

1 **Photovoltaic Panel Cooling by Atmospheric Water Sorption-Evaporation Cycle**

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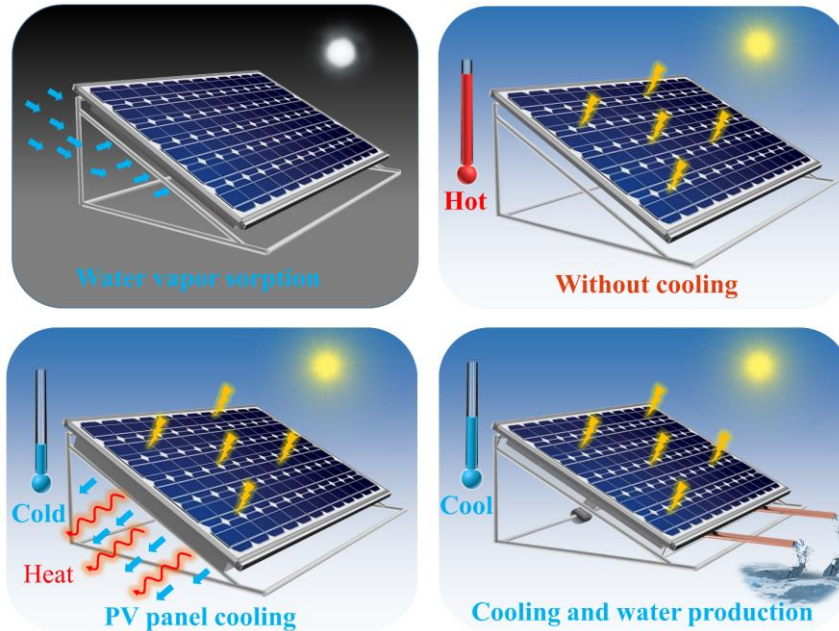
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9 **Abstract**

10 More than 600 gigawatts (GW) photovoltaic (PV) panels are currently installed worldwide, with
11 the predicted total capacity increasing very rapidly every year. One essential issue in PV
12 conversion is massive heat generation of PV panel under sunlight, which represents 75% to 96%
13 of the total absorbed solar energy and thus greatly increases temperature and decreases the energy
14 efficiency and lifetime of the PV panel. In this work, we demonstrate a new and versatile PV panel
15 cooling strategy that employs sorption-based atmospheric water harvester (AWH) as effective
16 cooling component. The AWH based PV cooling provides an averaged cooling power of 295 W/m²
17 and lowers temperature of PV panel by at least 10 °C under 1.0 kW/m² solar irradiation in lab
18 conditions. It delivers 13% to 19% increase in electricity generation of the commercial PV panels
19 in outdoor field tests conducted in winter and summer in Saudi Arabia. The AWH based PV panel
20 cooling strategy has little geographical constrain in its application and is promising in improving
21 electricity productivity of existing and future PV plants, which can be directly translated into less
22 CO₂ emission or less land occupation by PV setup. As solar power is taking a central stage in the
23 global fight against climate change, AWH based cooling represents a solid force toward
24 sustainability.



25

26 Solar energy is the most abundant, inexhaustible and clean renewable energy resource till date.
 27 A photovoltaic (PV) system converts solar energy into usable electricity and is currently the most
 28 popular means of solar energy utilization.^{1,2} In 2019, the total installed capacity of solar PV panels
 29 worldwide reached 600 gigawatts (GW) and it is projected that the global PV capacity will reach
 30 1500 GW by 2025 and 3000 GW by 2030.³ Generally, only 6%–25% of the absorbed solar energy
 31 is converted to electricity by commercial solar PV panels, with the rest inevitably converted to
 32 heat with a heat power of around 600 to 900 W/m² under one-sun illumination.^{4,5} The heat
 33 increases the temperature of the solar panel up to 40 °C above the ambient temperature.⁶ The raised
 34 temperature of the PV panel is detrimental to the energy conversion of the panel, with a reported
 35 0.4-0.5% energy efficiency loss for each degree of temperature rise.⁷⁻⁹ In addition, high
 36 temperature degrades the lifetime of the solar panel.^{10,11} Thus, effective and versatile cooling of
 37 the PV panel is highly important for effective and long-term power generation in existing as well
 38 as future solar power plants.

39 The current PV panel cooling technologies can be divided into two categories: active cooling and
 40 passive cooling.¹²⁻¹⁴ Active cooling uses coolant such as water or air to dissipate heat from the
 41 surface of PV panel.¹⁵⁻¹⁷ While it has high cooling efficiency, it requires complicated engineering
 42 design and needs energy to power up condenser and re-flux coolant.^{18,19} For example, water
 43 spraying PV cooling system can effectively reduce the PV temperature. However, a large quantity
 44 of liquid water is required and subsequently wasted during cooling. Forced airflow circulation
 45 process can be used to cool down PV panel without the consumption of water, but a heatsink is
 46 required and turbulent airflow would make the heatsink highly unstable.¹³ On the other hand,
 47 passive cooling relies on such natural processes as heat radiation and natural ventilation to enhance
 48 the heat dissipation.²⁰⁻²⁵ While passive cooling has simple system design and small energy
 49 consumption, it provides relatively small cooling power.²⁶ For example, radiative cooling can
 50 provide a cooling power of 40-120 W/m² only under clear and cloudless weather. Thus,
 51 engineering simple, inexpensive, and effective design for solar panel cooling is highly sought after.

