 K Franck-Condon simulation. Both the experimental and simulated spectra are arbitrarily scaled. The former uses an expanded vertical scaling for clarity, with the result that the most intense peaks are clipped (see Fig. 3).

similarly flat, constant β parameter curves across the respective photoelectron band profiles.

The $\hat{A}^2 B_2$ band anisotropy parameters (Fig. 12) behave rather differently. Across the low binding energy region of the band, the β parameter remains relatively constant. A shallow increase in β in the $h\nu = 31$ eV example can in fact be rather trivially attributed to the concomitant fall in electron energy as one steps through vibrational levels at a fixed photon energy; as seen in Fig. 10, there is a rapid change in the electronic (vibrationally unresolved) β parameter due



FIG. 12. Examples of the β parameter measurements across the $\overline{A}/\overline{B}$ band region of the photoelectron spectrum, recorded at two different photon energies: 19 eV (top) and 31 eV (bottom). For direct comparison, the corresponding "magic angle" photoelectron spectra (PES) are plotted on the same energy scale. Also included for reference is the 0 K Franck-Condon simulation. Both experimental and simulated photoelectron spectra are arbitrarily scaled.



FIG. 13. β parameter measurement across the \tilde{D}/\tilde{E} band region of the photoelectron spectrum, recorded with 25 eV photon energy. The experimental "magic angle" photoelectron spectrum (PES) is plotted on the same energy scale for direct comparison with the measured β . Also included for reference is the 300 K Franck-Condon simulation. Both the experimental and simulated spectra are arbitrarily scaled.

to the Cooper minimum at this photon energy, causing β to rise as the kinetic energy falls. Hence, we can safely conclude that again the Franck-Condon prediction of decoupled electronic and vibrational motion is unchallenged in this region.

However, at 11.75 eV binding energy there are rapid changes in the β values. In the examples shown in Fig. 12 there occurs a step increase in β in the hv = 19 eV recording while at hv = 31 eV a sharp decrease can be observed, but such seemingly discontinuous changes are observed across the photon energy range measured. These step changes coincide with the onset of the perturbed intensities noted in the vibrational analysis of the band (Sec. IV A 2). Prompted by this observation, our analysis of the $\tilde{A}^2 B_2$ band β parameter photon energy dependence (Fig. 10) was repeated, but using a binding energy window with reduced width, spanning just the perturbed region (11.78–11.85 eV). This subset is included in the relevant panel of Fig. 10. At the lower photon energies used for our study, the perturbed region's β parameter now exceeds that determined for the full band, while at the higher photon energies it is less, much as seen at the two specific photon energies appearing in Fig. 12.

What is visually striking in Fig. 12, and indeed at the other examined photon energies, is that when β diverges in the perturbed binding energy region of the \tilde{A}^2B_2 band it adopts the same values as the adjacent two split \tilde{B}^2A_1 band peaks around 12 eV binding energy. The earlier theoretical work performed by Haller *et al.*⁸ predicted that the photoelectron anisotropy parameter associated with a peak gaining intensity through vibronic coupling should be similar to that of the state to which it is coupled rather than that characterizing the remainder of the photoelectron band. Thus, our experimental evidence strongly suggests that the \tilde{A}^2B_2 and \tilde{B}^2A_1 states couple vibronically. This inference, based on our experimental spectra, is consistent with our theoretical results in the companion paper¹² which show that the intensity associated with the peak at 11.8 eV is derived from the neighbouring \tilde{B}^2A_1 state, to which the

 \tilde{A} ²B₂ state is vibronically coupled, rather than from the \tilde{A} ²B₂ state.