52 In this work, we demonstrate a simple but effective new PV cooling strategy to enhance the power
53 output of commercial PV panels. The cooling component in the design is an atmospheric water
54 harvester (AWH). The AWH harvests atmospheric water vapor by the sorption-based approach in
55 the evening and at night, vaporizes and thus releases the sorbed water by utilizing the waste heat
56 from the PV panel as energy source during daytime.²⁷⁻³⁰ Due to the large enthalpy of water
57 vaporization (~2450 J/g), the evaporation of the sorbed water takes away large amount of heat and
58 thus keeps the PV panel in a relatively low temperature under solar irradiation.³¹ The AWH in this
59 work is directly attached on the backside of a commercial PV panel and extracts and stores large
60 quantity of water from air even with very low relative humidity (RH) (i.e. <35%) in the evening
61 and during night when the PV panel is not operating. During daytime, as the PV panel is heated
62 up and its conducts heat to the AWH cooling layer. The heat in turn drives evaporation of the
63 stored water in the AWH accordingly, leading to a lowered PV panel temperature. Our results
64 show the AWH can provide an averaged cooling power of 295 W/m² when the solar cell is exposed
65 under one Sun illumination, leading to >10 °C decrease in the temperature and up to 15% increase
66 in electricity generation of the solar cell relative to the case in the absence of the AWH in lab
67 conditions. The outdoor field tests conducted in summer and winter achieved an electricity
68 generation enhancement of ~19% and ~13%, respectively, and also confirmed its performance
69 stability. The advantages of this AWH based PV cooling approach are (1) it has a simple
70 engineering design; (2) its application is globally suitable and not restricted by availability of liquid
71 water; (3) the cooling process is solely waste-heat based and does not entail any additional energy
72 consumption; (4) the design of the system allows for flexibility so that the evaporated water can
73 be easily collected as fresh water for other foreseeable beneficial uses.

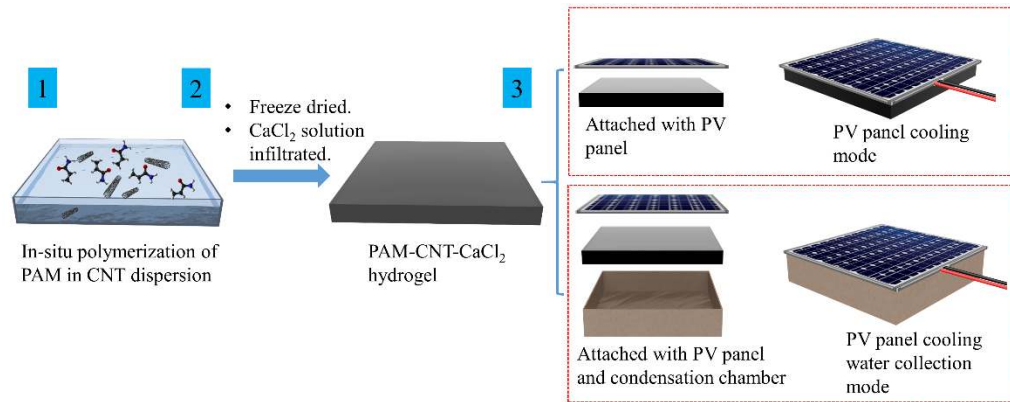
74 **Results and discussion**

75 In this work, a proof-of-concept of the atmospheric water sorption and evaporation based cooling
76 strategy is provided by using a commercial PV panel combined with a hydrogel based AWH,
77 consisting of CNT-embedded cross-linked polyacrylamide (PAM) as substrate and calcium
78 chloride as water vapor sorbent (PAM-CNT-CaCl₂). The hygroscopic salt CaCl₂ in the PAM
79 hydrogel makes the AWH able to capture a large quantity of water vapor from the air due to its
80 extremely high water affinity. The PAM hydrogel framework endues the composite with a gel-like
81 solid form, which holds CaCl₂ and its sorbed water at night and allows for controlled evaporation
82 taking place during daytime.

83 Theoretical modeling (Supplementary note 1) was conducted first to qualitatively evaluate the
84 mass and heat transfer within such a PV-cooling system, which guided the experimental design
85 later on. The details and results of the modelling can be found in SI. Based on the heat transfer
86 model, increasing emissivity of cooling material can further increase the cooling performance
87 through thermal radiation. Thereby, CNT was present in the hydrogel matrix to increase the
88 emissivity and to promote heat dissipation efficiency through thermal radiation of the AWH
89 cooling layer (Supplementary Figure 1).³²

90 **Synthesis of PAM-CNT-CaCl₂ hydrogel cooling layer**

91 The PAM-CNT-CaCl₂ hydrogel was fabricated following a similar procedure reported in literature,
 92 which is schematically presented in Figure 1.²⁹ The PAM-CNT hydrogel matrix was fabricated by
 93 in-situ polymerization of AM monomer in a CNT aqueous dispersion (step 1 of Figure 1). Freeze-
 94 drying was used to remove water solvent and to build a porous structure in the PAM-CNT hydrogel.
 95 The freeze-dried hydrogel was then infiltrated by a CaCl₂ aqueous solution to impregnate CaCl₂
 96 into the hydrogel matrix (step 2 of Figure 1). The PAM hydrogel gives the composite a gel-like
 97 solid form even after a large amount of water is sorbed. Finally, the hydrogel was attached to the
 98 backside of the PV panel to serve as the cooling layer (step 3 of Figure 1).