The $\tilde{D}^{2}B_{1}$ band β parameters show evidence for a rather different deviation from the expected Franck-Condon behavior with a clear variation observed across the photoelectron band profile. There is distinct structuring in the β curve in Fig. 13 that correlates with the assigned vibrational peaks in the photoelectron spectrum; similar observations apply across the photon energy range studied. Any interpretation of the variation in the β parameters will be affected by the overlap between the \tilde{D} ²B₁ and \tilde{E} ²B₂ state photoelectron bands, and the division at 14.1 eV implied in the regions listed in Table II is purely nominal. Hence some more gradual transitions in observed β values between those associated with the \tilde{D} ²B₁ and \tilde{E} ²B₂ states are fully expected. However, the irregularities noted in the \tilde{D} ²B₁ state vibrational analysis (Sec. IV A 3) hint at vibronic coupling between these two states.

In Fig. 14, we display the β parameters resolved to the first nine principal peaks in the band and assigned to \tilde{D} ²B₁ vibrational modes as indicated in Fig. 7. These peaks belong to short progressions in v₅⁺ in combination with the excitation of 0 or 1 quanta of v₆⁺ and to short progressions in v₆⁺



FIG. 14. Vibrationally resolved $\tilde{D}^2 B_1$ state β parameters, labeled according to assignments in Fig. 7. For clarity, these measurements are separately displayed as progressions in v₅⁺ in combination with 0 or 1 quanta in v₆⁺ (upper panels) and v₆⁺ in combination with 0 or 1 quanta in v₅⁺ (lower panels). Alongside the pure progression in v₆⁺, the anisotropy of the hot band $\bar{7}_1^0$ is also shown. Error bars are estimated from the Poisson counting statistics.

in combination with the excitation of 0 or 1 quanta of v_5^+ . Examining Fig. 14, it may be seen that the excitation of successive levels of v_5^+ (C–Cl stretching) results in a significant shift in the associated β parameter that exceeds our estimated statistical error bars. In contrast, the excitation of v_6^+ (skeletal flexing) does not lead to any significant variation in β . It is, however, interesting to note that the 7^0_1 hot band does appear to have β parameters with significantly lower anisotropy than that observed for excitations from the vibrationless level of the neutral. Both these v_5^+ and hot band observations contradict the Franck-Condon expectation of vibrational modes that are fully decoupled from the electronic degrees of freedom, but from foregoing comments it is clear that a full account of the vibronic coupling needs to be invoked for better understanding of the \tilde{D} $^{2}B_{1}$ - \tilde{E} $^{2}B_{2}$ state photoionization dynamics.

V. CONCLUSIONS

Plane polarized synchrotron radiation has been used to measure photoelectron spectra encompassing the outer valence electronic states of *cis*-dichloroethene, allowing for the analysis of the vibrational structure, photoelectron angular distributions, and electronic state branching ratios. The adiabatic approximation appears to be valid for the \tilde{X} ²B₁ state, and the vibrational structure observed in the experimental spectrum has been interpreted successfully using Franck-Condon simulations. Similar vibrational analyses have been attempted for the low energy portions of the \tilde{A} ²B₂ and \tilde{D} ²B₁ state bands, although perturbations and intensity irregularities are noted, providing a first hint of vibronic interactions with adjacent states.

Using two polarization orientations at multiple selected photon energies across the 19-90 eV range, the energy dependence of photoelectron anisotropy parameters and branching ratios has been determined. The β parameters associated with the $9b_2$, $10a_1$, $2a_2$, and $2b_1$ orbitals, all of which possess a significant Cl 3p character, exhibit an energy dependence reminiscent of that expected for atomic chlorine. Specifically, the β parameters and branching ratios pass through a minimum, at a photon energy of ~40 eV, associated with the Cooper minimum, in good agreement with theoretical calculations for this molecule. Dips in the electronic branching ratio of the $\tilde{A}^{2}B_{2}, \tilde{B}^{2}A_{1}$, and $\tilde{C}^{2}A_{2}$ states, again attributable to the Cooper minimum, also agree well with theoretical predictions. However, the expected dip in the \tilde{D} ²B₁ state branching ratio is not observed. Neither the $\tilde{D}^2 B_1$ nor the $\tilde{E}^2 B_2$ state branching ratio agrees well with adiabatic calculations, further hinting at $\tilde{D} - \tilde{E}$ state vibronic interactions.

Perhaps the most interesting experimental finding is that the photoelectron anisotropy parameter associated with a peak occurring at a binding energy of 11.8 eV displays the same photon energy dependence as that for the \tilde{B}^2A_1 state, rather than that for the low energy portion of the \tilde{A}^2B_2 state, from which we can now infer a strong possibility of vibronic coupling⁸ between the \tilde{A}^2B_2 and \tilde{B}^2A_1 states.

Although of a slightly different character, the photoelectron anisotropy measurements made in the low binding energy region of the $\tilde{D}^2 B_1$ state band also provide evidence of vibronic interaction. Both these observations of vibrational level dependent β values and the perturbations noted in the photoelectron spectrum reveal a breakdown of the adiabatic Franck-Condon approximation for this state.

As is evident from the comparison between our experimental results, Franck-Condon simulations of vibrational structure, and calculated anisotropy parameters and branching ratios, a full account of the vibronic coupling in *cis*-dichloroethene is required to address the discrepancies between our measured spectra and theoretical predictions. The theoretical modeling of these vibronic interactions is the subject of our companion paper.¹²

SUPPLEMENTARY MATERIAL

See supplementary material for further experimental details and description of the experimental data analysis procedures and tables of measured and calculated vibrational energies.

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Supplementary Material:

Photoionization dynamics of cis-dichloroethene from investigation of vibrationally resolved photoelectron spectra and angular distributions

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Contents

Experimental Apparatus and Procedure	2
Photoelectron Angular Distributions	3
Electronic State Branching Ratios	4
Table S-I	5
Table S-II	6
Table S-III	7
Table S-IV	8
References	.10

Experimental Apparatus and Procedure

The photoelectron spectra were recorded with a VG Scienta R4000 hemispherical electron energy analyser mounted on the soft X-ray undulator-based PLÉIADES beamline at the SOLEIL synchrotron radiation facility. Detailed descriptions of the beamline, monochromator, electron spectrometer and the experimental procedure have been reported previously,^{1,2} and are summarised below.

Synchrotron radiation from a 256 mm period, polarization selectable, electromagnetic undulator was focussed into a modified Petersen type monochromator.^{3,4} The plane of linear polarization could be chosen to lie either parallel or perpendicular to the plane of the electron orbit in the storage ring. The degree of linear polarization of the radiation entering the electron spectrometer is estimated to be > 99%. All the spectra were measured using a varied line spacing (VLS), varied groove depth (VGD) 400 lines/mm grating and a monochromator exit slit width of 30 μ m. The resulting theoretical optical resolution varied between 0.9 meV at hv = 20 eV and 5.0 meV at hv = 90 eV.

-2-

The photoelectron spectrometer was mounted in a fixed position, with the electron detection axis lying perpendicular to the electron orbital plane in the storage ring. Photoionization occurs inside a differentially pumped cell equipped with electrodes to compensate for the plasma potentials created by the formation of ions along the photon beam axis.⁵ Photoelectron spectra were recorded using either parallel or perpendicularly plane polarized radiation, with an analyser pass energy of 10 eV and a 0.2 mm curved entrance slit. Under these conditions, the predicted spectrometer resolution is 5 meV.

Photoelectron Angular Distributions

The photoionization differential cross section in the electric dipole approximation, assuming randomly oriented target molecules and electron analysis in a plane perpendicular to the photon propagation direction, can be expressed in the form ⁶

$$\frac{d\sigma}{d\Omega} = \frac{\sigma_{\text{total}}}{4\pi} \left[1 + \frac{\beta}{4} (3P\cos 2\theta + 1) \right]$$
 Eq. S1

where σ_{total} is the angle integrated cross section, β is the photoelectron anisotropy parameter, θ is the photoelectron ejection angle relative to the major polarization axis, and P is the degree of linear polarization (assumed to be unity) of the incident radiation.