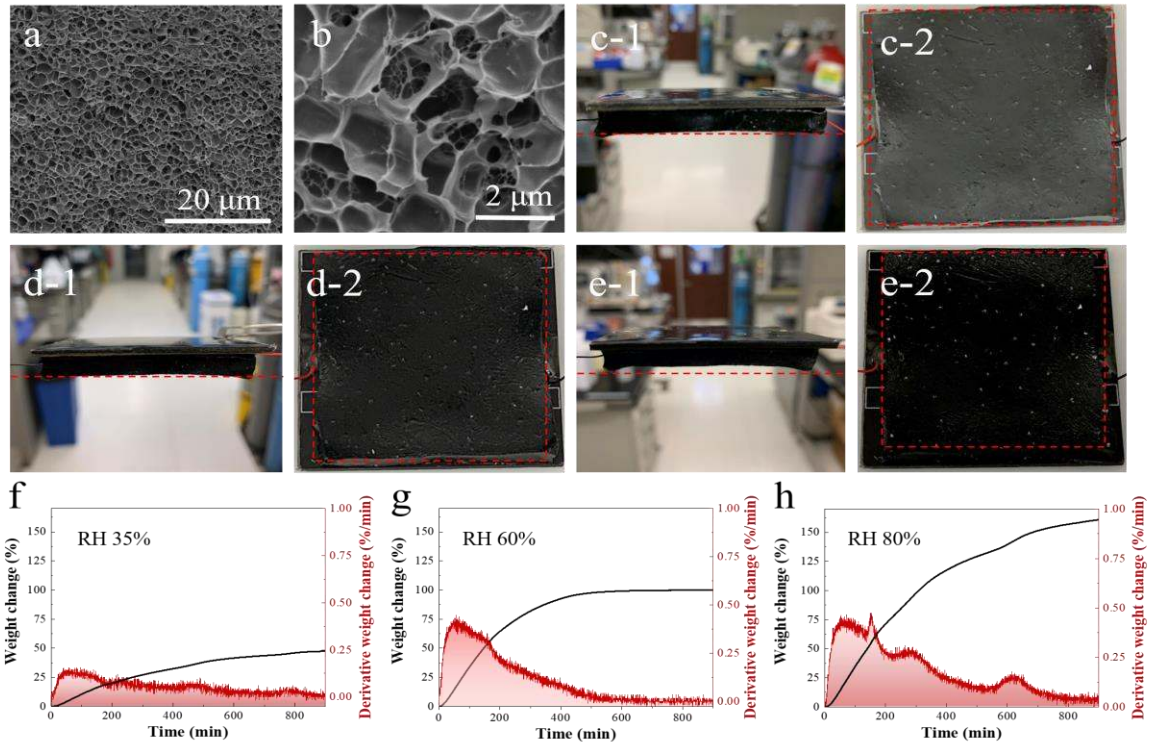


99

100 **Figure 1.** Schematic of PAM-CNT-CaCl₂ cooling layer synthesis process (step 1 and 2) and two
 101 PV panel working modes (step 3).

102 The SEM images of the freeze-dried PAM-CNT hydrogel (Figure 2a and b) clearly reveal the
 103 porous structure of the hydrogel. The average pore diameter is estimated to be around 3 μm and
 104 the wall thickness is ~100 nm. The size and thickness of the PAM-CNT-CaCl₂ hydrogel can be
 105 easily adjusted according to the shape and scale of PV panel by using a pre-fabricated mold or just
 106 by knife cutting. Figure 2 (c-e) shows the digital photos of a PAM-CNT-CaCl₂ hydrogel that was
 107 cut by a knife and attached on the backside of a commercialized PV panel, before, during, and
 108 after PV cooling test. The hydrogel can automatically and firmly attach to the backside of PV panel
 109 without additional assistance due to the rich hydrogen bonds of the hydrogel.

110 The water vapor sorption property of the PAM-CNT-CaCl₂ hydrogel under static RH mode was
 111 conducted on a simultaneous thermal analyzer (STA) coupled with module humidity generator
 112 (Figure 2 f-h). The sorption curves of static RH test present the weight change profiles of the
 113 sample during the tests, while the derivative weight change curves reflect the instant sorption rates
 114 during the test. As shown in Figure 2 (f), when RH was at a low level (i.e. 35%), the final water
 115 uptake value was ~50% compared with the original weight of the dehydrated PAM-CNT-CaCl₂
 116 hydrogel. The sorption rate increased to its peak value (~1.5 mg/g min) within 1 hour, and then
 117 slowly dropped back to zero at the end of the test. A similar trend of the sorption rate could be
 118 found when the RH was set at 60 and 80%, respectively. The amount of absorbed water by PAM-
 119 CNT-CaCl₂ at RH 60% and 80% was ~99% and ~160% in 15 hours, respectively. However, the
 120 peak values of derivative weight change of the hydrogel at RH 60% and 80% are quite similar (i.e.
 121 4.4 vs 4.6 mg/g min, Figure 2 g and h). The full data plot of STA test, including both sorption and
 122 desorption processes, can be found in Supplementary Figure 2.



123

124 **Figure 2.** (a) SEM image and (b) magnified SEM image of freeze-dried PAM-CNT hydrogel.
 125 Digital photo of PAM-CNT-CaCl₂ hydrogel attached on the backside of a commercialized PV
 126 panel (c) before water vapor desorption; after water vapor desorption for (d) 1.5 h and (e) 3 h under
 127 simulated sunlight strength of 1 kW/m². The hydrogel was cut into a squared shape by knife. The
 128 dash lines indicate the thickness change of the hydrogel during test, and the dash square indicates
 129 the dimension change of the hydrogel during the test. (f)-(h) Water vapor sorption properties of
 130 PAM-CNT-CaCl₂ hydrogel under different RH conditions at 25 °C.

131

132 PV panel cooling performance under simulated lab conditions

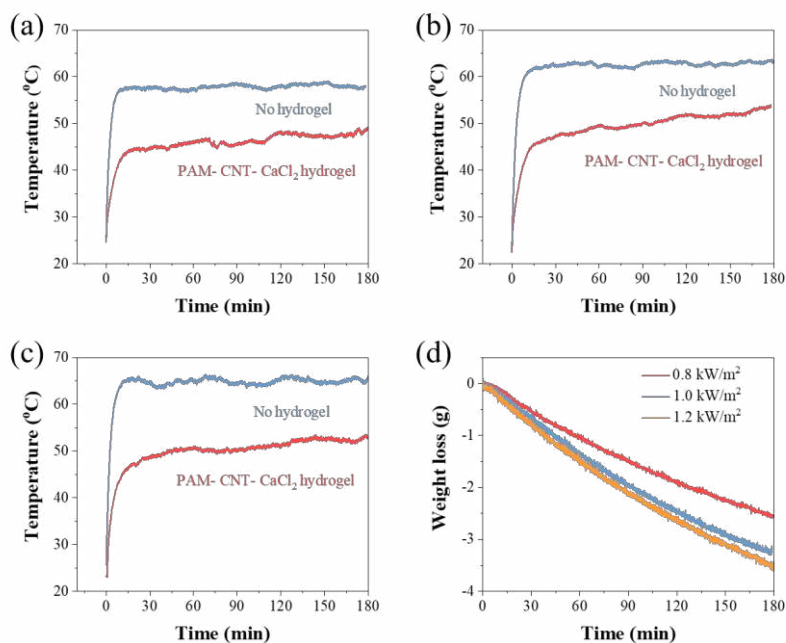
133 PV panel cooling experiments were performed under simulated sunlight first to investigate the
 134 effectiveness of the AWH cooling layer. Based on the I-V curves obtained, the characteristics (i.e.
 135 open circuit voltage (V_{oc}), filling factor (FF) efficiency, and maximum power output (P_{max})) of the
 136 PV panel were calculated and compared.

137 As can be seen in Figure 3 (a), the equilibrium temperature of the PV panel under 0.8 kW/m²
 138 sunlight irradiation was ~57 °C (without cooling layer) and 45 °C (with cooling layer). The
 139 temperature profiles of the PV panel under 1.0 W/m² and 1.2 W/m² sunlight irradiation are shown
 140 in Figure 3 (b), and (c), where a temperature drop of 10-15 °C was achieved when the AWH
 141 cooling layer was employed. The weight loss profile (Figure 3 d) of the hydrogel shows the amount
 142 of the water evaporated during the test. As can be seen clearly, the water evaporation rate (the
 143 slope of the curves) was highly dependent on the irradiated light intensity. For the conditions of
 144 0.8, 1.0, and 1.2 kW/m² light irradiation, around 2.75, 3.25, and 3.65 g of water was evaporated

145 throughout the period of test. The averaged cooling power (P) induced by water evaporation under
 146 each condition was calculated based on following equation.

$$147 \quad P = \frac{\Delta H_{vap} \times \Delta m}{t \times A} \quad (1)$$

148 Where ΔH_{vap} is the enthalpy of vaporization of water (2450 J/g), Δm is the weight loss of hydrogel
 149 due to water evaporation, t is test time (3 hours), and A is surface area of hydrogel (25 cm²). Thus,
 150 the cooling power of AWH cooling layer under 0.8, 1.0, and 1.2 kW/m² sunlight irradiation is
 151 249.5, 294.9, and 331.2 W/m², respectively. These values are much higher than the radiative
 152 cooling (i.e. 40-120 W/m²).³³



153
 154 **Figure 3.** Temperature profile of the PV panel with/without AWH cooling layer under (a) 0.8
 155 kW/m², (b) 1.0 kW/m², and (c) 1.2 kW/m², (d) weight loss profile of the hydrogel during the test.
 156 The original weight of hydrogel was 11.0 g.

157 The I-V curves of the PV panel with and without AWH cooling layer under 0.8 kW/m² sunlight
 158 irradiation are shown in Supplementary Figure 7 (a) and (b). In both occasions, the currents
 159 decreased with increasing voltages and the I-V curves corresponding to longer sunlight irradiation
 160 occurred below the curves corresponding to shorter irradiation time, indicating a poorer
 161 performance under longer irradiation. Similar I-V curves were observed when the light intensity
 162 was increased to 1.0 kW/m². Besides, the I-V curves in the presence of AWH cooling layer
 163 presented a delayed performance drop while the one without quickly reached its equilibrium of its
 164 minimized performance (Supplementary Figure 8 a and b).

165 In the equatorial or low latitude regions such as the Middle East and Africa, the peak sunlight
 166 intensity can be more than 1.2 kW/m² during the mid-day of summer.³⁴⁻³⁶ Under 1.2 kW/m² solar
 167 irradiation, the current of PV panel quickly dropped to its minimal value within 15 minutes,
 168 indicating a quick performance drop under strong light condition. In the presence of the AWH

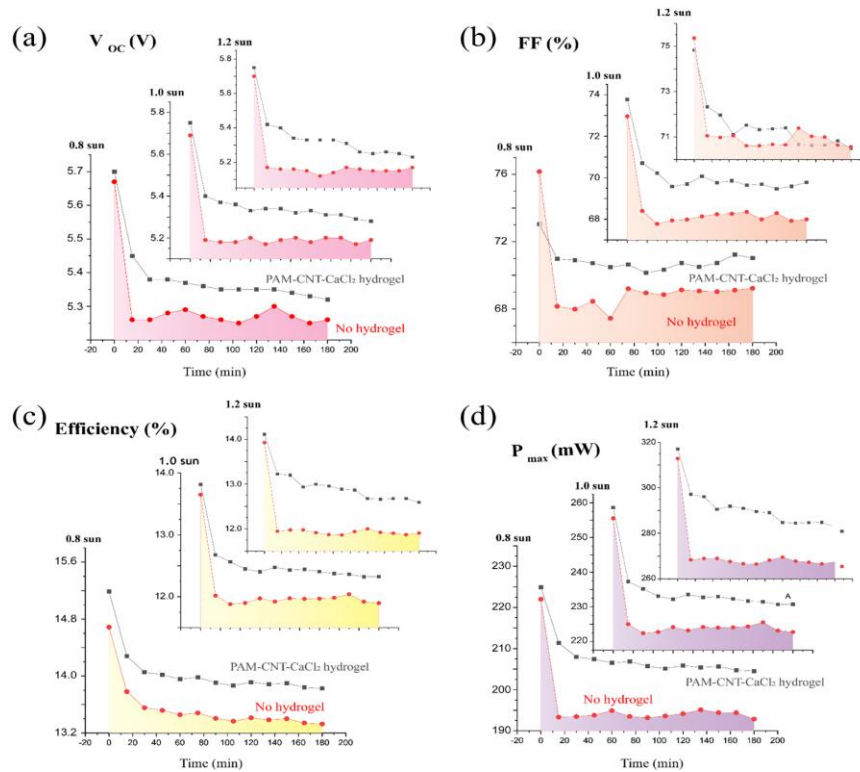
169 cooling layer, the IV curve shows a postponed performance drop (Supplementary Figure 9 a and
170 b).

171 The characteristics of the PV panel (i.e. V_{oc} , FF, P_{max} , and efficiency) were recorded under varied
172 test conditions (Figure 4). When the PV panel was operating without the AWH cooling layer, V_{oc}
173 of the PV panel under 0.8, 1.0, and 1.2 kW/m² sunlight irradiation quickly dropped from 5.68 V
174 to 5.25 V, 5.70 to 5.20 V, and 5.71 V to 5.18 V by the end of the first 15 minutes, followed by a
175 plateau region thereafter. In contrast, when the AWH cooling layer was present, the V_{oc} of the PV
176 panel dropped from 5.70 V to 5.45 V, 5.75 to 5.40 V, and 5.76 V to 5.41 V by the end of the first
177 15 minutes, followed by a slow drop to 5.33 V, 5.30, and 5.26 V after 3 hours test (Figure 4 a).