The β parameter can be deduced from spectra (S_{par}) recorded at θ = 0° and spectra (S_{perp}) recorded at θ = 90° using

$$\beta = \frac{2(S_{par} - S_{perp})}{(S_{par} + 2S_{perp})}$$
 Eq. S2

For the particular instrumental arrangement on the PLÉIADES beamline, S_{par} and S_{perp} correspond to normalised spectra recorded with the plane of linear polarization lying perpendicular or parallel, respectively, to the electron orbital plane in the storage ring. For θ = 54.7°, the so-called "magic angle", the photoelectron intensity (Eq. 1) becomes independent of β . We denote such spectra, which may here be reconstructed through $S_{MA} = S_{par} + 2S_{perp}$, as

magic angle spectra $(S_{MA})^2$ and will take advantage of their β -independence when focussing on other aspects of our results.

Electronic State Branching Ratios

The branching ratio for a particular state is defined as the photoelectron intensity in the band associated with that state divided by the summation of the photoelectron intensity in all the accessible bands. The evaluation of this ratio requires knowledge of the transmission efficiency of the electron analyser. This efficiency was determined as described by Jauhianen et al.⁷ The spectra were also normalized to the sample pressure, the photon flux, and the acquisition time. Our analysis concerns only those photoelectron bands with binding energies below 16.2 eV, that is, photoelectron bands due to the 8 a_1 and more tightly bound orbitals are ignored. Thus, the branching ratios for the seven outermost valence states sum to unity.

Table S-I.

The neutral \tilde{X} ¹A₁, and cation \tilde{X} ²B₁, \tilde{A} ²B₂ (C_{2v}) state calculated harmonic frequencies (cm⁻¹).

Mada	Symmetry ^a	Vibrational frequency (cm ⁻¹)					
would		Neutral $\widetilde{X} {}^{1}A_{1} {}^{b}$	Ion $\widetilde{X}^2B_1{}^b$	Ion $\widetilde{A}^2B_2^c$			
1	<i>a</i> ₁	3220.75	3185.98	3249.37			
2	<i>a</i> 1	1647.95	1442.73	1660.37			
3	<i>a</i> ₁	1213.39	1225.46	1118.95			
4	<i>a</i> ₁	704.73	794.60	714.08			
5	<i>a</i> ₁	165.38	179.59	216.65			
6	<i>a</i> ₂	910.76	916.84	1161.99			
7	<i>a</i> ₂	422.61	286.13	562.35			
8	<i>b</i> 1	712.82	740.99	737.51			
9	<i>b</i> ₂	3199.68	3172.34	3227.98			
10	<i>b</i> ₂	1318.92	1373.93	1223.69			
11	<i>b</i> ₂	846.43	950.14	773.35			
12 b ₂		571.55	586.67	534.06			

^a The calculated equilibrium geometries of the neutral (\widetilde{X} ¹A₁ state), and of the cation

 \tilde{X} ²B₁ and \tilde{A} ²B₂ states is C_{2v}.

^b B3LYP/cc-pVTZ

^cTD-B3LYP/cc-pVTZ

Table S-II

The neutral \tilde{X} ¹A₁ and cation \tilde{D} ²B₁ / ²B calculated harmonic frequencies (cm⁻¹).

Mode	Neutra	al $ ilde{X}$ 1 A $_1$	Ion ^a \widetilde{D} ² B ₁ (C _{2v}) / ² B (C ₂)			
	Symmetry ^a	Frequency ^b	Symmetry ^c	Frequency ^d	Frequency ^e	
1	<i>a</i> ₁	3225.91	а	3068.98	3063.02	
2	<i>a</i> 1	1661.76	а	1411.40	1398.69	
3	<i>a</i> 1	1194.16	а	1082.04	1098.32	
4	<i>a</i> 1	710.01	а	814.37	849.03	
5	<i>a</i> 1	164.01	а	591.86	588.74	
6	<i>a</i> ₂	886.11	а	386.97	393.1	
7	<i>a</i> ₂	416.07	а	151.16	152.35	
8	<i>b</i> 1	705.14	b	3092.36	3088.26	
9	<i>b</i> ₂	3203.93	b	1189.28	1178.27	
10	<i>b</i> ₂	1294.12	b	986.70	986.94	
11	<i>b</i> ₂	845.41	b	667.79	676.22	
12	<i>b</i> ₂	569.62	b	555.44	555.76	