178 A similar trend was observed for FF, with the AWH cooling layer delivering a better FF than
179 without (i.e. 71% vs. 69%, 70% vs. 68% for 0.8 and 1.0 kW/m² light irradiation, respectively).
180 Under 1.2 kW/m² sunlight irradiation, the one with AWH cooling layer shows a higher FF value
181 than that without until the end of 105 min. However, by the end of the 3 hours test, the FF values
182 of both cases were similar (Figure 4 b).

183 In the absence of the AWH cooling layer, within the first 30 minutes, the efficiency of the PV
184 panel quickly dropped from 14.8% to 13.5%, 13.7% to 11.8% and 14% to 11.9% for the conditions
185 of 0.8, 1.0, and 1.2 kW/m² sunlight irradiation, respectively. In contrast, in the presence of the
186 AWH cooling layer, the efficiency of the PV panel preserved the value of 14.0%, 12.6% and 12.75%
187 by the end of the experiment. (Figure 4 c).

188 The maximum power output, which indicates the electrical power generation ability of the PV
189 panel, is also compared. The P_{max} of PV panel without the cooling layer quickly dropped from 221
190 mW to 193 mW (0.8 kW/m²), 255 mW to 225 mW (1.0 kW/m²), and 315 mW to 270 mW (1.2
191 kW/m²) within the first 15 minutes. The P_{max} of PV panel cooled by the AWH layer slowly dropped
192 to 208 mW (0.8 kW/m²) and 222 mW (1.0 kW/m²) after 30 minutes sunlight irradiation and these
193 values remained almost unchanged until the end of the 3 hours test. For the condition of 1.2 kW/m²
194 sunlight irradiation, the P_{max} slowly dropped from 317 mW to 290 mW by the end of the first hour
195 and further to 283 mW by the end of the test. Therefore, the AWH cooling layer is beneficial to
196 promote the performance of PV panel under various sunlight strength (Figure 4 d).



197

198 **Figure 4.** Parallel comparison of (a) V_{oc} , (b) FF, (c) efficiency, and (d) P_{max} , under light intensity
 199 of 0.8, 1.0, and 1.2 kW/m^2 , respectively. The data points of $t=90$ min and $t=180$ min are also
 200 summarized in Supplementary Table 2.

201 With the AWH cooling layer, the averaged energy efficiency improvement of the PV panel is
 202 calculated to be 5.2, 5.0, and 7.3% under 0.8, 1.0 and 1.2 kW/m^2 condition, respectively and P_{max}
 203 is enhanced by 6.7, 5.3, and 7.4%, respectively. All these results clearly demonstrate that the
 204 temperature of the solar panel can be significantly decreased by evaporation of water from the
 205 AWH, which effectively diminishes the effect of the temperature increase and results in a better
 206 energy conversion performance.

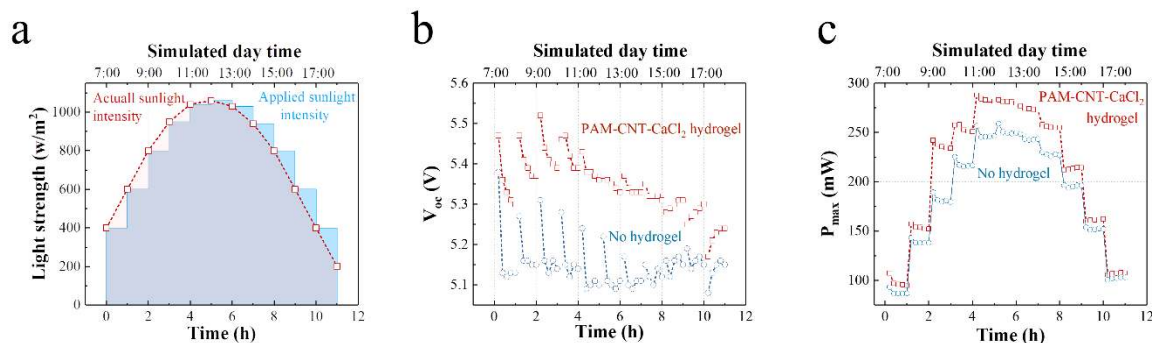
207 When the light intensity was 1.2 kW/m^2 , the enhanced PV performance was distinctly higher than
 208 those recorded under 0.8 and 1.0 kW/m^2 , which is explained as follows. In this work, hygroscopic
 209 salt CaCl_2 is used to capture water vapor from the air and the cooling of the PV panel relies on the
 210 evaporation of water from CaCl_2 aqueous solution inside hydrogel. In other words, the CaCl_2
 211 solution is being gradually concentrated in the course of the evaporation-induced cooling, which
 212 requires a gradually increasing temperature to drive further evaporation. Under a weak light
 213 condition, the evaporation rate of CaCl_2 solution is suppressed due to a low temperature of the PV
 214 panel. Under a strong light condition (i.e. 1.0-1.2 kW/m^2), due to the higher temperature of the PV
 215 panel, the evaporation rate is enhanced.

216 It is worth pointing out that, after most of the water in the AWH is released, the evaporation rate
 217 is slowed down and the heat dissipation process is suppressed accordingly. Thus, towards the end
 218 of the test, the temperature difference between the presence and absence of the AWH cooling layer

219 is reduced (Supplementary Figure 9 (c)-(f)). It is expected, when a larger quantity of AWH is
220 applied, the effect of the AWH assisted cooling will be more pronounced for longer period.

221 A simulated full-day experiment was conducted. Figure 5 (a) compares the actual sunlight strength
222 in real condition (red color, data obtained from NOAA)³⁹ and the applied sunlight in the test,
223 showing a high level of similarity. Figure 5 (b) shows the V_{oc} change of the PV panel with and
224 without the AWH cooling layer during the test. For the V_{oc} change curves obtained before the
225 simulated time of 15:00 (i.e., 8 hours test), the absolute value of voltage decreased, which was
226 resulted from the increased temperature of PV panel. For the curves after the simulated time of
227 15:00, an increment of V_{oc} was observed. This is because the light was adjusted to a lower strength,
228 which lowered the PV panel's temperature. The averaged V_{oc} value of the PV panel with the AWH
229 cooling layer showed 0.2 V higher than that without cooling layer. The P_{max} with the AWH cooling
230 layer was constantly higher value than that without (Figure 5c). Due to the fact that the heat
231 generated from the PV panel is not significant under weakened sunlight (i.e. 400, 600 W/m²), the
232 differences in temperature and P_{max} values between the PV panel with and without the cooling
233 layer were not considerable.

234 When the light intensity is strengthened (i.e. 800-1060 W/m²), the AWH is heated up, the
235 evaporation process is enhanced, the heat is compensated through evaporation process, and the
236 temperature of the PV panel is reduced. In the absence of the AWH, the increased PV panel
237 temperature led to a quiet distinct P_{max} value under strong sunlight irradiation (3-9 h in the test)
238 than the one in the presence of the AWH. Based on P_{max} , an overall 15.2% increase in power output
239 of the PV panel was obtained with the AWH cooling layer, indicating its capability of consistently
240 cooling down PV panel for elongated period.



241
242 **Figure 5.** (a) Actual sunlight strength during a day (data obtained from NOAA) and simulated
243 sunlight strength during the experiment. (b) V_{oc} and (c) P_{max} of PV panel with/without hydrogel
244 cooling layer under simulated daylight.

245 PV panel cooling performance under real outdoor conditions

246 To demonstrate the performance versatility and stability of the AWH-based PV panel cooling
247 system, two complete sorption-evaporation outdoor tests in two seasons (summer and winter of
248 Saudi Arabia) and a 7-day outdoor stability test were performed. All three outdoor tests were
249 conducted inside KAUST campus, Saudi Arabia (GMT+3, 22°19'06.1"N 39°06'15.0"E). The
250 details of the outdoor tests, including experimental setup (photos), solar irradiance, weather

251 conditions, hydrogel weight change, temperature and power of PV panel, etc. were monitored real-
252 time and can be found in Supplementary note 2. The performance improvement was calculated by
253 following equation:

$$254 \quad E = \left(\frac{\eta_{with\ hydrogel}}{\eta_{without\ hydrogel}} - 1 \right) \times 100\% \quad (2)$$

255 Where E denotes the efficiency enhancement, $\eta_{with\ hydrogel}$ and $\eta_{without\ hydrogel}$ are electricity
256 conversion efficiency of PV panel with and without hydrogel cooling layer attached, respectively.
257 The first test was conducted on August 14 and 15, 2019 (summer), during which the comparison
258 of the same PV with and without AWH cooling layer was conducted. Considering the daily
259 difference in solar irradiance, an efficiency enhancement of ~19.0% was obtained. This value is
260 slightly higher than the one observed in the lab condition (i.e., 15.2%), presumably due to outdoor
261 wind-field-facilitated evaporation and heat dissipation in the field conditions. The second outdoor
262 test was conducted from November 30 to December 5, 2019 (winter) with a similar experimental
263 setup (Supplementary note 2). A 13.5% efficiency enhancement was achieved with the AWH
264 cooling layer in comparison to the one without. This is a considerable improvement given that the
265 PV worked in a lower ambient temperature than in the first summer field test. The above two
266 outdoor test conducted in different seasons both produced satisfactory performances and thus
267 successfully demonstrate the feasibility and versatility of the concept of utilizing atmospheric
268 water as coolant to cool down the PV panel.

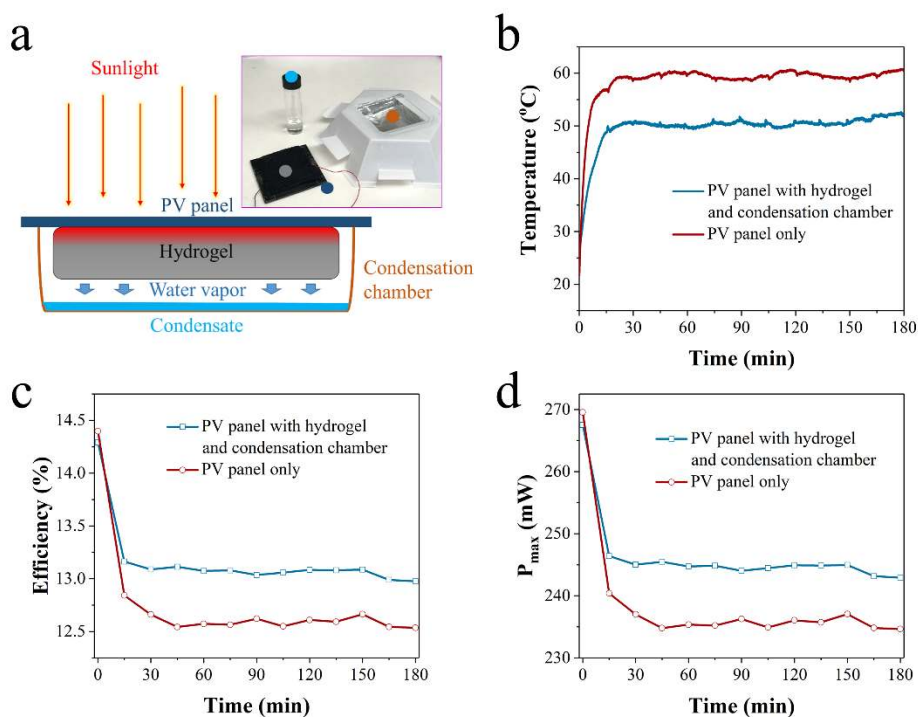
269 In addition, a 7-cycle outdoor field test was performed to confirm the stability and recyclability of
270 such a system, including the water sorption and desorption capacities of the hydrogel, its contact
271 with PV panel, its cooling performances, along with the resultant electricity generation
272 performance by PV. The test was performed between December 27, 2019 at 00:00 and January 2,
273 2020 and the results successfully demonstrate both structural and performance stability of the
274 cooling layer design (Supplementary note 2).