^a The calculated geometry of the cation in the \tilde{D} ²B₁ state is C₂. Thus the electronic state becomes ²B. The neutral vibrational modes labelled in Table IV as of a_1 or a_2 representation (C_{2v} point group) transpose to *a* representation, and those modes labelled as of b_1 or b_2 transpose to *b* representation in the lower symmetry C₂ point group common to both ion and neutral.

 c Calculated equilibrium geometry of $\widetilde{\textit{D}}\,\,^{2}\text{B}_{1}$ state is C_{2}

^d TD-B3LYP/cc-pVDZ calculation

^e TD-B3LYP/cc-pVTZ calculation

^b B3LYP/cc-pVDZ calculation. See Table III for B3LYP/cc-pVTZ neutral calculation results.

Table S-III.

Binding energies and tentative assignments for structure observed in the \tilde{X} ²B₁ state photoelectron band of cis-dichloroethene.

Vibrational level	Binding energy (eV)	Vibrational level	Binding energy (eV)	
00	9.659	4 ²	9.859	
41	9.759	4 ² 2 ¹	10.034	
4 ¹ 2 ¹	9.935	4 ² 2 ²	10.206	
4 ¹ 2 ²	10.110	4 ² 2 ³	10.376	
4 ¹ 2 ³	10.281			
4 ¹ 2 ⁴	10.451	3 ¹ 4 ¹	9.907	
		3 ¹ 4 ¹ 2 ¹	10.082	
31	9.809	3 ¹ 4 ¹ 2 ²	10.256	
3121	9.984	3 ¹ 4 ¹ 2 ³	10.426	
3 ¹ 2 ²	10.156			
3 ¹ 2 ³	10.329	3 ²	9.959	
3 ¹ 2 ⁴	10.496	3 ² 2 ¹	10.131	
		3 ² 2 ²	10.304	
21	9.835	3 ² 2 ³	10.474	
2 ²	10.010			
2 ³	10.182			
24	10.354			
2 ⁵	10.521			

Table S-IV.

Binding energies of vibrational progressions observed in the $\tilde{A} {}^{2}B_{2}$, $\tilde{D} {}^{2}B_{1}$ and $\tilde{F} {}^{2}A_{1}$ state photoelectron bands of cis-dichloroethene.

Electronic state	Progression Member	Binding energy (eV)	Electronic state	Progression Member	Binding energy (eV)
$(9b_2)^{-1}\widetilde{A}^2B_2$	1	11.498	$(2b_1)^{-1}\widetilde{D}^{-2}B_1$	1	13.677
	2	11.521		2	13.751
	3	11.549		3	13.825
	4	11.575		4	13.897
	5	11.601	P4	5	13.968
D1	6	11.627		6	14.041
PI	7	11.654		7	14.106
	8	11.681			
	9	11.707	$(9a_1)^{-1}\widetilde{F}{}^2A_1$	1	15.531
	10	11.734		2	15.605
	11	11.760		3	15.682
				4	15.756
$(2b_1)^{-1}\widetilde{D}^2B_1$	1	13.586	Р5	5	15.826
	2	13.659		6	15.899
	3	13.731		7	15.970
	4	13.801			
P2	5	13.871	$(9a_1)^{-1}\widetilde{F}^{2}A_1$	1	15.555
	6	13.944		2	15.630
	7	14.014		3	15.704
	8	14.086	Р6	4	15.779
				5	15.846
$(2b_1)^{-1}\widetilde{D}^2B_1$	1	13.631			

	2	13.705		
	3	13.776		
	4	13.848		
Р3	5	13.921		
	6	13.994		
	7	14.064		

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