275 Based on the outdoor test results as well as the model results, further performance improvement
276 of AWH-based PV cooling can start with AWH with enhanced vapor sorption capacity and
277 kinetics. To this end, hosting sorbents within macroporous substrates deserves research attention.
278 Based on the heat transfer model, increasing emissivity of cooling material can further increase
279 the cooling performance through thermal radiation. In the meanwhile, proper increase of thermal
280 conductivity of the cooling layer will further enhance the overall cooling performance.

281 **PV panel cooling and atmospheric water collection**

282 The AWH based PV panel cooling can be modified to produce clean water by integrating the
283 hydrogel cooling layer within a water condensation chamber with enlarged heat dissipation surface
284 area (Figure 6 a). In one experiment, an aluminum condensation chamber was attached right
285 beneath the AWH cooling layer (dimension $5 \times 5 \times 0.5\text{ cm}^3$), which led the surface temperature of
286 the panel stable at ~50 °C during the test (Figure 6b). As can be seen in Figure 6 (c), the efficiency
287 of PV panel in the presence of the condensation chamber decreased from 14.6% to 13.25% at the
288 first 15 min, and finally was stabilized at 13%, which was higher than the efficiency of the original
289 PV panel at steady state (12.5%). Meanwhile, the P_{max} of PV panel with the condensation chamber

290 decreased from 267 mW to 246 mW by the end of the experiment, which was higher than the
 291 steady P_{\max} of original PV panel (236 mW, Figure 6d). At the end of the experiment, around 2.0 g
 292 of water was collected and the calcium concentration in the collected water was below the
 293 detection limit of the instrument (i.e. 0.5 ppb). It has to be mentioned that the PV panel used in
 294 this experiment was different from the one in the previous section and thus a direct performance
 295 comparison was not made. The results demonstrate that the AWH-based PV panel cooling system
 296 can be extended to produce liquid water.



297
 298 **Figure 6.** (a) Schematic and digital photo of test setup for PV panel cooling and water collection.
 299 The components with the colored dots in the digital photo indicate the corresponding sections in
 300 the scheme with the same color. (b) The temperature profile of the PV panel top surface.
 301 Comparison of efficiency (c) and P_{\max} (d).

302 In arid and semi-arid regions that have frequent dust storms or dusty conditions, the surface of PV
 303 panel is typically and constantly covered with a layer of dust which blocks solar irradiation. It has
 304 been reported that a single dust storm can reduce the power output of PV panel by 20% and more
 305 than 50% of power output reduction may be resulted in if PV panel fails to be properly cleaned
 306 over six months.³⁷ Water spray is one of the most commonly used methods for PV panel cleaning
 307 and this work provides a solution of harvesting atmospheric water and thus cleaning PV panel in
 308 dry regions where getting liquid form water is challenging. Furthermore, fresh water supply in the
 309 arid and semi-arid regions largely relies on uneconomical and inefficient long-distance water
 310 transportation.³⁸⁻⁴⁰ This work provides an attractive solution for solar farms to cool down PV
 311 panels and simultaneously produce clean water for dust cleaning for PV panel and/or potable
 312 purpose.

313 **Conclusion**

314 In conclusion, this work successfully applies atmospheric water sorption-desorption cycle to cool
315 down PV panel. A cooling power of 295 W/m^2 under 1000 W/m^2 solar irradiation was achieved
316 and thus reduces temperature of PV panel by at least $10 \text{ }^\circ\text{C}$ during operation under lab conditions.
317 The outdoor field tests in summer and winter at Saudi Arabia show that the AWH cooling leads to
318 a $\sim 19\%$ and $\sim 13\%$ increase in the electricity generation by commercial PV respectively. As the
319 global PV installation capacity is predicted to reach 1500 GW by 2025, there would be $>150 \text{ GW}$
320 more electricity to be produced should all PV panels be cooled by this approach by then. This
321 amount of additional electricity can be directly translated into $>8.52 \times 10^7$ metric tons less of coal
322 per year, $>1.48 \times 10^8$ tons of CO_2 emission reduction per year, or 750 km^2 ($\sim 186,000$ acres) of land
323 to be freed from PV occupation, assuming 20% of solar PV electricity generation efficiency.
324 Further improvement of AWH-assisted PV cooling should include enhancing water vapor
325 sorption/desorption kinetics and thus capacity by the AWH, material corrosion, etc. The AWH
326 assisted PV cooling is widely applicable to all kinds of PV installation at any scale, from a single
327 household to large industrial PV farm, and thus is a meaningful contribution to the ongoing effort
328 to fight climate change.

329

330 **Methods**

331 **Chemical and Materials**

332 Acrylamide monomer (AM, 99%), N,N'-methylenebis(acrylamide) (MBAA, 99%), N,N,N',N'-
333 Tetramethylethylenediamine (TEMED, 99%), carbon nanotube (CNT, multi-walled), nitric acid,
334 and calcium chloride (CaCl₂, 99%) were purchased from Sigma-Aldrich. Potassium persulfate
335 (KPS, 99%) was purchased from Acros Organics. All chemicals were used without further
336 purification. Deionized (DI) water (18.2 MΩ, from Milli-Q system) was used throughout the
337 experiments. Commercial photovoltaic (PV) panel was purchased from SUNTHING GOOD Co.,
338 Ltd.

339 **Lab conditions:**

340 Both the room temperature and relative humidity (RH) were controlled by heating, ventilation, and
341 air conditioning (HVAC) system. The room temperature was maintained at 22 °C, and the RH was
342 maintained at 60% (absolute humidity: 11.6 grams of water per cubic meter of air (g/m³)).

343 **Fabrication of Polyacrylamide (PAM)-CNT-CaCl₂ Hydrogel**

344 **Pretreatment of CNT.** The pretreatment of CNT was followed with a previous work.³⁰ Briefly,
345 6.0 g of as-purchased CNT was dispersed in a mixture of sulfuric acid (97%, 180 mL) and nitric
346 acid (70%, 60 mL). After refluxed for 4 h at 70 °C, the dispersion was sonicated for 2 hours,
347 filtrated and washed by DI water thoroughly before use.

348 **Preparation of PAM-CNT-CaCl₂ hydrogel.** 25.0 g of AM was dissolved in 125 mL of 0.5
349 mg/mL CNT aqueous dispersion, followed by purging with nitrogen to eliminate dissolved oxygen.
350 Then, 0.125 g of KPS and 0.048 g of MBAA were added into the dispersion as initiator and cross-
351 linking agent, respectively. Finally, 2 mL of TEMED was added as cross-linking accelerator. After
352 settling the mixture overnight under a nitrogen atmosphere, the PAM-CNT hydrogel was obtained.
353 To load CaCl₂ into the hydrogel, the as-prepared PAM-CNT hydrogel was first freeze-dried at -80
354 °C (FreeZone 2.5 plus, LABCONCO), followed by immersing the dry hydrogel in 125 mL CaCl₂
355 solution (0.4 g/mL) for 48 hours under ambient condition to fabricate the PAM-CNT-CaCl₂
356 hydrogel.

357 **Material Characterization**

358 Scanning electron microscopy image was obtained by a Zeiss Merlin field emission scanning
359 electron microscope (FE-SEM). The metal content in the condensed water was examined by
360 inductively coupled plasma-optical emission spectroscopy (ICP-OES, Agilent 5100) equipped
361 with charge-coupled device (CCD detector) and charge injection device (CID detector).

362 Water vapor sorption and desorption tests were conducted on a simultaneous thermal analyzer
363 (STA, Jupiter STA-449, NETZSCH) coupled with a modular humidity generator (MHG,
364 ProUmid). The humidity generator was programmed to output nitrogen flow (100 mL/min) with
365 predefined relative humidity (RH) and equilibration time. In more details, the STA furnace was
366 programmed to simulate the temperature during the night (25 °C, water vapor sorption process).
367 The flow temperature and flow rate of nitrogen carrier were set to be 25°C, 100 mL/min while the

368 pre-designated RH was set to be 35, 60, and 80%, respectively. The selection of humidity
369 parameter was based on the averaged global humidity distribution map. The relative humidity of
370 35, 60, and 80% represents most of the low (i.e., Central Australia, North Africa, Middle east, etc.),
371 medium (i.e., Southern Africa, Northern Asia, Central Europe, North America, etc.), and high (i.e.,
372 South America, South Asia, Northern Europe, etc.) humidity regions around the world. Absolute
373 humidity of STA in the water vapor sorption tests (25 °C, RH 35, 60, and 80%) was 8.1, 13.8, and
374 18.4 g/m³, respectively. The STA furnace was first heated up to 85 °C to release the sorbed water
375 and then cooled down to 25 °C for the water vapor sorption test.

376 **Device assembly**

377 The as-obtained PAM-CNT-CaCl₂ hydrogel was directly attached to the backside of
378 commercialized PV panel due to its self-adhesion property. For PV panel cooling, the hydrogel
379 attached PV panel was directly mounted on a homemade polystyrene frame, and the water
380 evaporated from the hydrogel was directly released to the ambient air. For PV panel cooling with
381 water collection, an additional condensation chamber was attached to cover the hydrogel and to
382 collect the released water.

383 **PV panel cooling test**

384 The performance of the PV panel under different working conditions was tested on a Keithley-
385 2400 source meter. The hydrogel-attached PV panel was first placed in the ambient with RH of
386 60% and temperature of 22 °C for 17 hours. Then the PV panel was directly exposed to the
387 simulated sunlight (Oriel 94023A solar simulator, AM 1.5 filter). The mass change of the PV panel
388 with the hydrogel was measured and recorded by an electronic balance connected to a computer.
389 An IR camera (FLIR 655sc) was used to monitor and record the temperature change of the PV
390 panel.

391 The simulated daylight was used to test the performance of the hydrogel-based cooling layer. The
392 light intensity of solar simulator was tuned to reflect the daily variation of light strength during a
393 day. The PV panel used in this work was composed of 9 pieces of solar cell in series combination.
394 The effective area of the PV panel was 18.2 cm². Based on the mass transfer model (Supplementary
395 note 1), the time required for the water vapor sorption process is highly dependent on the thickness
396 of hydrogel. To ensure an effective sorption within a reasonable time frame, the attached hydrogel
397 cooling layer was designed to be 5 × 5 × 0.5 cm³ in dimension and was exposed to the ambient
398 condition (RH 60%, 22 °C) for 17 hours prior to the test. The experiments were performed under
399 simulated sunlight with different light intensities (0.8, 1.0, and 1.2 kW/m²) for 3 hours.

400 A simulated full-day experiment was conducted; the experiment was performed for 11 hours,
401 starting from the simulated daytime of 7:00 am to 18:00 pm. The solar simulator was carefully
402 adjusted every hour to reflect the daily light intensity variation while the data points were collected
403 every 15 minutes. Three pieces of hydrogel with a dimension of 5 × 5 × 0.6 cm³ were stacked and
404 attached to the PV panel. The detailed parameters can be found in Supplementary Table 3.³⁶

405 The simulated solar intensity for the PV panel cooling and water collection mode was 1.0 kW/m².
406 Dry weight of the hydrogel before water vapor sorption (i.e. weight of polyacrylamide and CaCl₂

407 along with its non-removable crystallized water) was 7.9 g, and the dimension of the hydrogel
408 before water vapor sorption was $\sim 4.2 \times 4.2 \times 0.3 \text{ cm}^3$. Water vapor sorption process was conducted
409 by placing the dried hydrogel in lab condition for 17 hours. After the water vapor sorption process,
410 the dimension of the hydrogel changed to $\sim 5 \times 5 \times 0.5 \text{ cm}^3$, and the weight of the hydrogel was 14.1
411 g.

412

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416 **Author contributions**

417 Peng Wang supervised the project; Renyuan Li, Yusuf Shi and Peng Wang conceived the idea and
418 designed the experiments; Renyuan Li and Mengchun Wu performed the materials synthesis,
419 characterization and performance investigation; Seunghyun Hong performed the graphic work;
420 and Renyuan Li and Peng Wang co-wrote the paper. All authors discussed and commented on the
421 manuscript.

422 **Corresponding Author**

423 Correspondence to Peng Wang

424 **Data availability**

425 The data that support the findings of this study are available from the corresponding author on reasonable
426 request.

427 **Competing interests**

428 Peng Wang, Renyuan Li and Yusuf Shi have a patent application related to the work presented in
429 this paper.

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